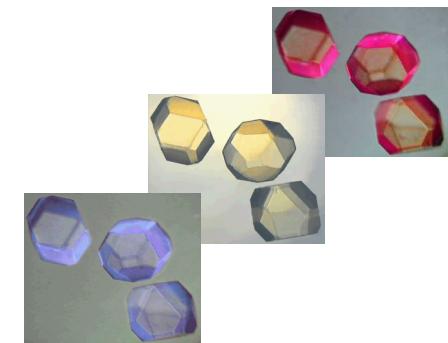
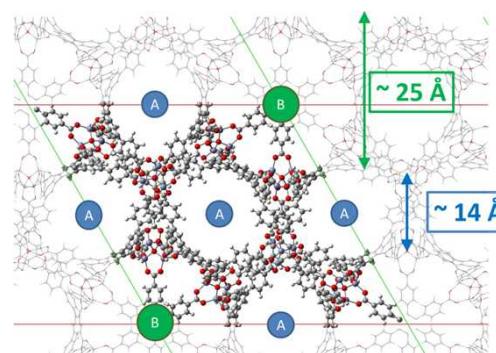
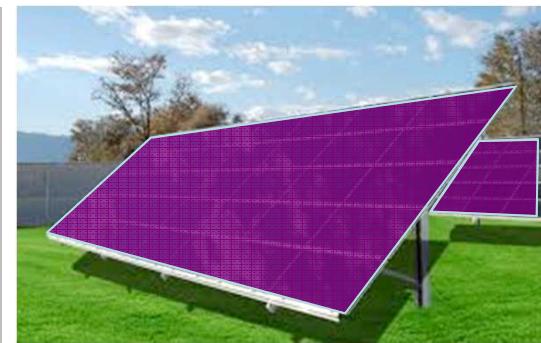


*Exceptional service in the national interest*



## Light-Harvesting Metal-Organic Frameworks: Long-Range Ordered Structures for Next Generation Dye Sensitized Solar Cells

**Erik D. Spoerke**

Leo Small, Jill Wheeler, Tim Lambert  
Michael Foster, Vitalie Stavila, Kirsty  
Leong, and Mark D. Allendorf

Symposium R: Photoactive Nanoparticles and  
Nanostructures

Materials Research Society Spring Meeting  
San Francisco, CA  
April 6-10, 2015

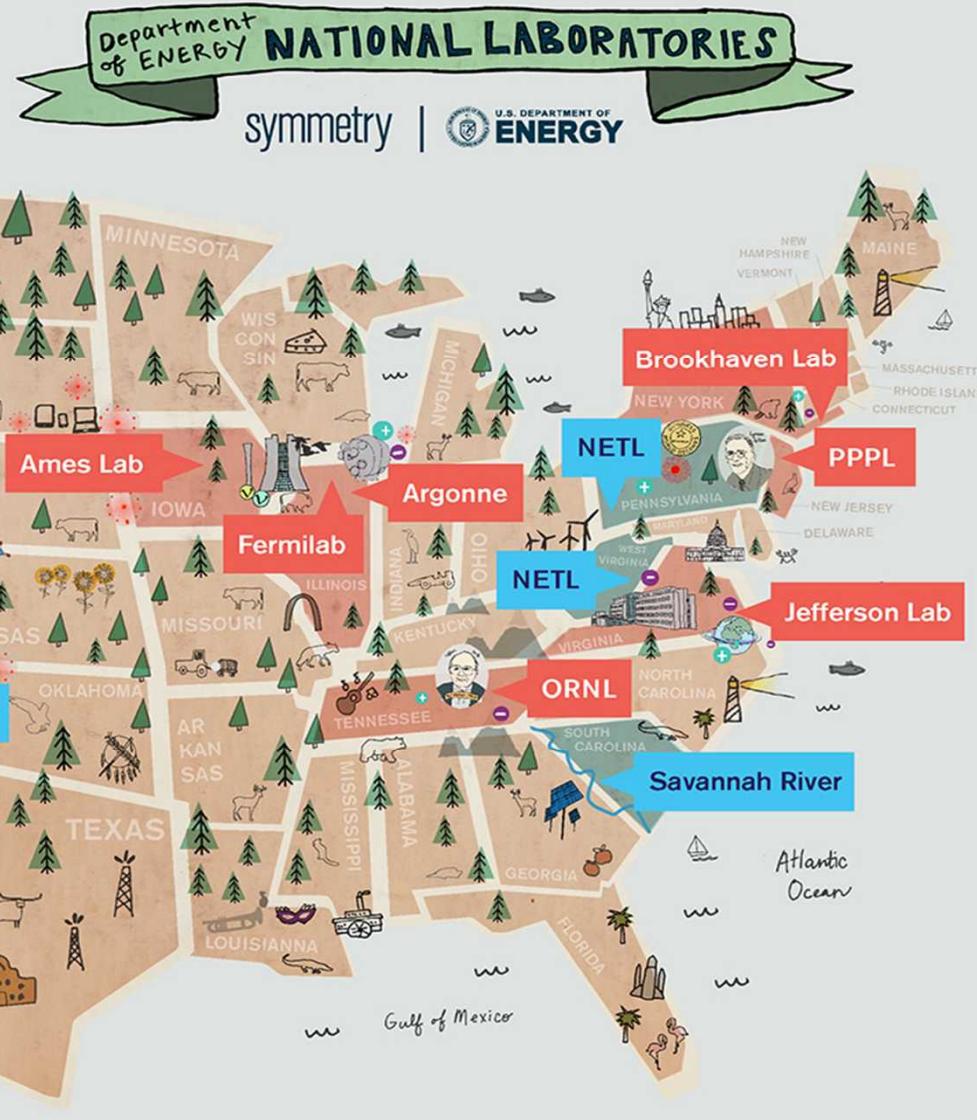


Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

*Exceptional service in the national interest*



Sandia  
National  
Laboratories



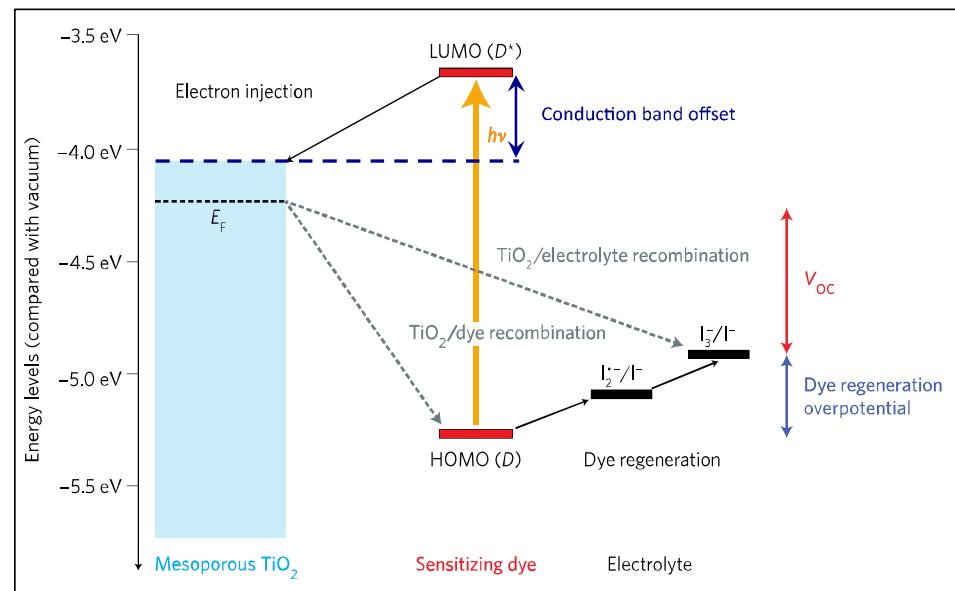
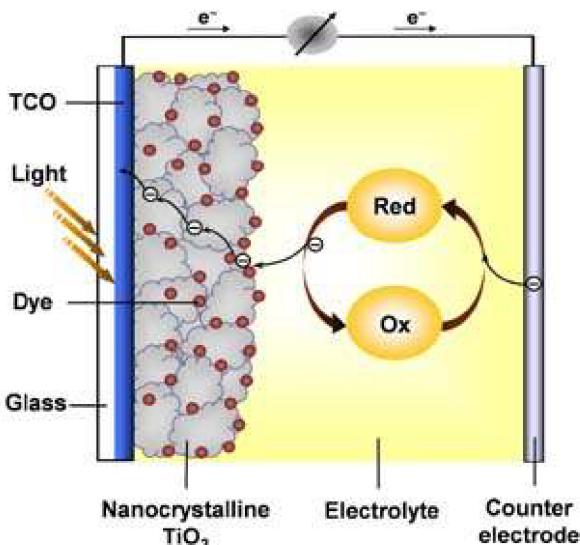
# Technical Challenge

## Dye-Sensitized Solar Cells

(O'Regan, B. & Grätzel, M. (1991). *Nature*, **353**, pp 737.)

World record is >15% efficiency...but there are still some critical challenges:

- Limited light harvesting
  - Spectral range
  - Dye concentration (\*without dye aggregation)
- Carrier lifetimes
- Band offset overpotentials
- Stability/Reliability



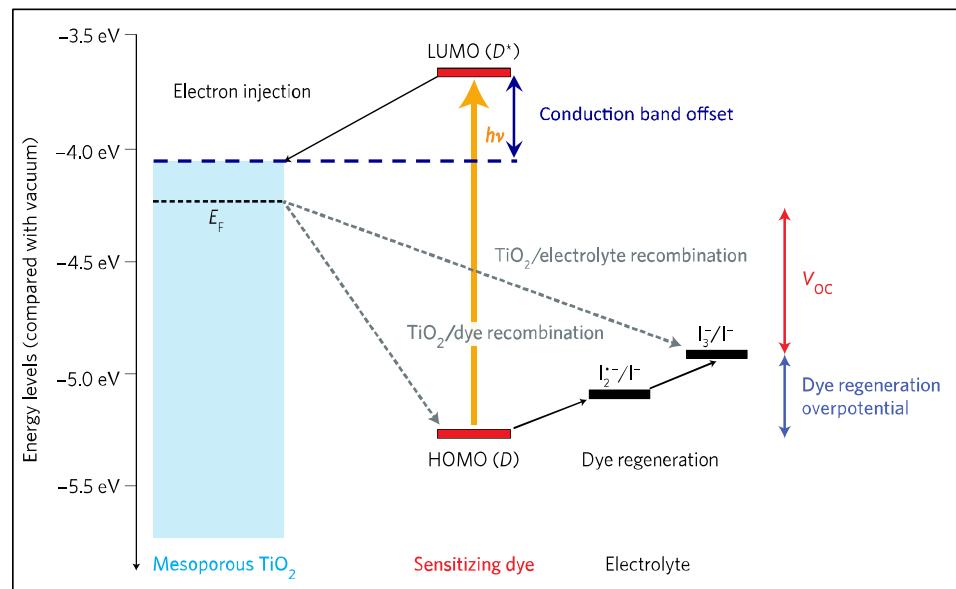
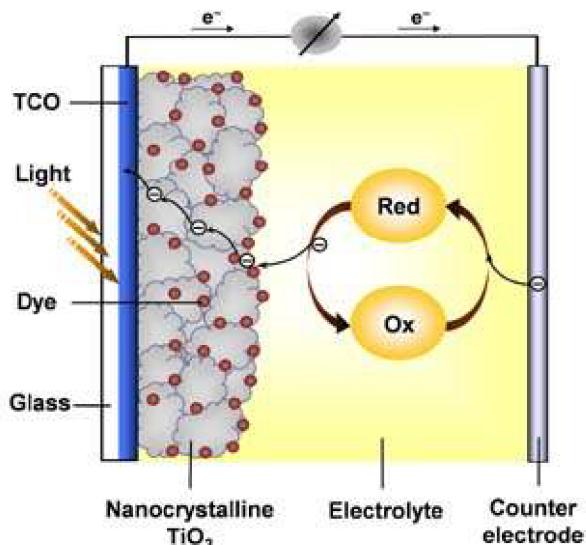
Adapted from B. Hardin, et al. (2012) *Nature Photonics*, **6**, 162-169.

# Technical Challenge

## Dye-Sensitized Solar Cells

(O'Regan, B. & Grätzel, M. (1991). *Nature*, **353**, pp 737.)

*Can we use Metal-Organic Frameworks (MOFs) to address these challenges?*

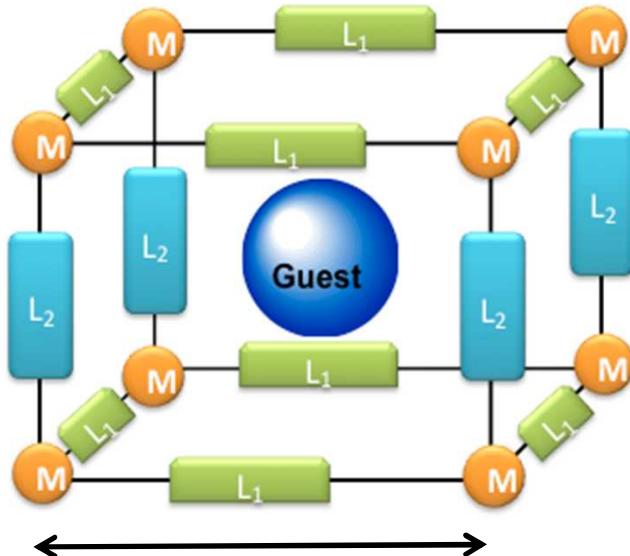


Adapted from B. Hardin, et al. (2012) *Nature Photonics*, **6**, 162-169.

# What are Metal-Organic Frameworks?

## Metal-Organic Frameworks (MOFs)

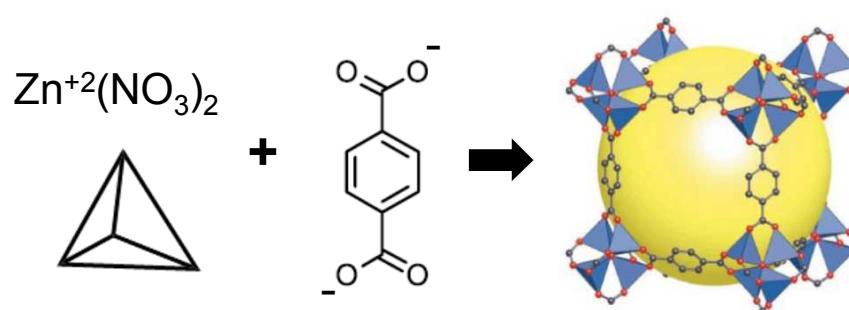
A subset of coordination polymers



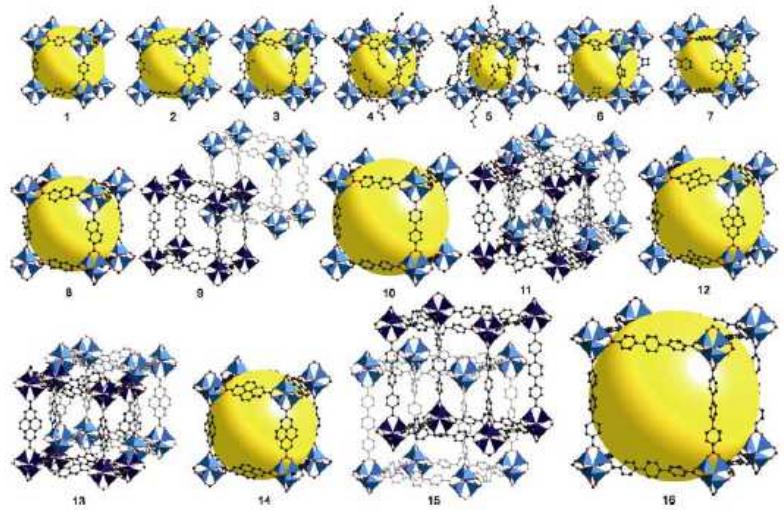
Crystalline MOF structures are composed of metal nodes (M), linkers ( $L_1$ ) and pillars ( $L_2$ ).

The nanoporous character of the MOF allows incorporation of molecular guests, organized on the nanoscale.

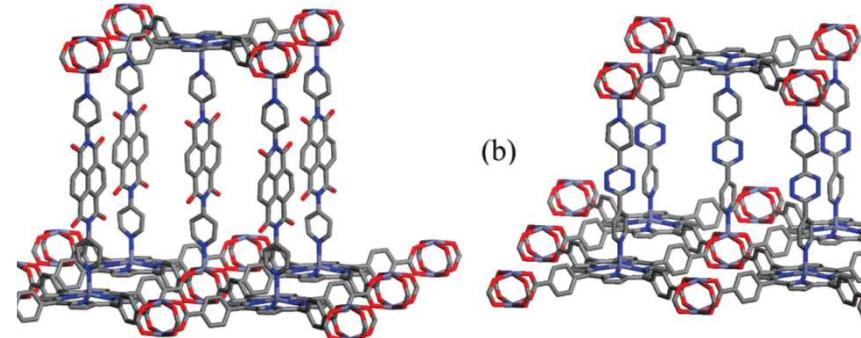
This chemically “modular” system allows for tuning of the structure, properties, and function of these materials.



# MOF Building Block Chemistry Determines Crystal Structure

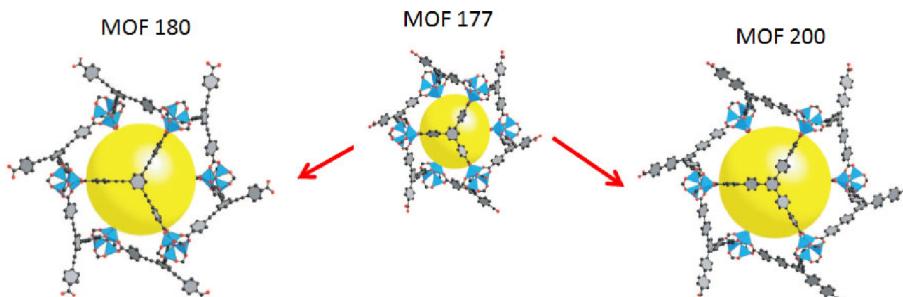


<http://yaghi.berkeley.edu/research-MOF.html>



Side view of PPF-18

Chung *et al.* *Crystal Growth & Design*, Vol. 9, No. 7, 2009

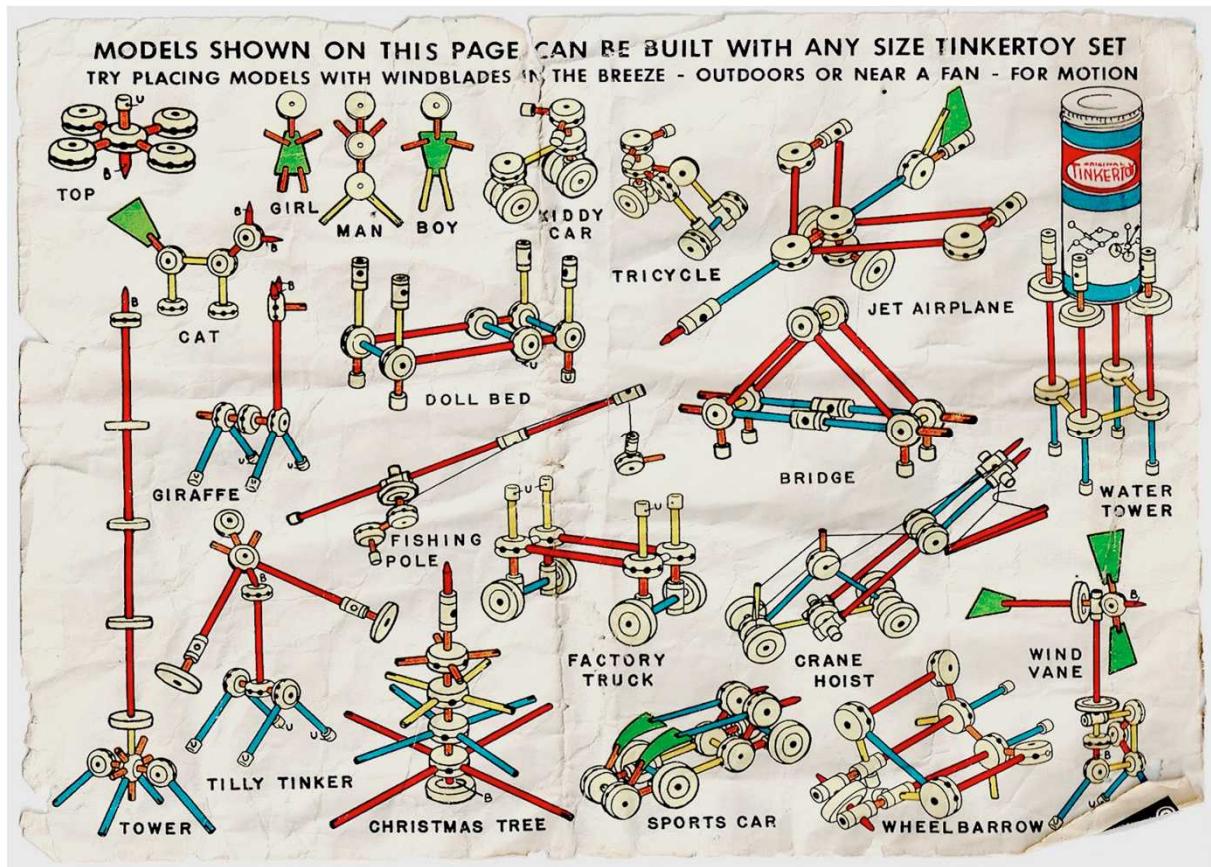


<http://www.cchem.berkeley.edu/molsim/teaching/fall2011/CCS/Group7/structure.htm>

Varying the “modular” composition of MOFs allows for tremendous flexibility of structure and function.

# MOFs: Supramolecular “Tinker Toys”

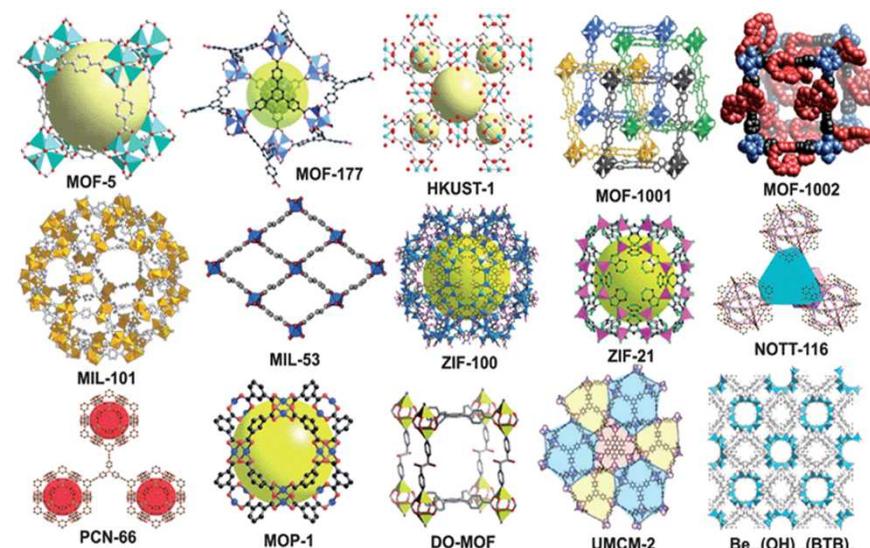
*MOFs are modular materials, diverse in form and function!*



# MOFs Properties

MOFs offer a number of valuable properties, often in unique combinations...

