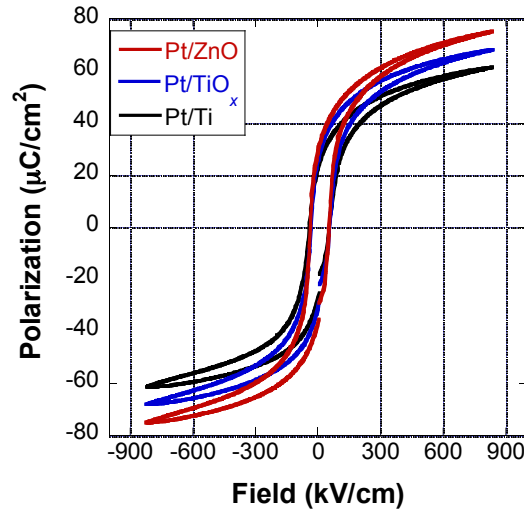


Solution Chemistry, Substrate, and Processing Effects on the Chemical Heterogeneity in PZT Films



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Collaborators

Sandia National Labs

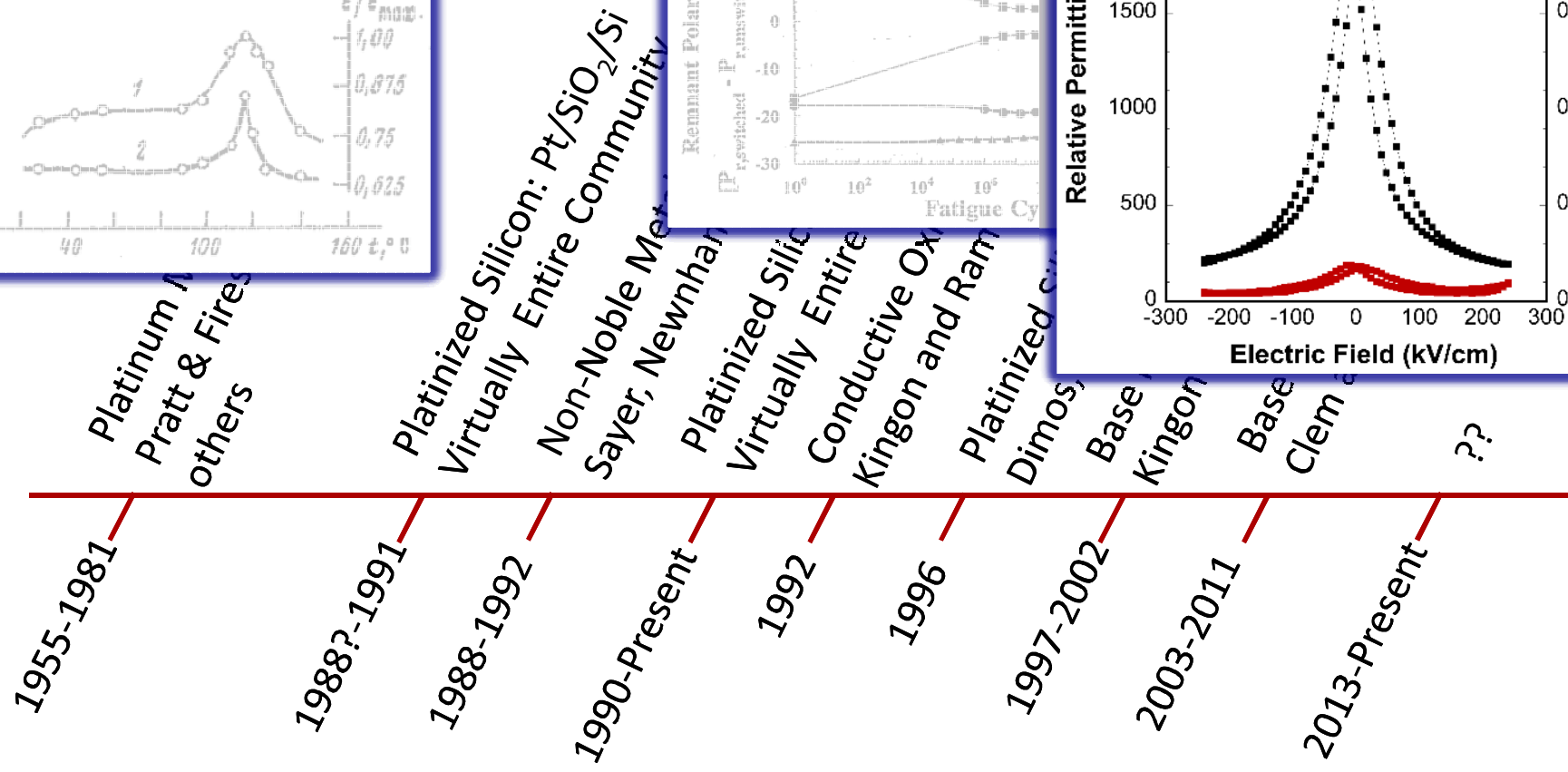
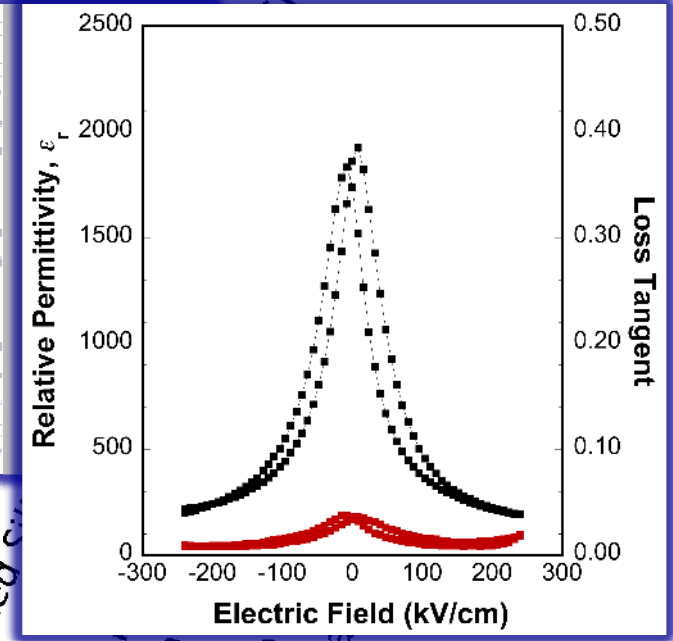
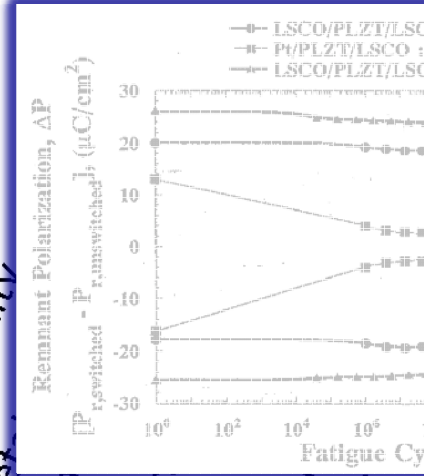
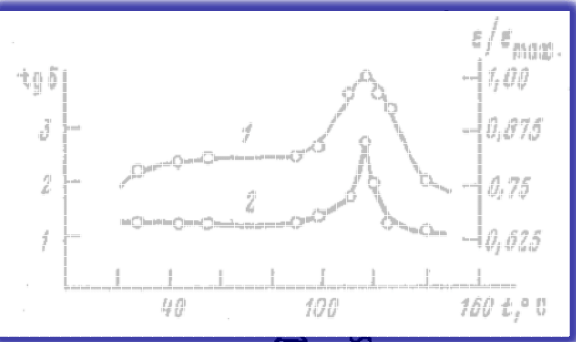
Microscopy

- Paul Kotula
- Bryan Gauntt
- Ping Lu
- Bonnie McKenzie
- Amy Allen
- Michael Rye

Solution/Film Preparation

- Chris T. Shelton
- Geoff Brennecke
- Dara V. Gough
- Erik Spoerke
- Kelsey Meyer
- Harlan Brown-Shaklee

A brief history of ferroelectric thin film substrates*



Lessons Learned (and forgotten?)

Fox, *et al.* 1994

Kim, *et al.* 2000

Tani and Payne 1992

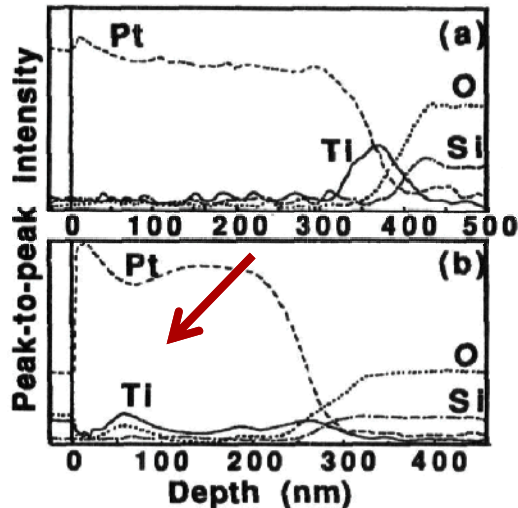


Fig.2 AES of A substrates (a) before and (b) after annealing at 650 °C for 30 min.

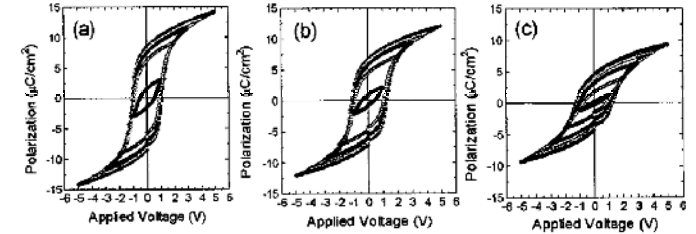
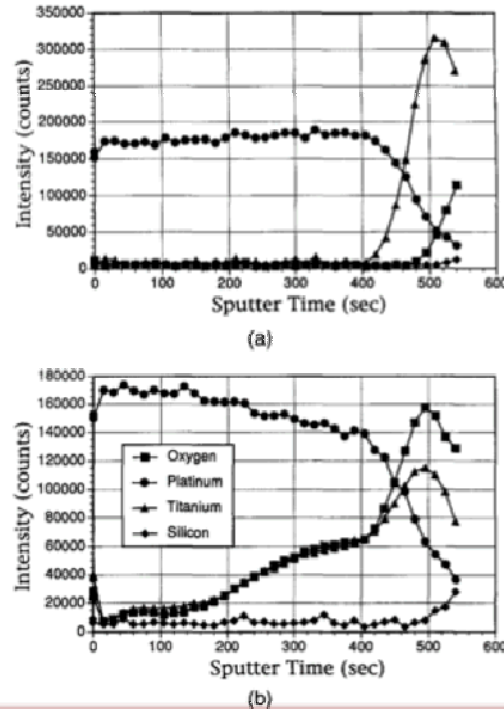
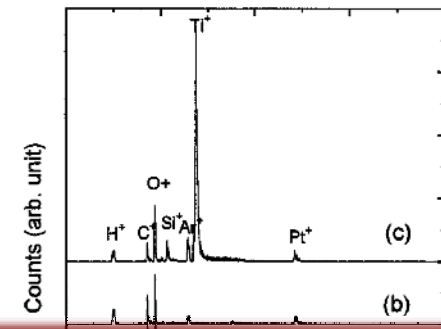
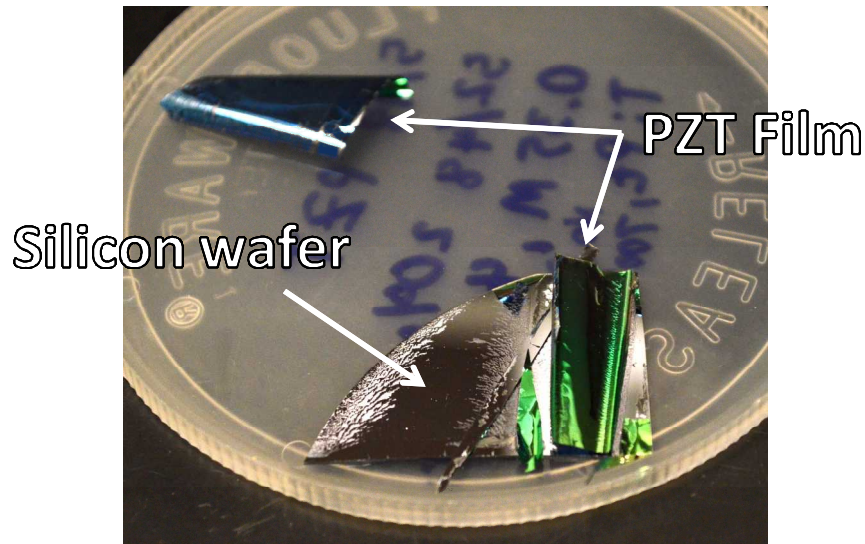
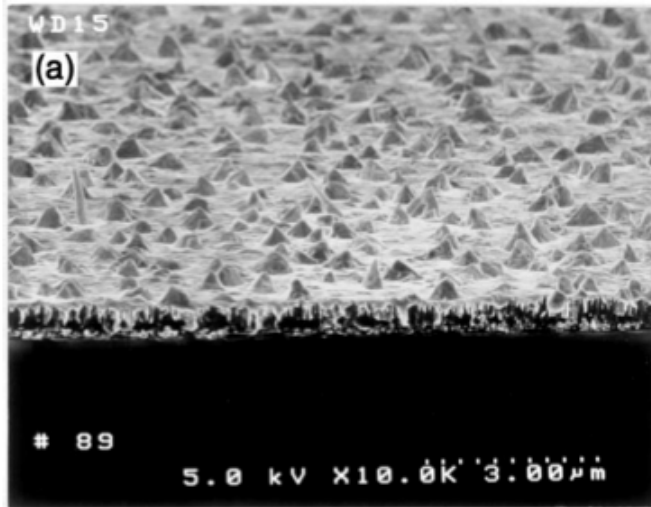


FIG. 2. $P-V$ hysteresis loops of SBT thin films (a) on Pt/ZrO₂/SiO₂/Si, (b) Pt/TiO₂/SiO₂/Si, and (c) Pt/Ti/SiO₂/Si substrates.



- Electrode/adhesion layer stacks cannot be considered as inert bystanders!
- Titanium-based adhesion layers are known to diffuse through platinum

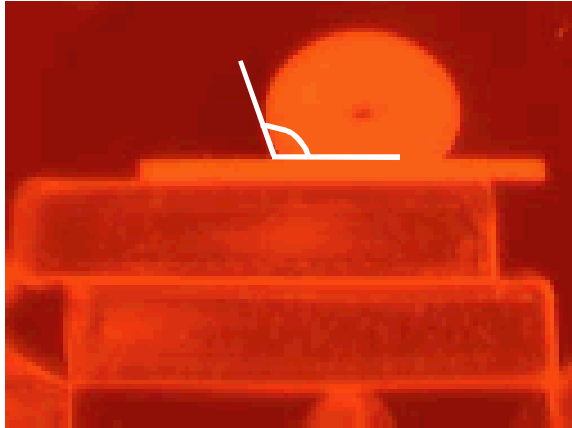
Pt/Ti remains state of the art in platinized silicon:



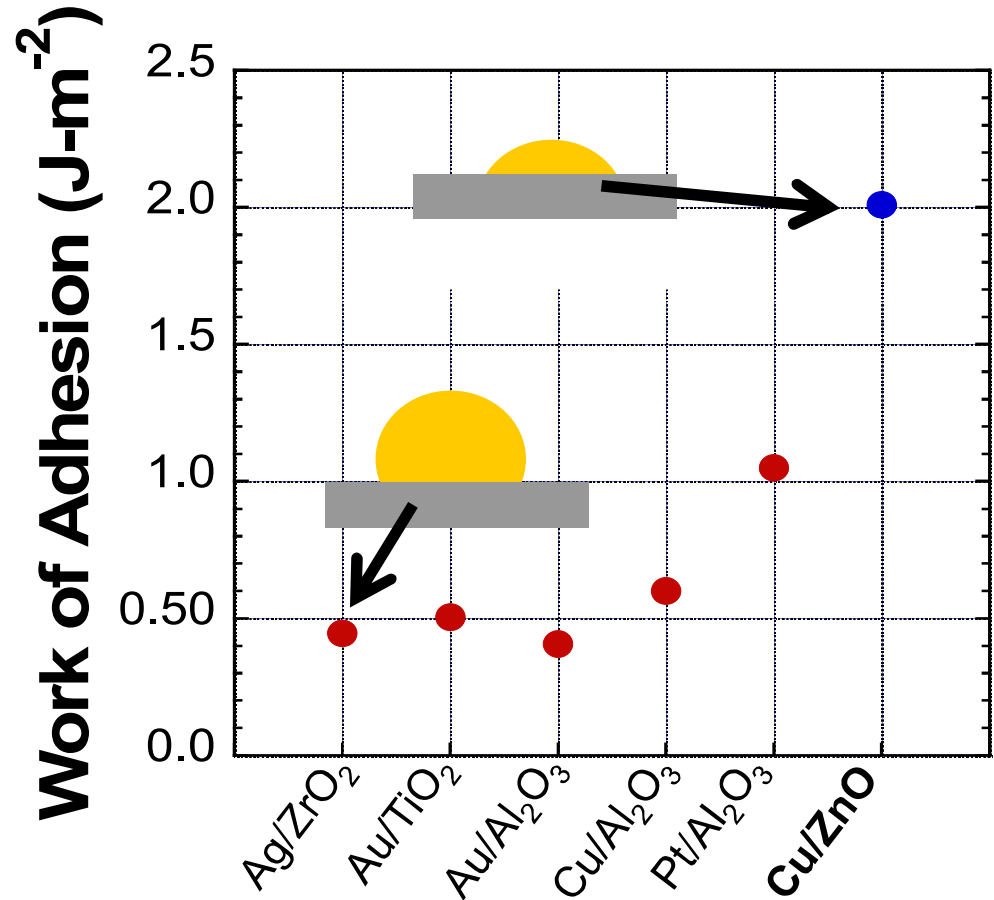
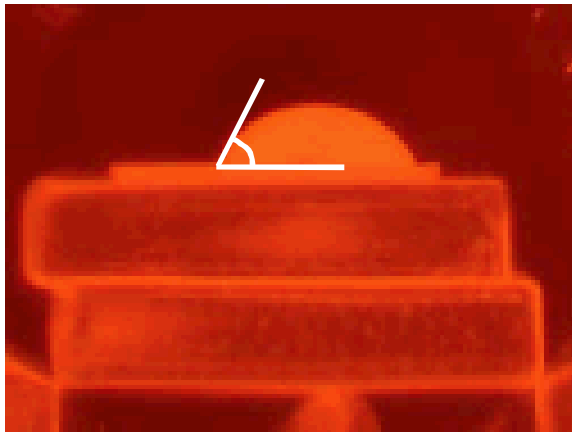
- Titanium 'adhesion' layers by far the most common
 - Adhesion of Pt/ZrO₂ and Pt/TiO₂ not as good as Pt/Ti
- Titanium expands upon oxidation causing hillocks
- Adhesion layer still fails with large temperature swings (>700°)
- **>20 years of research and small improvements made**
- ***There must be a better solution!***

