

Oxygen Firing of Chem-Prep PZT 95/5

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Background

Extrinsic pore formers are added:

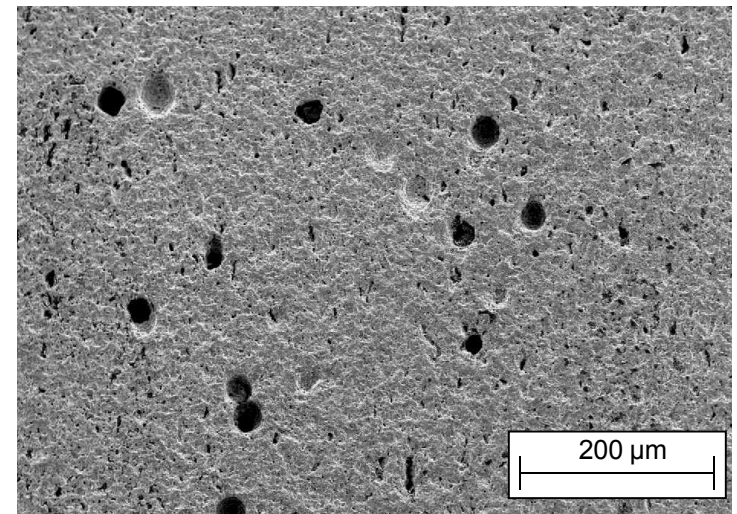
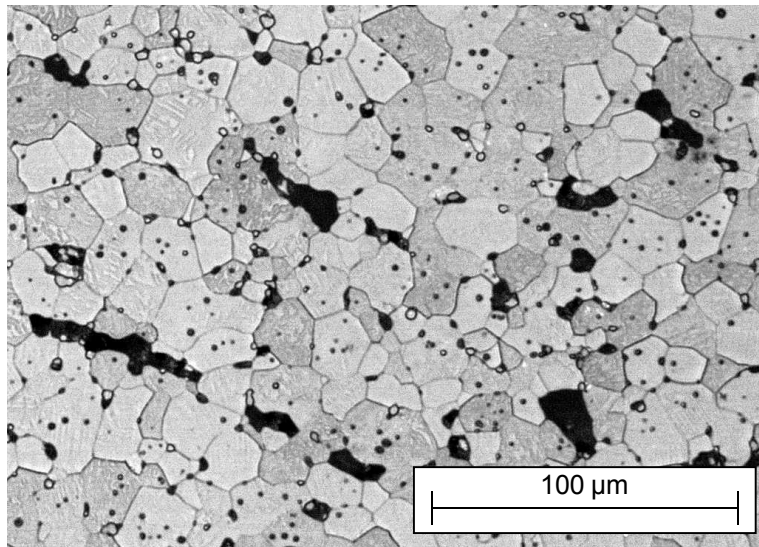
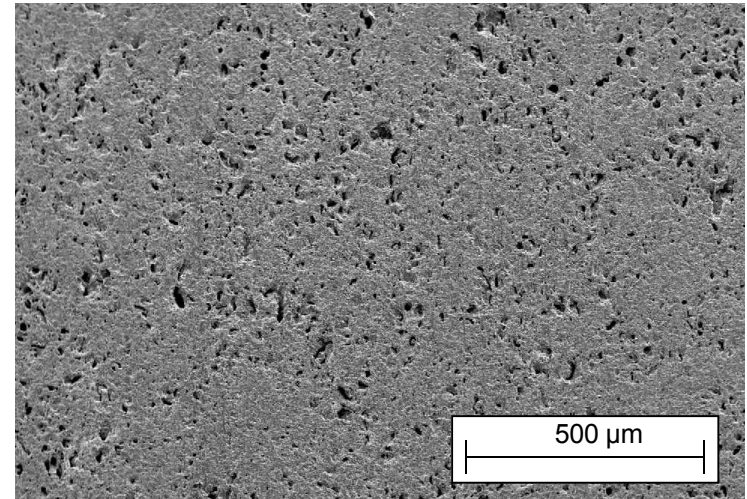
- Control (limit) density
- FE properties are effected by pore former type

PZT also contains Intrinsic porosity

- Intergranular
- Intragranular
- Effect of Intrinsic porosity is not understood

Oxygen firing promotes density in other PZT compositions

- Increased grain boundary and lattice diffusion rates





Sample Fabrication, Testing, and Analyses

Chem-Prep PZT 95/5:

- Metal acetate/alkoxide solution process*
- Avg. particle size: 5 μm , typ.
- Particle size range: $<0.1 \mu\text{m}$ to 40 μm
- Surface area: 1.3 m^2/g , typ.

Powder Granulation and Compaction :

- 2 wt.% acrylic binder (Glascot HA4, Ciba Specialty Chemicals)
- Twin-shell blender
- Die pressed at 14,000 psi, Green density: 4.4 g/cm^3 , typ.

