

# **Characterization of devices fabricated from electrostatically transferred graphene: comparison with epitaxial based devices**

**Stephen W. Howell**

**L.B. Biedermann, T. Ohta, W. Pan, A.J. Ross, T.E. Beechem and D.C. Trotter**

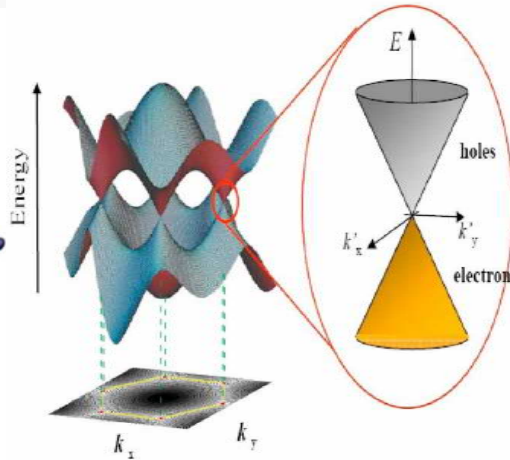
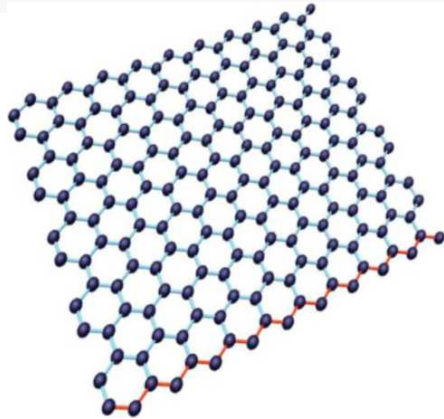
**Sandia National Laboratories**

**October 20, 2010**

Sandia is a multiprogram laboratory operated by Sandia Corporation, a Lockheed Martin Company, for the United States Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

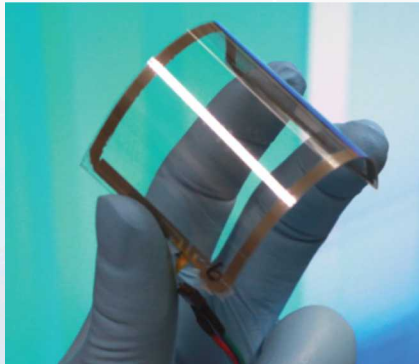


# Why graphene?



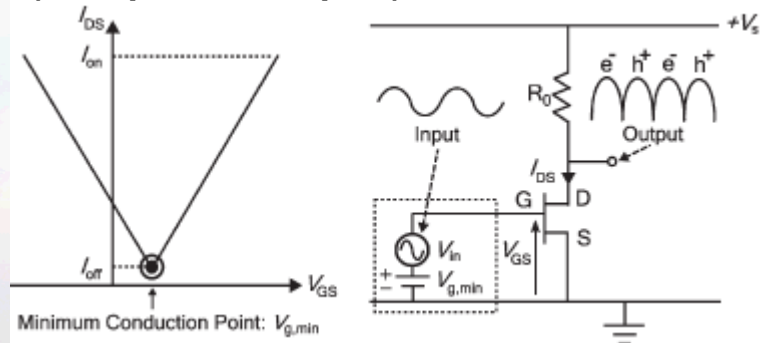
Mobility,  $\mu$ , up to 250,000 cm<sup>2</sup>/Vs  
(suspended exfoliated graphene)  
Ambipolar, zero-bandgap  
Current densities up to  $5 \times 10^8$  A/cm<sup>2</sup>  
Elastic modulus  $\sim 1$  TPa

## Transparent electrodes



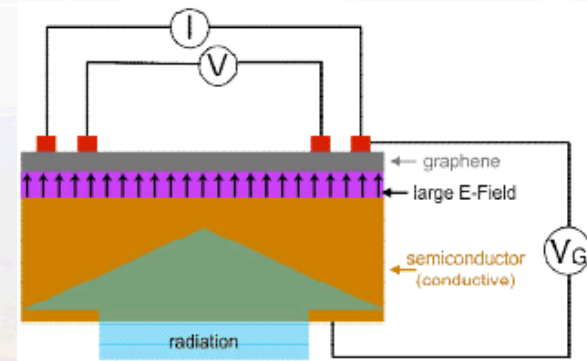
S. Bae et al, *Nature Nano.*,  
5, 574-578 (2010)

## Frequency multiplier (ambipolar transport)



H. Wang et al., *IEEE Elec. Dev. Lett.*,  
30, 547-549 (2009).

## Read-out for radiation detection



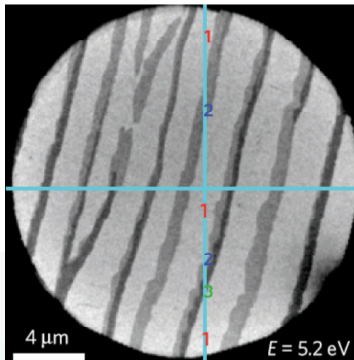
M. Foxe et al., *IEEE Trans. Nuclear Sci.*, submitted (2010).



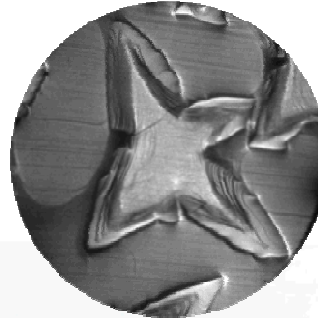
# Sandia's graphene effort

- Sandia is in the 3rd year of an internally funded project that is focused on developing the scientific biases required for synthesis of high quality graphene

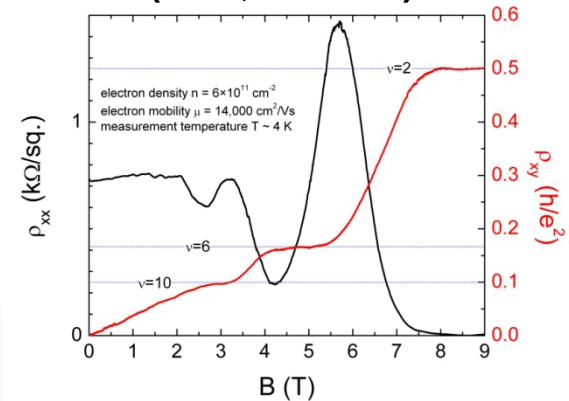
Thermal Decomposition  
of SiC (Ohta)



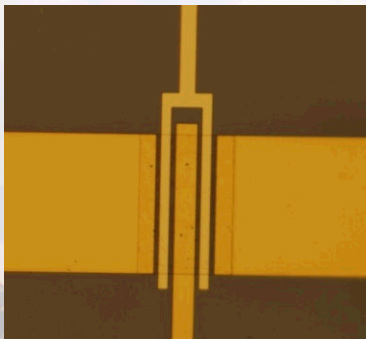
CVD Dep on Metals  
(McCarty)



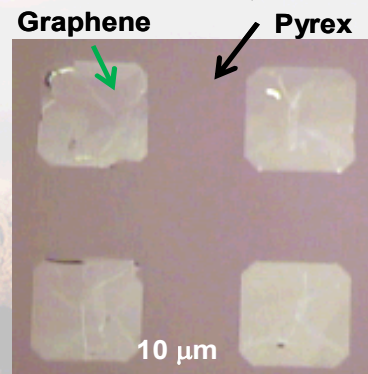
Transport Measurements  
(Pan, Howell)



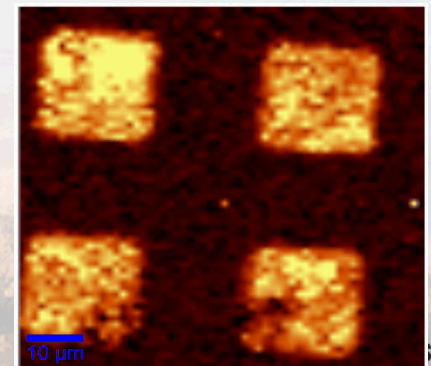
GFET Development  
(Howell)



Transfer  
(Biedermann, Howell)



Raman Mapping  
(Beechem, Ohta)







# Motivation for transferring epitaxial graphene (EG) from SiC

## ■ Why use EG on SiC?

- EG grown on SiC has the potential for high carrier mobility
- Current synthesis approaches allow for tight control of graphene thickness (monolayer and bilayer coverage)

## ■ Development of a scalable transfer technology could:

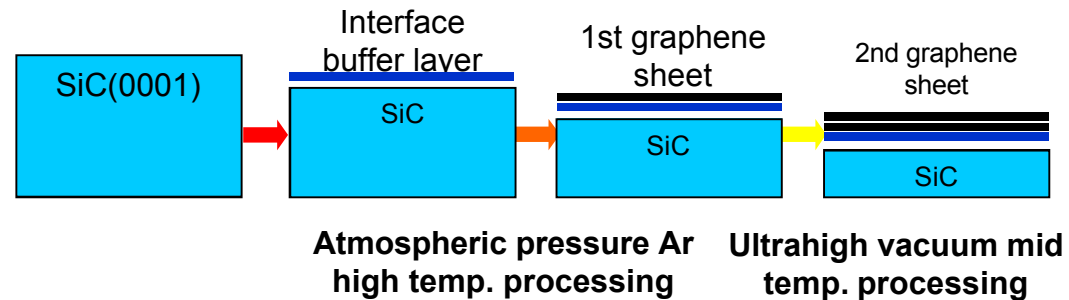
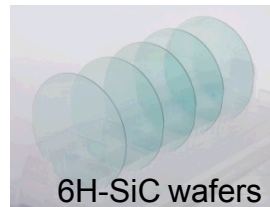
- Enable EG integration with current CMOS processes
- Enable new types of MEMS device structures



