



Cell Level Abuse Tests Of HEV\PEV Li-Ion Chemistries

SAND2008-6034C

Advanced Technology Development

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**1st International Conference on Advanced Lithium
Batteries for Automobile Applications
Argonne National Laboratory
September 15-17, 2008**

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Li-Ion Safety Issues

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Abuse Tolerance is a Critical Requirement for Li-Ion Battery Development and Application

There are two main modes for failure:

➤ Field Failures

- Failures during normal use
 - Rare (less than one in a million)
 - Not predictable
 - Abrupt, high-rate reactions often resulting in cell disassembly and fire
 - Cause usually attributed to manufacturing defects

➤ Abuse Failures

- Failures due to external factors exceeding normal design limits
 - Mechanical (crush, nail penetration, shock)
 - Thermal (over temperature)
 - Electrical (short circuit, overcharge)



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Cell Chemistries Evaluated in DOE Advanced Technology Development (ATD) Program

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➤ Cathode materials:

- ☐ LiCoO_2
- ☐ $\text{LiNi}_{0.85}\text{Co}_{0.15}\text{O}_2$ (Gen1)
- ☐ $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2$ (Gen2);
- ☐ $\text{LiNi}_{1/3}\text{Co}_{1/3}\text{Mn}_{1/3}\text{O}_2$; $\text{Li}_{1.1}(\text{Ni}_{1/3}\text{Co}_{1/3}\text{Mn}_{1/3})_{0.9}\text{O}_2$ (Gen3)
- ☐ LiMn_2O_4 (Spinel)
- ☐ LiFePO_4

➤ Anode materials:

- ☐ MCMB (Gen1 and 3)
- ☐ MAG10 (Gen2)
- ☐ GDR (Gen2)

➤ Electrolytes/salts:

- ☐ EC:EMC (3:7) 1.2M LiPF_6
- ☐ EC:PC:DMC (1:1:3) 1.2M LiPF_6
- ☐ LiBOB, LiBETI

➤ Additives:

- ☐ SEI enhancer – Vinyl ethylene carbonate (VEC); Vinylene carbonate (VC)
- ☐ Flame retardants– e.g. Phosphazene-based “Phoslyte”
from Bridgestone; phosphate TPP; ...



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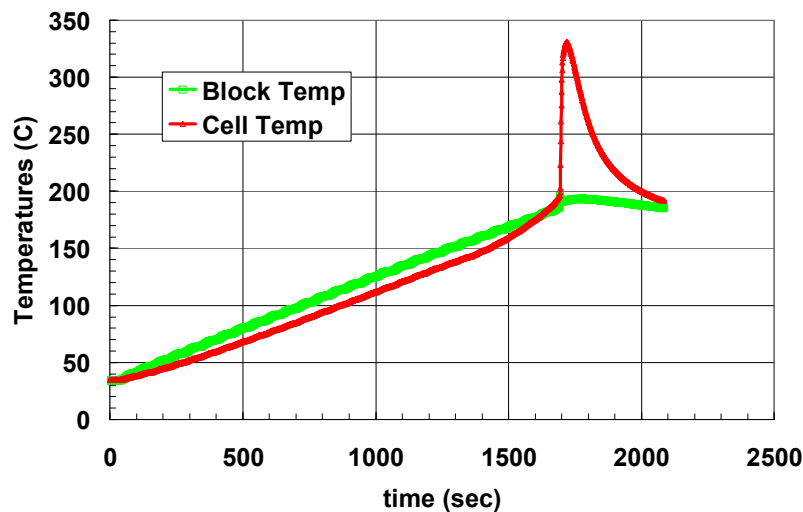
Thermal Runaway Can Be Grouped Into Three Major Temperature Regimes

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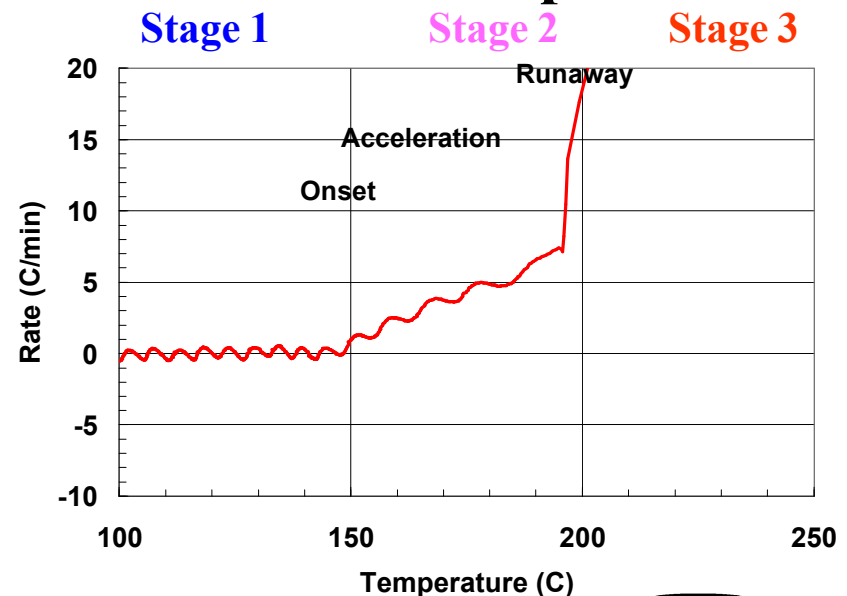
Thermal Ramp Response (100% SOC)

- Stage 1: Room Temperature to 150°C – Onset of thermal runaway
- Stage 2: 150°C - 180°C – Venting and accelerated heating (smoke)
- Stage 3: 180°C and above – Explosive decomposition (flame)

Ramp Temperatures



Differential Temperature Rate



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Thermal Runaway Characterization by Accelerating Rate Calorimetry (ARC)

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ARC Apparatus for Measurement
of Full Cells with Gas Capture



