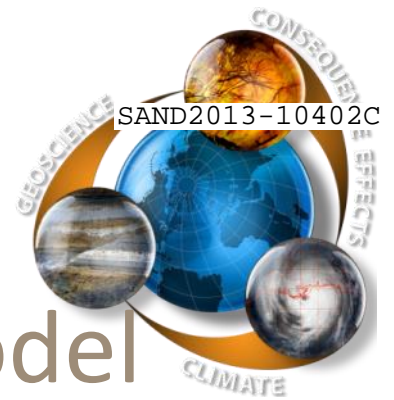


Upscaling of reaction rates in reactive transport using pore-scale reactive transport model



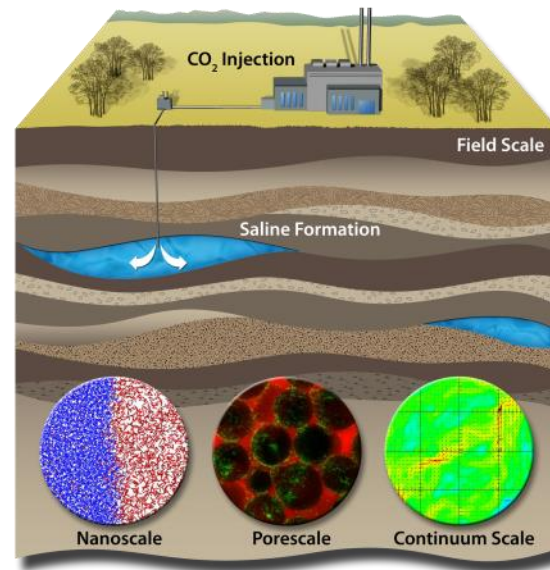
Hongkyu Yoon

December 11, 2013

Collaborators:

Thomas Dewers, Bill Arnold (SNL)

Jon Major, Peter Eichhubl (UT)



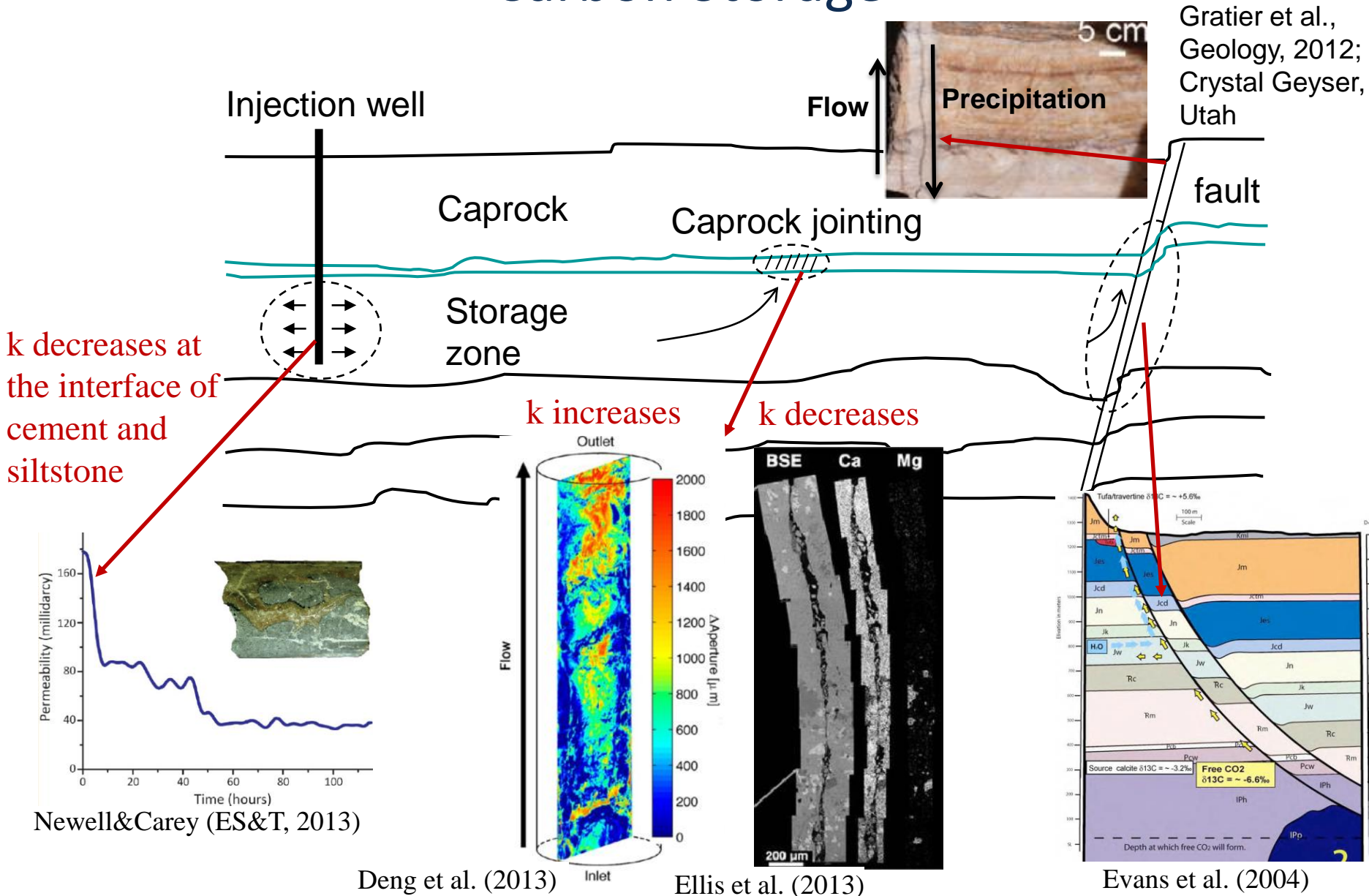
*Exceptional
service
in the
national
interest*

Acknowledgment: This work is supported as part of the Center for Frontiers of Subsurface Energy Security, an Energy Frontier Research Center funded by the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences under Award Number DE-SC0001114.



Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

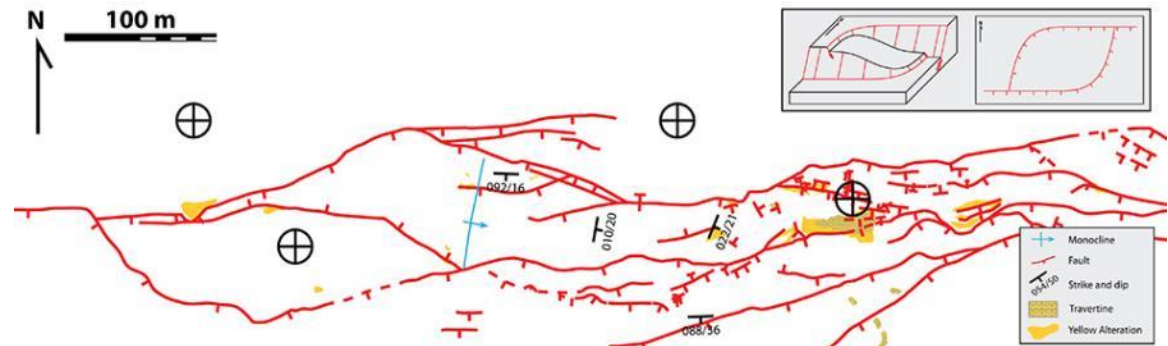
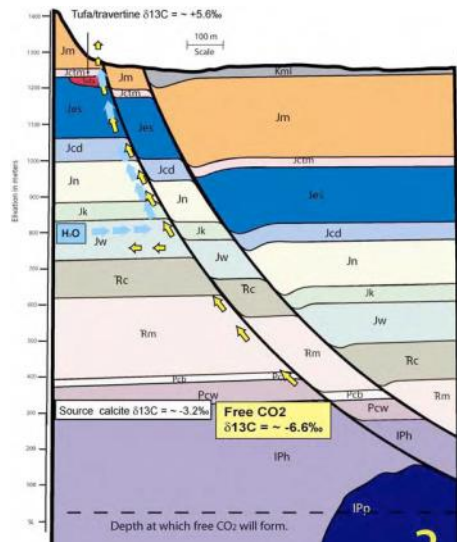
Carbon Storage



