

Mechanical properties of muscovite as a function of pressure and temperature

Clay Mineral Society 2011

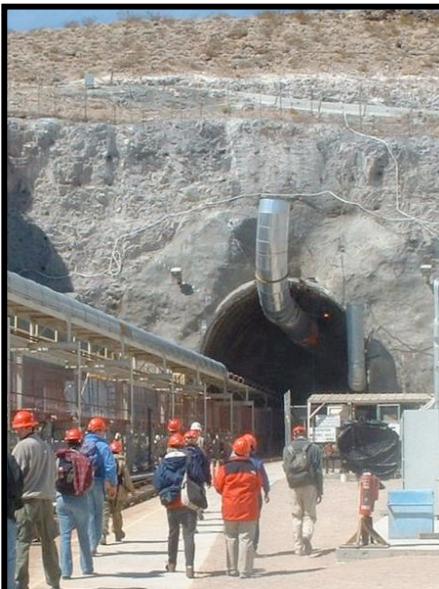
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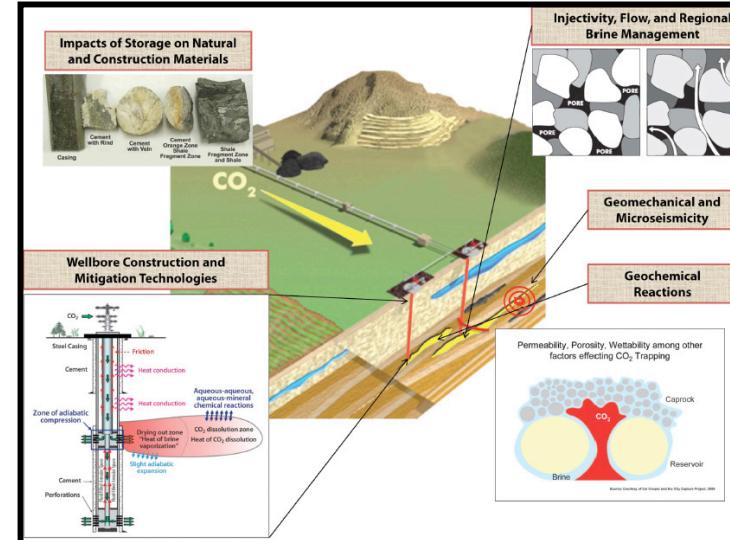
Areas of application of mechanical properties

Waste repositories



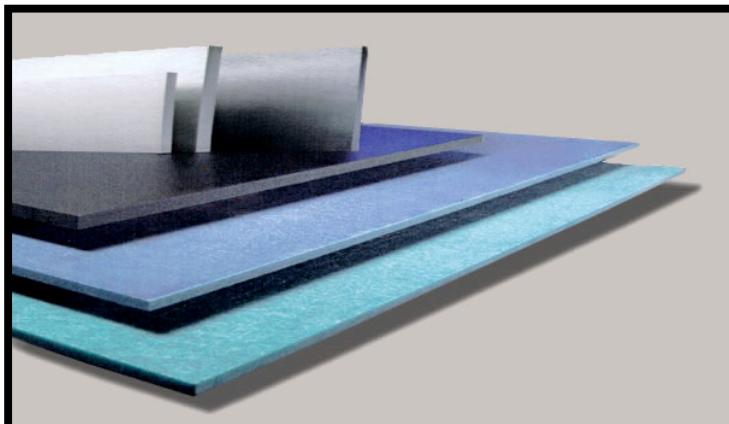
http://en.wikipedia.org/wiki/Yucca_Mountain_nuclear_waste_repository

Carbon sequestration



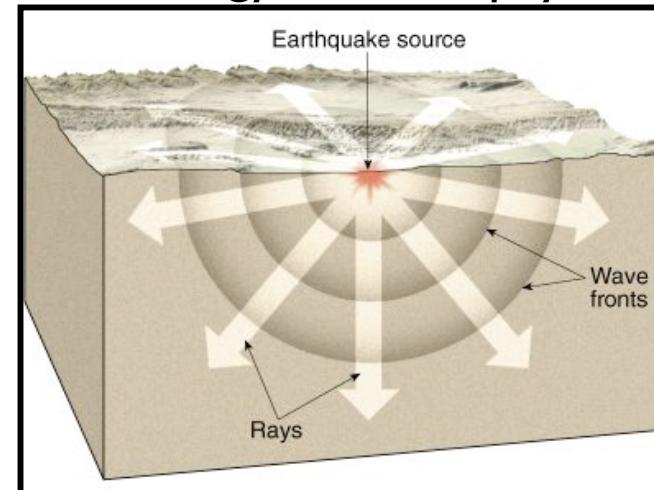
http://www.netl.doe.gov/technologies/carbon_seq/corerd/storage.html#geologic

Composite material properties



<http://www.pentagon-ems.com/products/material/esd-composite-material.html>

Seismology/ Mineral physics



http://geophysics.ou.edu/solid_earth/notes/seismology/seismo_interior/seismo_interior.html

