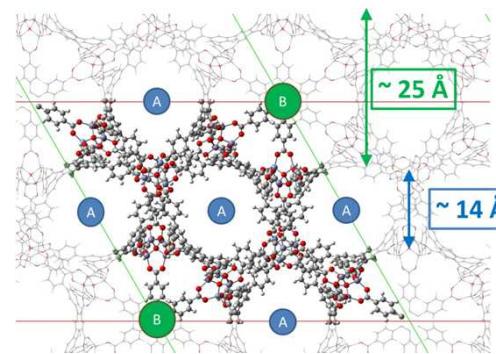


Exceptional service in the national interest



Solar MOFs: Supramolecular Materials to Enable Next Generation Dye-Sensitized Solar Cells

Erik D. Spoerke

Leo Small, Steven Wolf, Jill Wheeler

Michael Foster, Vitalie Stavila, Kirsty Leong, and Mark D. Allendorf

Energy Materials Nanotechnology 2015
on Photovoltaics



Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

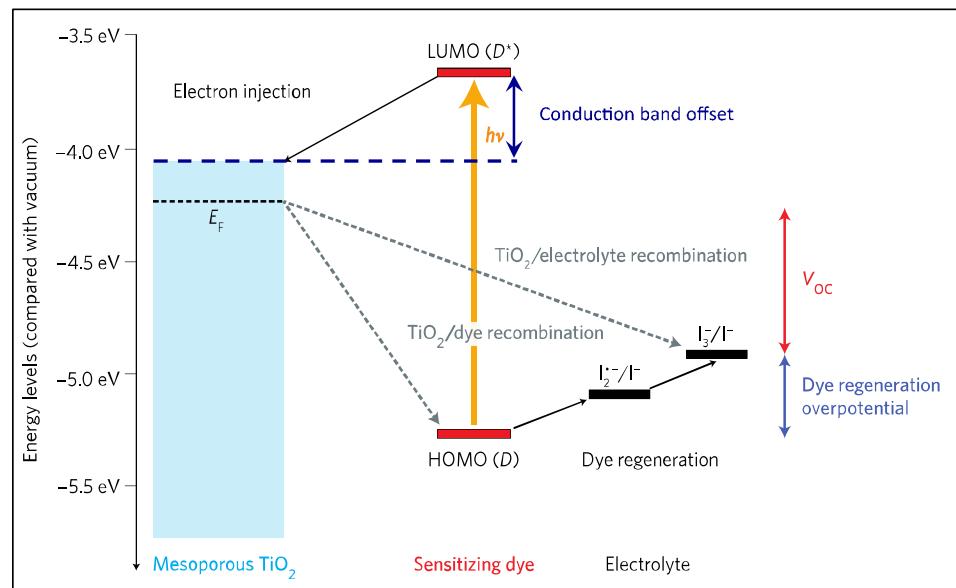
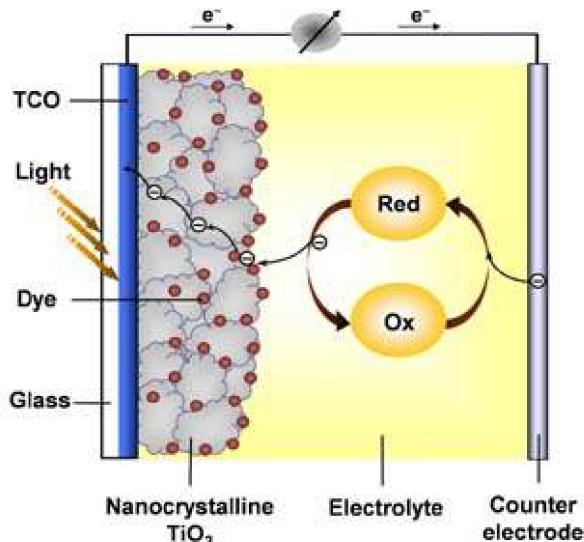
Technical Challenge

Dye-Sensitized Solar Cells

(O'Regan, B. & Grätzel, M. (1991). *Nature*, **353**, pp 737.)

World record is >15% efficiency...but there are still some critical challenges:

- Limited light harvesting
 - Spectral range
 - Dye concentration (*without dye aggregation)
- Carrier lifetimes
- Band offset overpotentials

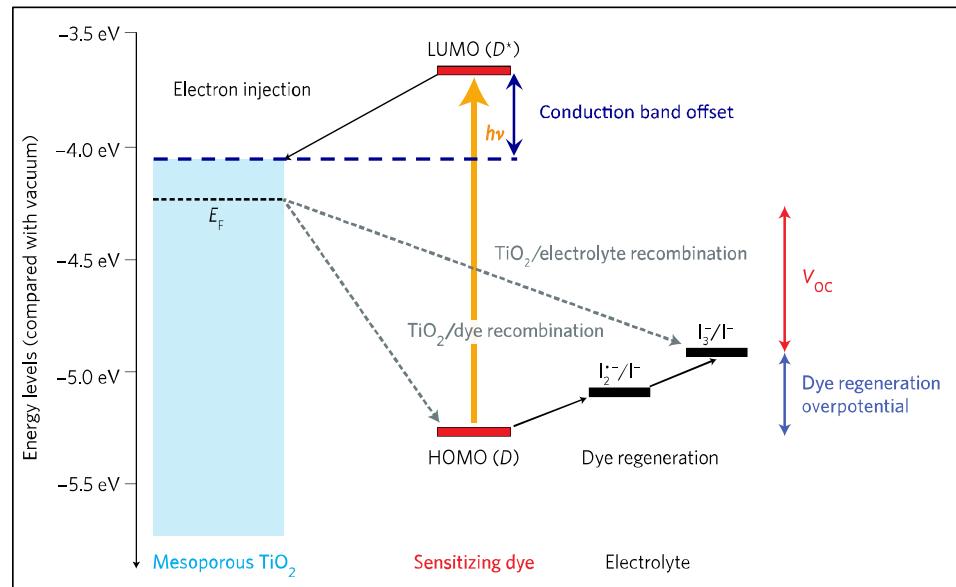
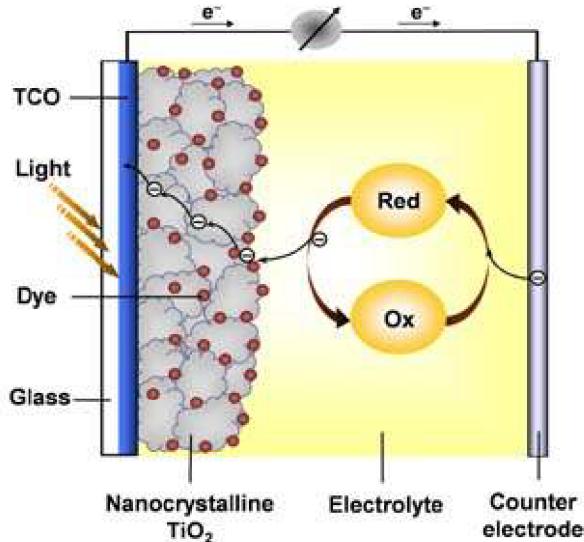


Technical Challenge

Dye-Sensitized Solar Cells

(O'Regan, B. & Grätzel, M. (1991). *Nature*, **353**, pp 737.)

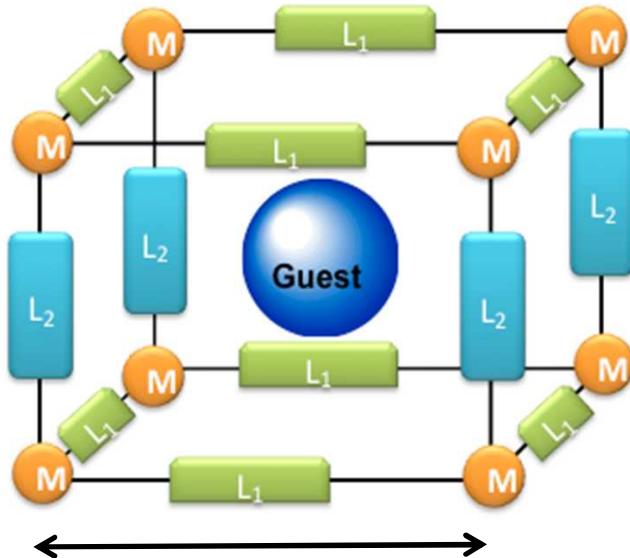
Can we use Metal-Organic Frameworks (MOFs) to address these challenges?



What are Metal-Organic Frameworks?

Metal-Organic Frameworks (MOFs)

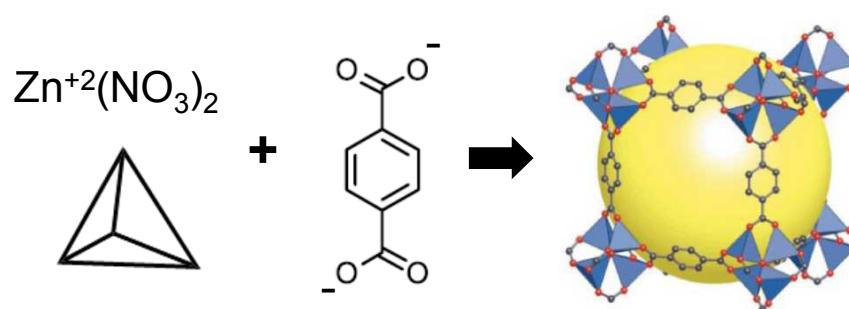
A subset of coordination polymers



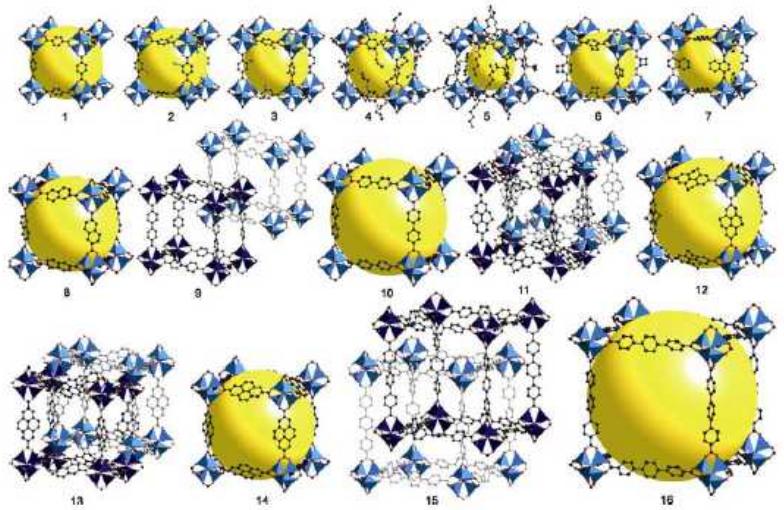
Crystalline MOF structures are composed of metal nodes (M), linkers (L_1) and pillars (L_2).

The nanoporous character of the MOF allows incorporation of molecular guests, organized on the nanoscale.

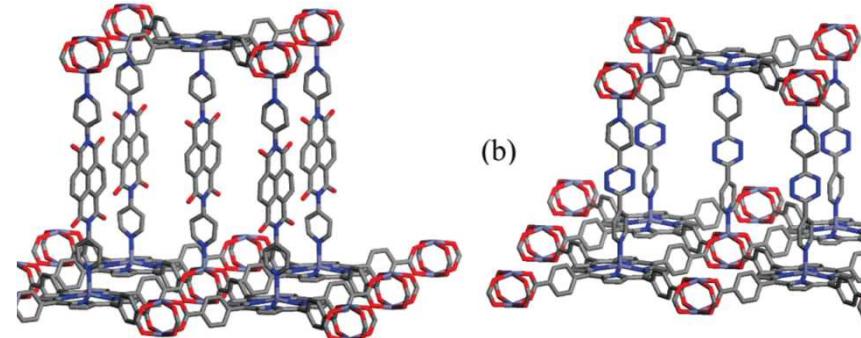
This chemically “modular” system allows for tuning of the structure, properties, and function of these materials.



MOF Building Block Chemistry Determines Crystal Structure

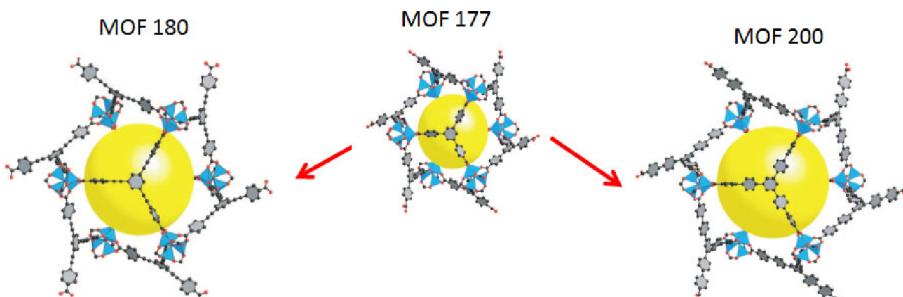


<http://yaghi.berkeley.edu/research-MOF.html>



Side view of PPF-18

Chung *et al.* *Crystal Growth & Design*, Vol. 9, No. 7, 2009

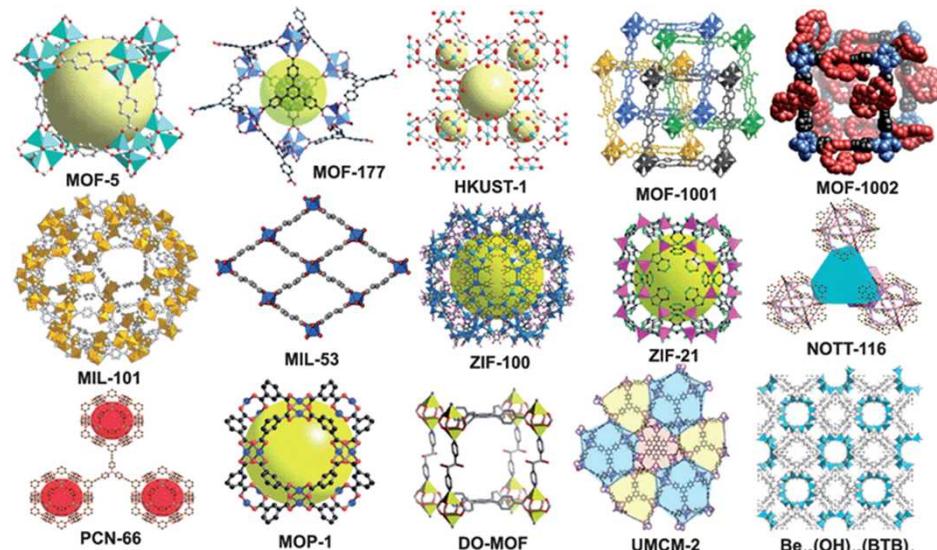


<http://www.cchem.berkeley.edu/molsim/teaching/fall2011/CCS/Group7/structure.htm>

Varying the “modular” composition of MOFs allows for tremendous flexibility of structure and function.

MOF Properties

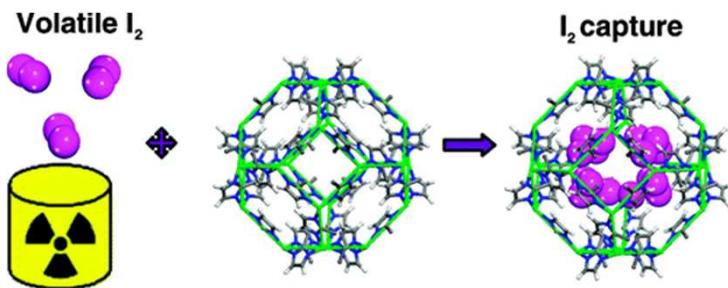
- Crystalline Order
- Nanoporosity (Guest/Host capability)
- Ultrahigh surface area (record ~ 7000 m²/g)
- Chemical, thermal, and structural stability*
- High chemical reactivity*
- Photoactivity*
- Charge transfer*



*MOF dependent

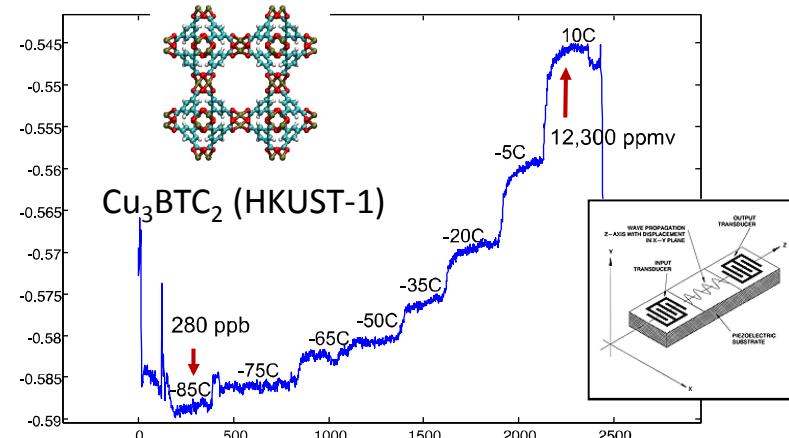
MOF Applications Span A Wide Range of Fields

Gas Sorption (I₂ capture)



Sava, et al. *JACS*, 2011, 133 (32), pp 12398–12401

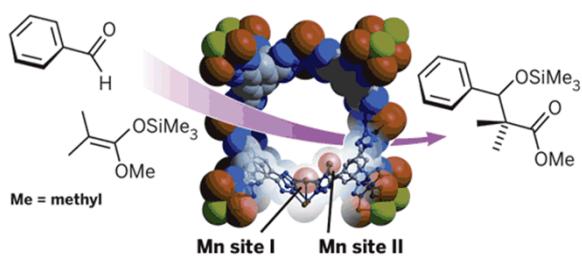
Sensing (H₂O sensing)



Robinson et al. *Anal. Chem.* 84 (2012), 7043

Catalysis (Mukaiyama aldol synthesis)

