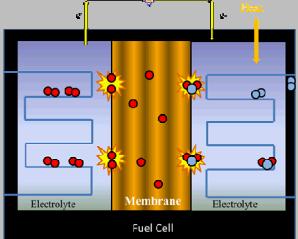


“ex-situ”

SAND2014-2033C

Characterization of Heterogeneous Diffusion and Degradation in Fuel Cell Polymer Membranes using PFG HRMAS NMR

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Todd M. Alam

Sandia National Laboratories, Albuquerque, NM
87185

247th ACS National Meeting
Dallas, TX
March 19, 2014

Acknowledgements: Dr. Michael Hibbs and Kim Childress



U.S. DEPARTMENT OF
ENERGY

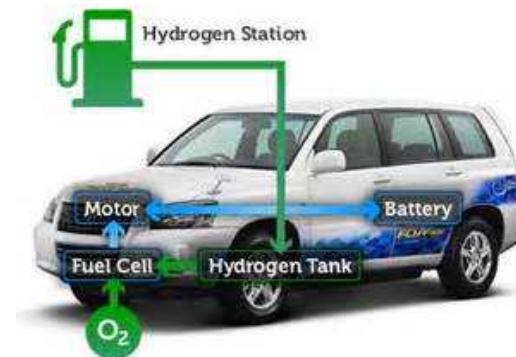
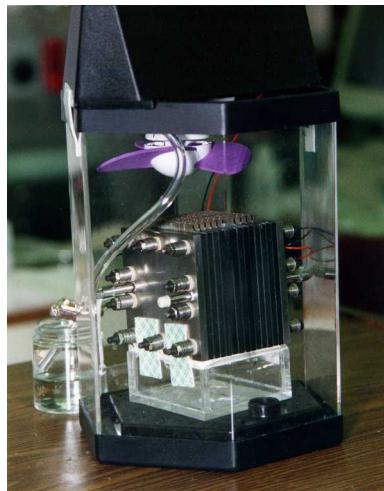


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Fuel Cells – Emerging Technology

“Old Technology – Material Advances Lead the Way”

- Convert chemical energy (fuel) to electricity using oxygen.
- Different types of fuels (hydrogen, methanol, ethanol...).
- Can produce electricity as long as there is fuel (unlike batteries).
- Power generation (backup), including remote sites, military, automobile.
- Higher efficiency (60 – 85%) than combustion systems (30%).



MeOH Fuel Cells - AEMs



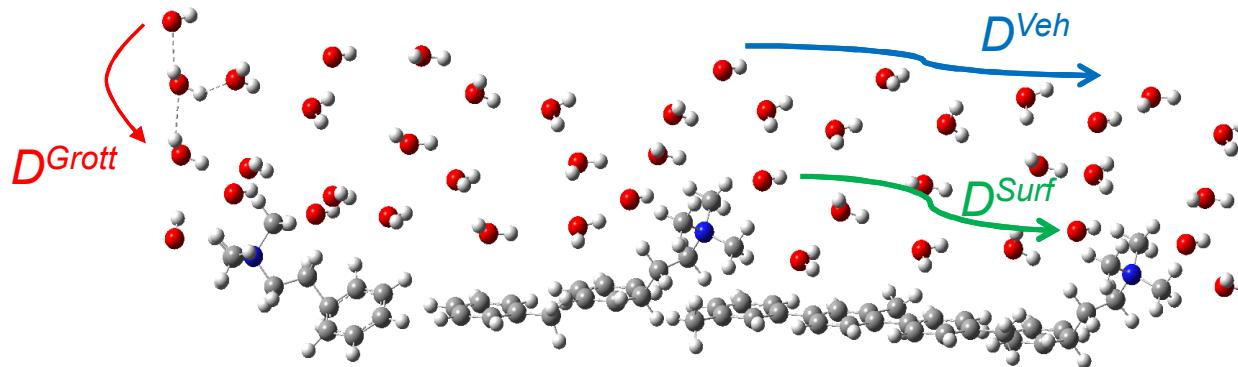
Development of new membranes materials for a wide range of technological applications ultimately based on fundamental understanding of transport.

