

Degradation studies of Nylon and Kevlar

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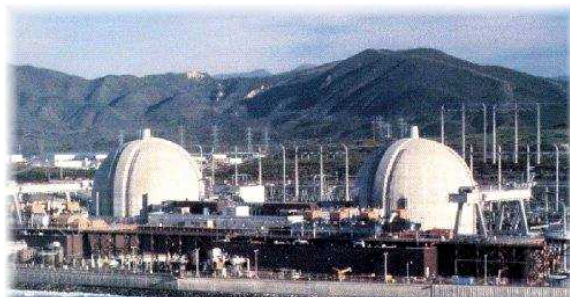
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Organic Materials Aging and Degradation



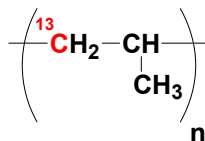
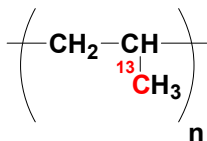
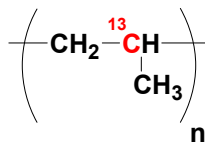
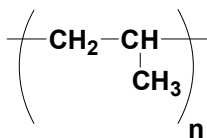
O-rings



Nuclear Power Plant Cable Insulation



Aircraft Wire Insulation



Labeled Polypropylene



Textiles



Deception!

Results from initial high temperature, short times (even out to 1 year) accelerated aging can be misleading.

Chemistry / mechanisms must be understood.



Approach/Goals

Macroscopic level

Physical property

Tensile Property

Permeation

Elongation

Dimensional changes

Molecular Level

Chemical Property

UV-VIS
Spectroscopy

Surface Analysis

Differential Scanning Calorimetry

Additives

Molecular Weight Analysis

Density

X-Ray Analysis

Infra-red
Spectroscopy

Nuclear Magnetic Resonance

GPC

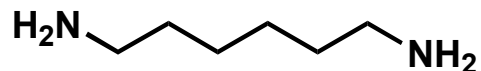
Goals

- Prediction of physical properties vs. time
 - Predict remaining lifetime of field materials
- Develop condition monitoring method



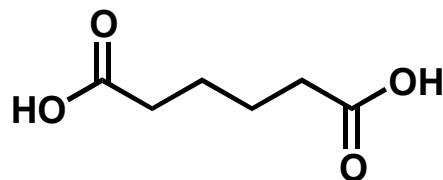
Nylon Structure

Vulnerable to degradation via oxygen and water

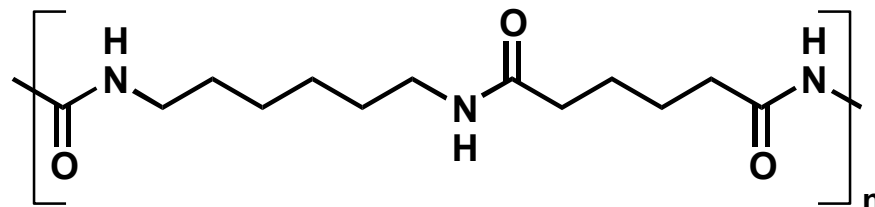
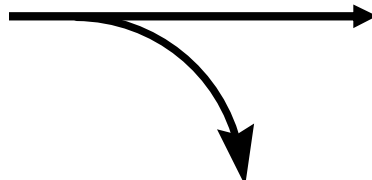


1,6 -Hexanediamine

+



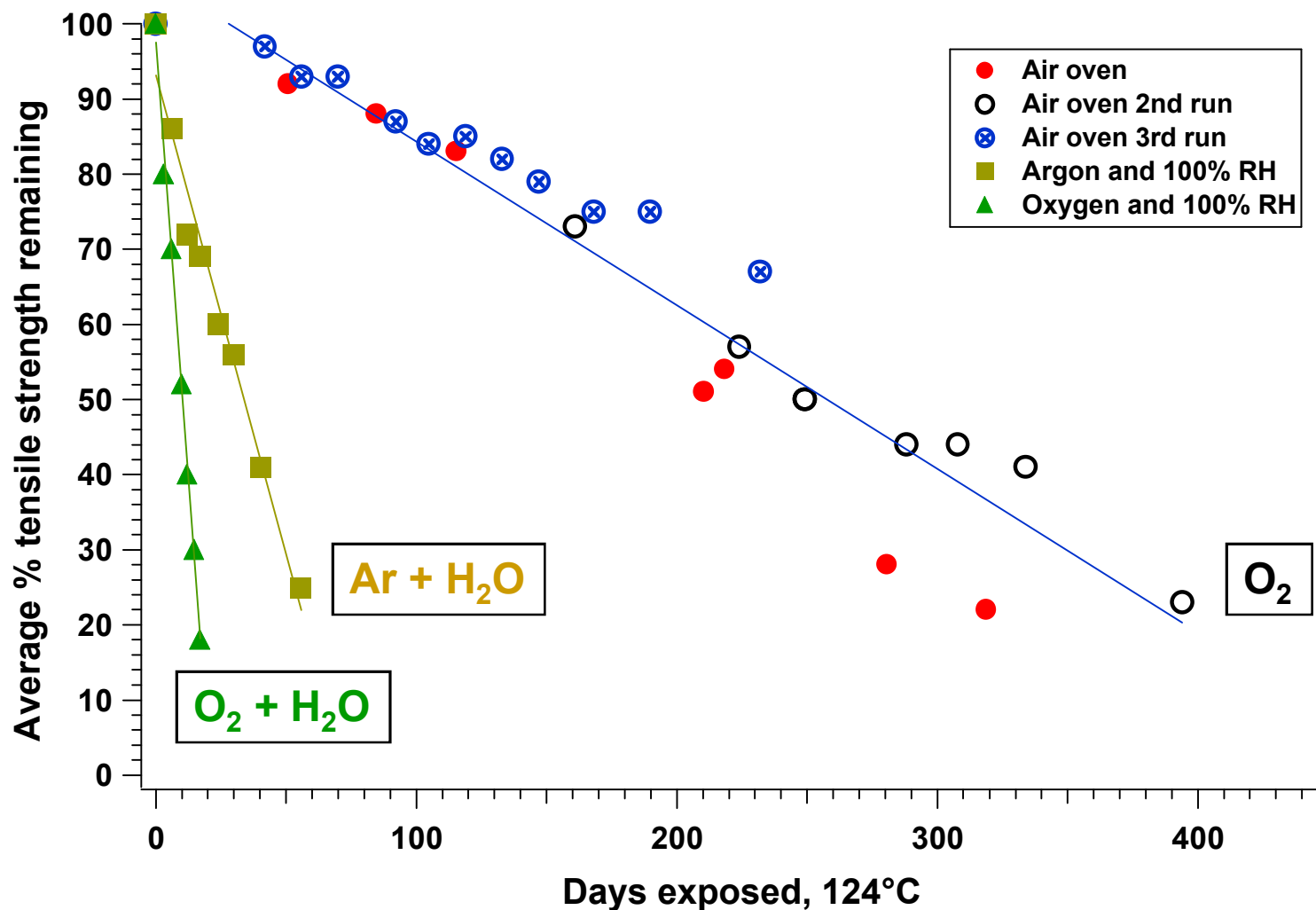
Adipic Acid



Nylon 6.6



Nylon Tensile Comparison



(1) Bernstein, R.; Derzon, D. K.; Gillen, K. T. Polymer Degradation and Stability, Nylon 6.6 accelerated aging studies: thermal-oxidative degradation and its interaction with hydrolysis 2005, 88, 480-488.
 (2) Bernstein, R.; Derzon, D. K.; Gillen, K. T. Polymer Preprints (American Chemical Society, Division of Polymer Chemistry), Nylon degradation studies: Humidity and aging 2002, 43, 1349.



Nylon Tensile Comparison

Conditions

Normalized Rate (Slope)

124 °C

Rate for Thermal and Oxygen

1.0

Rate for Thermal, **Argon**, and 100% R_H

7.4

8.4

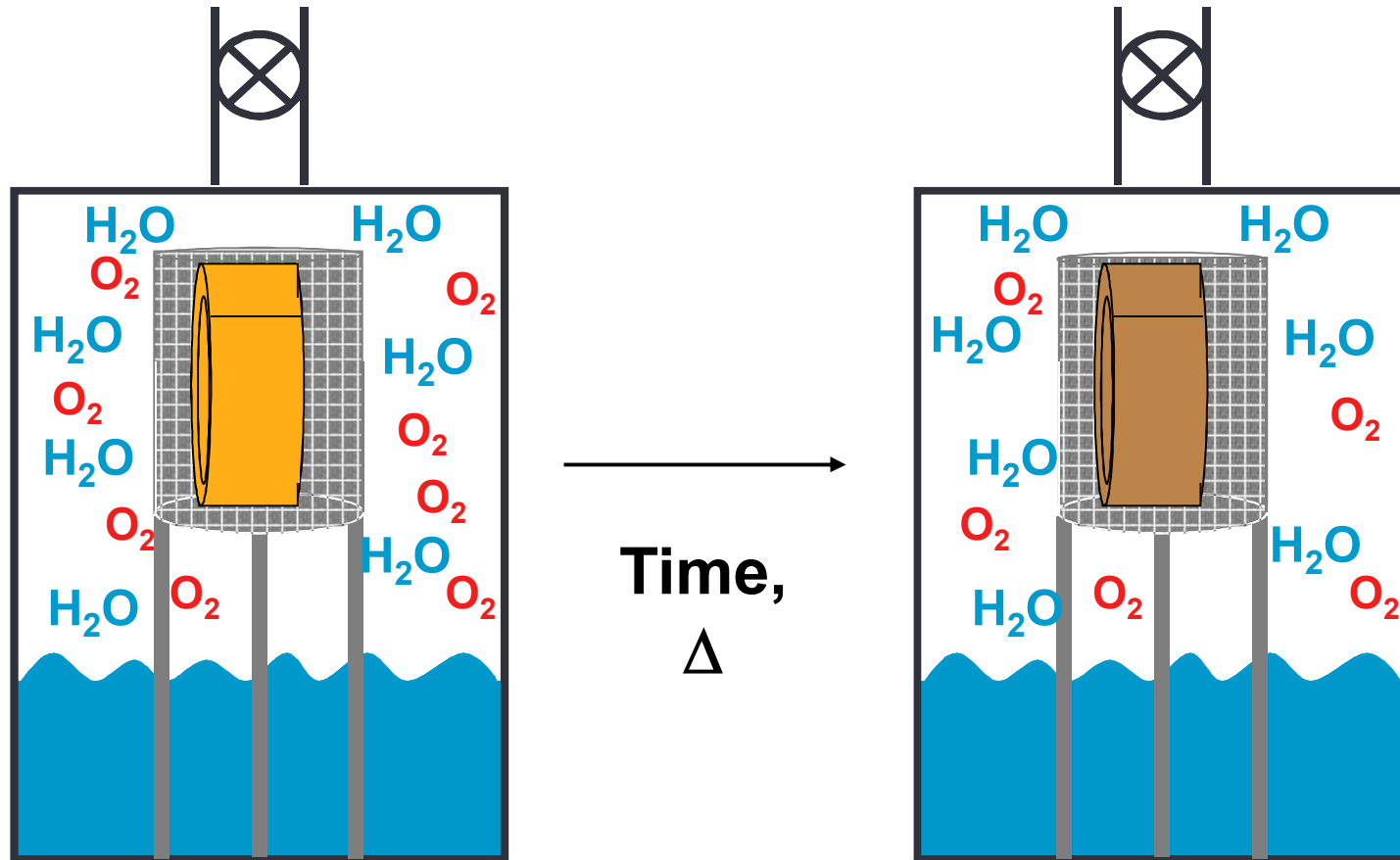
Rate for Thermal, **Oxygen**, and 100% R_H

22.9

Combined rate ca. 3 times sum of individual rates



Humidity Aging Schematic



Humidity Aging Hardware



Time-Temperature Superposition

Does mechanism change as a function of temperature?

