

Directed Self-Assembly of 1D Microtubule Nano-Arrays

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Stevens, and George D. Bachand

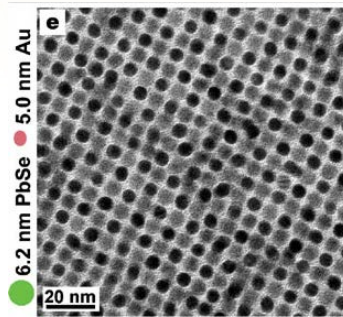
2014 MRS Fall Meeting

Symposium F: Reverse Engineering of Bioinspired Nanomaterials

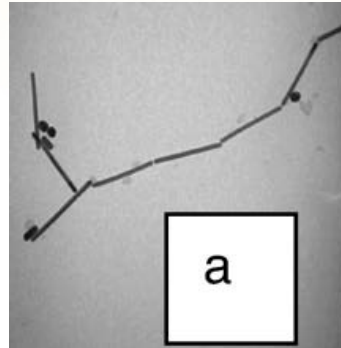
December 3, 2014

Self-Assembly of Nanoscale Building Blocks

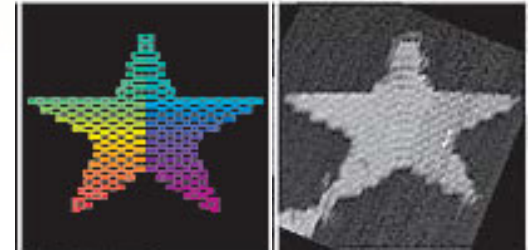
Large numbers of short-range interactions can be tailored to self-assemble nanoscale materials increasing structural and functional complexity.



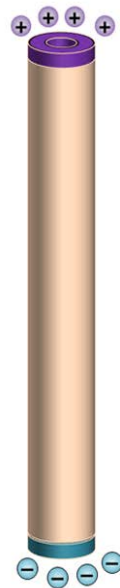
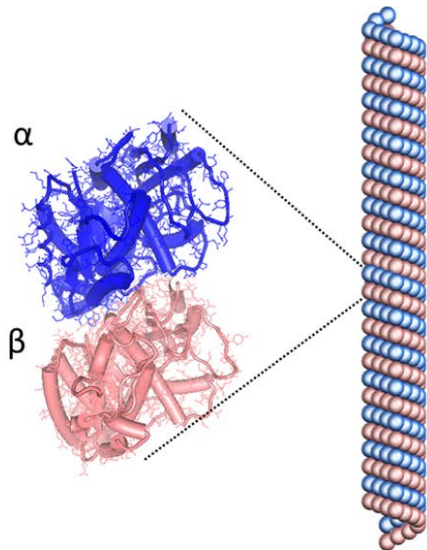
Shevchenko et al., *Nature* **439**, 55 (2006)



Caswell et al., *JACS* **125**, 13914 (2003)



Rothmund, *Nature* **440**, 297 (2006)



Microtubules (MTs):

- Hollow filament – 25 nm diameter, 10s of microns in length; intrinsic anisotropy/polarity
- Persistence length $\sim 5 \text{ mm}$; linear charge density of $\sim 260e^- \mu\text{m}^{-1}$
- Highly dynamic; stabilized *in vitro* w/ taxol
- Electrostatically repulsive rigid rods with attractive ends

Higher Order Assemblies of Microtubules

Lateral self-assembly

MTs bundles form in the presence of multivalent metal ions

- Short-range attractions caused by counterions form microtubule bundles

Hexagonal Bundle

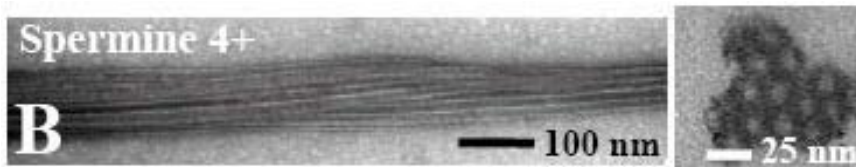
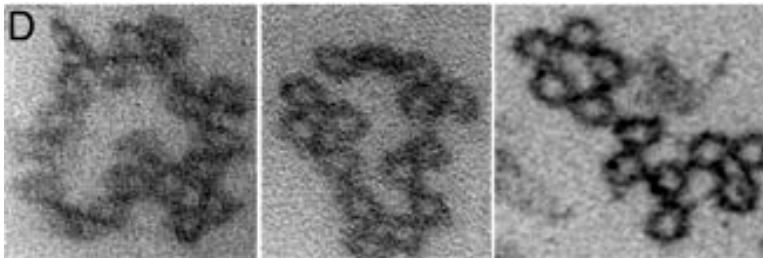


