

Resistive Switching In Aluminum Nitride

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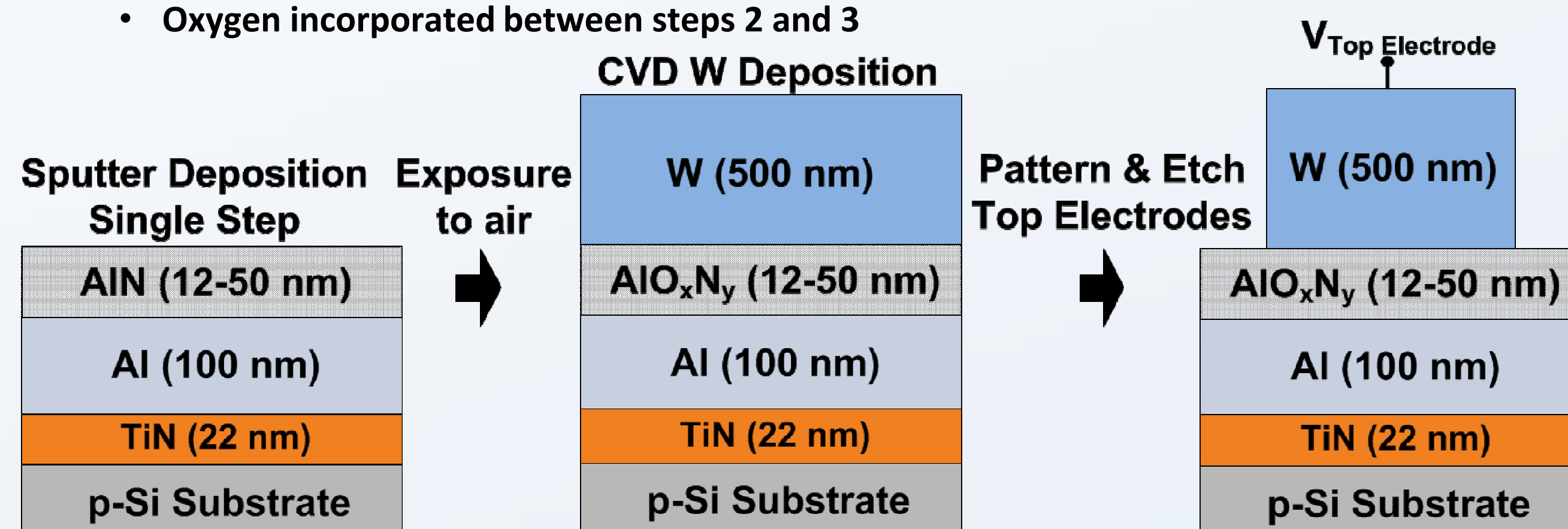
Introduction

- The Resistive or Redox RAM (ReRAM) is one of the most promising replacements for Flash, DRAM, and even SRAM memories [1]
- Many transition metal oxides exhibit bipolar resistance switching based on two reduction/oxidation mechanisms [2]
 - Valence change mechanism (VCM) – anion motion
 - Electrochemical change mechanism (ECM) – cation motion
- Search for idea resistance switching material is still underway
- Desire fab compatible, high endurance ($>10^{15}$), long retention (10 y), high speed (<10 ns W/E) nonvolatile memory
- Resistance switching in aluminum nitride recently discovered [3]-[5]

Experimental Details

Fabrication procedure (illustrated below)

- Fabricated in Sandia's MESA Microelectronics Development Lab on 6 inch silicon wafers:
 - Deposit blanket TiN with reactive PVD
 - Deposit Al with PVD @ 350°C
 - Deposit AlN with reactive PVD @ 350°C
 - Deposit W top electrode
 - Pattern ReRAM devices
 - Wet etch W to define device area
 - Deposit backside Al
- Oxygen incorporated between steps 2 and 3



