

# STAR Fellowship in CINT's Electron Microscopy Lab



Erin Meadors  
Center for Integrated Nanotechnologies  
Katie Junjohann, Mentor  
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This work was performed at the Center for Integrated Nanotechnologies (CINT), a U.S. DOE Office of Basic Energy Sciences user facility. Sandia National Laboratories is a multiprogram laboratory managed and operated by Sandia Corporation, a wholly-owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC0494AL85000.



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# CINT



<http://sstp.org/wp-content/uploads/CINT.jpg>

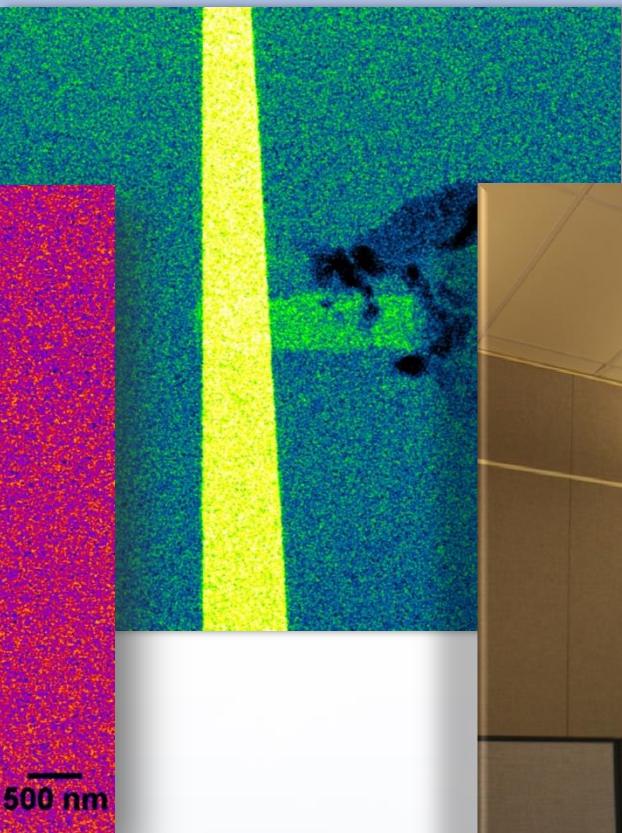
## Department of Energy Nanoscale Science Research Center



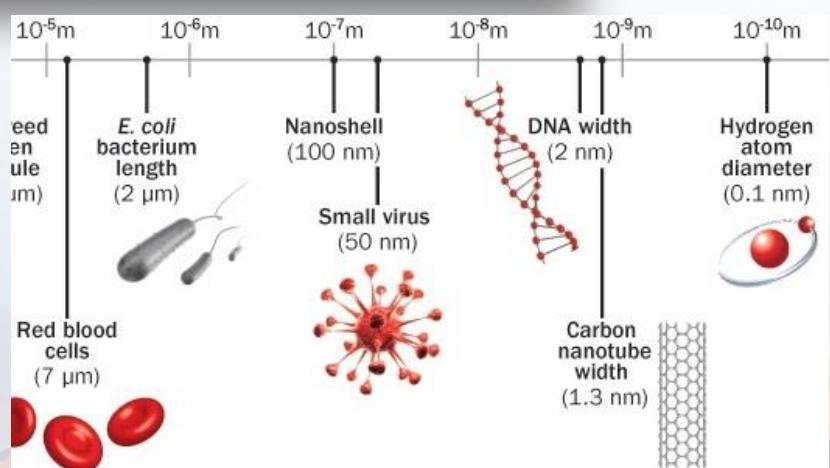
<https://nsrportal.sandia.gov/>



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# TEM

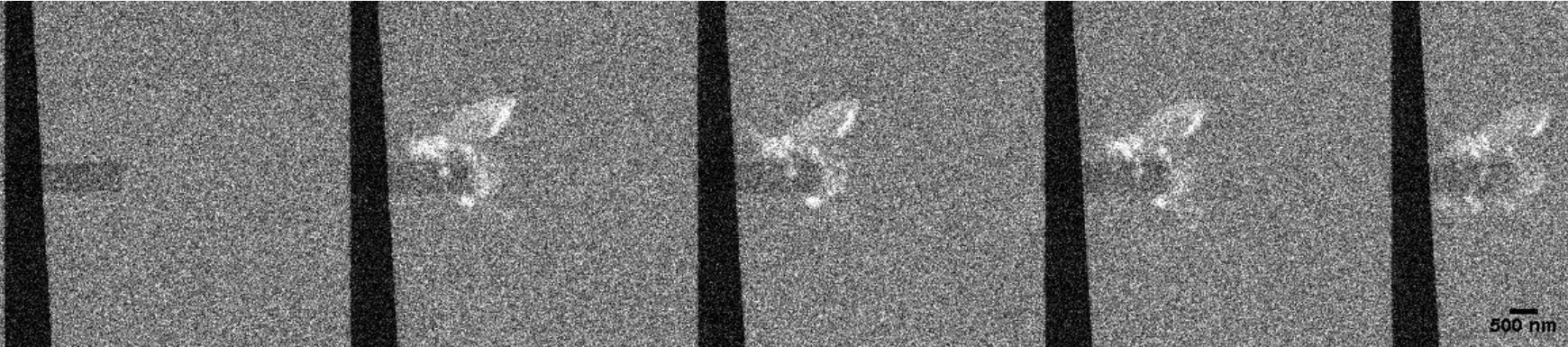


[https://www.sciencenews.org/pictures/120410/nanotubes\\_sizes\\_zoom.gif](https://www.sciencenews.org/pictures/120410/nanotubes_sizes_zoom.gif)