# Synthesis approach for epitaxial graphene (EG) on SiC

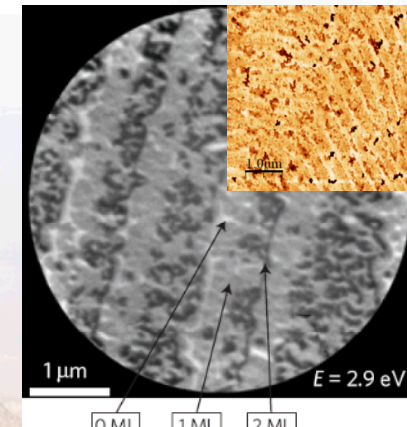
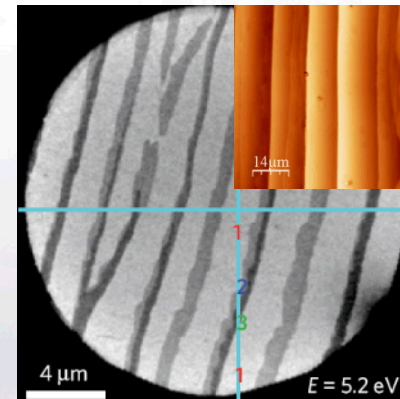
## ■ Graphitization of SiC:

- Sublimation of Si at high temperature ( $>1200\text{ }^{\circ}\text{C}$ ) leaves graphene layer at SiC surface



## • Argon-assisted graphene formation yields large ( $\sim 100\text{ }\mu\text{m}^2$ ) graphene domains

- Samples prepared using Ar atmosphere at atmospheric pressure and high temp
- Using this method we have exquisite control of :
  - Domain size
  - Percentage coverage of mono/bilayers



K. V. Emtsev et al., Nature Mater. 8, 203 (2009).

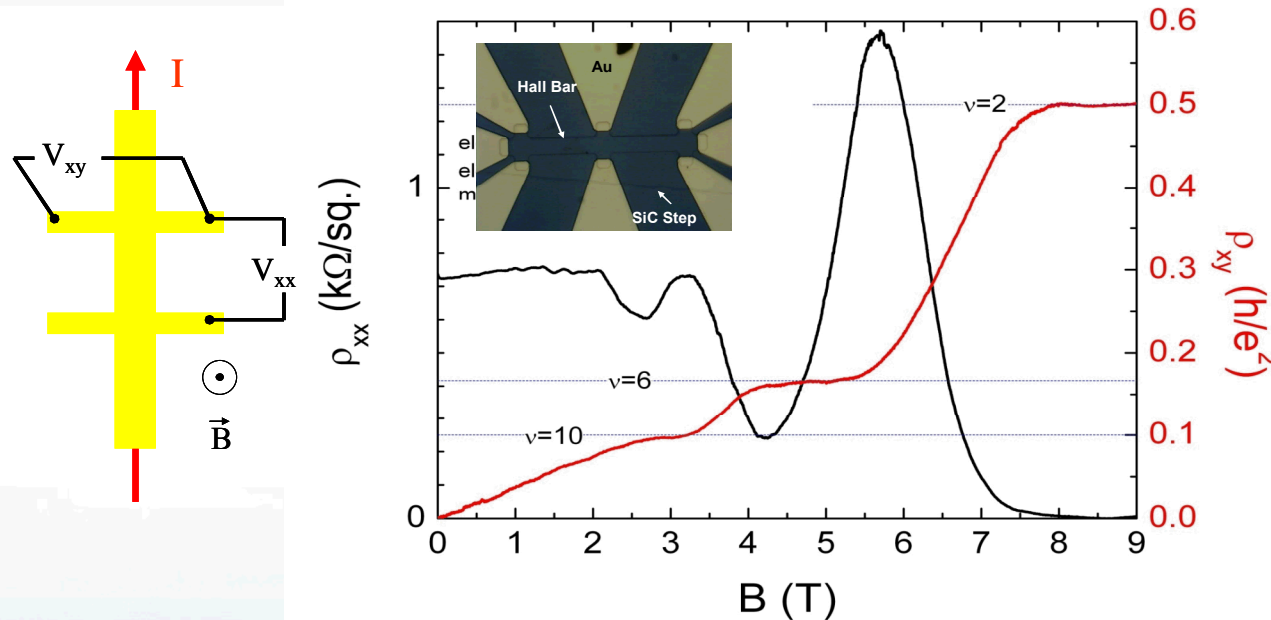
C. Virojanadara et al., Phys. Rev. B 78, 245403 (2008)



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# Electronic characterization of EG grown on SiC

## Low Temp Transport Measurements



•Epi-graphene electron mobility:  
14,000  $cm^2/Vs$

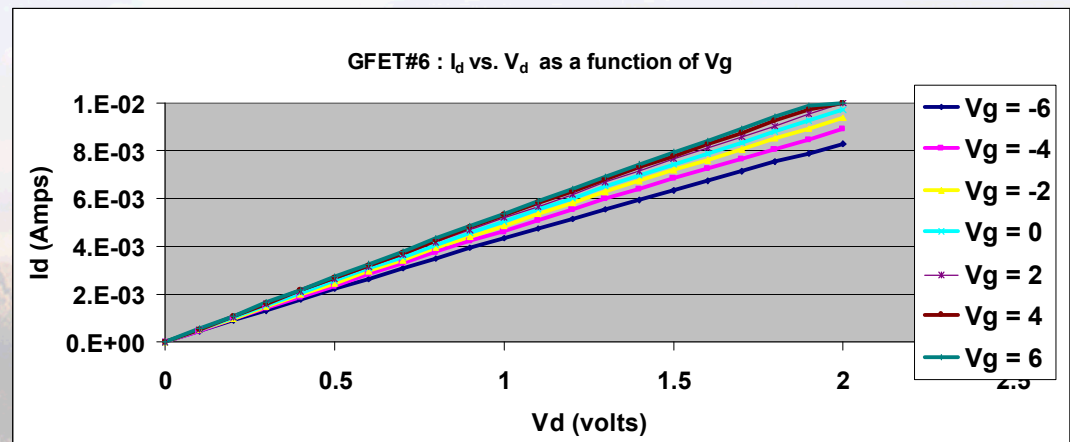
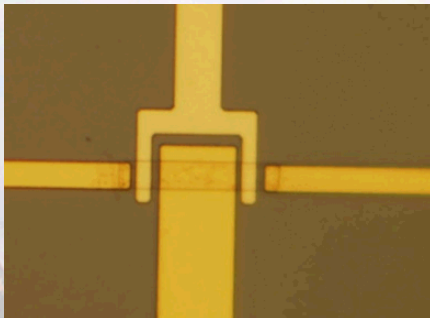
•Electron density:  $6 \times 10^{11} cm^{-2}$

•Graphene sheet resistance:  
 $\sim 1600 \Omega/sq$  (average from 12 devices)

•Observed IOHE on 3 devices on the same chip

Pan *et al.*, accepted by APL

## GFET Development

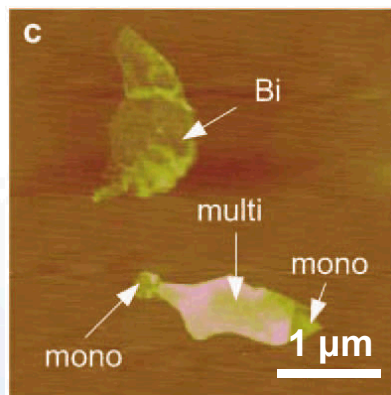




# Methods to transfer EG to SiO<sub>2</sub>

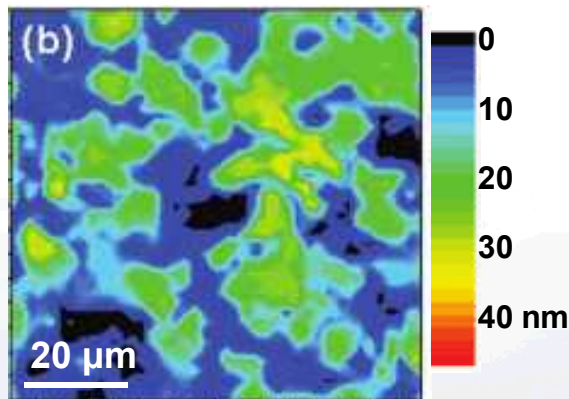


**Thermal release tape  
+ 5 N/mm<sup>2</sup> pressure**



Si-face EG  
Small flakes

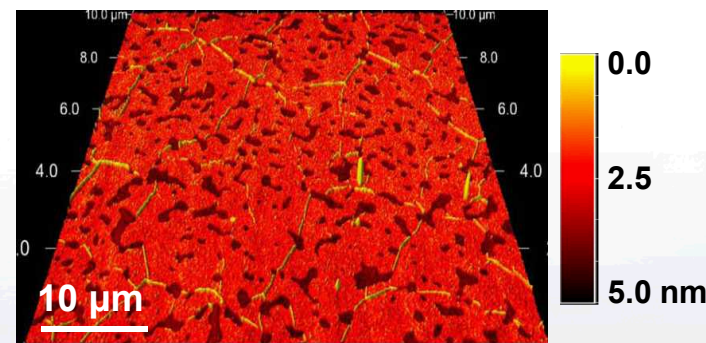
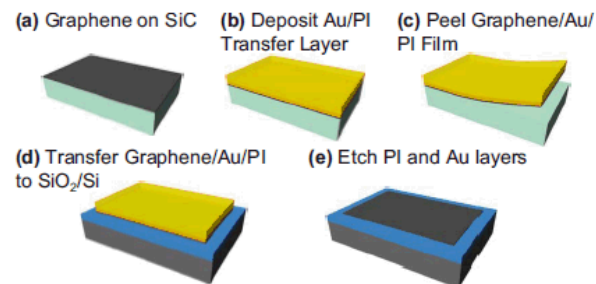
D. Lee *et al.*, *Nano Lett.* **8**, 4320-5 (2008).