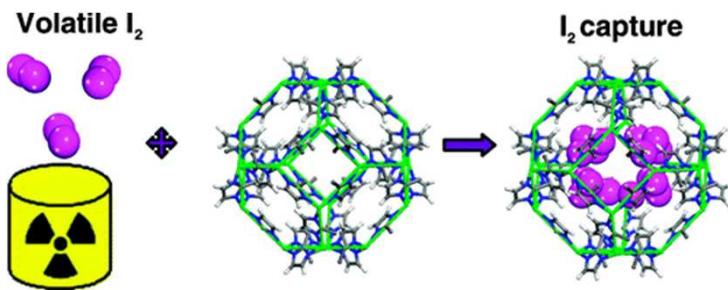
- Crystalline Order
- Nanoporosity (Guest/Host capability)
- Ultrahigh surface area (record  $\sim 7000 \text{ m}^2/\text{g}$ )
- High chemical reactivity\*
- Chemical, thermal, “irradiation” stability\*
- Photoactivity\*
- Charge/energy transfer\*



\*MOF dependent

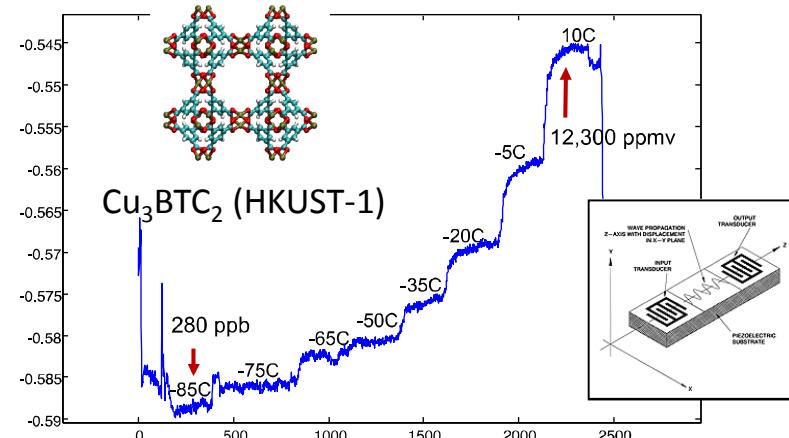
# MOF Applications Span A Wide Range of Fields

## Gas Sorption (I<sub>2</sub> capture)



Sava, et al. *JACS*, 2011, 133 (32), pp 12398–12401

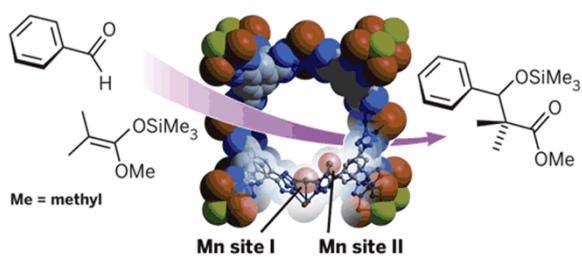
## Sensing (H<sub>2</sub>O sensing)



Robinson et al. *Anal. Chem.* 84 (2012), 7043

## Catalysis (Mukaiyama aldol synthesis)

### Mn-BTT



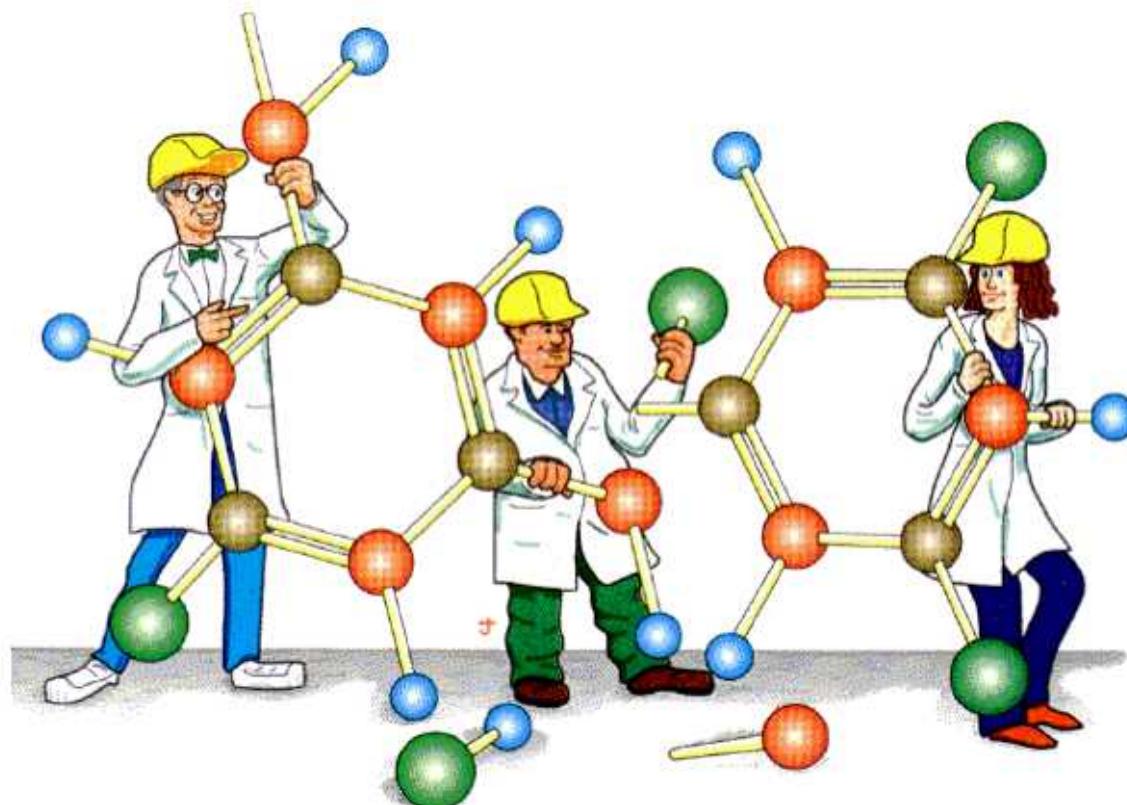
Horike, et al. *JACS*, 2008, 130 (18), pp 5854–5855

## Electronics/Optoelectronics

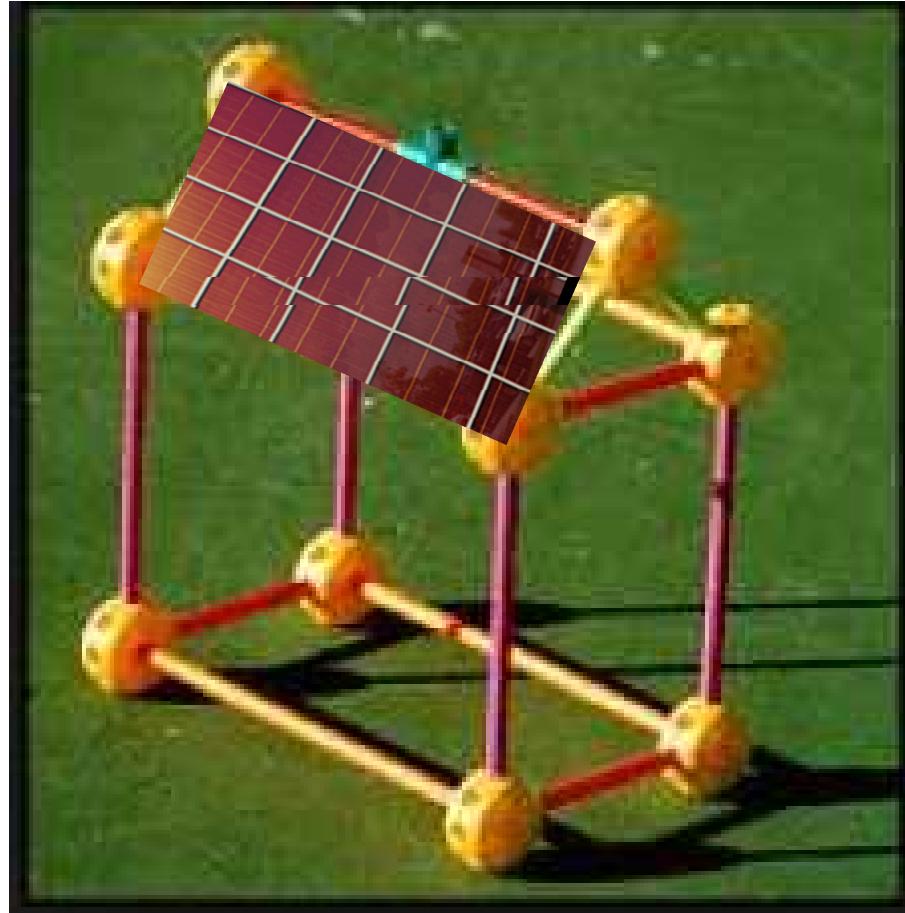


# Realizing Supramolecular Materials...

*Can we manipulate and assemble these supramolecular building blocks to improve DSSCs?*



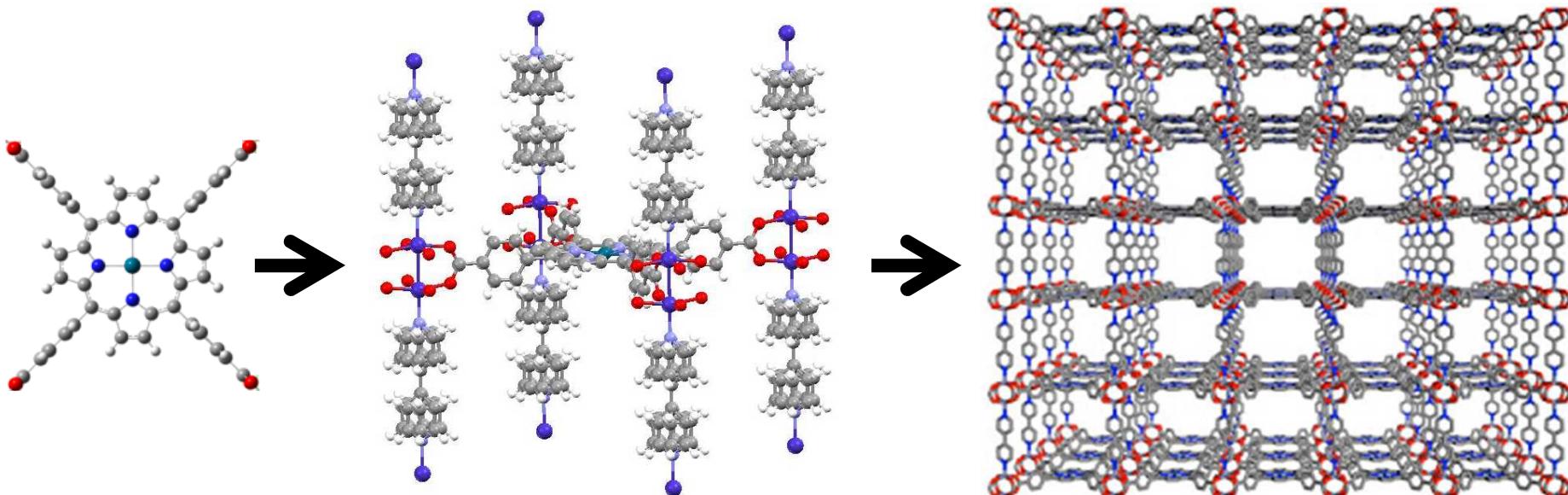
# Merging “Modular” Chemistry and Photovoltaics...



## Consider Pillared Porphyrin Frameworks (PPFs)

In PPF MOFs, transition metal cations coordinate the assembly of photoactive metalloporphyrins into sheets, stacked atop molecular pillars.

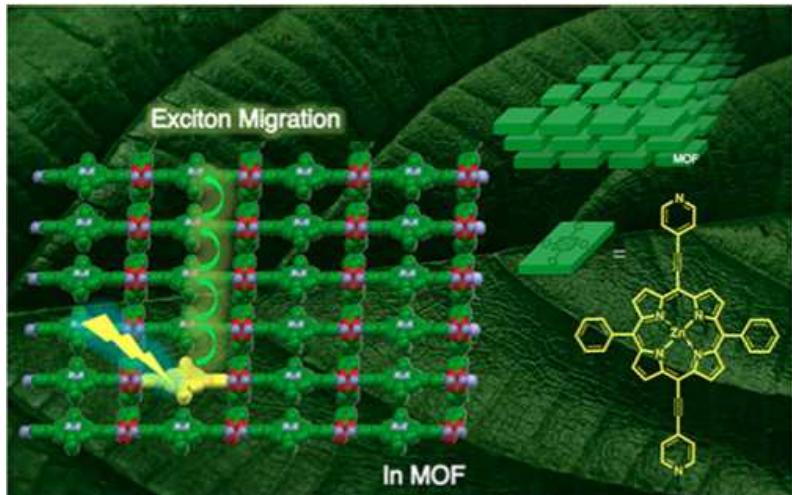
PPF-5



# Building Photoactive MOFs

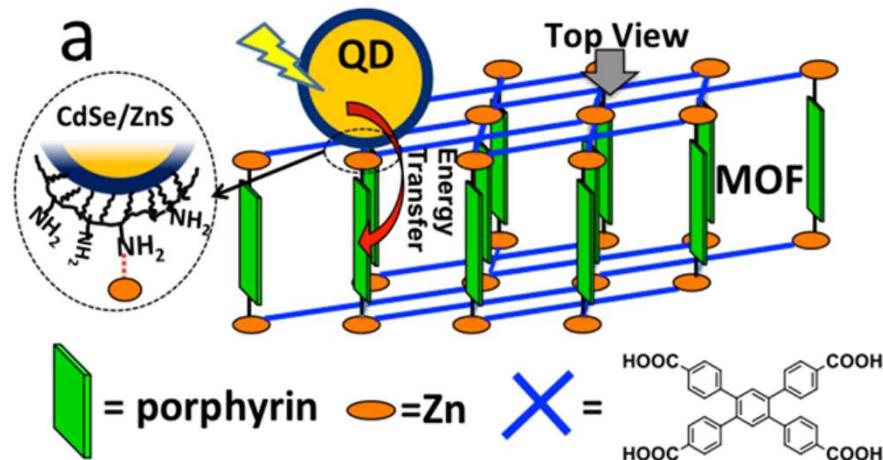
*Energy transfer is viable in porphyrin-based MOFs*

*Fast exciton transport between porphyrins*



H.-J. Son, et al. JACS (2013) **135**. 862-869.

*Energy transfer between MOFs and semiconductors*

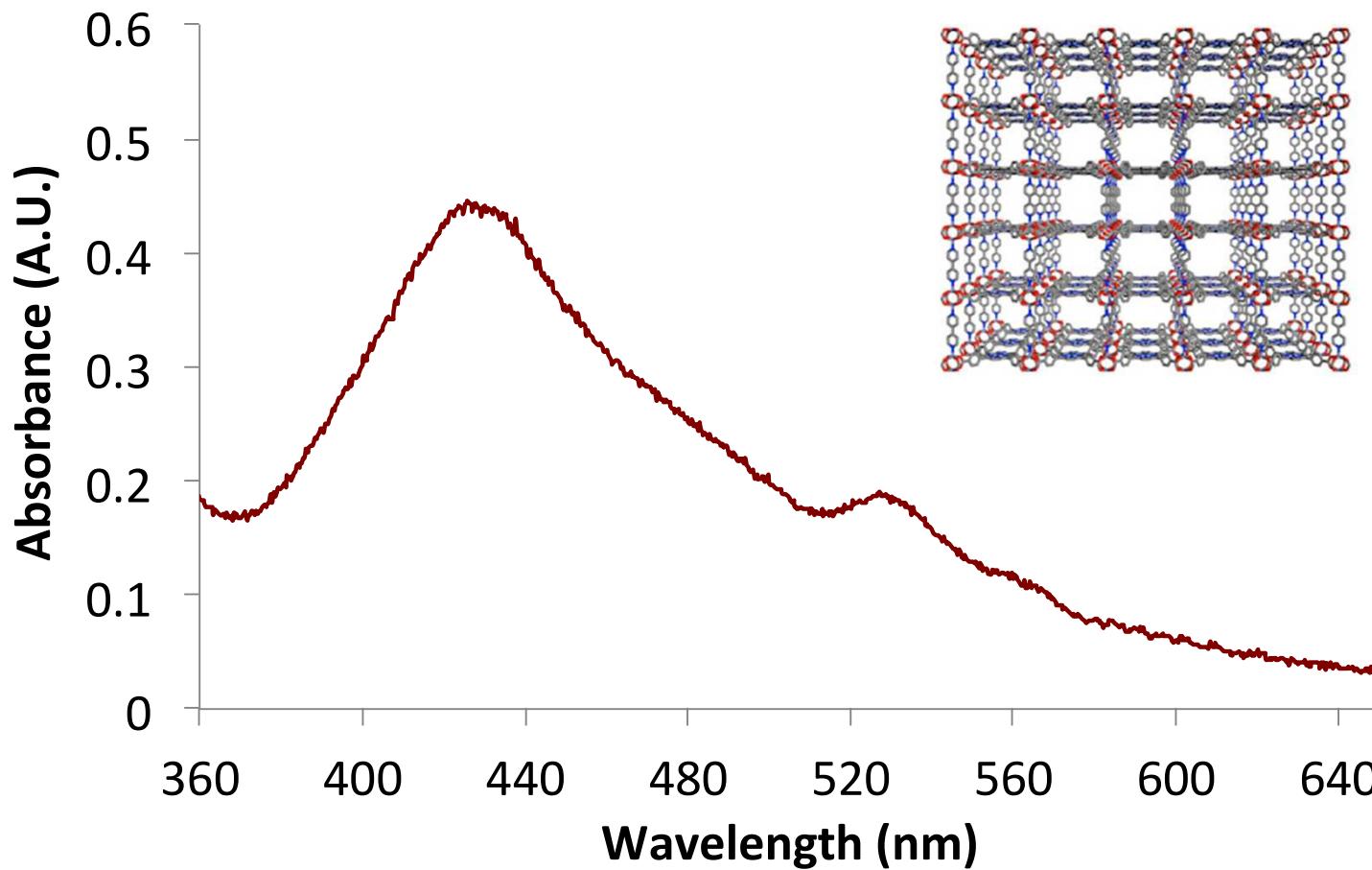


S.Jin, et al. JACS (2013) **135**. 955-958.

Precedents for using porphyrins in DSSCs...

1. Kay and Grätzel, J. Phys. Chem. (1993) **97**, 6292.
2. Walter, et al. J. Porphyrins and Phthalocyanines. (2010) **14**, 759.
3. M. J. Griffith and A. J. Mozer (2011), Available from: <http://www.intechopen.com/books/solar-cells-dye-sensitized-devices/porphyrin-based-dye-sensitized-solar-cells>

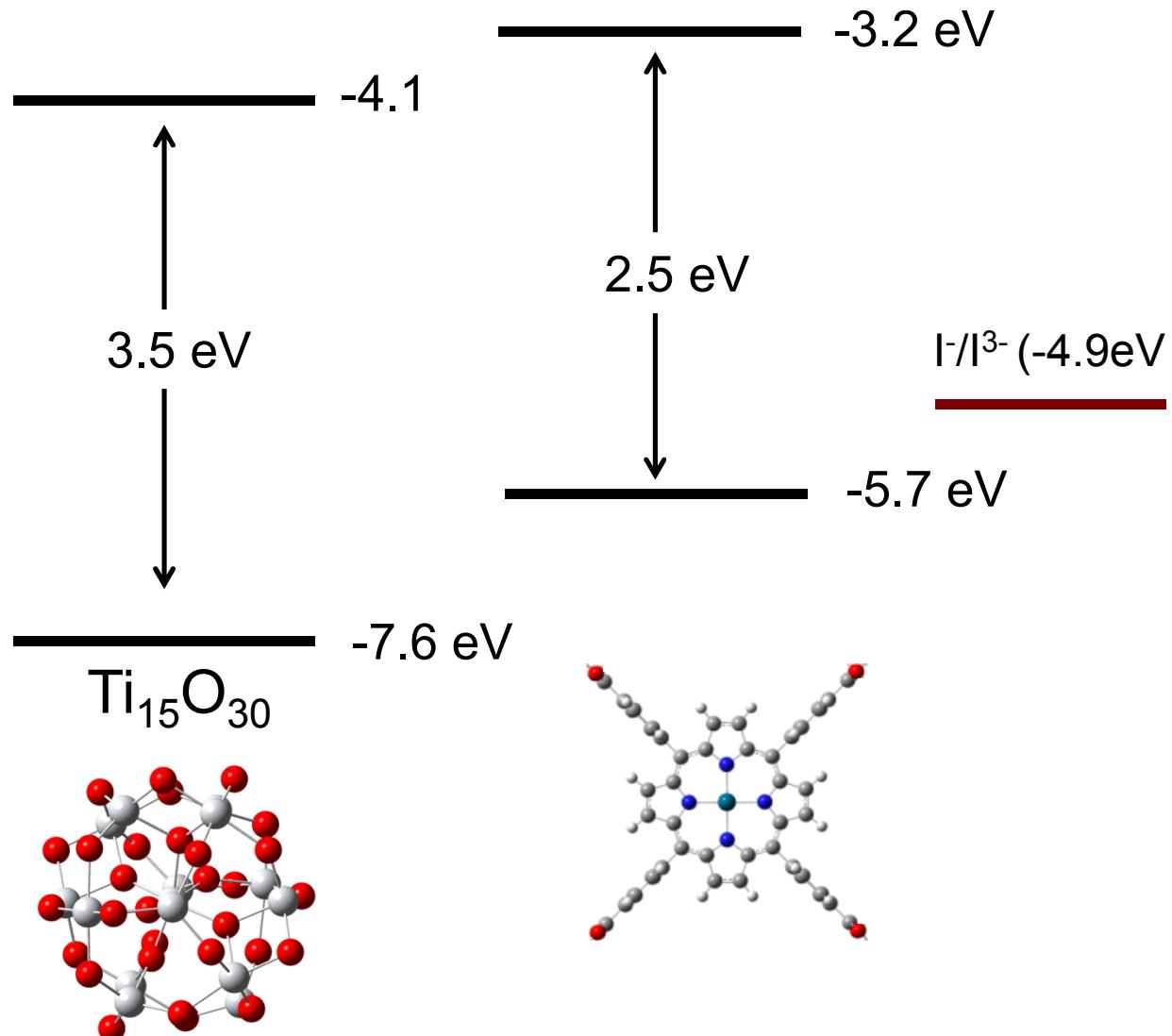
# PPF-5 Optical Absorbance



*PPF-5 absorbs meaningful visible light.*

# DFT Predictions of Band Alignment

Density Functional Theory (DFT) predicts reasonable band alignment between the PPF-5 porphyrin and a  $\text{TiO}_2$  electron acceptor.



