

# Novel Particle Detector Materials Based on Zinc-Stilbene Metal Organic Frameworks

---

F. P. Doty,<sup>a</sup> C. A. Bauer,<sup>b</sup> P. G. Grant,<sup>c</sup>  
B. A. Simmons,<sup>a</sup> A. J. Skulan,<sup>a</sup> M. D. Allendorf<sup>a</sup>

The Electrochemical Society  
211<sup>th</sup> National Meeting  
Chicago, Illinois  
May 6 – 10, 2007

<sup>a</sup>Sandia National Laboratories  
Livermore, California

<sup>b</sup>Georgia Institute of Technology  
Atlanta, Georgia

<sup>c</sup>Lawrence Livermore National Laboratory  
Livermore, California



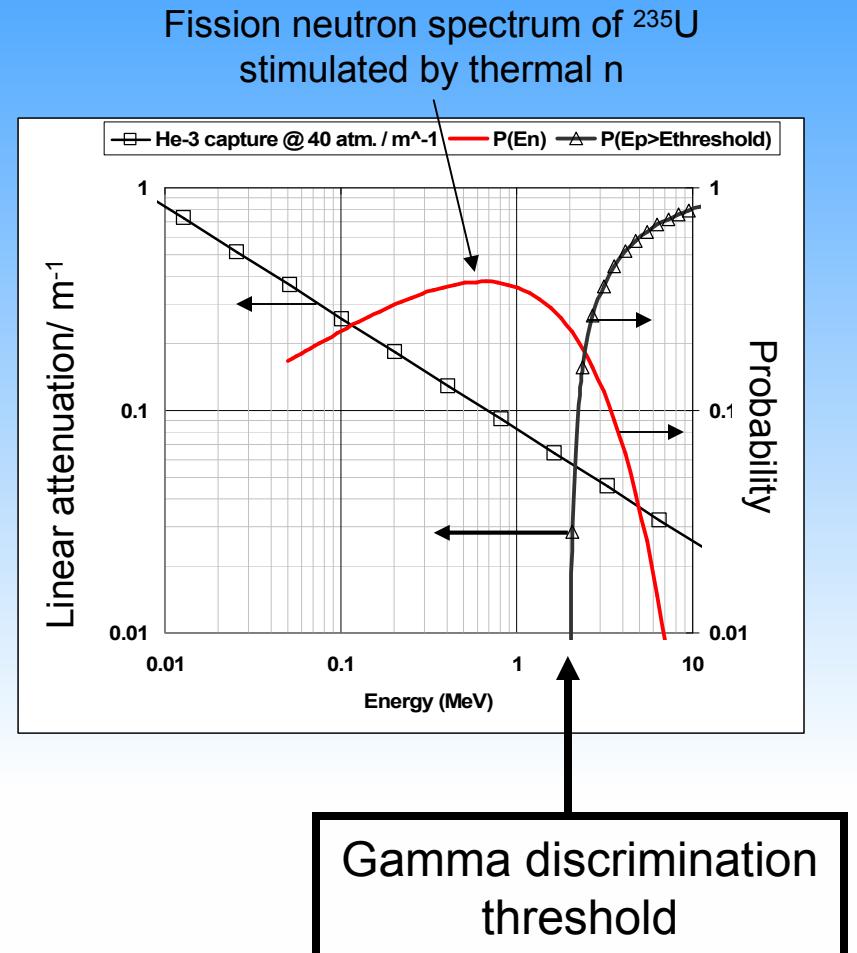
Sandia National Laboratories

# Direct detection of fission-spectrum neutrons is inefficient using existing sensor materials

- Fission neutrons:
  - Sources:  $^{235}\text{U}$ ,  $^{241}\text{Pu}$
  - Spectra peak < 1 MeV (Maxwell dist.)
- Applications:
  - Detecting nuclear proliferation
  - Long term monitoring
  - Cargo screening
  - Neutron radiography
- Detection methods:  
 $^3\text{He}$  capture (**Moderator Required**)
  - Efficient only at  $E_n < 0.01$  MeV
  - Direct detection impractical (> 1m, 40 atm)  
→ Particle direction and energy are lost

Liquid organic scintillators (**High Threshold**)

- Low detection efficiency at  $E_n < 10$  MeV
- Combustible
- Toxic solvents



New scintillator materials are needed to enable practical detection of these particles