3+, 4+, 5+
Large Cation



2+
Small Cation

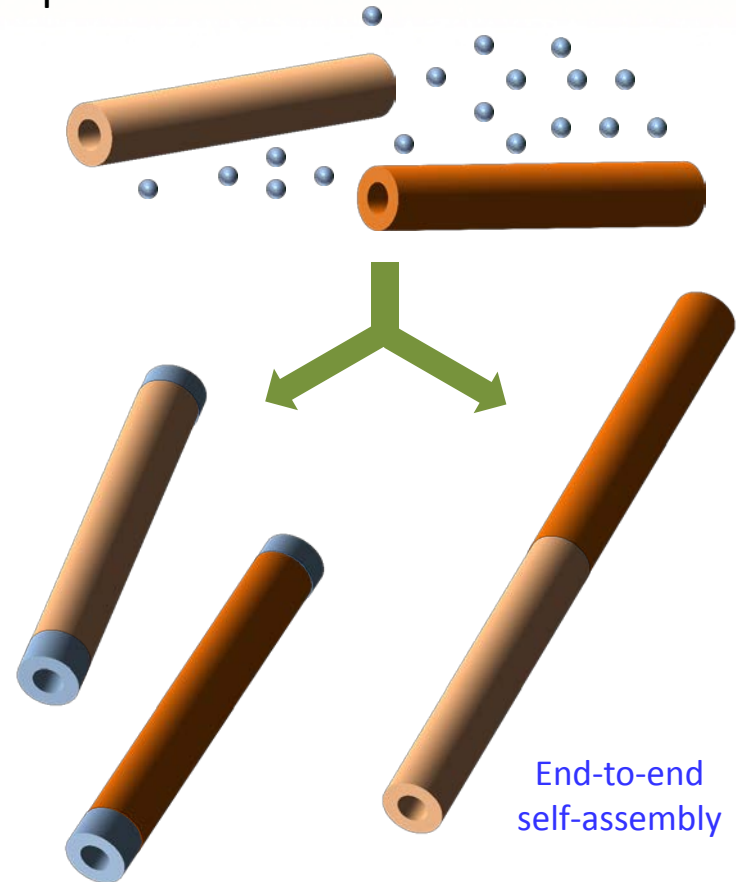
Living Necklace Bundle



Needleman et al., *Proc. Nat. Acad. Sci.* **101**, 16099 (2004)

Longitudinal self-assembly?

End-to-end self-assembly of MTs may also be possible



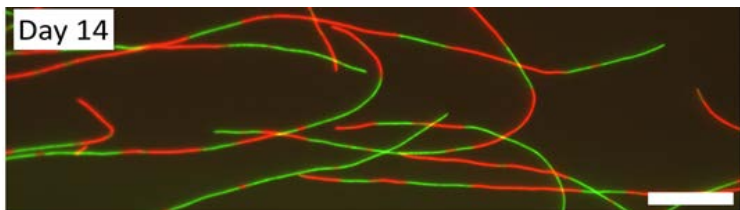
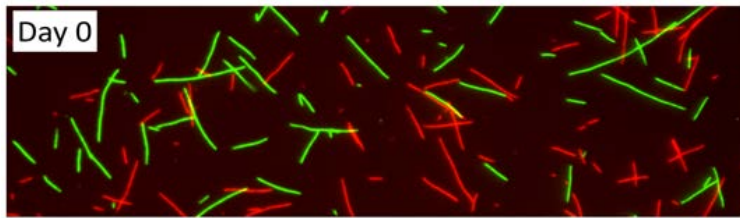
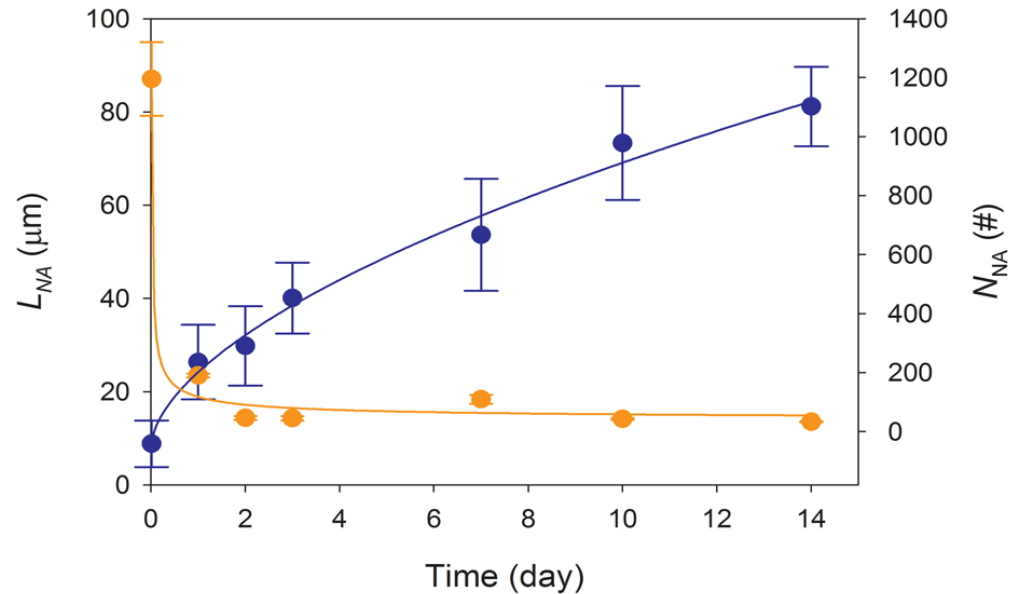
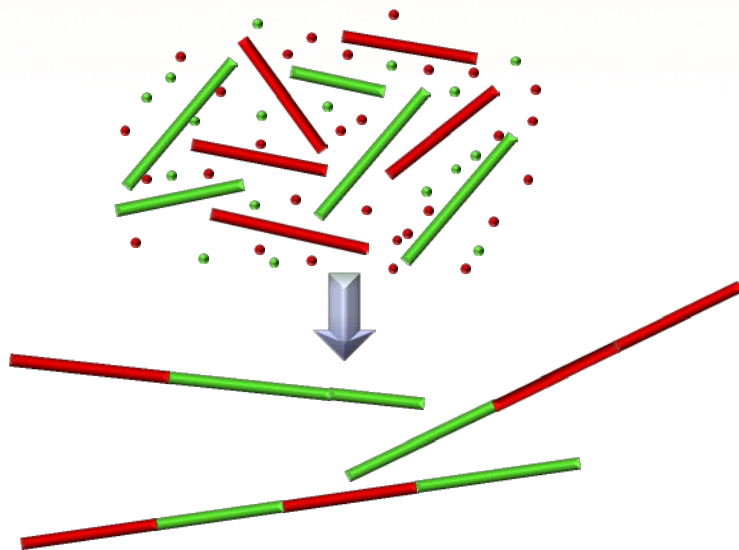
Seeded
polymerization



