

Final Report (2002-2014) for DOE grant (DE-FG02-02ER45999)
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Summary of Training

Over the course of this period of funding, the PI published 62 papers and graduated 15 Ph.D. students supported by the grant. In addition 17 undergraduate students, 10 postdocs and 16 Visiting Scholars were also involved in this research. Over 150 talks at national and international conferences on the topic were given by the PI and her research group.

Summary of Broader impacts

In addition to the training of undergraduate students, graduate students, Visiting Professors and Visiting Scholars, the PI and her group participated in numerous activities encompassing outreach and community service. The development and education of students is of utmost importance to the field of chemistry. Fostering a diverse and positive learning environment is vital for the development and future of science. The versatility of skills required to work on this project represented a golden opportunity for students to learn the synthesis and characterization of organic and inorganic compounds as well as essential methods of characterization techniques. The students worked closely with the PI and the collaborators to learn first-hand how to perform all of the measurements and to interpret magnetic and conductivity data. The PI sent her students to work with both domestic and international collaborators one-on-one, so that students could obtain a better understanding of the techniques and methods of characterization that could not be carried out at her home institution.

The PI has been committed to upholding the value of education and providing opportunities for her students to enhance their experience not only through collaborations, but through attending in-house seminars, meetings with visiting professors and participating in conferences, especially those that focus on their field of study. The PI has sent many students to special conferences/meetings/workshops throughout the past twelve years of DOE support. The PI has also supported the advancement of the education of her students by sending them to facilities such as the Advanced Light Source at Lawrence Berkeley National Laboratory and the Advanced Photon Source at Argonne National Laboratory in order to gain hands-on experience, efforts that have been very important for their education. These opportunities are invaluable for their future success in careers in both academics or industry. The PI has encouraged the students to apply for and has helped them to prepare Fellowship applications to the NSF, Ford Foundation, DOE and the [Louis Stokes Alliances for Minority Participation](#) (LSAMP), many of which have been awarded.

The PI is also committed to recruiting students from underrepresented backgrounds. The PI wrote a local section ACS grant several years to start a chapter of NOBCChE (National Organization of Black Chemists and Chemical Engineers) at Texas A&M which will help in the recruitment and retention of African American undergraduate and graduate chemistry students. The PI and her students are also strongly committed to reaching out to younger

generations in order to enhance their passion for science and enriching their education by volunteering in the community. These efforts consist of development of educational materials, giving demonstrations at several events and local schools, offering free tutoring and acting as coaches and Specific instances are Coaches Clinic Lab, SACNAS free tutoring service since 2011- <http://sacnas.tamu.edu/node/15>, Community Outreach Bryan School District- Chemistry and Nanotechnology Workshops, Organized chemistry demonstrations for the National Chemistry Week for the CHEM 116 showroom (since 2004), Science fair judge at Harmony Science Academy, Science Bowl judge, Science Olympiads Chem Lab C, Expanding your horizons in Science and Mathematics.

Geometrical and Electronic Control of Anisotropy in Mononuclear Complexes.

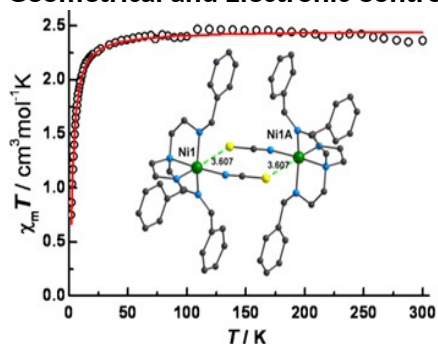


Figure 1. Molecular structure and DC magnetic data for $[\text{Ni}(\text{trenbn,H})(\text{NCS})][\text{ClO}_4]\text{MeOH}$.

