

# EVALUATION OF ELASTOMERS FOR GEOTHERMAL WELL APPLICATIONS

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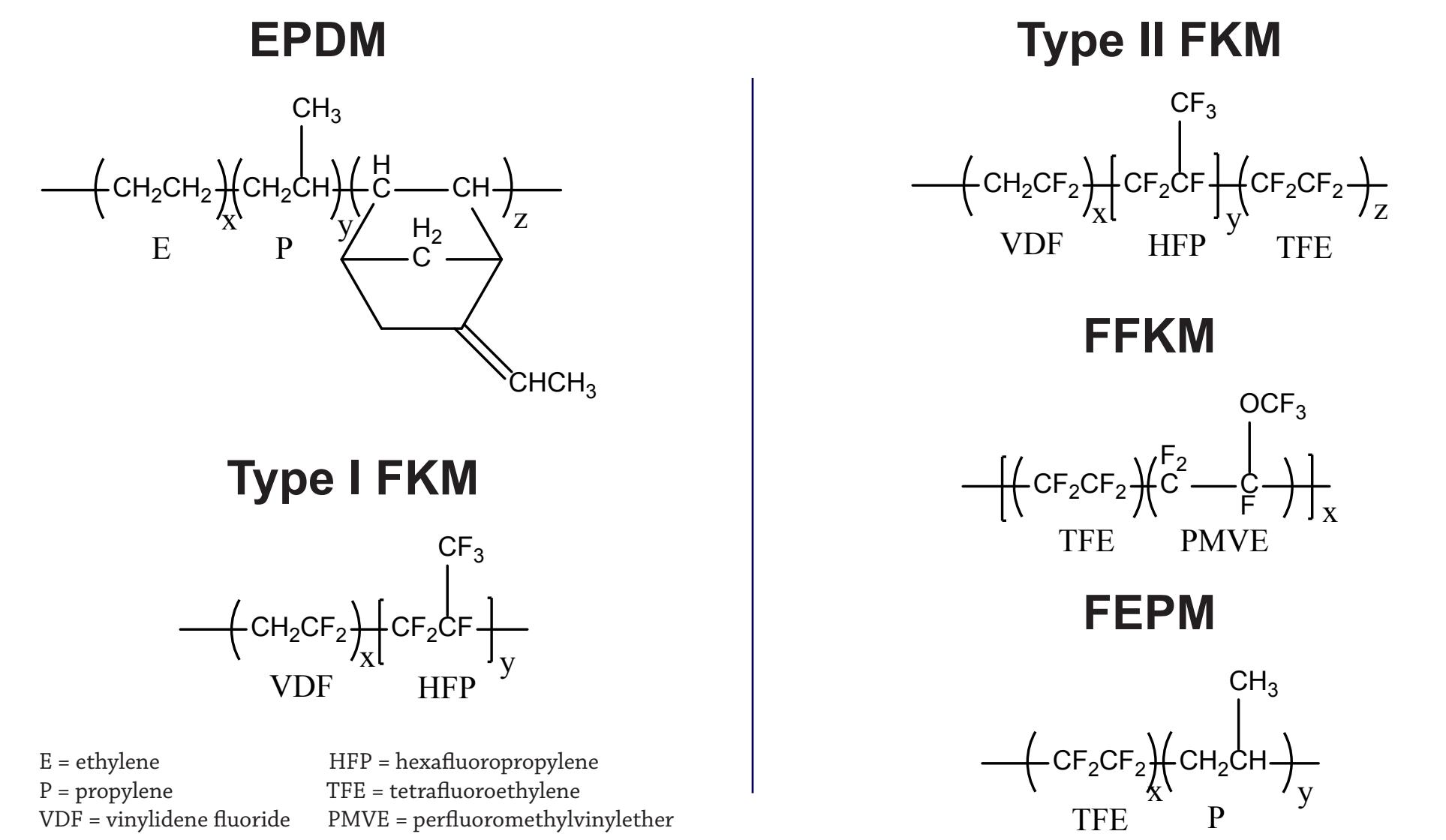
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## 1. INTRODUCTION

The emphasis on sustainable energy has brought with it challenges associated with materials performance. In particular, geothermal wells push the boundaries of elastomer stability due to harsh environmental conditions, where temperatures around 300°C and pressures of 5000 psi or greater are not uncommon. Additionally, well brines and drilling fluids subject these materials to very severe chemical environments, which also impacts elastomer degradation and stability. The aim of this study is to understand how commercially available elastomers perform under geothermal well-like conditions and make recommendations to the community based on these results. This poster highlights the degradation chemistry and mechanical performance of several elastomers after aging in several different environments.