Bisque and High Firing:

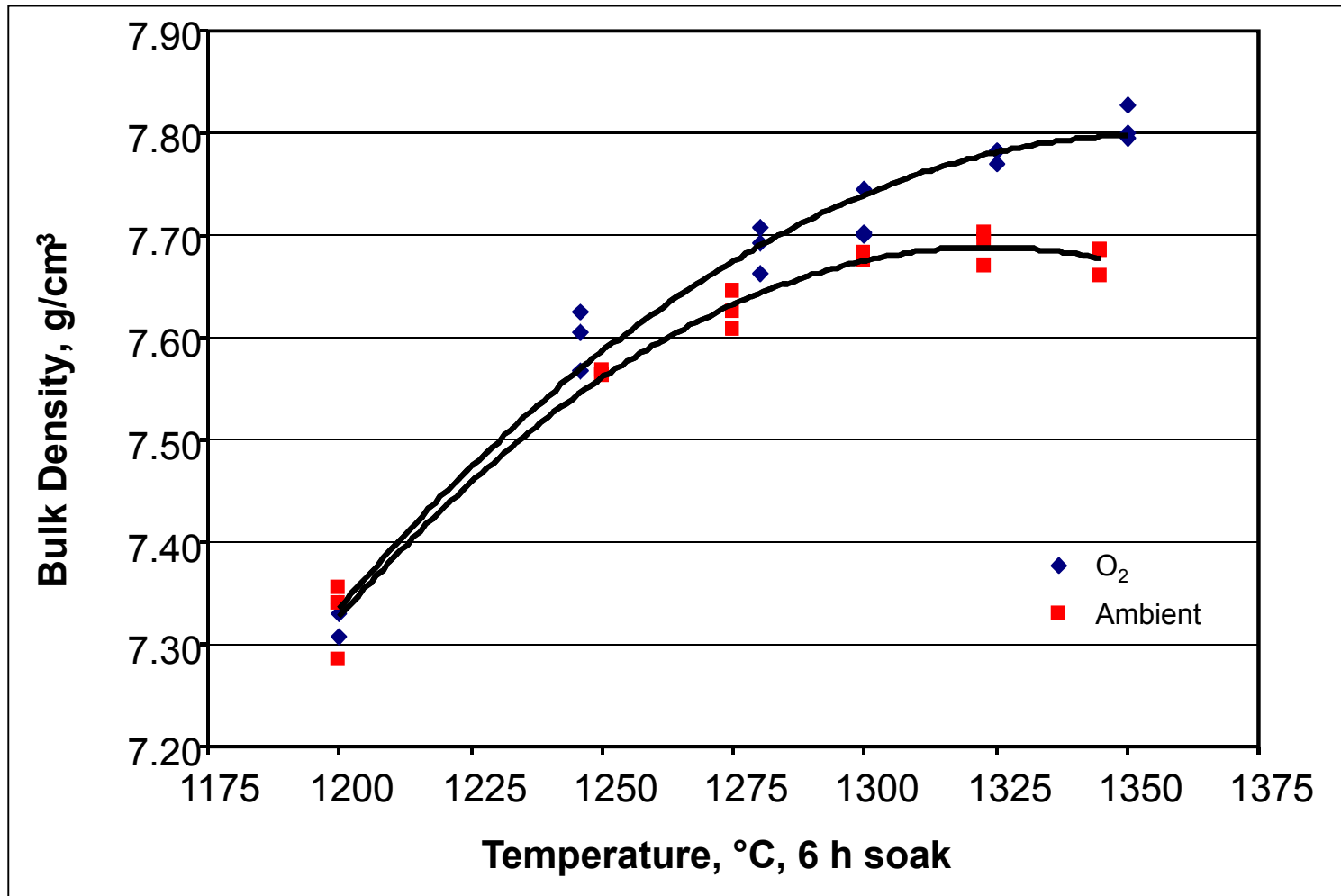
- Bisque at 750°C/4 hr soak, flowing air, density: 4.35 g/cm^3 , typ.
- Tube furnace, 1200 - 1350°C, with soak times up to 12 h.
- Flowing O_2 or ambient environments
- Double alumina crucible system, Wt. Loss range: 0.2 – 1.2 wt.%.

Testing and Analyses:

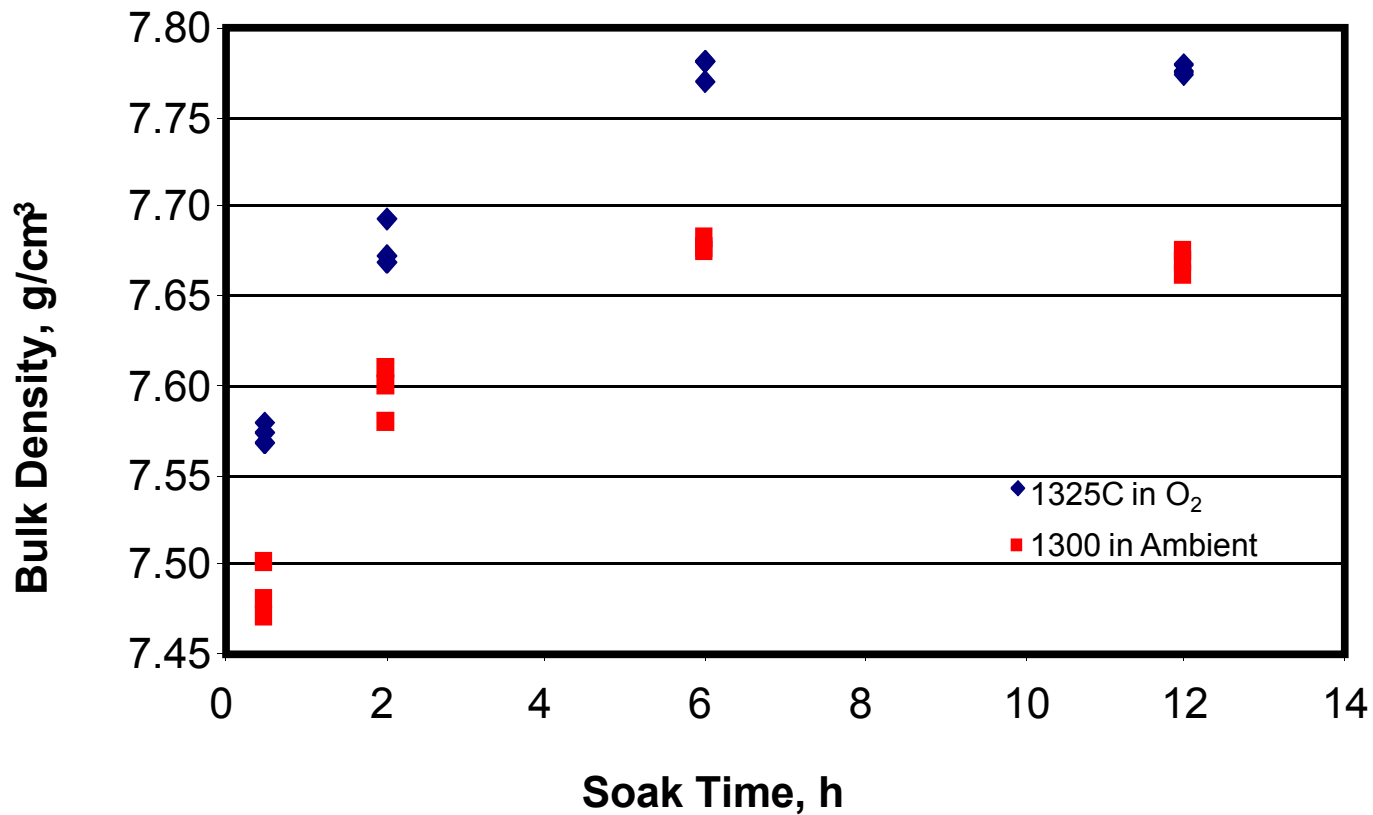
- Bulk Density (immersion)
- Ag Metallize, and polarize
- Ferroelectric hysteresis loop
- Hydrostatic depoling pressure
- Microstructural analysis (polish, etch, image analyses)

*J. A. Voigt, D. L. Sipola, B. A. Tuttle, M. T. Anderson, "Nonaqueous solution synthesis process for preparing oxide powders of lead zirconate titanate and related materials," U.S. Patent No. 5,908,802, June 1, 1999.

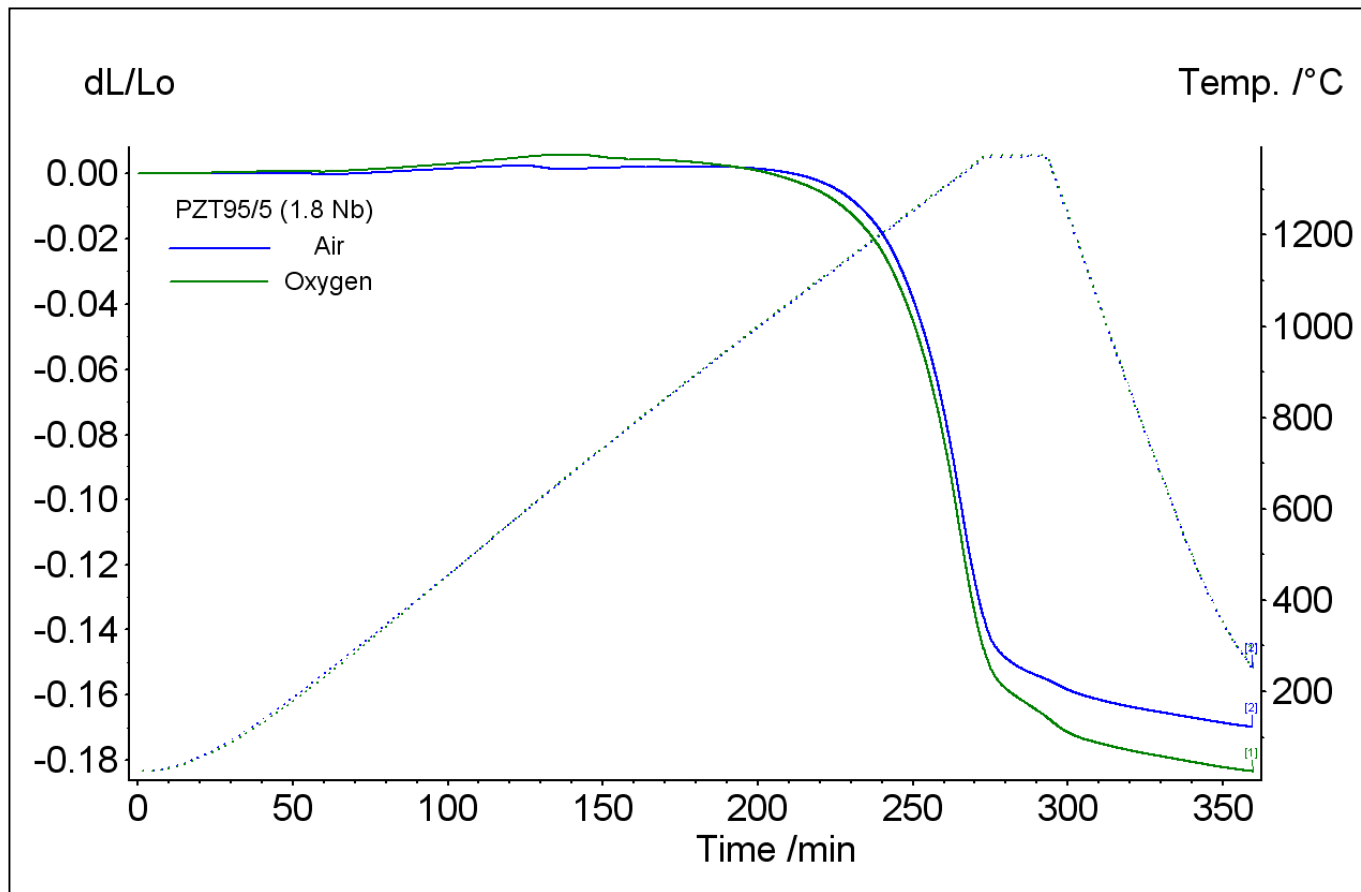
O₂ Environments Significantly Increase the Bulk Density of PZT 95/5