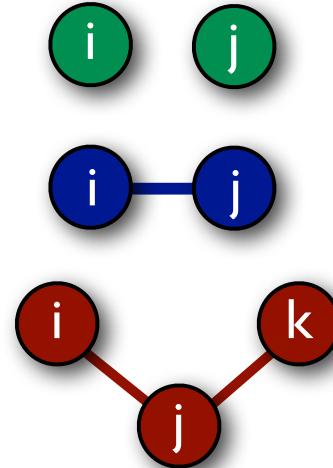
How to model muscovite in a simulation

Accurate description of how atoms interact

Clayff: Flexible model for clays, oxides, aqueous ions, brines

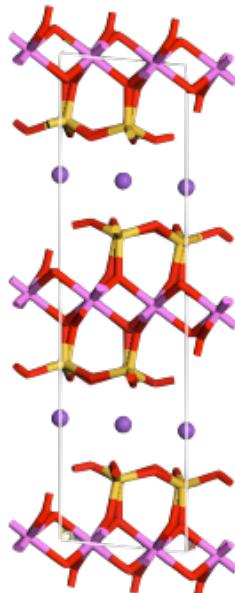
$$U_{ij} = \sum_i \sum_j \left(\frac{A_{ij}}{r_{ij}^{12}} + \frac{B_{ij}}{r_{ij}^6} + \frac{q_i q_j}{\epsilon r_{ij}} \right) + \sum_{ij} \frac{1}{2} k_b (r_{ij} - r_o)^2 + \sum_{ijk} \frac{1}{2} k_\theta (\theta_{ijk} - \theta_o)^2$$

Short-range terms Electrostatic terms Bonded terms Angle terms



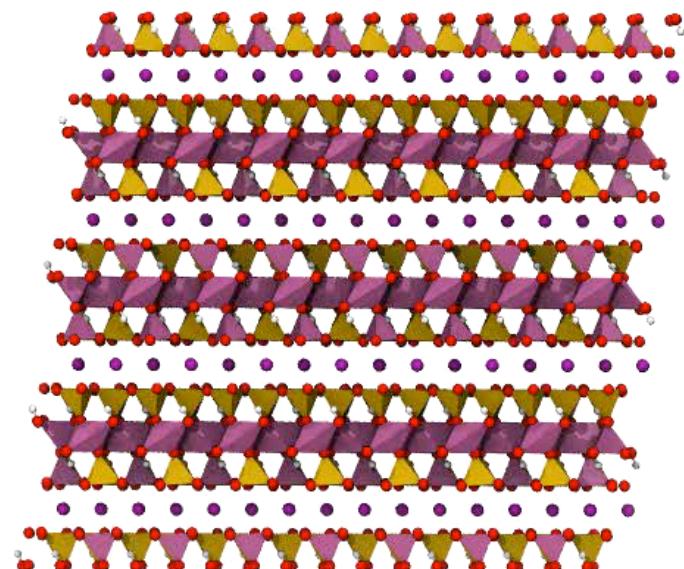
Cygan, Liang, and Kalinichev (2004) *Journal of Physical Chemistry B*

Accurate chemical and structural description



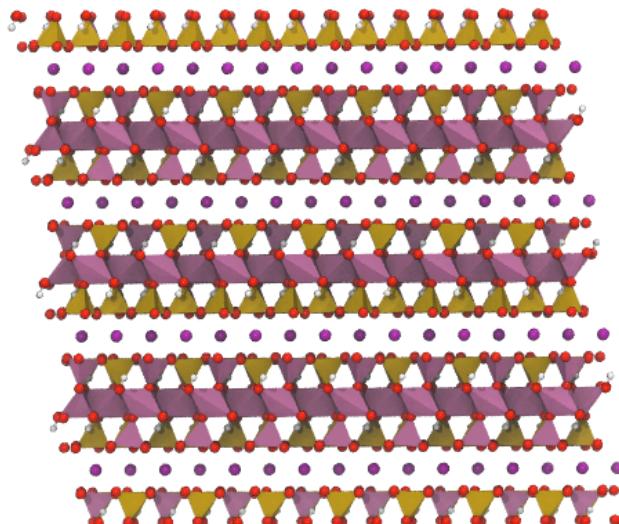
Muscovite unit cell
 $KAl_2(Si_3Al)O_{10}(OH)_2$

Monoclinic C2/c 2M1 unit cell
Supercell $8 \times 4 \times 2$ (5376 atoms)
 $4.1 \text{ nm} \times 3.6 \text{ nm} \times 4.0 \text{ nm}$

