

Used Fuel Disposition Campaign

Evaluation of the Frequency for Gas Sampling for the High Burnup Confirmatory Data Project

Christine Stockman
Sandia National Laboratories

UFD Working Group

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**Used
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Authors

Christine T. Stockman (SNL)

Halim A. Alsaed (ENS)

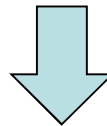
Charles R. Bryan (SNL)

Steven C. Marschman (INL)

John M. Scaglione (ORNL)

5 Questions and a Strategy

- 1. What are the possible gases?**
- 2. Could detection limits impact the gas sampling frequency strategy?**
- 3. Could gases segregate within the cask?**
- 4. What are the degradation mechanisms that could change the gas composition?**
- 5. Can changes in gas composition provide information about these degradation mechanisms that may be useful for informing component performance predictive capabilities?**



Gas Sampling Frequency Strategy

1. What are the possible gases

- Expected - helium backfill at 2.2 atm
 - *233 moles (930 g) in project cask*
- Possible – residual water vapor and/or its decomposition products (H_2 and O_2) at less than 0.25 volume % (suggested limit in NUREG-1536).
 - *Equivalent to 10.5 ml of liquid water.*
- Not expected – residual air, residual water vapor in excess of 0.25 volume %, fission product gases

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Table 2-2. Calculated ^{85}Kr and ^{134}Xe content of the 25 Sister Rods at possible loading and unloading times

Assembly				Rod	2/1/2016	2/1/2027	Stable
ID	Enrichment	Burnup (MWD/MTU)	Cooling Time (years)	ID	^{85}Kr (Ci)	^{85}Kr (Ci)	^{134}Xe (g)
30A	4.55	52.0	5.4	G-9	22.540	11.092	4.255
				K-9	22.496	11.072	4.244
				D-5	22.924	11.284	4.351
				E-14	22.980	11.312	4.365
				P-2	21.432	10.548	3.984
5K7	4.55	53.3	9.9	P-2	16.472	8.108	4.119
				C-5	17.800	8.760	4.562
				K-9	17.152	8.440	4.343
				O-14	16.852	8.296	4.244
6U3	4.45	52.7	11.7	I-7	15.164	7.464	4.360
				M-9	15.312	7.536	4.416
				K-9	15.216	7.488	4.380
				L-8	15.328	7.544	4.422
				O-5	15.928	7.840	4.655
				M-3	15.792	7.772	4.603
3F9	4.25	52.3	14.4	P-16	14.180	6.980	3.993
				N-5	12.620	6.212	4.361
				D-7	12.288	6.048	4.210
3D8	4.20	54.9	17.4	P-2	11.692	5.756	3.943
				E-14	10.992	5.412	4.717
3A1	4.00	50.0	21.4	B-2	9.796	4.824	4.044
				B-16	7.236	3.562	3.884
F35	3.59	57.9	26.9	F-5	7.572	3.726	4.124
				P-17	5.572	2.743	4.620
				K-13	5.572	2.743	4.620

2. Could detection limits impact gas sampling frequency strategy?



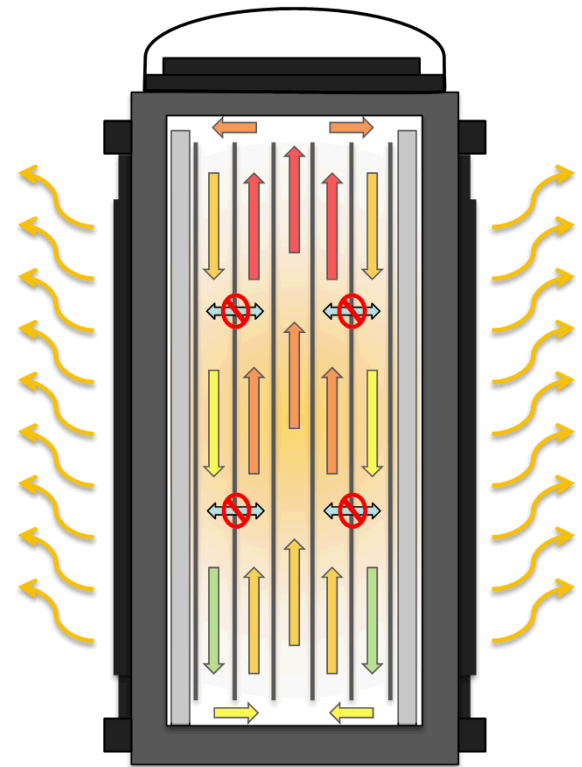
No

- Evaluated ^{85}Kr , ^{134}Xe , H_2O , H_2 and O_2
- Modern detection efficiency is orders of magnitude better than needed for all of these gases
 - *^{134}Xe can be measured using modern gas mass spectroscopy with a detection efficiency as low as 10 ppb. Release of 1% of ^{134}Xe from a single rod would result in a mass concentration of 46 ppm.*
 - *^{85}Kr can be measured using gamma spectroscopy, mass spectroscopy, and liquid scintillation with a detection efficiency on the range of one pCi/cm³. Release of 1% of ^{85}Kr from a single rod would result in a concentration of 22 nCi/cm³.*
 - *H_2O detection limits are in the ppm-range. 0.25% water vapor equates to 11 parts per thousand.*

3. Could gas segregate due to gravitational settling or thermal diffusion?

➤ **No**

- Minimal gravitational settling may occur immediately after gas release, but a compositional gradient cannot be maintained even in a static gas column, let alone in a cask with a convecting gas column.
- Significant compositional gradients due to thermal diffusion require comparable initial concentrations of the two gas components, large thermal gradients, and a geometry that allows gas diffusion between the rising and falling arms of the convection cells. None of these exist in the project cask.