$\text{TiO}_2$  Molecular Clusters  
DFT (B3LYP/LanL2DZ)

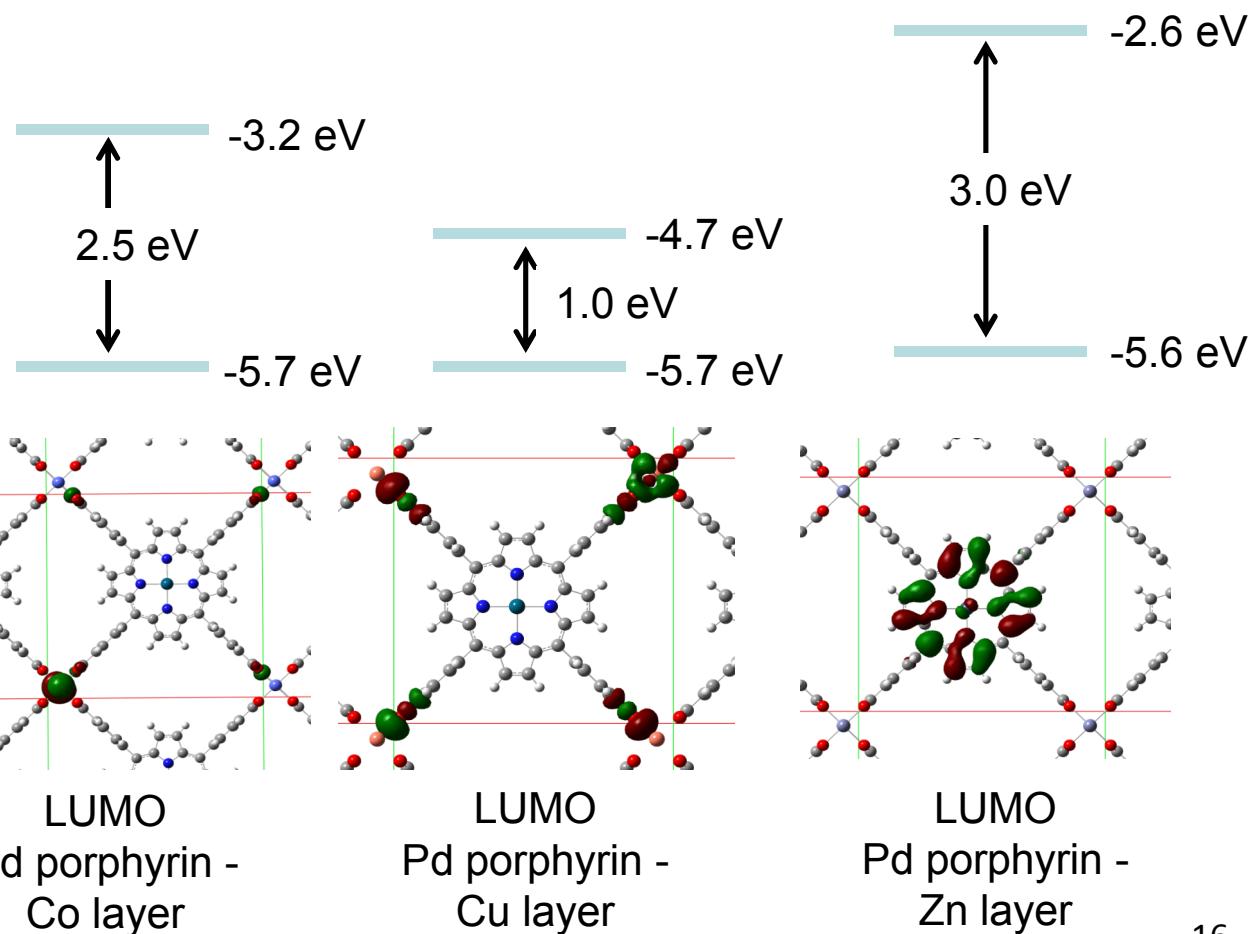
PPF-5 2D  
DFT(B3LYP/ CEP-31G)

# Band Structure Tailoring

*Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.*

## Varying transition metal ions

2D periodic  
optimization -  
DFT(B3LYP/ CEP-  
31G)

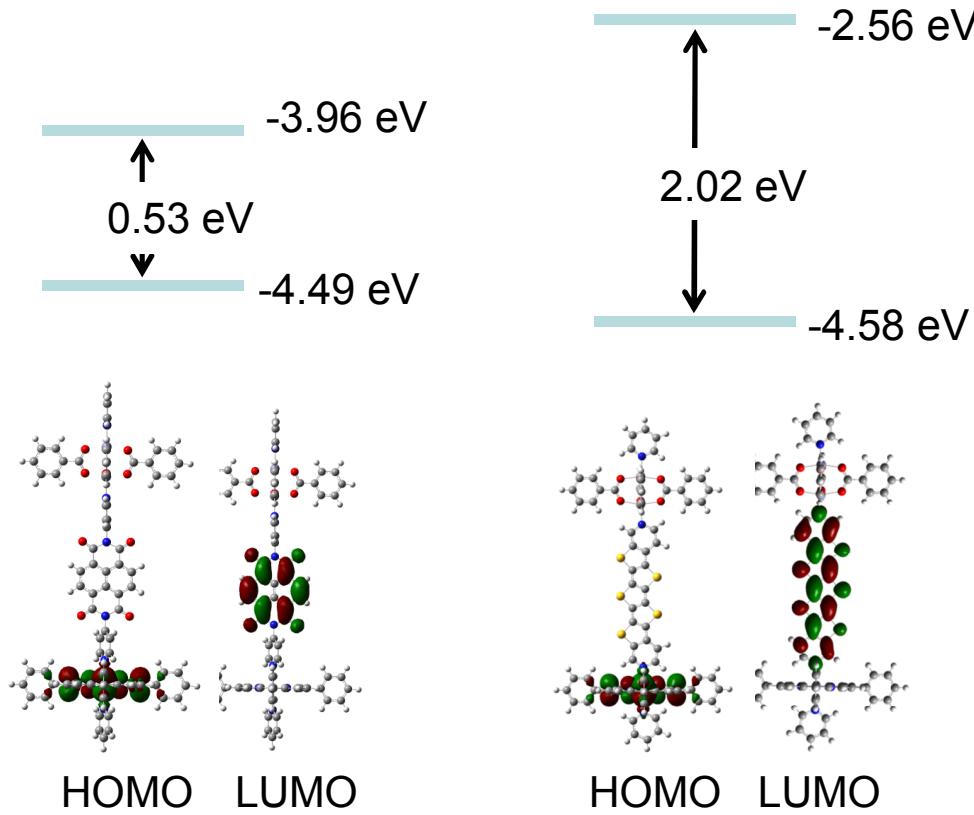


# Band Structure Tailoring

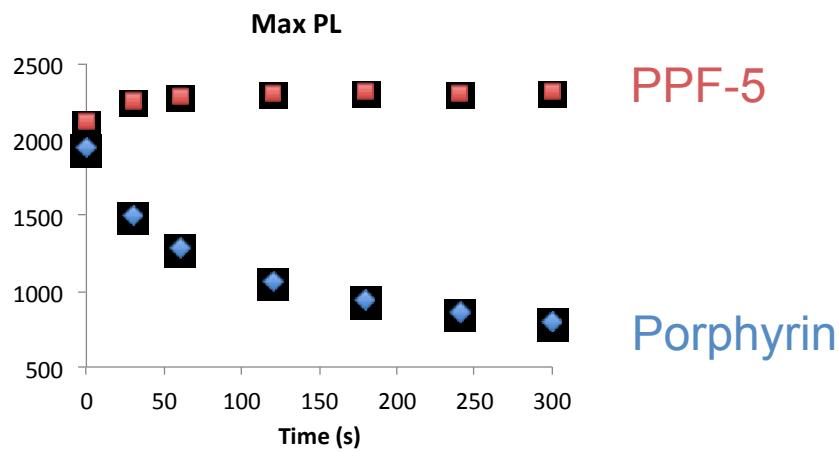
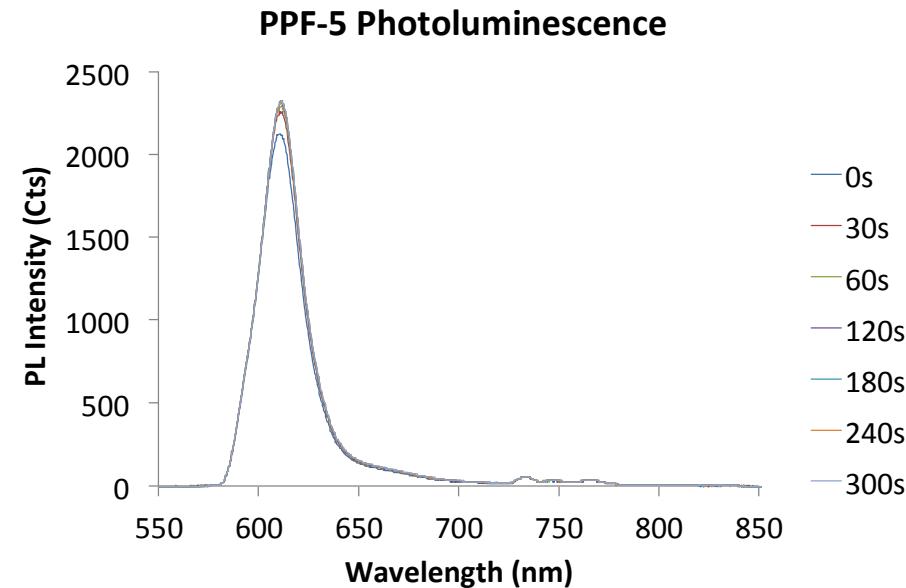
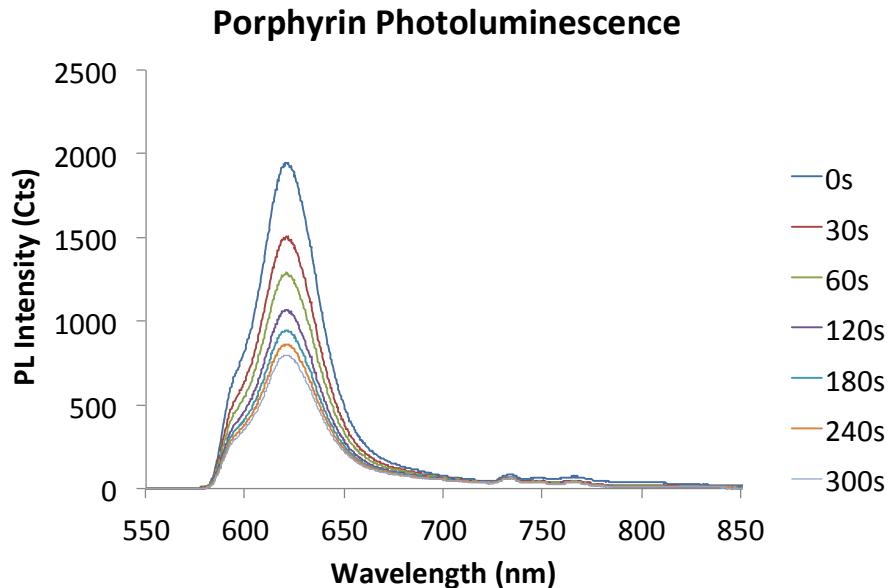
*Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.*

## Varying organic pillars

2D periodic optimization -  
DFT(B3LYP/ CEP-31G)



# Molecular Photostability in PPF-5

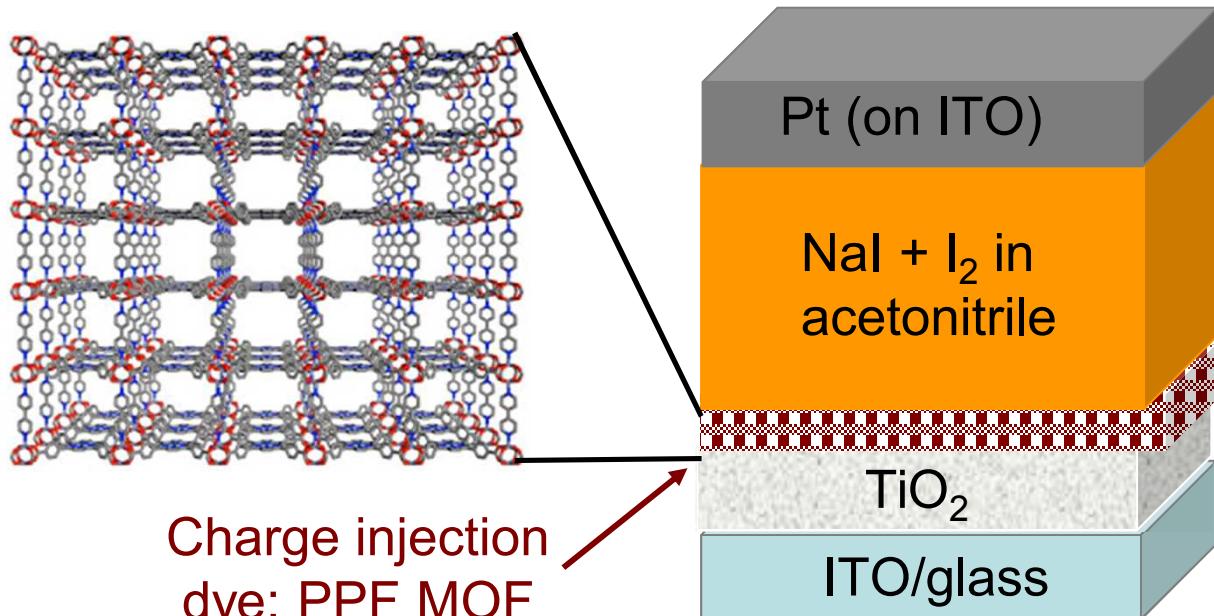


While “free” porphyrin is susceptible to photodegradation, incorporation of the porphyrin into PPF-5 leads to remarkably photostability.

# Integration of PPF-5 in DSSCs

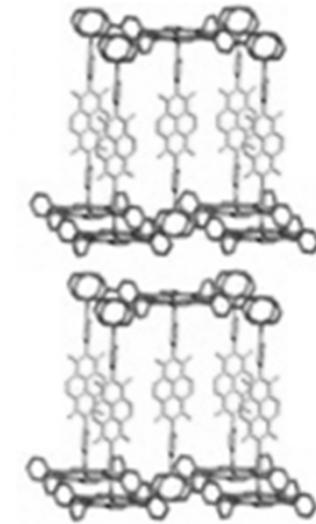
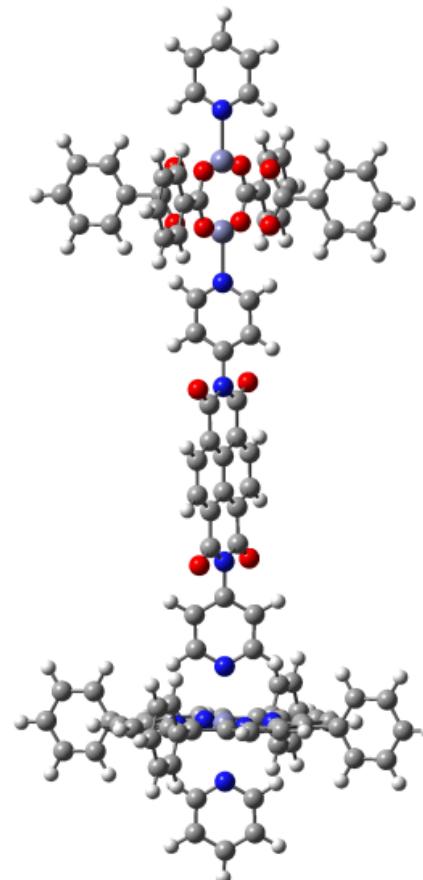
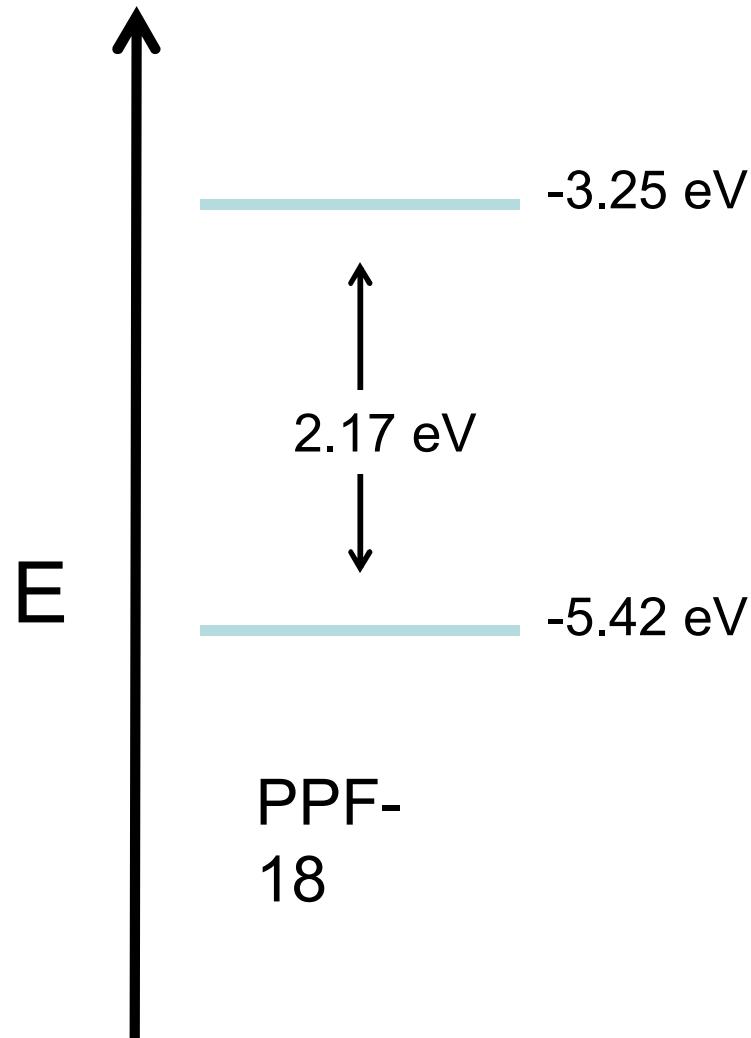
Potential advantages of PPF-5 active layer:

- Visible light absorption
- Reasonable, potentially tunable band alignment
- Ordered charge transport pathways
- Non-aggregated dye assembly
- Porosity for electrolyte access
- Promising photostability



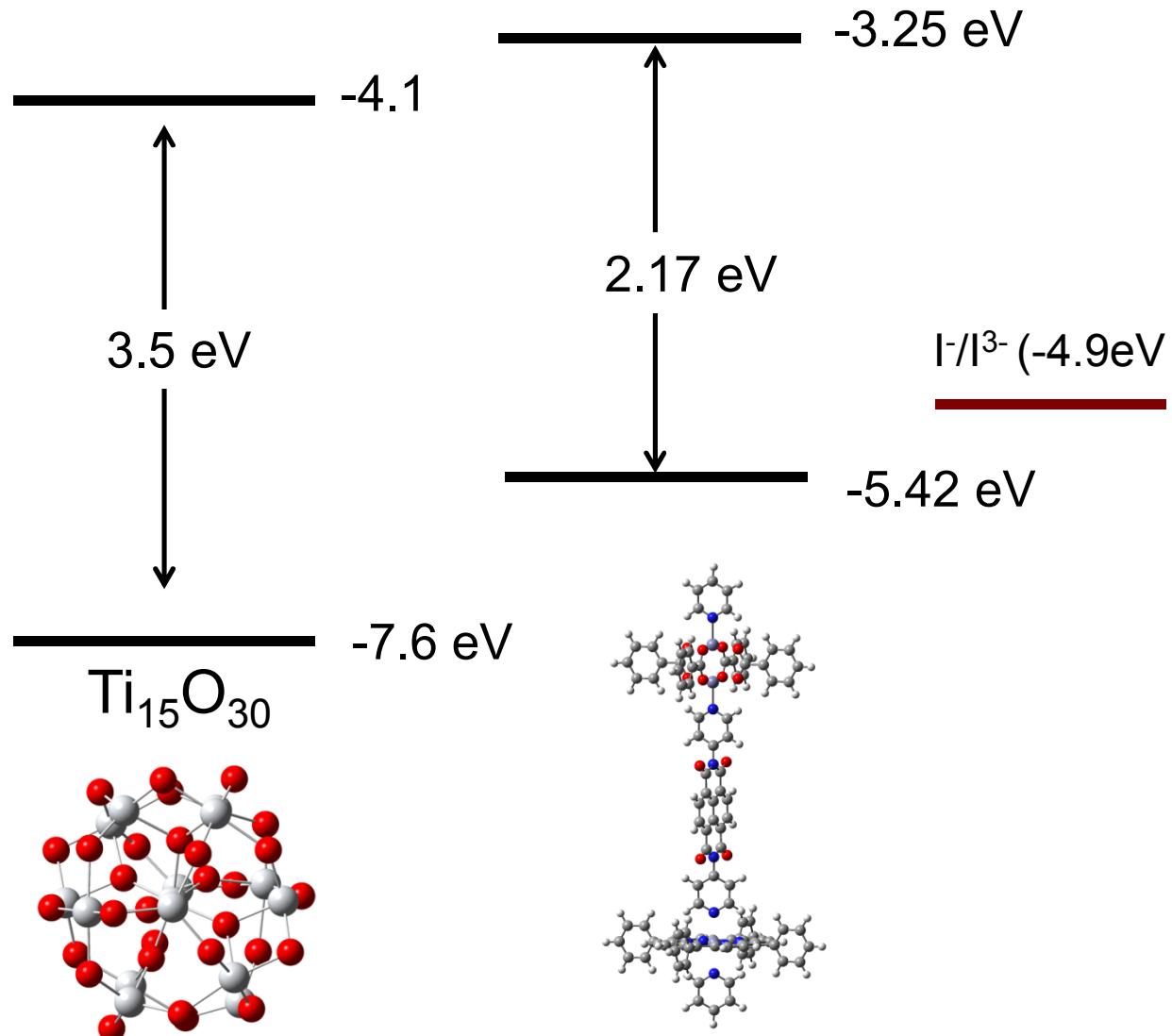
*Can we grow PPF-5 onto the  $\text{TiO}_2$  anode for integration into a “standard” DSSC configuration?*

# PPF-18 Band Structure



# DFT Predicts Slightly Better Band Alignment for PPF-18

Density Functional Theory (DFT) predicts reasonable band alignment between the PPF-5 porphyrin and a  $\text{TiO}_2$  electron acceptor.



$\text{TiO}_2$  Molecular Clusters  
DFT (B3LYP/LanL2DZ)

PPF-5 2D  
DFT(B3LYP/ CEP-31G)

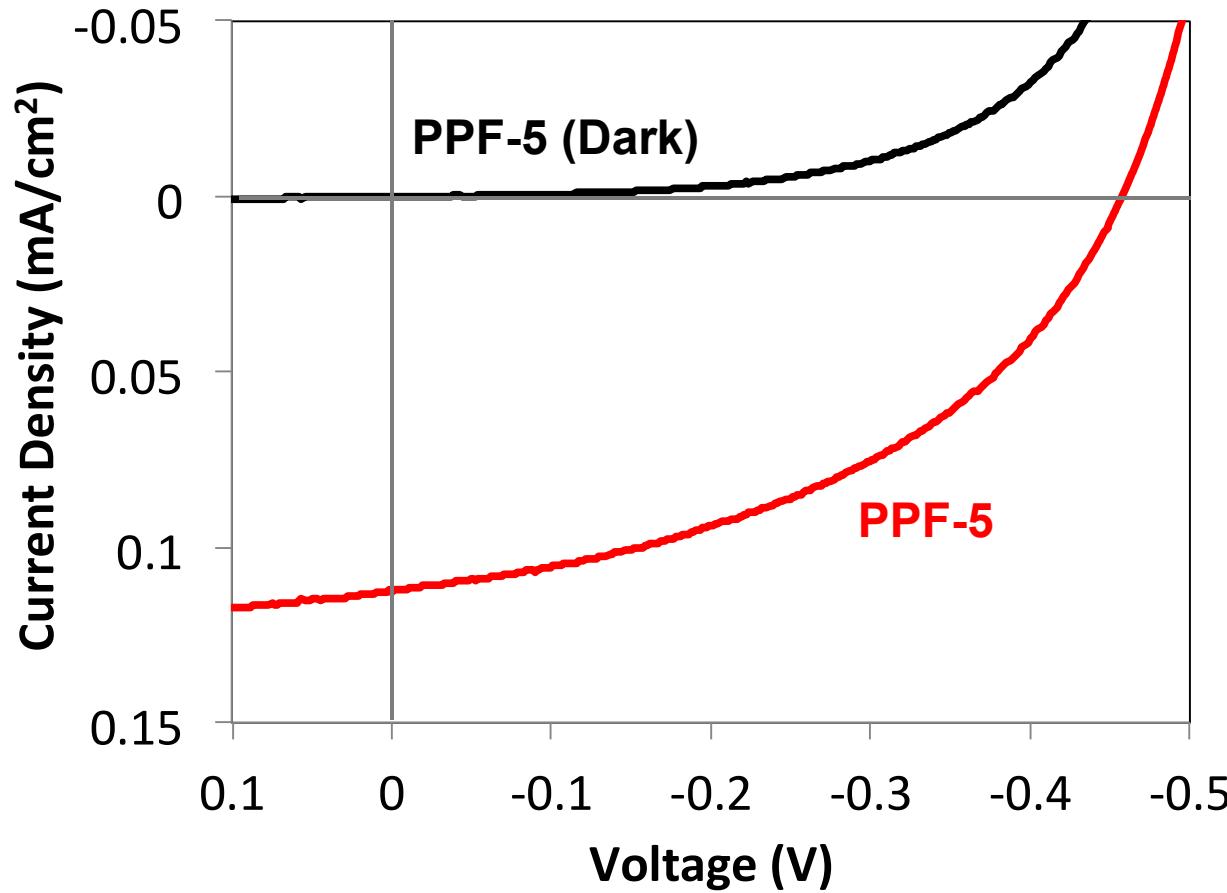
# Performance of PPF-5 in a DSSC?



*Can the PPF-5 MOF be used in a  
functioning DSSC?*

# Performance of PPF-5 in a DSSC

PPF-5 DSSC produces measurable photocurrent!



