

# Gas permeability of graphene oxide membranes

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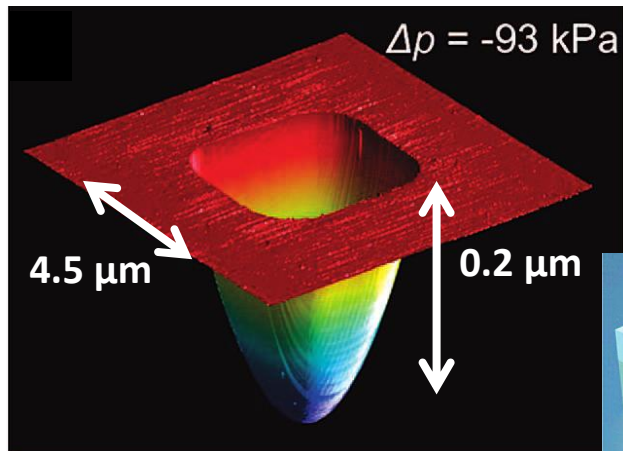
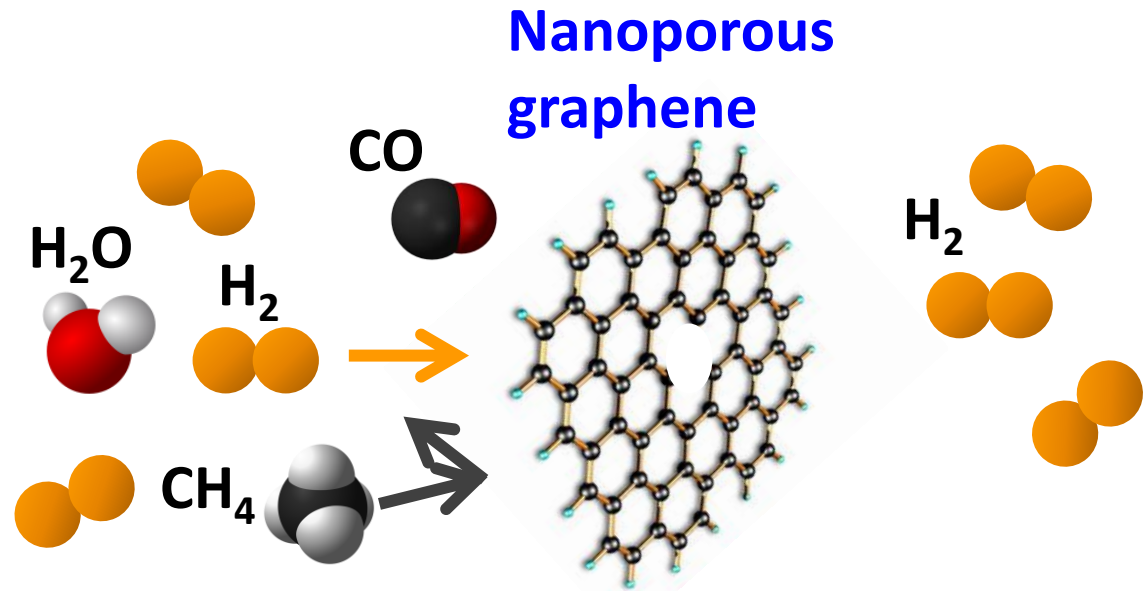
**MRS Fall Meeting, Symposium T**

**November 28<sup>th</sup>, 2012**

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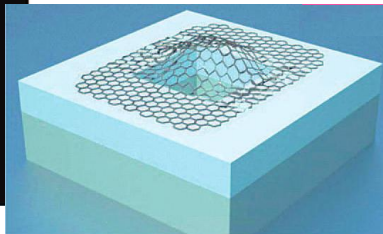
# Permselective gas membranes enable energy technology

Steam reforming of natural gas:



J. Bunch *et al.*, *Nano Lett* **8**, 2459-62 (2008).

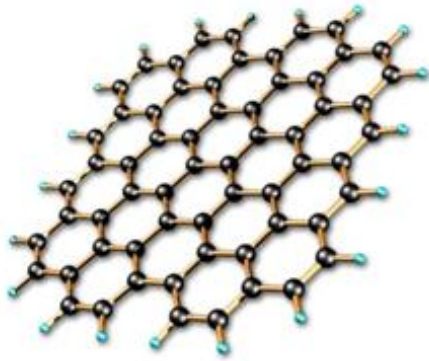
A pristine graphene membrane is impervious to gasses



AFM image of a graphene monolayer under  $\Delta p = 700 \text{ torr}$

# Graphene oxide is a one-atom thin solution-processable membrane material.

**Graphene**  
sp<sup>2</sup> bonded carbon atoms

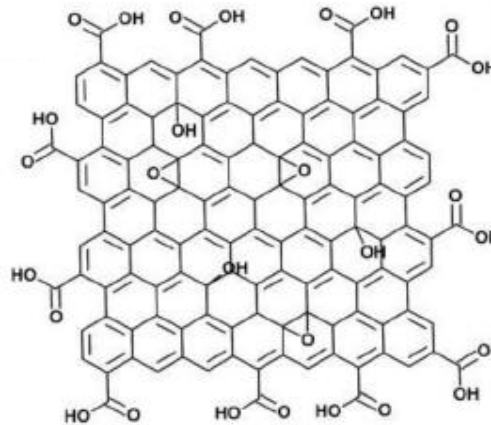


Natural graphite



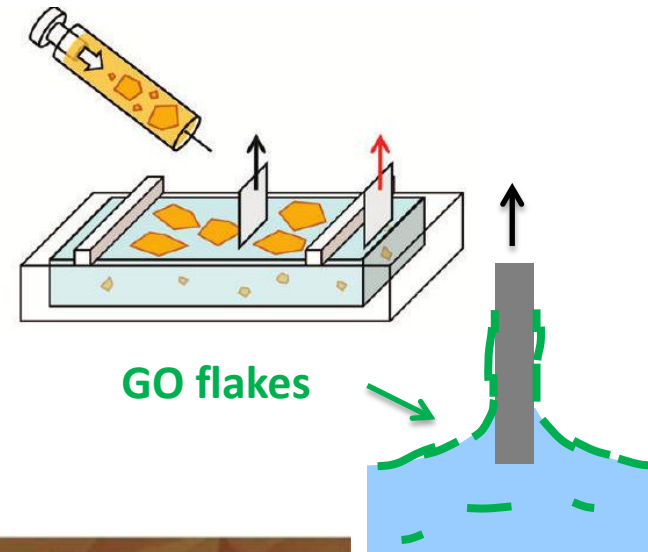
G. Eda and M. Chhowalla, *Adv. Mater.* 2010

**Graphene Oxide (GO)**

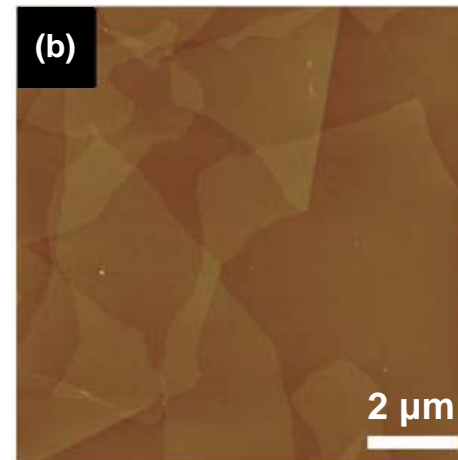


Chemical  
oxidation

**Langmuir-Blodgett deposition**

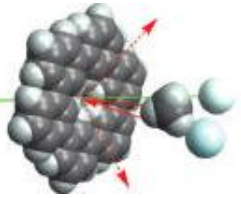


GO flakes

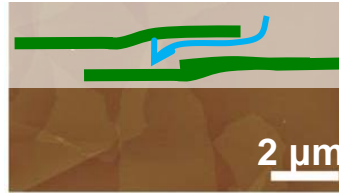


L. Cote *et al.*  
*Soft Matter* 2010.

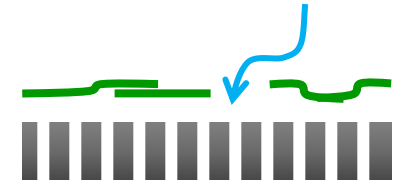
# Permeation pathways through a supported GO membrane