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Directed Self-assembly (DSA) of MTs

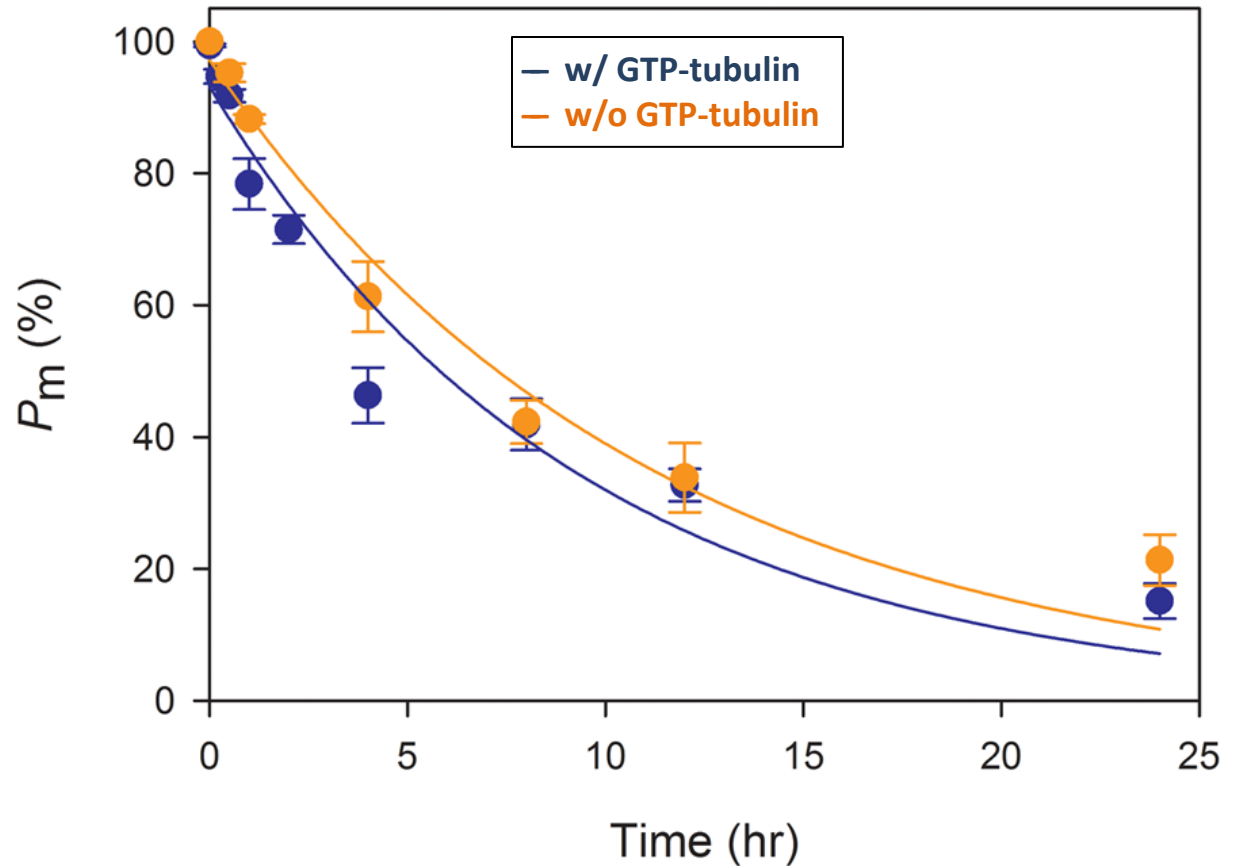
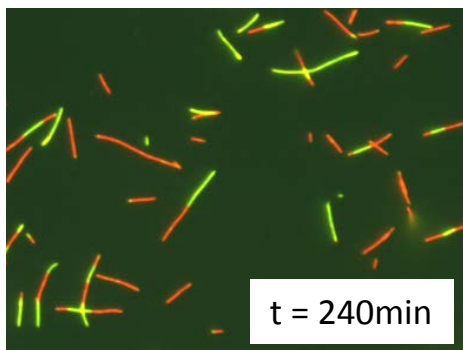
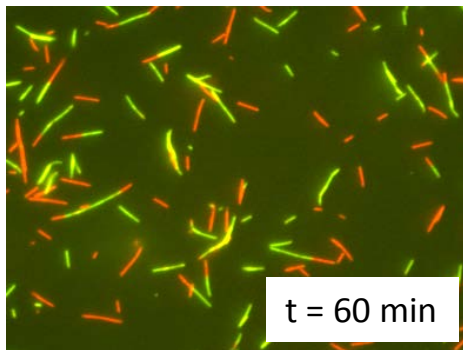
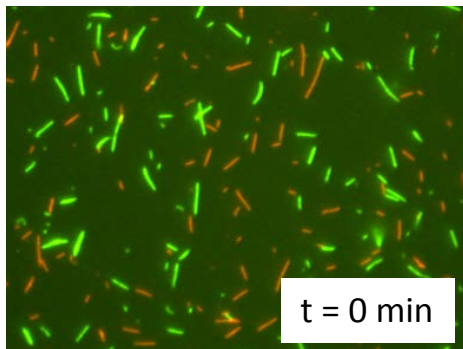
Taxol-stabilized MTs self-assemble head-to-tail to form extended 1D oligomeric nano-arrays (NAs)



