

SAND2012-7064C

The Four Pillars of Inherent Chemical Safety / Security

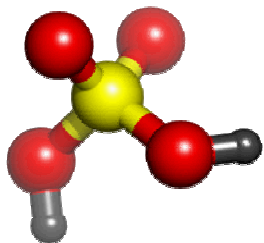
Nairobi, Kenya
September, 2012



SAND No. 2012-7064C

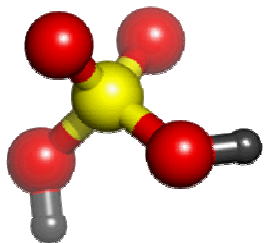
Sandia is a multi-program laboratory operated by Sandia Corporation, a Lockheed Martin Company,
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Part of formal IST definition

- ▶ IST is an iterative process that considers such options, including eliminating a hazard, reducing a hazard, substituting a less hazardous material, using less hazardous process conditions, and designing a process to reduce the potential for, or consequences of, human error, equipment failure, or intentional harm.



Pillar One

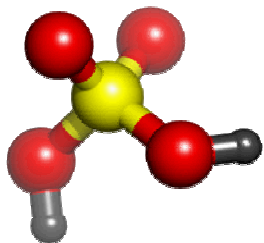
Minimize

Substitute

Moderate

Simplify

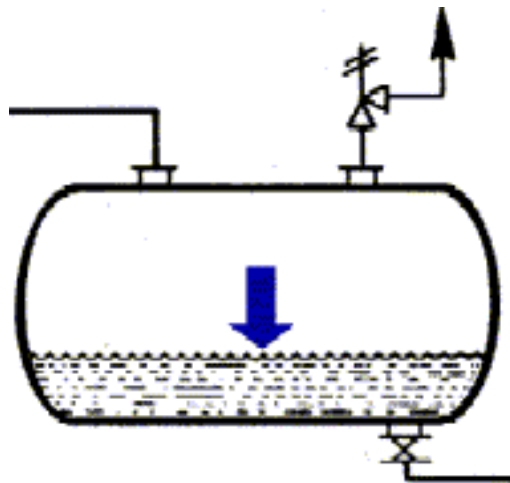


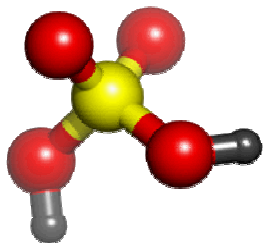


Minimize

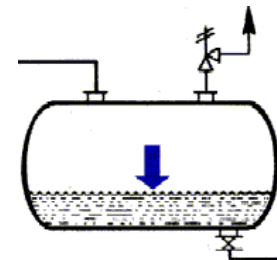
To *minimize* is to reduce the amount of potential energy present (get the system closer to a zero energy state),

this reduces the potential impacts if containment or control of the hazard is lost.





Minimize



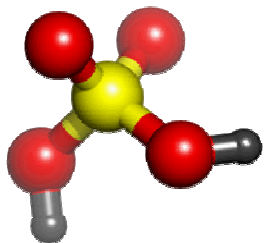
Some strategies for making a process inherently safer by *minimization*:

▶ Inventory reduction:

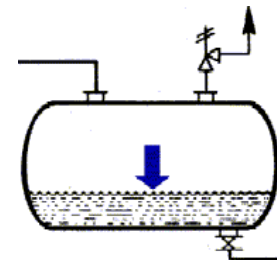
- less material stored ← *requires administrative control*
- fewer tanks; just-in-time delivery
- less vapor volume
- generate on demand (chlorine, methyl isocyanate, ammonia, hydrogen...)
- receive by pipeline instead of by truck or rail

▶ Process intensification

▶ Process operation closer to ambient conditions

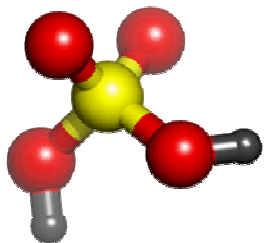


Minimize



Ultimate case:

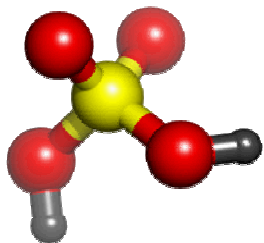
- ▶ Elimination of the hazard:
 - Eliminating use of a particular hazardous material
 - Operating the system at lower temperatures (with respect to a particular hazard)
 - Shutting down the process
 - Using another company to provide processing (*risk transfer*)