If same mechanism:

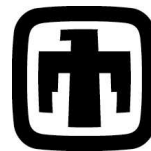
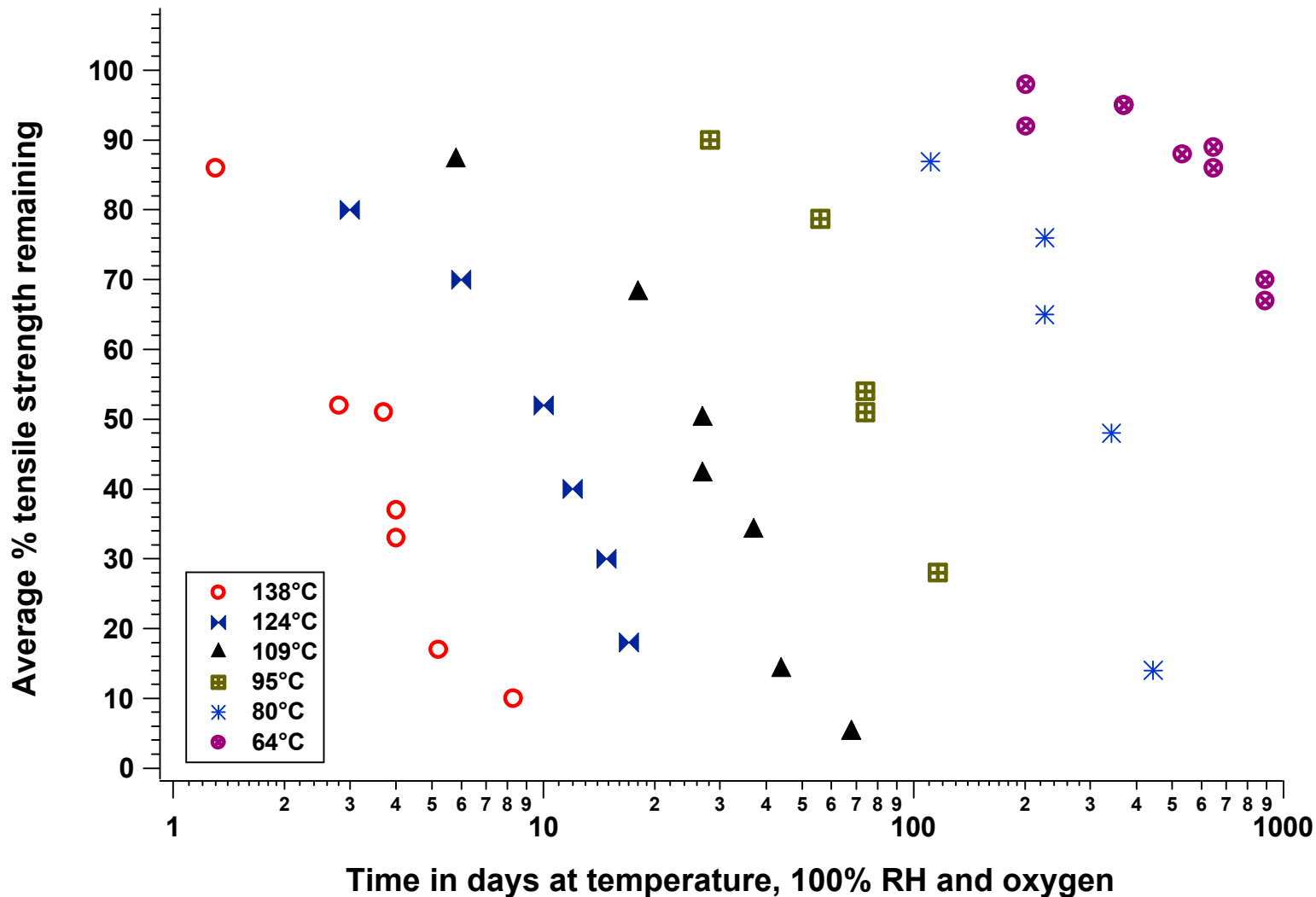
- same shape (log graph)
- should be constant acceleration (multiple)

1. Pick a reference temperature
2. Multiply the time at each temperature by the constant that gives the best overlap with the reference temperature data
3. Define that multiple as ' a_T ' ($a_T = 1$ for ref. temp.)
4. Find a_T for each temperature

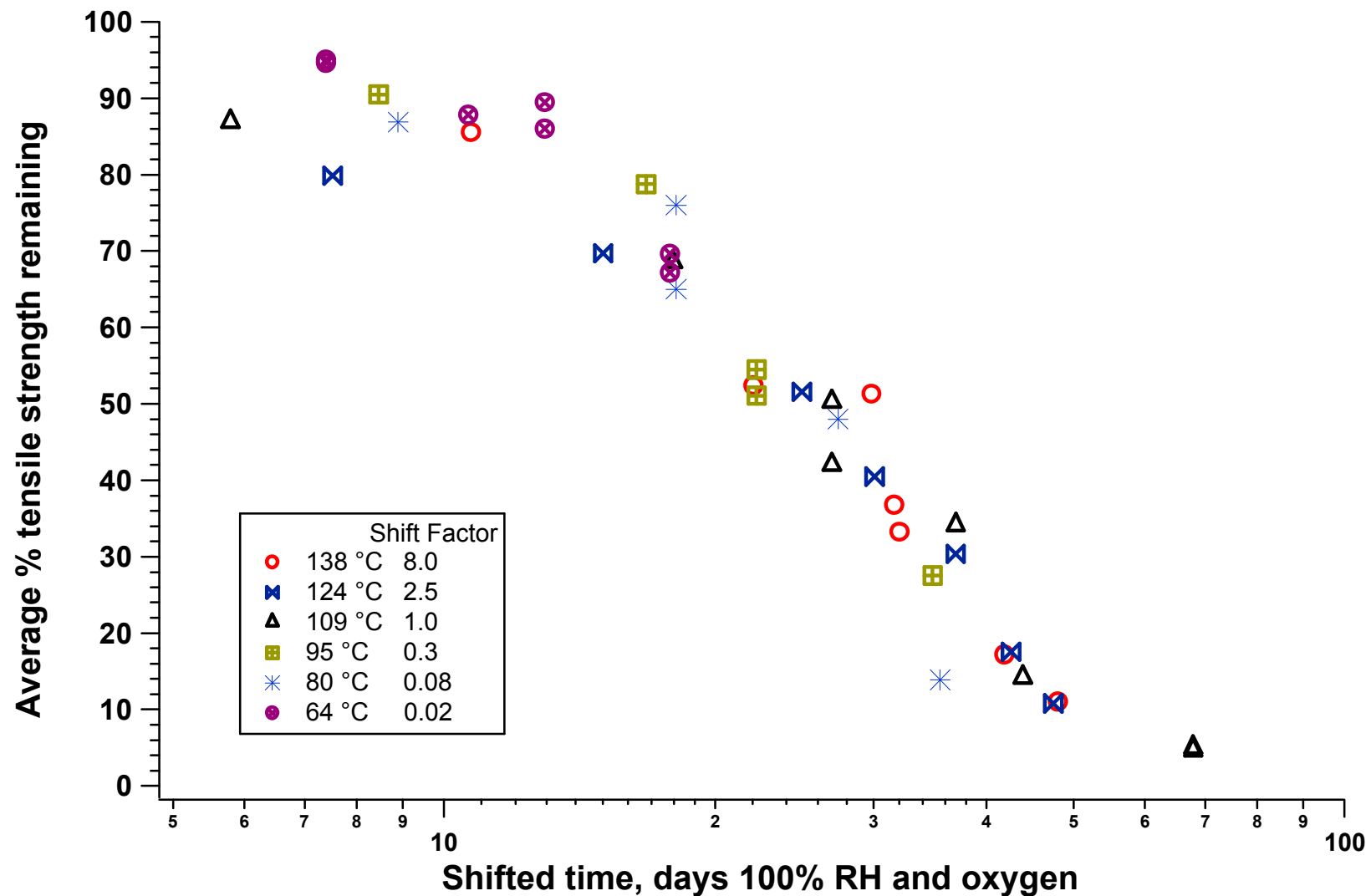
Plot $\log(a_T)$ vs $1/T$ linear if Arrhenius



