

Ion Discrimination by Nanoscale Design

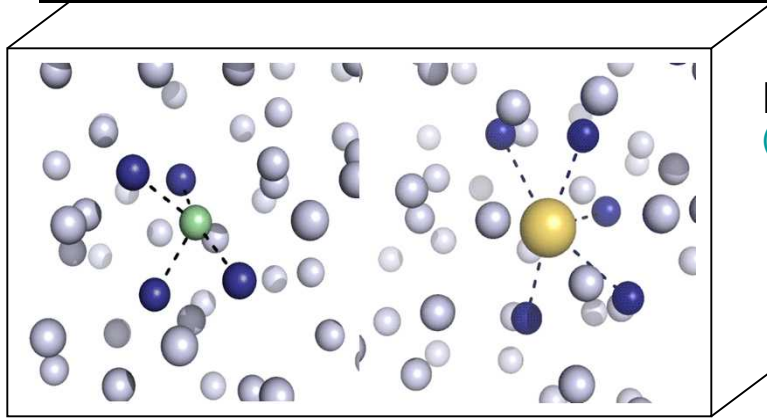


water tower: www.pbase.com/mescaleroman/images/57217408

Susan Rempe

Sandia National Labs, Albuquerque, NM

Natural protein channels: Not just simple holes!



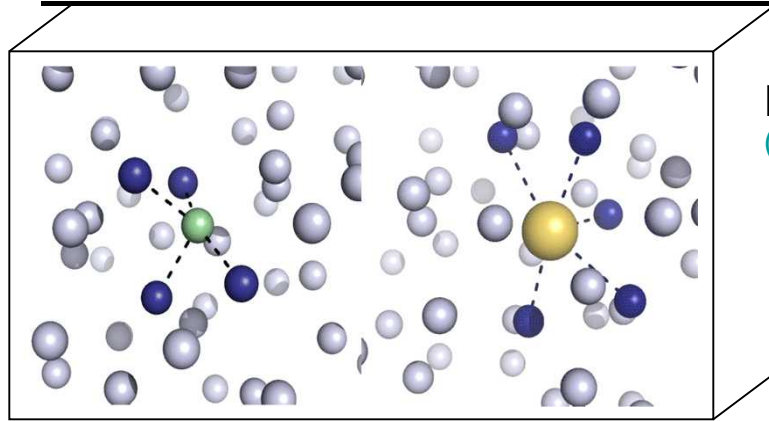
Ions in water
(Varma & Rempe, 2007)

- Ions are 'happy' **coordinated** with water **ligands** in liquid water

Bare ion		Ion-water bond
.95 Å	Na ⁺	2.4 Å
1.4 Å	H ₂ O	2.8 Å
1.3 Å	K ⁺	2.8 Å

- K⁺/Na⁺ exquisite discrimination:
 - same charge
 - same size at kT
 - larger ion transported, fast!

Natural protein channels: Not just simple holes!



Ions in water
(Varma & Rempe, 2007)

- Ions are 'happy' **coordinated** with water **ligands** in liquid water

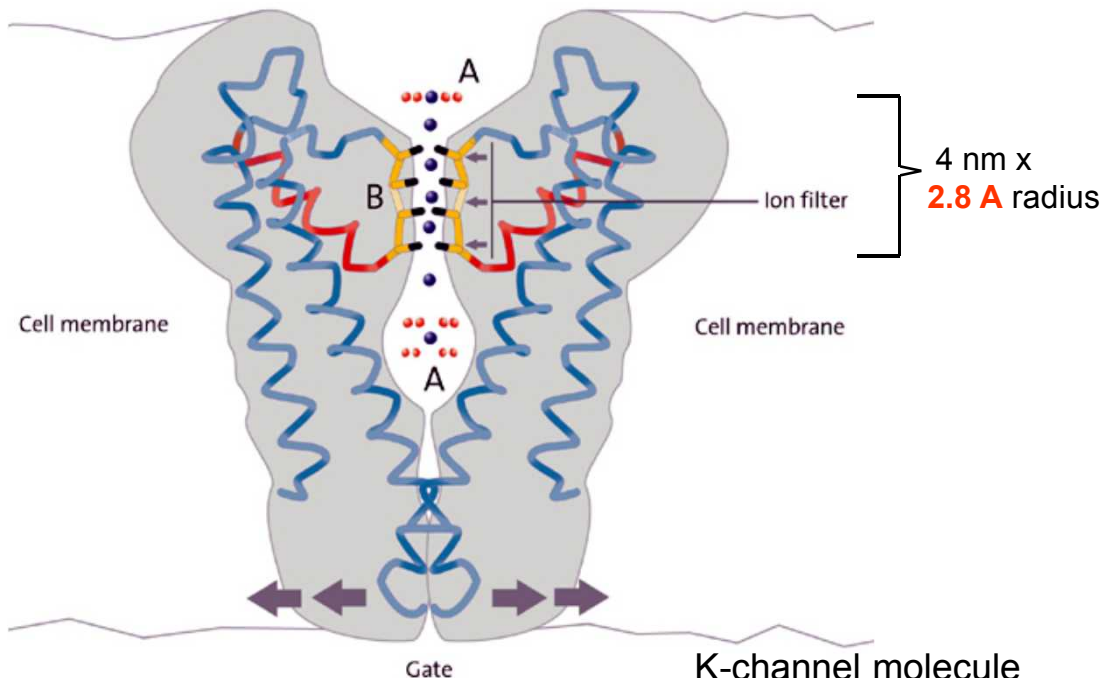
Unresolved Questions:

- How do channels work?

Smaller Na^+
ions rejected

Larger K^+ ions
permeate fast

Ion channel



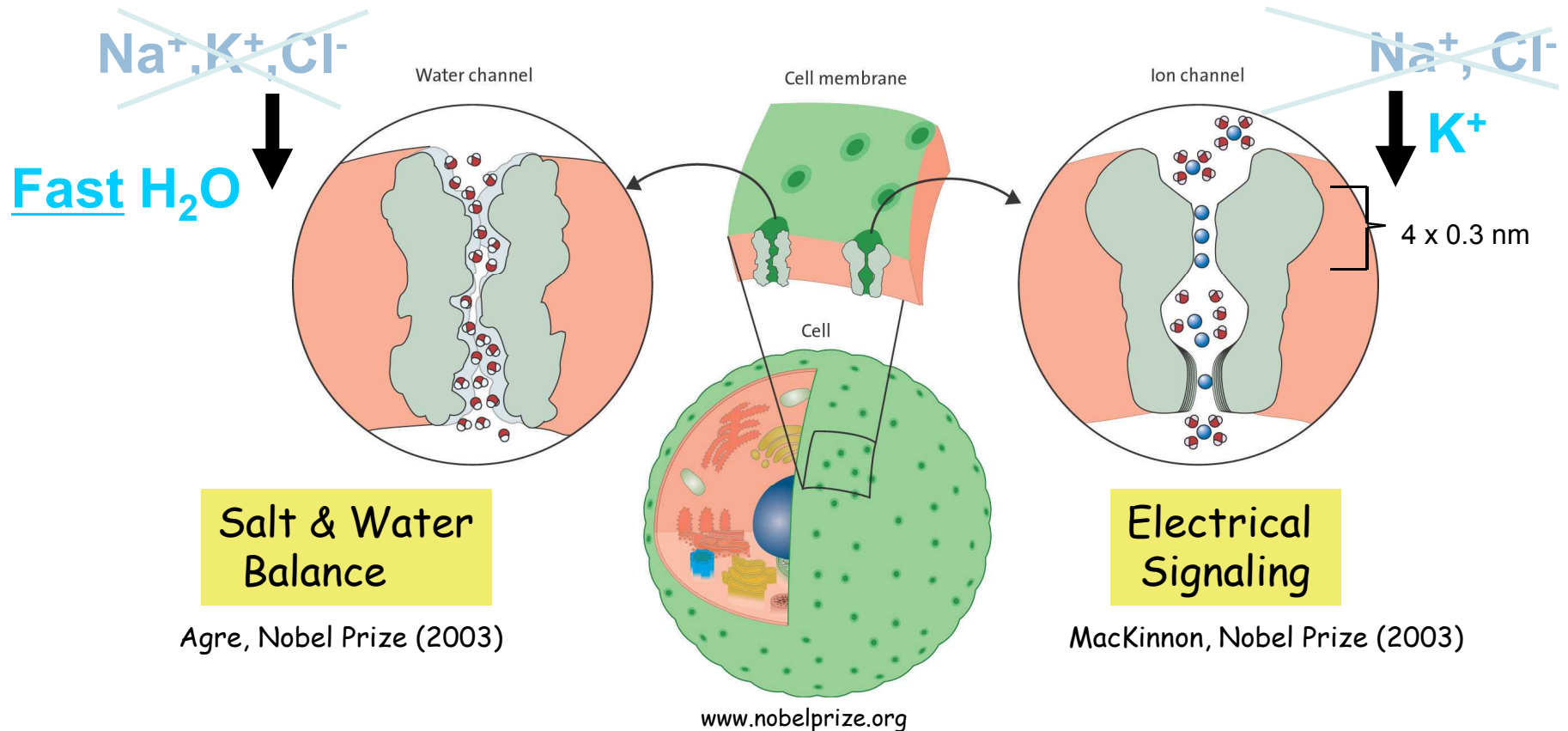
K-channel molecule
(www.nobelprize.org)

Bare ion		Ion-water bond
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1.3 Å	K^+	2.8 Å

- K^+/Na^+ exquisite discrimination:
 - same charge
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 - larger ion transported, fast!