Averaged metrics:

$V_{oc}$  (V) =  $0.452 \pm 0.029$

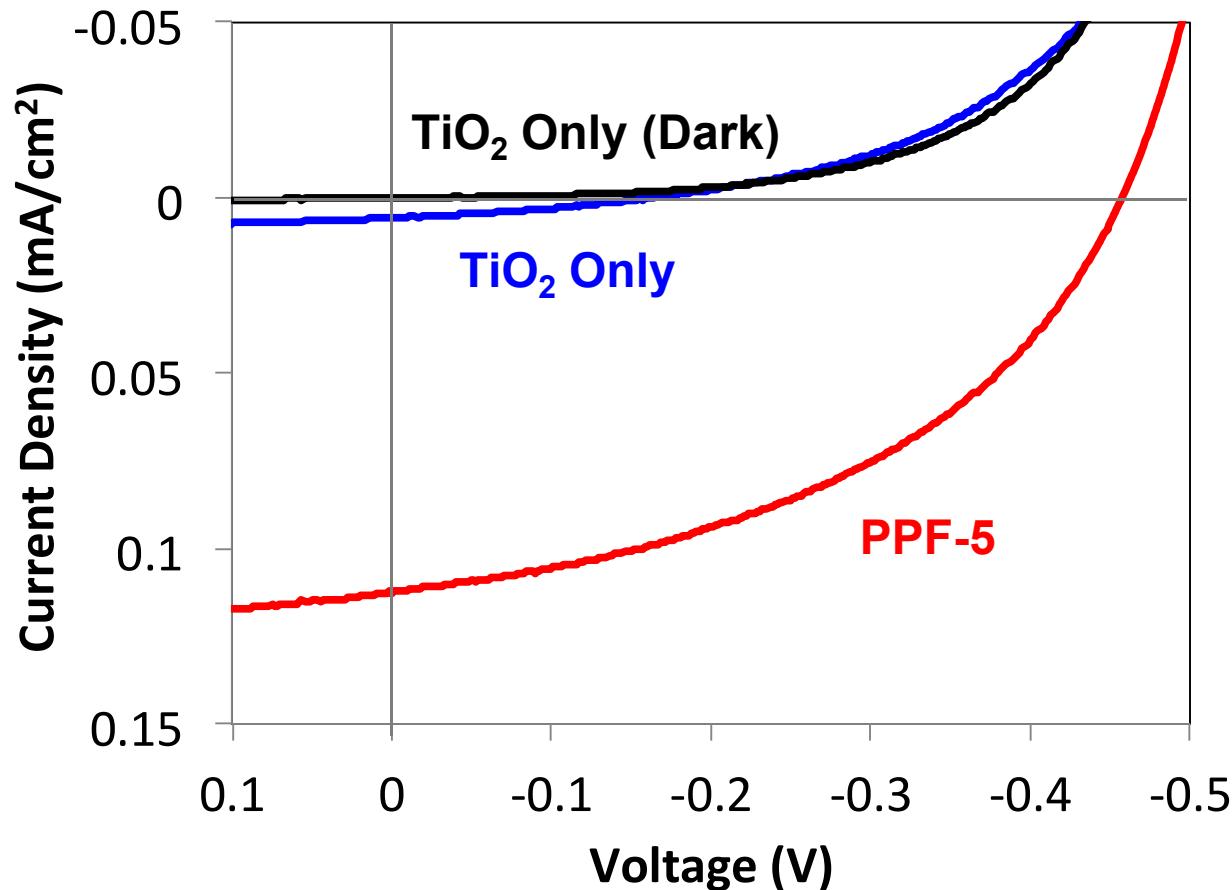
$J_{sc}$  (mA/cm<sup>2</sup>) =  $0.097 \pm 0.014$

FF =  $0.47 \pm 0.031$

$\eta$  (%) =  $0.026 \pm 0.0038$

# Negative Control: No PPF-5

Control experiments containing no PPF-5 produce negligible photocurrent.



In the absence of dye:

$$V_{oc} (V) = 0.16$$

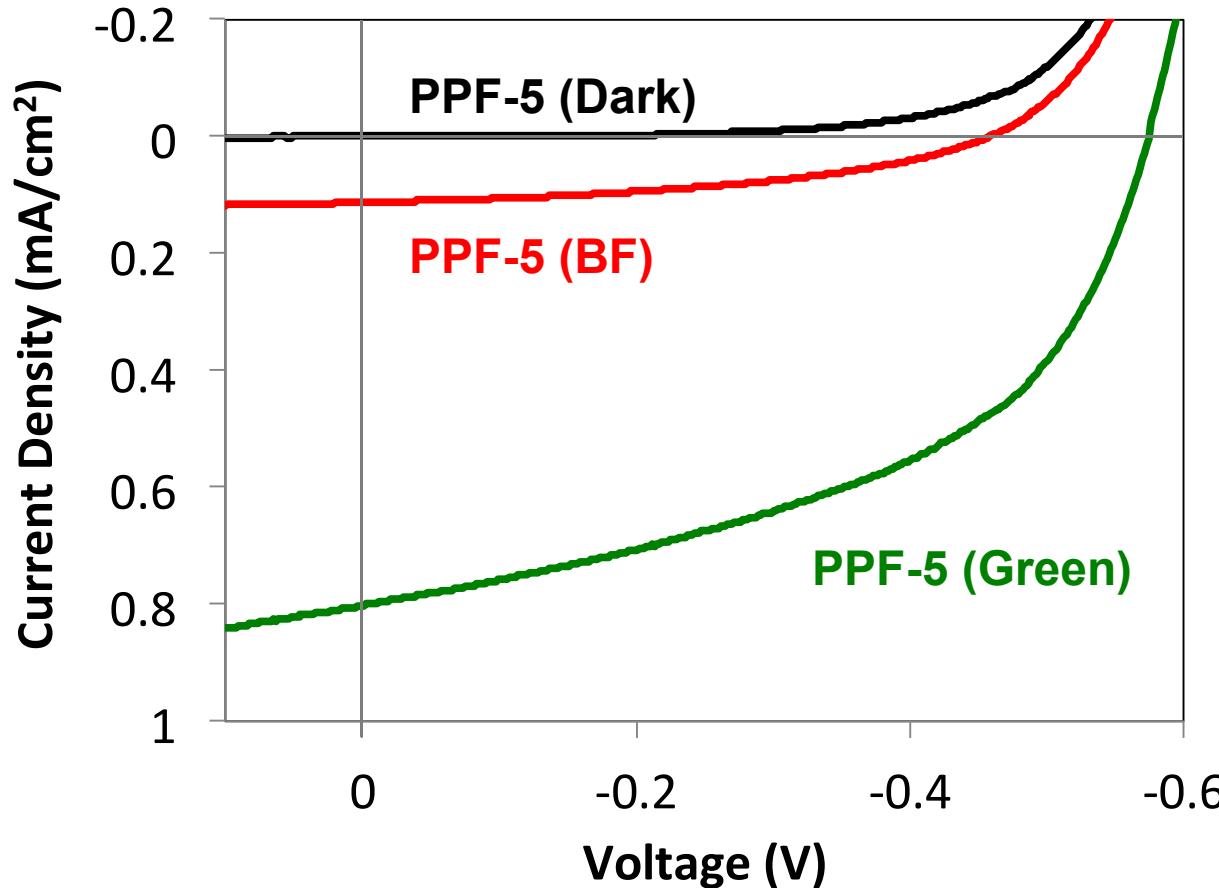
$$J_{sc} (\text{mA/cm}^2) = 0.0058$$

$$FF = 0.31$$

$$\eta (\%) = 0.000378$$

# Enhancing PPF-5 Performance with Green Light

Selective excitation with Green light produces enhanced photocurrent, confirming contribution from PPF-5 absorber.



Averaged metrics:

V<sub>oc</sub> (V) = 0.452 ± 0.029

J<sub>sc</sub> (mA/cm<sup>2</sup>) = 0.097 ± 0.014

FF = 0.47 ± 0.031

V<sub>oc</sub> (V) = 0.563 ± 0.018

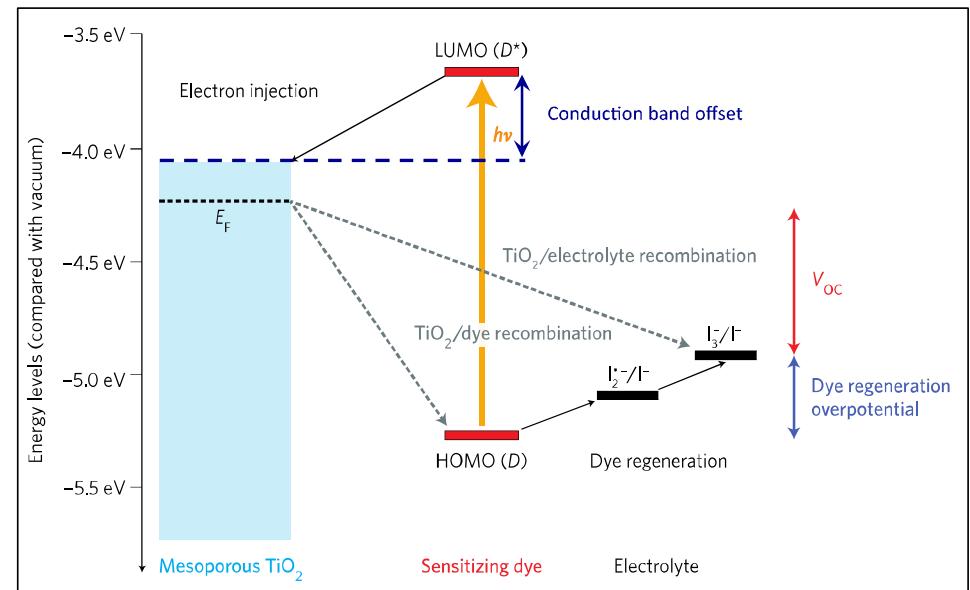
J<sub>sc</sub> (mA/cm<sup>2</sup>) = 0.73 ± 0.083

FF = 0.537 ± 0.07

# What Next?

How to build on this initial demonstration?

- Improve interfacial loading on  $\text{TiO}_2$
- Optimize band alignments to reduce loss in potential
- Increase spectral range of absorber
- Consider Guest-host interactions
- Explore stability/reliability



# Wrapping up

- MOFs are highly porous, multifunctional composites crystals, assembled from “modular” molecular building blocks.
- Grown by Layer-by-Layer processes, *PPF-5 integrated into DSSC devices can serve as a functional active absorber!*
- Tuning of MOF composition and structure are expected to improve DSSC device performance.
- This preliminary demonstration shows that this electrochemical configuration is a feasible platform to explore the diversity of MOF chemistry in solar applications.

*This is just the beginning!*

# Thank you!!

## Sandia (Albuquerque)

Jill S. Wheeler

Dr. Dara Van Gough

Steven Wolf

**Dr. Leo Small**

Dr. Tim Lambert

Bonnie McKenzie

## Funding

- U.S. DOE Energy Efficiency and Renewable Energy (Sunshot): Next Generation PV I and II
- Sandia Laboratory Directed Research and Development Program

## Sandia (California)

Dr. Michael Foster

Dr. Vitalie Stavila

Dr. Kirsty Leong

Dr. Mark D. Allendorf



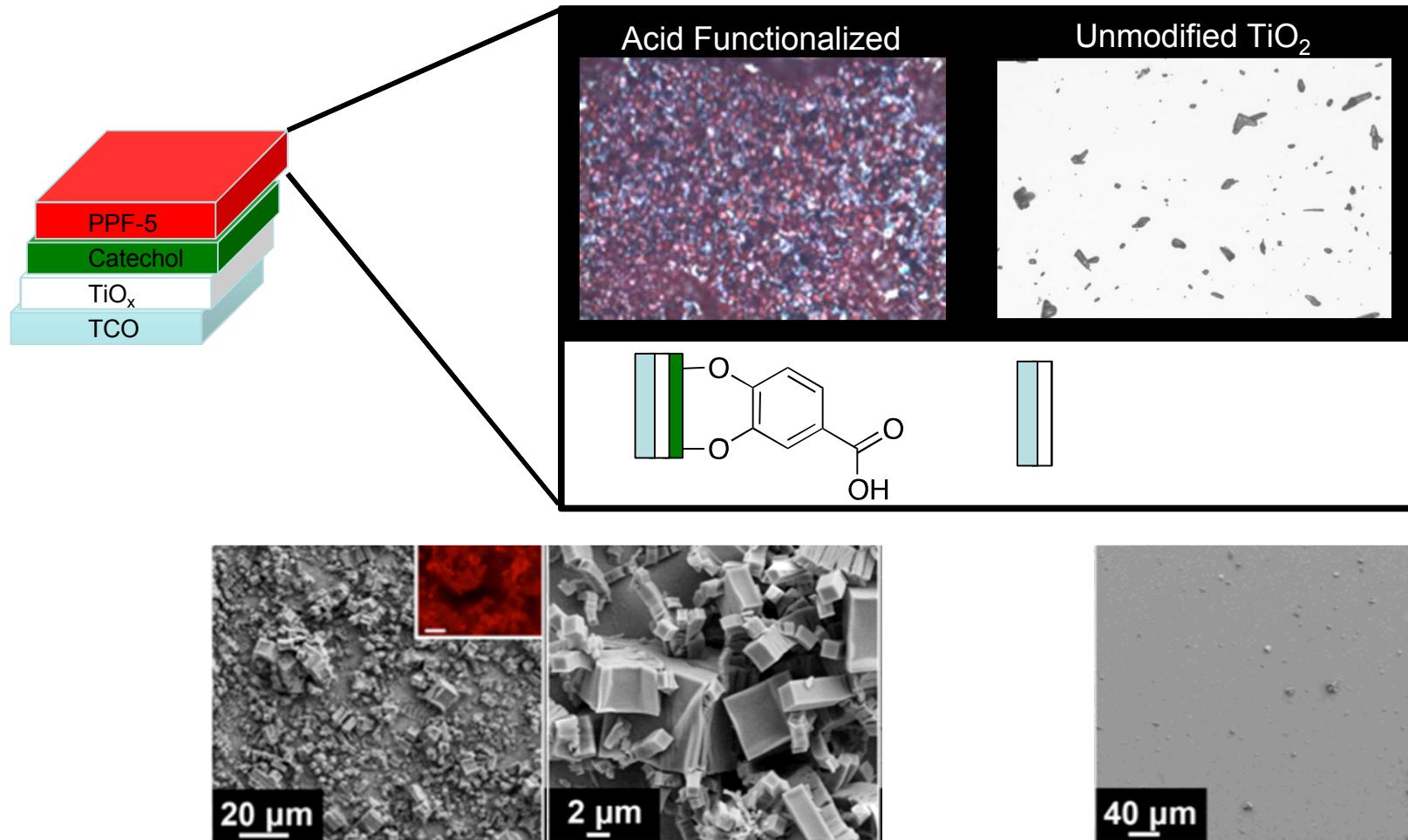
Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000

# Backup Slides



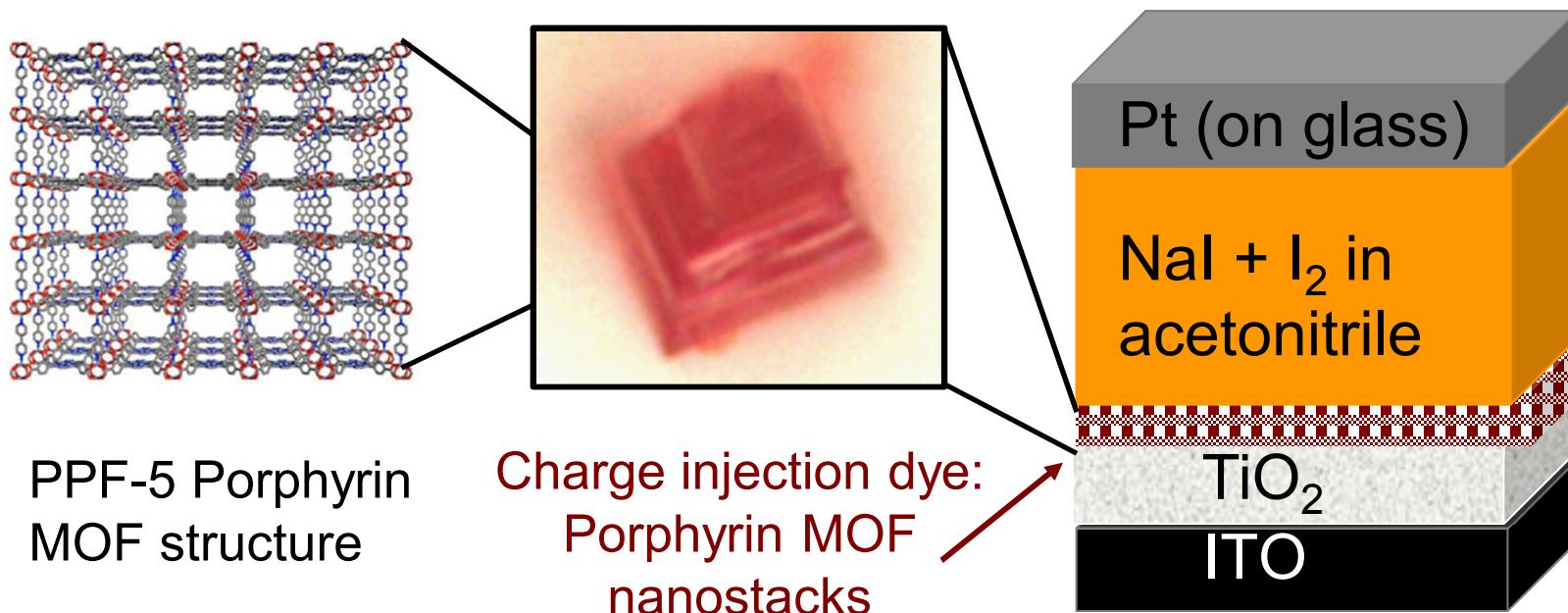
# Introduction to MOFs

Solvothermal PPF-5 growth on acid-functionalized surfaces promotes surface-nucleated PPF-5 crystal growth.



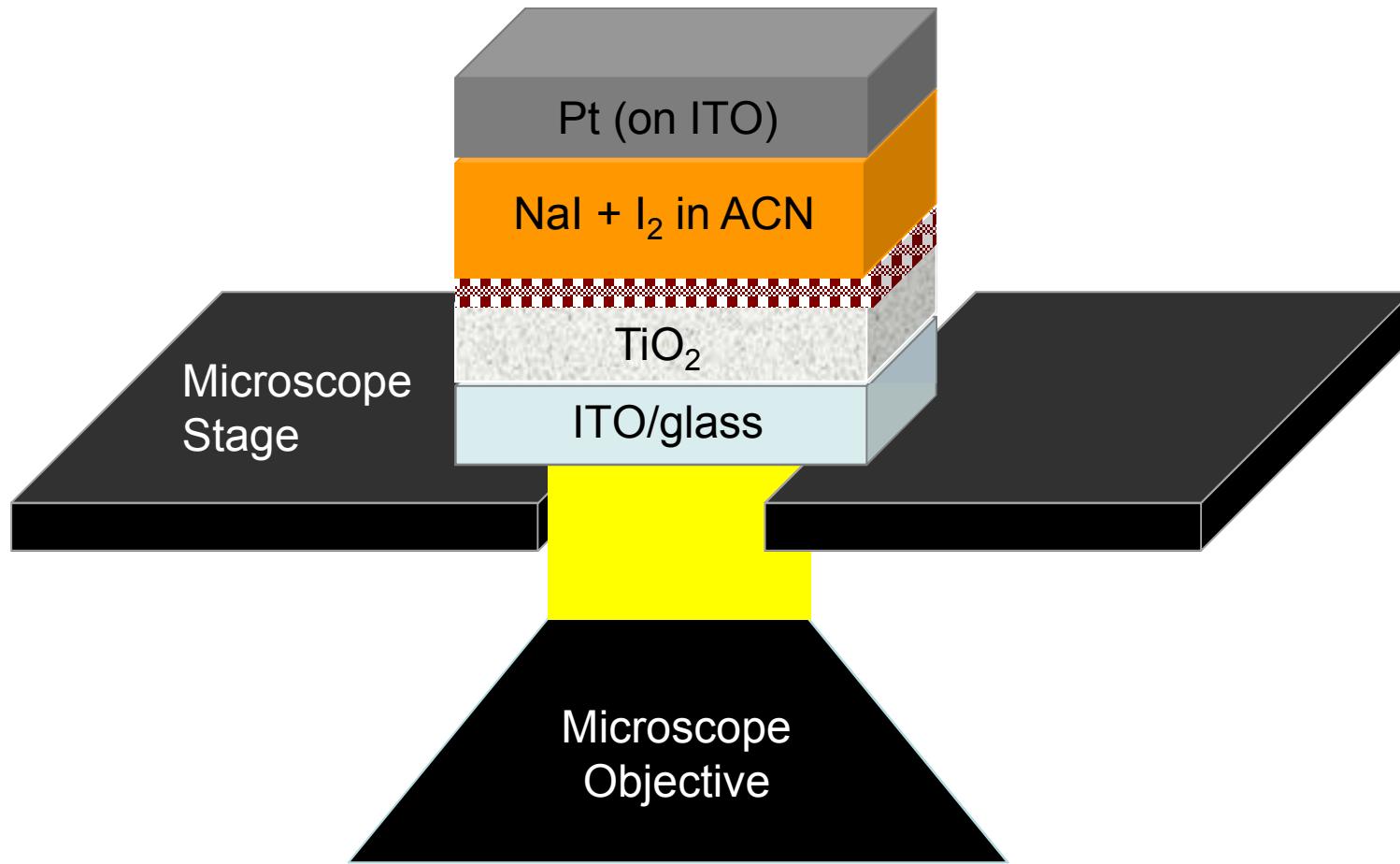
# Integration of PPF-5 in DSSCs

PPF-5 should be readily incorpoated into a DSSC device configuration...



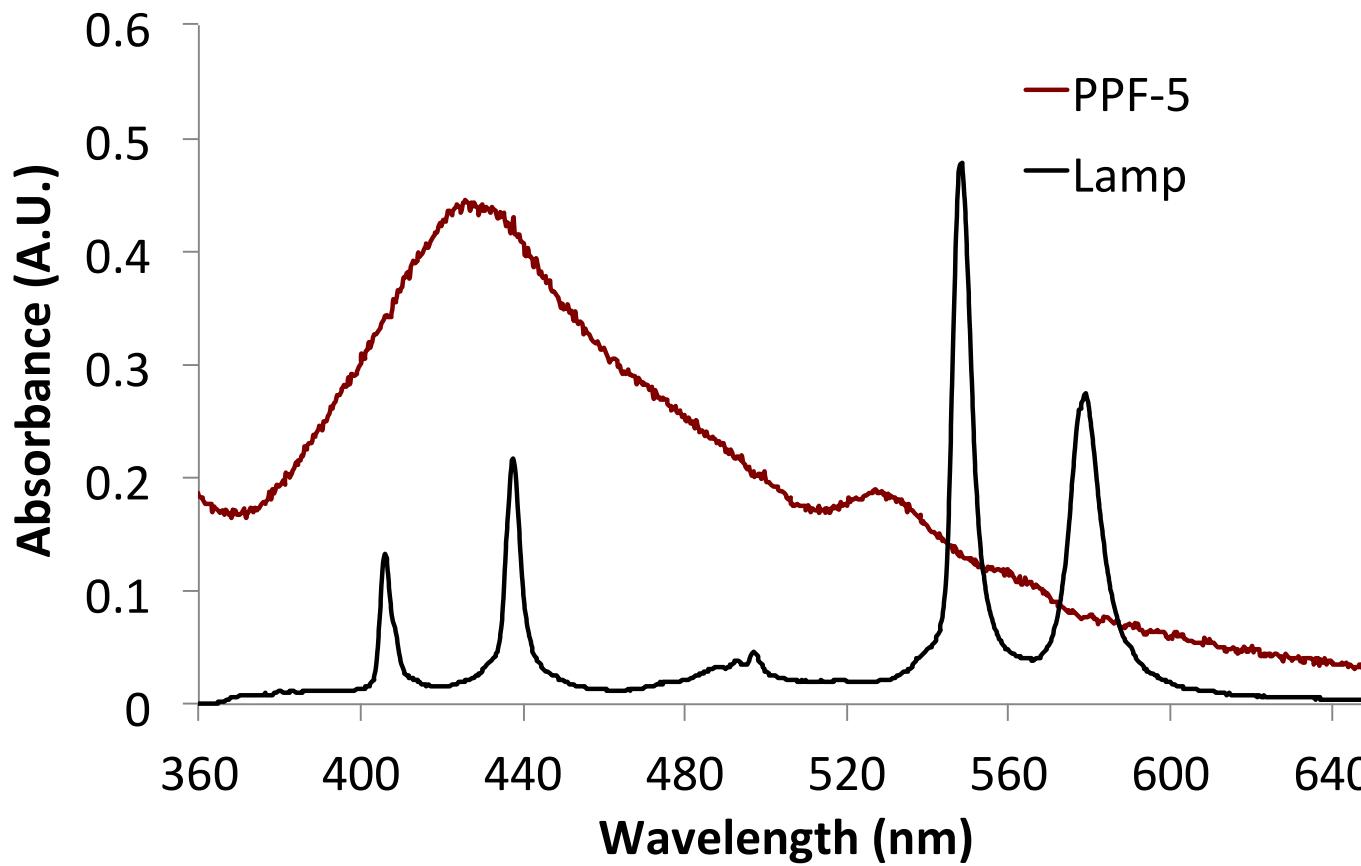
# PV Measurement Setup

DSSC devices assembled and tested on microscope stage with UV-filtered Hg-arc lamp.