$d \sim 0.3 \text{ nm}$   
 $Kn \gg 10$

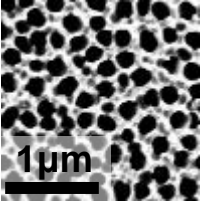


$d \sim 1 \text{ nm}$   
 $Kn > 10$

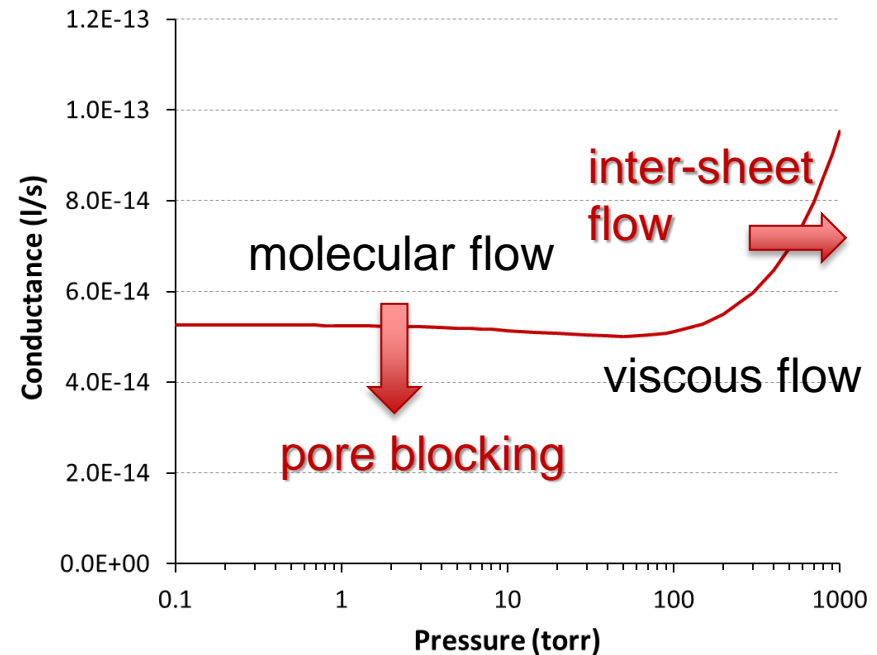
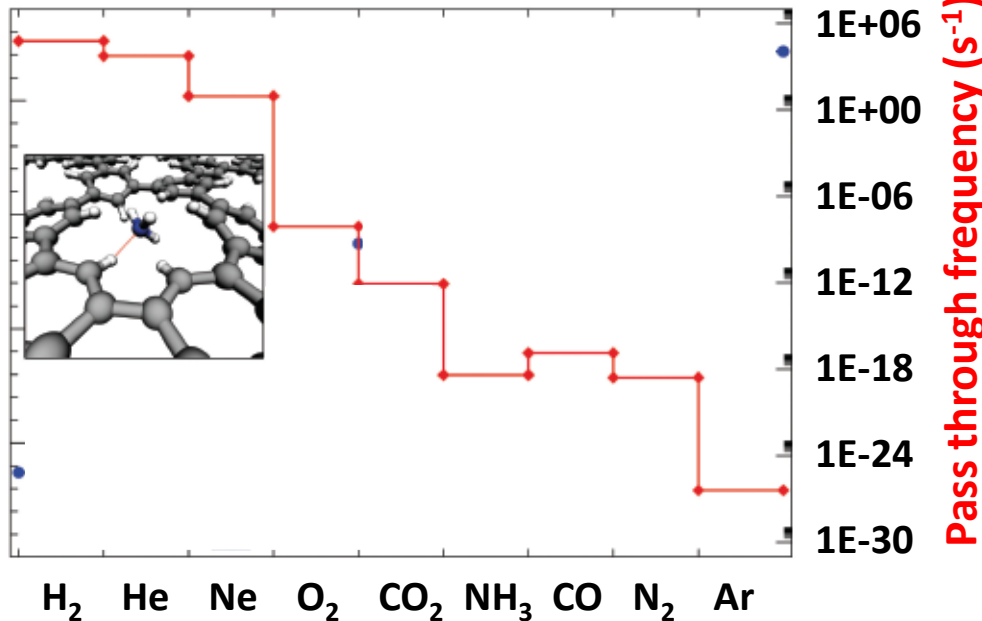


$d \sim 20\text{--}200 \text{ nm}$   
 $Kn \sim 10$

Anodisc



For  $\text{H}_2/\text{CH}_4$  and  $\text{He}/\text{CH}_4$ , selectivity  $> 10^{20}$



[1] S. Blankenburg *et al.* *Small* **6** (2010); [2] D. Jiang *et al.* *Nano Lett.* **9** (2009); [3] J. Schrier, *J. Phys. Chem. Lett.*, **1** (2010).

# Membrane relaxation experiments have shown size-selective permeation

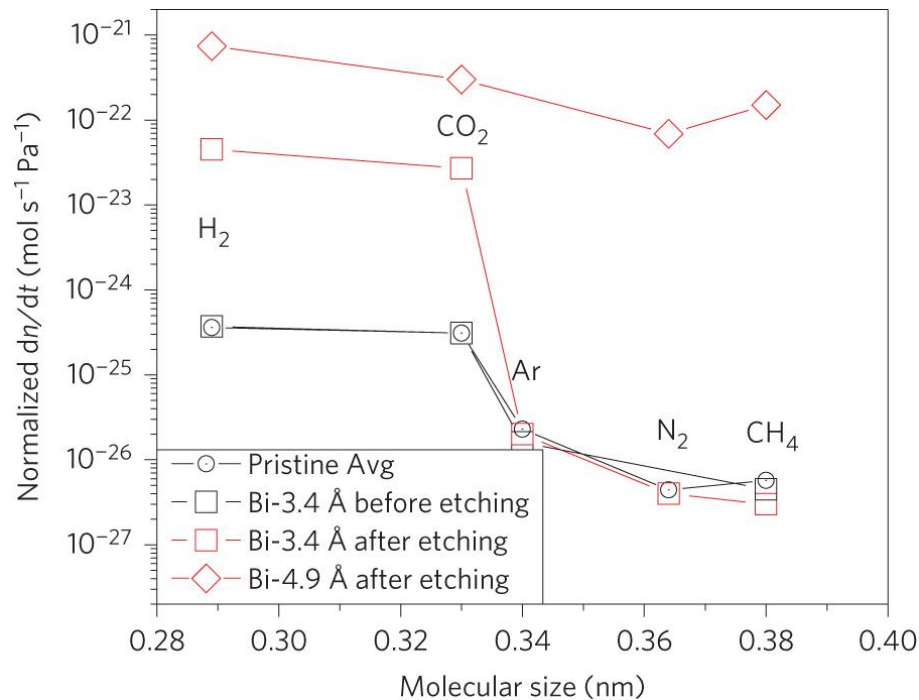
## LETTERS

PUBLISHED ONLINE: 7 OCTOBER 2012 | DOI: 10.1038/NNANO.2012.162

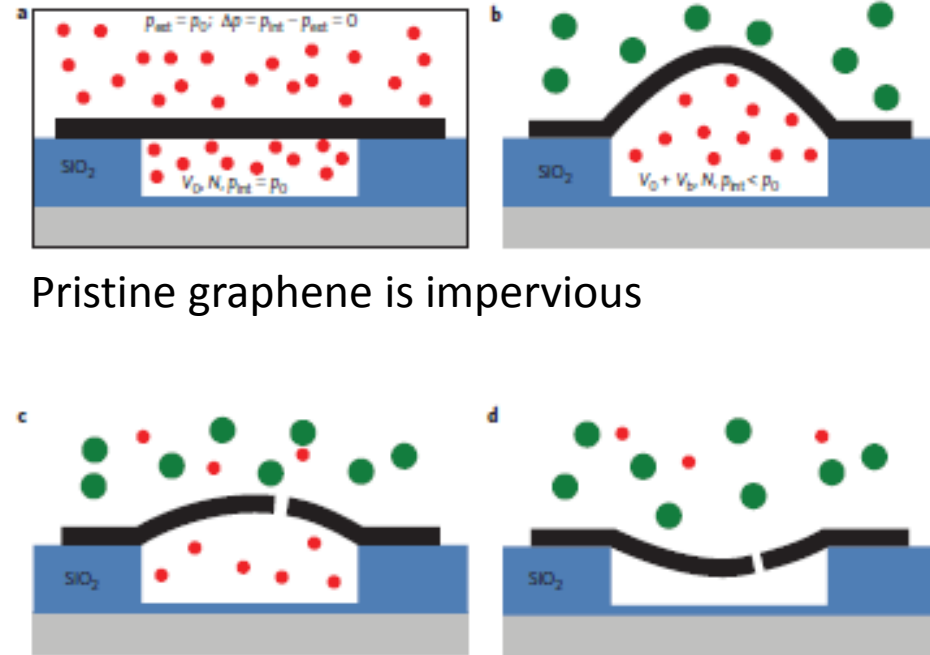
nature  
nanotechnology

## Selective molecular sieving through porous graphene

Steven P. Koenig, Luda Wang, John Pellegrino and J. Scott Bunch\*



Oxidation generates nanopores, permitting size-selective permeation

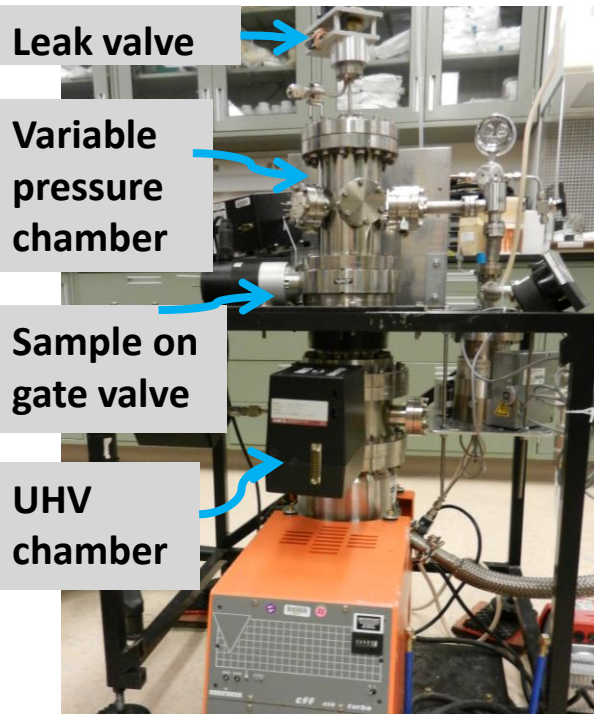


Pristine graphene is impervious

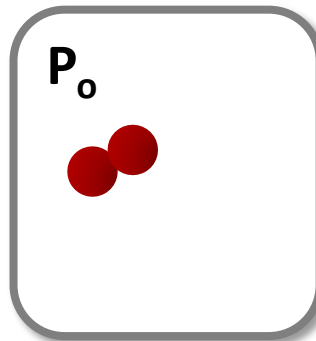
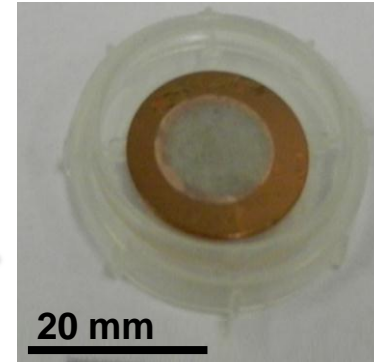
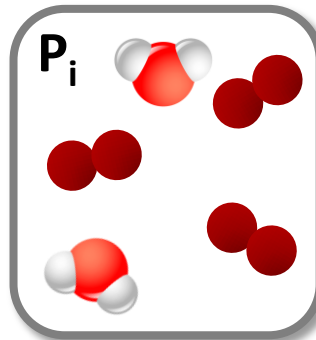
Small gases permeate through nanopores