We found that the treatment of $\text{tren}^{\text{Bn,H}}$, $\text{Ni}(\text{H}_2\text{O})_6[\text{ClO}_4]_2$ and NaNCS in acetonitrile affords the compound $[\text{Ni}(\text{tren}^{\text{bn,H}})(\text{NCS})][\text{ClO}_4]\cdot\text{MeOH}$ (Figure 1). The Ni ion is located in a distorted tetragonal pyramid instead of the expected trigonal bipyramidal geometry due to close intermolecular $\text{Ni}\cdots\text{S}$ contacts (3.607 Å). As a result, a small D_{Ni} value (-0.23 cm^{-1}) was found. Bulkier substituents are required for rendering the equatorial plane more crowded in order to maintain the TBP geometry.

Incorporation of Titanium Ions into Magnetic

Materials

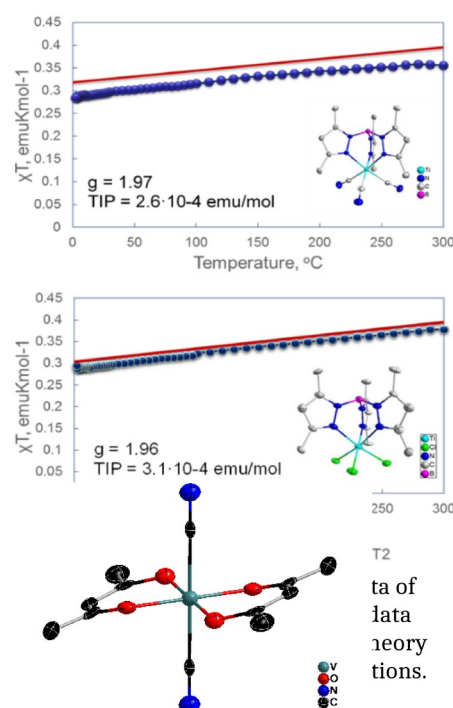


Figure 4. $[(\text{acac})_2\text{V}(\text{CN})_2]^-$ anion.

Table 1. Calculated ligand field energies in cm^{-1} .

State	$(\text{Et}_4\text{N})[\text{Tp}^*\text{TiCl}_3]$	$(\text{Et}_4\text{N})[\text{Tp}^*\text{Ti}(\text{CN})_3]$
$^4\text{E}_g$	22910	22920
$^4\text{A}_g$	1642	

Figure 3. The $\{(\text{Tp}^*\text{Ti}(\text{CN})_2)_2(\mu\text{-OAc})_4(\mu\text{-O})_2\text{Mn}\}^-$ anion.

In the last year of this DOE grant we continued working on the Ti^{III} building block, $(\text{Et}_4\text{N})[\text{Tp}^*\text{Ti}(\text{CN})_3]$ and initiated a collaboration with Dr. Alessandro Soncini from the University of Melbourne who performed *ab initio* CASSCF calculations on $(\text{Et}_4\text{N})[\text{Tp}^*\text{Ti}(\text{CN})_3]$ and $(\text{Et}_4\text{N})[\text{Tp}^*\text{TiCl}_3]$. Using the atom

coordinates, the ligand field spectra and magnetic properties were predicted from these calculations. As shown in Figure 2, the theory results are in excellent agreement with experimental data. Table 1 shows the calculated ligand field energies of the starting materials. From this we can see that the ground state of this compound is non-degenerate, which is not ideal for having first-order spin-orbit coupling (emanating from a degenerate ground state in a trigonally distorted systems). Replacing the methyl substituents on the Tp^* ligand for a bulkier group such as phenyl would likely promote a trigonal elongation and stabilize the degenerate ^2E state. The $[\text{Et}_4\text{N}](\text{Tp}^*\text{Ti}(\text{CN})_3)$ forms trinuclear molecules of general formula $\{(\text{Tp}^*\text{Ti}(\text{CN})_2)_2(\mu\text{-OAc})_4(\mu\text{-O})_2\text{M}\}^{0/1-}$ ($\text{M} = \text{Cr}^{\text{III}}, \text{Co}^{\text{II}}, \text{Mn}^{\text{II}}$) Figure 3 in the presence of trace H_2O . The chromium analog contains an $[\text{Et}_4\text{N}]^+$ cation in its structure, which means that at least one of the titanium centers remains trivalent. This represents a rare example of a Ti^{III} containing molecule; magnetic measurements are currently underway. The Mn compounds contain only Ti^{IV} . The $\text{Tp}^*\text{Ti}^{\text{IV}}$ therefore acts as a diamagnetic cap and would be an ideal system for testing isolated pseudo-octahedral environments for mononuclear SMMs. According to recent theory, d^1 , d^2 , d^6 , and d^7 transition metal ions have the possibility for exhibiting large anisotropy in octahedral environments. Because of this, particular focus is on the Fe^{II} and Co^{II} analogs.