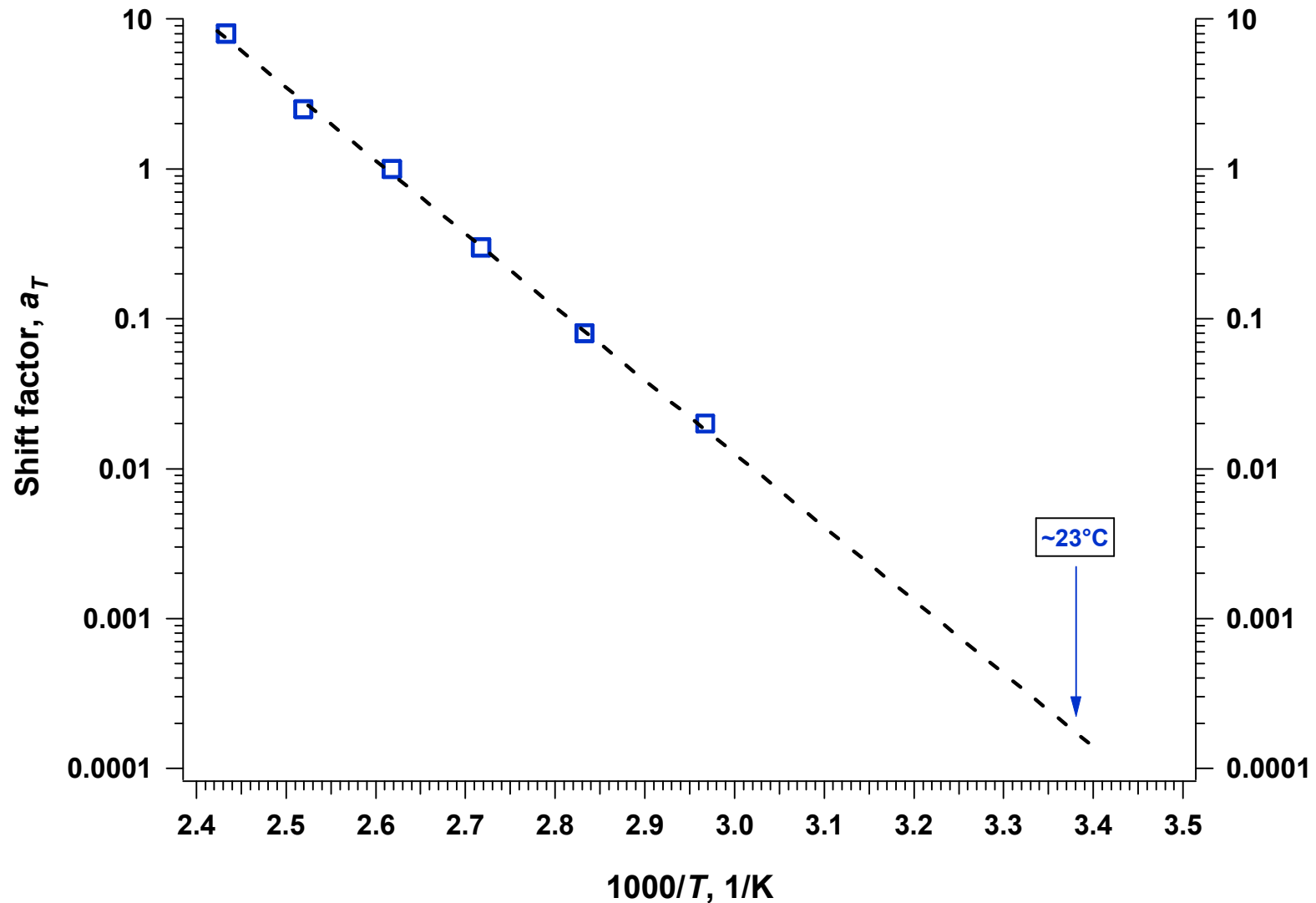
Nylon 100% RH O₂



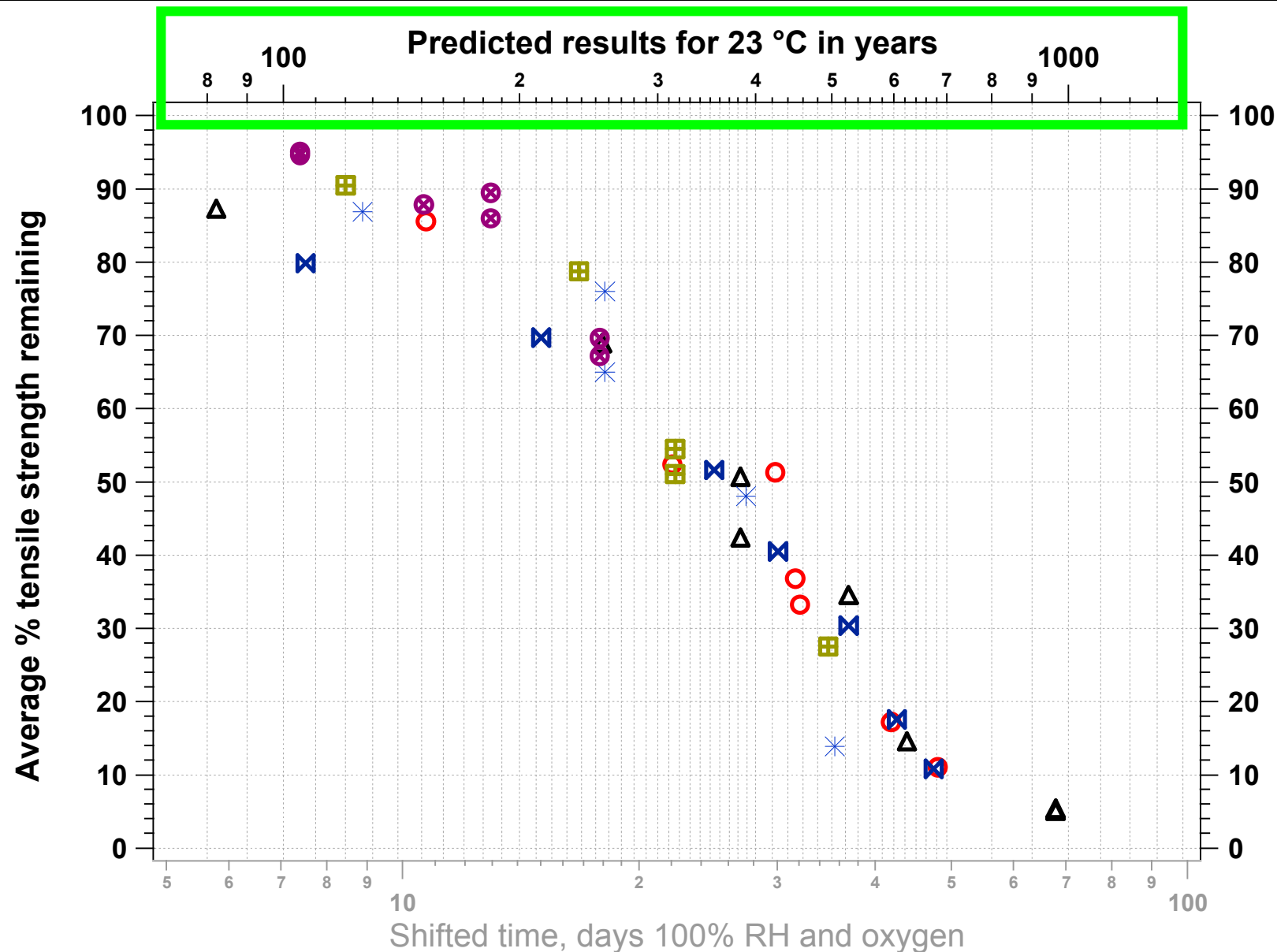
Shifted Nylon 100% RH O₂



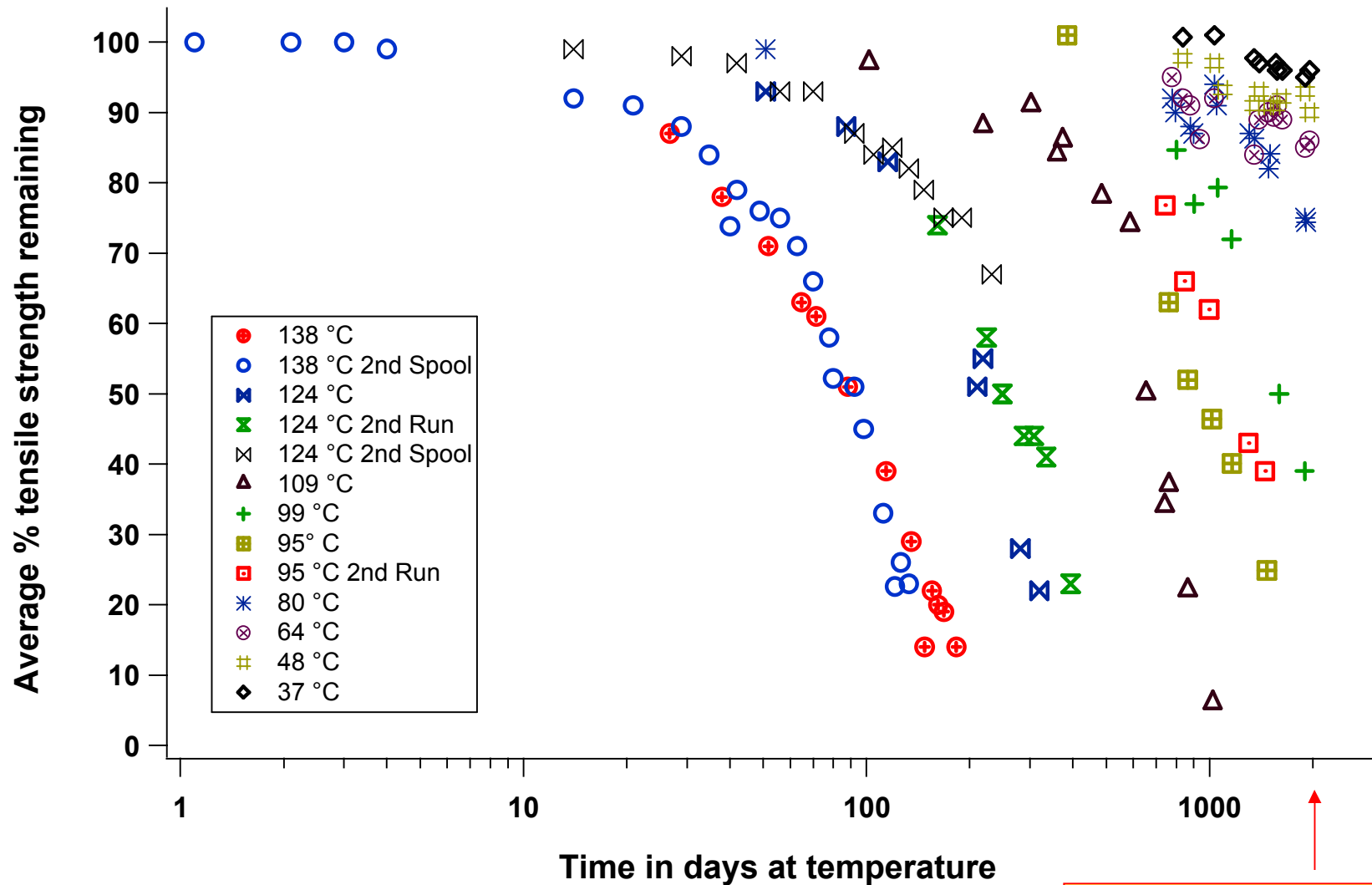
Shift Factors Nylon 100% RH O₂



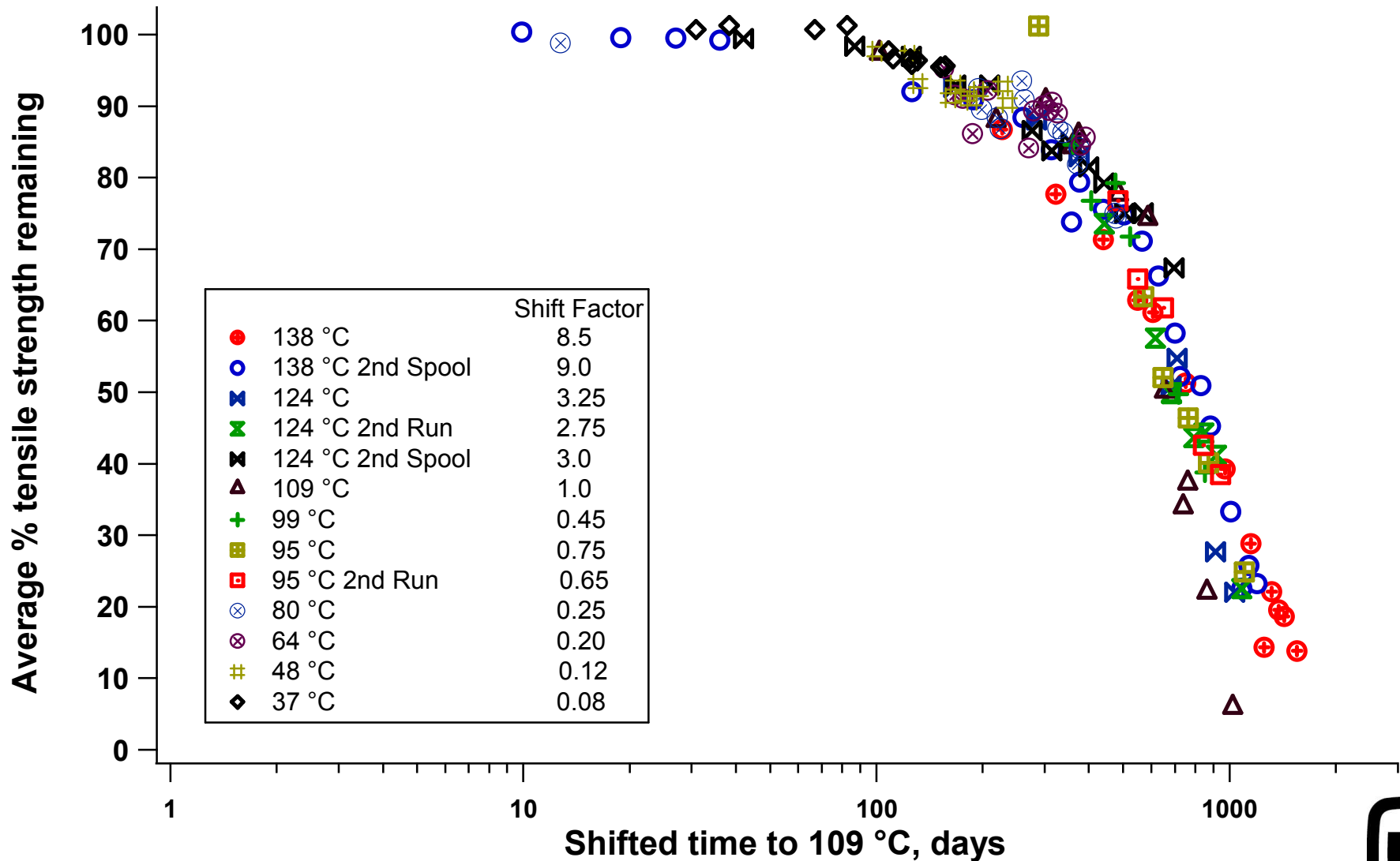
Predicted Tensile Strength; Nylon 100% RH O₂ at 23 °C



