

Optical Spectroscopy and Scintillation Properties of a Cerium Activated $\text{Cs}_2\text{NaYBr}_6$ Crystal

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Introduction

The quest for high performance scintillators is crucial in supporting the development of uncooled, advanced gamma ray spectroscopy and radiation detection applications. New scintillators should have high light yield, short lifetime, good proportionality and stopping power. Scintillators with these characteristics can ultimately replace benchmark scintillators such as NaI(Tl) and CsI(Tl) while providing better energy resolution. In this study, we explore a cerium doped $\text{Cs}_2\text{NaYBr}_6$ scintillator for radiation detection applications. The result of its optical and radioluminescent properties are presented.

Crystal Growth and Characterization

The cerium doped $\text{Cs}_2\text{NaYBr}_6$ crystal was grown by a vertical Bridgman technique. Cerium was used as an effective activator to produce higher light yield and faster response. A 10 mm sample of a cerium doped $\text{Cs}_2\text{NaYBr}_6$ single crystal (12.7 mm dia.) was used for scintillation characterization (Fig. 1). Luminescence and scintillator properties, including photo-excitation and emission, photoluminescence quantum yield, gamma energy spectrum, and radioluminescence decay time, and linearity response, were investigated.

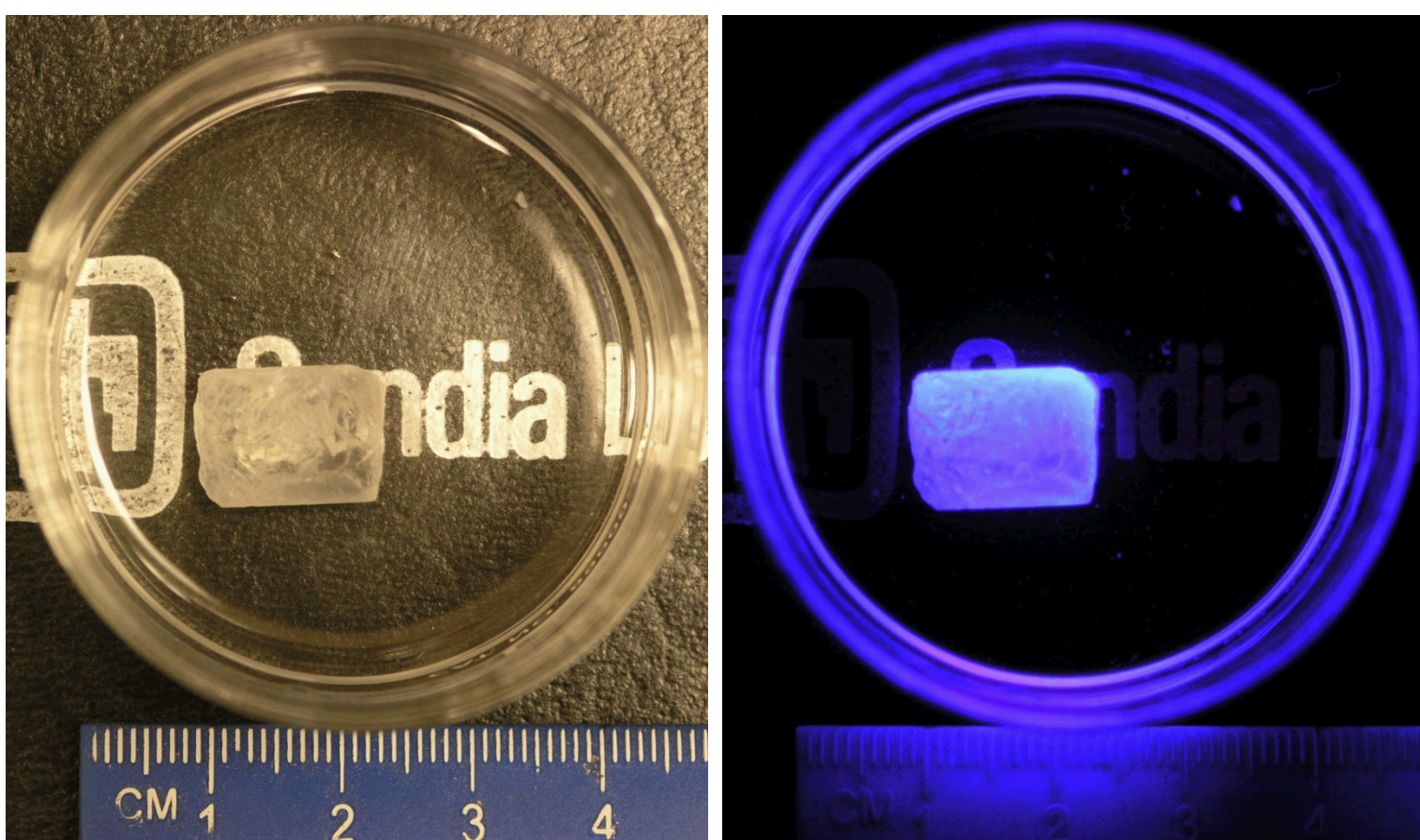


Fig. 1. The $\text{Cs}_2\text{NaYBr}_6$ crystal used in radioluminescent characterization. Photographs on left and right are taken under ambient and black light conditions.