Trans-dicyanide building blocks for chains. In an effort to expand the library of SCM cyanide building blocks, $[\text{V}(\text{acac})_2\text{Cl}(\text{THF})]$ was used to prepare $\text{PPN}[(\text{acac})_2\text{V}(\text{CN})_2]\cdot\text{PPNCl}$ (Figure 4) and $[\text{V}(\text{Salphen})\text{Cl}(\text{DMF})]$ was reacted with CN^- to give $\text{PPN}[\text{V}(\text{Salphen})(\text{CN})_2]$. The analogous

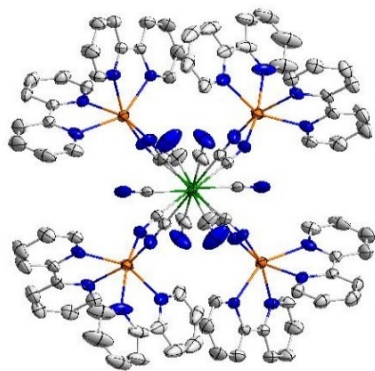


Figure 5. Molecular structure of the $\{[\text{Mn}^{\text{II}}(\text{bpy})_2]_4[\text{Mo}^{\text{III}}(\text{CN})_7]_2\}$ octahedral molecule.

salen-based building blocks $\text{PPN}[\text{V}(\text{MeOSalen})(\text{CN})_2]$ and $\text{PPN}[\text{V}(\text{Salen})(\text{CN})_2]$ were also isolated.

Cyanometallate Chemistry of Highly Anisotropic Metal Ions

We found an interesting alternative to TBP product formation by using 2,2'-bipyridine (bpy) instead of tmphen. The reaction of $[\text{Mo}^{\text{III}}(\text{CN})_7]$ with $[\text{Mn}^{\text{II}}(\text{bpy})_2]^{2+}$ in a $\text{H}_2\text{O}/\text{MeOH}$ mixture produces the octahedral molecule $\{[\text{Mn}^{\text{II}}(\text{bpy})_2]_4[\text{Mo}^{\text{III}}(\text{CN})_7]_2\}$ as confirmed by single crystal X-ray data (Figure 5). The self-assembly with bpy appears to be solvent independent and selective for octahedral molecules. Interestingly the magnetic data revealed strong ferromagnetic interactions between the Mo^{III} and Mn^{II}

centers with a Weiss constant of +43 K. The compound exhibits the onset of frequency dependence in the out-of-phase AC susceptibility data consistent with SMM behavior.

Compounds with Dianionic Organocyanide Ligands

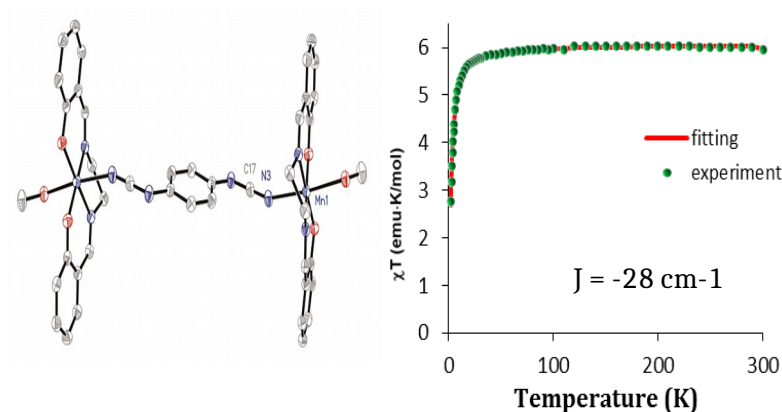


Figure 6. Structure and plot of χT versus T for 1.