Electrical Characterization

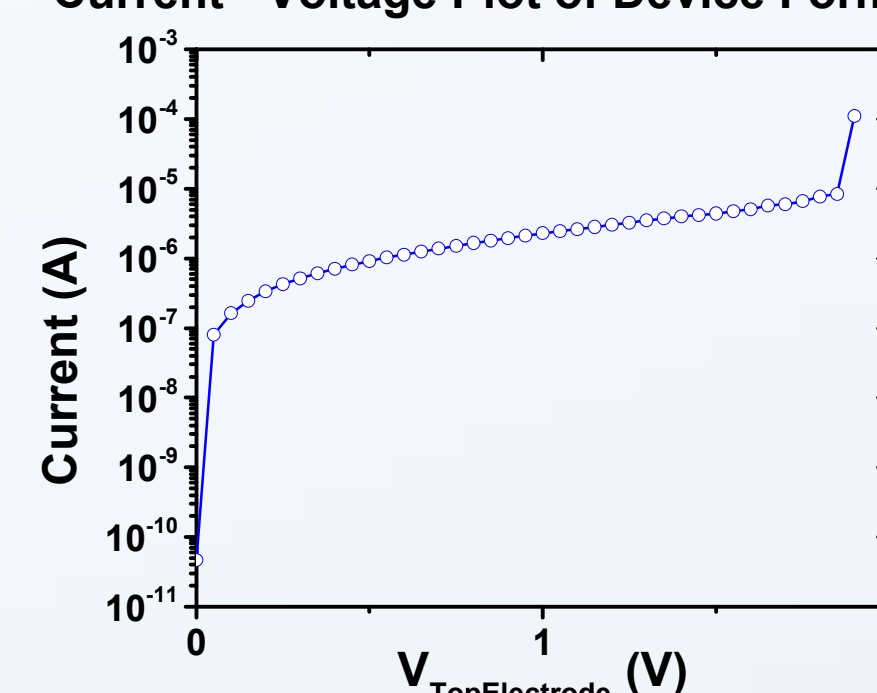
- Cascade low noise probe station
- Current-voltage (I-V) and pulsing with Agilent 4156C
- Pulsing expander box capable of 1 μs pulse width
- Two terminal connection to V_{TopElectrode} and substrate

