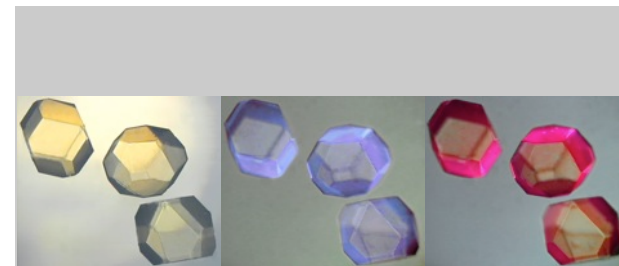
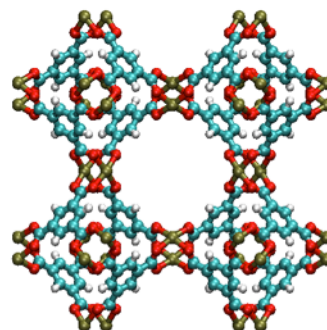
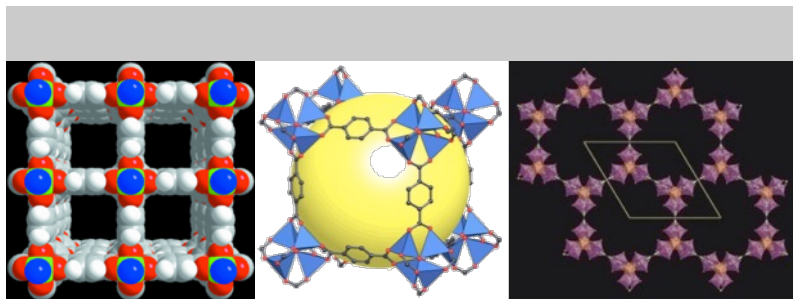


*Exceptional service in the national interest*



# Interactions of Light and Charge with Nanoporous Metal-Organic Frameworks

*IEEE SFBA Nanotechnology Council Chapter*  
July 22, 2015

**Mark Allendorf, Senior Scientist**

**mdallen@sandia.gov**

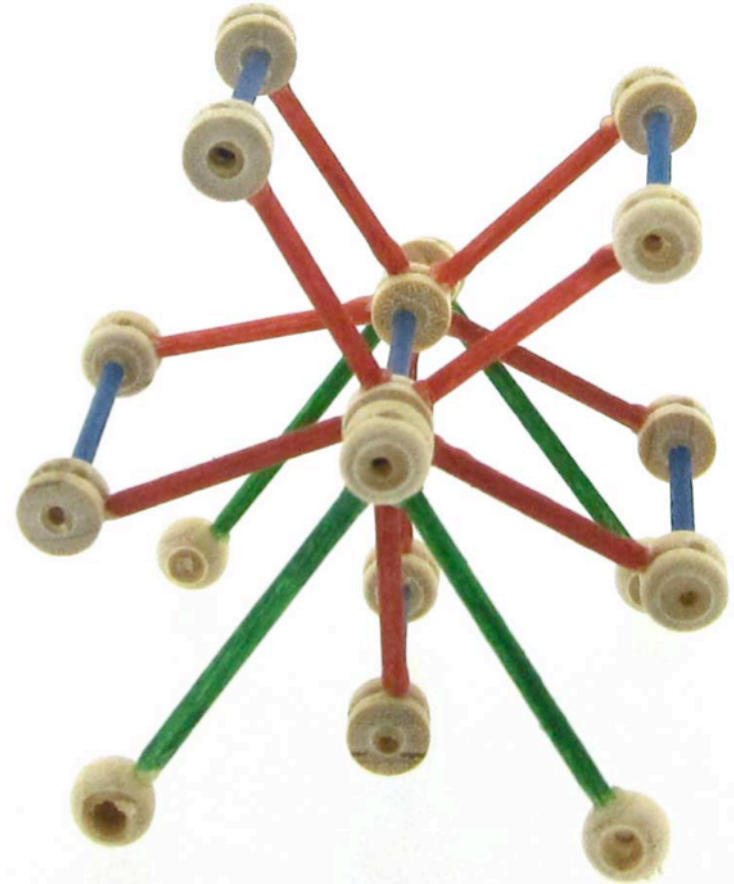
**Transportation Energy Center**

**Sandia National Laboratories, Livermore, CA**



Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

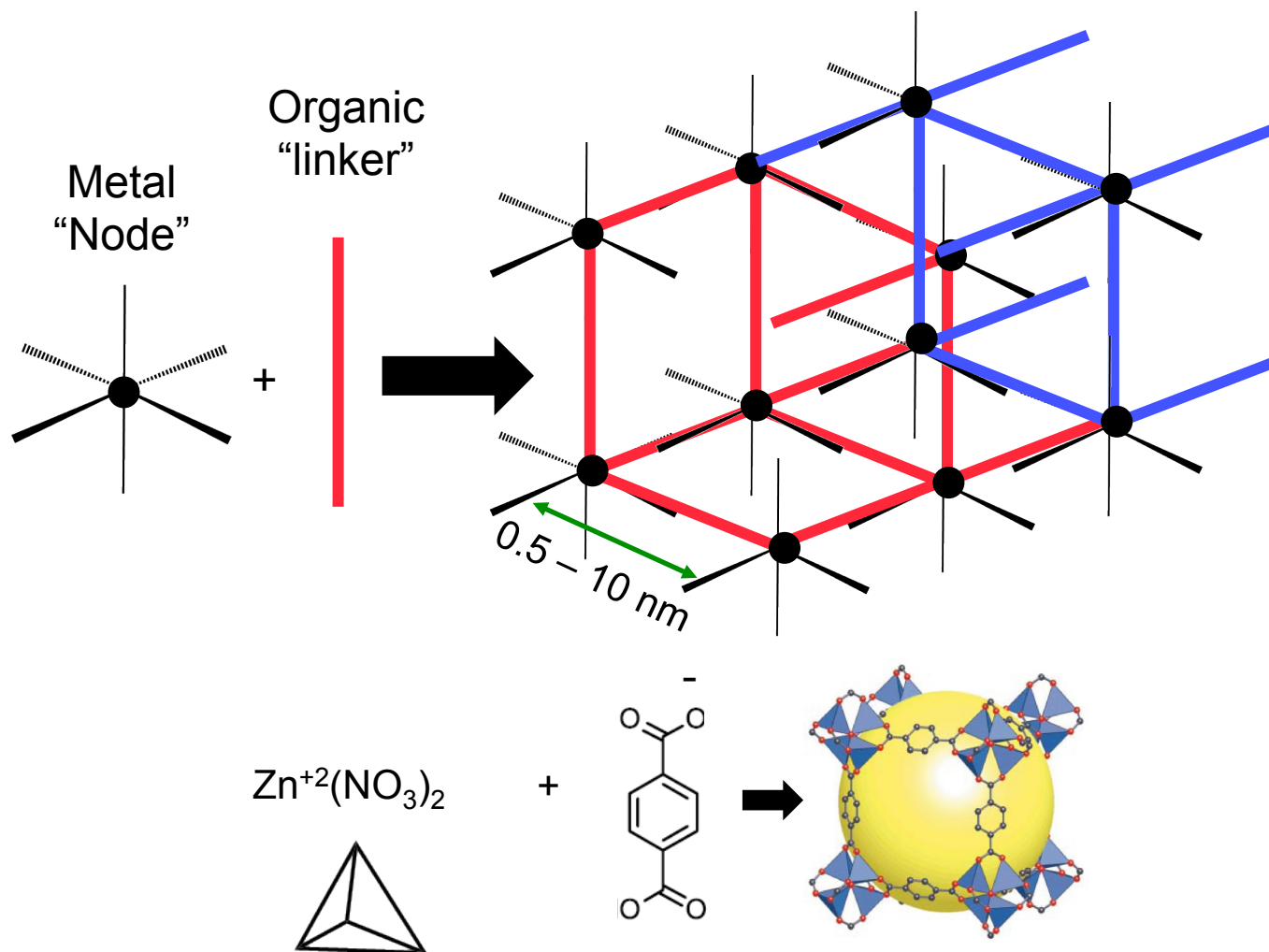
# Remember these?



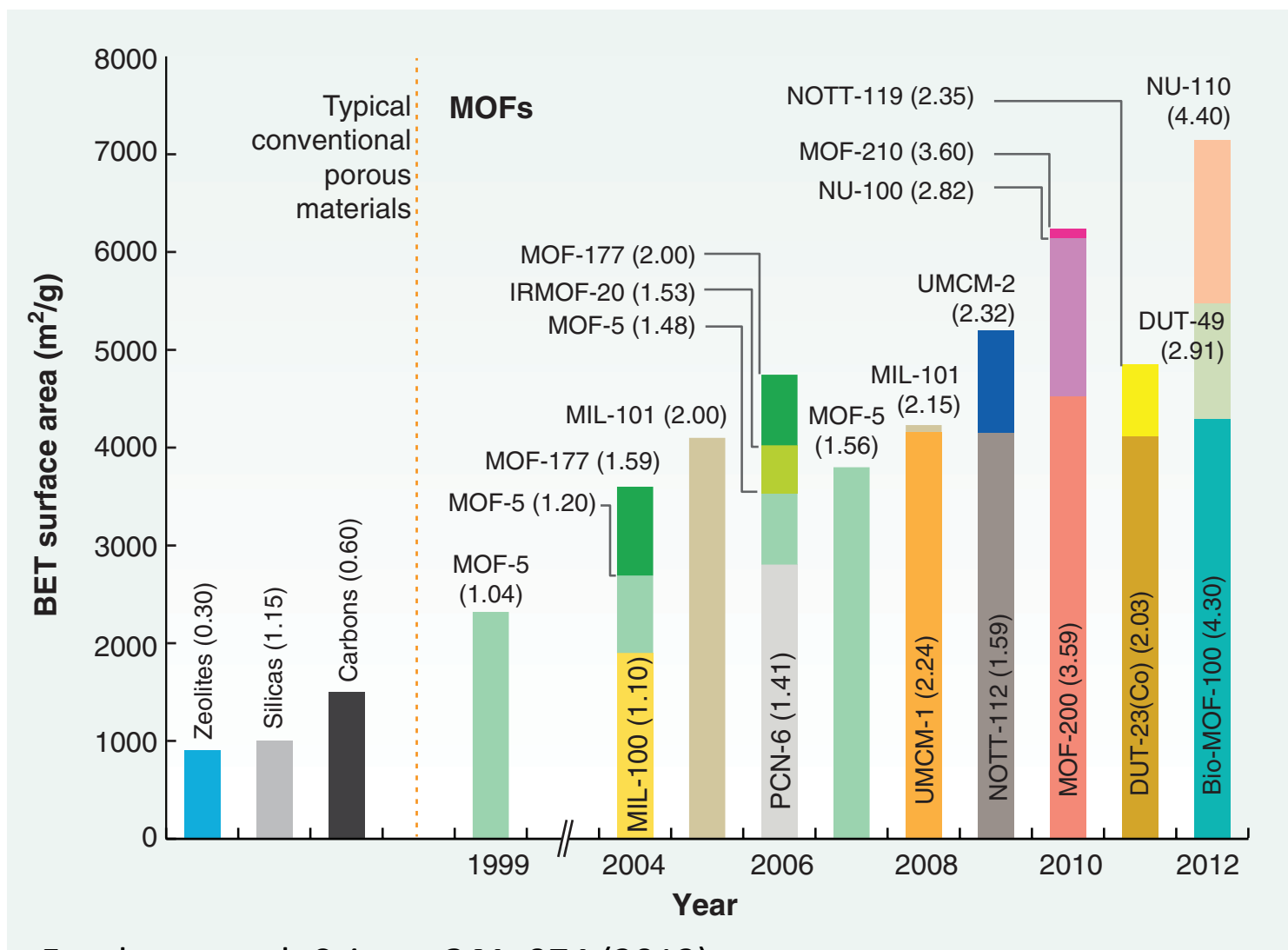
**Knowing structure is POWER because it correlates with function**

# What is a Metal-Organic Framework?

*Crystalline (therefore ordered), self-assembled, nanoporous structure*



# Record surface areas



Furukawa et al. *Science* **341**, 974 (2013)

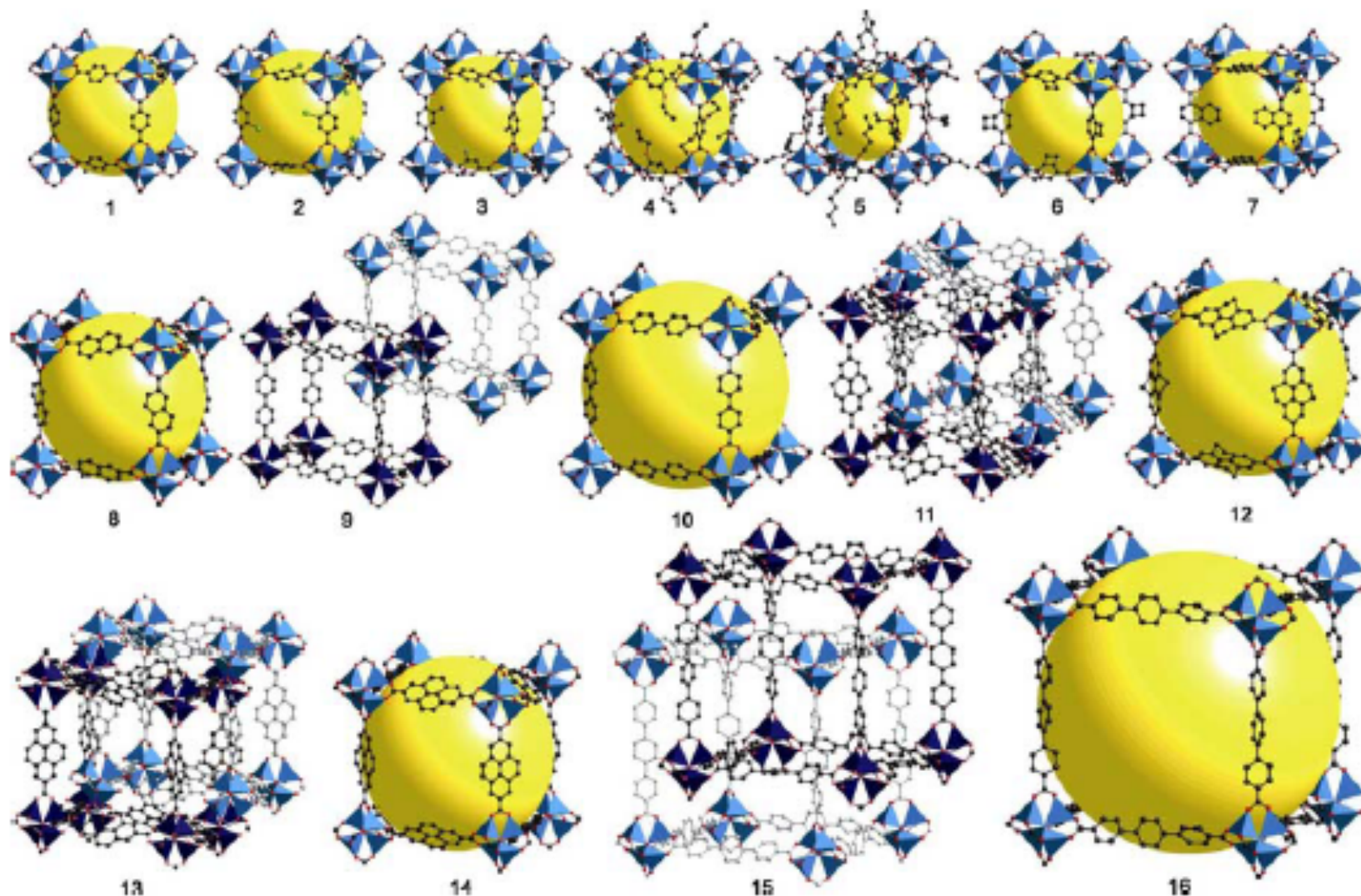


**1 football field = 5350 m<sup>2</sup>**



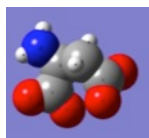


# MOFs are self-assembled, nanoporous materials with tunable pore size and properties

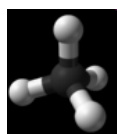


# Guest molecule + MOF → ordered, tunable platform for controlling interactions at the nanoscale

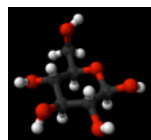
Amino acids



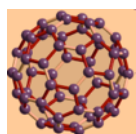
$C_3H_8$   
4.3 Å



Glucose  
(~ 9 Å)