A series of proposed model dinuclear compounds bridged by para- and ortho-dicyanamidobenzene dianions were synthesized and their magnetic properties studied:

{Mn(salen)}₂(μ-DCNQI) (1),

{Mn(salen)}₂(μ-anti-N(cyano),N(amido)-DCYD) (2), {Cr(salen)}

(CH₃OH)}₂(μ-DCTL) (3), and {Mn(salen)}

(CH₃OH)}₂(μ-DM-DCNQI) (4) (DCNQI²⁻ =

1,4-dicyanamidobenzene dianion, DM-DCNQI²⁻ =

2,5-dimethyl-1,4-dicyanamidobenzene dianion,

DCYD²⁻ = 1,3-dicyanamidobenzene dianion and

DCTL²⁻ = 2,6-dicyanamidotoluene). The premise of this work was to test computational results

that predicted strong ferromagnetic and antiferromagnetic exchange coupling interactions

between trivalent 3d transition metals, on the

order of ± 75 -150 cm⁻¹ for these diamagnetic linkers. Unlike what was predicted by theory, it

was found that the coupling between the trivalent Mn^{III} centers was much weaker than predicted (Figure 6). We attribute this situation to the strong Jahn-Teller distortion of the

Mn^{III} d⁴ unit which leads to long Mn-N bond distances (2.242 Å). In the case of DCYD²⁻, an unexpected structure was obtained with one of the Mn^{III} ions coordinated to an amido-N instead of the terminal cyano-N (Figure 7). Therefore, a methyl group was employed to block the free rotation to obtain DCTL²⁻, and even with the Jahn-Teller inactive octahedral Cr^{III} ion in place of Mn^{III}, only very weak antiferromagnetic interactions are observed.

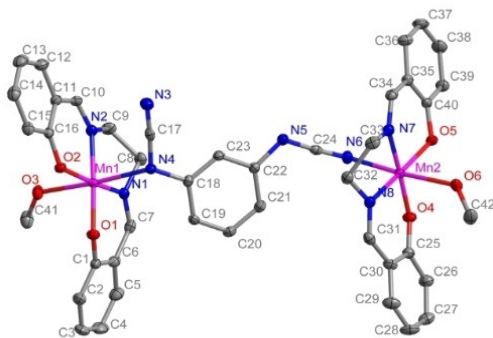


Figure 7. Molecular structure of 2.

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DOE Publications Supported by DE-FG02-02ER45999

1. 1,3-Dithiolan-2-one. Eric Reinheimer, John Bacsá and Kim R. Dunbar *Acta Cryst. Section E*, 2004, E60, 1206-1207.

We gratefully acknowledge the service provided by Dr Joseph Reibenspies and the X-ray facility at Texas A&M University. We thank the National Science Foundation for PI and NIRT grants (CHE-9906583 and DMR-0103455) and for equipment grants for the CCD X-ray equipment (CHE9807975). We thank the Department of Energy for a PI grant (DE-FG03-02ER45999). Support from the Welch Foundation (A-1449) and from a Telecommunications and Informatics Task Force (TITF) Grant from Texas A&M University is also gratefully acknowledged.

2. Bis(pyridine-2-ylmethanolato- κ^2 N,O) bis(trifluoroacetato)nickel(II). John Bacsá, Hanhua Zhao and Kim R. Dunbar *Acta Cryst.* **2004**, E60, m1040-m1042.

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3. Syntheses, Structure, and Magnetic Properties of New Types of Cu(II), Co(II), and Mn(II) Organophosphonate Materials: Three-Dimensional Frameworks and a One-Dimensional Chain Motif. Deyuan Kong, Yang Li, Xiang Ouyang, Andrey V. Prosvirin, Hanhua Zhao, Joseph H. Ross, Jr., Kim R. Dunbar and Abraham Clearfield *Chem. Mater.* **2004**, 16, 3020-3031.