Structure and Thermal Properties

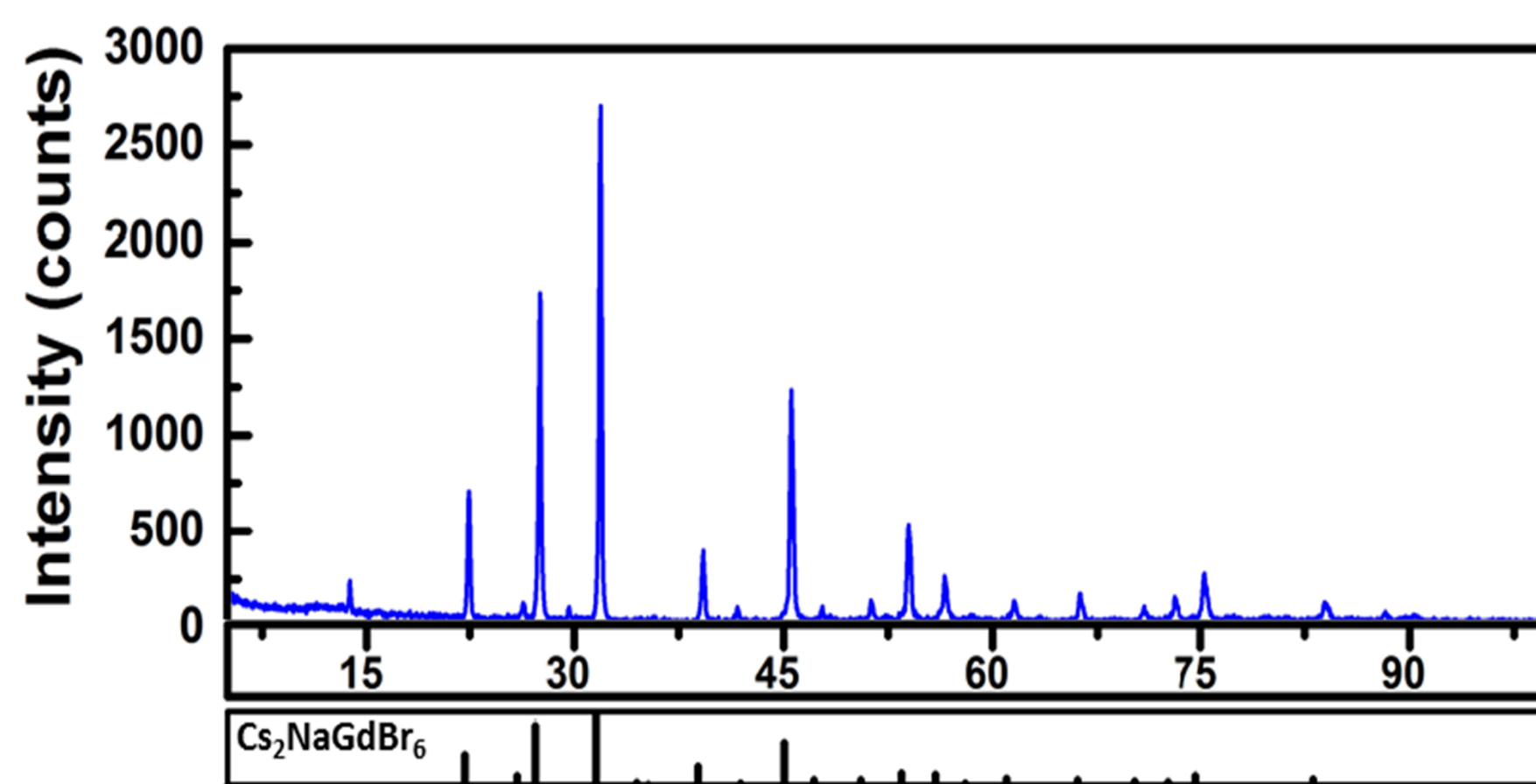


Fig. 2. X-ray powder diffraction pattern for the cerium doped $\text{Cs}_2\text{NaYBr}_6$.