Sandia National Laboratories

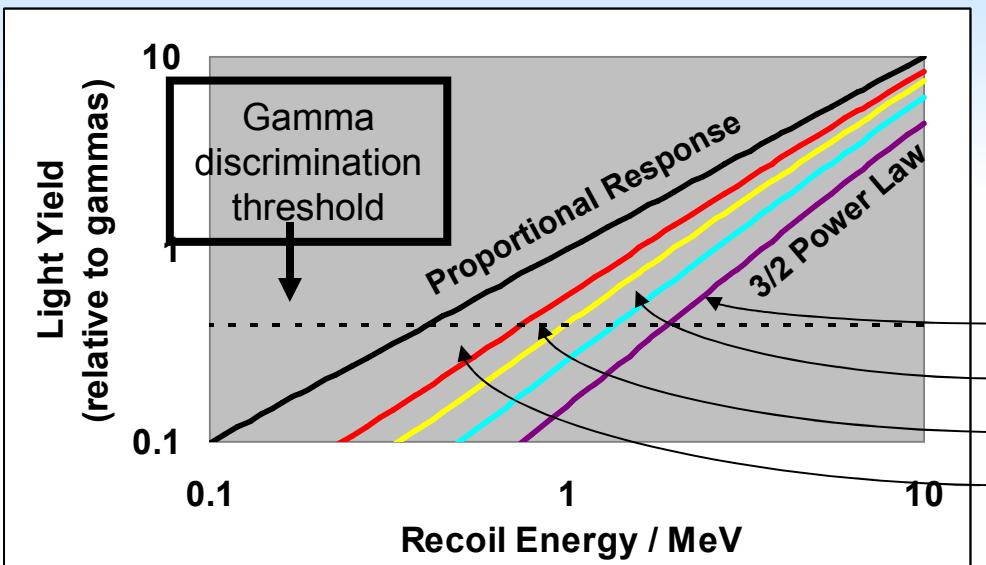
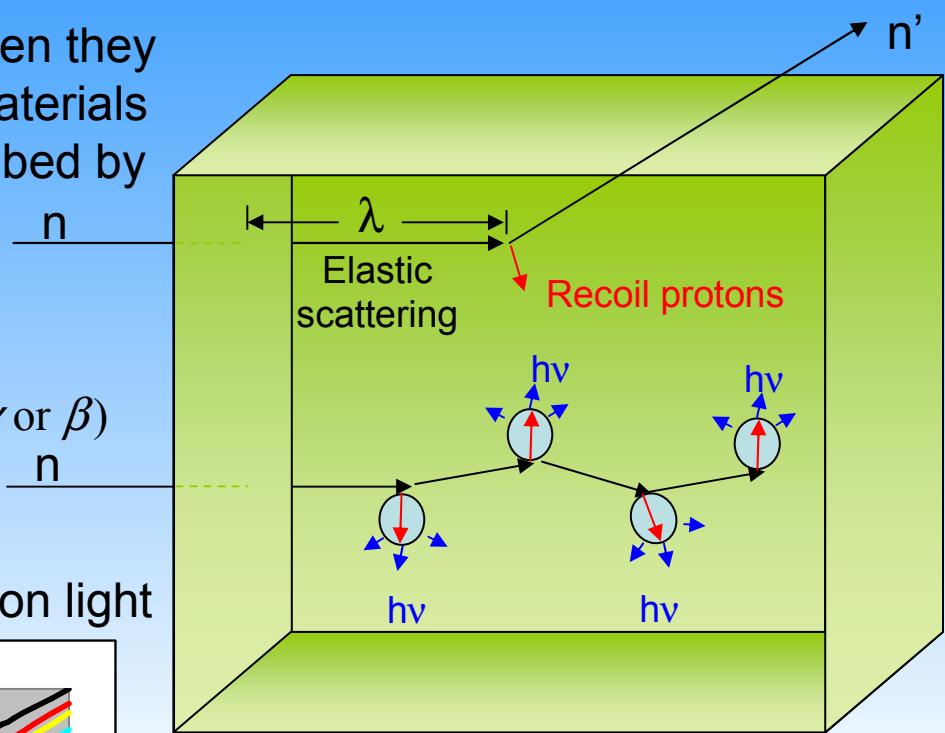
## Scintillation light is partially quenched at high $dE/dx$

- Neutrons create ionizing protons when they elastically scatter in hydrogenous materials
- Light output from ion tracks is described by Birks' equation:

$$L(E) = S \int_0^E \frac{dE}{1 + kB(dE / dx)}$$

$L(E) = SE$  in the limit of low  $dE / dx$  ( $\gamma$  or  $\beta$ )

- $dE/dx \propto$  density
- Predicts decreasing density will decrease quenching of scintillation light



**Commercial Scintillator EJ301:**  
Birks eq. fits observed  $L(E) \sim E^{3/2}$   
Density  $0.88 \text{ g/cm}^3$

Density  $0.44 \text{ g/cm}^3$

Density  $0.22 \text{ g/cm}^3$

Density  $0.11 \text{ g/cm}^3$



Sandia National Laboratories

# $dE/dx$ Scintillation Quenching

## *A case of static fluorescence quenching*

Fluorescence quenching is modeled by :



Where

$[M^*]$  = Fluorescent state concentration

$[Q]$  = Quenching species

$[MQ^*]$  = Nonradiative complex

$$K = \frac{[MQ^*]}{[M^*][Q]} \longrightarrow [MQ^*] = K[Q][M^*]$$

$$[M_{total}^*] = [M^*] + [MQ^*]$$

$$[M_{total}^*] = [M^*] + K[Q][M^*]$$

$$\frac{[M_{total}^*]}{[M^*]} = \frac{L_0}{L} = 1 + K[Q]$$

Stern-Volmer  
equation

Along a particle track we assume :