C-face EG  
Very thick graphene  
 $\mu \sim 1350 \text{ cm}^2/\text{Vs}$

J. Caldwell *et al.*, *ACS Nano.* **4**, 118-14 (2010).

## Gold/polyimide film handle



Si-face EG  
Damaged graphene  
 $\mu \sim 100 \text{ cm}^2/\text{Vs}$

S. Unarunoai *et al.*, *APL.* **95**, 202101 (2009).



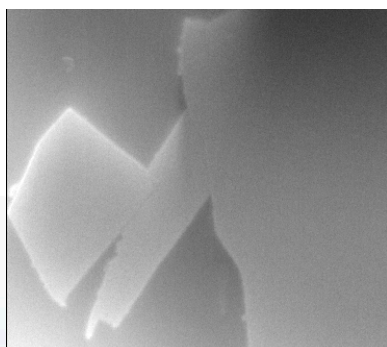
# Voltage-driven exfoliation of graphite



Graphite to Pyrex  
**1.2 – 1.7 kV**

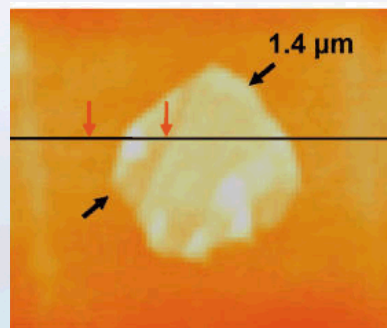
$\mu \sim 10^4 \text{ cm}^2/\text{Vs}$

A. Shukla *et al.*, *Solid State Comm.* **149**, 718-21 (2009).



HOPG to  $\text{SiO}_2$   
**3-30 kV**

A. Sidorov *et al.*,  
*Nanotechnology* **18**,  
135301 (2007).

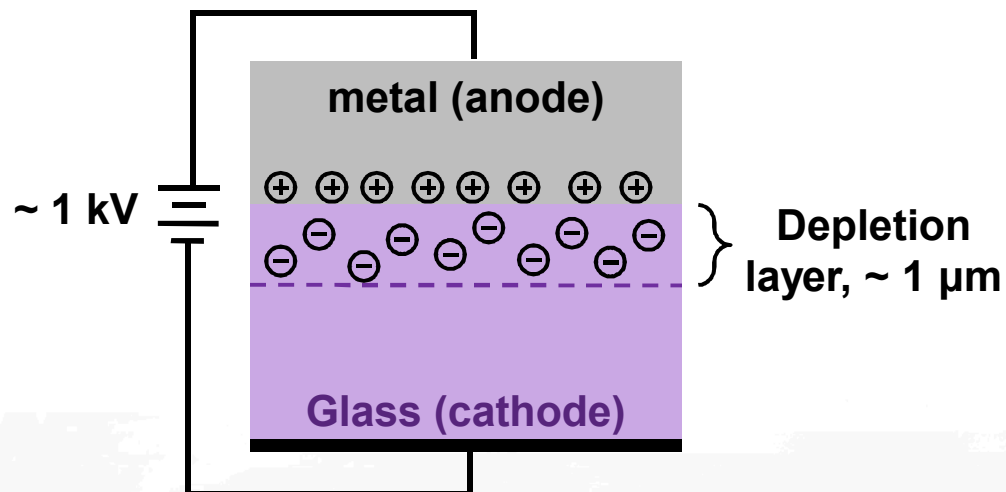


Pre-patterned  
HOPG to  $\text{SiO}_2$   
**8.5 V**

$\mu \sim 1050 \text{ cm}^2/\text{Vs}$

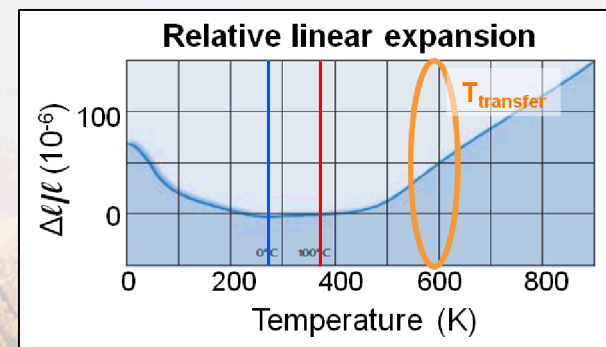
X. Liang *et al.*, *Nano Lett.* **9**, 467-72 (2009).

Shukla's method is derived from anodic bonding.



Common glasses are Pyrex, borosilicate glass, and Zerodur.

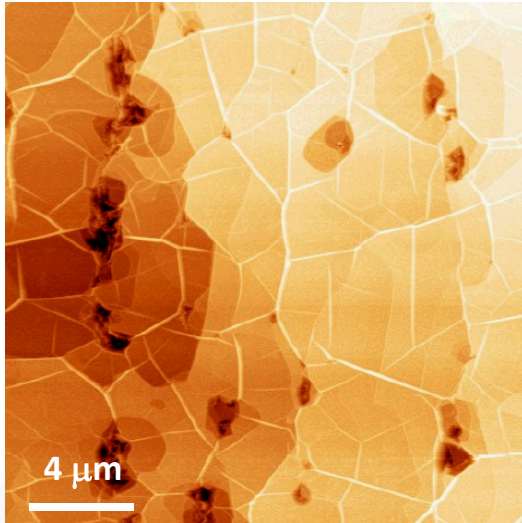
Zerodur is a  $\text{Li}_2\text{O}_3$ -  
 $\text{Al}_2\text{O}_3$ - $\text{SiO}_2$  glass  
ceramic



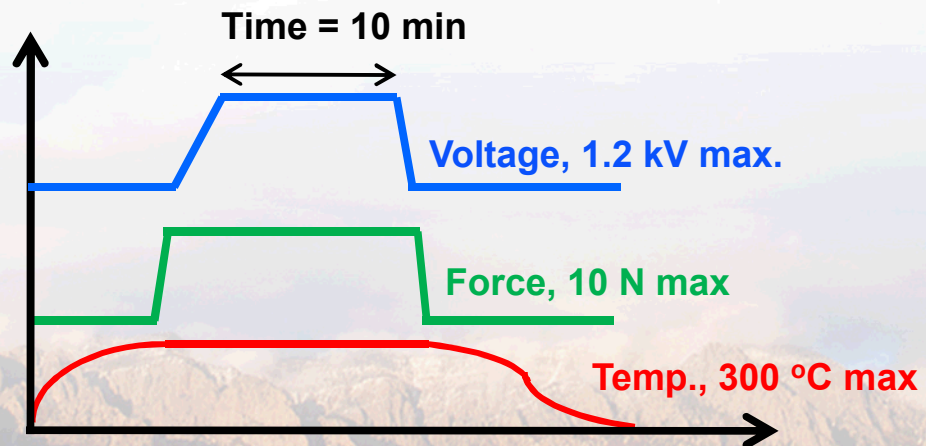
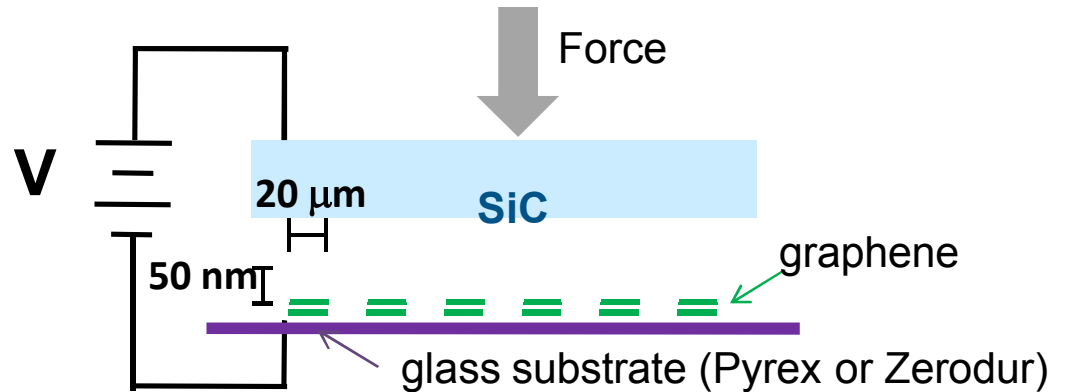
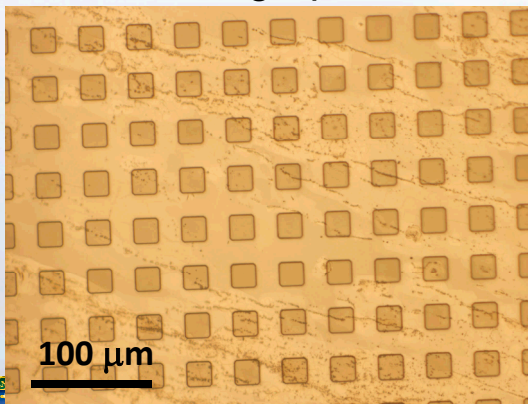