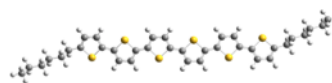
$C_{60}$   
(~ 10 Å)



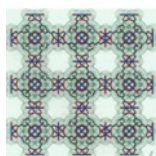
DNA (~ 20 Å)



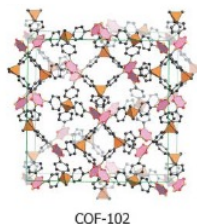
Thiophene oligomers  
(up to 37 Å)



HKUST-1

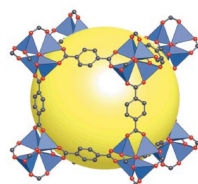


COF-102



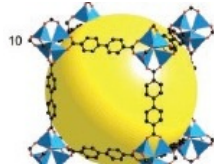
1.0 nm

IRMOF-1

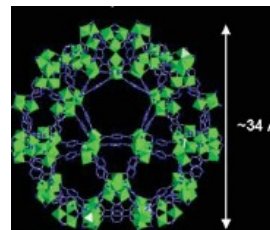


2.0

IRMOF-10

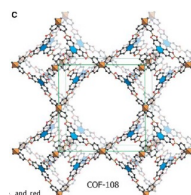


MIL-101

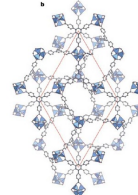


3.0 nm

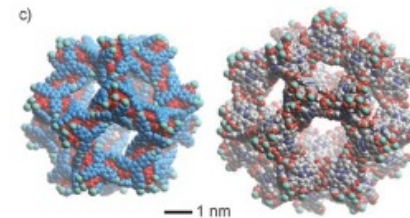
COF-108



MOF-177



$\{Tb_{16}(TATB)_{16}\}$



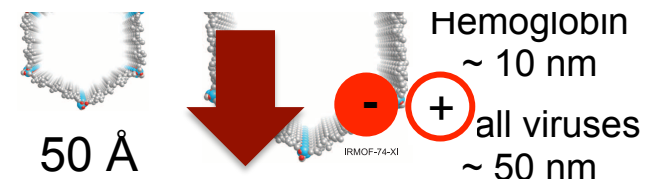
4.0 nm

5.0 nm

Interior pore diameter

MOF-74-XI

Exciton Diffusion Distance



50 Å

10.0 nm

# Device applications: the new frontier for MOFs

## ■ Optical

- Luminescent
- Non-linear optical

## ■ Electronic and magnetic

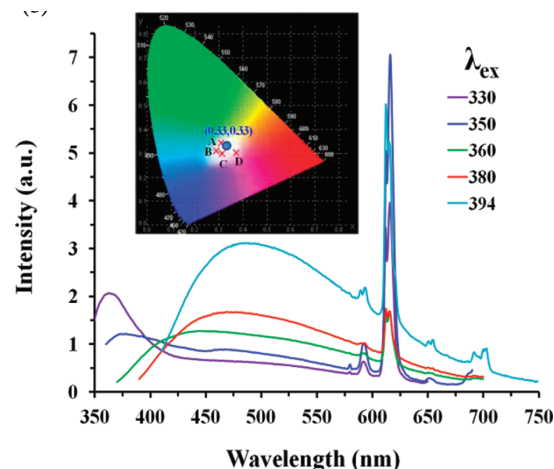
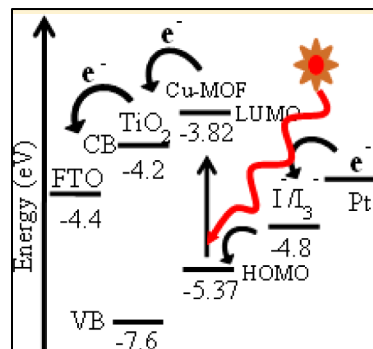
- Dielectric
- Electrically conducting
- Ion conducting
- Ferroelectric
- Ferromagnetic
- Antiferromagnetic
- Spin crossover
- Ion conducting

## ■ Mechanical flexibility

### Photovoltaics

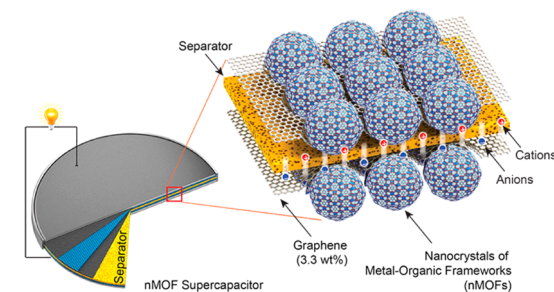
D. Y. Lee et al.

*J. Phys. Chem. C* **118**, 16328



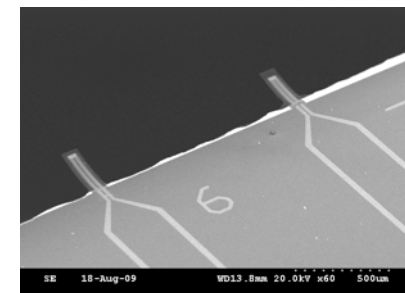
### Solid-state lighting

D.F. Sava et al. *JACS* 2012, 134, 3983



### Supercapacitors

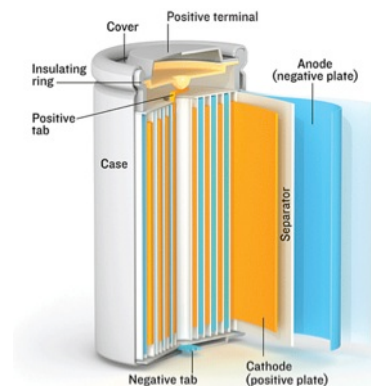
K. M. Choi et al. *ACSNano* 2014



### Sensors

Allendorf et al.

*JACS* **2008**, 130, 14404



### Batteries and Fuel Cells

G. Shimizu et al. *Chem. Soc. Rev.* 2014,**43**, 5913

# Our work

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- Thin films
- Energy transfer/photovoltaics
- Radiation detection
- Chemical sensing
- Electrically conducting MOFs

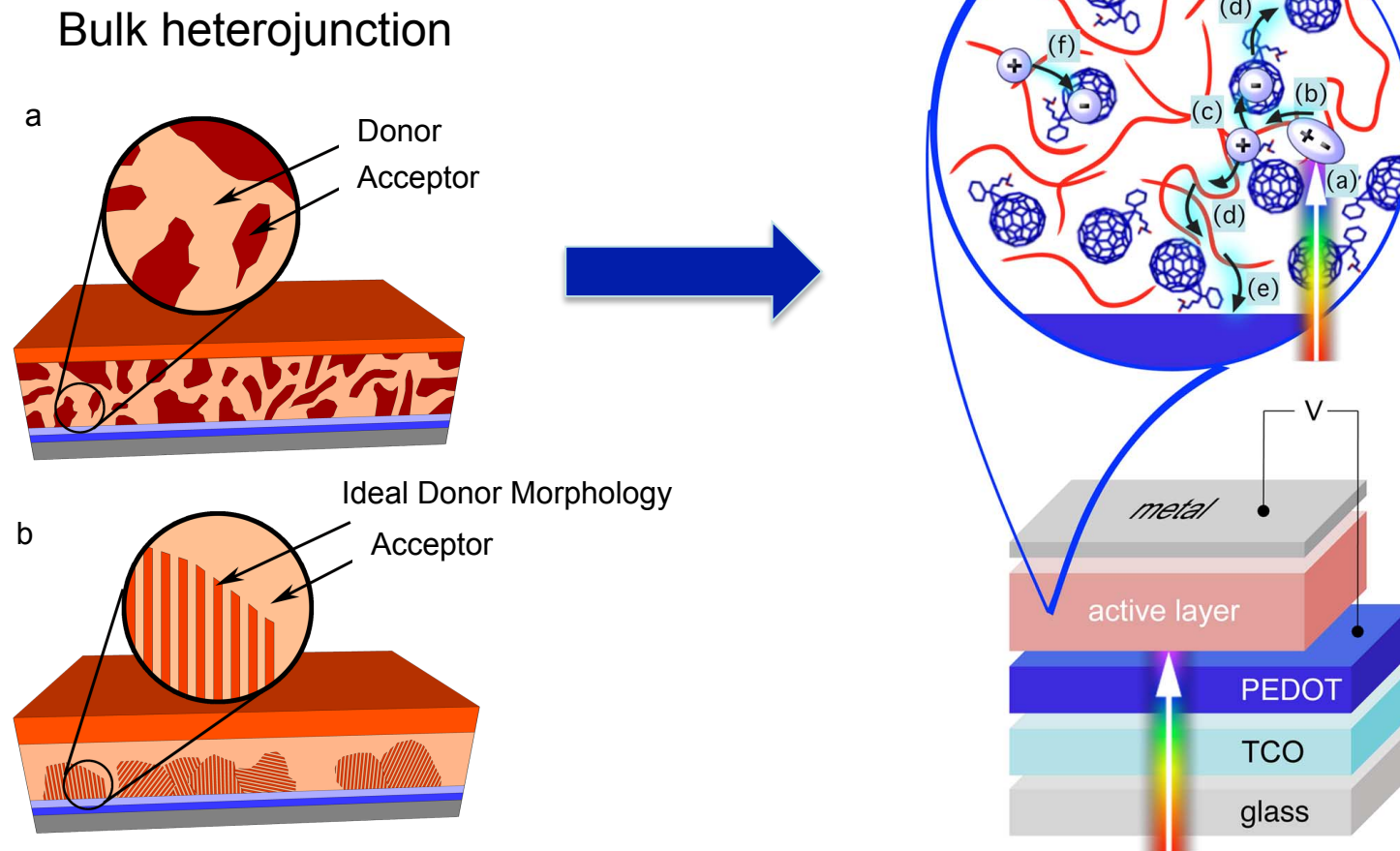
**Today's presentation**

- Nanoreactors/catalysis
- Hydrogen storage
- Nanoparticle and nanowire templates

**Other areas of  
research**



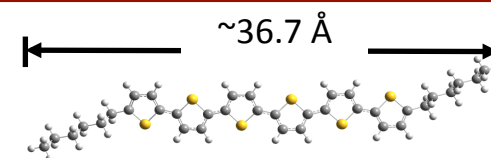
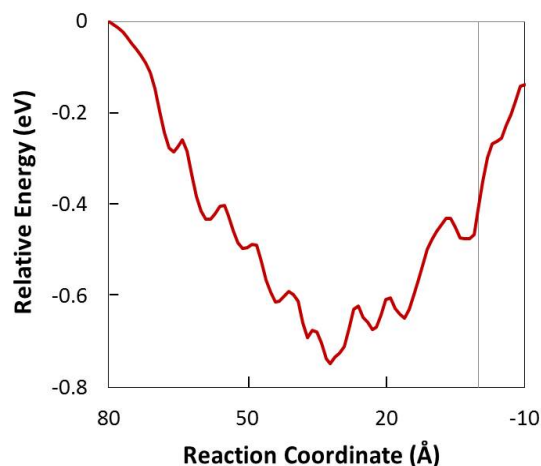
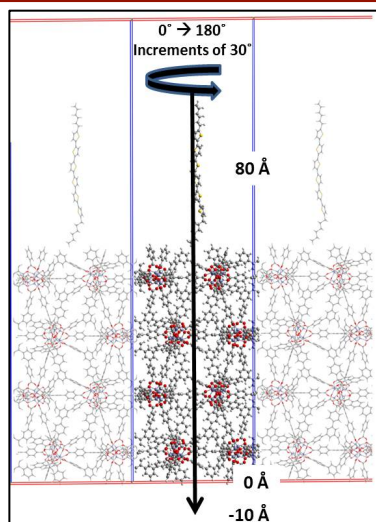
# Disorder is the enemy of efficiency



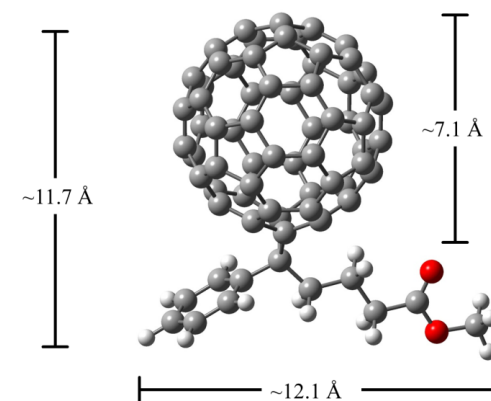
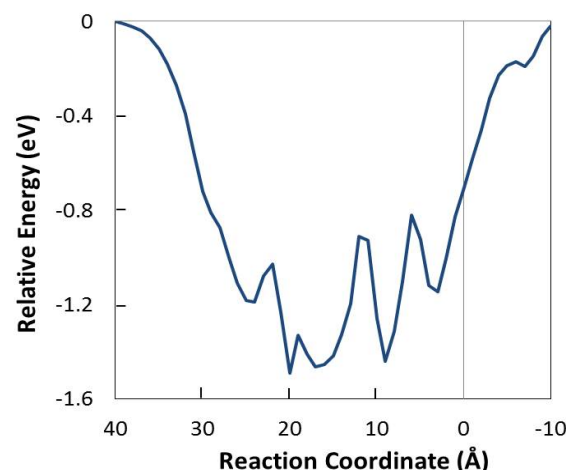
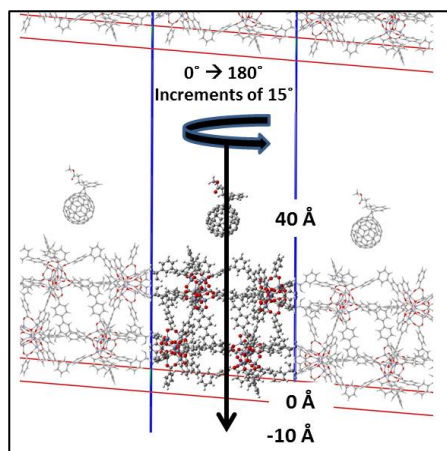
C. Deibel et al. IEEE Journal Of Selected Topics In Quantum Electronics,  
Vol. 16, No. 6, November/December 2010



# Infiltrating MOF-177 pores with thiophenes and PCBM proceeds without an energy barrier



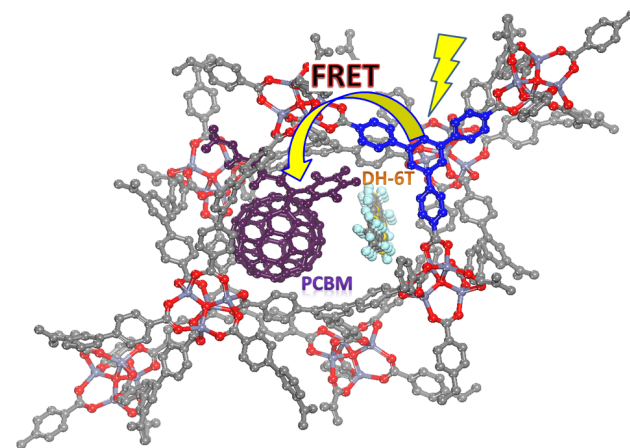
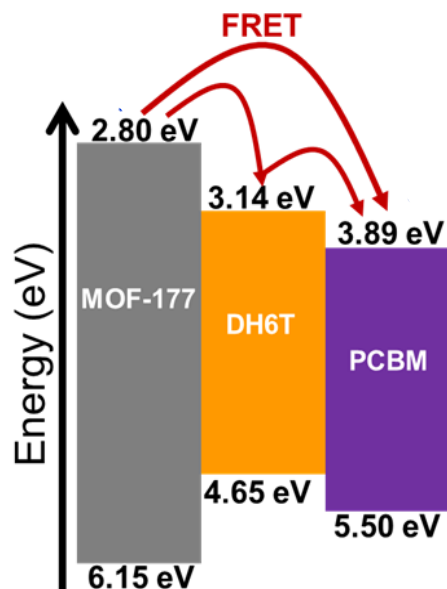
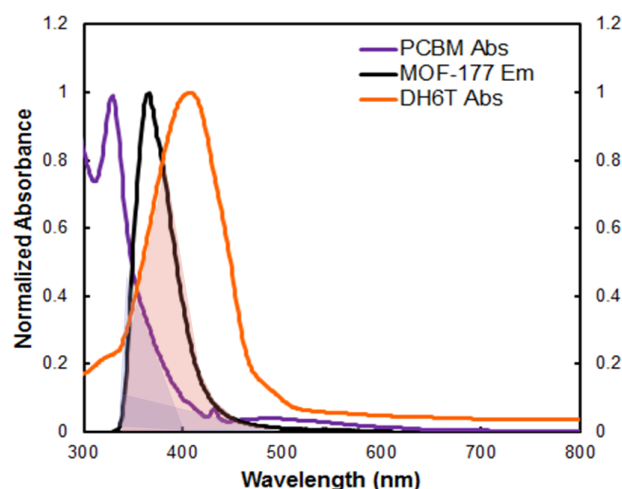
**Sexithiophene  
Electron Donor**  
17 kcal/mol B.E.



