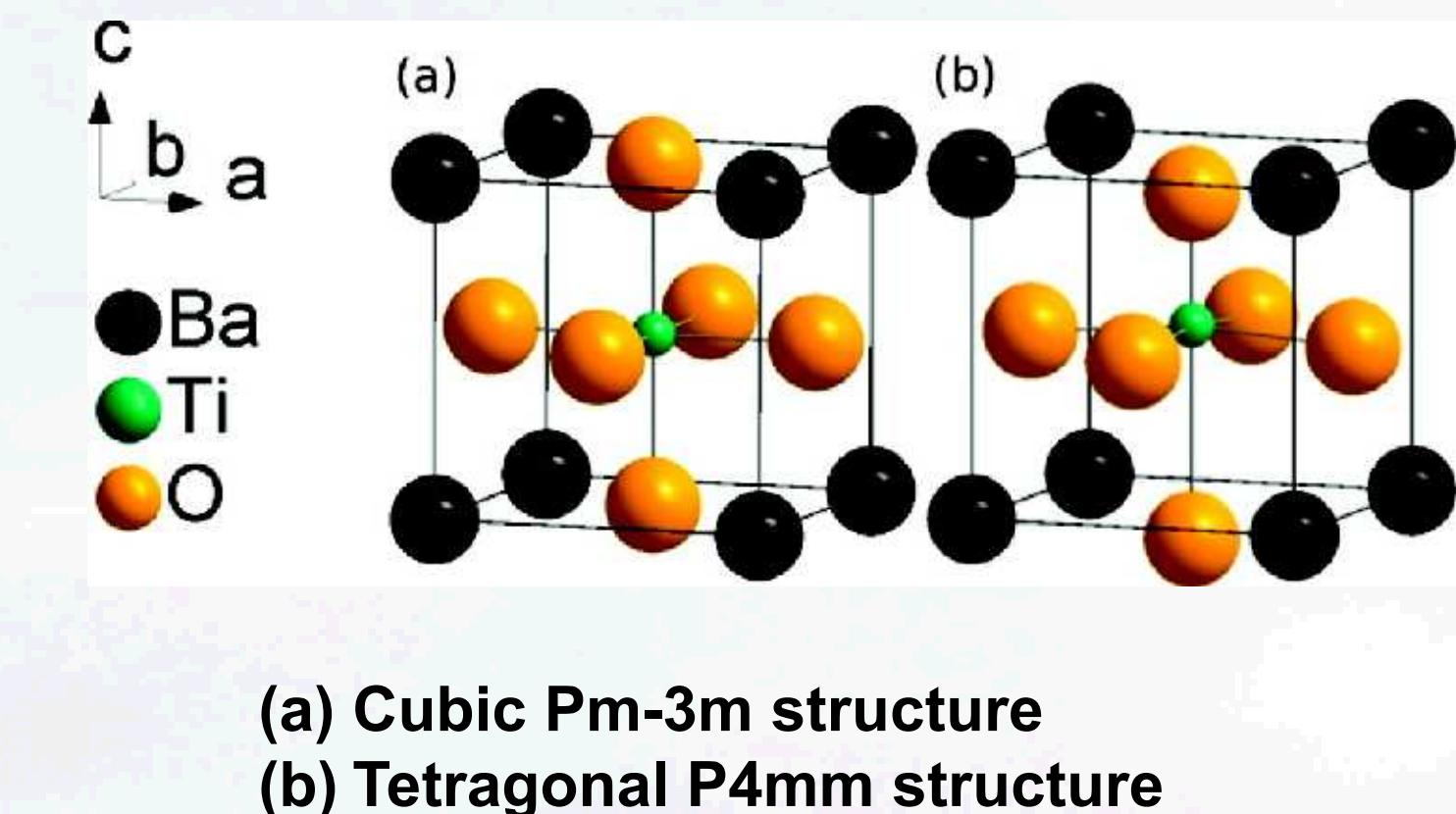


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## Introduction

Ferroelectric nanoparticles, in particular barium titanate (BTO), exhibit bewildering behavior at the nanoscale. In most sintered, fully dense devices, their permittivity drops off with decreasing grain size from a maximum at  $\sim 1\text{ }\mu\text{m}$  and larger grain sizes. However, when BTO particles remain unsintered and are studied in solution or some other matrix material, their permittivity has been shown (although these results remain controversial) to increase dramatically as a critical diameter is approached, after which the particles' permittivity falls precipitously with further decreases in size. Even below this critical size where measured permittivity values of BTO nanoparticles are extremely low and approach zero, some BTO samples have been shown to have tetragonal crystallographic distortions at the local level. However, these tetragonal distortions are not maintained coherently across the entire width of the particle, resulting in a macroscopic crystal structure that resembles a cubic lattice. To date, little is known about how synthesis methods, passivating ligands, and other physical properties of BTO nanoparticles affect their behavior at the nanoscale.