**Top
Heater**



**Clam Shell
Blast Shield**

**Gas Tight
Cell Holder**

Heating Unit

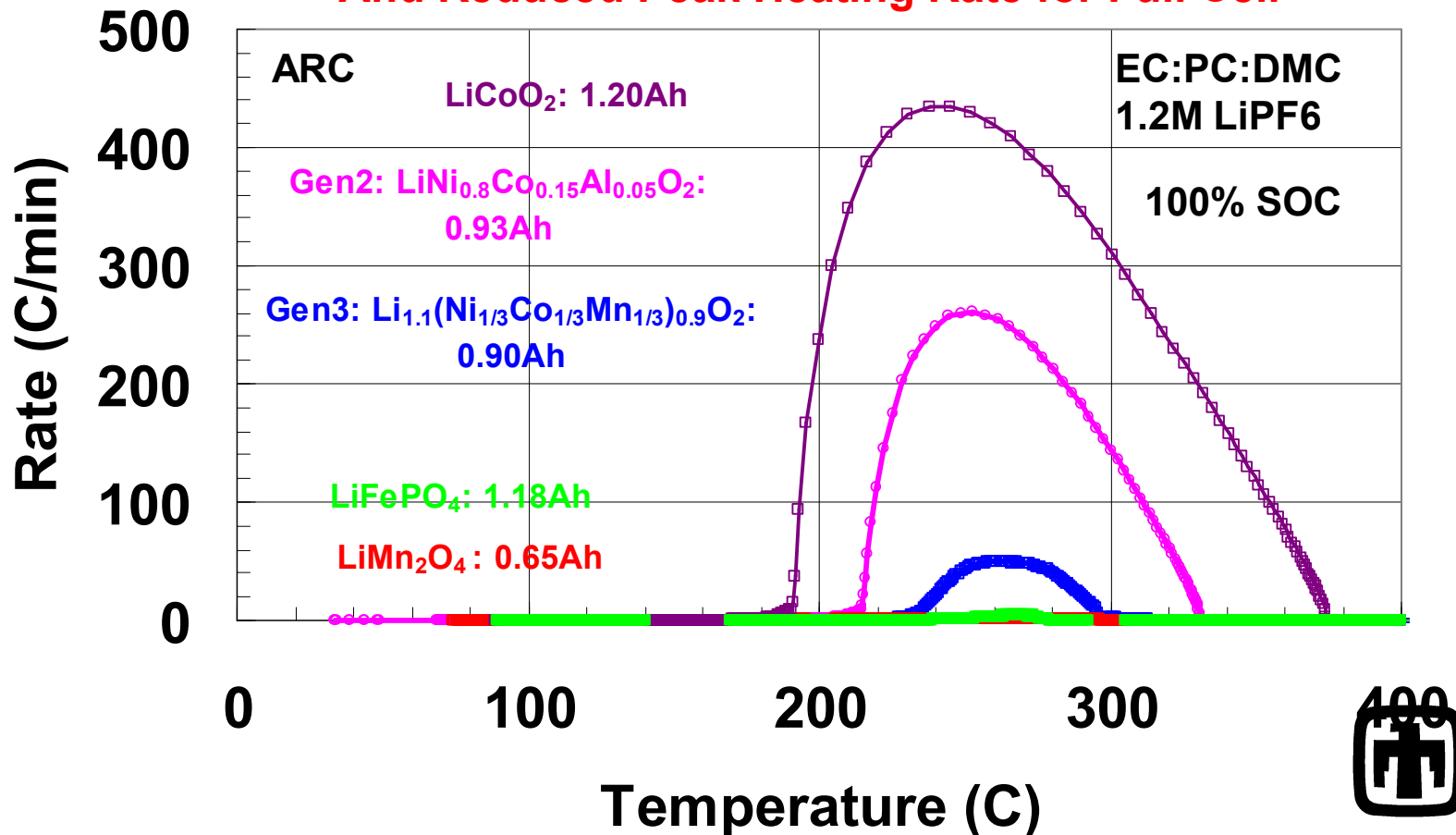


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Thermal Runaway Cathode Comparisons

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Improved Cathode Stability Results in
Increased Thermal Runaway Temperature
And Reduced Peak Heating Rate for Full Cell

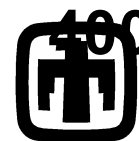
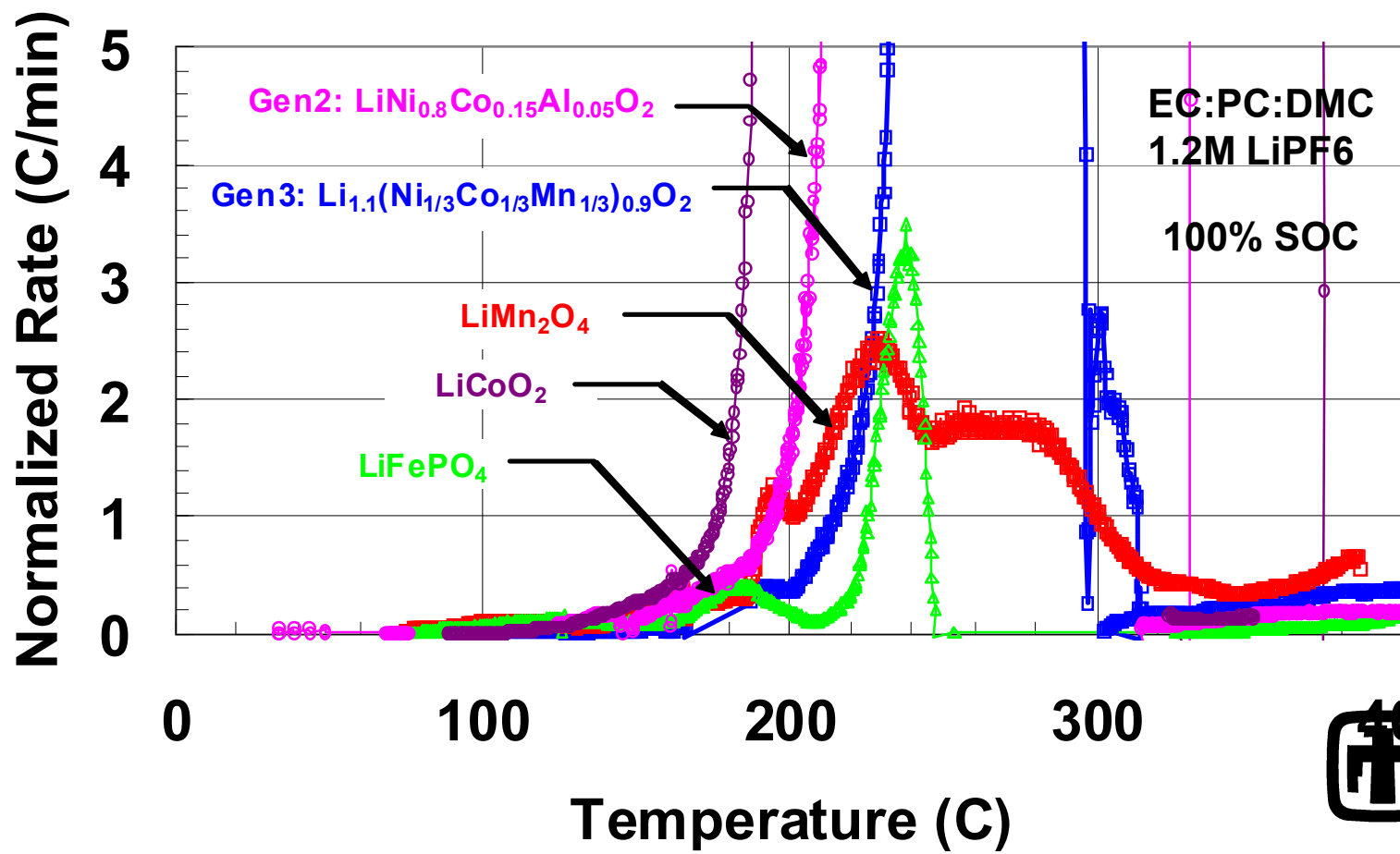


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Thermal Runaway Cathode Comparisons (Expanded View)

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Initial Onset Reactions Dominated by Anode