Natural channel proteins:

Small dimensions, large biological impact



**Channels are fundamental to life:
severe Health consequences if disrupted.**

Nobel Prizes underscore Health significance of Channels....



Time and Nobel Prize Field	Laureates	Specific Research
<u>1963 Nobel Prize in Medicine</u>	<u>Alan L. Hodgkin, Andrew F. Huxley</u>	<u>Ion mechanisms of nerve impulse</u>
1985 Nobel Prize in Medicine	Michael S. Brown, Joseph L. Goldstein	Regulation of cholesterol metabolism
1988 Nobel Prize in Chemistry	Johann Deisenhofer, Robert Huber, Hartmut Michel	Structure of photosynthetic reaction center
<u>1991 Nobel Prize in Medicine</u>	<u>Erwin Neher, Bert Sakmann</u>	<u>Function of single ion channels</u>
1997 Nobel Prize in Chemistry	Jens C. Skou	Membrane-bound turnover of ATP
1999 Nobel Prize in Medicine	Günter Blobel	Principles of protein compartmentalization
2003 Nobel Prize in Chemistry	Roderick MacKinnon, Peter Agre	Molecular structural analysis of membrane channels, existence of water channels
2004 Nobel Prize in Medicine	Richard Axel, Linda B. Buck	Odorant receptors

Global Problem: Desalination

Clean water: **a global precious commodity**

- water is recyclable
- but RO is expensive, produces sterile water

Salinity Levels:

Seawater: ~35 g/l (0.6 M)

Brackish: ~1-5 g/l (0.08 M)

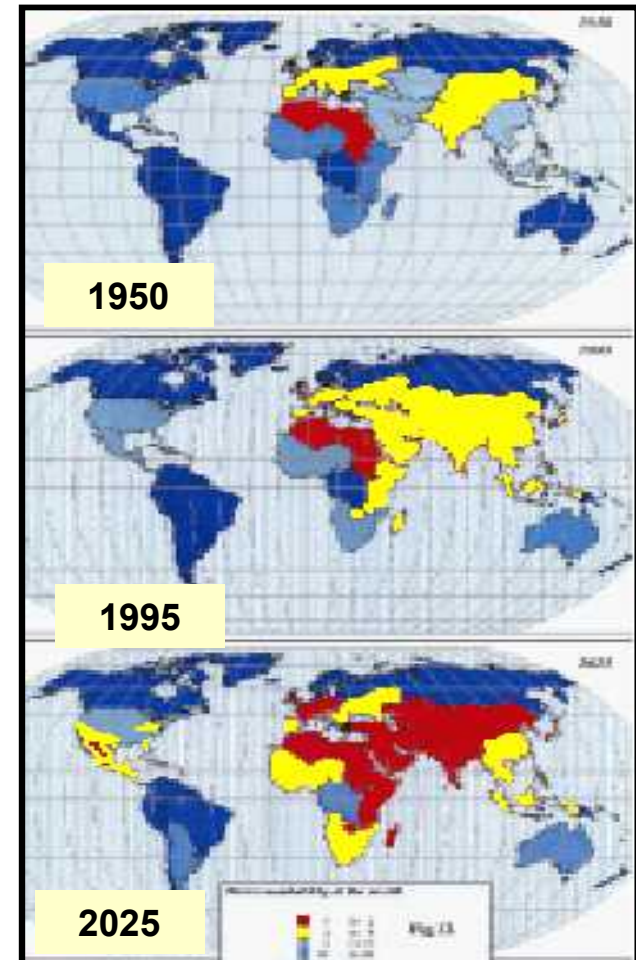
Potable: <0.5 g/l (0.008 M)

Efficient membranes: critical challenge

- fast (barrierless) water transport
- select ion exclusion (mineral water)

“Water promises to be to the 21st century what oil was to the 20th century: the precious commodity that determines the wealth of nations.”

Fortune Magazine, May 15, 2000

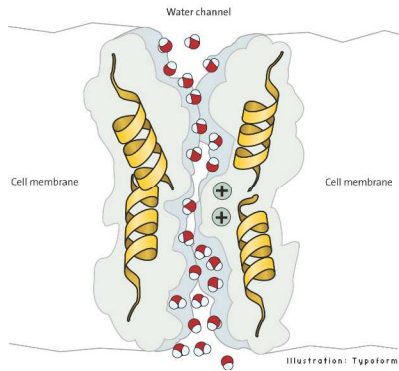


shortage

Efficient Membranes:

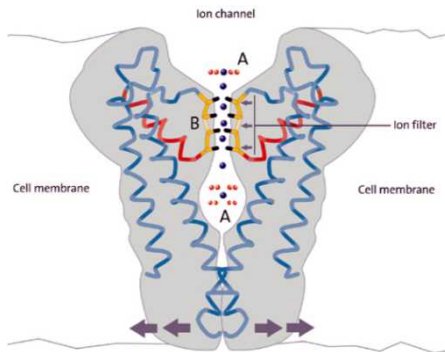
Understand, design, engineer nano-channels for desalination

Water channels



- transport H_2O fast

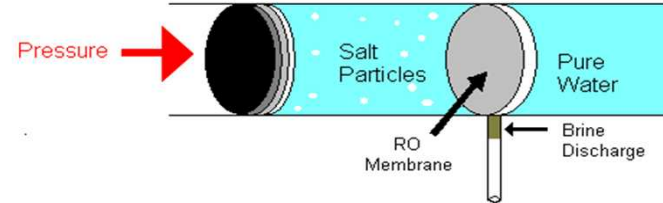
Ion channels



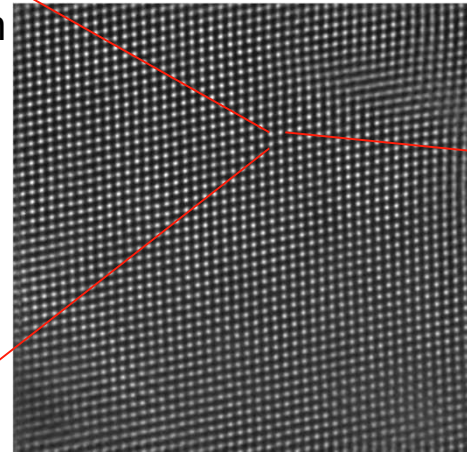
- transport select minerals

- Bio-inspired design

- Theory & modeling



- Inorganic membrane synthesis



Brinker Lab

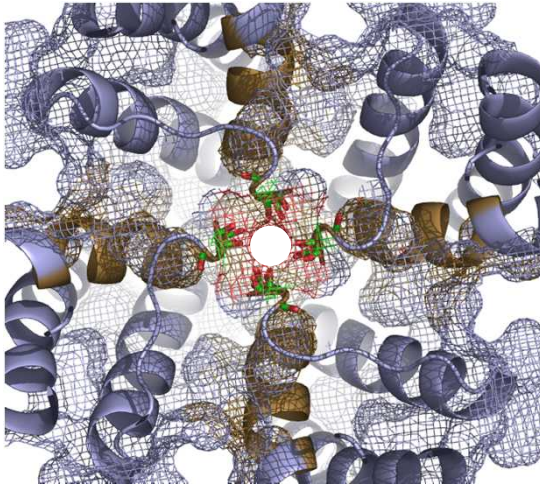


Solution: Harness molecular biomechanisms.
Gain 10x in water flux + minerals.

What parameters do we give our engineers?

Natural Channels

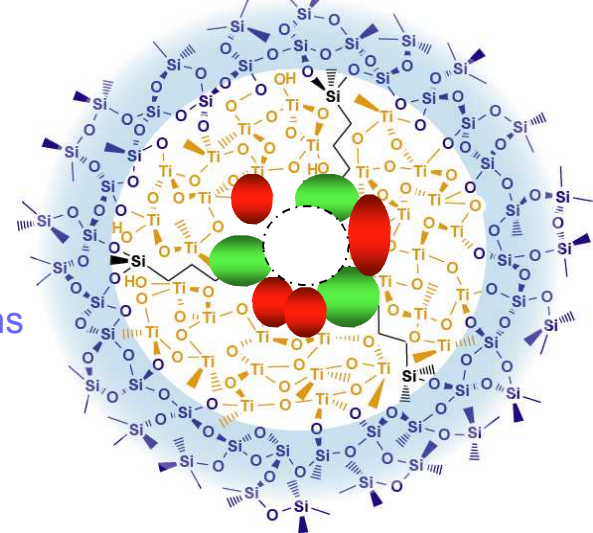
4 x 0.3 nm
Active site



Bio mechanisms

Engineering solutions

Inorganic Channels



Critical Channel Design Issues:

What's significant about (1) Mouth?