35 kcal/mol B. E.  
**PCBM**  
**Electron Acceptor**

# MOF-guest spectral overlap → Fluorescence Resonant Energy Transfer (FRET)

*Efficient energy transfer via “FRET cascade” is feasible*



The MOF has three functions:

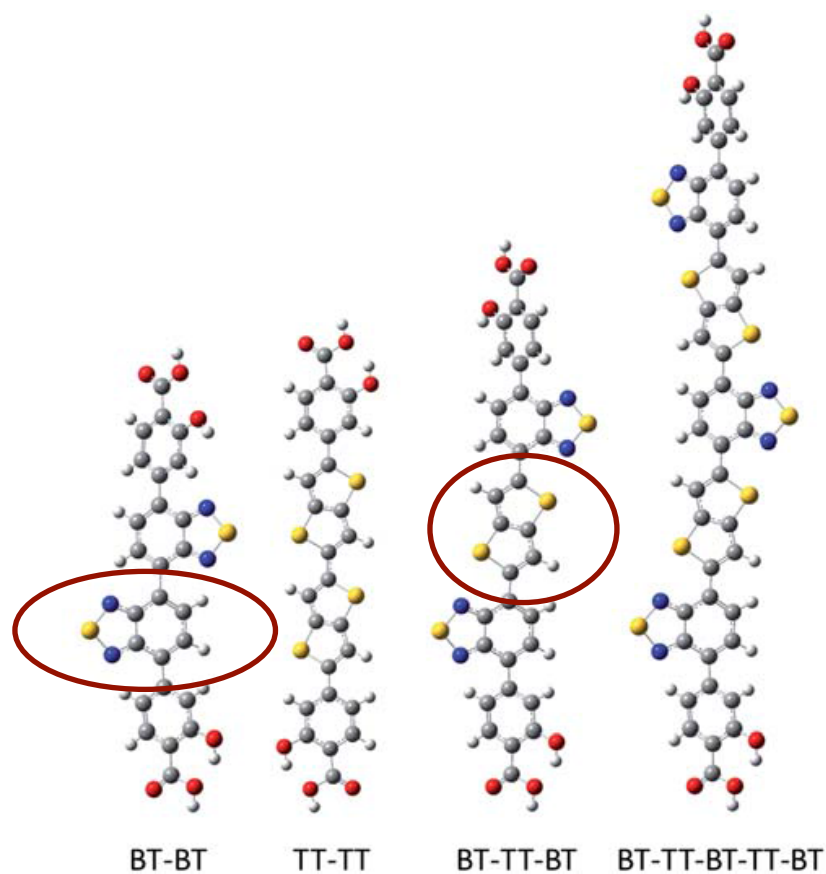
- Confines/stabilizes donor and acceptor
- Prevents phase segregation
- Serve as a photon antenna

K. Leong et al. *J. Mater. Chem. A*, **2** (2014), 3389

# New Donor-Acceptor IRMOF-74 linkers for improved solar coverage

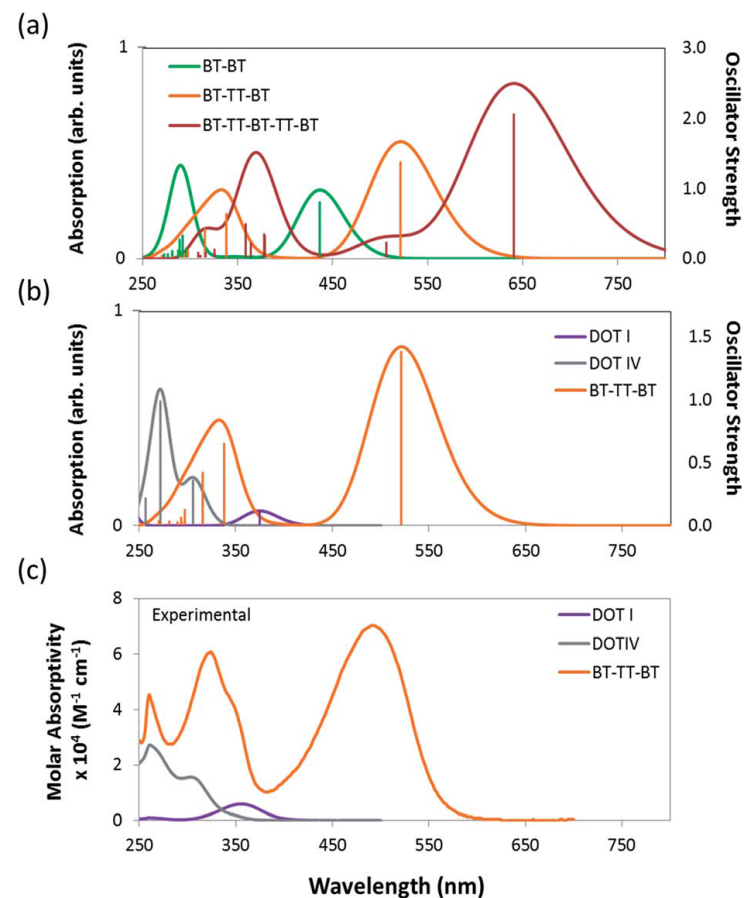
Electron deficient: benzo[c][1,2,5]thiadiazole (BT)

Electron rich: thieno[3,2-b]thiophene (TT)



Foster, Allendorf et al. *Chem. Sci.*, 2014, **5**, 2081

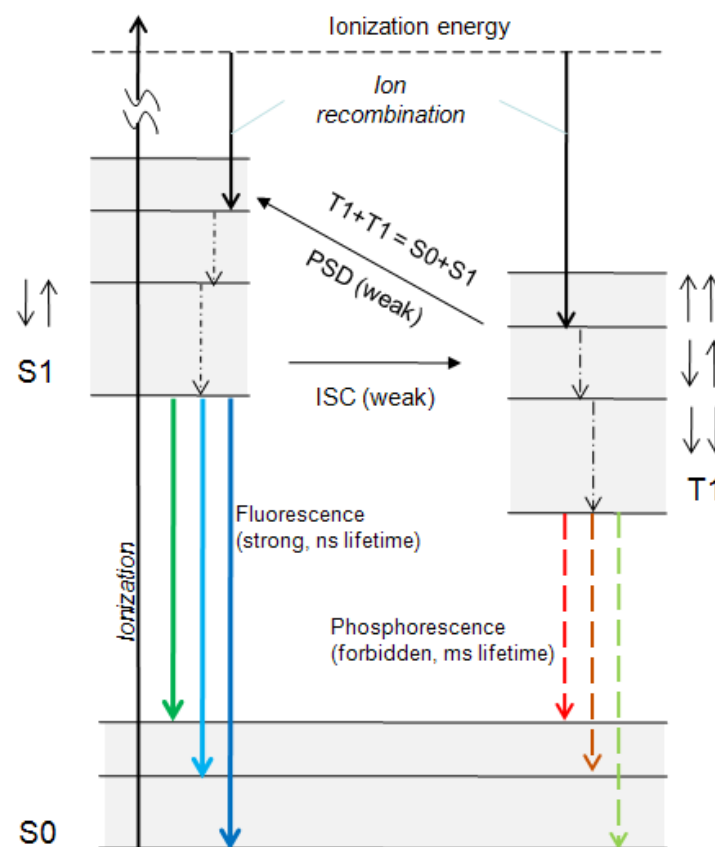
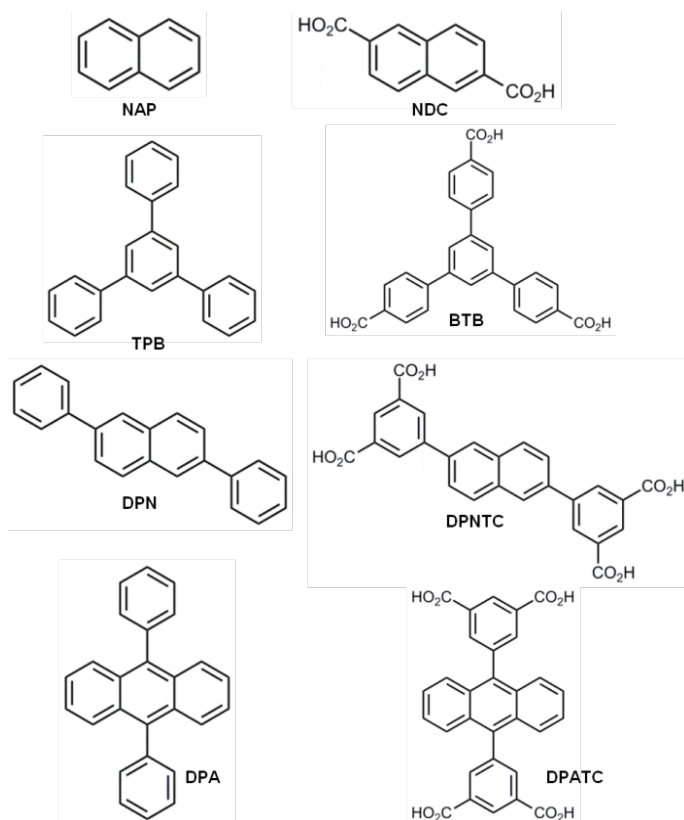
(a,b) Long-range corrected DFT\*



\*Non-empirically tuned functional/  
time-dependent DFT

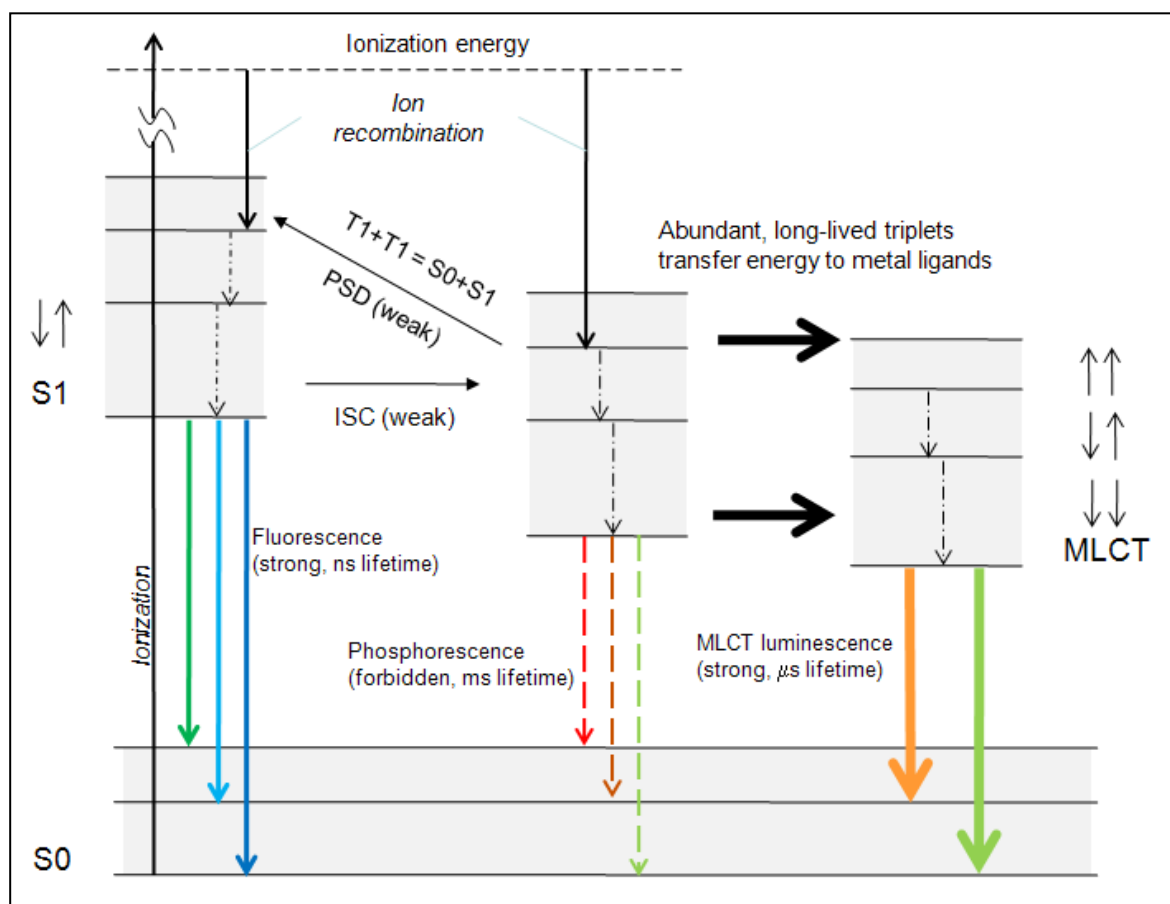
# Basics of radiation detection using organic scintillators

## Typical luminescent MOF linkers



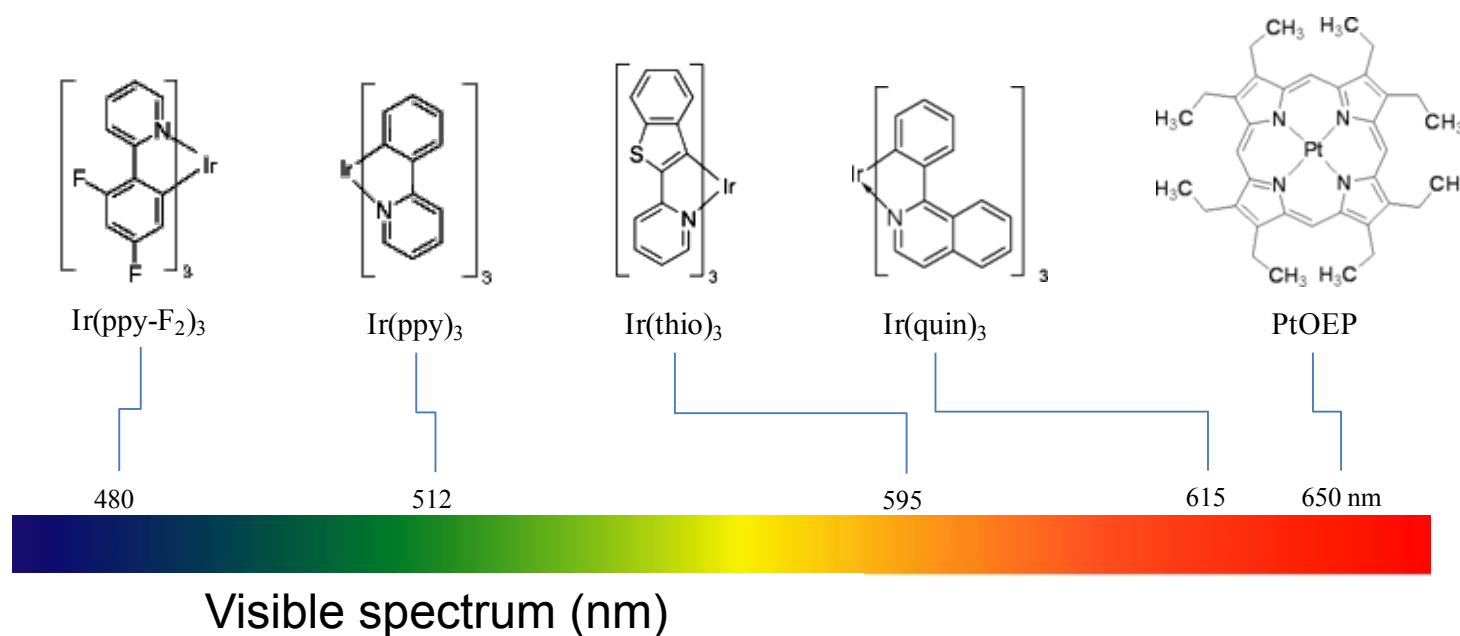
# Guest molecules in MOFs create new approach to radiation detection