Conductivity Related to Diffusion Properties

via Nernst-Einstein Equation

$$\sigma_i = \frac{F^2}{RT} \left(\sum_{\substack{j=1 \\ j \neq i}}^n \frac{1}{D_{ij}^{eff}} + \frac{1}{D_{iM}^{eff}} \right)^{-1} C_i$$

$$\sigma = \frac{F^2}{RT} \left(D_{\text{OH}^-}^{\text{Surf}} C_{\text{OH}^-}^{\text{Surf}} + D_{\text{OH}^-}^{\text{Grotthuss}} C_{\text{OH}^-}^{\text{Grotthuss}} + D_{\text{OH}^-}^{\text{Veh}} C_{\text{OH}^-}^{\text{Veh}} \right)$$

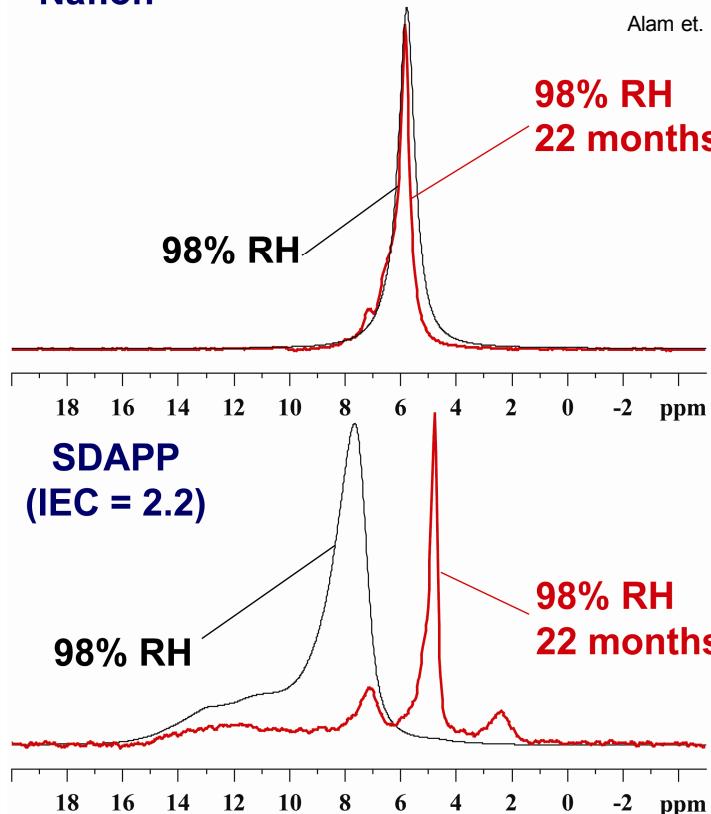


The transport of MeOH can also be discussed in terms of different diffusion environments.

PFG NMR and Site Resolution

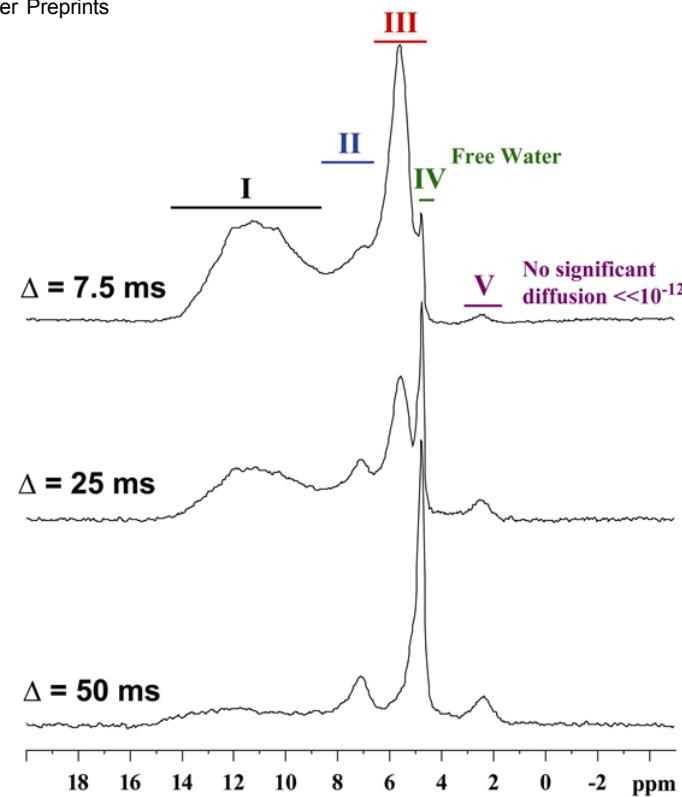
Pulse Field Gradient (PFG) NMR used to measure solvent diffusion.