4. What are the degradation mechanisms that could change the gas composition?

- Corrosion of internal components can consume H_2O and O_2
- Presence of water logged components can increase H_2O and O_2 concentration
- Breach of cladding can release fission gases - not expected
 - *Early degradation mechanisms: DHC, fission product attack, creep, oxidation*
 - *Long-term degradation mechanisms: fuel restructuring/swelling, hydride embrittlement and reorientation*

5. Can changes in gas composition provide information about these degradation mechanisms that may be useful for informing component performance predictive capabilities?
- Can provide total consumption rates of H_2O , H_2 and O_2
 - Cannot provide the fraction that is consumed by each component or the amount of degradation experienced by each component
 - Can indicate cladding breach and provide changes in total concentration of fission gases.
 - Cannot indicate number of failed rods or mechanism(s) that caused breach

Gas Sampling Frequency Strategy

■ Adaptive

- based on measured temperatures, initial (in-building) sampling, and one early sampling (within a year).

■ If conditions are as expected

- then the sampling frequency is not driven by a predictable technical basis, and an operationally based interval is recommended.

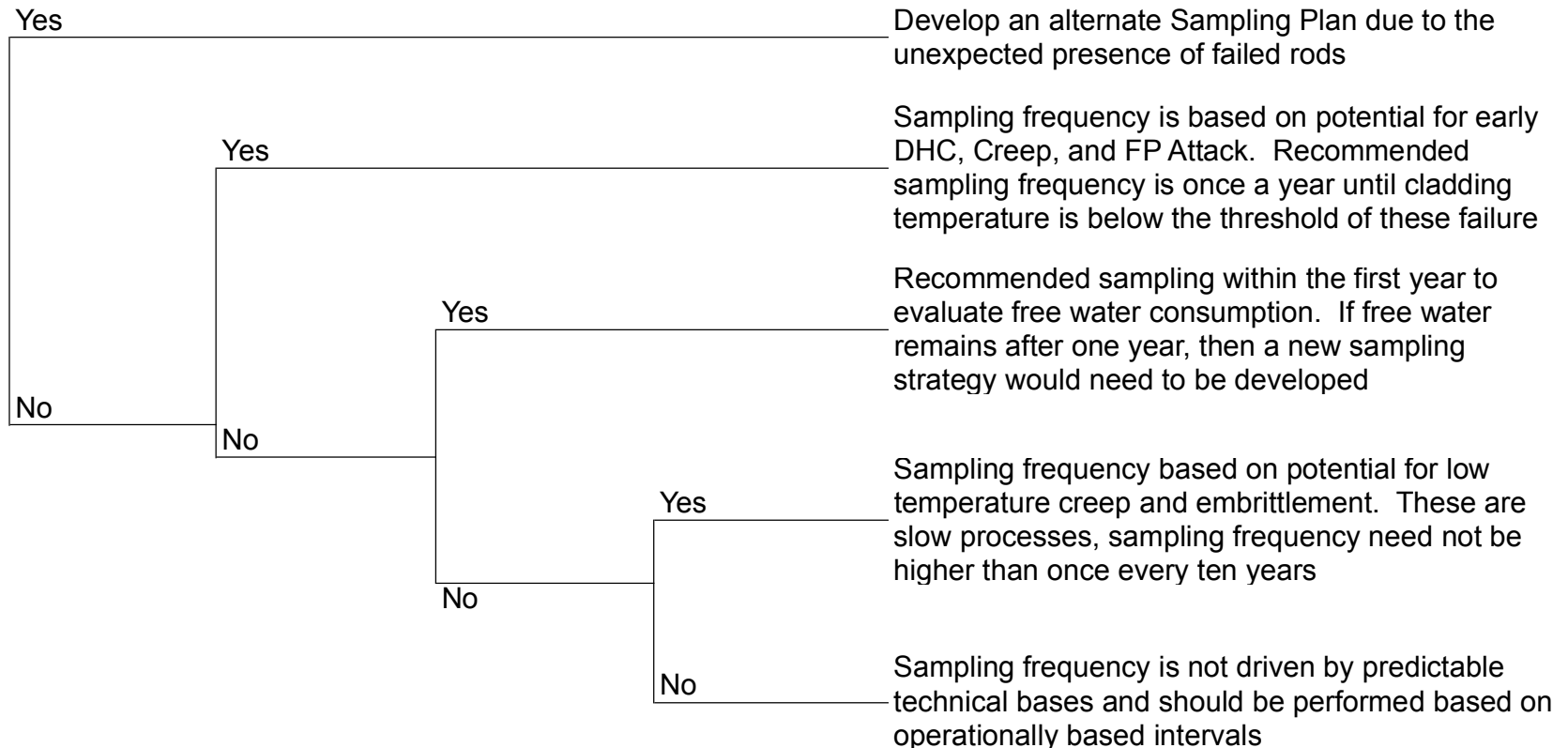
■ Sampling strategy should be reevaluated if:

- Fission product gases are detected
- Cladding temperatures are above 400° C
- ≥ 0.25 volume percent water and/or oxygen is present at 1 year sampling
- Any off-normal, accident, or natural phenomena event occurs

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Gas Sampling Frequency Strategy

Initial sampling indicates presence of fission gases	Cladding temperature is significantly higher than expected	Initial sampling indicates presence of H ₂ O, H ₂ , or O ₂	Cladding is below ductile-to brittle transition temperature (DBTT)	Recommended Sampling Frequency
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■ **Sample the cask a minimum of four times**

- Shortly after loading
- At about one year
- Approximately three years after the one year sampling
- At ten years prior to shipping the cask

■ **If there are any indications of a fuel rod breach from the first sampling after loading, more samples may be taken while the cask is in the North Anna decontamination building.**