

# Synthesis and Thermal Stability of Gold-Zinc Oxide Nano-Composite Thin Films for Electrical Contacts

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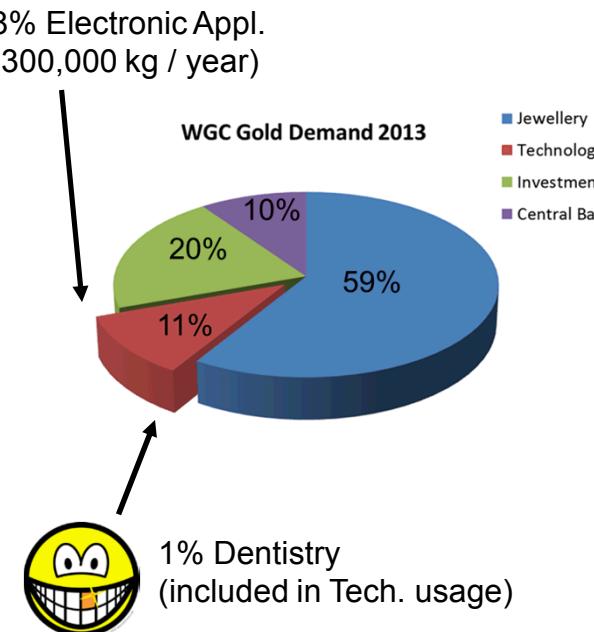
# Technical Application

## Electronic connectors, switches and relays

Electronics Applications:  
Chemically Inert and low electrical  
resistance

World Gold Council,  
Gold Demand Report Feb. 2014

Tonnes	2012	2013 <sup>1</sup>
Jewellery	1896.1	2,209.50
Technology	407.5	404.8
Electronics	284.5	282.4
Other Industrial	84.4	85
Dentistry	38.6	37.3
Investment	1,568.10	773.3
Total bar and coin demand	1,289.00	1,654.10
Physical Bar demand	962.7	1,266.90
Official Coin	213	283.4
Medals/Imitation Coin	113.4	103.8
ETFs & similar products <sup>3</sup>	279.1	880.8
Central bank net purchases	544.1	368.6
Gold demand	4,415.80	3,756.10
London PM fix, US\$/oz	1,669.00	1,411.20



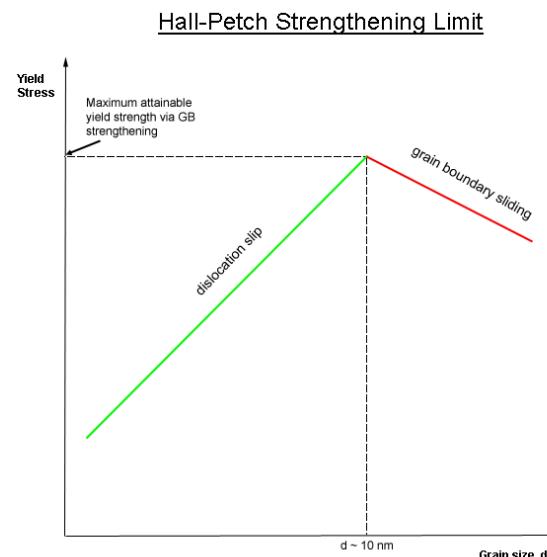
- Pure gold is soft and has unacceptable amount of friction and wear
- Gold is typically hardened by alloying with small amounts of Ni or Co (referred to as hard gold) to achieve the desired balance between friction, wear and ECR
- Current practice is to apply hard gold by electrodeposition
- Nickel underplating is used any time the substrate alloy contains copper to prevent diffusion of Cu into Au coating.

# Hard Gold

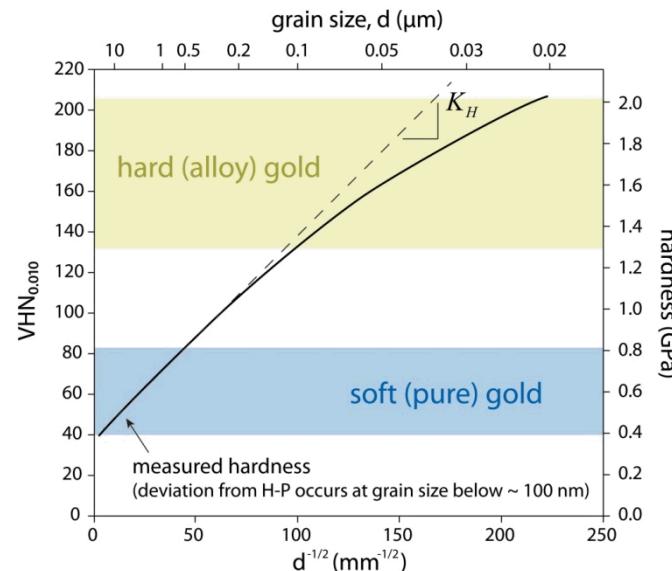
## ASTM Types

From ASTM B488-11 / MIL-DTL-45204D:

(mass % Au)	type	suggested applications (ASTM)
> 99.7% Au (hardest)	I	general-purpose, high-reliability electrical contacts
> 99.0% Au	II	general-purpose, wear resistance; low temperature only
(softest) > 99.9% Au	III IIIA	soldering; limits impact of oxidation of codeposited material semiconductor components, nuclear eng., high temperature



From: Lo, Augis, and Pinnel, JAP (1979)



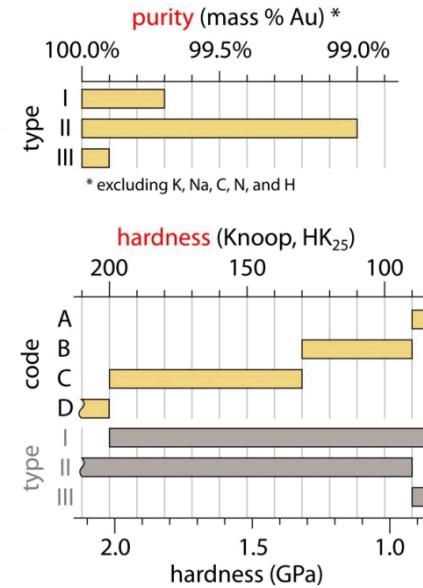
Hall-Petch relationship:

$$H = H_o + K_H d^{-1/2}$$

$H$  - hardness

$d$  - avg. grain diameter

Strengthening via grain refinement  
Theoretical maximum strength at  $\sim 10\text{nm}$



# Electrochemical Deposition

Problems: Toxic plating baths & surface oxidation

## Electroplating:

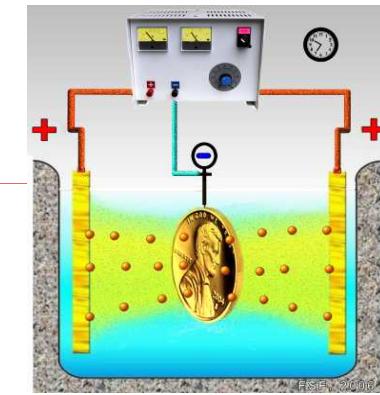
Hard Gold is typically plated from cyanide solution



Bath modifiers of As, Pb, Cd or Ti

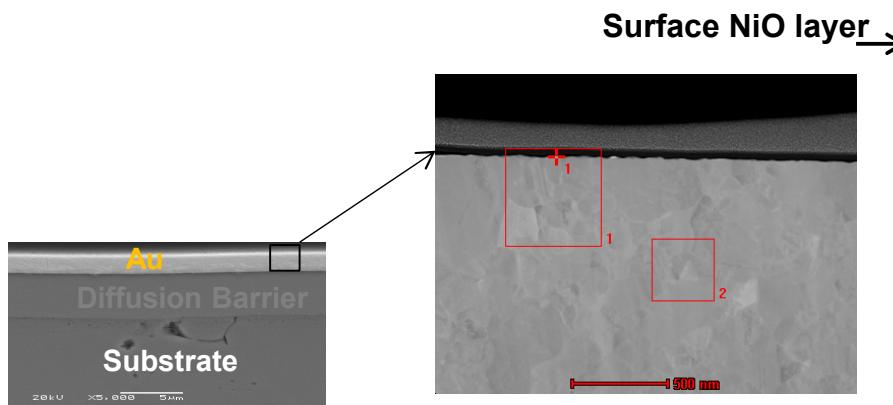
Small amounts hardening agent such as Ni or Co

Nickel barrier layer applied first



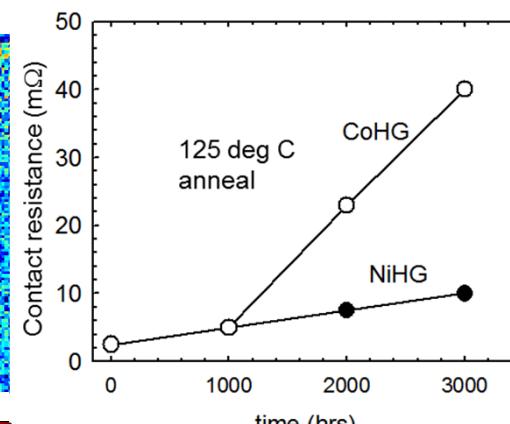
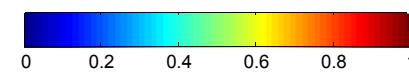
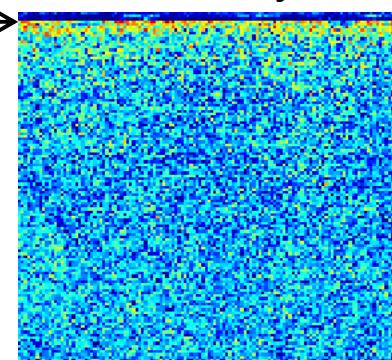
## Diffusion and Surface Oxidation:

Diffusion and segregation of Ni or Co hardeners to the surface over time, ECR degradation



Cross-sectional TEM and Spectral Imaging

Relative Ni concentration  
After 35+ years

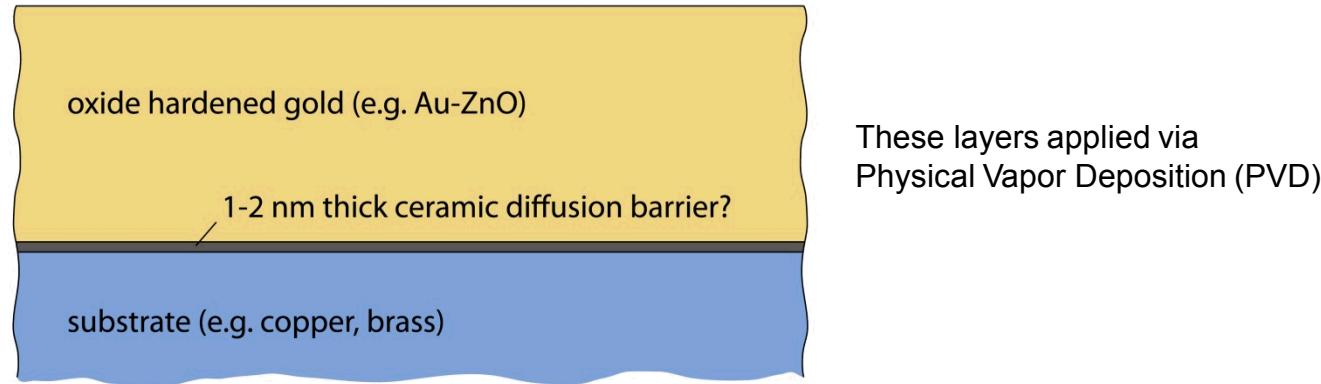


Y. Okinaka and M. Hoshino, Gold Bulletin, 31(1), 3 (1998).

# PVD Hard Gold Synthesis

## Routes to Mitigate Diffusion

- *Not limited to electrochemically compatible materials*
- *Shut-down diffusion by co-deposition of thermodynamically stable ceramic material*
- *Deposition of barrier under-layer would eliminate diffusion of substrate material*
- *Semiconducting ceramic chosen to minimize potential impact on electrical resistance*



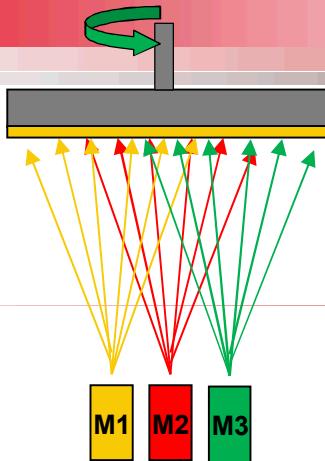
- *Au-ZnO co-deposition by PVD (e-beam evaporation or sputtering)*
- *PVD is environmentally friendly*

# Co-Deposition

## Available PVD Options

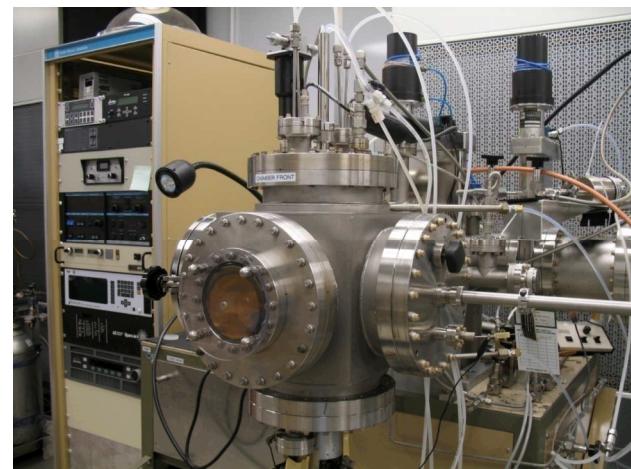
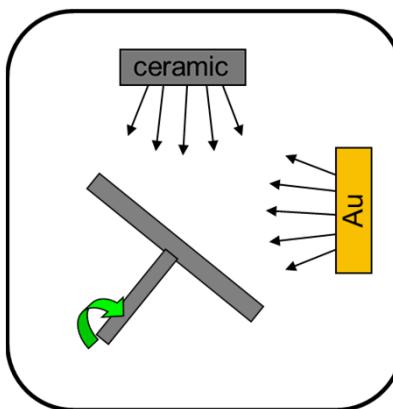
### Evaporation:

- Triad e-beam evaporation of ternary alloy thin films
- Shutter in front of substrate for consistent composition, graded or layered films
- Independent QCM control of material deposition rate
- Compositional control to < 0.1%



### Sputtering:

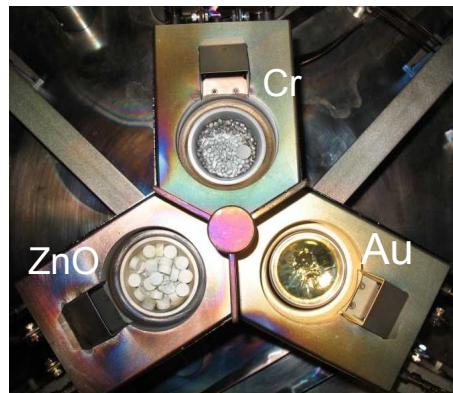
- Co-deposition of elements, alloys and compounds
- Composition control to ~1-2%, limited by minimum power required for plasma ignition
- Composite targets expensive and limits experimental compositional range



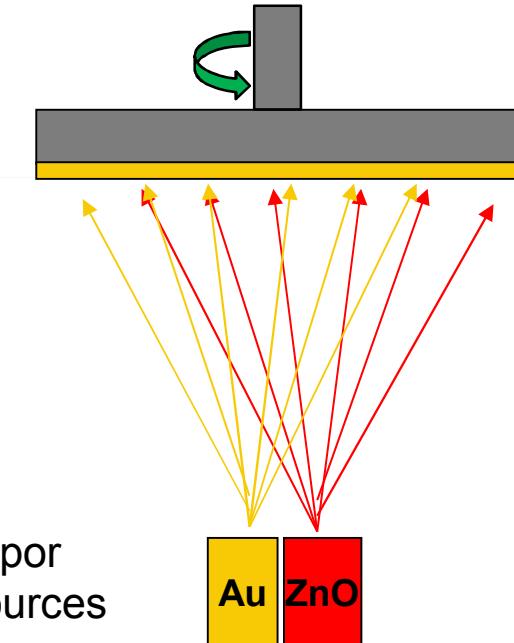
# Triad E-Beam Deposition



E-beam approach chosen for compositional control



Substrate  
 $Au_xZnO_y$  Film

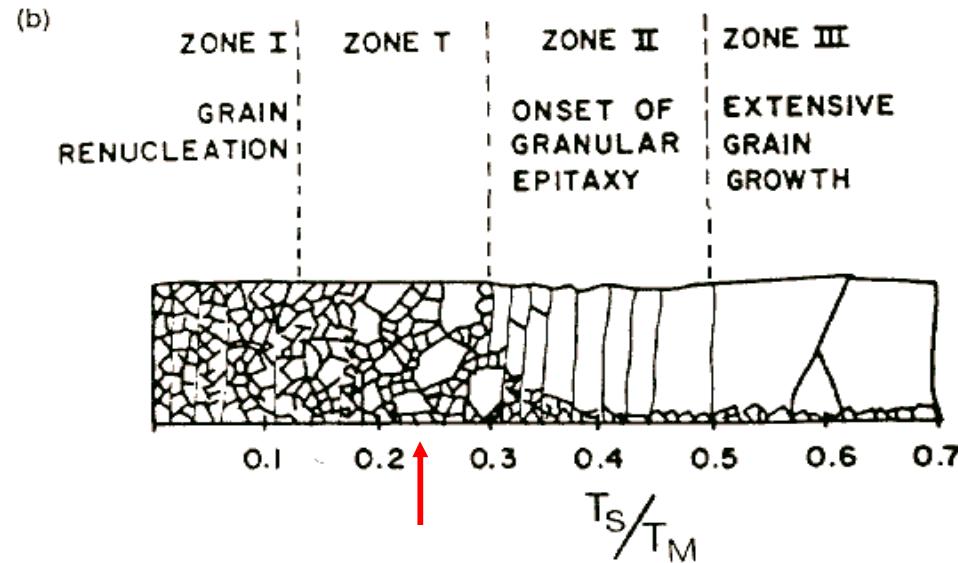


- Triad electron gun for E-beam evaporation
- Co-deposition of Au-ZnO composite thin films
- Third pocket used for adhesion or barrier layer material
- Shutter in front of substrate for consistent composition
- Substrate rotation for improved uniformity
- Line of sight shields on QCMs eliminate cross-talk