Ionization creates singlet and triplet excitons, but typically most triplets decay by non-radiative processes



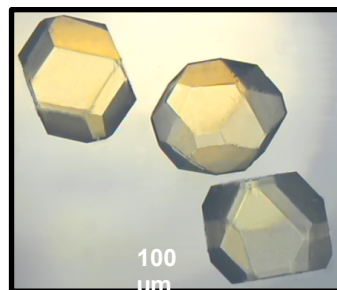
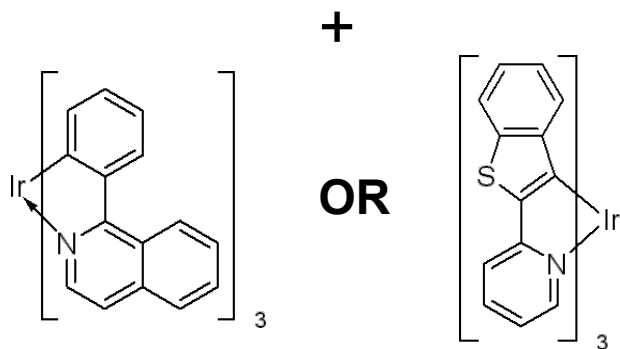
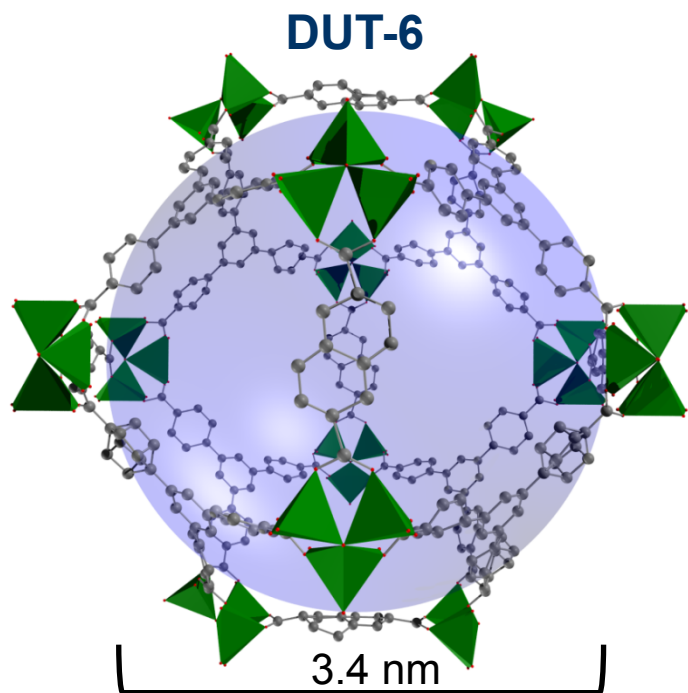
**~300% increase in  
light yield  
theoretically  
possible**

# Triplet harvesting heavy-metal complexes are well known in OLED technology

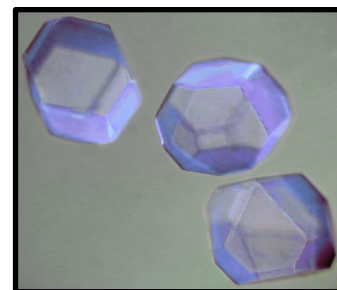




# Extrinsic (Triplet) luminescence: Ir organometallic guest molecules



Bright-Field (Vis)

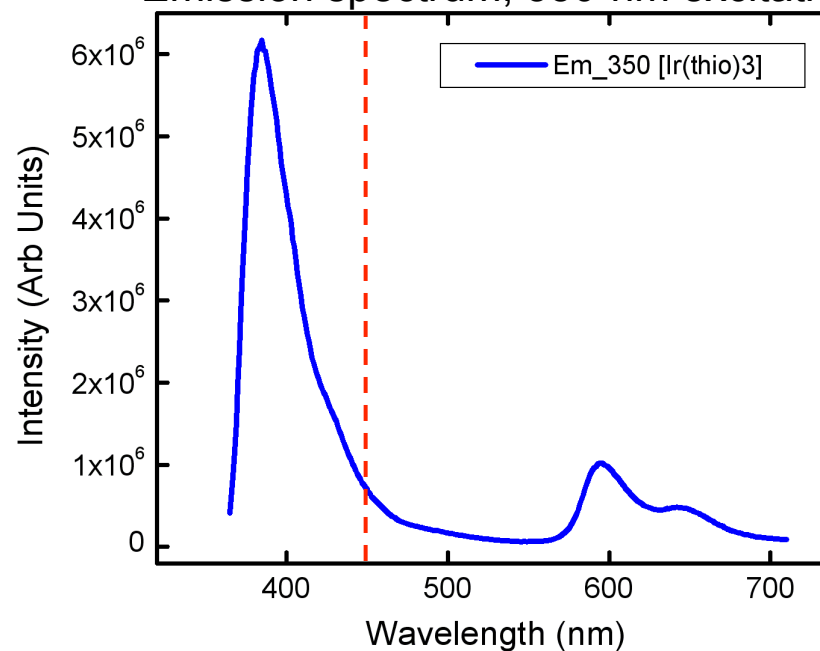


450 nm short-pass

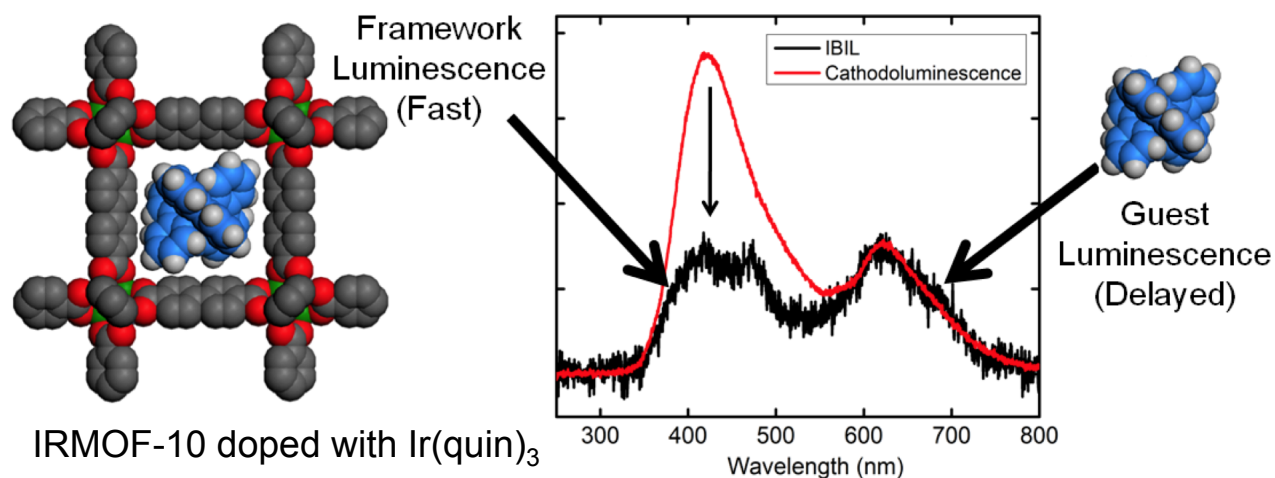


450 nm long-pass

Emission spectrum, 350-nm excitation

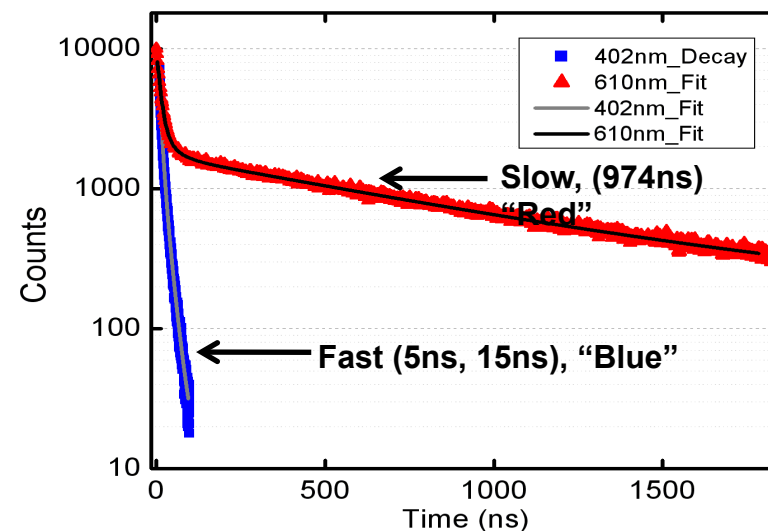


# Radiation detection by spectral shape discrimination: IRMOF-10 doped with Ir(quin)<sub>3</sub>



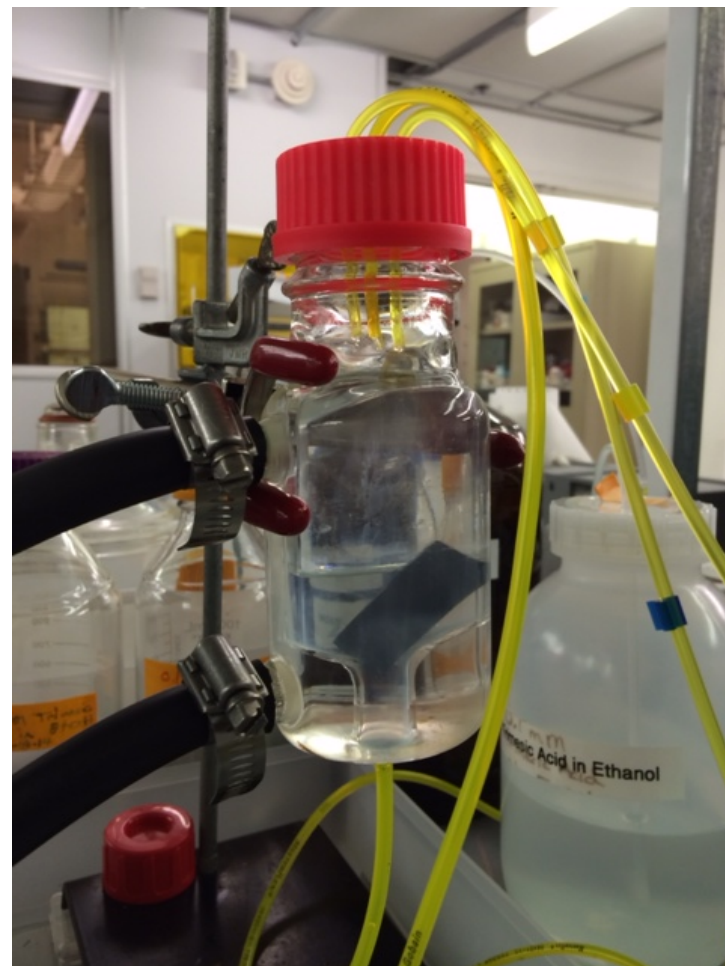
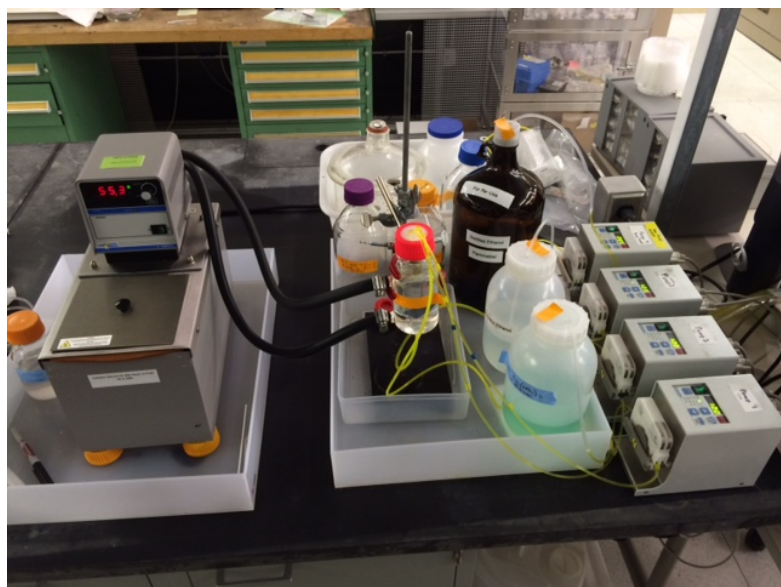
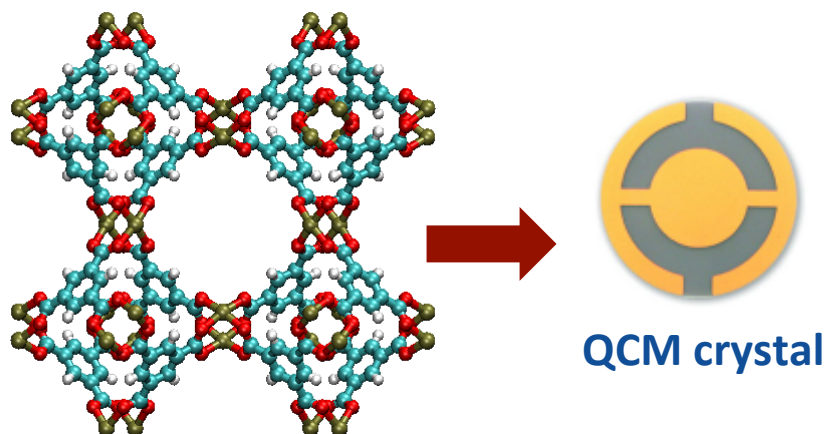
$$\begin{aligned} [I_{614}/I_{410}]: \text{PL} &= 0.11 \\ [I_{614}/I_{410}]: \text{CL} &= 0.39 \\ [I_{614}/I_{410}]: \text{IBIL} &= 1.08 \end{aligned}$$

- Sterically matched host-guest combination
- Exponential PL decay times
- Dichroic wavelength selection (SSD) vs. decay time (pulse-shape discrimination)