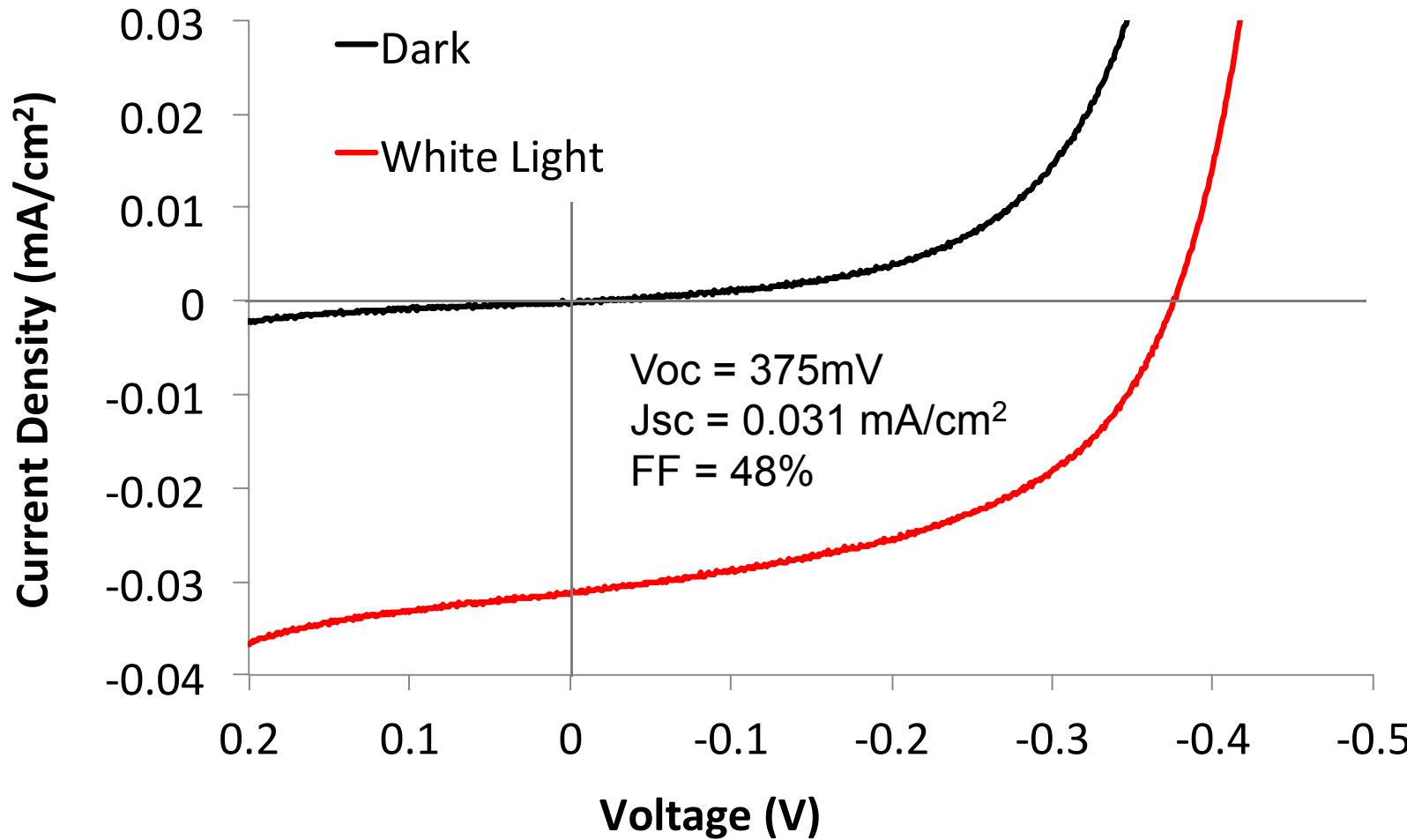


# PPF-5 Optical Absorbance

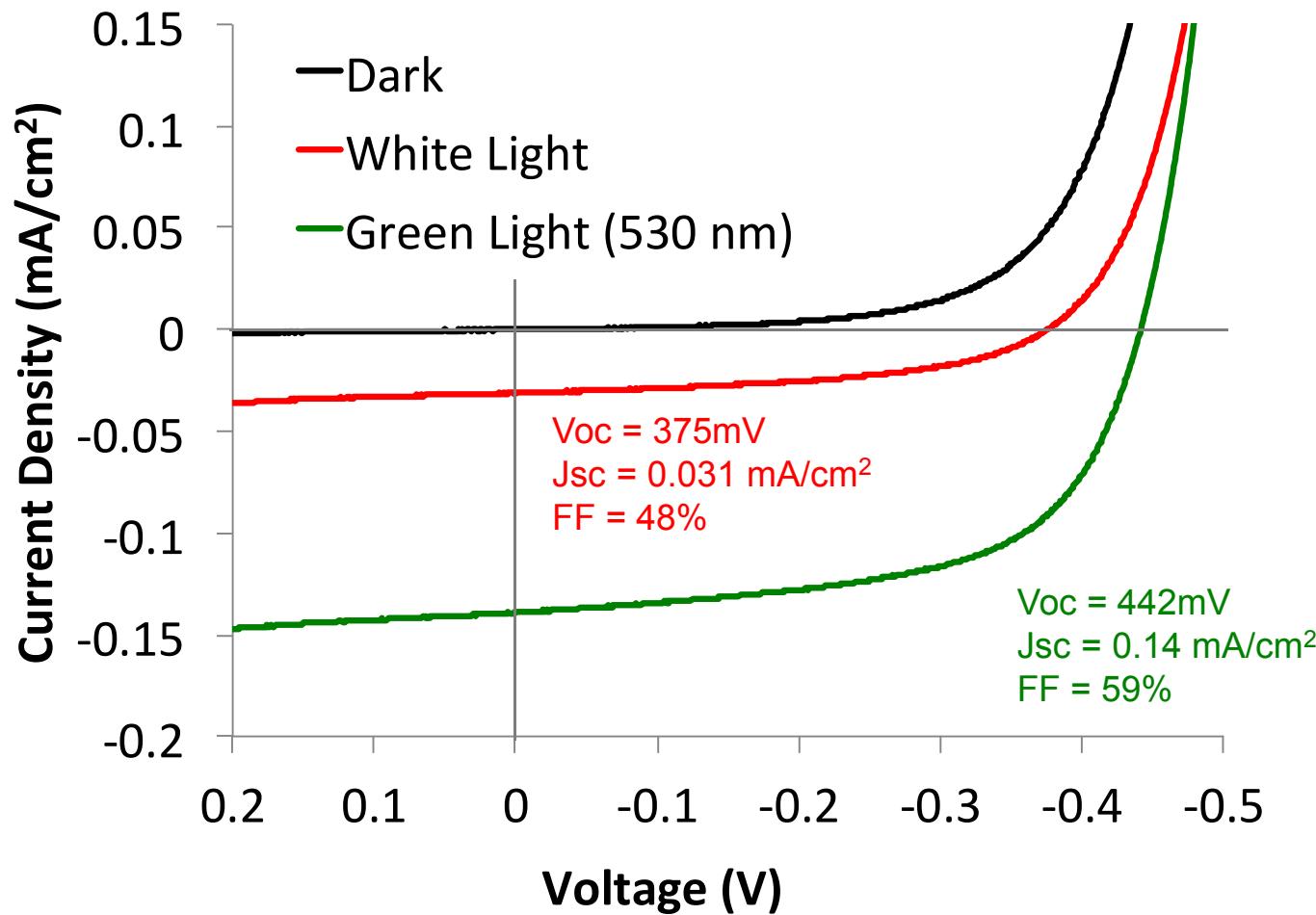
PPF-5 absorbance aligns reasonably UV-filtered Hg-arc lamp.



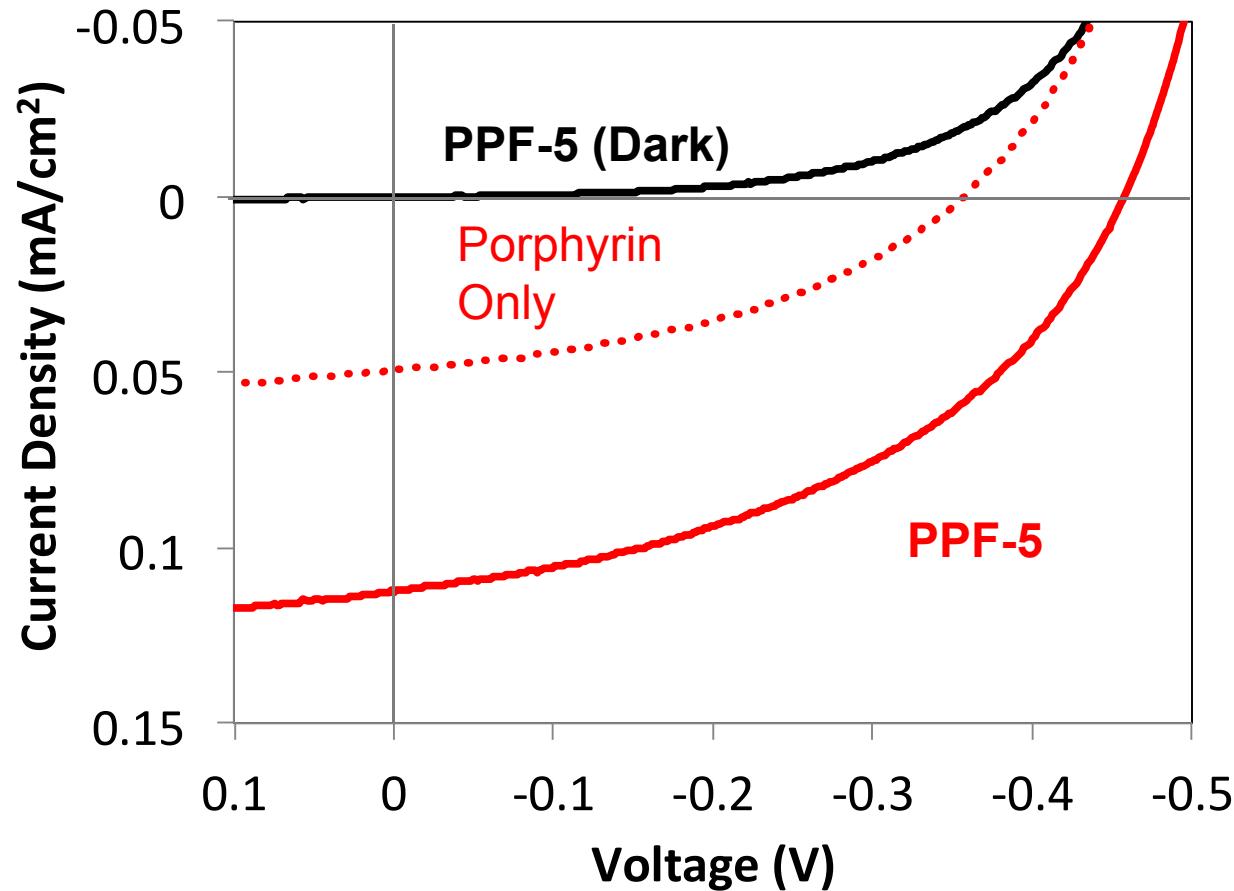
# PPF-5 PV Performance in a DSSC



# Optical “cheating”



# Performance of PPF-5 in a DSSC



Averaged metrics:

$\text{Voc (V)} = 0.401 \pm 0.057$

$\text{Jsc } (\text{mA}/\text{cm}^2) = 0.061 \pm 0.017$

$\text{FF} = 0.45 \pm 0.039$

$\eta \text{ (\%)} = 0.015 \pm 0.0073$

$\text{Voc (V)} = 0.452 \pm 0.029$

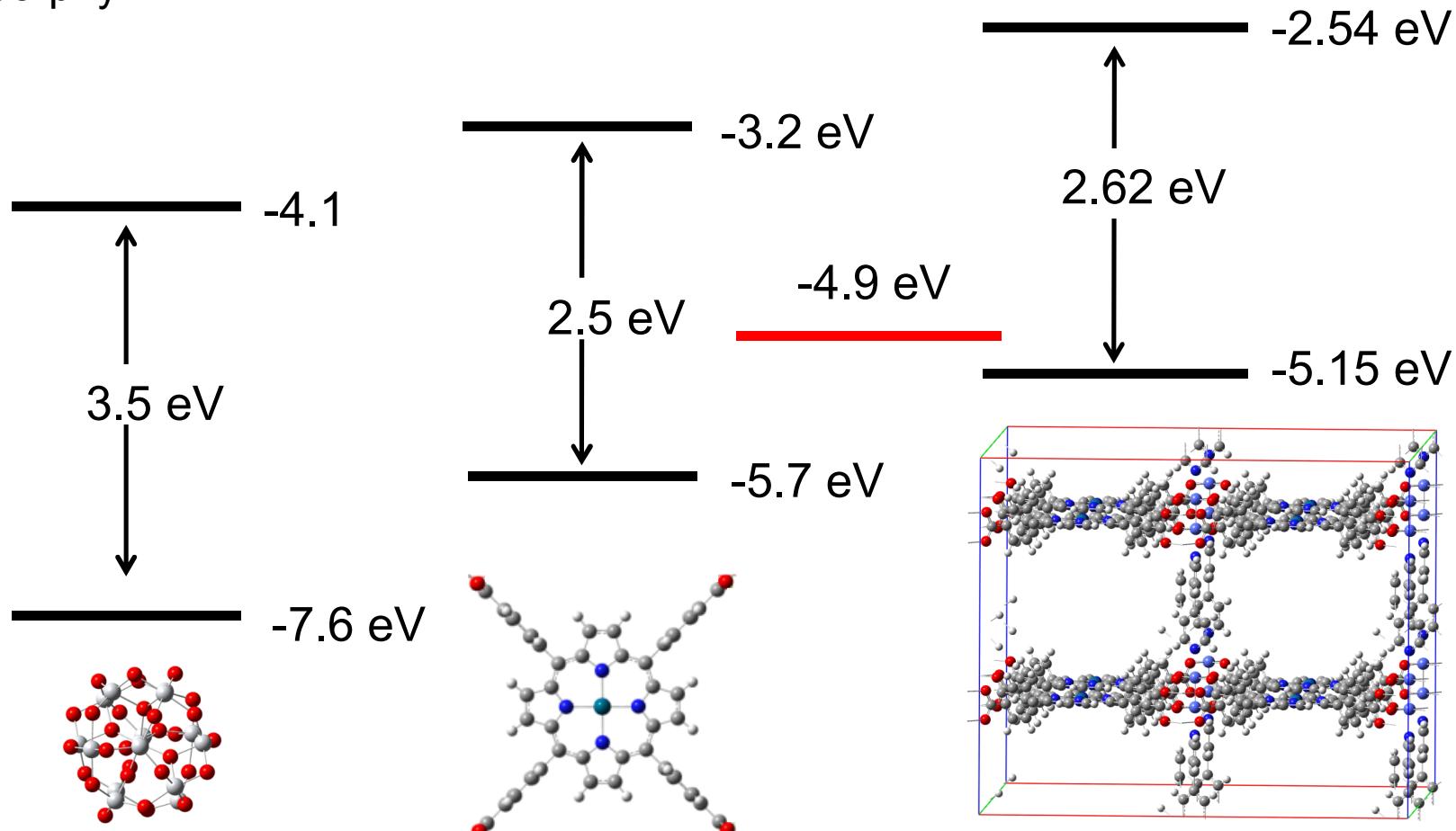
$\text{Jsc } (\text{mA}/\text{cm}^2) = 0.097 \pm 0.014$

$\text{FF} = 0.47 \pm 0.031$

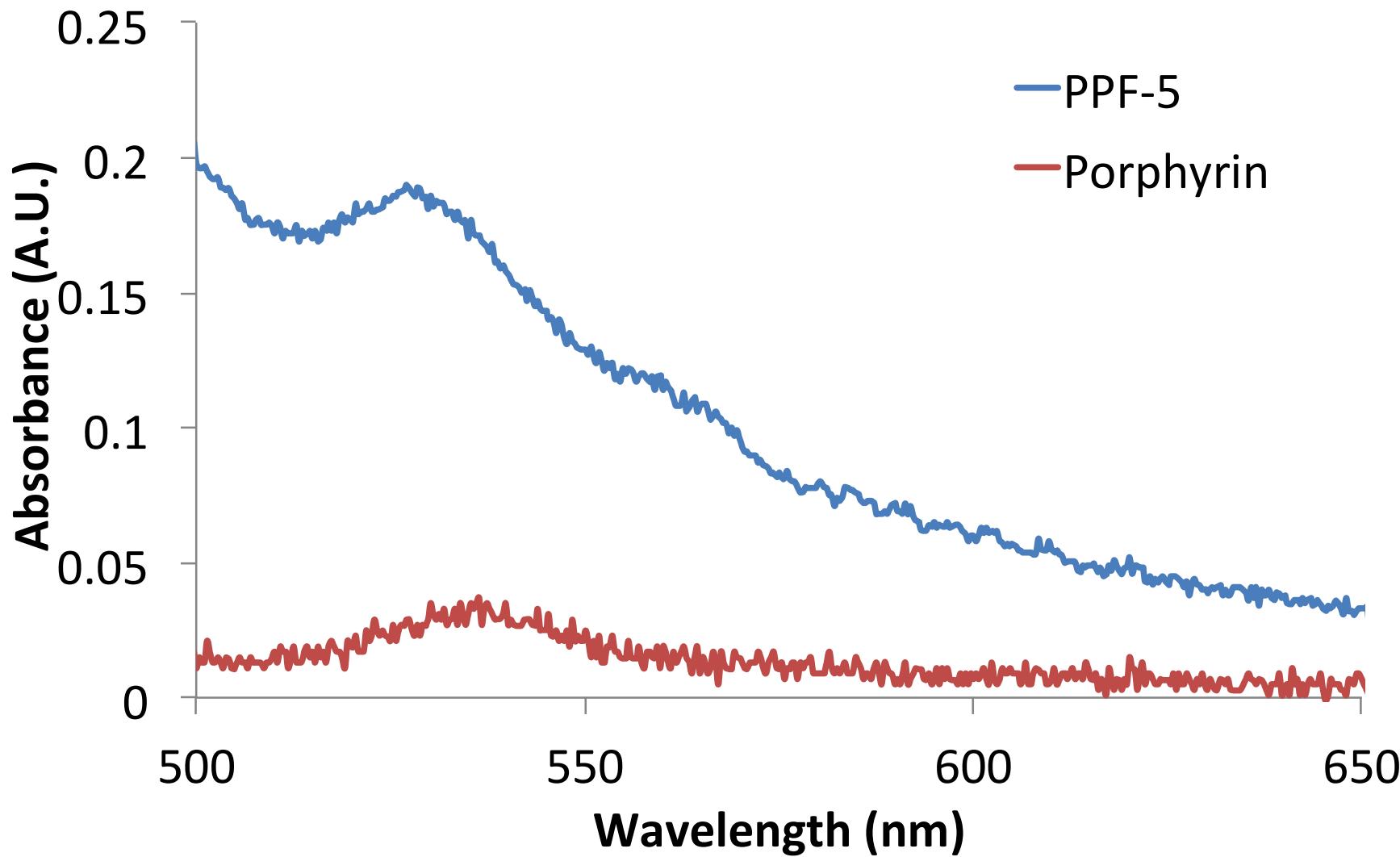
$\eta \text{ (\%)} = 0.026 \pm 0.0038$

# DFT Predictions of Band Alignment

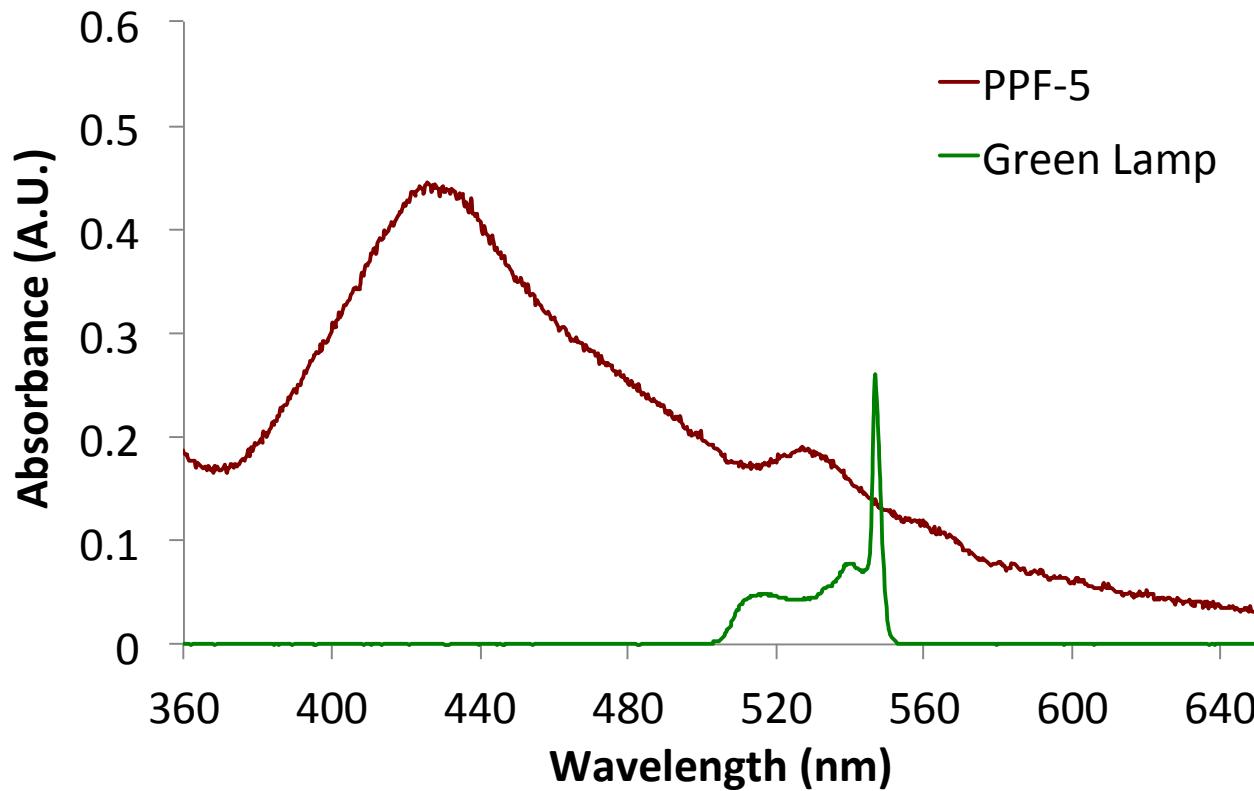
Density Functional Theory (DFT) predicts a shift in band alignment of PPF-5 relative to the Pd-porphyrin.