Metal Wetting and Adhesion

Molten Cu on Al_2O_3



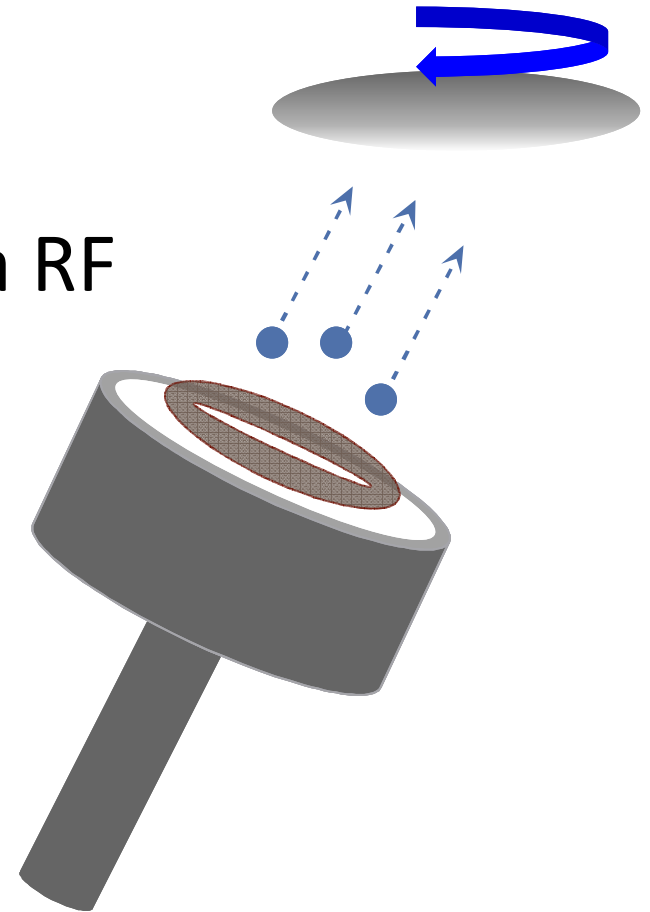
Molten Cu on ZnO



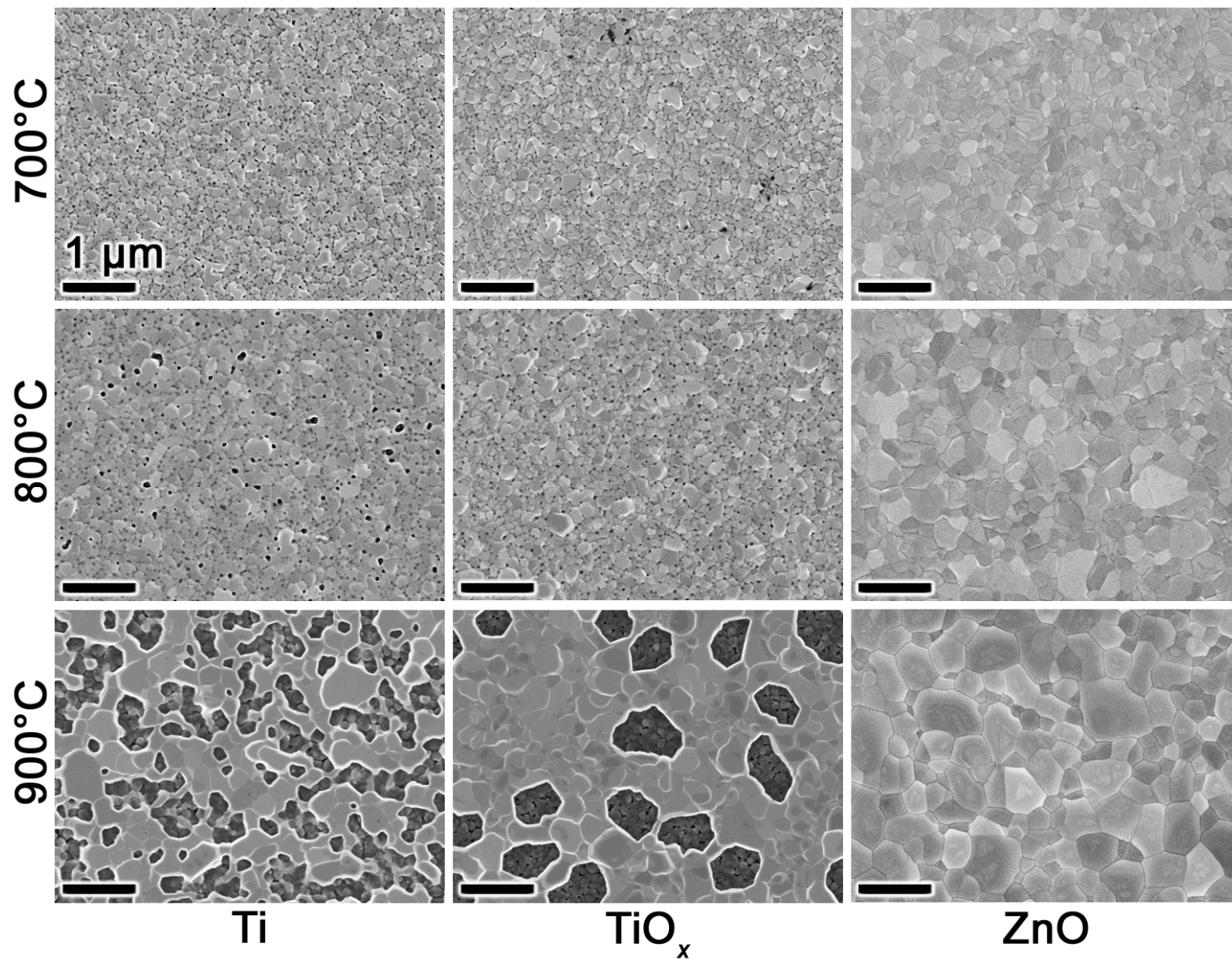
Can we use lessons learned from wetting experiments to guide selection of improved adhesion layers?

Pt/Si Substrate Preparation

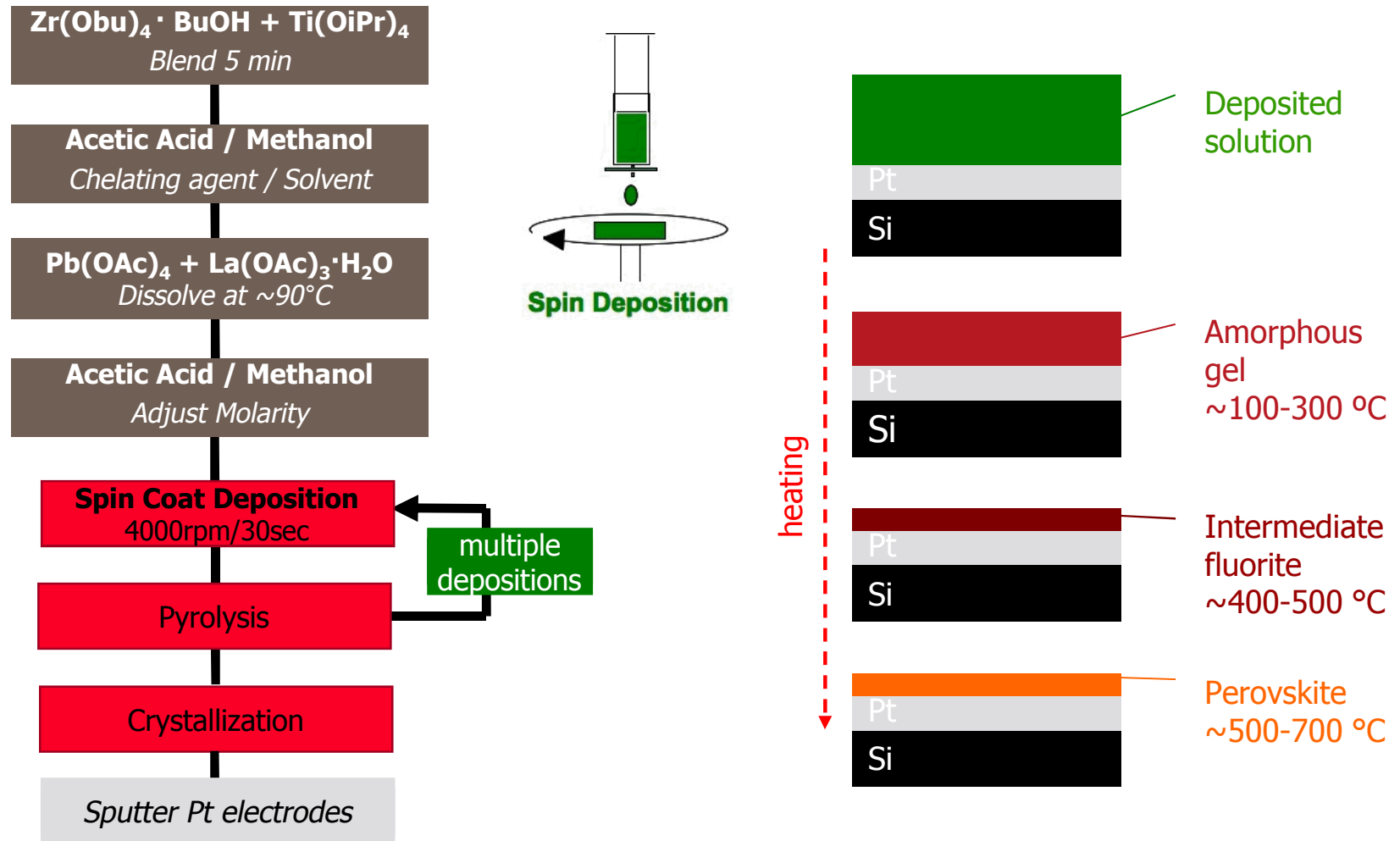
- 100 nm Pt/40 nm buffer/400 nm SiO_2/Si
- Ti, TiO_x , ZnO, and Pt prepared via RF magnetron sputtering
 - 30° off-axis geometry
 - Ti: Ar atmosphere
 - TiO_x : Ar sputter, O_2 anneal
 - ZnO: 1:1 Ar: O_2 atmosphere
 - Pt: Ar Atmosphere