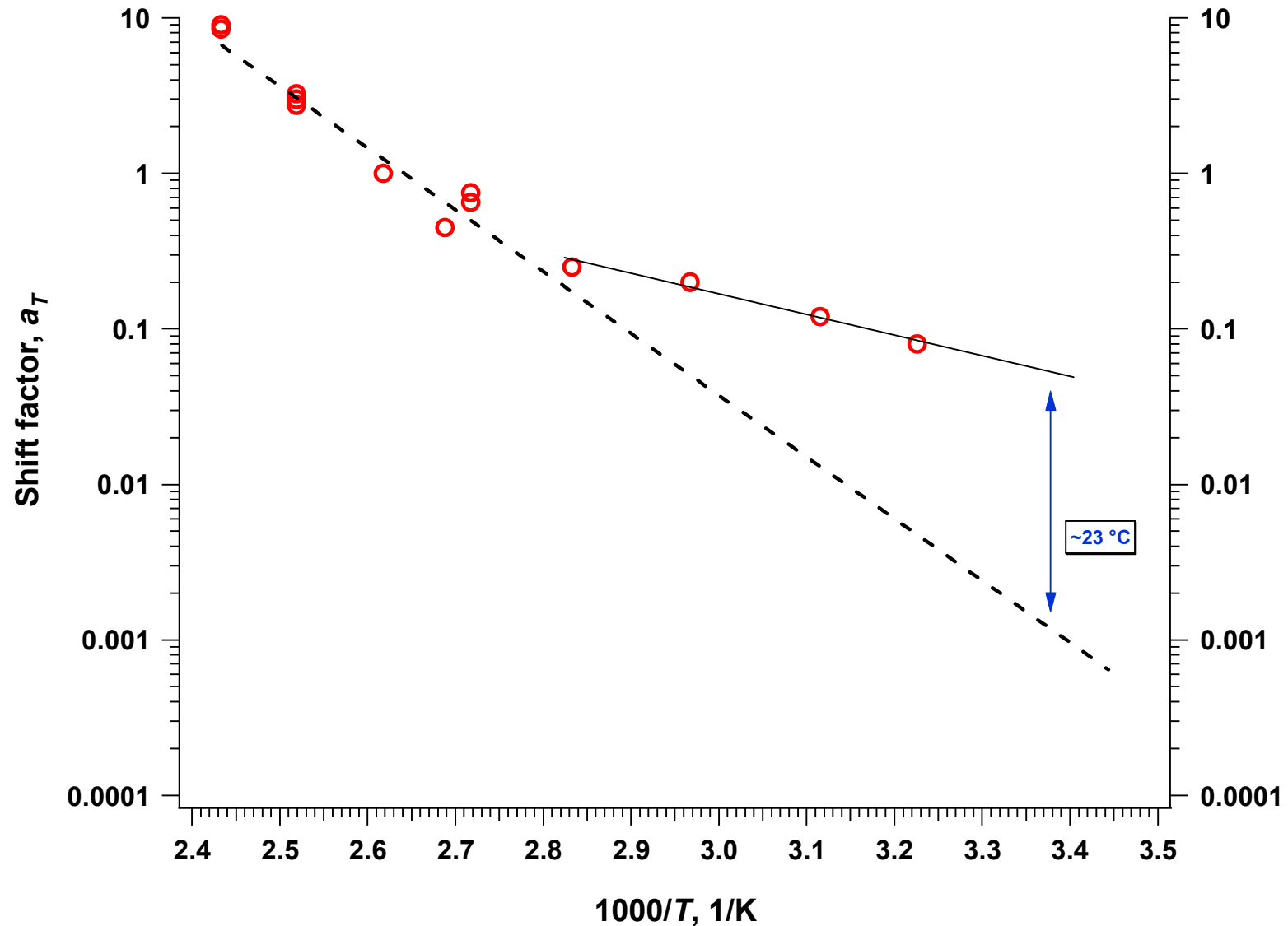
Thermal-oxidative Aging: Nylon



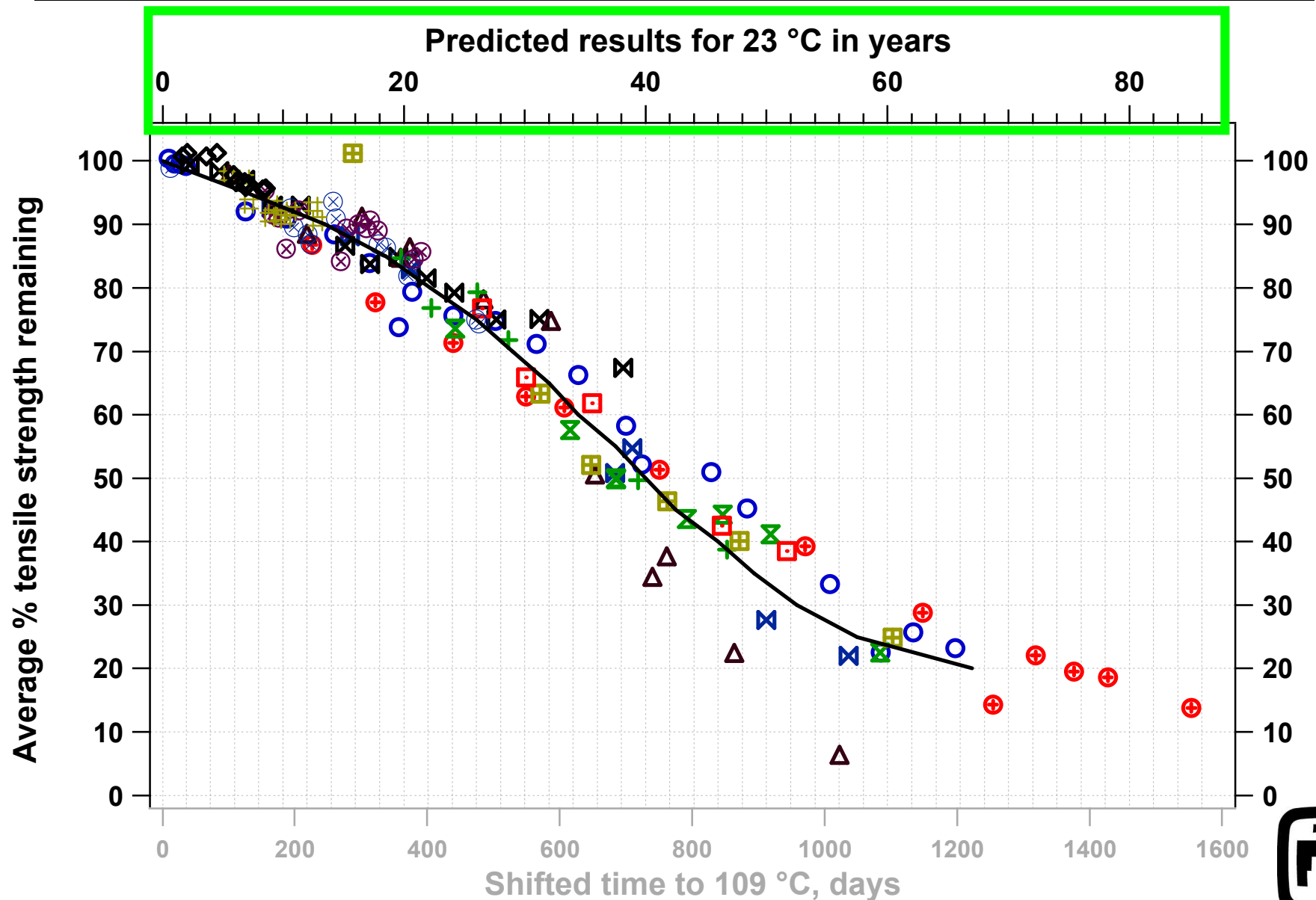
Thermal-oxidative Aging: Nylon Shifted Data



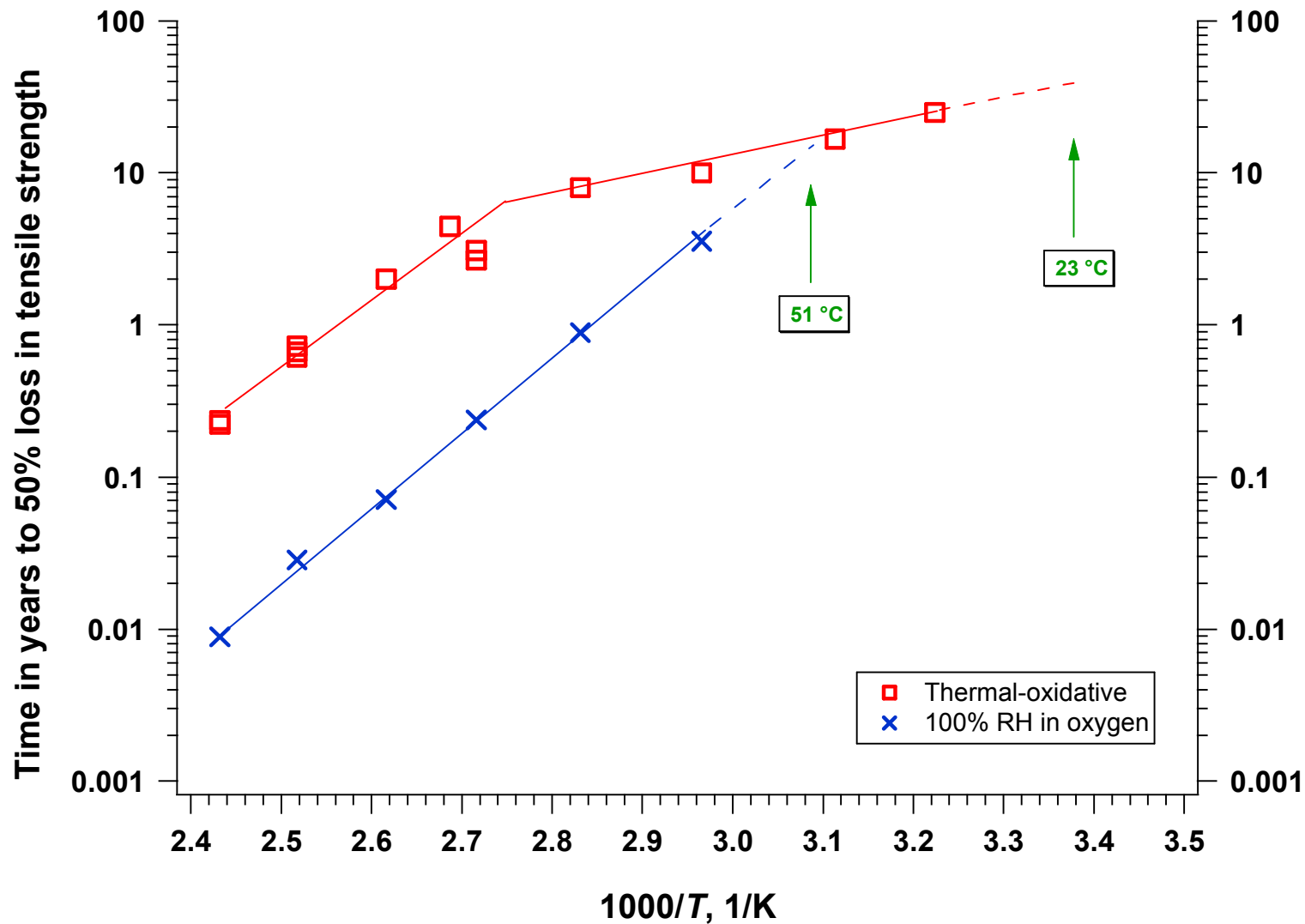
Thermal-oxidative Aging: Nylon Shift Factor Graph