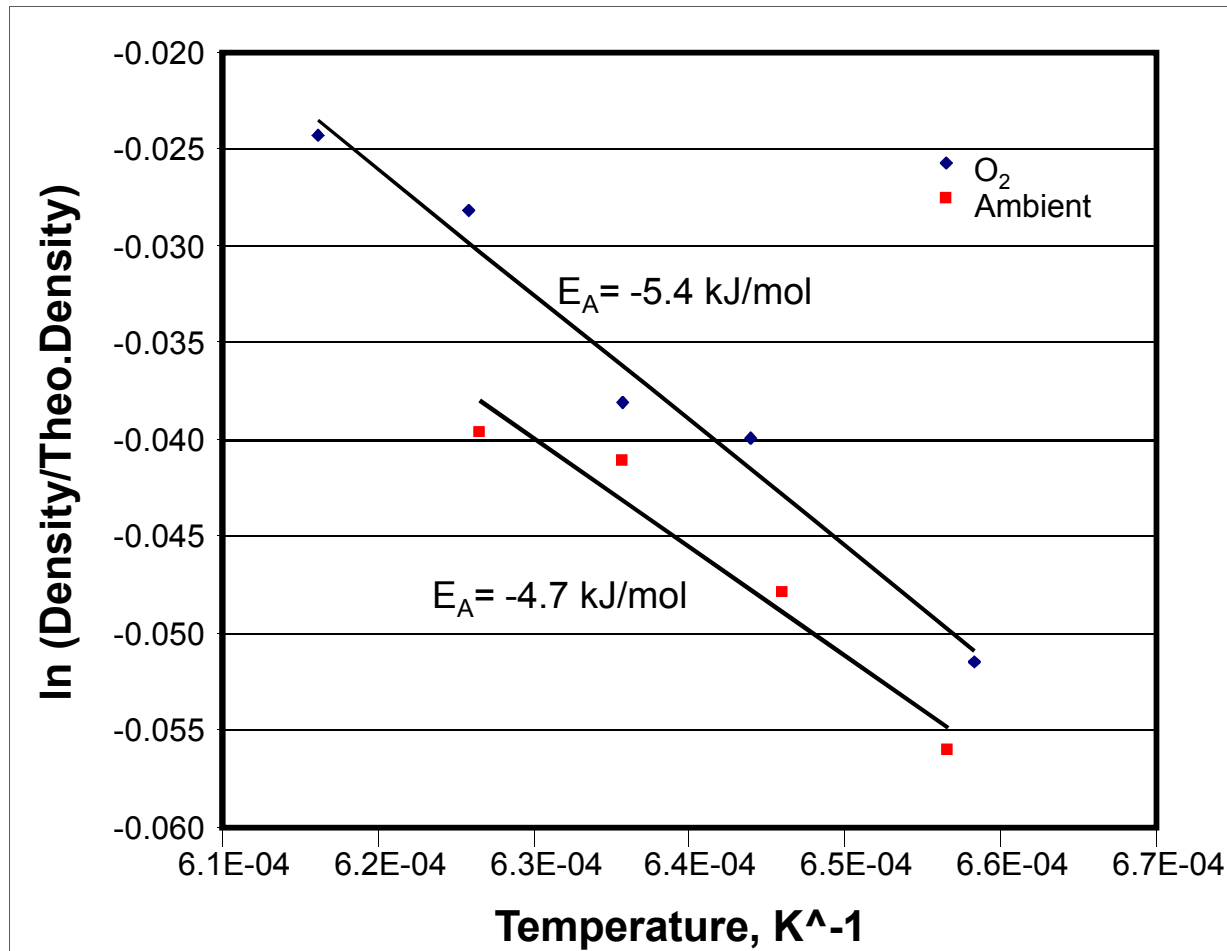
Six Hour Soak Periods are Required to Reach Maximum Density