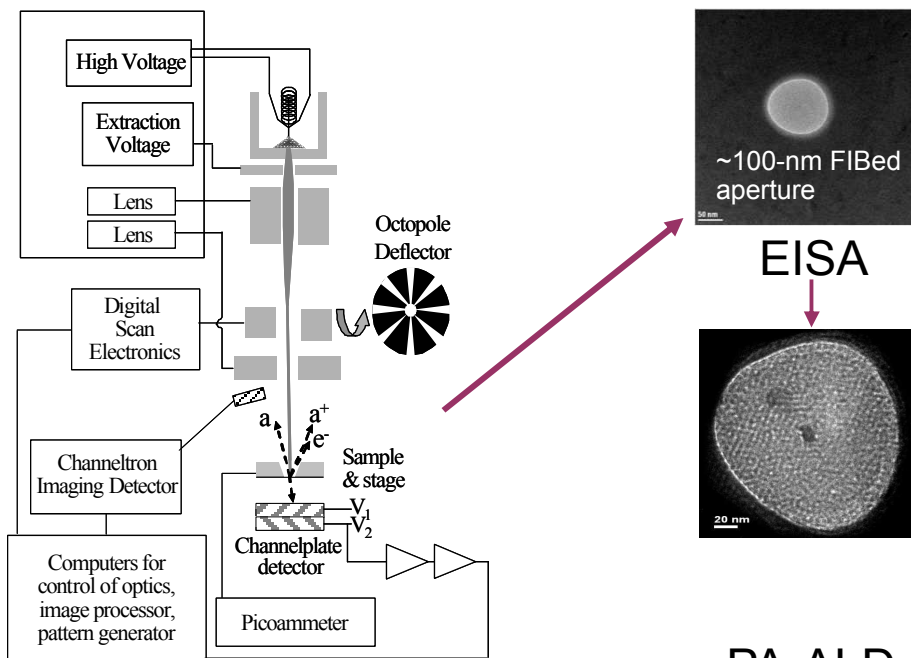
(2) Narrow filter?

(3) Chemistry, architecture?

Subtle, challenging questions demand molecular precision:

—————→ **Molecular Modeling** + Molecular Synthesis

Platforms for Experiments & Modeling



FIB experimental set-up

Special holder device

- TEM observation
- Patch Clamp
- PA-ALD
- Iterate

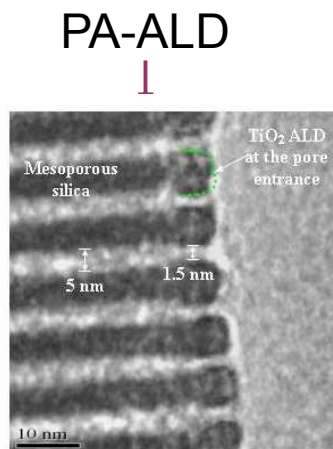
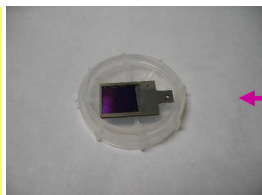
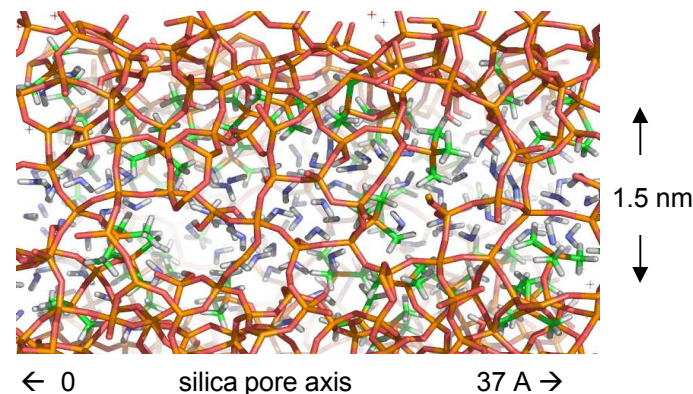


Fig. 1a

- Multiscale modeling **essential** to understand combined effects of pore size, structure, chemistry and charge.



- **Classical** (large, long times)
- **Quantum** (accuracy)
 - **Thermodynamics** (work)
 - **Dynamics** (transport)

- Experimental platform allows successive modification (pore size, and surface chemistry), imaging and transport measurements on identical sample.

Modeling Approach: Molecular + quantum accuracy

- Quantum (ab initio) interactions:

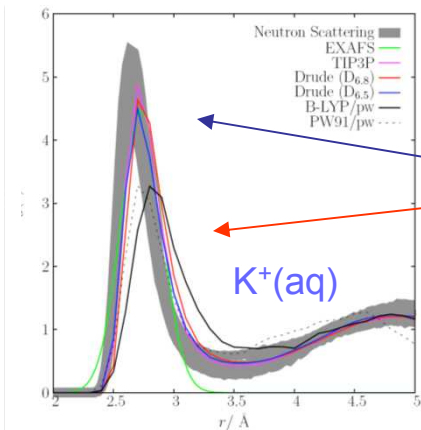
- expensive; describe complex interactions
(ex. ion-ligand chemical bonds)

- Classical molecular interactions:

- inexpensive; simplified, parameterized

- Example 1

More ligands bind classical ions??

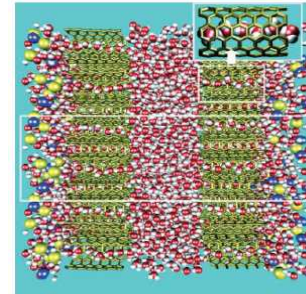


classical
quantum

Whitfield, Rempe, et al. *JCTC* (2007)

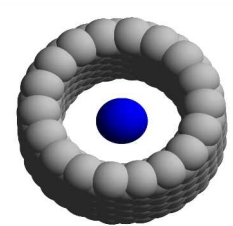
- Example 2

Classical nanotubes
Exclude ions??



Hummer & co *PNAS* (2003)

Quantum tubes
Admit ions!!

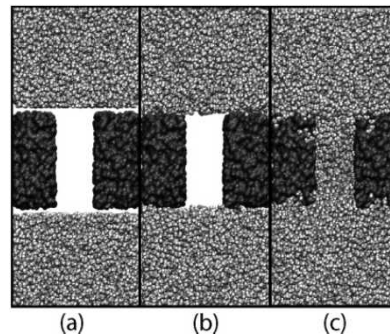


$Na^+ : -210 \text{ kJ/mol}$

Leung, Marsman *JCP* (2007))

- Example 3

<10% change in classical parameters:
Water fills/empties?



Leung, Rempe, Lorenz *PRL* (2006)
channel construction

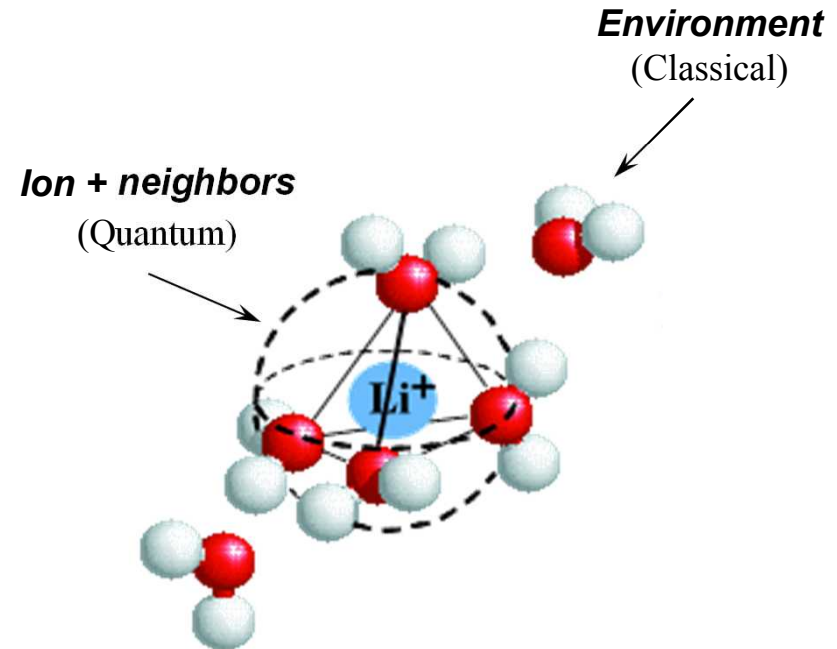
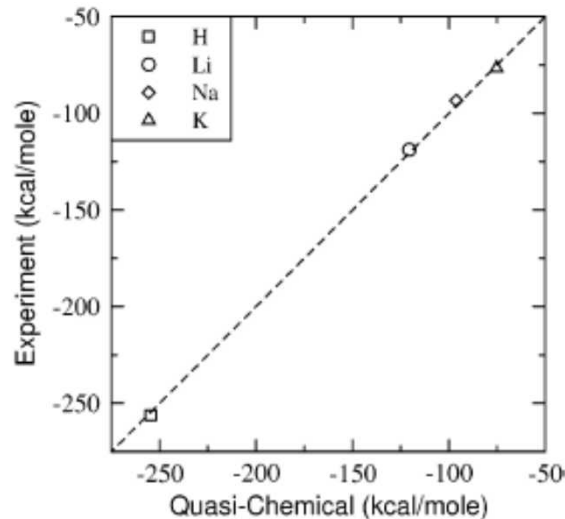
Cruz-Chu et al *JPCB* (2006)