Pillar Two

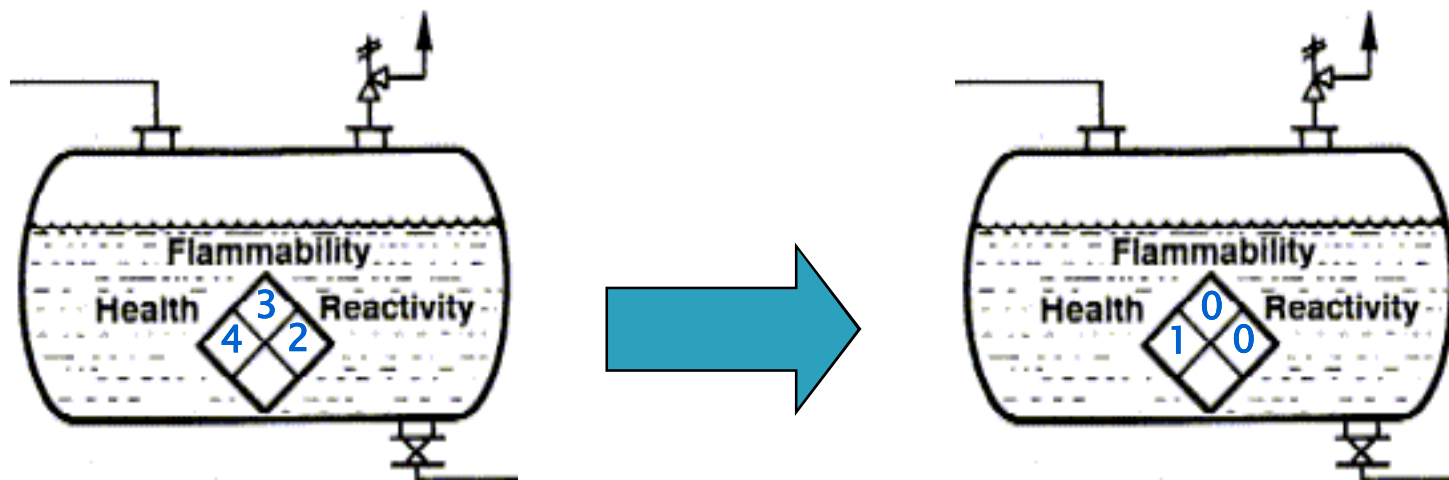
Minimize
Substitute
Moderate
Simplify

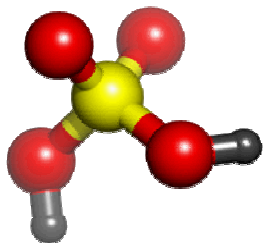




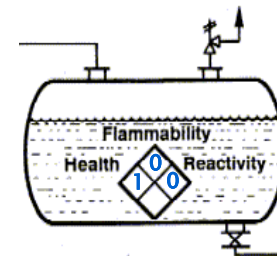
Substitute

To *substitute* is to replace with a less hazardous material or condition.



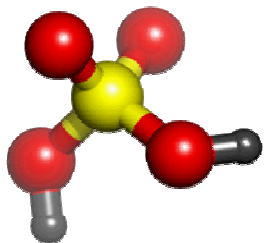


Substitute

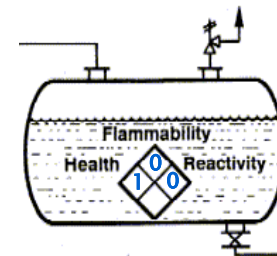


Some strategies for making a process inherently safer by *substitution*:

- ▶ Commercially available alternatives
- ▶ Alternative raw material or intermediate that can be transported and stored more safely (example ethylene glycol is toxic, propylene glycol is non-toxic)
- ▶ Alternative chemistry- Biosynthesis routes
- ▶ Oleum alternative: Sulfur burning to generate SO_3 on demand



Substitute



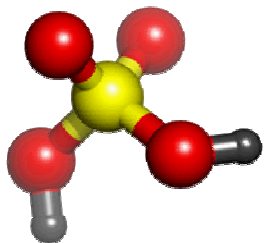
Solvent substitutes:

- Water-based paints, adhesives
- Aqueous cleaning systems
- Less volatile solvents; higher flash point
- Dibasic esters for paint stripping instead of dichloromethane