Mn-BTT



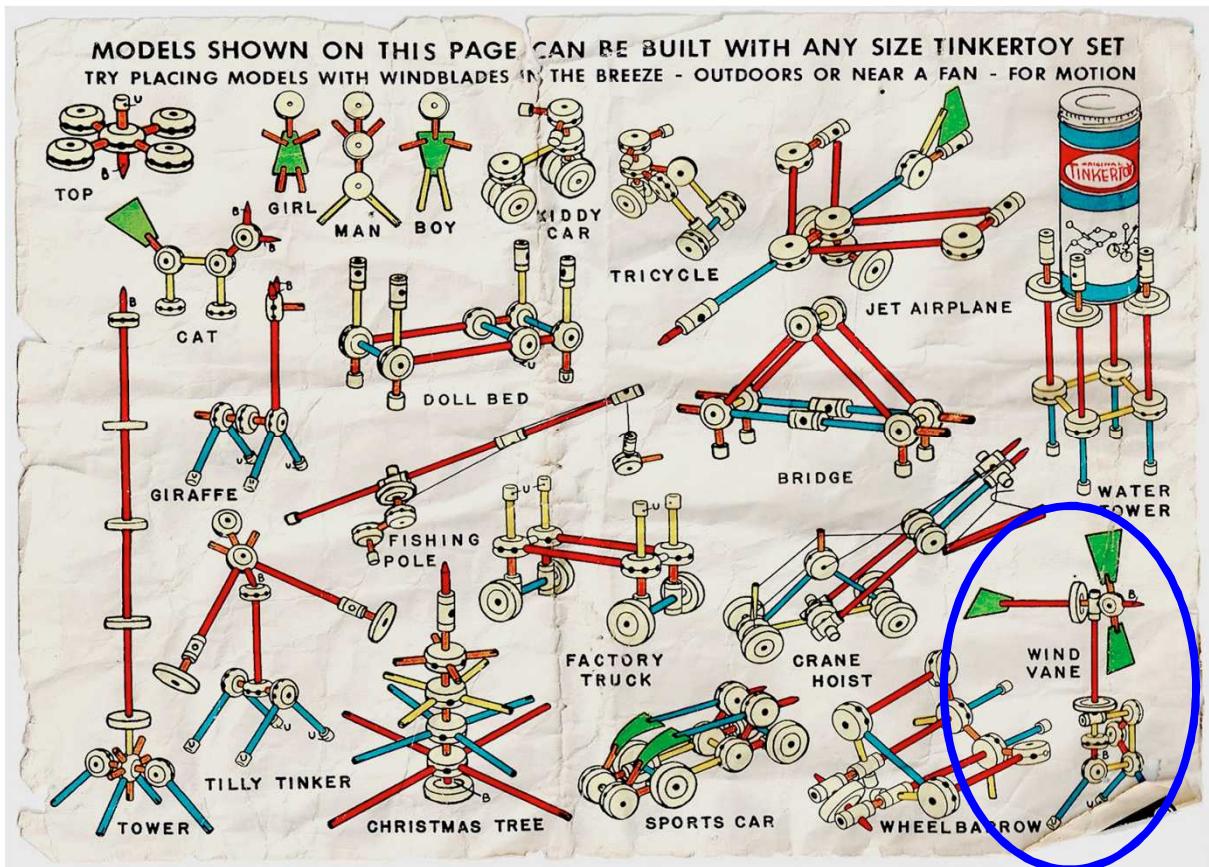
Horike, et al. *JACS*, 2008, 130 (18), pp 5854–5855

Electronics/Optoelectronics



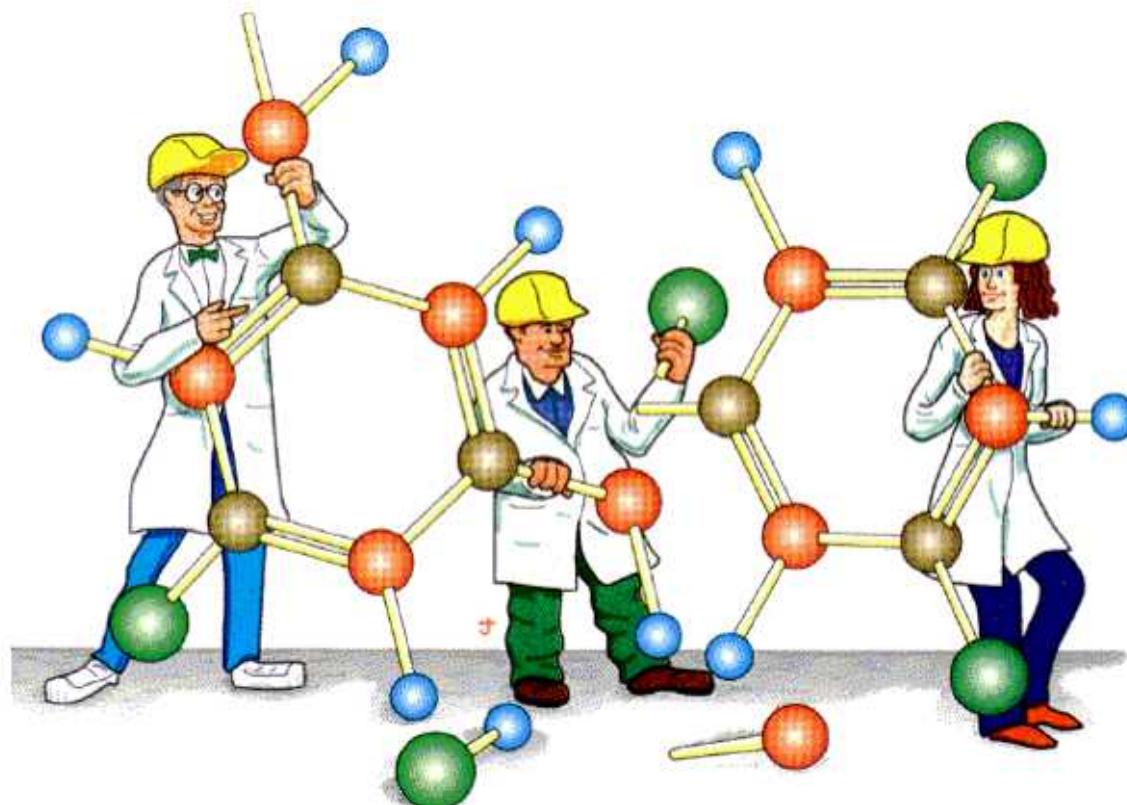
MOFs: Supramolecular “Tinker Toys”

A modular materials system for renewable energy?

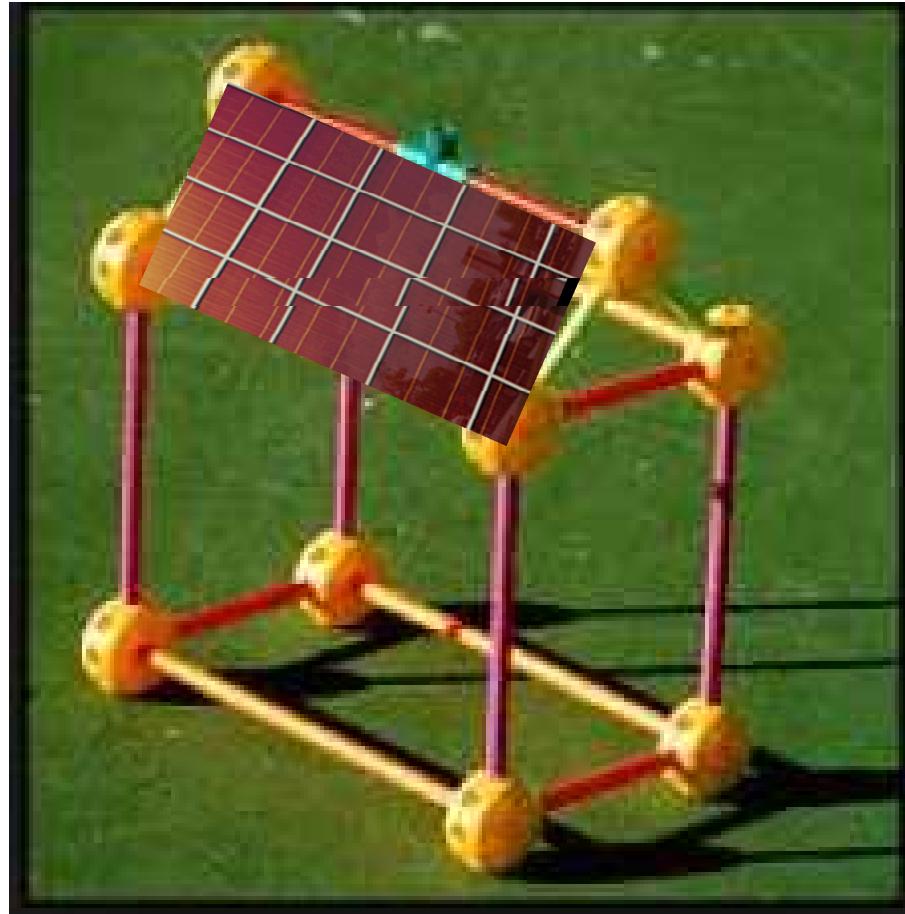


Realizing Supramolecular Materials...

Can we manipulate and assemble these supramolecular building blocks to improve DSSCs?



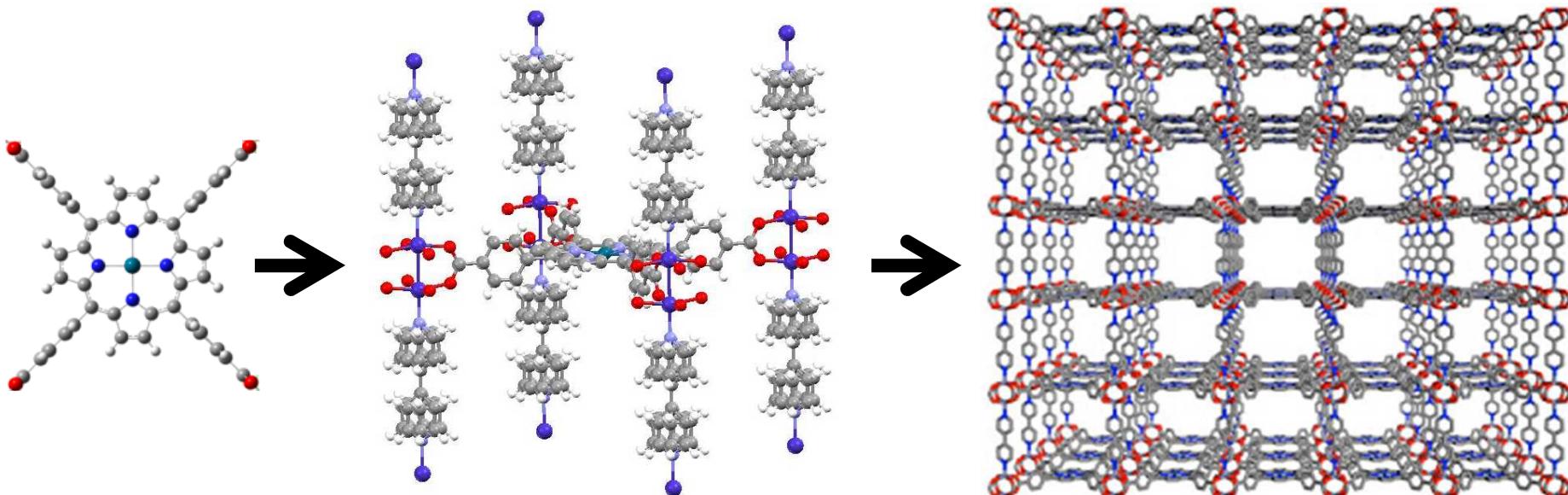
Merging “Modular” Chemistry and Photovoltaics...



Consider Pillared Porphyrin Frameworks (PPFs)

In PPF MOFs, transition metal cations coordinate the assembly of photoactive metalloporphyrins into sheets, stacked atop molecular pillars.

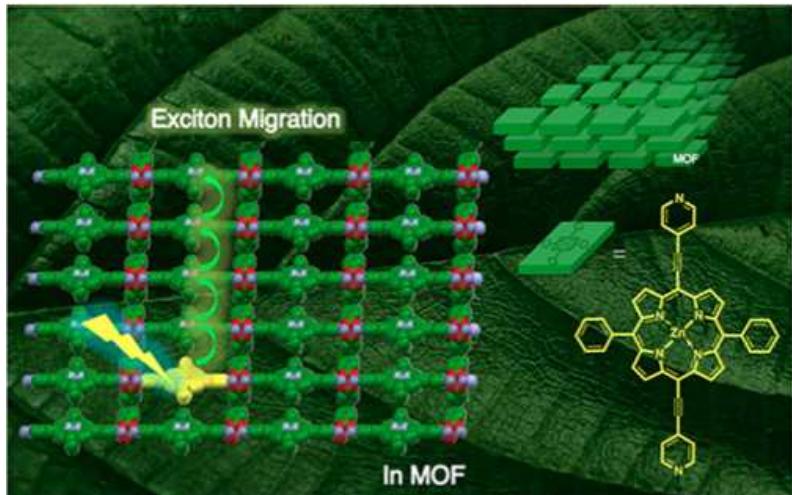
PPF-5



Building Photoactive MOFs

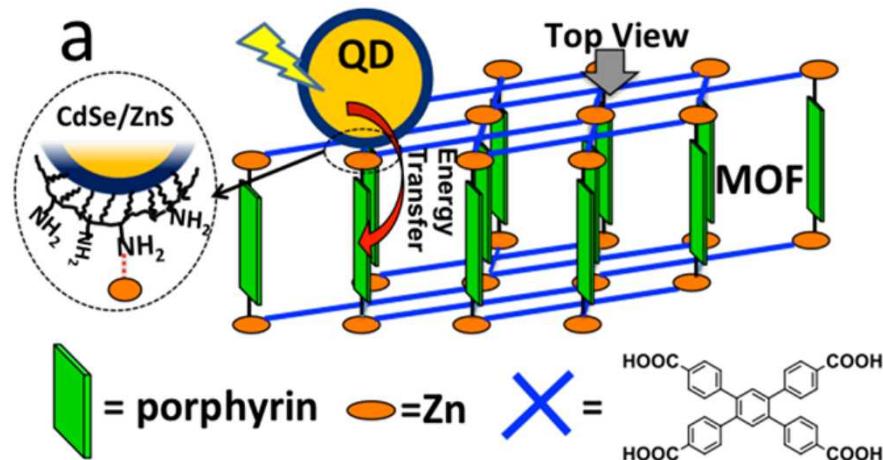
Energy transfer is viable in porphyrin-based MOFs

Fast exciton transport between porphyrins



H.-J. Son, et al. JACS (2013) **135**. 862-869.

Energy transfer between MOFs and semiconductors

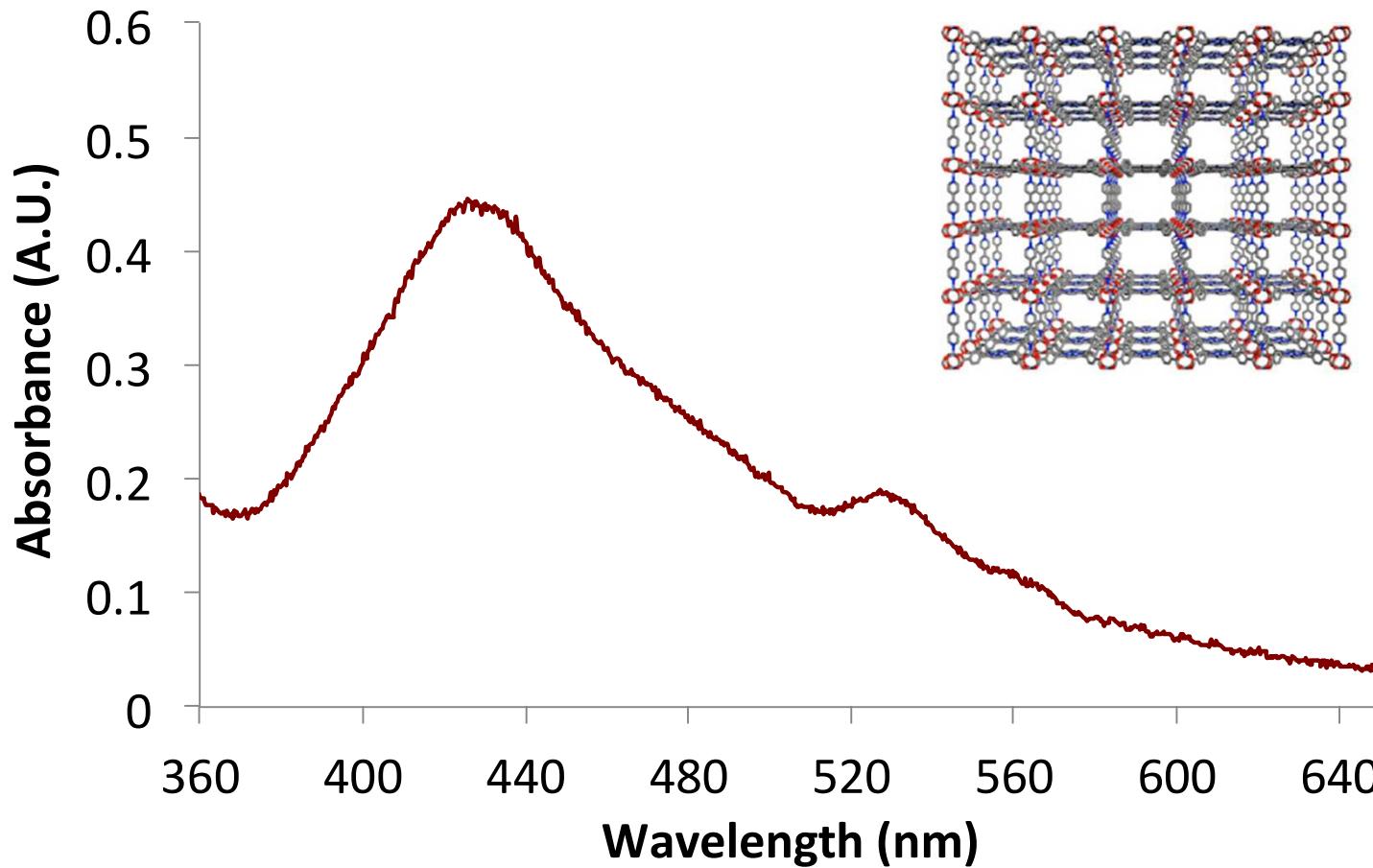


S.Jin, et al. JACS (2013) **135**. 955-958.

Precedents for using porphyrins in DSSCs...