Structural refinement based on X-ray powder diffraction data (Fig. 2) indicates that $\text{Cs}_2\text{NaYBr}_6:\text{Ce}^{3+}$ has a cubic elpasolite structure (space group Fm-3m (225)) with a lattice parameter of 11.3003(3) Å, which is consistent with our embedded ion model prediction. Differential scanning calorimetry (DSC) measurement showing the melting point of this new compound is at 702 °C, and the enthalpy change associated with the melting process is 59.65 J/g (Fig. 3).

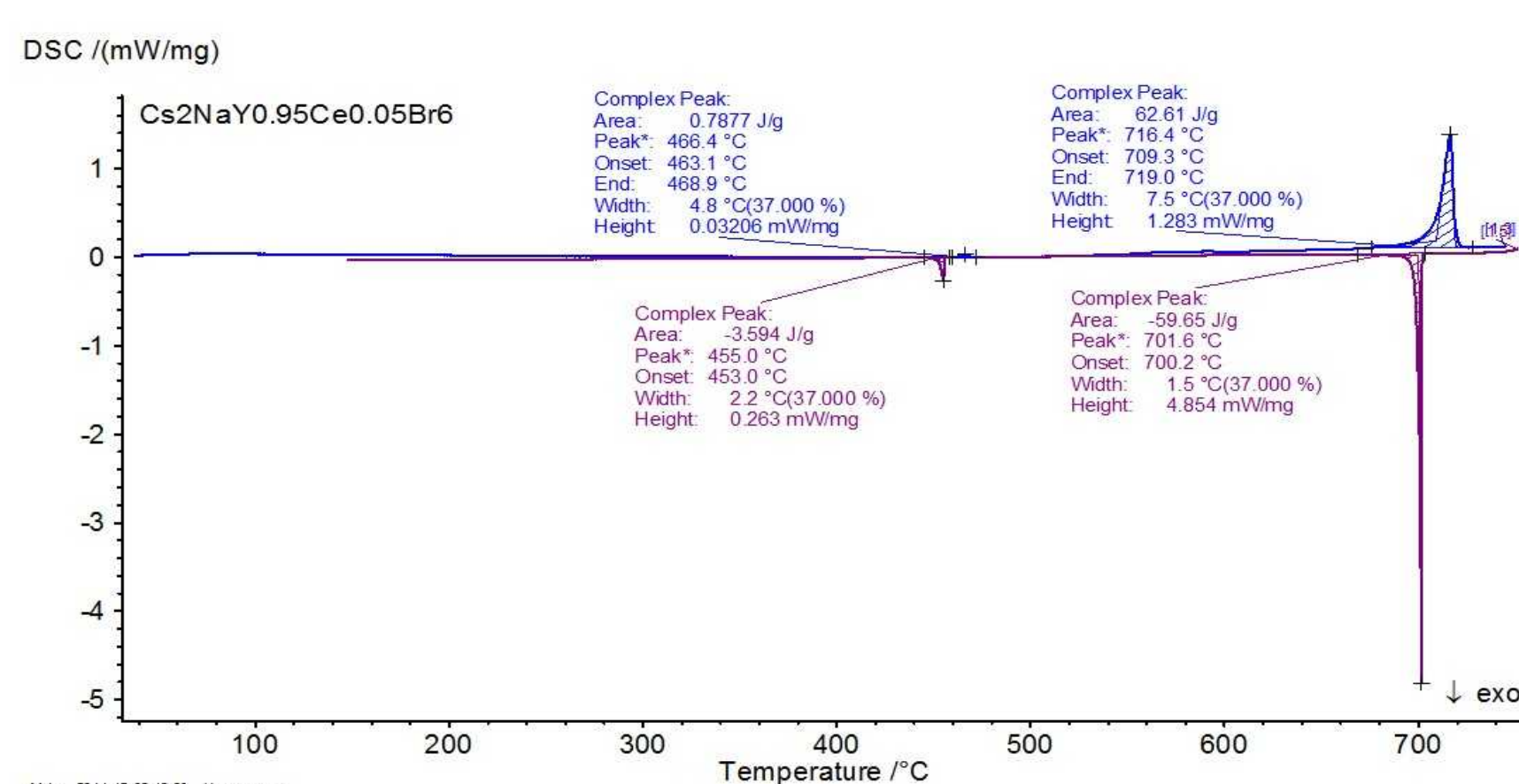


Fig. 3. The DSC measurement for the cerium doped $\text{Cs}_2\text{NaYBr}_6$. Data were collected under flowing argon with a heat and cooling rate of 5°C/min.