O₂ Promotes Density during Final Stage Sintering



Activation Energy for Sintering is Comparable Between the Two Atmospheres

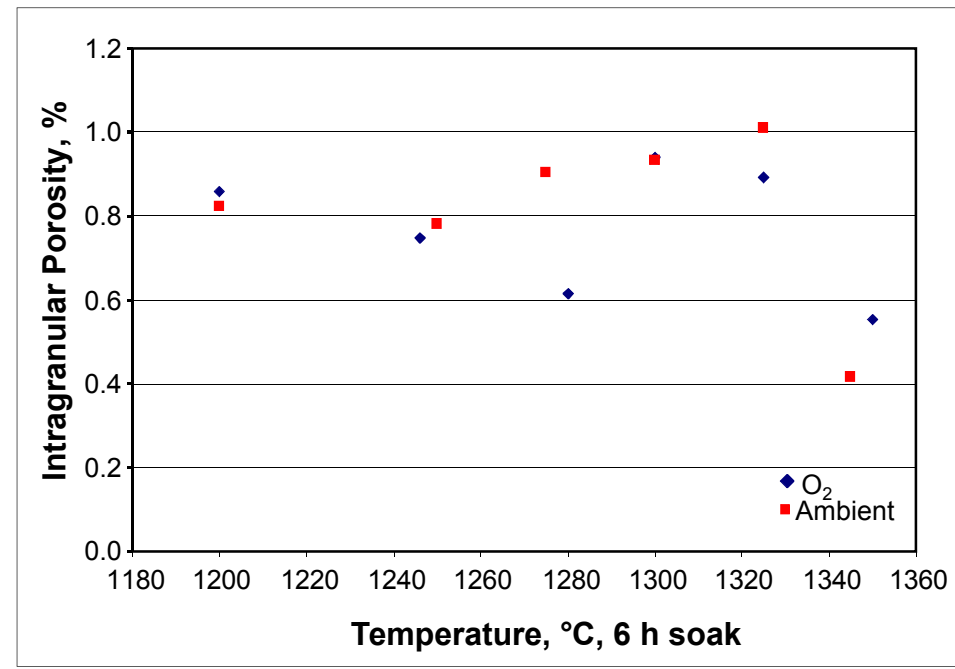
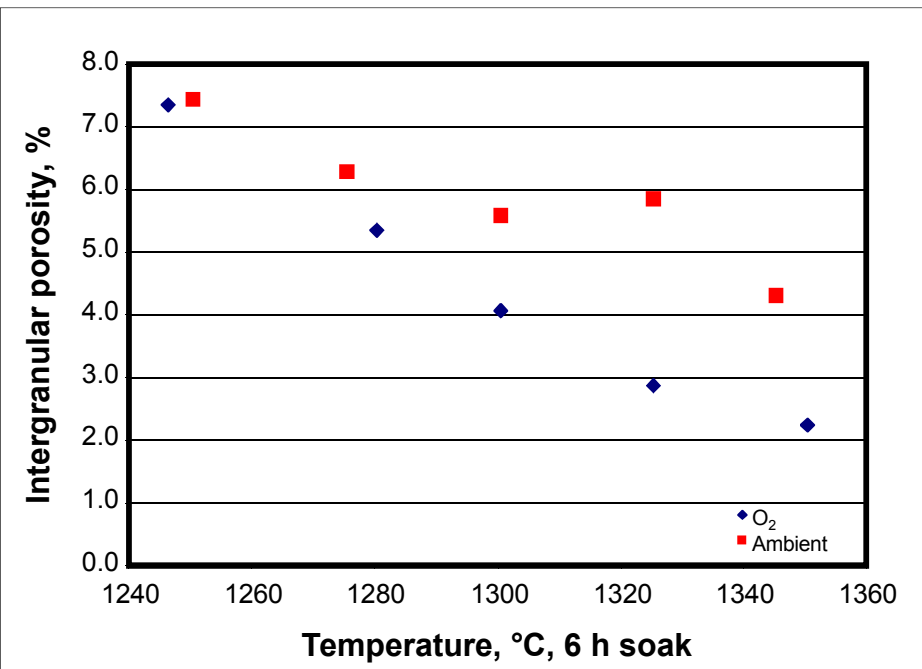


From the Arrhenius Equation:

$$k = Ae^{-\left(\frac{E_a}{RT}\right)}$$

$$E_a \equiv -R \left[\frac{\partial \ln\left(\frac{D}{D_{rh}}\right)}{\partial T^{-1}} \right]$$