Nafion



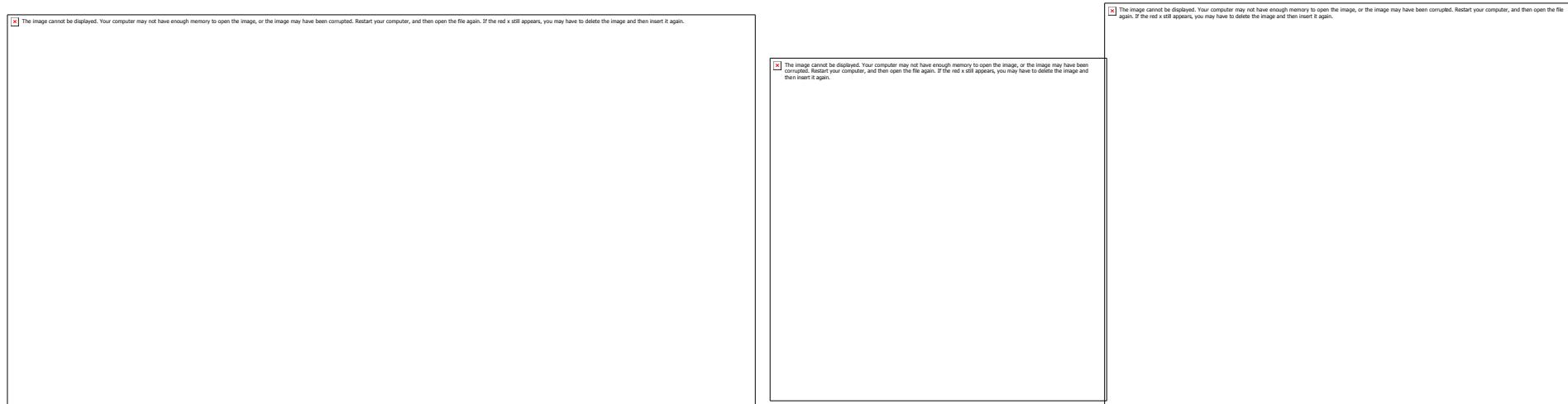
Alam et. al (2008) ACS Polymer Preprints

SDAPP



Under static PFG NMR we have occasionally observed different water environments, but the lack of resolution was never considered an issue!

Site Resolution in MeOH Fuel Cell Membranes



Different water environments in polymers

- Water in hot pressed Nafion, Jeong and Han, Bull. Korean Chem. Soc. (2009), 30, 1559.
- Water in PEEK, Baias *et al.*, Chem. Phys. Lett. (2008), 456, 227; (2009) 473, 142. MAS with SSB with no chemical shift resolution.
- Mele *et al.*, J. Incl. Phenom. Macrocycl. Chem. (2011), 69, 403. HRMAS resolution.

HRMAS PFG NMR and Site Resolution

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$T_1 = 850 \text{ ms}$
 $D = 5.4 \times 10^{-10} \text{ m}^2/\text{s}$

$T_1 = 1.13 \text{ s}$
 $D = 2.3 \times 10^{-10} \text{ m}^2/\text{s}$

$T_1 = 2.0 \text{ s}$
 $D = 1.8 \times 10^{-9} \text{ m}^2/\text{s}$

$T_1 = 2.1 \text{ s}$
 $D = 1.6 \times 10^{-9} \text{ m}^2/\text{s}$

Resolution is always exciting! Can ask questions about differences between MeOH and water association with the membrane.

AEM Membranes Investigated

 The image cannot be displayed. Your computer may not have enough memory to open the image, or the image may have been corrupted. Restart your computer, and then open the file again. If the red x still appears, you may have to delete the image and then insert it again.

Alkane spacers (C_6) added for higher mobility, increased water content, alkaline stability.

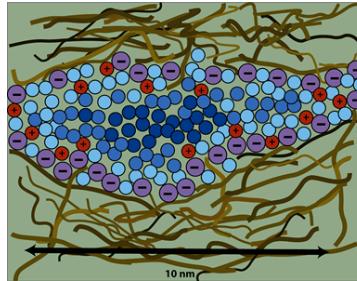
What is impact on diffusion and types of environments?

¹H HRMAS NMR of Different AEM Membranes



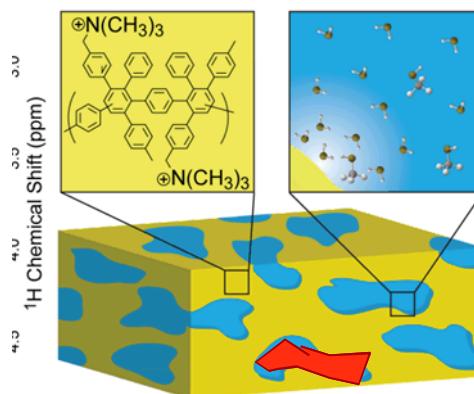
2D ^1H - ^1H Exchange/NOESY Studies

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- Free and associated domains exist.
- These domains show some exchange.
- Associated water and MeOH in close contact with membrane.