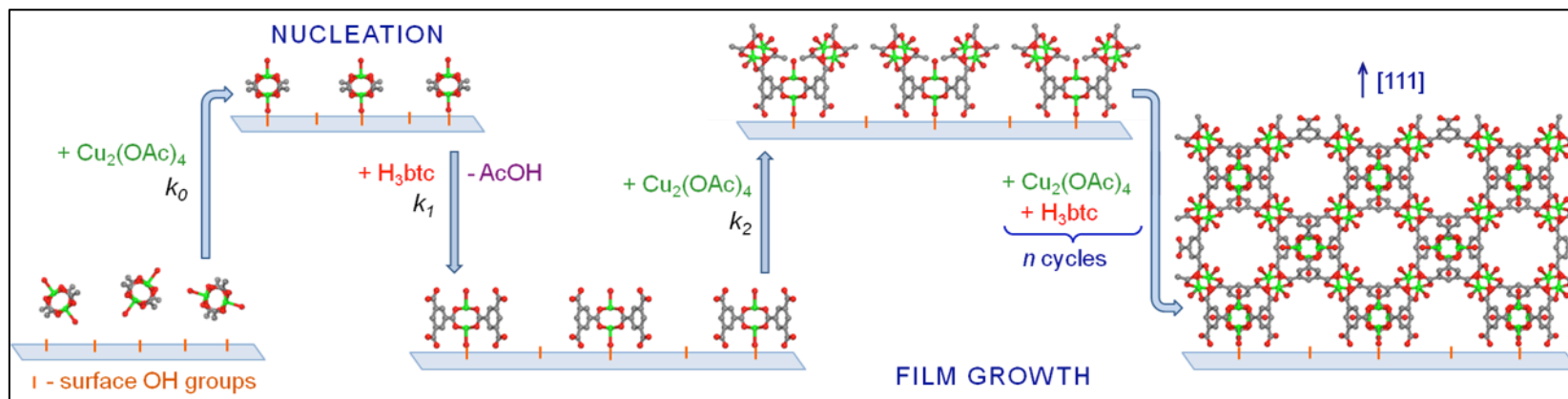
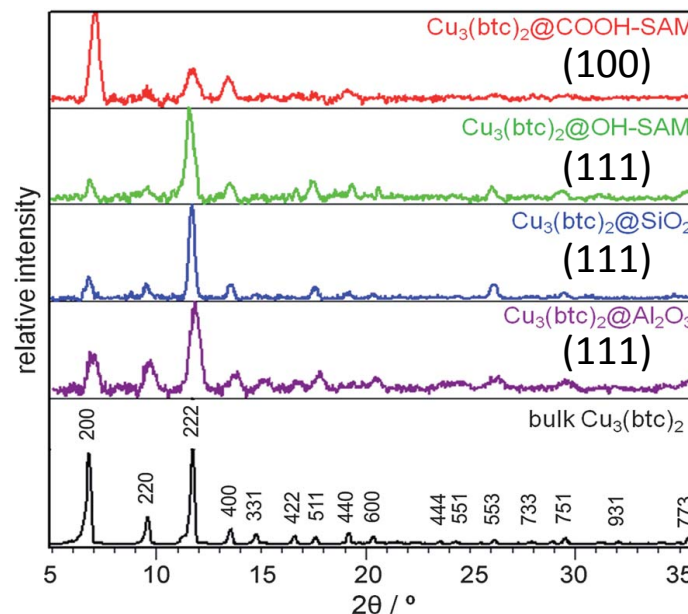
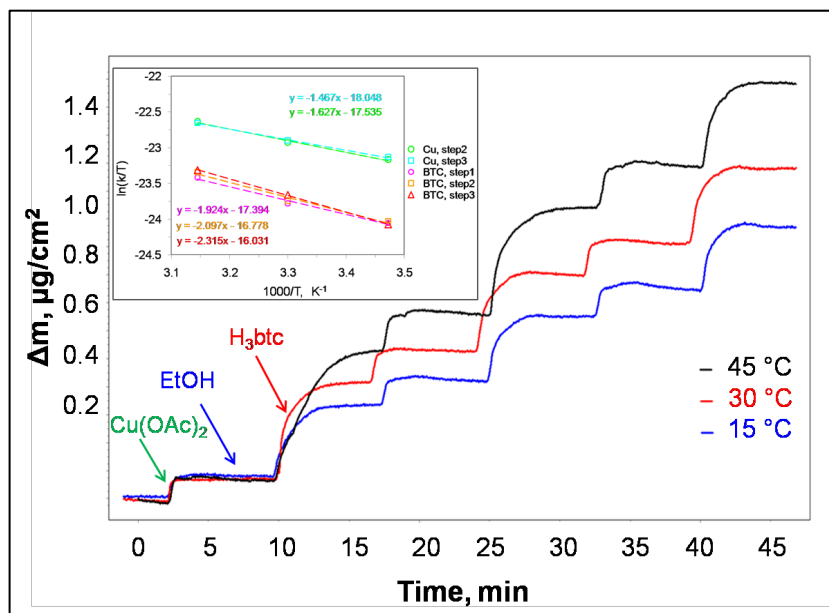
# Layer-by-layer MOF film growth: liquid-phase epitaxy Sandia National Laboratories

HKUST-1



Automated MOF film growth with QCM capability

# Layer-by-layer deposition: self-limiting growth with controllable orientation

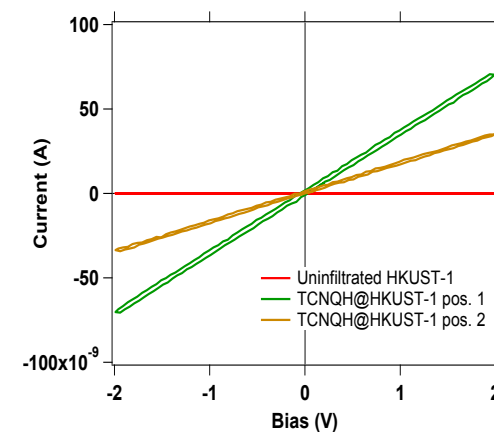
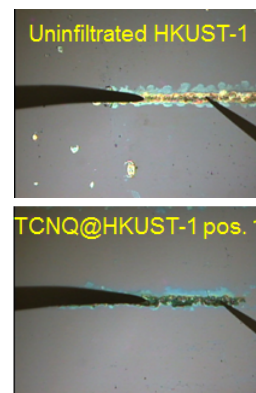
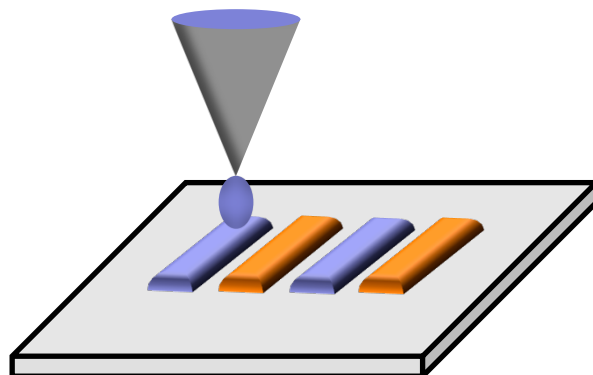
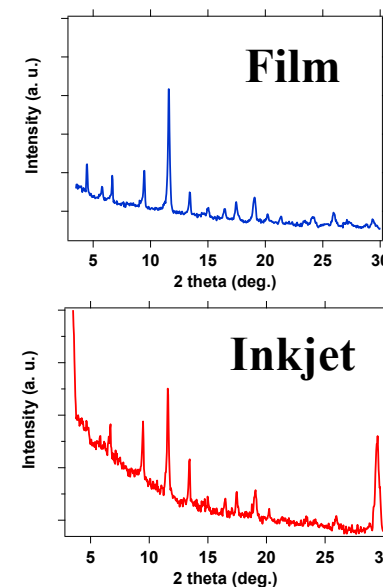
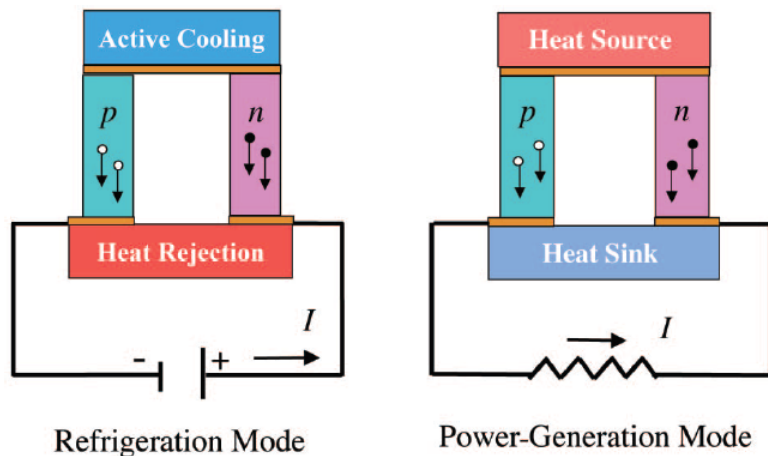


V. Stavlia et al. *Chem. Sci.* **3** (2012), 1531–1540

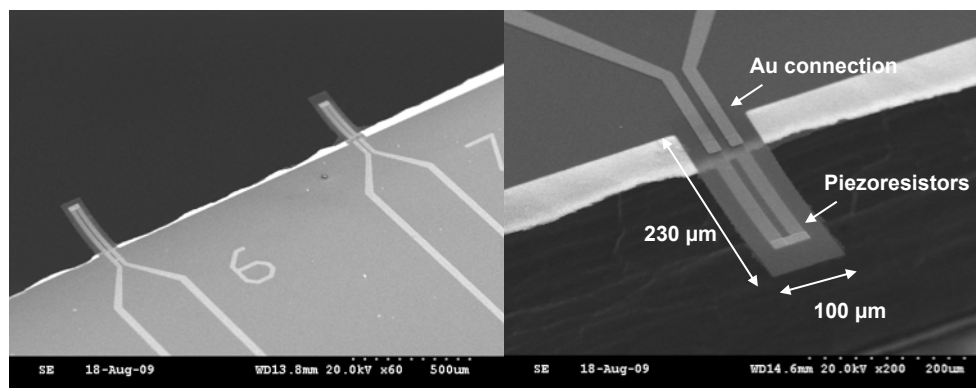


# Inkjet deposition of conducting MOF

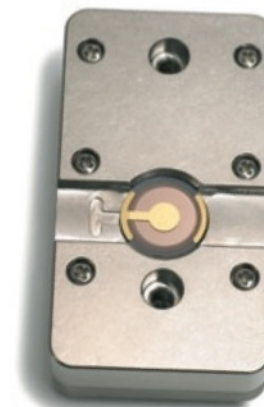
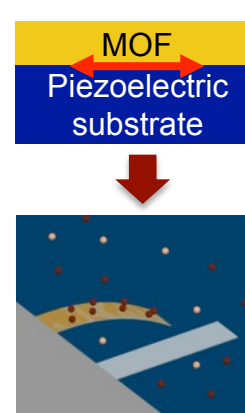
*Practical TEs require thick, n- and p-elements*



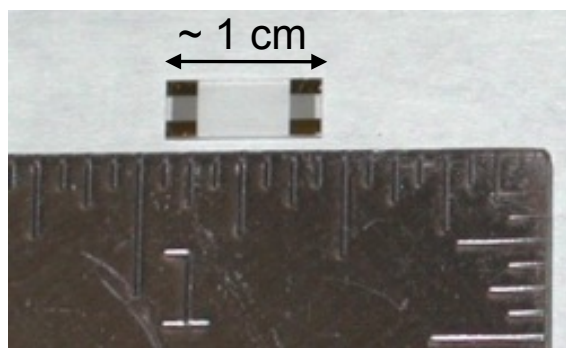
# Sensing platforms for detection by mass uptake



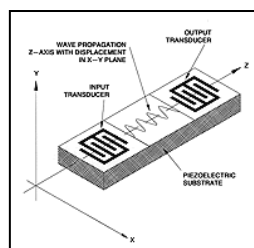
Microcantilevers (fg sensitivity)



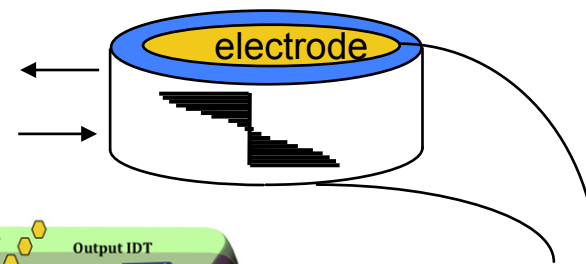
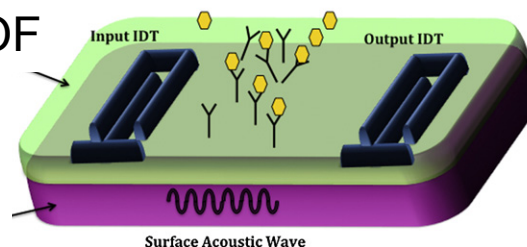
Quartz crystal microbalance (QCM) ( $\sim 1 \text{ ng/cm}^2$  sensitivity)



Surface acoustic wave (SAW) sensors ( $\sim 0.1 \text{ ng/cm}^2$  sensitivity)



MOF



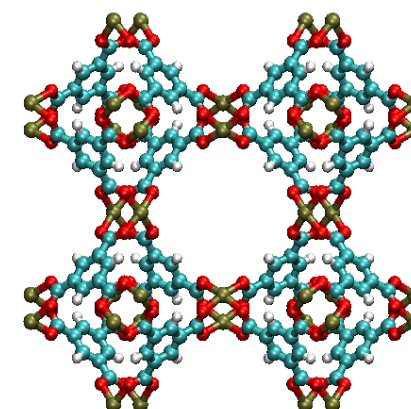
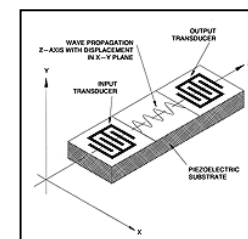
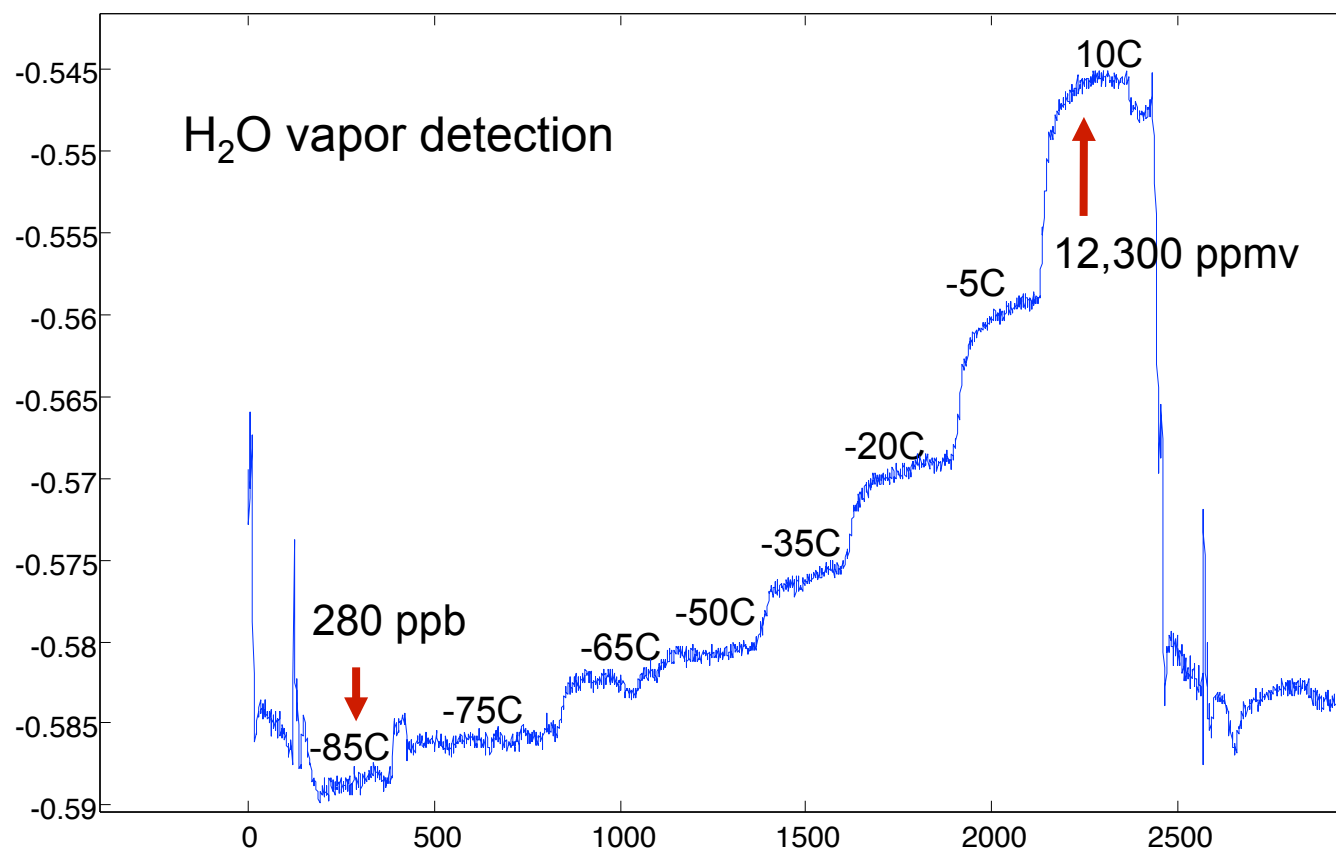
**Recognition chemistries are required to enhance sensitivity and impart selectivity to the device**