1. Kay and Grätzel, J. Phys. Chem. (1993) **97**, 6292.
2. Walter, et al. J. Porphyrins and Phthalocyanines. (2010) **14**, 759.
3. M. J. Griffith and A. J. Mozer (2011), Available from: <http://www.intechopen.com/books/solar-cells-dye-sensitized-devices/porphyrin-based-dye-sensitized-solar-cells>

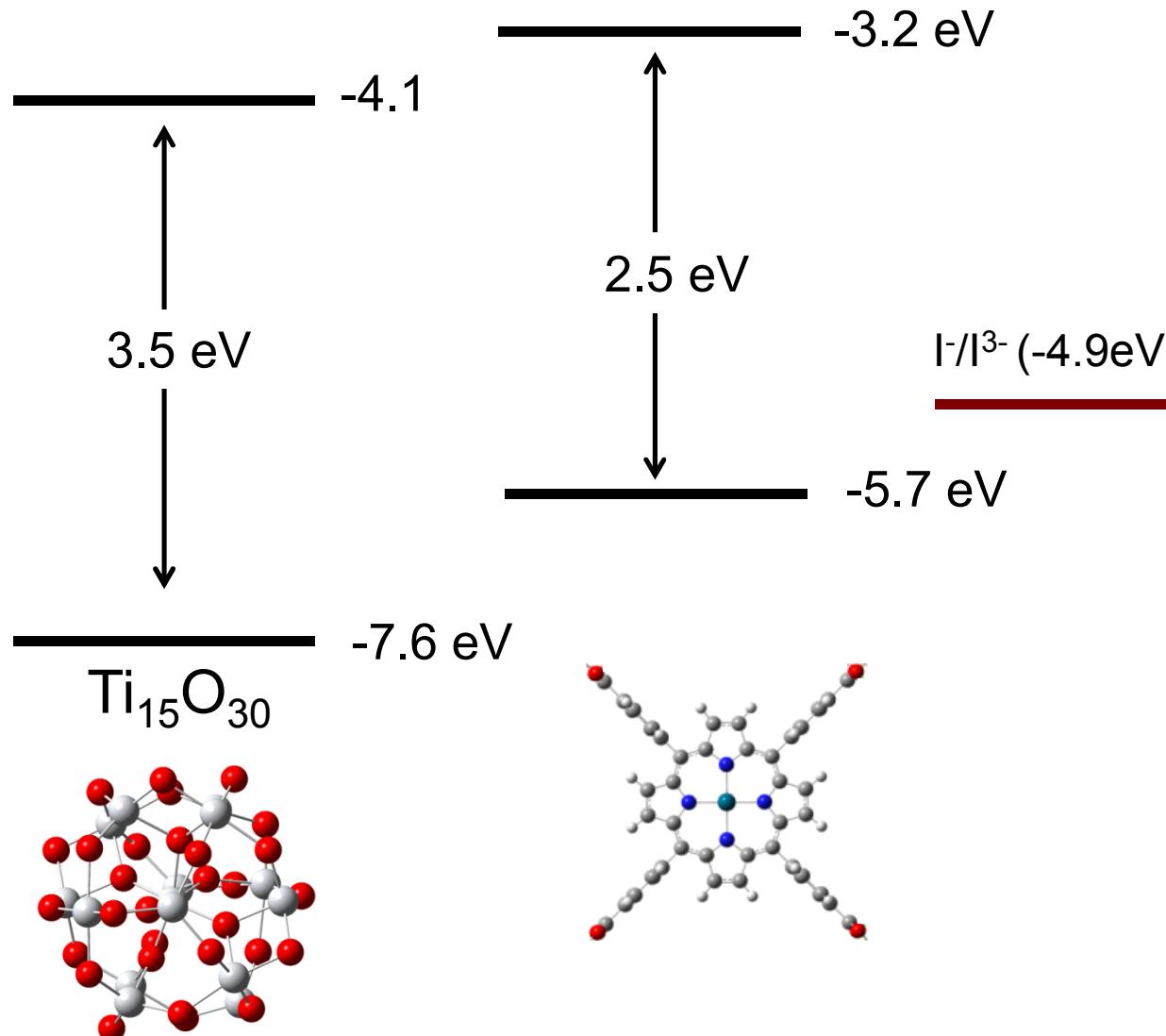
PPF-5 Optical Absorbance



PPF-5 absorbs meaningful visible light.

DFT Predictions of Band Alignment

Density Functional Theory (DFT) predicts reasonable band alignment between the PPF-5 porphyrin and a TiO_2 electron acceptor.



TiO_2 Molecular Clusters
DFT (B3LYP/LanL2DZ)

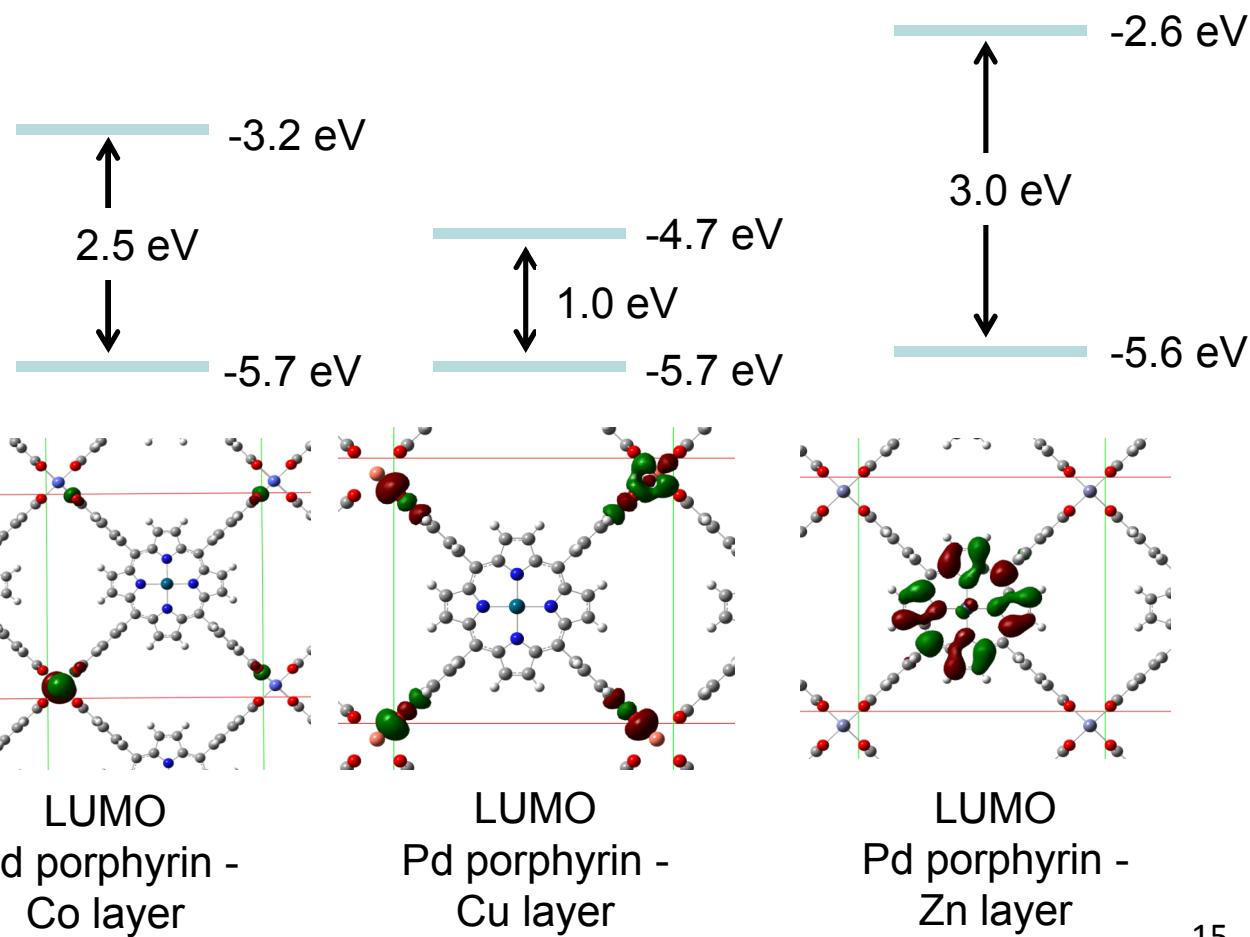
PPF-5 2D
DFT(B3LYP/ CEP-31G)

Band Structure Tailoring

Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.

Varying transition metal ions

2D periodic
optimization -
DFT(B3LYP/ CEP-
31G)

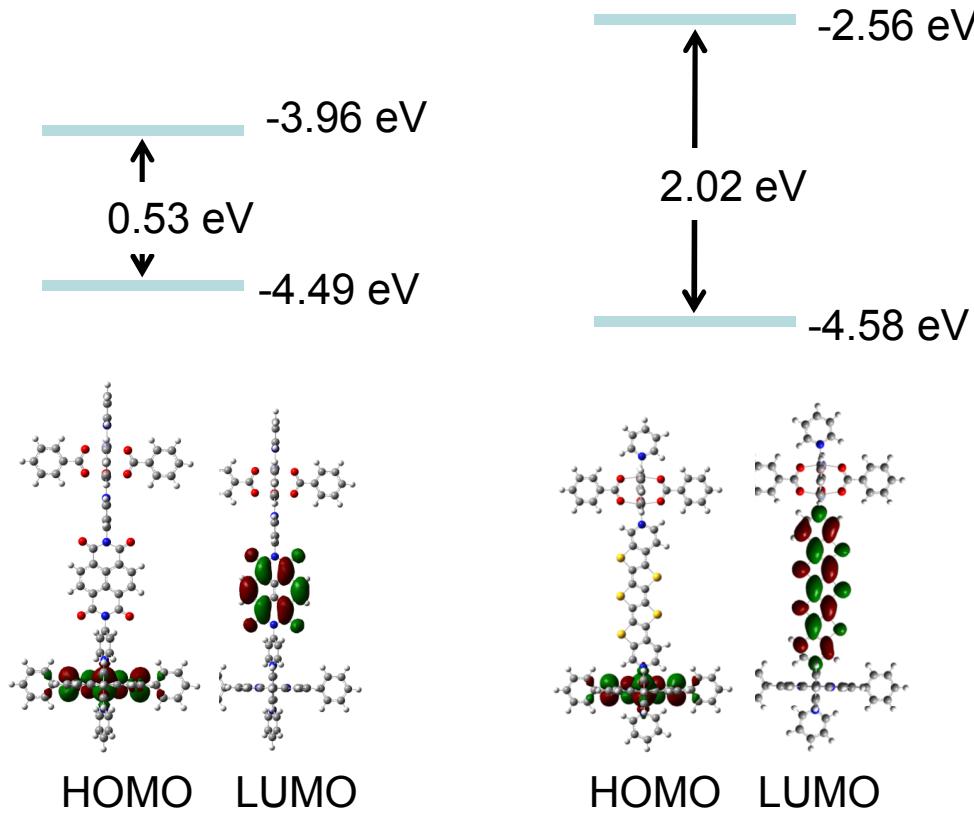


Band Structure Tailoring

Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.

Varying organic pillars

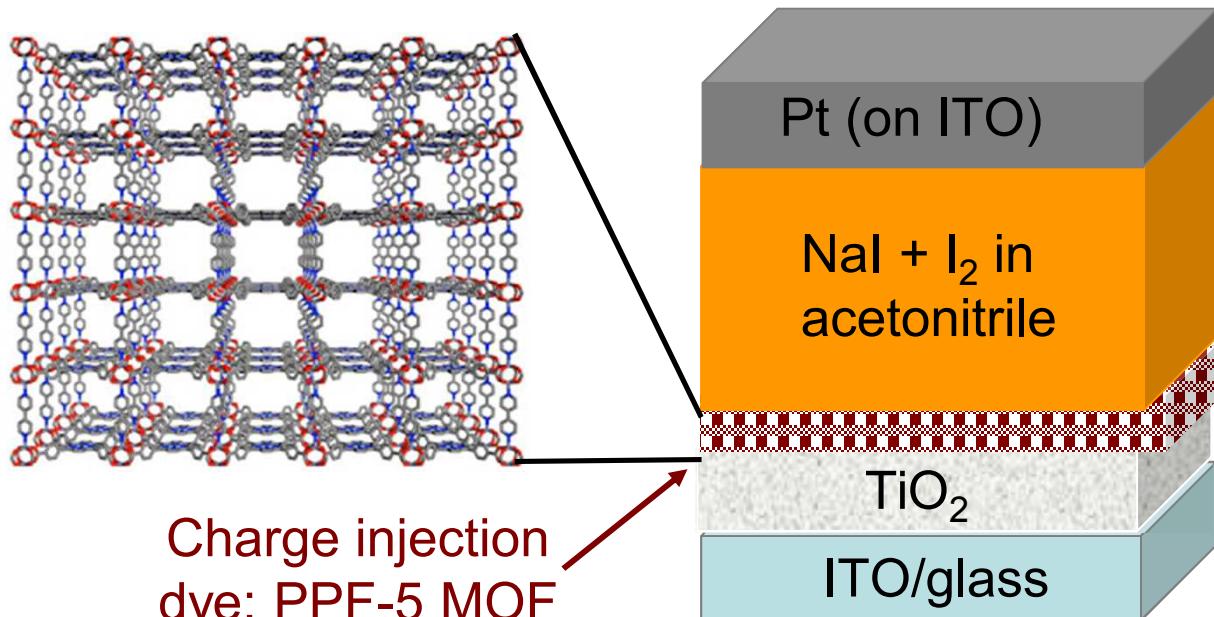
2D periodic optimization -
DFT(B3LYP/ CEP-31G)



Integration of PPF-5 in DSSCs

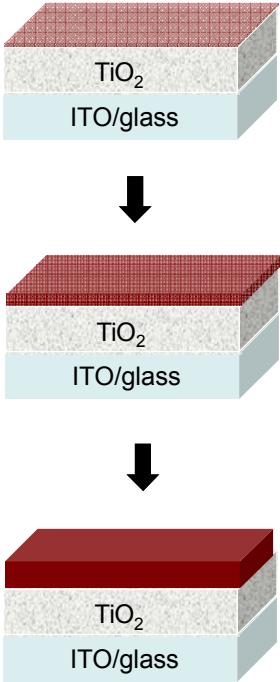
Potential advantages of PPF-5 active layer:

- Visible light absorption
- Reasonable, potentially tunable band alignment
- Ordered charge transport pathways
- Non-aggregated dye assembly
- Porosity for electrolyte access

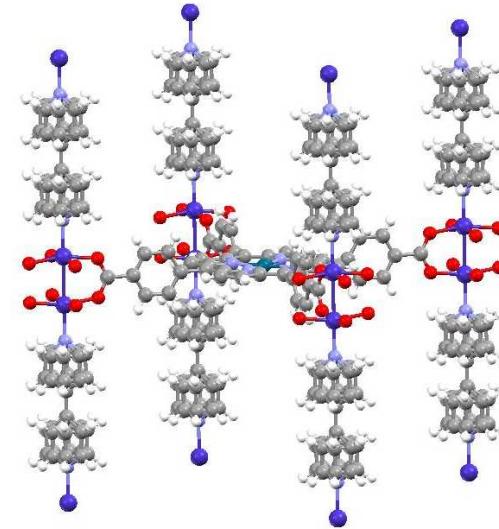
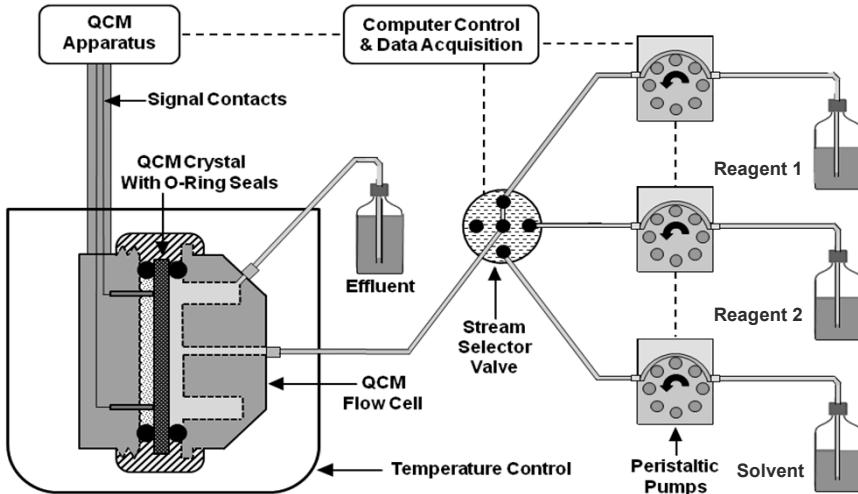


Can we grow PPF-5 onto the TiO₂ anode for integration into a “standard” DSSC configuration?

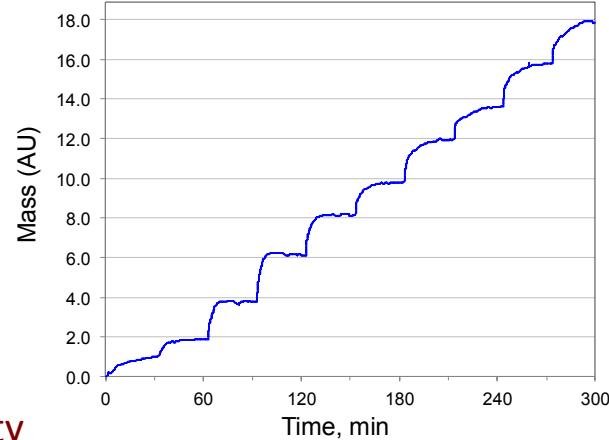
Layer-by-Layer MOF Growth



1. Deposit/anneal TiO_2 nanoparticle slurry (DeGussa P25) on ITO/glass
2. Pretreat TiO_2 with Co^{2+}
3. Sequentially introduce
 - Co^{2+} -ligated porphyrin linker
 - Solvent wash
 - Bipyridine pillar
 - Solvent wash



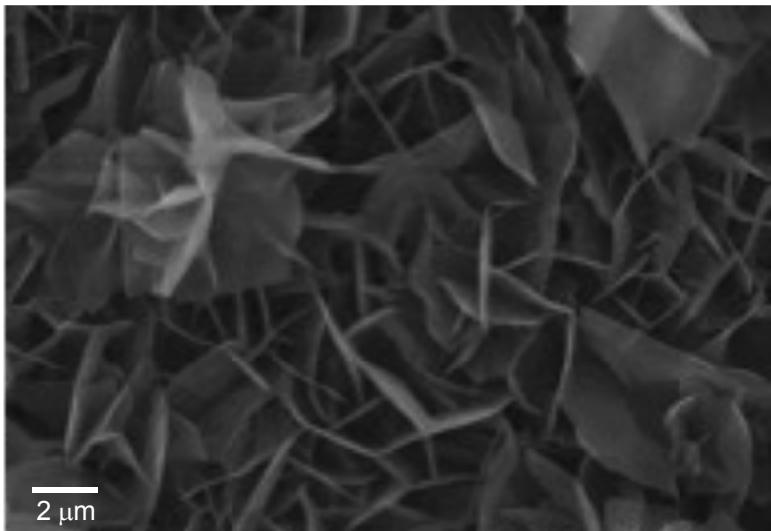
Step by step PPF-5 growth



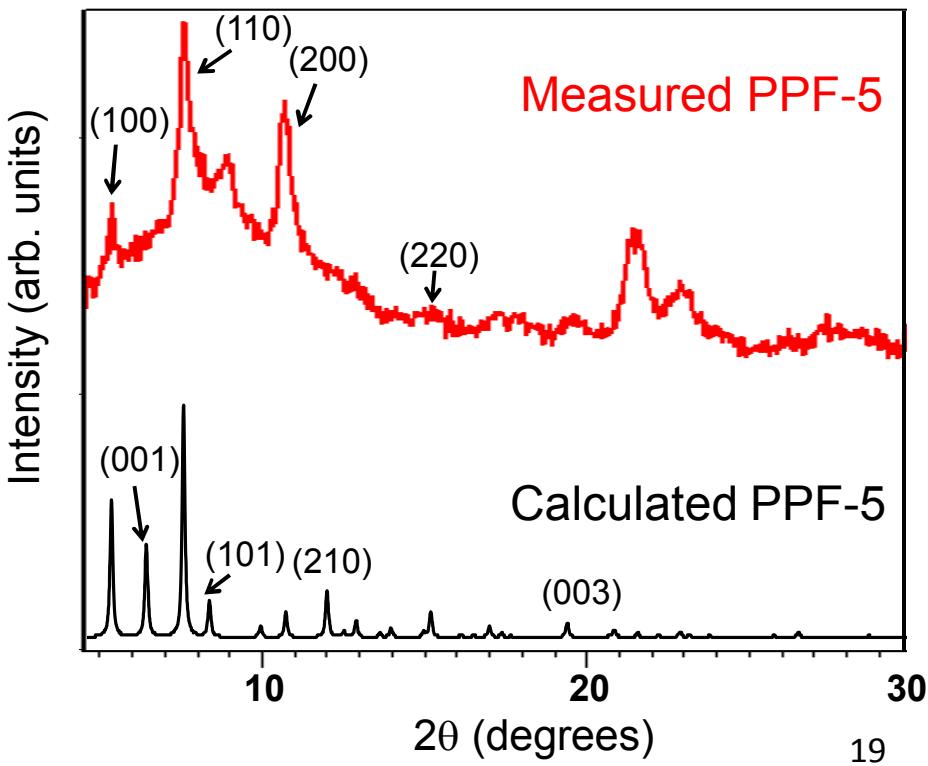
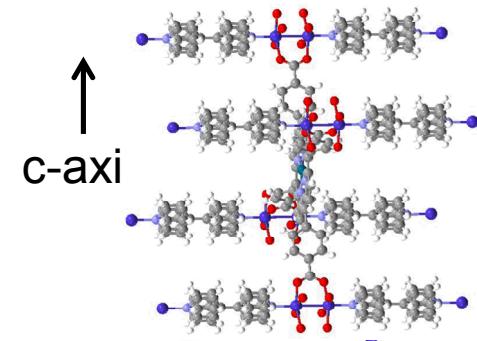
Schematic representation of automated MOF film growth/QCM capability

Characterization of PPF-5 Films on TiO_2 /ITO

LBL-growth produces a porous array of PPF-5 crystals exhibiting crystallographic orientation on the TiO_2 surface.