Results – Electrical

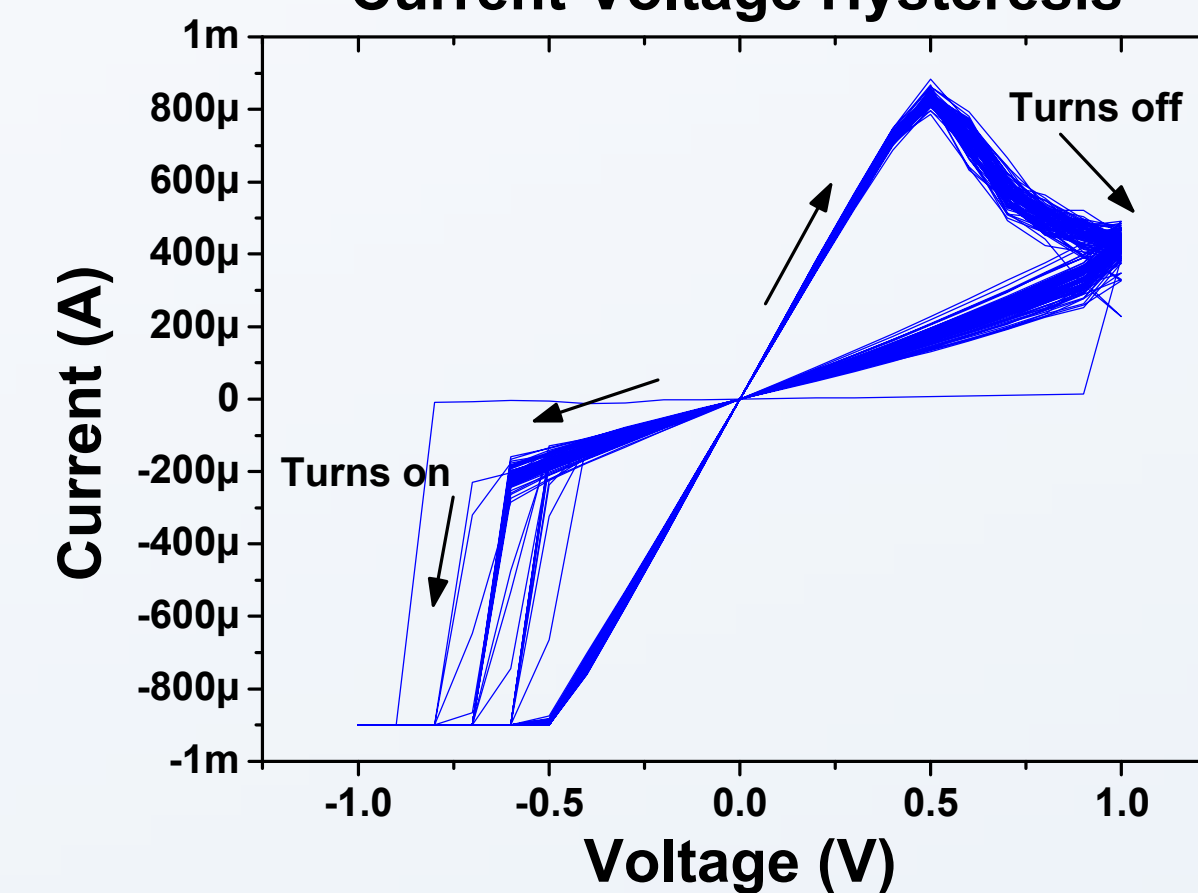
Electrical Characterization

- Positive electroforming at 100 μA
- Highly linear high resistance and low resistance states (HRS, LRS)
- Ohmic conduction
- $R_{\text{on}}/R_{\text{off}} \approx 6-7$

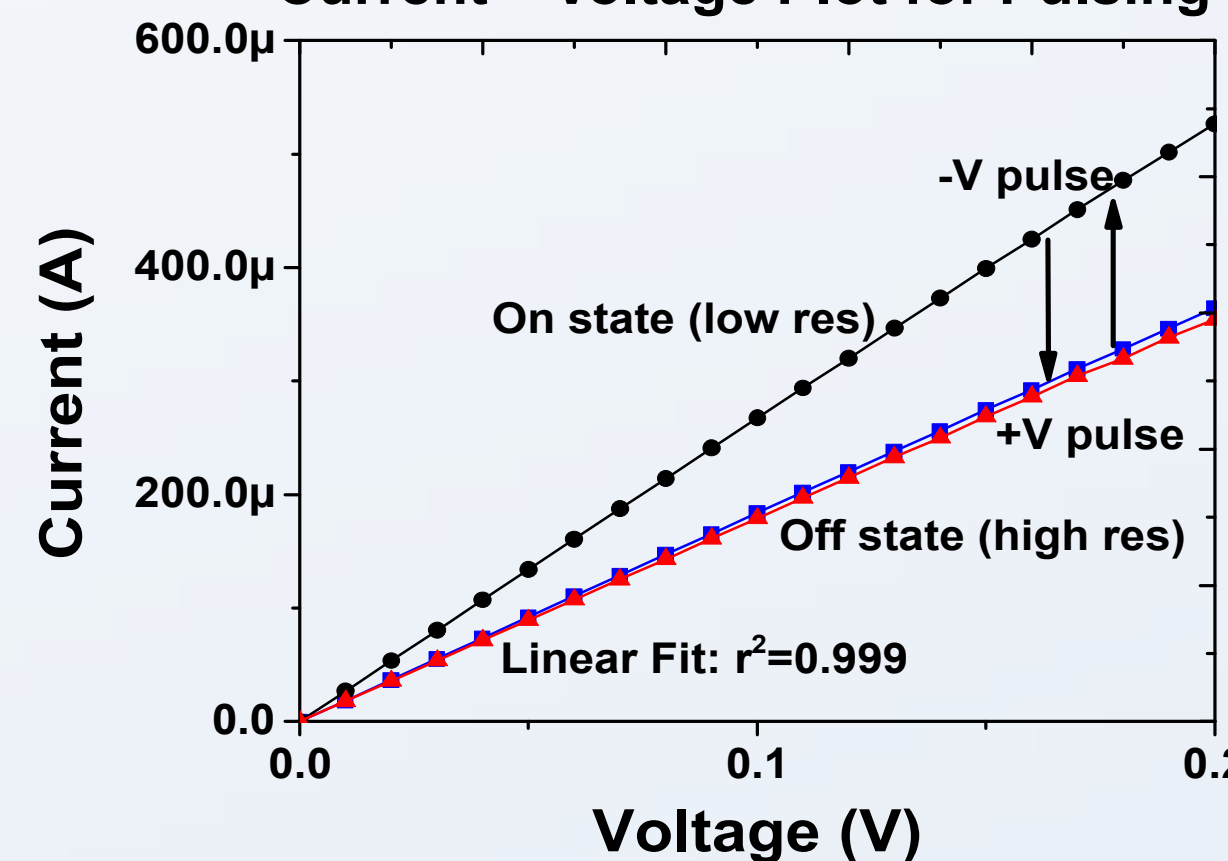
Current - Voltage Plot of Device Forming