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4. Molecular Cube of Re^{II} and Mn^{II} That Exhibits Single-Molecule Magnetism. Eric J. Schelter, Andrey V. Prosvirin and Kim R. Dunbar *J. Am. Chem. Soc.* **2004**, 126, 15004-15005.

K.R.D. gratefully acknowledges the Department of Energy (DE-FG03-02ER45999) for funding of this research. The SMART APEX CCD diffractometer and the SQUID magnetometer were purchased by funds provided by the National Science Foundation (Grants NSF-9807975 and NSF-9974899). Highfrequency AC susceptibility experiments were performed at the National High Magnetic Field Laboratory in Tallahassee, FL. We thank Dr. Alexei Souslov for his assistance with these measurements. Use of the TAMU/LBMS-Applications Laboratory and help from Ms. Vanessa Santiago are also acknowledged.

5. Films of Mn_{12} -acetate deposited by low-energy laser ablation. J. Means, V. Meenakshi, R. V. A. Srivastava, W. Teizer, Al. A. Kolomenskii, H. A. Schuessler, H. Zhao and K. R. Dunbar *J. Magnetism & Magnetic Mater.* **2004**, 284, 215-219.

We thank C. Berlinguette, V. Pokrovsky, N. Sinitsyn and W. Lackowski for helpful discussions. J. Means is supported by a Doctoral Research Fellowship from Sandia National Laboratories. The authors gratefully acknowledge the following support: (1) Robert A. Welch Foundation, (2) Texas Advanced Research Program (010366-0038-2001), (3) The US Department of Energy (DOE-DE-FG03-02ER45999), (4) National Science Foundation for support from Nanoscale Science and Engineering (NIRT) Grant (DMR-0103455) and for an equipment grant to purchase a SQUID magnetometer (NSF-9974899).

6. Mn_{12} -acetate film pattern generated by photolithography methods. K. Kim, D. M. Seo, J. Means, V. Meenakshi, W. Teizer, H. Zhao and K. R. Dunbar *Appl. Physics Lett.* **2004**, 85, 3872-3874.

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7. The step-wise assembly of an undecanuclear heterotrimetallic cyanide cluster. Curtis P. Berlinguette and Kim R. Dunbar *Chem. Comm.* **2005**, 2451-2453.

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8. Structural and magnetic properties of iron(II) complexes with 1,4,5,8,9,12-hexaazatriphenylene (HAT). Mikhail Shatruk, Abdellatif Chouai, Andrey V. Prosvirin and Kim R. Dunbar *Dalton Trans.* **2005**, 1897-1902.

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9. Control of the Barrier in Cyanide Based Single Molecule Magnets $\text{Mn(III)}_2\text{Mn(II)}_3$: Theoretical Analysis. Boris S. Tsukerblat, Andrew V. Pali, Sergei M. Ostrovsky, Sergei V. Kunitsky, Sophia I. Klokishner and Kim R. Dunbar *J. Chem. Theory Comput.* **2005**, 1, 668-673.

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10. Synthesis, crystal structures and magnetic properties of two new coordination polymers based on the tricyanoethenolate ligand: $\{\text{Fe}(\text{C}_5\text{N}_3\text{O})_2(\text{CH}_3\text{CN})_2\}_\infty$ and $\{\text{Co}(\text{C}_5\text{N}_3\text{O})_2(\text{C}_4\text{H}_4\text{O})_2\}_\infty$. Hanhua Zhao, John Bacsá, Andrey Prosvirin, Nazario Lopez and Kim R. Dunbar *Polyhedron* **2005**, 24, 1907-1912.

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11. Enhanced magnetic anisotropy of Mn_{12} -acetate. D.M. Seo, V. Meenakshi, W. Teizer, H. Zhao, K.R. Dunbar *J. Mag. Mag. Mat.* **2006**, 301, 31-36.