# Graphene transfer procedure

As-grown C-face graphene

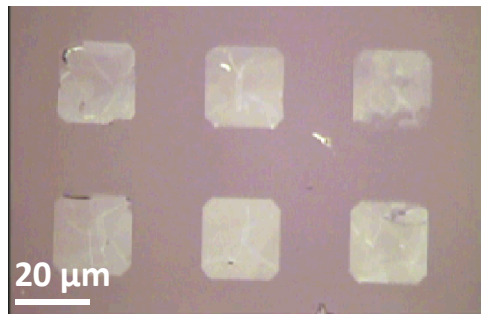


Patterned graphene

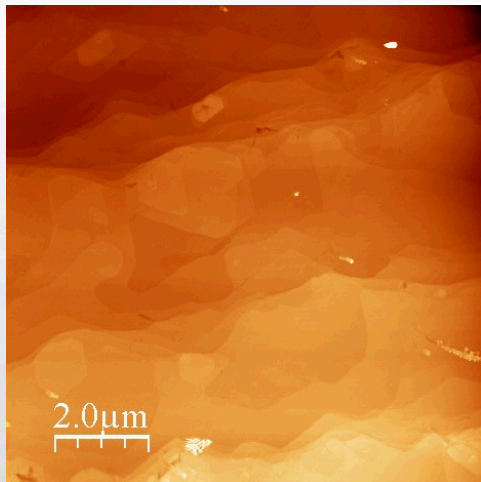


# How to judge graphene transfer

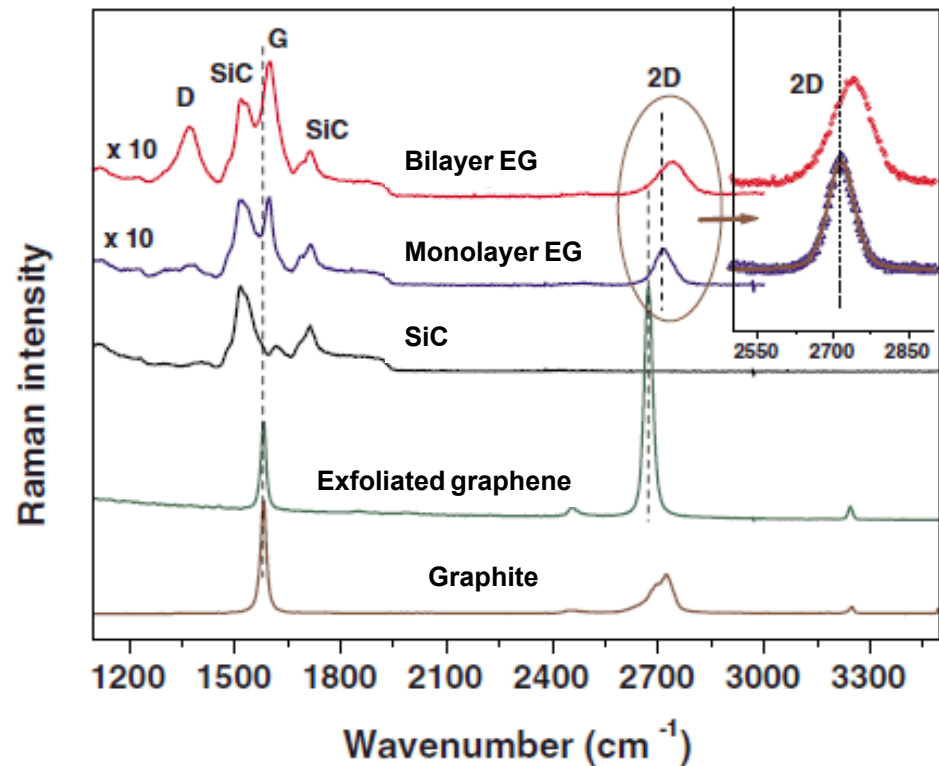
Optical microscopy



Atomic force microscopy



Raman spectroscopy



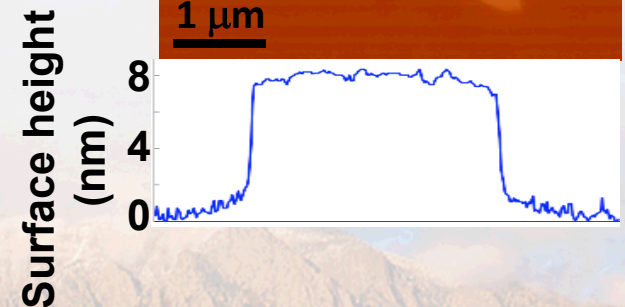
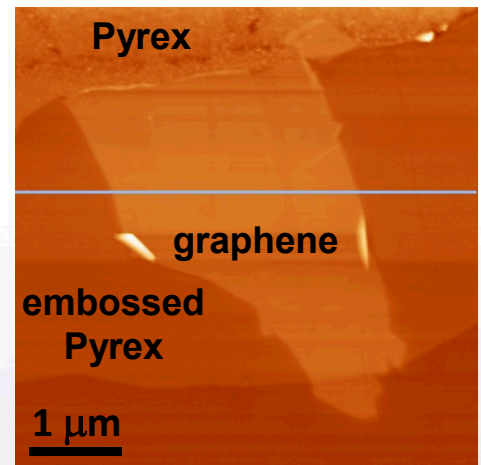
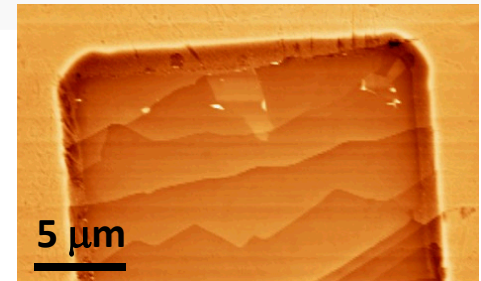
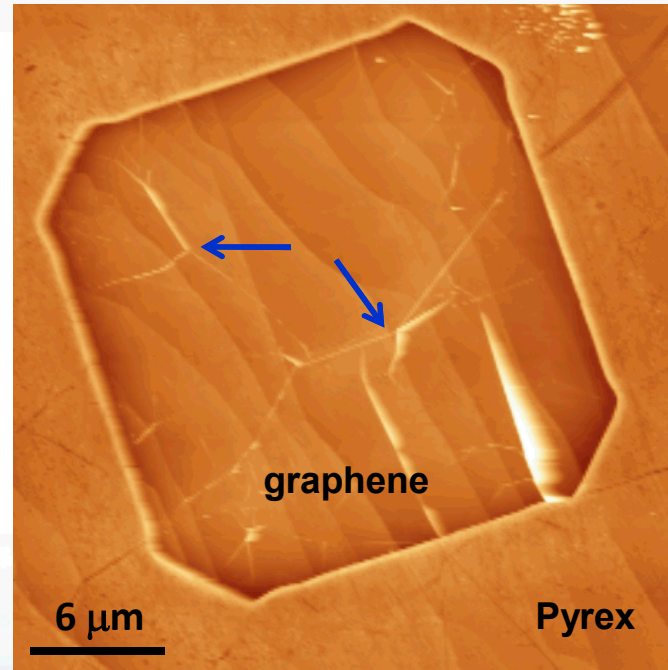
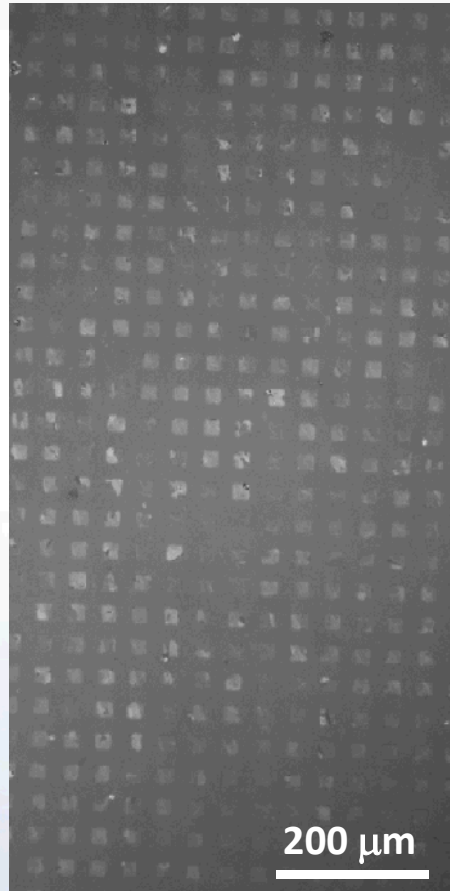
D-band:  $\sim 1350 \text{ cm}^{-1}$

G-band:  $\sim 1580 \text{ cm}^{-1}$

2D-band:  $\sim 2700 \text{ cm}^{-1}$



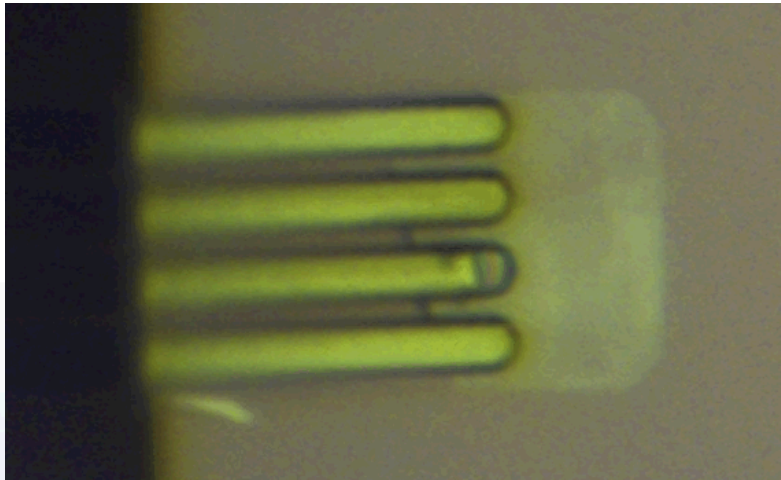
# C-face graphene squares transferred to Pyrex





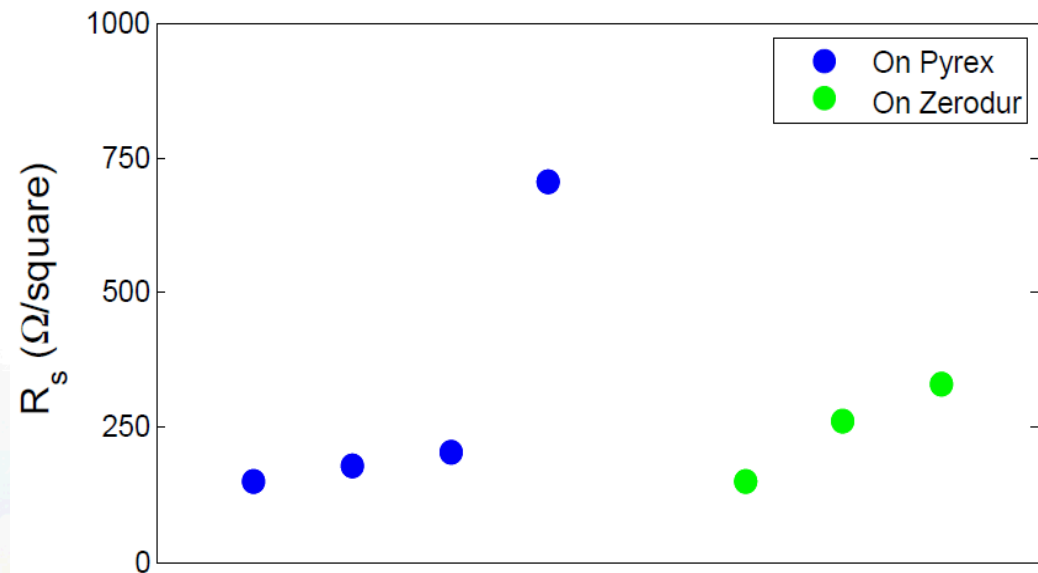
# Electronic characterization of transferred graphene from SiC (C-face)

Graphene transferred from the C-face has  $R_s$  as low as  $150 \Omega/\square$ .