# MOF-coated MEMS devices can compete with state-of-the-art sensors

SAW sensor coated with HKUST-1 ( $\text{Cu}_3(\text{BTC})_2$ )

Sensor response > 4 orders of magnitude in  $\text{H}_2\text{O}$  concentration

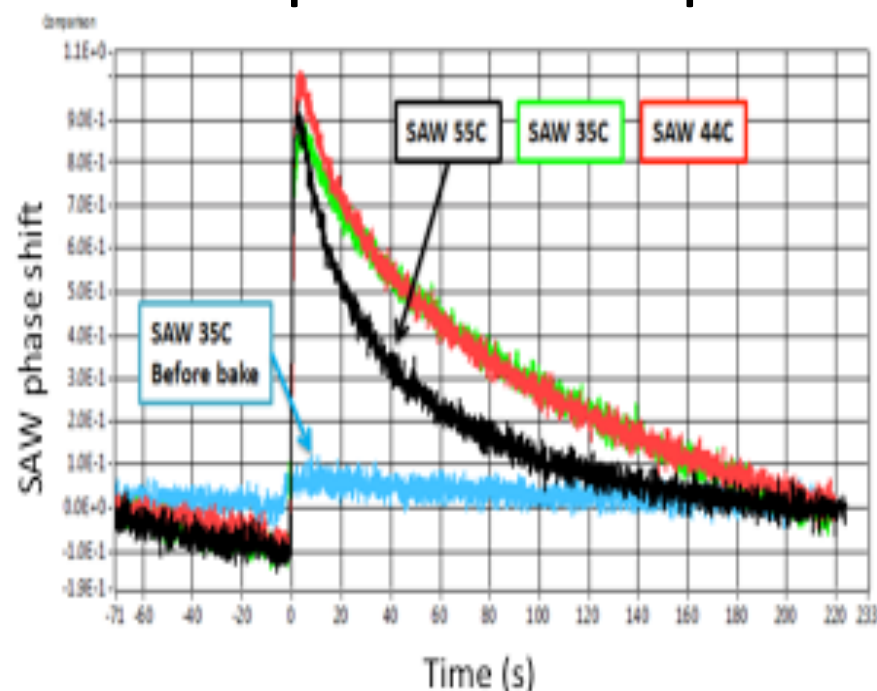


$\text{Cu}_3\text{BTC}_2$  (HKUST-1)

# Response to acetone: ZIF-8 vs. HKUST-1

## HKUST-1 (~100 nm film on SiO<sub>2</sub>)

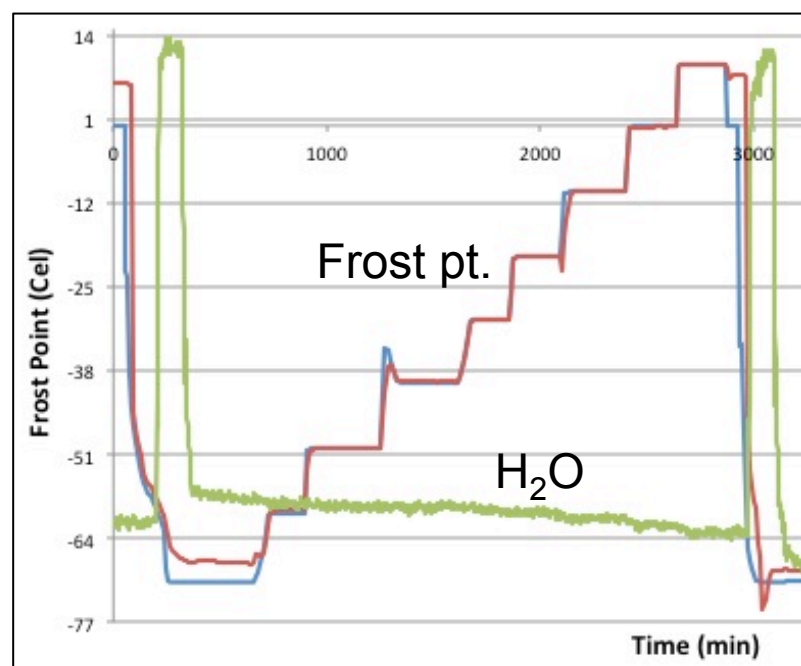
### Response to acetone pulse



- Fast response to H<sub>2</sub>O
- H<sub>2</sub>O LOD: < 1 ppm
- No response to N<sub>2</sub>, CH<sub>4</sub>, CO<sub>2</sub>

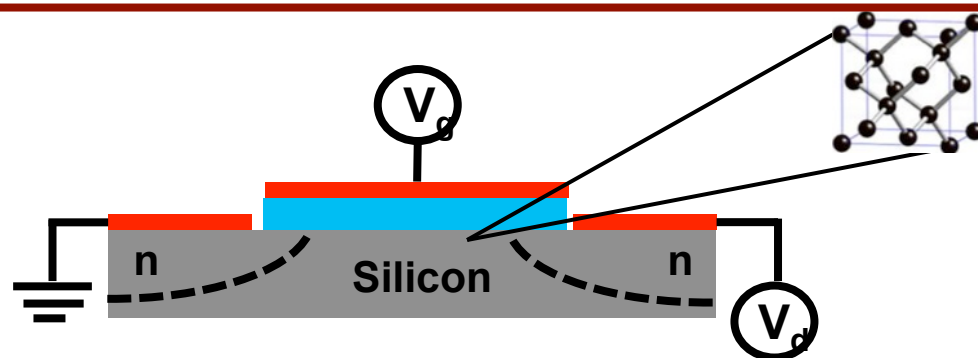
## ZIF-8 (450 nm film on SiO<sub>2</sub>)

### Response to humidity



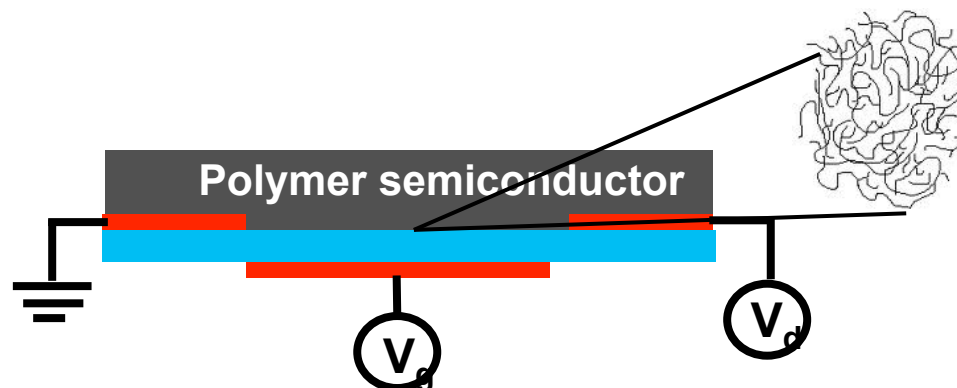
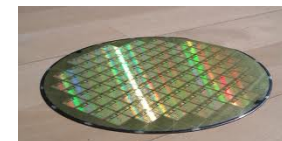
- Acetone LOD: ~ 30 ppm
- No response to water vapor (above)
- No response to N<sub>2</sub>, CH<sub>4</sub>, CO<sub>2</sub>

# MOFs as electronic materials combine features of inorganic and organic conductors



## Crystalline inorganic semiconductor

- High mobility
- Stability
- High cost
- Non-flexible
- Radiation damage

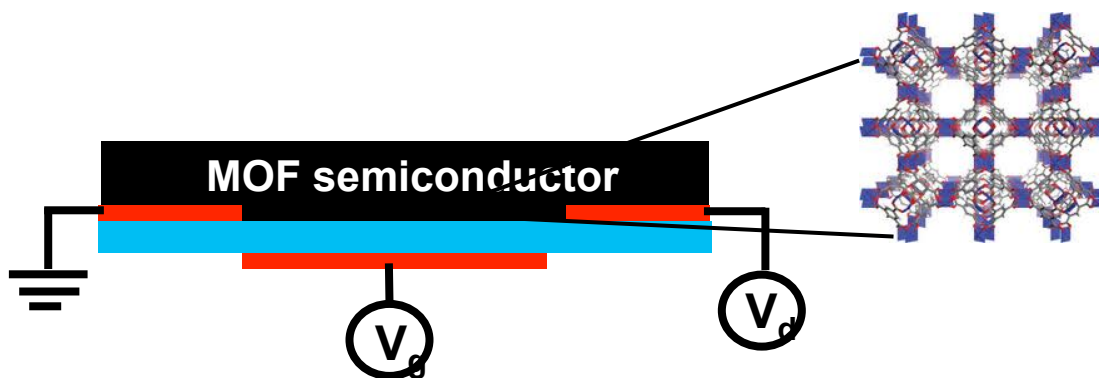


## Disordered organic semiconductor

- Flexible
- Tunable w/ chemistry
- Low cost fabrication
- Poor mobility
- Instability
- Low free carrier densities



+

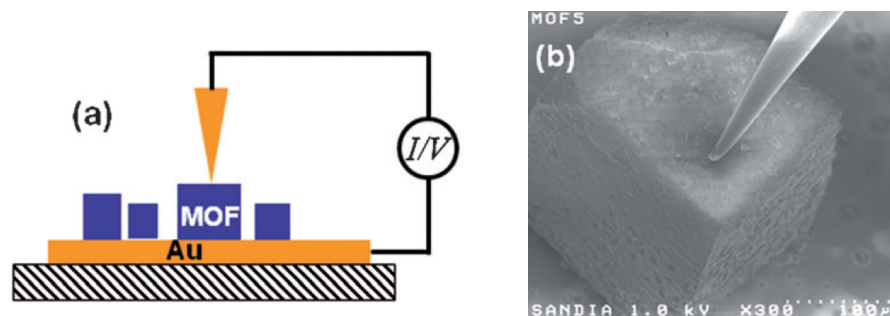
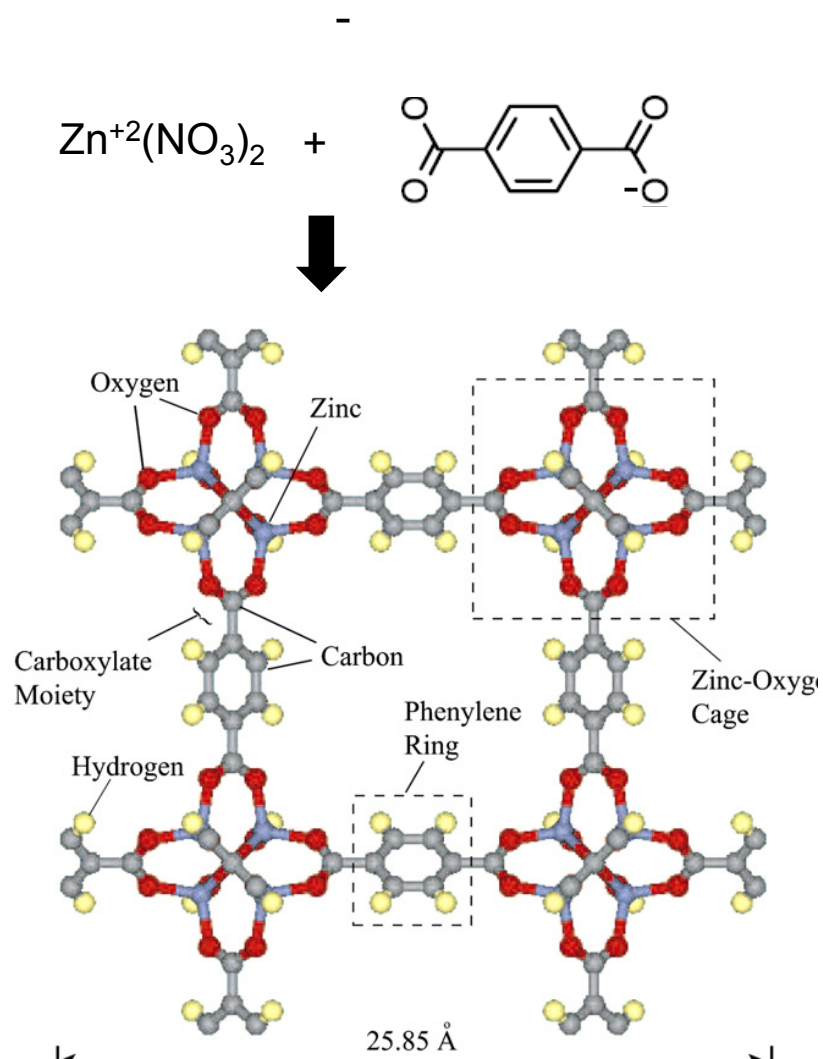


## Crystalline MOF semiconductor

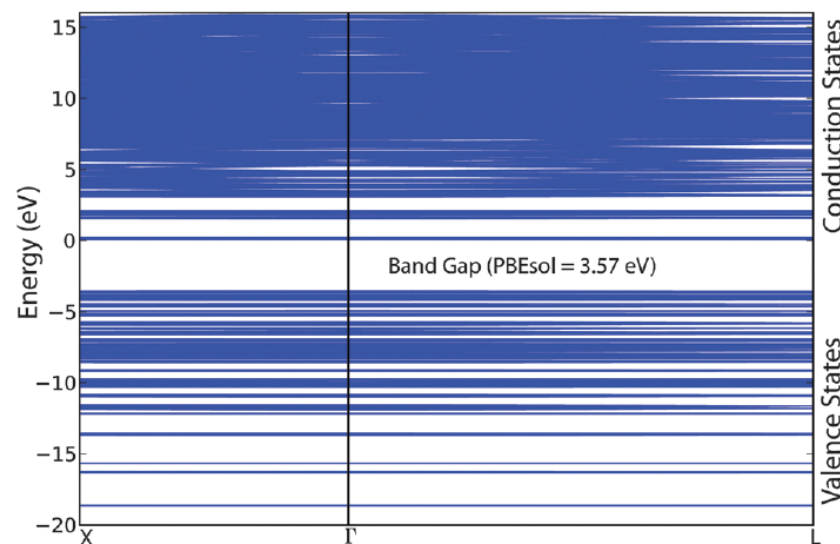
- Structurally flexible
- Tunable w/ chemistry
- Scalable to nanometers
- Low cost fabrication
- Reconfigurable electronics
- Rad-hard
- Novel electronic material