$[M^*] \propto$  Light yield / unit track length

$[M_{total}^*], [Q] \propto$  Energy / unit track length

$$[M^*] \propto (dL / dx)$$

$$[Q] \propto (dE / dx)$$

$$[M_{total}^*] \propto (dE / dx)$$

*Substituting into Stern-Volmer*

$$S \frac{(dE / dx)}{(dL / dx)} = 1 + kB(dE / dx)$$

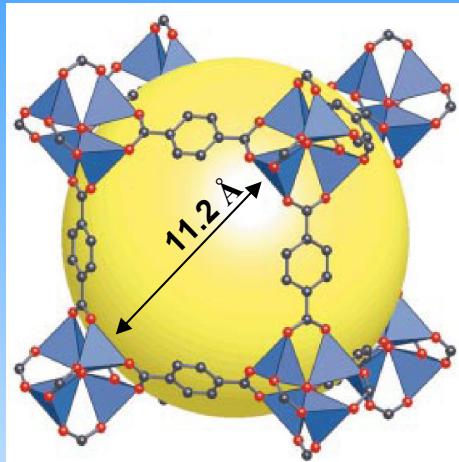
$$\frac{dL}{dx} = S \frac{(dE / dx)}{1 + kB(dE / dx)}$$

Birks'  
equation

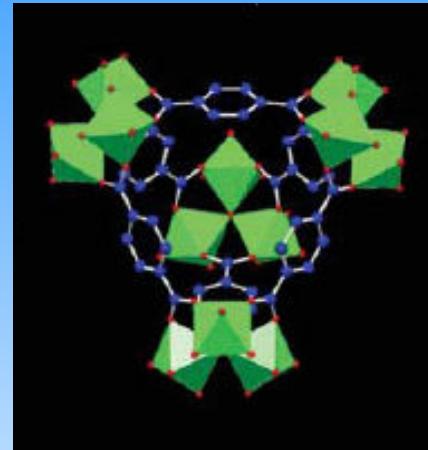


Sandia National Laboratories

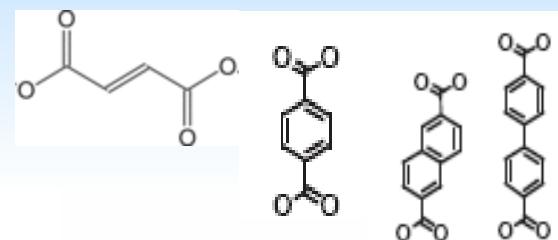
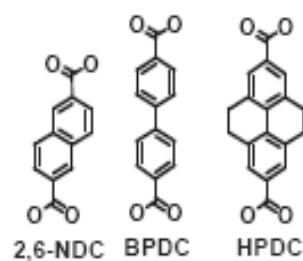
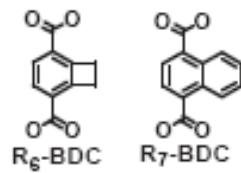
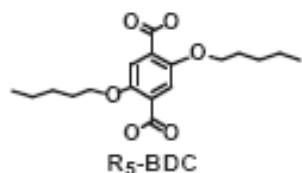
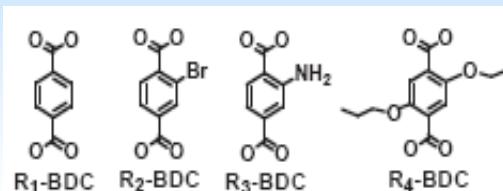
# Nanoporous coordination polymers provide an opportunity for tunable scintillator density



