
Crystallization Kinetics and Microstructure Evolution of Vapor-Deposited Hexanitroazobenzene (HNAB) Films

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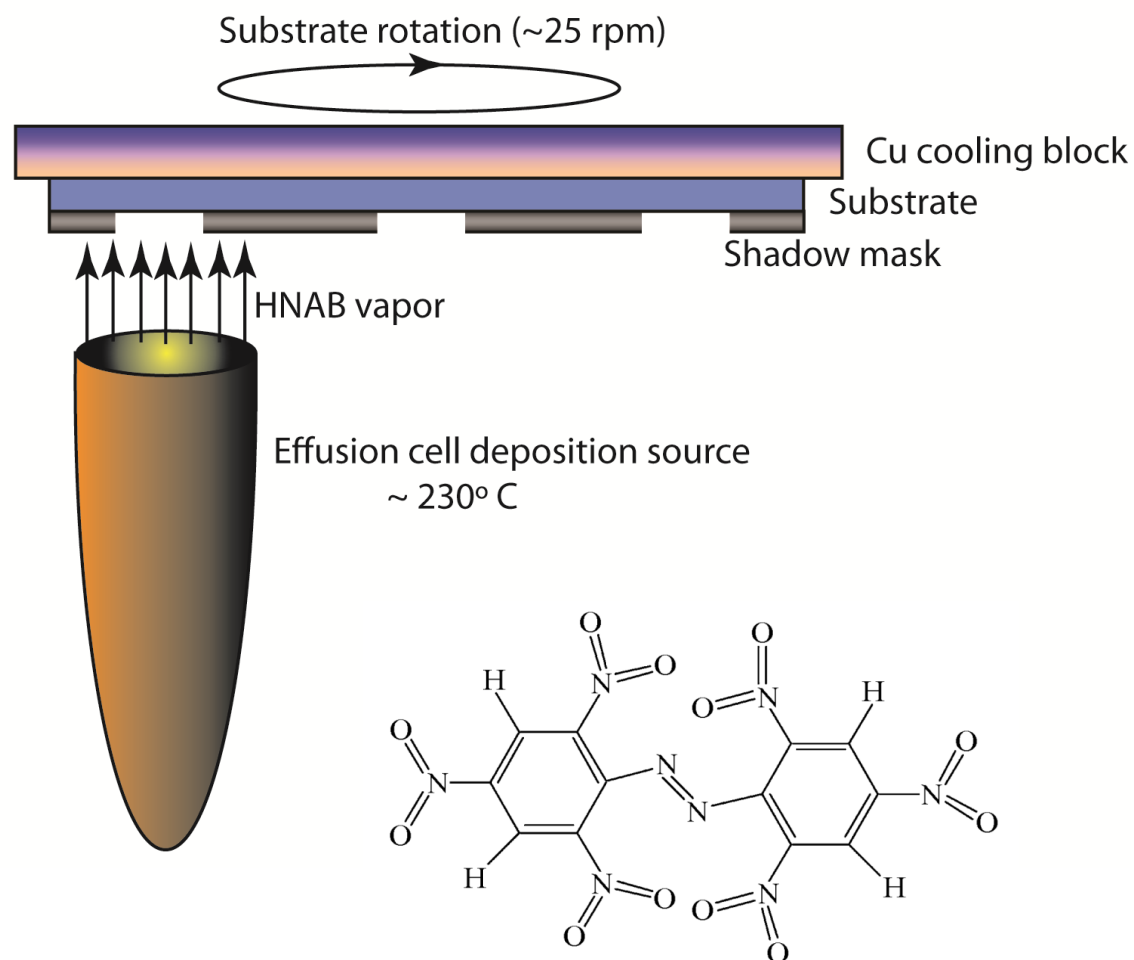
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Motivation

- **Vapor-deposited hexanitroazobenzene (HNAB) is an interesting material for studying explosive behavior at sub-mm geometries**
 - Relatively small critical thickness for detonation to propagate ($< 200 \mu\text{m}$)
 - If substrate is kept sufficiently cool, deposition leads to a dense (non-porous), amorphous structure – independent of substrate or small variations in deposition conditions
- **Amorphous structure is not stable**
 - At RT, crystallization occurs over several weeks – faster at elevated temperatures
 - Effects of temperature on morphology and crystallization kinetics?