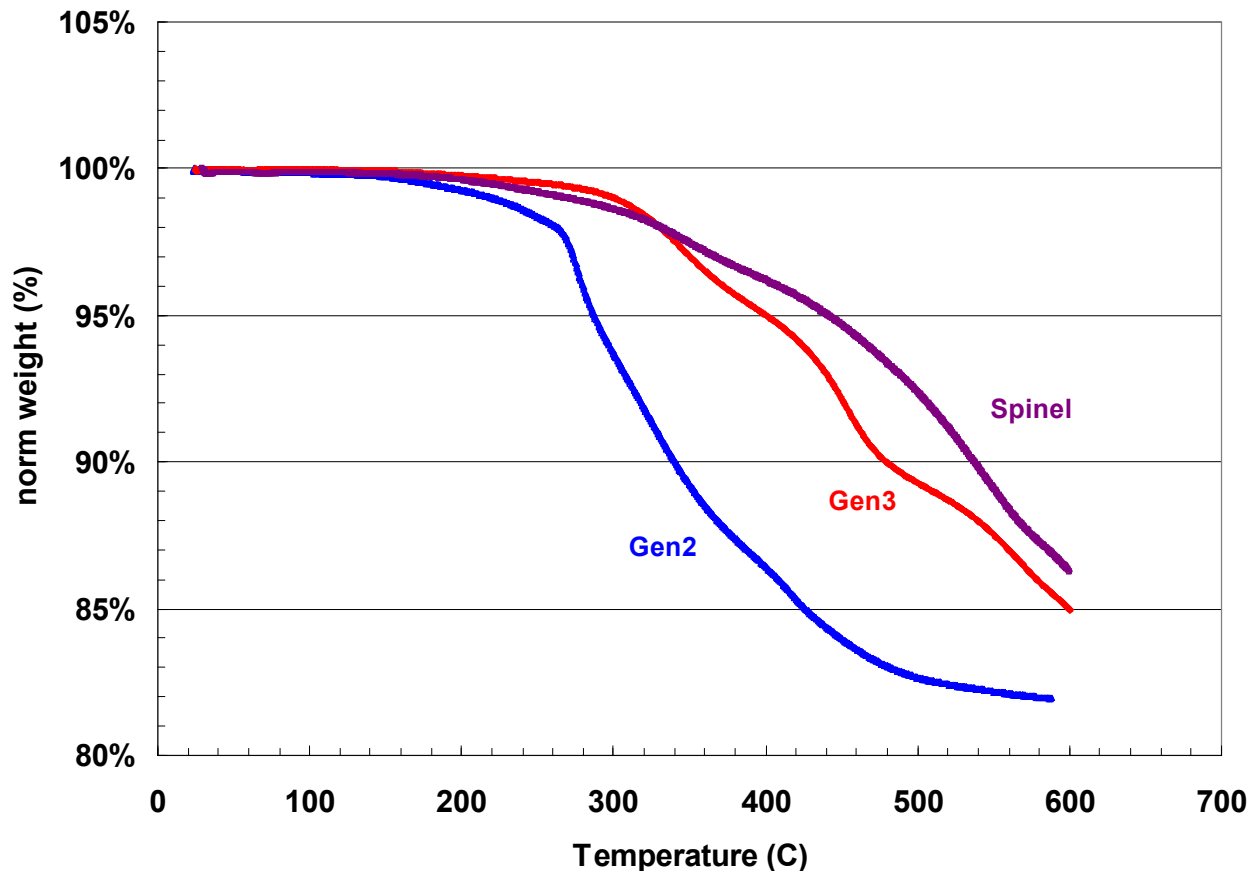


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TGA Profiles

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Mass Loss From Oxygen Release
From Decomposing Cathodes



Oxygen Generation
Correlates With
ARC Reaction Rate
and Enthalpy

Gen2 Shows Highest
Mass Loss and
Greatest Reaction Rate

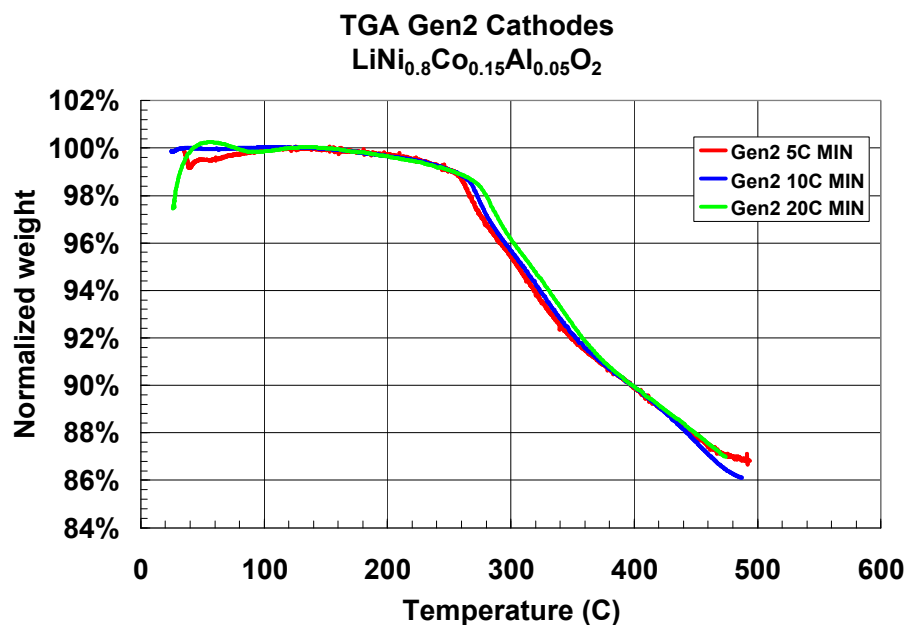


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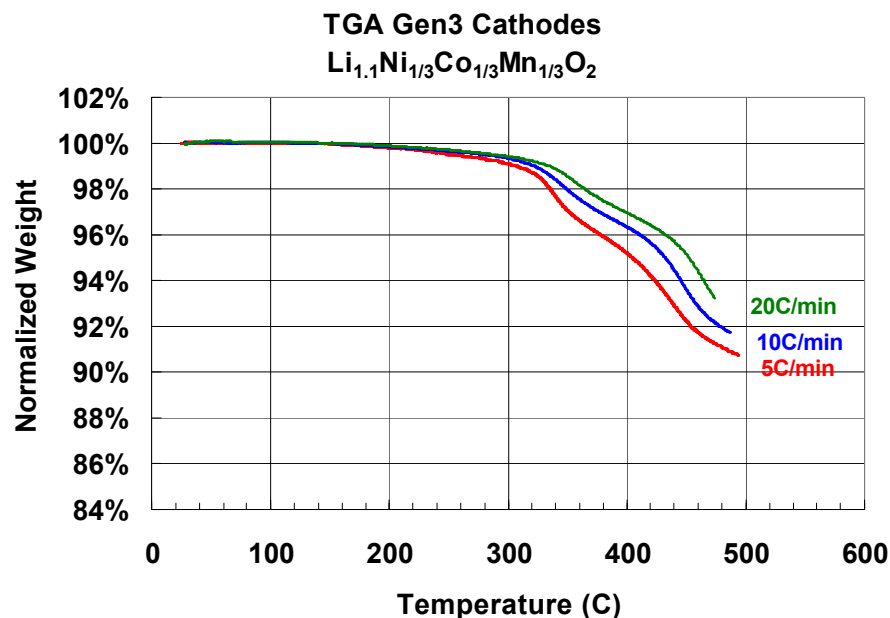
Reaction Kinetics Also Determine Peak Runaway Response

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TGA Curves Show Fast Gen2 Decomposition Kinetics