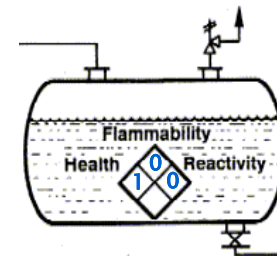
Web resources are available

- “Substitutes in Non-Aerosol Solvent Cleaning,”

www.epa.gov/ozone/snap/solvents/solvents.pdf

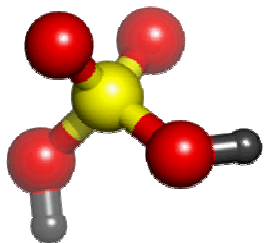


Substitute



Some chlorine alternatives: Cl_2

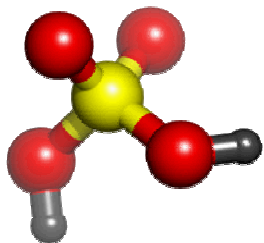
- ▶ Sodium hypochlorite
- ▶ Calcium hypochlorite
- ▶ Hydrogen peroxide
- ▶ Chlorine dioxide
- ▶ Bromine
- ▶ Mixed oxidants
- ▶ Other technologies (UV radiation)



Pillar Three

Minimize
Substitute
Moderate
Simplify



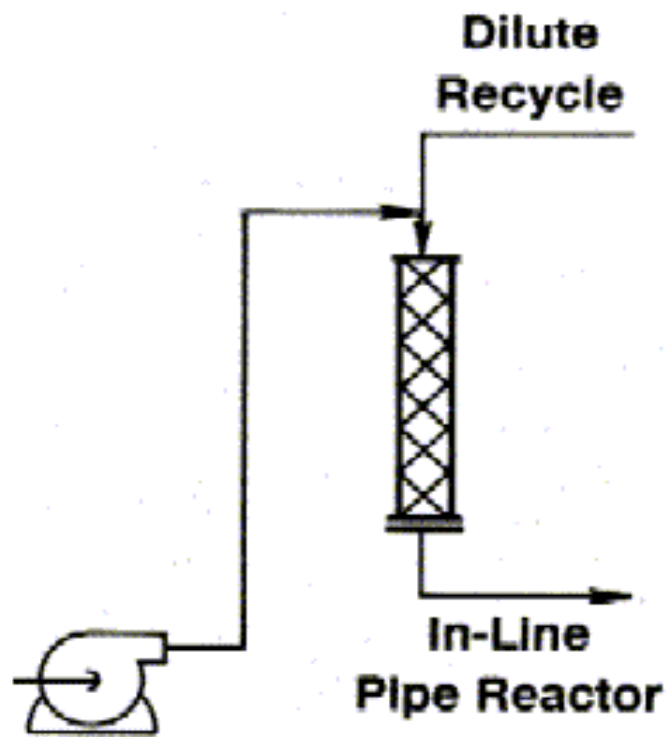


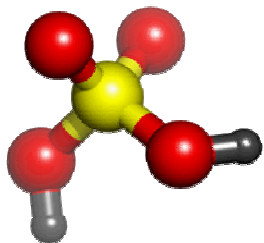
Moderate

To *moderate* (or *attenuate*) is to handle a material under less hazardous process conditions.

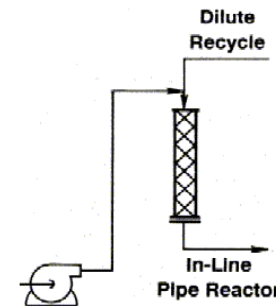
Dilution
Refrigeration
Less severe processing conditions

Note: Available energy may be the same, but potential loss event impacts can be reduced





Moderate



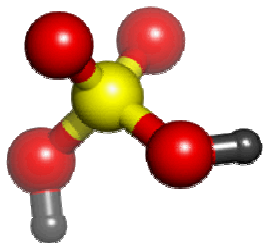
Some strategies for making a process inherently safer by *moderation* :

▶ **Dilution:**

- using in aqueous instead of anhydrous form
- Using in solution such that the solute would boil off before a runaway reaction temperature was achieved
- Lower concentration of benzoyl peroxide in paste
- Mixing coal dust with rock dust

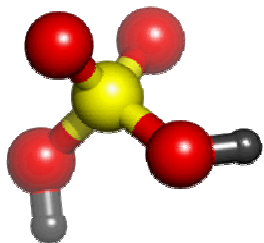
▶ **Refrigeration:**

- storing anhydrous ammonia as a refrigerated liquid instead of as a liquefied gas