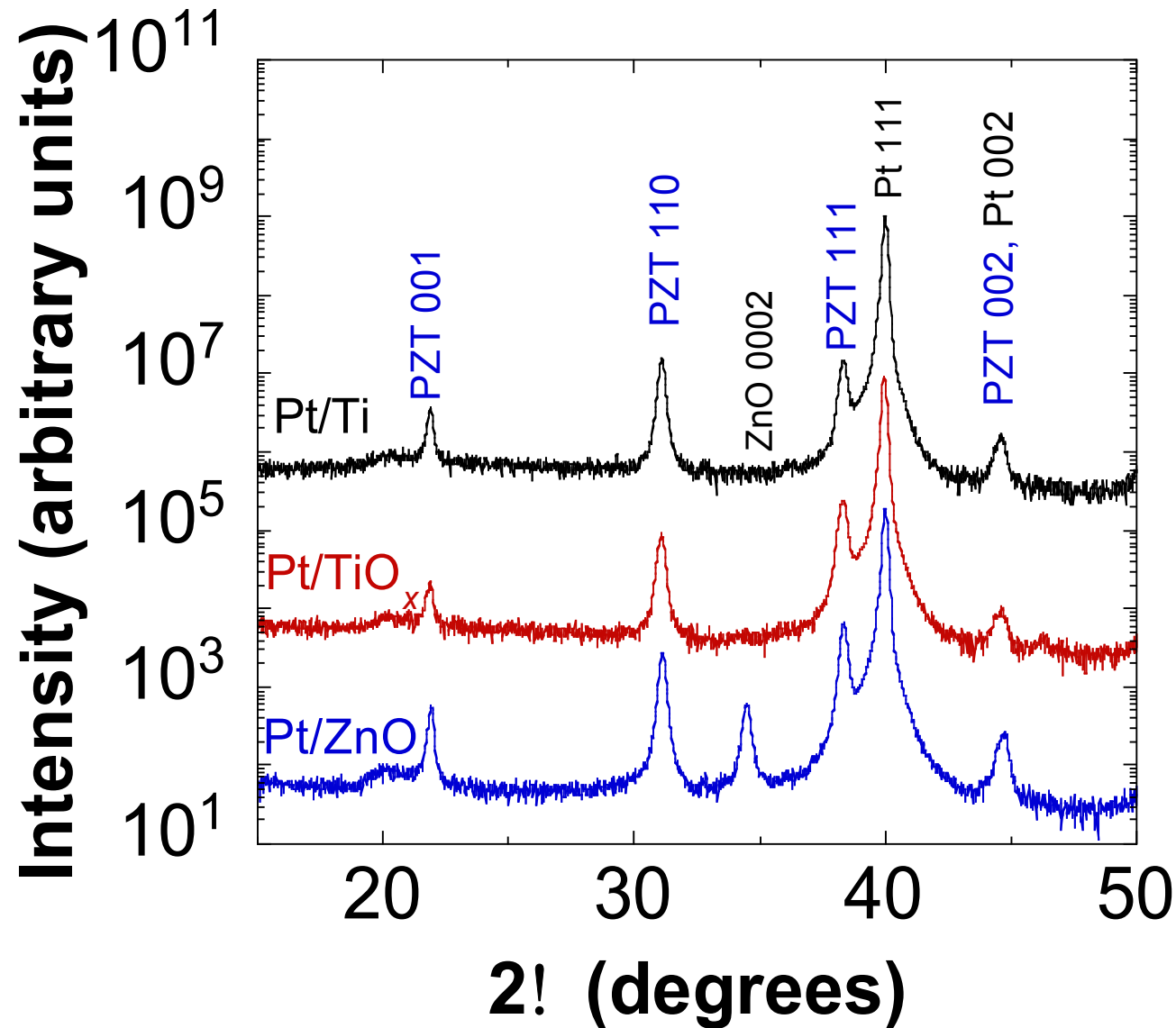
Electrode Temperature Stability



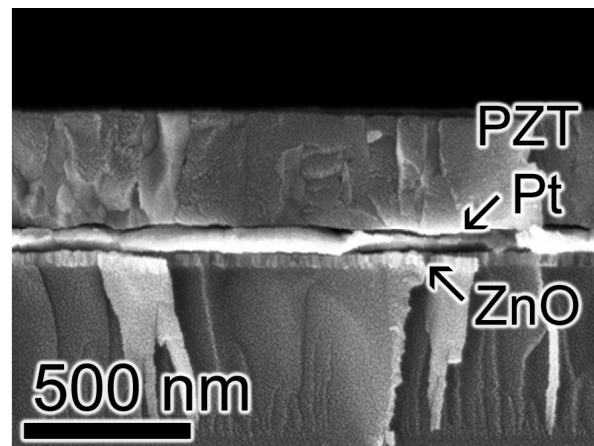
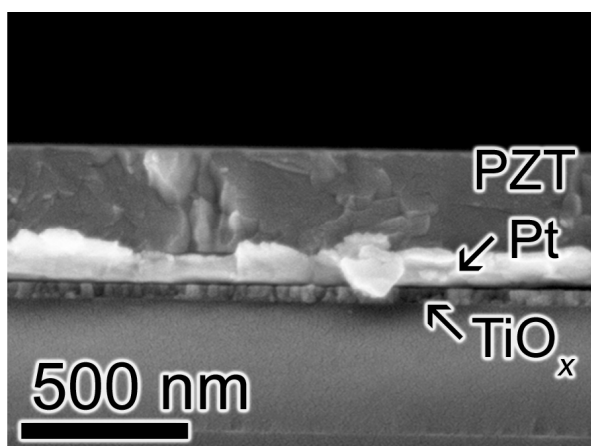
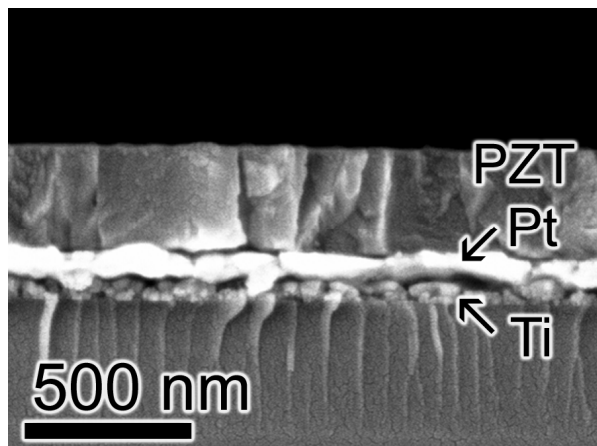
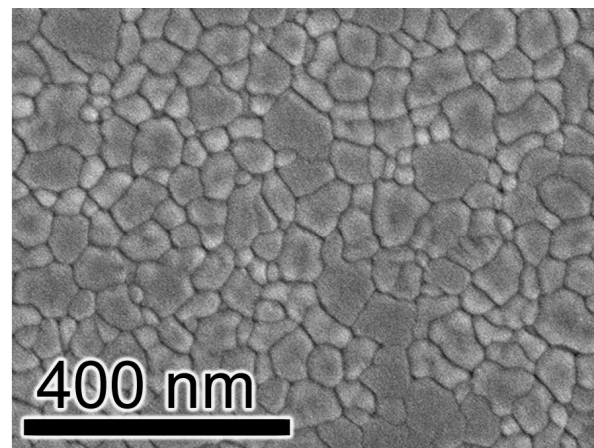
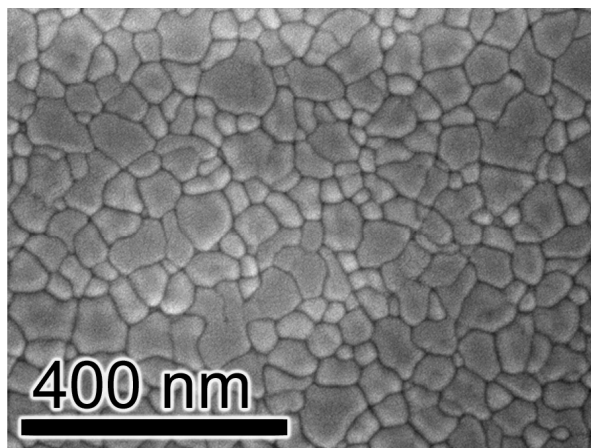
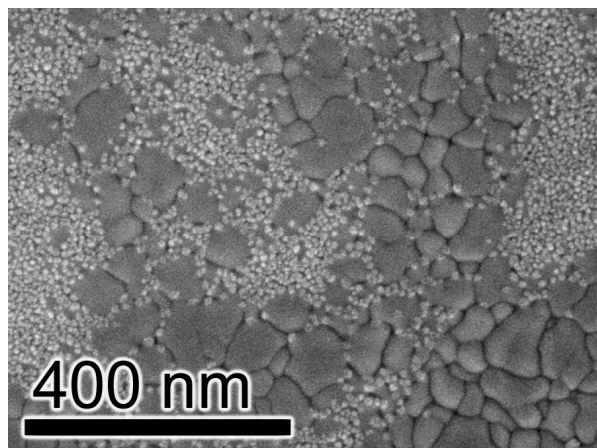
SNL IMO PZT CSD Film Preparation



Film Phase/Crystallographic Properties

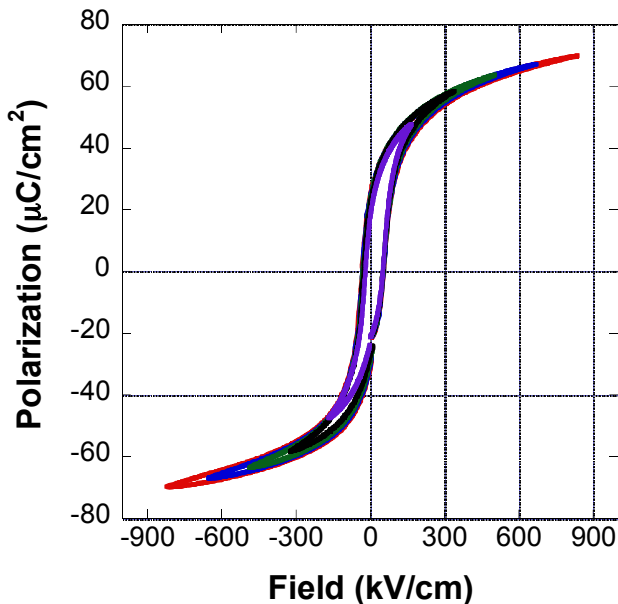


PZT Film Microstructure

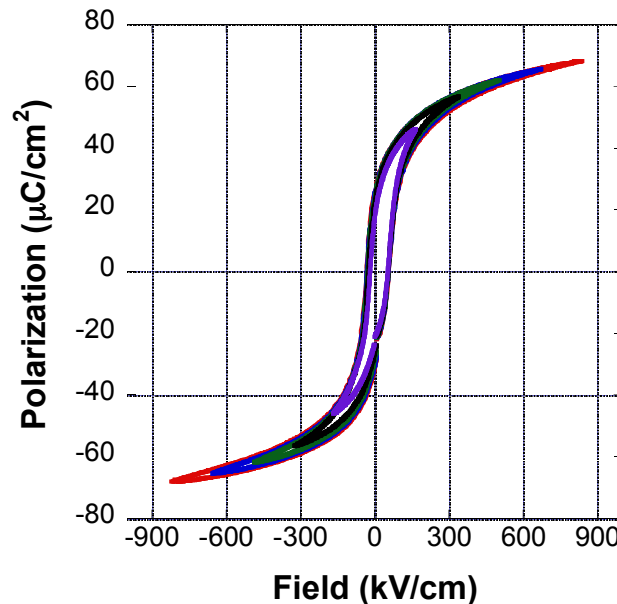


Dielectric and Ferroelectric properties

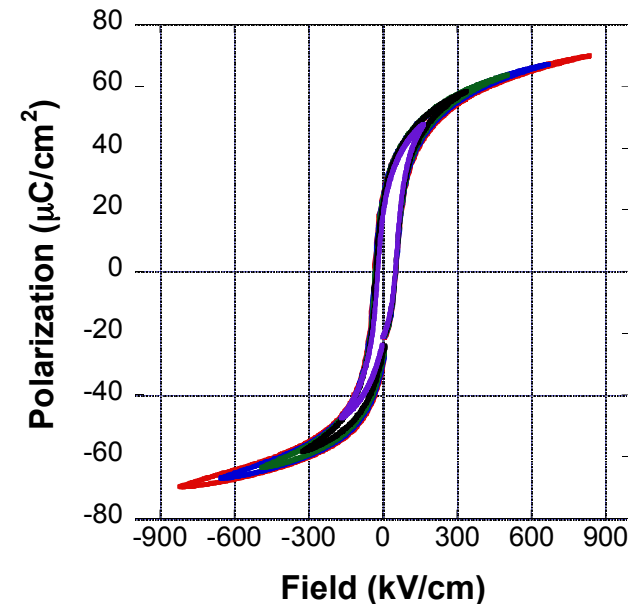
PZT/Pt/Ti



PZT/Pt/ TiO_x



PZT/Pt/ZnO

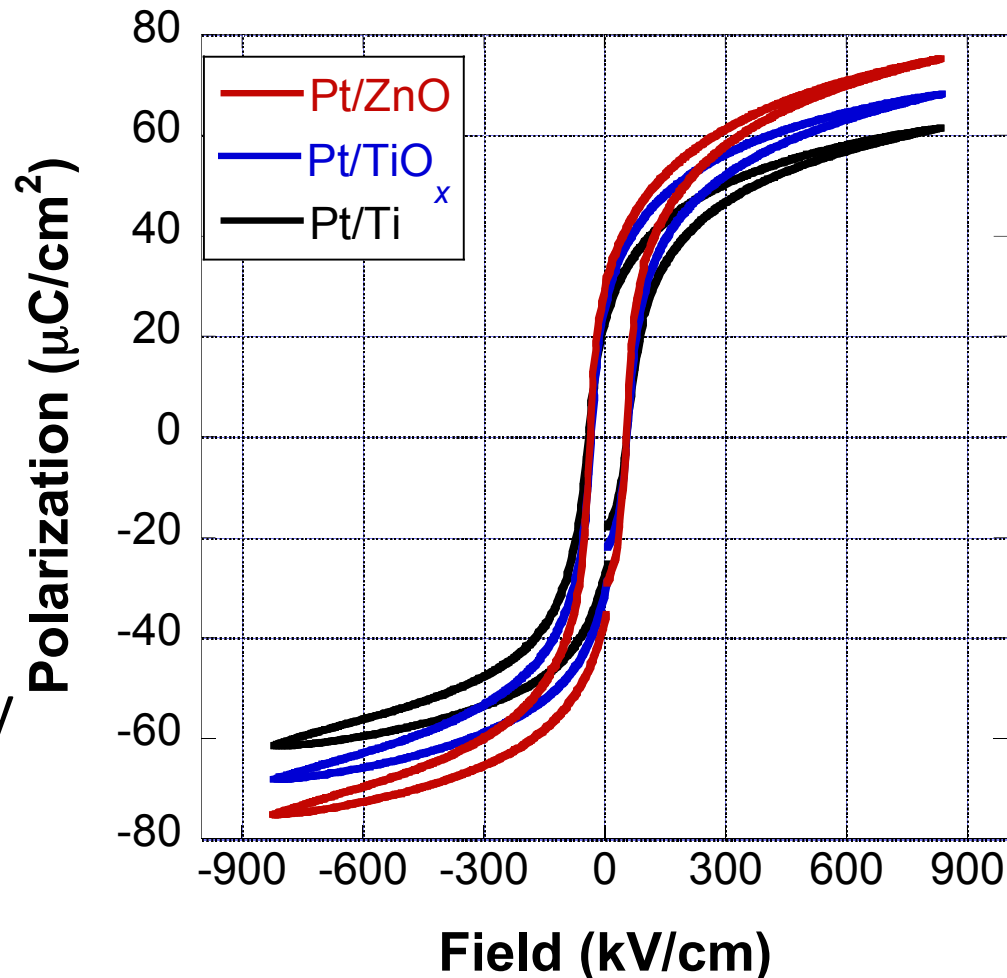


- Measurements conducted at 10 Hz
- All films saturate well
- Low leakage

PZT Films Ferroelectric Properties: Pt/ZnO Substrates Superior

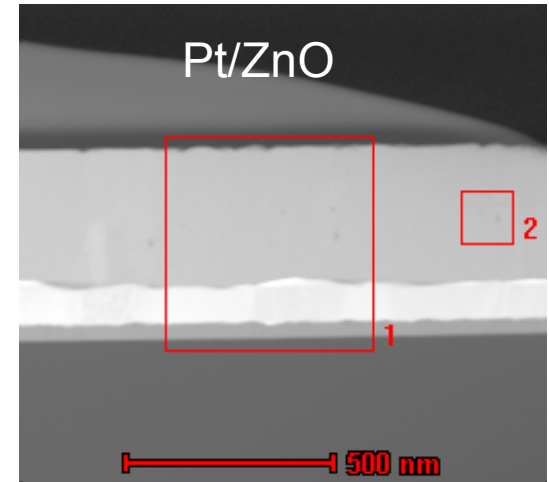
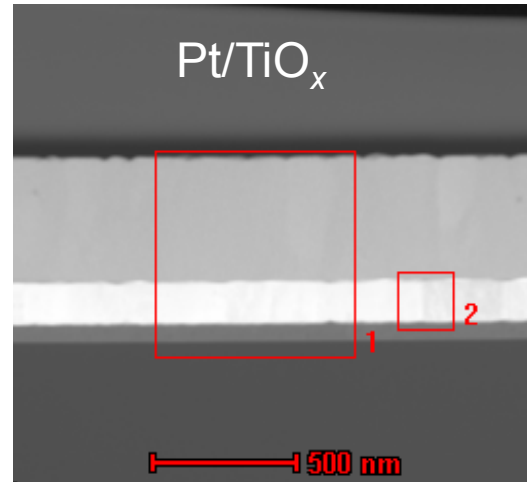
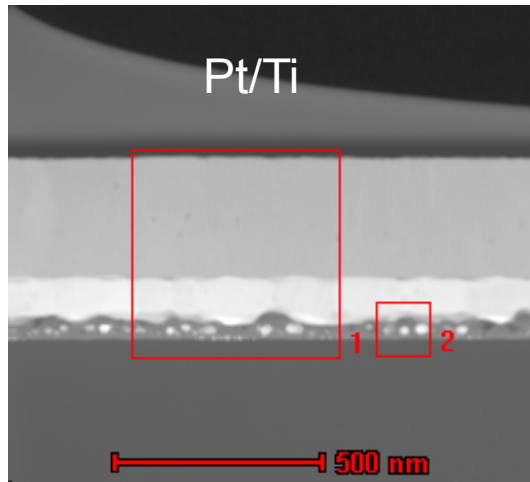
	$P_{\max}$ ($\mu\text{C}/\text{cm}^2$)	P_r ($\mu\text{C}/\text{cm}^2$)	E_c (kV/cm)	K
Pt/Ti	61.5	23.8	47.0	1440
Pt/TiO _x	68.2	27.0	44.1	1620
Pt/ZnO	75.3	32.2	43.8	1740

- Both P_r and K increase on ZnO-buffered substrates
- Coercive field decreases slightly for PZT on ZnO-buffered substrates

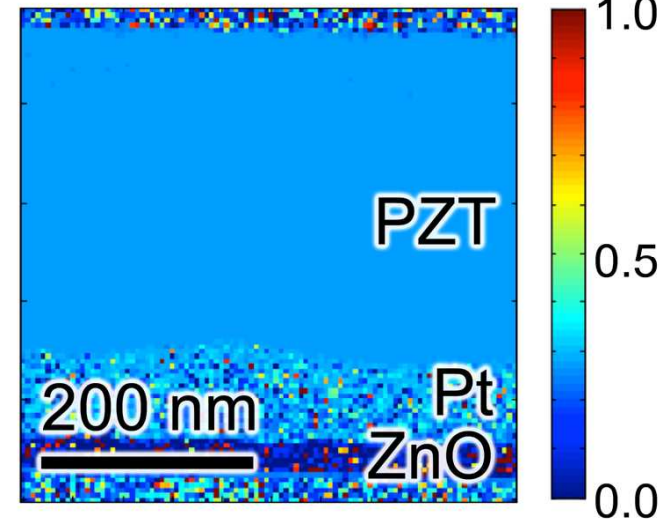
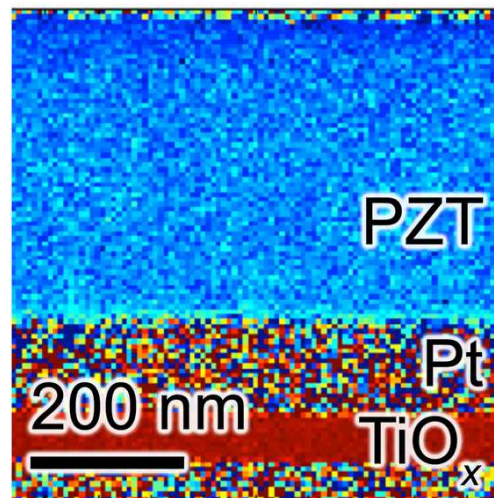
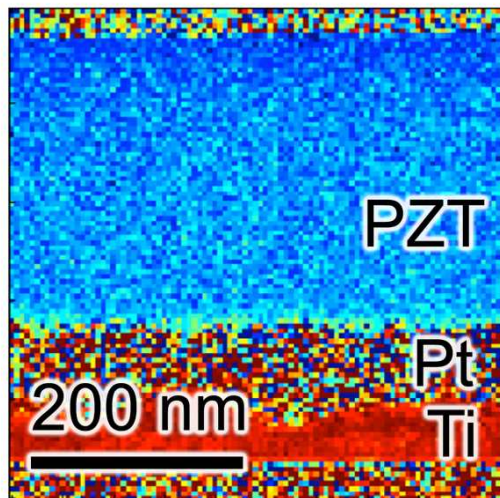


PZT Composition Analysis:

No Composition Gradient on Pt/ZnO

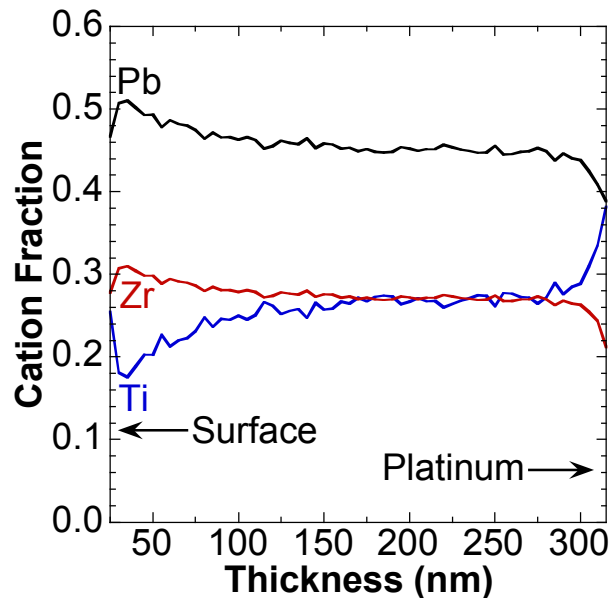


Titanium Distribution

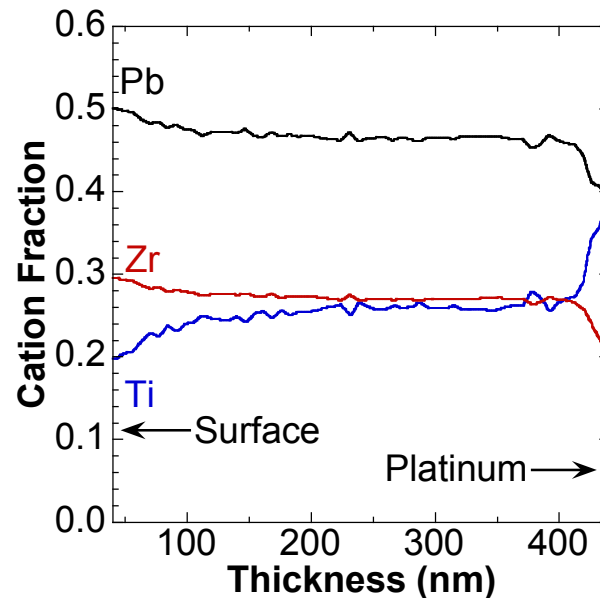


PZT Composition Analysis: No Composition Gradient on Pt/ZnO

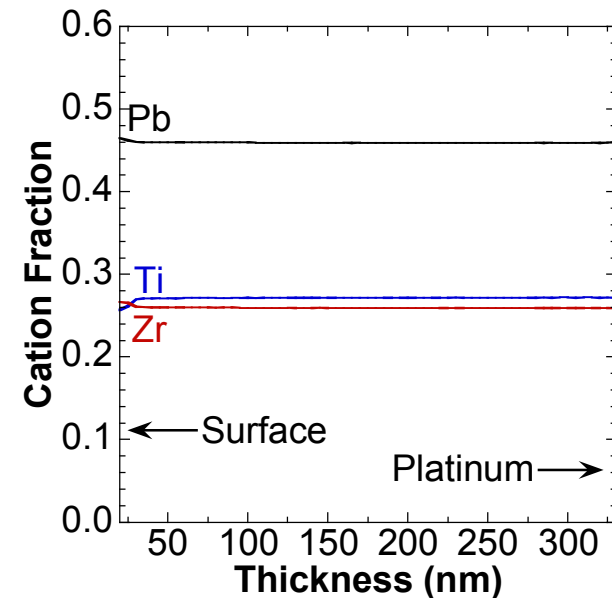
Pt/Ti



Pt/TiO_x

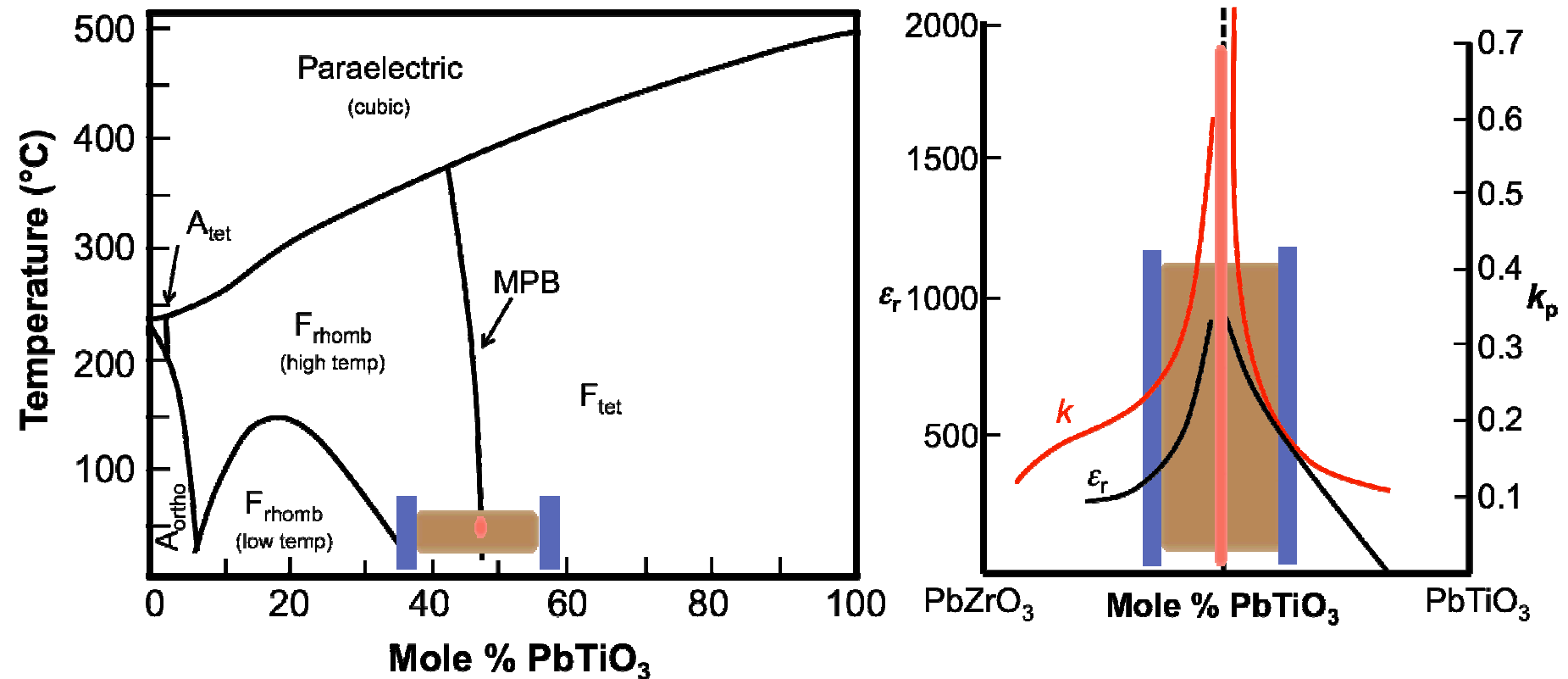


Pt/ZnO



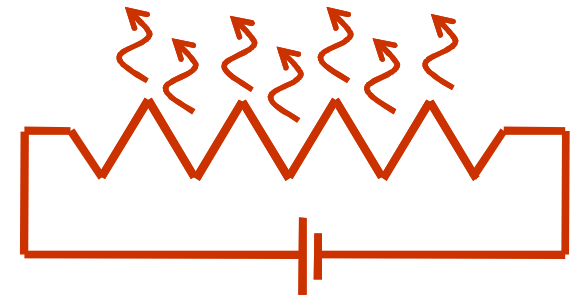
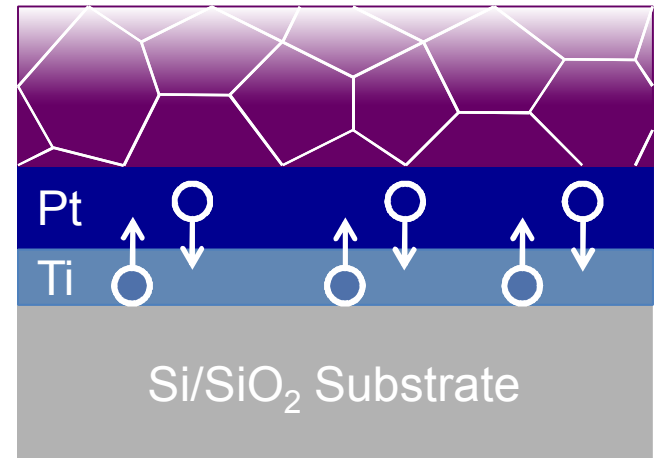
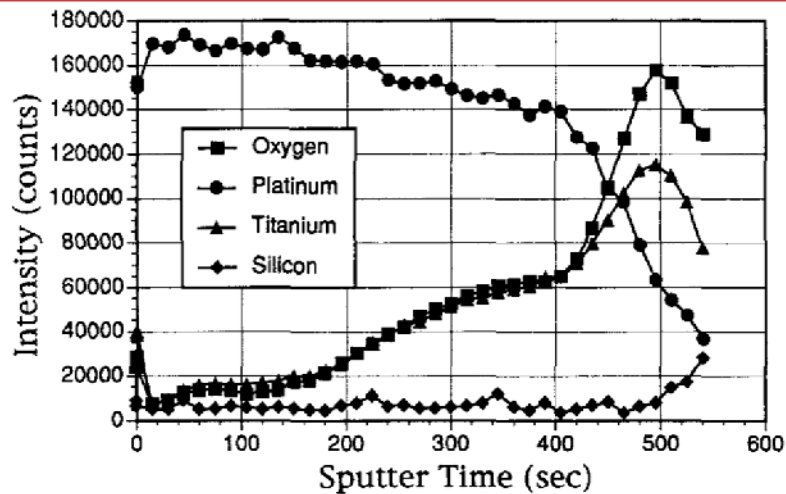
***Substantially improved chemical
homogeneity in films on Pt/ZnO substrates***

Chemical Homogeneity Effects on Electronic Properties

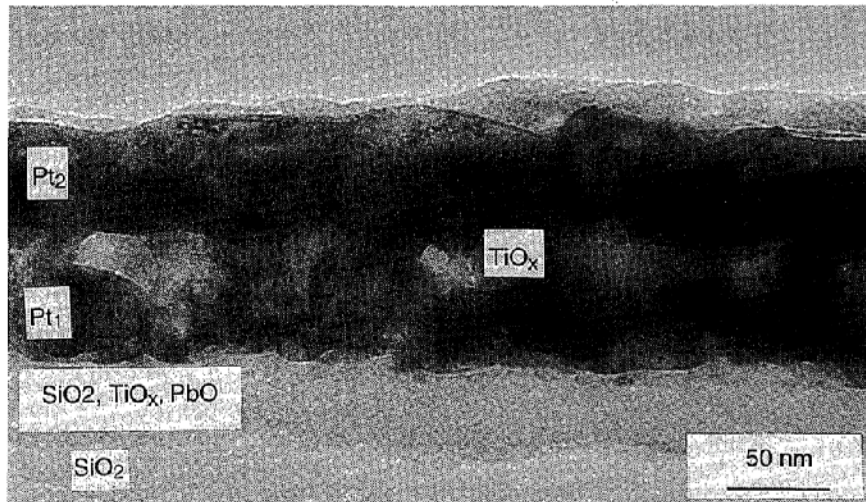


Increased dielectric and ferroelectric response directly correlates with improved homogeneity*

Revising prior art: Ti diffuses into Pt!