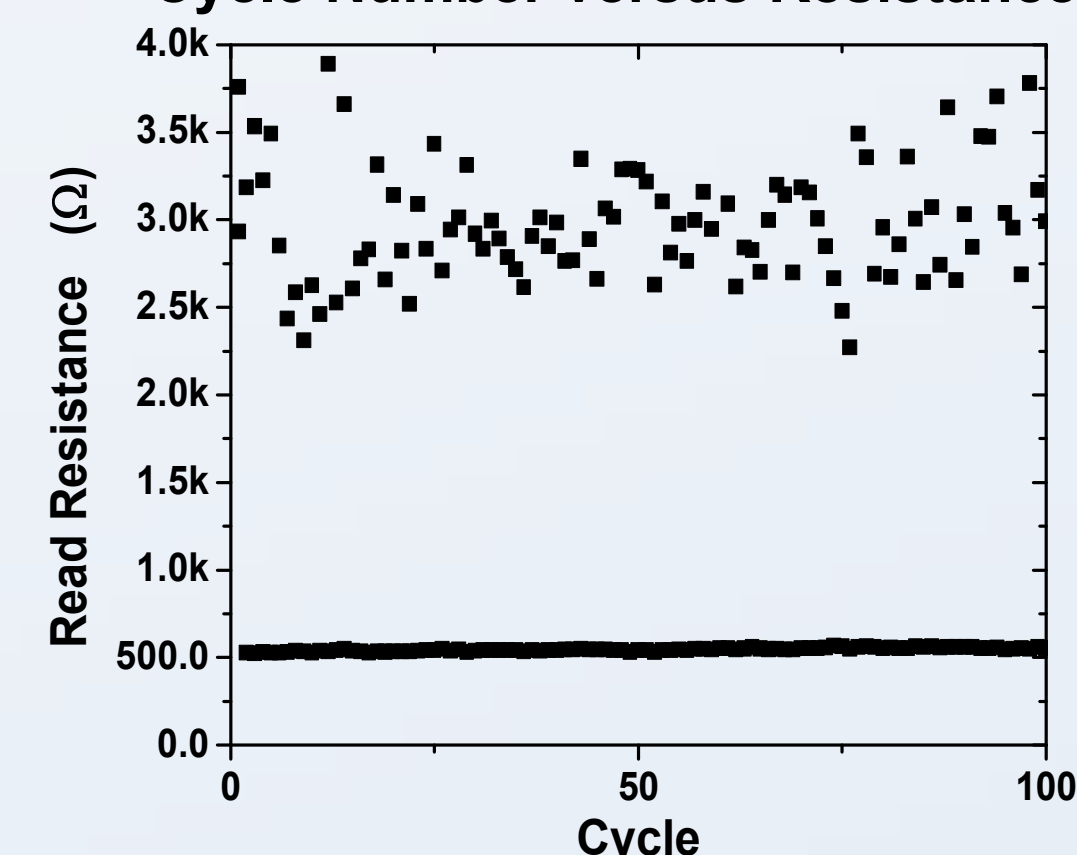
Current-Voltage Hysteresis



Current - Voltage Plot for Pulsing



Cycle Number versus Resistance



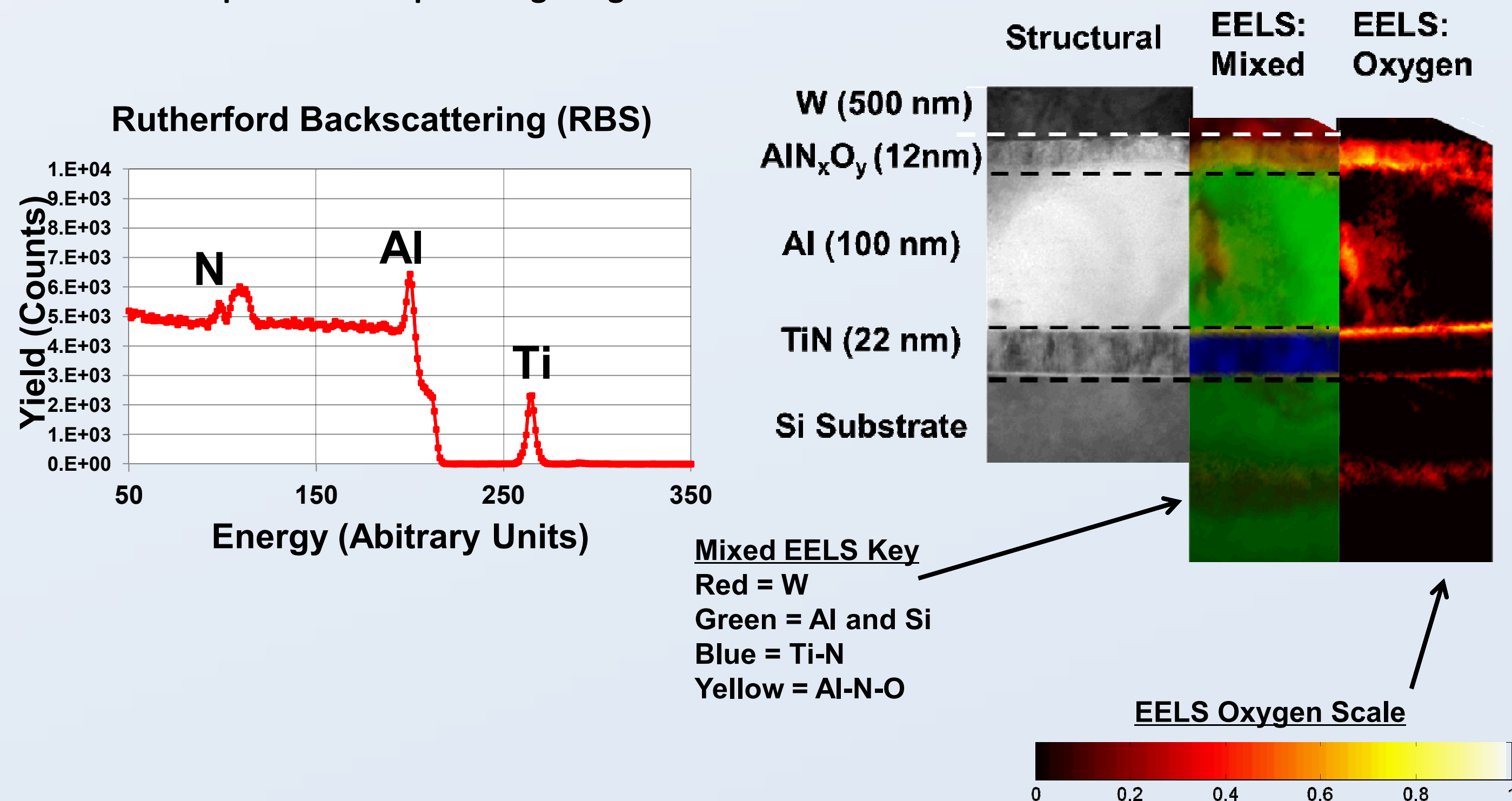
Results – Physical Characterization

Rutherford Backscattering (RBS)

- Very common elemental analysis technique using an ion beam
- Does not accurately show the incorporation of oxygen in the AlN film
- Secondary ion mass spectrometry (SIMS) (not shown) also gives inaccurate oxygen quantification due to preferential sputtering of light elements

Electron Energy Loss Spectroscopy (EELS)