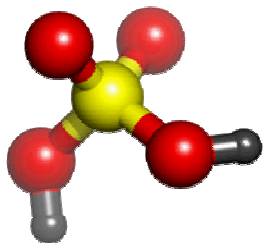
Moderate

- ▶ Aqueous ammonia instead of anhydrous
 - Anhydrous is hazardous gas
- ▶ Aqueous HCl in place of anhydrous HCl
 - Anhydrous is a hazardous gas
- ▶ Sulfuric acid in place of oleum
 - Oleum has dissolved SO_3 (a toxic gas)
- ▶ Dynamite instead of nitroglycerine
 - The nitroglycerin in dynamite is stabilized on a solid carrier (diatomaceous earth)



Less severe processing conditions

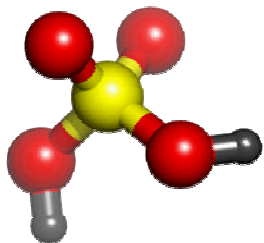
- ▶ Ammonia manufacture
 - 1930s - pressures up to 600 atmospheres (atm)
 - 1950s - typically 300-350 atm
 - 1980s - plants operating at pressures of 100-150 atm were being built
- ▶ Result of understanding and improving the process
- ▶ Lower pressure plants are cheaper, more efficient, as well as safer



Pillar Four

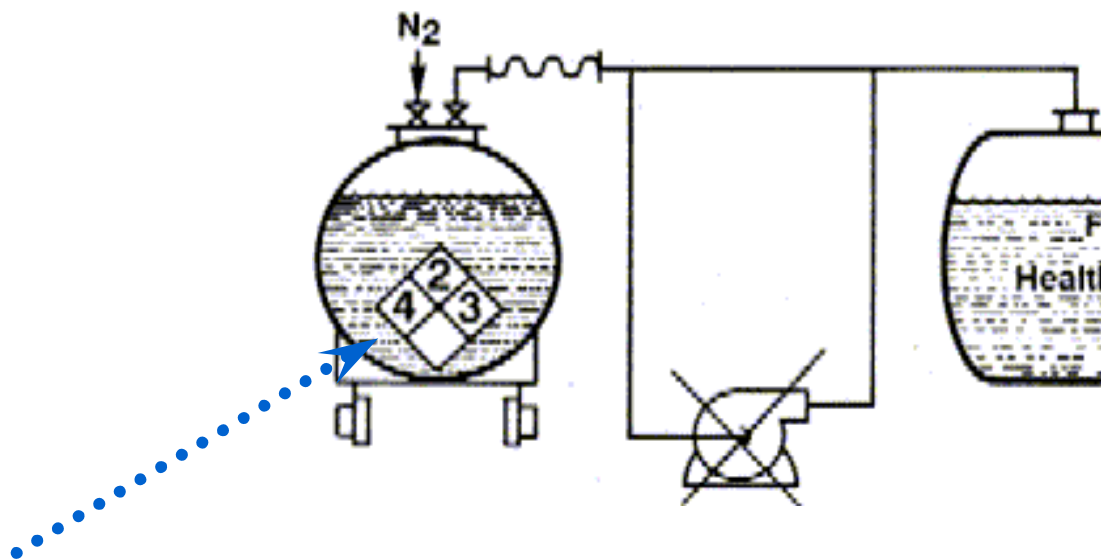
Minimize
Substitute
Moderate
Simplify



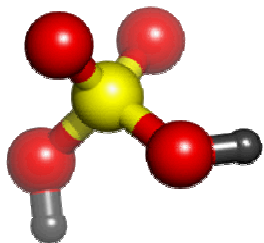


Simplify

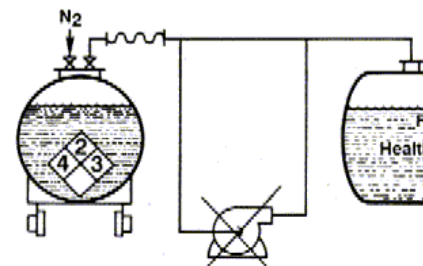
To *simplify* is to eliminate unnecessary complexity.



(Not “first-order” inherent safety, since the underlying hazard is still there.)