# Zone Model for Evaporated Films



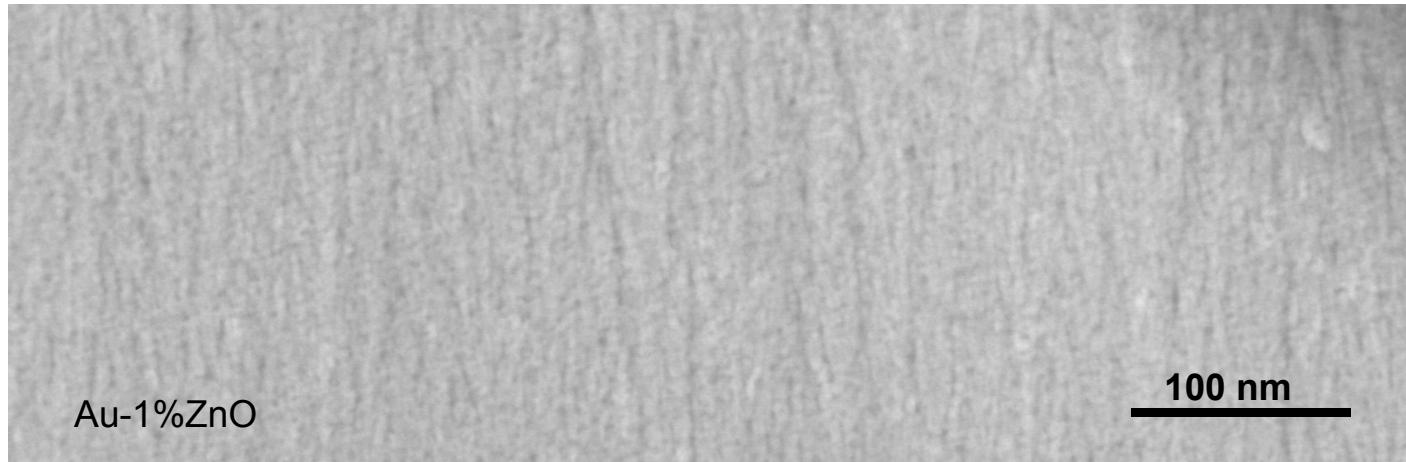
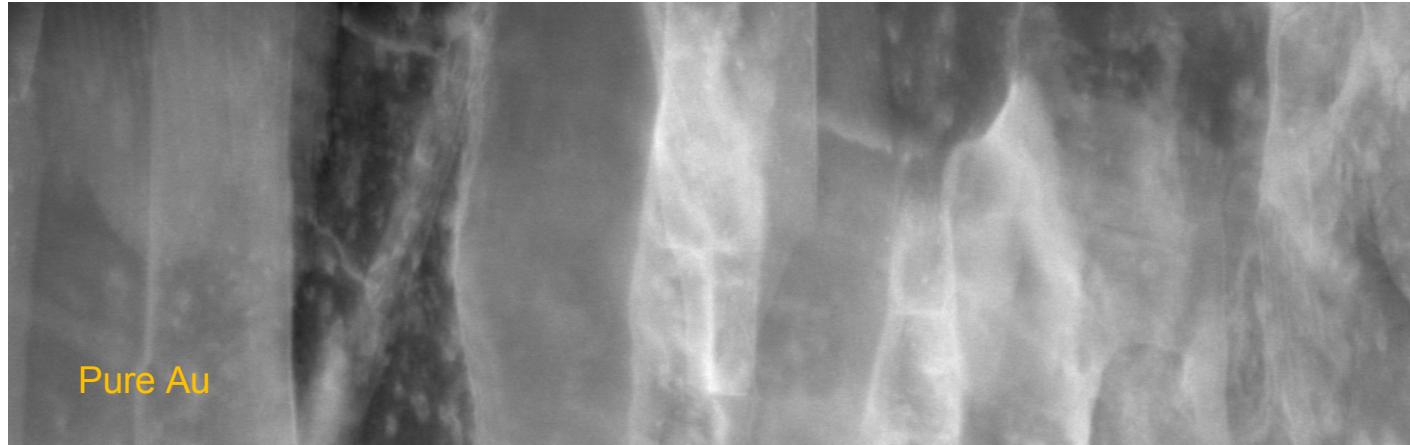
## RT Au Transitional Zone



Fraction of $T_m$	Au Temp (°C)	Au Temp (K)	ZnO Temp (°C)	ZnO Temp (K)
$T_m$	1064	1337	2248	2521
$0.5 T_m$	396	669	988	1261
$0.4 T_m$	262	535	735	1008
$0.3 T_m$	128	401	483	756
$0.22 T_m$	25	298	281	554

# Cross-Sectional TEM

(TEM Specimens prepared by FIB)

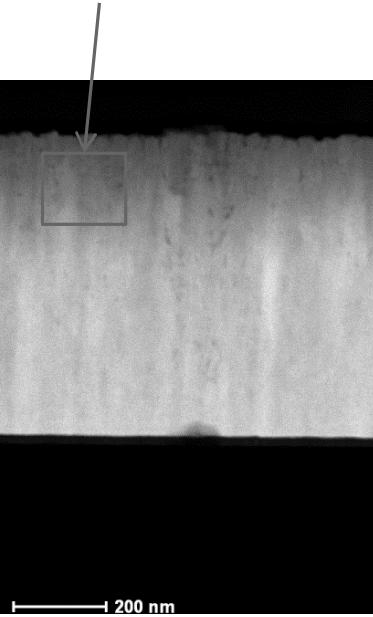


*Columnar grain structure is clearly observed with a significant reduction in grain size for Au + ZnO (1 vol%) thin film*