Optical Characterization

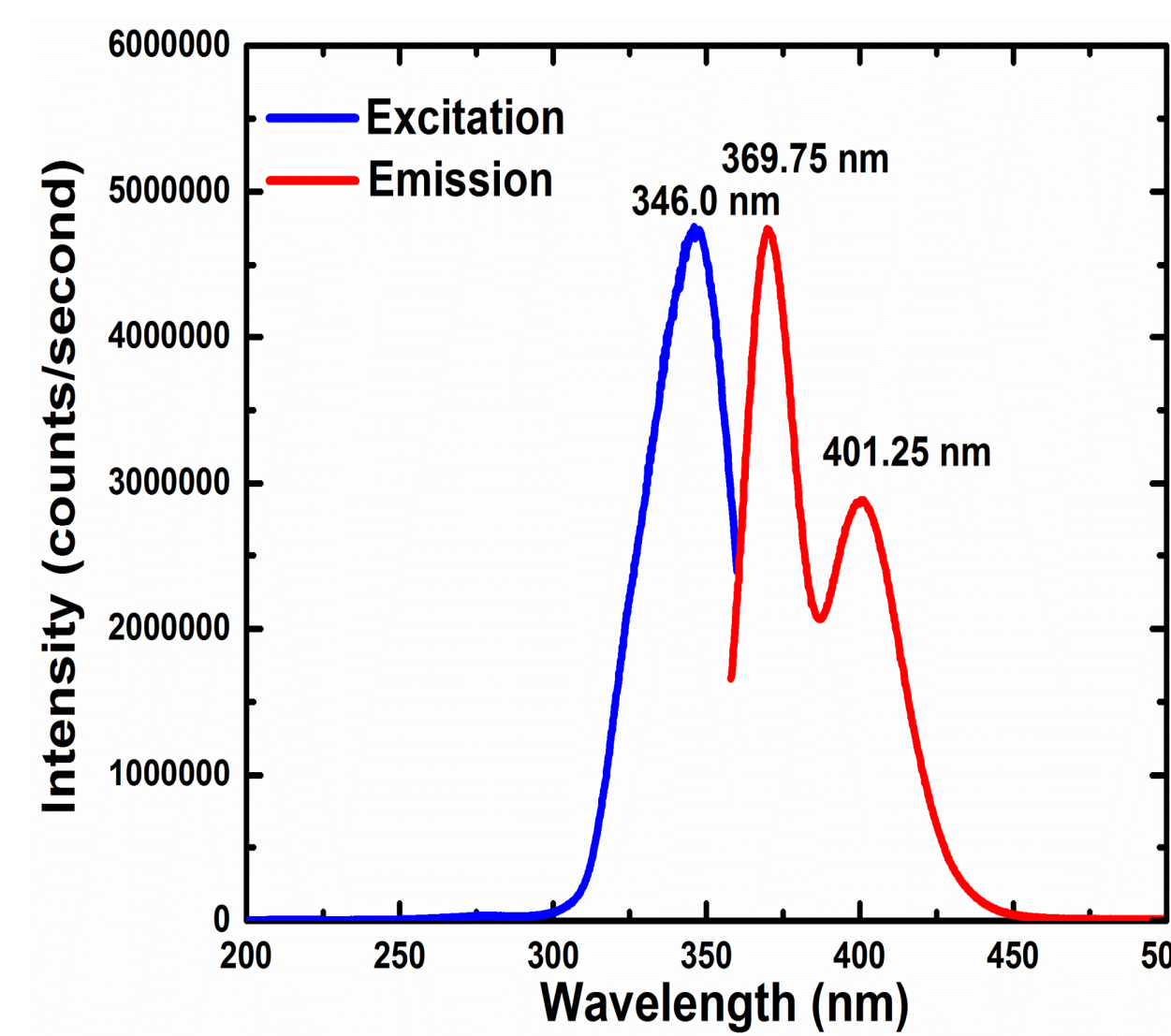


Fig. 4. Photoluminescence for the cerium doped $\text{Cs}_2\text{NaYBr}_6$.

Photoluminescence decay was measured by a time-correlated-single photon counting technique, using a pulsed 340 nm LED as the light source. The optical decay can be best fitted by two exponentially decay components (Fig. 5). The first and second lifetimes are 31.46 ns and 20.46 ns, respectively. While the first, 31.46 ns, component dominates the decay behavior (71.12%) the second component occupied almost three tenths of the decay population (28.88%).