- Average length increased by >10-fold over 14 days
- Δ in the number & length of NAs inversely correlated (Spearman rank correlation = -0.893; $P < 0.001$)
- Δ in length = [DSA](#), not seeded polymerization

Reaction Rate of Self-Assembly

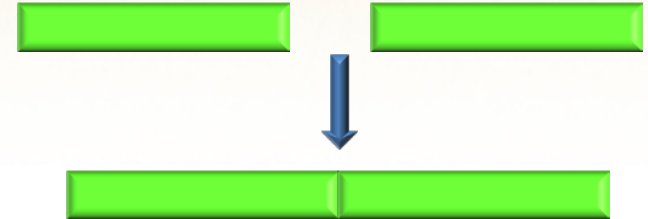
Kinetics of NA assembly characterized by disappearance of “monomers” over time. Data suggest that DSA follows **second-order reaction kinetics**.



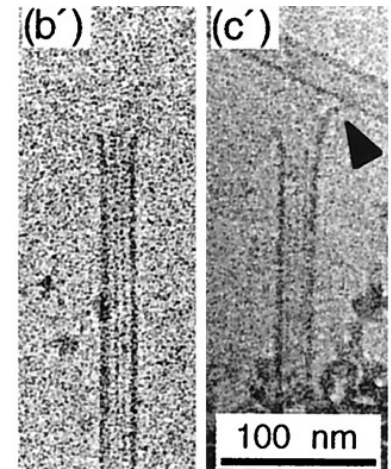
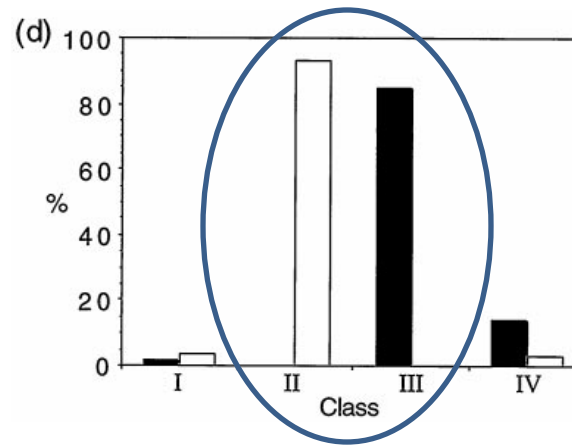
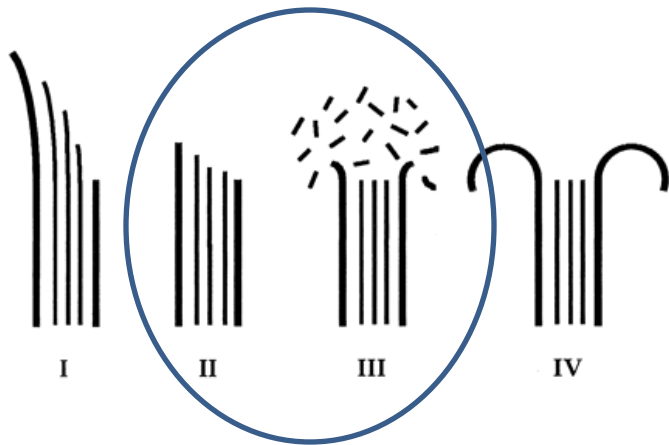
Understanding the Mechanism of DSA

Simplest mechanism:

- Microtubule ends collide via thermal fluctuation
- Short-range interactions join two segments
- Stable junction formed b/w segments