**IRMOF-1:** Rigid, open framework with tunable pore size, chemistry (Yaghi et al., *Science* 2002)



**Cr and Fe MIL:** 6000 m<sup>2</sup>/g surface area (Férey et al., *Science* 2005)

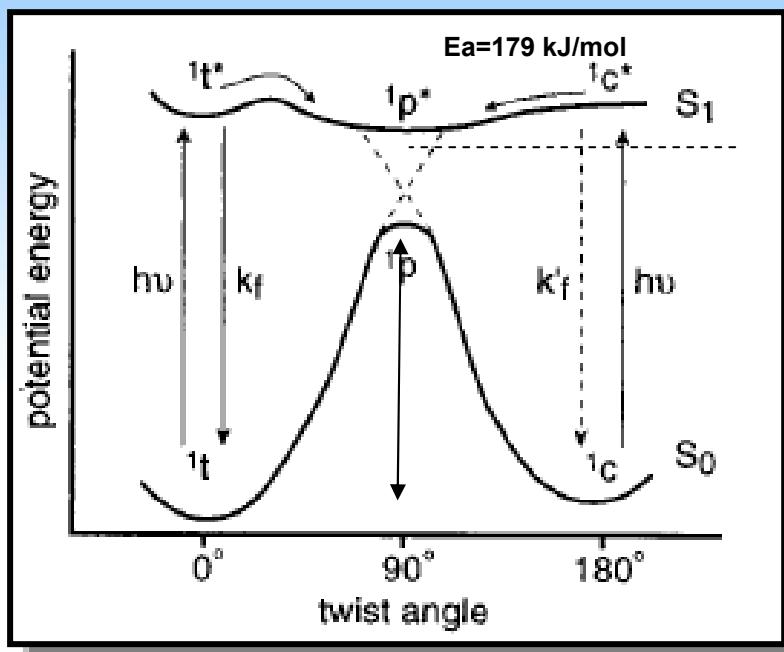


Sandia National Laboratories

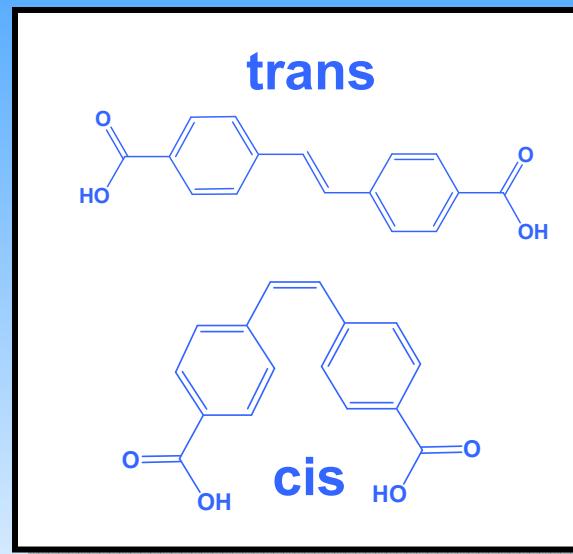
# Fluorescent MOFs: entry point for creating scintillators based on coordination polymers

Stilbene: a fluorescent linker for MOF synthesis

## Potential energy surface



A. Simeonov, *Science*, 2000



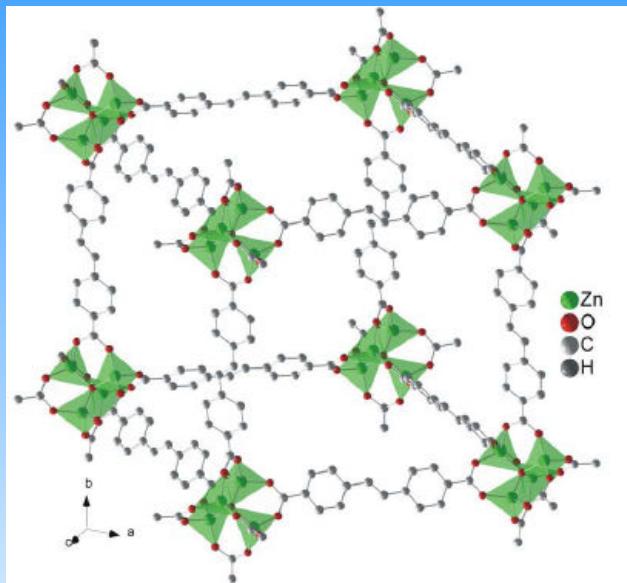
## Stilbene linker offers:

- Potential for low density frameworks
- Constrained linker conformation
  - High fluorescence quantum efficiency
- Zn-based compounds: transparent
- Low self adsorption

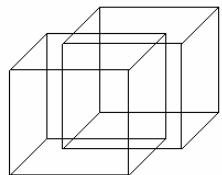


Sandia National Laboratories

# New stilbene MOFs: multiple structures result from the same starting materials under different synthetic conditions



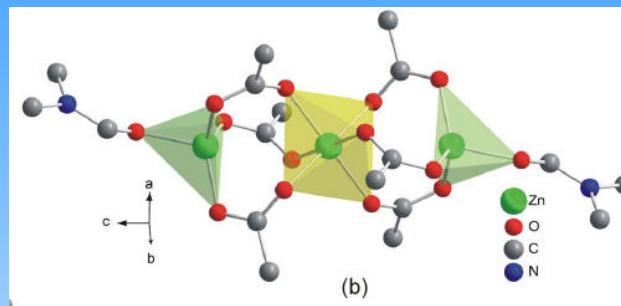
Formula unit:  $\text{Zn}_4\text{OL}_3$



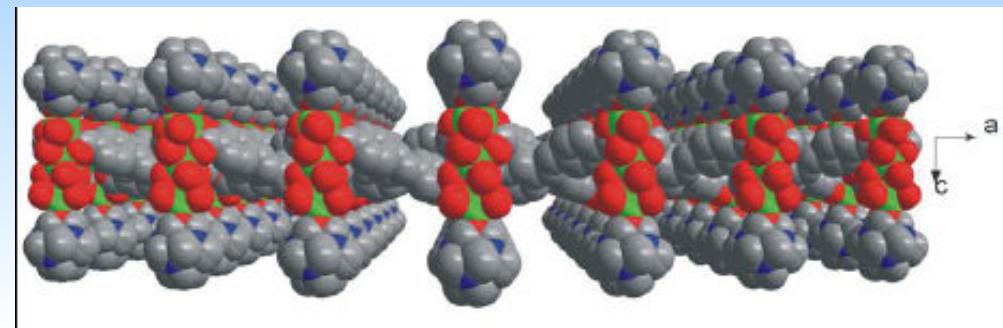
## "3D" cubic structure

- DEF, 105 °C/ 16 hours
- Interpenetrated, IRMOF-type structure
- Open porosity
- Surface area = 580 m<sup>2</sup>/g

Bauer et al., *J. Amer. Chem. Soc.*, in press, 2007.