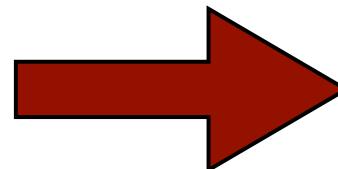


Simulation methods to determine mechanical properties

Molecular dynamics
Sandia's LAMMPS code

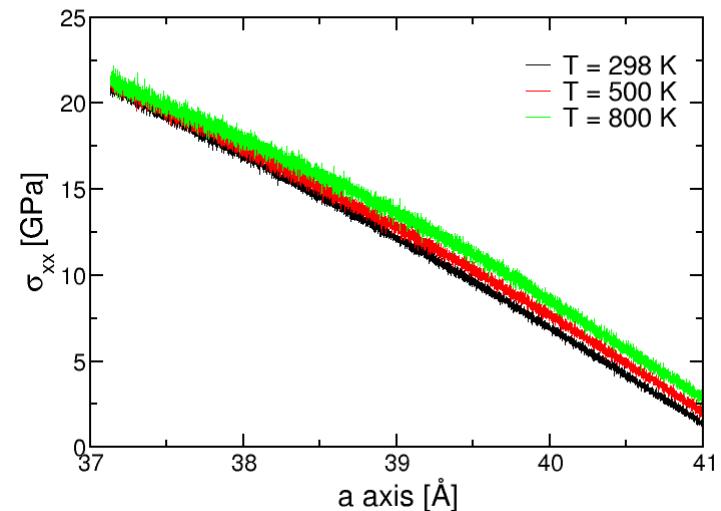


Simulation evolves

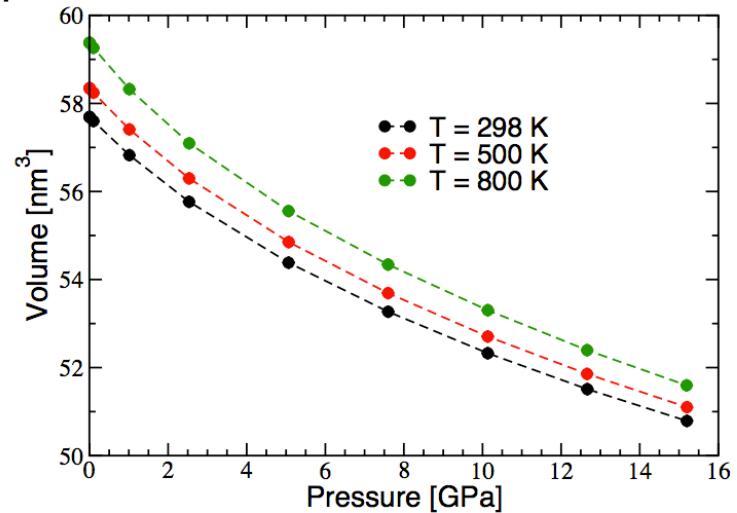


$T = 298, 500, 800 \text{ K}$

Non-equilibrium NPT: $\varepsilon = -10 \text{ to } 10\%$

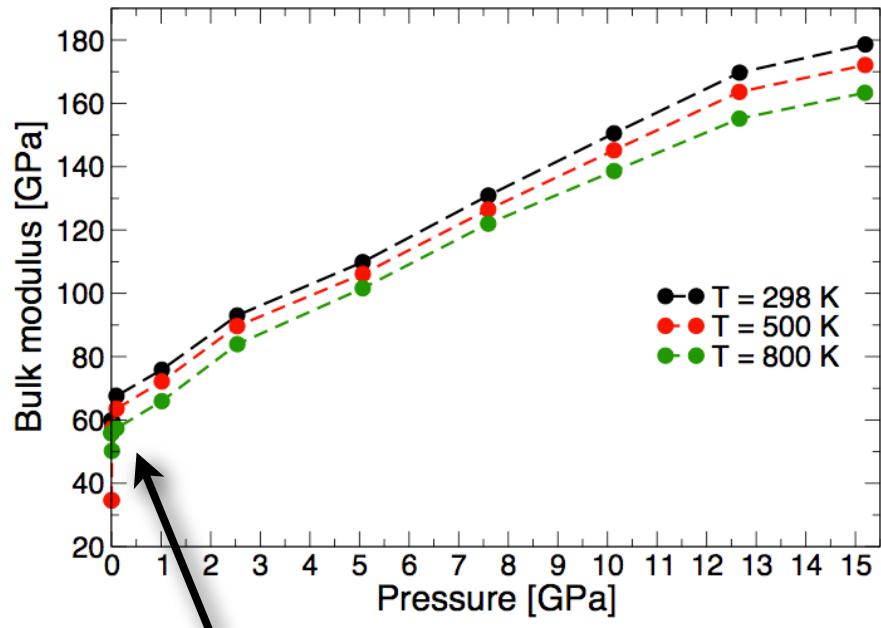