Predictions:

Ion transfer thermodynamics using liquid state theory

- Calculate work to transfer ions (water \rightarrow channels), efficiently & accurately
 - 'quasi-chemical' theory¹

- Theory well-tested for ion hydration:²

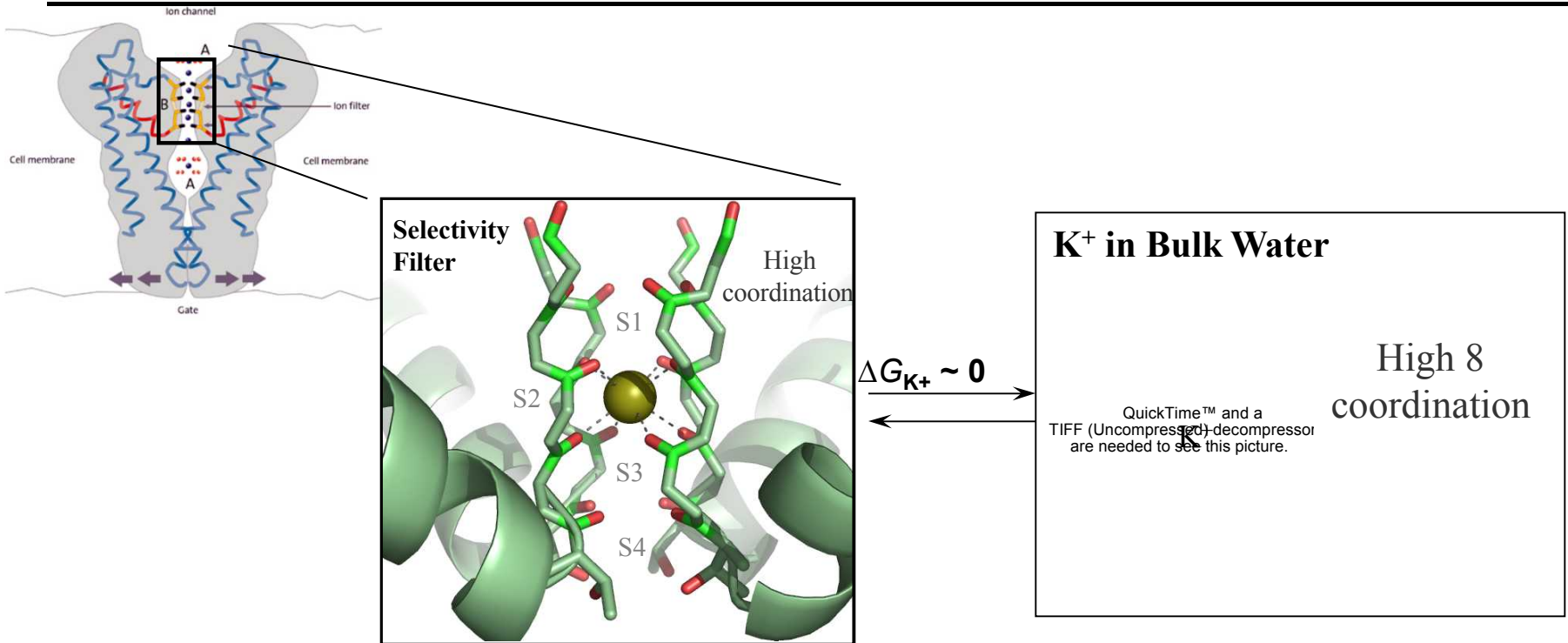


- Apply to Bio/Inorganic channels for Nanoscale Design Parameters^{3,4,5}

- ¹Pratt & Rempe (1999); Sabo, Rempe, *et al JPCB* (2008)
- ²JACS (2004), PCCP (2004), FPE(2001), JACS (2000)
- ³Varma & Rempe *Biophys J* (2007)
- ⁴Varma, Sabo, Rempe *J Molec Bio* (2008)
- ⁵Leung, Rempe, Lorenz *PRL* (2006)

K⁺/Na⁺ Ion Discrimination Problem:

How do K-selective channels work? Prevailing Views

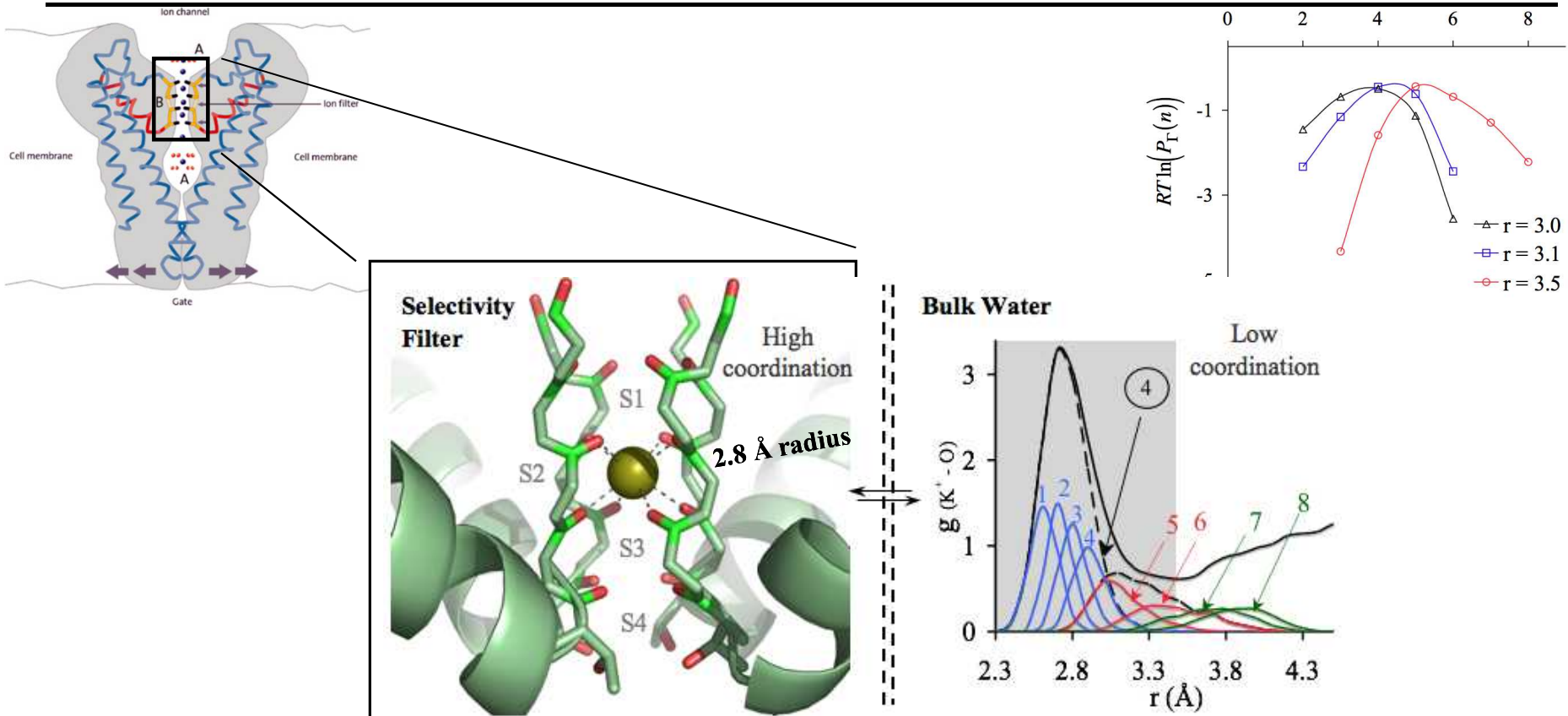


1. Mimic K⁺ ion hydration structures for fast K⁺ transfer^{1,2,3}
2. Specific cavity size fits K⁺^{1,2} vs Liquid-like flexibility³

¹Bezanilla & Armstrong (1972), ²Zhou *et al.* (2001), ³Noskov *et al.* (2004)

K⁺/Na⁺ Ion Discrimination Problem:

How do K-selective channels work? New View #1



1. **Not Mimic** of K⁺ ion hydration structures for fast K⁺ transfer^{1,2}
 ==> Work avoided by special channel environment (no H-bonders)³

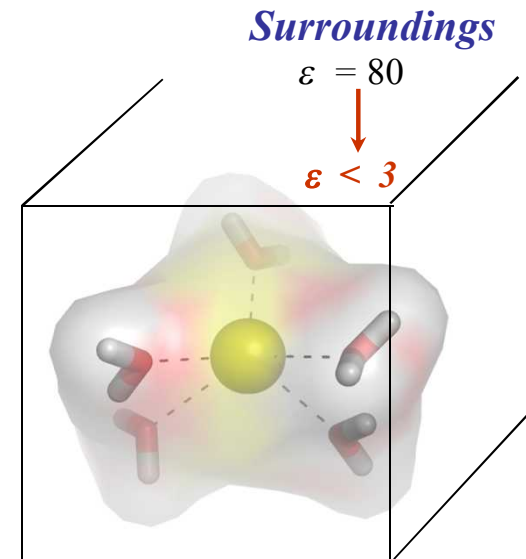
Special Channel Environment:

No local competitors for O-ligands stabilizes high ion coordinations

- Coordination preferences controlled by ligand desolvation penalty



In this “quasi-liquid” environment,
preferred ion coordination is higher



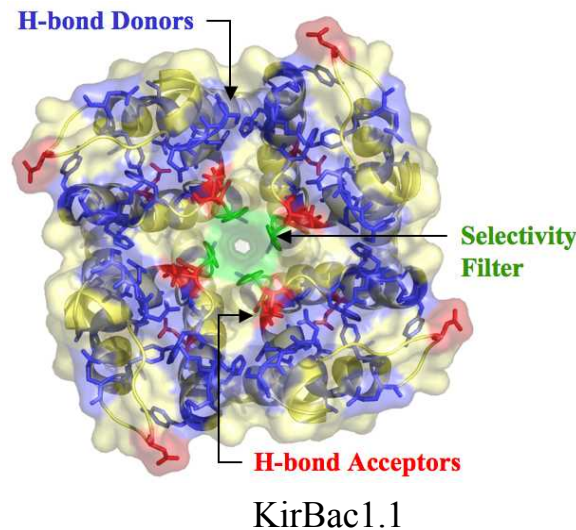
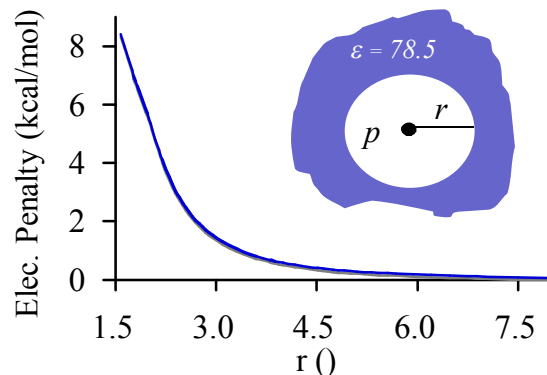
Special Channel Environment:

No local competitors for O-ligands stabilizes high ion coordinations

- Coordination preferences controlled by ligand desolvation penalty



- Ligand desolvation penalty eliminated locally (6 Å)

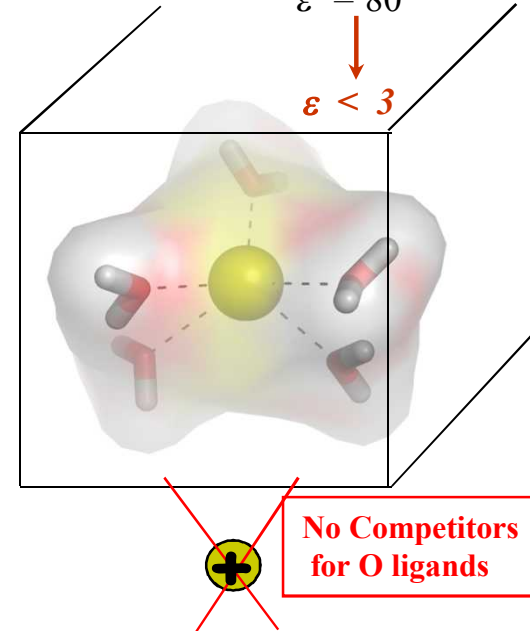


(His, Arg, Lys, Gln, Asn, Trp, Tyr, Cys > 6.8 Å away from binding site ligands)

Surroundings

$\epsilon = 80$

$\epsilon < 3$



Highly selective K-channels lack H-bond donors locally

- explains how 8 ligands bind K^+ with no work

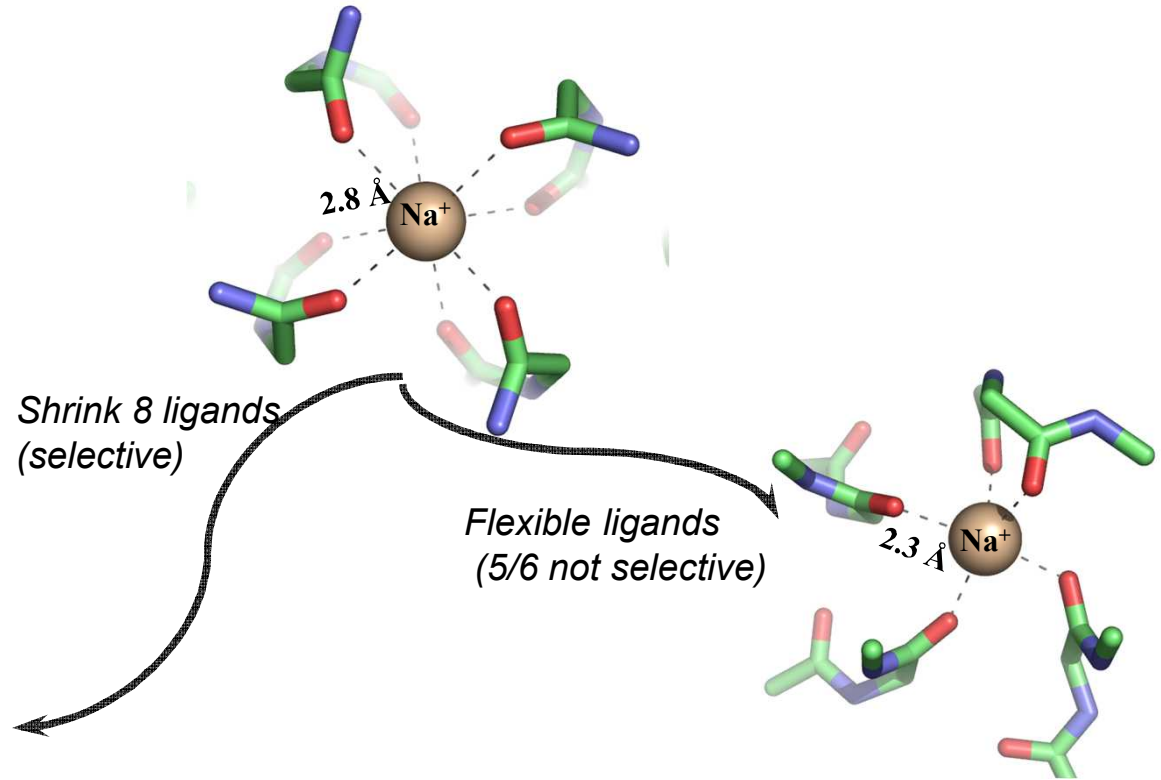
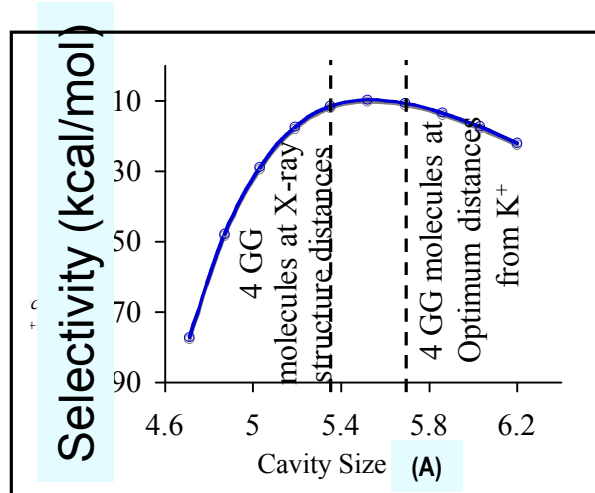
Special Channel Environment:

Structural transitions in ion coordination driven by
changes in competition for ligand binding

QuickTime™ and a
TIFF (Uncompressed) decompressor
are needed to see this picture.