Formula unit:  $\text{ZnL}_3(\text{DMF})_2$   
 $\text{L} = 4,4'$ -stilbenedicarboxylic acid



## "2D" hexagonal structure

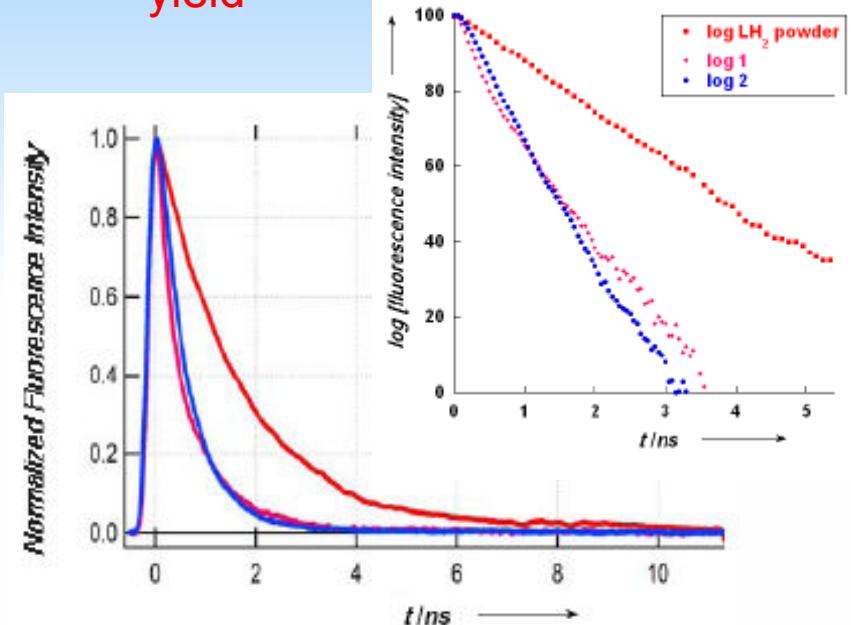
- DMF, 70 °C/16 hrs; 85 °C/ 4 hrs
- Dense material



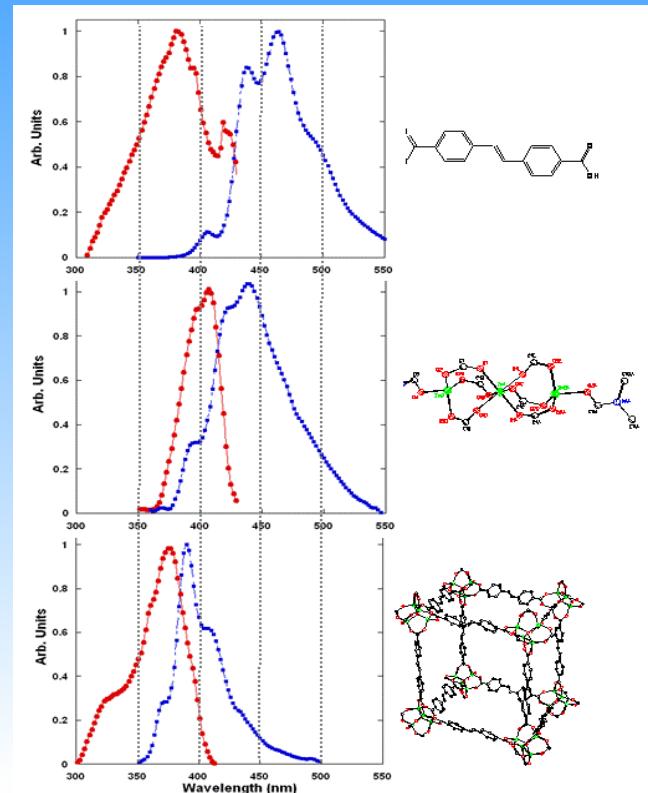
Sandia National Laboratories

# Constrained orientation of stilbene group in MOF increases fluorescence light yield

- **Increased fluorescence lifetimes**
  - Stilbene in solution:  $\tau_1 < 100$  ps
  - $\text{LH}_2$ :  $\tau_1 = 0.73$  ns,  $\tau_2 = 2.49$  ns
  - 2D MOF:  $\tau_1 = 0.2$  ns,  $\tau_2 = 0.95$  ns
  - 3D MOF:  $\tau = 0.50$  ns
  - ☞ Implies higher quantum yields
  - ☞ Potential for high scintillation light yield



Decreasing linker interactions



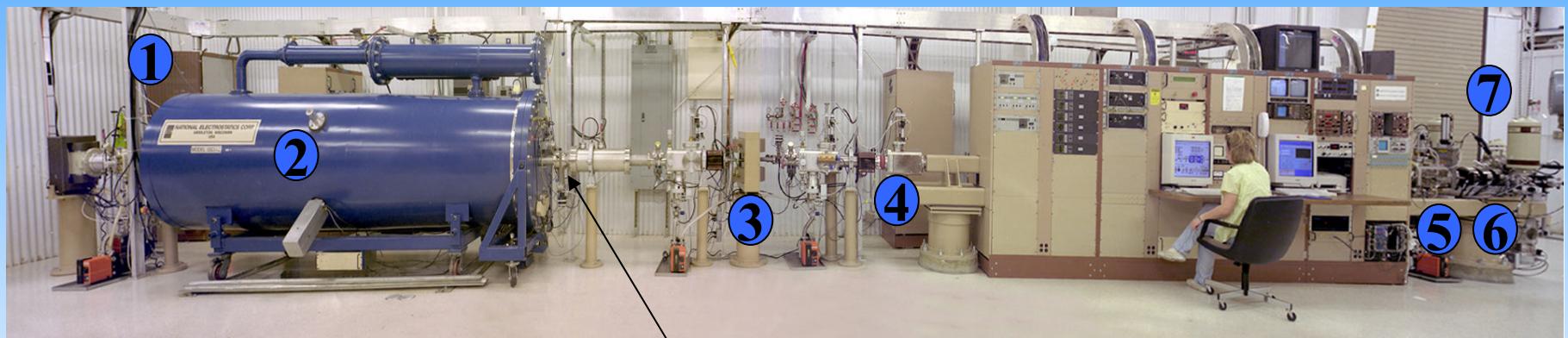
Single crystal fluorescence and excitation spectra