Equilibrium NPT: $P = 1\text{E}-3 \text{ to } 15 \text{ GPa}$

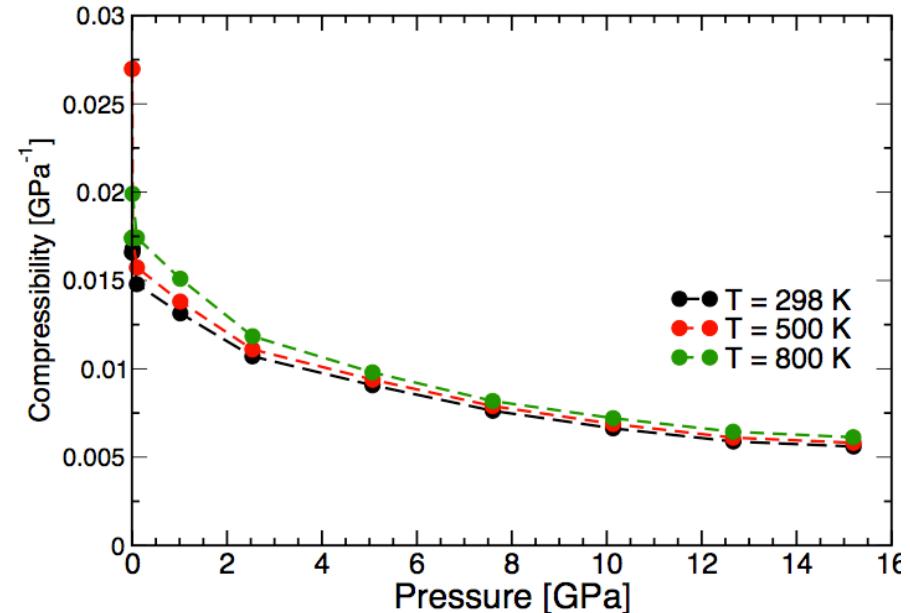


Bulk Modulus and Compressibility

$$\text{Bulk modulus} = -V(\partial P/\partial V)_T$$



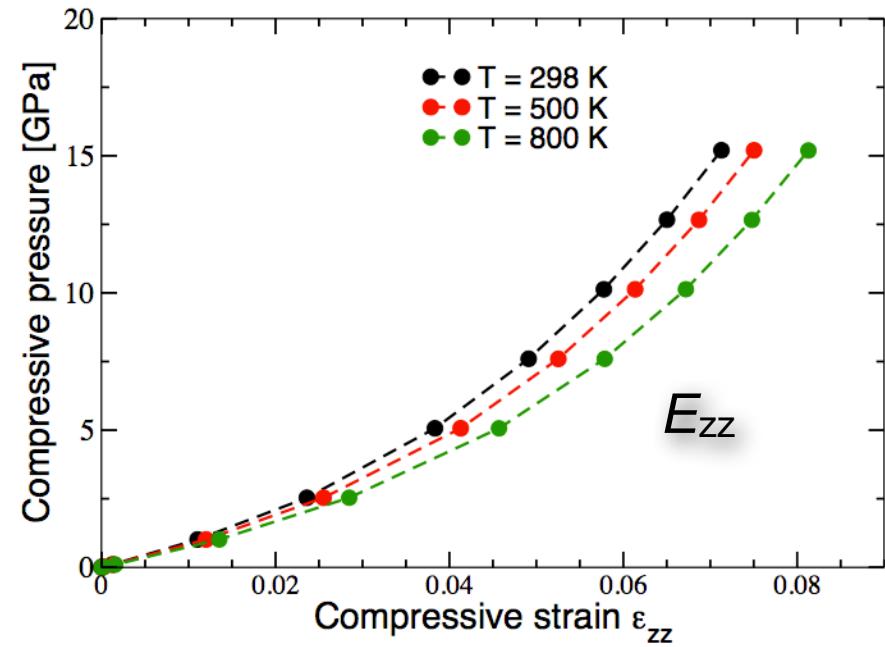
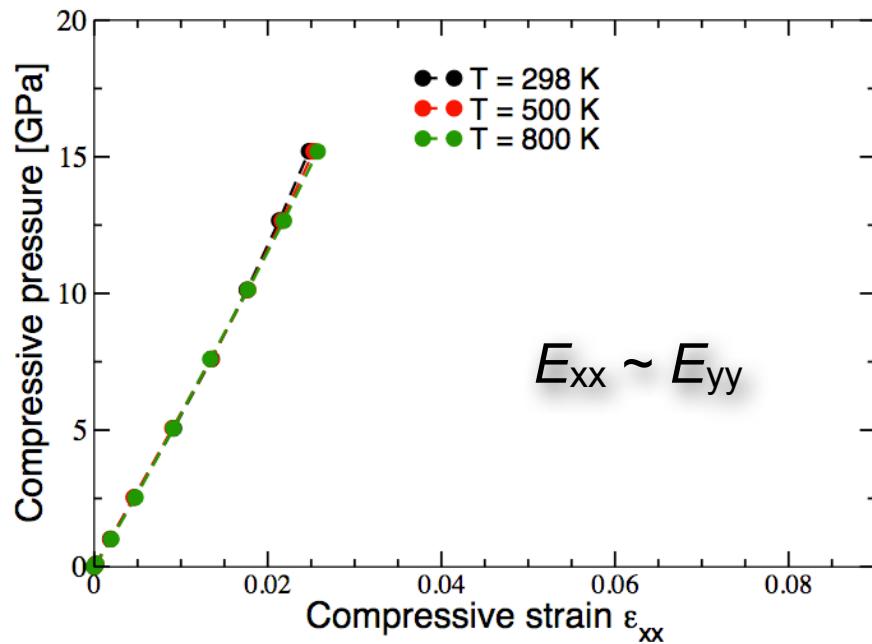
$$\text{Compressibility} = -1/V (\partial V/\partial P)_T$$