ORNL Oak Ridge National Laboratory

# Image processing

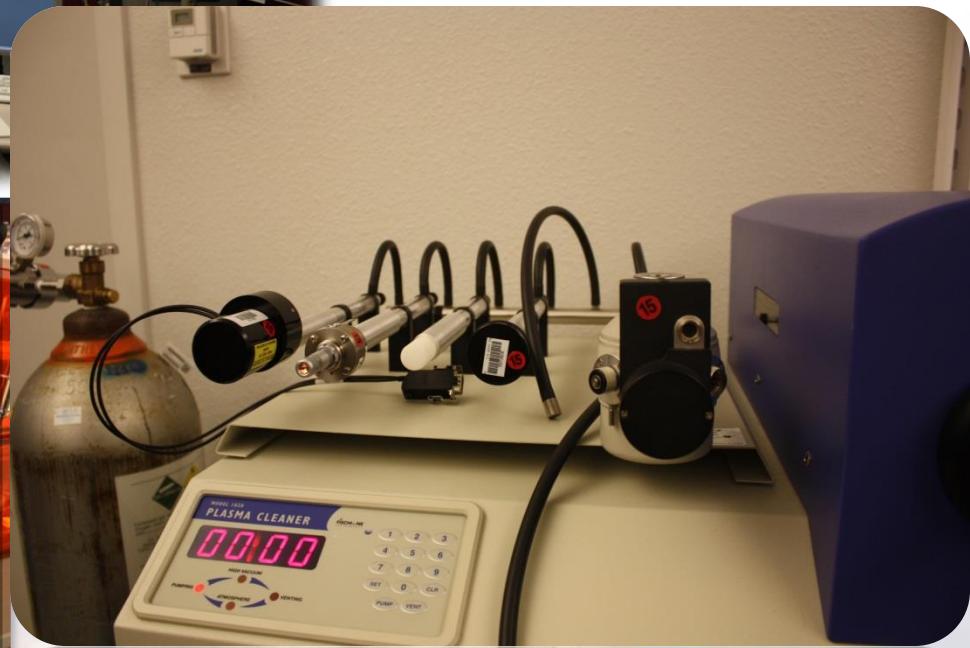
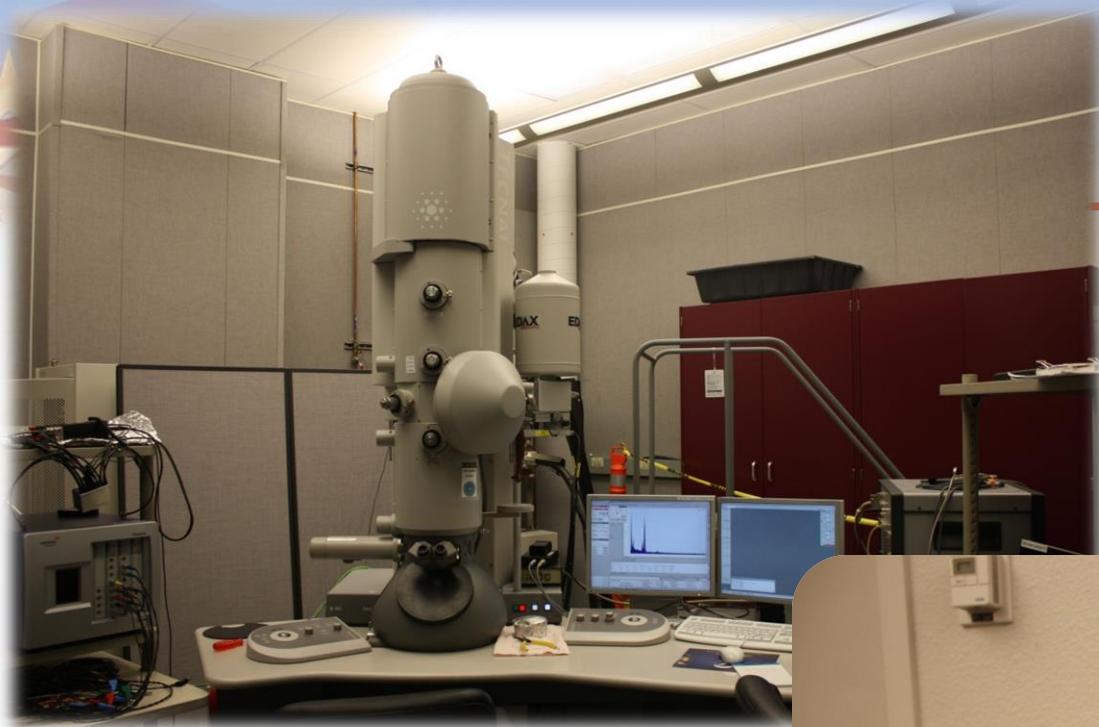
**Bright Field**



**Dark Field**



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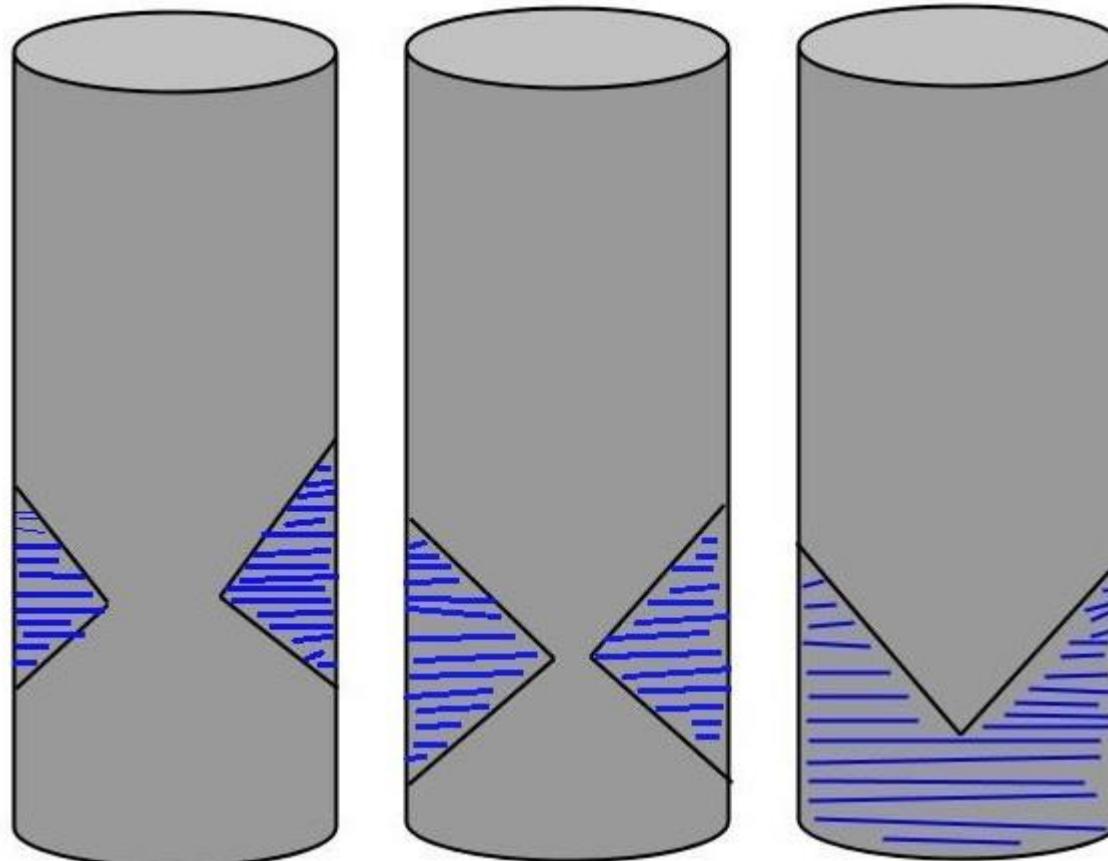


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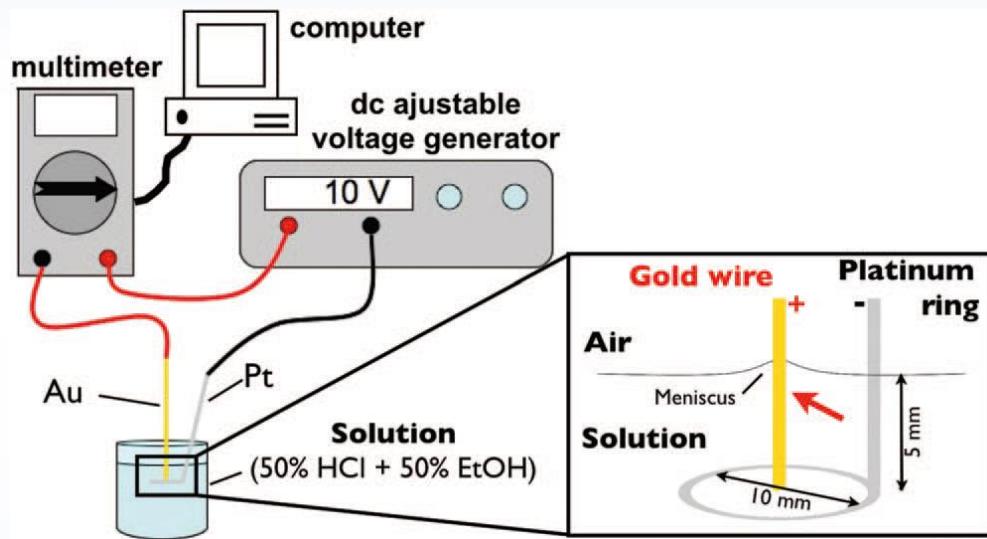
# Electrochemical Etching of Probes