TGA Curves Show Slower Gen3 Decomposition Kinetics



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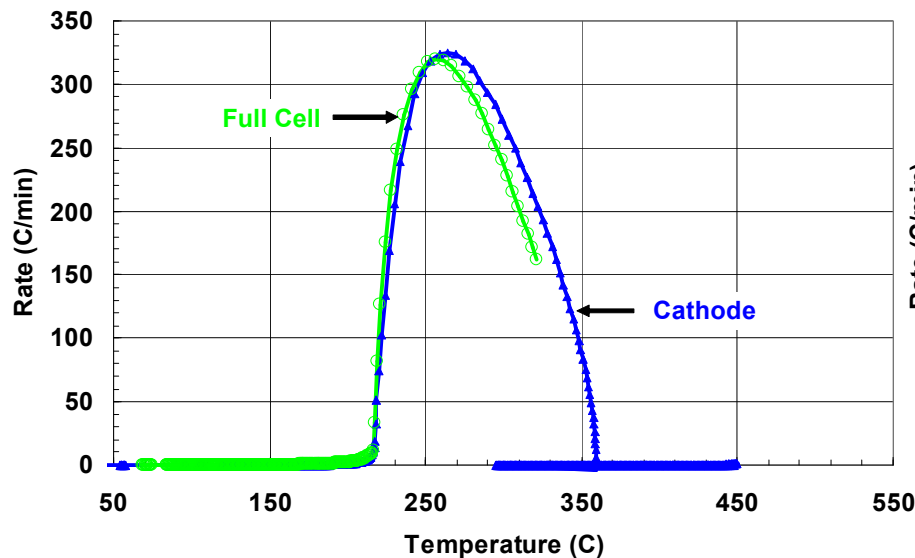
Comparison of Gen2 and Gen3 Cathodes Full Cell ARC Thermal Runaway Profiles

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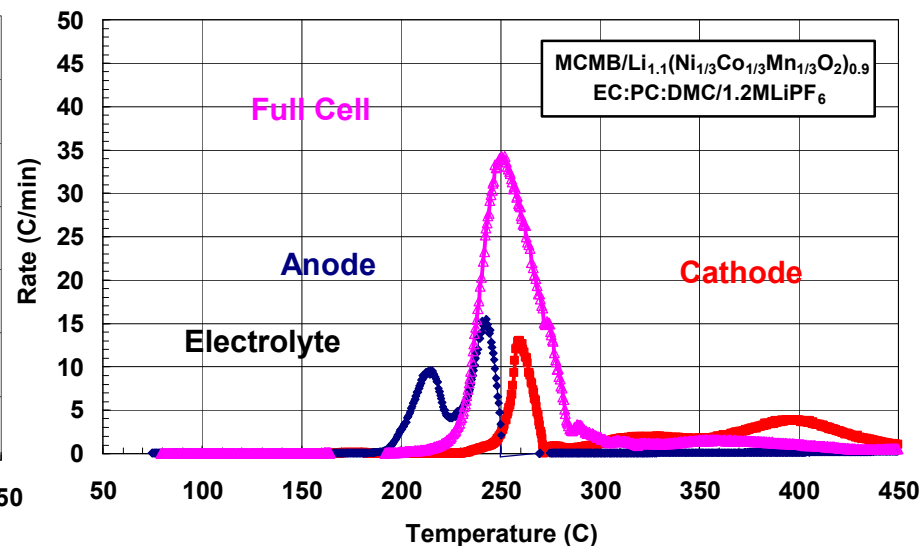
Peak Thermal Runaway Profile Determined By Cathode Reactions

$\text{LiNi}_{0.5}\text{Co}_{0.45}\text{Al}_{0.05}\text{O}_2$ (Gen2)

SNL Gen2 Cathode in 18650 Can 100%SOC



$\text{Li}_{1.1}(\text{Ni}_{1/3}\text{Co}_{1/3}\text{Mn}_{1/3})_{0.9}\text{O}_2$ (Gen3)



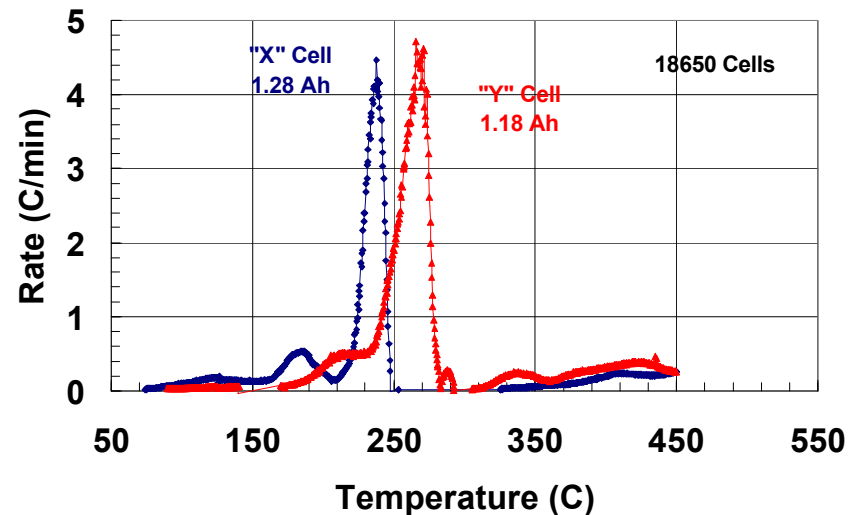
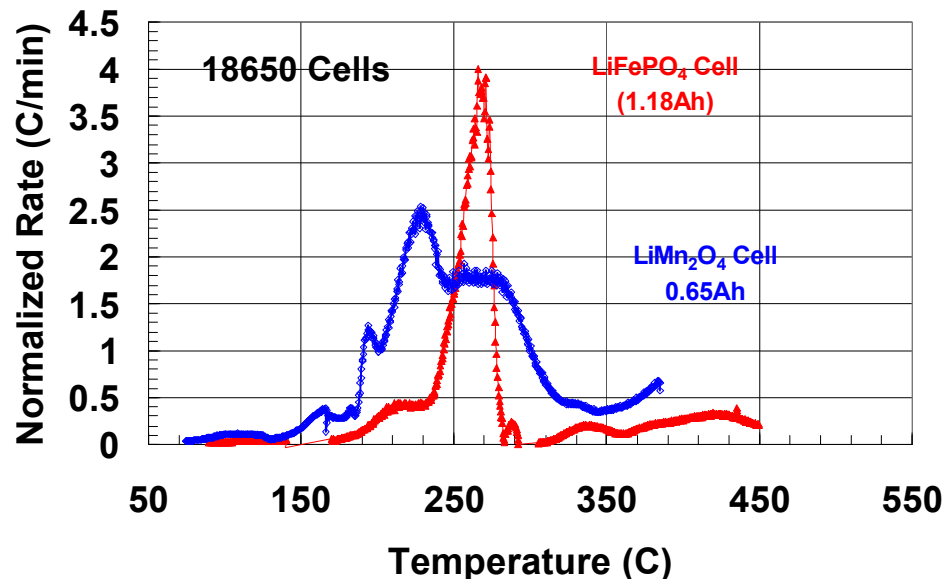
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LiFePO₄ Cells Show Lower Reaction Kinetics and Reaction Enthalpy Compared to LiMn₂O₄ Cells

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LiFePO₄ Does Not Release Oxygen and Shows the Lowest Thermal Reactions