# High dynamic range of permeability allows flow channel differentiation

## Permeation System



GO membrane on porous alumina oxide  
(Whatman Anodisc)



A precision permeation system allows for high pressure differentials

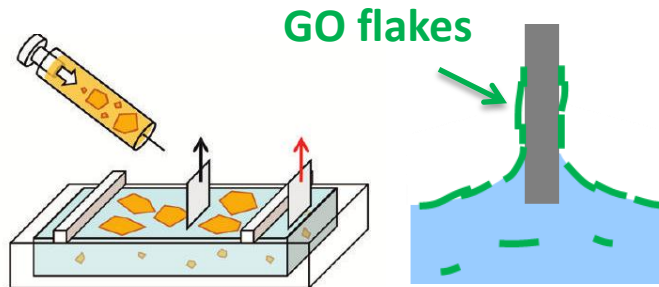
- $P_i/P_o \sim 10^{11}$  Torr
- $P_i \sim 1000$  Torr

# Tunability of Langmuir-Blodgett deposition allows GO assembly on varied surfaces

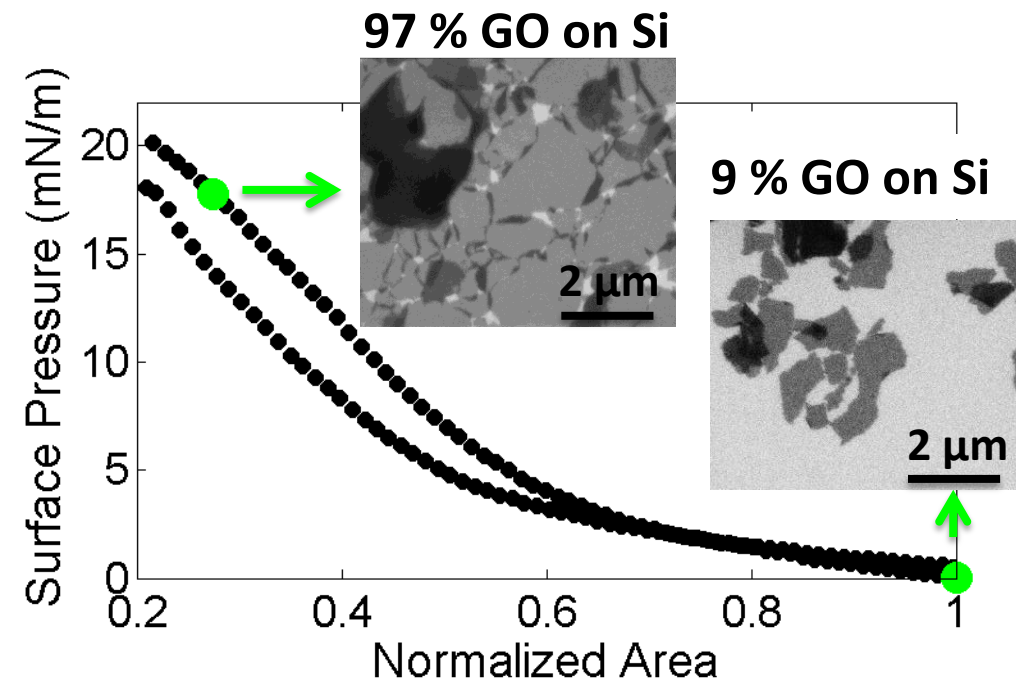
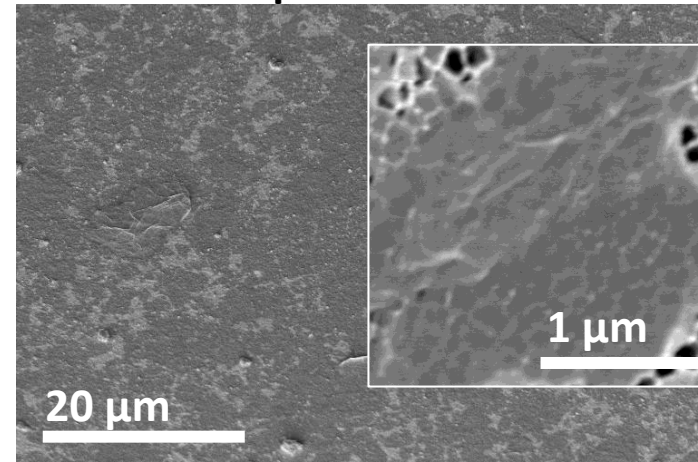
Spreading solution:

1:5 Aqueous GO in methanol

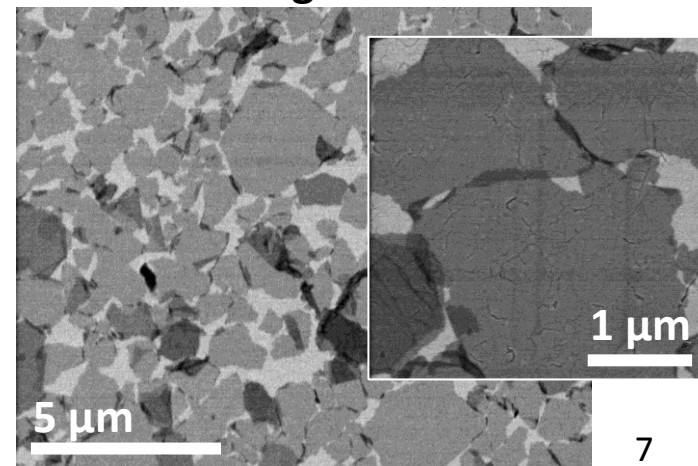
DI water subphase



86 % GO on porous Anodisc

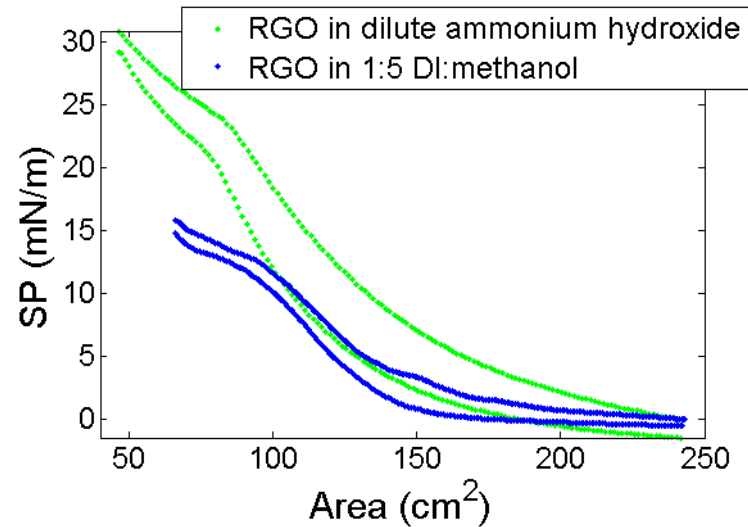
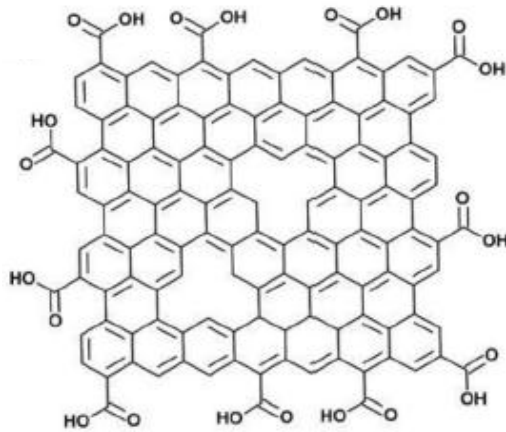