Thermal-oxidative Aging: Nylon Shifted Data

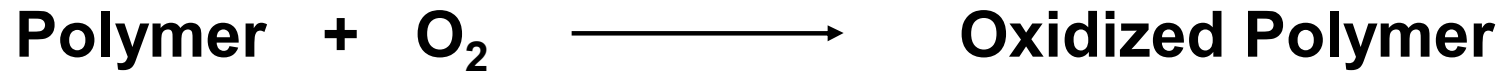


Thermal-oxidative and Humidity Comparison



Thermal Exposure

Thermal-Oxidation

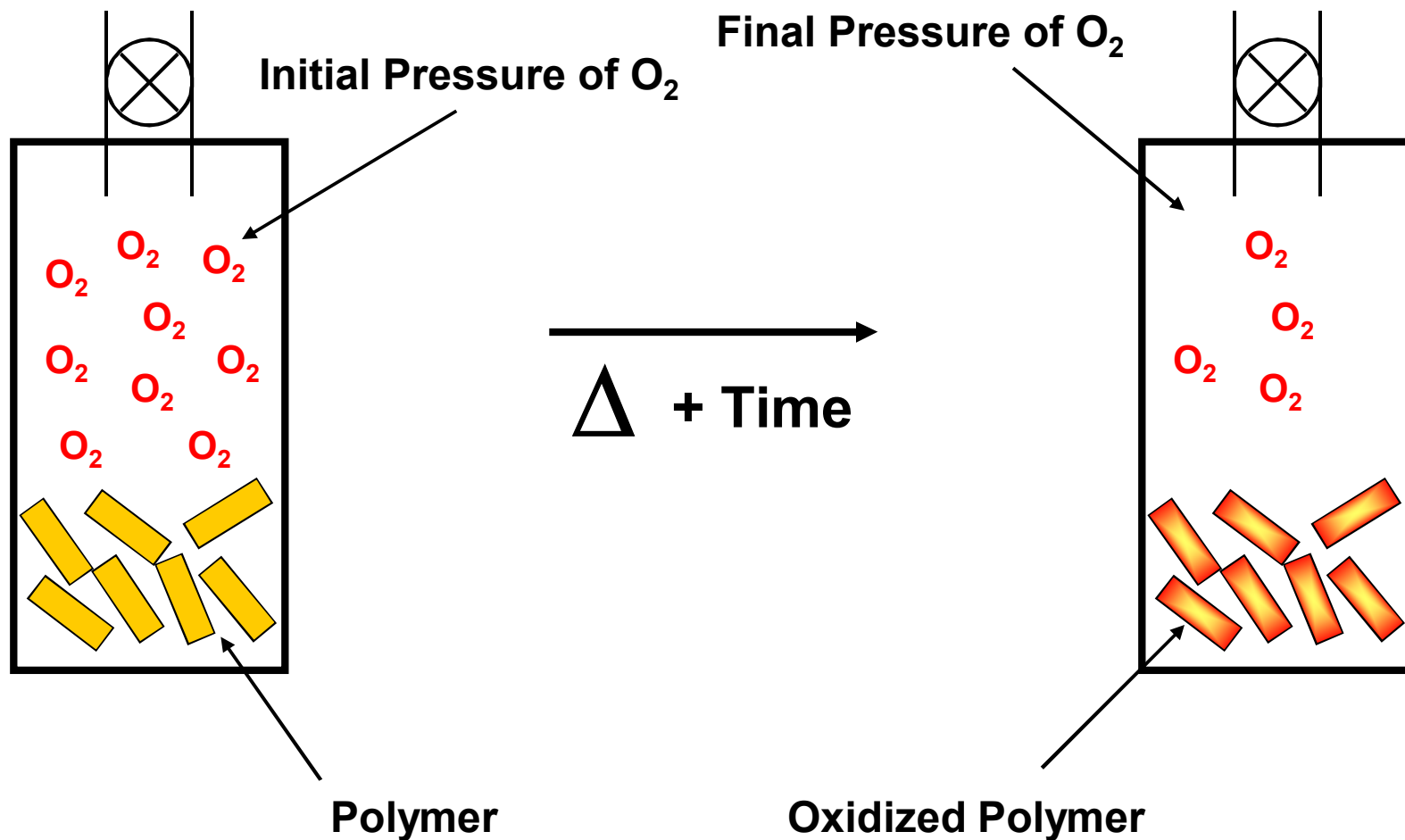


Quantify amount of oxygen consumed

- **Simple in theory**
- **Difficult in practice**
- **Amazingly sensitive**



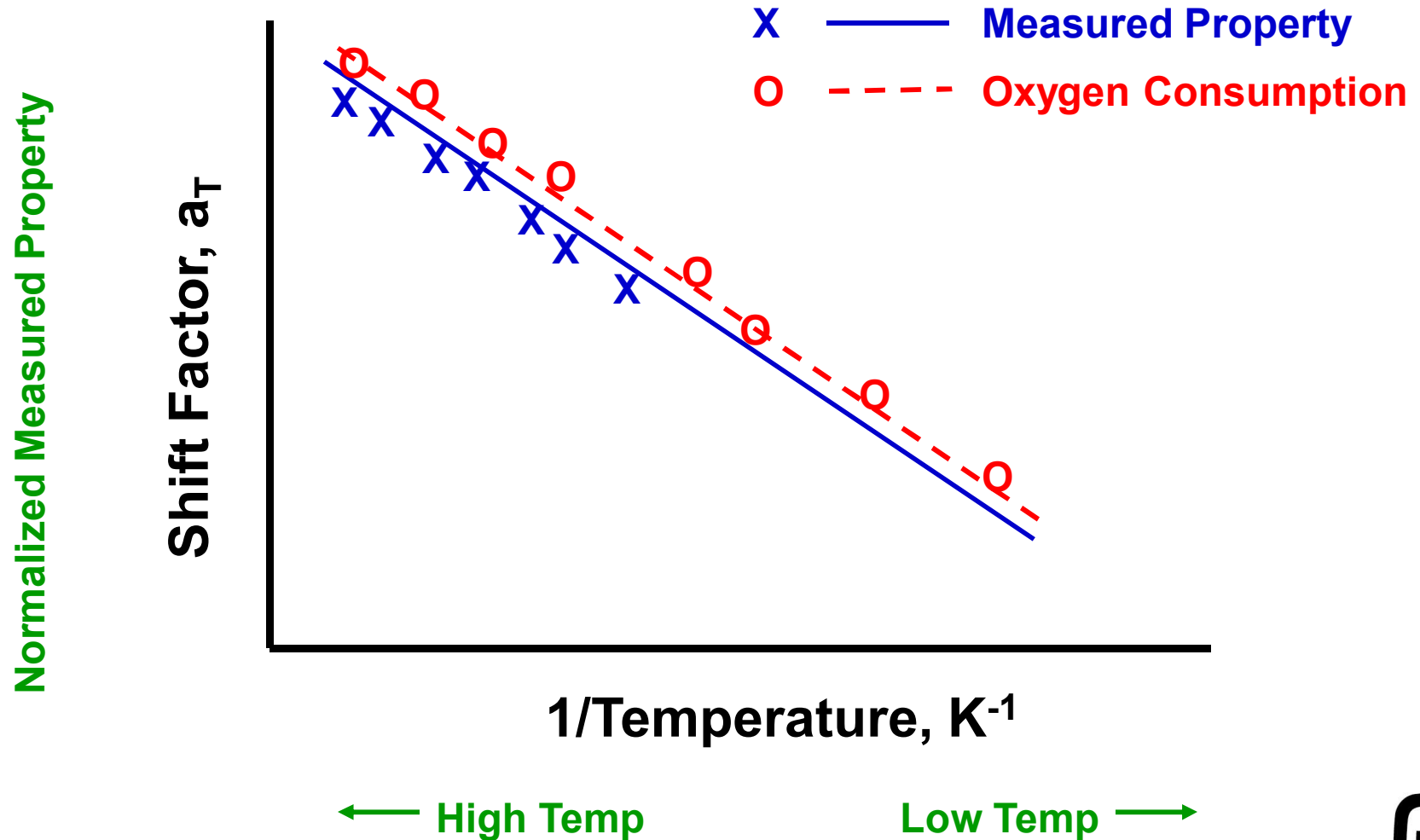
Schematic of Oxuptake



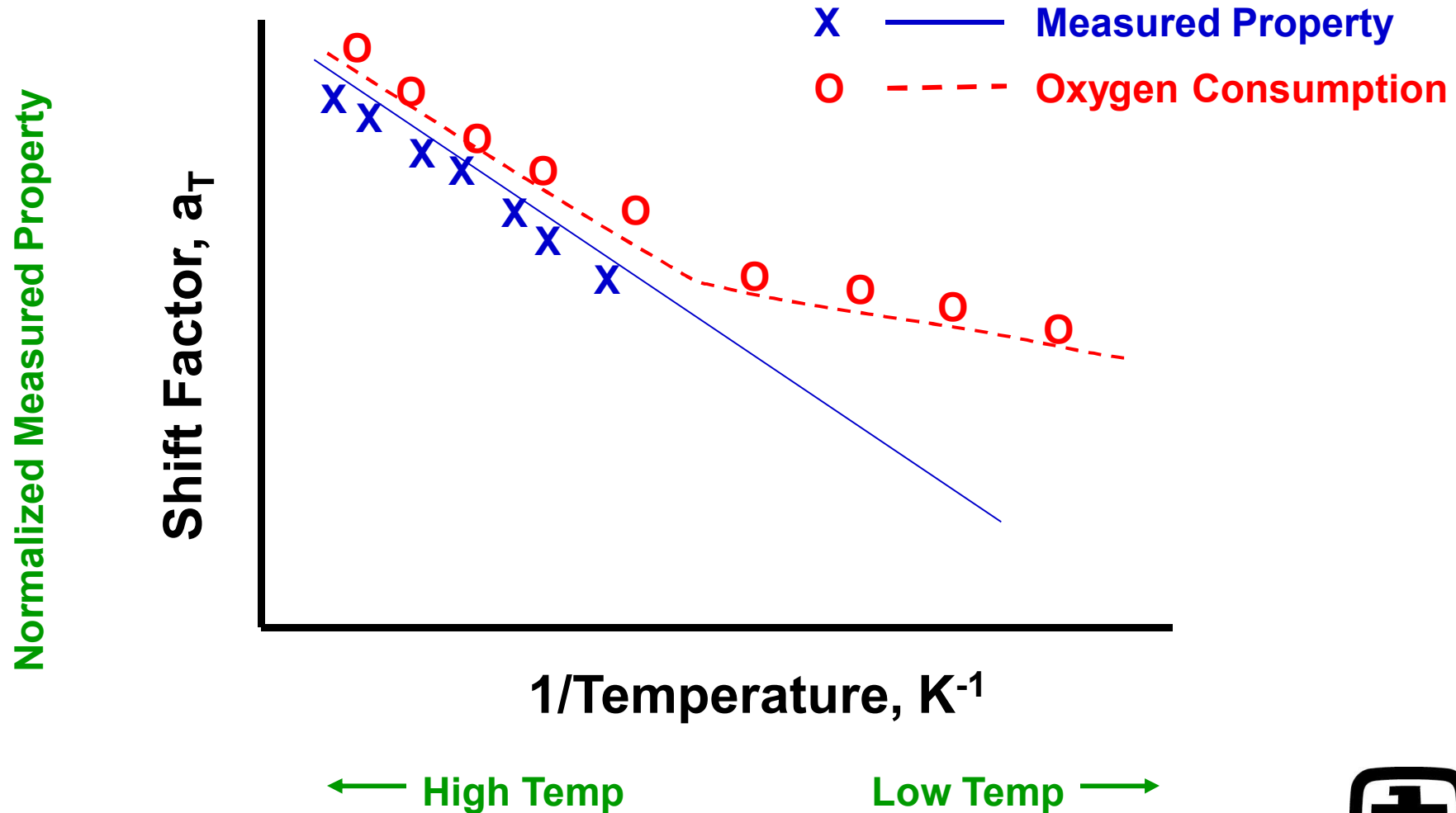
Oxygen Consumption



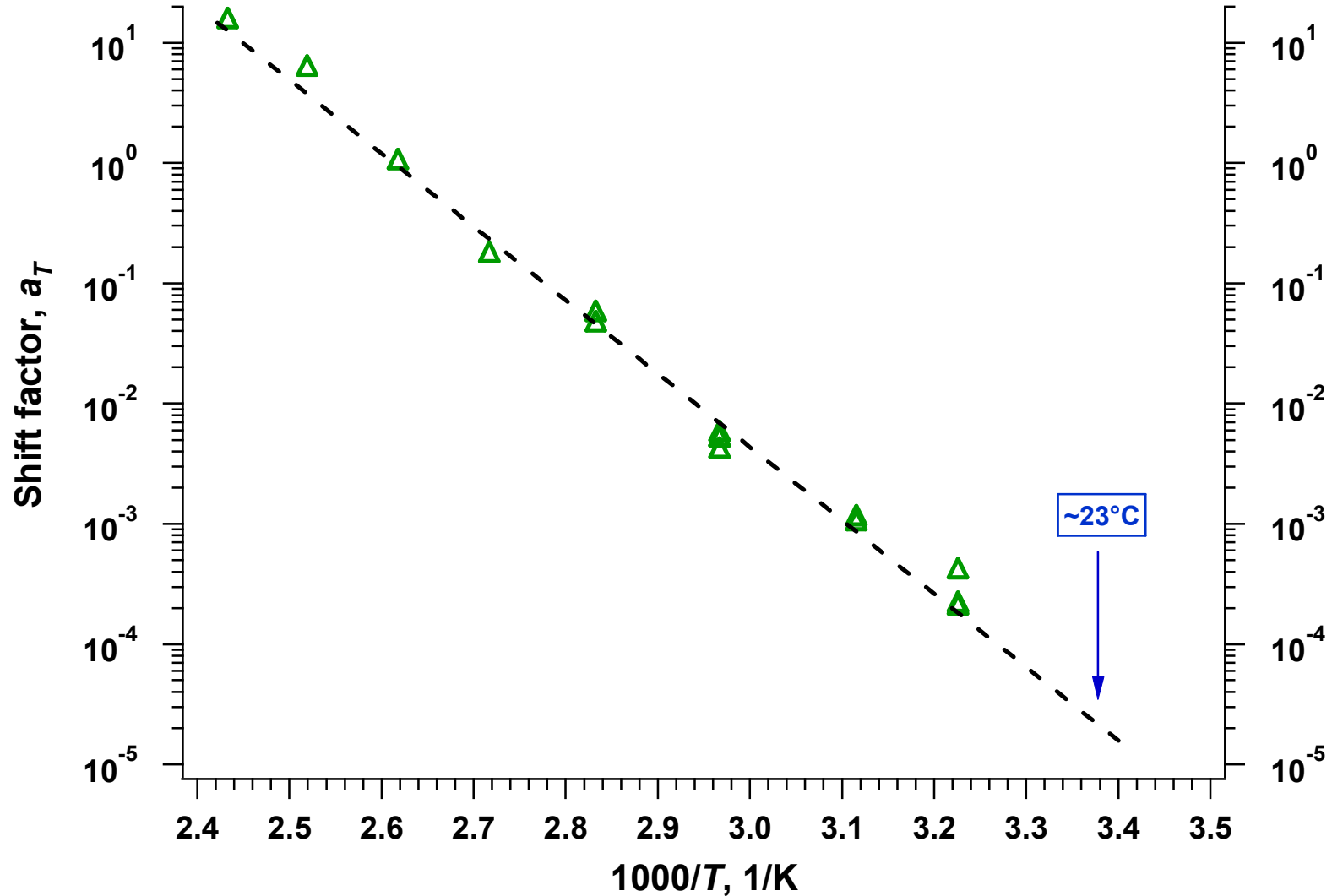
Enhanced Extrapolation 'Good'



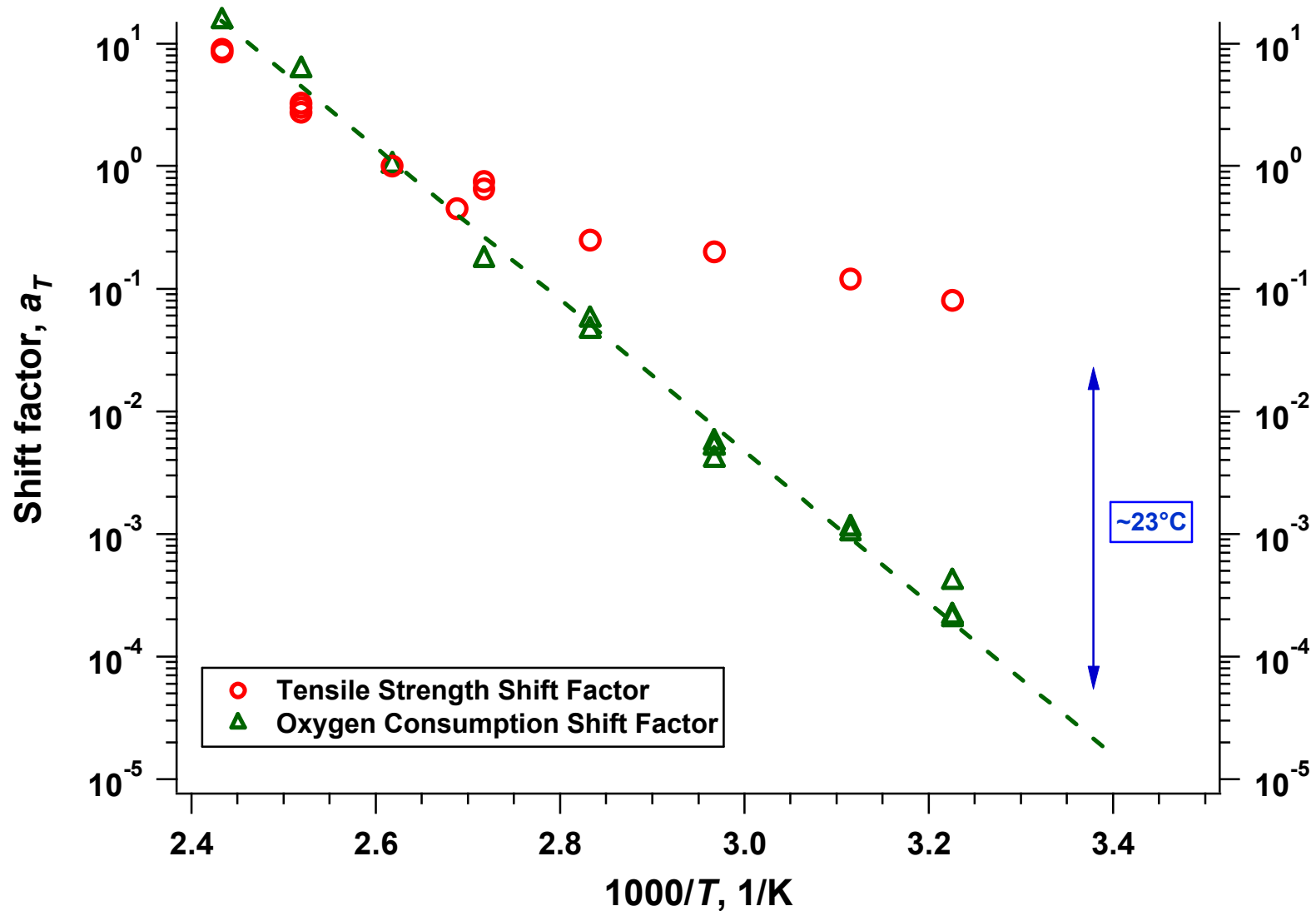
Enhanced Extrapolation: 'Bad'