Film Deposition



***Schematic of deposition from effusion cell source
and molecular structure of HNAB***

- Base pressure $\sim 10^{-6}$ Torr
- (100) Si substrates
- 10 mm source-substrate distance
- ~ 100 μm thick films after 30 minute deposition

Time-Lapse Optical Microscopy (RT–65°C)



*Time-lapse video of crystallization at 40°C
1.5 hour time-lapse, 3.5 days elapsed*



*Time-lapse video of crystallization at 65°C
1.5 min time-lapse, 2.5 hrs elapsed*

- Similar nucleation and growth process at temperatures 65°C and below
 - ⇒ Orange regions = HNAB-II, Yellow = unknown polymorph
 - ⇒ Higher temperatures lead to faster crystallization
 - ⇒ Greater amount of HNAB-II with increasing temperature

Time-Lapse Optical Microscopy (75°C)

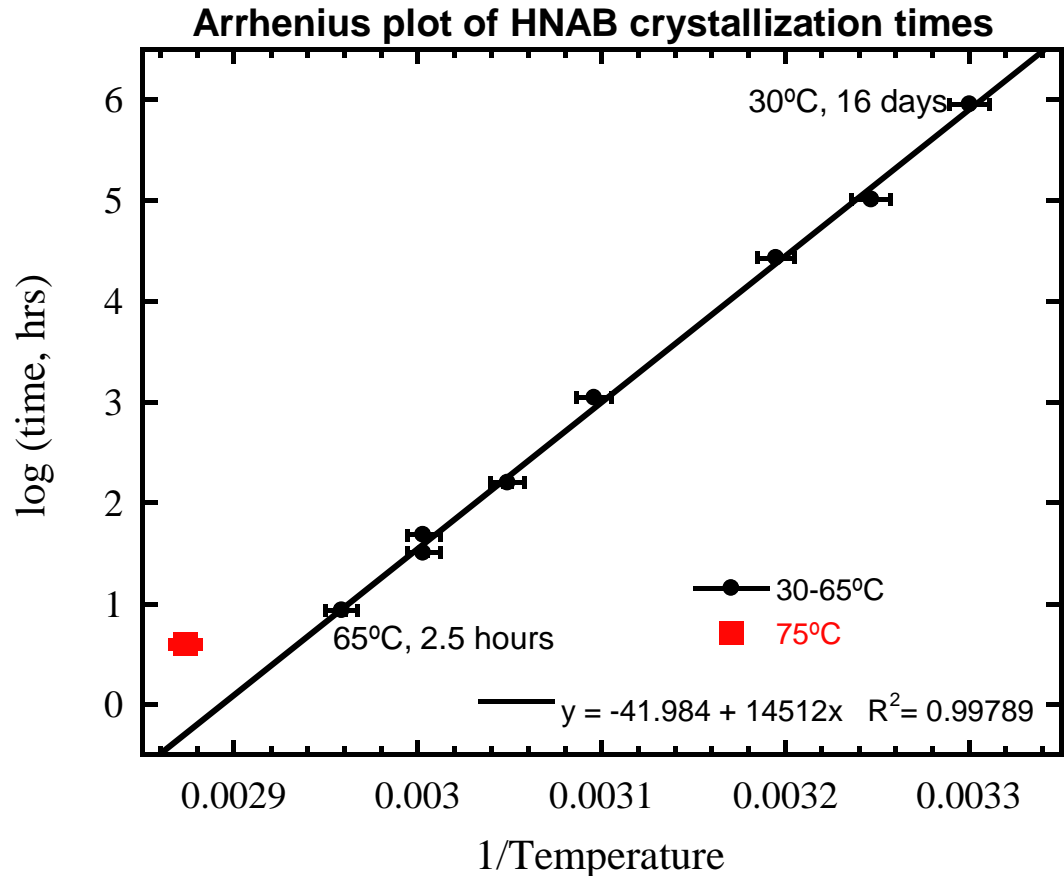


*Time-lapse video of crystallization at 75°C
15 sec (initial darkening)/2 min time-lapse,
2 hrs elapsed*