Cell from Two Manufacturers (X and Y) with LiFePO₄ Cathodes Both Show Similar Responses

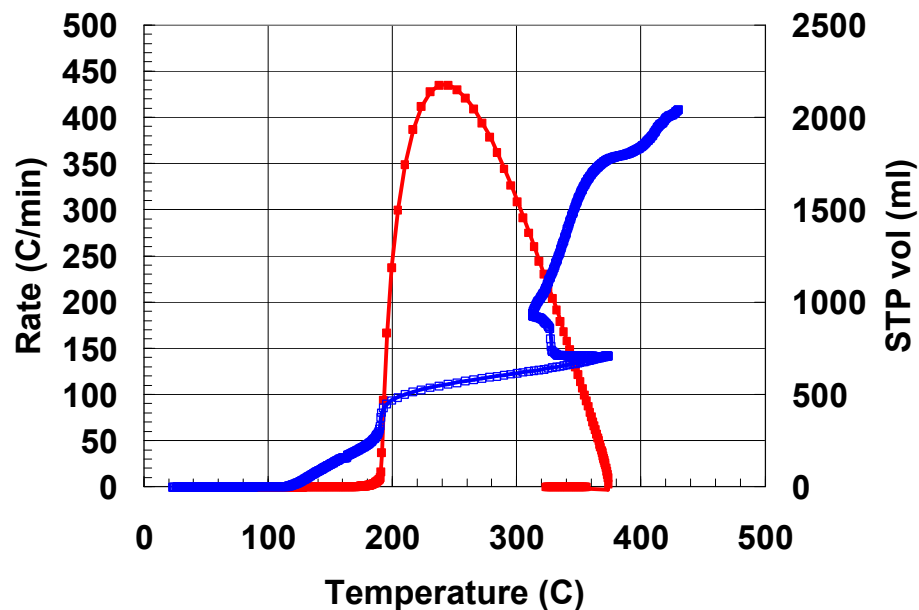
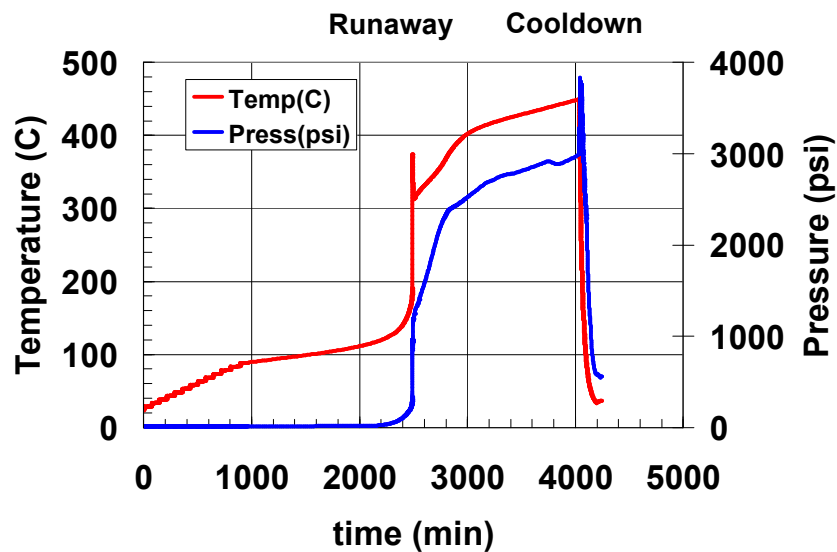


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Gas Evolution is a Critical Property Affecting Safety

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ARC



1.2 Ah MCMB/LiCoO₂ EC:PC:DMC/1.2M LiPF₆



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Gas Generation During Thermal Runaway

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Thermal Ramp Test to Runaway ($\sim 200^{\circ}\text{C}$)



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Gas Evolution and Composition

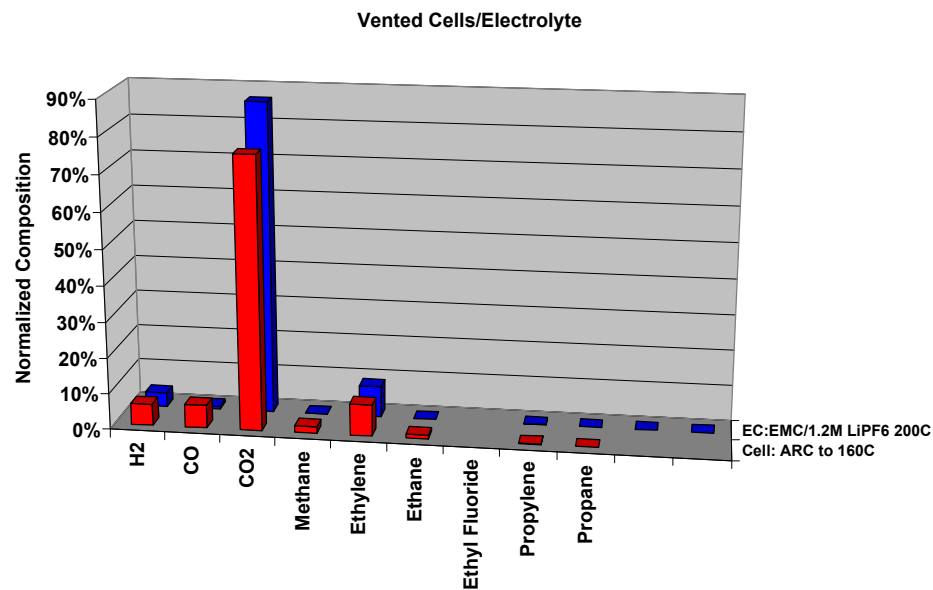
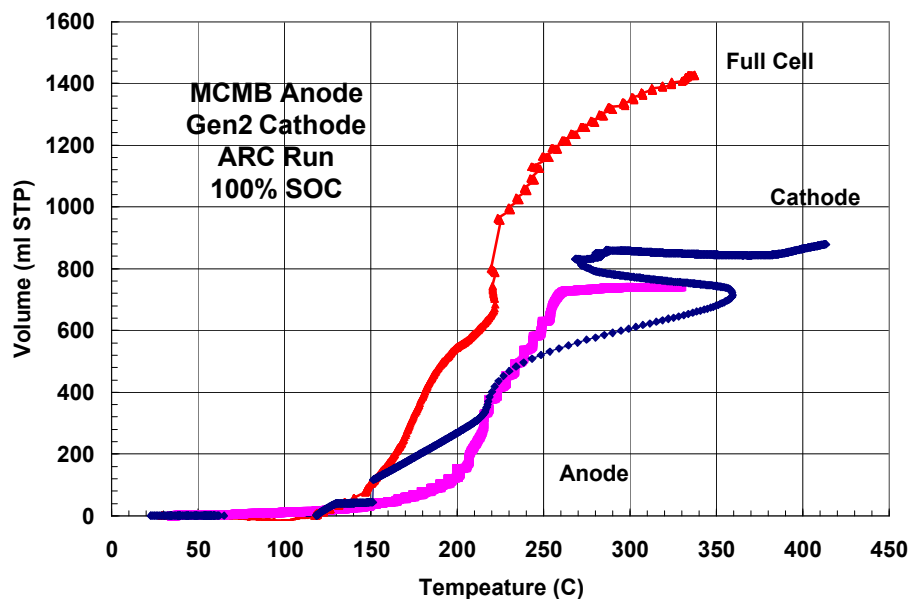
Cathode and Anode Removed from Full Cell at 100%SOC
Resealed in 18650 Cans with Electrolyte