- Using a mixture of chemicals and applying a current to the solution to etch away the material from a thin wire to form a very sharp point

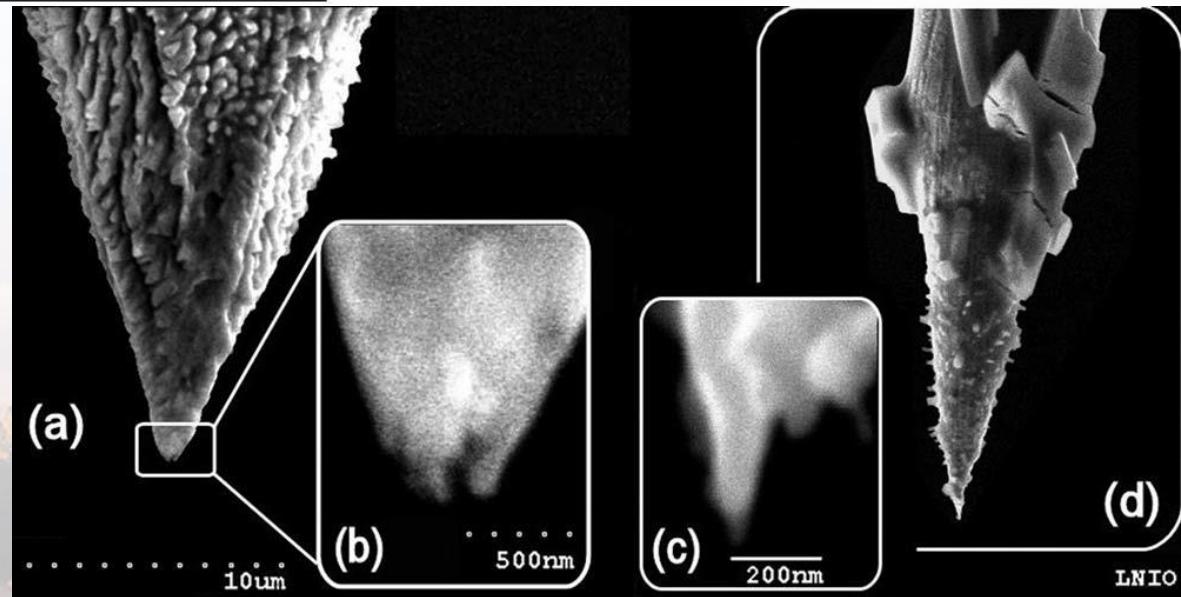


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# Background

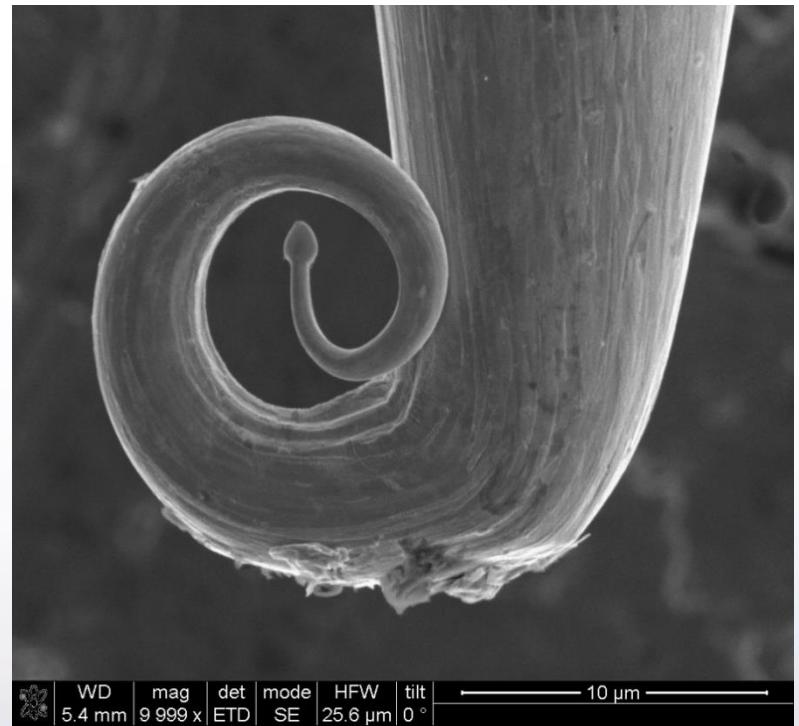
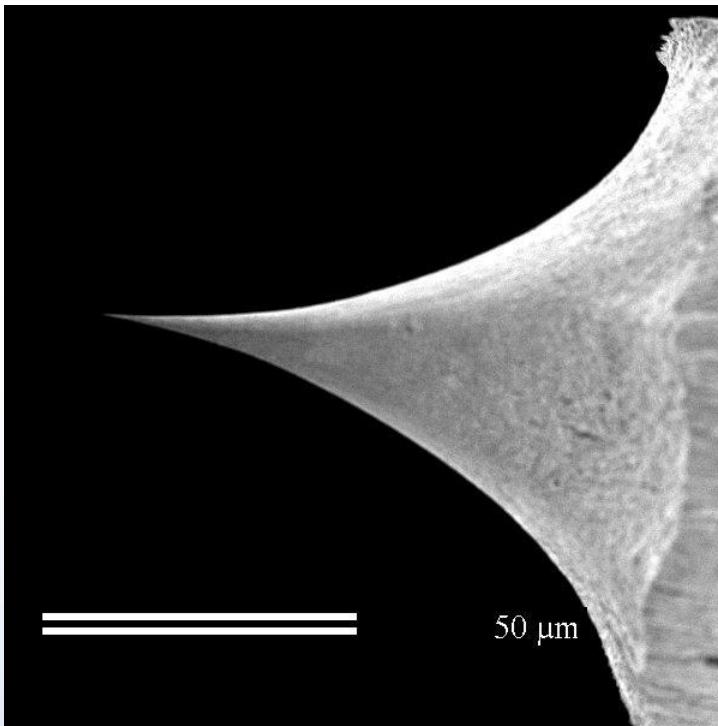


Review of Scientific Instruments **84**, 073702 (2013);  
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# Expectations and Reality