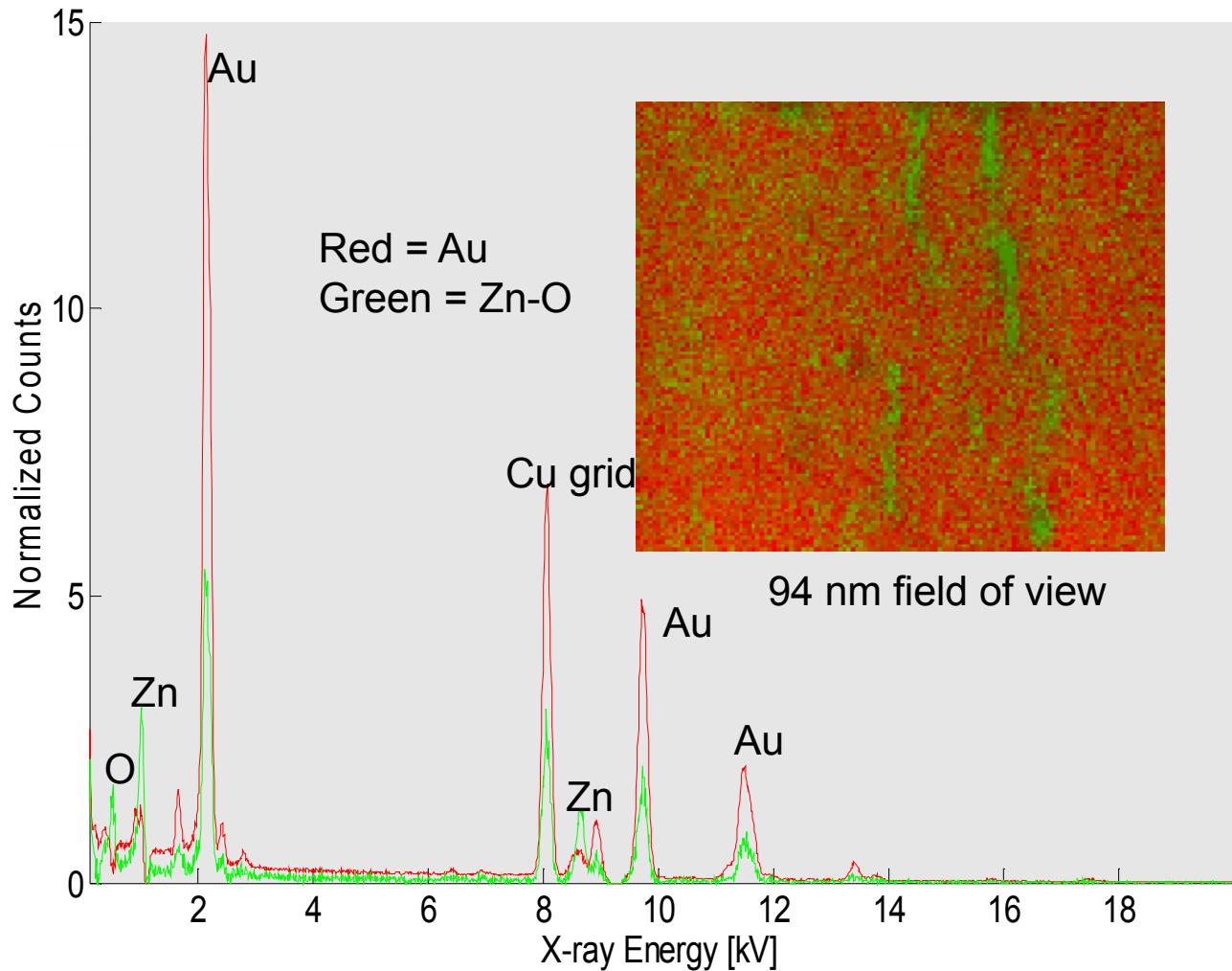
# STEM Spectral Analysis

X-Sect Au-ZnO film

Region of spectral analysis

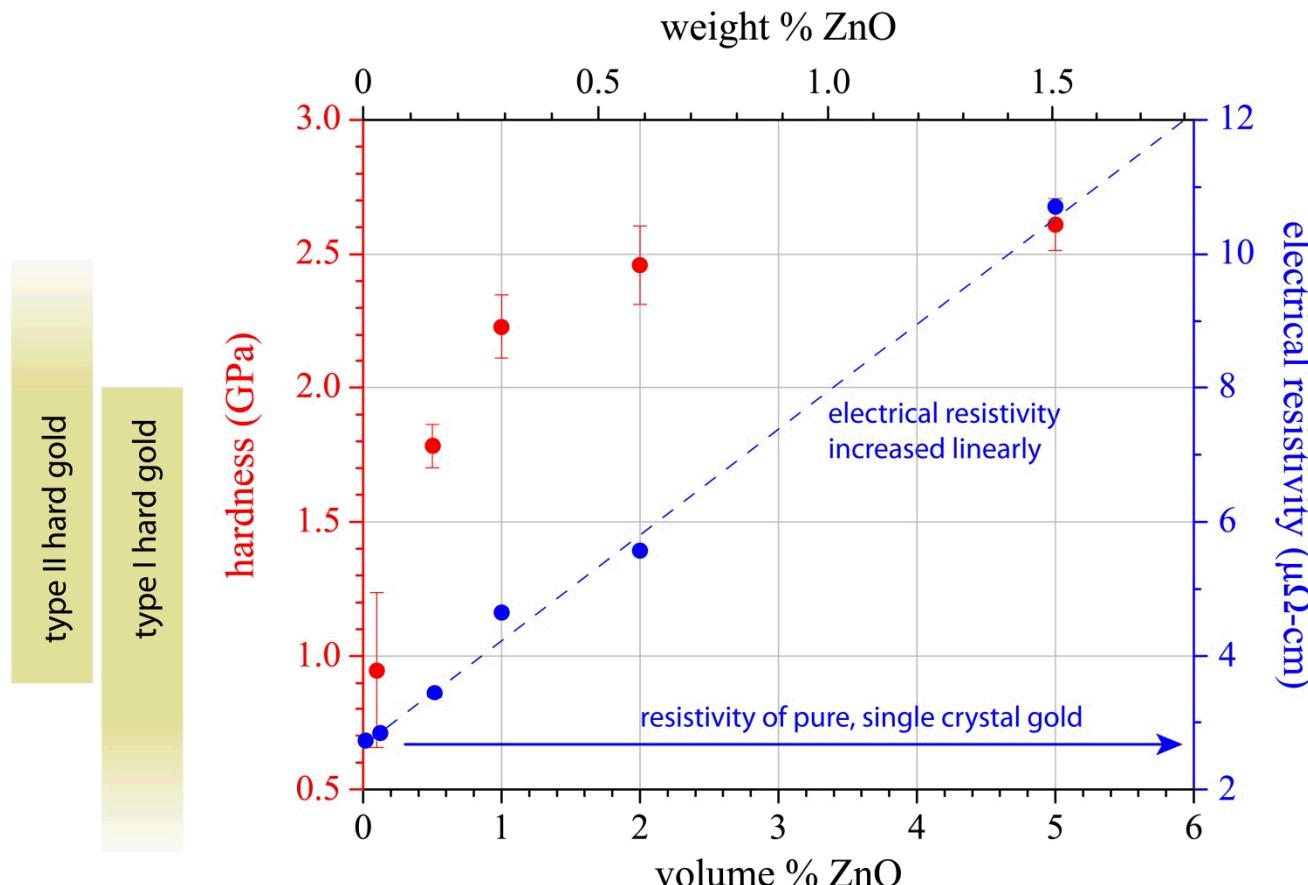


STEM cross-section of  
Au+ZnO (5 vol%)



# Hardness and Electrical Resistivity

## Comparison to hard gold specification



for reference:

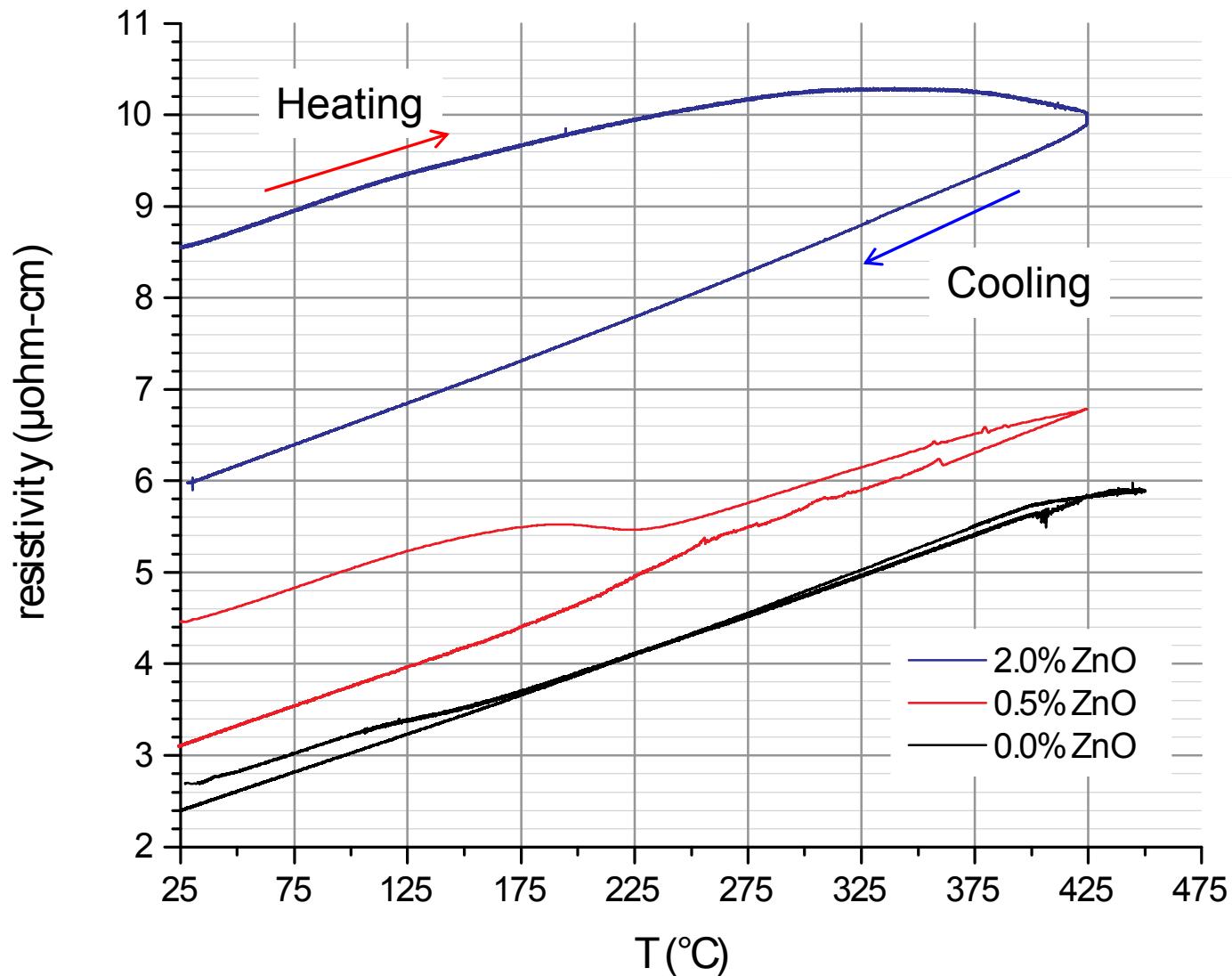
0.7 vol % Ni is ~ 0.3 wt % (type I, best ECR performance)

2.2 vol % Ni is ~ 1 wt % (type II, max allowed)

Electrical resistivity measured via van der Pauw method -- square Si wafers pieces coated with composite, no adhesion layers

# Temperature Dependent Resistivity

TCR Stability (0, 0.5, 2.0 vol% ZnO)



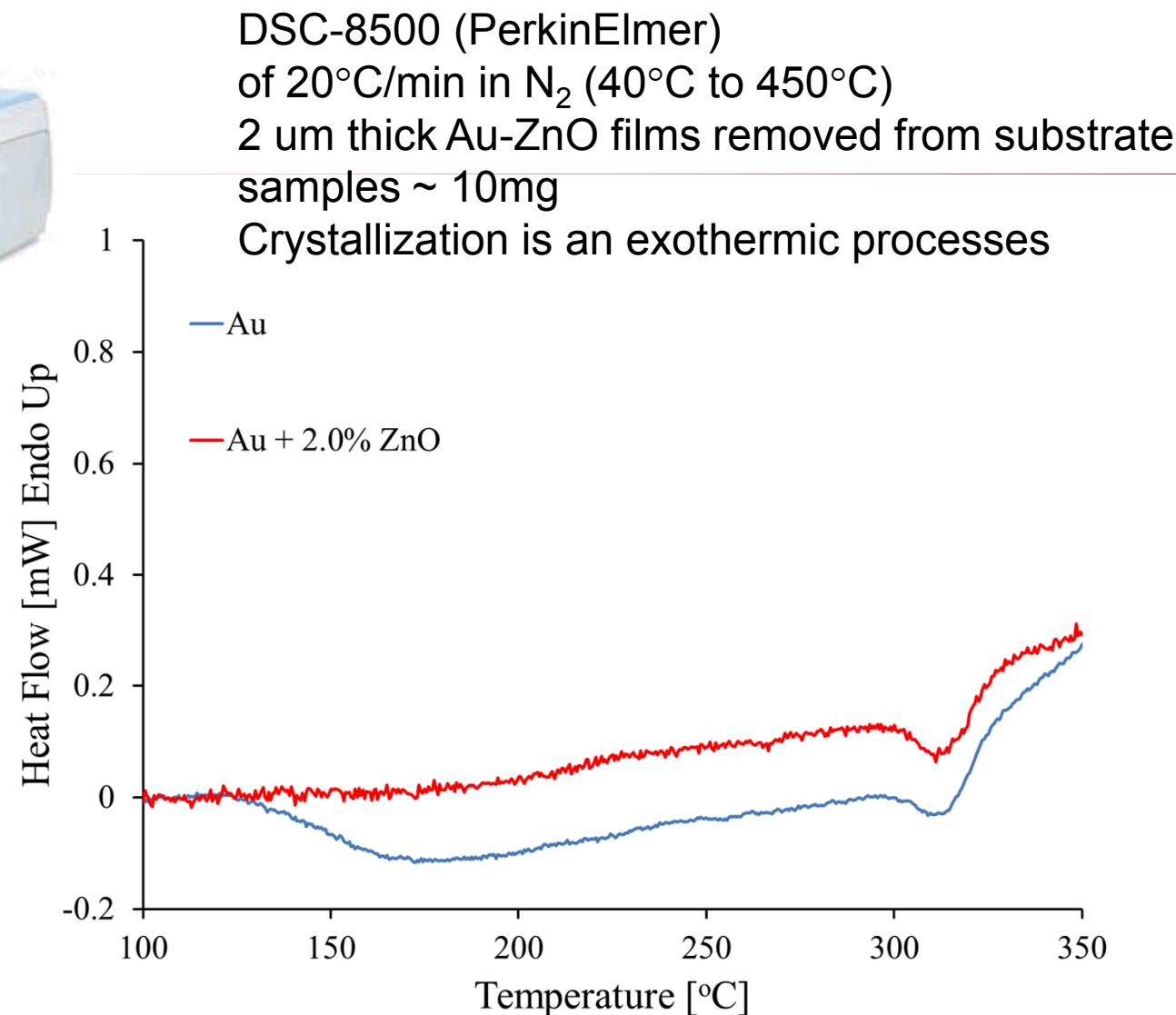
$$\text{TCR} \quad \alpha = (1/R)\Delta R / \Delta T \quad (\text{}/\text{°C})$$