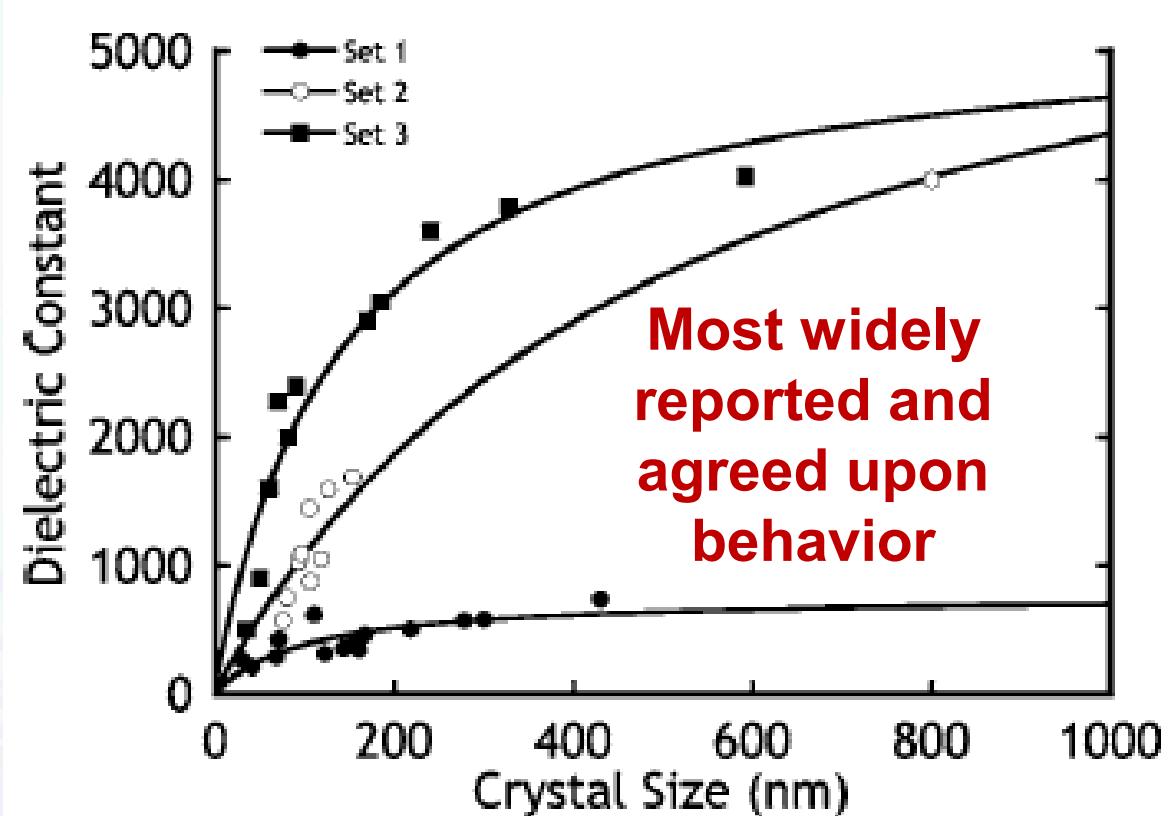


(a) Cubic Pm-3m structure  
(b) Tetragonal P4mm structure

Smith et. al., J. Am. Chem. Soc. 130 (2008)  
6955-6963

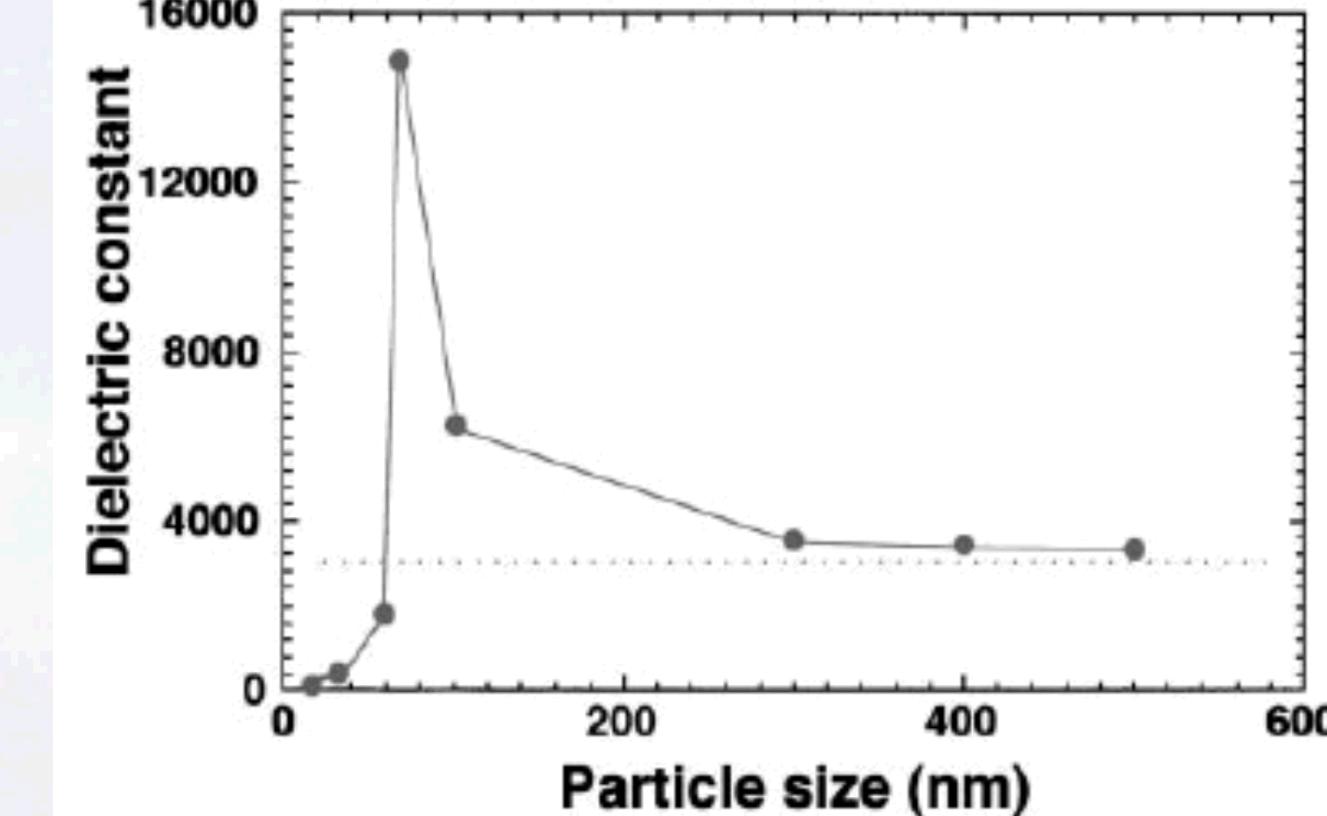
Bulk BTO exhibits a tetragonal phase at room temperature. Spontaneous polarization of the unit cell is due to both a stretching of one axis and a displacement of the  $\text{Ti}^{4+}$  ion. A tetragonal (T) to cubic (C) phase transition occurs at  $125^\circ\text{C}$ . For nanoparticles, T and C phases may coexist.