- Initial surface darkening observed at temperatures above 70°C
- Leads to very different crystallization behavior, progressing primarily from the edges towards the center and forming primarily the unknown polymorph
- Onset of darkening delayed by heating in flowing Ar atmosphere

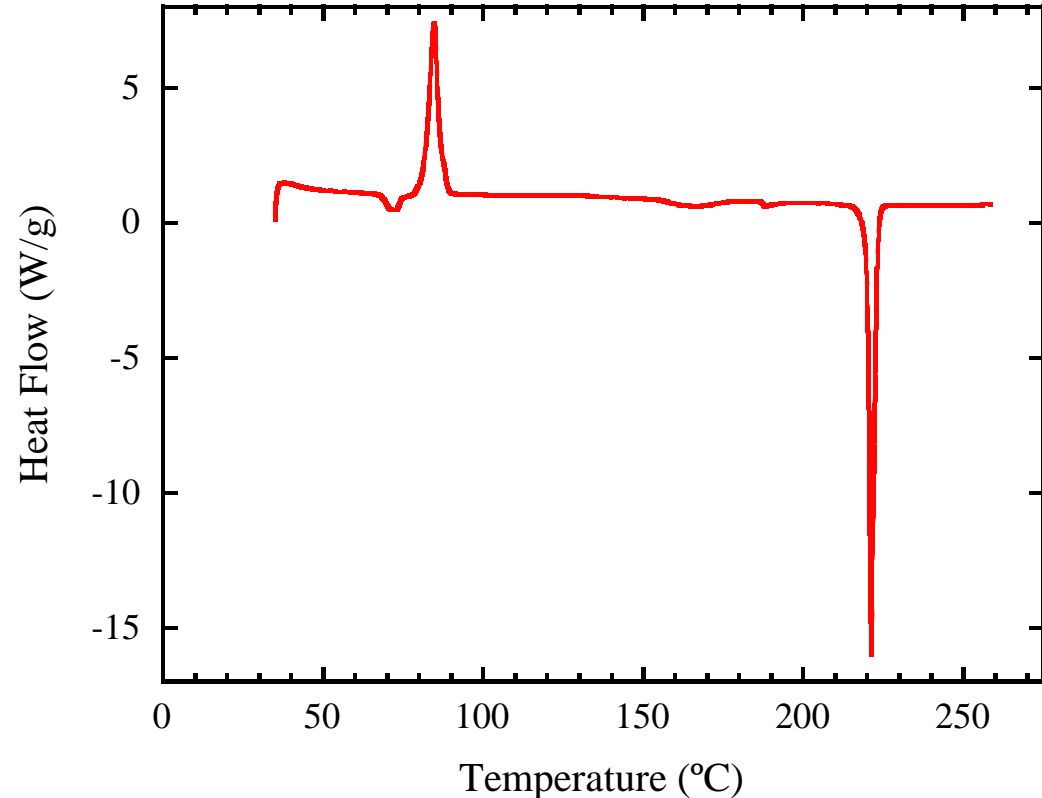
Crystallization Kinetics

- Despite variation in spatial distribution of nucleation sites, time for complete crystallization falls on an Arrhenius plot
- Linear fit appears to be valid for temperatures $\leq 65^{\circ}\text{C}$ (prior to surface darkening)
- Crystallization rate inhibited at higher temperatures (where surface darkening occurs)
- Extrapolates a crystallization time of ~ 8 years at 0°C