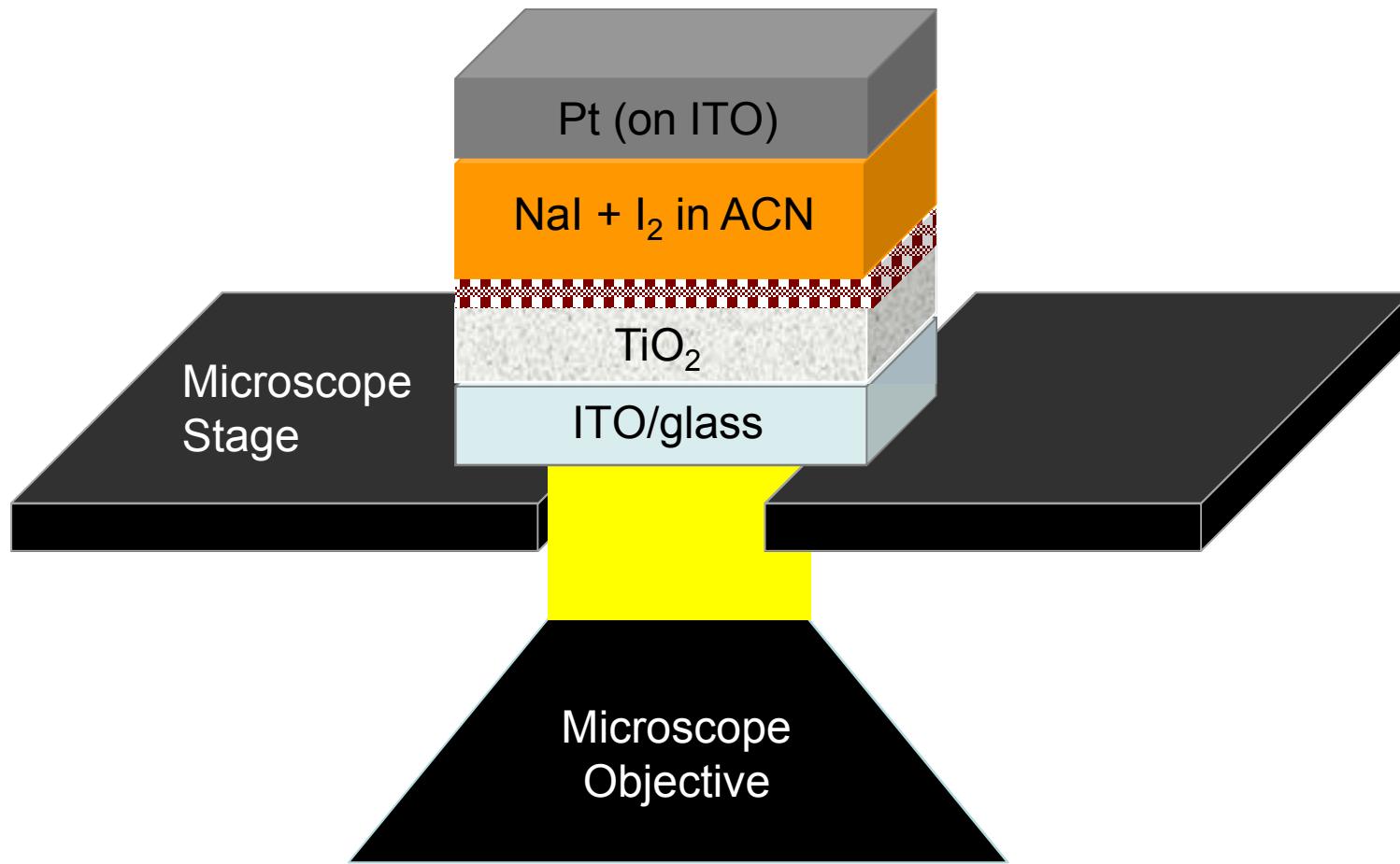


Scanning electron micrograph shows PPF-5 "platelets" extending off the TiO_2 surface.



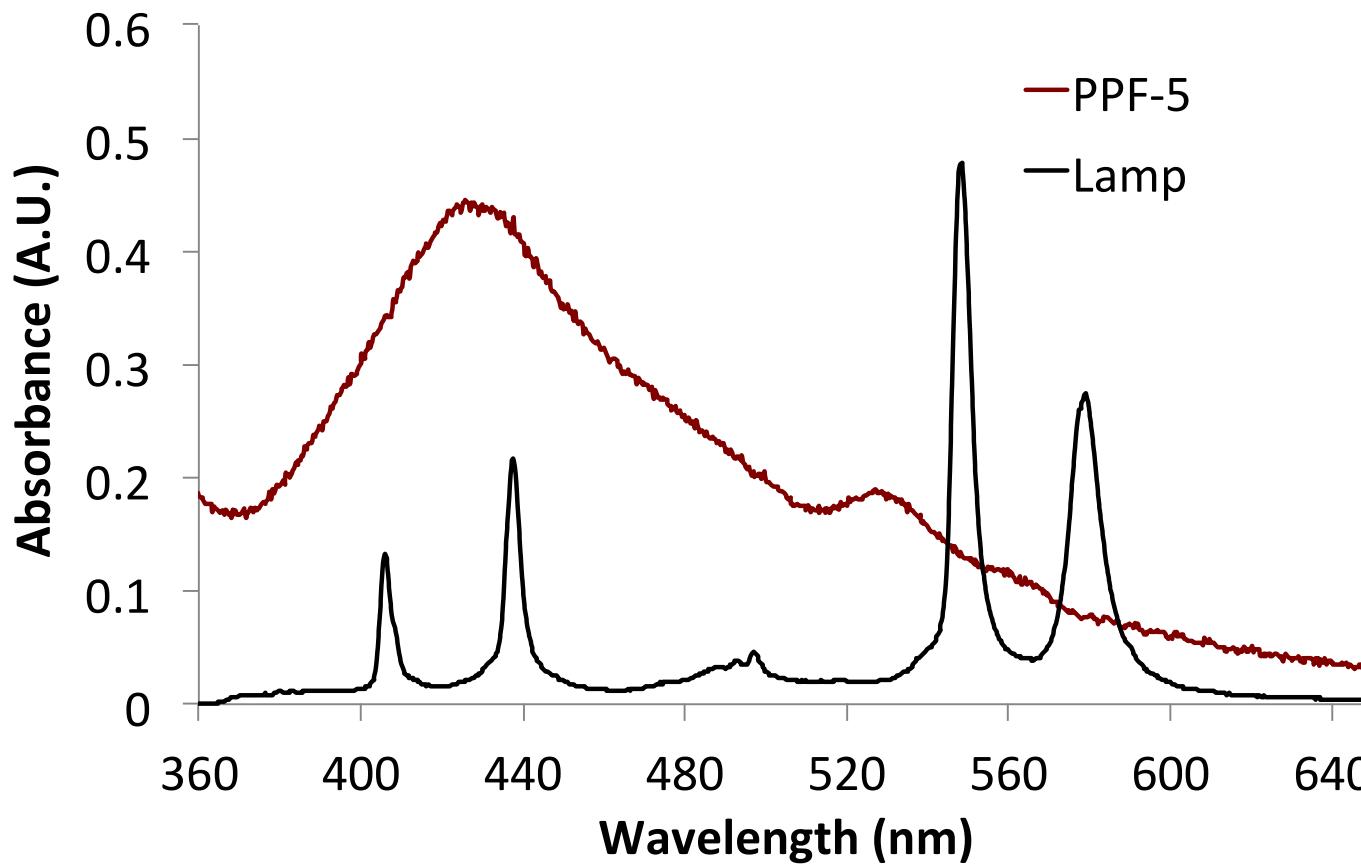
PV Measurement Setup

DSSC devices assembled and tested on microscope stage with UV-filtered Hg-arc lamp.



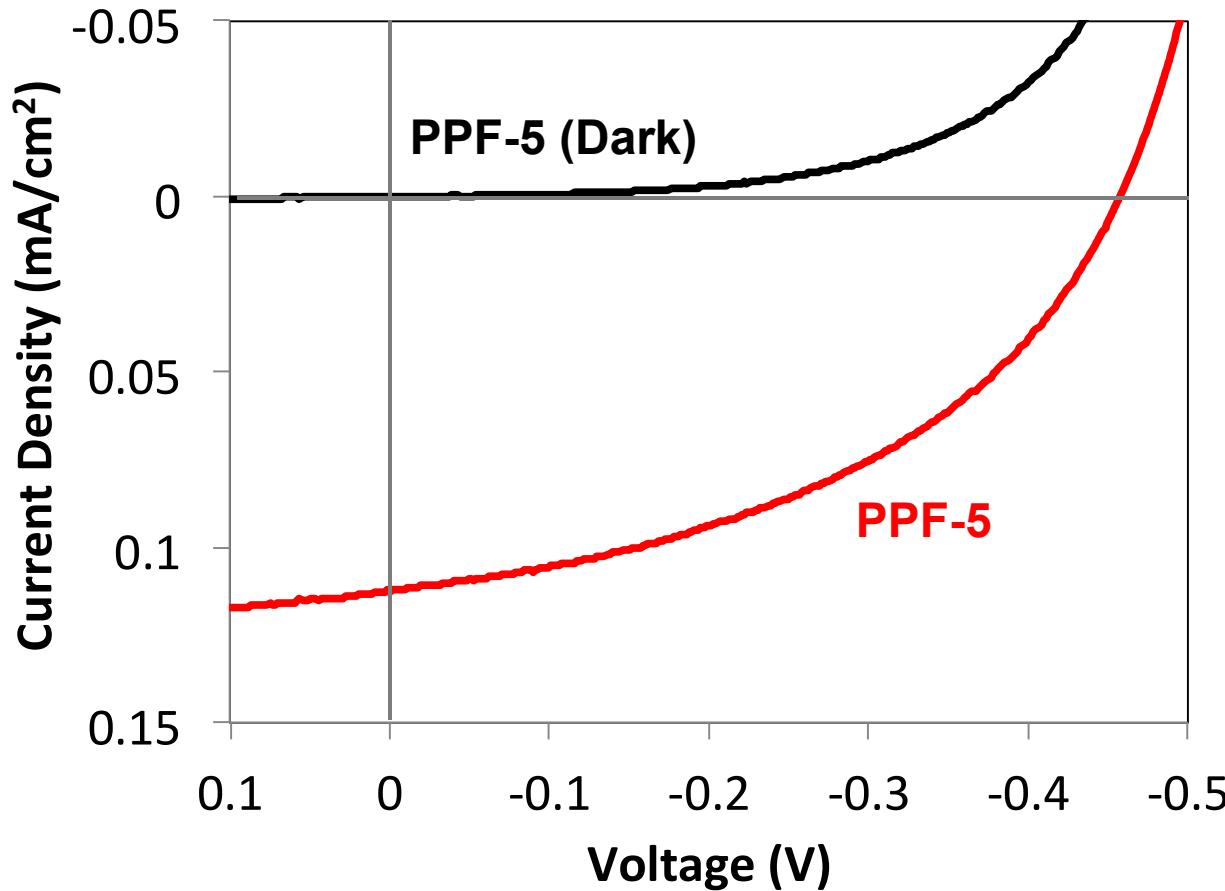
PPF-5 Optical Absorbance

PPF-5 absorbance aligns reasonably UV-filtered Hg-arc lamp.



Performance of PPF-5 in a DSSC

PPF-5 DSSC produces measurable photocurrent!



Averaged metrics:

V_{oc} (V) = 0.452 ± 0.029

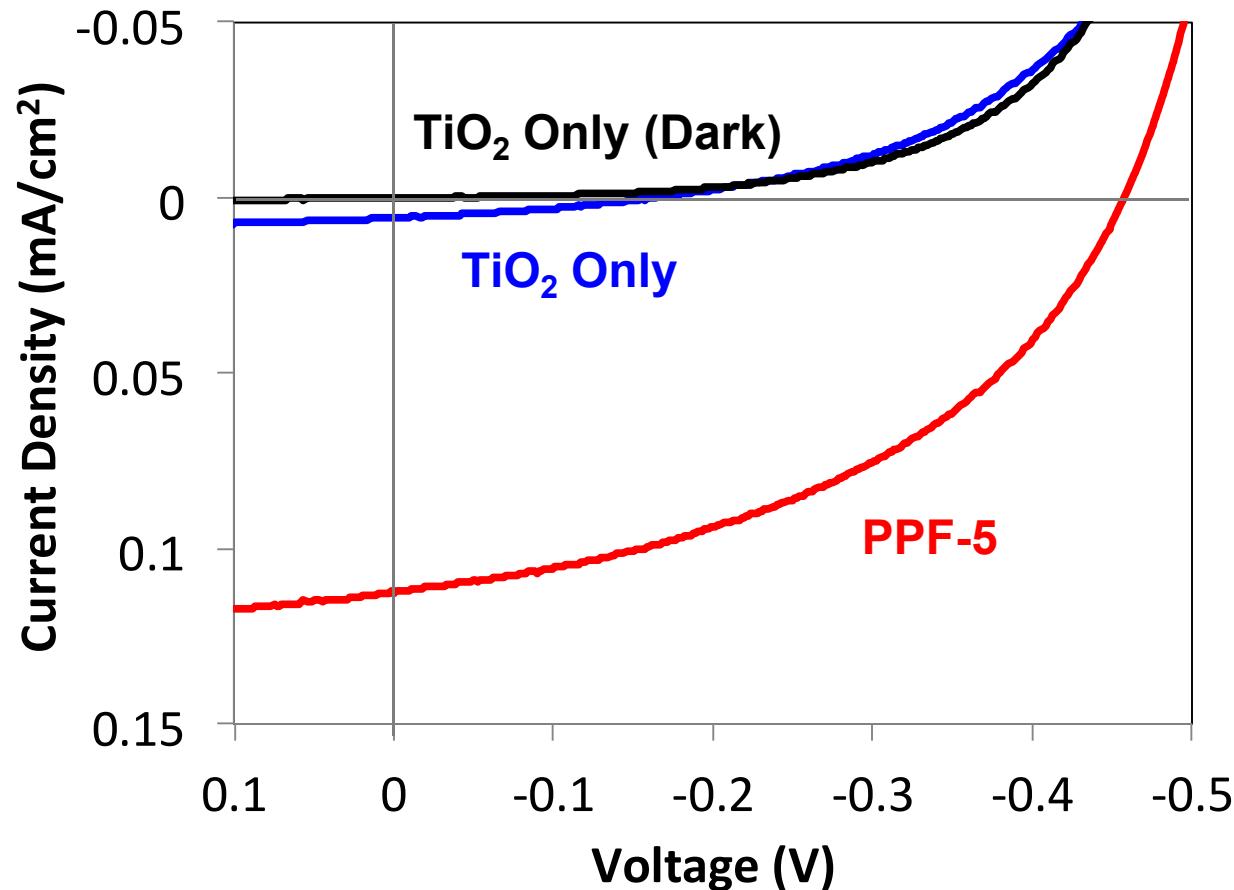
J_{sc} (mA/cm²) = 0.097 ± 0.014

FF = 0.47 ± 0.031

η (%) = 0.026 ± 0.0038

Negative Control: No PPF-5

Control experiments containing no PPF-5 produce negligible photocurrent.



In the absence of dye:

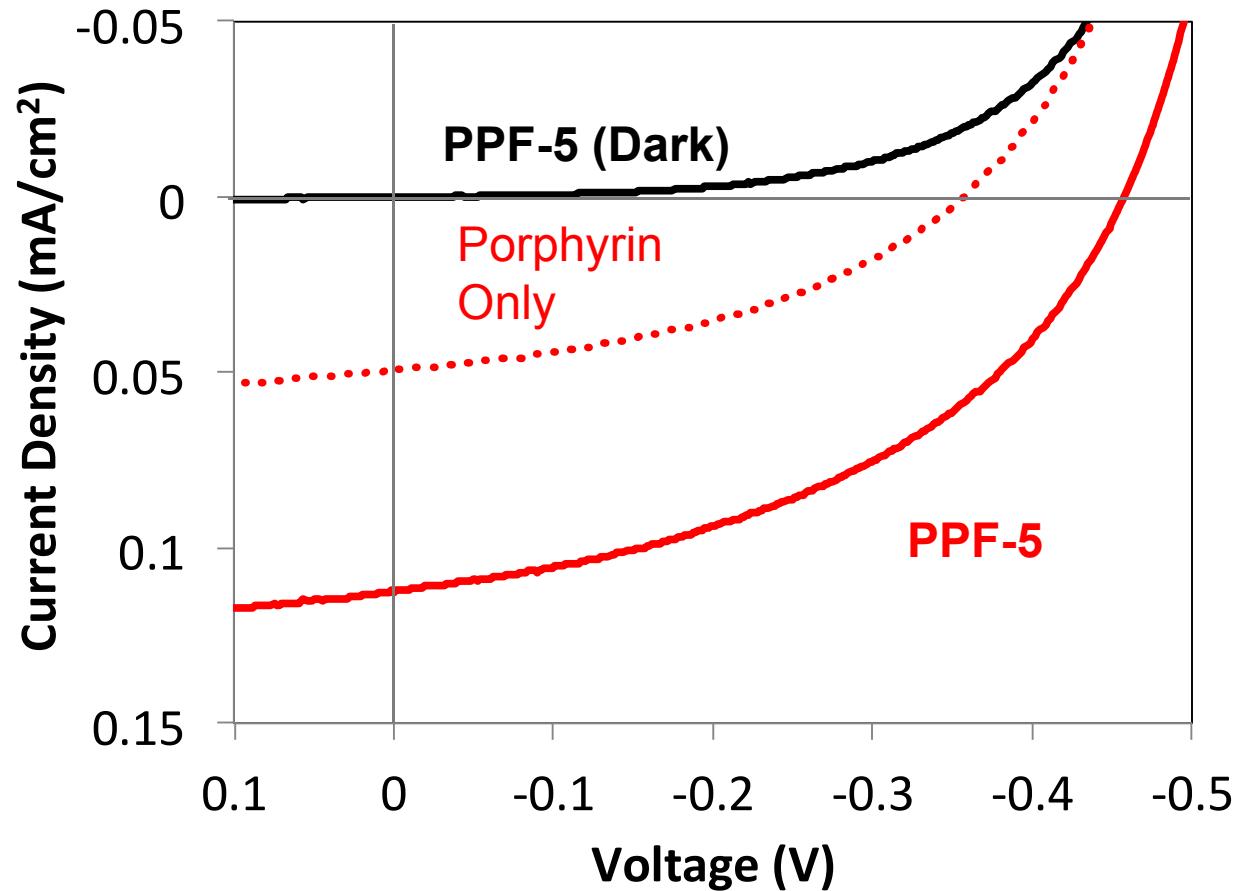
$$V_{oc} (V) = 0.16$$

$$J_{sc} (\text{mA/cm}^2) = 0.0058$$

$$FF = 0.31$$

$$\eta (\%) = 0.000378$$

Performance of PPF-5 in a DSSC



Averaged metrics:

$\text{Voc (V)} = 0.401 \pm 0.057$

$\text{Jsc } (\text{mA}/\text{cm}^2) = 0.061 \pm 0.017$

$\text{FF} = 0.45 \pm 0.039$

$\eta \text{ (\%)} = 0.015 \pm 0.0073$

$\text{Voc (V)} = 0.452 \pm 0.029$

$\text{Jsc } (\text{mA}/\text{cm}^2) = 0.097 \pm 0.014$

$\text{FF} = 0.47 \pm 0.031$

$\eta \text{ (\%)} = 0.026 \pm 0.0038$

Wrapping up

- MOFs are highly porous, multifunctional composites crystals, assembled from “modular” molecular building blocks.
- Grown by Layer-by-Layer processes, *PPF-5 integrated into DSSC devices can serve as a functional active absorber!*
- Tuning of MOF composition and structure are expected to improve DSSC device performance.
- This preliminary demonstration shows that this electrochemical configuration is a feasible platform to explore the diversity of MOF chemistry in solar applications.