$\alpha=0.004$   
agrees with literature

# Differential Scanning Calorimeter

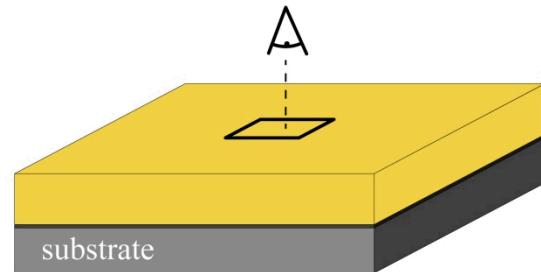


DSC

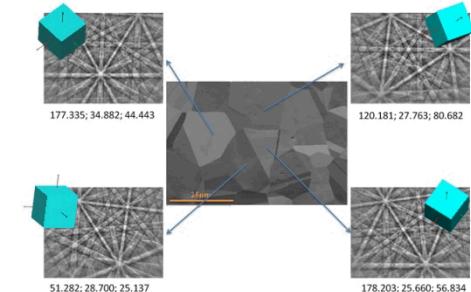


## Grain size and texture impact

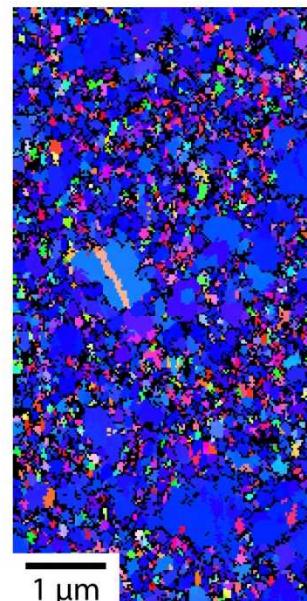
surface normal EBSD maps



Oxford Instruments - <http://www.ebsd.com>

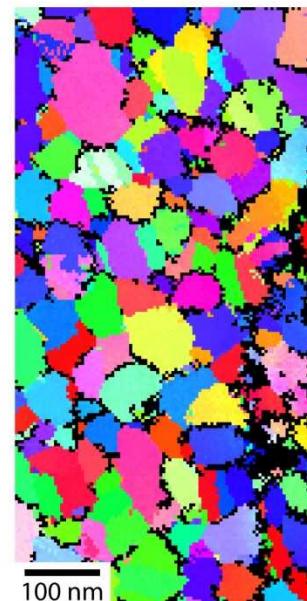


pure Au film:

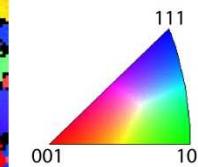


large avg. grain size (> 250 nm)  
preferential <111> surface texture  
bimodal distribution  
hardness ~ 900 MPa

2.0 vol. % ZnO film



equiaxed & non-textured  
avg. grain size ~ 50 nm  
hardness ~ 2.5 GPa

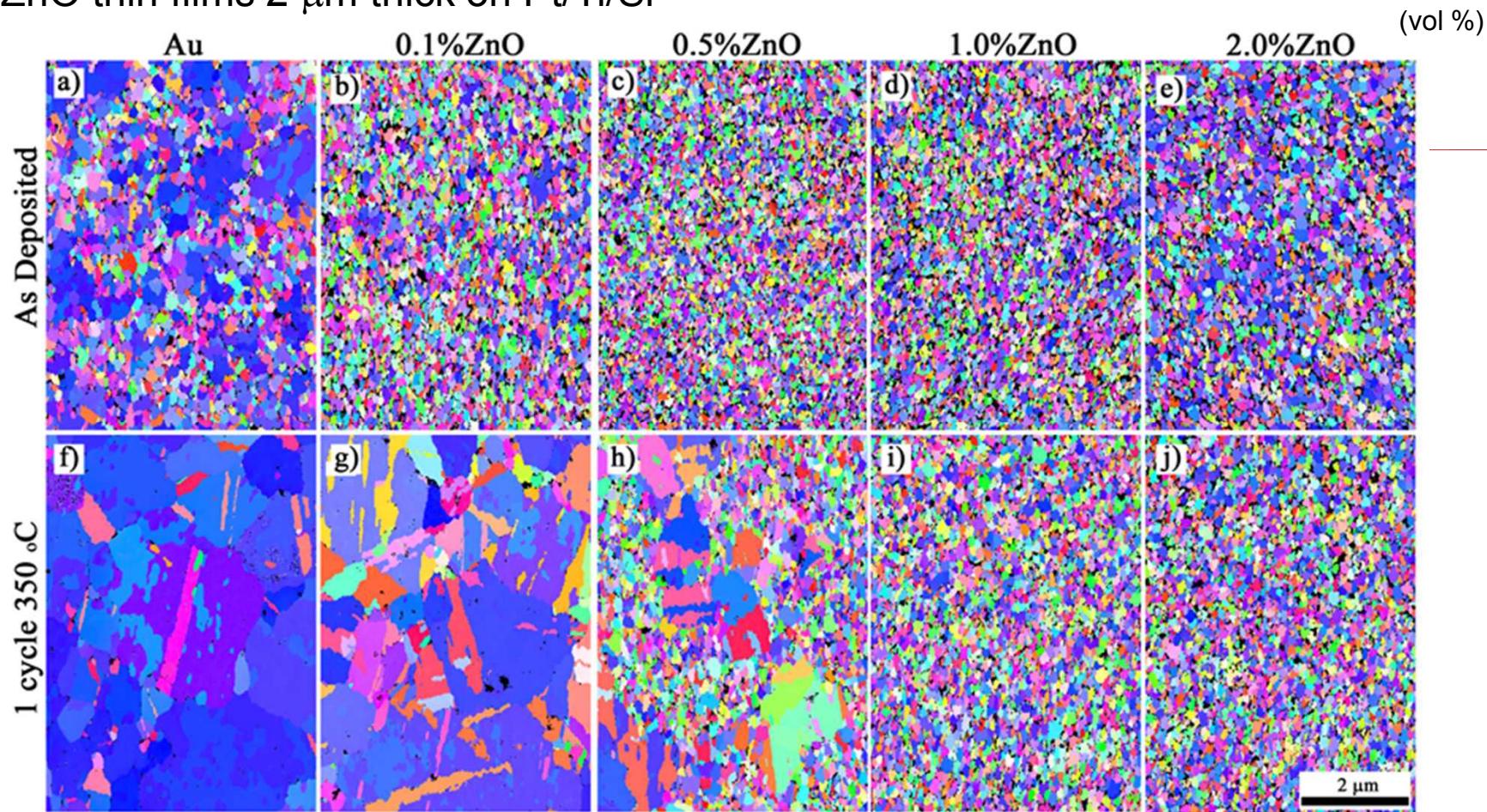


\* note the difference in scale bars \*

# Thermal Stability

## EBSD

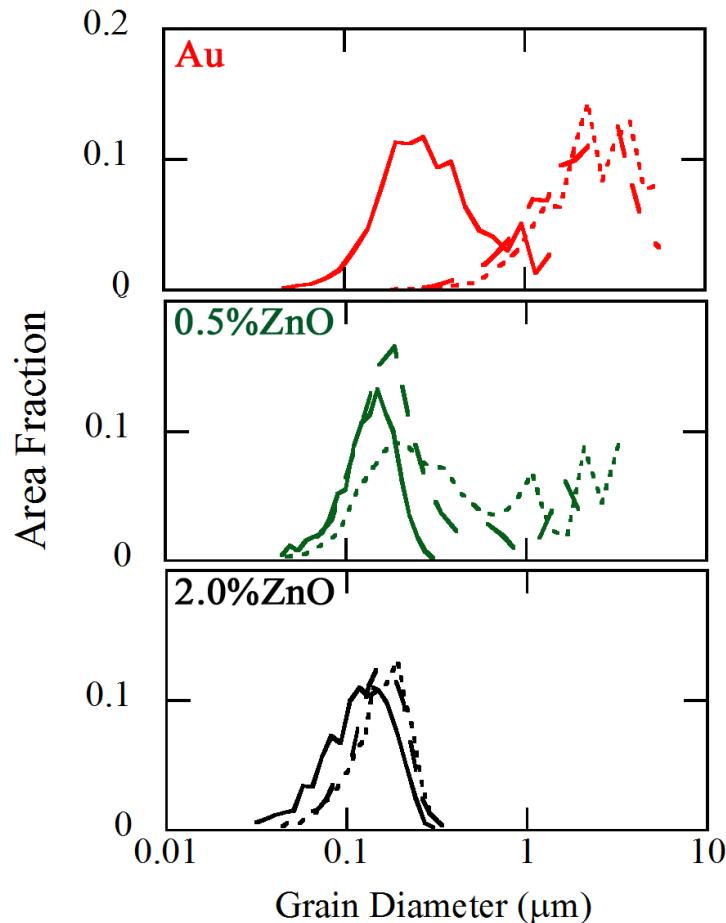
Au-ZnO thin films 2  $\mu\text{m}$  thick on Pt/Ti/Si