## Behavior of Nanosized $\text{BaTiO}_3$



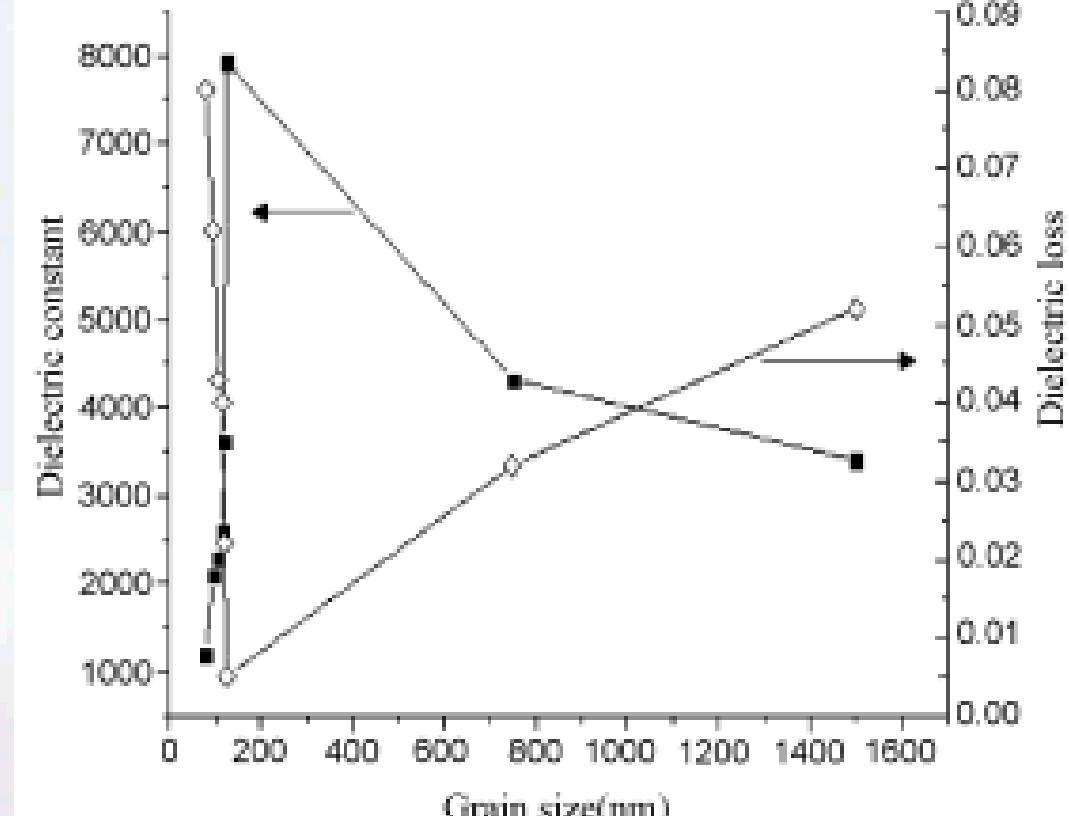
Sintered  $\text{BaTiO}_3$

Aygün et. al., J. Appl. Phys. Vol. 109 (2011)  
034108



BaTiO<sub>3</sub> particles in solution

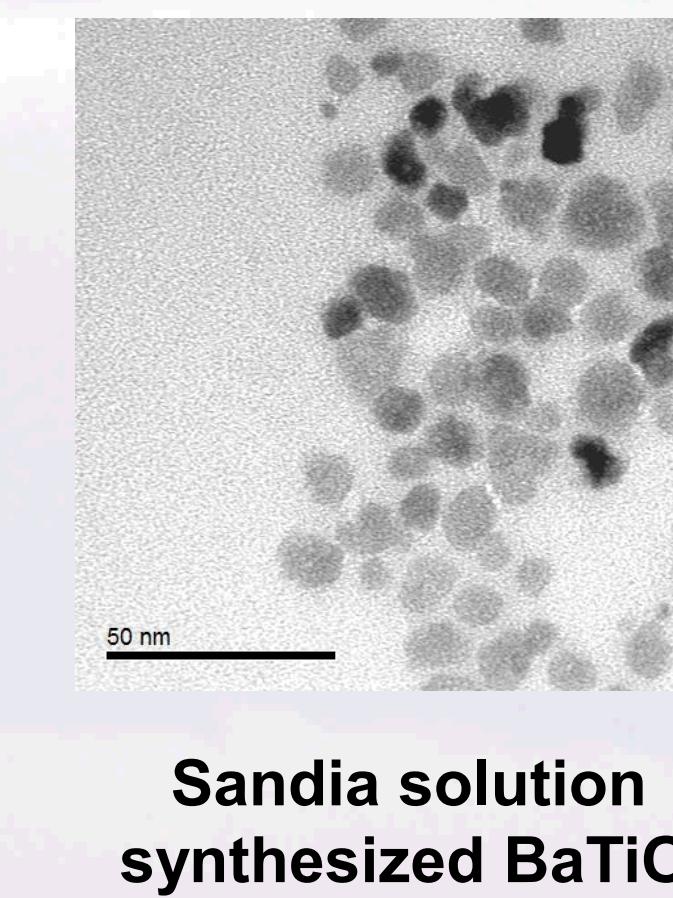
Wada et. al., Jpn. J. Appl. Phys. Vol. 42 (2003) 6188-6195



Sintered  $\text{BaTiO}_3$

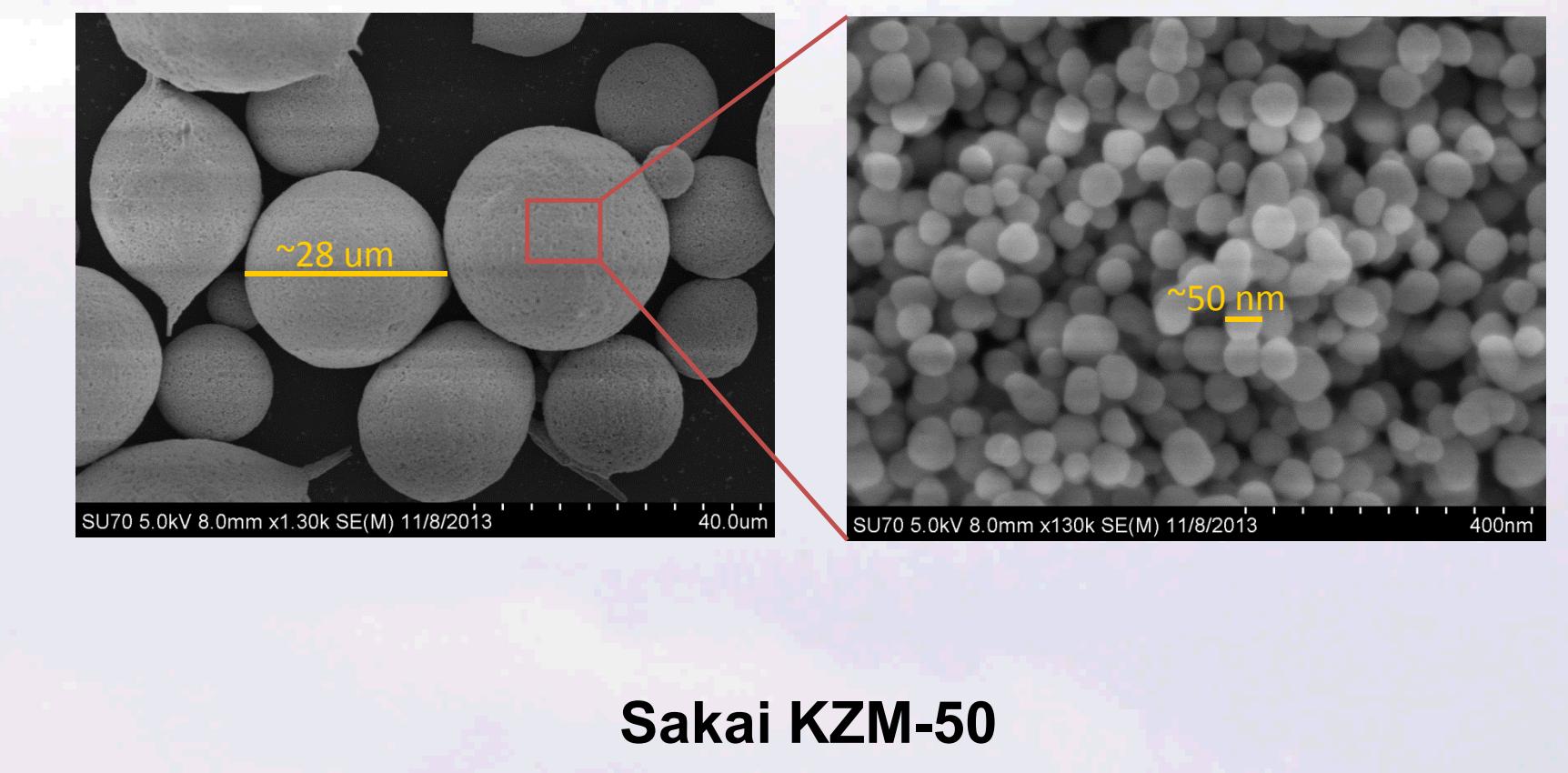
Ying and Hsieh, Materials Science and Engineering B 138 (2007) 241-245