# Absorbance Changes in PPF-5



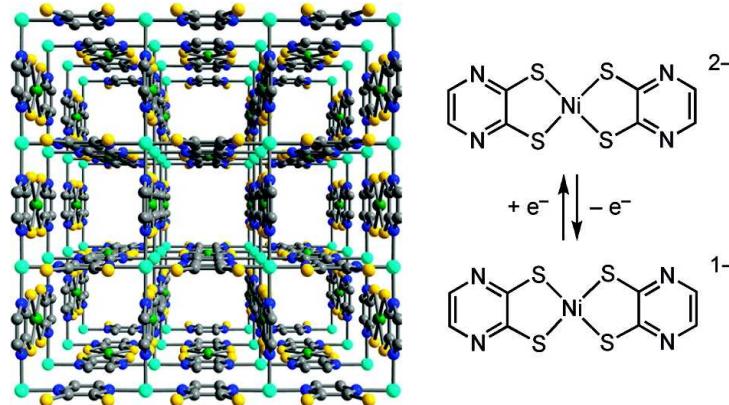
# Optical “Cheating”



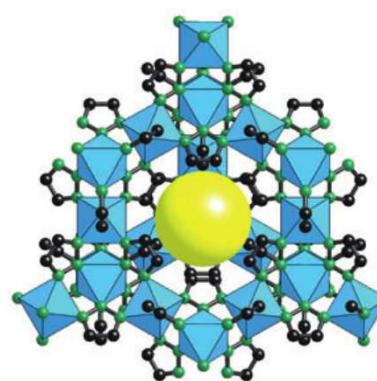
*PPF-5 absorbance aligns reasonably with Hg-arc lamp.* 39

# Electrically conducting porous MOFs are rare

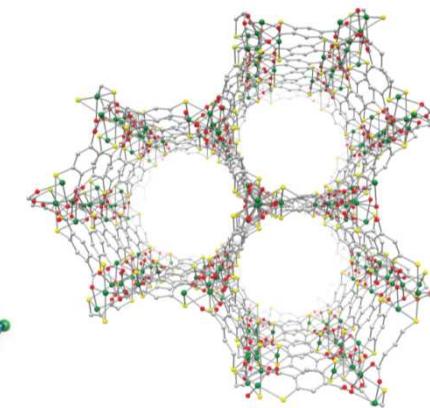
- **p-type Cu-Ni Dithiolene MOF**
  - First semiconducting, porous MOF
  - Conductivity increases with oxidative doping
  - Original Cu-Cu version is not porous  
(*Inorg. Chem.* 2009, 48, 9048)
- **Other examples**
  - MET-3 (Fe-triazolate MOF)
  - Mn(thiophenol) MOF:  $(-\text{Mn}-\text{S}-)^\infty$  Chains
- **Strategies for conducting MOFs:**
  - Charge delocalization
  - 2<sup>nd</sup>- and 3<sup>rd</sup> row transition metals
  - Redox-active ligands (e.g., TCNQ)
  - Soft ligands (e.g. S-containing molecules)



Y. Kobayashi et al. *Chem. Mater.* 2010, 22, 4120



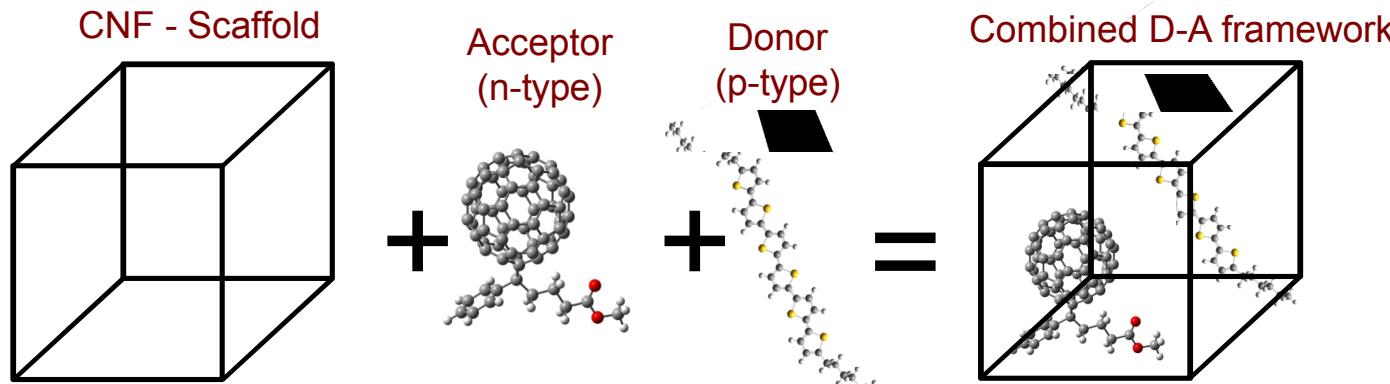
MET-3 (Fe)  
Gándara et al.  
*Chem. Eur. J.* 2012,  
18, 10595



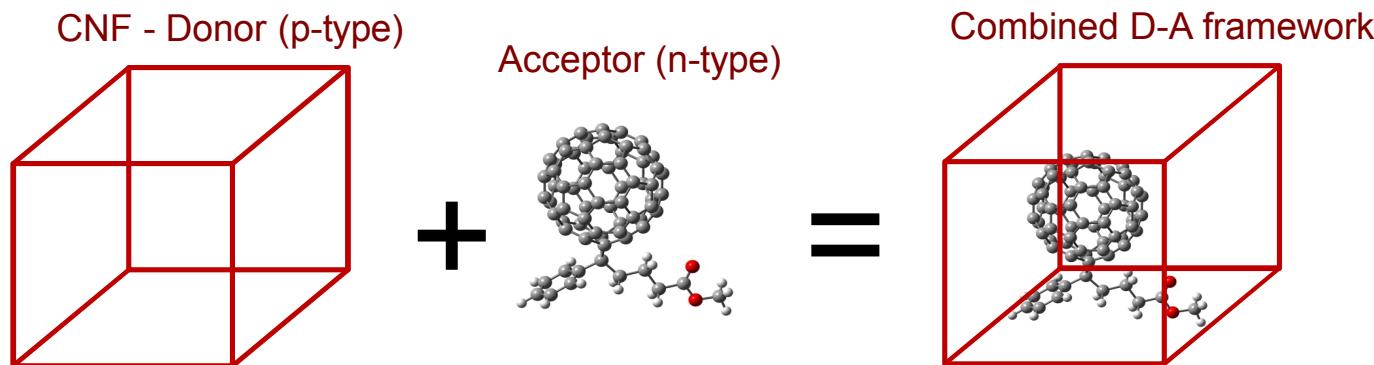
Mn(thiophenol) MOF  
L. Sun et al.  
*J. Am. Chem. Soc.*  
2013, 135, 8185 40

# Composite MOF Scaffolds for OPV

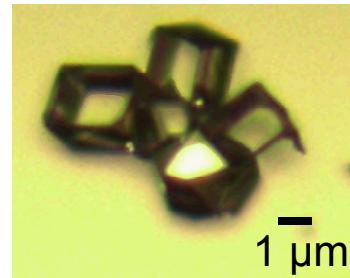
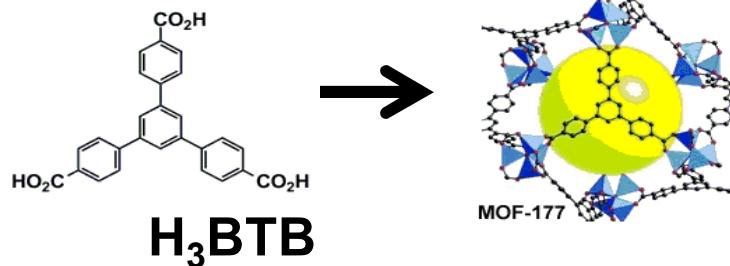
**Passive Scaffolds** - the CNF simply functions to order the donor/acceptor materials and plays no active role in the PV energy conversion process.



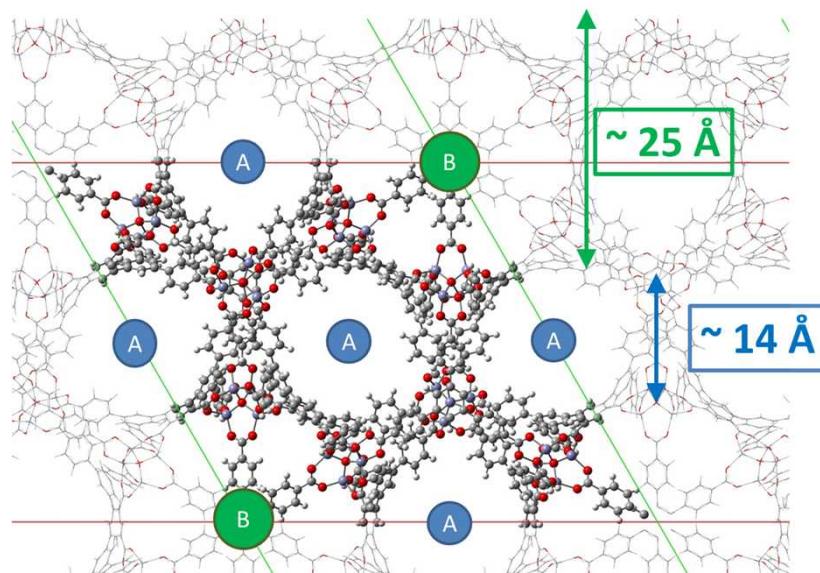
**Active Network** – the CNF is designed to play an active role in the PV energy conversion process by functioning as the donor or acceptor material.



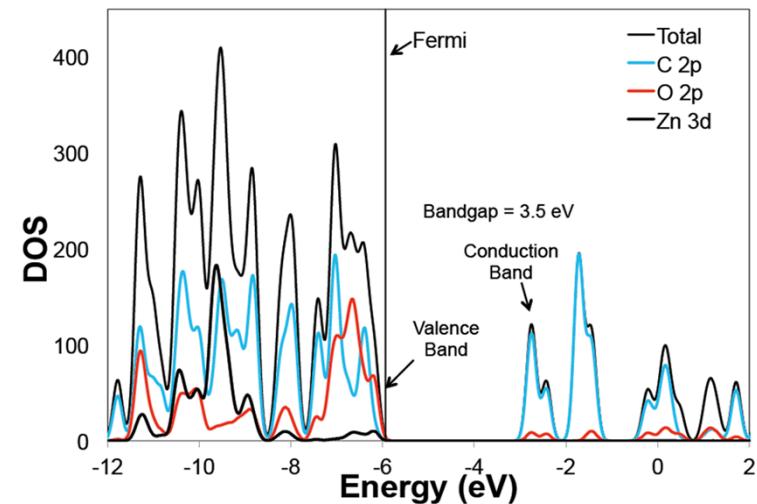
# MOF177: A Passive Scaffold for D-A Assembly



Optical image  
of MOF-177  
crystals.



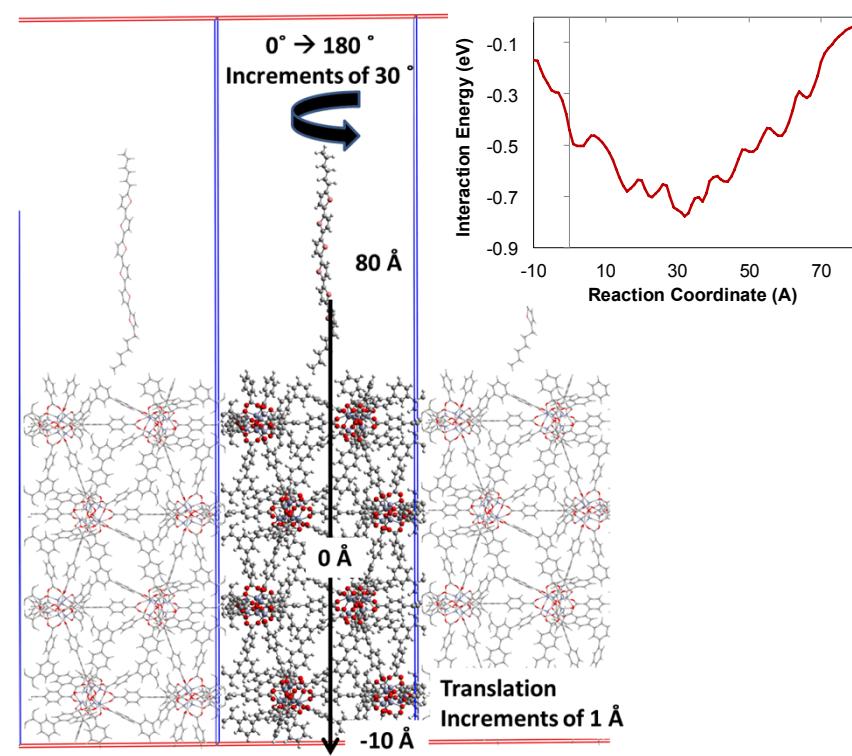
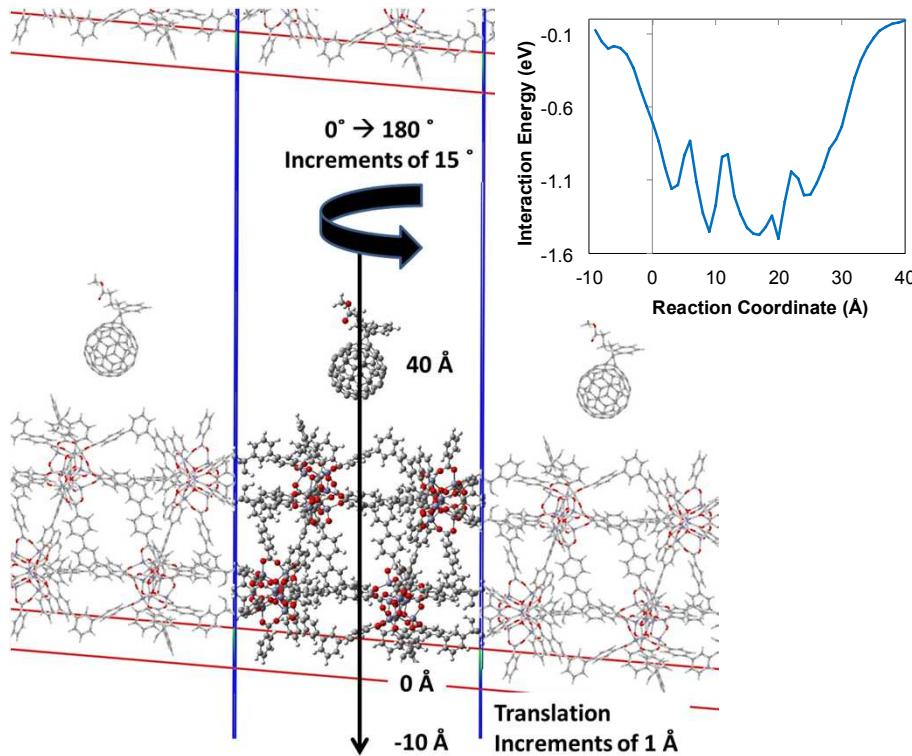
MOF-177 – sites “A” and “B” denote unique cavities; DFTB optimized structure.



Partial Density of States (PDOS) -  
Density Functional Tight-Binding (DFTB)  
calculations of MOF-177.

# Predicting Guest Infiltration

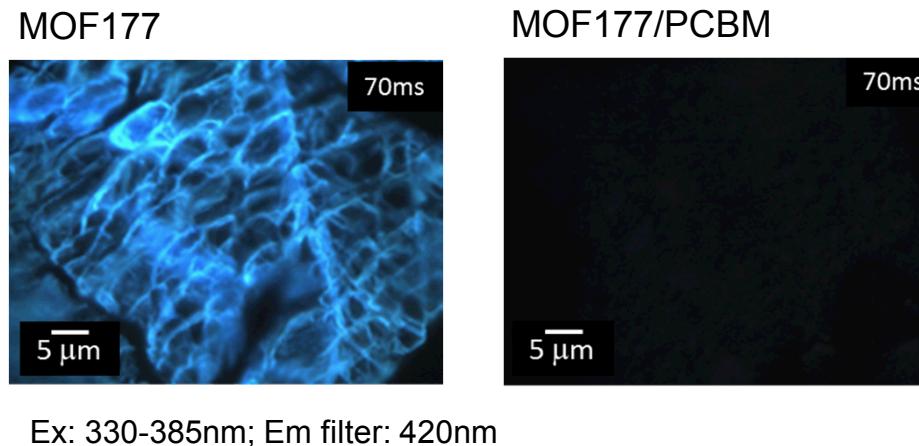
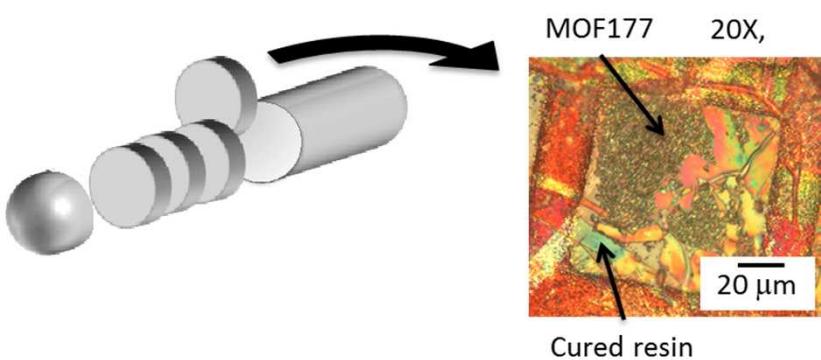
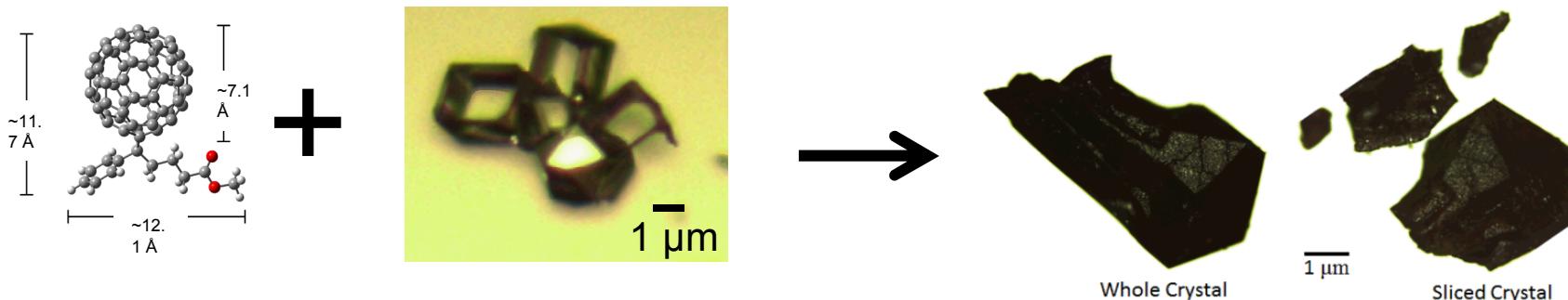
*Density Functional simulations show that infiltration of both PCBM (A) and Sexithiophene (D) in MOF177 are enthalpically favored.*



>600 structures generated for each case; 10 step geometry optimization performed to remove close contacts. Interaction energies determined using Density Functional based Tight Binding (DFTB) method.

# PCBM Integration into MOF177

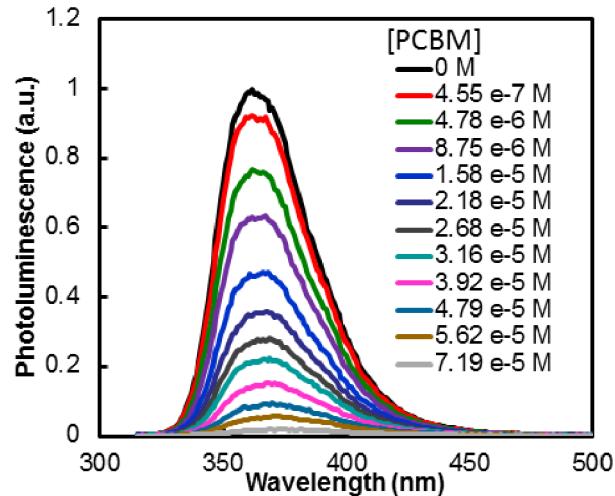
*Incubation of MOF177 crystals in concentrated PCBM solutions leads to PCBM infiltration*



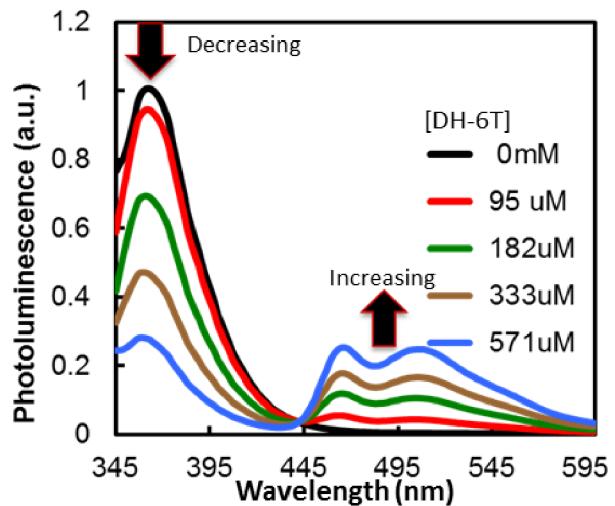
Spectroscopic characterization of PCBM@MOF177 cross-sections shows significant quenching of MOF177 fluorescence.

# Introduction to MOFs

*Spectral characterization reveals complex energy transfer between MOF177, PCBM, and DH6T*

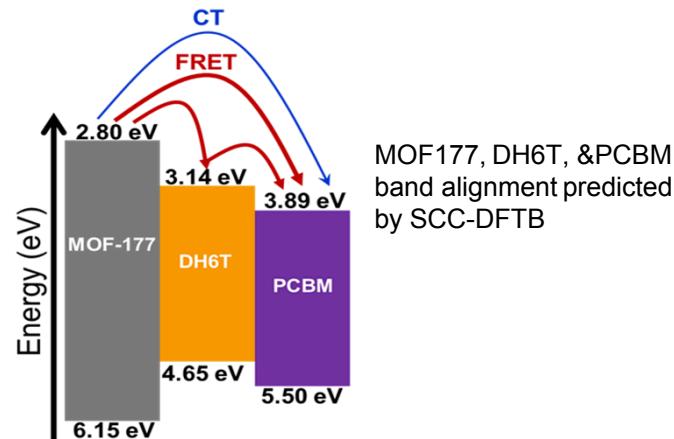
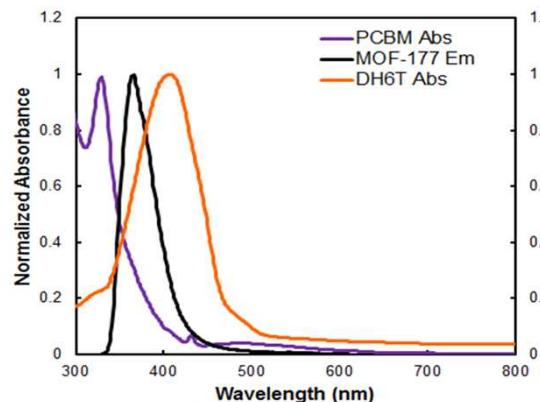


MOF177 and  
DH-6T are  
quenched by  
PCBM.



MOF177  
transfers  
energy to DH-  
6T.

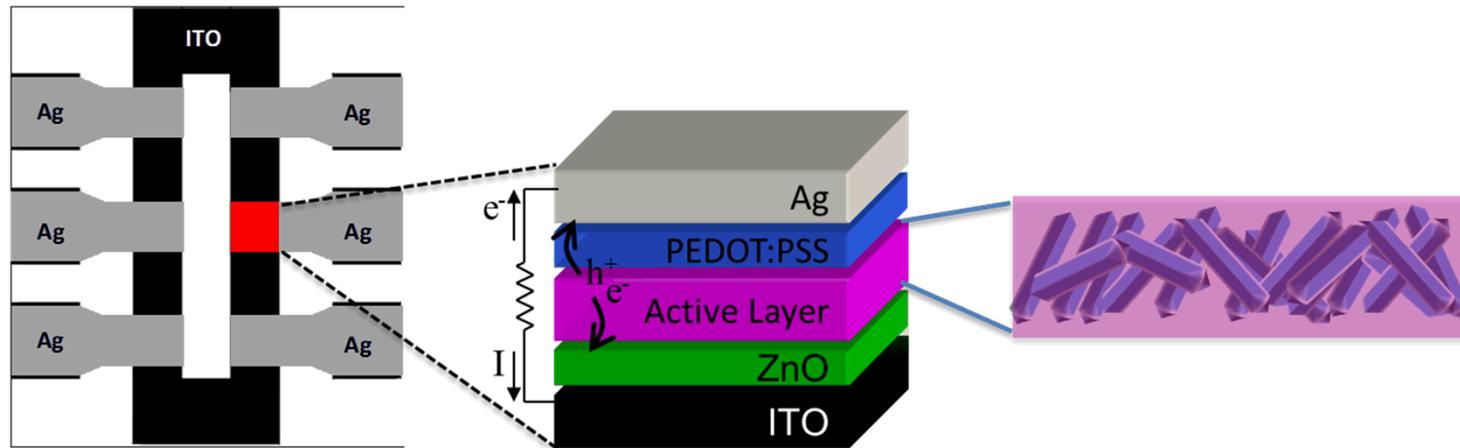
Spectral overlap between DH-6T, PCBM, and MOF177 inform multiple optoelectronic relationships.



MOF177, DH6T, &PCBM  
band alignment predicted  
by SCC-DFTB

# Device Integration

PCBM@MOF177 were incorporated into hybrid OPV active layers to evaluate the influence of MOF templating on PV performance.