Micro four probe on  $10 \times 10 \mu\text{m}^2$  square of transferred graphene

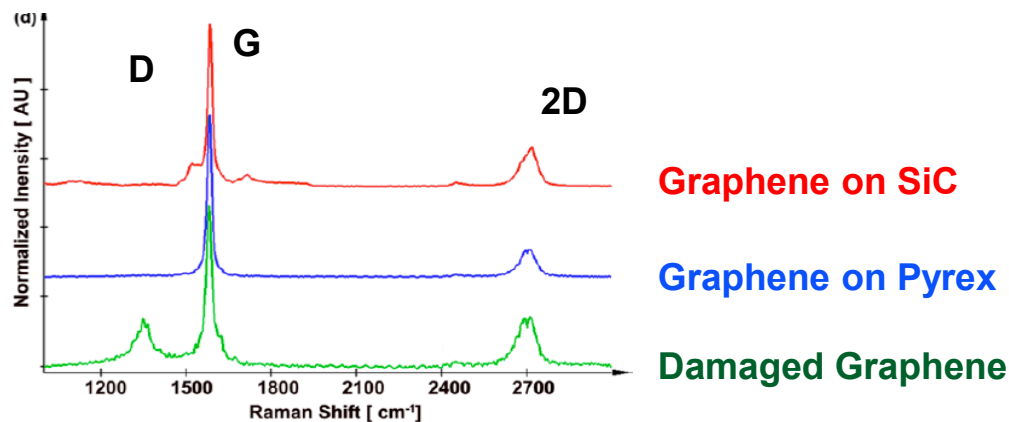
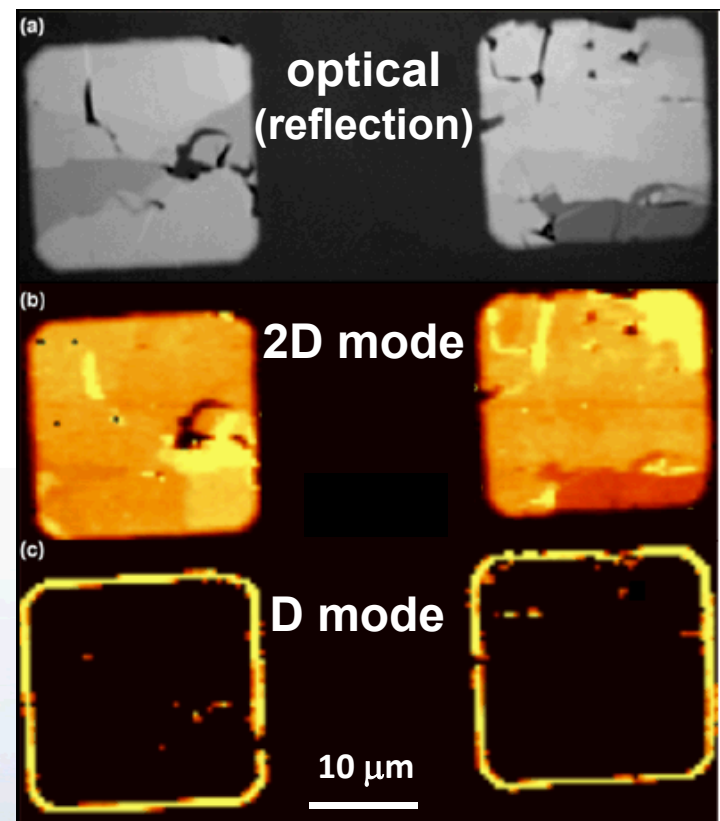
$$R_s = \frac{\pi}{\ln 2} \frac{V}{I}$$



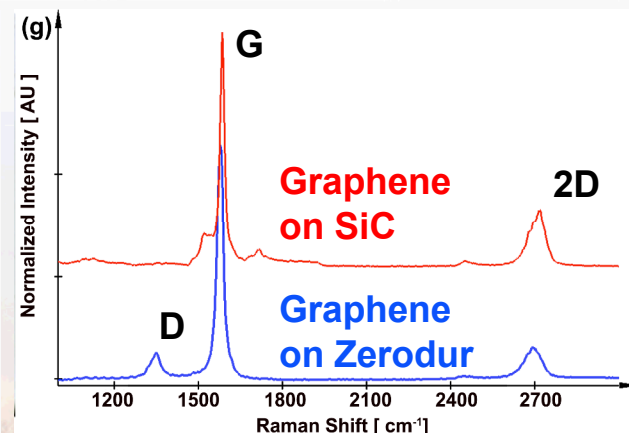
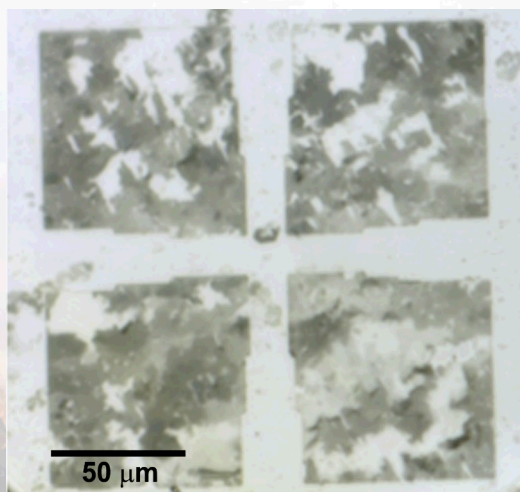
L. Biedermann, T. Beechem, A. Ross, T. Ohta, and S. Howell,  
submitted to *New J. of Physics*.

# Raman analysis confirms graphene transfer

## Graphene on Pyrex



## Graphene on Zerodur

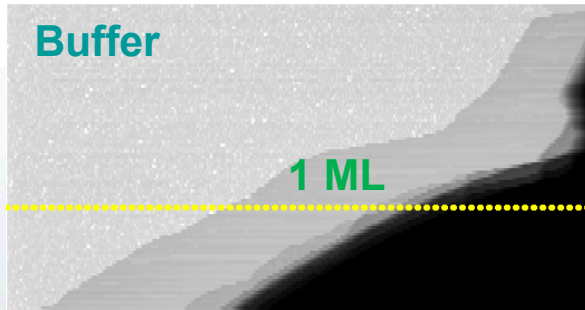
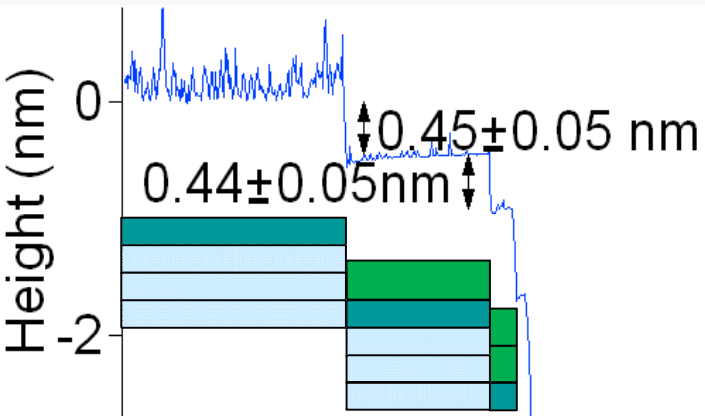


L. Biedermann, T. Beechem, A. Ross, T. Ohta, and S. Howell,  
submitted to *New J. of Physics*.



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# About-face: Si-face epitaxial graphene

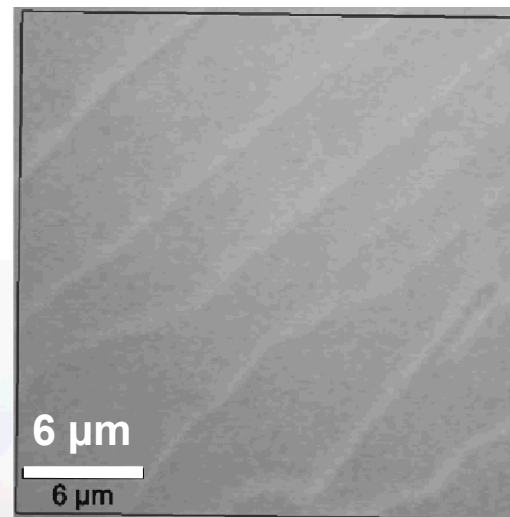


T. Ohta *et al.* *Phys Rev B* **81**, 121411(R) (2010).

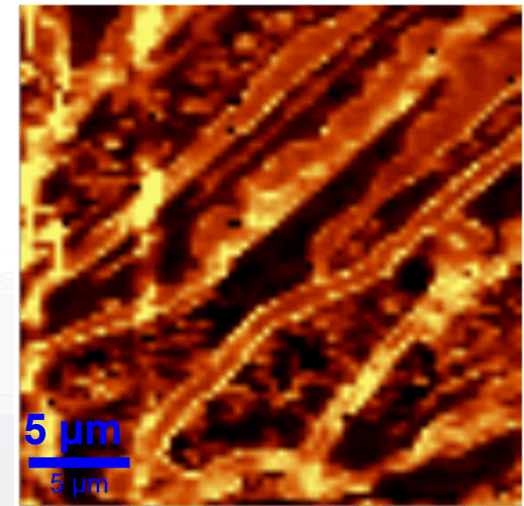
Sample 090728 #27

60 % 1 ML, 30 % 2 ML, 10 %  $\geq 3$  ML

**Optical**



**2D FWHM**



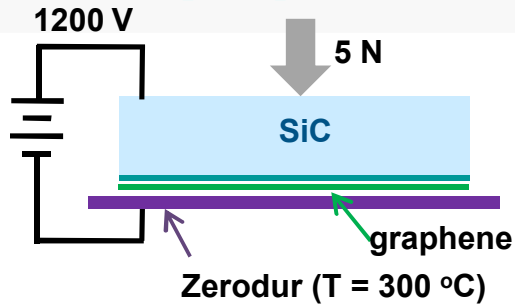
**Mobility at 4 K:  $\mu = 14000 \text{ cm}^2/\text{Vs}$**