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Onset of Gas Generation at
Cathode Around 150°C

Equal Gas Volumes for Reactions at
Cathode and Anode

Gas Composition Largely CO₂
Both for Full Cell and
Electrolyte Decomposition

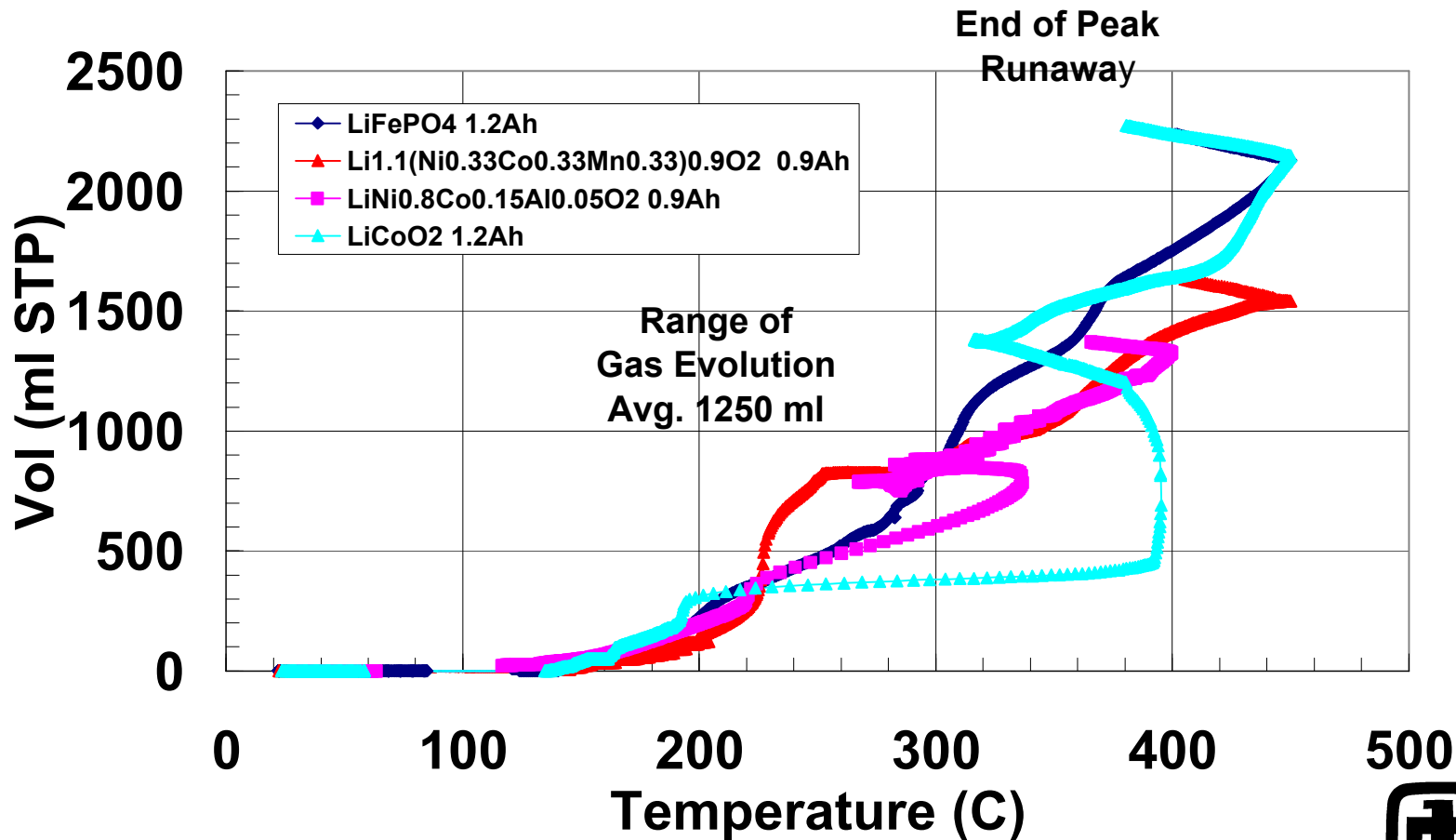


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Gas Evolution for Different Cathode Chemistries

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Gas Evolution Largely Determined by Quantity of Electrolyte



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Thermal Ramp Apparatus

Ramp to runaway in air with external ignition

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65mm

18mm

Ramp at 6 °C/min

190 °C

Thermal Block

Cell TC
Block TC

Cell Voltage
Leads

Heater Cartridge

Insulated
Cell

Copper
Block



Heat Block with External Ignition Sources
Cell has vented and is about to enter
explosive decomposition stage.

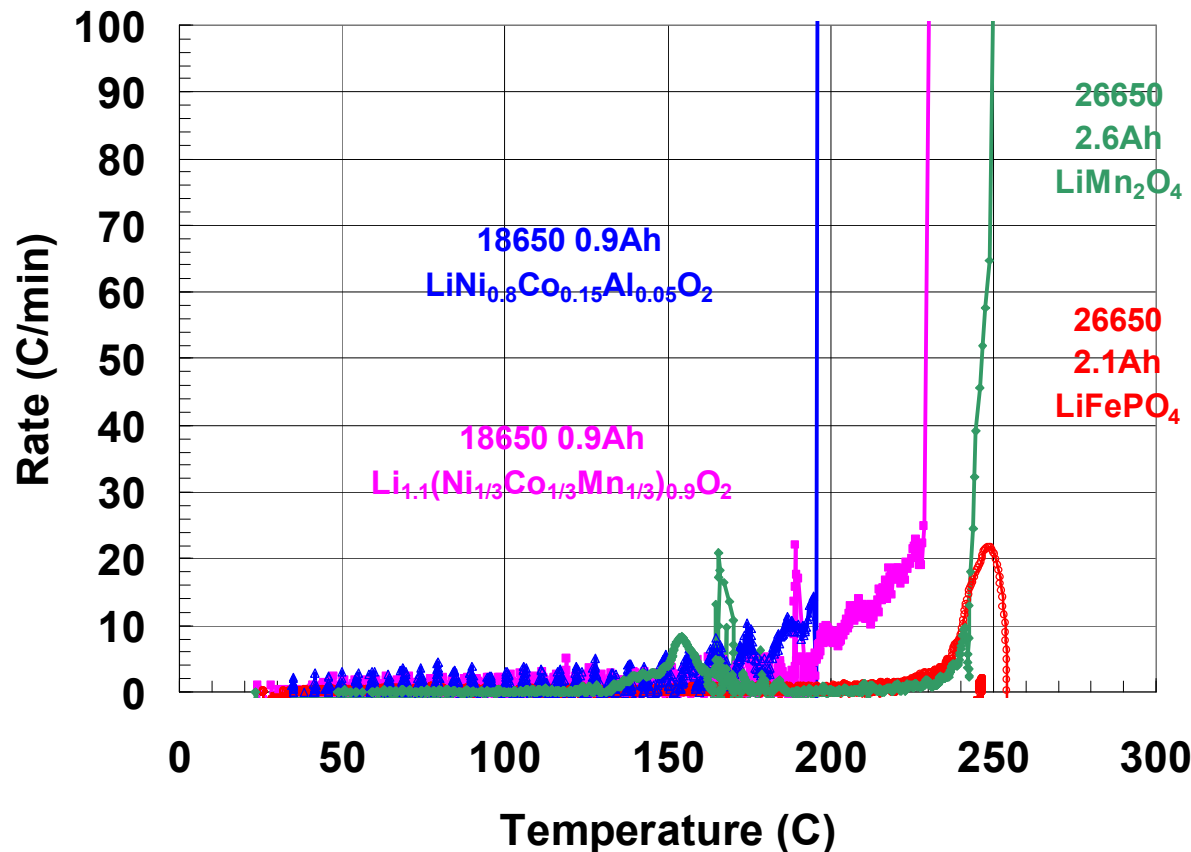


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Greater Abuse Tolerance Shown for New Chemistries with Reduced or No Cathode Oxygen Release