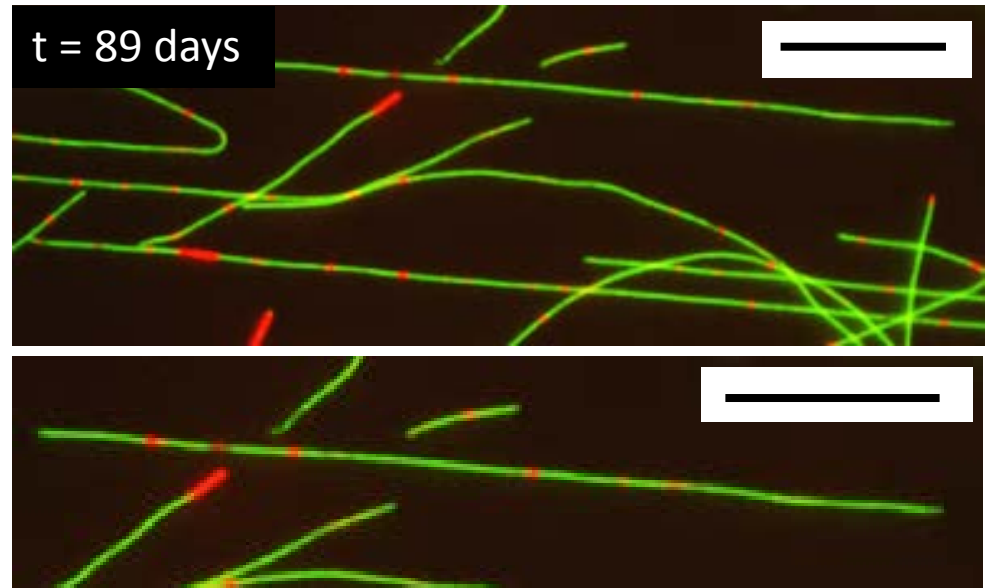
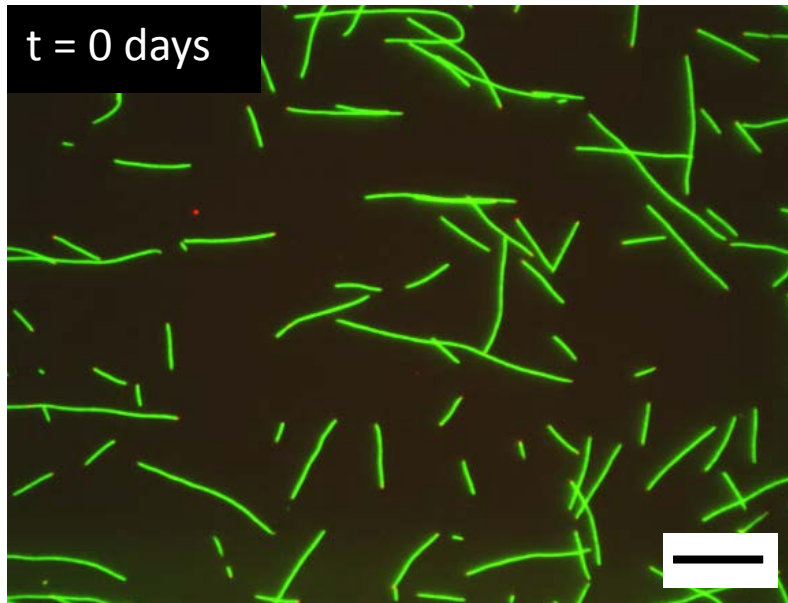
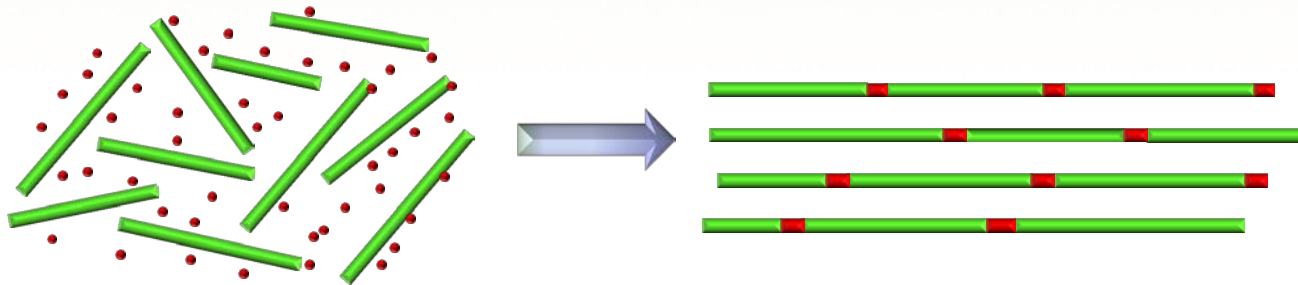
Microtubule ends are rarely blunt end. What happens at junctions between assembling segments?



Muller-Reichert et al., *Proc. Nat. Acad. Sci.* **95**, 3661 (1998)

Two-Step Process for MT DSA

Addition of free red tubulin to green microtubules supports two-step mechanism in the overall fusion process.



BUT...what is the exact mechanism?