Anion Exchange Membrane



Diffusion Analysis of Individual Species

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$$\langle R^2(\Delta) \rangle = -6 \ln [E(q, \Delta) / E(0, \Delta)] / q^2$$

$$\langle R^2 \rangle = 6Dt$$

$$\langle z_M^2(\Delta) \rangle = -2 \ln [E(q, \Delta) / E(0, \Delta)] / q^2$$

$$\langle z_M^2 \rangle = 2D_\alpha t^\alpha$$

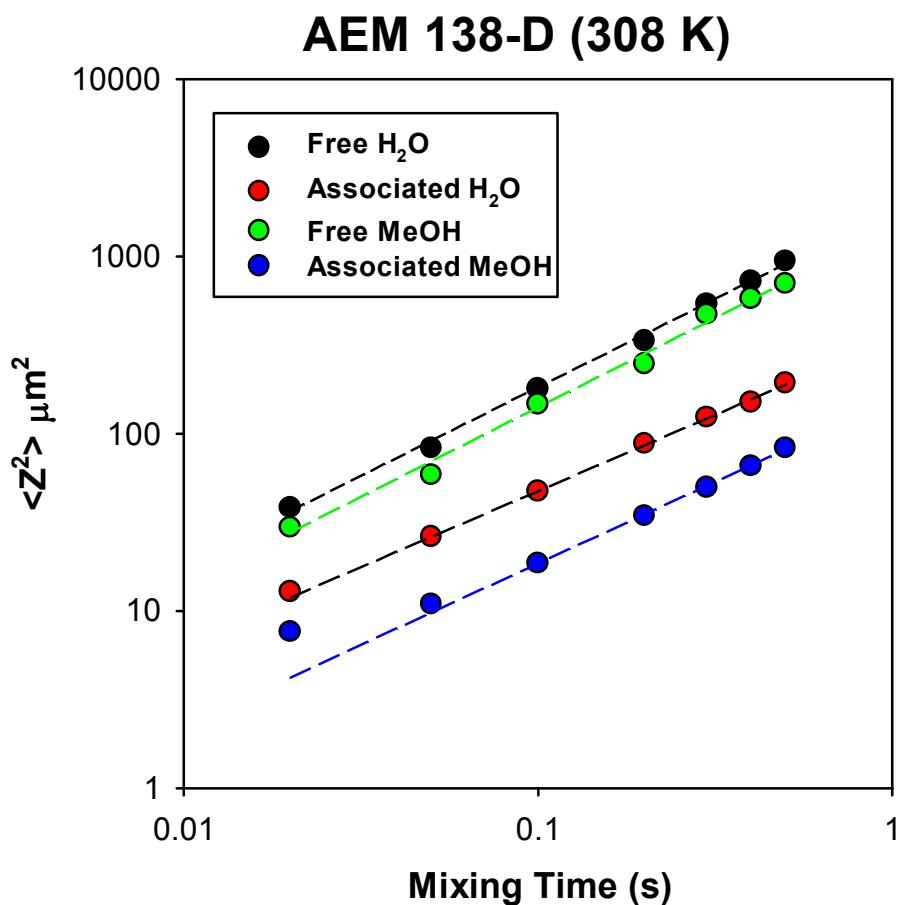
- **Associated diffusion is an order of magnitude slower than free species (Water and MeOH).**
- **MeOH diffusion slower than Water in both environments.**
- **The ratio of $D_{\text{assoc}}/D_{\text{free}}$ is much smaller for MeOH, suggesting preferential association with membrane.**

Diffusion Analysis of Individual Species

The image cannot be displayed. Your computer may not have enough memory to open the image, or the image may have been corrupted. Restart your computer, and then open the file again. If the red x still appears, you may have to delete the image and then insert it again.

- Extract Δz^2 from multiple different
- Δ delays in PFG NMR
- Evaluate possibility of anomalous diffusion ($\alpha \neq 1$).
- Most systems show normal diffusion. As expected in these membranes.
- Associated water environment reveal fractal diffusion at lower hydration/temperatures.

Anomalous Diffusion in Associated Environments?