K^+/Na^+ Ion Discrimination Problem:

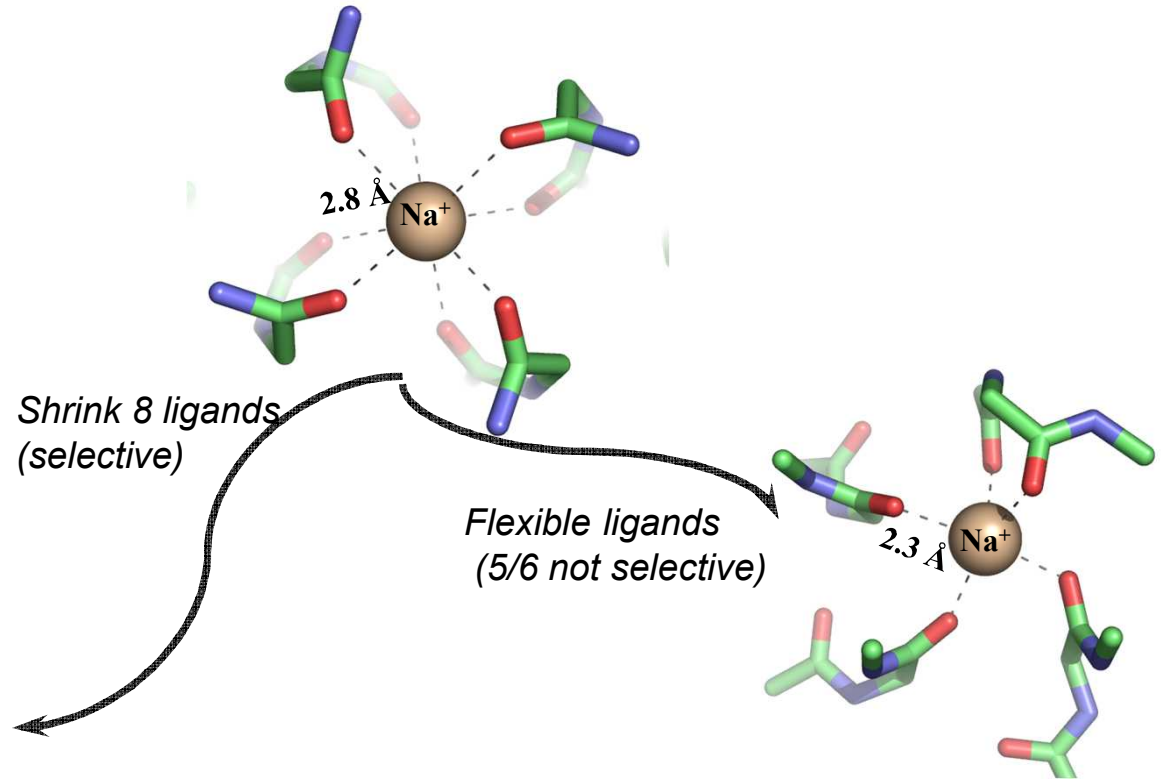
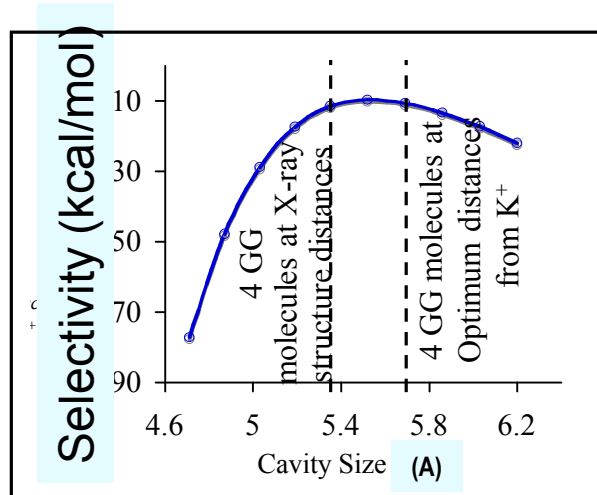
How do K-selective channels work? New View #2



1. Cavity size important
2. Flexibility important

K⁺/Na⁺ Ion Discrimination Problem:

How do K-selective channels work? New View #2



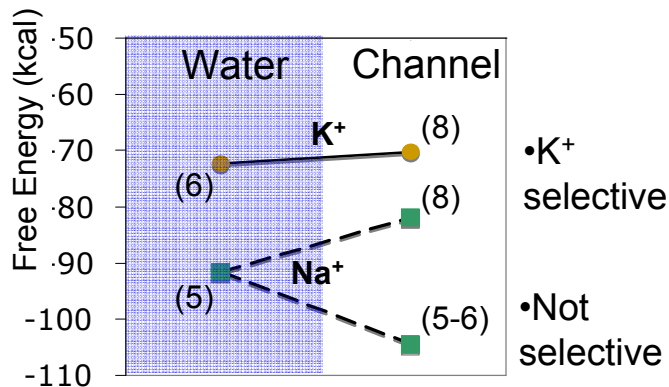
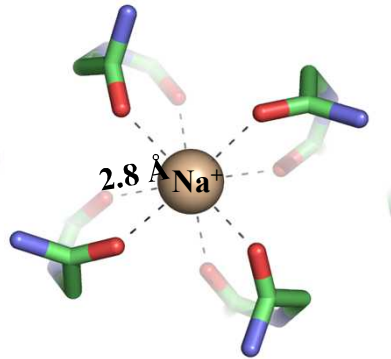
1. Cavity size important, **but specific size not necessary**
 2. Flexibility important, **but can reduce ligands & eliminate selectivity**
- ==> Rigidity¹⁻⁵ important to maintain Over-Coordinated¹⁻⁴ ions (>6 ligands)**

¹Varma & Rempe *Biophys J* (2007); ²Varma, Sabo, Rempe *J Molec Bio* (2008);

³Bostick & Brooks *PNAS* (2007); ⁴Thomas *et al. Biophys J* (2007); ⁵Asthaagiri *et al. JCP* (2006)

Fast K^+ / Na^+ discrimination: Mechanism & Translation Strategy #1

Natural Channels¹



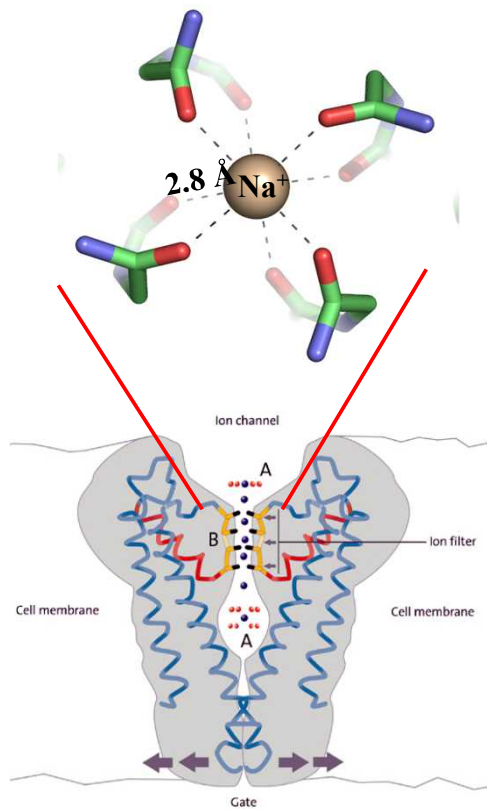
Nanoscale design parameters (natural channel):

“The caress of the surroundings,² the crowding of the ligands”

¹Varma & Rempe *Biophys J* (2007); ²Jordan *Biophys J* (2007); ³Jiang, Brinker, et al. *JACS* (2006)

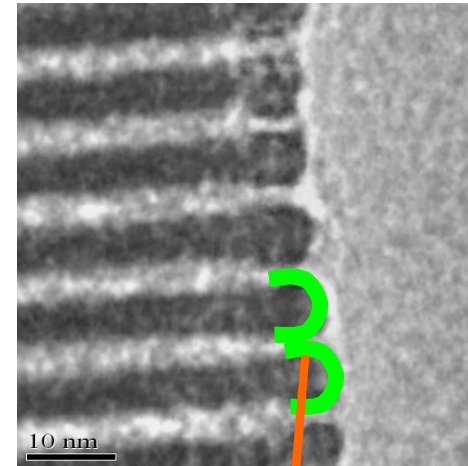
Fast K^+/Na^+ discrimination: Mechanism & Translation Strategy #1

Natural Channels¹

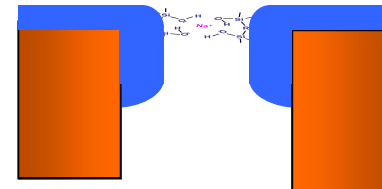
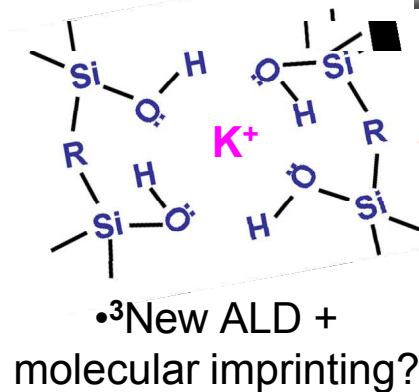


Inorganic Channels

Bio mechanisms
→
Engineered Mimicry



4 nm x 1.5 nm



Nanoscale design parameters (natural channel):

“The caress of the surroundings,² the crowding of the ligands”

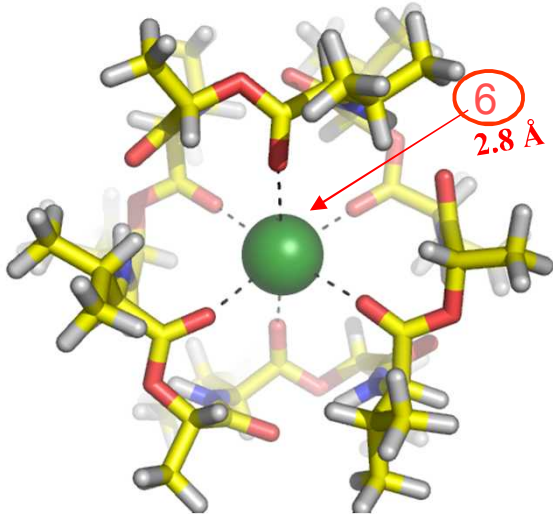
¹Varma & Rempe *Biophys J* (2007); ²Jordan *Biophys J* (2007); ³Jiang, Brinker, et al. *JACS* (2006,2007)

Fast K^+/Na^+ discrimination:

Mechanism #2

Natural K-selective Molecule

- 6 C=O ligands, not >6?