Temperature cycled to 350  $^{\circ}\text{C}$  at 10  $^{\circ}\text{C}/\text{min}$

# Thermal Stability

## Grain Size Distribution



Large grain growth

Abnormal grain growth, large grain sintering observed only in regions

Addition of ZnO impeded grain sintering

— As deposited  
- - - 1 cycle  
- · - 5 cycles

Temperature cycled to 350°C at 10°C/min

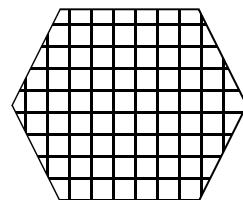
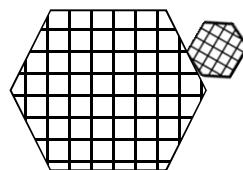
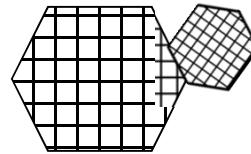
# Sintering

## Grain Growth Mechanisms

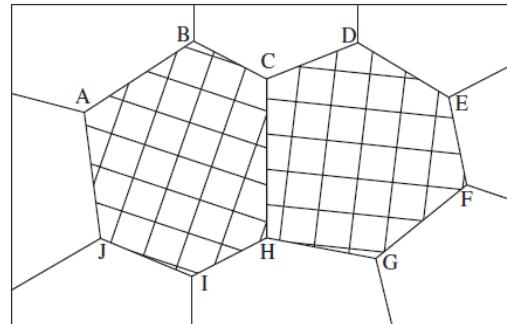
The two common mechanisms for grain growth are:

1. Ostwald Ripening – Diffusion of atoms from smaller grain to larger grain
2. Coalescence – adjacent crystallites join to become one larger grain

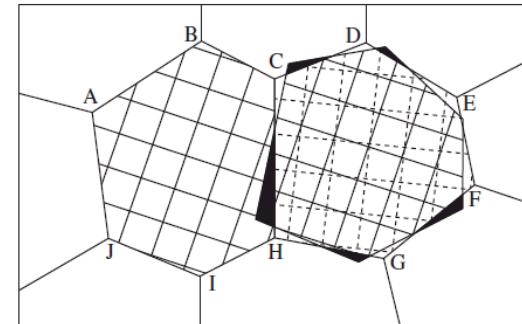
Ripening:



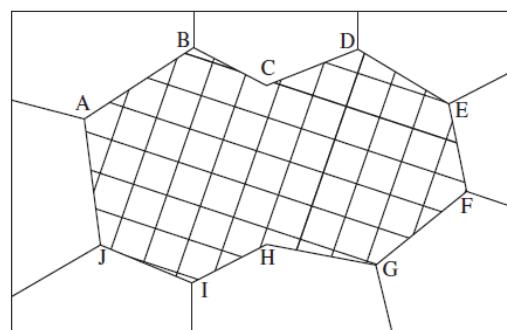
Coalescence:



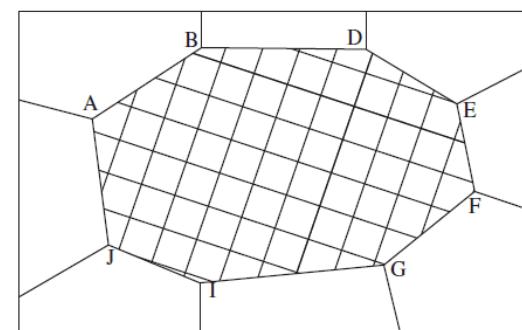
(a)  
The original subgrain structure before coalescence



(b)  
One subgrain is undergoing a rotation



(c)  
The subgrain structure just after coalescence

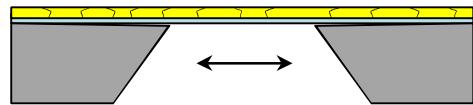


(d)  
The final subgrain structure after some subboundary migration

# In Situ Heating

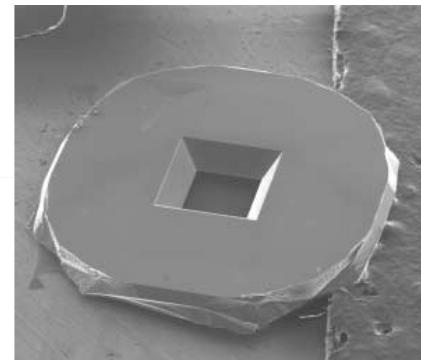
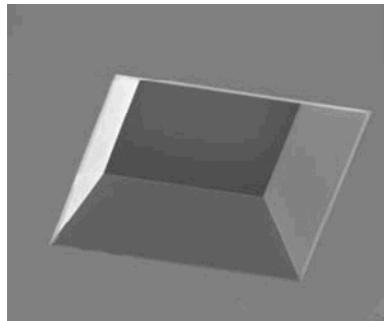
## TEM Study

Au – ZnO Thin Films Deposited Directly on SPI  $\text{Si}_3\text{N}_4$  TEM Membranes



100 nm  
window

50 nm Au + ZnO  
20 nm  $\text{Si}_3\text{N}_4$   
200  $\mu\text{m}$  Si

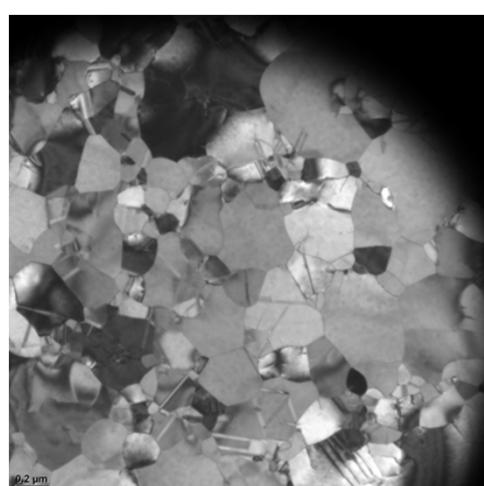
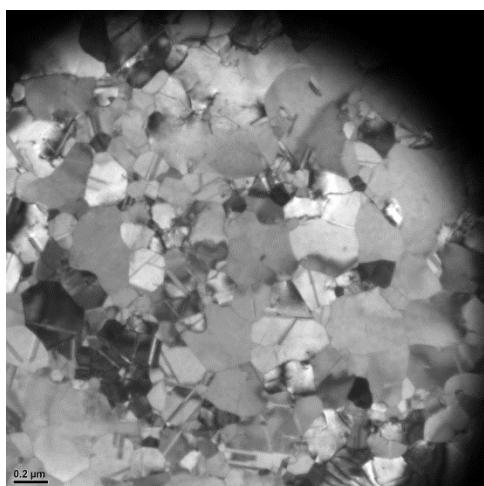
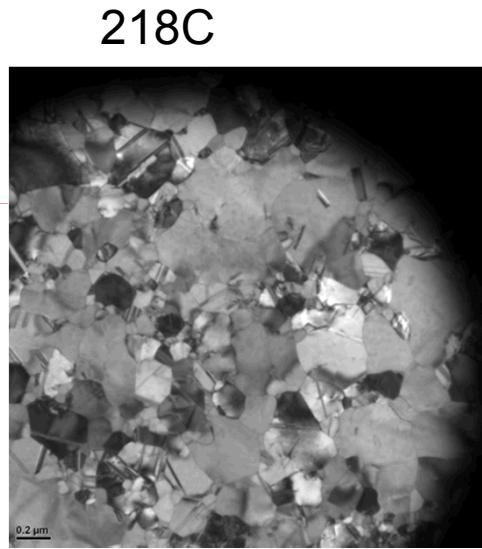
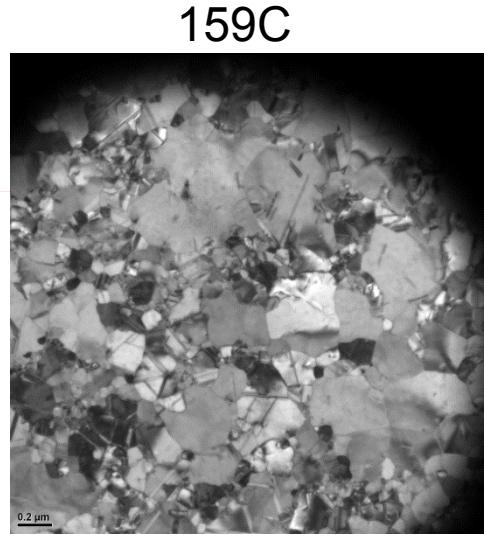
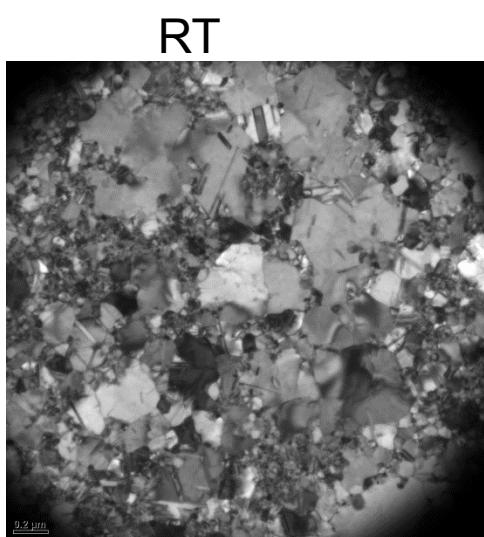


