

Mechanistic Studies on the CVD Modification of Zeolitic Pores In MFI Membranes

Zhong Tang¹, Seok-Jhin Kim¹, Junhang Dong^{1,*}, Tina M. Nenoff²

¹Department of Chemical & Materials Engineering, University of Cincinnati, Cincinnati, Ohio 45221

²Surface and Interfacial Science, Sandia National Laboratories, PO Box 5800, MS 1415, Albuquerque, New Mexico 87185

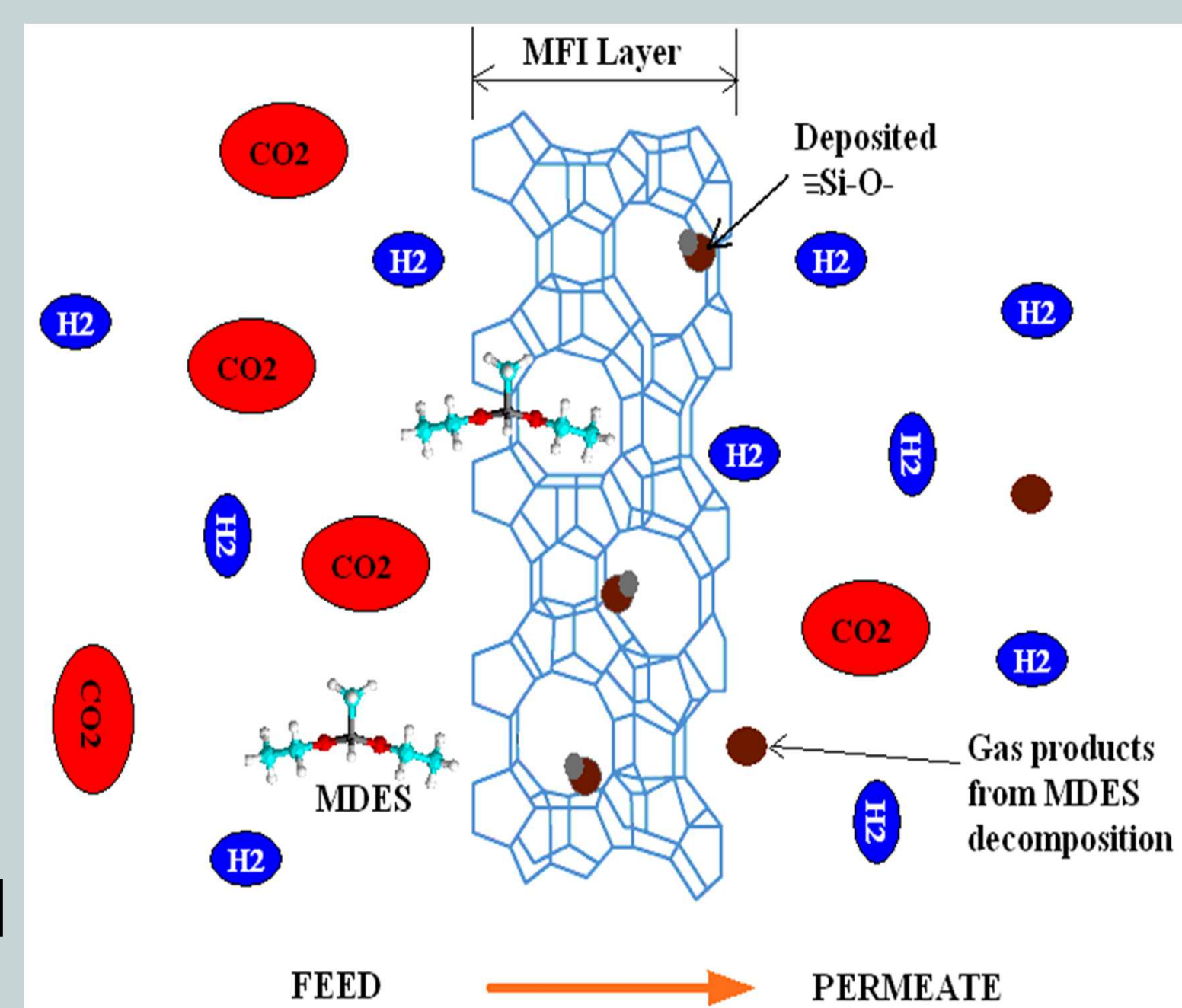
*Phone: (513) 556-3992; Email: junhang.dong@uc.edu.