Source	Name	Primary Particle Diameter (nm)	Synthesis Method
Sandia	SNL	10	80°C solution
TPL	HPB-1000	50	hydrothermal
Sakai	KZM-50	50	hydrothermal
Sakai	BT-01	100	hydrothermal
Sakai	BT-02	200	hydrothermal
Sakai	BT-03	300	hydrothermal
Sakai	BT-04	400	hydrothermal
Sakai	BT-05	500	hydrothermal

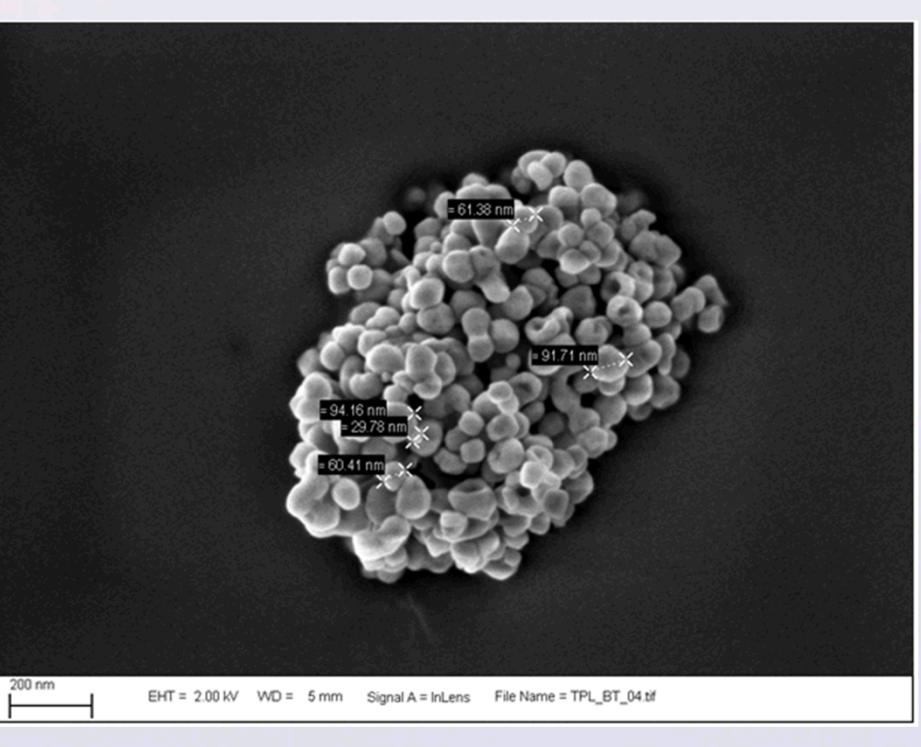


Sandia solution synthesized  $\text{BaTiO}_3$

## The Particles

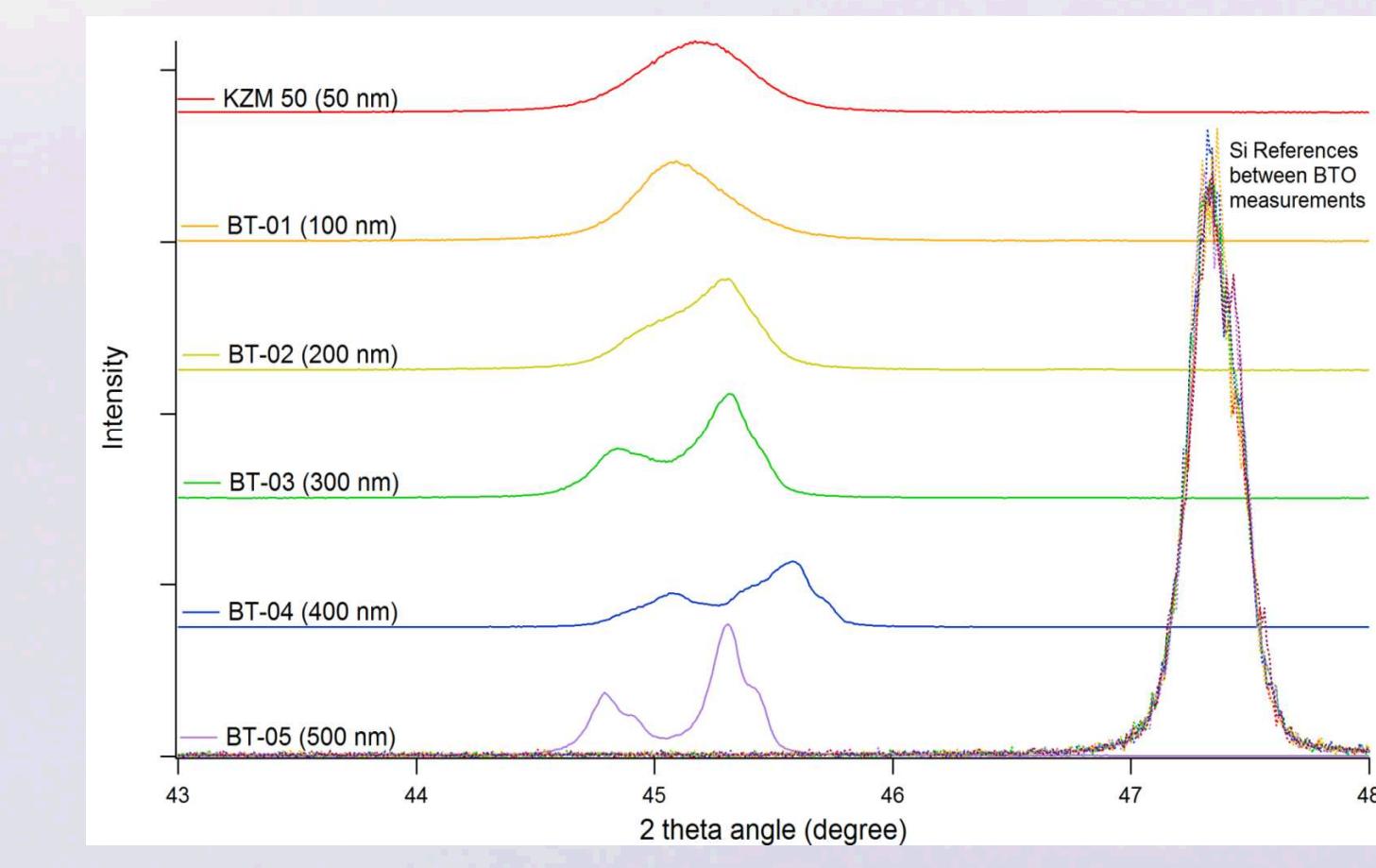


Sakai KZM-50



TPL NanOxide HPB-1000

## Powder X-Ray Diffraction (XRD)



XRD measures a cubic structure for 50 and 100 nm particles. For particles larger than 200 nm, the [200] peak is split, indicating an asymmetry along one axis (tetragonality)