Imaged on Philips CM30  
Transmission Electron  
Microscope in Bright Field

Heating in 10C steps  
from 150C to 350C

# In Situ Heating TEM

Pure Au - 50nm thick



FOV 2.78 μm

As dep. Grain size 50 – 200nm

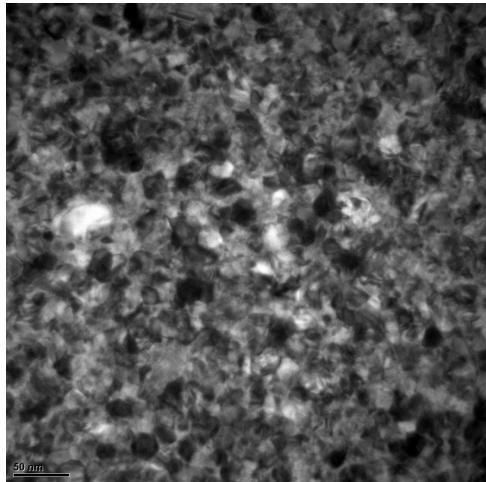
Small grains gone by 159°C

@ 352°C grain size 100 – 400nm

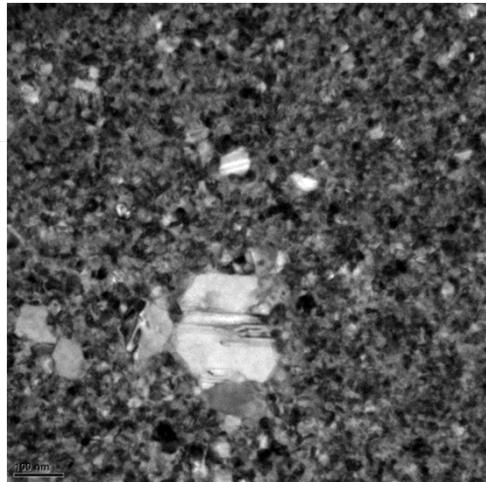
# In Situ Heating TEM

Au – 2% ZnO 50nm thick

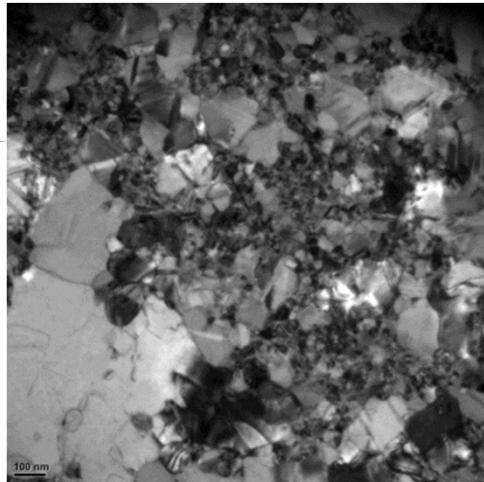
RT



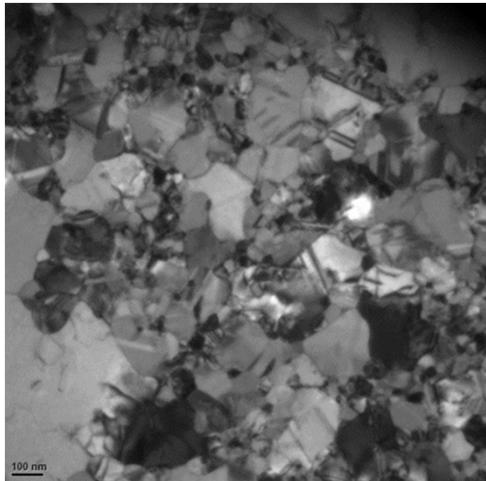
155°C



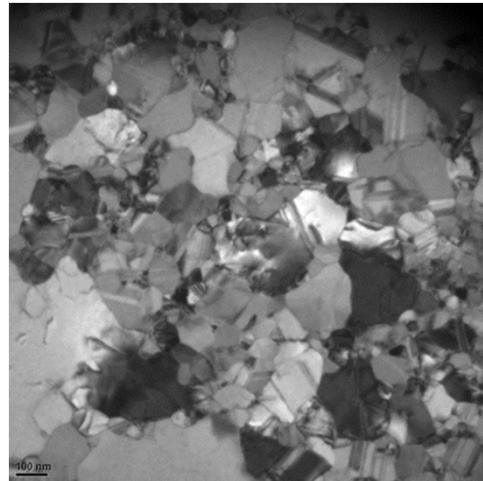
222°C



300°C



347°C



FOV 1.5  $\mu$ m (1<sup>st</sup> image 450nm)

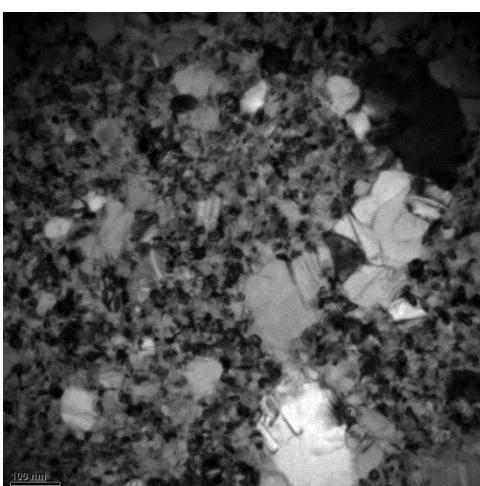
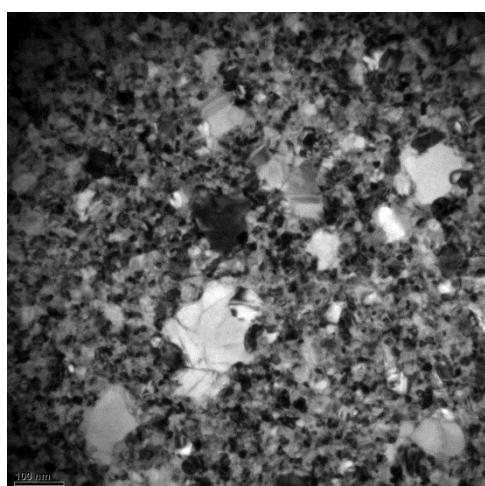
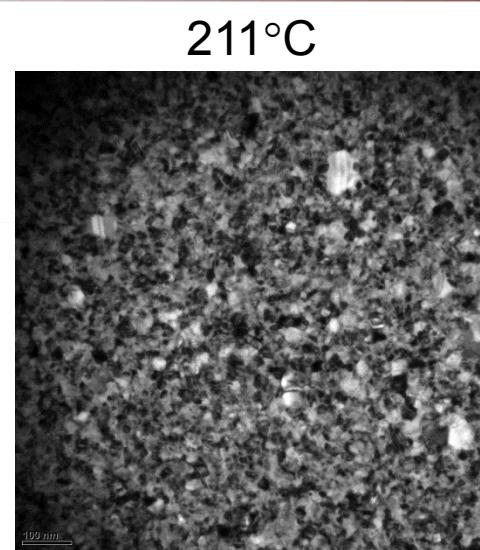
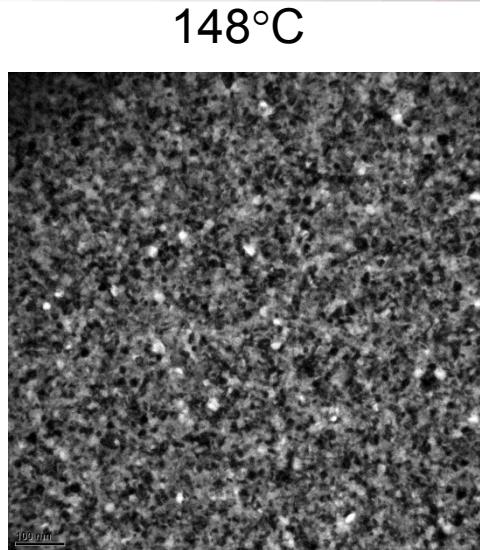
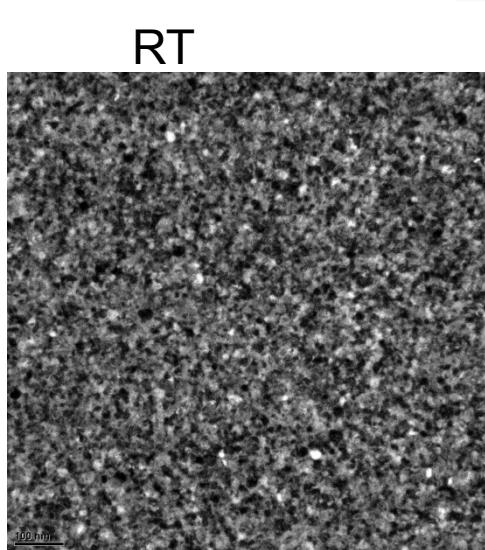
As dep. Grain size 10 – 20nm