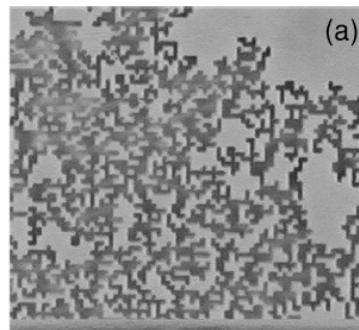
Anomalous diffusion can be expressed through the power law.

$$\langle z^2 \rangle = 2D_\alpha \Delta^\alpha$$

$\alpha = 1$, normal diffusion

$\alpha < 1$, sub-diffusive

$\alpha \sim 0.7$ 2D fractal



Klemm, Metler, Kimmich, *Phys. Rev. E* (2002), 65, 021112

Disappears with increasing temperature and hydration.

Solvent Diffusional Activation Energies

The image cannot be displayed. Your computer may not have enough memory to open the image, or the image may have been corrupted. Restart your computer, and then open the file again. If the red x still appears, you may have to delete the image and then insert it again.

Sample (IEC)	E _a (kJ/Mol)			
	F-H ₂ O	A-H ₂ O	F-MeOH	A-MeOH
1N MeOH	26.0	--	27.0	--
ATMPP (1.48)	20.0	44.0	28.3	23.6
ATMPP (1.79)	29.7	24.5	26.2	29.4
ATMPP (2.35)	26.7	28.7	27.0	29.2
TMAC ₆ PCC ₆ (2.13)	37.6	33.3	38.6	30.6
TMAC ₆ PCC ₆ (2.27)	--	23.2	--	16.5
TMAC ₆ PCC ₆ (2.60)	37.4		37.5	

- Results similar to Nafion and Nafion composites.
- No direct comparison because individual sites not investigated.

Ratios of Diffusion Rates

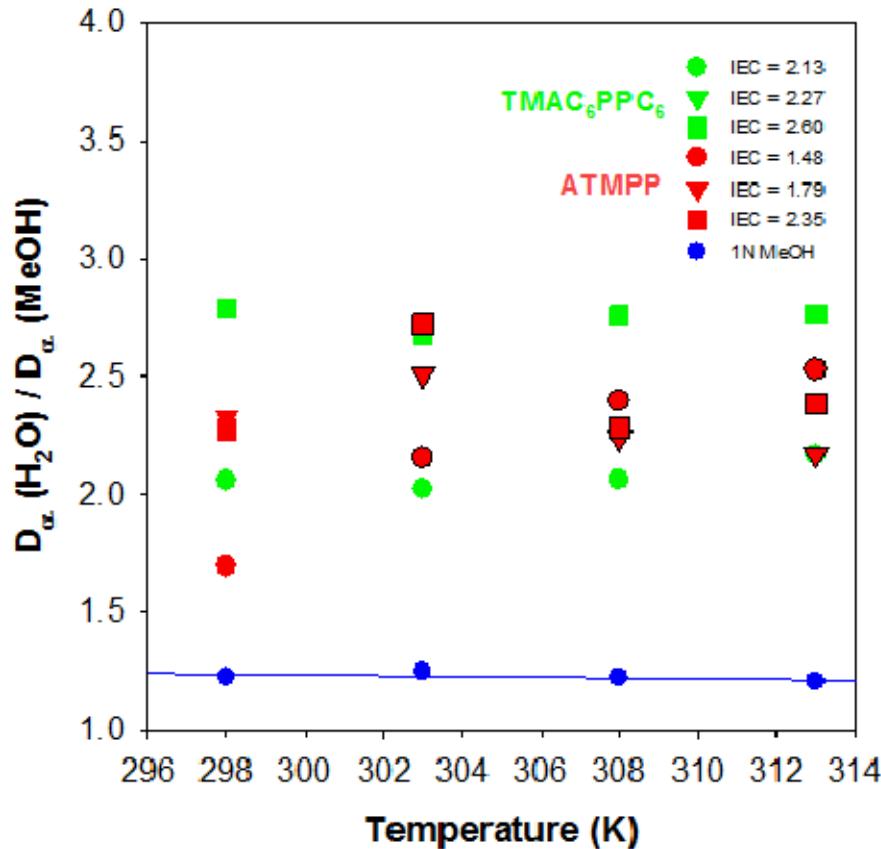
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**Associated Sites ~5 times
slower than 1N MeOH.**