^aExperimental value (T = 300 K) ~ 61.0

Temperature effects are small over pressure ranges studied

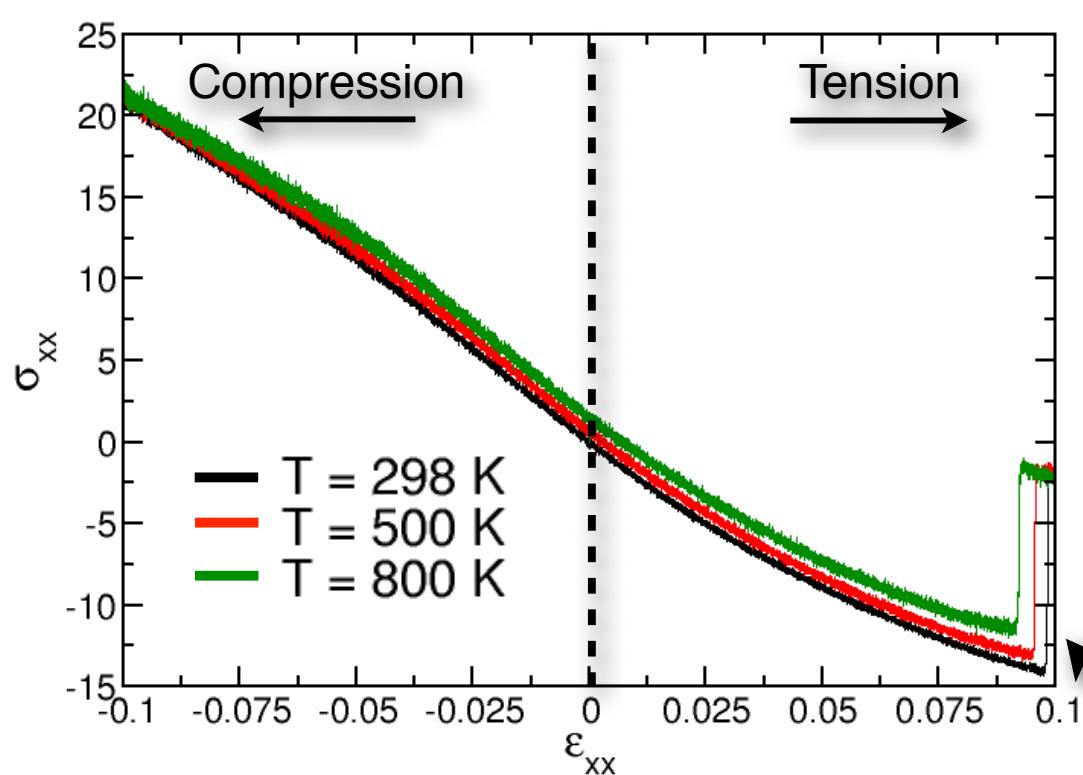
Equilibrium stress-strain curves



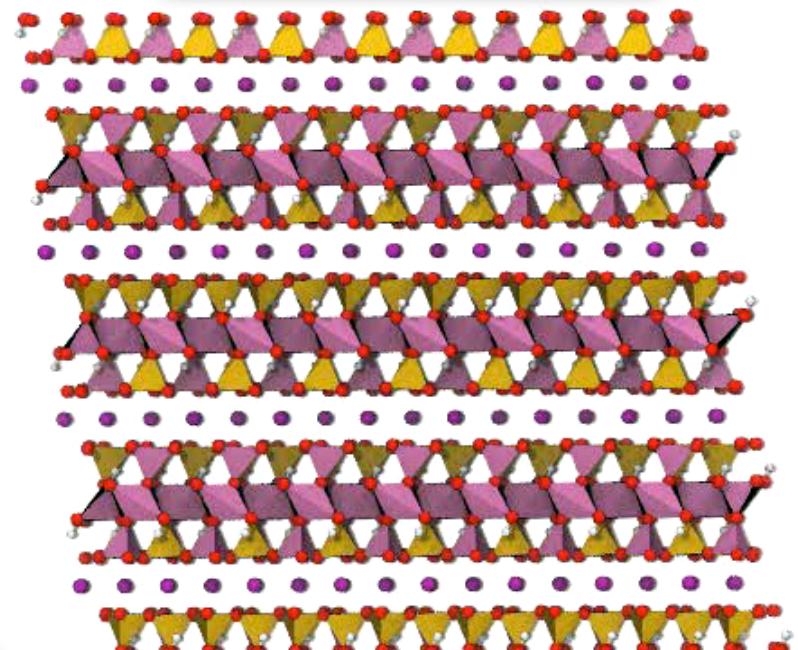
Temperature effects are greater in the z direction perpendicular to the cation interlayer