1. INTRODUCTION

MFI membrane modification by catalytic thermal cracking deposition (CTCD) of silane in the zeolitic pores can dramatically enhance high temperature H₂/CO₂ separation^[1,2]. In this research, we study the mechanisms of this new CVD modification process.

2. APPROACH

Modification was conducted at 450°C during the permeation of a 50/50 H₂/CO₂ gas mixture, with continuous online monitoring of the modification effect. Si modification of zeolite pore, reduces the pore size such that transport of H₂ and CO₂ is changed to **configurational diffusion** mechanism, resulting in a dramatic enhancement of H₂ selectivity. H₂ permeance is only moderately affected because deposition occurs to only a small depth of the channel length.

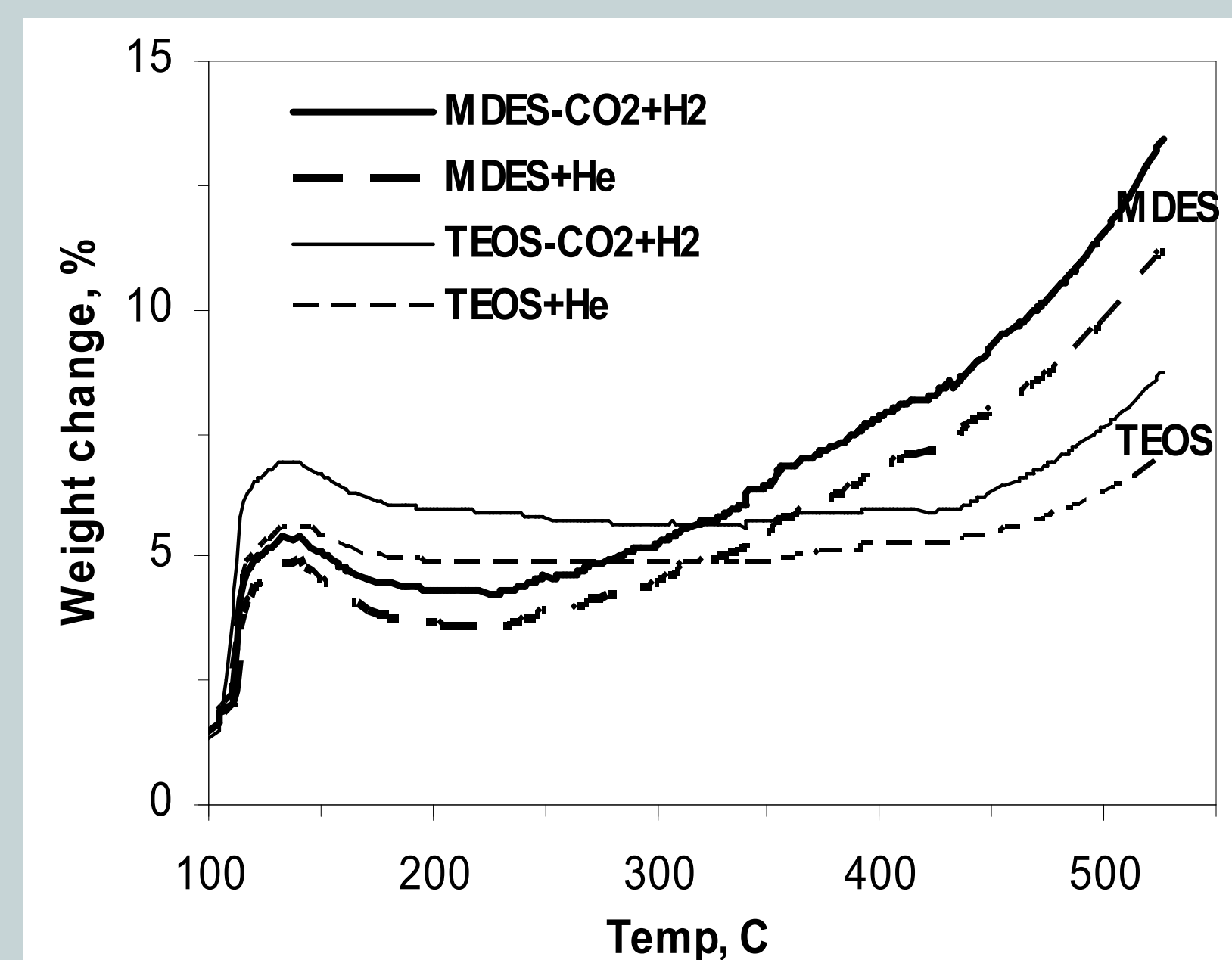
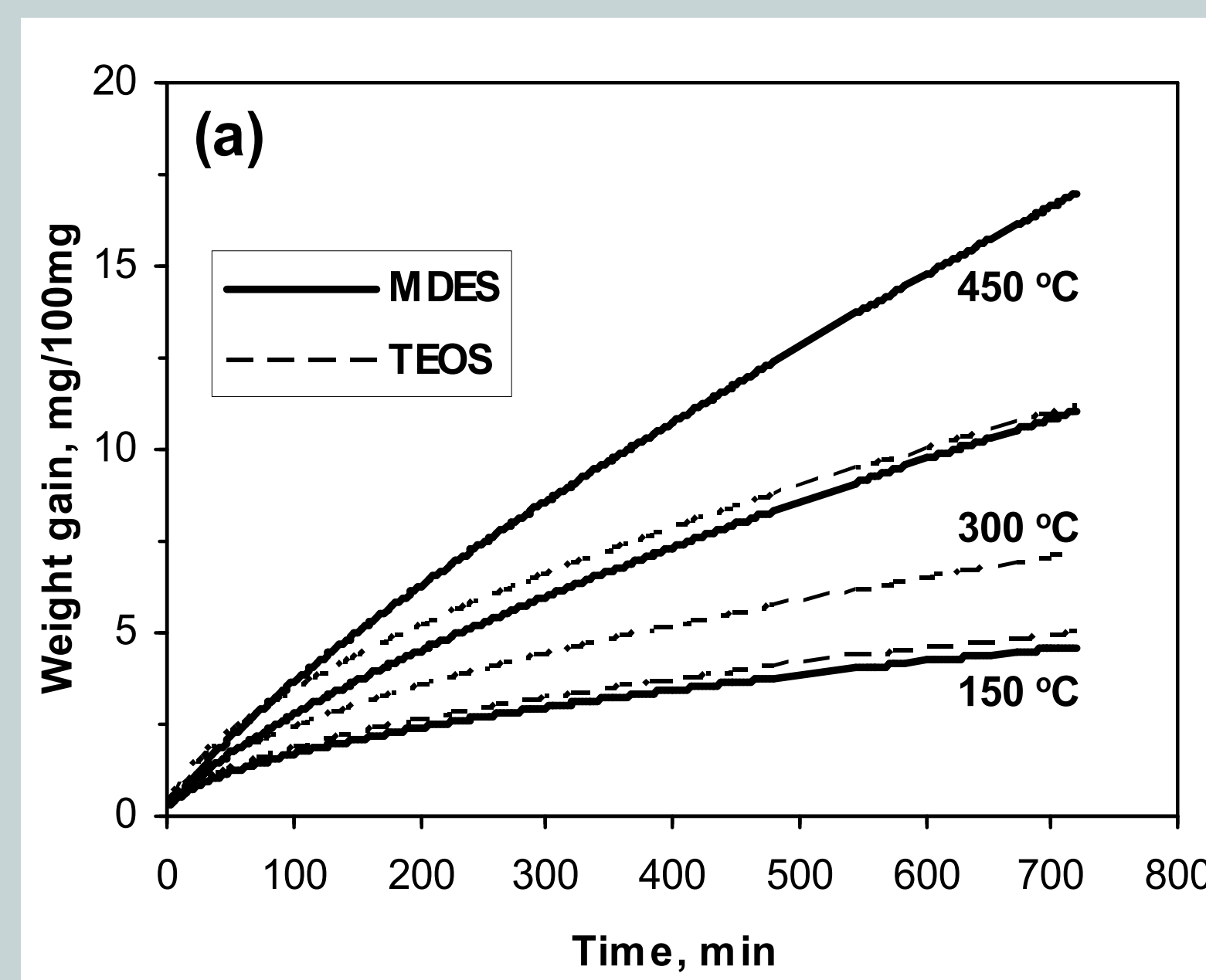