## 2. MATERIALS



## 3. EXPERIMENTAL METHODS

**Aging in Steam, N<sub>2</sub>**  
5 cycles of steam/cooling in N<sub>2</sub>.  
One cycle = 300°C steam for 24 hrs, quench to 25 °C at 1200 psi.

**Aging in Steam, Air**  
5 cycles of steam/cooling in air.  
One cycle = 300°C steam for 24 hrs, quench to 25 °C at 1200 psi.

**Aging in Drilling Fluid:**  
300°C, 7 days, ~1200 psi submerged in a drilling fluid mimic with pH 9-10.  
Drilling fluid composition:

| Major Components      | Percent | Major Components | Compounds                        | Percent |
|-----------------------|---------|------------------|----------------------------------|---------|
| Water                 | 74-83   | Cl <sup>-</sup>  | NaCl                             | 13.5    |
| Barite                | 10-15   | Na <sup>+</sup>  | NaCl                             | 6       |
| Bentonite             | 5-7     | Ca <sup>2+</sup> | CaCl <sub>2</sub>                | 2       |
| Caustic soda          | 0.3     | K <sup>+</sup>   | KCl                              | 1.5     |
| Soda ash              | 1       | Mg <sup>2+</sup> | MgCl <sub>2</sub>                | 0.9     |
| Polyanionic cellulose | 0.3-1.2 | PPM              |                                  |         |
| Xanthan gum           | 0.3-0.5 | CO <sub>2</sub>  | NaHCO <sub>3</sub>               | 15,000  |
| Starch                | 0.5-1   | Fe <sup>2+</sup> | FeCl <sub>2</sub>                | 1000    |
|                       |         | Mn <sup>2+</sup> | MnCl <sub>2</sub>                | 930     |
|                       |         | Li <sup>+</sup>  | LiCl                             | 410     |
|                       |         | Zn <sup>2+</sup> | ZnCl <sub>2</sub>                | 370     |
|                       |         | B <sup>3+</sup>  | H <sub>3</sub> BO <sub>3</sub>   | 330     |
|                       |         | Si <sup>4+</sup> | Na <sub>2</sub> SiO <sub>3</sub> | 250     |
|                       |         | Ba <sup>2+</sup> | BaCl <sub>2</sub>                | 130     |
|                       |         | H <sub>2</sub> S | H <sub>2</sub> SO <sub>4</sub>   | 70      |

**Thermal Cycle Aging:**  
24 hours at 300°C with water quenching to 25°C and hold for 5 hours - repeated five times.

**Thermogravimetric Analysis (TGA):**  
Sample sizes ranged from approximately 10 - 50 mg. Ramp 20°C/min to 700°C on a TA Instruments TGA Q50 V20.10.

**Fourier Transform Infrared Spectroscopy (FTIR):**  
Attenuated Total Reflection FTIR was used to identify any potential chemical changes in the elastomers after exposure to the different aging environments.

**Modulus Profile Testing:**  
Modulus profiles were taken using a home-built instrument. The machine operates by scanning the surface with a parabolic tip at user-defined intervals (0.2 mm) and using displacement from a known force applied to each point on sample to calculate modulus. Samples are cross-sectioned and embedded in epoxy prior to running the experiment.