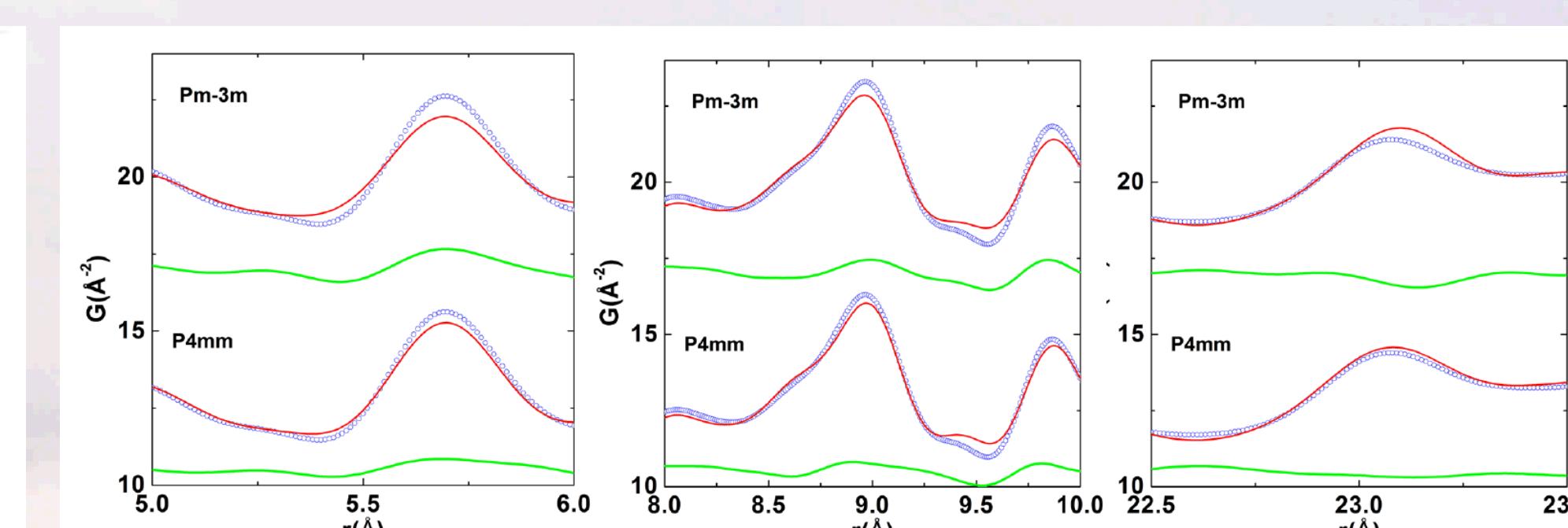
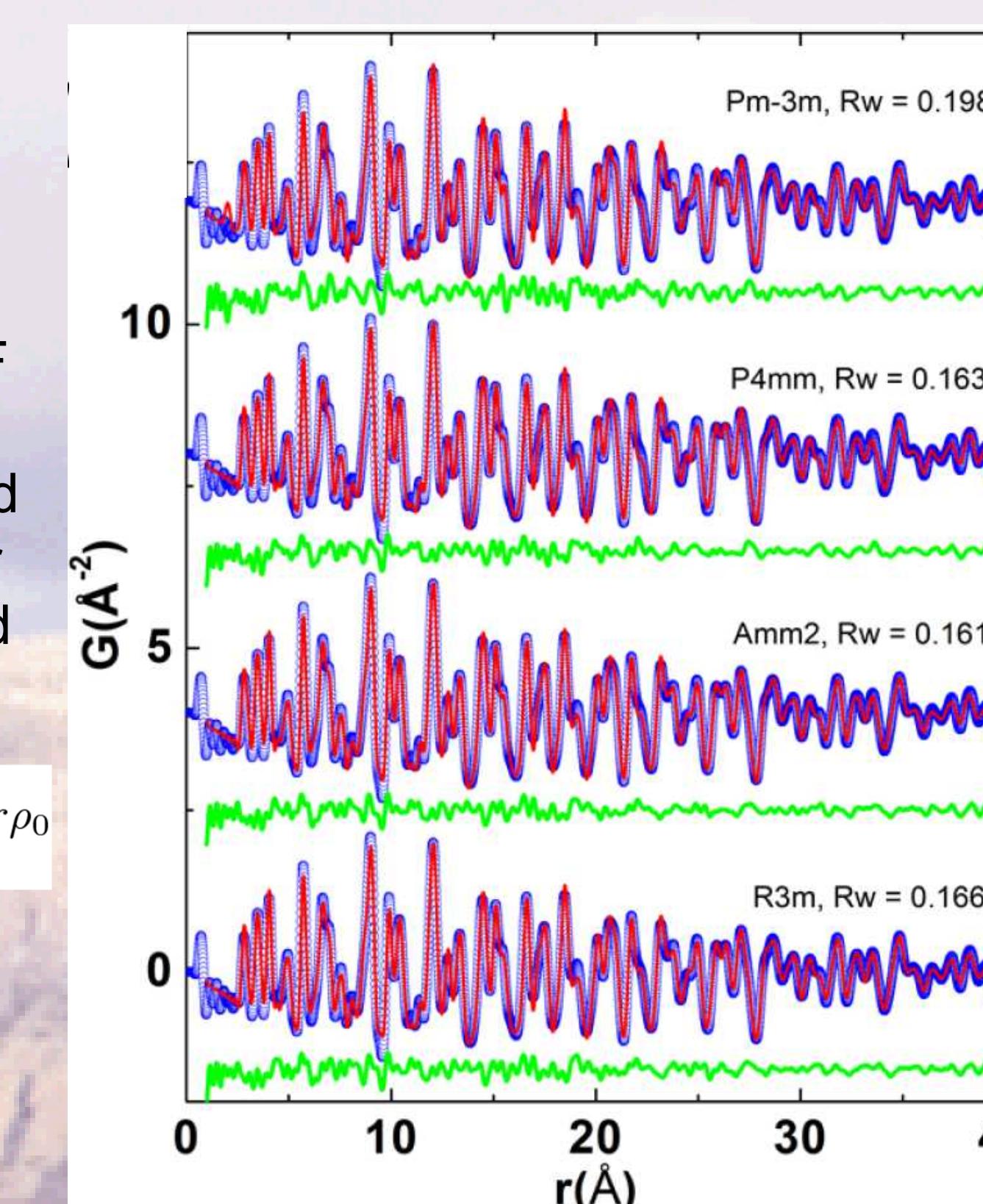
## High Energy X-Ray Scattering at NSLS (Brookhaven National Lab), analyzed via the atomic Pair Distribution Function (PDF)