This research was supported by the Texas Advanced Research Program (010366-0038-2001), the Robert A. Welch Foundation (A-1585), the National Science Foundation (DMR-0103455 and NSF-9974899), DOE (DOE-DE-FG03-02ER45999), and the Telecommunications and Informatics Task Force (TITF 2001-3) at Texas A&M University. Use of the TAMU/CIMS Materials Characterization Facility is acknowledged. We thank D. Naugle for helpful discussions.

12. Influence of anions on the dimensionality of extended networks based on Cu¹ cations and 1,4,5,8,9,12-hexaazatriphenylene (HAT) ligands. Mikhail Shatruk, Abdellatif Chouai and Kim R. Dunbar *Dalton Trans.* **2006**, 2184-2191.

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13. Highly Anisotropic Orbitally Dependent Superexchange in Cyano-Bridged Clusters Containing Mn(III) and Mn(II) Ions. Andrew Pali, Sergey M. Ostrovsky, Sophia I. Klokishner, Boris S. Tsukerblat and Kim R. Dunbar *Chem. Phys. Chem.* **2006**, 7, 871-879.

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14. Syntheses and reactivity studies of solvated dirhenium acetonitrile complexes. Jitendra K. Bera, Eric Schelter, Sanjib K. Patra, John Bacsá and Kim Dunbar, *Dalton Trans.* **2006**, 4011-4019.

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15. Syntheses, Structural Studies, and Magnetic Properties of Divalent Cu and Co Selenites with Organic Constituents. Mei-Ling Feng, Andrey V. Prosvirin, Jiang-Gao Mao and Kim R. Dunbar, *Chem. Eur. J.* **2006**, 12, 8312-8323.

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16. Long-Range Ordered Magnet of a Charge-Transfer $\text{Ru}_2^{4+}/\text{TCNQ}$ Two-Dimensional Network Compound. Hitoshi Miyasaka, Toru Izawa, Nao Takahashi, Masahiro Yamashita, and Kim R. Dunbar, *J. Am. Chem. Soc.*, **2006**, 128, 11358-11359.

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17. Origin of the Single Chain Magnet Behavior of the $\text{Co}(\text{H}_2\text{L})(\text{H}_2\text{O})$ Compound with a 1D Structure. Andrei V. Palii, Sergei M. Ostrovsky, Sophia I. Klokishner, Oleg S. Reu, Zhong-Ming Sun, Andrei V. Prosvirin, Han-Hua Zhao, Jiang-Gao Mao, and K.R. Dunbar, *J. Phys. Chem. A.*, **2006**, 110, 14003-14012.

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18. Lanthanide-3d cyanometalate chains $\text{Ln}(\text{III})-\text{M}(\text{III})$ ($\text{Ln} = \text{Pr}, \text{Nd}, \text{Sm}, \text{Eu}, \text{Gd}, \text{Tb}$; $\text{M} = \text{Fe}$) with the tridentate ligand 2,4,6-tri (2-pyridyl)-1,3,5-triazine (tptz): evidence of ferromagnetic interactions for the $\text{Sm}(\text{III})-\text{M}(\text{III})$ compounds ($\text{M} = \text{Fe}, \text{Cr}$). Hanhua Zhao, Nazario Lopez, Andrey Prosvirin, Helen T. Chifotides and Kim R. Dunbar, *Dalton Trans.*, **2007**, 878-888.

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19. Systematic Investigation of Trigonal-Bipyramidal Cyanide-Bridged Clusters of the First Row Transition Metals. Mikhail Shatruk, Kristen Chambers, Andrey V. Prosvirin, Kim R. Dunbar, *Inorg. Chem.*, **2007**, 46, 5155-5165.

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20. Enhanced alignment of Mn_{12} -acetate micro-crystals. D. Seo, W. Teizer, H. Zhao, and K.R. Dunbar, *J. Mag. Mag. Mat.*, **2007**, 312, 205-209.

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21. A Family of Mixed Metal Cyanide Cubes with Alternating Octahedral and Tetrahedral Corners Exhibiting a Variety of Magnetic Behaviors Including Single Molecule Magnetism. Eric J. Schelter, Ferdi Karadas, Carolina Avendano, Andrey V. Prosvirin, and Kim R. Dunbar, *J. Am. Chem. Soc.*, **2007**, 129, 8139-8149.