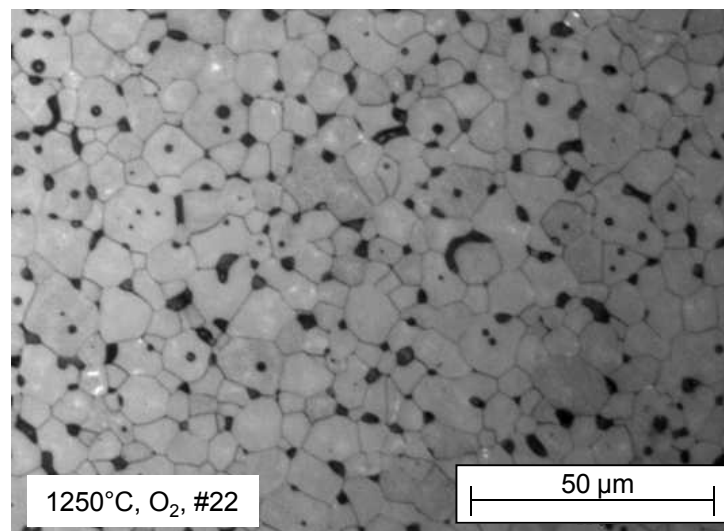
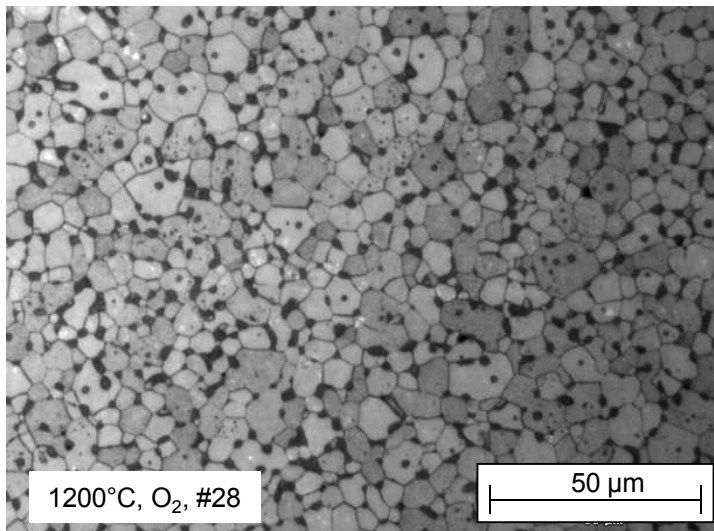
Image Analysis Measurements of Inter- and Intra-Granular Porosity



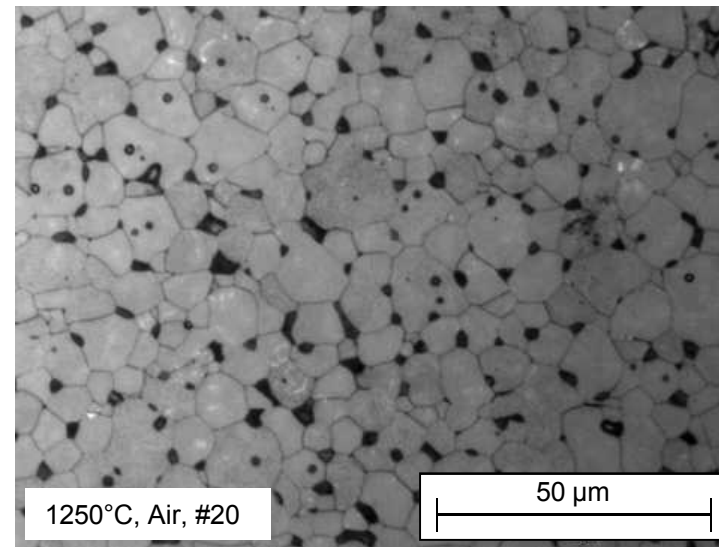
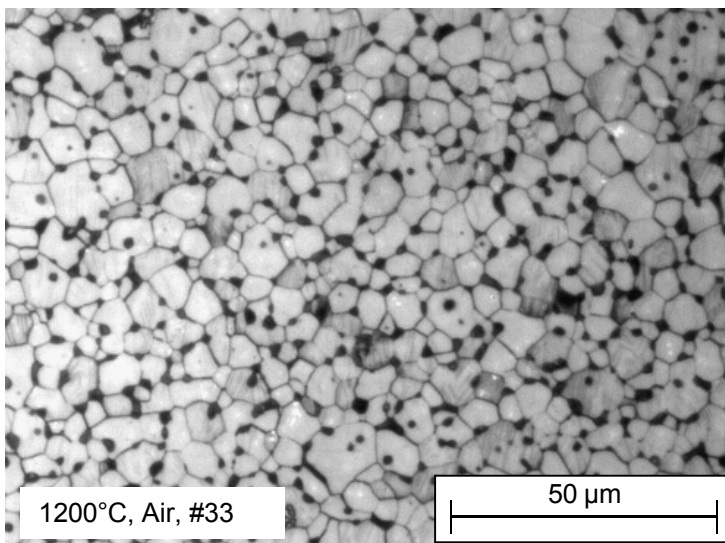
- Samples polished and etched (HCl, 70°C, 15 sec)
- analySIS FIVE (www.olympus.co)