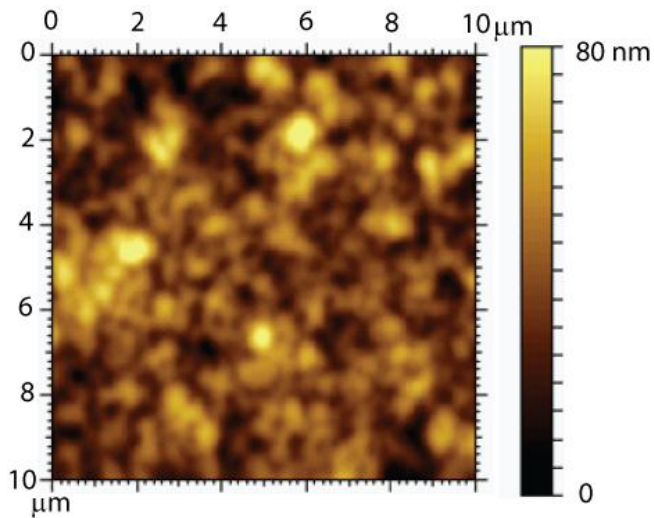
DSC Analysis

- Suggests that glass transition occurs at $\sim 70^{\circ}\text{C}$
 \Rightarrow Same temperature as the onset of surface darkening
- Crystallization exotherm occurs shortly after T_g
 \Rightarrow Heat of crystallization $\sim 55 \text{ J/g}$
- Large melting endotherm at 221°C , consistent with literature
- Smaller endotherms at ~ 166 and 188°C attributed to polymorphic changes



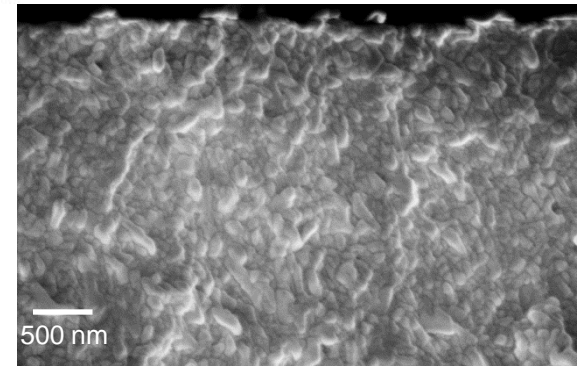
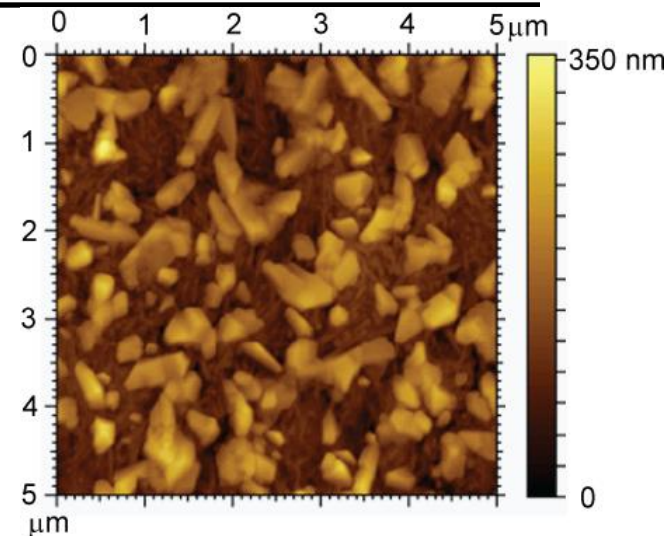
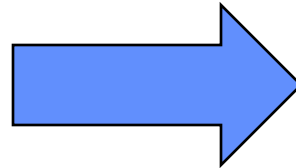
*DSC data from an amorphous HNAB film
heated from 40–250°C at 5°C/min.*