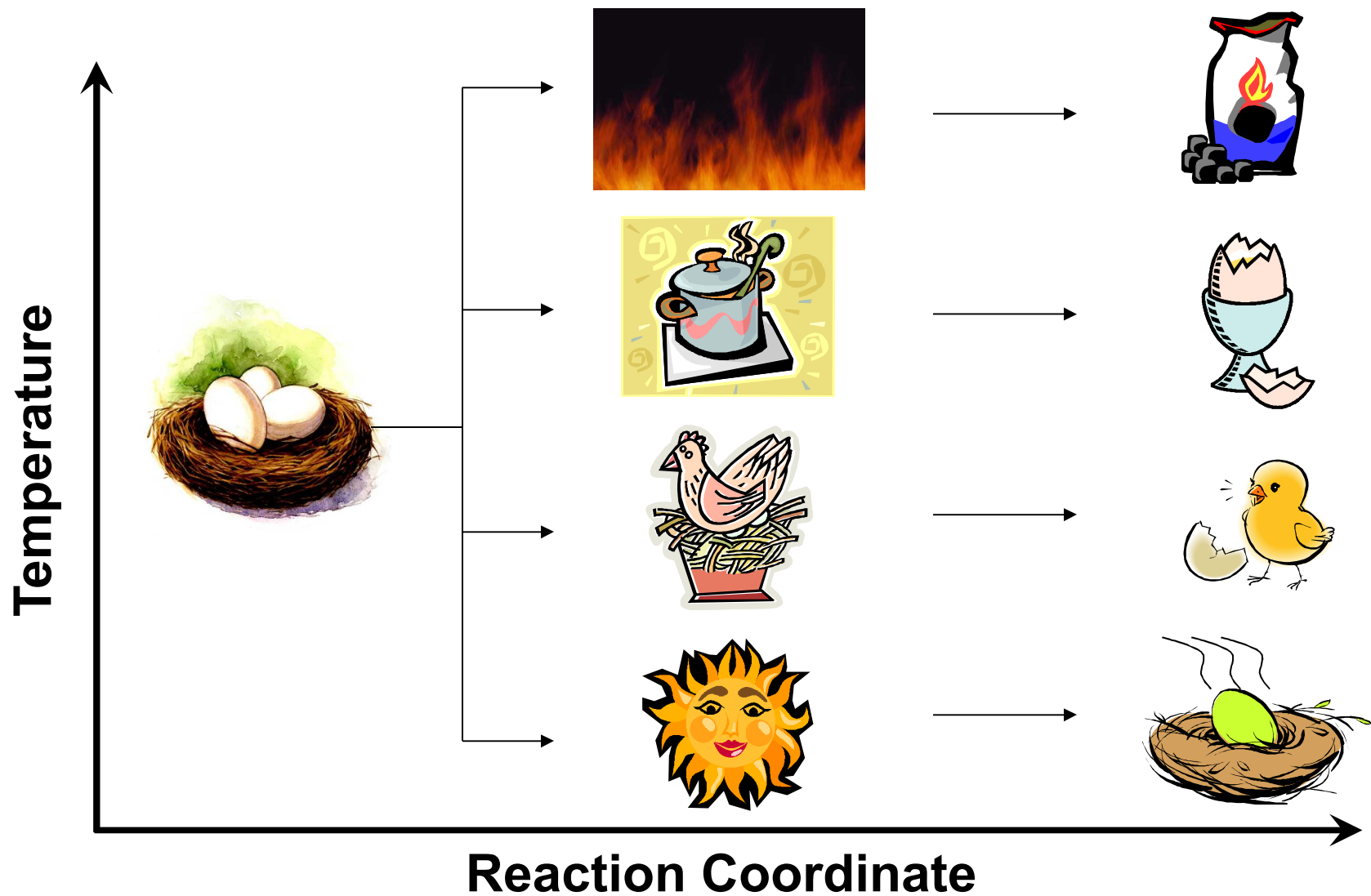
Nylon Oxygen Consumption: Shift Factor Plot



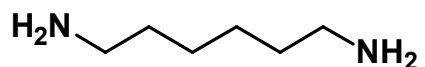
Nylon: Tensile versus Oxygen Consumption



'Accelerated Aging'

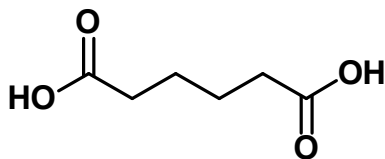


Nylon vs. Kevlar: Aliphatic vs. Aromatic

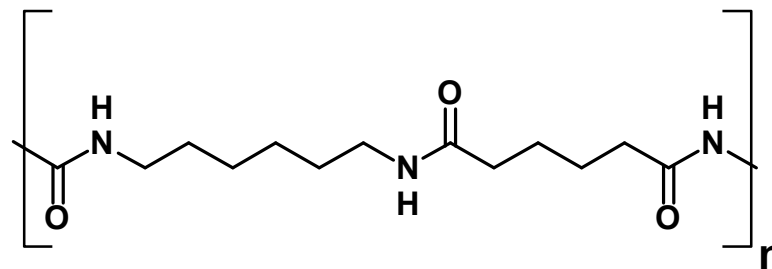
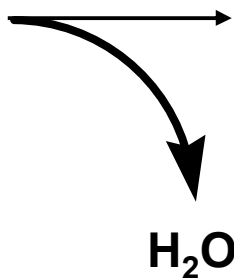


1,6 -Hexanediamine

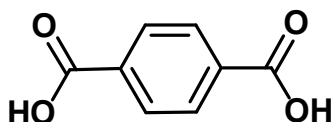
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Adipic Acid

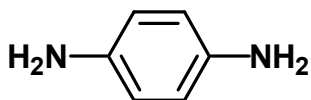


Nylon 6.6

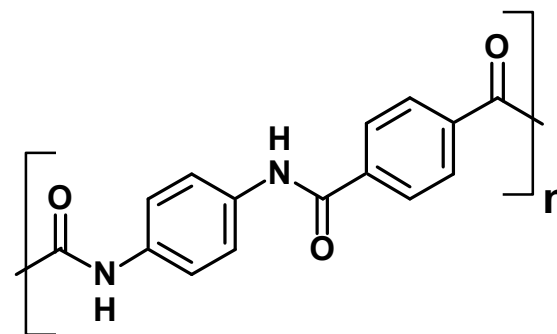
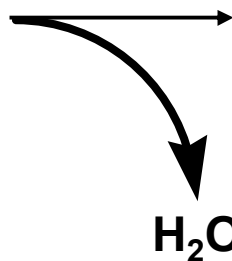


Terephthalic Acid

+



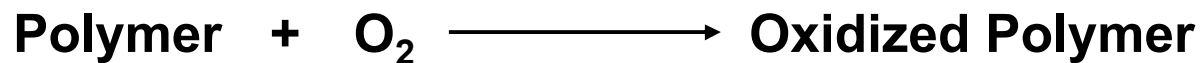
1,4 Phenylenediamine



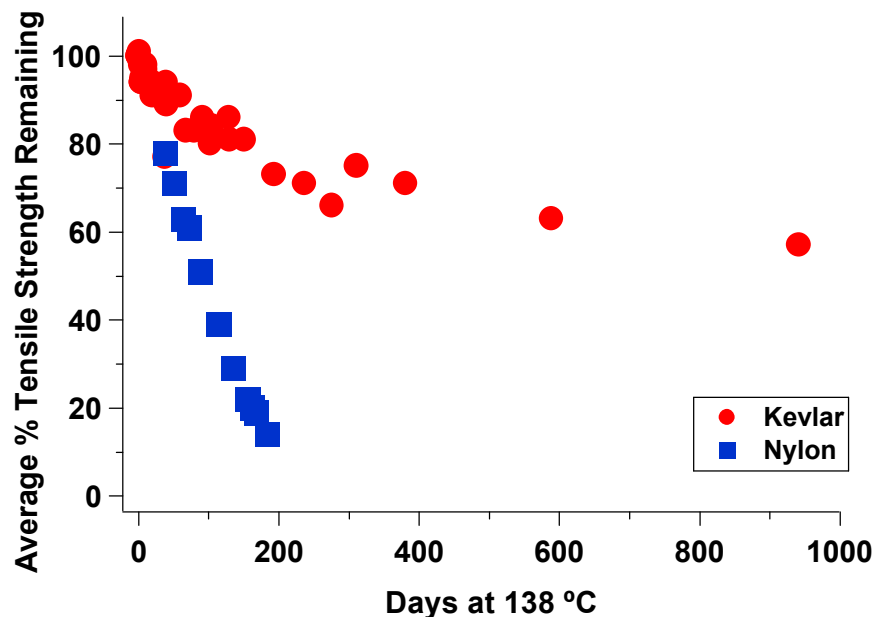
Kevlar



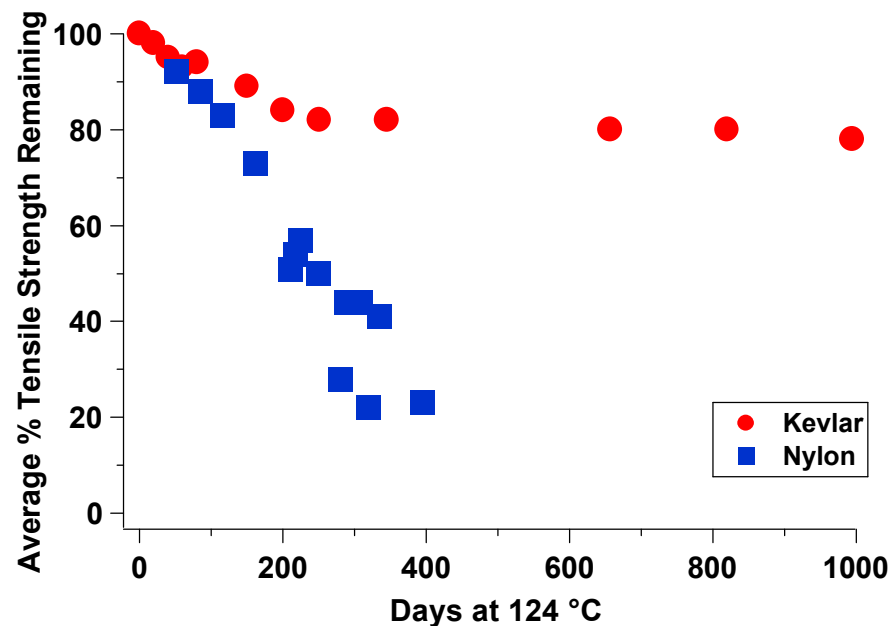
Kevlar vs. Nylon: Thermal-Oxidative Degradation