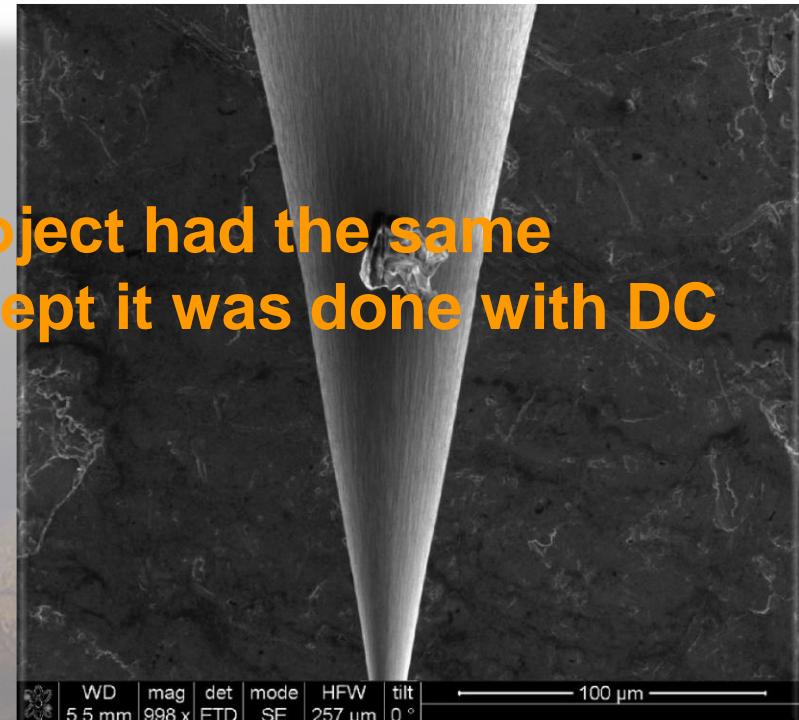


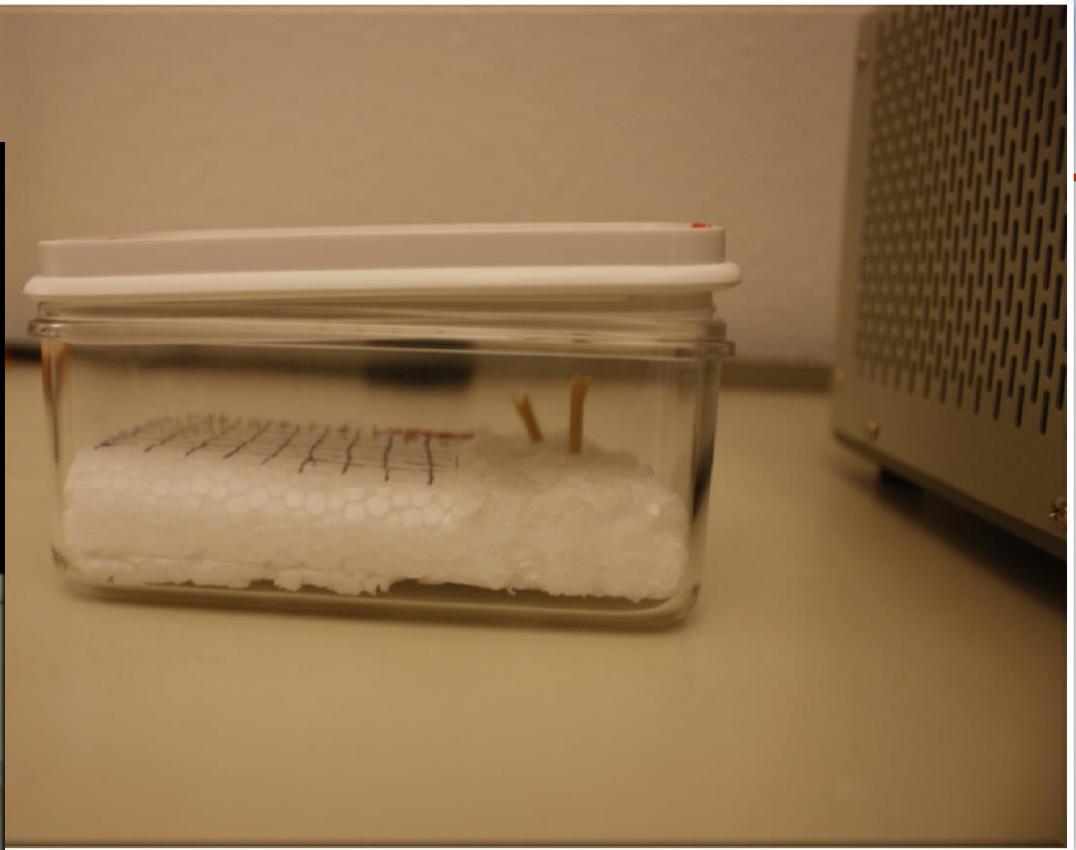
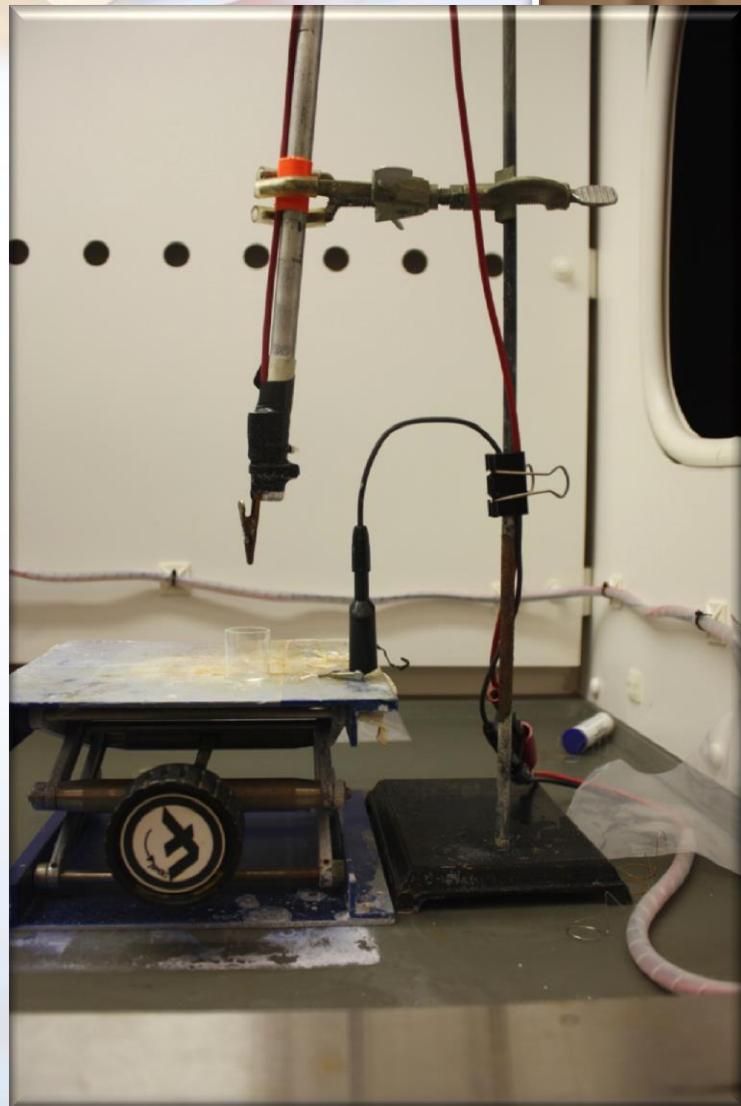
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# My Project

A test on 10 Tungsten probes etched at constant AC voltages: 20, 15, 10, 5, and 2.5 in a solution of NaOH (sodium hydroxide) and NH<sub>4</sub>OH (ammonium hydroxide)