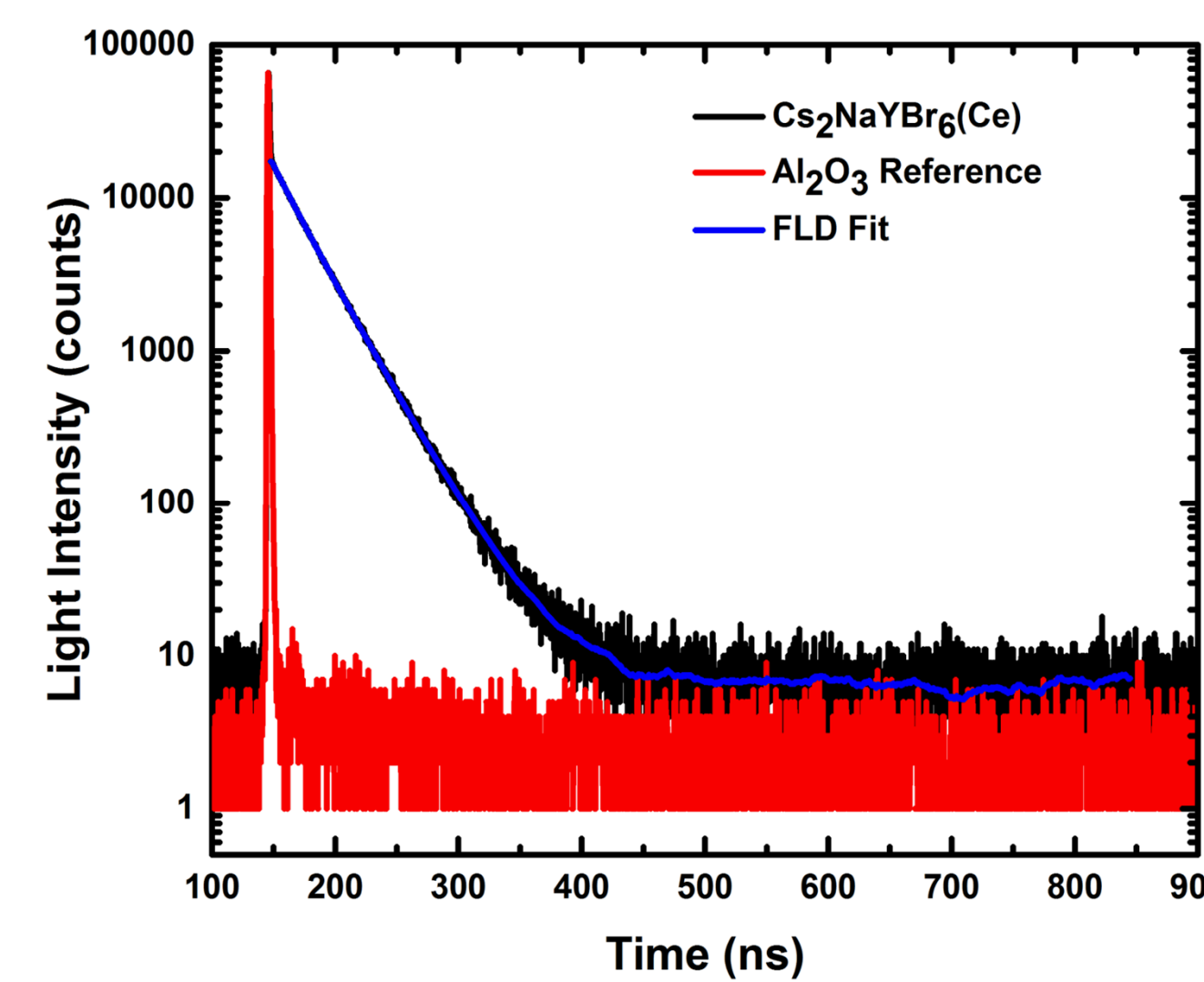


Fig. 5 Photoluminescence decay measurement for the cerium doped $\text{Cs}_2\text{NaYBr}_6$.

There is no evidence to suggest the presence of afterglow for this new compound as all decay components are within 50 ns. With a short photoluminescence lifetime, this new material is promising to have a fast response time when excited with gamma rays.

Scintillation Properties

Radioluminescence decay measured by a coincidence technique was fitted with two components (Fig. 6). The first component is 43.95 ns which makes up most of the decay events (99.98%), while the second component is much slower with a decay time of 537.7 ns which only made up 0.02% of the decay events. The cause of this long lasting afterglow is still not clear and will be studied in the future.