138 °C



124 °C

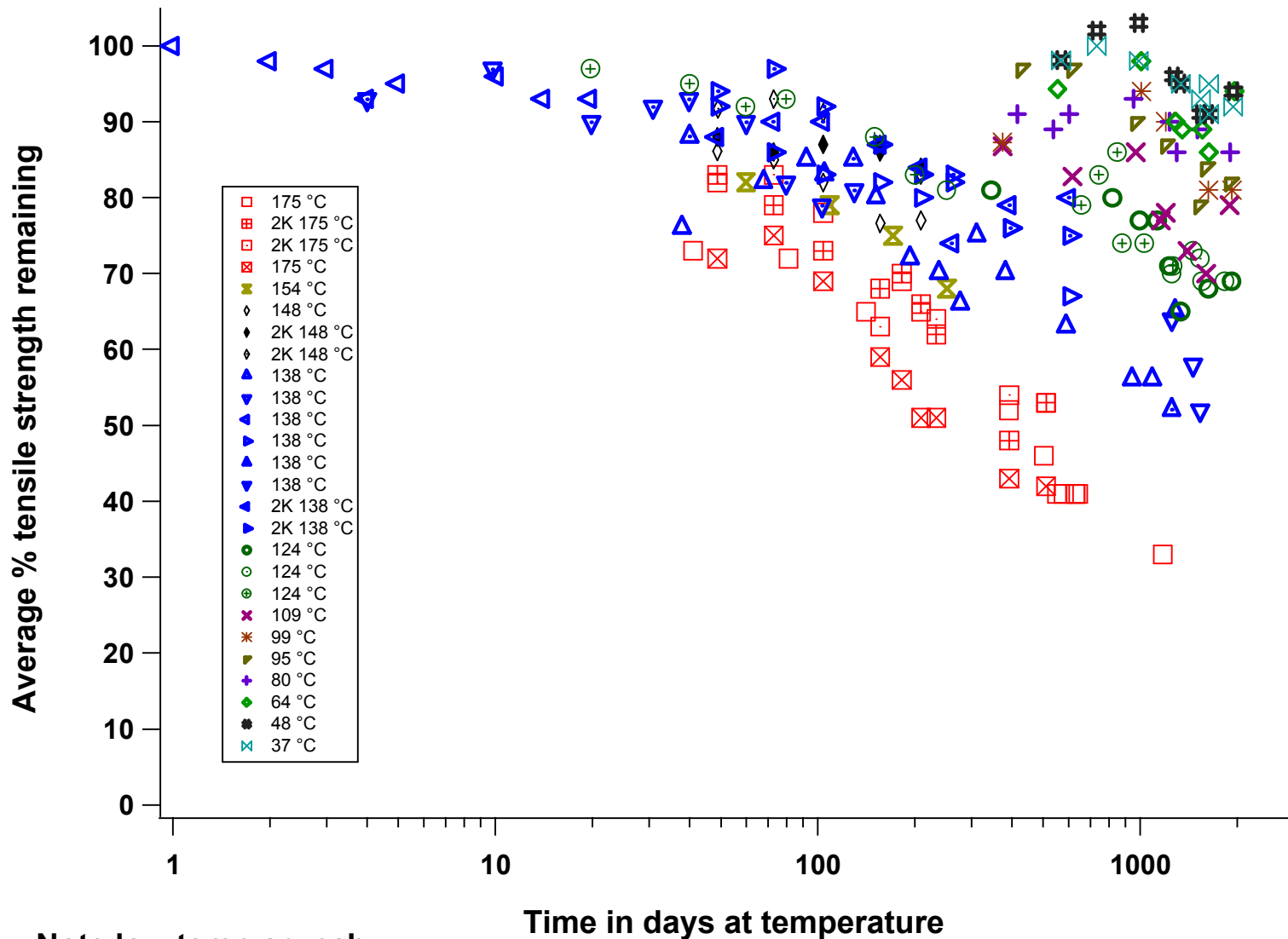


Aromatic vs. Aliphatic backbone

Kevlar more resistant to thermal-oxidative aging than Nylon



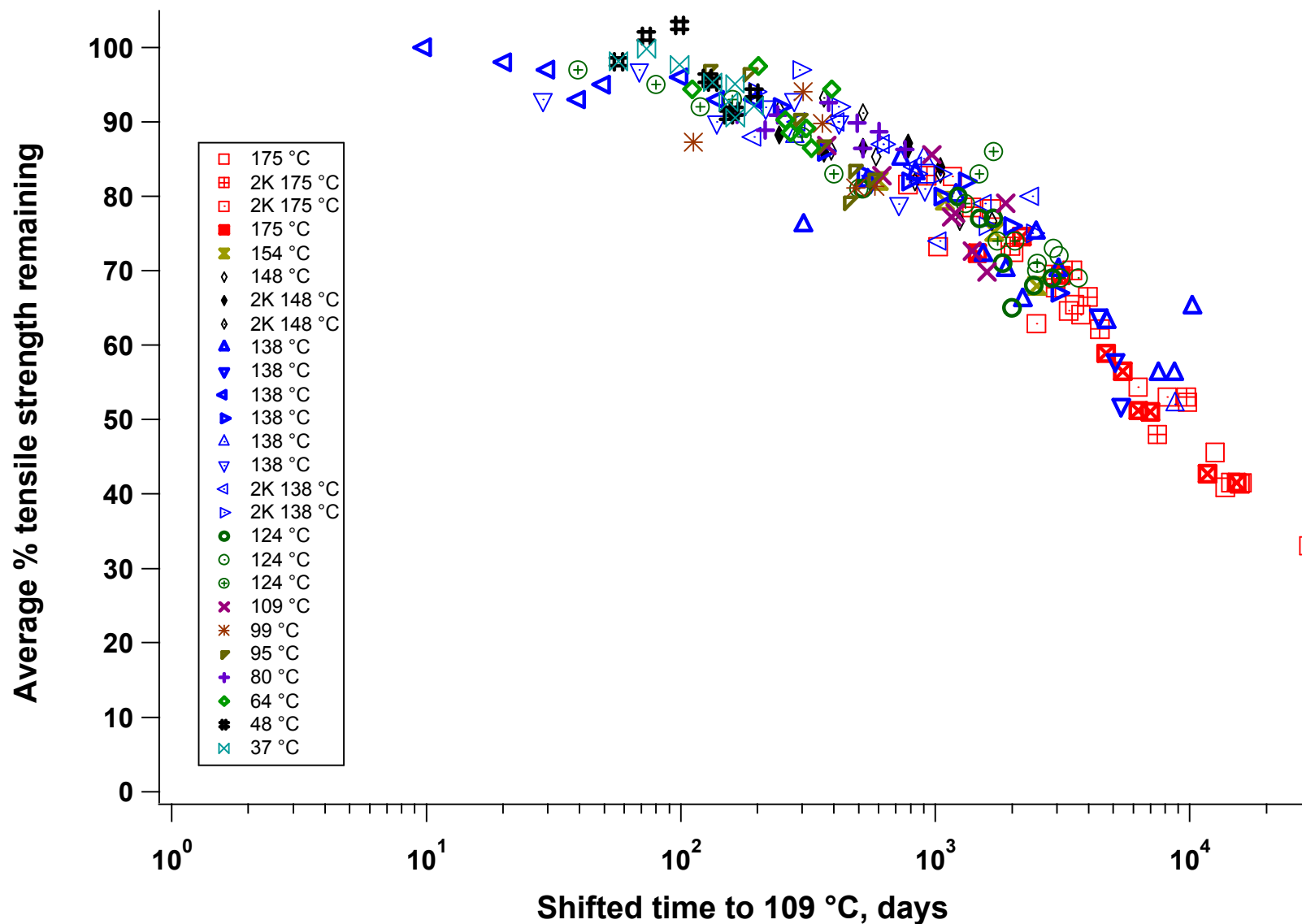
Kevlar Thermal oxidative Aging



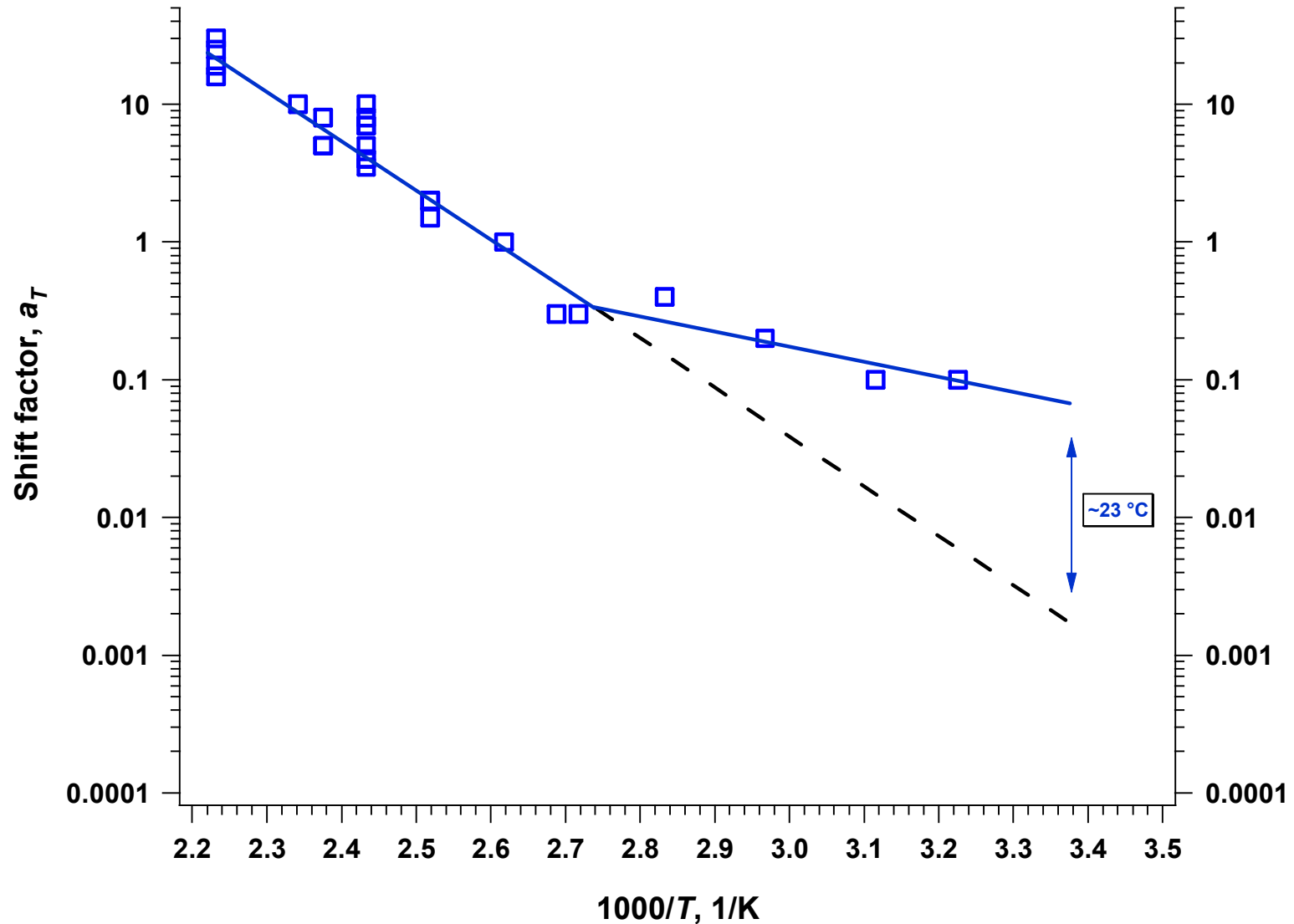
Note low temp crunch....
1K vs. 2K no difference