The second part of my project had the same conditions as the first except it was done with DC voltage instead of AC





Set up for etching is inside of fume hood to ensure safety during the experiment



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**Nail Polish  
Marker**

**Tungsten Wire**

**NH<sub>4</sub>OH**

**Platinum  
Ring**

**NaOH  
Pellets\***

\*the pellets should be entirely dissolved in the solution before the Pt ring or W wire are added and the electrolysis process begins



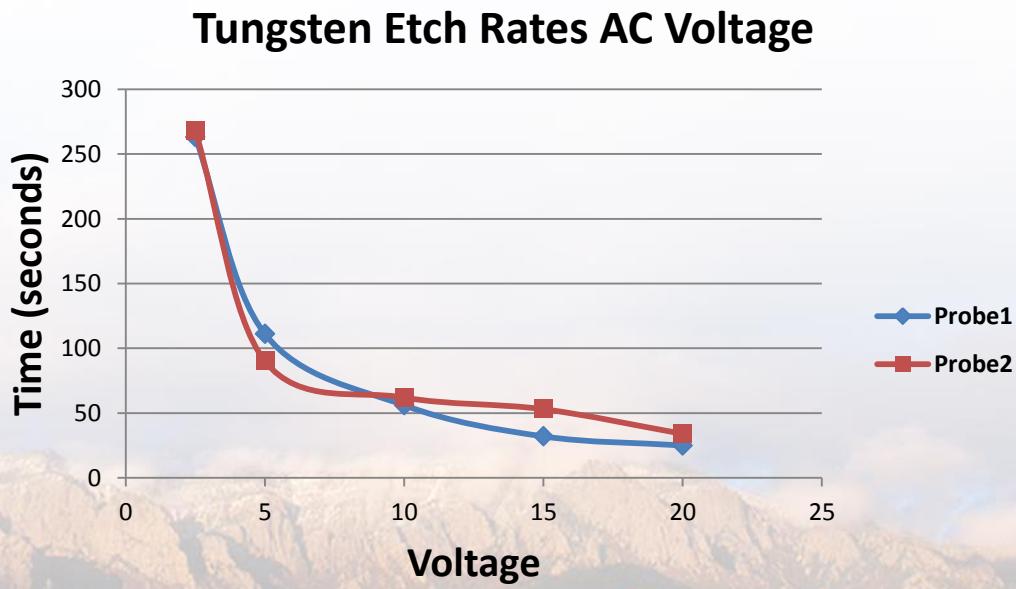
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# Tungsten Data AC

Voltage (Volts)	Time for First Probe (sec)	Time for Second Probe (sec)
20	25	34
15	32	53
10	56	62
5	111	90
2.5	263	268

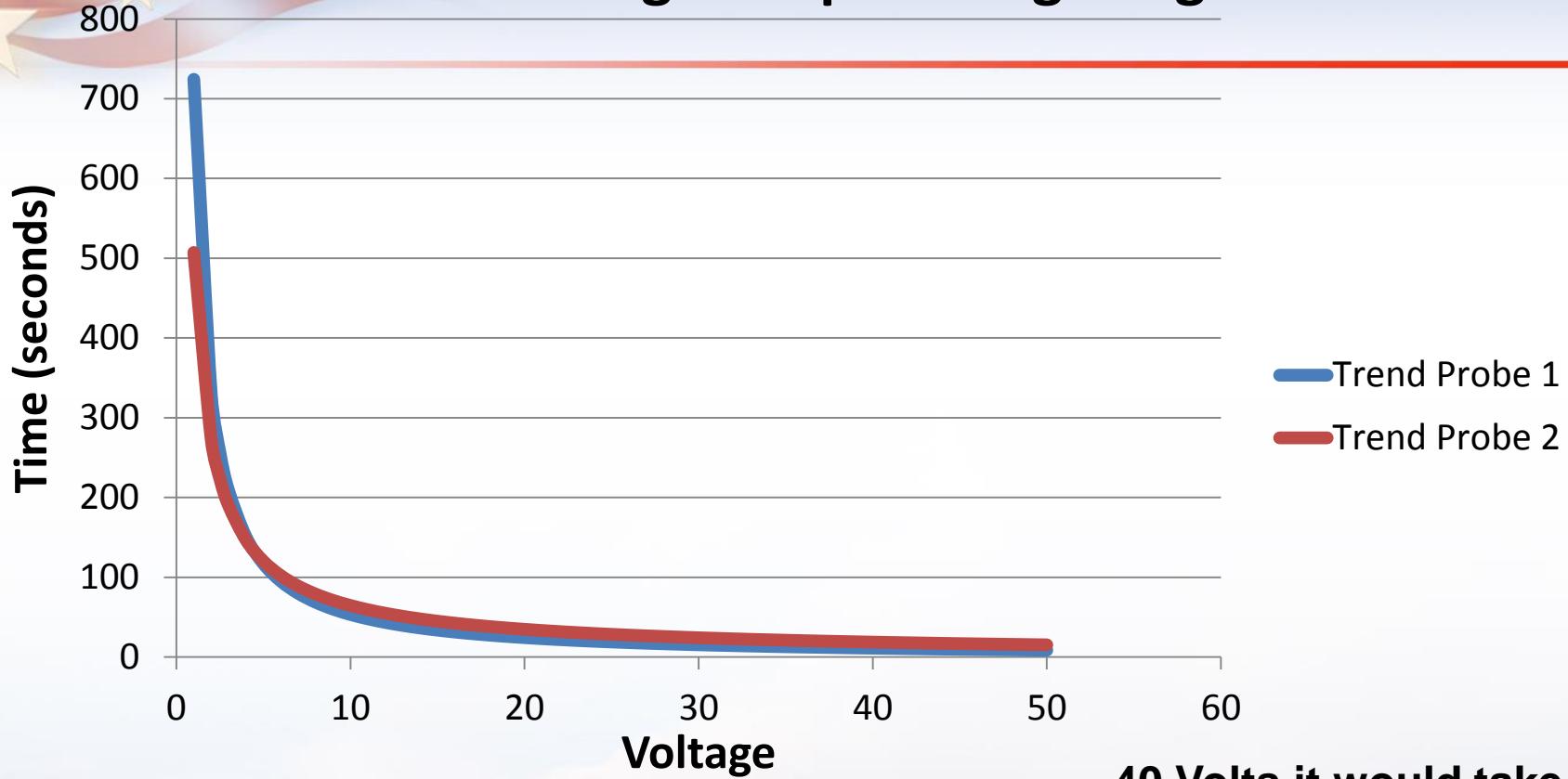
A trend can be seen from the data as the amount of time it took the tips to fall off increased as the Voltage was decreased

The graph shows the data starting from the lowest voltage, showing how much longer it took to etch



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# Estimation For Single Step Etching Tungsten AC