Microstructure Evolution – Low Temperature



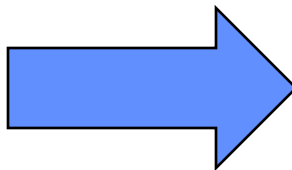
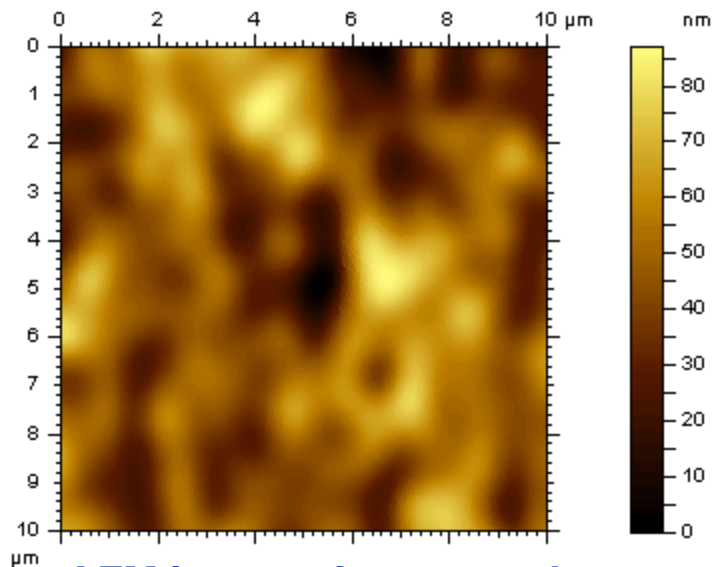
AFM image of an as-deposited amorphous HNAB film

- Crystallization at temperature has only small effects on resultant microstructure below 65°C
- Surface roughness increases from ~15 nm to 50 nm during crystallization
- Grain size after crystallization below 65°C ~100 nm



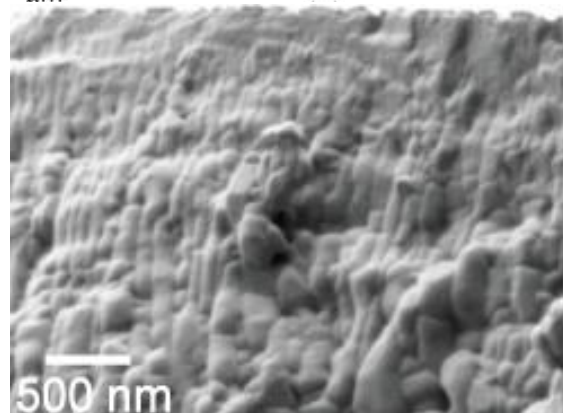
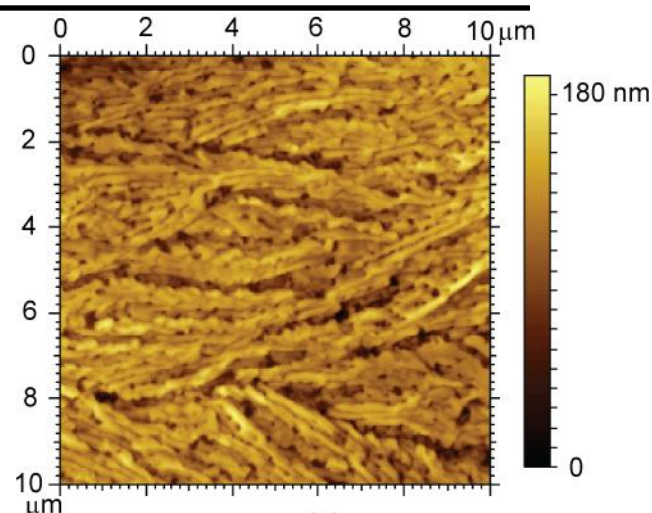
AFM and SEM images following crystallization at 60°C

Microstructure Evolution – Higher Temperatures



AFM image of an amorphous HNAB film after heating to 75°C

- Major changes in microstructure of crystallized HNAB observed at temperatures above 70°C
- Smoother, vermicular surface morphology (~25 nm surface roughness)
- Larger grain size in interior of film (~300nm – 1μm)



AFM and SEM images following crystallization at 75°C



Conclusions

- Vapor-deposited HNAB films form a dense amorphous structure that crystallizes over time to a mixture of HNAB-II and an unknown crystal structure
- At temperatures below $\sim 70^{\circ}\text{C}$, crystallization times follow an Arrhenius behavior and microstructure varies only slightly with crystallization temperature
- At higher temperatures, surface darkening leads to...
 - A different polymorph becoming dominant
 - A very different surface morphology evolving
 - A substantial decrease in the rate of crystallization