84 % GO on gold

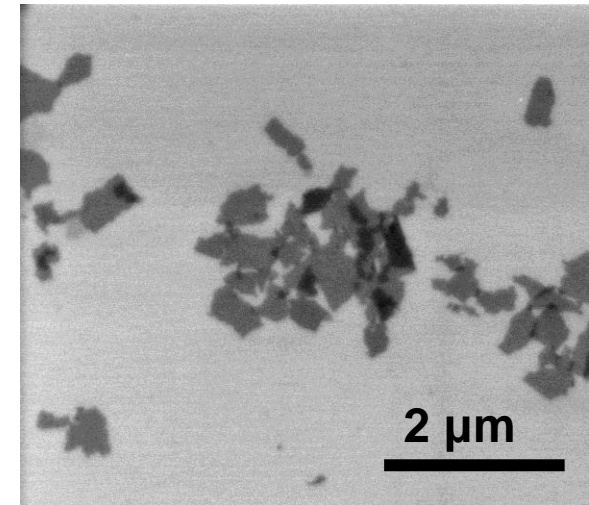


# RGO surface phase is stable, yet deposition is challenging

## RGO structure



With  $\text{NH}_4\text{OH}$ , stable RGO monolayer, yet poor deposition

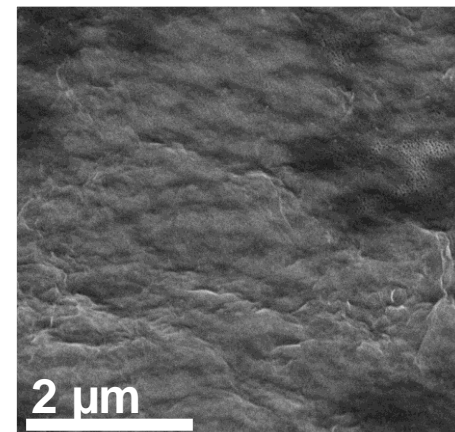
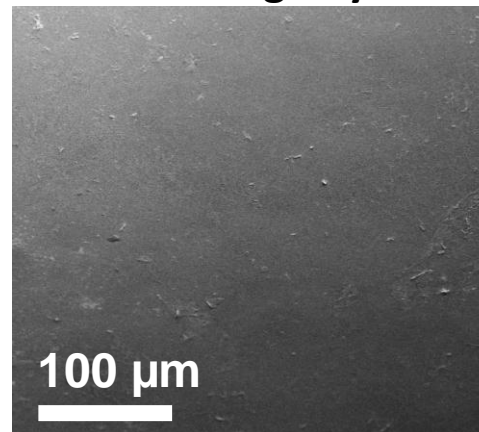
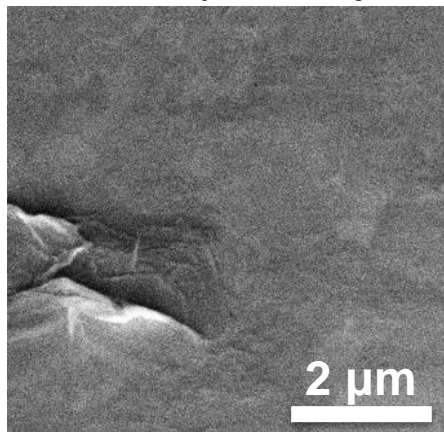
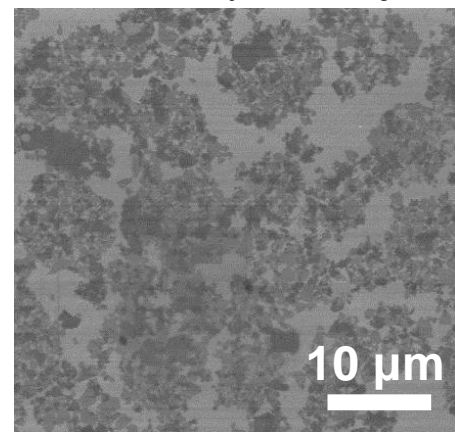


With methanol, RGO flakes cluster together, increasing deposition on Si and Anodisc.

RGO on Si, 0 mN/m

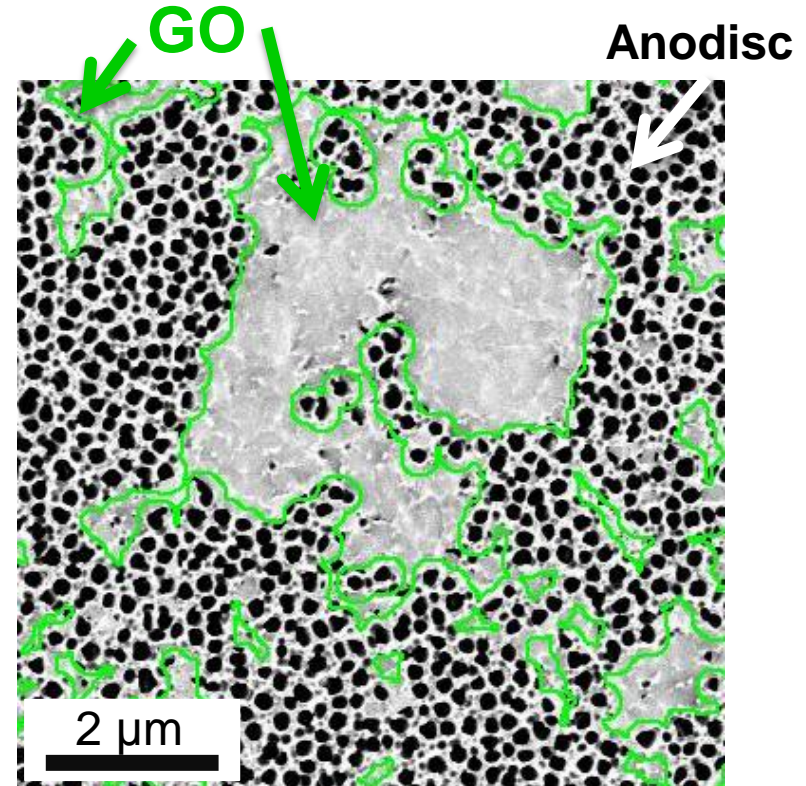
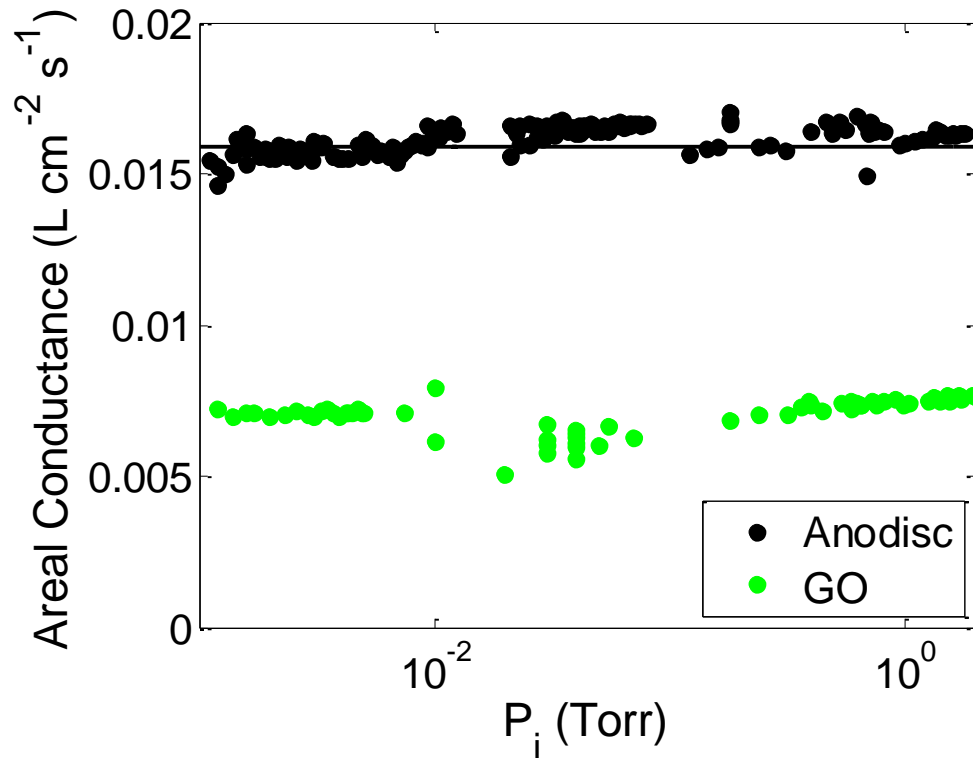
RGO on Si, 10 mN/m

>95 % coverage by RGO on 200-nm Anodisc



# Preliminary permeation results demonstrate decreased conductance due to GO membrane

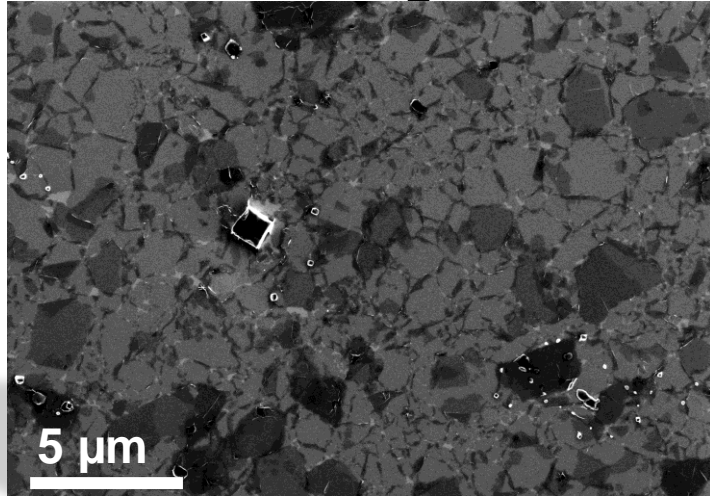
40 % decrease in conductance



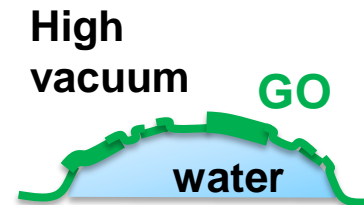
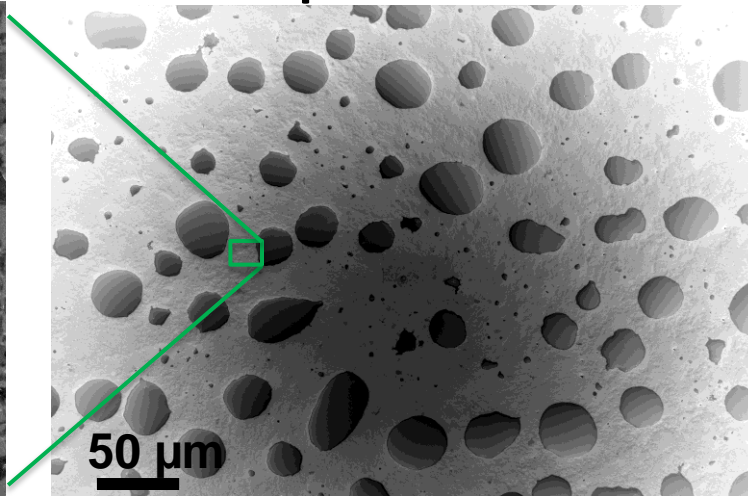
**15  $\pm$  5 % GO coverage**  
on this side of the sample.