Wei Pan, Sandia

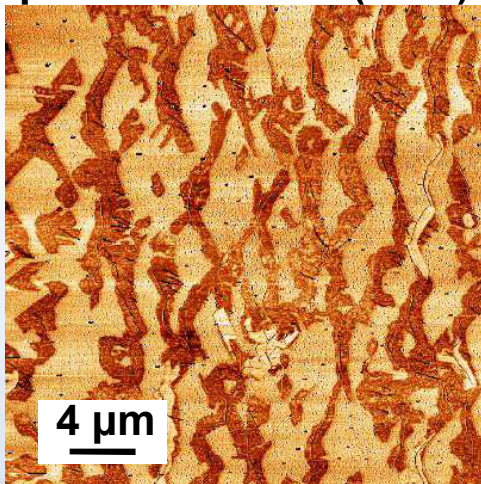




# Chip-scale transfer of Si-face graphene to Zerodur

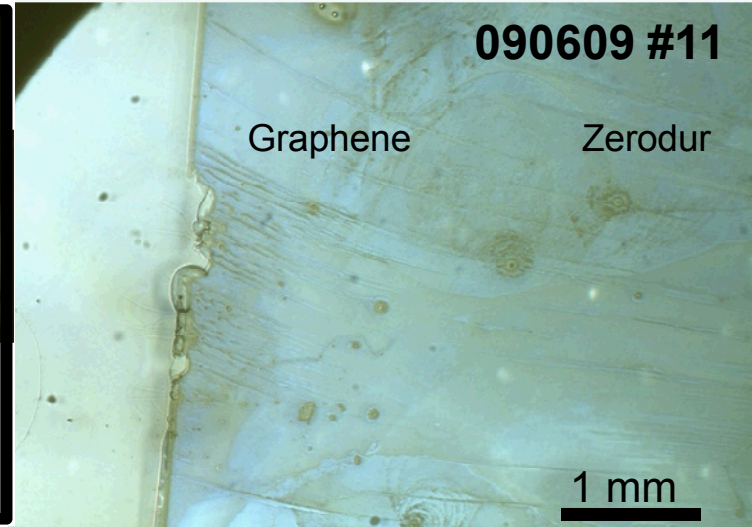
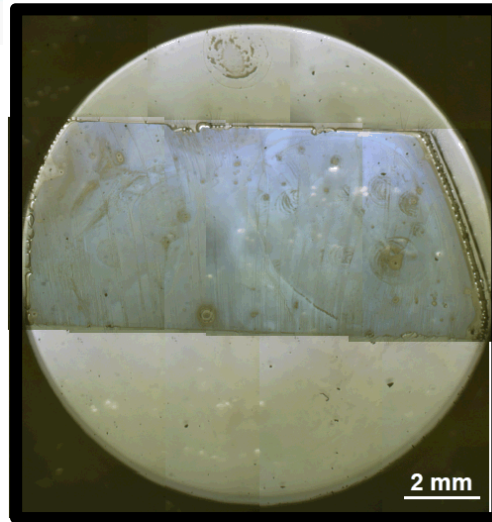


Graphene on SiC,  
prior to transfer (AFM)

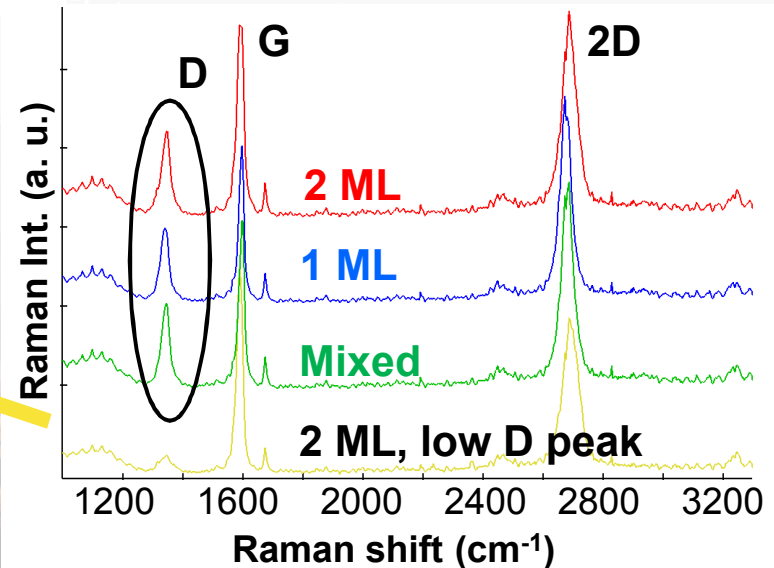
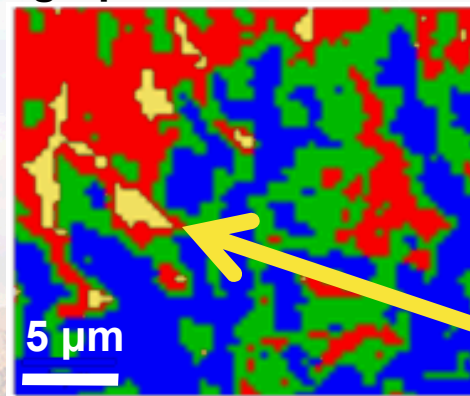


Bright phase = 1 ML

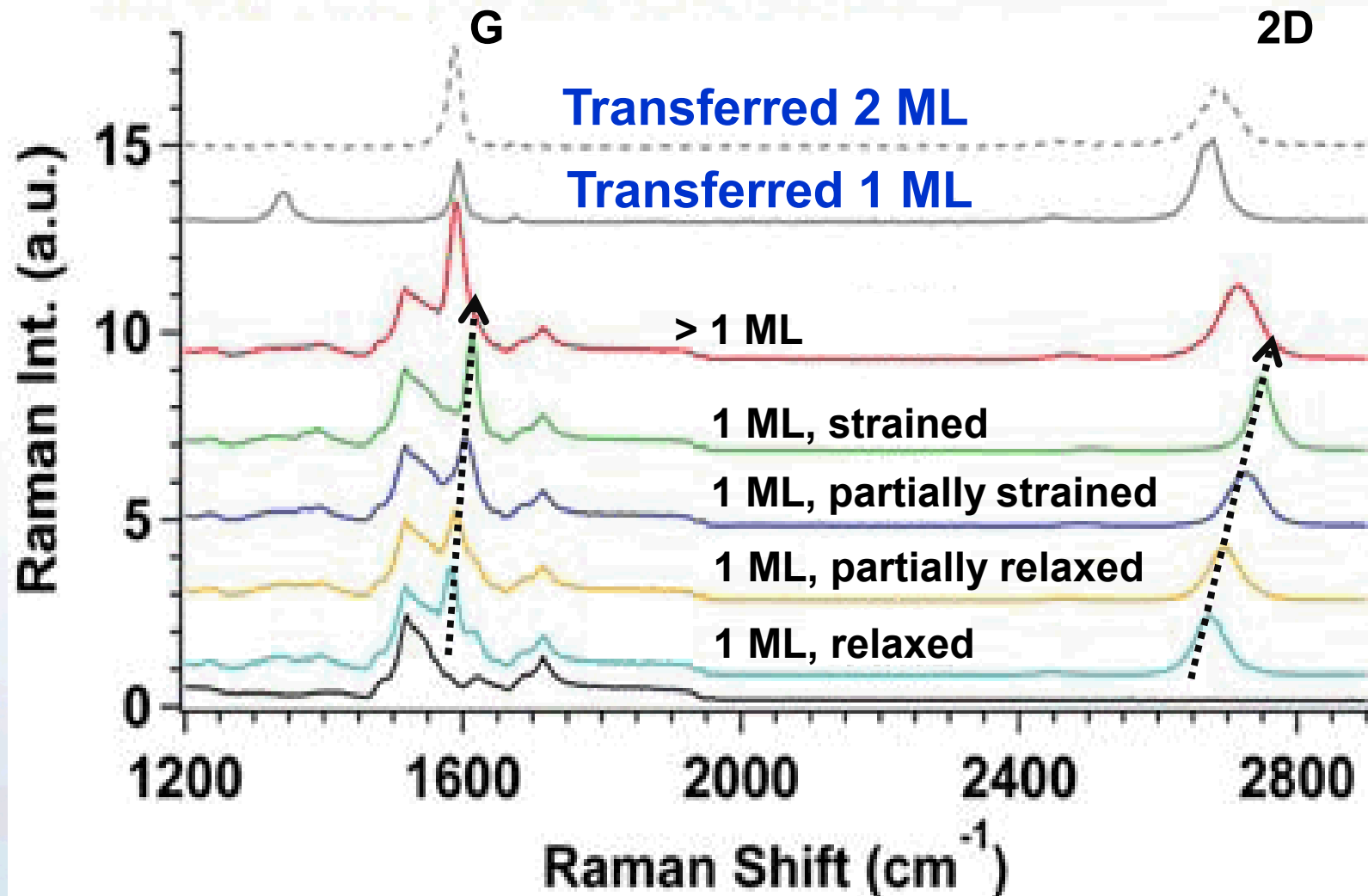
Dark phase = 2 ML



Cluster analysis of  
graphene on Zerodur



# Transfer relaxes strain inherent in epitaxial graphene



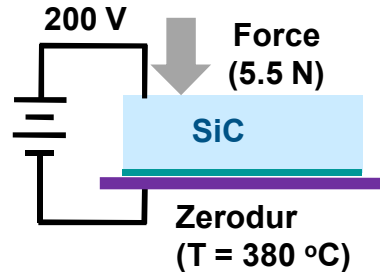
D. Schmidt *et al.*, submitted to *Nano Letters* (2010).