Trend 1:  $y = 724.3x^{-1.133}$

$R^2 = 0.9975$

Trend 2:  $y = 507.22x^{-0.895}$

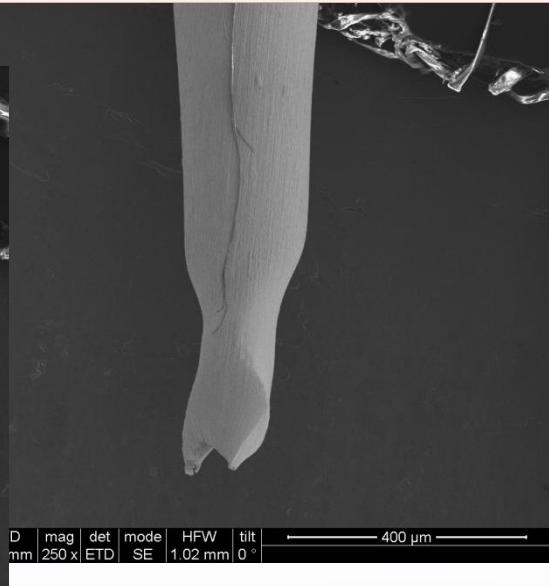
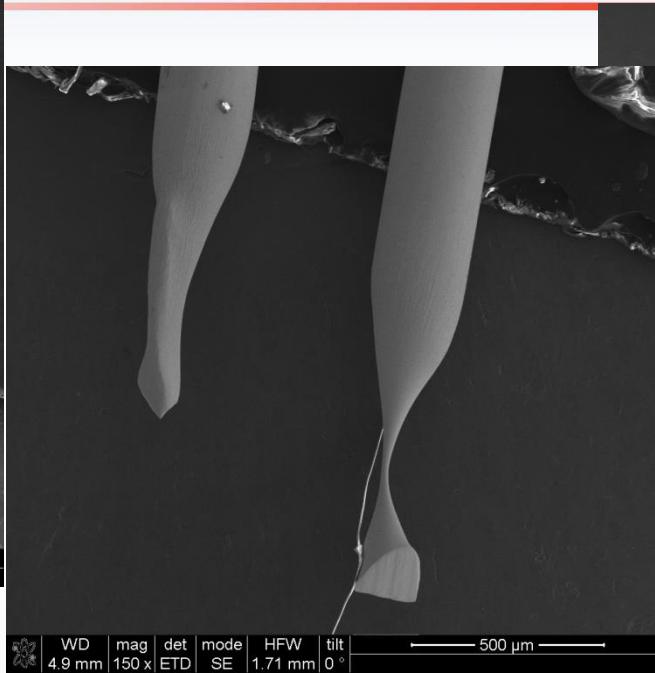
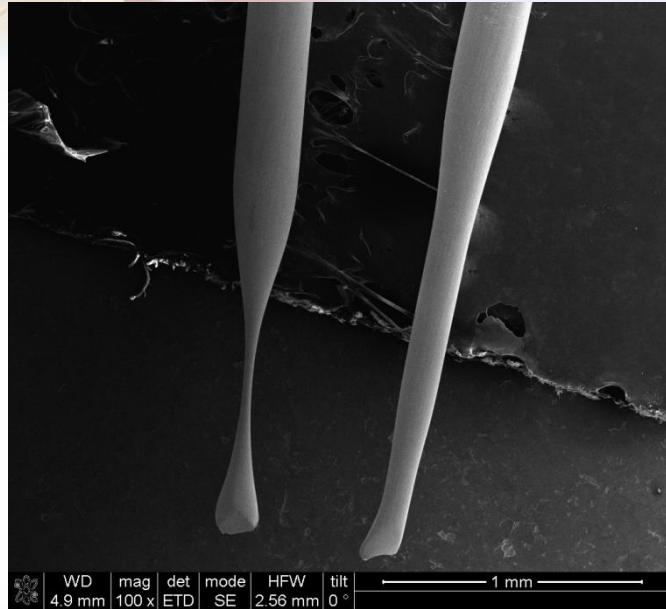
$R^2 = 0.9401$

40 Volts it would take approximately 18 seconds, and at 1 Volt it would take around 12 minutes



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# Finished AC Tungsten Probes



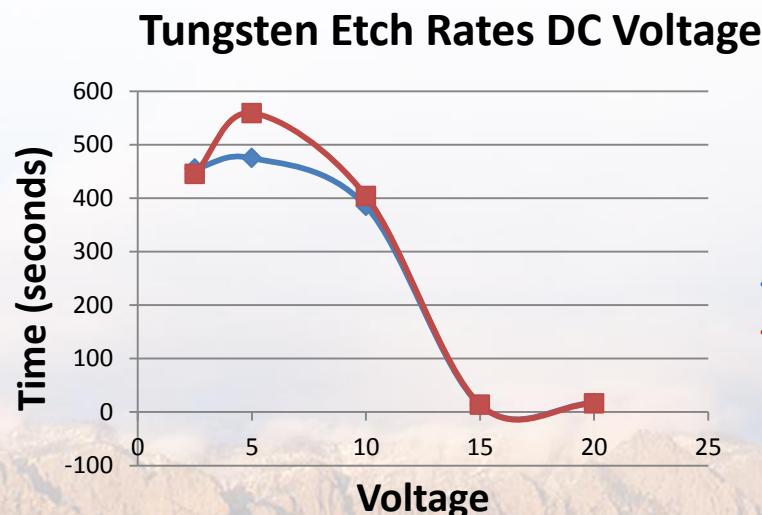
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# Tungsten Data DC

Voltage (Volts)	Time for First Probe	Time for Second Probe
20	17 sec	16 sec
15	13 sec	14 sec
10	6 min 25 sec	6 min 44 sec
5	7 min 55 sec	9 min 19 sec
2.5	7 min 35 sec	7 min 25 sec

The results from the DC voltage test were greatly varied, ranging from seconds to several minutes, however overall the DC voltage took longer on average than the AC voltage tests

There may have been a few outliers, it was often hard to know when the tip would drop since times were so varied during the experiment



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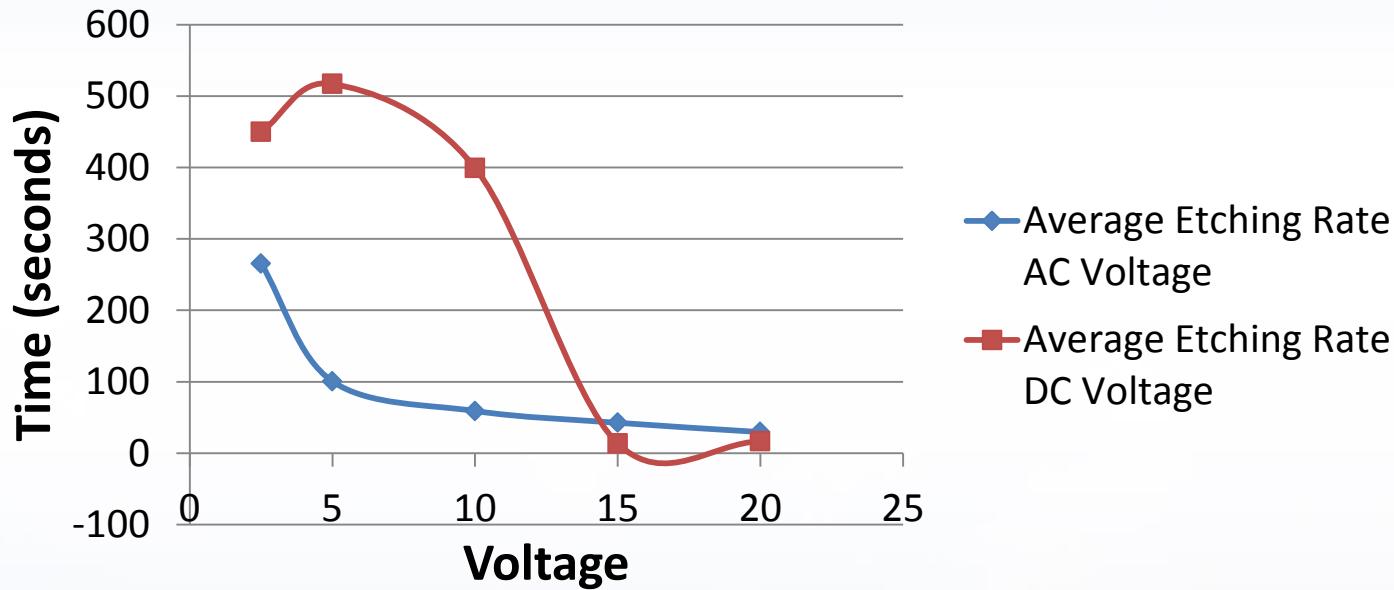
# Finished DC Tungsten Probes