# Ordered assembly of GO necessary to prevent permeation of water

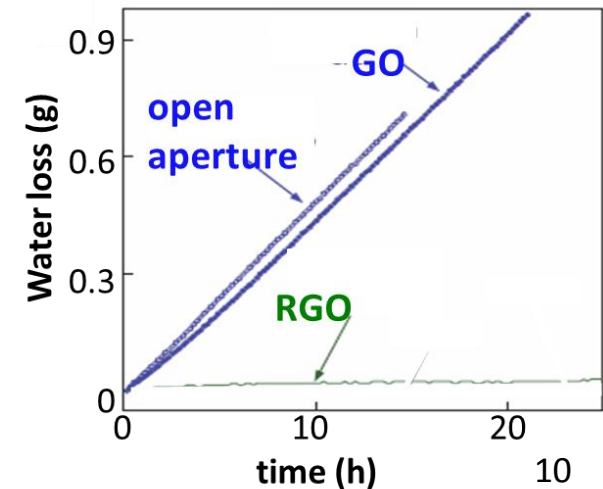
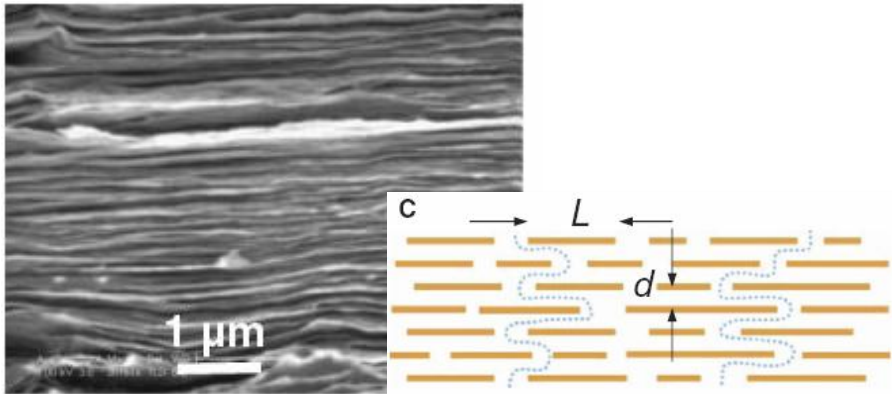
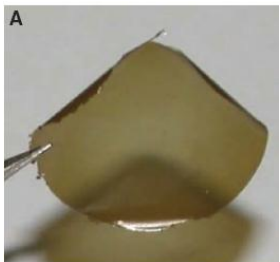
98 % GO coverage on Si.



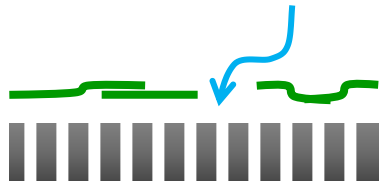
GO overlaps to encase bubbles



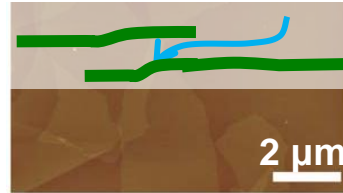
Unimpeded Permeation of Water Through Helium-Leak-Tight  
Graphene-Based Membranes  
R. R. Nair *et al.* *Science* 335, 442 (2012);



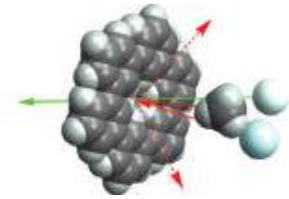
# Optimizing permeation pathways through a GO membrane



Transition regime



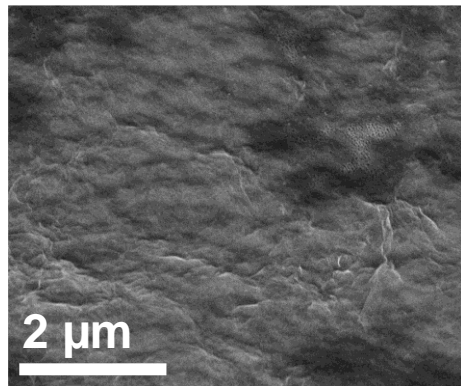
Free-molecular regime



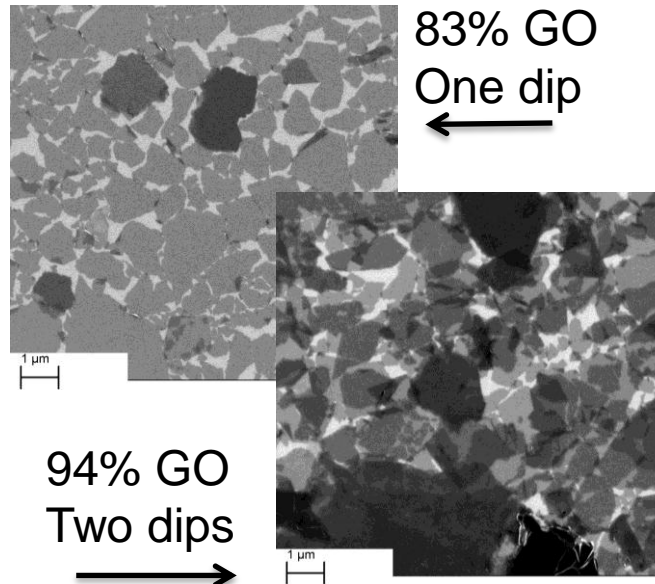
Size-selective permeability

## Optimized substrates

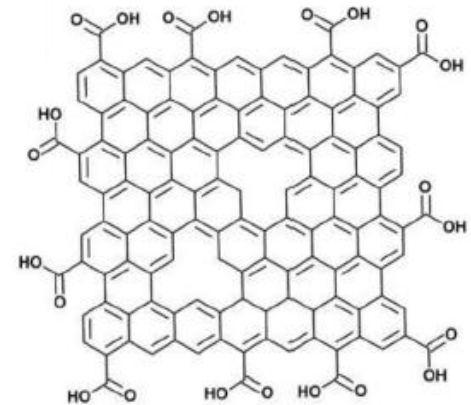
Microchannel plate?  
Polycarbonate track  
etched filter?



## Tunable assembly



## UV-oxidative etching Chemical reduction, **GO → RGO**



# Conclusions

Demonstrated L-B assembly of GO and RGO films on a porous substrate

- Pathway for supported membrane structures

Achieved Langmuir-Blodgett deposition of RGO membranes, which hasn't been reported before.

- RGO membrane assembly requires a network of overlapping flakes.

Discovered smoothly overlapping graphitic membranes restrict water permeation.