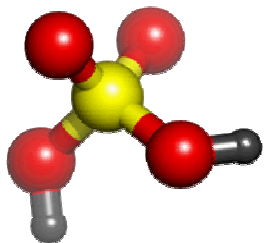


Simplify

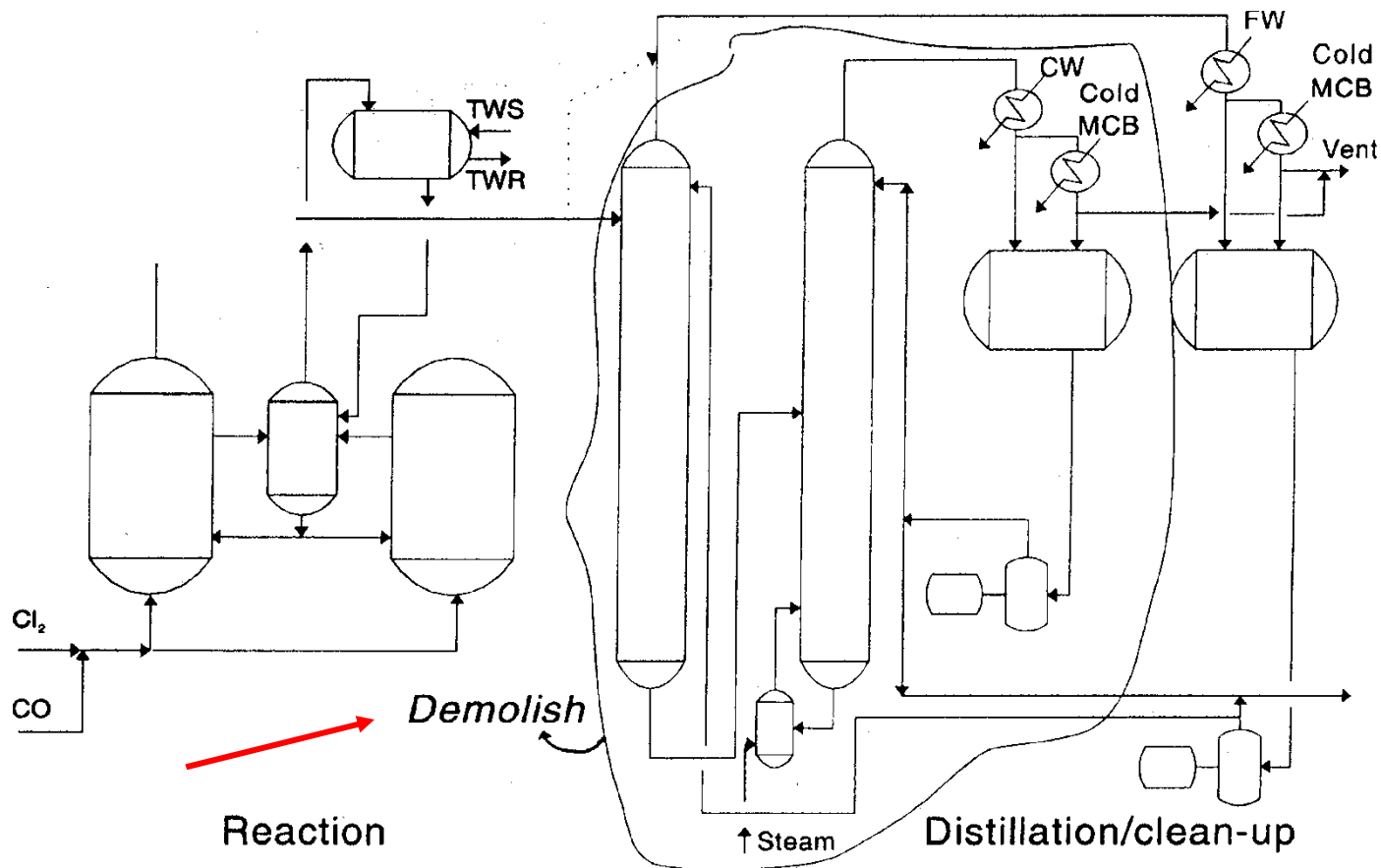


Some *simplification* strategies:

- ▶ **Use simpler equipment arrangement:**
 - Gravity flow instead of pumping
 - Natural convection
 - Collocation of shutoff valves
- ▶ **Eliminate interconnections** to reduce the likelihood of inadvertent mixing
- ▶ **Minimize number of flanges, connections, and other potential leak locations**

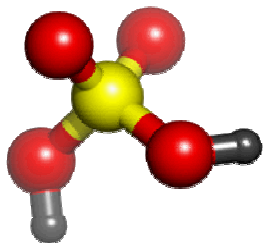


Simplify...