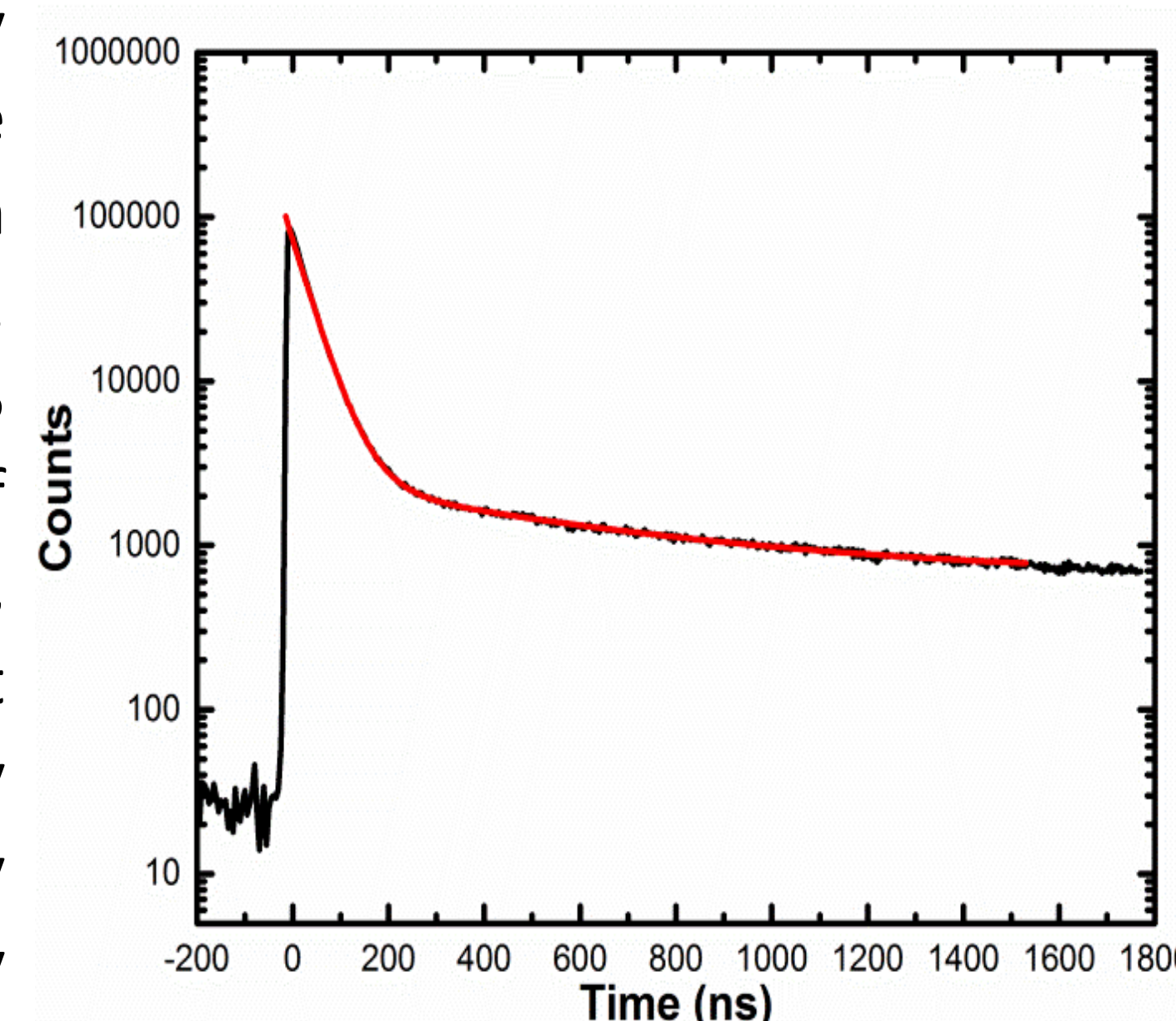


Fig. 6. Radioluminescence decay measurement for the cerium doped $\text{Cs}_2\text{NaYBr}_6$. Cs-137 is used for the coincidence measurement. ($\chi^2 = 0.9835$).