*MOFs combine features of inorganic and organic materials*

# Most MOFs are Insulators, Lack Delocalized p-Network

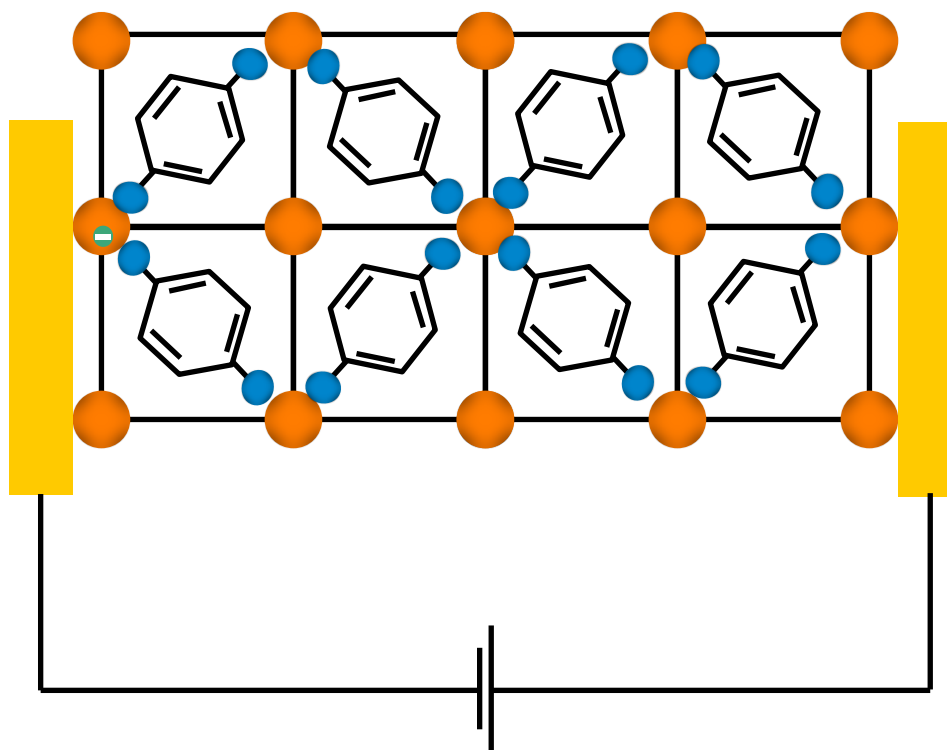


M. D. Allendorf, A. Schwartzberg, V. Stavila, A. A. Talin, *Chemistry – A European Journal* 17, 11372 (2011).



C. H. Hendon, D. Tiana, A. Walsh, *Phys. Chem. Chem. Phys.*, 2012, **14**, 13120

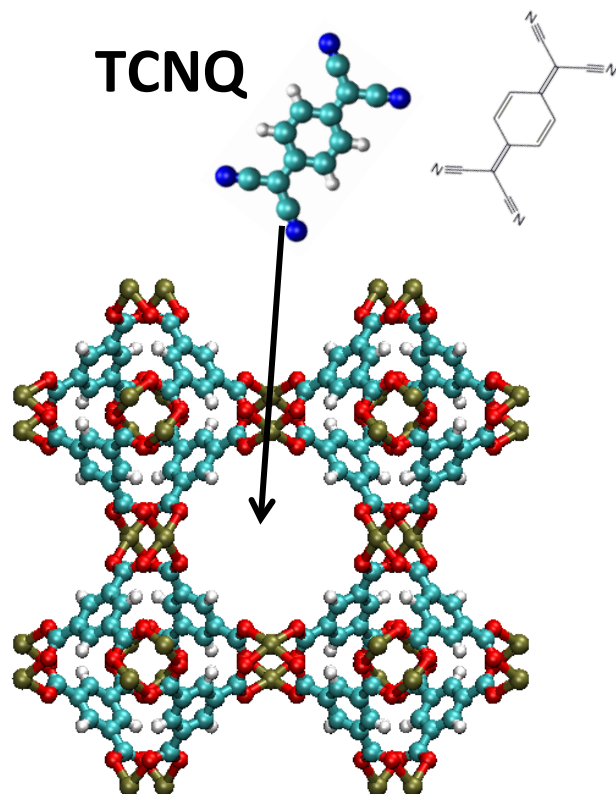
# Can a 'guest' provide coupling between SBUs to create a conduction path?



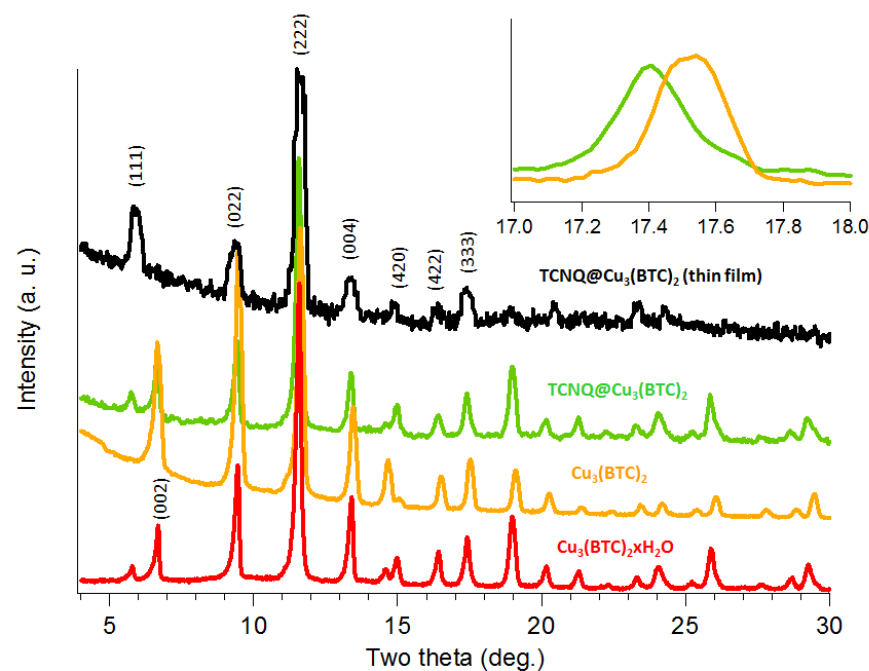
Requirements:

- Open metal sites
- Delocalized  $\pi$ -electrons
- Strong ligand-metal coupling

# Guest@MOF: Emergent properties by infiltrating with guest molecules?



$\text{Cu}_2(\text{BTC})_3$   
(HKUST-1)



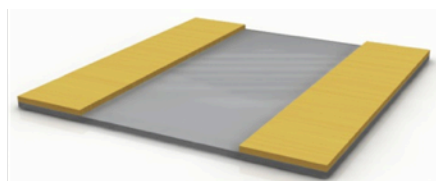
TCNQ loading: ~ 1 molecule/large pore



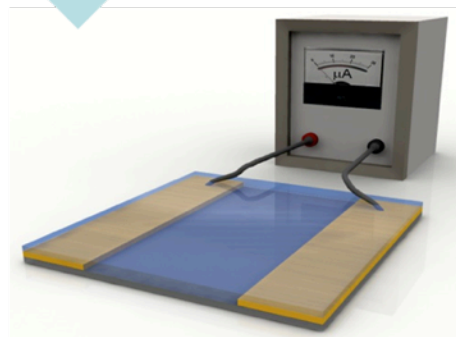
# TCNQ $\rightarrow$ $\text{Cu}_2(\text{BTC})_3$ leads to color change...

MOF film grown by layer-by-layer method

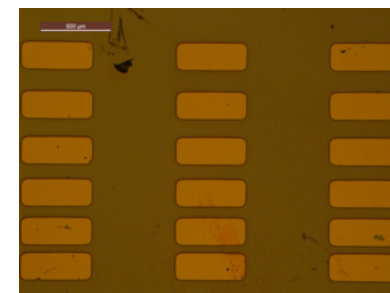
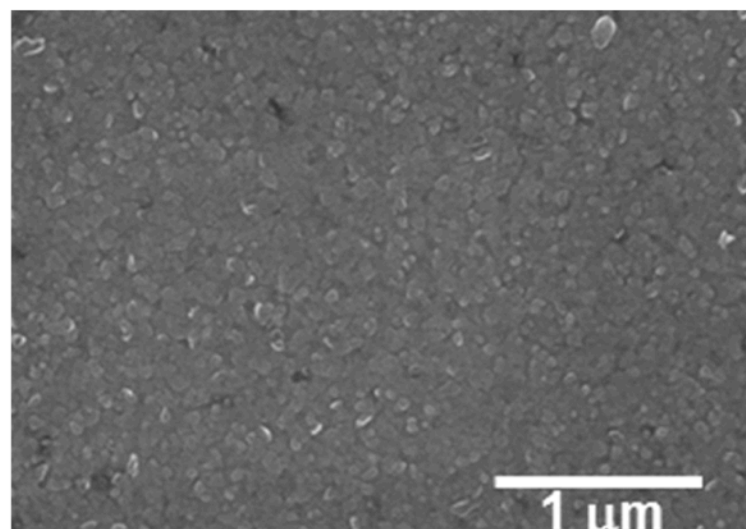
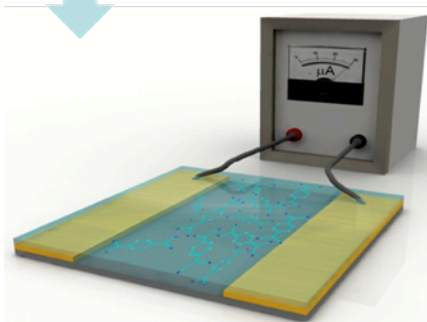
MOF film on  $\text{SiO}_x$  with Pt electrodes



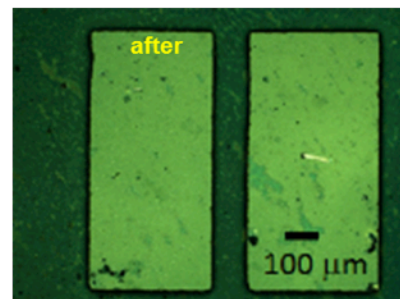
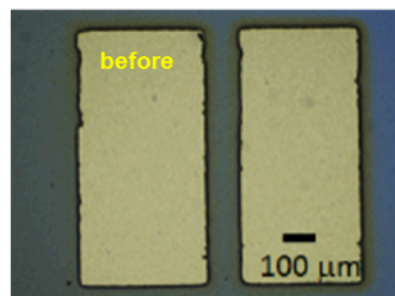
MOF growth



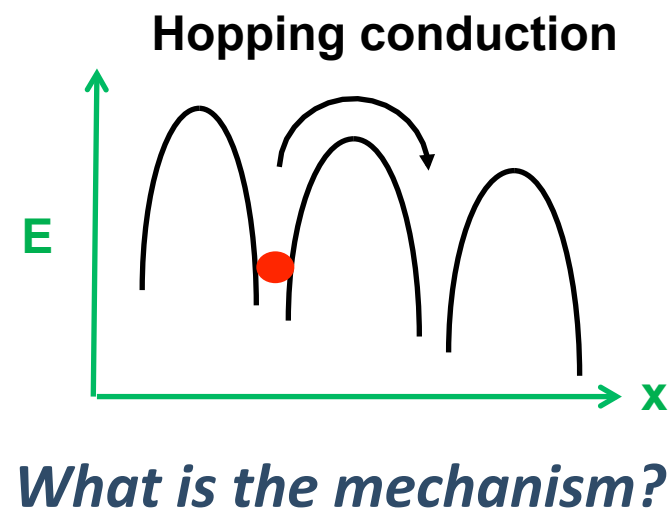
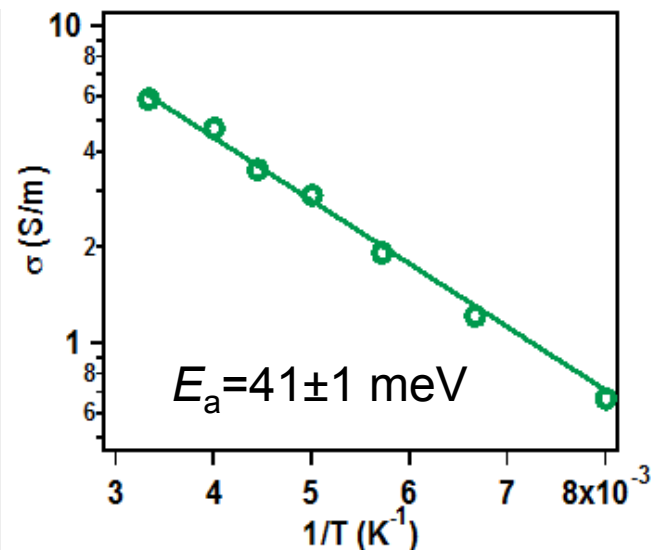
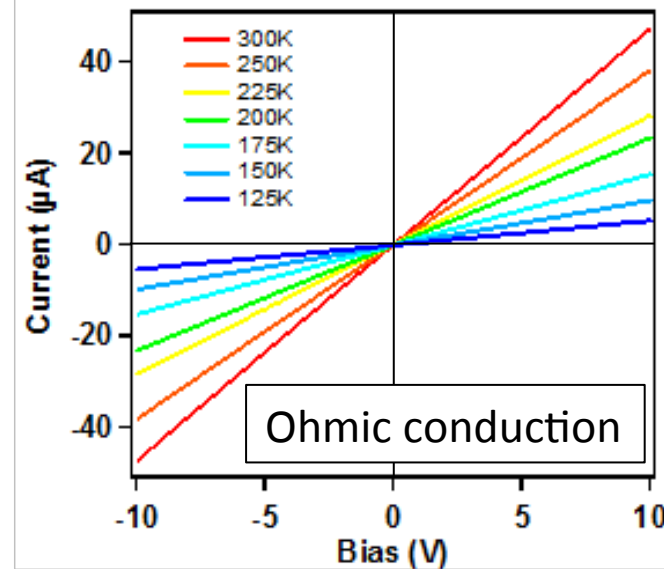
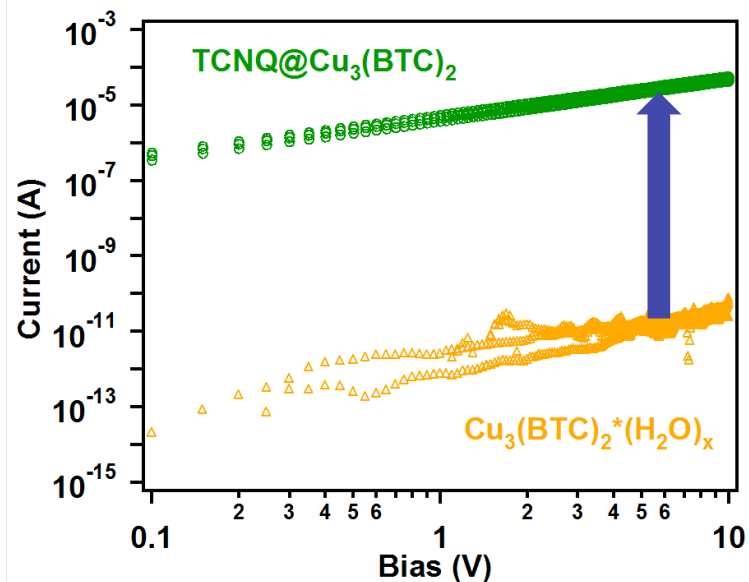
Molecule infiltration



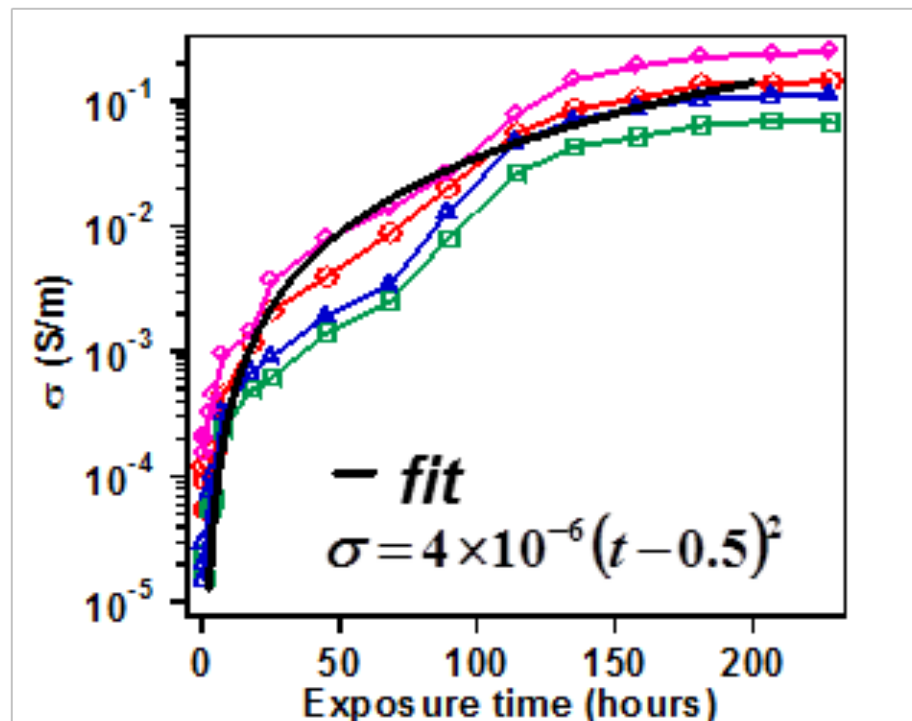
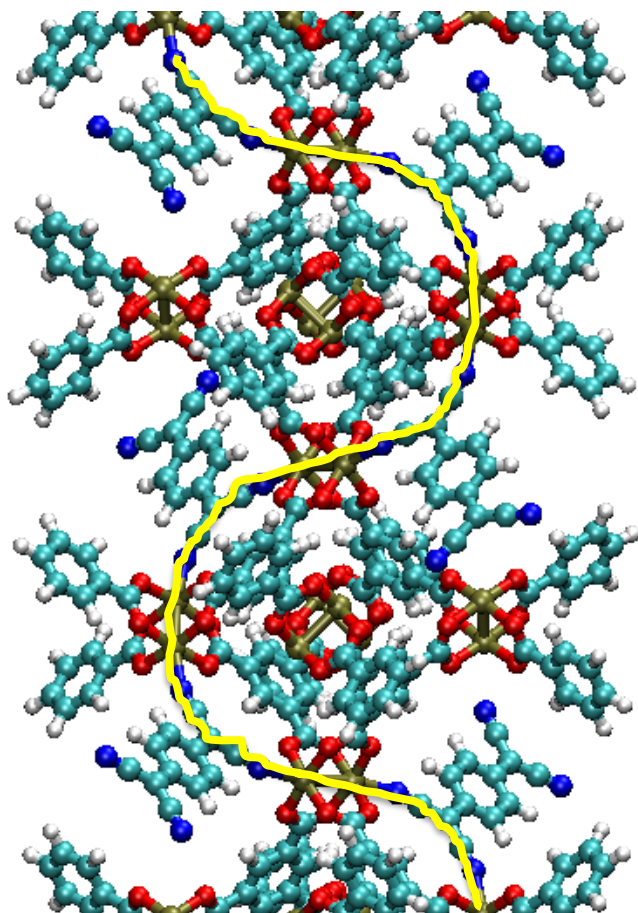
MOF film + TCNQ