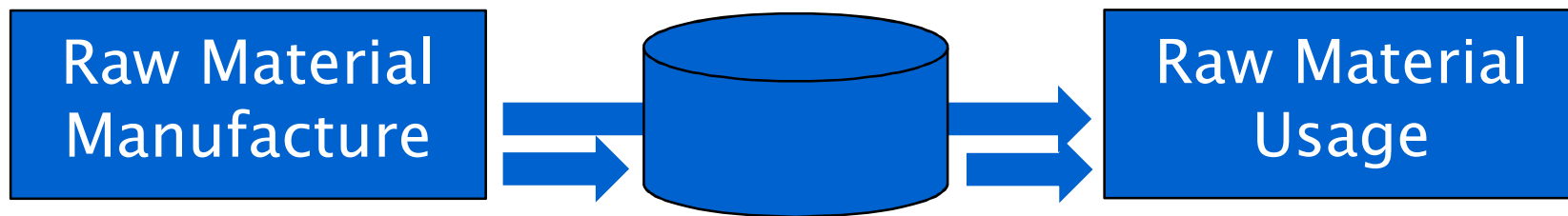
Simplification of Dow Phosgene Unit for MDI Production

R. Gowland, "Applying Inherently Safer Concepts to a Phosgene Plant Acquisition," *Process Safety Progress* 15(1), 57

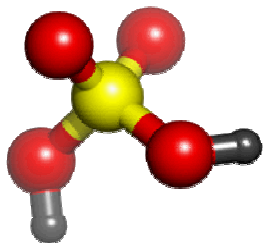


DISCUSSION

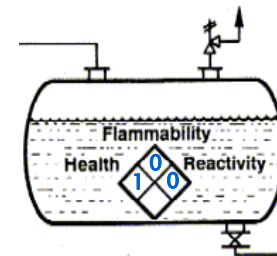
- ▶ An inherent safety review recommends eliminating intermediate storage of a hazardous raw material:



- ▶ What are the inherent safety benefits?
 -
- ▶ What are the possible drawbacks?
 -

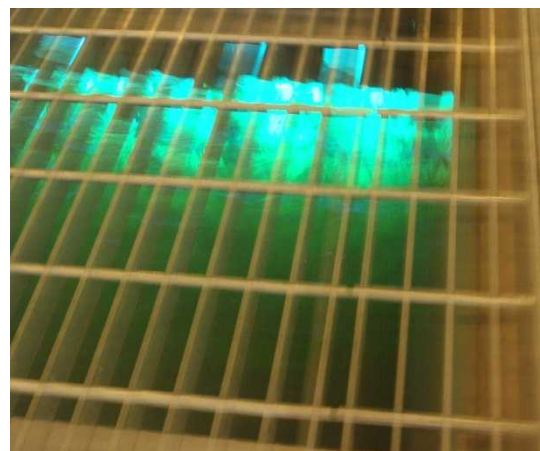


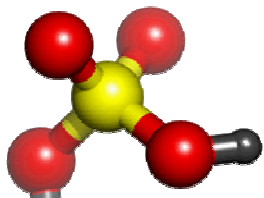
Case Study - Substitute Wastewater Plant