**ATMC₆PPC₆ higher
diffusion.**

**“Free Sites” NOT really like
Bulk 1N MeOH!**

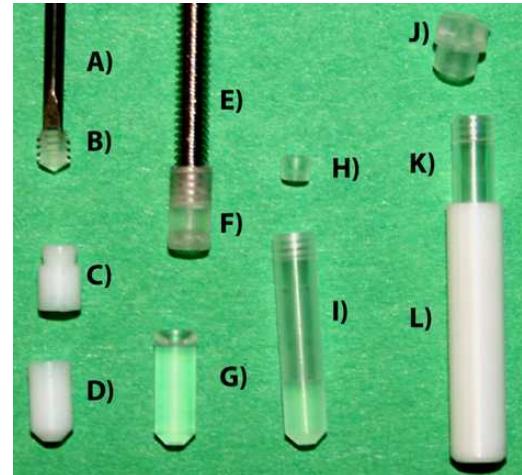
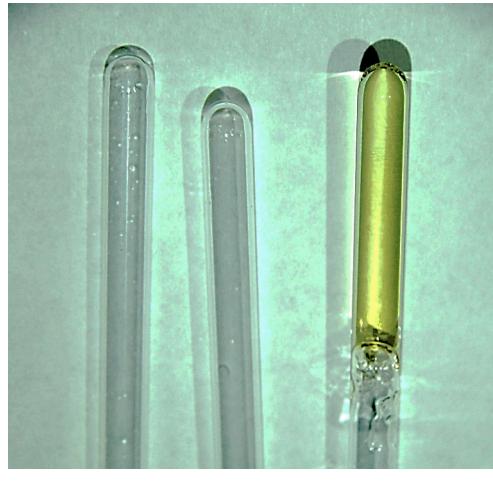
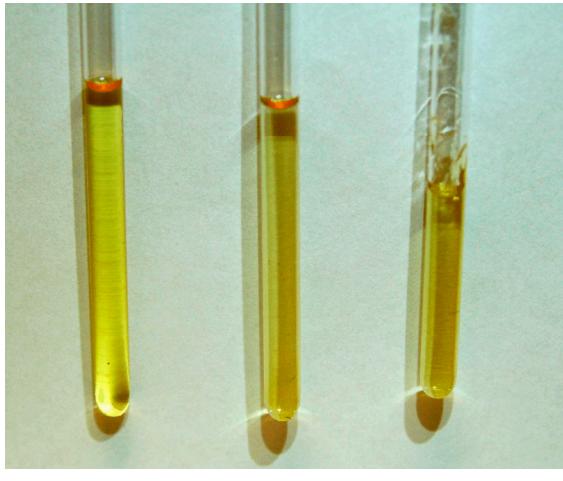
Ratios of Diffusion ($\text{H}_2\text{O}/\text{MeOH}$)



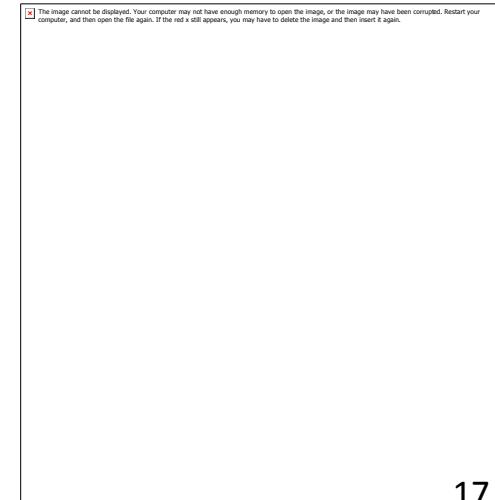
- All of these membranes show a preferential reduction of MeOH.
- Helps reduce MeOH cross-over in membranes.