## Acknowledgements

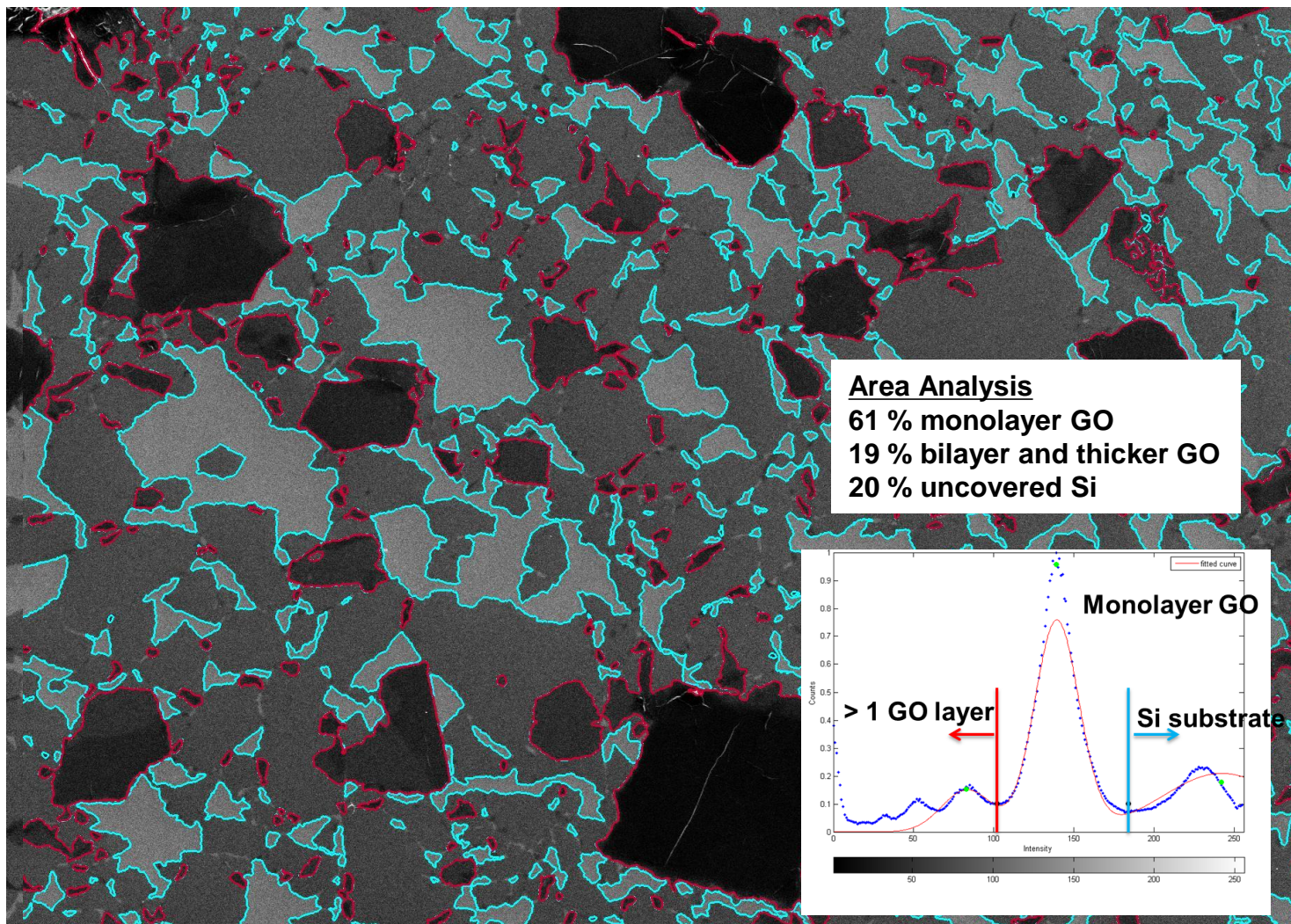
**Graphene oxide:** Tim Lambert

**Experimental assistance:** Susan Brozik, Ron Goeke, Stephen Rose, Cody Washburn, and Dave Wheeler

**Scanning electron and optical microscopy:** Bonnie McKenzie and Alice Kilgo

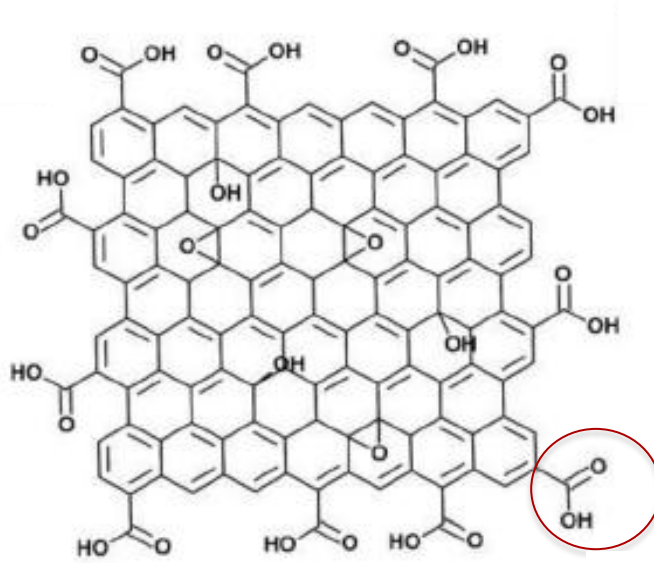


# Determination of Graphene Coverage

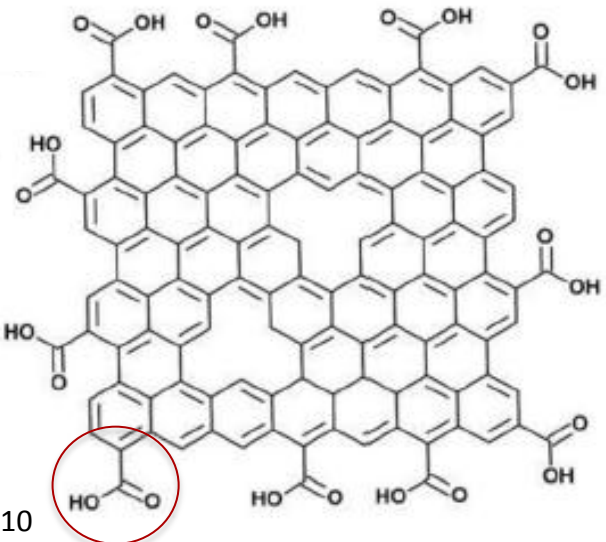


# Compare properties of GO and RGO films

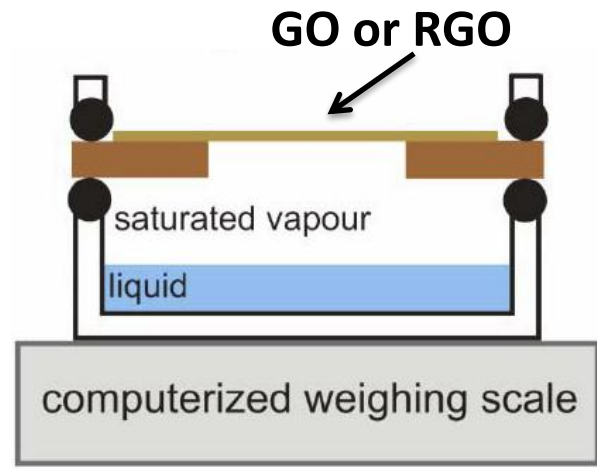
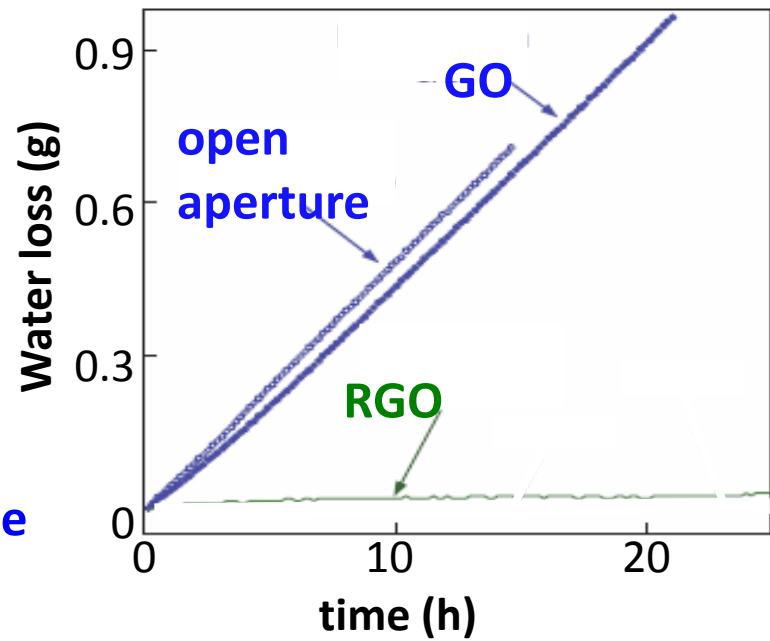
## Graphene Oxide (GO)



## Reduced Graphene Oxide (RGO)



+ 80 °C and  
Ascorbic acid  
(Vitamin C)



G. Eda and M. Chhowalla, *Adv. Mater.* 2010  
M. Fernandez-Merino *et al.*, *J. Phys. Chem. C* 2010

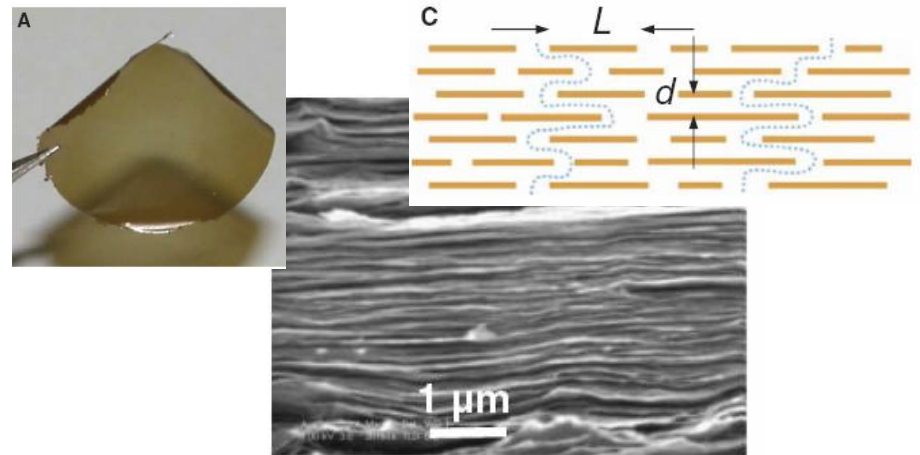
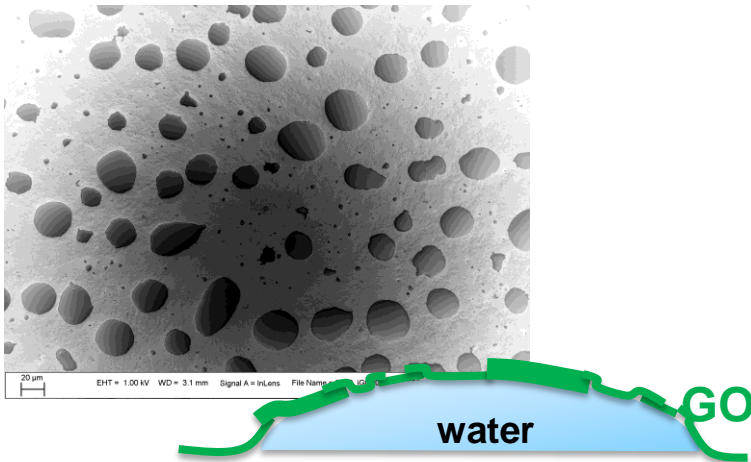
R. Nair *et al.* *Science* 2012

# Conclusions

Optimized deposition of GO on a variety of substrates, including porous alumina.

Achieved Langmuir-Blodgett deposition of RGO membranes, which hasn't been reported before.

Discovered both GO and RGO membranes restrict water permeation.



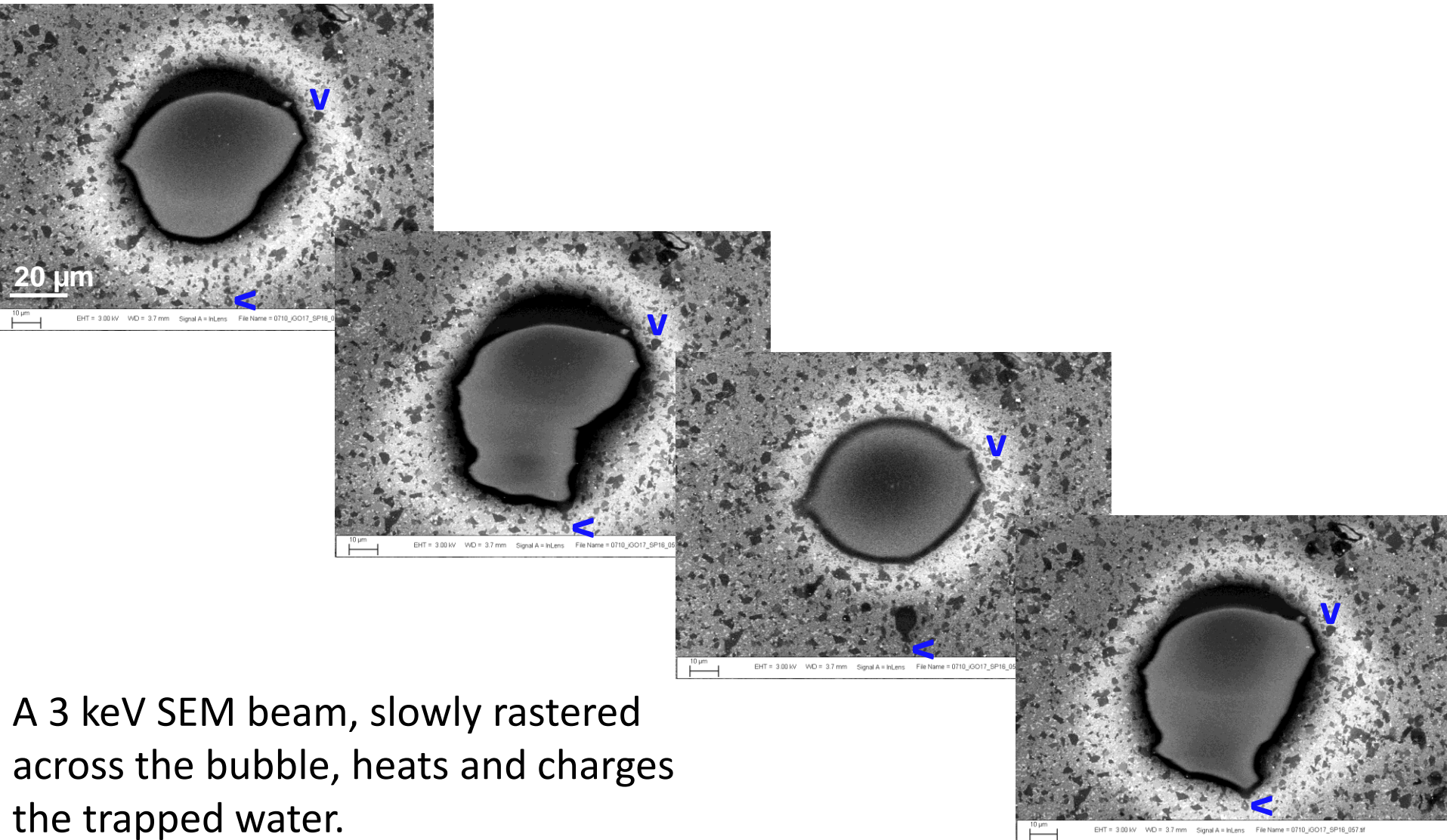
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**Graphene oxide:** Tim Lambert