Chlorine alternative : Cl_2

- ▶ **Alternative Disinfection (UV radiation)**





Wastewater plant chlorination process- Substitute



2-25 ton Cl₂



Cl₂ Evaporators



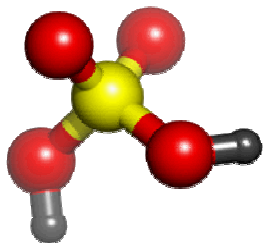
Chlorinator



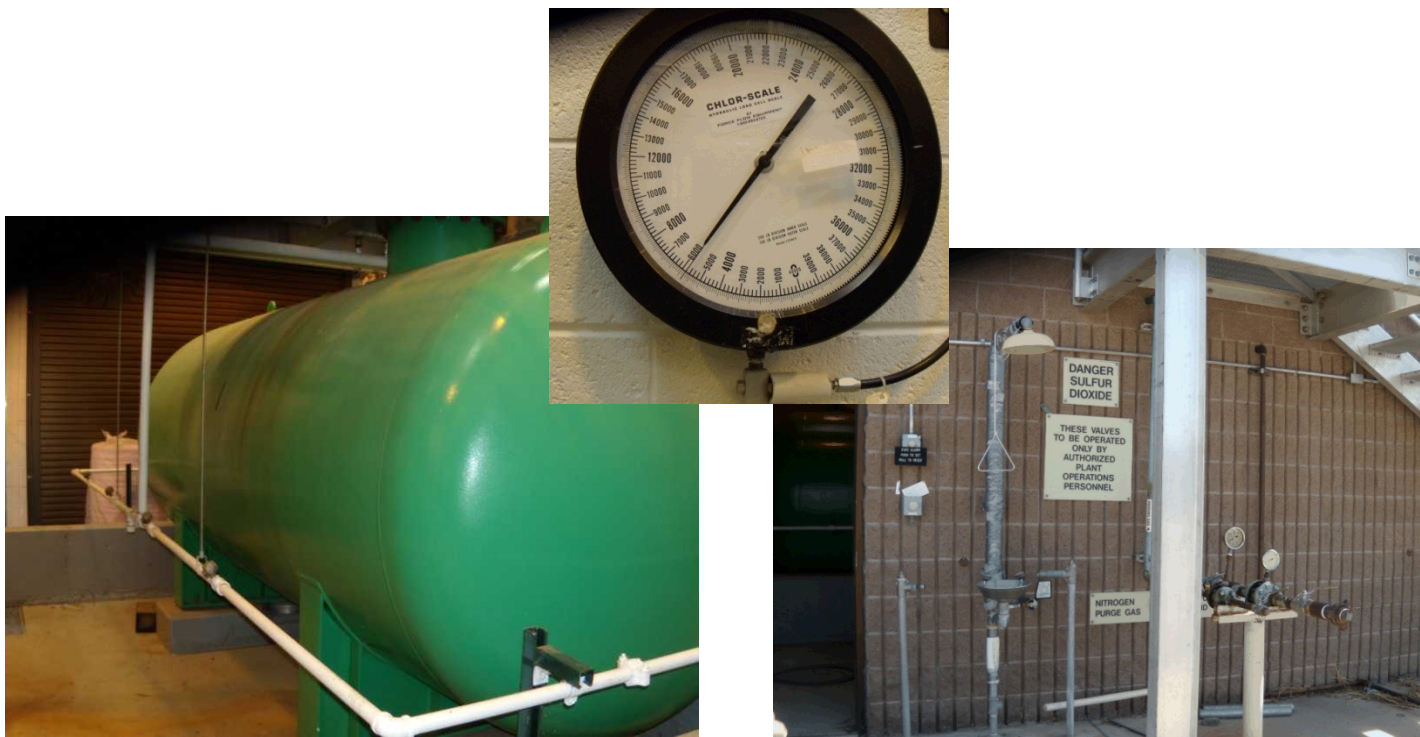
Shower/Eyewash



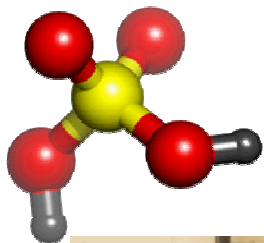
Chlorine contactor



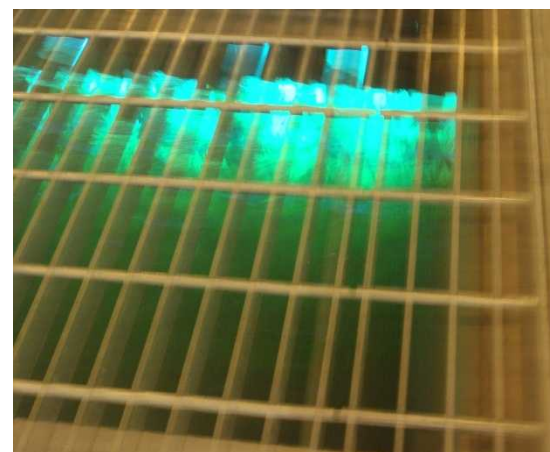
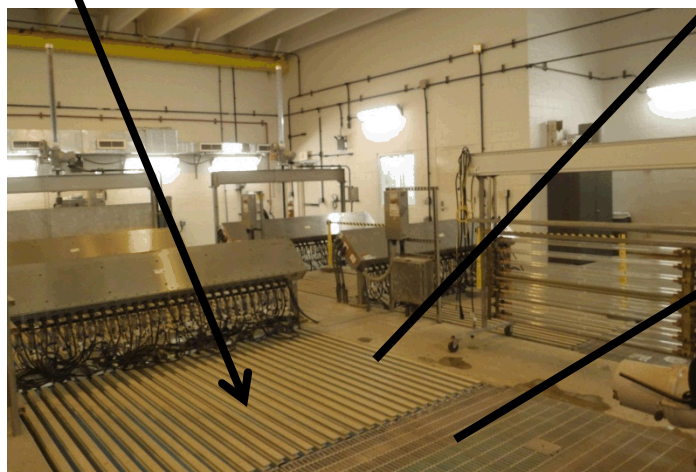
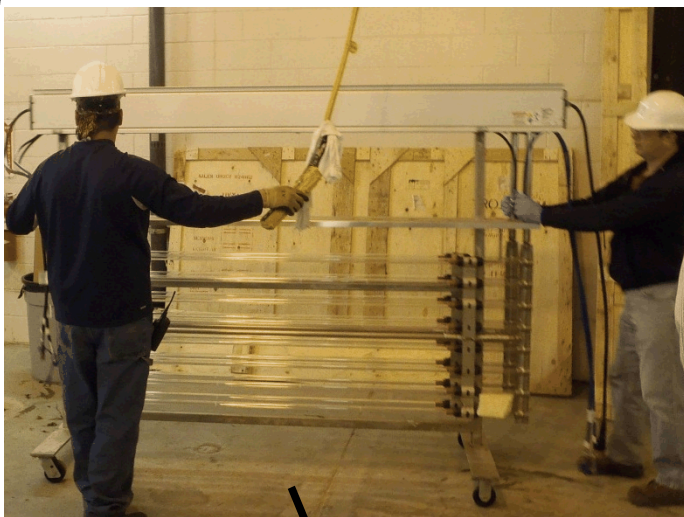
SO₂ used to remove excess chlorine before discharge - Substitute