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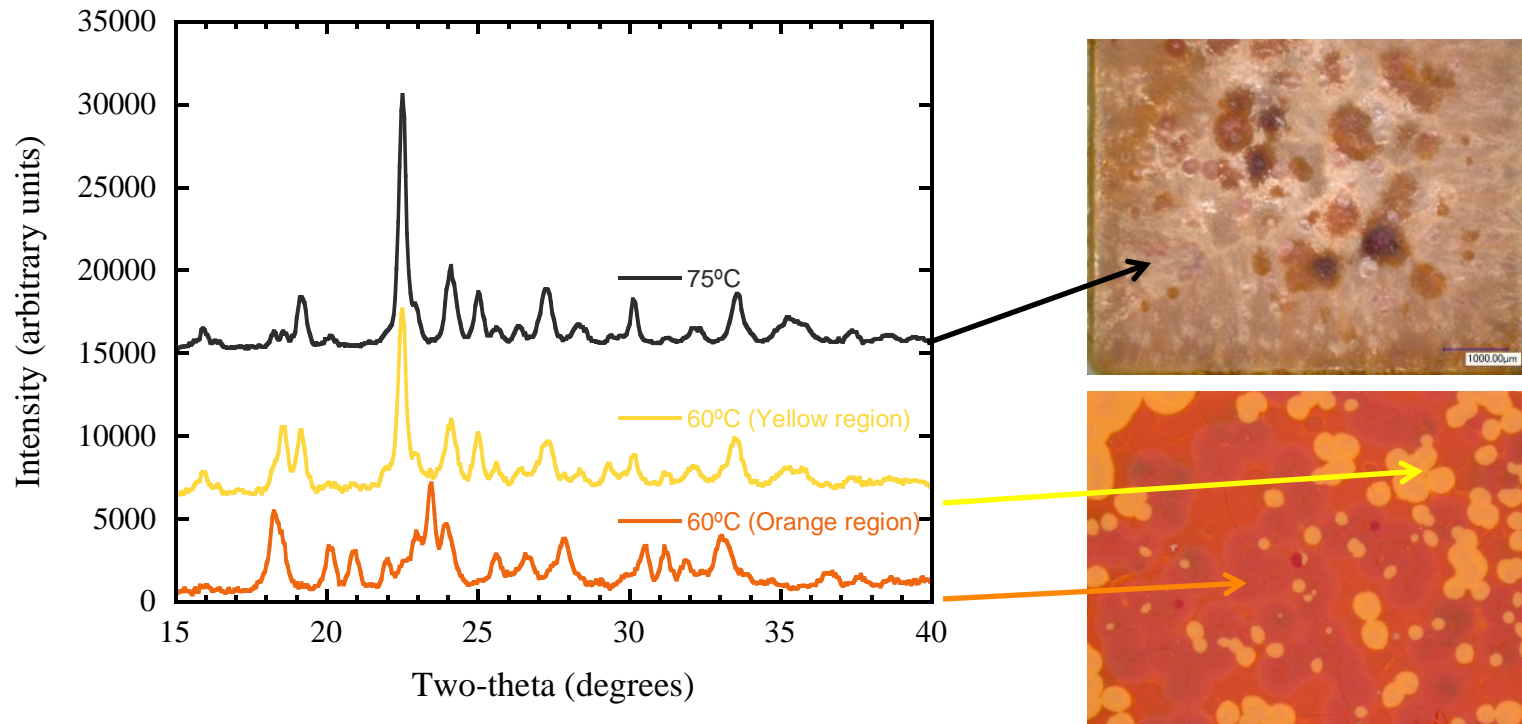
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Robert Patton