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Anode Reactions Still Contribute to Runaway



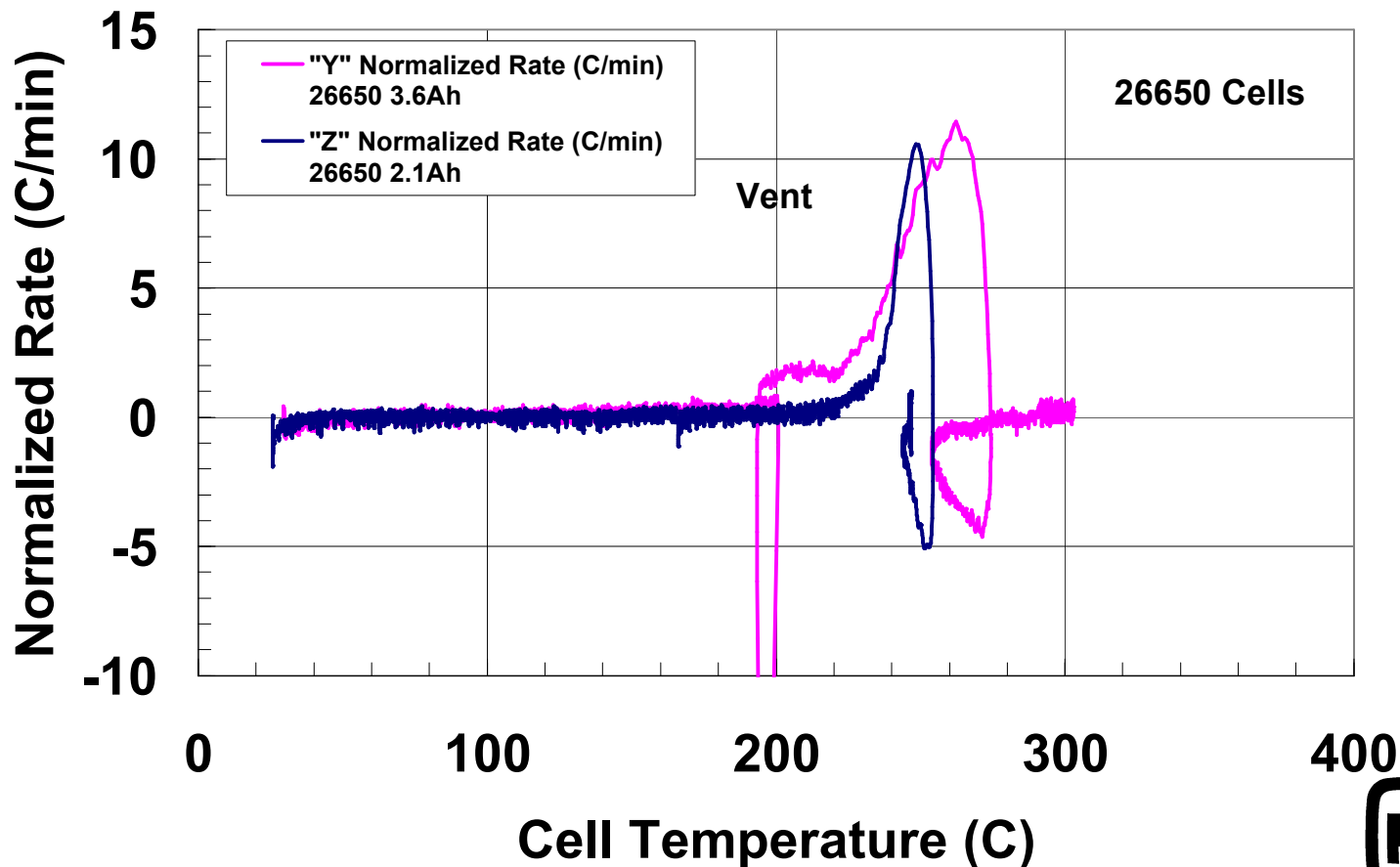
LiFePO_4 olivine cathodes show the greatest reduction in heating rate and increased onset temperature for runaway



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Comparison of Two Commercial LiFePO_4 Cells

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Movie Clips of Cell Thermal Runaway Showing Improved Response of LiFePO_4

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50% Li/Ni/Co/Mn and
50% LiMn_2O_4 Cell



A123 LiFePO_4 Cell



**Explosive Combustion of
Vent Gases (mostly solvent vapors)**

100% SOC

26650 Commercial Cells



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II. Overcharge Response

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- **Overcharge is one of the most energetic abuse conditions**
 - Highly reactive, unstable cathode
 - Highly lithiated anode
- **High levels of heat generation**
 - Separator shutdown and possible internal short
 - Initiation of thermal decomposition runaway
- **Flammable gas generation**
 - Hydrogen
 - Venting of solvent vapors



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Overcharge Gas Generation

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We have previously shown that explosive gases are generated during overcharge: Hydrogen and flammable solvent vapors



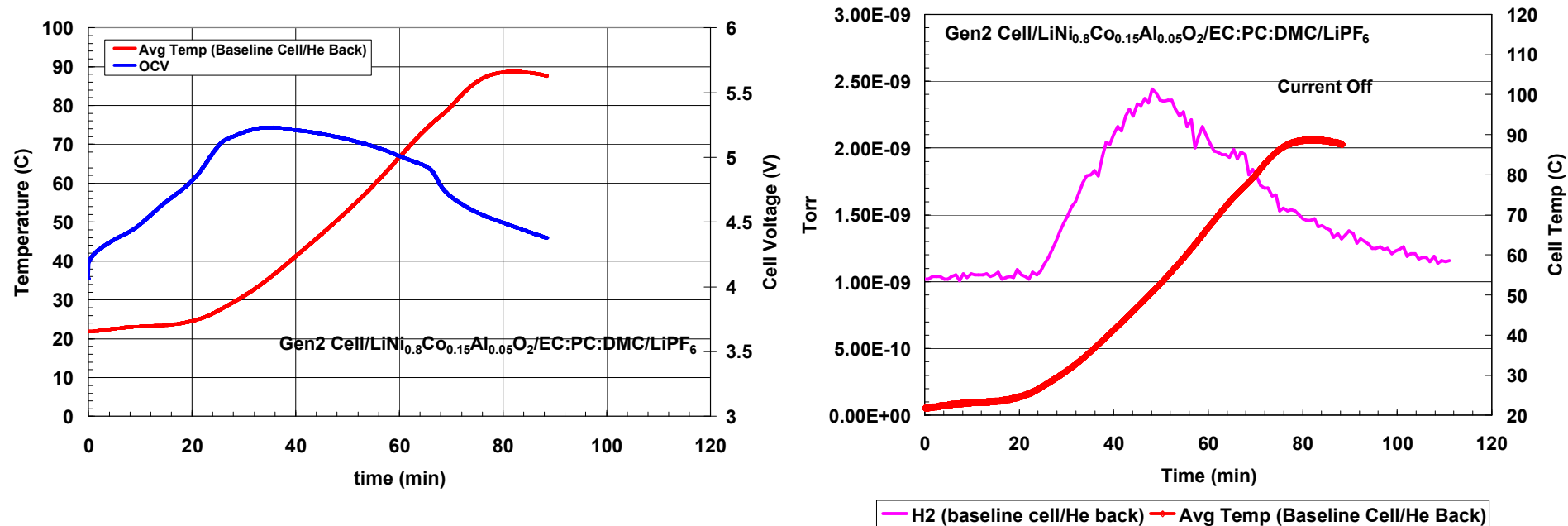
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Sequence of Gas Generation During Overcharge