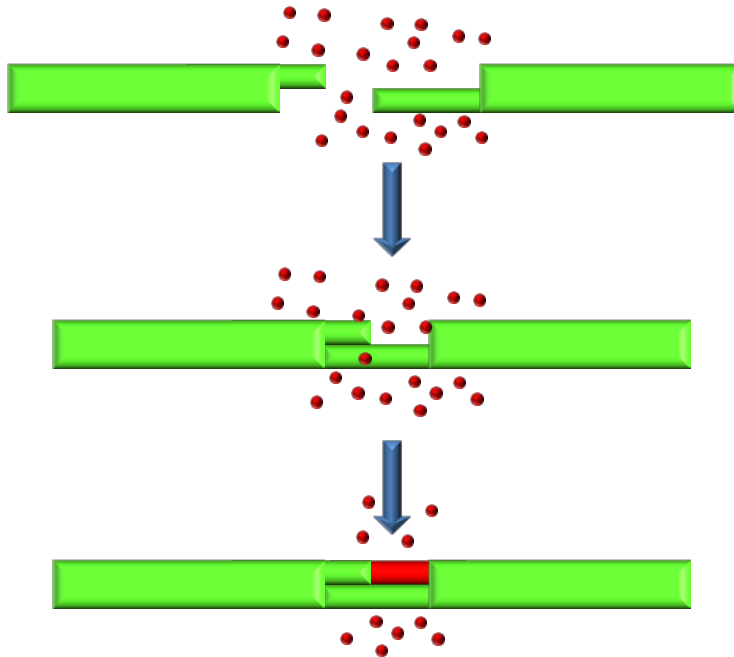
Bachand et al., *RSC Adv.* **4**, 51641 (2014)



Proposed Mechanisms for MT DSA

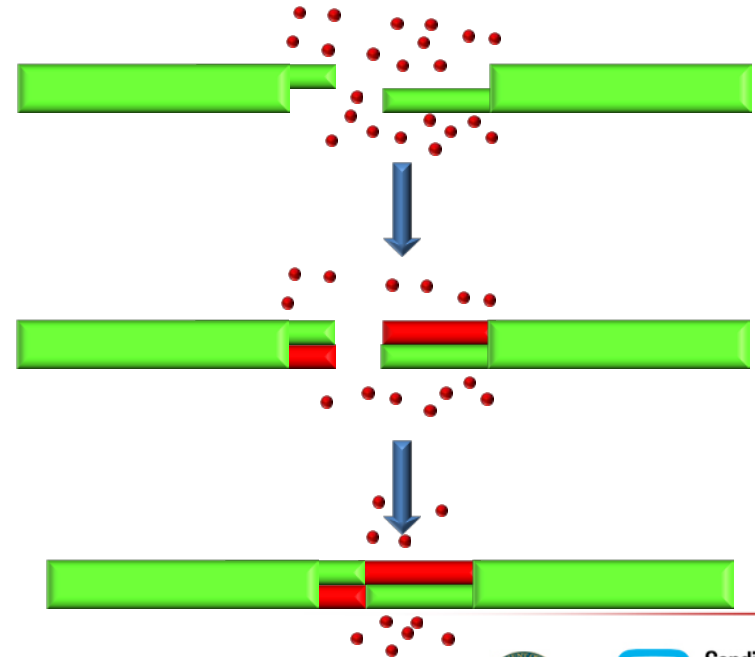
1 – Vacancy Filling

- MT ends collide via thermal fluctuation
- Short-range interactions hold segments together weakly (with defect)
- Free GTP-tubulin dimers fill lattice vacancies and stabilize the junction



2 – Blunt ending

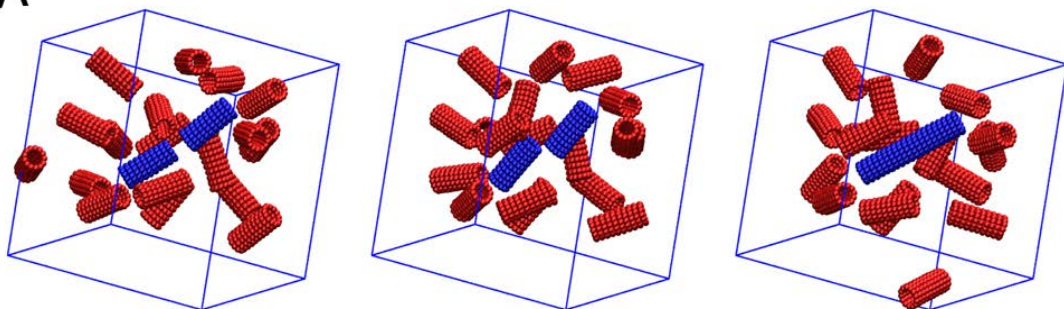
- Free GTP-tubulin dimers polymerize at the MT ends forming a blunt end
- MT ends collide via thermal fluctuation
- Short-range interactions hold segments together until junction forms