### Introduction to PDF

The atomic Pair Distribution Function (PDF) is the probability of one atom finding a neighboring atom at a distance "r". The PDF makes no assumption about periodicity, evaluating all intensities from both Bragg and diffuse scattering. It is obtained by a Fourier transformation of powder diffraction data and its relationship to a material's structure is:

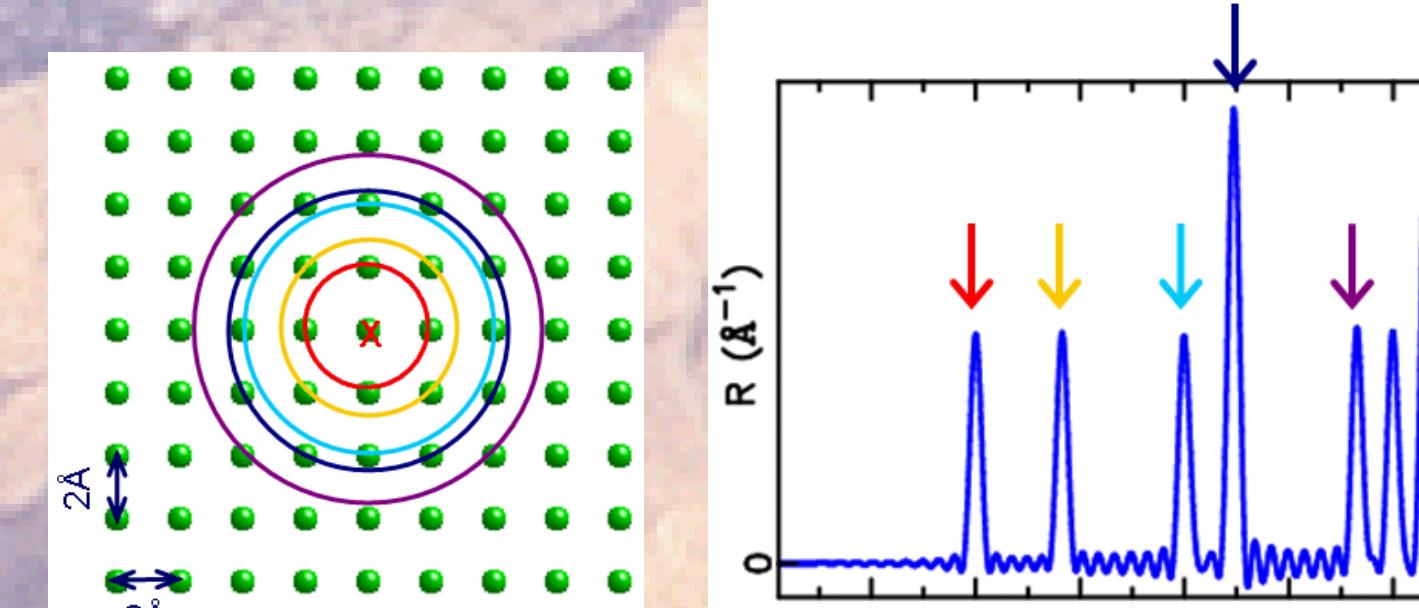
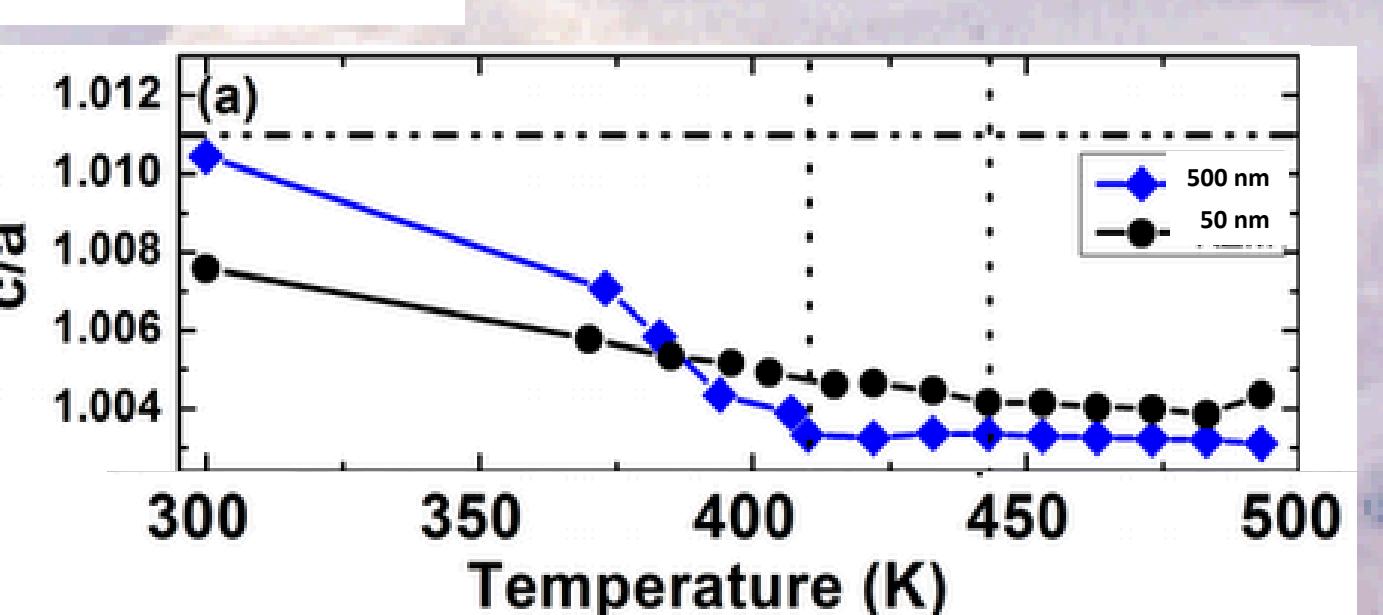
$$G(r) = 4\pi r [\rho(r) - \rho_0] = \frac{1}{Nr} \sum_{m \neq n} \frac{b_m b_n}{\langle b \rangle^2} \delta(r - r_{mn}) - 4\pi r \rho_0$$

Farrow and Billinge, Acta Cryst. A Vol. 65 (2009)  
232-239



Expanded PDF fits (red) to experimental PDF data (blue) of SNL  $\text{BaTiO}_3$  nanoparticles at 300 K using cubic (top) and tetragonal (bottom) phase groups. The tetragonal fit is visibly superior.