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- 18650 Gen2 Cell/ $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2/\text{EC:PC:DMC}/\text{LiPF}_6$
- Helium background to keep temperature below separator shutdown
- Cell can cut open to allow gas escape during run
- Real-Time Mass Spectrometry Analysis

Hydrogen Generation at Onset of Heat Generation



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Abuse Response Improved with Electrolyte Additives

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- Additives to improve:
 - SEI
 - VEC, VC
 - Electrolyte stability
 - LiBOB, LiBETI
 - Overcharge
 - **Shuttle Additives**
 - Flammability
 - Phosphazenes
 - Phosphates



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Overcharge is One the Most Abusive Conditions for Li-ion Cells:

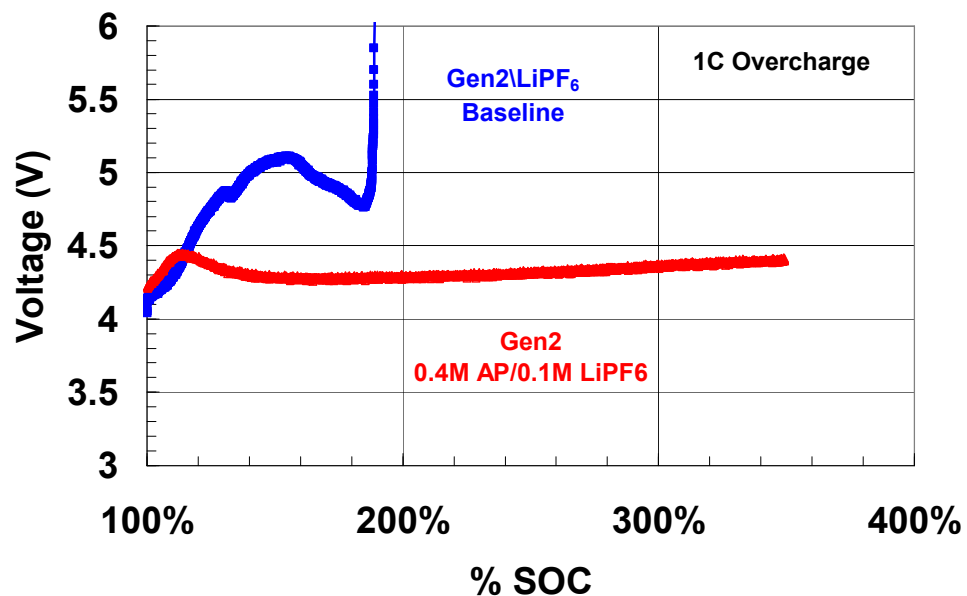
Improved Response with Overcharge Additive

0.4M $\text{Li}_2\text{B}_{12}\text{F}_9\text{H}_3$ and 0.1M LiPF_6 in EC/DEC (3:7)

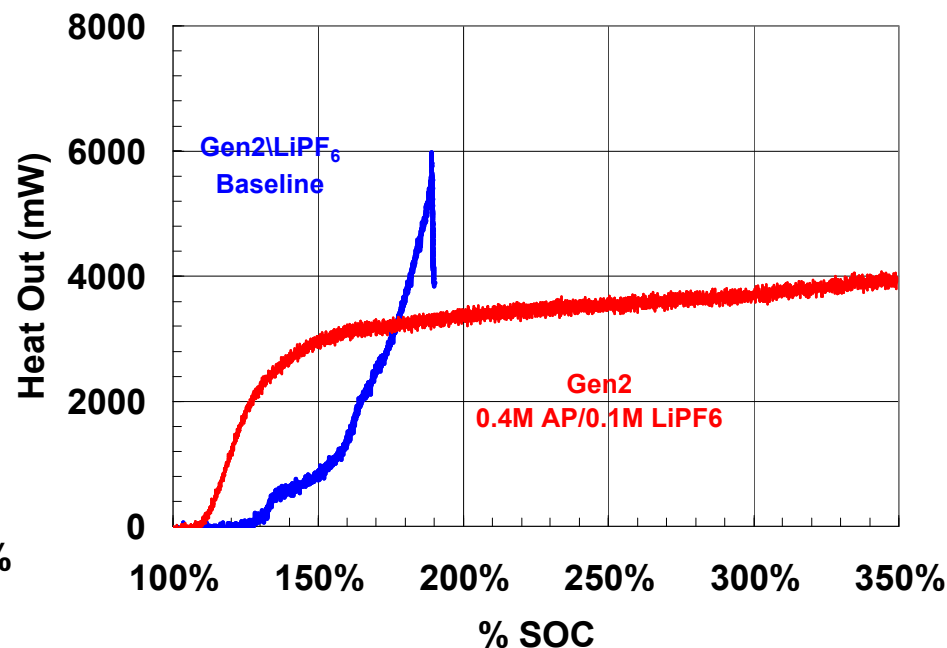
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Shuttle Mechanism Limits Cell Voltage and Peak Heating Rate During Overcharge

Separator Shutdown



Separator Shutdown



Material developed by Air Products and Chemicals, Inc.



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Thermal Abuse Observations

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- **Steady Improvements in Abuse Tolerance Have Been Achieved**
- **Flammable Electrolyte and Gas Generation Are Still an Issue**

➤ *Increased thermal stability has been demonstrated with more*

stable cathodes ($\text{Li}_{1.1}(\text{Ni}_{1/3}\text{Co}_{1/3}\text{Mn}_{1/3})_{0.9}\text{O}_2$; LiMn_2O_4 Spinel and LiFePO_4)

– Improved stability results from decreased oxygen generation

➤ *Anode reactions are still important to provide better abuse tolerance*

➤ *Some additives have shown improvements in thermal and overcharge abuse response*



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Acknowledgements

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Thanks to Dave Johnson, Craig Carmignani, Jill Langendorf, Lorie Davis, Mike Russell and Mani Nagasubramanian for performance of this work.

This work was performed under the auspices of DOE FreedomCAR & Vehicle Technologies Office through the Advanced Technology Development (ATD) High Power Battery Development Program.