3. EXPERIMENTAL

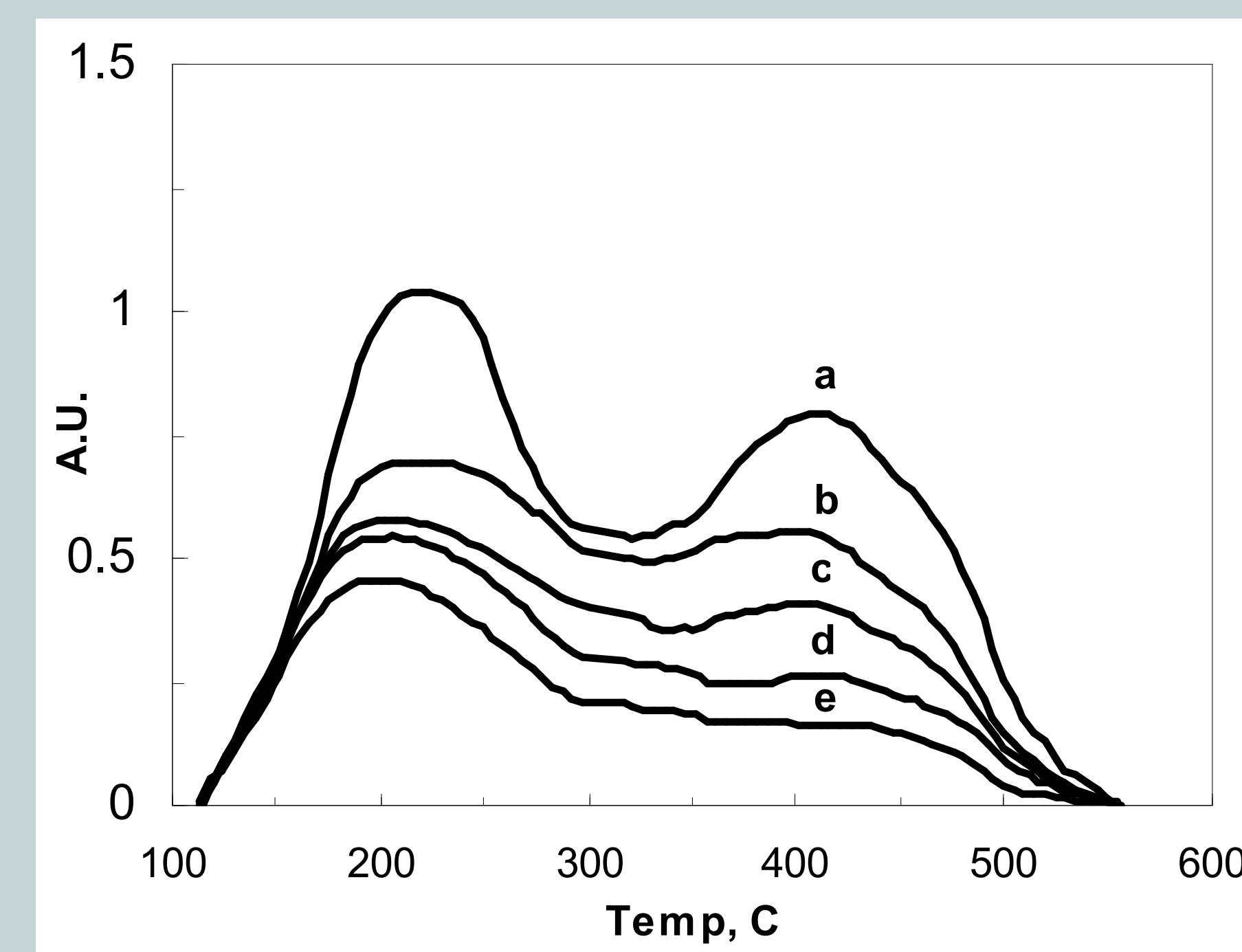
3.1 Zeolite membranes and modification

MFI zeolite particles were obtained by hydrothermal synthesis; Zeolite membranes were synthesized on alumina supports by microwave heating^[3]. CTCD was studied for both MDES (methyl-diethoxysilane, 0.4x0.9nm) and TEOS (tetraethyl orthosilicate, 0.96x0.96nm).