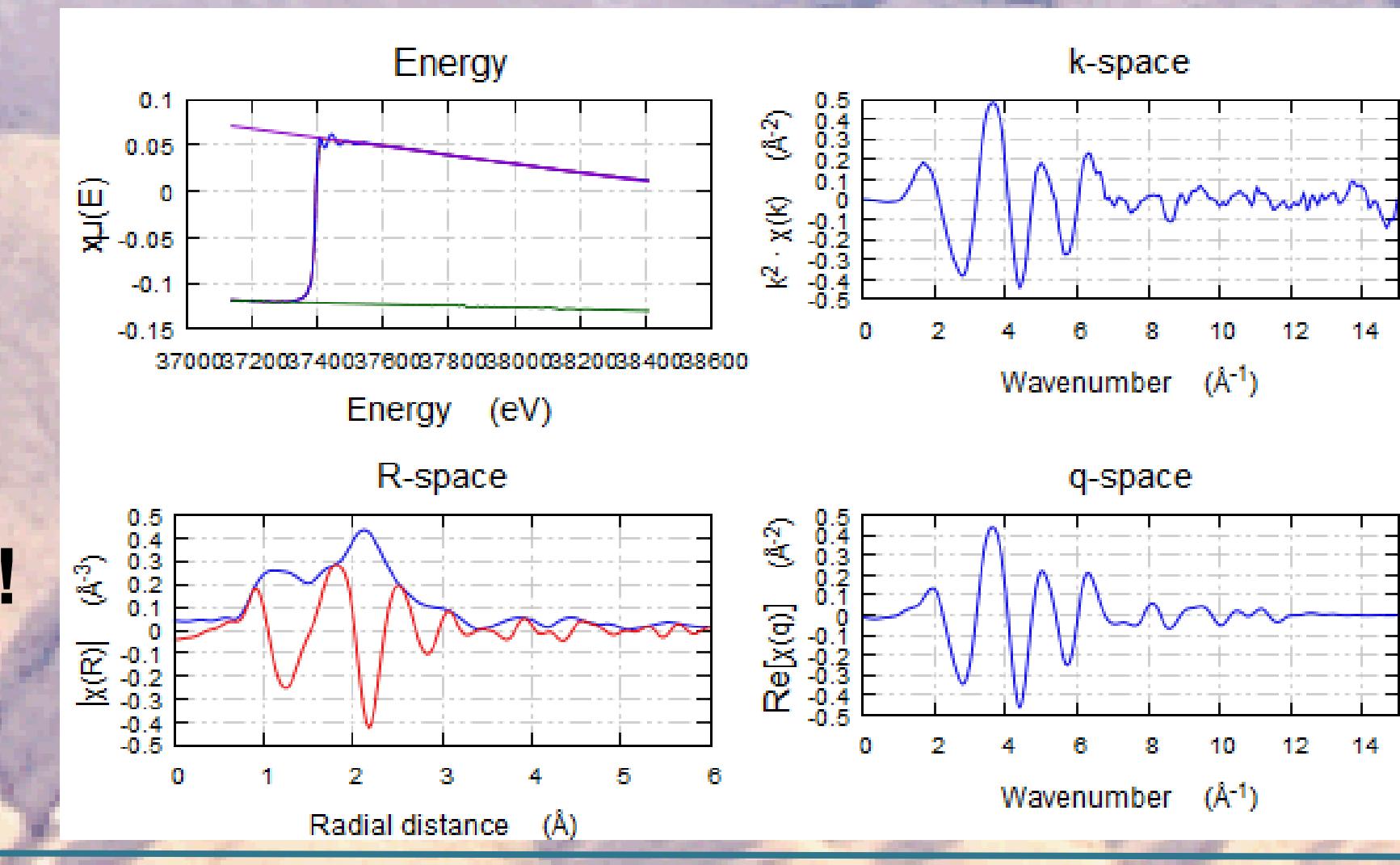
Measurements from the National Synchrotron Light Source (BNL) demonstrate reduced tetragonality at room temperature for 50 nm particles and confirm the diffuse and incomplete  $T \rightarrow C$  transition suggested in Raman.



PDF model fits to Sandia  $\text{BaTiO}_3$  nanoparticles measured at 300 K with four different candidate space groups. Blue circles are the measured data, red curves are the calculated PDFs of the best-fit structural models, and the green curves offset below are difference curves.

## XAFS at APS (Argonne National Lab):

Collected but still being fit!



## Acknowledgments

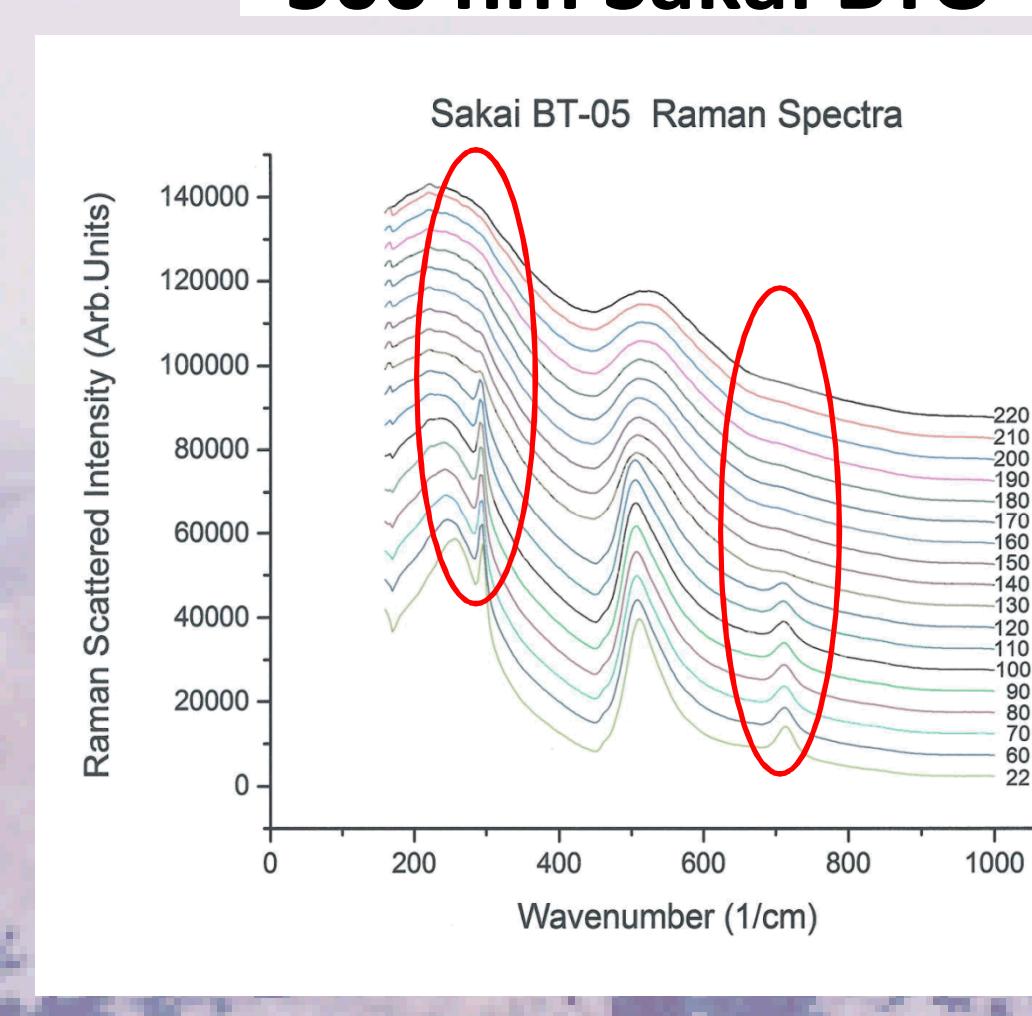
Prof. Charles Taylor (Pomona College)  
Prof. Robert Gaines (Pomona College)  
Prof. David Tanenbaum (Pomona College)