Little Grand Wash Fault

Crystal Geyser, Utah

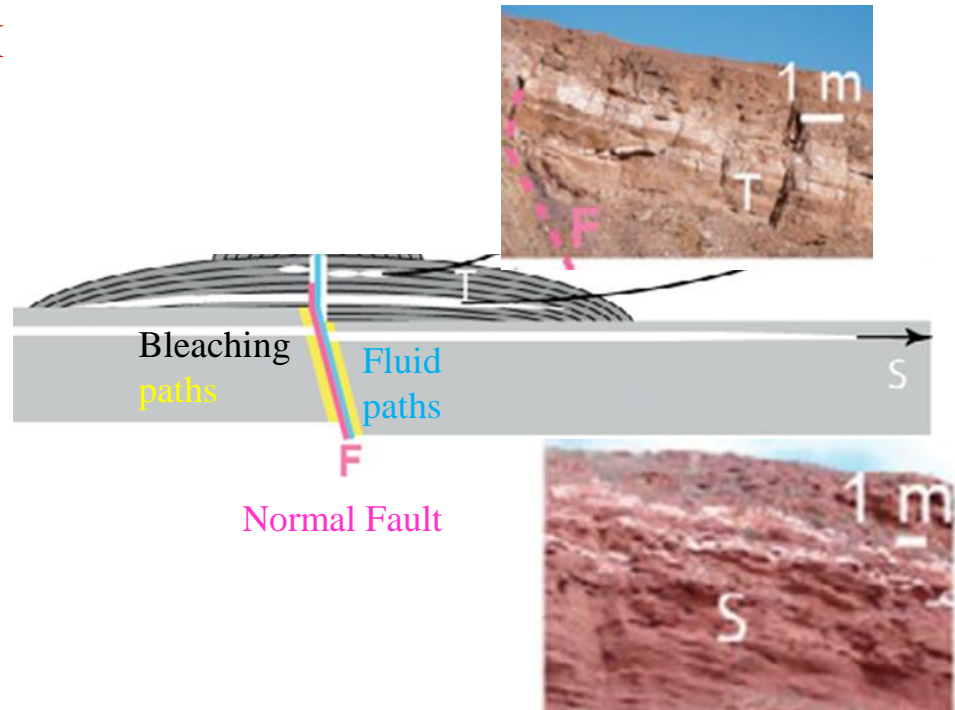
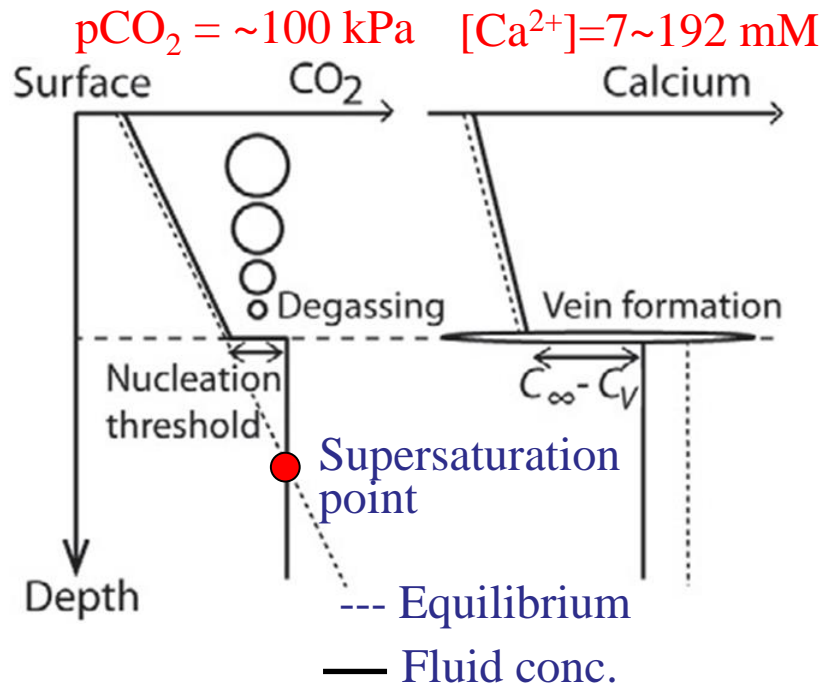
- Observations along the surface exposure of the Grand Wash fault indicate alteration zones of 10-50 m width with spacing on the order of 100 m
- Different flow conduits active at different times
- Locations of conduits controlled by fault-segment intersections and/or topography
- Sandstone permeability reduced by 3 to 4 orders of magnitude in alteration zones by carbonate cementation



From Peter Eichhubl

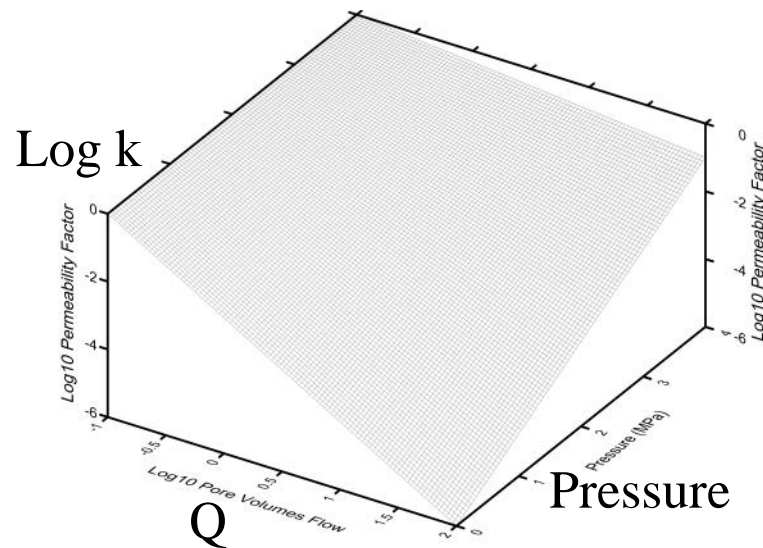
Succinct chemistry & mineralogy

- Spring water: Low temperature and Mg concentrations
- Aragonite (metastable CaCO_3 phase) precipitation indicates large, sudden increase of supersaturation with respect to CaCO_3 in solution \rightarrow high CO_2 degassing rate
- Springs: high concentrations of dissolved CO_2 ($p\text{CO}_2 = \sim 100 \text{ kPa}$)
- Both calcite and aragonite are present
- For some horizontal travertine veins, veins grow top to bottom (U-Th dating)