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# Comparison of Etch Rates for AC and DC Voltage



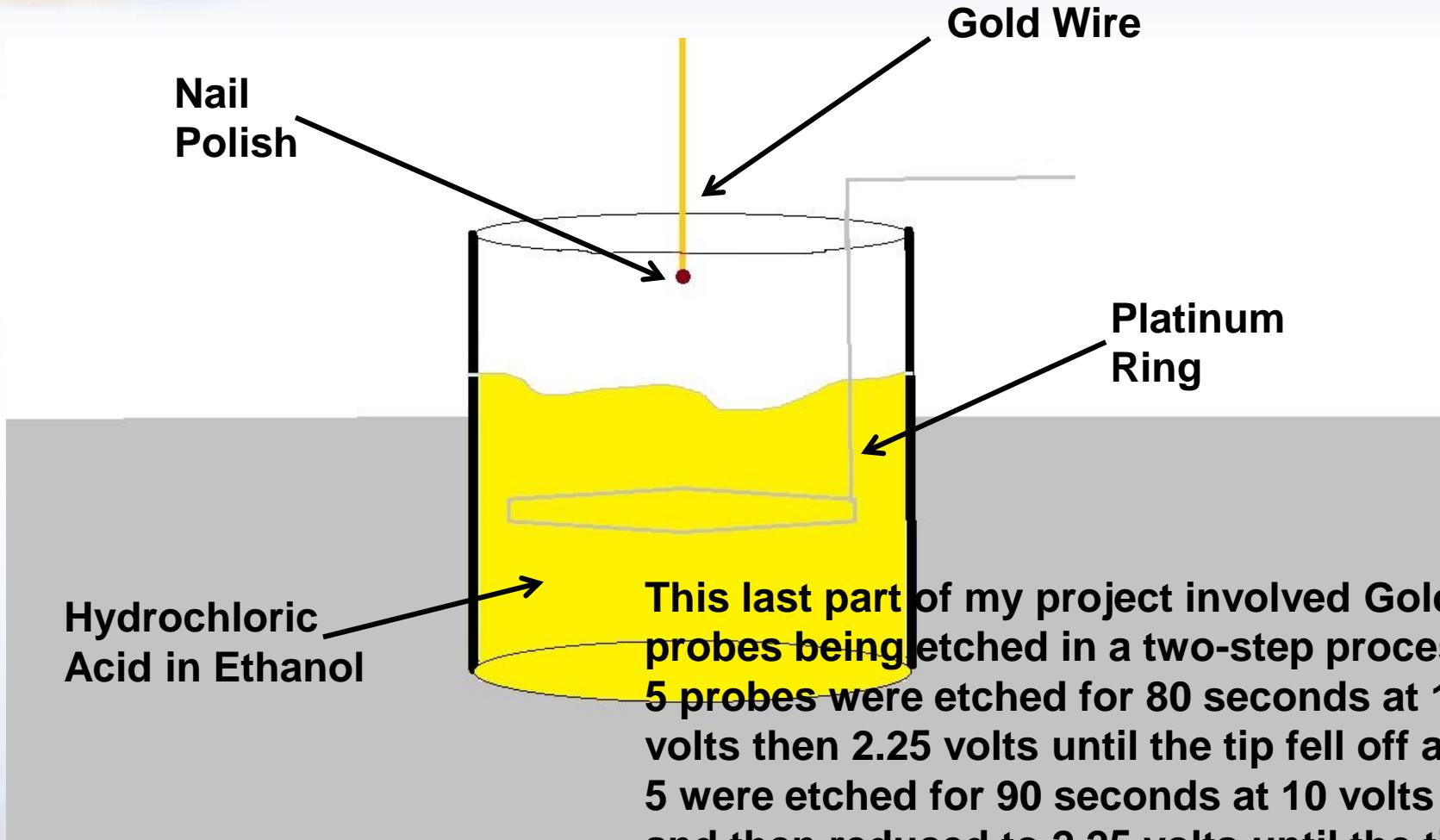
AC Voltage

DC Voltage



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# Electrochemical Etching of Gold Probes