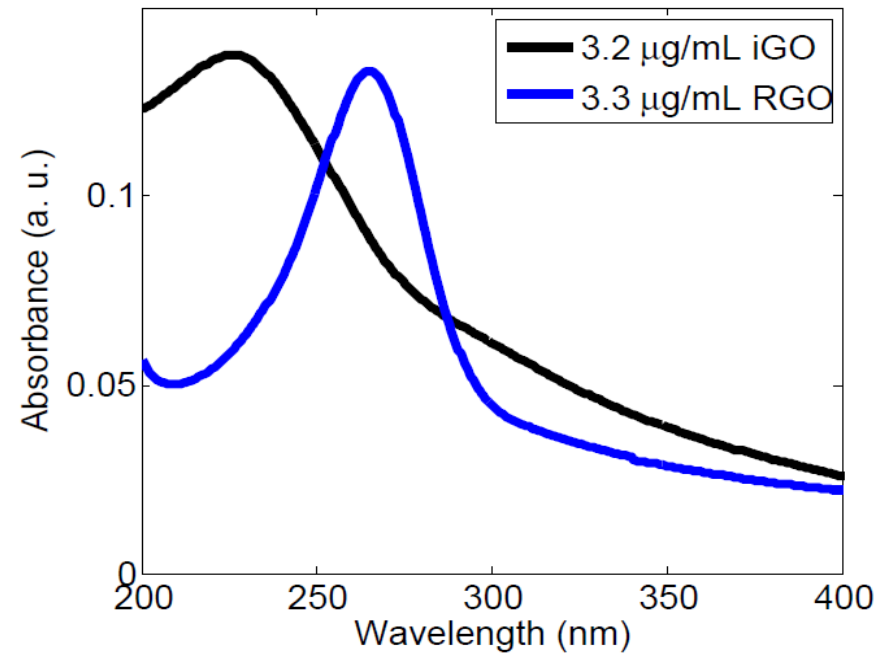
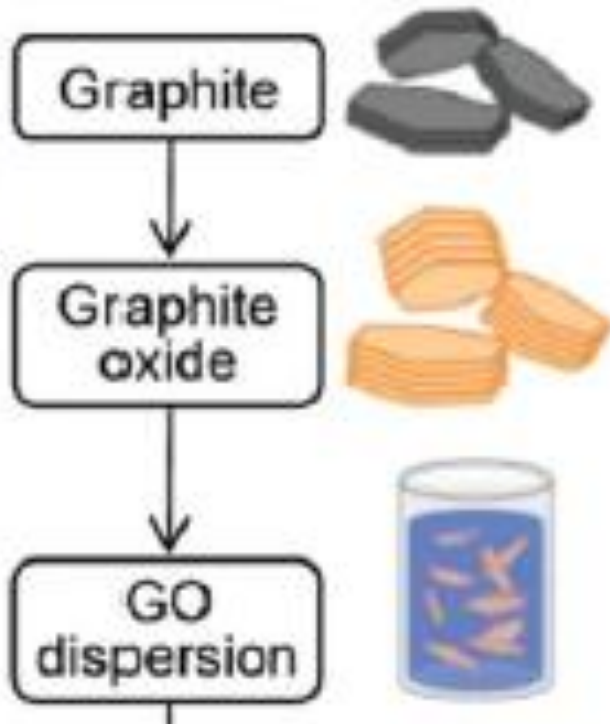
**Experimental assistance:** Susan Brozik, Cody Washburn, and Dave Wheeler

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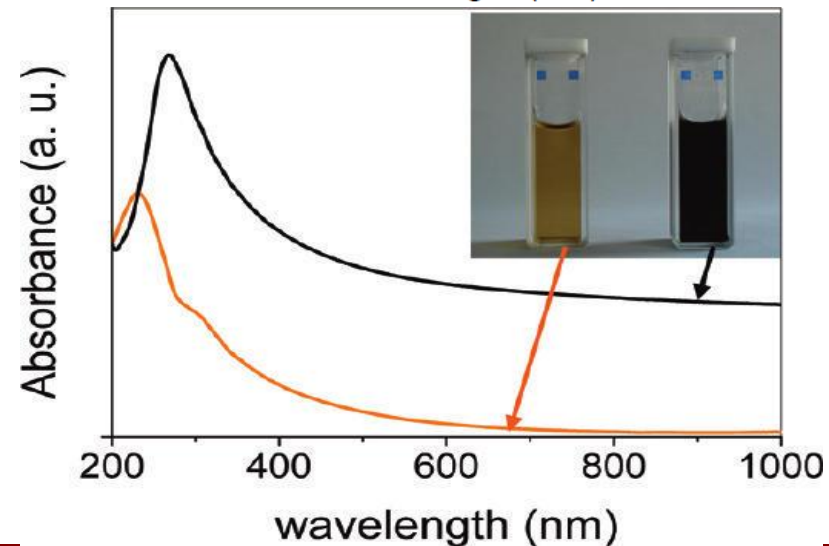
# Composite GO membranes can stretch without bursting



# Natural graphite is oxidized via Hummer's method

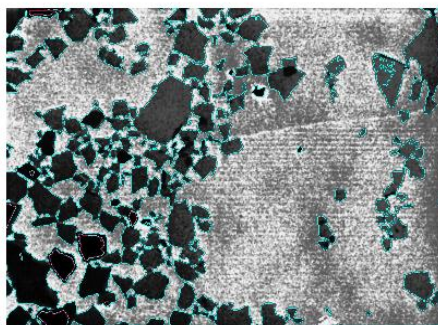
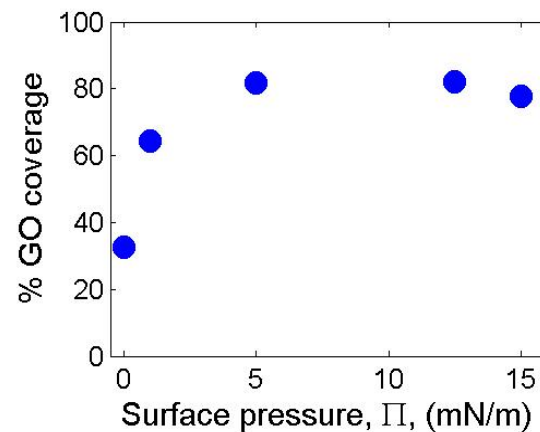
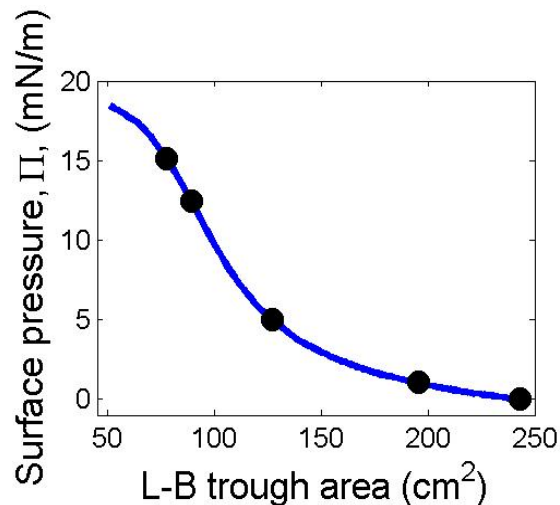
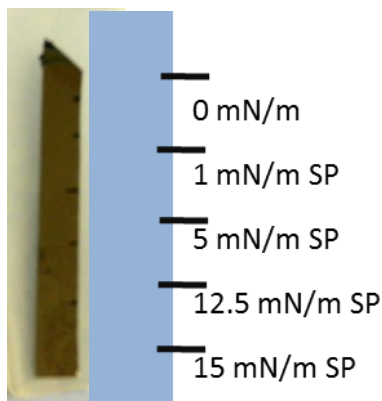


Oxidized using a mixture of  $\text{NaNO}_3$ ,  $\text{H}_2\text{SO}_4$ , and  $\text{KMnO}_4$ .  
Diluted and heated with  $\text{H}_2\text{O}$  and  $\text{H}_2\text{O}_2$   
Filtered and rinsed with  $\text{HCl}$

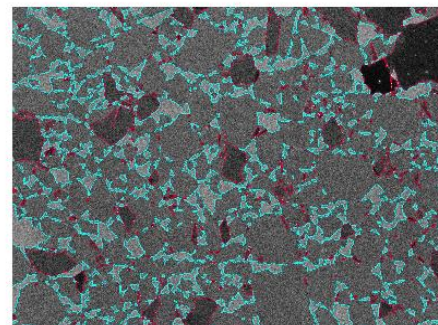


Paredes *et al.*, *Langmuir* 2009.