Oxygen and Ambient Fired Samples

O₂

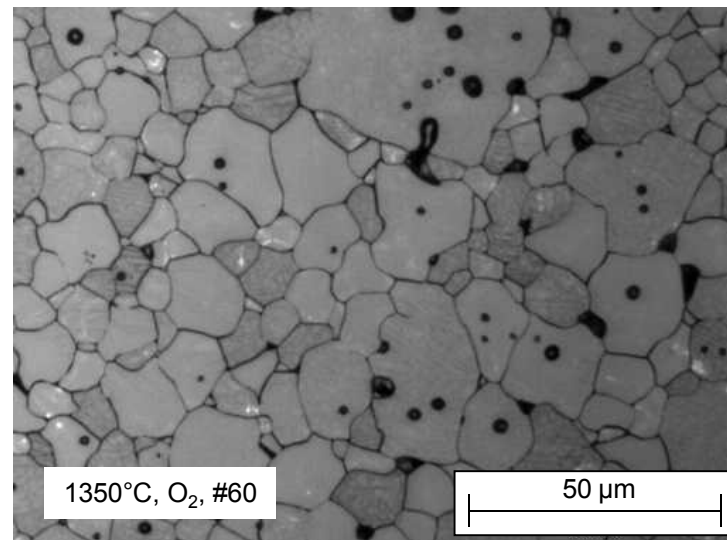
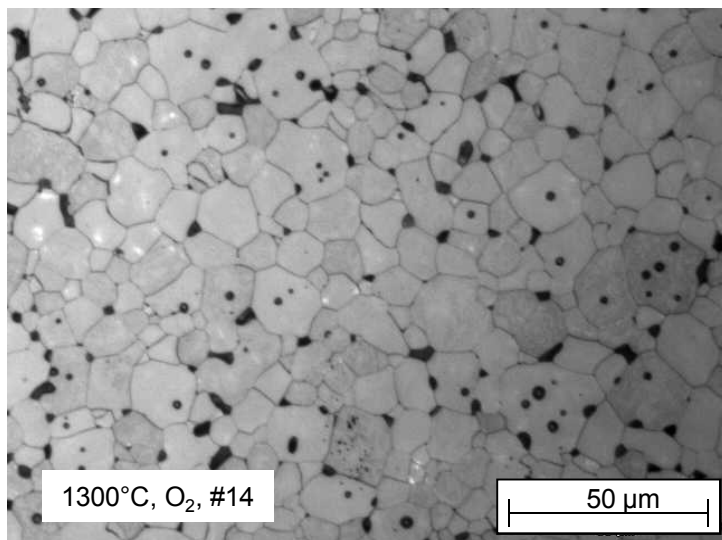


Ambient

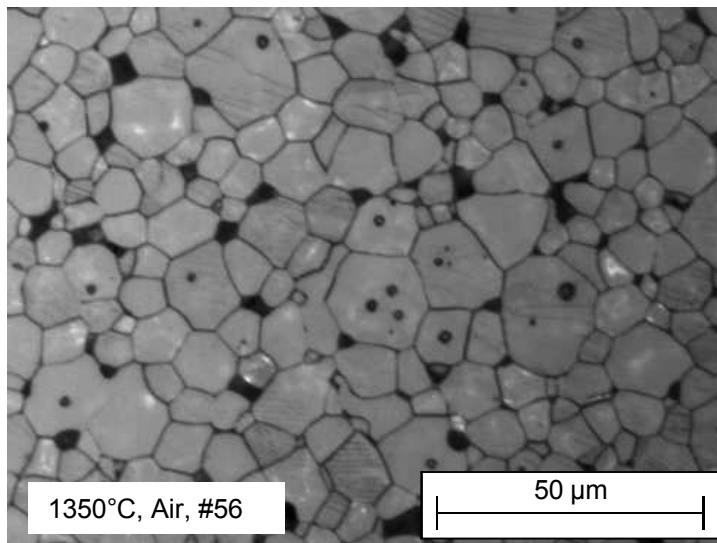
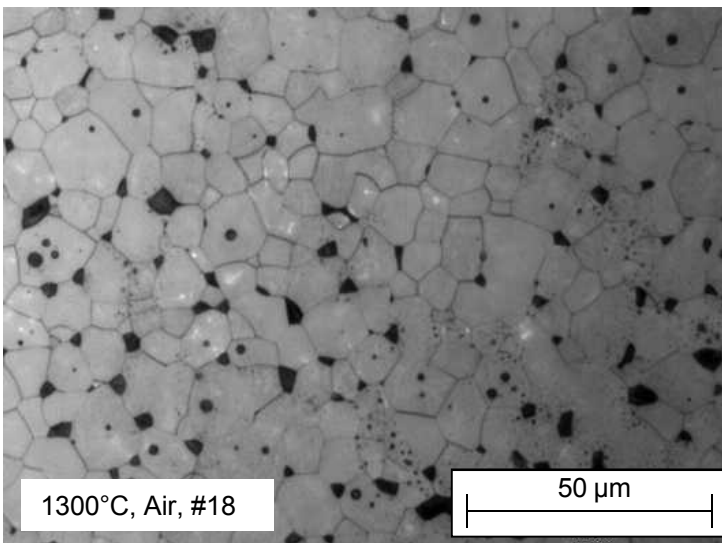