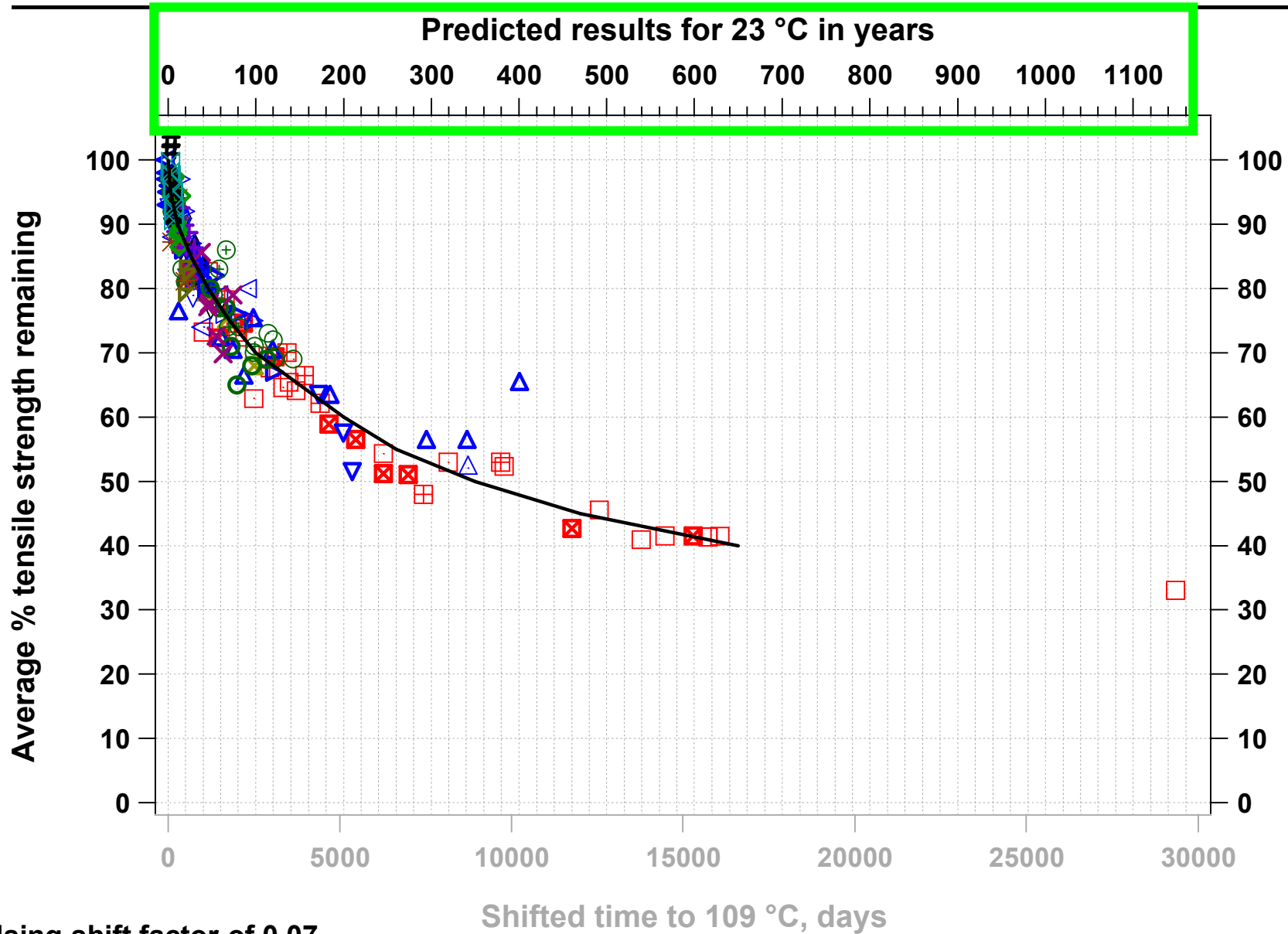
Kevlar Thermal oxidative -Shifted



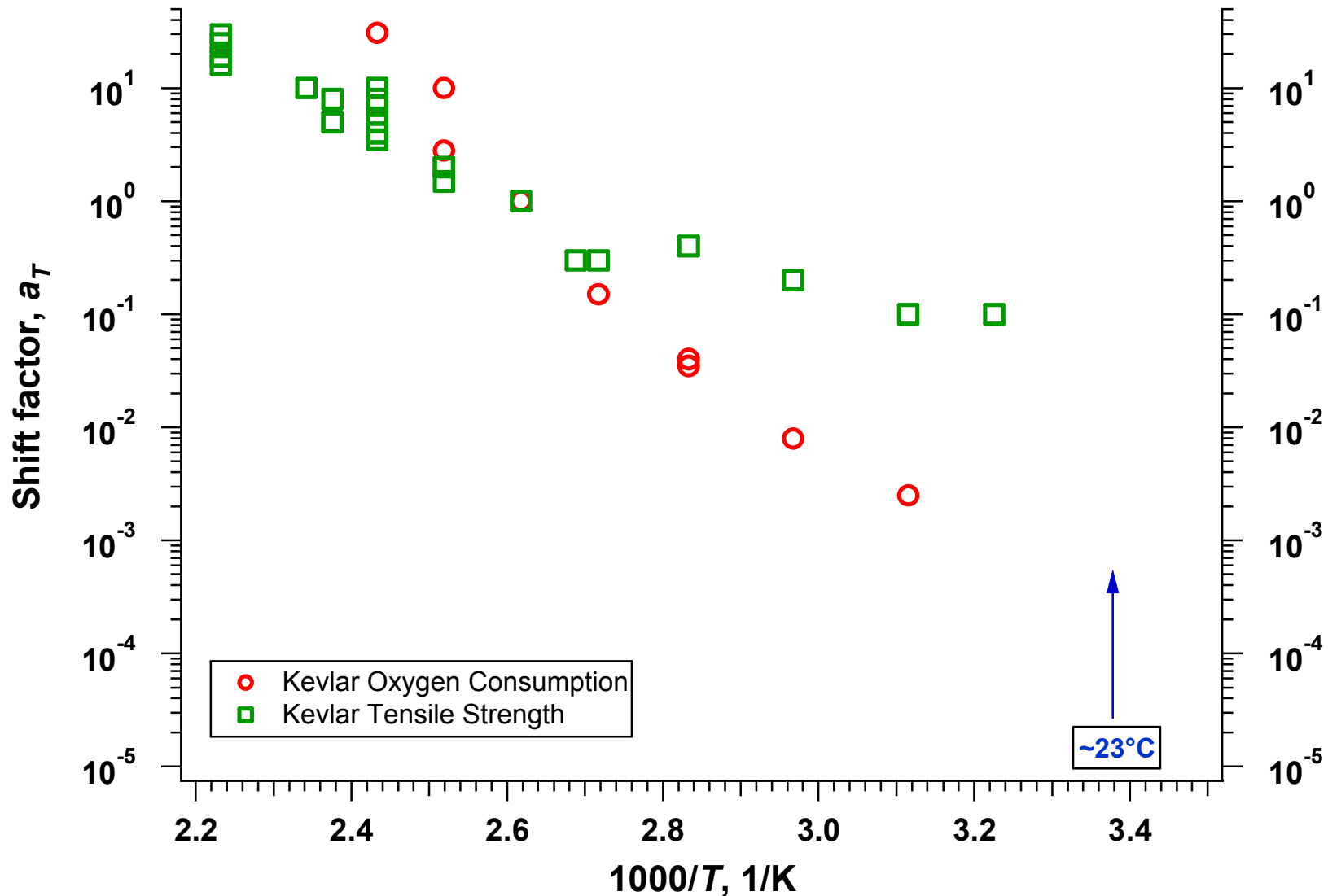
Kevlar Thermal oxidative –Shifted Factor plot



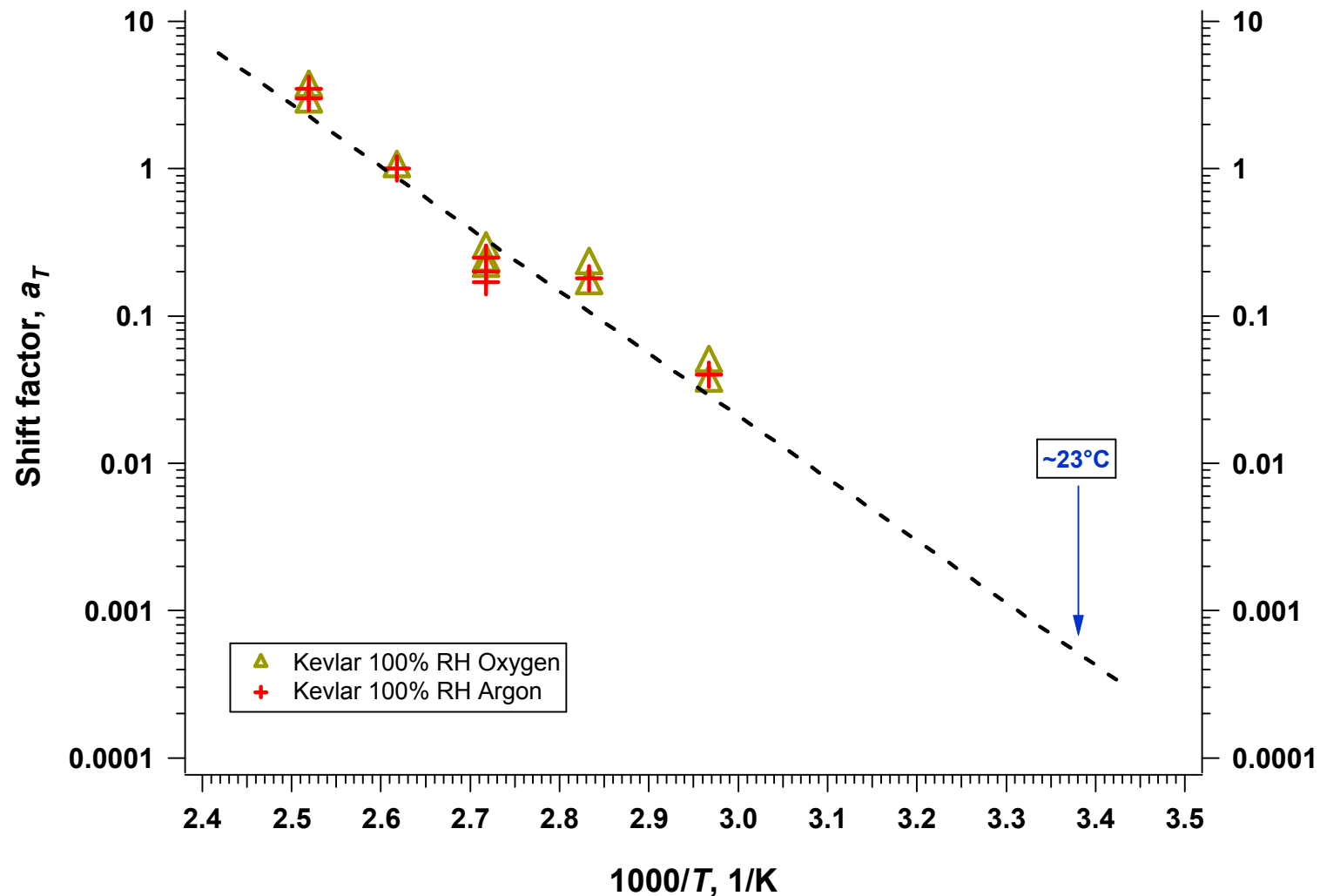
Kevlar Thermal oxidative –RT Prediction



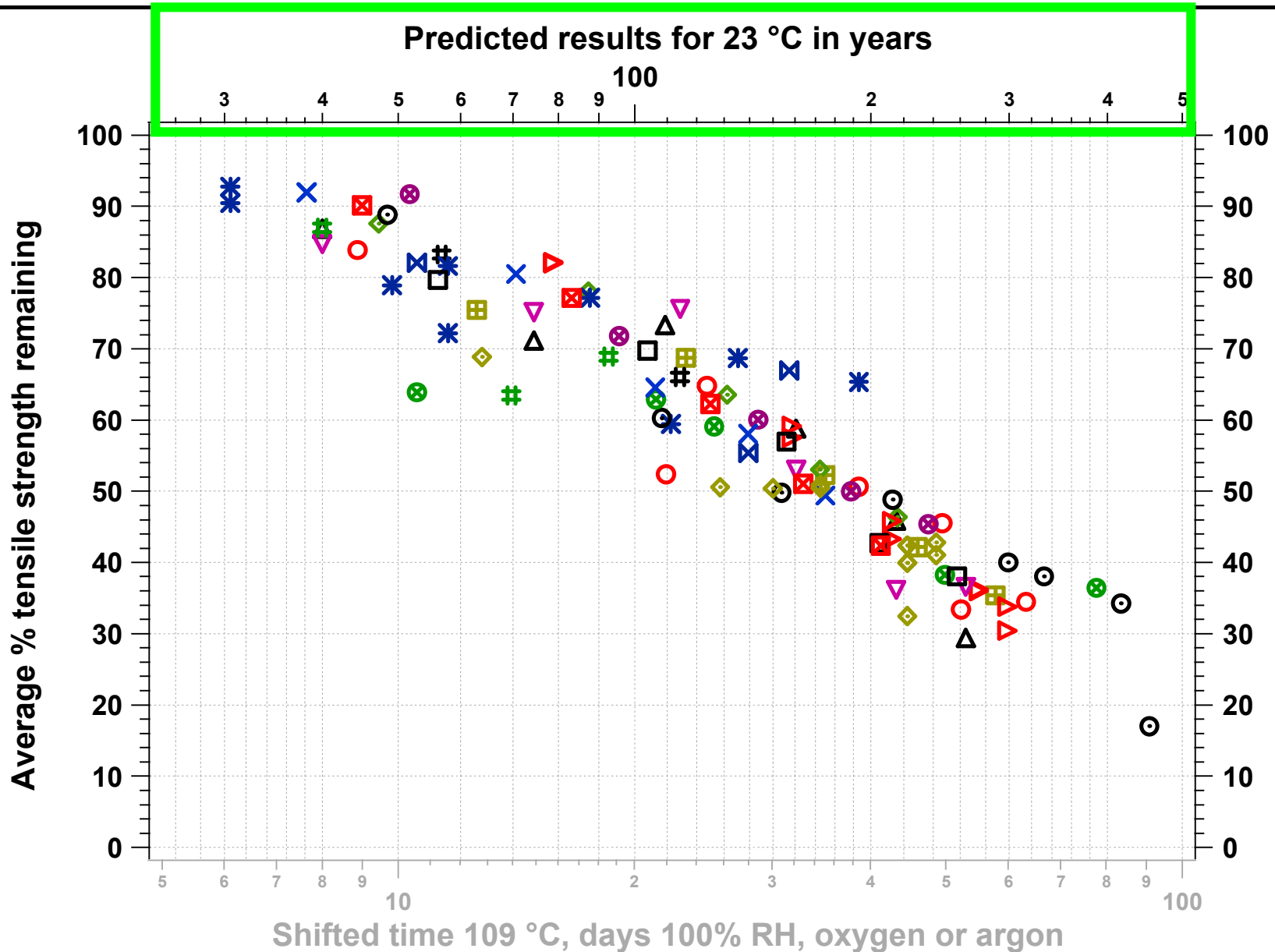
Kevlar Oxygen Consumption vs. Tensile



Kevlar 100% RH Argon and Oxygen –linear fit



Kevlar 100% RH Argon and Oxygen



Conclusions

Nylon linear oxygen consumption

Nylon significant curvature in tensile

**Nylon humidity not relevant at lower temperatures
-High temperature data was deceiving**

**Kevlar has much better aging characteristics but also
has significant curvature in tensile**



Acknowledgements

-DOE \$\$\$\$\$

Sandia is a multiprogram laboratory operated by Sandia Corporation, a Lockheed Martin Company, for the United States Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.



Shift Factors Nylon 100% RH