Uniaxial stress strain curves

Compression and tension simulations are conducted at constant strain rate



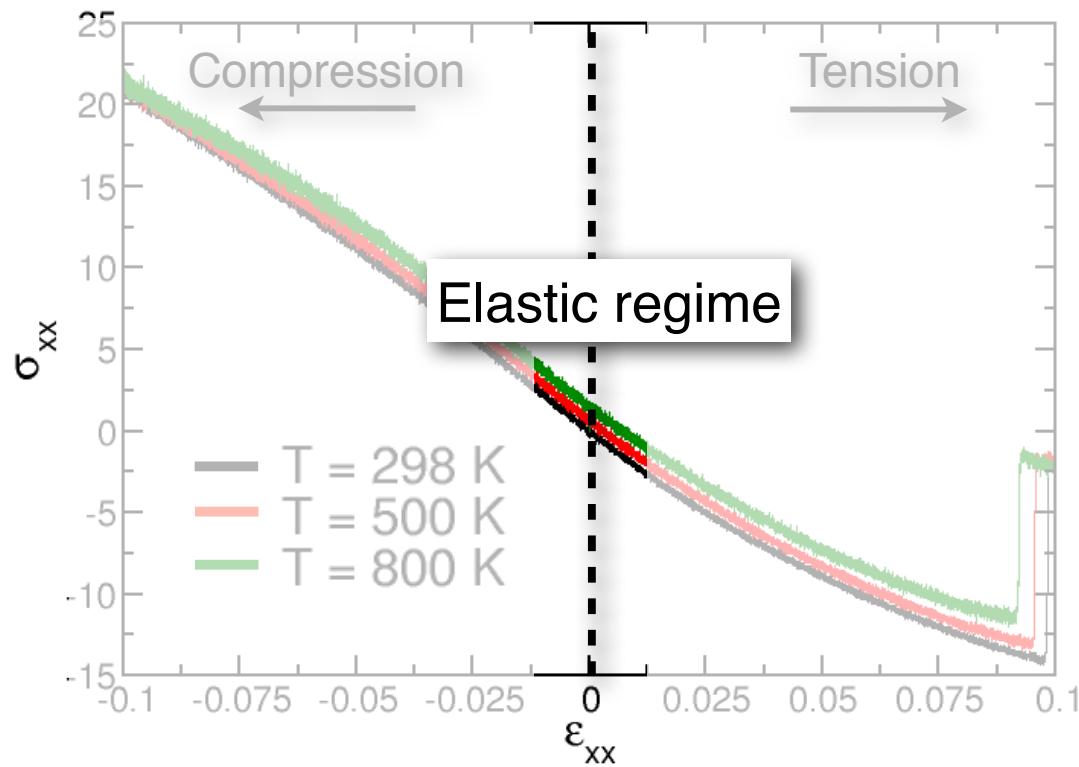
$T = 298 \text{ K, tension}$



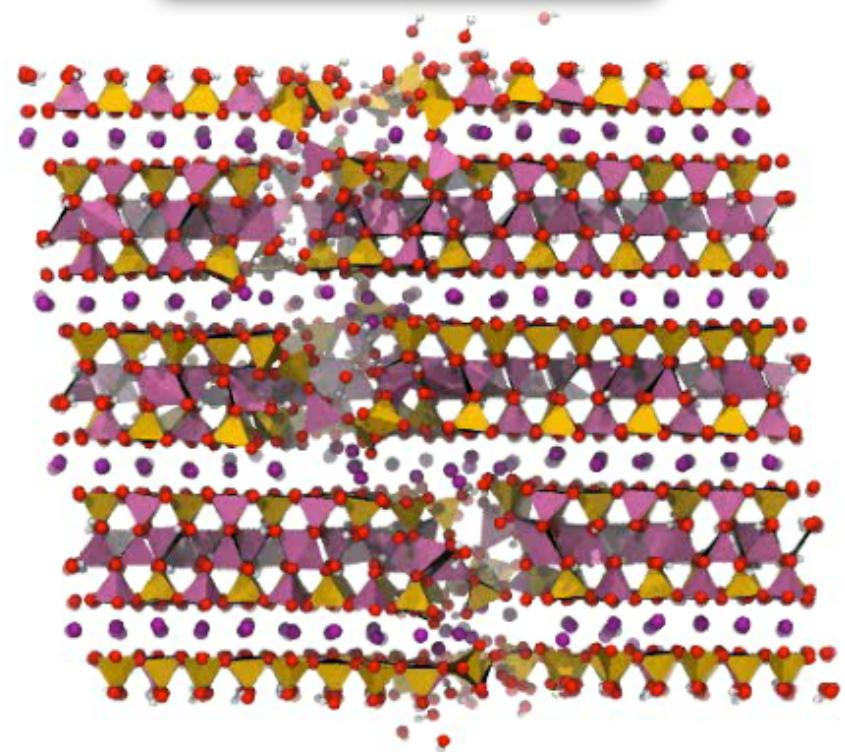
Model breaks apart at high strain

Uniaxial stress strain curves

Compression and tension simulations are conducted at constant strain rate

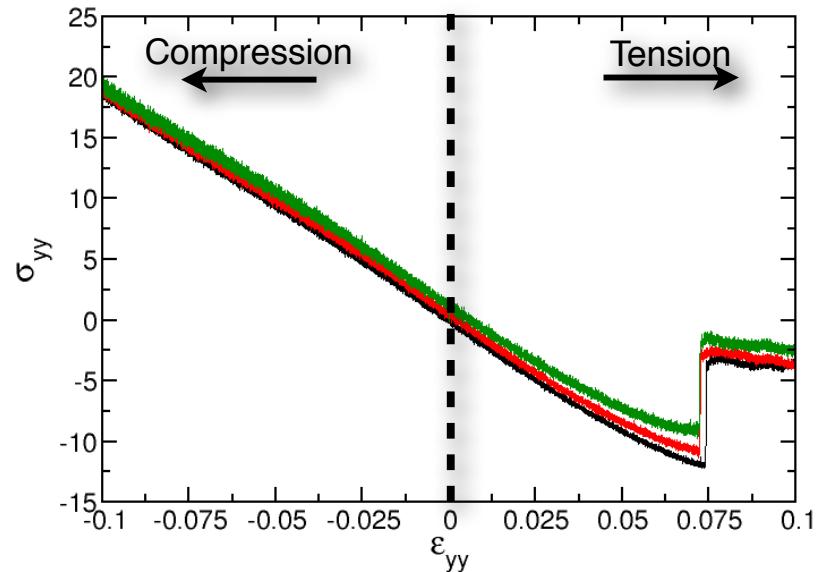
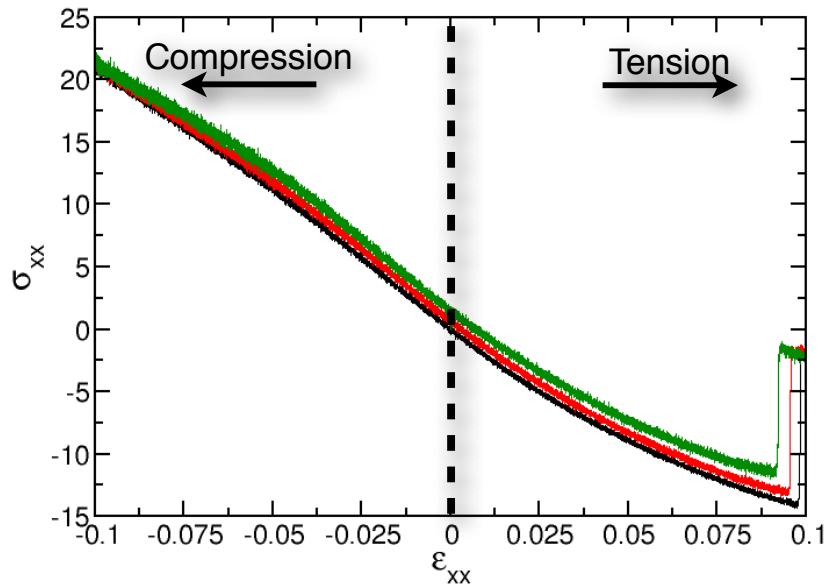


$T = 298 \text{ K, tension}$



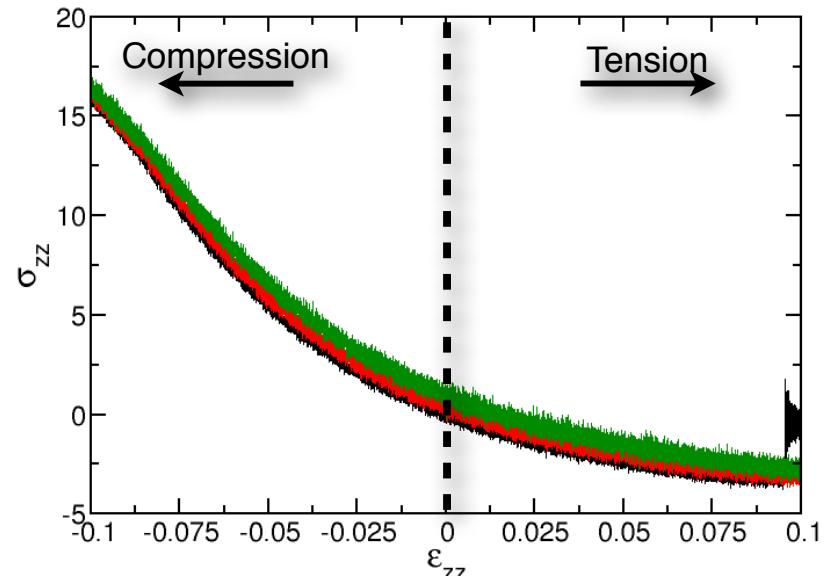
Calculate Young's moduli and shear moduli from data within the elastic regime