O₂ Atmosphere is More Effective at Removing Intergranular Porosity

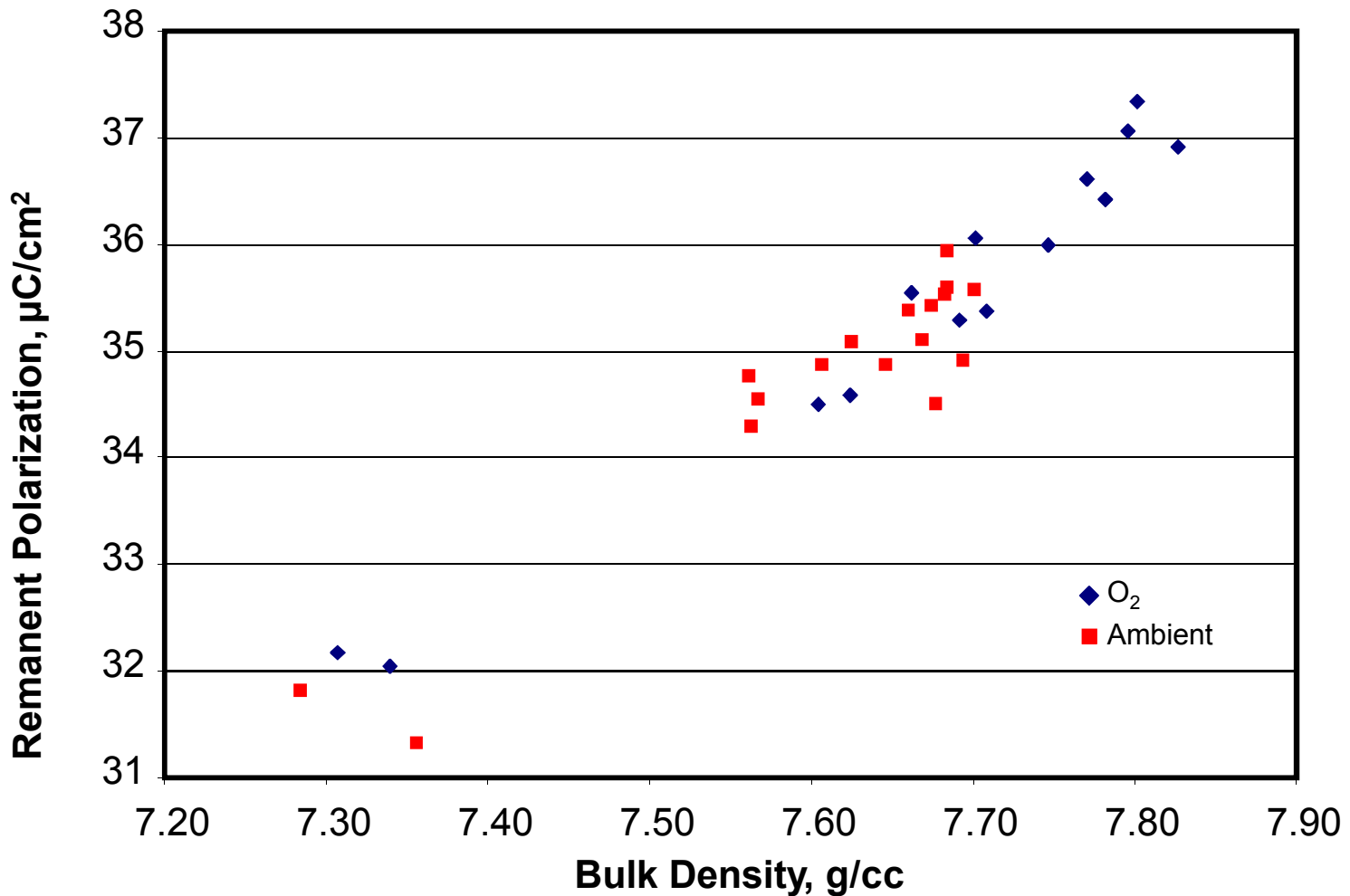
O₂



Ambient



Polarization Increased with Density, Regardless of Atmosphere





Summary

- Firing PZT 95/5 under O₂ atmospheres increases bulk density by 0.1 g/cm³
- Increased density leads to ~4% increase in polarization
- Firing shrinkage, density, and microstructural analyses show that the onset and progression of sintering are very similar.
- However, with oxygen, more intergranular porosity is eliminated during the final stage of sintering, most likely due to the higher diffusion rate of oxygen in PZT.
- Intragranular porosity varies little with firing atmosphere and persists regardless of firing conditions.

Acknowledgements

- George Burns
- Ted Montoya
- Tom Pehr

$$E_a \equiv -R \left[\frac{\partial \ln \left(\frac{D}{D_{Th}} \right)}{\partial T^{-1}} \right] \quad \text{From the Arrhenius Equation:}$$

$$k = Ae^{-\left(\frac{E_a}{RT}\right)}$$

Activation Energy for Sintering:

$$k = Ae^{-\left(\frac{E_a}{RT}\right)}$$

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$$E_a \equiv -R \left(\frac{\partial \left(\frac{\ln D}{D_{Th}} \right)}{\partial (T^{-1})} \right)$$

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$$k = Ae^{-\left(\frac{E_a}{RT}\right)}$$