Two-Step Process for DSA of MTs

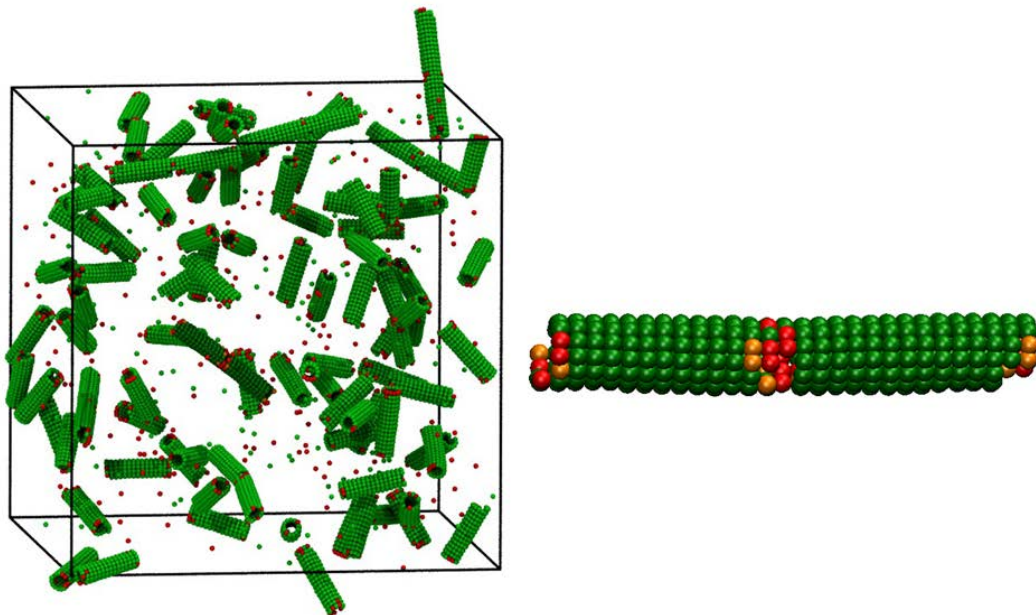
MD simulations of fusion support all of the mechanisms, suggesting that all may play a role in the direct assembly process.

A



MD simulation demonstrates that blunt-ended MTs will self-assemble into extended 1D structures

B



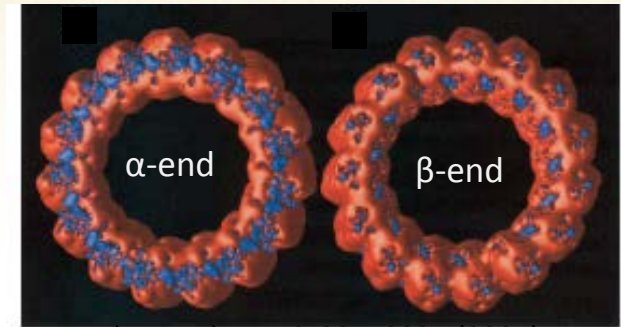
Simulations with jagged-ended MTs demonstrate that:

- Seeded polymerization occurs at ends
- Jagged-ends MTs will assemble
- Vacancies filled by free & dissociated dimers

Role of Short-Range Electrostatics

Bonds among dimers in MTs:

- ΔG longitudinal bond > ΔG lateral bonds
- Hydrophobic and electrostatic interactions
- α - and β -ends are electrostatically attractive



Baker et al. *PNAS*, **98**, 10037 (2001)

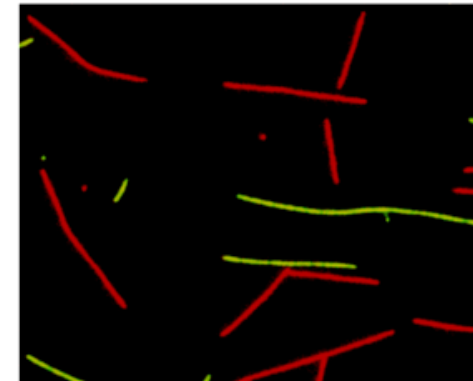
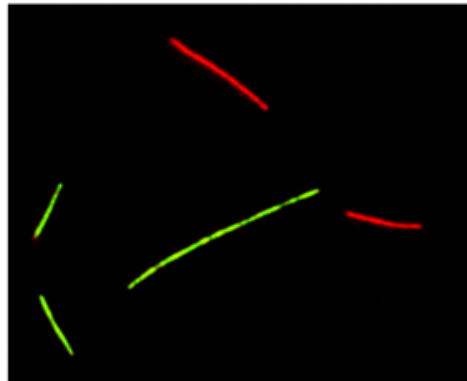
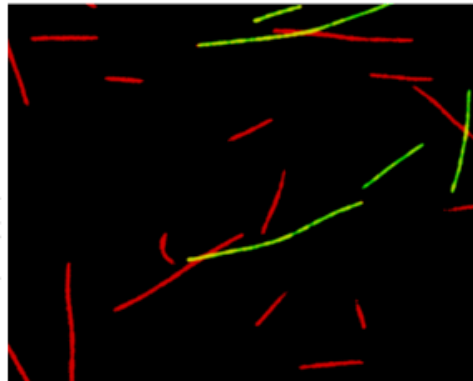
Will MTs self-assemble if the Coulomb interactions are screened?

0M NaCl

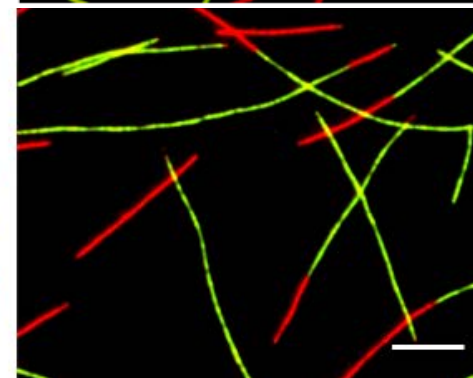
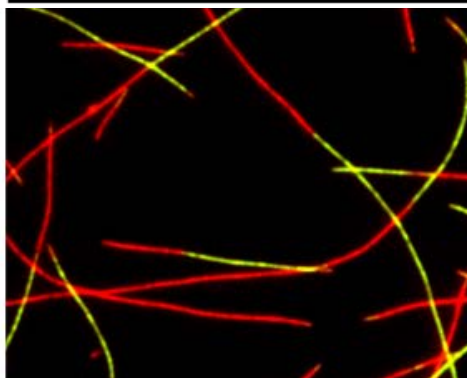
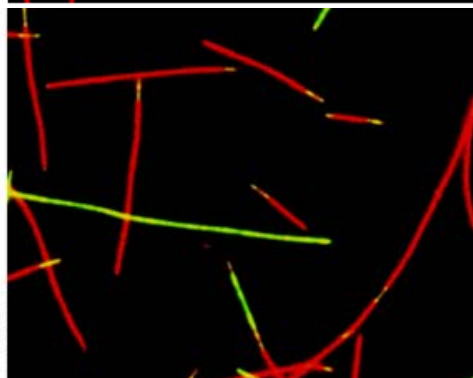
100mM NaCl