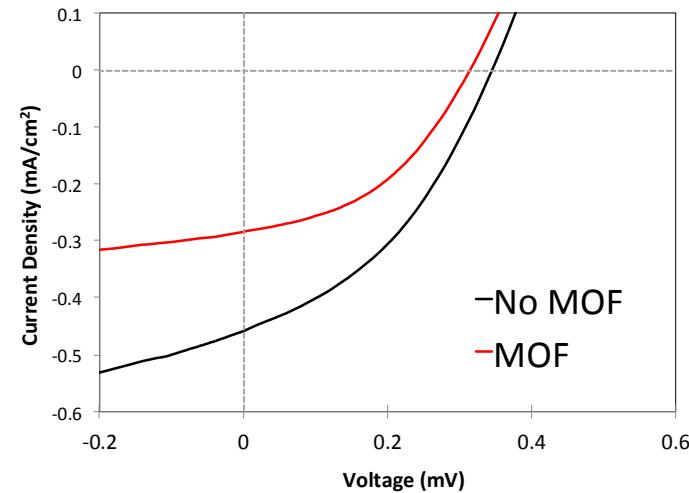
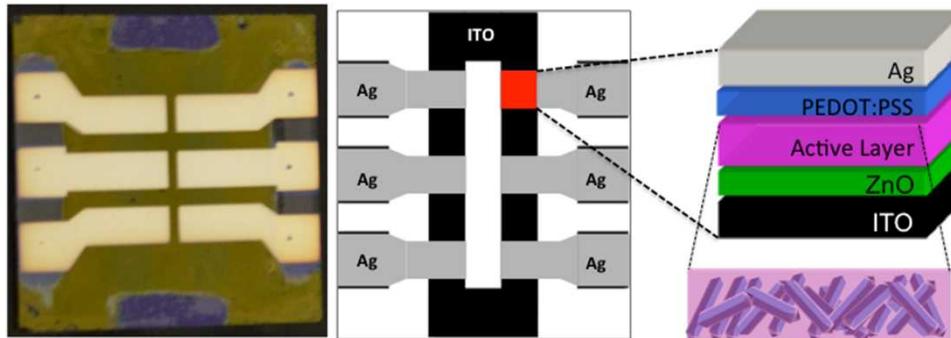


Patterned ITO and silver electrodes create controlled  $0.1\text{cm}^2$  active areas.

Inverted device configuration

Infiltrated (e.g., oligothiophene, PCBM) MOF177 particles incorporated into polythiophene active layer.

# Device Testing



	V <sub>OC</sub> (mV)	J <sub>SC</sub> (mA/cm <sup>2</sup> )	FF (%)	Efficiency (%)
With MOF177	320 +/- 23	0.290 +/- 0.018	40 +/- 0.3	0.04 +/- 0.003
P3HT Alone	340 +/- 26	0.460 +/- 0.029	39 +/- 1.9	0.06 +/- 0.01

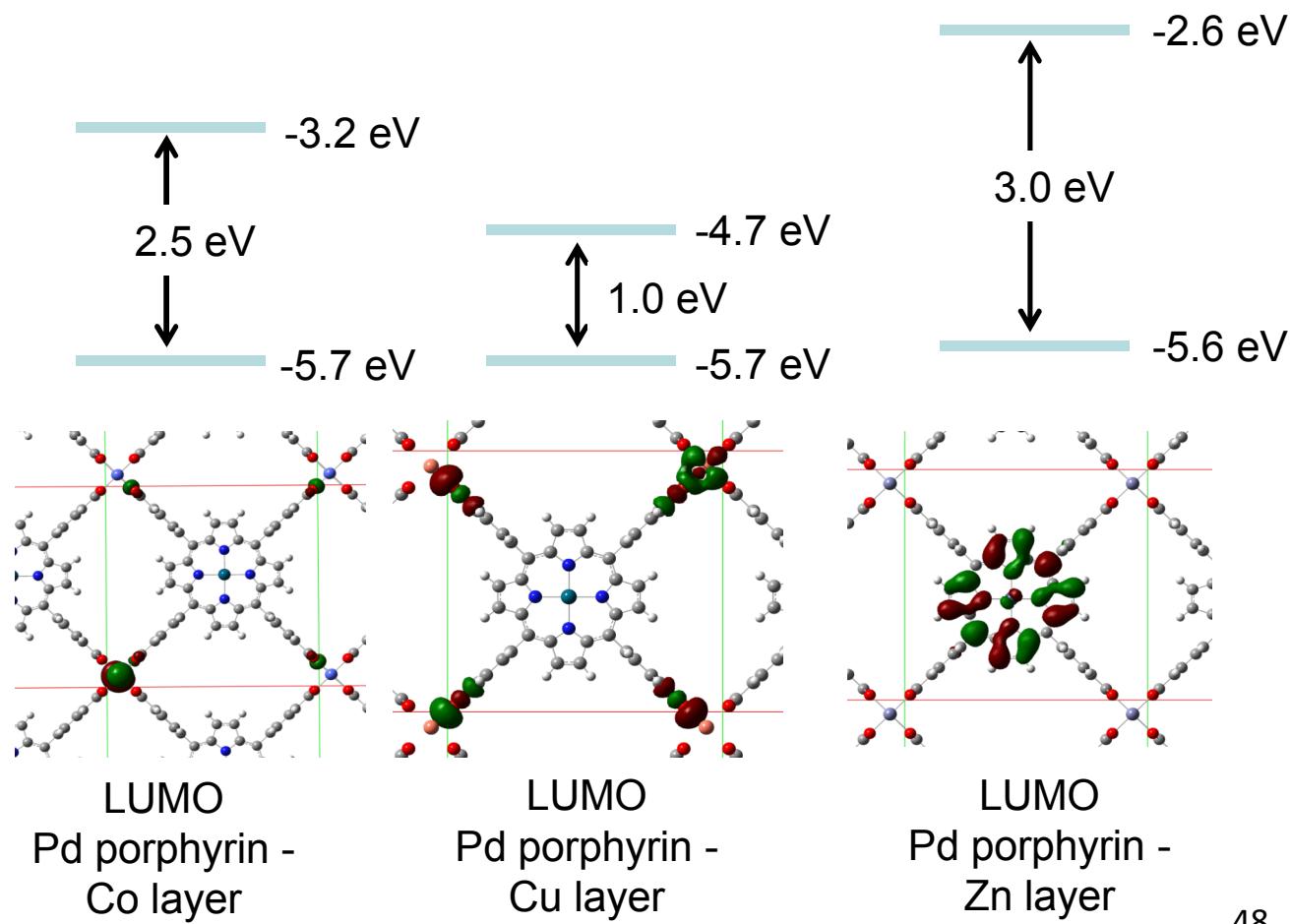
PV testing with 1 sun illumination shows reduced current, possibly from reduced active volume or charge trapping in suspended PCBM@MOF177 composites.

# Band Structure Tailoring

*Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.*

## Varying transition metal ions

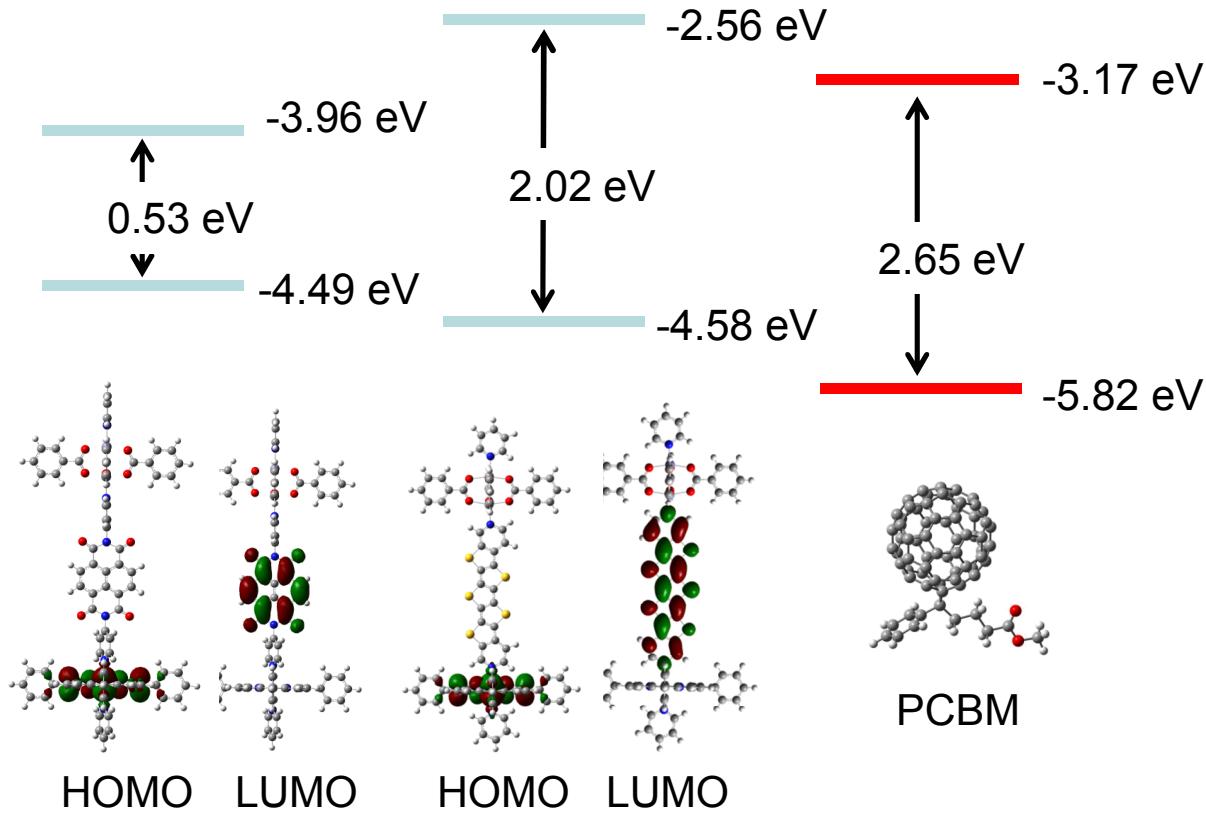
2D periodic  
optimization -  
DFT(B3LYP/ CEP-  
31G)



# Band Structure Tailoring

*Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.*

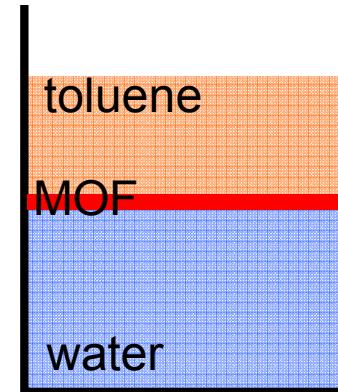
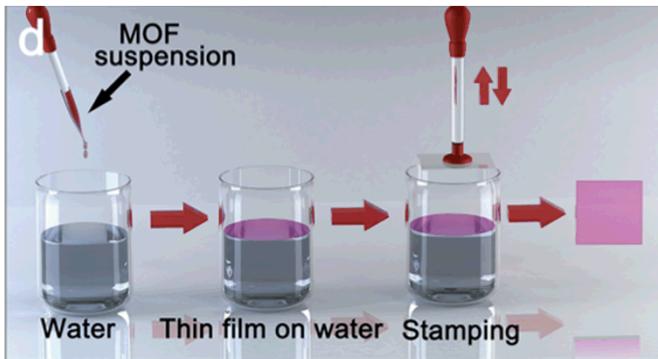
## Varying organic pillars



2D periodic optimization -  
DFT(B3LYP/ CEP-31G)

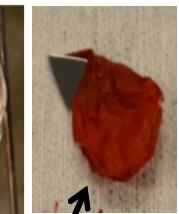
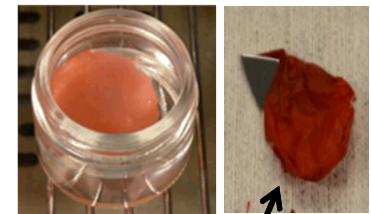
# Revisiting Layered Growth

*Can we deposit a thicker active layer to increase our current?*



G. Xu, et al., JACS 2012, 134, 16524

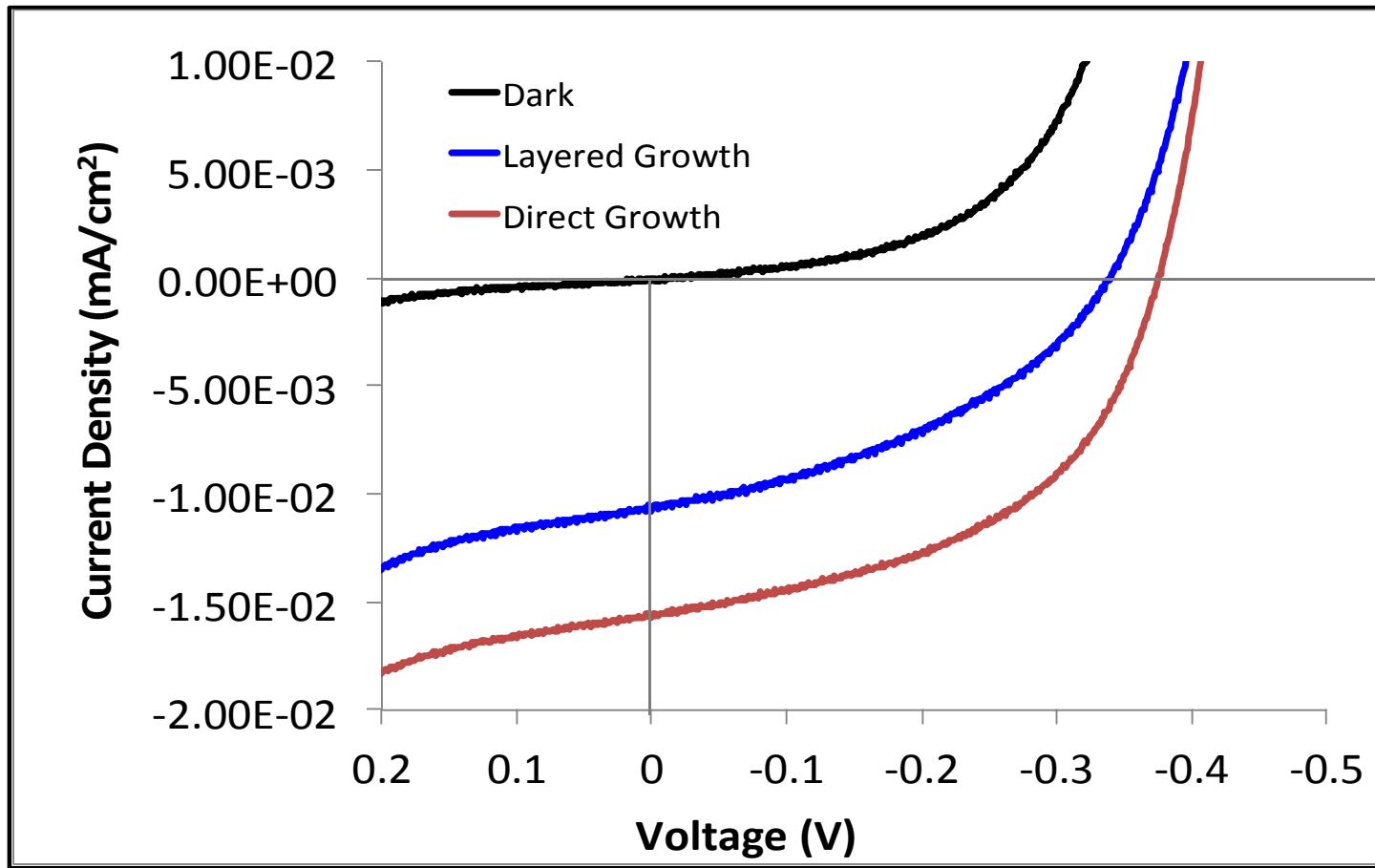
Stamped onto  
OTS/APS-treated  
Si  
(toluene +  
acetone, 1:4)  
Untouched on  
water surface for  
48 hours  
(toluene + acetone,  
1:4)



Self-supporting film!

# PPF-5 Integration is Important!

*PPF-5 grown directly on  $TiO_2$  yields greater PV response than PPF-5 deposited by layered growth*



## Requirements for efficient OPV system:

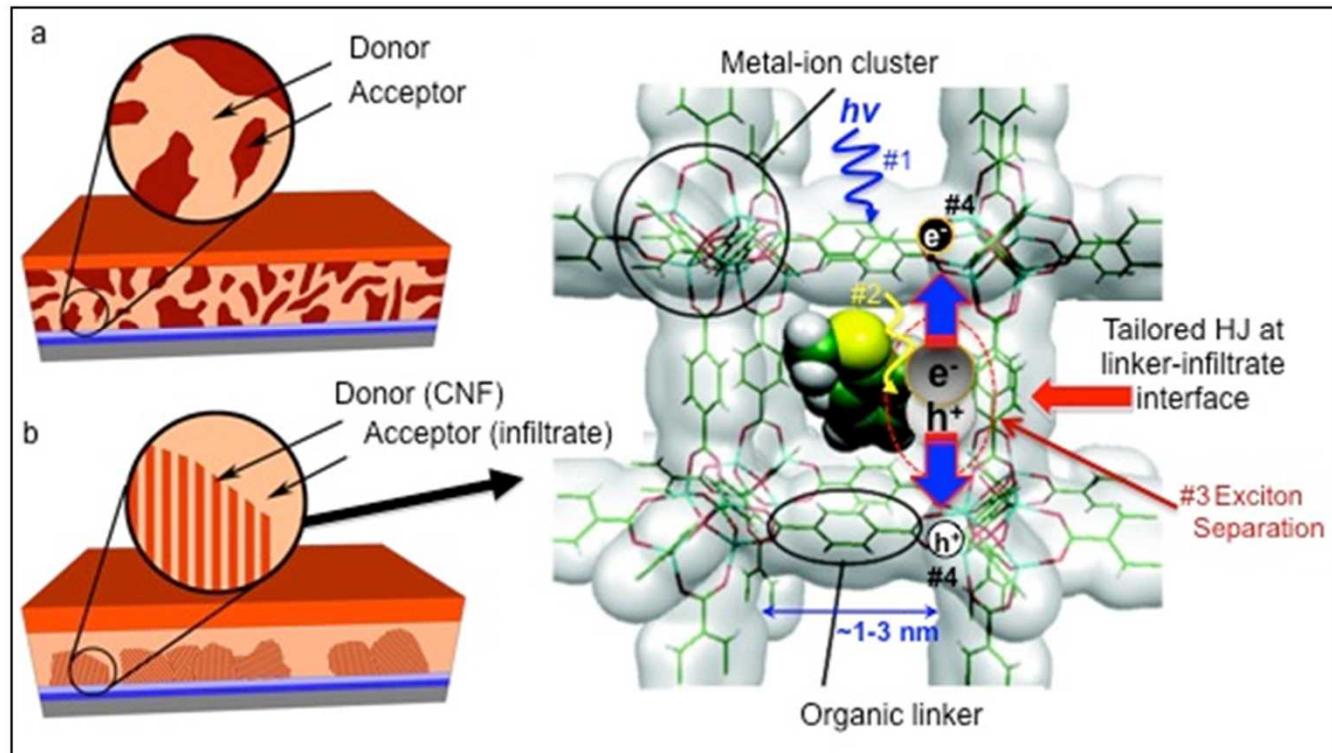
- Good p-type absorber
- Short exciton diffusion distances
- Ordered molecular charge separation interfaces
- Proper band alignment (D-A interface) for charge separation/transfer
- Facile incorporation into device architectures!

# A Supramolecular Approach to PV Integration

## Order vs. disorder: creation of nano-heterojunctions

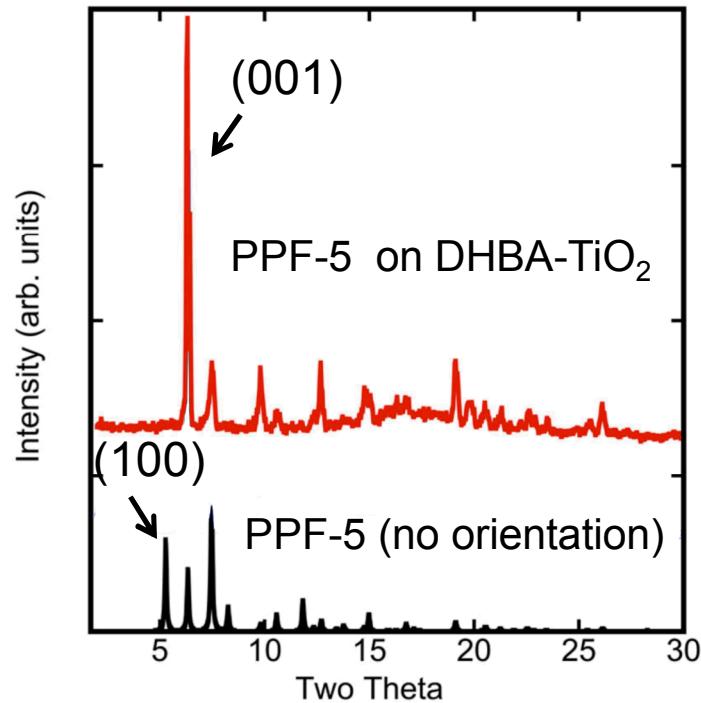
Conventional  
disordered BHJ

Highly ordered  
“Nano-HJ” using  
CNF platform

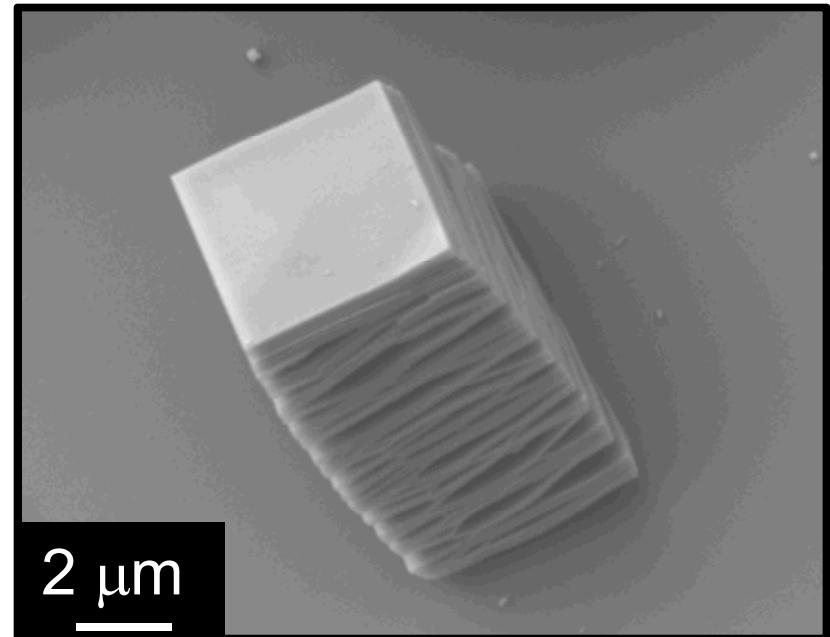


# Oriented Solvothermal PPF-5

*Solvothermal PPF-5 is oriented, appearing as stacks of MOF-sheets.*

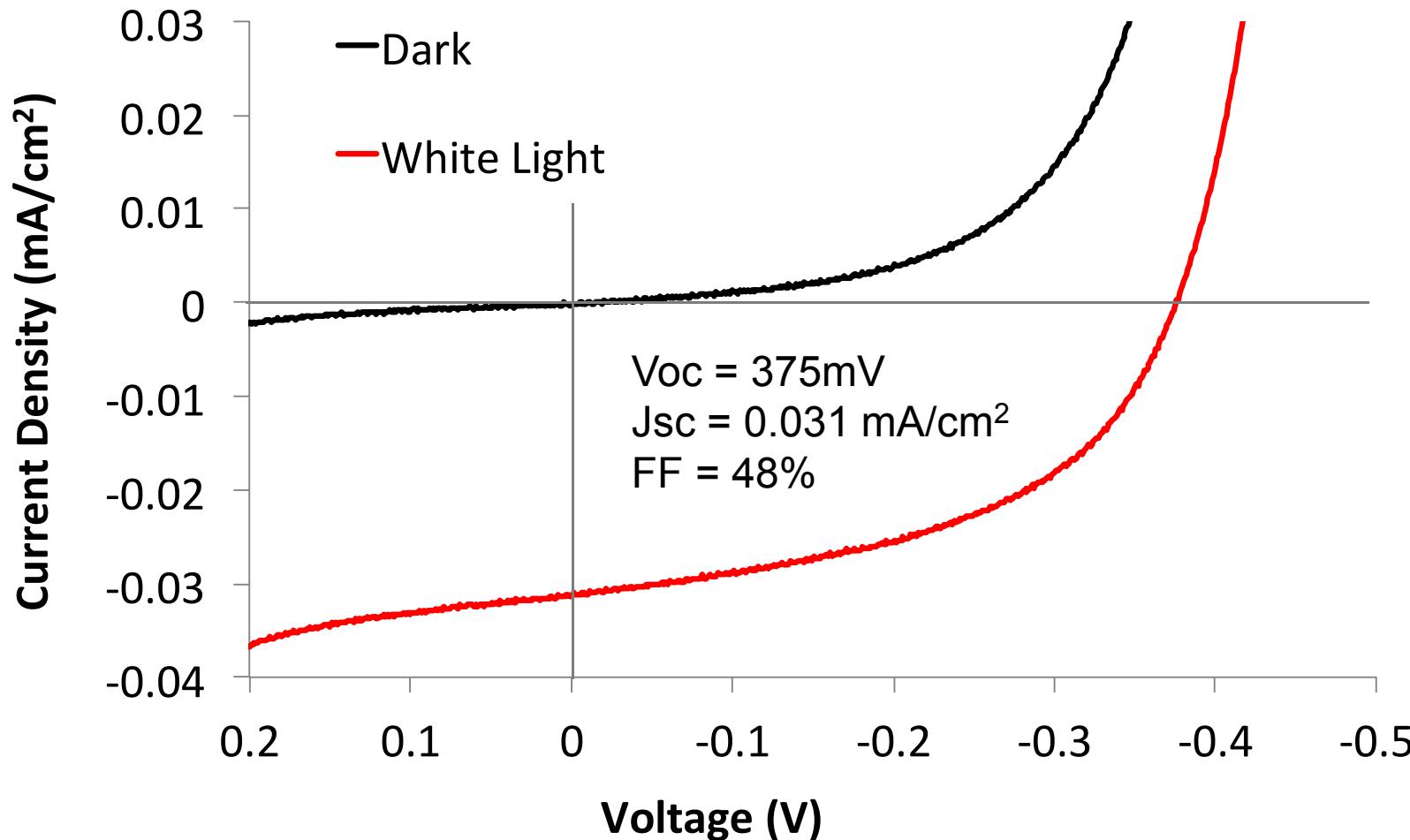


Grazing Incidence XRD shows growth of PPF-5 with preferential (001) orientation (red).