3.2 Characterization

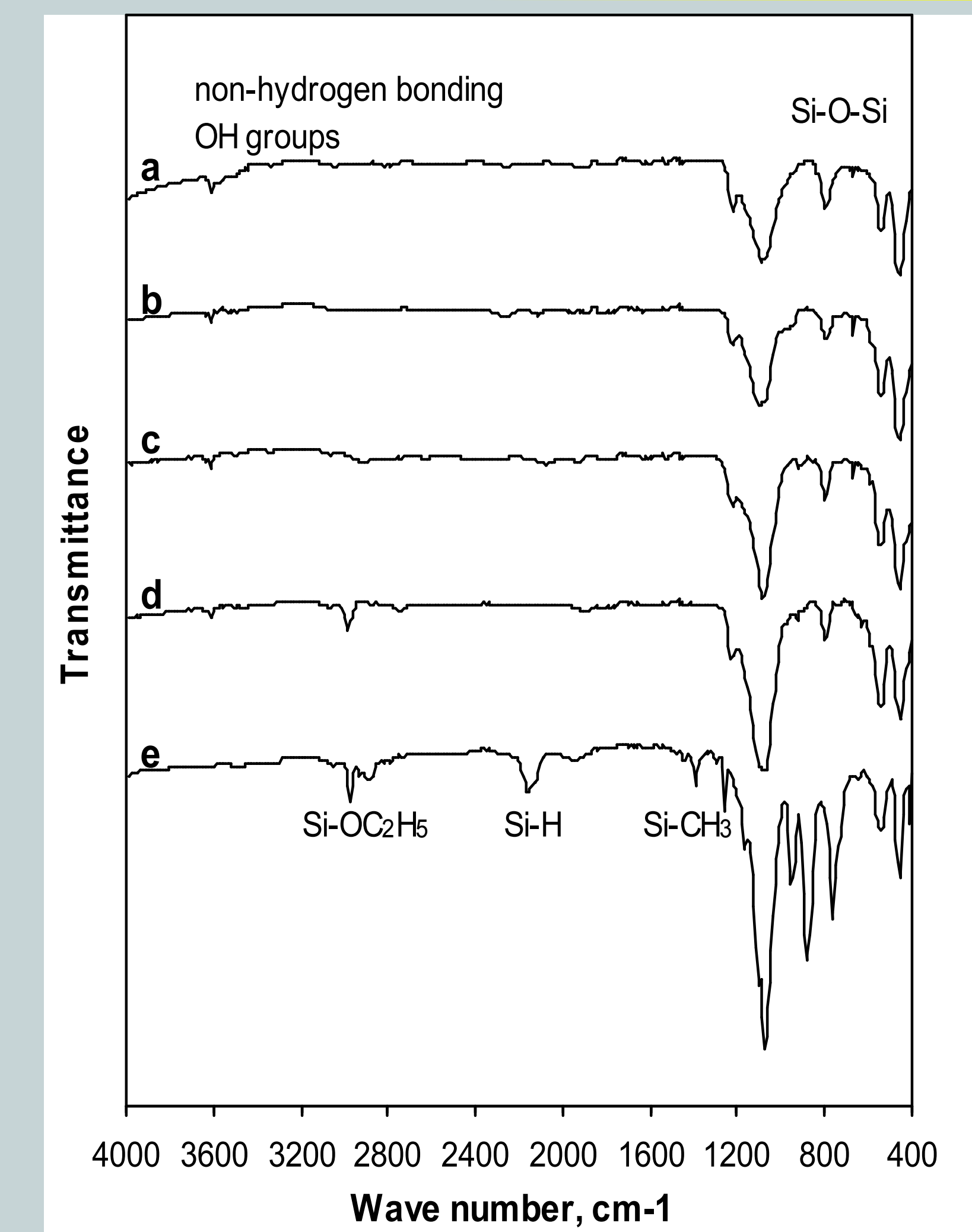


TGA tests showed that large TEOS was unable to enter and deposit in side the zeolitic pores while the **smaller MDES was able to modify the international surface of MFI zeolites**. The CTCD modification was more effective in H₂/CO₂ carrier gas than in inner He carrier because the former generates H₂O which can facilitate the catalytic thermal cracking.

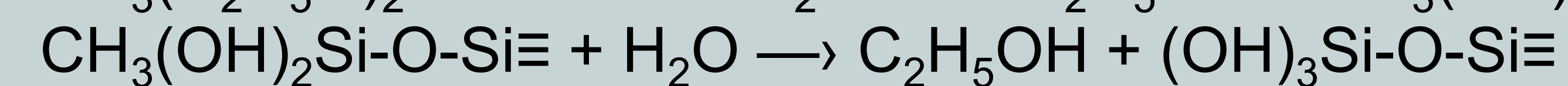
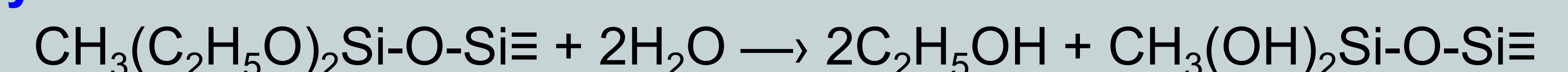


NH₃-TPD results indicated that the deposition occurred inside zeolite pores: (a) sample S-1, (b) 150°C, (c) 300°C, (d) 450°C, (e) 23~450°C