... and  $>10^7$  increase in conductivity, air stable  $> 1$  year



# Proposed conductivity mechanism: $\text{Cu}_2(\text{btc})_4$ paddlewheels bridged by TCNQ

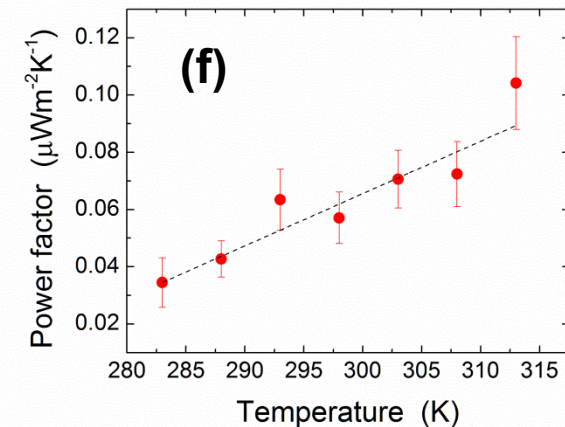
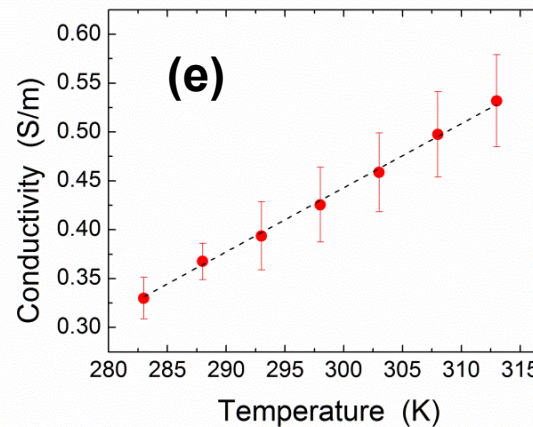
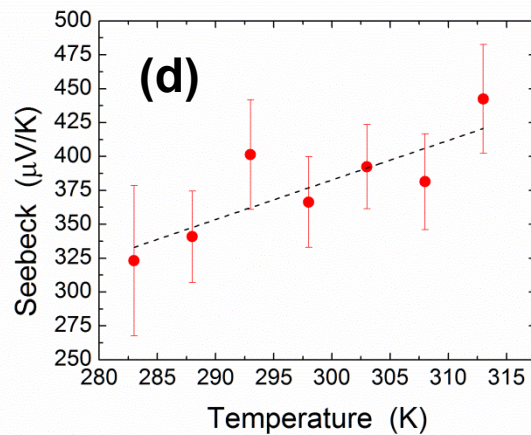
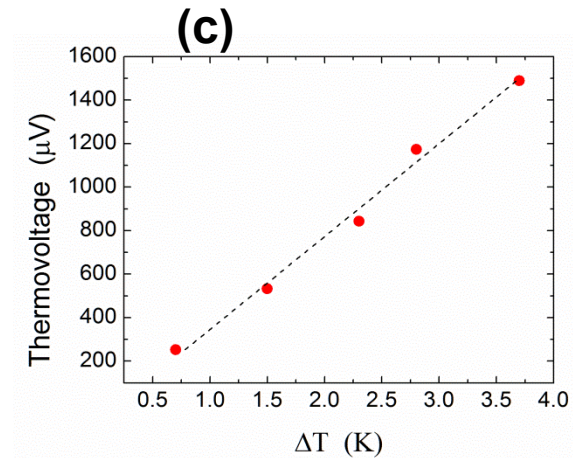
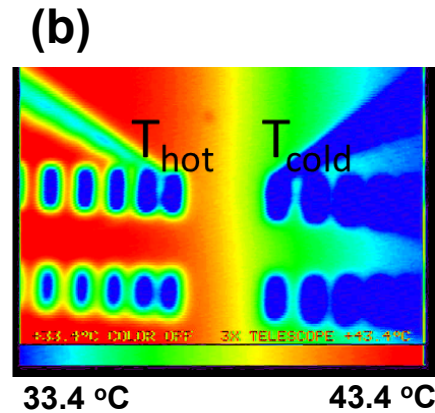
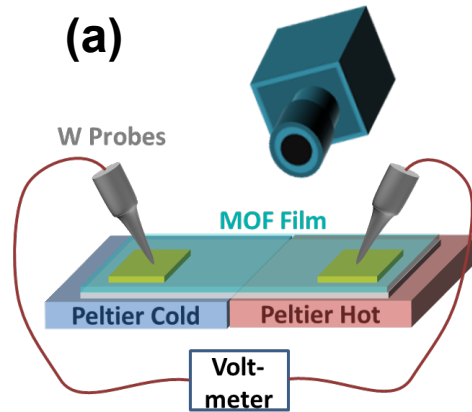


Percolation model

Continuous TCNQ@CuBTC pathway is achievable with 4 TCNQs  
Experimental loading = 8 TCNQs/unit cell  $\rightarrow$  two continuous pathways are possible



# $S$ , $\sigma$ increase with temperature



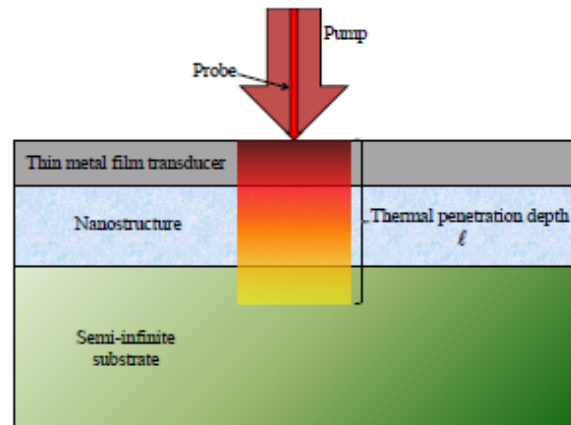
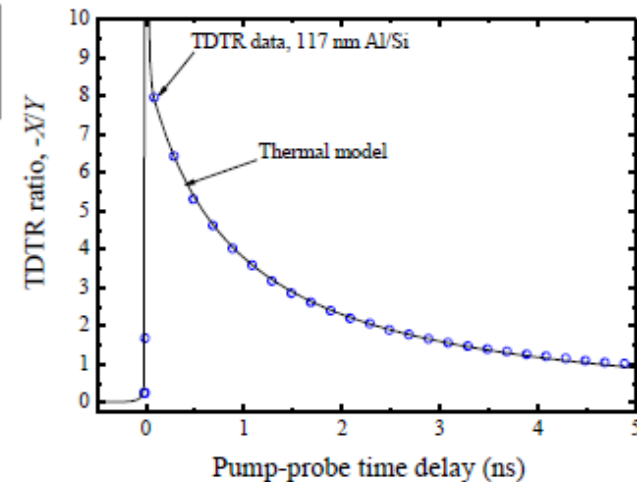
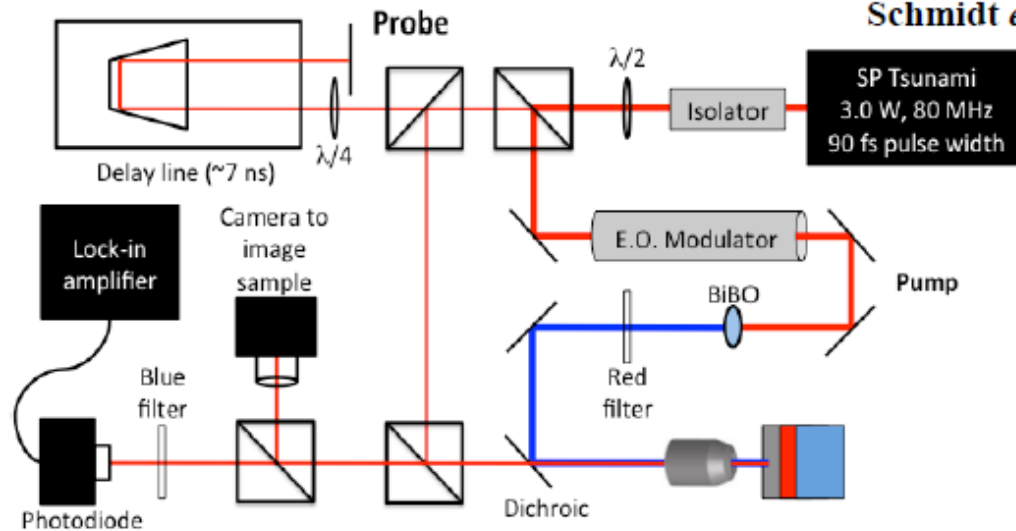
K. J. Erickson, F. Léonard, V. Stavila, M. E. Foster, C. D. Spataru, R. E. Jones, B. M. Foley, P. E. Hopkins, M. D. Allendorf, and A. A. Talin, *Advanced Materials*, accepted

# Thermal conductivity measured by Time Domain ThermoReflectance (TDTR) (P. Hopkins group)

Hopkins *et al.*, *J. Heat Trans.* 132, 081302 (2010)

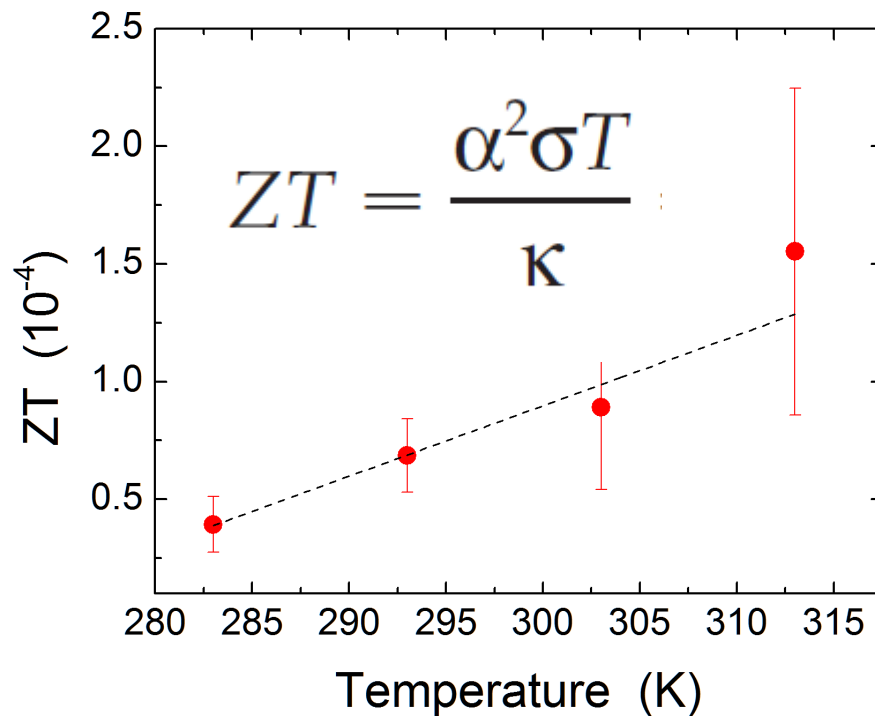
Cahill, *Rev. Sci. Instr.* 75, 5119 (2004)

Schmidt *et al.*, *Rev. Sci. Instr.* 74, 114902 (2008)

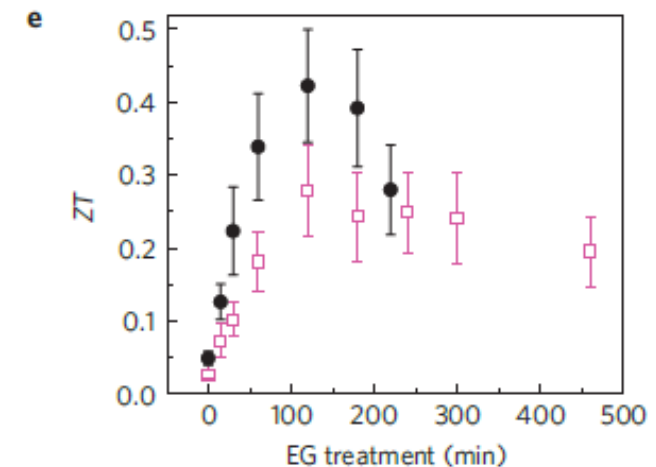
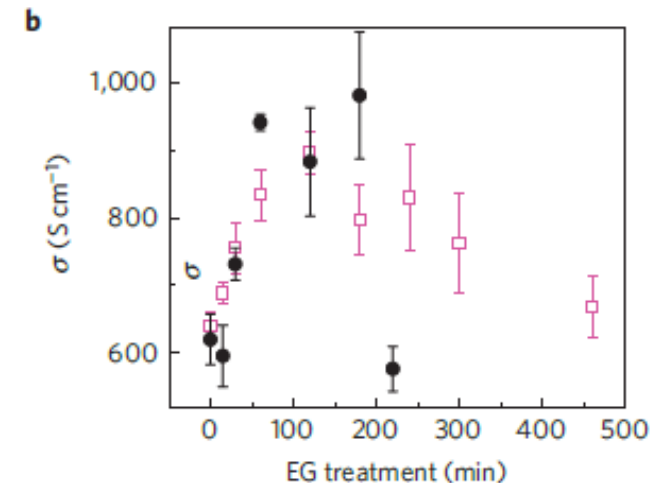


- Can measure thermal conductivity of thin films and substrates ( $\kappa$ ) separately from thermal boundary conductance ( $h_K$ )
- Nanometer spatial resolution ( $\sim 10$ 's of nm)
- Femtosecond to nanosecond temporal resolution
- Noncontact

# Large Seebeck, low $\kappa$ , but $\sigma$ still too low...



K. J. Erickson, F. Léonard, V. Stavila, M. E. Foster, C. D. Spataru, R. E. Jones, B. M. Foley, P. E. Hopkins, M. D. Allendorf, and A. A. Talin, *Advanced Materials*, accepted




G. H. Kim, L. Shao, K. Zhang, K. P. Pipe. *Nat. Mater.* **2013**, 12, 719



# Four basic roles for MOFs in devices

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- Adsorbant: provide sensitivity and selectivity (sensing)
  - Electronic: passive (low-k dielectric)
  - Electronic: active (sensing; thermoelectrics; logic)
  - Light absorbing/emitting: energy harvesting, radiation detection
  - Template or scaffold: ordered structures for improving exciton harvesting
- 
- 

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- Vitalie Stavila
- Alec Talin
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