This is just the beginning!

Acknowledgements

Sandia (Albuquerque)

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Dr. Leo Small

Bonnie McKenzie

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- **U.S. DOE Energy Efficiency and Renewable Energy (Sunshot)**
- **Sandia Laboratory Directed Research and Development Program**

Sandia (California)

Dr. Michael Foster

Dr. Vitalie Stavila

Dr. Kirsty Leong

Dr. Mark D. Allendorf



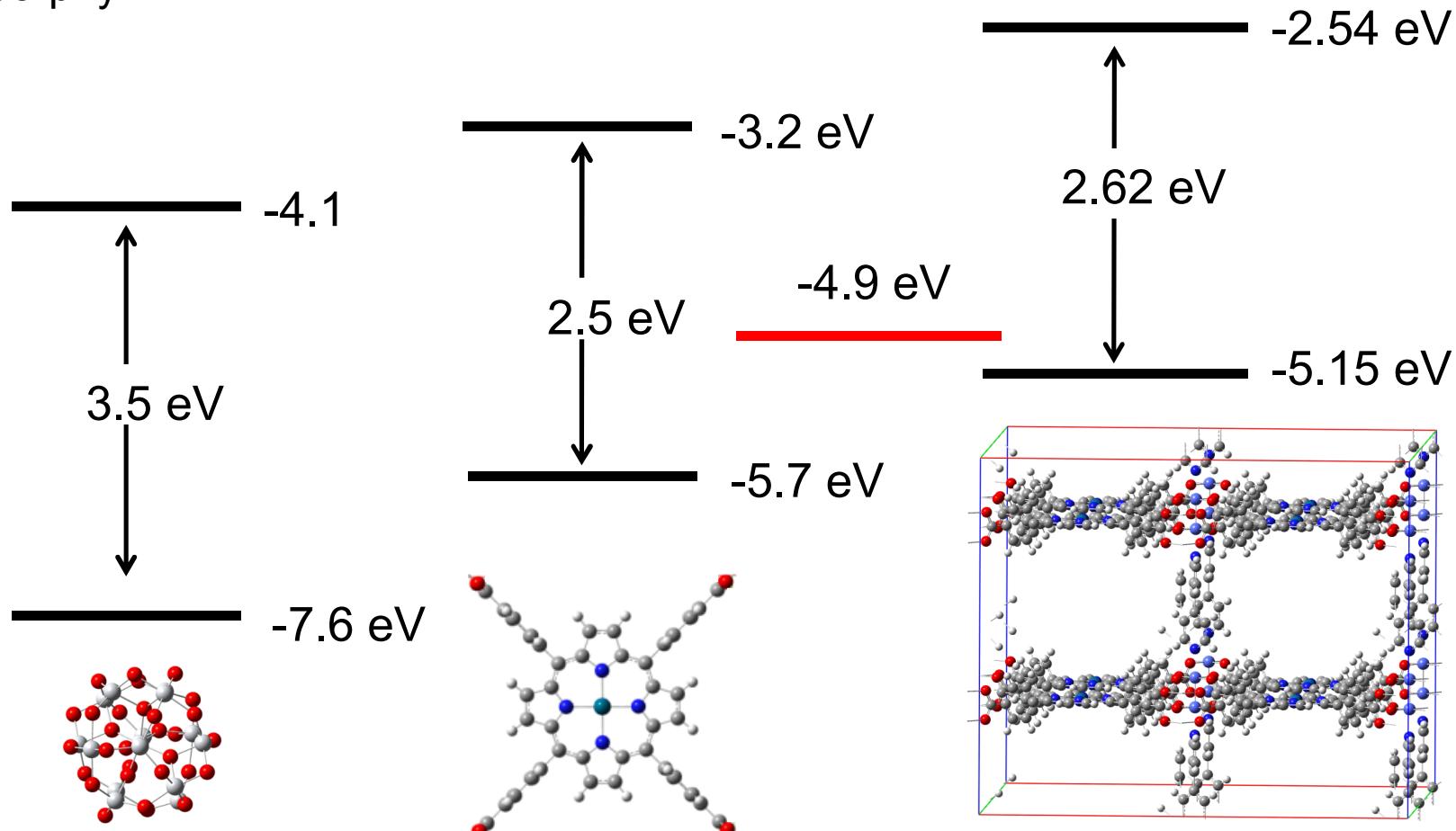
Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000

Backup Slides

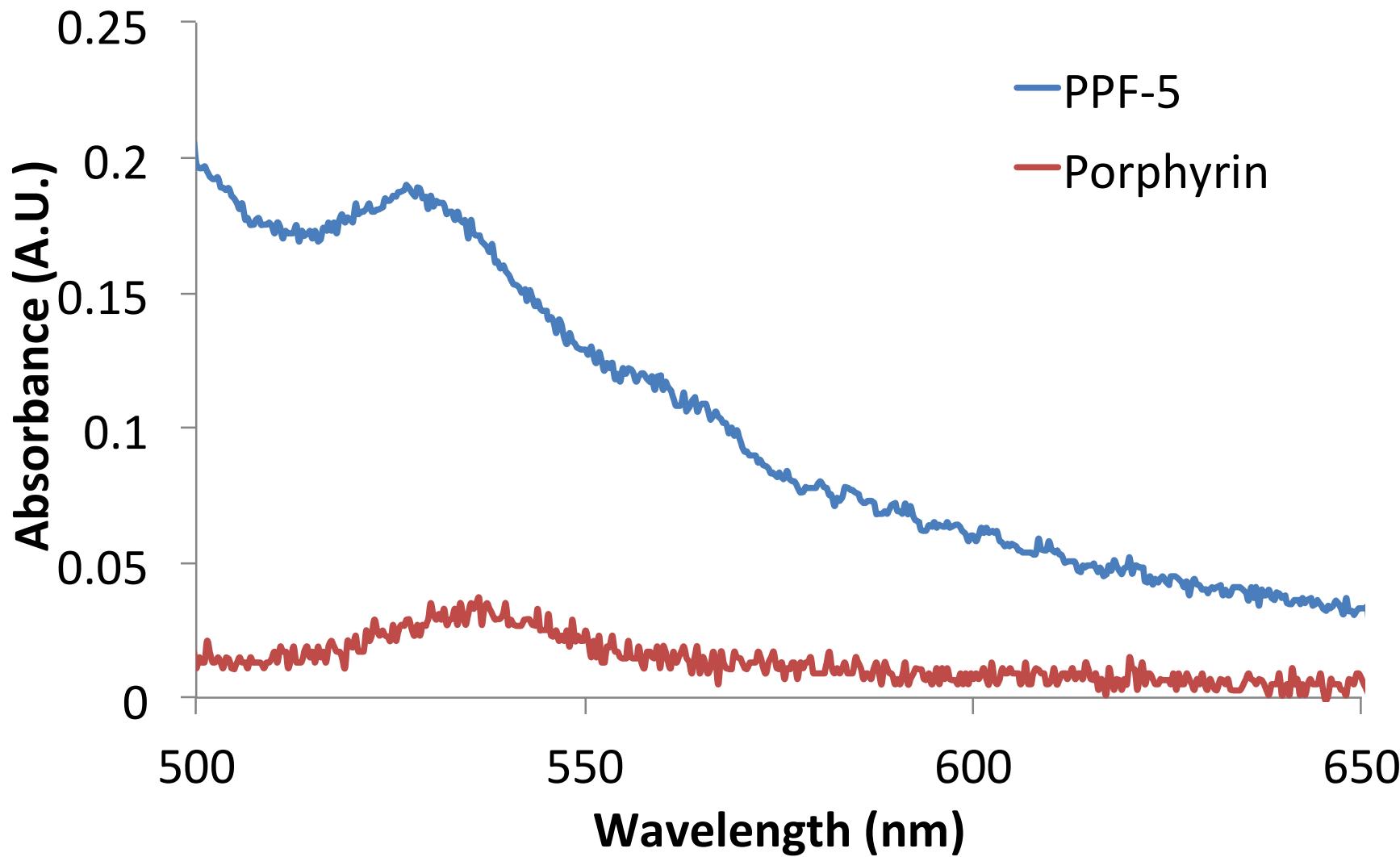


DFT Predictions of Band Alignment

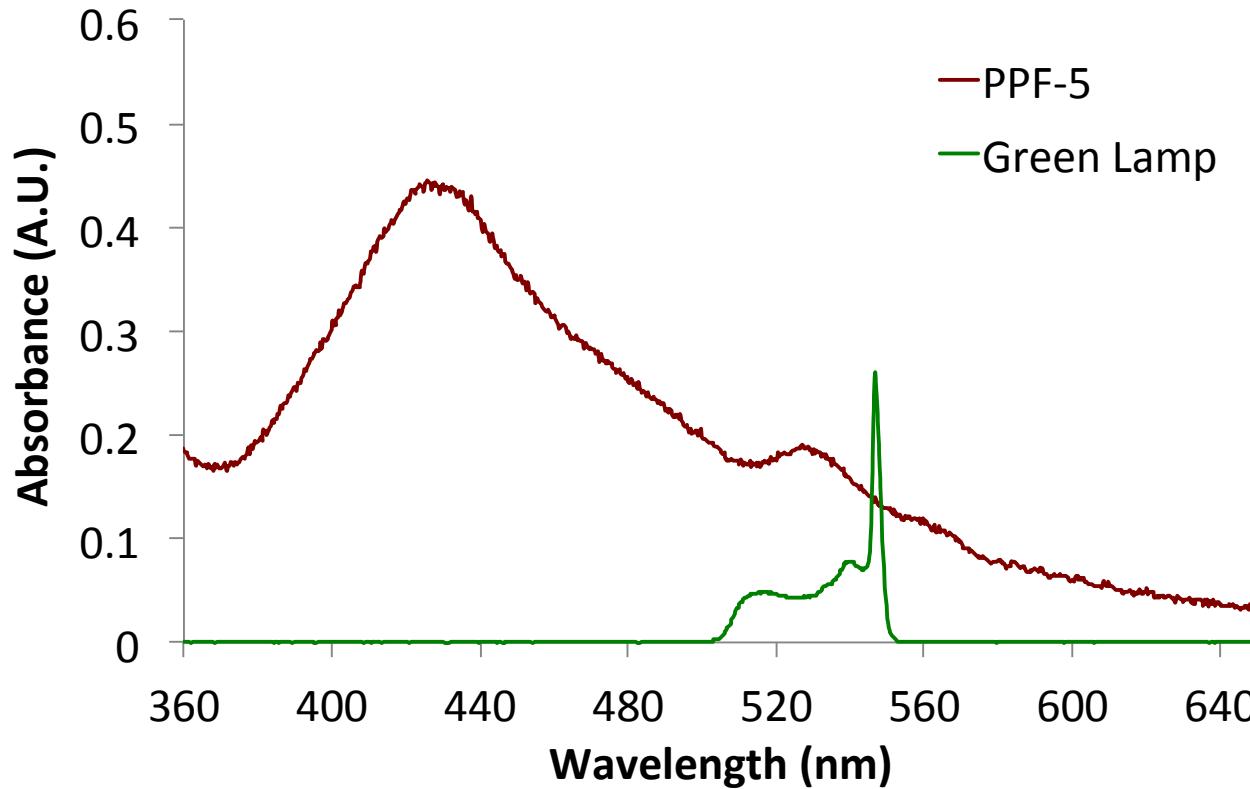
Density Functional Theory (DFT) predicts a shift in band alignment of PPF-5 relative to the Pd-porphyrin.



Absorbance Changes in PPF-5



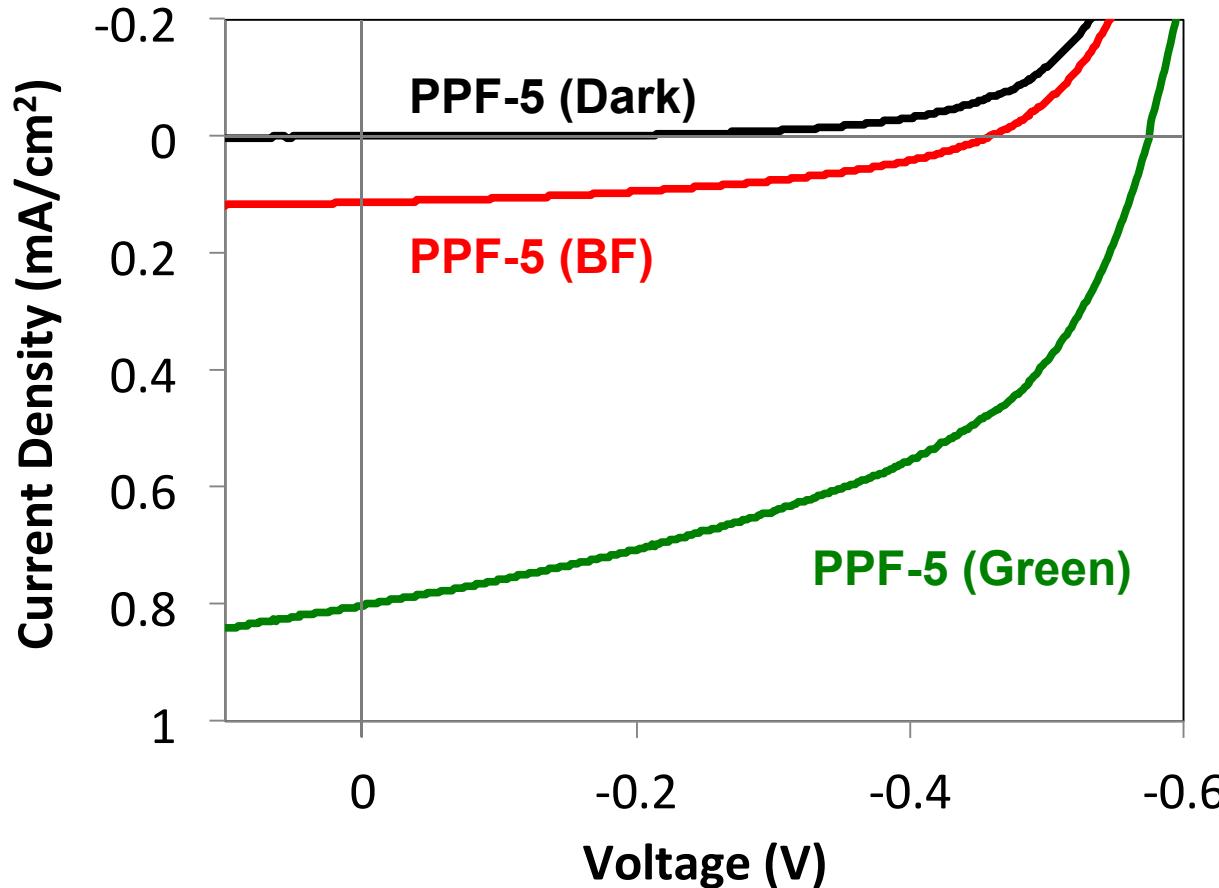
Optical “Cheating”



PPF-5 absorbance aligns reasonably with Hg-arc lamp.

Enhancing PPF-5 Performance with Green Light

Selective excitation with Green light produces enhanced photocurrent, confirming contribution from PPF-5 absorber.