Why Use HRMAS For Aged Membranes

Aged ATMPP (4M KOH) swollen in DMSO



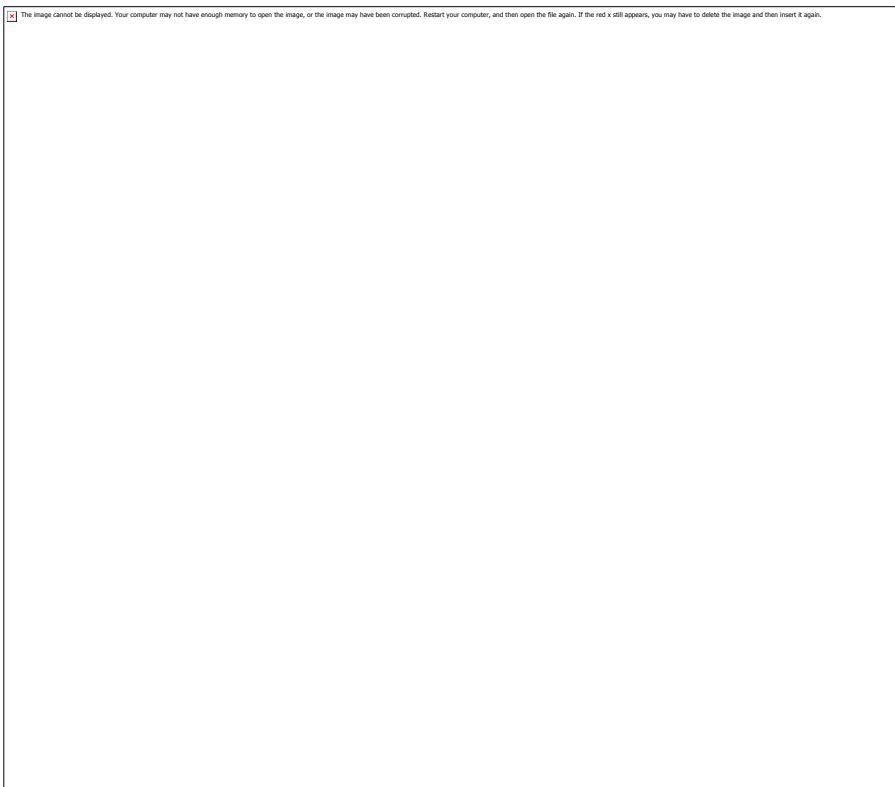
- Aged samples no longer dissolve in DMSO, become gels. AEMs also become brittle with aging.
- Solution NMR becomes broad....can address with HRMAS NMR.
- “Liquid like samples” need to retain liquid under MAS.
- Need to consider centrifugation effects under MAS.