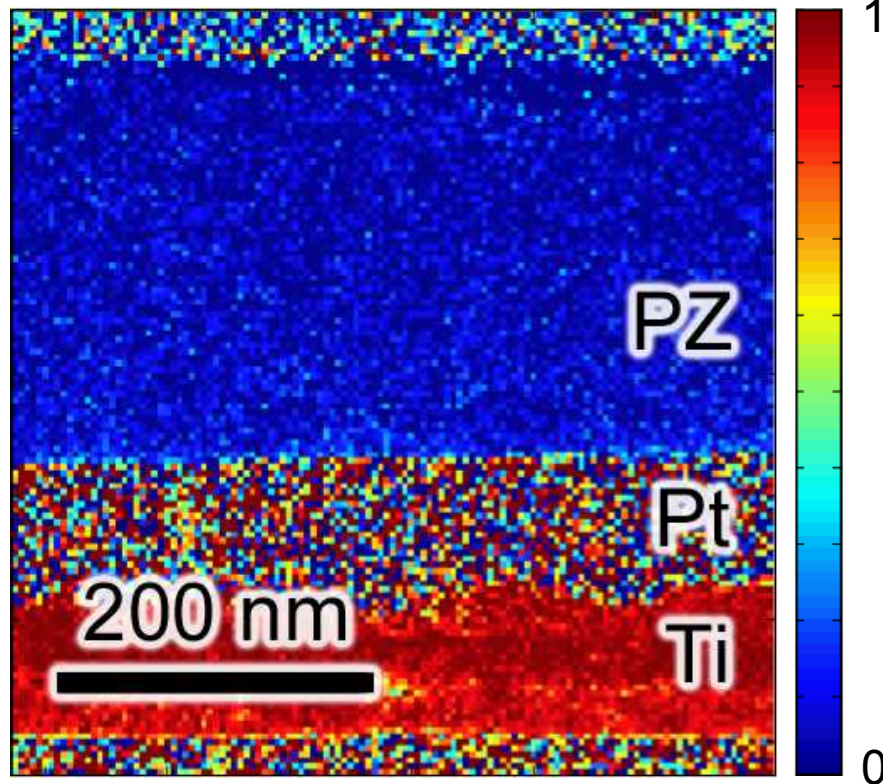


Ti diffusing through Pt can incorporate into the PZT Film

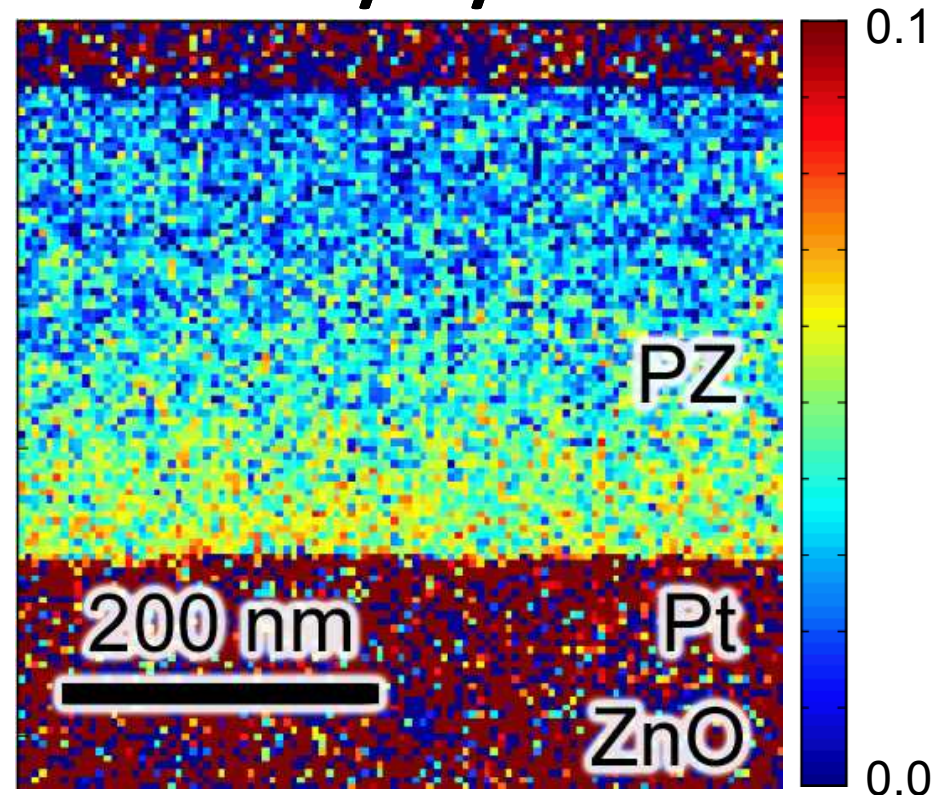


Test for Gradient Mechanism: PbZrO_3

PZ/Pt/Ti

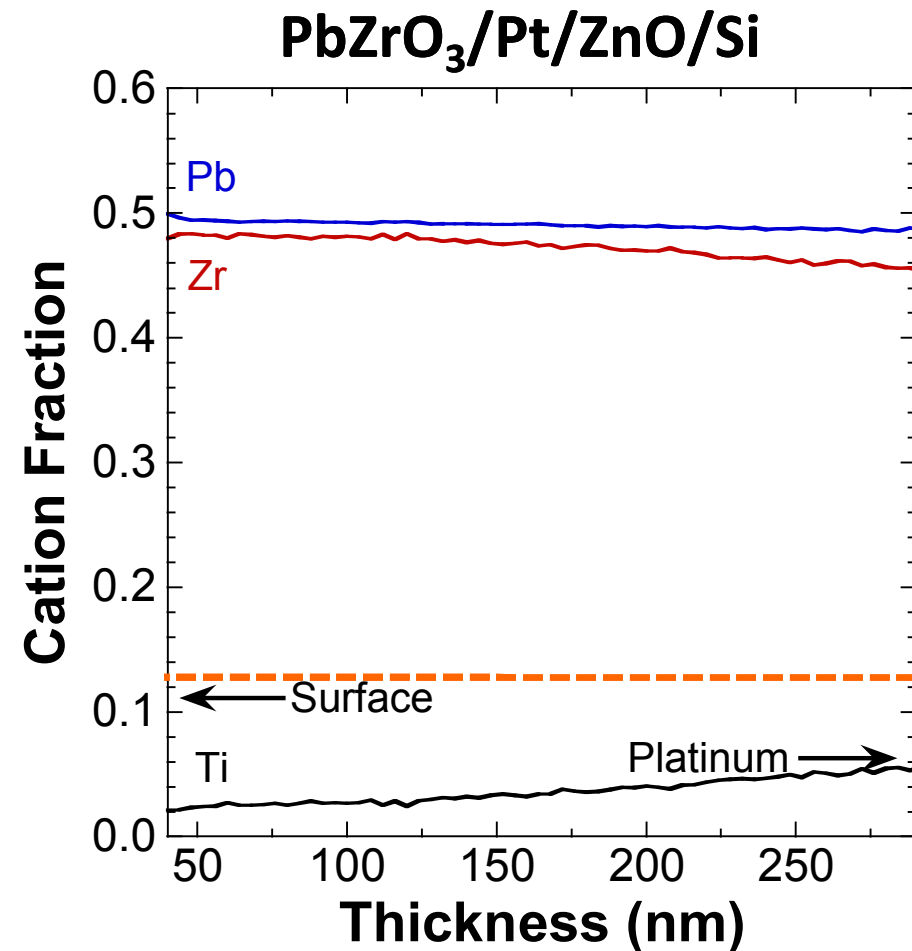
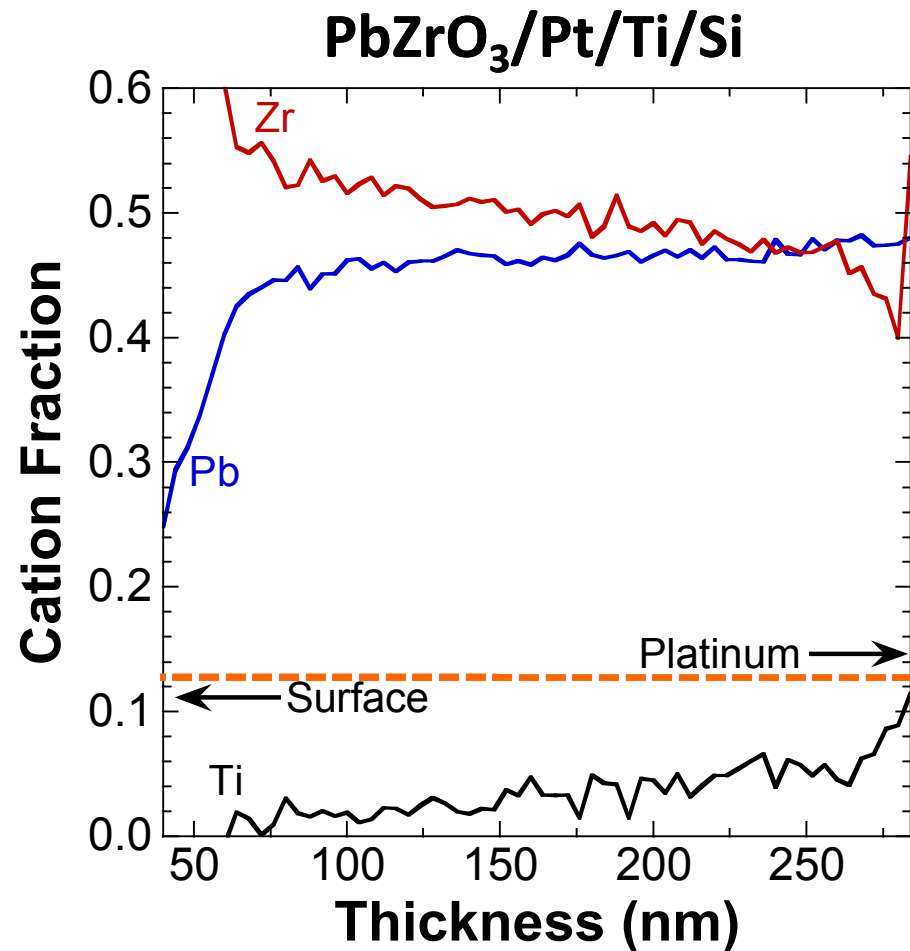


PZ/Pt/ZnO



Note Scale Difference

Mechanism: Ti Diffusion from Adhesion Layer: Test with PbZrO_3

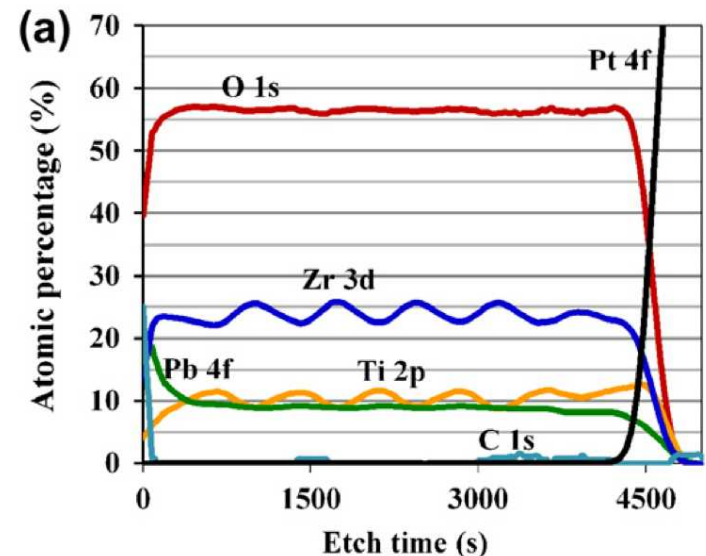
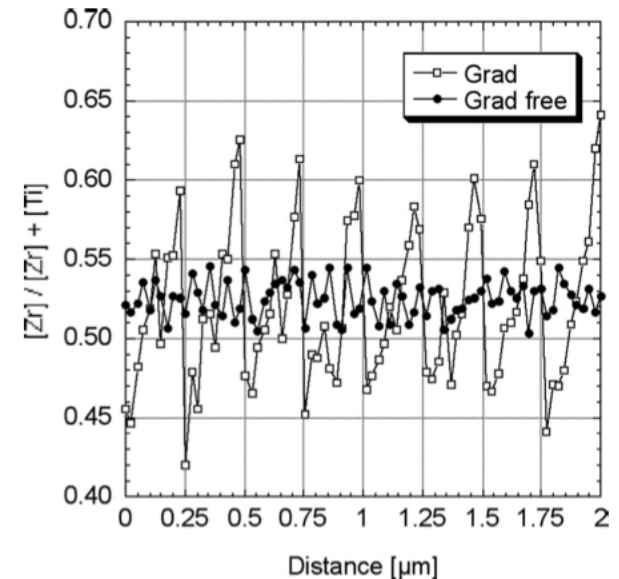


Less Ti and smaller gradient on Pt/ZnO

Solution Chemistry Effects?

Open Issues:

- This model cannot explain 'saw-tooth' profiles observed by other groups*
- What roles do solution chemistry and processing conditions play in gradient formation?
- More experiments necessary to isolate controlling variables
 - Substrate: Pt/Ti vs Pt/ZnO
 - Chemistry: IMO vs 2-MOE[‡]
 - Pyrolysis: 300, 350, 400°C
 - Heating Method/Rate: 15, 50, 200°C/s



*Amanuma, *et al.*, *Appl. Phys. Lett.* **65**, 3140 (1994)

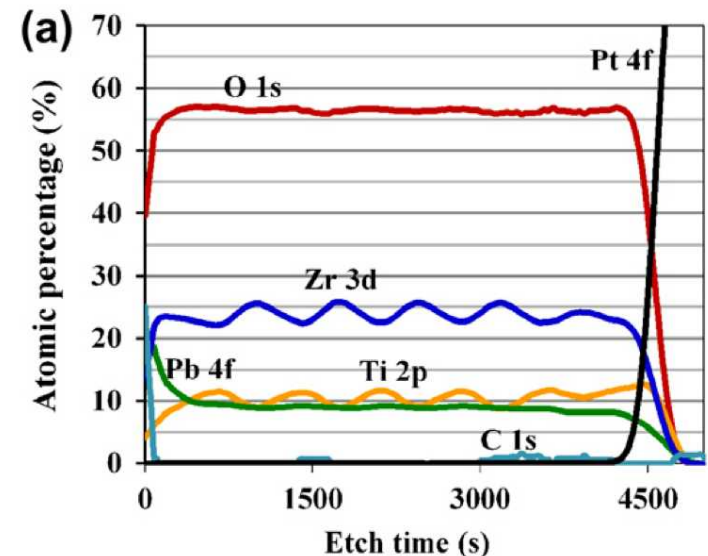
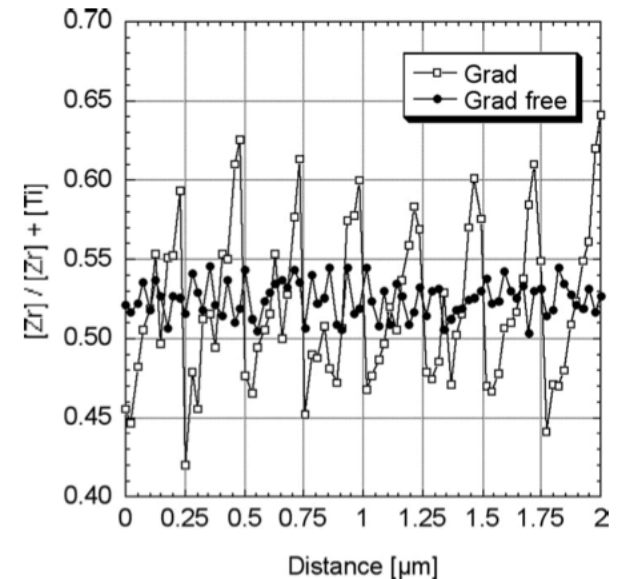
Calame and Muralt, *Appl. Phys. Lett.* **90**, 062907 (2007), Bastani and Bassiri-Gharb, *Acta. Mater.* **60**, 1346 (2012)

[‡]Based upon: C.D.E. Lakeman and D.A. Payne, *JACerS*, **75(11)**, 3091 (1992)

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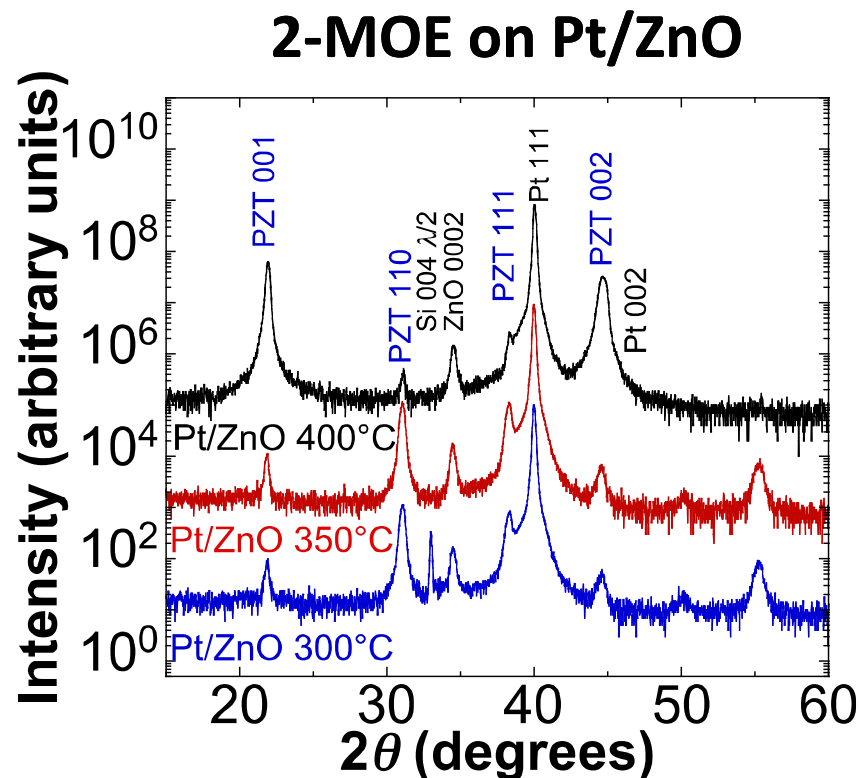
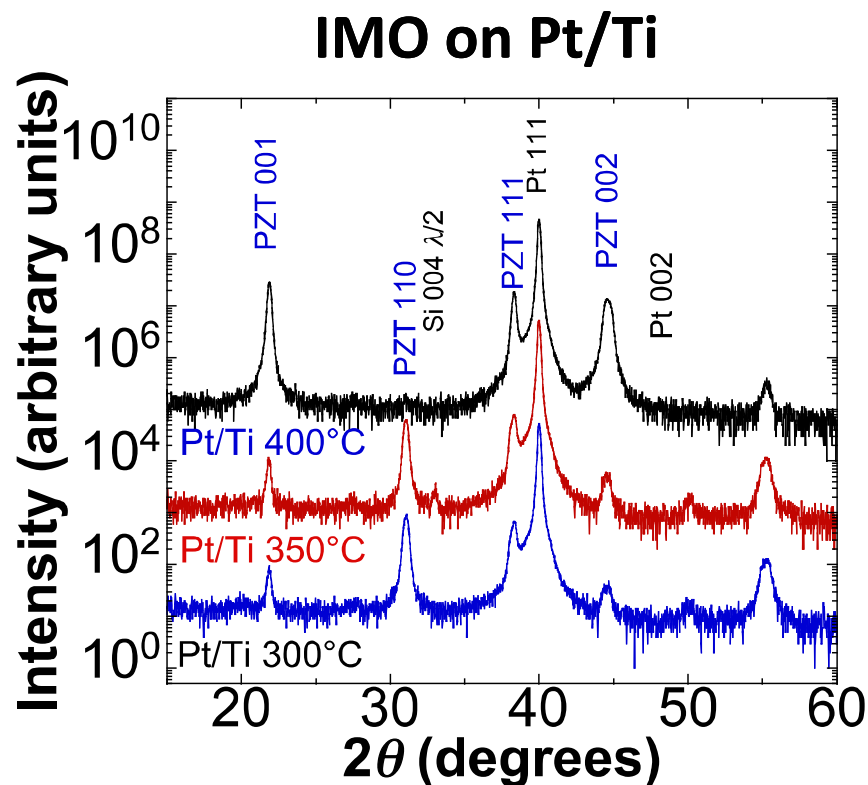
*Amanuma, *et al.*, *Appl. Phys. Lett.* **65**, 3140 (1994)

Calame and Muralt, *Appl. Phys. Lett.* **90**, 062907 (2007), Bastani and Bassiri-Gharb, *Acta. Mater.* **60**, 1346 (2012)

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J.F. Ihlefeld, US-Japan 2013, November 4, 2013

Pyrolysis Effects: Texture

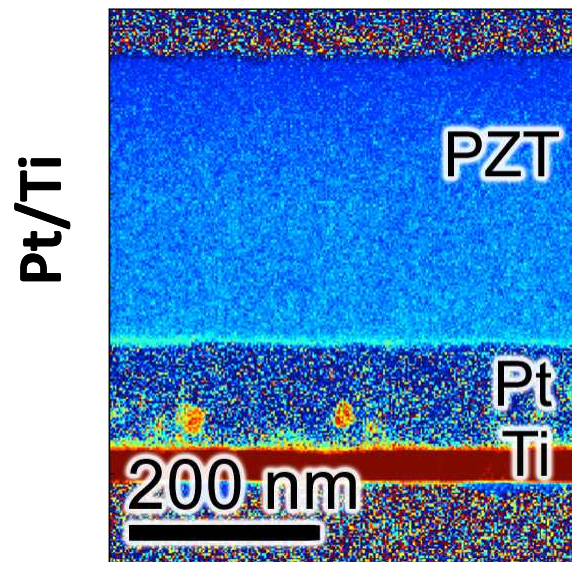


- Pyrolysis temperature affects texture
- No apparent difference with substrate or chemistry
- Consistent with previous work*

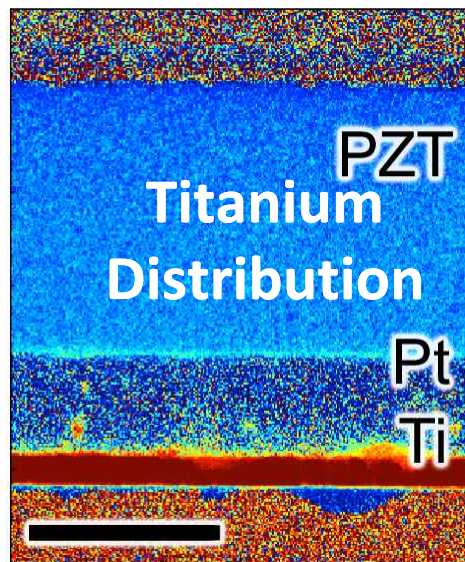
*Brennecka, et al., *Proc. 14th US-Japan*, 136 (2009) *Chen and Chen, *JACerS*, **81** (1) 97-105 (1998)