Averaged metrics:

V_{oc} (V) = 0.452 ± 0.029

J_{sc} (mA/cm²) = 0.097 ± 0.014

FF = 0.47 ± 0.031

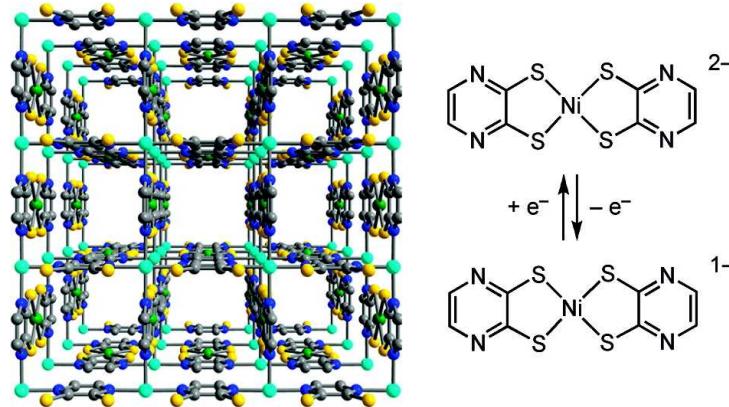
V_{oc} (V) = 0.563 ± 0.018

J_{sc} (mA/cm²) = 0.73 ± 0.083

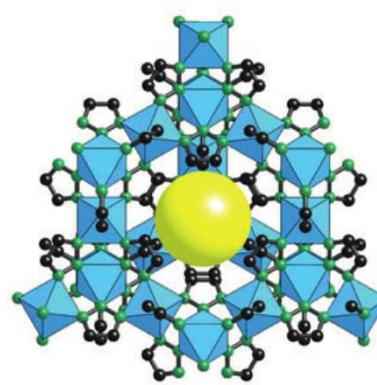
FF = 0.537 ± 0.07

Electrically conducting porous MOFs are rare

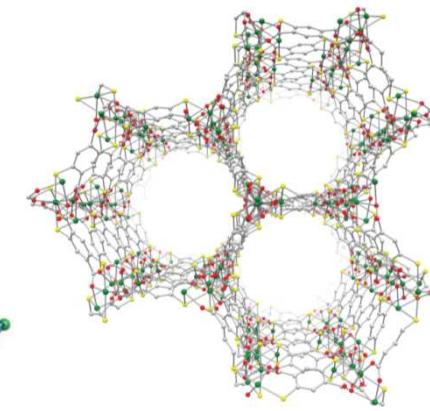
- **p-type Cu-Ni Dithiolene MOF**
 - First semiconducting, porous MOF
 - Conductivity increases with oxidative doping
 - Original Cu-Cu version is not porous
(*Inorg. Chem.* 2009, 48, 9048)
- **Other examples**
 - MET-3 (Fe-triazolate MOF)
 - Mn(thiophenol) MOF: $(-\text{Mn}-\text{S}-)^\infty$ Chains
- **Strategies for conducting MOFs:**
 - Charge delocalization
 - 2nd- and 3rd row transition metals
 - Redox-active ligands (e.g., TCNQ)
 - Soft ligands (e.g. S-containing molecules)



Y. Kobayashi et al. *Chem. Mater.* 2010, 22, 4120



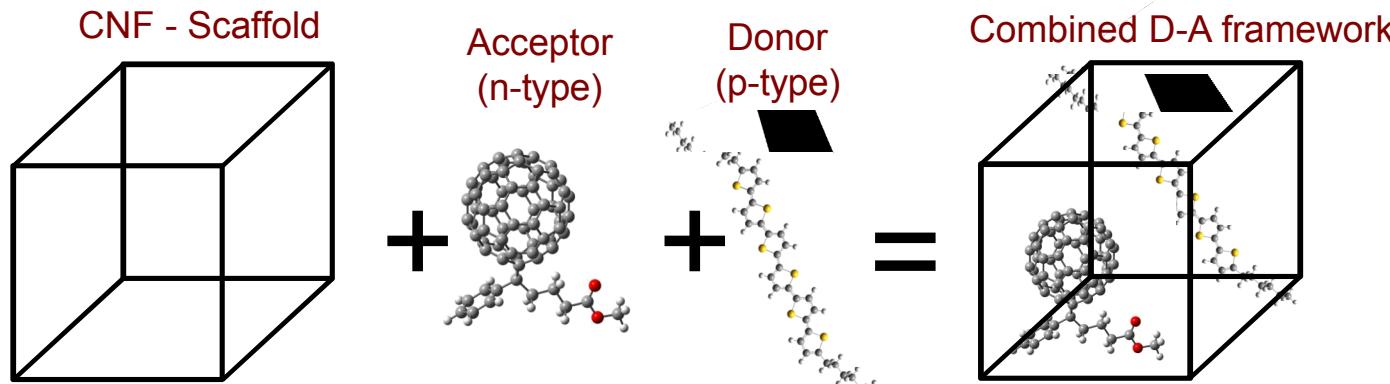
MET-3 (Fe)
Gándara et al.
Chem. Eur. J. 2012,
18, 10595



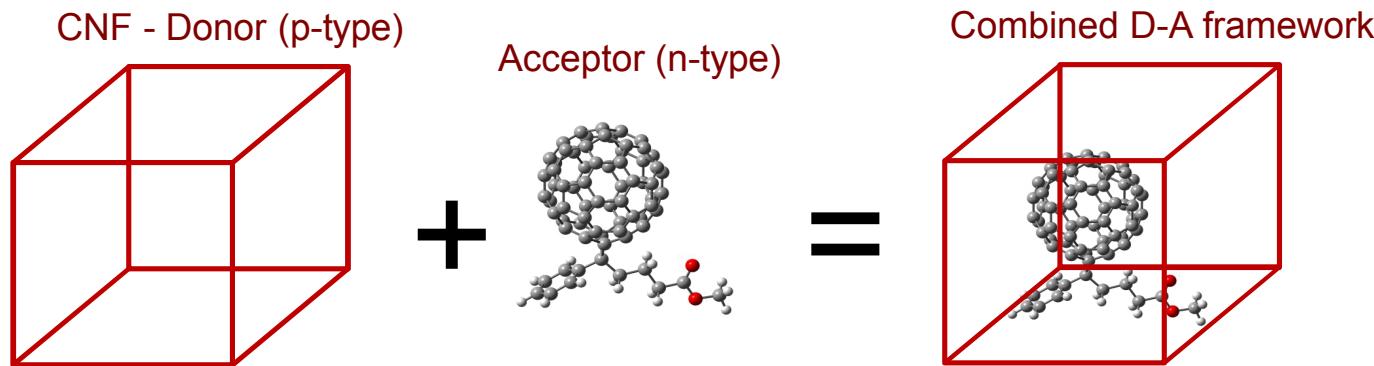
Mn(thiophenol) MOF
L. Sun et al.
J. Am. Chem. Soc.
2013, 135, 8185

Composite MOF Scaffolds for OPV

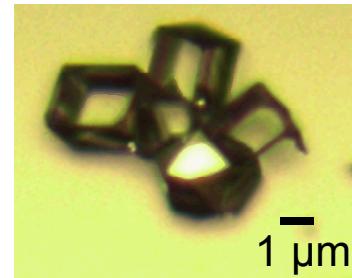
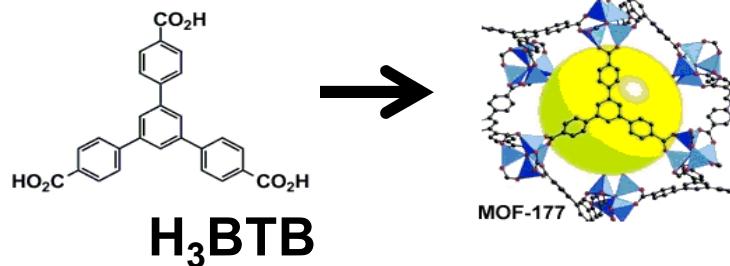
Passive Scaffolds - the CNF simply functions to order the donor/acceptor materials and plays no active role in the PV energy conversion process.



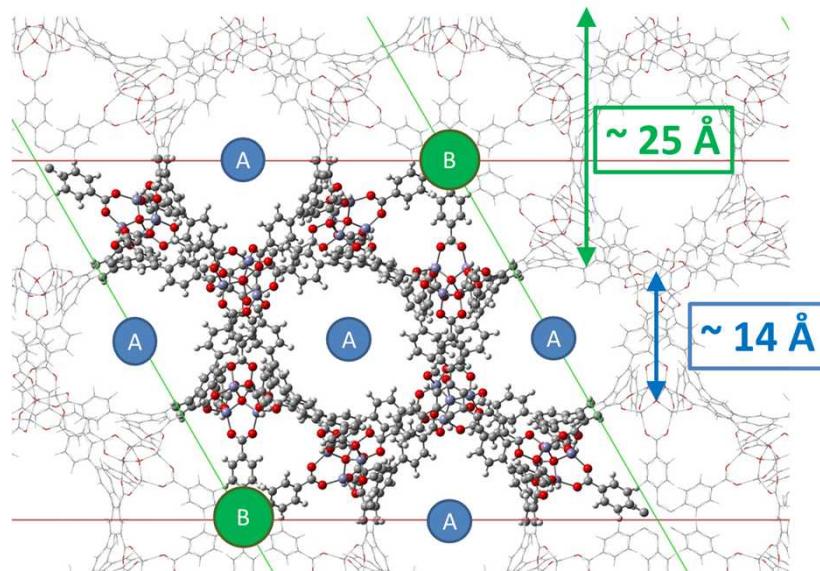
Active Network – the CNF is designed to play an active role in the PV energy conversion process by functioning as the donor or acceptor material.



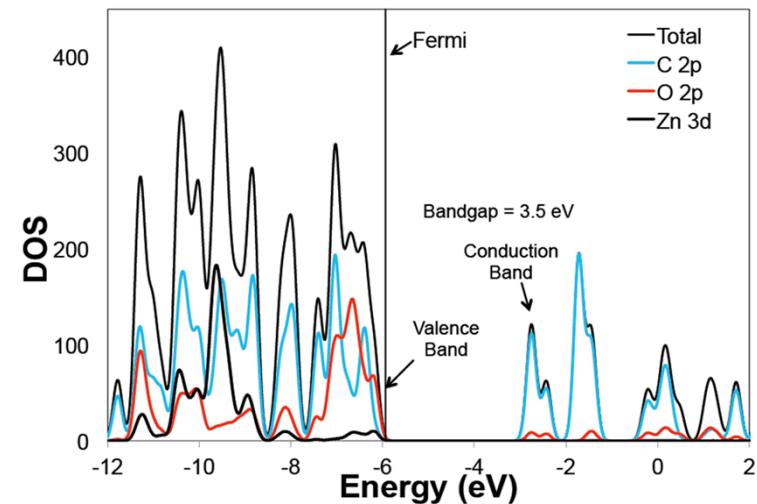
MOF177: A Passive Scaffold for D-A Assembly



Optical image
of MOF-177
crystals.



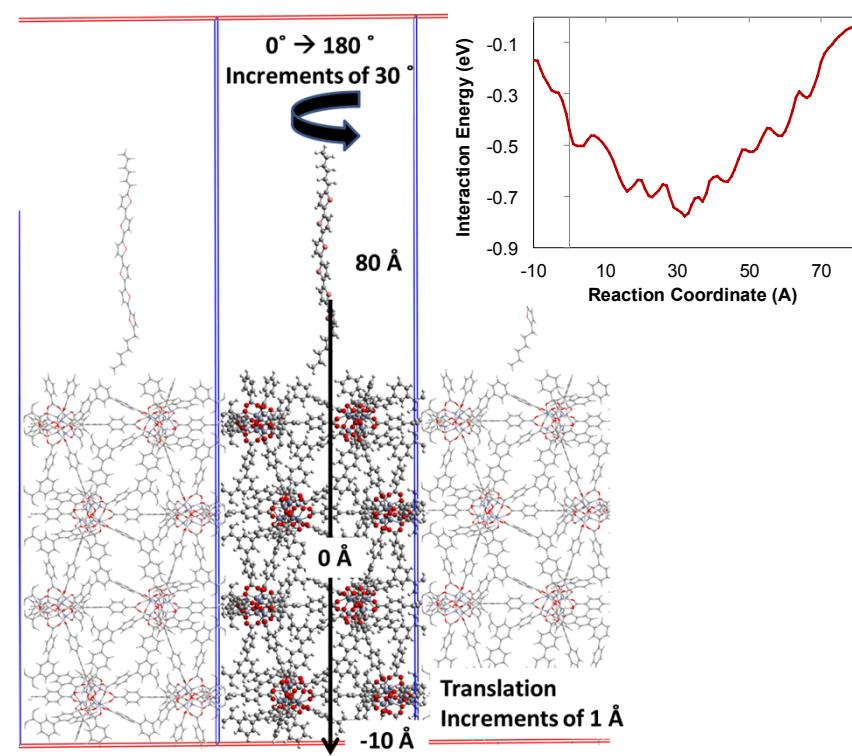
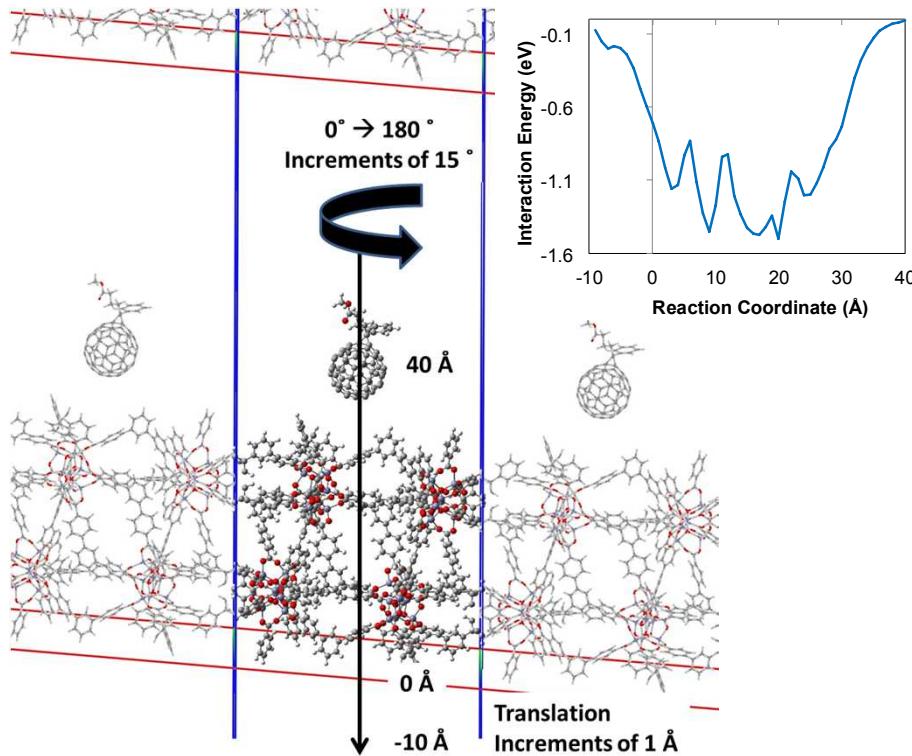
MOF-177 – sites “A” and “B” denote unique cavities; DFTB optimized structure.



Partial Density of States (PDOS) -
Density Functional Tight-Binding (DFTB)
calculations of MOF-177.

Predicting Guest Infiltration

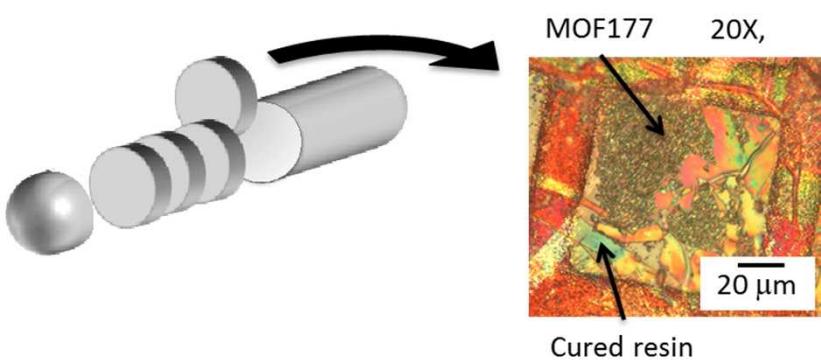
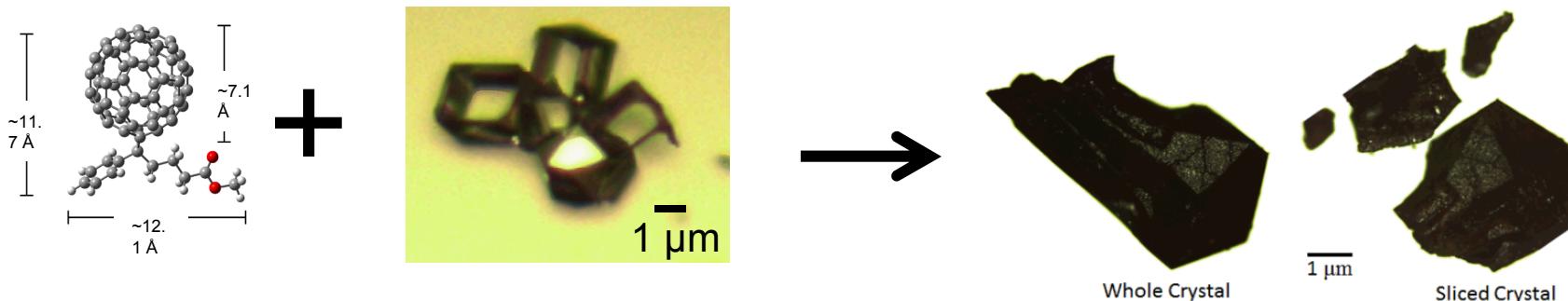
Density Functional simulations show that infiltration of both PCBM (A) and Sexithiophene (D) in MOF177 are enthalpically favored.



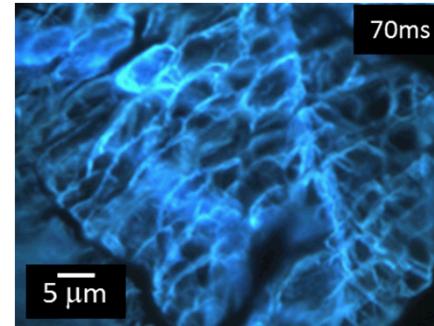
>600 structures generated for each case; 10 step geometry optimization performed to remove close contacts. Interaction energies determined using Density Functional based Tight Binding (DFTB) method.

PCBM Integration into MOF177

Incubation of MOF177 crystals in concentrated PCBM solutions leads to PCBM infiltration

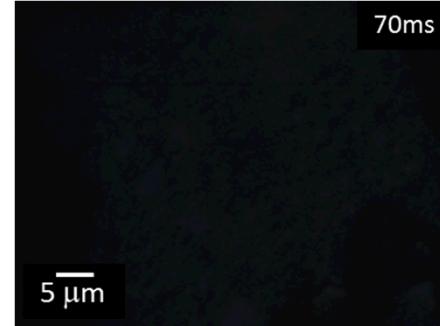


MOF177



Ex: 330-385nm; Em filter: 420nm

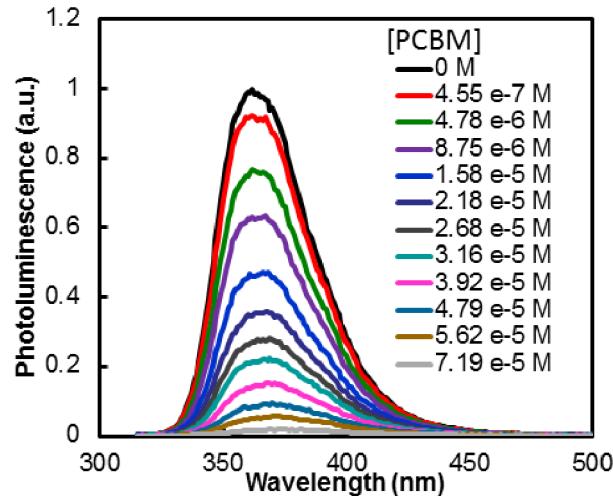
MOF177/PCBM



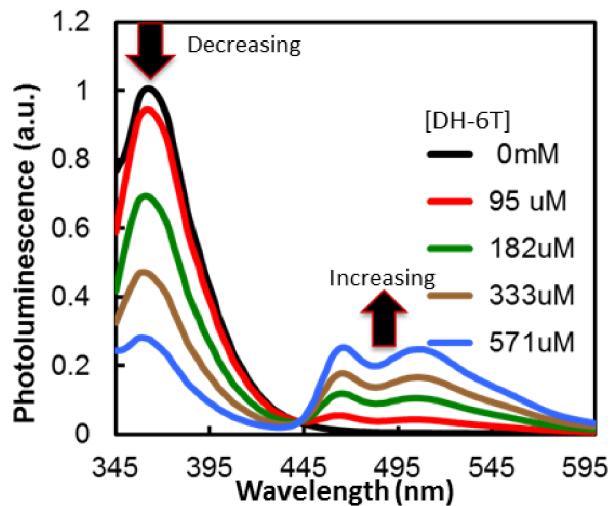
Spectroscopic characterization of PCBM@MOF177 cross-sections shows significant quenching of MOF177 fluorescence.