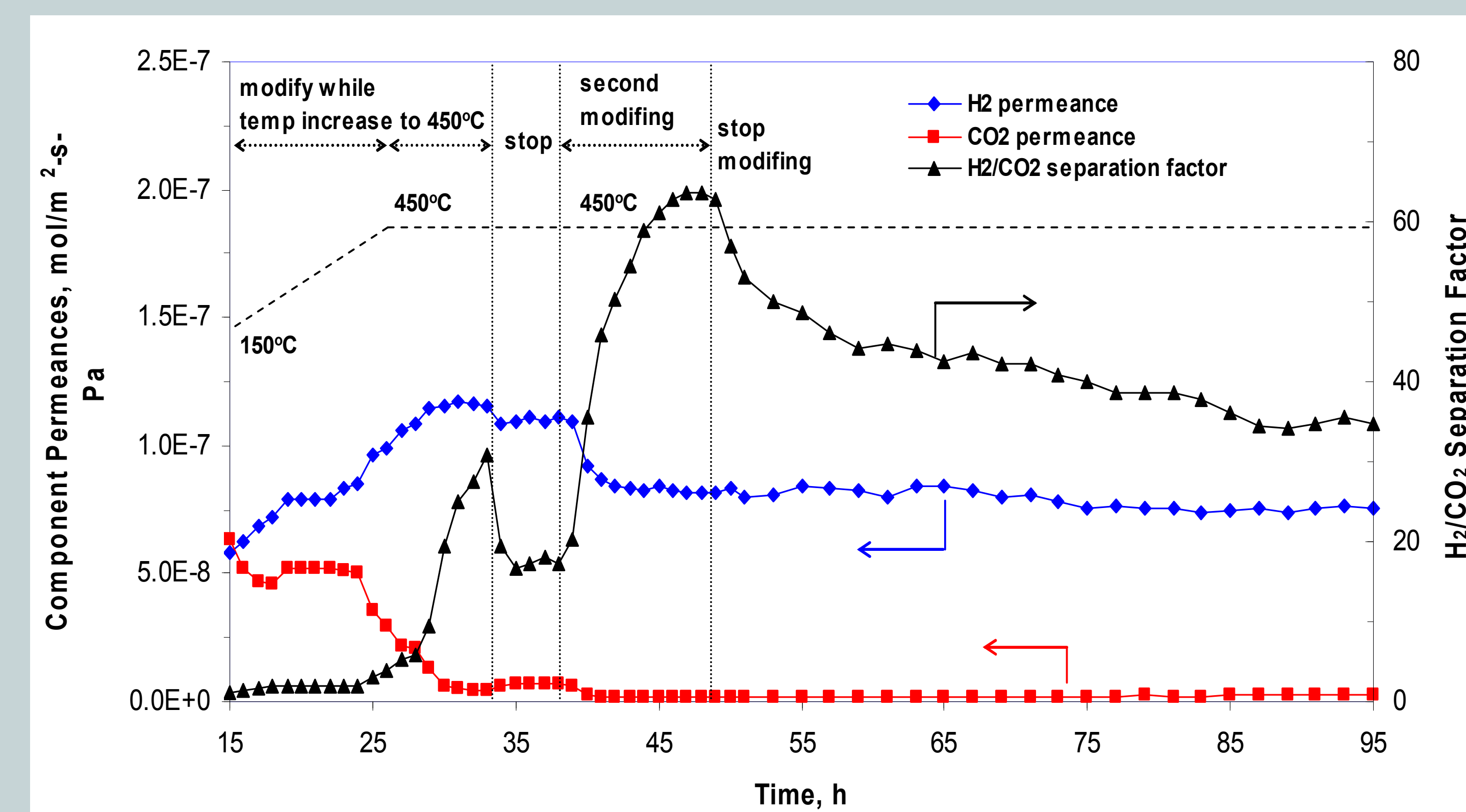
FTIR spectra showing that SiO₂ was the final deposit: (a) sample S-1, (b) modified from 150~450°C, (c) 450°C, (d) 300°C, (e) 150°C



Key Reactions of CTCD for MDES



3.3 H₂ separation enhancement



Results of online monitoring of membrane modification showing elimination of intercrystalline pores by CTCD of TEOS and zeolitic pore reduction by CTCD of MDES.

H ₂ /CO ₂ , 450 °C	Separation Factor
Before modification	3.8
1 st round	15
2 nd round	34

4. CONCLUSIONS

MFI zeolite channels can be effectively modified by CTCD of MDES to achieve high selectivity for H₂ from CO₂. The CTCD of MDES resulted in SiO₂ deposition in side the zeolitic channels. Using H₂/CO₂ as carrier gas not only allows for great process control but facilitates the CTCD process.

References: [1] T. Masuda, N. Fukumoto, M. Kitamura, *Micropor Mesopor. Mat.* 48 (2001) 239; [2] X. Gu, Z. Tang, J. Dong, *Micropor. Mesopor. Mat.* 111 (2008) 411; [3] X. Gu, J. Dong, T.M. Nenoff, D.E. Ozokwelu, *J. Member. Sci.* 280 (2006) 624.

Acknowledgement: Research supported by NETL/DOE (DE-FG36-GO15043) & SNL LDRD. Sandia is a multi-program lab operated by Sandia Corp., a Lockheed Martin Co., for the US DOE's NNSA, DE-AC04-94-AL85000.