HRMAS NMR For Degradation Studies

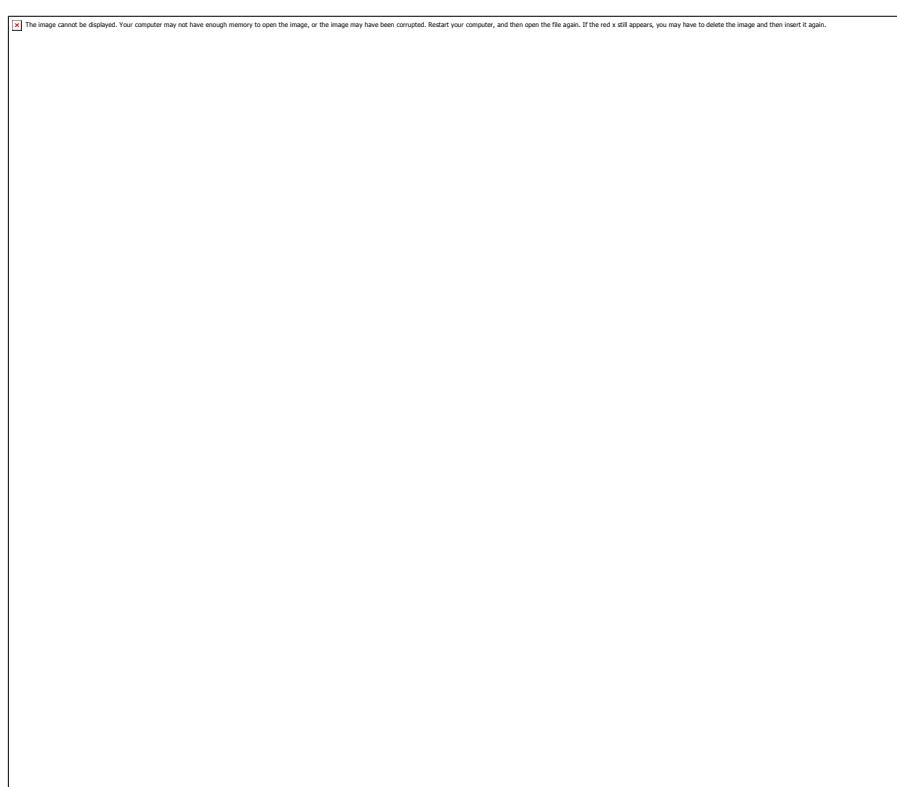
Solution NMR

Aged ATMPP (4M KOH) swollen in DMSO



HRMAS NMR

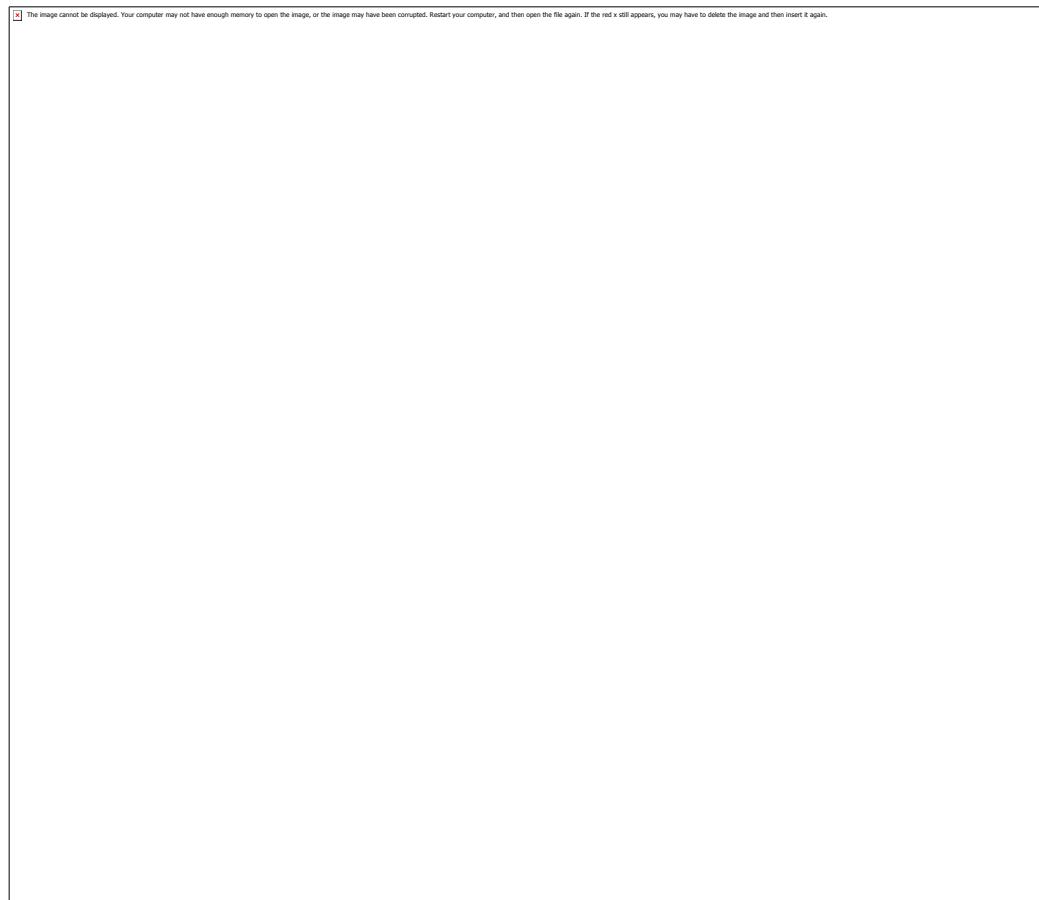
10-Day Aged ATMPP (4M KOH) swollen in DMSO



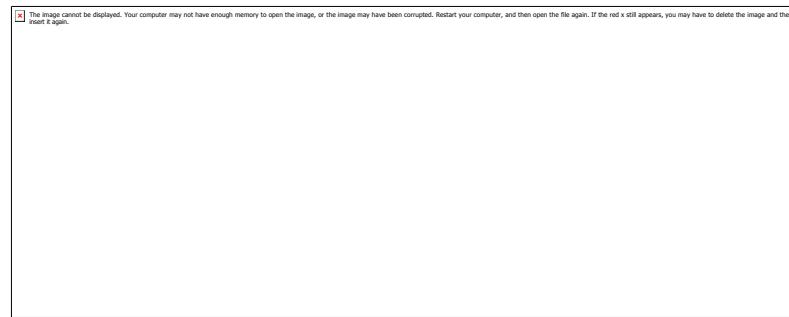
- HRMAS removes magnetic susceptibility at low MAS speeds (> 1 kHz).
- Solution like resolution obtained at RT.
- Lower S/N as 30 μ L inserts used for HRMAS.

HRMAS NMR For Degradation Studies

2D NOESY ^1H HRMAS ($\tau_{\text{mix}} = 100$ ms)



Conclusions



- HRMAS NMR provides increased resolution in MeOH fuel cell membranes.
- Combined with PFG resolves diffusion for 4 unique solvent environments.
- This system may represent a special case, not a generalized solution for all membrane systems.
- The diffusion measurements of low water content membranes reveals associated water and MeOH have sub-diffusion process (motion on membrane surface).
- There is a large membrane influence on the MeOH versus water species.
- Synthetic modification improved solvent diffusion with respect to original AEM membranes.
- HRMAS PFG NMR does provide a novel tool to study these heterogeneous diffusion processes in polymer membranes.

Thanks for your attention!