Introduction to MOFs

Spectral characterization reveals complex energy transfer between MOF177, PCBM, and DH6T

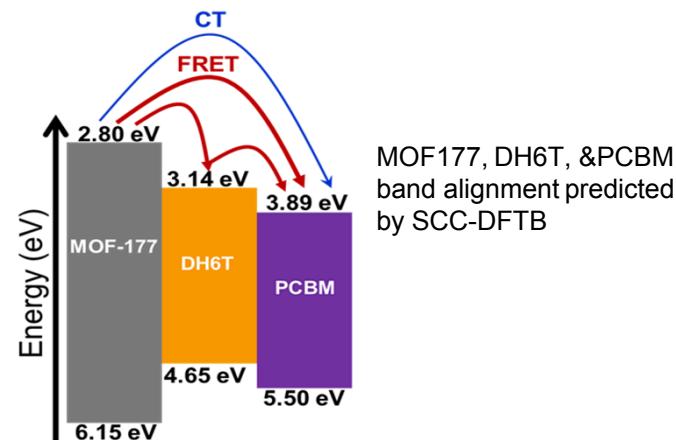
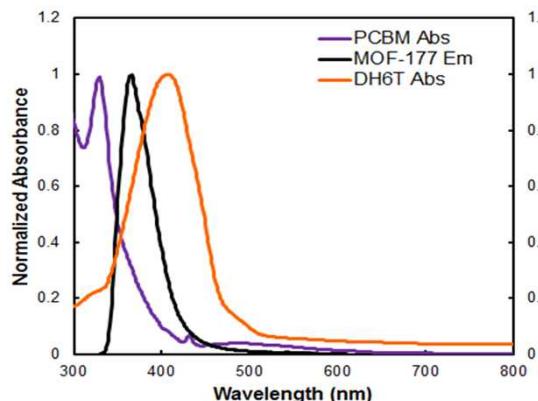


MOF177 and DH-6T are quenched by PCBM.



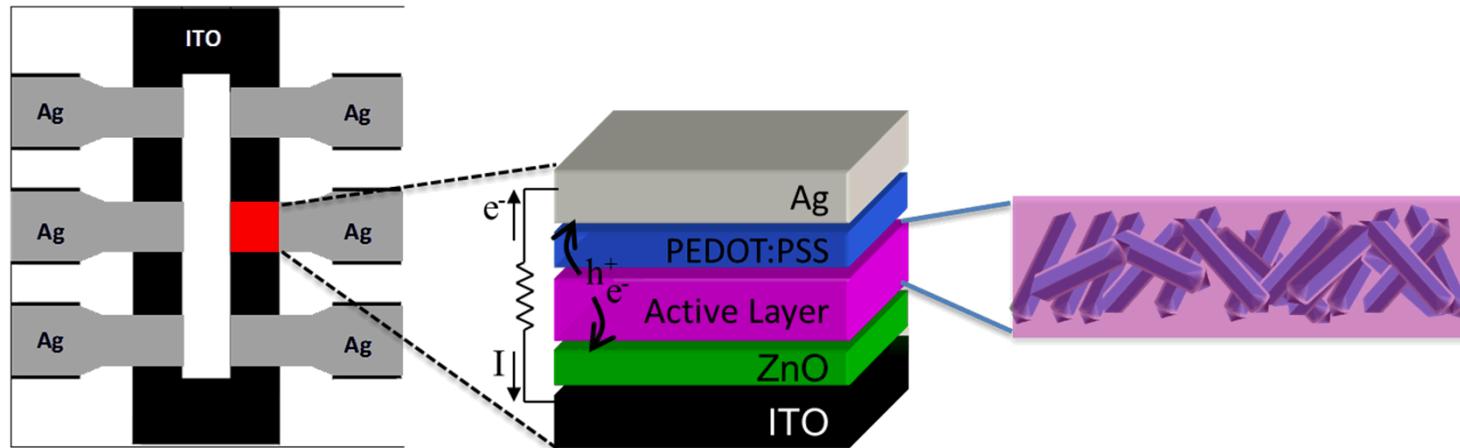
MOF177 transfers energy to DH-6T.

Spectral overlap between DH-6T, PCBM, and MOF177 inform multiple optoelectronic relationships.



Device Integration

PCBM@MOF177 were incorporated into hybrid OPV active layers to evaluate the influence of MOF templating on PV performance.

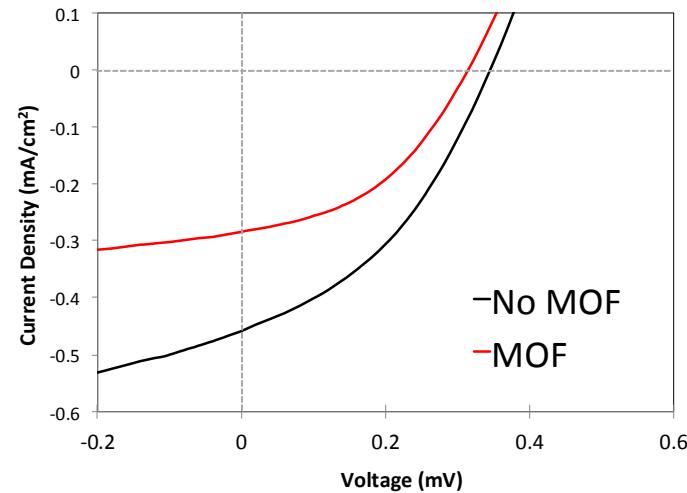
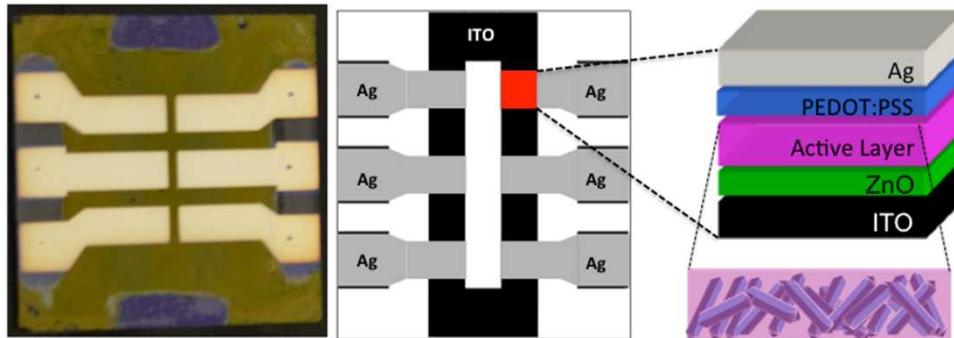


Patterned ITO and silver electrodes create controlled 0.1cm² active areas.

Inverted device configuration

Infiltrated (e.g., oligothiophene, PCBM) MOF177 particles incorporated into polythiophene active layer.

Device Testing



	V _{OC} (mV)	J _{SC} (mA/cm ²)	FF (%)	Efficiency (%)
With MOF177	320 +/- 23	0.290 +/- 0.018	40 +/- 0.3	0.04 +/- 0.003
P3HT Alone	340 +/- 26	0.460 +/- 0.029	39 +/- 1.9	0.06 +/- 0.01

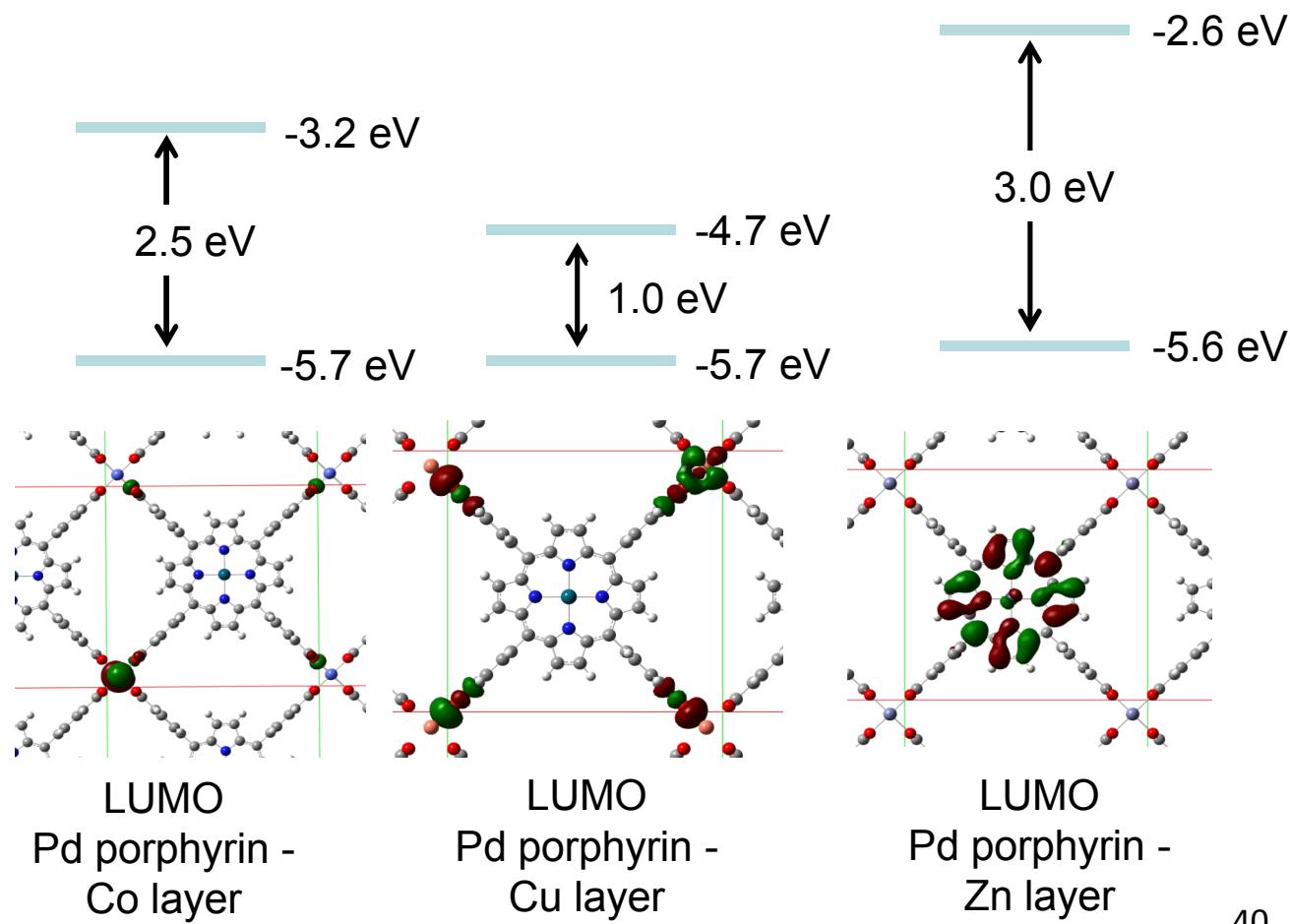
PV testing with 1 sun illumination shows reduced current, possibly from reduced active volume or charge trapping in suspended PCBM@MOF177 composites.

Band Structure Tailoring

Density Functional Theory (DFT) simulations show that by varying the composition of PPF molecular building blocks, it is possible to tune the electronic band structure of these MOFs.

Varying transition metal ions

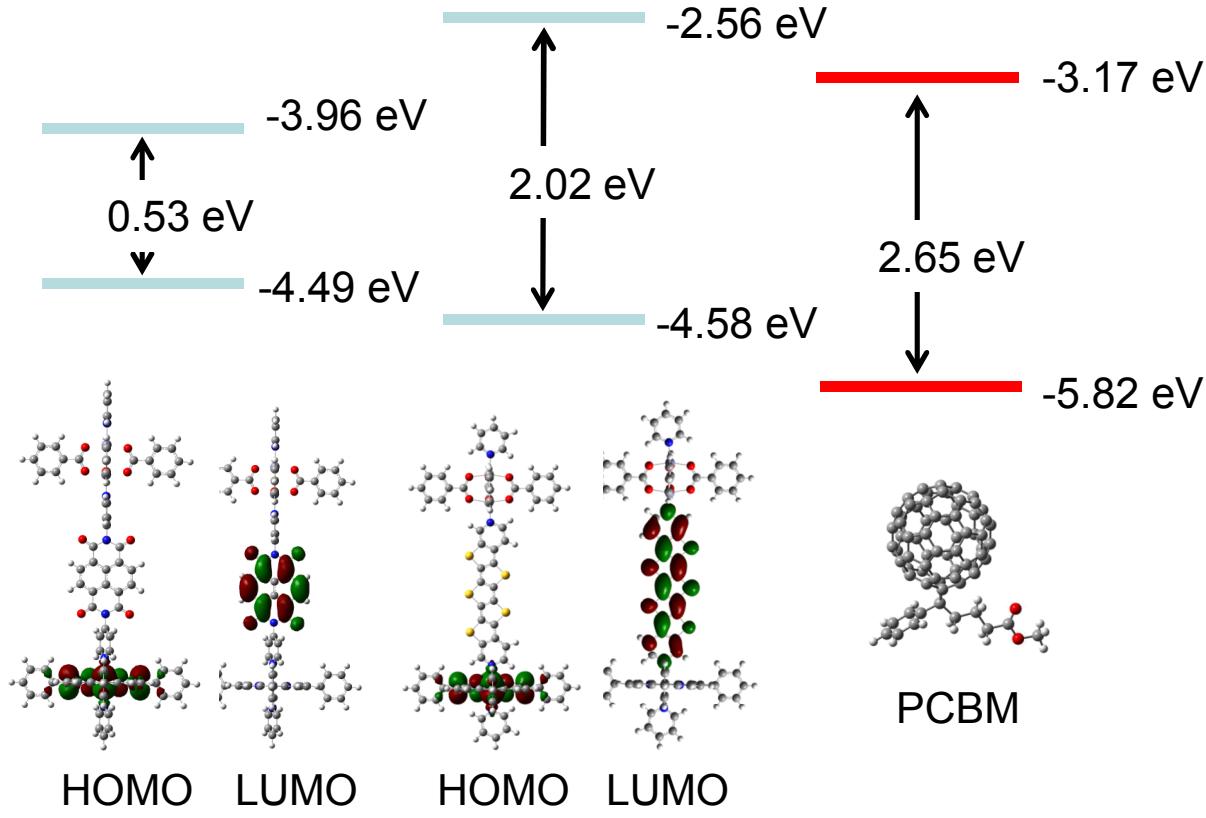
2D periodic
optimization -
DFT(B3LYP/ CEP-
31G)



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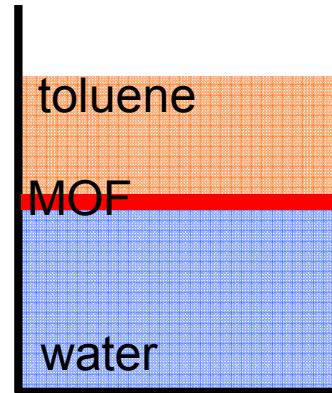
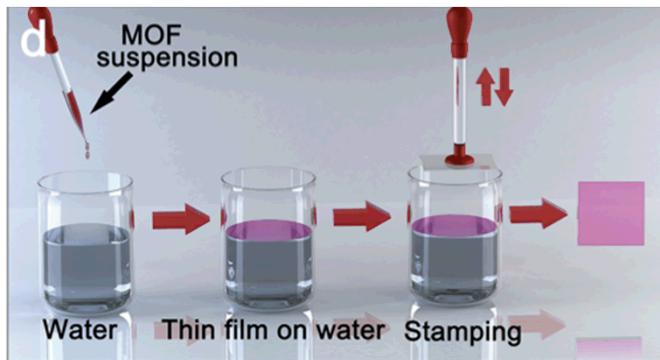
Varying organic pillars



2D periodic optimization -
DFT(B3LYP/ CEP-31G)

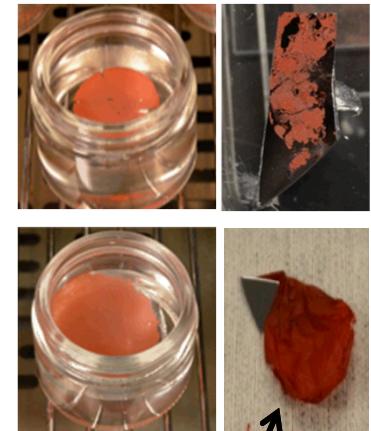
Revisiting Layered Growth

Can we deposit a thicker active layer to increase our current?



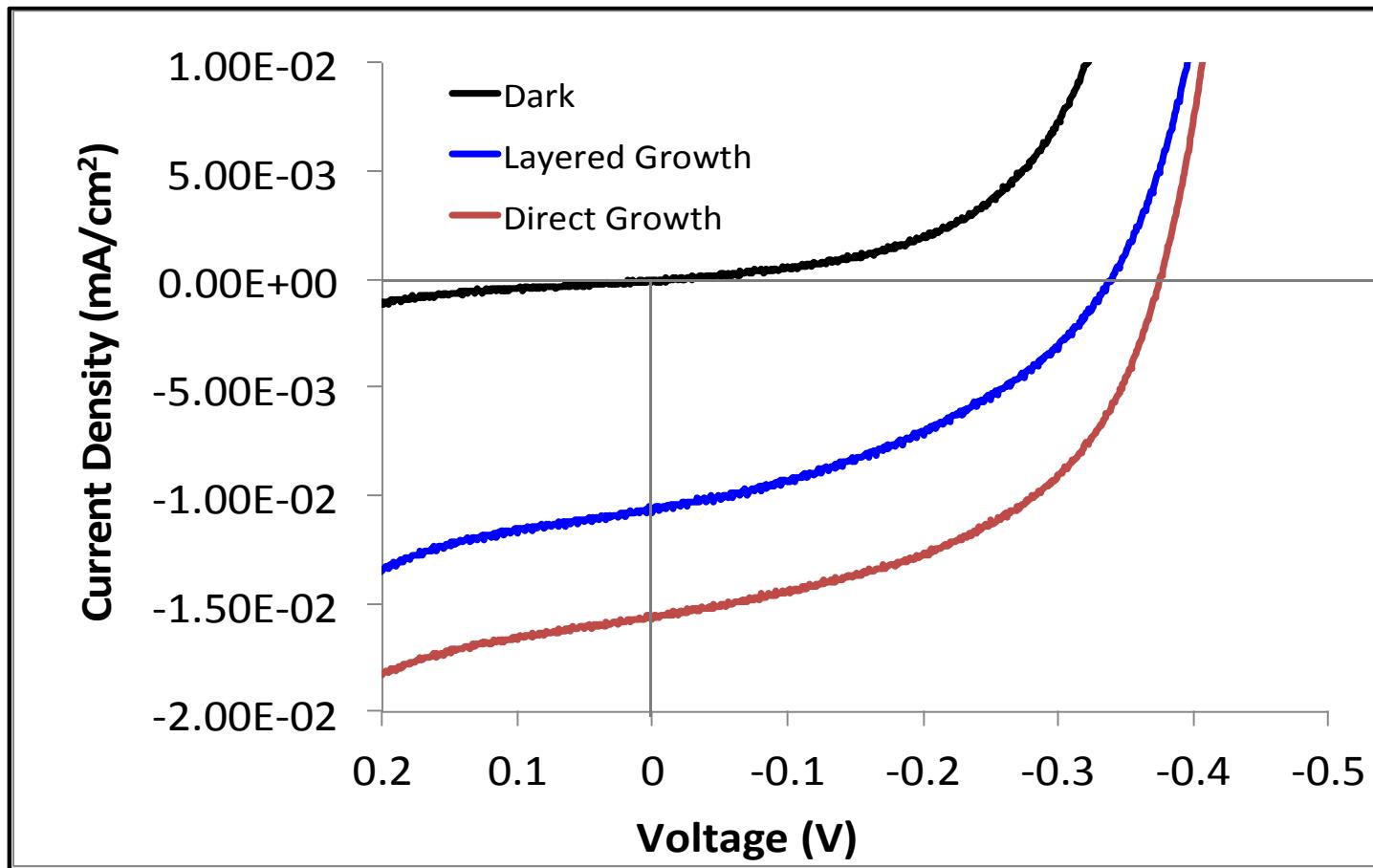
G. Xu, et al., JACS 2012, 134, 16524

Stamped onto
OTS/APS-treated
Si
(toluene +
acetone, 1:4)
Untouched on
water surface for
48 hours
(toluene + acetone,
1:4)



PPF-5 Integration is Important!