Solution Chemistry Effect: IMO films

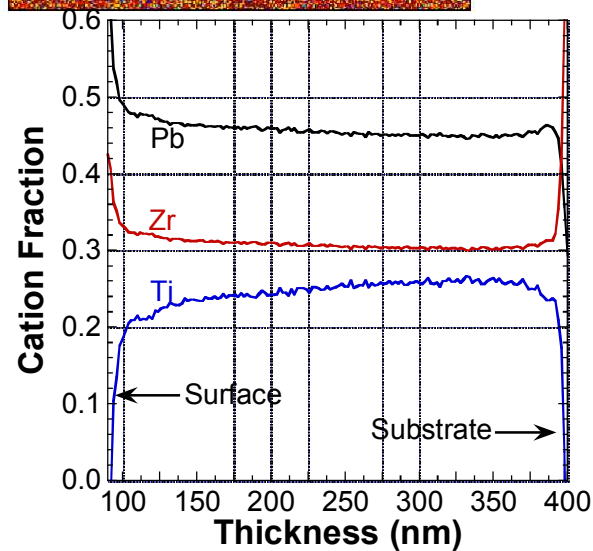
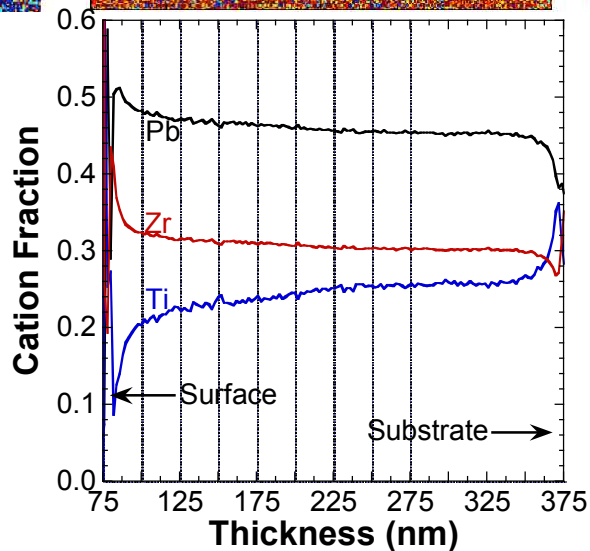
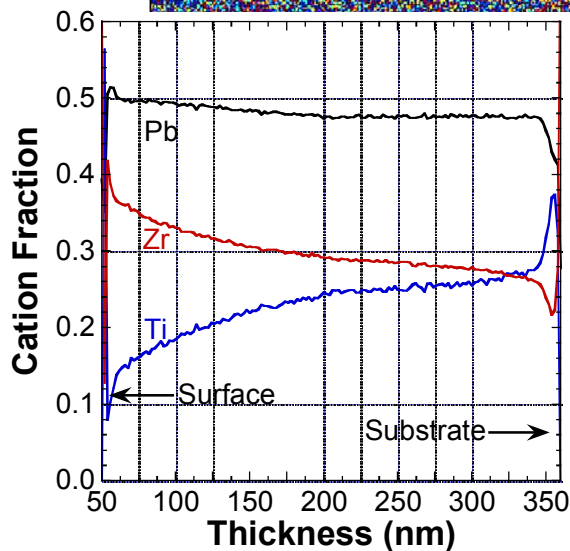
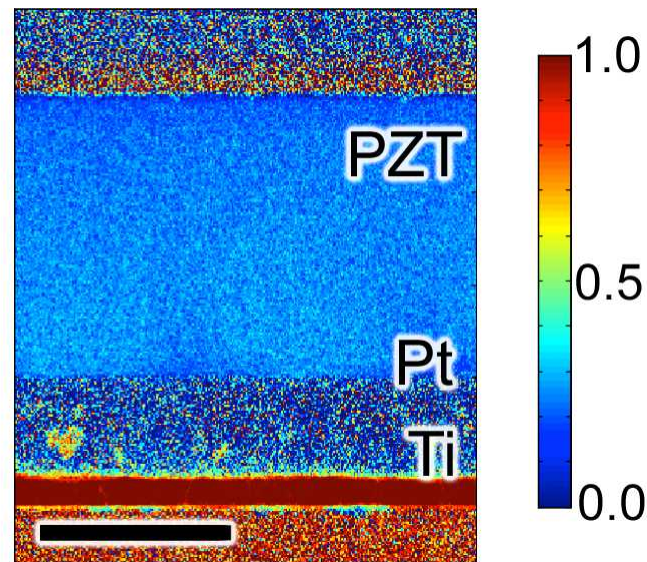
300°C



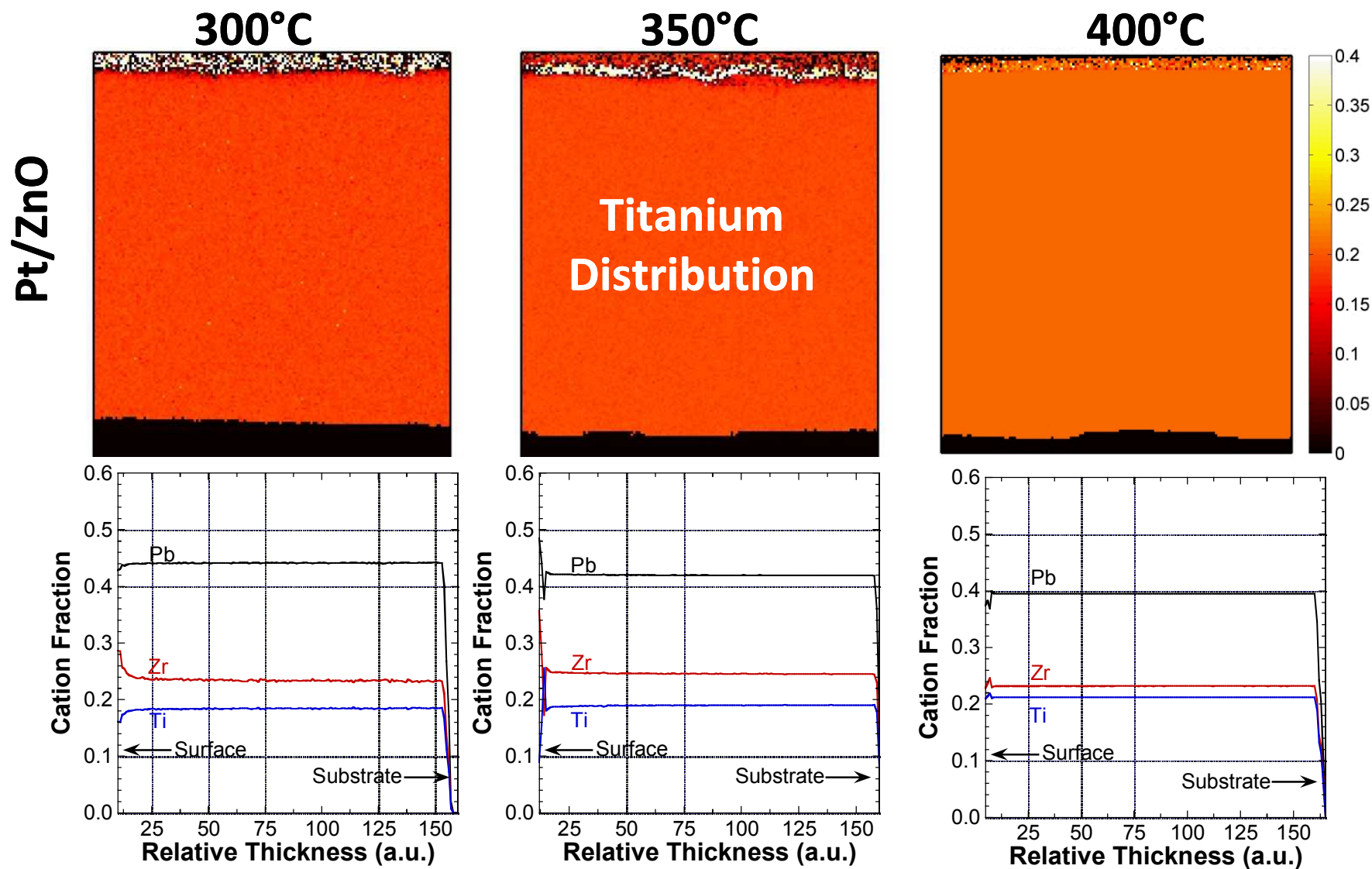
350°C



400°C

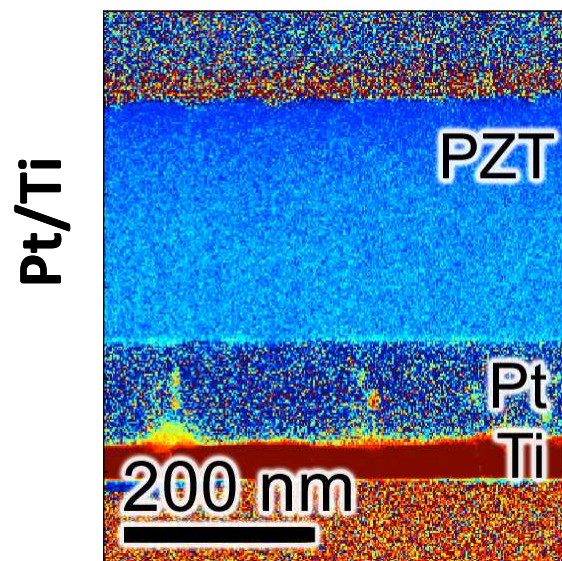


Solution Chemistry Effect: IMO Films

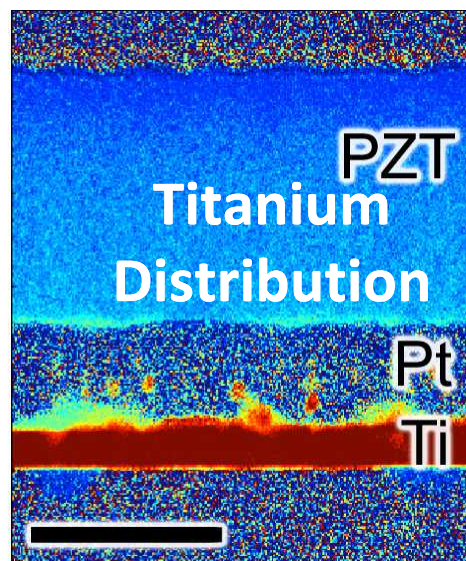


Solution Chemistry Effect: 2-MOE films

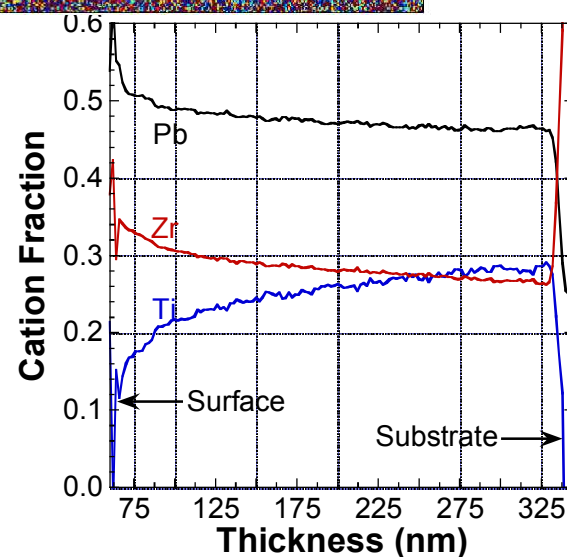
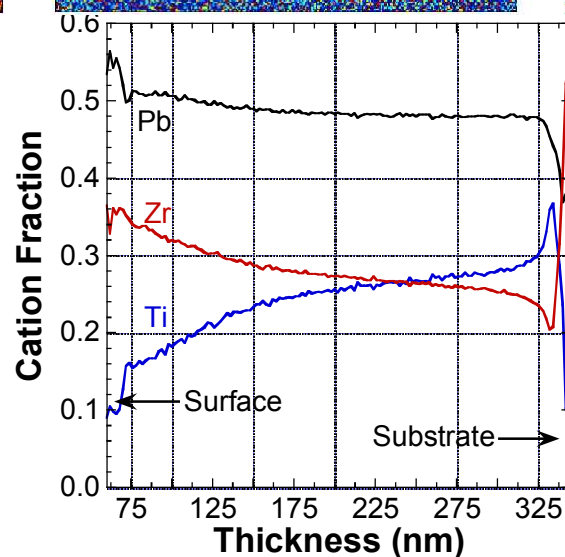
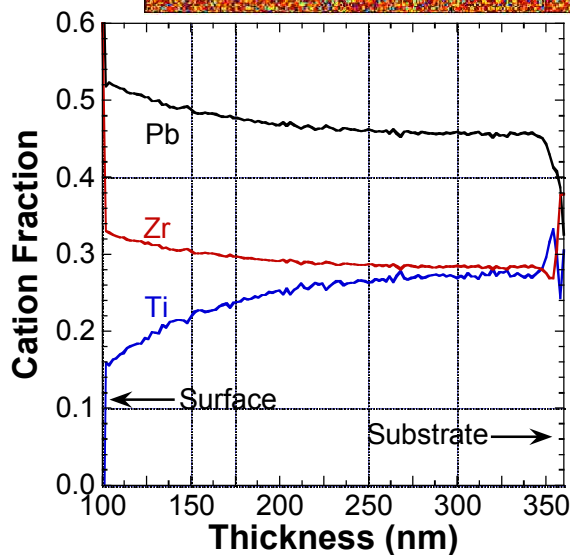
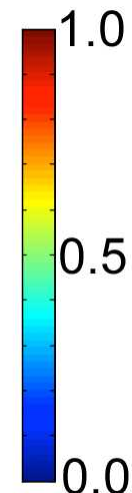
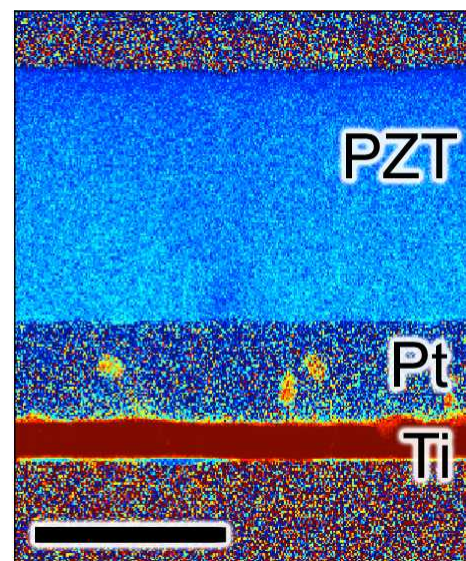
300°C



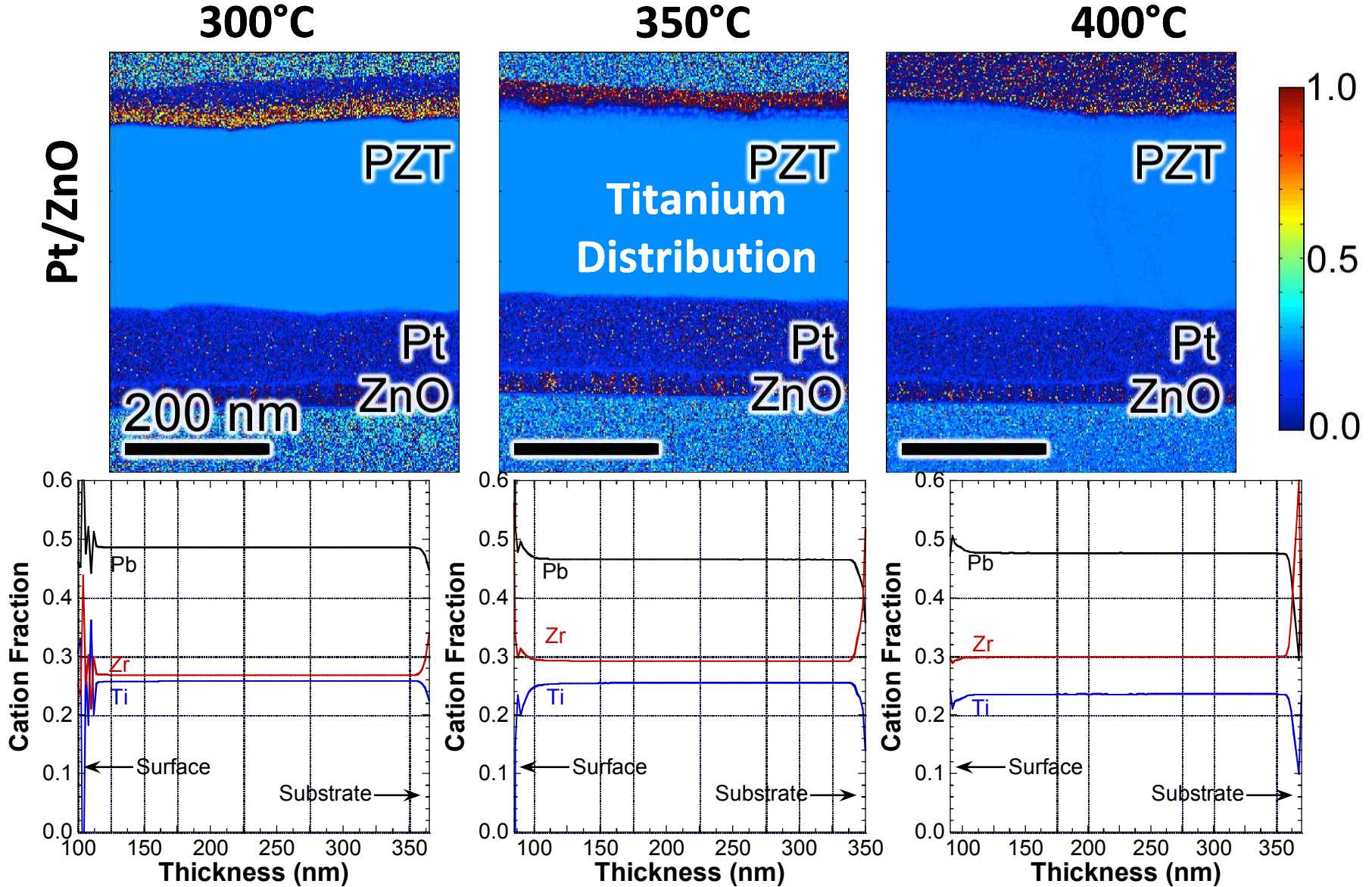
350°C



400°C



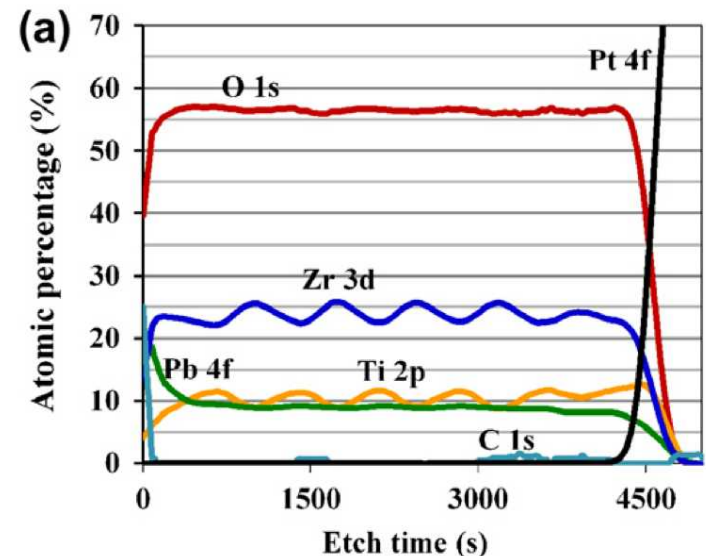
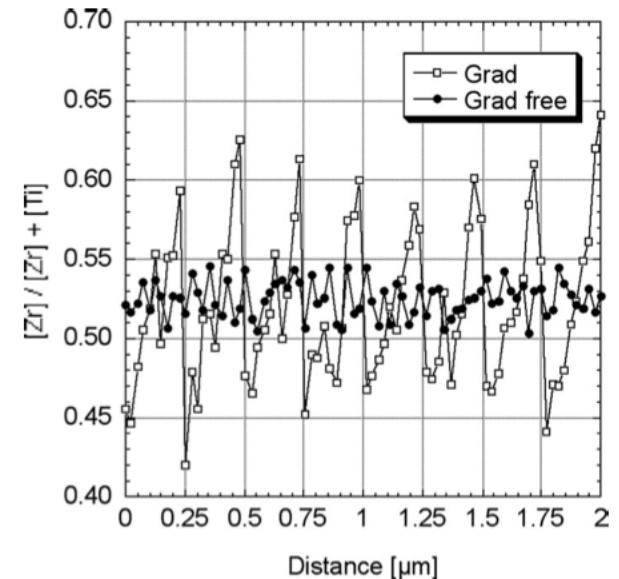
Solution Chemistry Effect: 2-MOE films