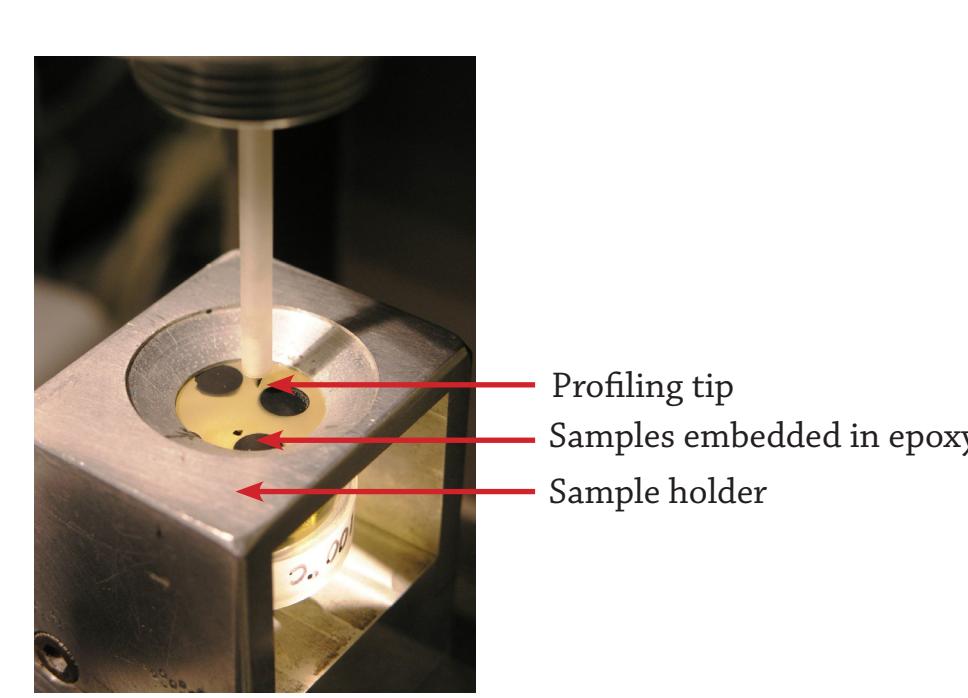


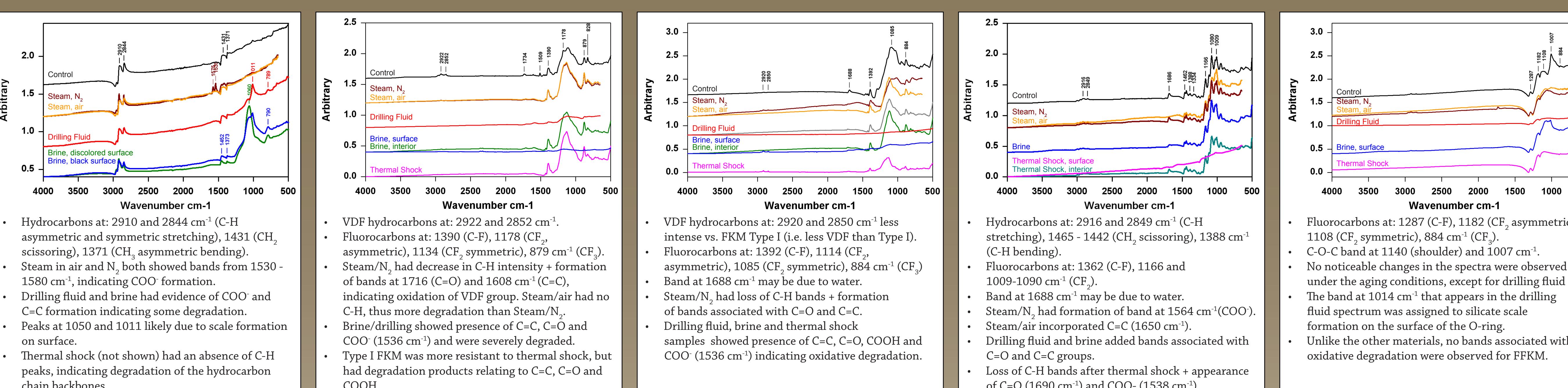
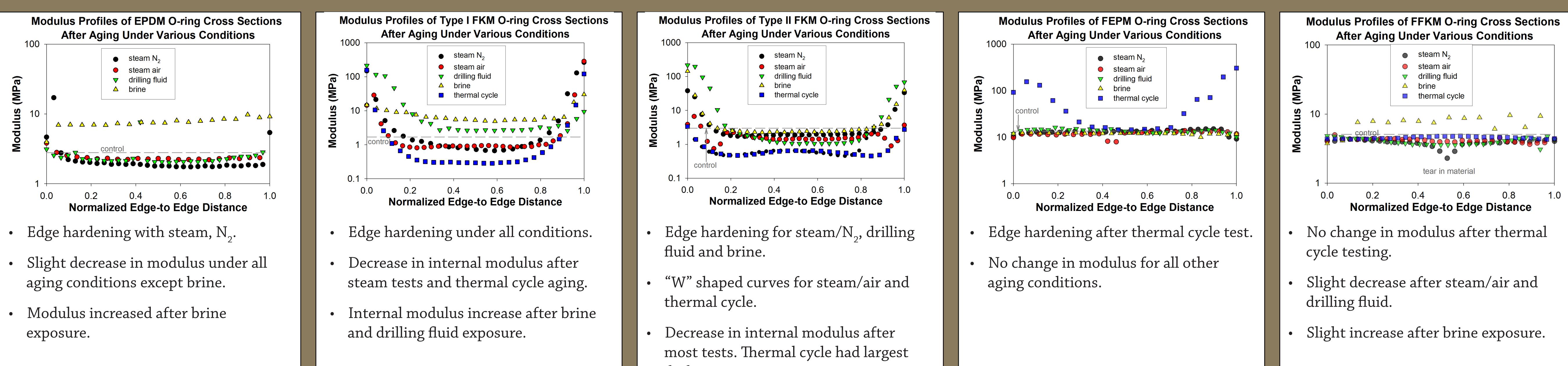
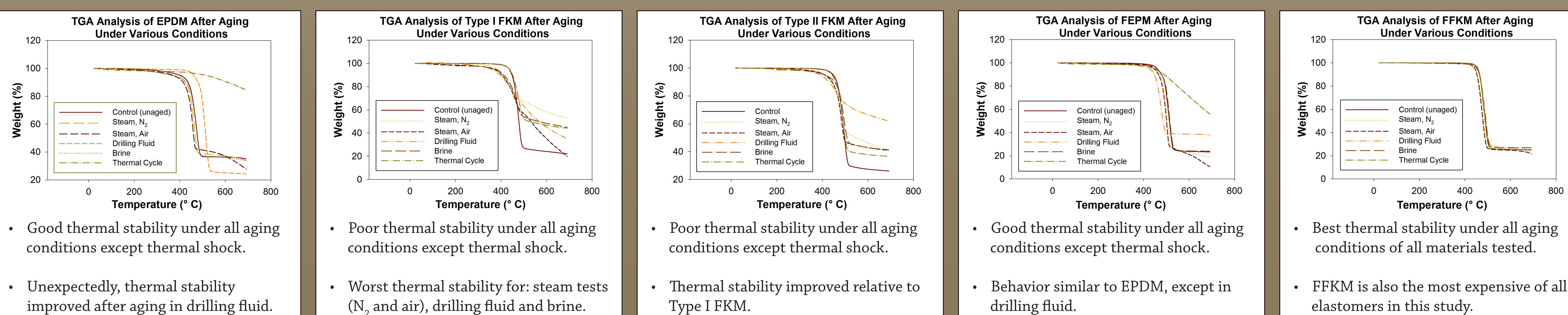
Figure 1. Close-up view of modulus profiler set-up.