Crystal Geyser Site: Grand Wash Fault Modeling

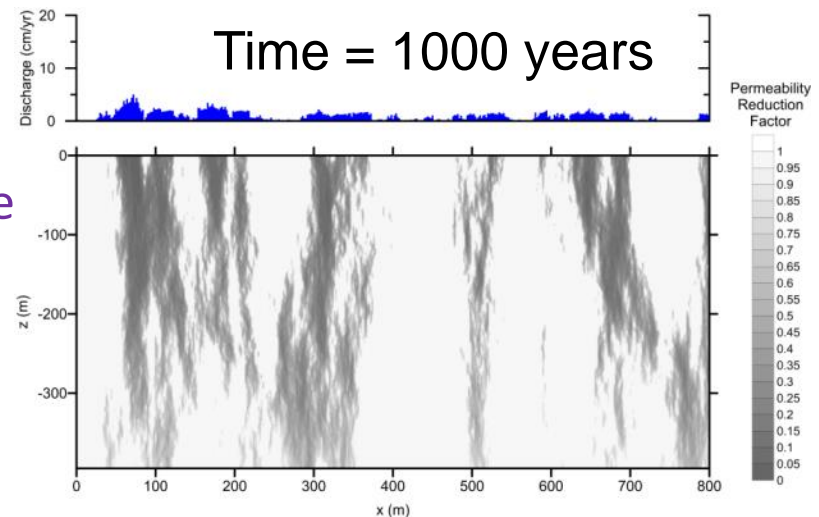
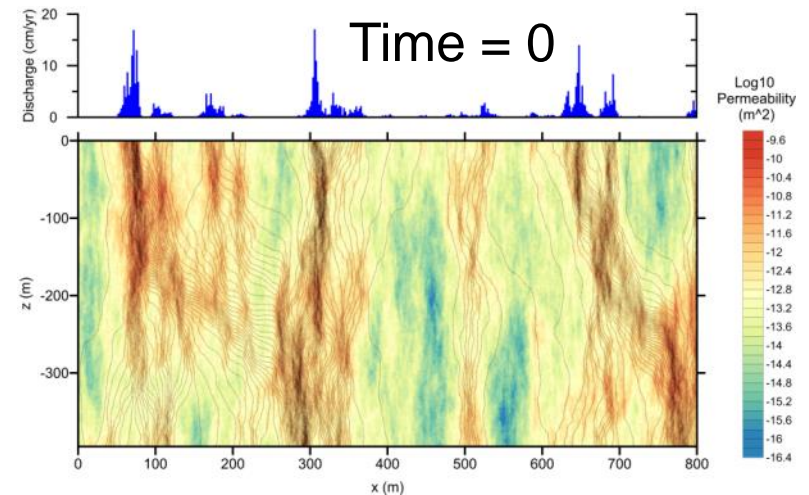
- Simplified, two-dimensional response surface
- Permeability reduction due to calcite precipitation is a function of cumulative pore volume (Q) of groundwater and fluid pressure (P)
- Fluid pressure is taken as a gross proxy for chemical conditions in which higher calcite solubility is associated with higher fluid pressure (greater depth)



Adapted from Mehmani et al. (2012)

Crystal Geyser Site: Grand Wash Fault Modeling

- Unconditional geostatistical simulation
- Initial simulated steady-state flow pattern is qualitatively similar to the spacing of seeps along the Grand Wash fault (~100's of m between locations of groundwater discharge)
- Transient flow simulation includes explicit updating of k field at each time step using the response surface (FEHM)
- k is reduced by several orders of magnitude by calcite precipitation, primarily in the shallower high-flow channels
- Evolution of the flow field results in more dispersed groundwater discharge at the surface



Response Function based on Pore Scale Simulations

- Vigorously tested pore-scale model can be used to develop a response function (or dimension reduction model) for continuum-scale permeability and porosity (k - ε) relationships
- k - ε and surface area- ε relationships will be developed over a range of solution chemistry, chemical reaction, and pore structure configurations in addition to Pe and Da numbers

Changes in porosity due to precipitation ~tortuosity and permeability by phenomenological power law relations

$$\tau(\varphi) = \tau_0 \left(\frac{\varphi}{\varphi_0} \right)^n \quad k(\varphi) = k_0 \left(\frac{\varphi}{\varphi_0} \right)^n \quad a_{cc} = a_{cc}^0 \left(\frac{\varphi}{\varphi_0} \right)^{2/3}$$

Response Function based on Pore Scale Simulations

- Pe & Da numbers

$$Pe(uL/D) = 0.08, 0.8, 8$$

$$Da(kL/(K_{sp}^{0.5} \times D)) = 0.002, 0.02, 0.1$$

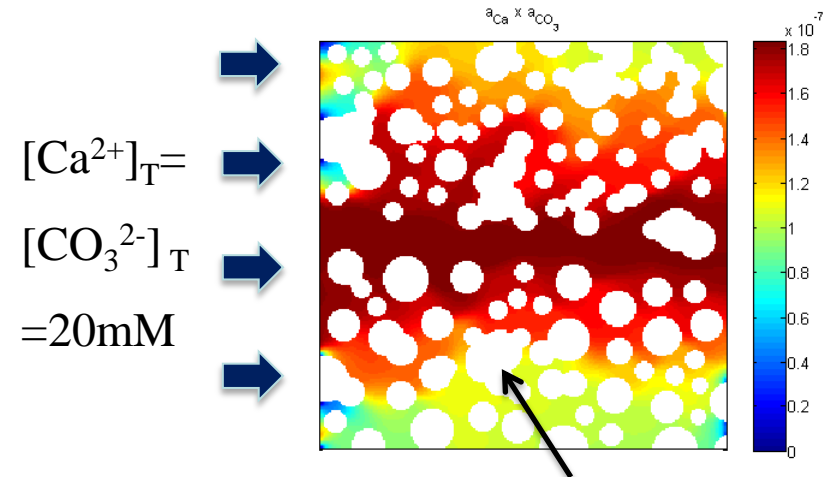
$$(K_{sp, \text{calcite}} = 3.3 \times 10^{-9} \text{ M}^2)$$