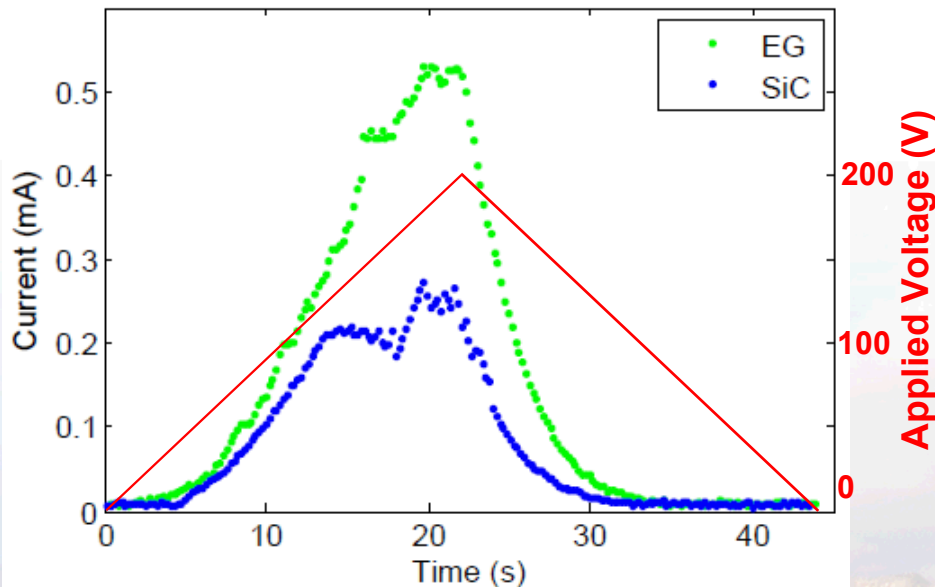
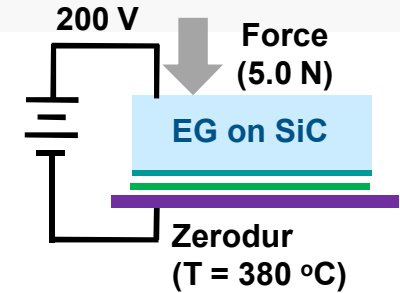
Sandia National Laboratories

# Electrochemical analysis suggests graphene is bonding to the Zerodur

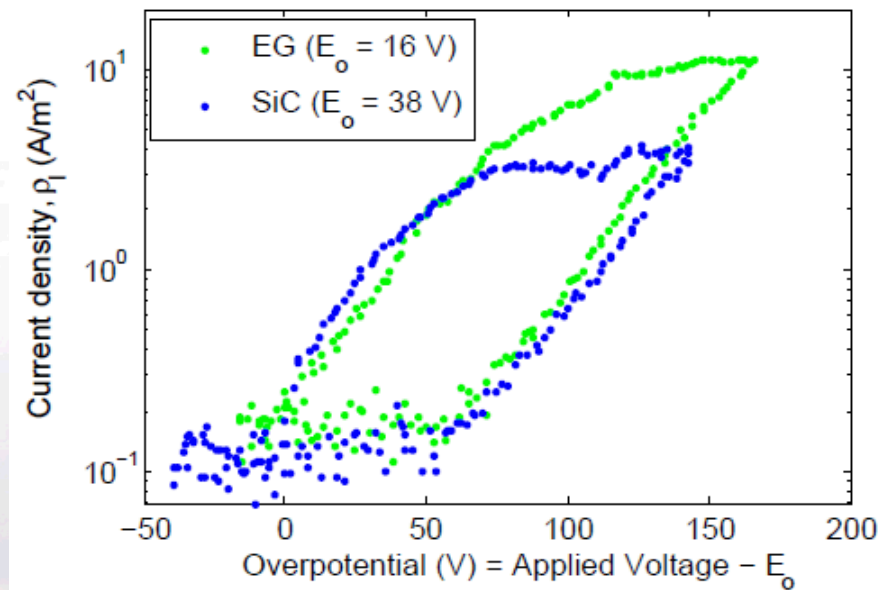
Anodic bonding of SiC to Zerodur



Transfer of EG to Zerodur



•Zerodur cannot be removed without breaking SiC



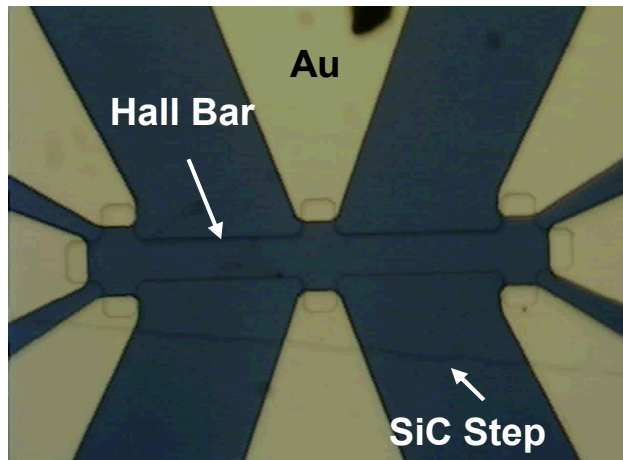
•Zerodur releases from graphene/Zerodur with little force





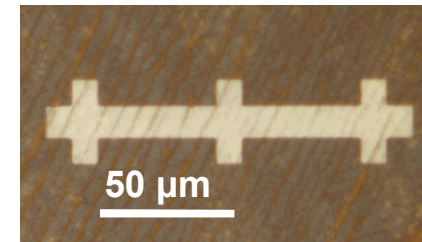
# Comparison of transferred Si-face graphene to EG

EG Hall bar on SiC(0001)



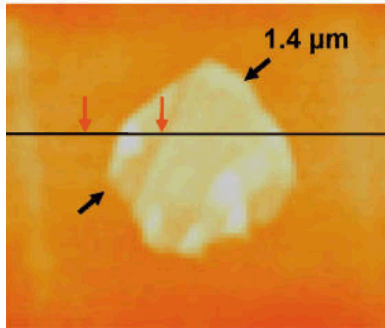
- Epi-graphene electron mobility: 14,000  $\text{cm}^2/\text{Vs}$
- Electron density:  $6 \times 10^{11} \text{ cm}^{-2}$
- Graphene sheet resistance:  $\sim 1600 \Omega/\text{sq}$  (average from 12 devices)

Hall bars structures, Si-face graphene on Zerodur



- Poor large scale transfer (most Hall bars are open circuit across their length)
- Measurements across small distances ( $< 10 \mu\text{m}$ ) do show some electrical activity ( $\sim 5 \text{ k}\Omega - 5 \text{ M}\Omega$ )
- The transferred material appears to be isolate ribbons of graphene
  - Breaks could be caused by SiC steps

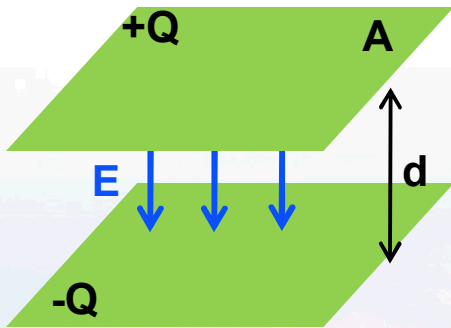
# Ongoing work: Electrostatic rather than electrochemical transfer



Pre-patterned  
HOPG to SiO<sub>2</sub>  
**8.5 V**  
 $\mu \sim 1050 \text{ cm}^2/\text{Vs}$

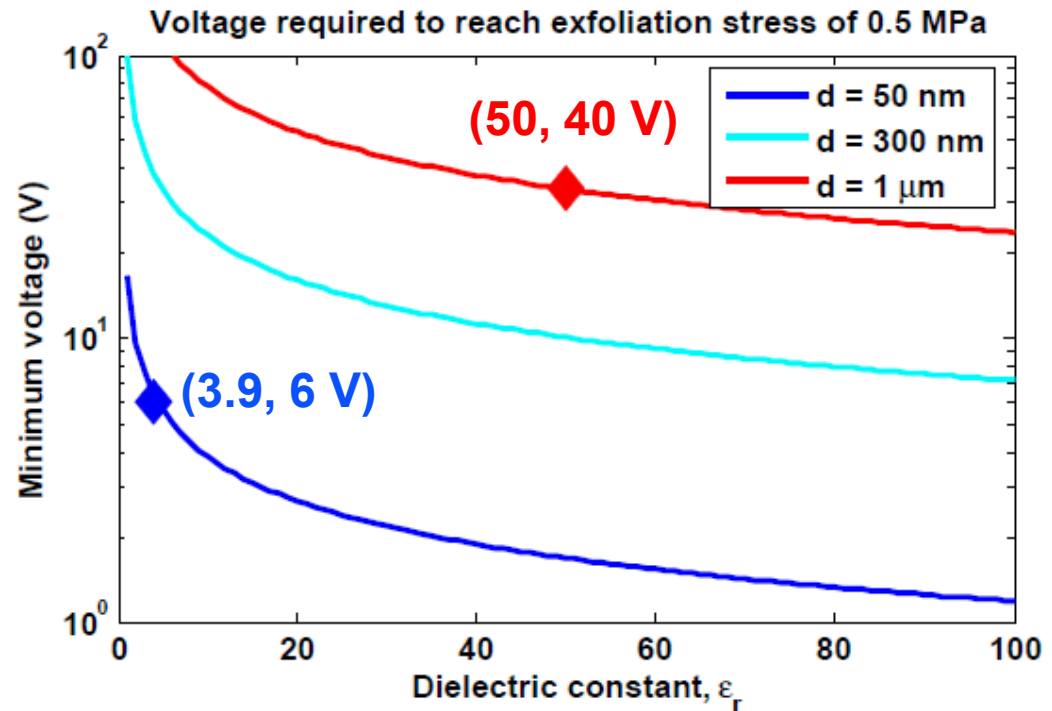
X. Liang *et al.*, *Nano Lett.* **9**, 467-72 (2009).