Sandia National Laboratories

# The interaction of MOF compounds with charged particles can be probed using a nuclear microprobe

*An accelerator produces charged particles that can be focused and scanned across the sample with high spatial resolution*



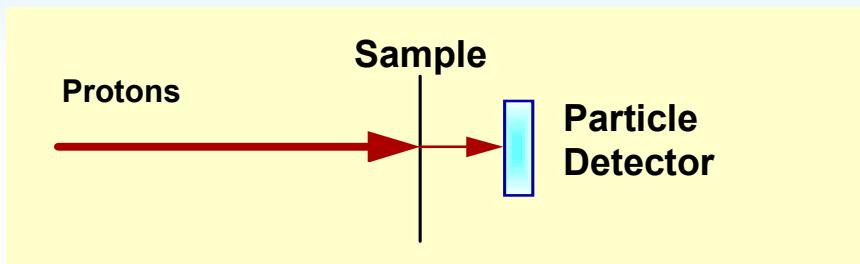
- 1 Ion Source ( $H^+$  or  $O^{3+}$ )**
- 2 Terminal 1.5 million volts**
- 3 Bending magnet**
- 4 Object slits (1 - 850  $\mu m$ )**
- 5 Image slits**
- 6 Magnetic Lens**
- 7 Sample Chamber**

Ion beam at this point:  
 $H^+$  accelerated to 3 MeV  
or  $O^{3+}$  accelerated to 6 MeV

# Microprobe diagnostic tools enable quantitative measurement of light output by MOFs

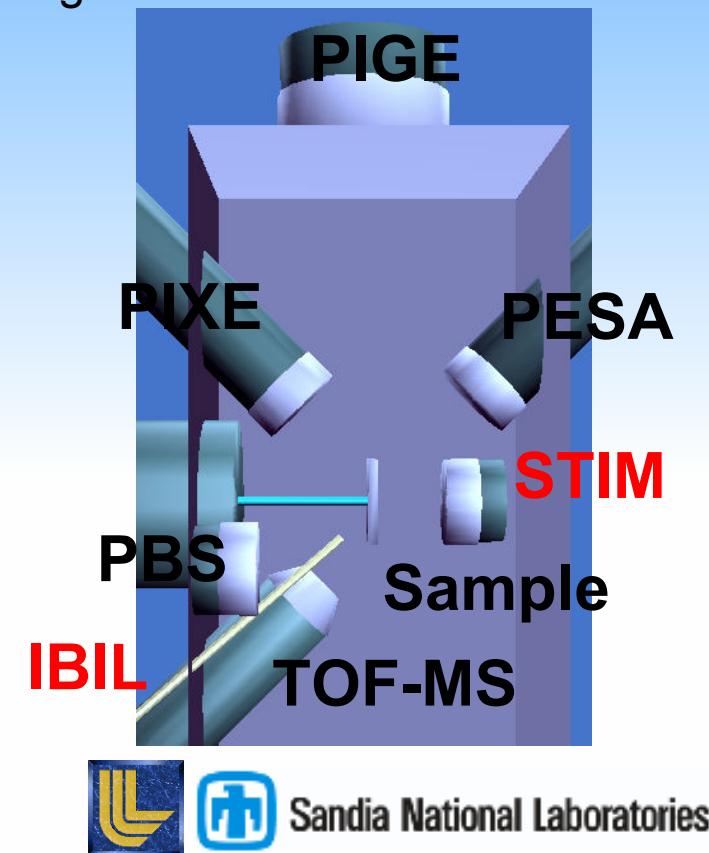
## Scanning Transmission Ion Microscopy (STIM)

- Energy loss through sample
- Dependent on mass
- Minimal dependence on composition
- Forms images of the sample
- Max. spatial resolution =  $0.5 \mu\text{m}$
- Use to measure crystal size, thickness