- Chemical speciation

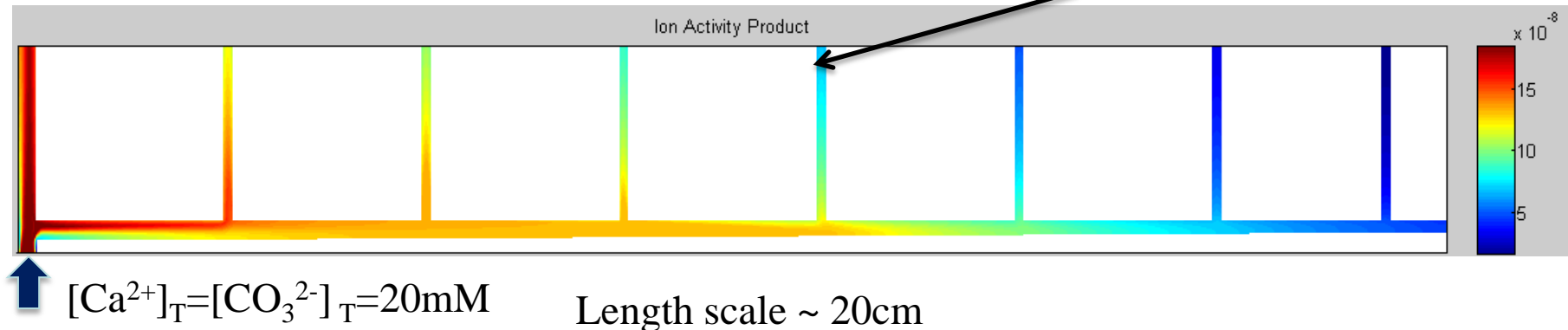
Speciation: Ca^{2+} , H^+ , CO_3^{2-} , HCO_3^- , H_2CO_3

No speciation: Ca^{2+} , CO_3^{2-}

- Pore structures

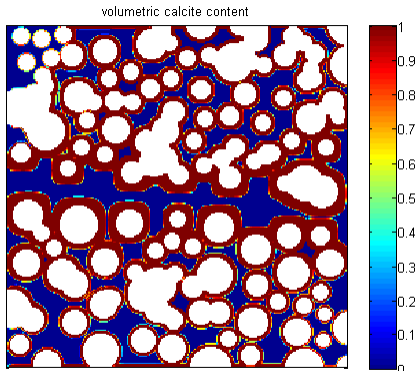


fracture network with varying apertures and patterns



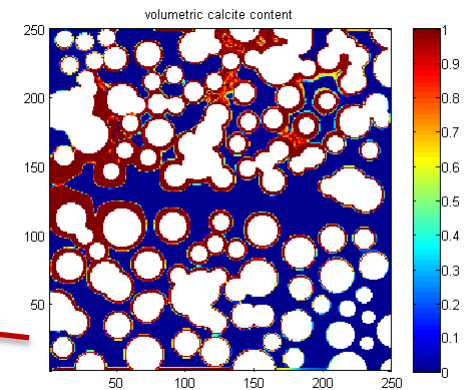
Permeability-Porosity Relationships

High Pe; Low Da

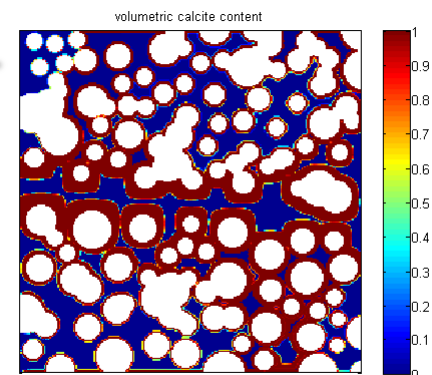
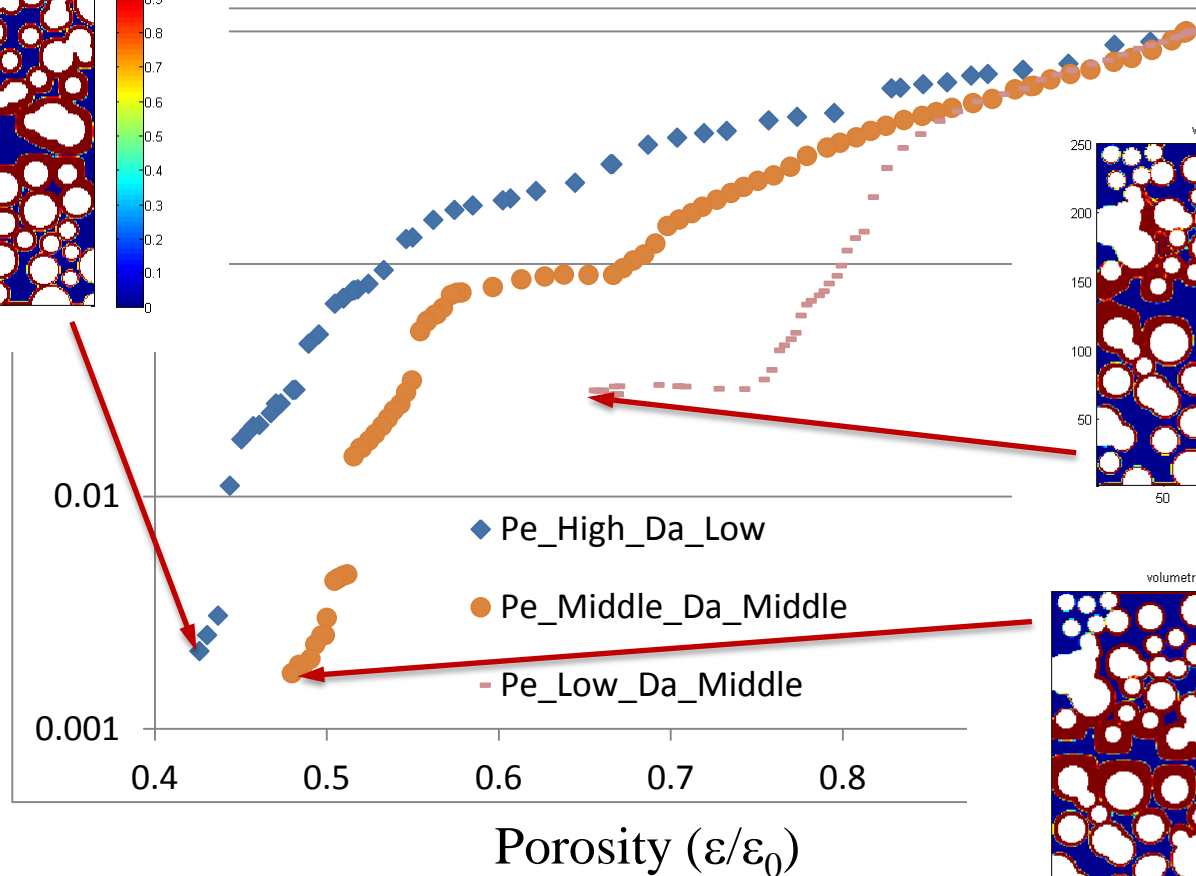