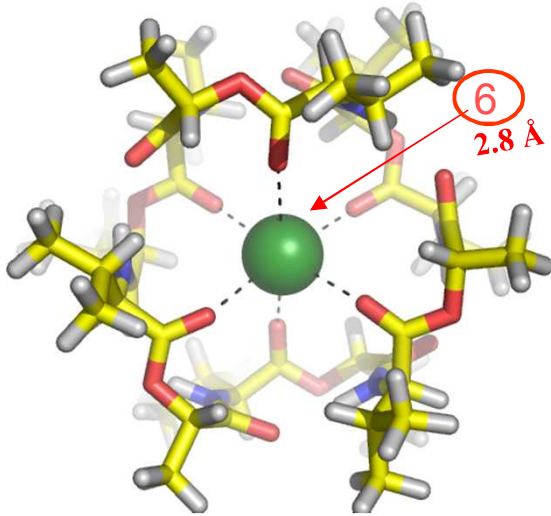


Fast K^+ / Na^+ discrimination:

Mechanism #2

Natural K-selective Molecule

- 6 C=O ligands, not >6?



QuickTime™ and a
TIFF (Uncompressed) decompressor
are needed to see this picture.

QuickTime™ and a
TIFF (Uncompressed) decompressor
are needed to see this picture.

$\Delta\Delta_{gn}(\epsilon)$: change in complexation energy for 2 values of ϵ , which represent dielectric constant lipid membranes. 6-fold coordination is also more stable for higher values of ϵ , such as $\epsilon=80$. $RMSD_n$ reflects change in backbone structure of K^+ -bound valinomycin due to complexation by n water molecules.

Fast K^+/Na^+ discrimination:

Mechanism #2

Natural K-selective Molecule

- Special C=O chemistry?

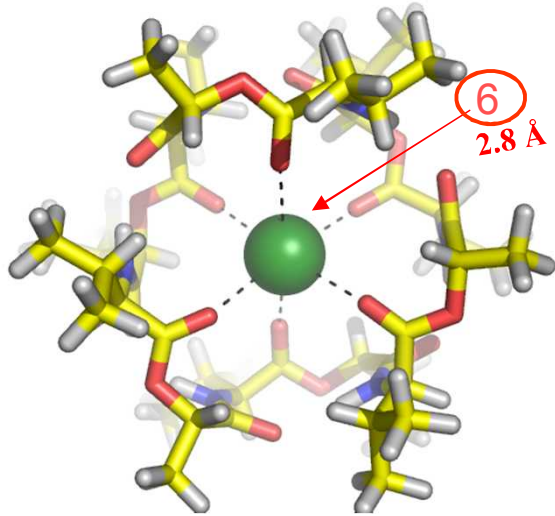
QuickTime™ and a
TIFF (Uncompressed) decompressor
are needed to see this picture.

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TIFF (Uncompressed) decompressor
are needed to see this picture.

Fast K^+ / Na^+ discrimination:

Mechanism #2

Natural K-selective Molecule



- H-bonds maintain cavity size

QuickTime™ and a
TIFF (Uncompressed) decompressor
are needed to see this picture.

¹Varma, Sabo, Rempe *J Molec Biol* (2008)

5.1 Å

2.3 Å

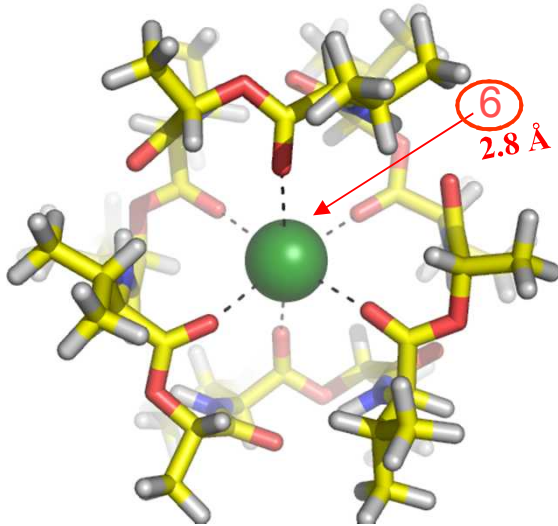
2.9 Å

H-bonds turned off by replacing proton acceptor atoms =O with =CH₂ groups. QC optimization results in small K⁺ complex changes, but large changes in Na⁺ complex. Absence of H-bonds also increases free energy difference between Na⁺ & K⁺ complexes from 12.3 to 17.8 kcal/mol, thus reducing K⁺/Na⁺ selectivity relative to liquid water.

Fast K^+ / Na^+ discrimination:

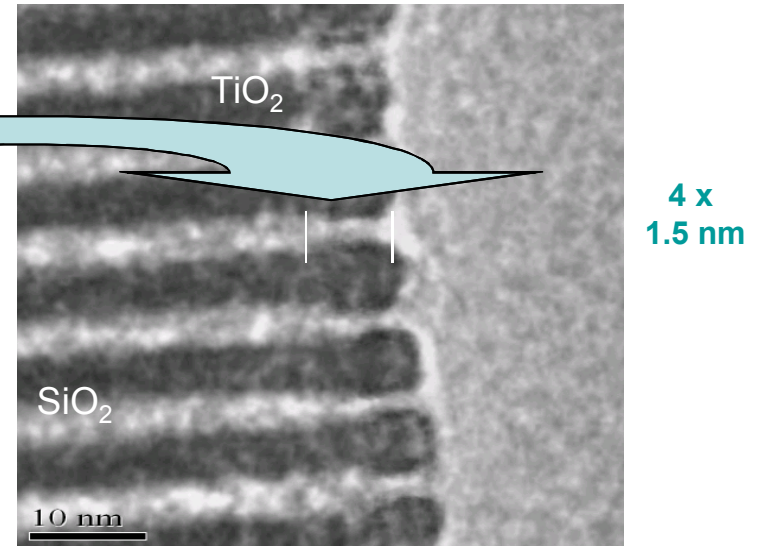
Mechanism & Translation Strategy #2

Natural K-selective Molecule



- **Specific cavity** fits K^+ , not Na^+
- Less selective in water

Inorganic Channels



- **Ion-selective aperture?**

• Outstanding questions:

- Queue of ion binding sites?
- Avoid block?
- Fast water transport + ion selectivity?



Fast size discrimination:

Translated (Fabricated) by Atomic Layer Deposition

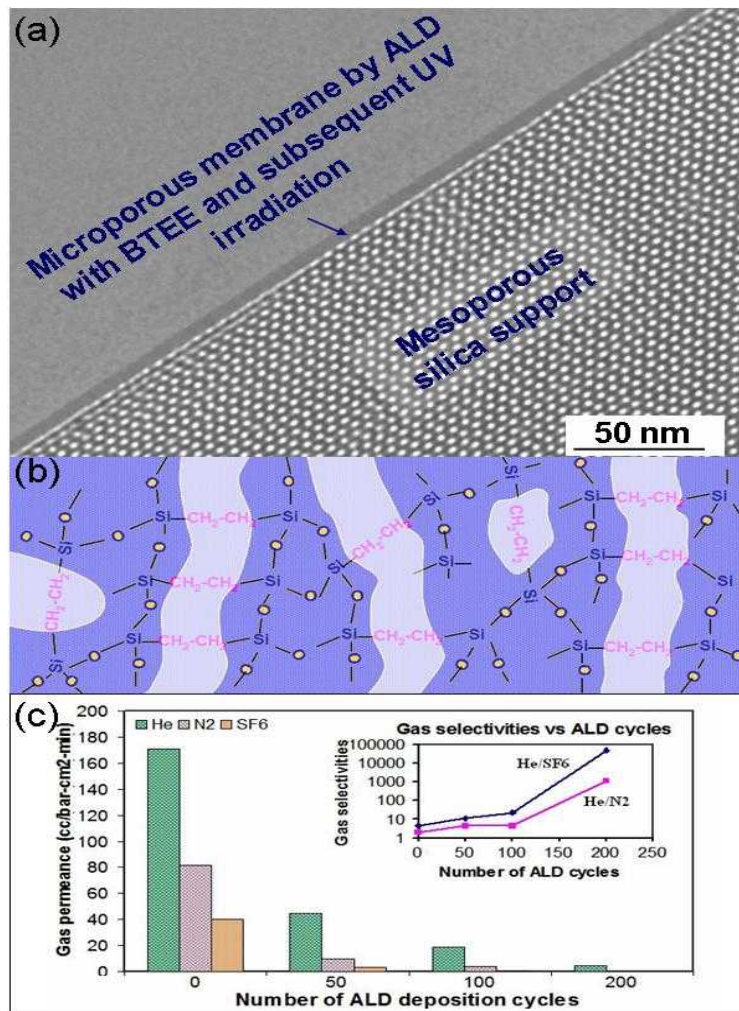
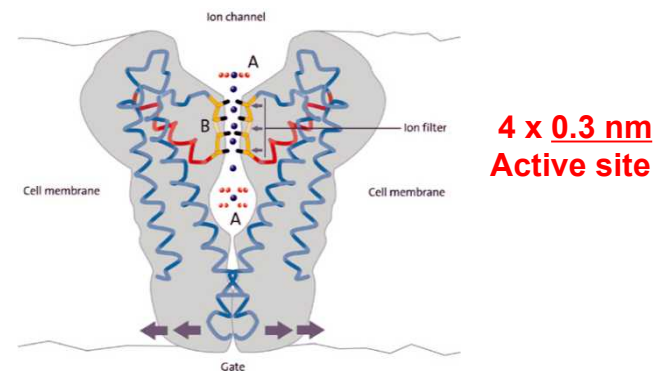