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22. Magnetic anisotropy in the octanuclear $(\text{Re}_4\text{Mn}_4)\text{-Mn}^{\text{II}}$ cluster exhibiting Single-Molecule Magnet behavior: Quantum-spin and classical-spin approaches. A. V. Palii, S. M. Ostrovsky, S. I. Klokishner, B.S. Tsukerblat, E. J. Schelter, A. Prosvirin and K. R. Dunbar, *Inorg. Chim. Acta*, **2007**, 360, 3915-3924.

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23. [Nanopatterning of Mn-12-acetate single-molecule magnet films](#). K. Kim, A. Ford, V. Meenakshi, W. Teizer, H. Zhao and K. R. Dunbar, *J. Appl. Phys.*, 2007, 102, 094306:1-5.

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This work was supported, in part, by the National Science Foundation (NSF) via CHE-1265993 (ASV), DMR-1213030 (GC), DMR-1202033 (MWM), DMR-1157490 (NHMFL), and CHE-0821346 (KAA). Research supported by the U.S. Department of Energy, Office of Basic Energy Sciences, Division of Materials Sciences and Engineering under Award DE-FG02-02ER45999 (KRD). ASV also acknowledges UF support and the resources provided by the UF High-Performance Computing Center. NVD is an HHMI International Student Research fellow. We thank Prof. A. Angerhofer

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This material is based on work at Texas A&M University supported by the U.S. Department of Energy, Materials Sciences Division, under Grant Nos. DE-FG02-02ER45999 and DE-SC0012582. D.P. acknowledges the financial support of the Polish Ministry of Science and Higher Education, within the Iuventus Plus Programme (Grant Agreement No. 0370/IP3/2015/73). This work was also supported by the Centre National de la Recherche Scientifique (CNRS), the University of Bordeaux, and the Conseil Régional d'Aquitaine. K.S.P. thanks The Danish Research Council for a DFF | Sapere Aude: Research Talent award (Grant 4090-00201). J.D. acknowledges financial support from the Swiss National Science Foundation, Grant No. PZ00P2_142474. This project has received funding from the European Union's Seventh Framework Programme for research, technological development and demonstration under the NMI3-II Grant Number 283883 for the INS studies.

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This material is based on work at Texas A&M University supported by the U.S. Department of Energy, Materials Sciences Division, under Grant Nos. DE-FG02-02ER45999 and DE-SC0012582. We would like to thank the Laboratory for Molecular Simulation at Texas A&M University for providing software and computer time.

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- | | | |
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5. Dr. Sergei Ostrovosky, Visiting Scholar, Inst. of Applied Physics, Chisinau, Moldova (9/04-12/04; 06/08)
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7. Prof. Sophia Klokishner, Inst. of Applied Physics, Academy Sciences, Kishinev, Moldova (7/06; 06/08)
8. Dr. Andrei Pali, Inst. of Applied Physics of the Academy of Sciences, Kishinev, Moldova (7/06; 06/08)
9. Dr. Akira Ota, Tokyo Institute of Technology, Ookayama, Japan (4/07-9/10)
10. Dr. Ahmed Youssef, Faculty of Science – Ain Shams U., Egypt (7/08-4/09).
11. Maria Fernanda Ballesteros Rivas, Visiting Ph.D. Student Intern, Mexico (9/09-9/10).
12. Ming Fang, Visiting Ph.D. Student Intern, Nankai, China (9/10-8/11).
13. Dr. Q ng-lun Wang, Visiting Chinese Scholar, Nankai University (4/11-2/12).
14. Prof. Anne Richards, Visiting Professor, LaTrobe University, Australia (1/13-4/13).
15. Zhaoyang Li, Visiting Ph.D. Student Scholar, Kyushu University, Japan (10/12-3/13)

16. Prof. Zhao-Xi Wang, Visiting Professor, Department of Chemistry, Shanghai University
(10/13-9/14)