PPF-5 grown directly on TiO_2 yields greater PV response than PPF-5 deposited by layered growth



Requirements for efficient OPV system:

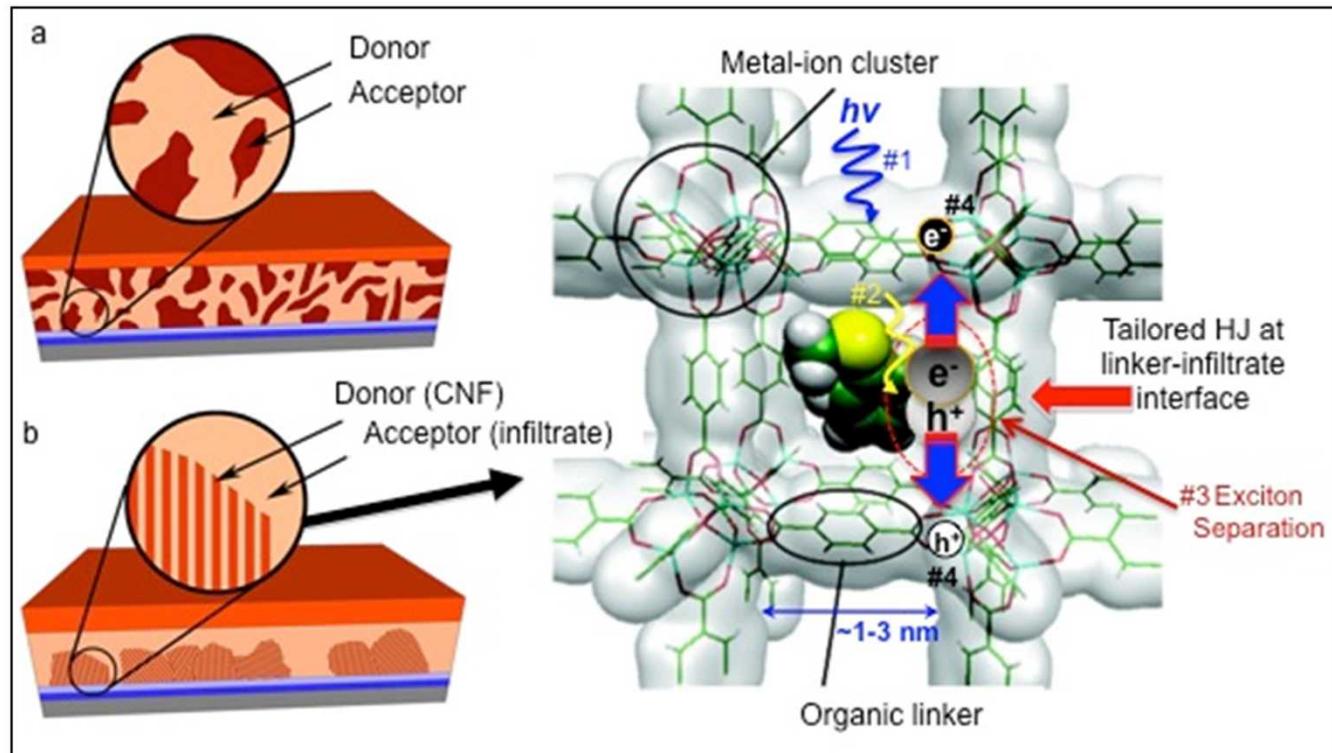
- Good p-type absorber
- Short exciton diffusion distances
- Ordered molecular charge separation interfaces
- Proper band alignment (D-A interface) for charge separation/transfer
- Facile incorporation into device architectures!

A Supramolecular Approach to PV Integration

Order vs. disorder: creation of nano-heterojunctions

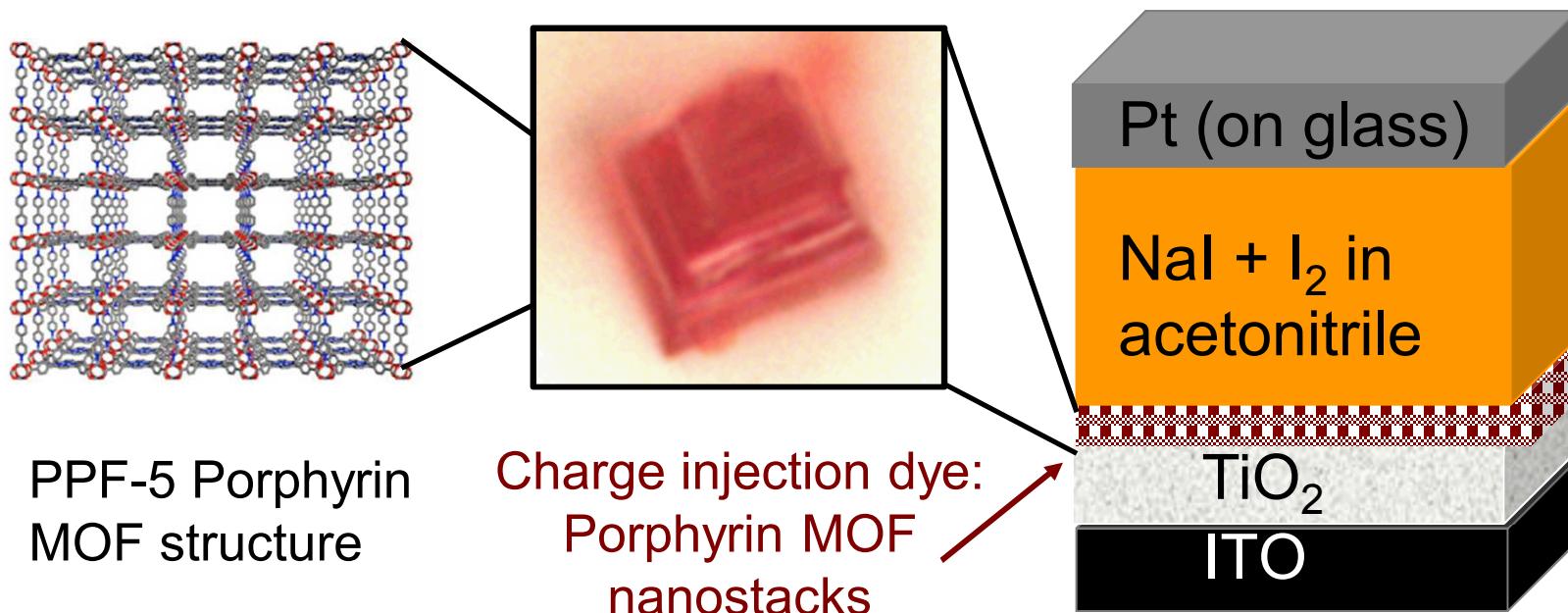
Conventional
disordered BHJ

Highly ordered
“Nano-HJ” using
CNF platform

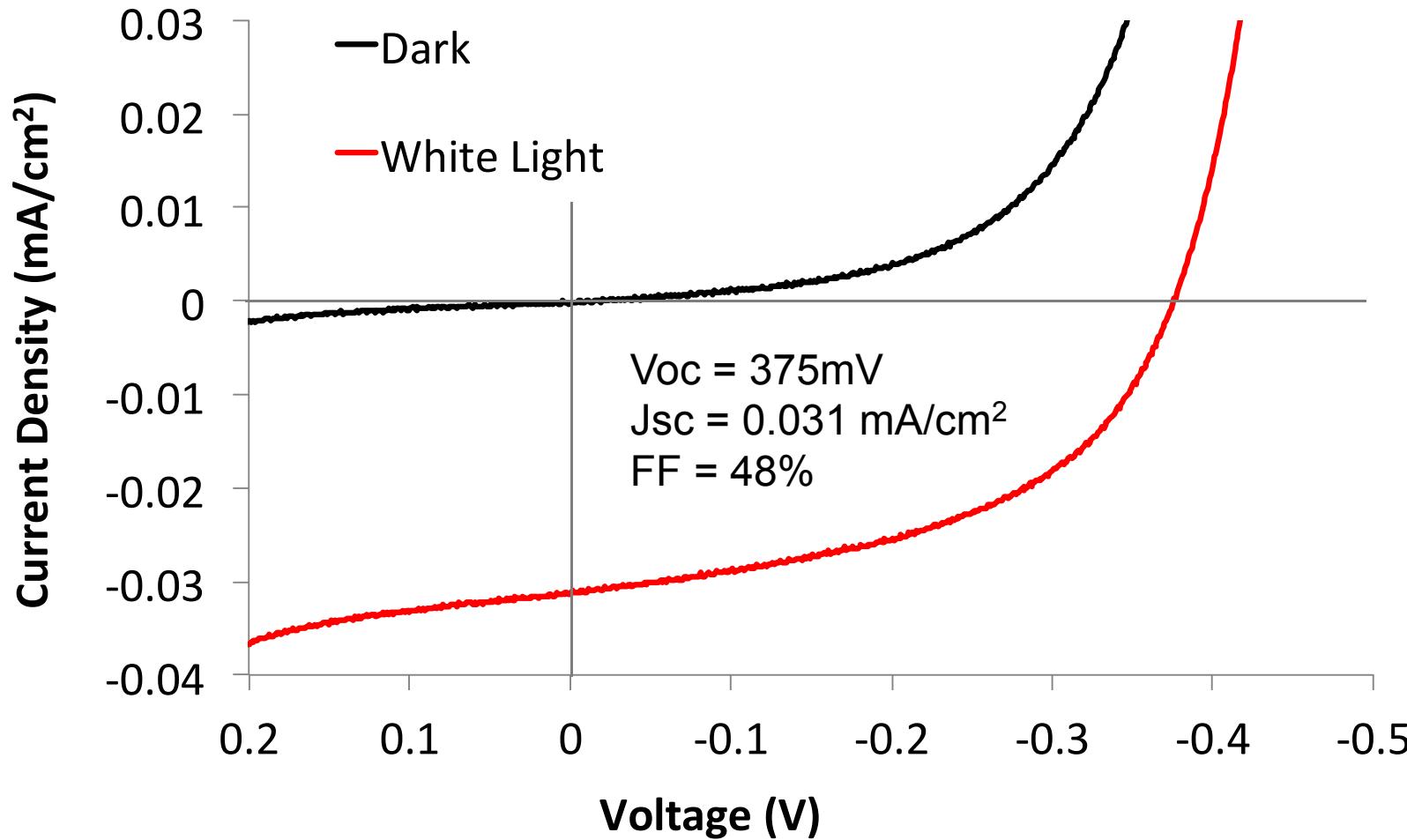


Integration of PPF-5 in DSSCs

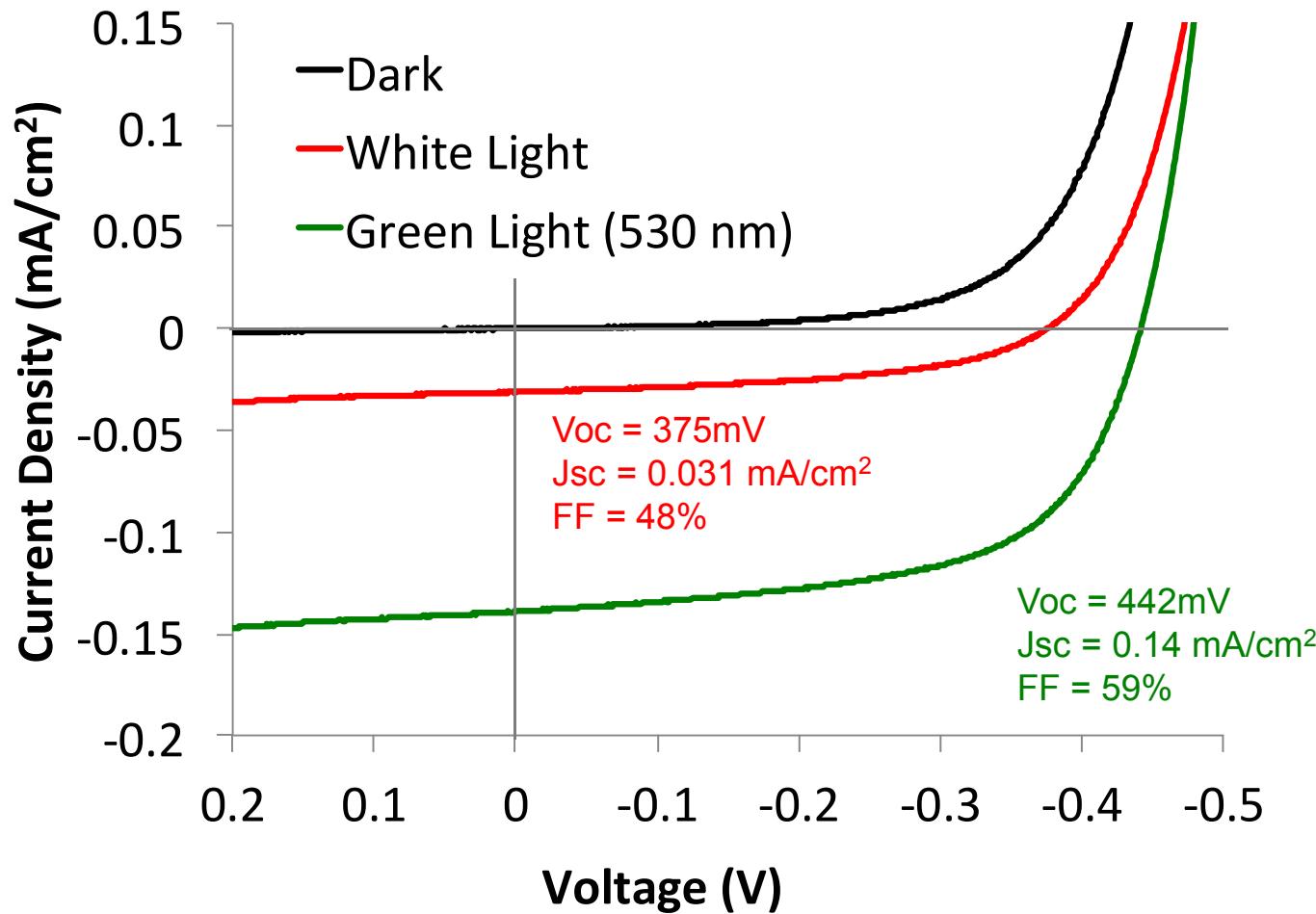
PPF-5 should be readily incorpoated into a DSSC device configuration...



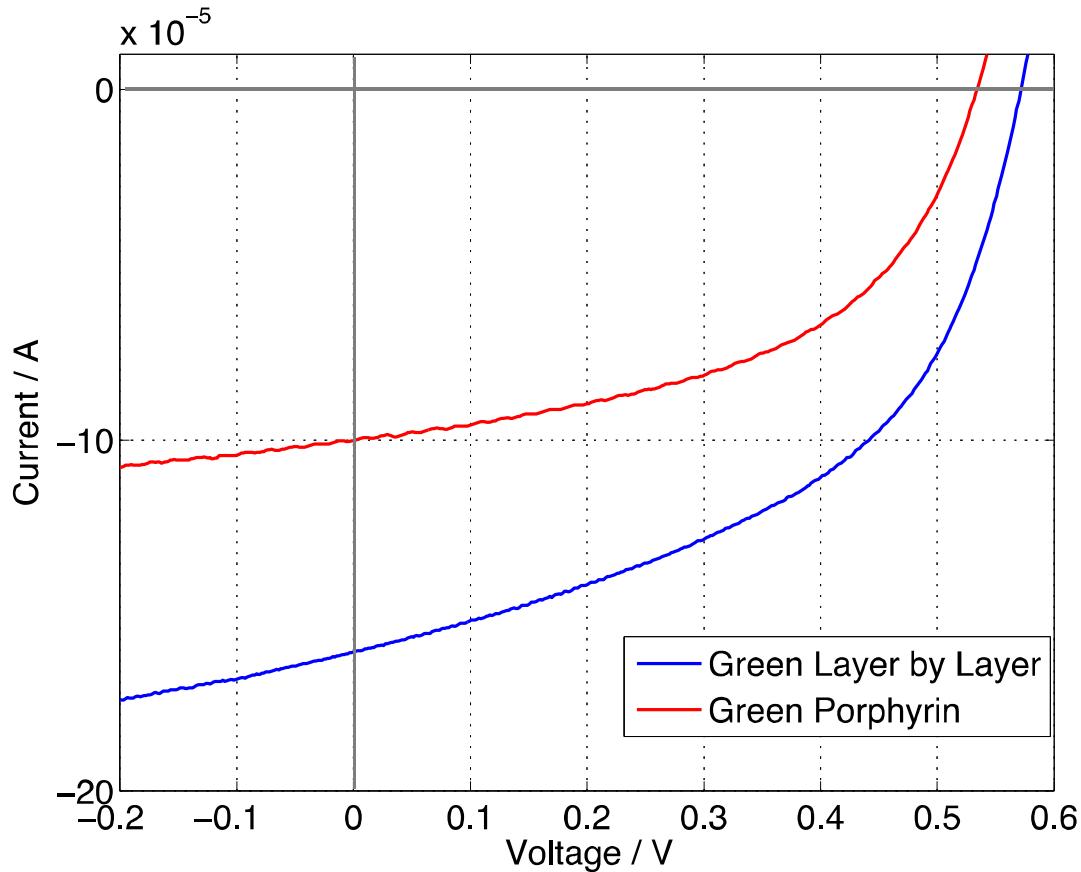
PPF-5 PV Performance in a DSSC



Optical “cheating”



Performance of PPF-5 in a DSSC



$V_{oc} = 0.57V$
 $J_{sc} = 1.6E-4 A$
 $FF = 0.49$

$V_{oc} = 0.532V$
 $J_{sc} = 1 E-4 A$
 $FF = 0.506$