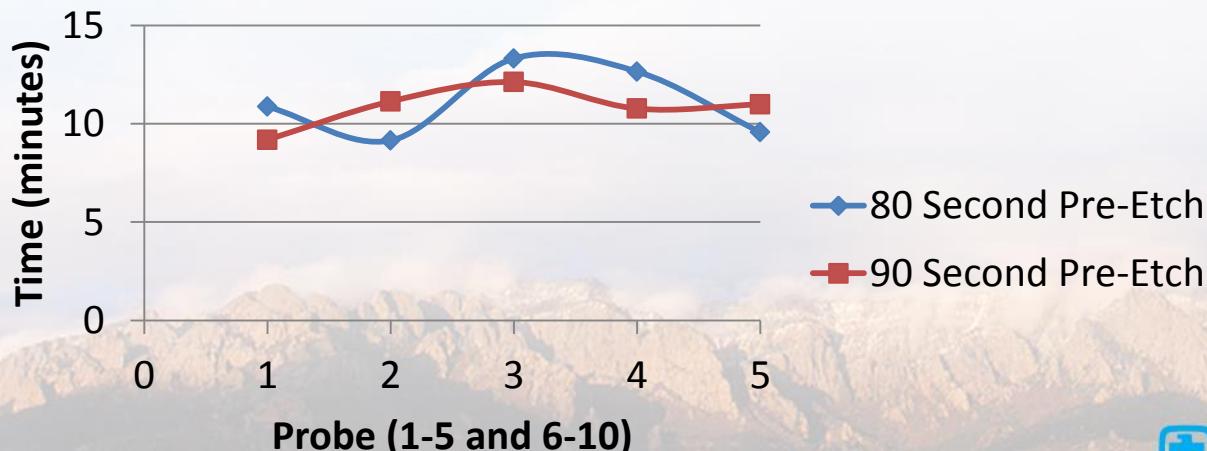


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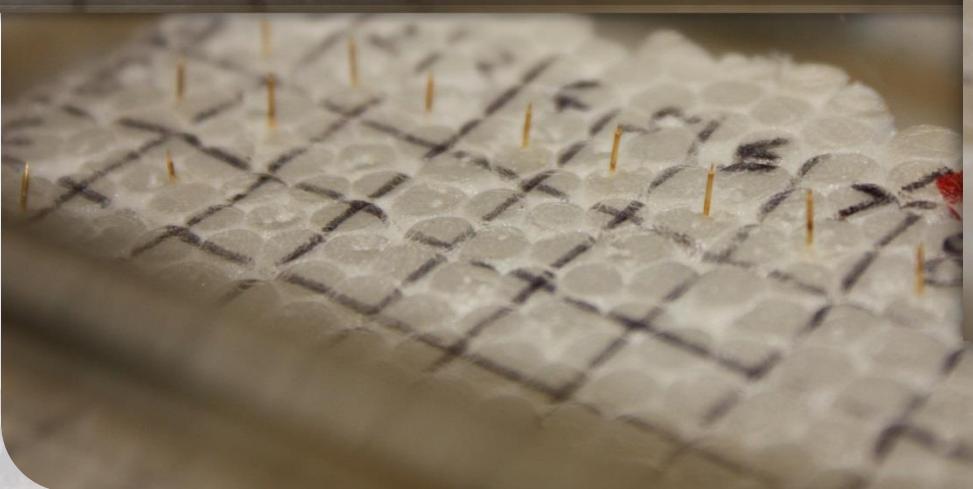
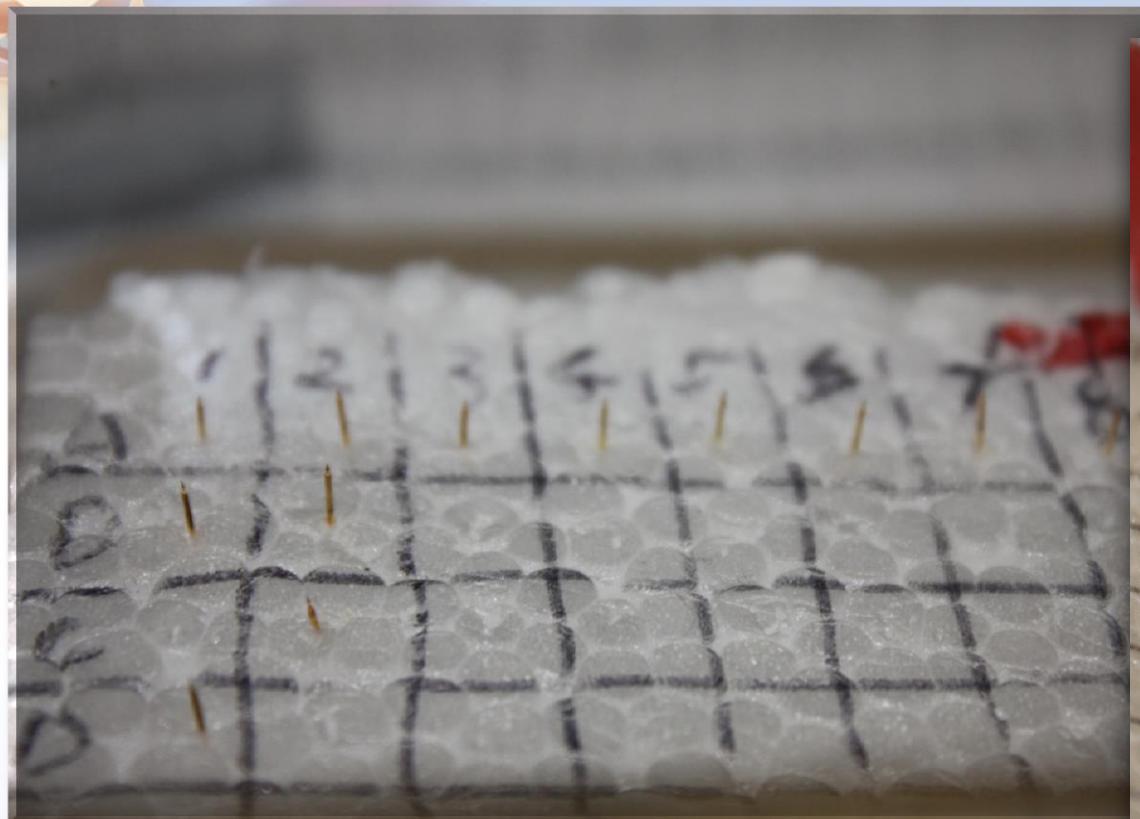
# Gold Data

Probes Pre-Etched for 80 seconds on 10 V	Probes Pre-Etched for 90 seconds on 10 V
10 min 52 sec	9 min 11 sec
9 min 9 sec	11 min 8 sec
13 min 19 sec	12 min 7 sec
12 min 39 sec	10 min 46 sec
9 min 34 sec	10 min 59 sec

**Comparison of 80 second and 90 second pre-etch**



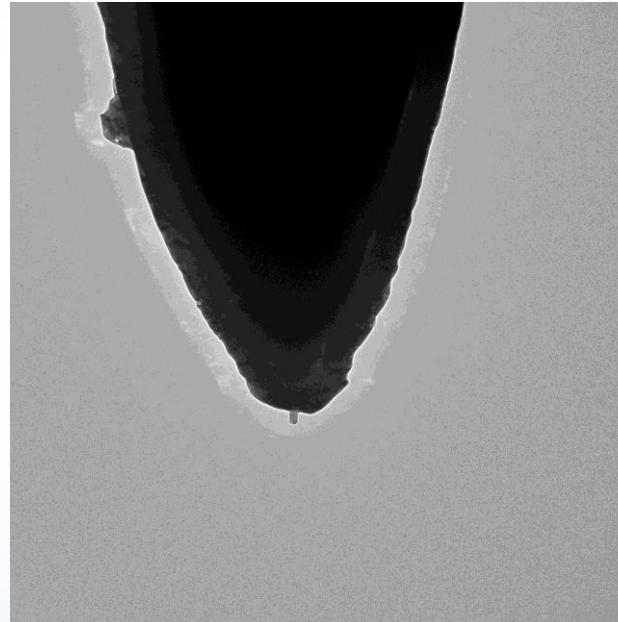
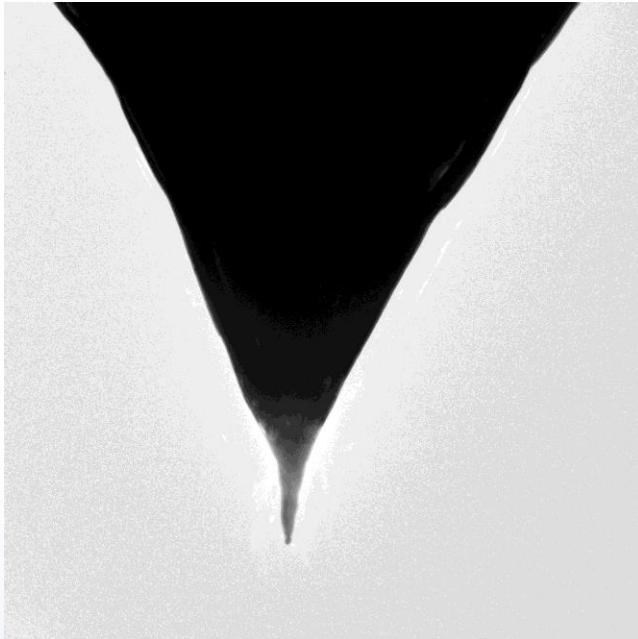
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# Finished Gold Probes



**Gold Probes Etched Using  
the Same Conditions**



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