Prof. Ronald Grimm (Harvey Mudd College)  
Tetsuya Ohtani (Sakai Chemical)

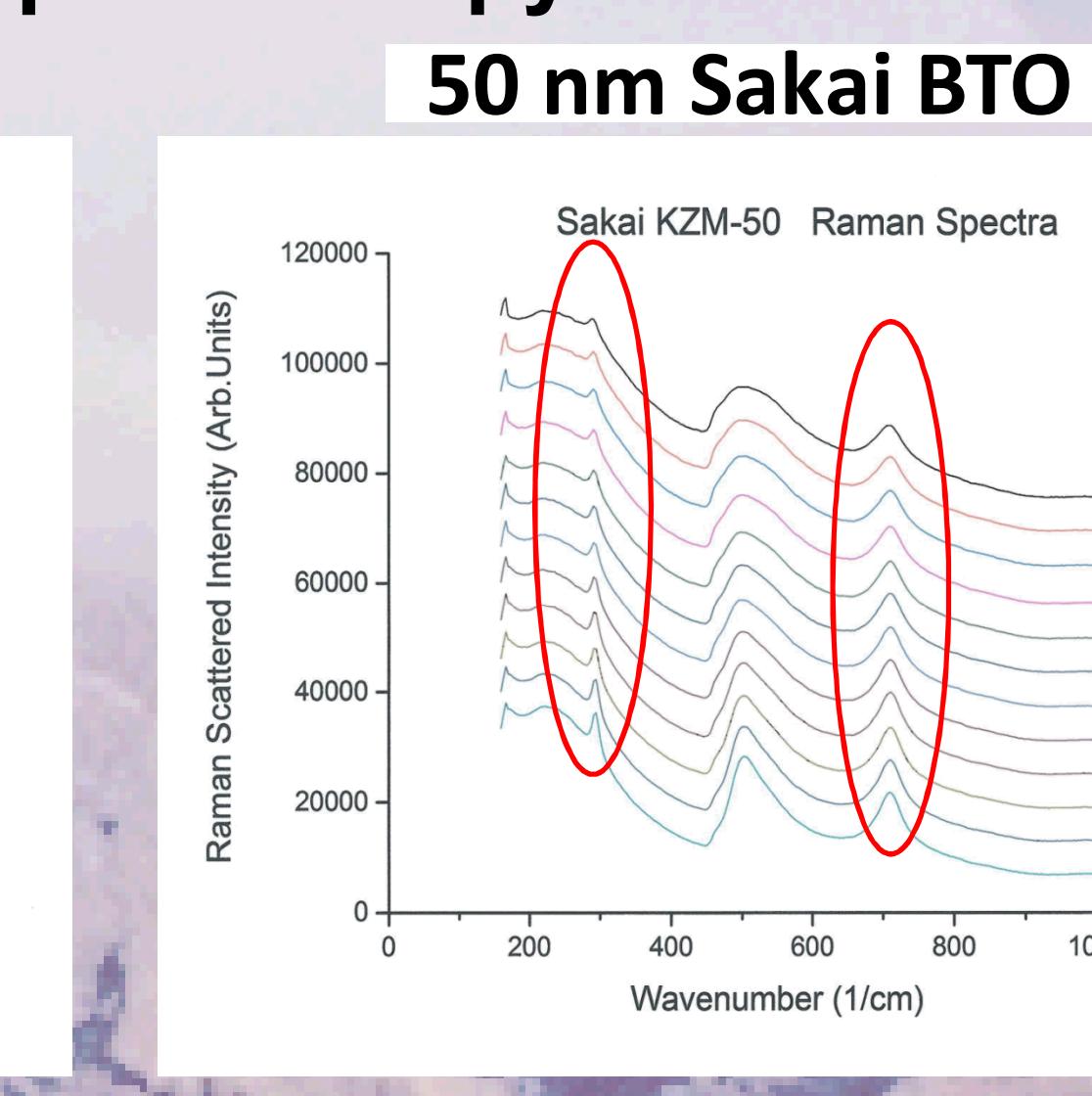
Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000. Use of the National Synchrotron Light Source, Brookhaven National Laboratory, was supported by the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences, under Contract No. DE-AC02-98CH10886. The DND-CAT located at the Advanced Photon Source (APS) is supported by E. I. DuPont de Nemours & Co., The Dow Chemical Co., and the State of Illinois. Use of the APS was supported by the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences, under Contract DE-AC02-06CH11357. Work in the Billinge group was supported by the Columbia University Energy Frontier Research Center (EFRC) funded by the U.S. Department of Energy, Basic Energy Sciences (DOE-BES), under grant DE-SC0001085.

## Raman Spectroscopy

### 500 nm Sakai BTO



For 500 nm BTO, as in bulk, the  $300\text{cm}^{-1}$  and  $710\text{cm}^{-1}$  peaks disappear sharply at the  $T \rightarrow C$  transition (125°C).



For 50 nm BTO, these peaks remain, suggesting persisting tetragonality at elevated temperatures.

## Conclusions:

- Particles  $> 200\text{ nm}$  behave metrically tetragonal
- Particles  $< 200\text{ nm}$  have a distorted tetragonal (or possibly other) structure
  - Diffuse phase transition
  - Possible disordered coexistence of phases

## Future Work:

- Analyze XAFS data for more insight into local BTO structure
- Study the effect of surface functionality (ligands) on the crystal structure of SNL BTO nanoparticles