Carbonate speciation

Low Pe;
Medium Da



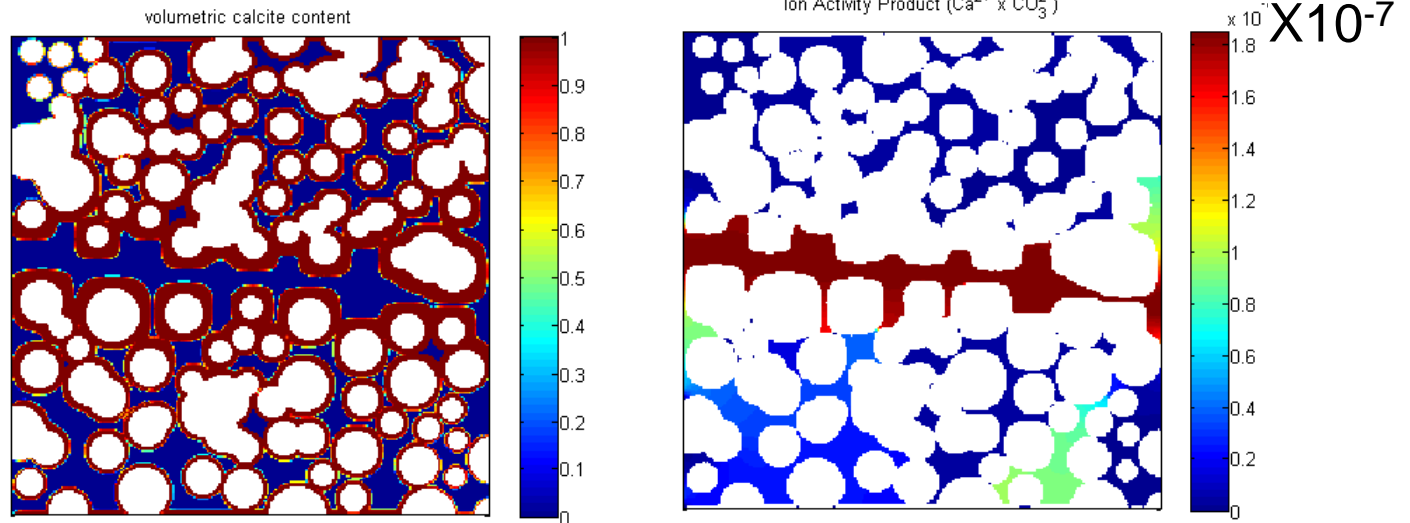
Permeability (k/k_0)



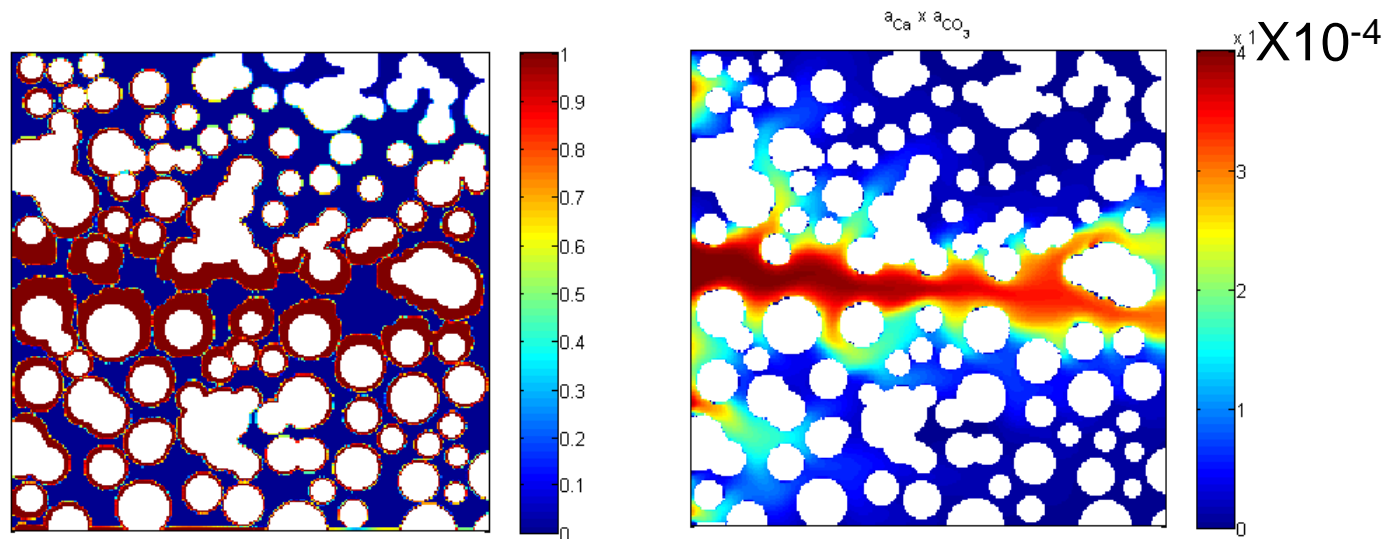
Medium Pe; Medium Da

High Pe & Low Da

Speciation



No speciation

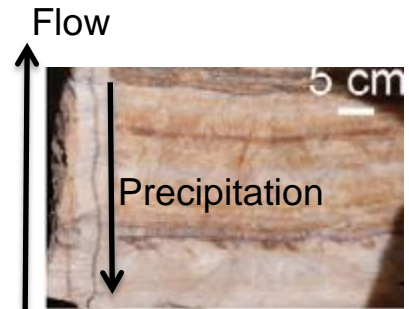
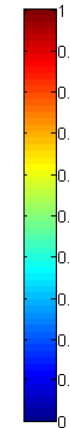
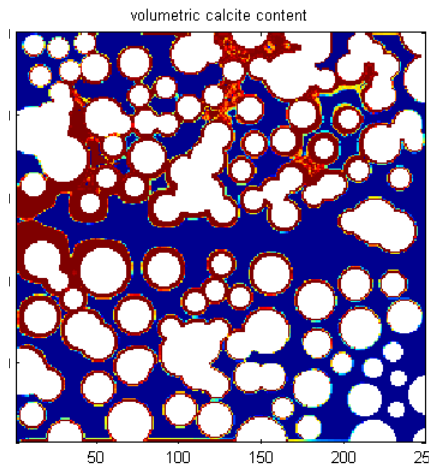
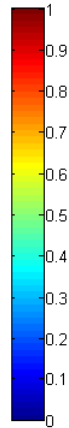
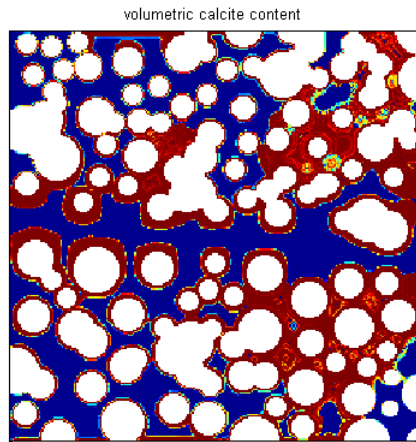


Pore clogging at the front

Medium Pe ; High Da

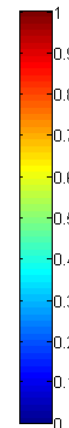
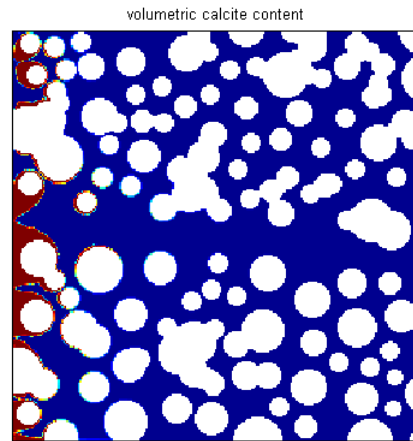
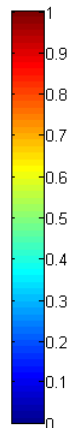
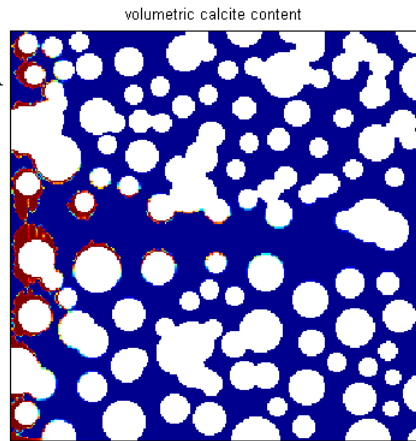
Low Pe ; Medium Da