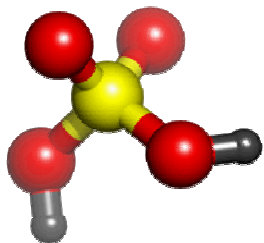


Two 15 ton SO₂

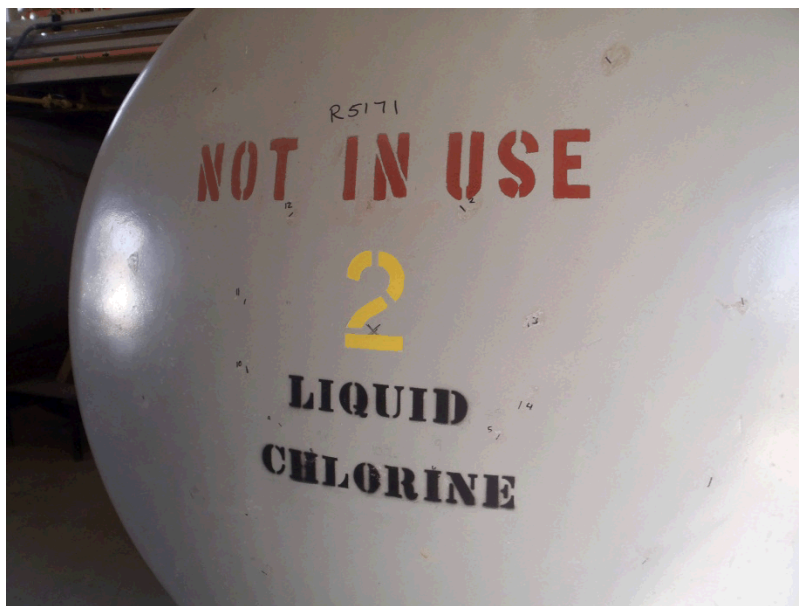


Ultraviolet disinfection replaces both Cl_2 and SO_2 - Substitute





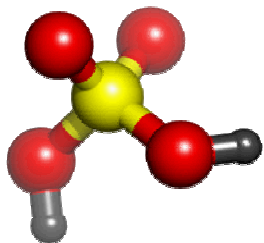
Bulk chemical removed from process - Substitute



2-25 ton Cl₂



2-15 ton SO₂



Discussion

- ▶ Can you think of applications of substitution to chemical management?
- ▶ What role does regulation play in the selection of technology?
- ▶ What may make it difficult to mandate Inherent Safety?