Young's modulus



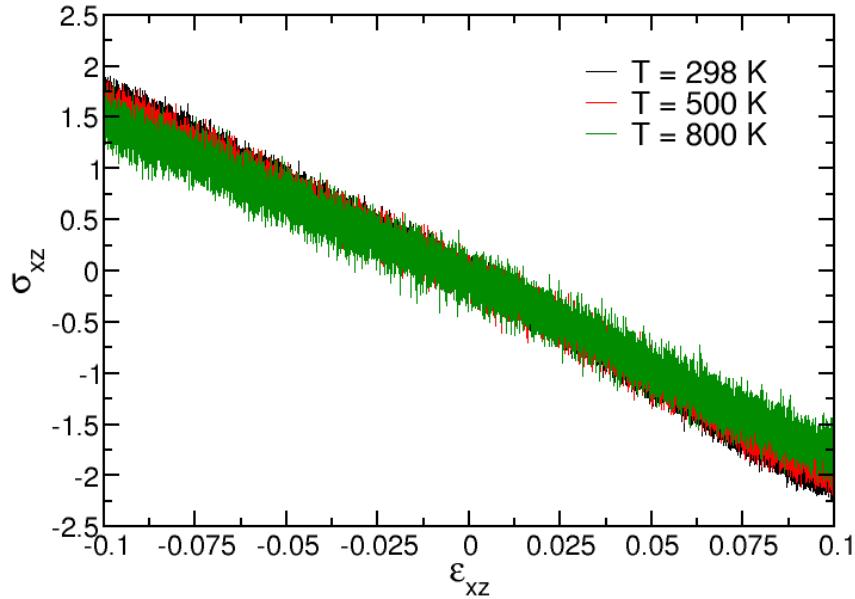
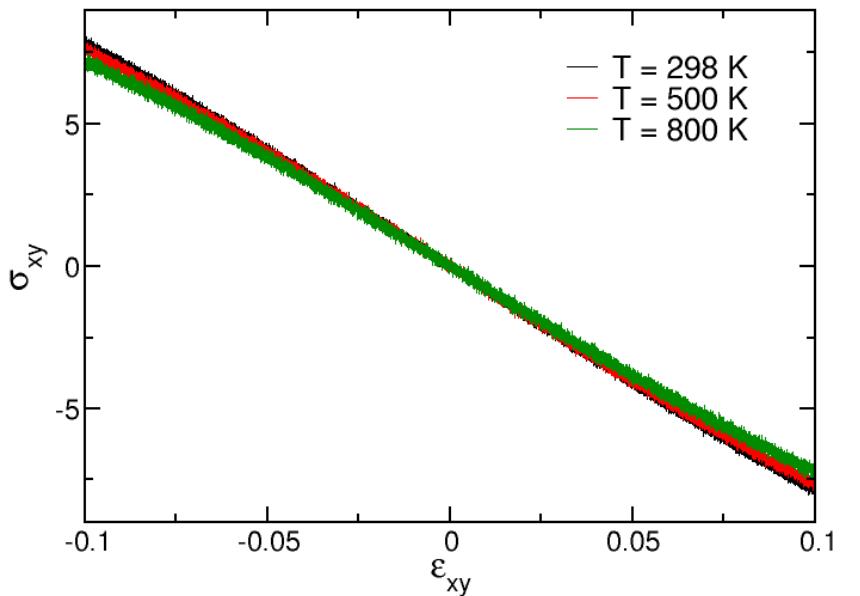
— T = 298 K
 — T = 500 K
 — T = 800 K

E [GPa] / T [K]	Exp	298	500	800
xx	160.9 ^a	220.7	219.9	215.4
yy	161.3 ^a	199.6	198.2	192.6
zz	77.4 ^b	69.4	69.7	74.3



^aVaughan *et al.*, Journal Geophysical Research, 1986; ^bZhang *et al.*, American Mineralogist, 2010