Speciation



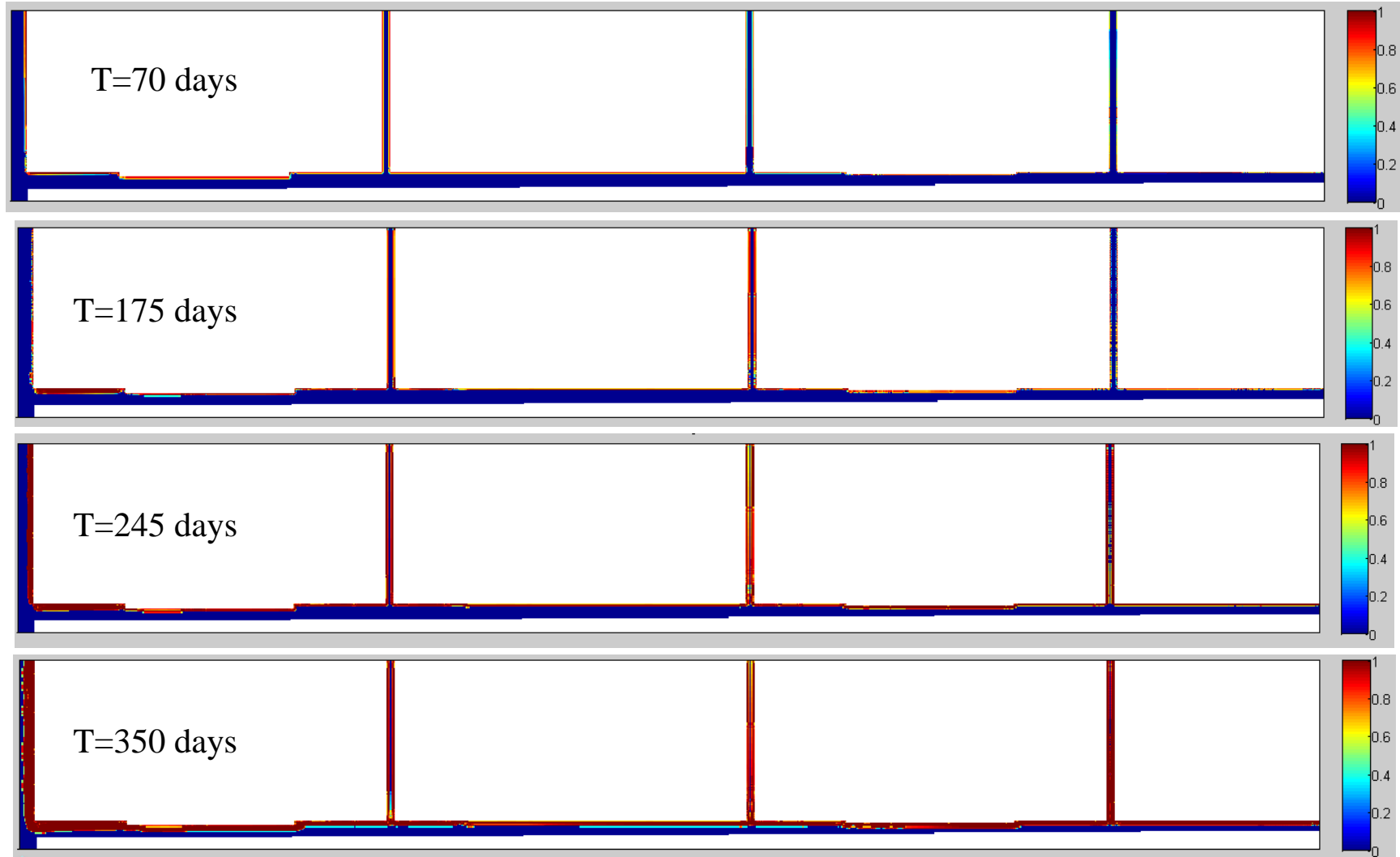
Gratier et al. (2012);
Crystal Geyser, Utah