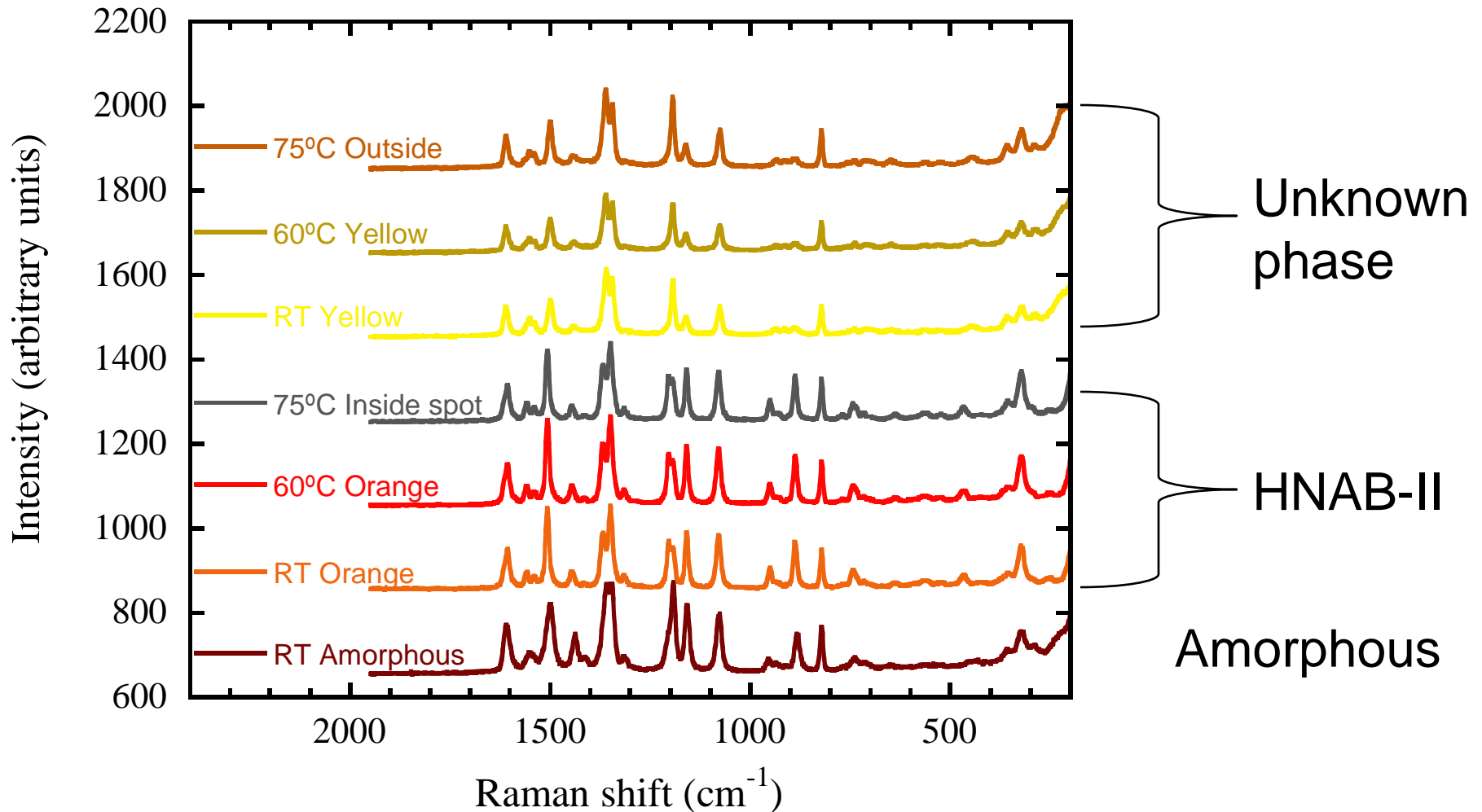
Bonus Slides

X-ray Analysis – Phase Identification



- Regions of different colors isolated using GADDs with 300 μm spot size
- Film at 75°C – Not identified, large peaks match HNAB-III, but others do not
- Film at 60°C –
 - Orange regions consistent with HNAB-II
 - Yellow regions very similar to film crystallized at 75°C

Raman Analysis – RT, 60°C, and 75°C





Future Work

- **Isolate/identify unknown crystal structure**
- **Further kinetics experiments**
 - Crystallization front velocity for each phase
 - Activation energy for crystallization of each phase
- **Effects of microstructure on explosive properties?**
 - Detonation velocity
 - Critical thickness for detonation