# GO coverage can be controlled through surface pressure



GO deposited at  $\Pi = 0$  mN/m

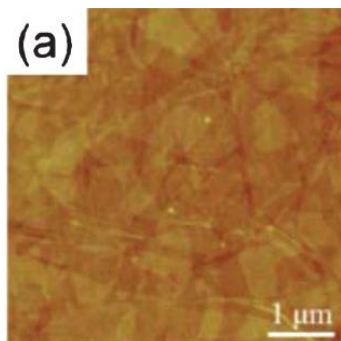
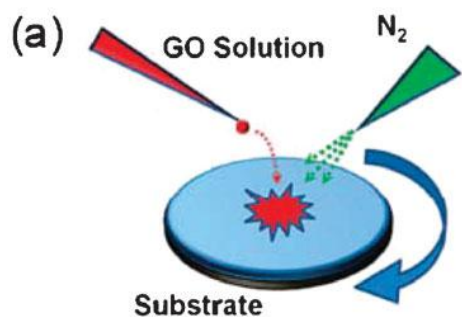


GO deposited at  $\Pi = 12.5$  mN/m

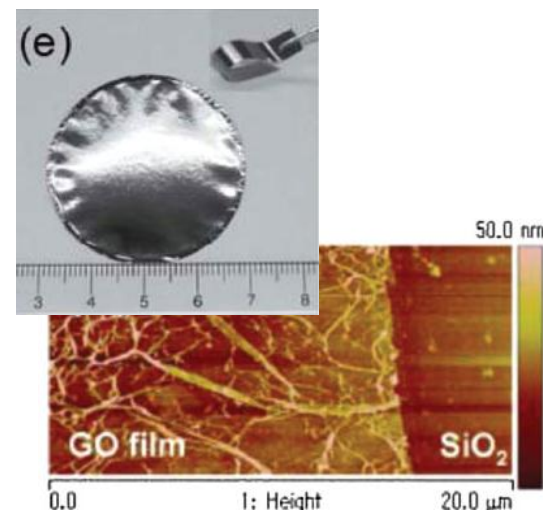
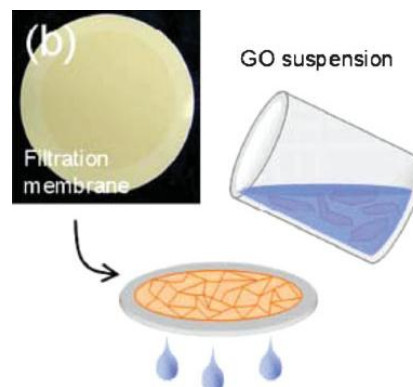
GO coverage plateaus at SP  $\approx 5$  mN/m

# Assembly of GO and RGO films

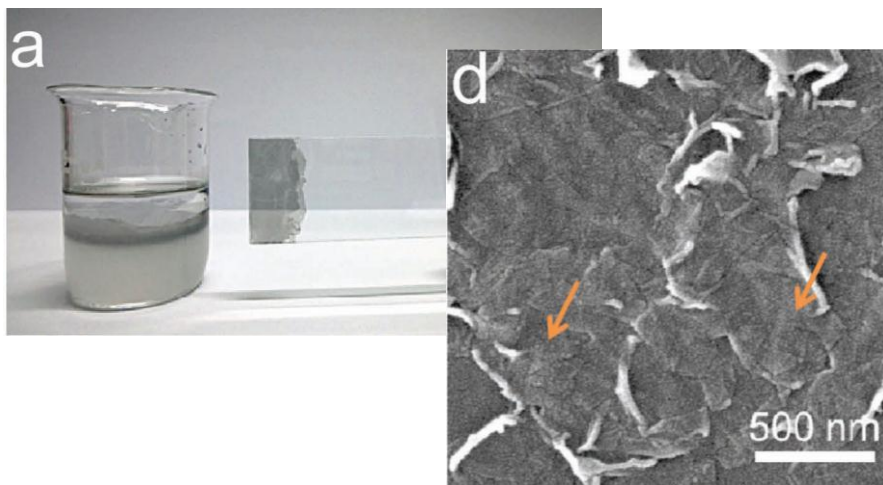
## Drop casting<sup>1</sup>



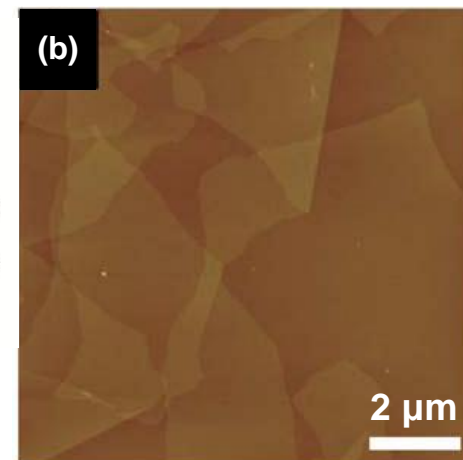
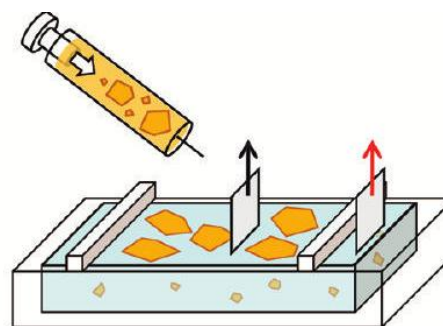
## Vacuum filtration<sup>2</sup>



## Assembly at an oil-water interface<sup>3</sup>



## Langmuir-Blodgett deposition<sup>4</sup>



[1] G. Eda and M. Chhowalla, *Adv. Mater.* 2010

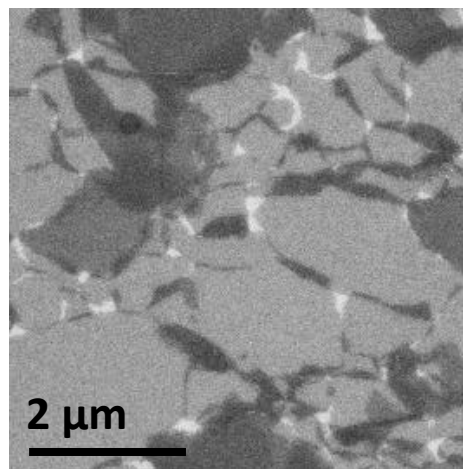
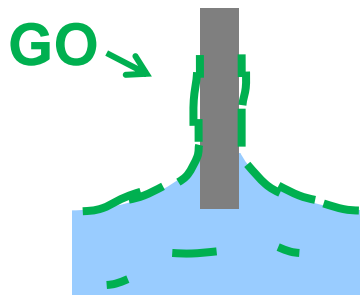
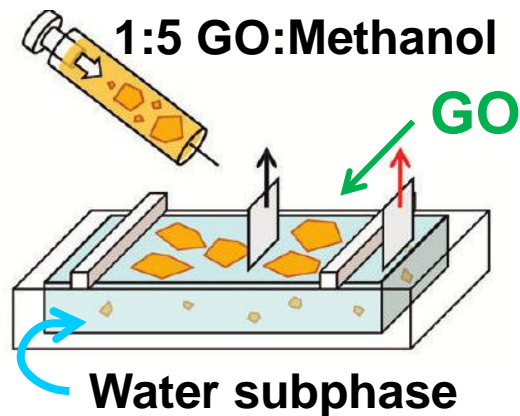
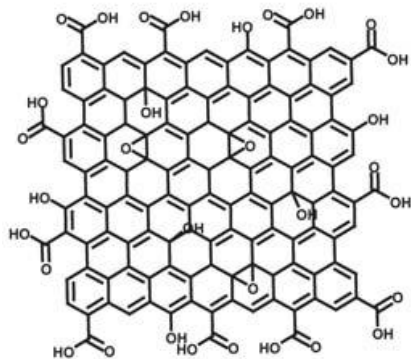
[2] S.-K. Lee *et al.* *Nano Lett.* 2012

[3] S. Gan *et al.* *Adv. Mater.* 2012

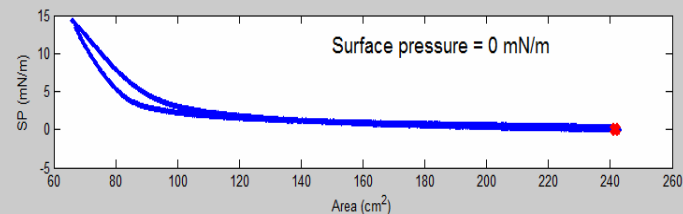
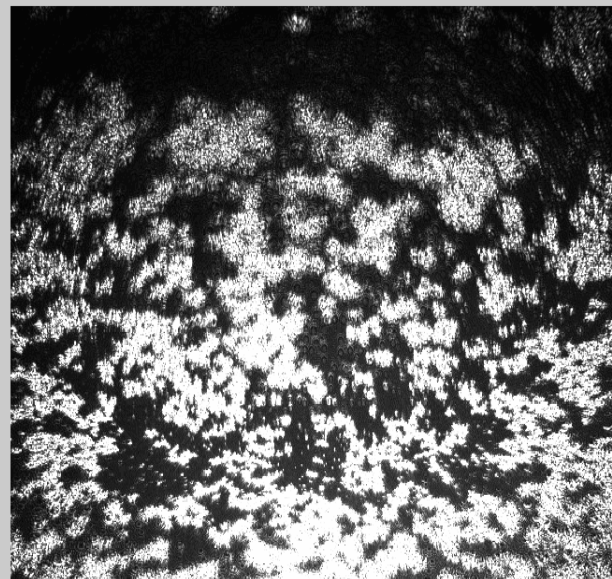
[4] L. Cote *et al.* *Soft Matter* 2010.

# Stable graphene oxide surface phase enables Langmuir-Blodgett assembly

## GO structure



08/08/12, iGO01: Compression at 10 mm/min



**Goal: Optimize deposition to generate continuous GO and RGO films**