No speciation



High Pe & High Da

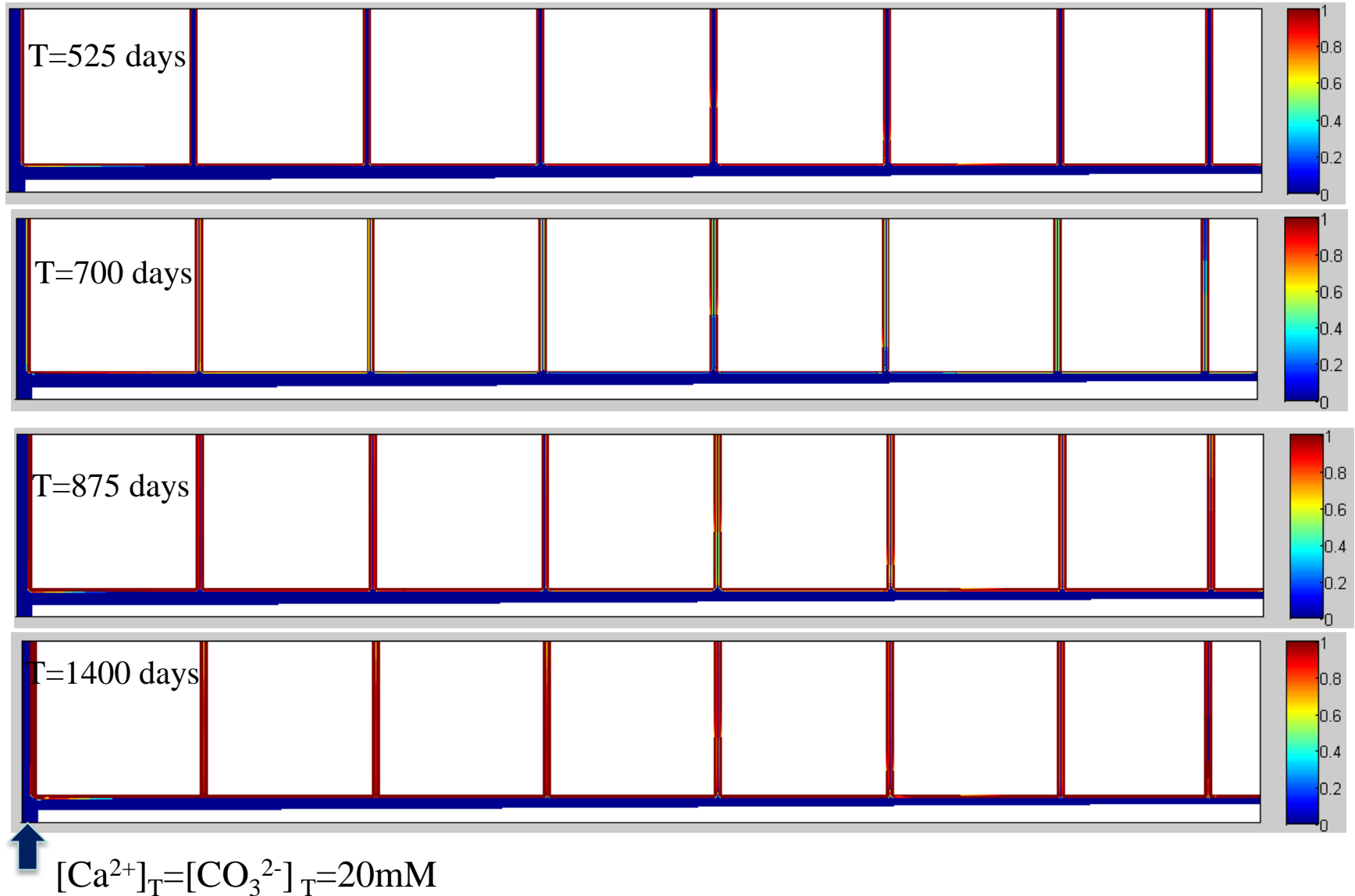
CaCO₃ volumetric content



$[\text{Ca}^{2+}]_{\text{T}} = [\text{CO}_3^{2-}]_{\text{T}} = 20\text{mM}$

High Pe & Medium Da

CaCO₃ volumetric content

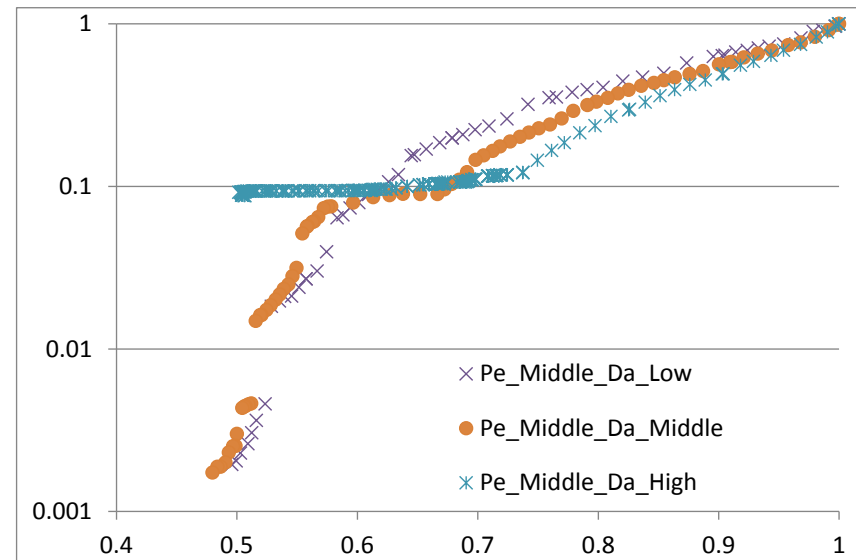
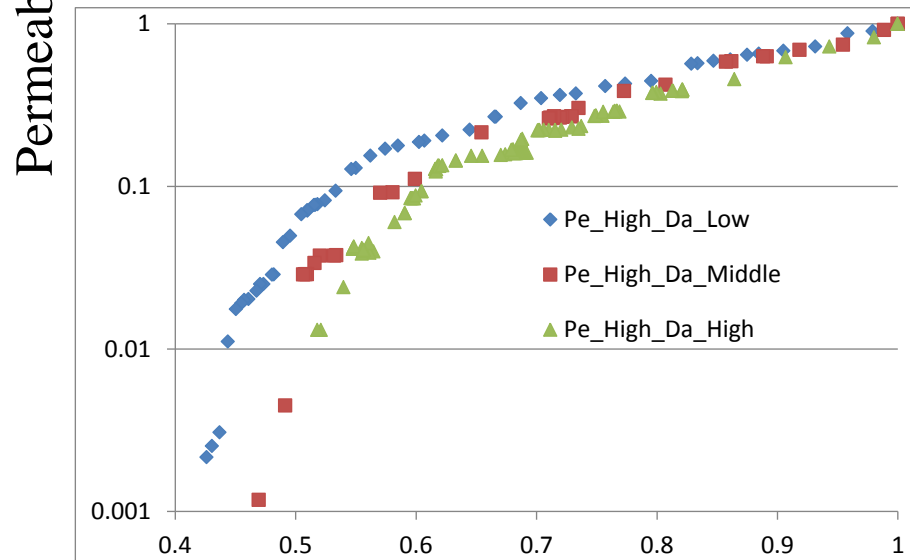
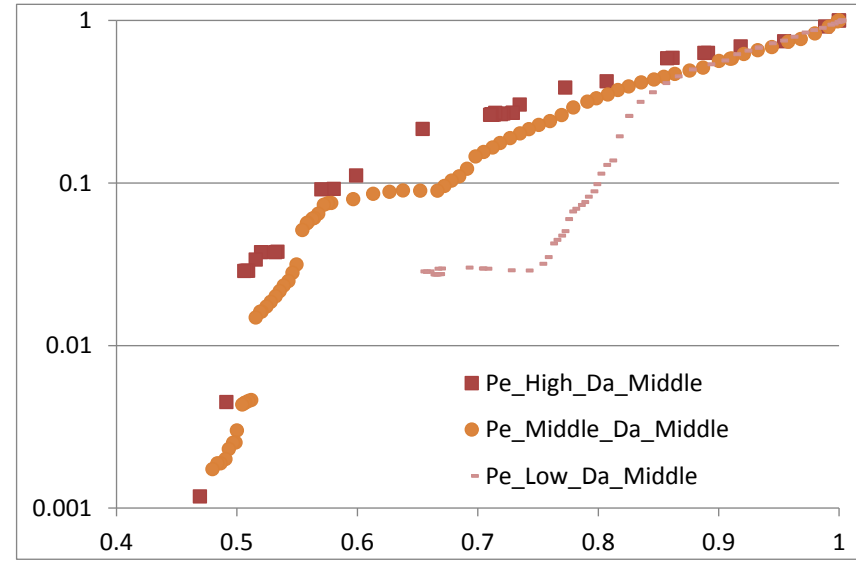
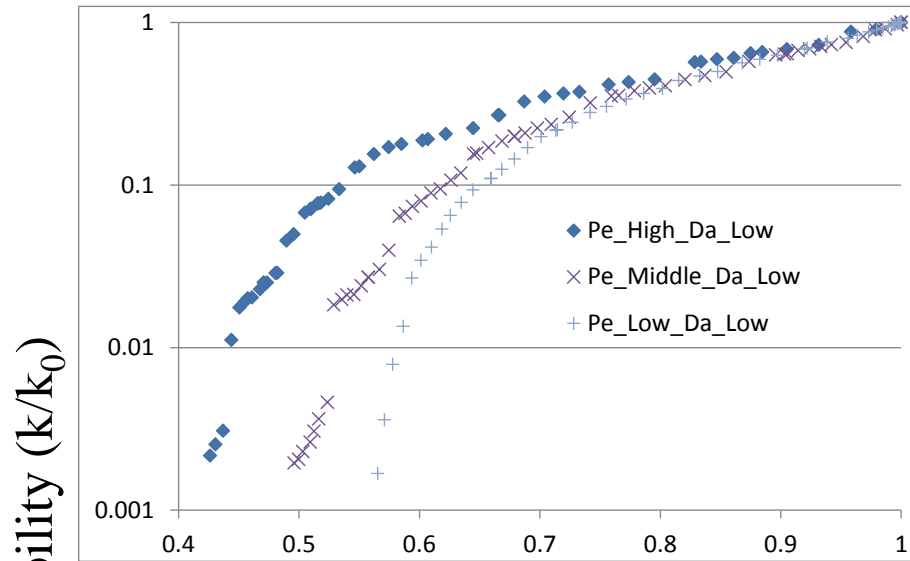