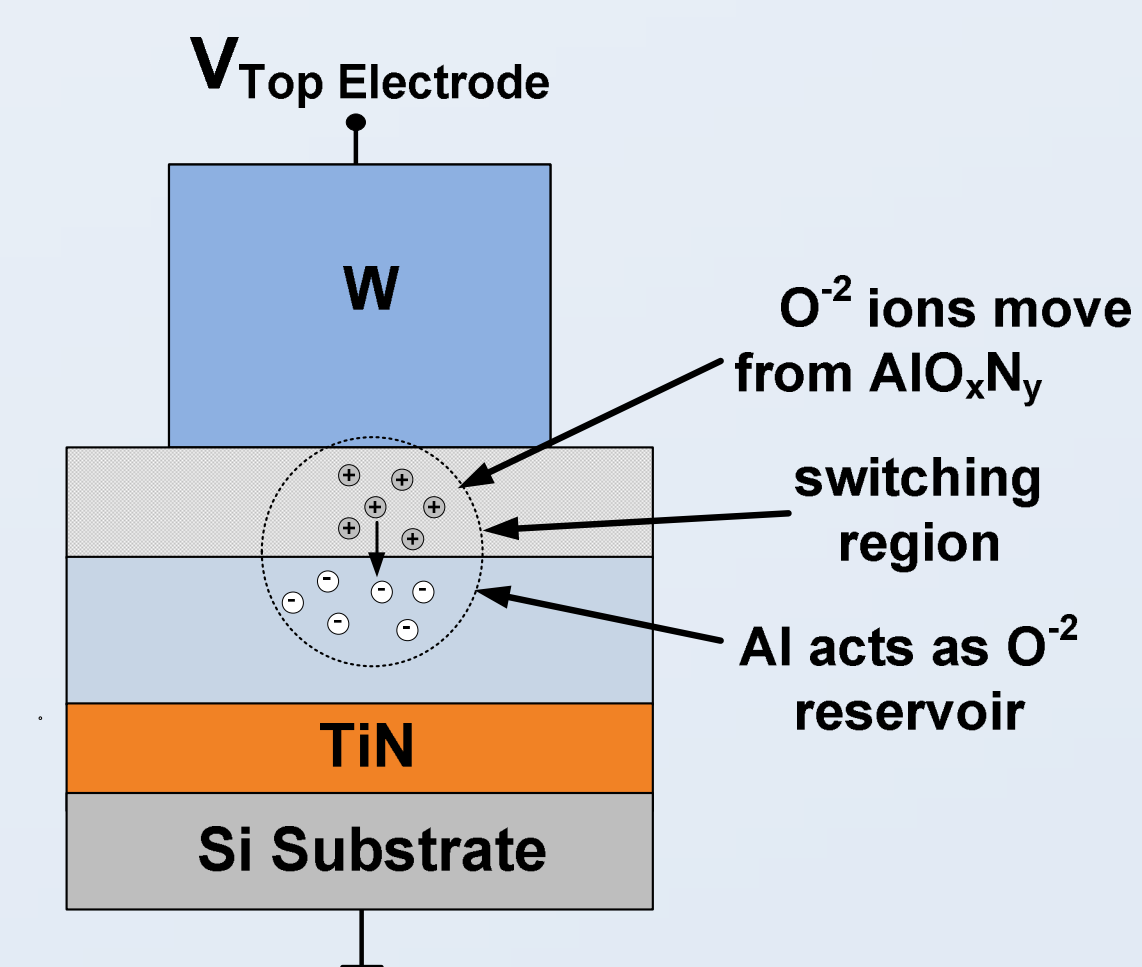
- Analysis technique used in conjunction with a TEM microscope to analyze chemical composition of a sample
- Excellent spatial resolution
- Used Sandia's FEI Titan Aberration-Corrected TEM
- Shows that our AlN film incorporates oxygen



Possible Mechanism

Redox Reaction

- HRS and LRS are highly linear:
 - Ohmic conduction in both
- Electroforming required – fits with redox model
 - Nitrogen not mobile in AlN [6]
 - Al is the mobile species
- Oxygen often substitutes for nitrogen in AlN (on the octahedral site)
- Oxygen is highly reactive with AlN
- Oxygen is deep donor in AlN [7]-[8]



Summary

- Resistive switching observed in oxygen doped AlN (AlO_xN_y) films
- Oxygen is most often incorporated into AlN films – this is difficult to prevent
- Oxygen-AlN redox reaction is a strong possibility as a mechanism
 - Oxygen substitution of octahedral N in AlN
- Future work includes investigating temperature dependence, endurance, and effect of oxygen concentration

References

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