## Ion Beam Induced Luminescence (IBIL)

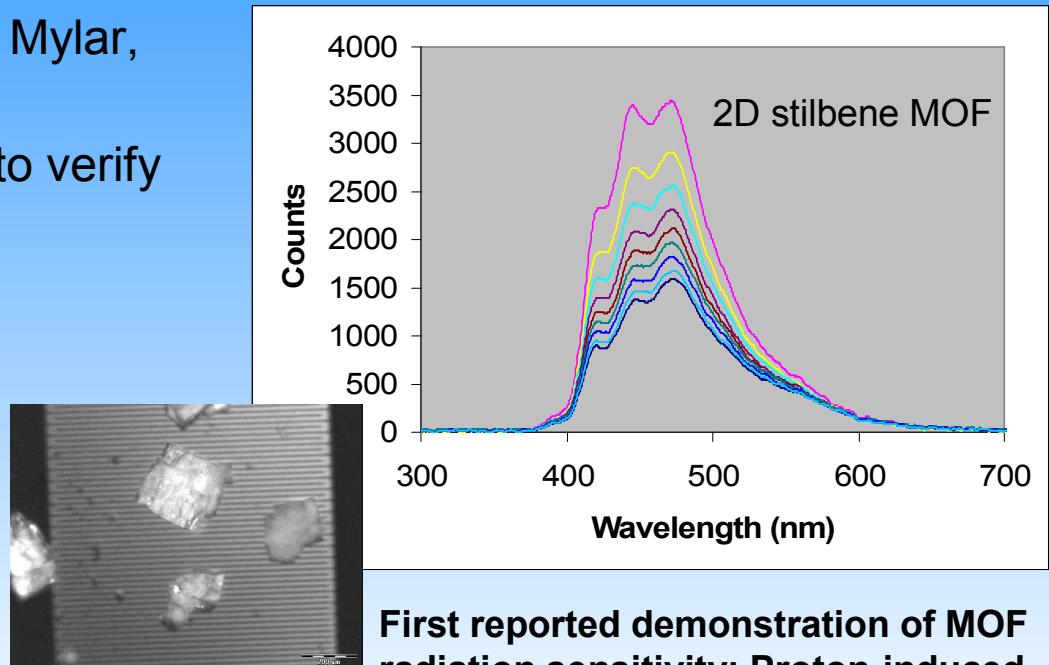
- Allows the differentiation of minerals or organics with similar elemental composition
- 1 nm wavelength resolution 180-1100 nm



Sandia National Laboratories

# IBIL spectra demonstrate that stilbene MOFs emit light when they interact with protons

- Samples mounted on gold foils, Mylar, or TEM grids
- SEM obtained before and after to verify integrity of crystals
- Beam energy: 3 MeV
- Beam dose rate:  $\sim$ 5 Mrad/s
- Data obtained for
  - Anthracene (standard)
  - 2d MOF
  - 3d MOF
- Note: IRMOF-1 exhibits no IBIL



**First reported demonstration of MOF radiation sensitivity: Proton-induced fluorescence.**

SAMPLE	Mass (mg)	Dose rate J/kg/s	Cts/J peak ch	% Anthracene
Anthracene	0.4	1.20E+04	1.13E+09	100%
2D MOF	0.38	9.33E+03	2.49E+08	22%
3D MOF 1	0.22	1.00E+04	9.62E+07	9%
3D MOF 2	0.49	9.33E+03	6.62E+07	6%
Stilbene				50%
BC422Q (commercial organic scintillator)				11%

Comparable to commercial scintillators



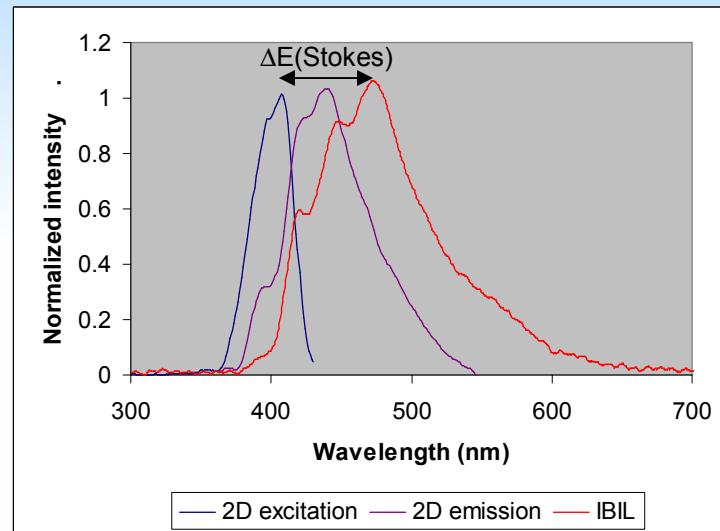
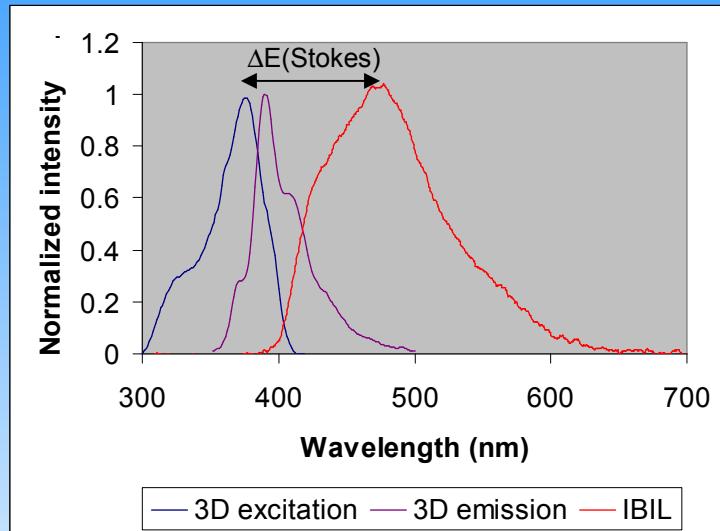
Sandia National Laboratories