Figure 2. Top: top-down, Bottom: side view of cross-sectioned o-rings embedded in epoxy.



## 4. RESULTS & DISCUSSION



## 5. CONCLUSIONS

EPDM is a low-cost alternative to fluoropolymers for geothermal well applications under all conditions but thermal shock. Additionally, the material has an unexpected improvement in thermal resistance in drilling fluid. However, products associated with oxidative degradation were observed for all tests. This material is the lowest cost of all materials in the study.

Type I FKM showed edge hardening and evidence of oxidative degradation in all of the conditions studied. The edge hardening and resulting flaking behavior could cause a reduction in the sealing force, making this material unsuitable for o-ring applications under the conditions studied. However, the material may be adequate for use at lower use temperatures.

Type II FKM was also observed to have edge hardening and oxidative degradation in all of the conditions studied, but to a lesser extent than Type I FKM. This is likely due to less VDF in Type II, as VDF undergoes preferential degradation to the other polymer segments. As with Type I FKM, this polymer may be adequate for use at lower temperatures.

FEPM performed similarly to EPDM. FEPM remained relatively stable after steam tests and aging in drilling fluid and brine. Oxidative degradation products were noted under all aging conditions, and the thermal shock sample had additional evidence of degradation associated with the propylene group (loss of C-H bands in FTIR spectrum).

FFKM remained the most stable under all aging conditions in this study, as evidenced by a lack of change in modulus, thermal stability, and FTIR spectra. However, the scale formation observed in the brine could lead to mechanical instability after further aging. While this is the best choice of all materials studied, it is also the most expensive.