Fig. 1 a) TEM: micropore membrane self-assembled on mesopore support;
b) UV/ozone removes C₂ bridging ligands (pore templates), from dense hybrid film;
c) gas permeance vs ALD cycle; after 200 cycles, N₂ excluded/He transports;
thus pore size ~0.3 nm.

- ¹biomimetic pores (**0.3 nm diameter**)
via new ALD & molecular templating
 - high flux + high size-selectivity (He/N₂)
- template-based approach for uniform molecular-sized pores established
- mimic biological K⁺/Na⁺ selectivity?

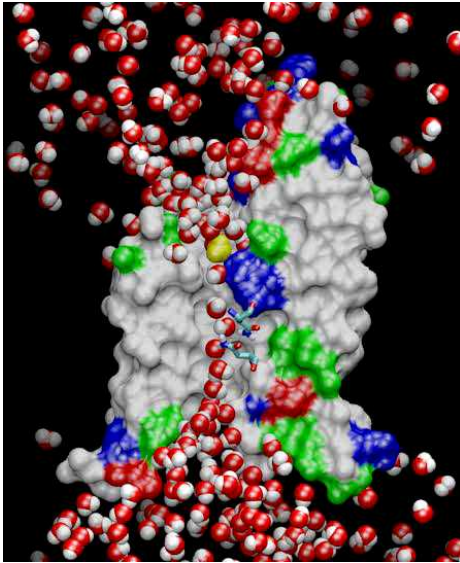


¹Jiang, Brinker, *et al JACS*, 2007

Summary:

Nanoscale Channel Structures for Big Problems

Natural Water Channel



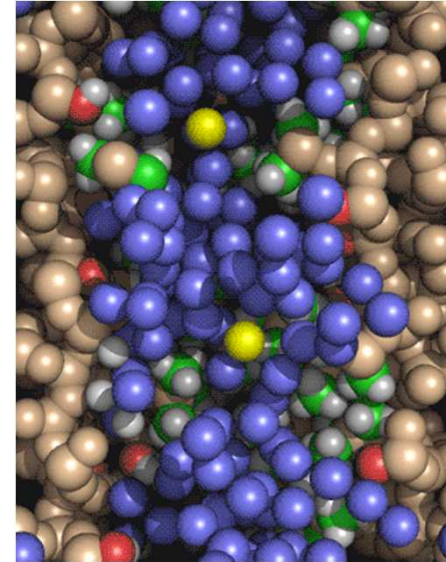
(Tajkhorshid & co)

Bio mechanisms



Engineering solutions

Inorganic Water Channel



(Desal Team, Sandia)

- **Channels:** molecular structures & subtle, important functions
- **Solve problems:**
 - **Health** (medicine + biodefense)
 - **Nanomedicine** (smart dialysis)
 - **Water-Energy** (mineral water, efficiently)

Biology ↔ **inorganic nanostructures**
Quantum Modeling ↔ **experiments**

Acknowledgements

- **Funding**

1. DOE: Sandia's Water Desalination Program
2. DOE: Sandia's LDRD program (BST, ERN)
3. NIH: National Nanomedicine Center

- **Compute time (~ 300,000 hours cpu)**

1. Sandia Computing: Thunderbird
2. National Center of Supercomputing Applications (NCSA), UIUC

- **Collaborative Science, Engineering, Technology Teams**



Desalination Team at Sandia

PI: Susan Rempe

- Jeff Brinker
- Kevin Leung
- Steve Plimpton
- Dubravko Sabo ([postdoc](#))
- Seema Singh
- Sameer Varma ([postdoc](#))
- Ying-Bing Jiang ([postdoc](#))
- Tom Mayer (project manager)

NIH Center for Design of Biomimetic Nanoconductors

(<http://www.nanoconductor.org>)

PI: Eric Jakobsson, UIUC

Senior Scientists

- | | |
|-----------------------------|---------------------------|
| • Narayan Aluru (UIUC) | • Atul Parikh (UC Davis) |
| • Hagan Baylay (U Oxford) | • Umberto Ravioli (UIUC) |
| • Jeff Brinker (SNL) | • Susan Rempe (SNL PI) |
| • Millicent Firestone (ANL) | • Benoit Roux (U Chicago) |
| • David LaVan (Yale) | • Marco Saraniti (IIT) |
| • Kevin Leung (SNL) | • Larry Scott (IIT) |
| • Steve Plimpton (SNL) | |

Ion Discrimination by Nanoscale Design



**Susan Rempe, Sameer Varma, Dubravko Sabo,
Kevin Leung, Ying-bing Jiang, Jeff Brinker**

Sandia National Labs

K⁺ transfer from Water into fully flexible K-channel binding sites



- Build molecular models and predict:
 - Optimized complexes; overlay X-ray structure
 - K⁺ transfer thermodynamics
 - Ion selectivity
- **Achieve a model that represents known data:**
 - reproduces measured ion channel properties
 - reveals **new determinants of selectivity:** **environment & coordination**

K⁺/Na⁺ Ion Discrimination Problem:

How do K-selective channels work?

Traditional Mechanism

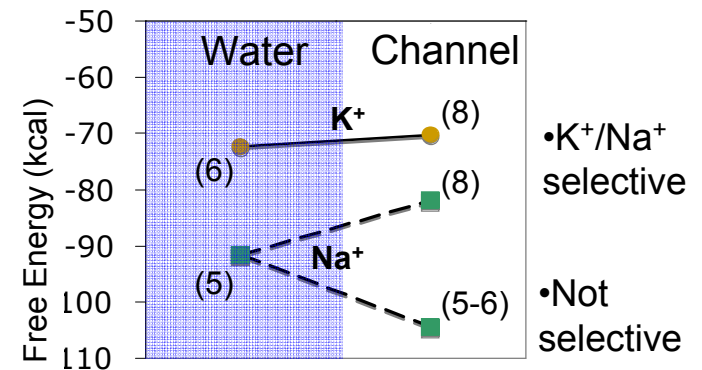
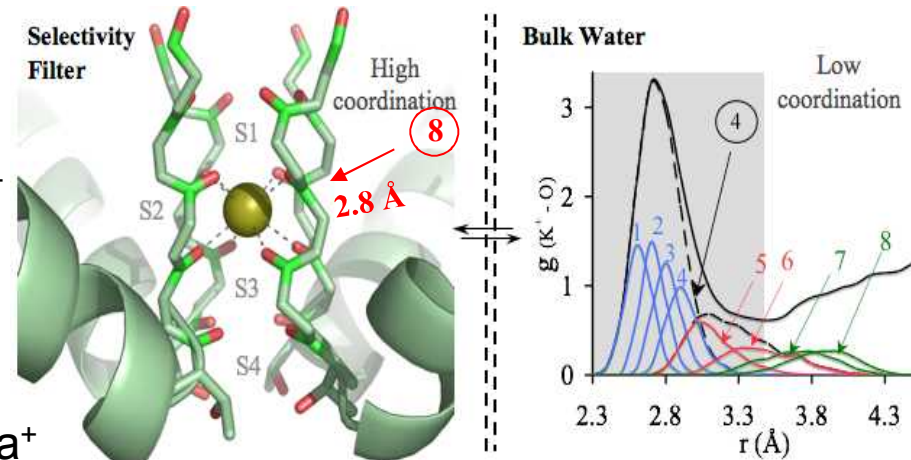
- Ion coordination assumed **fixed, mimicked**
- **Specific cavity size** fits permeant K⁺, not Na⁺

Over-coordination Mechanism

- Ion **coordination linked to environment**
- **Specific (C=O) ligand number** 'fits' K⁺, not Na⁺
- “The caress of the surroundings, the crowding of the ligands.” Jordan (2007), New & Notable *BJ*

Impact

- **New explanation** of K-channel selectivity
- **Engineering parameters**
- **Health, Water, Nanoengineering**



Varma & Rempe (2007). *Biophys J*
Varma & Rempe (2006). *Biophys Chem*
Varma, Sabo, Rempe (2008). *J Molec. Bio*