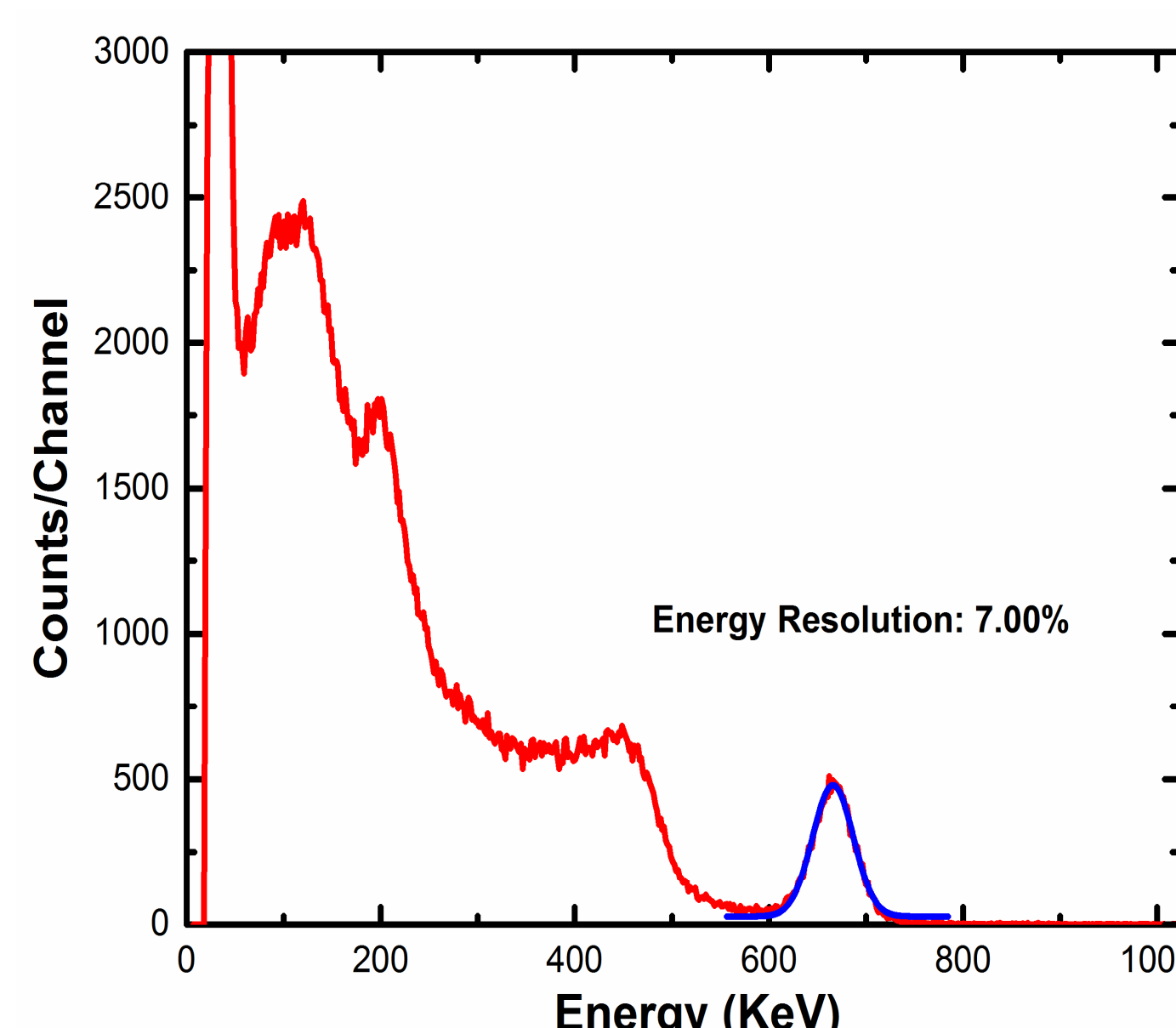


Fig. 7. Pulse Height Spectra for the cerium doped $\text{Cs}_2\text{NaYBr}_6$. Sample was irradiated by a Cs-137 source.

Excitation and emission spectra reveal a relatively small 23.75nm Stokes shift (Fig. 4). The small Stokes shift could potentially cause self-absorption in large crystals, which makes them more vulnerable to temperature change as strong self-absorption at high temperatures can diminish the scintillator performance. The double emission peaks on the right are characteristics of the 5d to 4f transition for Ce^{3+} activator. The cerium doped $\text{Cs}_2\text{NaYBr}_6$ crystal has a quantum yield of 41.7%.

Scintillation Properties Cont.