Abnormal grain growth  
starts @ 155°C

@ 347°C grain size 50 – 200nm

# In Situ Heating TEM

Au 5% ZnO - 50nm thick



FOV 960nm

As dep. Grain size ~10 nm

minor grain growth starts @ 211°C

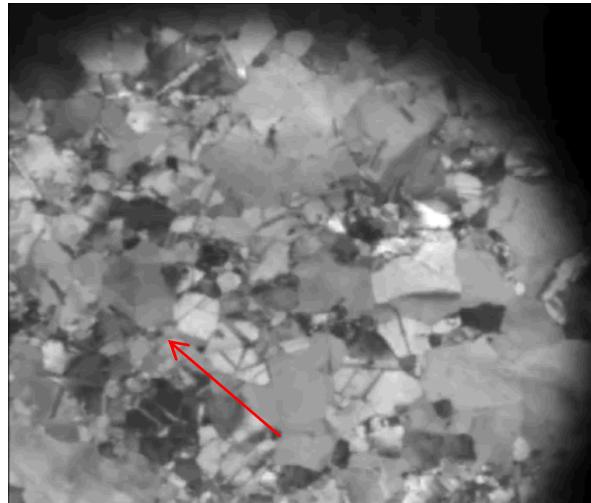
@ 349°C grain size ~10 & 100nm

# Sintering of Pure Au Film

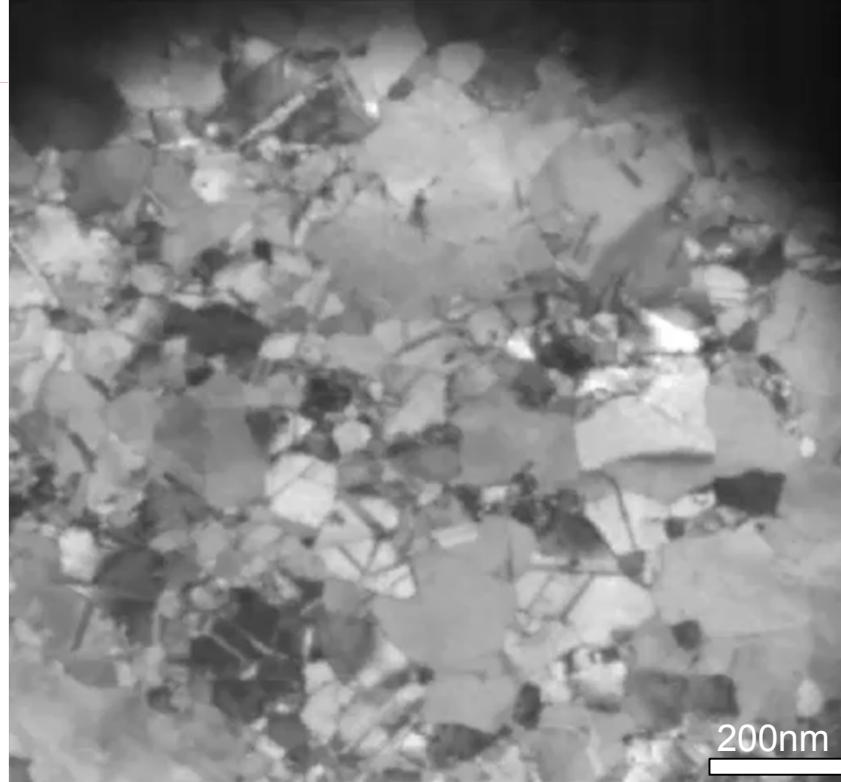


## Sintering movie

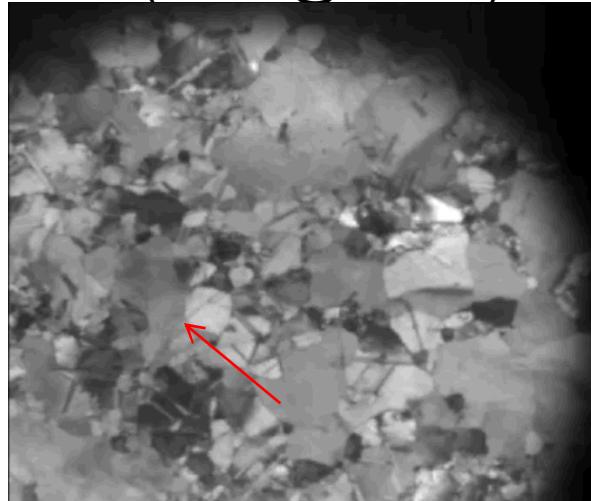
before



Speed in real time



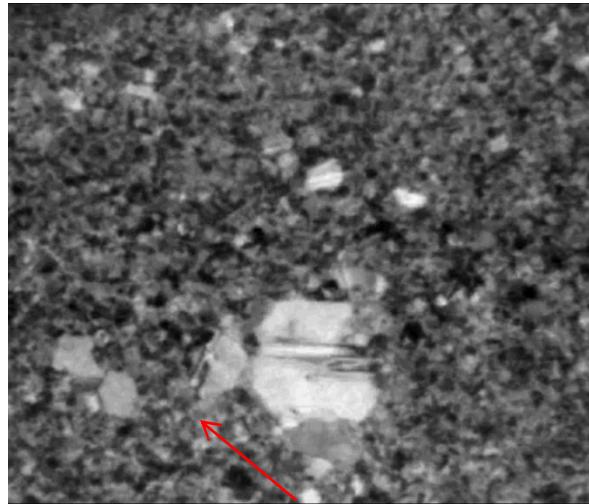
After (2 min @ 180°C)



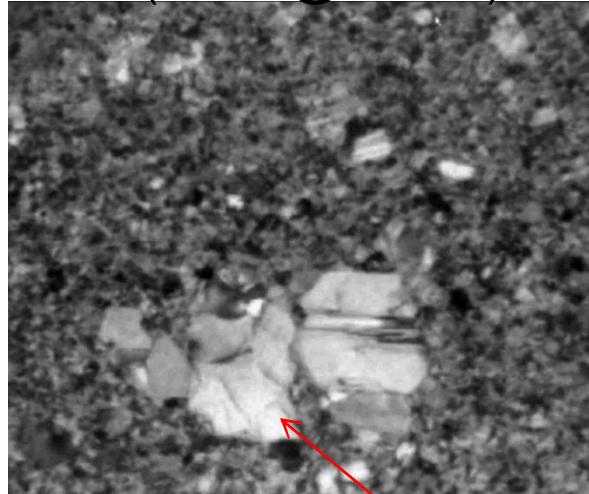
# Sintering of Au-2%ZnO

## Sintering movie

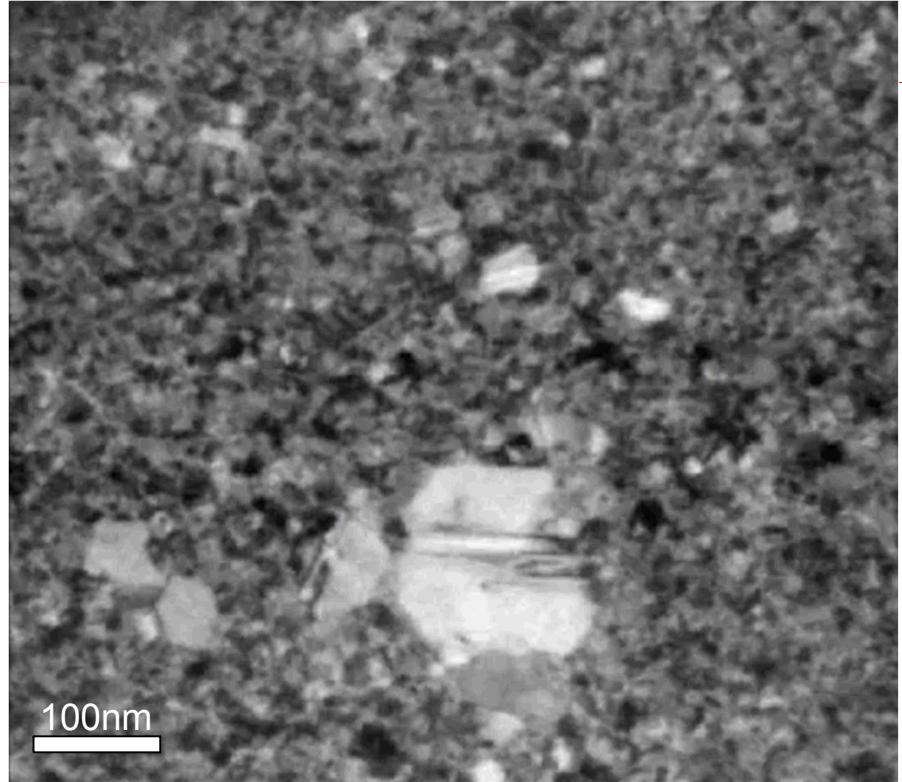
before



After (2 min @ 155°C)



Speed in 16X real time



Abnormal growth is when a few grains grow much larger than the remaining majority

# Conclusions

- A hard gold thin film has been synthesized by co-evaporation comprising nano-crystalline Au pinned by small amounts of ZnO
- Thermal stability of Au-ZnO(2 vol%) is significantly enhanced over pure gold and shows no surface layer formation which would impact electrical contact resistance.
- PVD processes are environmentally friendly compared with electroplating.

**E3-1-9 Thursday 10:40 AM – Nic Argibay**

## Acknowledgements:

Rachel Schoeppner (WSU & Purdue),  
Paul Kotula, Joe Michael for Microscopy  
Robert Reeves for DSC  
Carl Smith for thin film deposition.