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 - ~~Chemistry: IMO vs 2-MOE[‡]~~
 - ~~Pyrolysis: 300, 350, 400°C~~
 - **Heating Method/Rate: 15, 50, 200°C/s via RTA**



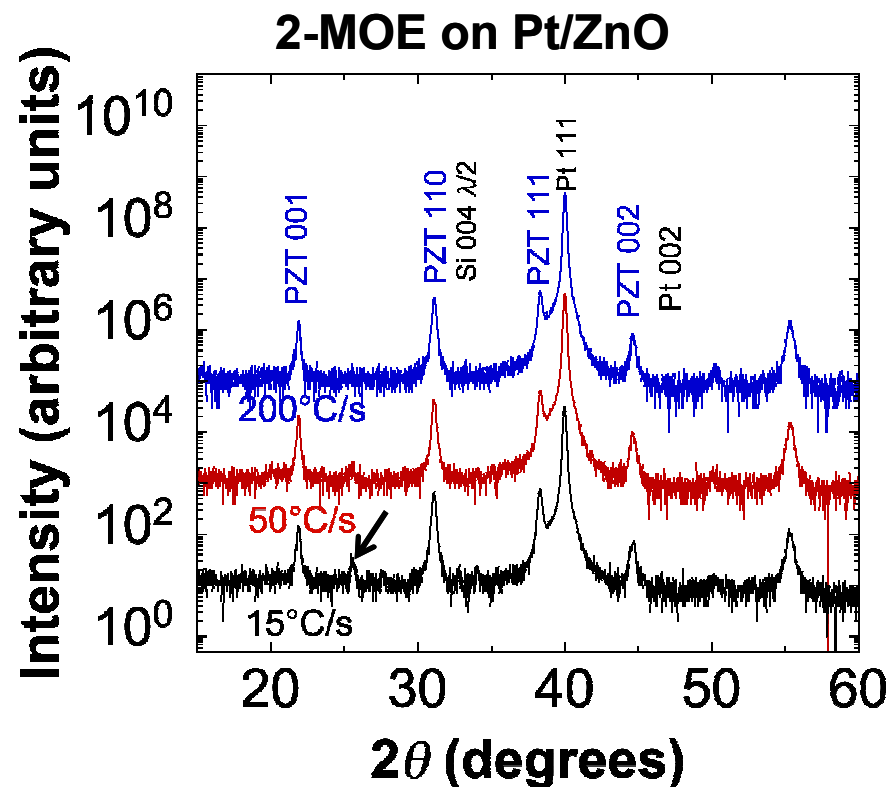
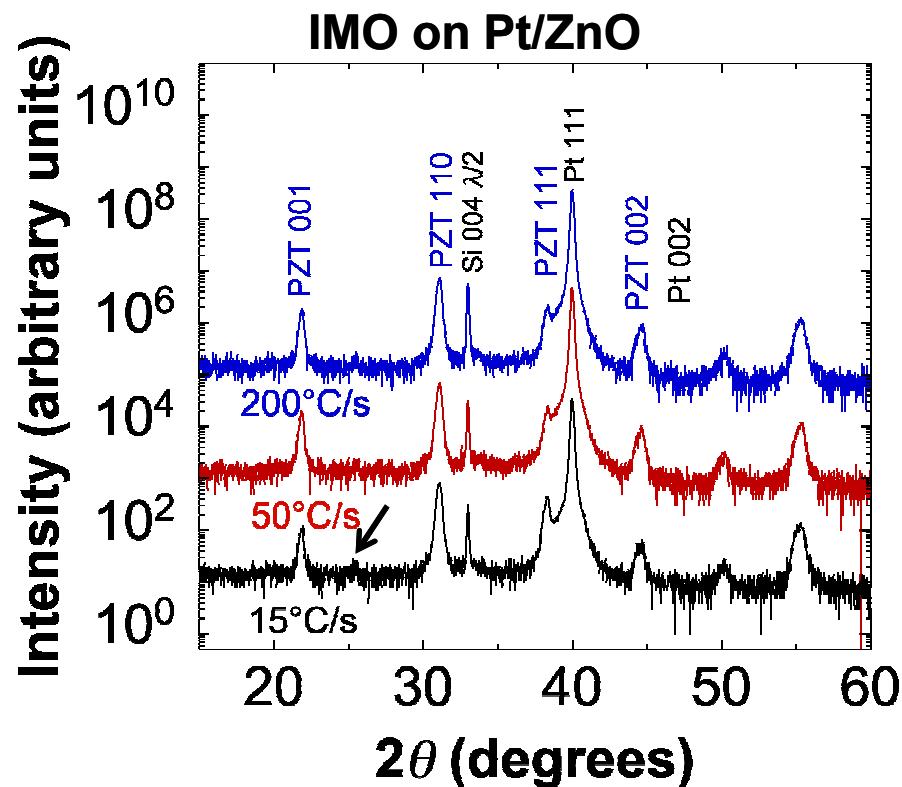
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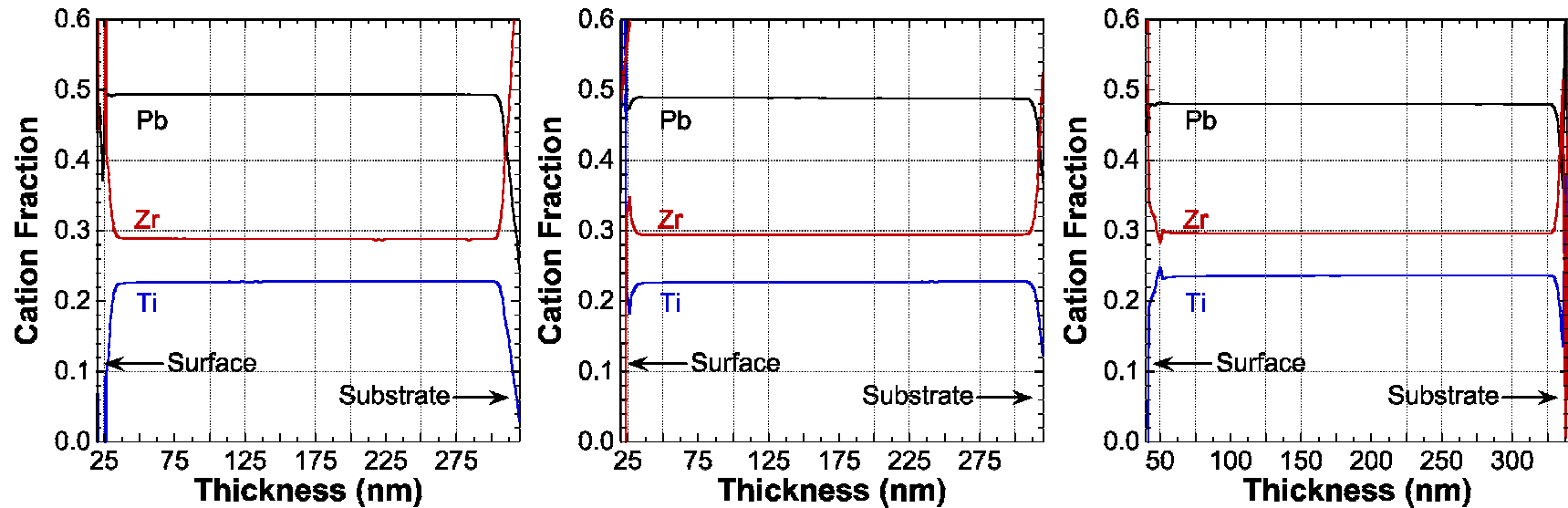
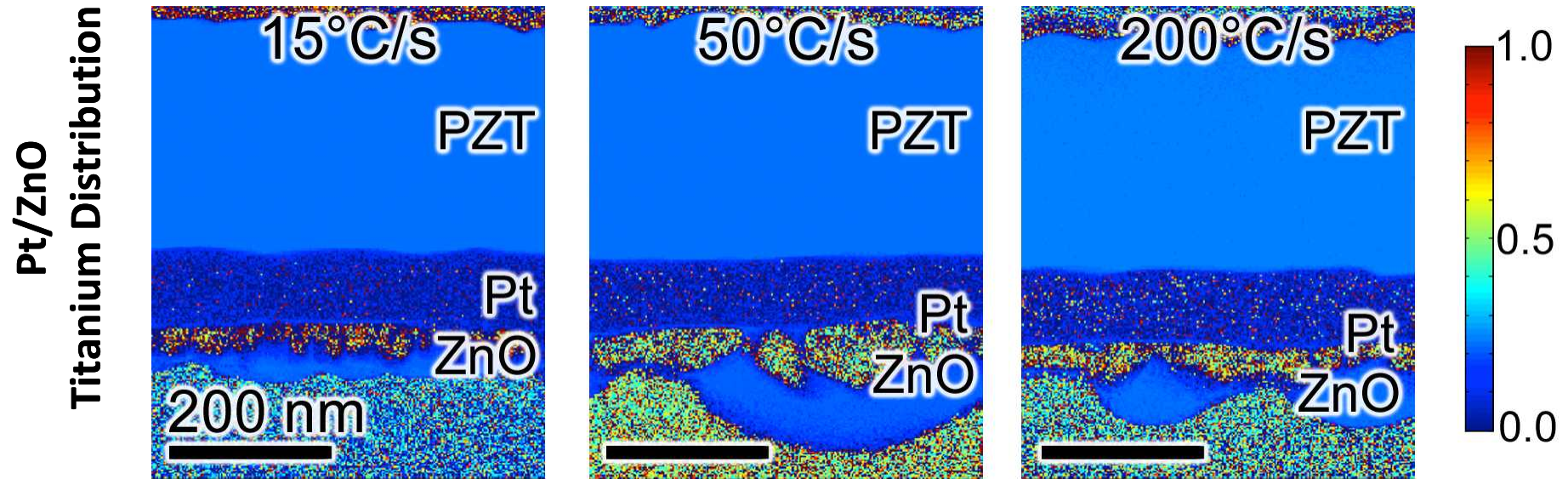
J.F. Ihlefeld, US-Japan 2013, November 4, 2013

RTA Heating Rate Effects: Texture



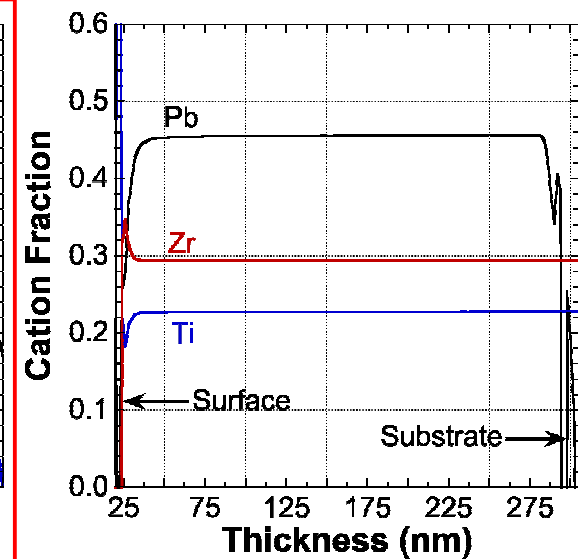
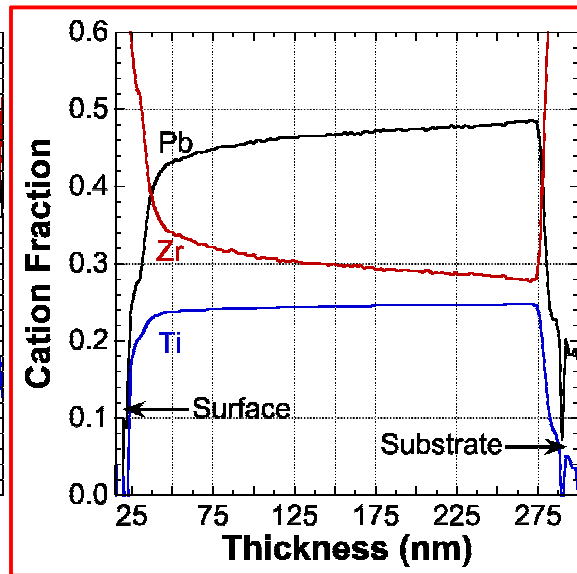
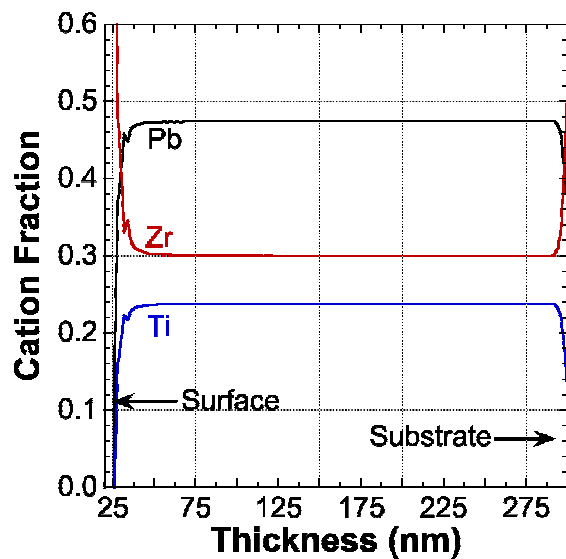
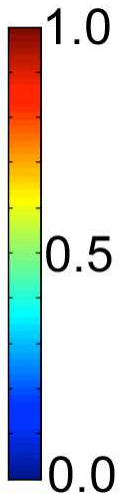
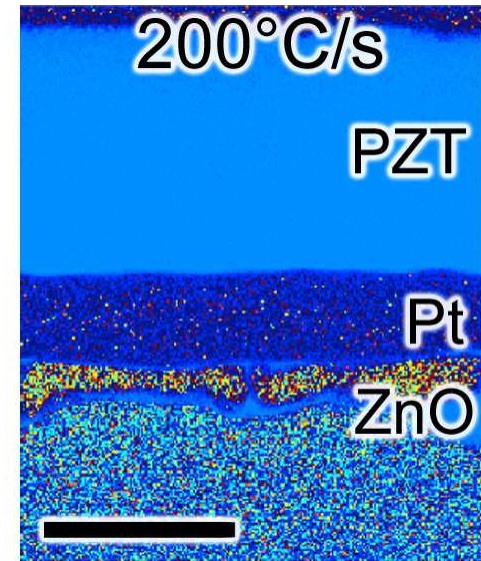
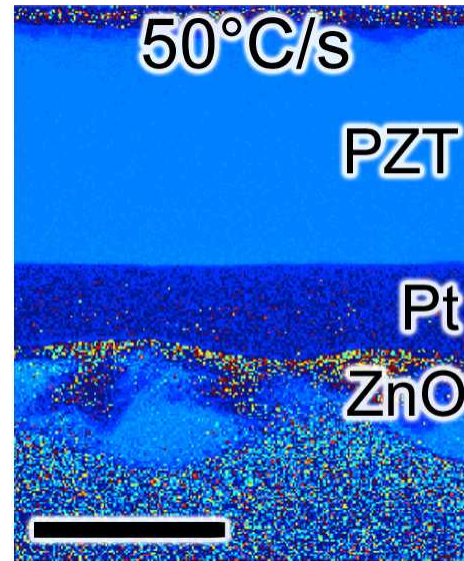
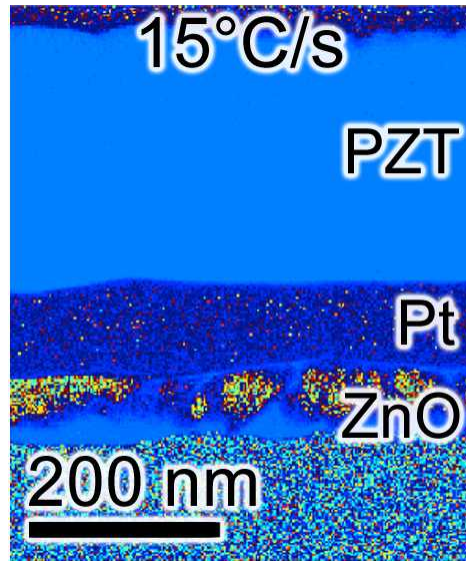
- No apparent heating rate dependence to texture
 - Consistent with previous studies*
- ZnO diffraction peaks no longer visible
- Additional peaks present in low ramp rate samples