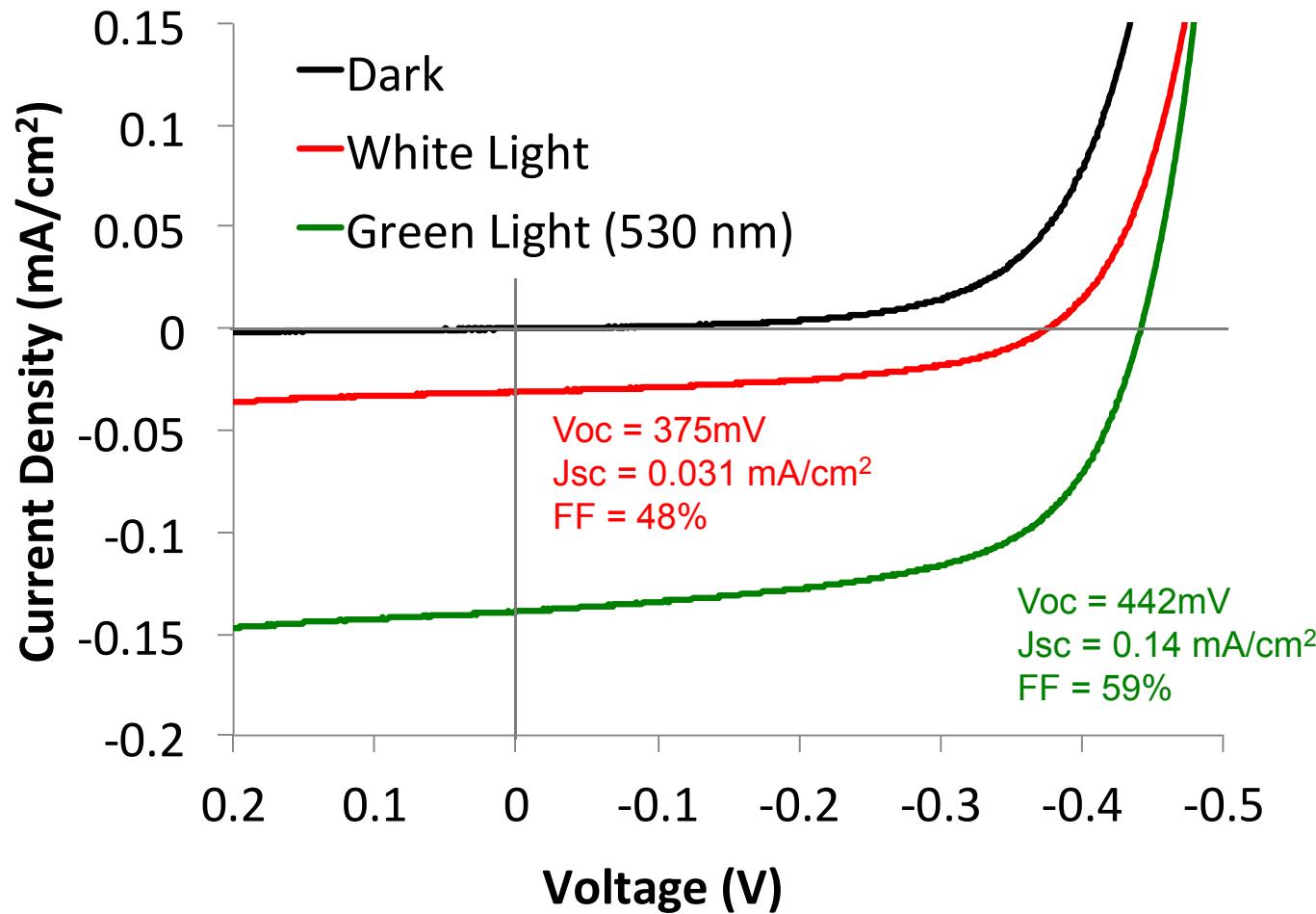


Scanning electron micrograph of PPF-5 stacks

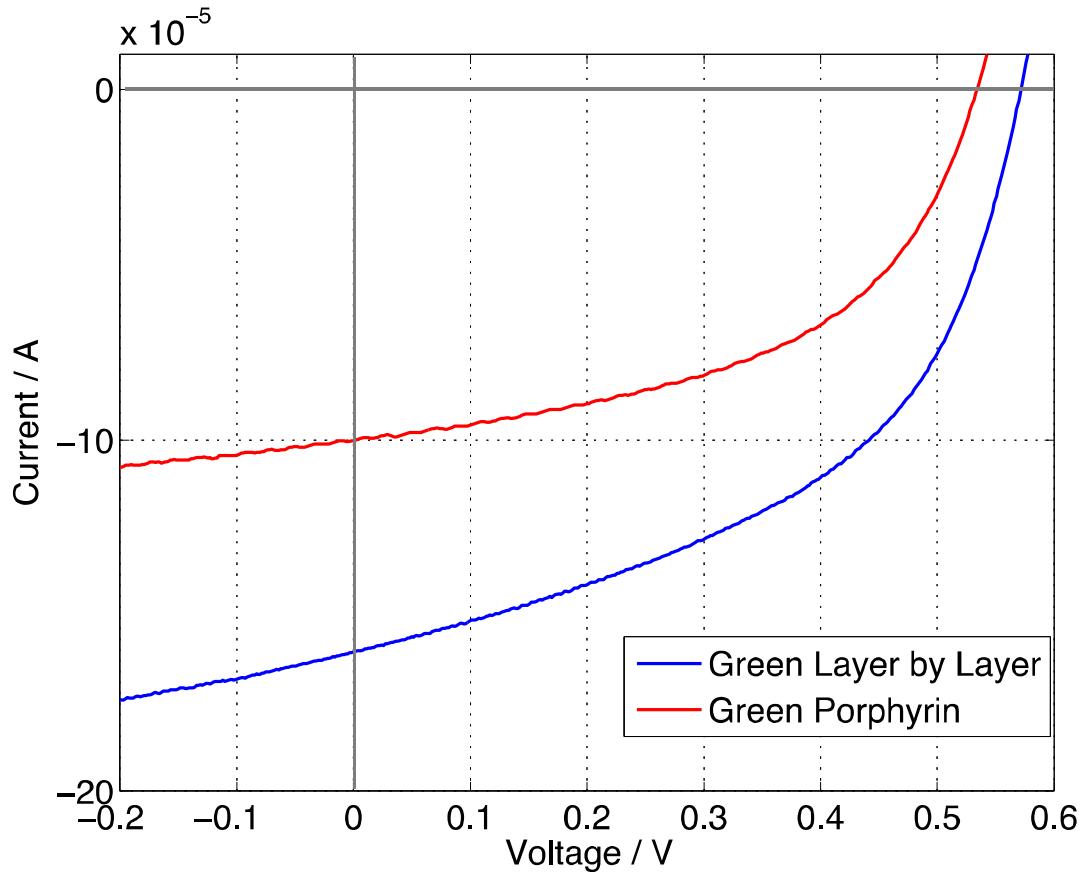
# PPF-5 PV Performance in a DSSC



# Optical “cheating”



# Performance of PPF-5 in a DSSC

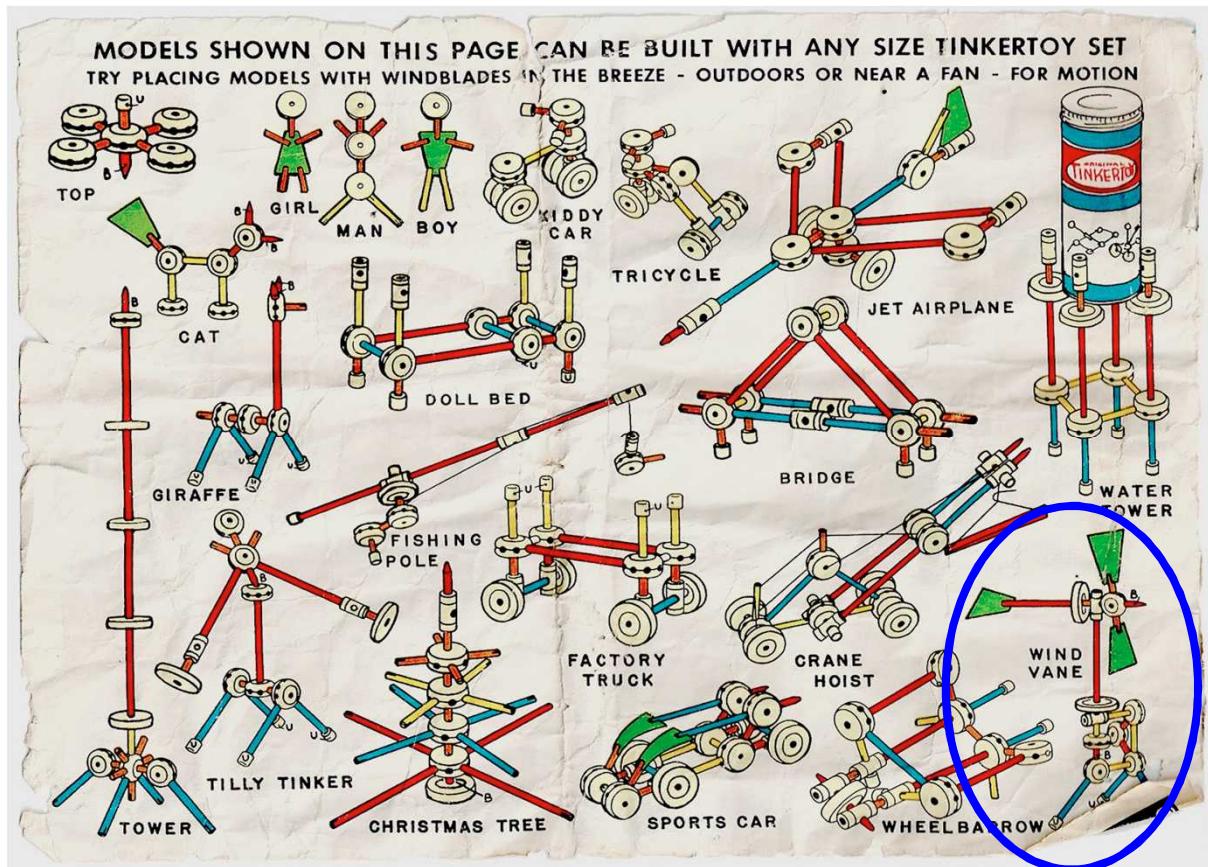


$V_{oc} = 0.57V$   
 $J_{sc} = 1.6E-4 A$   
 $FF = 0.49$

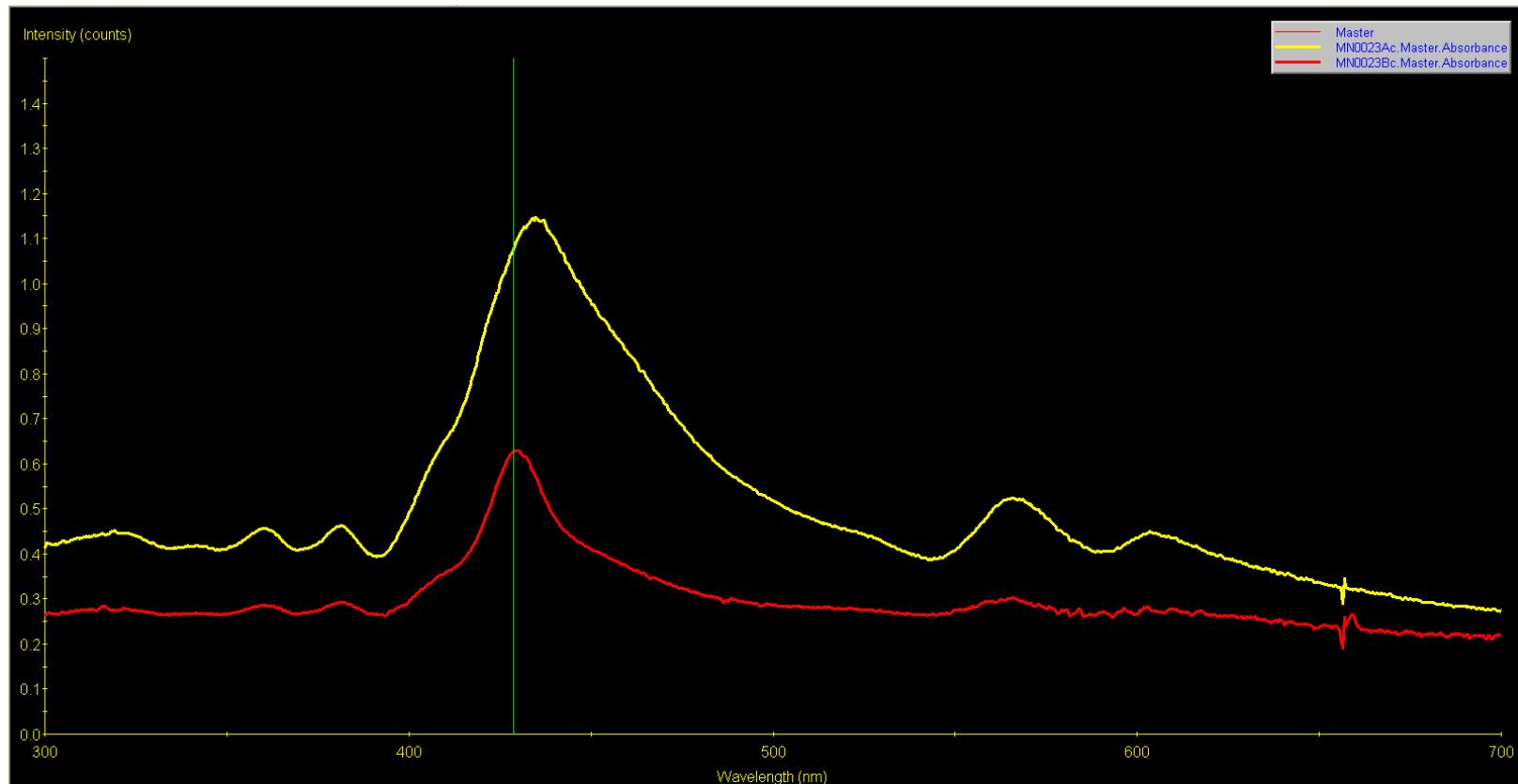
$V_{oc} = 0.532V$   
 $J_{sc} = 1 E-4 A$   
 $FF = 0.506$

# MOFs: Supramolecular “Tinker Toys”

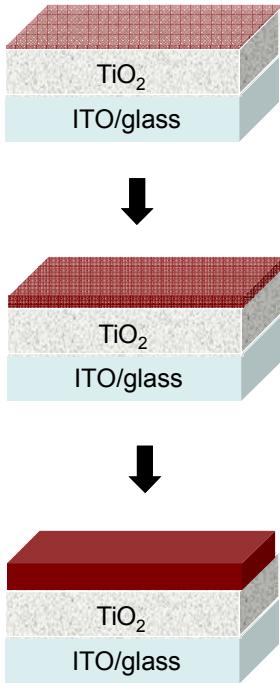
*Can MOFs be exploited for renewable energy?*



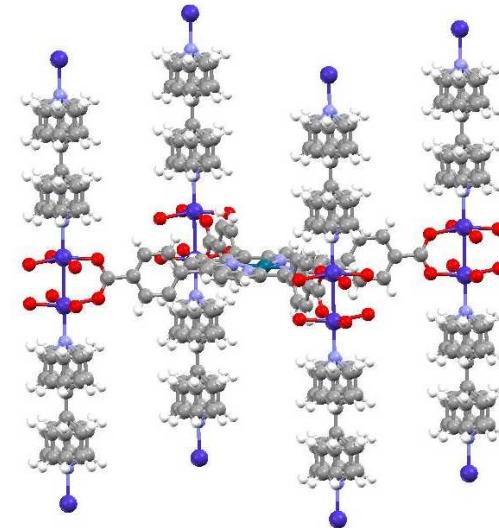
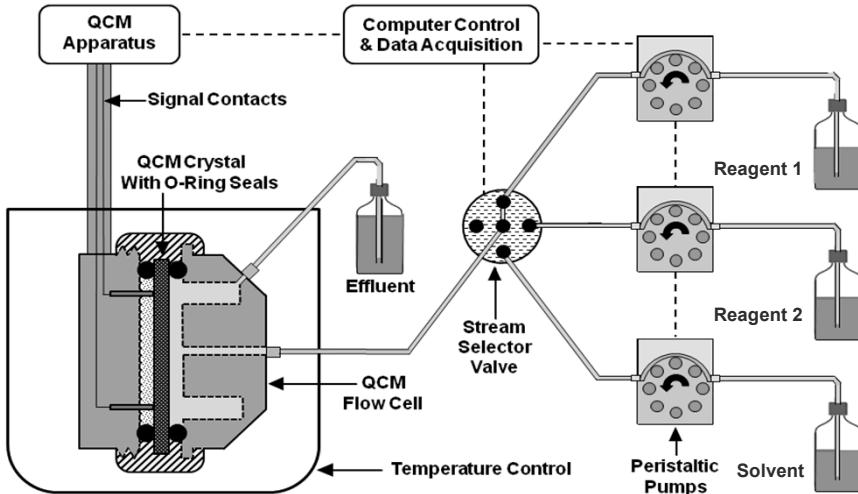
# PPF-18 Photoluminescence



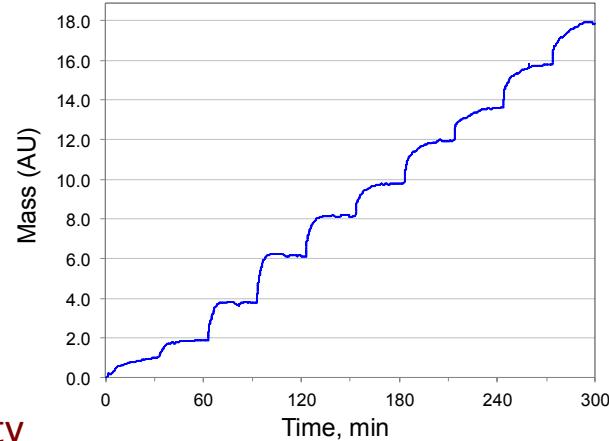
# Layer-by-Layer MOF Growth



1. Deposit/anneal TiO<sub>2</sub> nanoparticle slurry (DeGussa P25) on ITO/glass
2. Pretreat TiO<sub>2</sub> with Co<sup>2+</sup>
3. Sequentially introduce
  - Co<sup>2+</sup>-ligated porphyrin linker
  - Solvent wash
  - Bipyridine pillar
  - Solvent wash



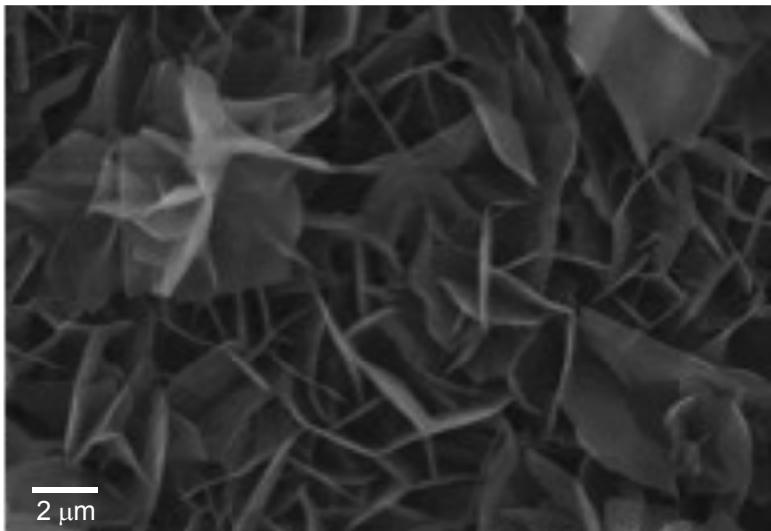
Step by step PPF-5 growth



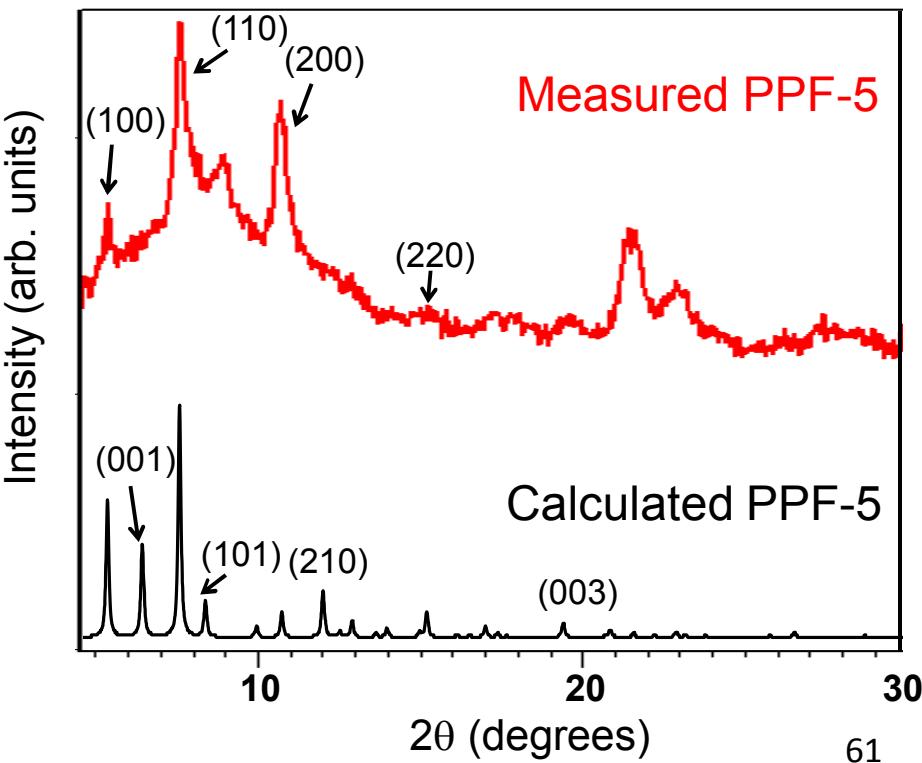
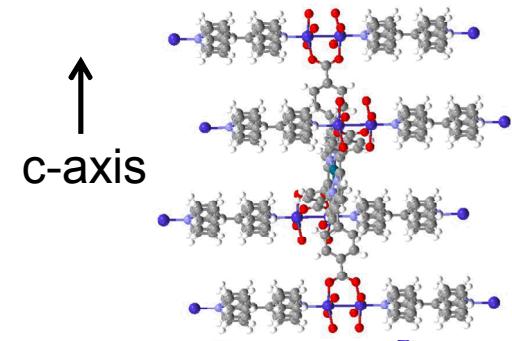
Schematic representation of automated MOF film growth/QCM capability

# Characterization of PPF-5 Films on $\text{TiO}_2$ /ITO

LBL-growth produces a porous array of PPF-5 crystals exhibiting crystallographic orientation on the  $\text{TiO}_2$  surface.



Scanning electron micrograph shows PPF-5 "platelets" extending off the  $\text{TiO}_2$  surface.



# Integration of PPF-5 in DSSCs

PPF-5 should be readily incorporated into a DSSC device configuration...