500mM NaCl

0 Hours



4 Hours



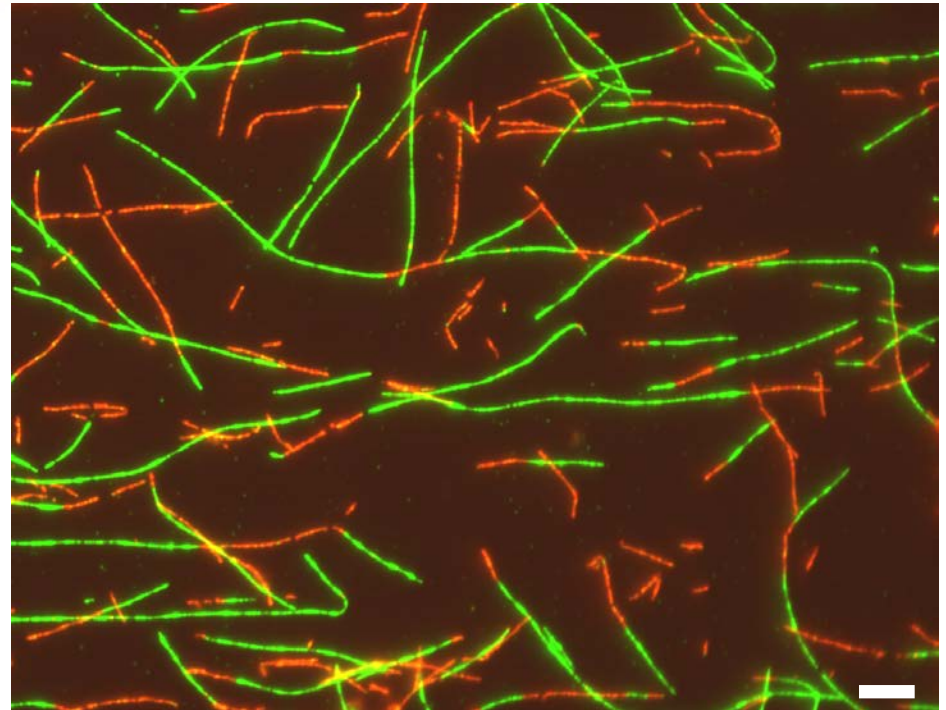
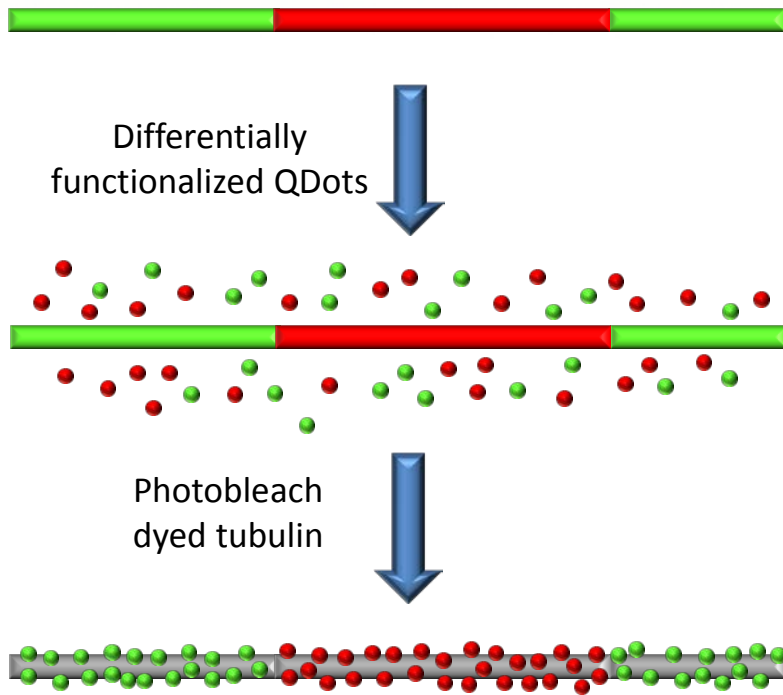
Charge neutralization appears to increase the rate of self-assembly



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Future: Heterostructured Nanowires

1D heterostructured nanowires (aspect ratio $>3,000>1$) can be formed from assembled microtubule structures where individual segments have unique chemical functionality.



Morphology of these nanowires can be tuned by altering the assembly process (e.g., shearing one segment type to achieve uniform size distribution).

Acknowledgments

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