Ramp Rate Effect: IMO Films

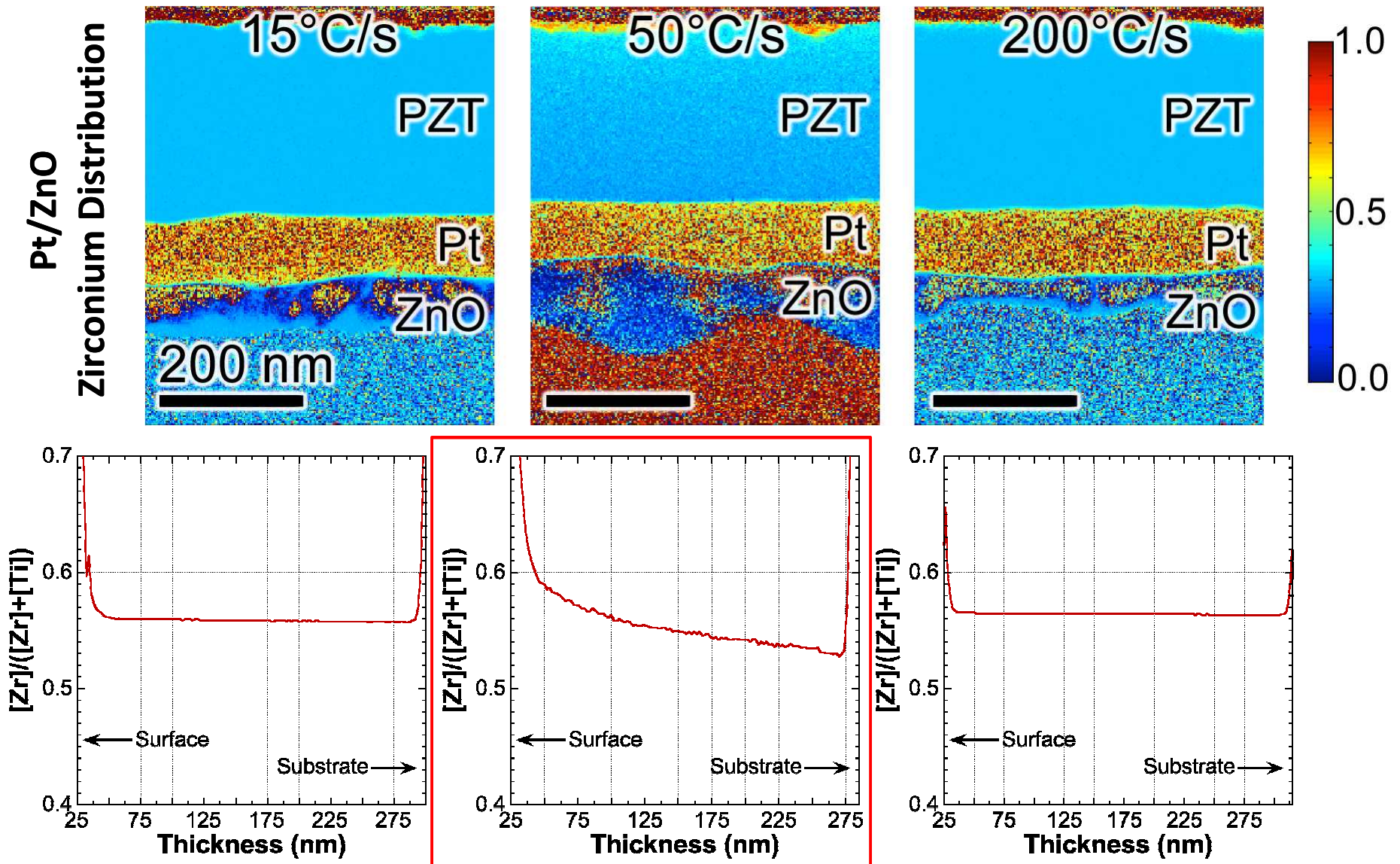


Ramp Rate Effect: 2-MOE Films

Pt/ZnO
Titanium Distribution

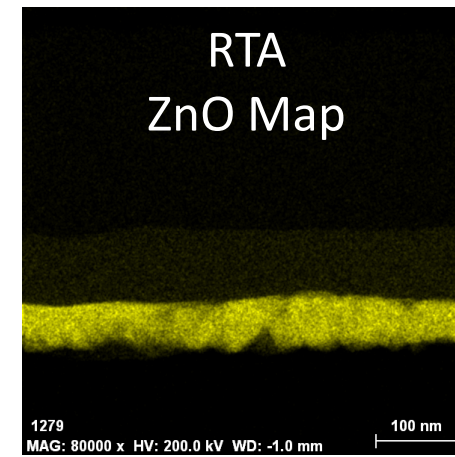
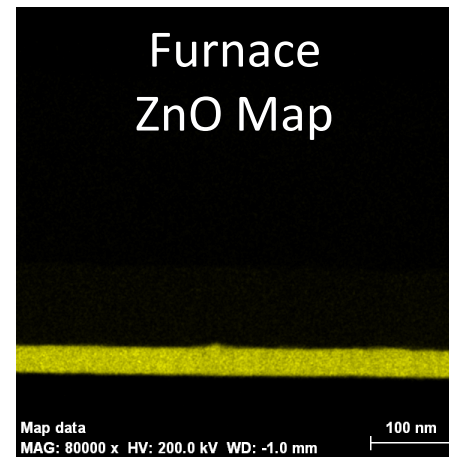
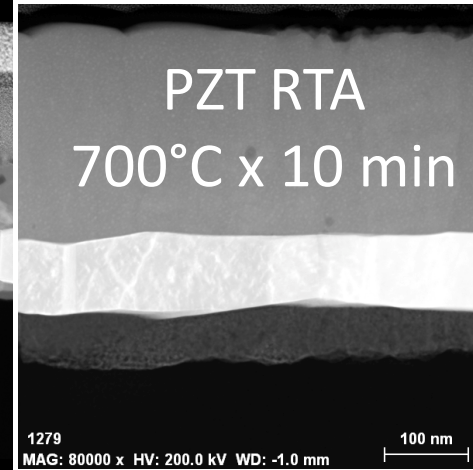
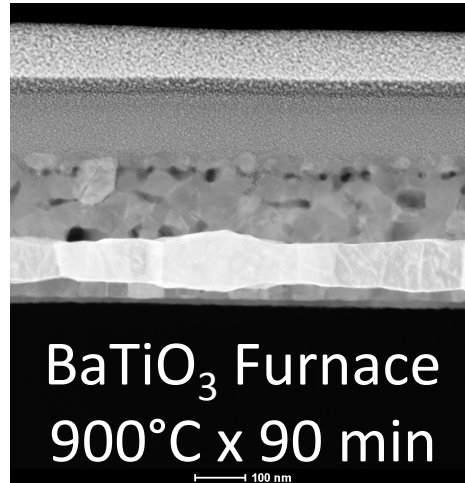
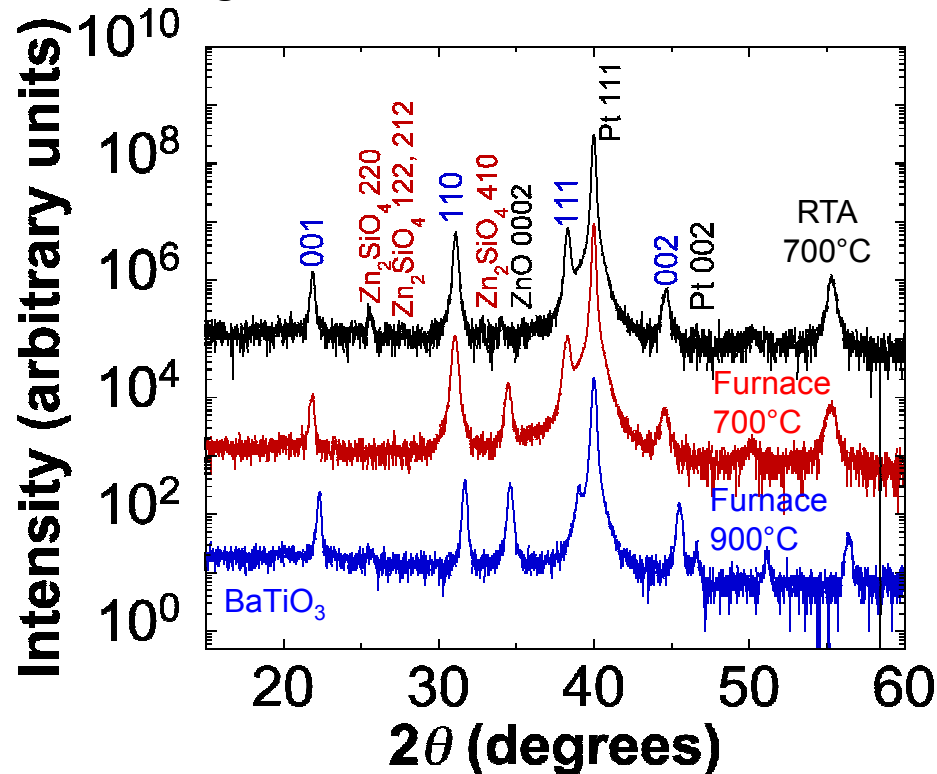


Ramp Rate Effect: 2-MOE Films



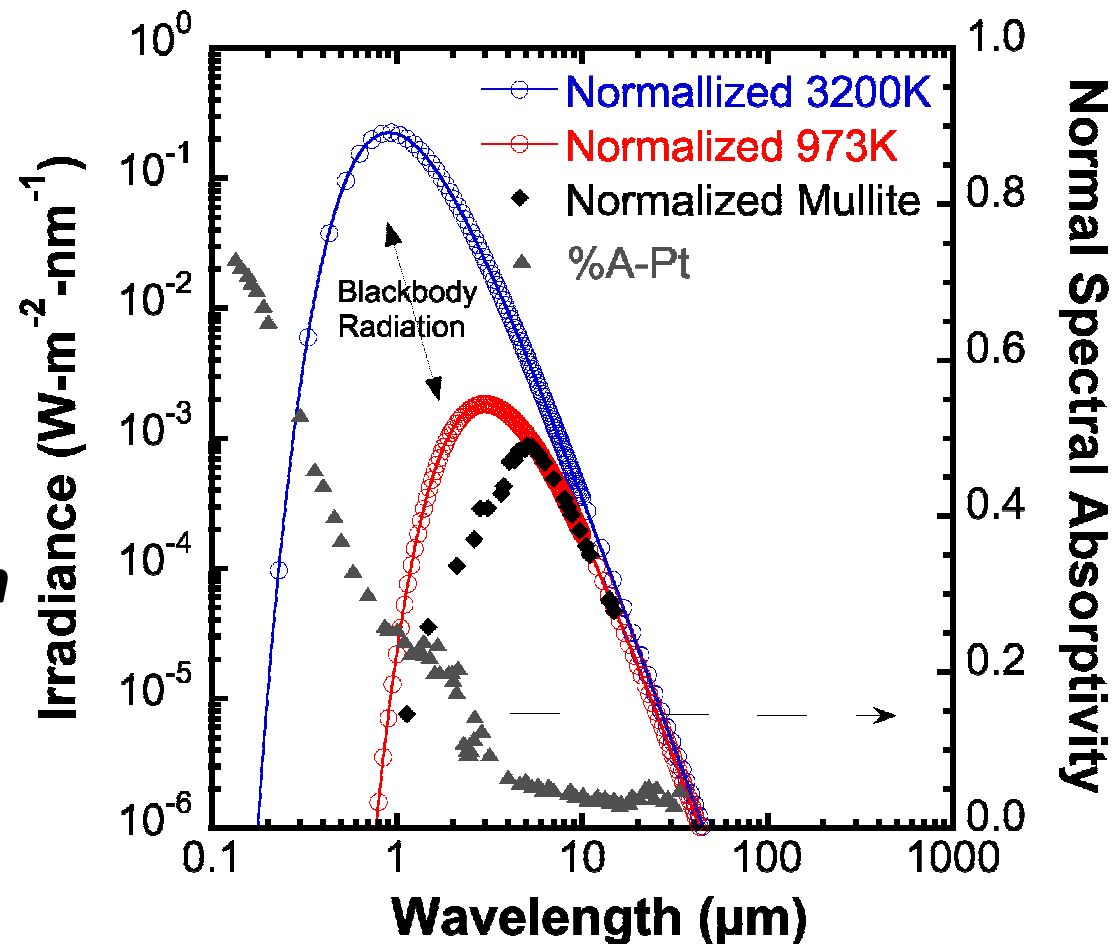
Interesting Observation: ZnO-SiO_2 Reaction

- Observe formation of secondary Zn_2SiO_4 phase in RTA samples
- Only previously observed this in films subjected to significantly higher thermal budgets



Interesting Observation: ZnO-SiO₂ Reaction

- Heating mechanism differs
 - RTA=Optical absorption
 - Furnace=optical+conductive
- ***Pt electrode may get substantially hotter than RTA thermocouple***
- ***Locally heating Pt likely affects film crystallization processes***

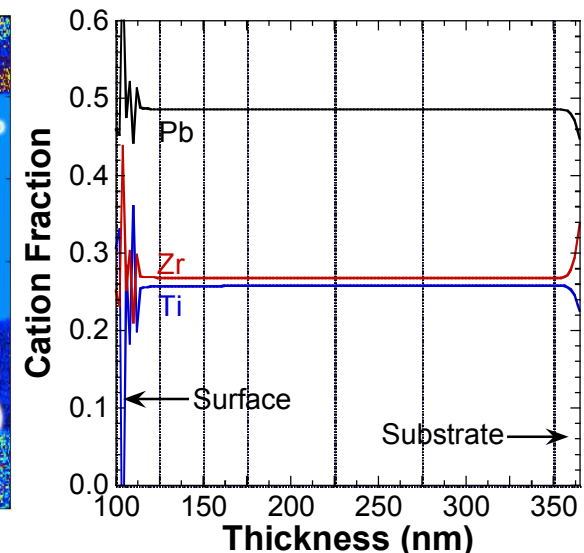
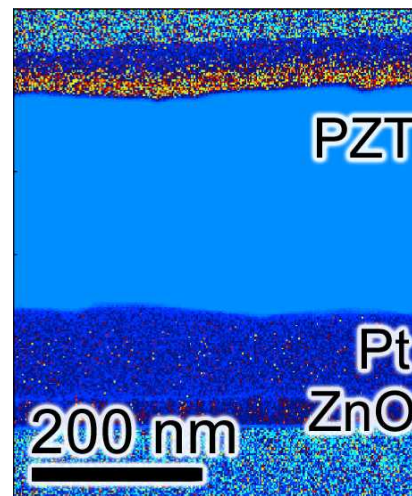
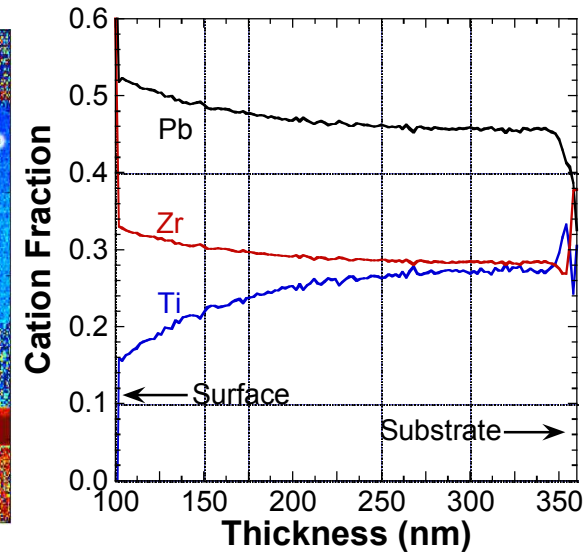
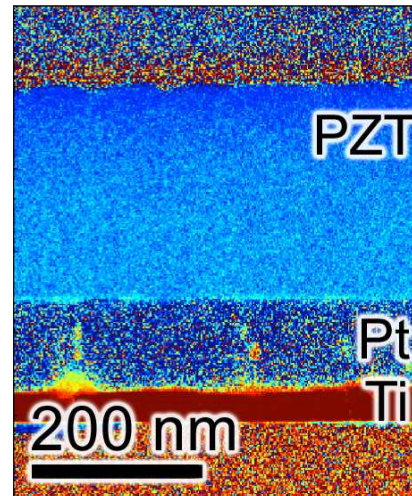


*Thermo-optical data calculated from: "Thermal Radiative Properties: Metallic Elements and Alloys," and "Thermal Radiative Properties of Nonmetallic Solids" by Y.S. Touloukian and D.P. Dewitt (1972)

Processing of PZT Film: Lessons Learned To Date

Open Issues:

- Pyrolysis temperature does not significantly effect gradient formation
- On substrates devoid of a titanium adhesion layer, gradient formation appears to be hindered – even for 2-MOE sol-gel chemistries
- Heating rate and method does appear to matter*
 - Slight gradient observed in RTA samples with 50°C/s ramp rate



Summary

- IMO PZT films prepared on Pt/ZnO substrates possess improved electronic properties compared to traditional substrates
- ZnO as an adhesion layer enables chemically homogeneous films for most processing conditions
- Titanium/TiO_x may be poor choices for Pt adhesion layers for the high performance device development
- *If a chemically homogeneous film is required, then furnace anneals are preferable*