The exfoliation strength of graphite is  $\sim 0.4 \text{ MPa}$ .



$$\vec{F}_{cap} = Q\vec{E}$$

$$P_{cap} = \frac{\vec{F}_{cap}}{A} = \frac{\epsilon_o \epsilon_r V^2}{d^2}$$





# Conclusions

- Demonstrated transfer of graphene to Pyrex and Zerodur from the C-face of SiC
- Graphene transferred from C-face was characterized with Raman and micro four probe
  - The average resistance was found to be **XXX**  $\Omega/\text{Sq}$
- Attempted transfer of graphene monolayer from Si-face of SiC
- Characterized with Raman Spectroscopy
  - Detected presence of defects
  - Showed relaxation in strain
- Electrical characterization determined that the transferred films were not continuous over large length scales (transfer of ribbons)
- Current monitor during transfer indicated possible chemical bonding to Zerodur







# Graphene collaborators

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**Laura Biedermann**, Transfer

**Thomas Beechem**, Raman spectroscopy

**Taisuke Ohta**, epitaxial graphene growth

**Anthony (TJ) Ross**, lithography and etching

**Jeff Stevens**, etching

**Graham Yelton**, electrochemistry advice





# Backup Slides



# Explored transfer parameter space; transfer still damages graphene

	Range	090609 #11	081209 #30	090609 #33	090728 #06	090401 #03	0907278 #18
Voltage (kV)	0.2 – 1.2	1200	530	525	525	550	550
Force (N)	2 – 10	5.0	4.0	4.2	3.9	8.9	7.7
Time (min)	1 – 10	10	4	3.5	1	4	5
Temp (°C)	300 – 390	300	370	370	370	390	370
% 1 ML	45 – 85	50	45	80	80	85	75
I(D) / I(G)		1.0	1.3	1.0	0.7	1.4	1.2

Metric for graphene quality:

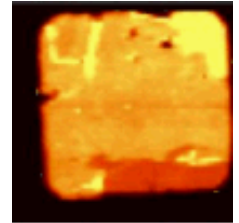
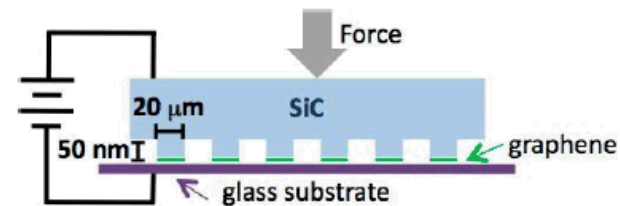
$$\frac{I(D)}{I(G)} = \frac{\text{D-peak intensity}}{\text{G-peak intensity}}$$





# Graphene transfer

- Scanning tunneling microscopy and atomic force microscopy
- Graphene transfer to insulating substrates
- Laser Doppler Vibrometry
- Casimir force measurements



## ■ LDRD 11-0876: Enabling Graphene Nanoelectronics

■ PI: Stephen W. Howell, PM: Carlos Gutierrez

■ **Core team:** Thomas Beechem, Tom Friedmann, Tim Lambert, Kevin Leung, Kevin McCarty, Taisuke Ohta, Wei Pan, Anthony Ross, Jeff Stevens, Cody Washburn, and David Wheeler

### ■ Publications submitted:

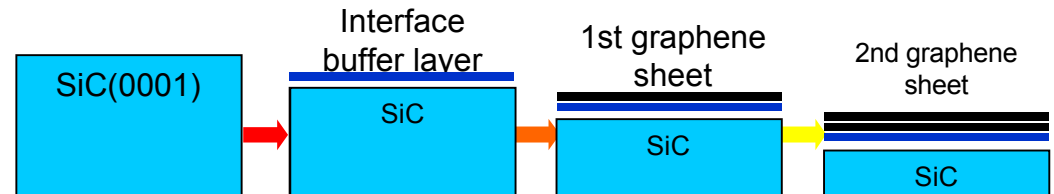
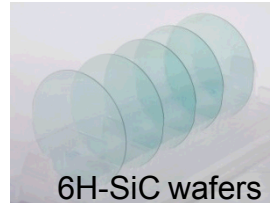
- L. Biedermann, T. Beechem, A. Ross, T. Ohta, and S. Howell, "Electrostatic transfer of patterned epitaxial graphene from SiC(000-1) to glass," submitted to *New J. of Physics*.
- D. Schmidt, T. Ohta, L. Biedermann, S. Howell, and G. Kellogg, "Inhomogeneous strain fields in epitaxial graphene," submitted to *PNAS*.



# Understanding Graphene Growth on SiC

## ■ Graphitization of SiC:

- Sublimation of Si at high temperature ( $>1200\text{ }^{\circ}\text{C}$ ) leaves graphene layer at SiC surface



## • Argon-assisted graphene formation yields large ( $\sim 100\text{ }\mu\text{m}^2$ ) graphene domains

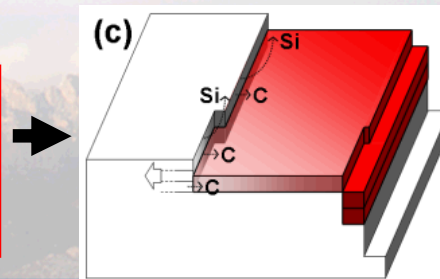
- Samples prepared using Ar atmosphere at atmospheric pressure and high temp
- Using this method we have exquisite control of :
  - Domain size
  - Percentage coverage of mono/bilayers

• Understanding the growth mechanism of graphene on SiC

- Growth morphology strongly depends on the step structure

### Step-flow growth

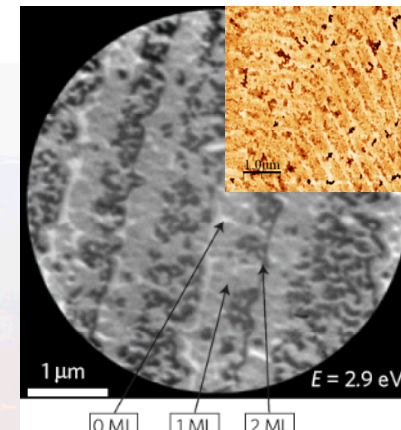
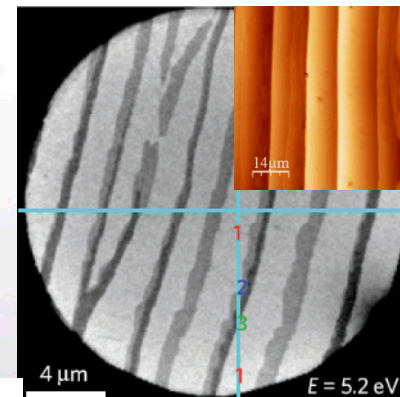
- Key for high large-area growth
- Real-time growth observations using LEEM



Atmospheric pressure Ar high temp. processing



Ultrahigh vacuum mid temp. processing



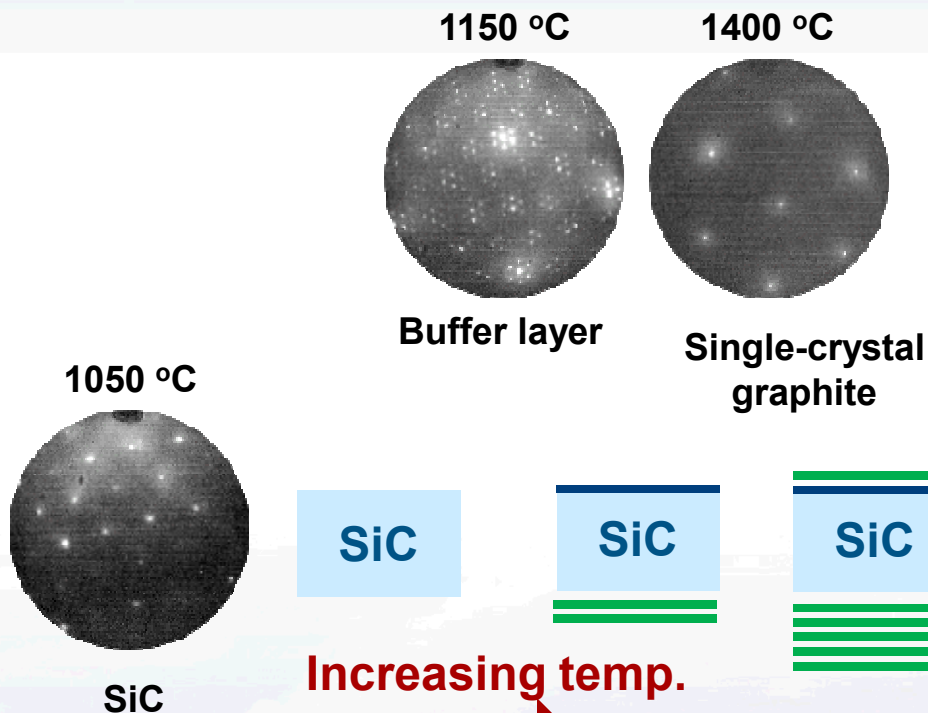
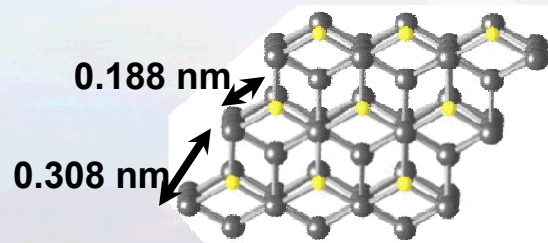
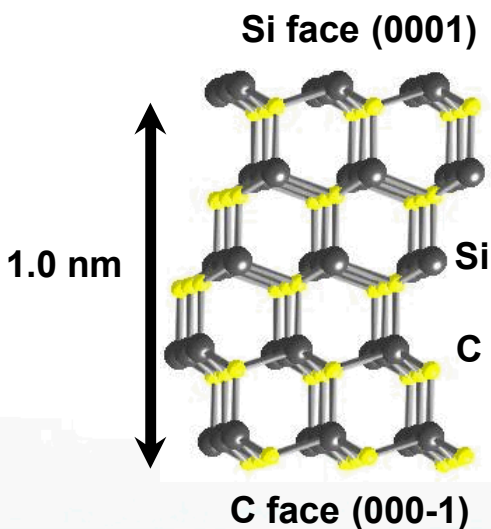
K. V. Emtsev et al., Nature Mater. 8, 203 (2009).

C. Virojanadara et al., Phys. Rev. B 78, 245403 (2008)

T. Ohta, N. C. Bartelt, S. Nie, K. Thürmer, G. L. Kellogg, PRB 81, 121411(R)(2010)

# Epitaxial graphene growth on SiC

4H - SiC



- 1 to few ML graphene
- Buffer layer between graphene and SiC



- Many-layer graphene common
- Disordered stacking