# Self absorption of luminescence should be minimal in stilbene MOFs

- $\Delta E(\text{Stokes}) = E_{\max}(\text{excitation}) - \Delta E_{\max}(\text{emis})$
- Fluorescence Stokes shifts:
  - 2D MOF:  $1215 \text{ cm}^{-1}$
  - 3D MOF:  $815 \text{ cm}^{-1}$
  - Linker(H)<sub>2</sub>:  $4088 \text{ cm}^{-1}$
- IBIL Stokes shifts:
  - 2D MOF:  $3415 \text{ cm}^{-1}$
  - 3D MOF:  $5618 \text{ cm}^{-1}$

→ IBIL emission is well separated from absorption

→ Long detector path lengths should be feasible
- Broad emission on the long-wavelength side may be an indication of radiation damage along the ion track



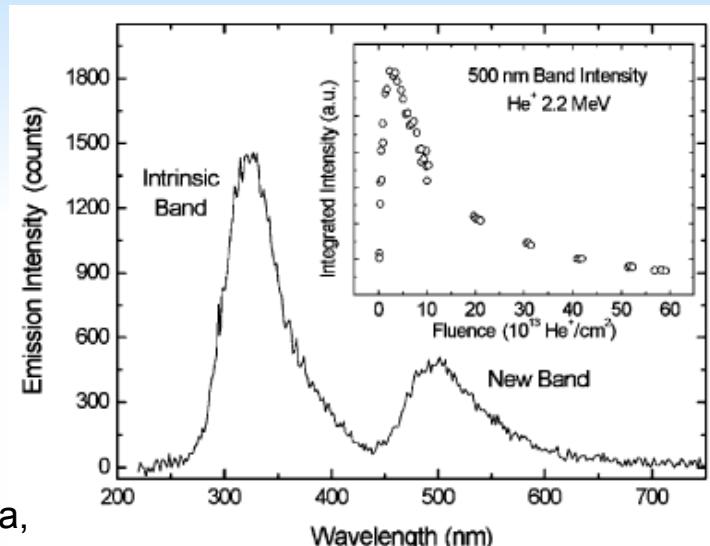
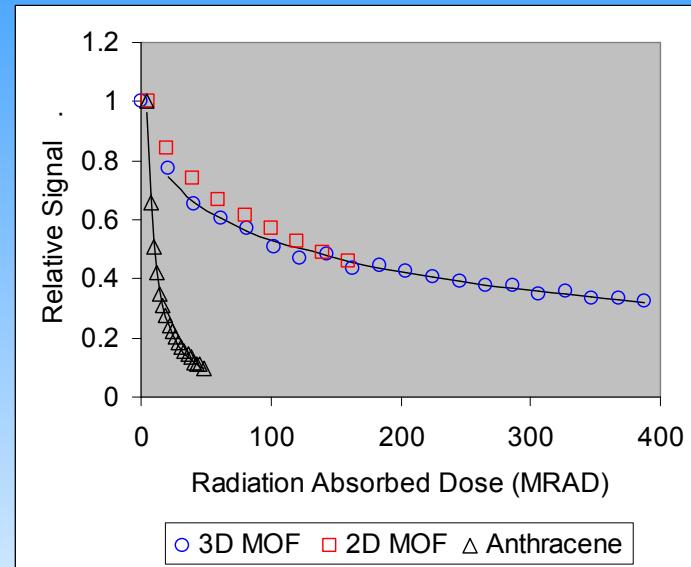
Sandia National Laboratories

# Stilbene MOFs are much more resistant to radiation damage than the anthracene standard

- Anthracene: **loses > 90% of initial luminescence after 50 Mrad dose**
- 2D and 3D stilbene MOFs lose only ~ 68% **after 400 Mrad dose**

1 rad = amount of energy absorbed/ mass  
= 100 ergs/g

- This dose is  $\gg$  expected lifetime dose
- Low density of 3D MOF should favor higher damage threshold:
  - Anthracene: 1.28 g/cm<sup>3</sup>
  - Stilbene: 0.97 g/cm<sup>3</sup>
  - 2D MOF: 1.52 g/cm<sup>3</sup>
  - 3D MOF: 0.50 g/cm<sup>3</sup>
- Clearly, factors beyond density must be accounted for
- Fluorescent polymers exposed to H<sup>+</sup> IBIL exhibit new emission
  - Attributed to radical formation leading to polymer crosslinking



IBIL spectrum of polyvinyltoluene, (A. Quaranta, *Nucl. Inst. Meth. Phys. Res. B*, 2005)



Stories

# Summary and conclusions

---

- Fluorescent metal organic frameworks constructed from stilbene dicarboxylates are luminescent when exposed to high-energy  $H^+$
- Very high radiation tolerance is exhibited
- Damage caused by the beam made lead to new emission bands
  - Time-dependent IBIL may resolve this
- Low density (0.5 g/mol) 3D stilbene MOF is expected to a more proportional response to particle energy
  - IBIL measurements as a function of beam energy in the 0.1 – 5 MeV energy range needed to verify this theory
- MOFs are a novel and potentially promising solution to the problem of detecting fission neutrons

---

## Acknowledgements

This work was funded by the Sandia National Laboratories Laboratory Directed Research and Development program