Summary

- Vigorously tested pore-scale model was used to develop a response function (or dimension reduction model) for continuum-scale permeability and porosity (k - ϵ) relationships
- Pore scale model was able to qualitatively capture pore clogging patterns observed at the Little Grand Wash Fault
- An adaptive strategy to couple pore- and continuum scale using a response function approach will be tested against travertine patterns observed in the Little Grand Wash Fault
- Algorithms developed in this work will be implemented into a continuum scale reactive transport model (p-FLOTRAN)

Questions?

Permeability-Porosity Relationships

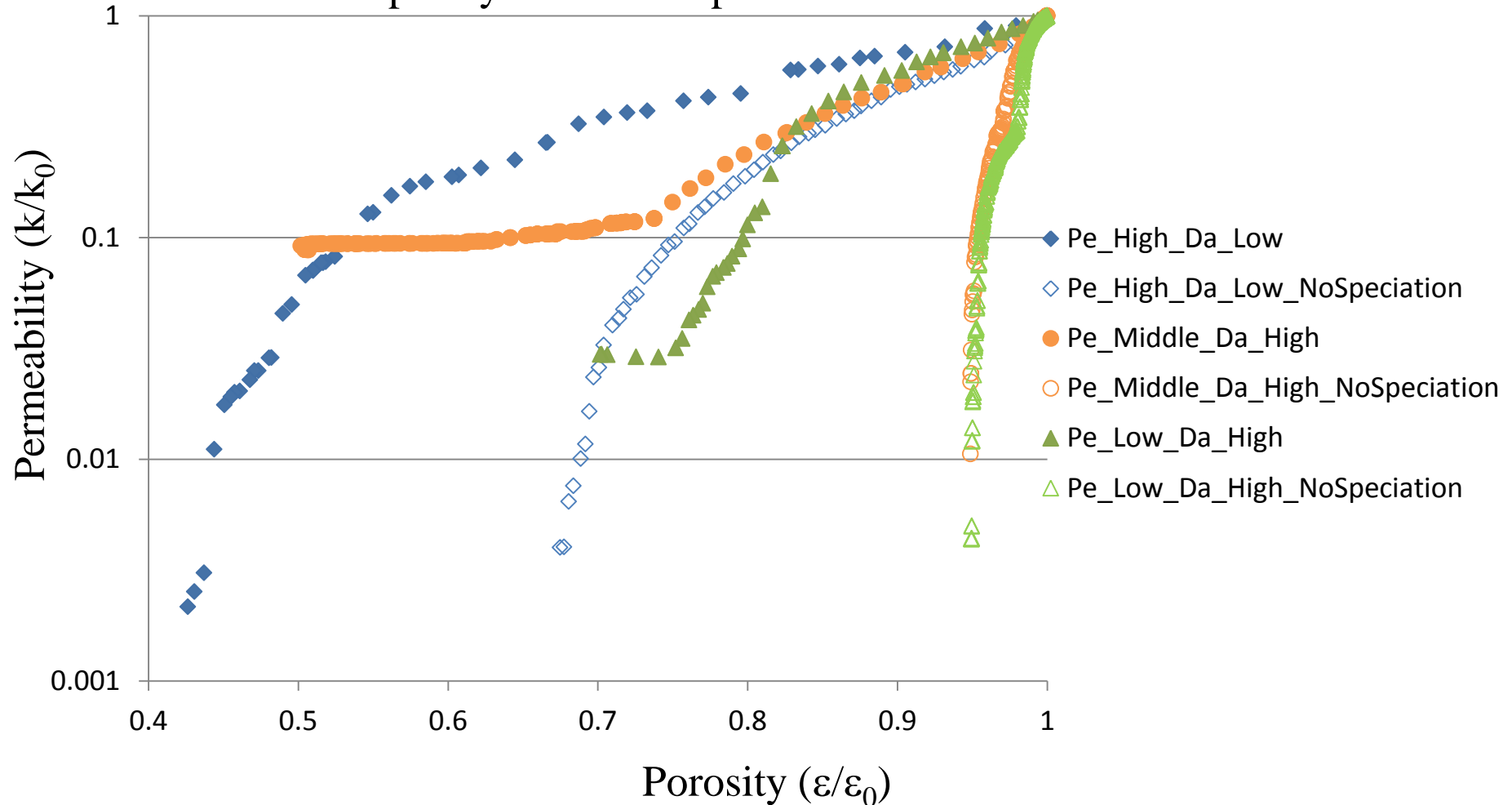


Porosity (ϵ/ϵ_0)

Chemical Speciation

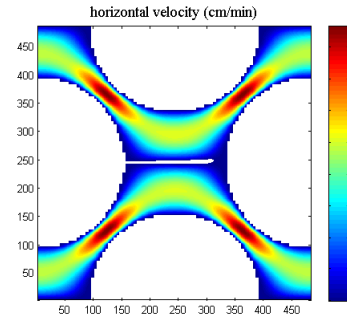
Solid symbols: Speciation

Open symbols: No speciation



Pore Scale Model Framework

Lattice Boltzmann Method:
Velocity field (u) at pore scale



Velocity at 1 micron resolution

Finite Volume Method: Reactive transport at pore scale

Δt

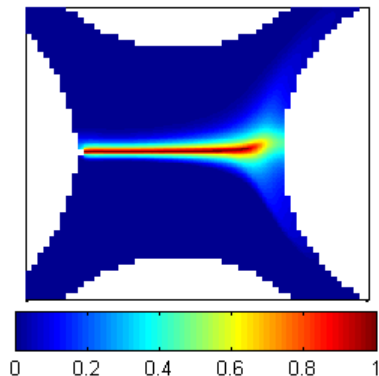
$\Psi_j = C_j + \sum_{i=1}^{N_{eq}} \nu_{ji} C_i$ Chemical equilibrium in bulk fluid (e.g., H^+ , HCO_3^- , ...)
Extended Debye-Hückel Equation for activity coefficients

$$D \frac{\partial \Psi_j}{\partial \mathbf{n}} = -I_m \quad \text{on reactive surface}$$

$$I_m = k_{cc} \left([\Omega]^n - 1 \right)^m \quad \Omega = \frac{Q_{CC}}{K_{sp}} \text{ or } \ln \left(\frac{Q_{CC}}{K_{sp}} \right)$$

Update of $CaCO_3$ volumetric content (V_m)

$$\frac{\partial V_m}{\partial t} = \overline{V}_m s_m k_{cc} \left(\left[\frac{a_{Ca^{2+}} a_{CO_3^{2-}}}{K_{sp}} \right]^n - 1 \right)^m$$



Mineral phase
volumetric content

Fracture network