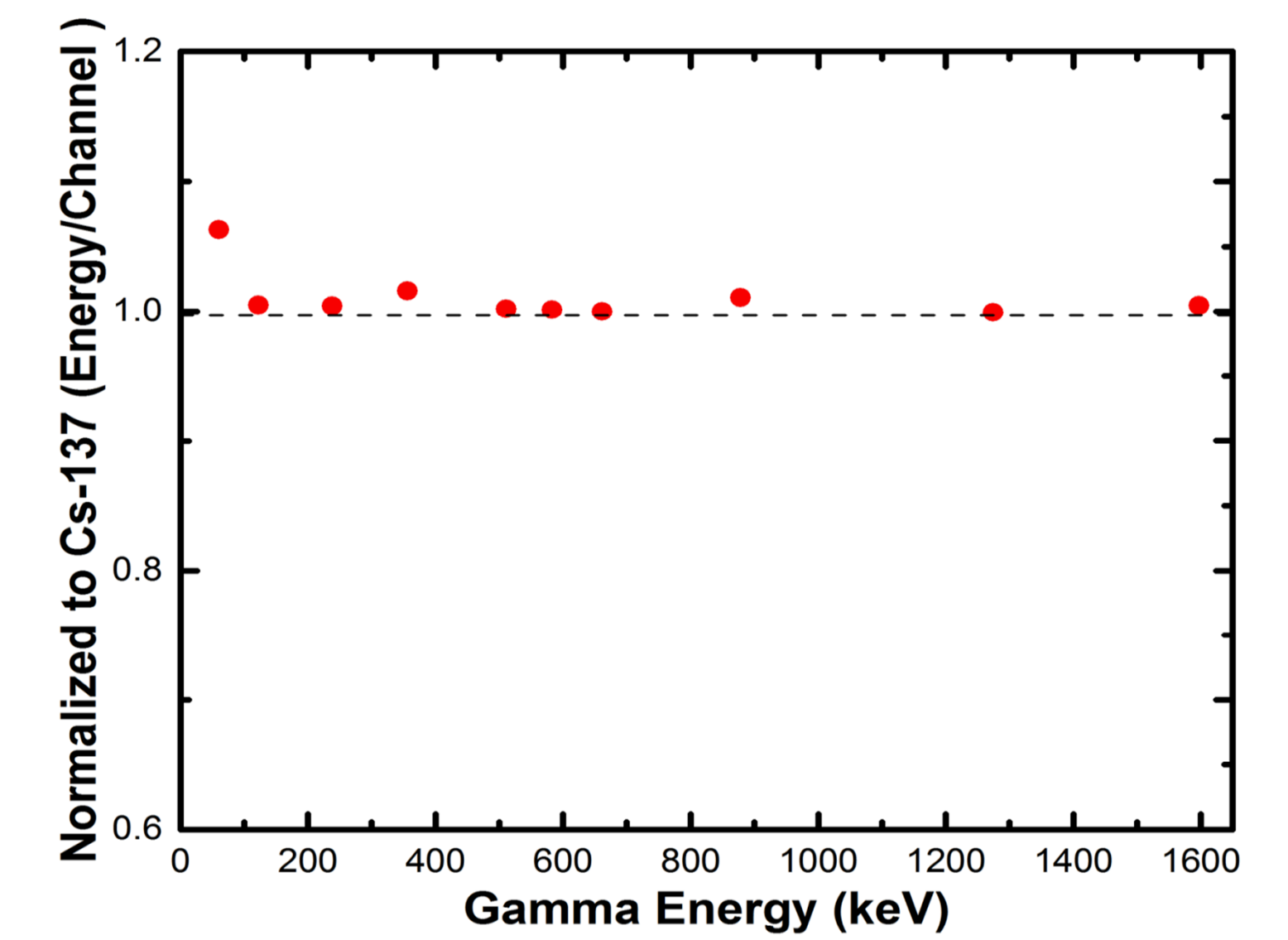


Fig. 8. Linearity measurement for the cerium doped $\text{Cs}_2\text{NaYBr}_6$. Data were collected from different isotopes and normalized to Cs-137.

This new compound exhibits good proportionality at high energy range (>500 keV) and starts to lose linearity at low energies (Fig. 8). A good proportional response is a general trait for the elpasolite halide crystals. A linearity of 6.0% is observed at the low energy range for this crystal, which is better than many well established scintillators such as NaI:Tl (14%), CsI:Tl (10%), and LSO (11%). The linearity response can be further improved with a higher quality crystal through a refined growth process and the arts of detector package techniques.

Conclusion

A cerium doped $\text{Cs}_2\text{NaYBr}_6$ crystals was grown using a vertical Bridgman technique for optical and radioluminescent characterization. Optical excitation and emission spectra indicate that cerium is an effective activator for this new compound. In addition, cerium produces a fast response time for optical and radioluminescence responses. The energy resolution at 662 keV for this crystal is ~7%, with a satisfactory linearity response in the 62 keV to 1597 keV energy range. These preliminary results suggest that with improved crystal quality this new compound is promising for gamma ray spectroscopy applications.

Acknowledgements

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