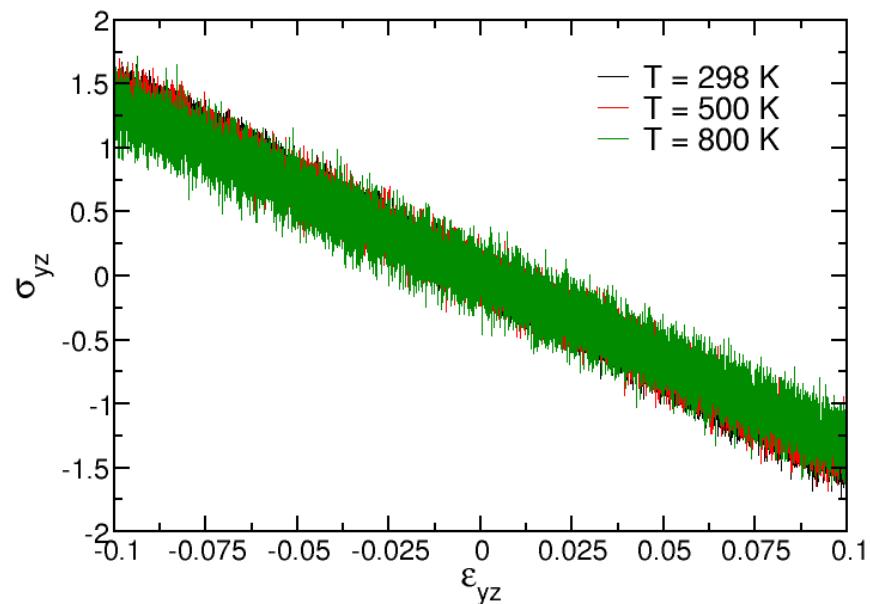
Shear modulus



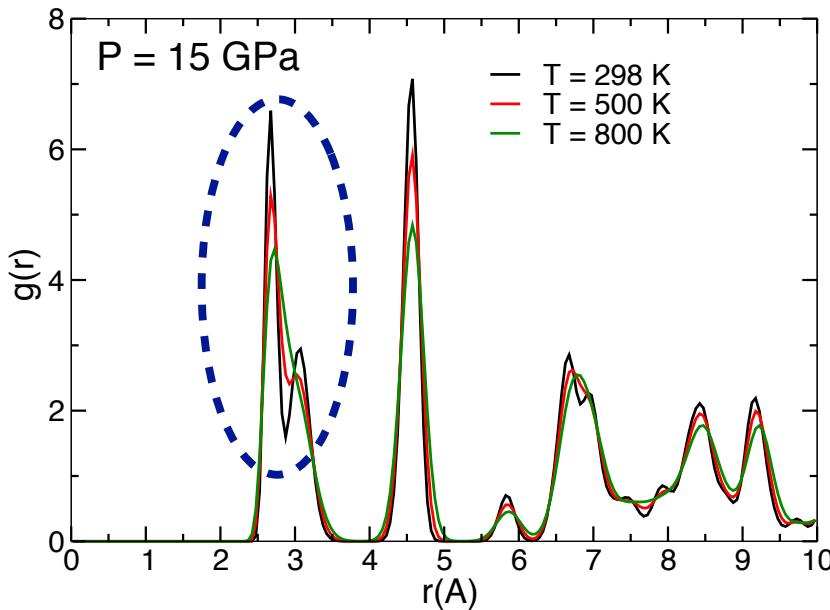
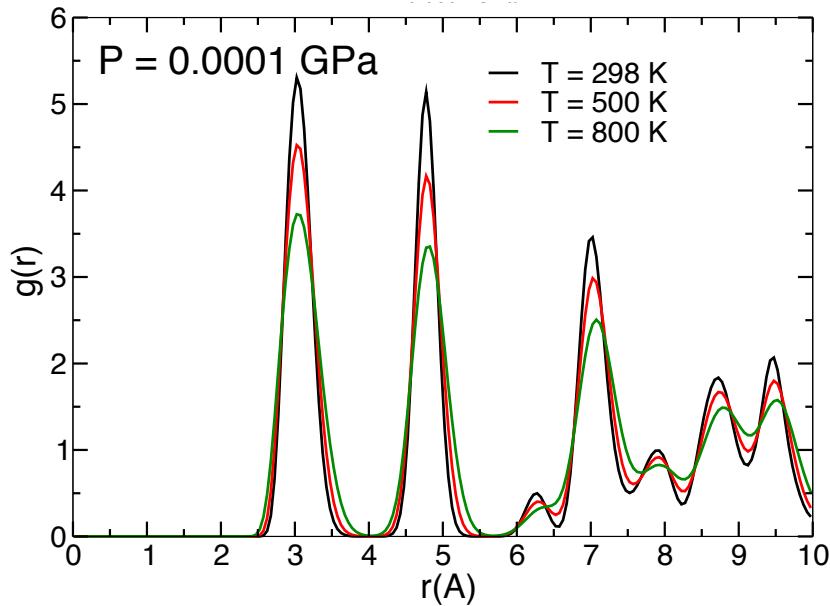
Legend:

- T = 298 K
- T = 500 K
- T = 800 K

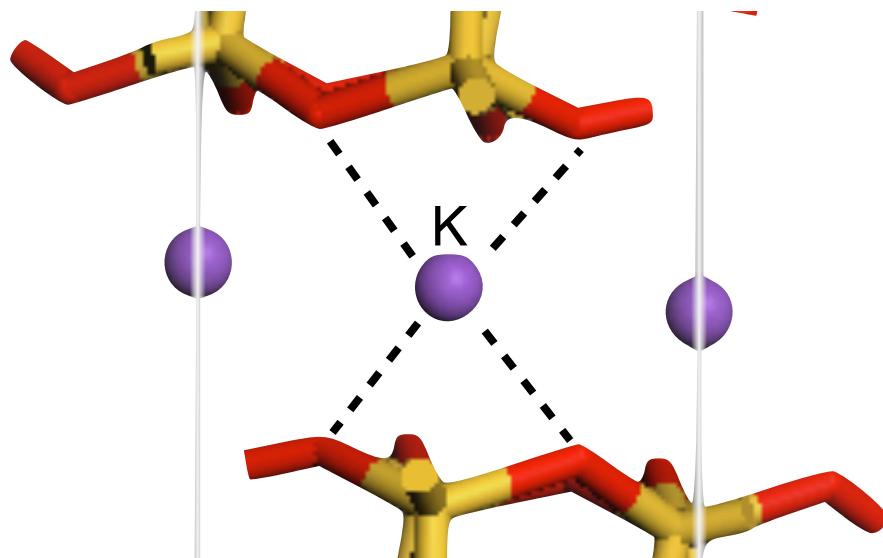
E [GPa] / T [K]	Exp	298	500	800
xy	71	83.9	81.7	78.7
xz	17	18.8	17.7	16.4
yz	15	15.1	13.6	12.7



Structural changes with deformation



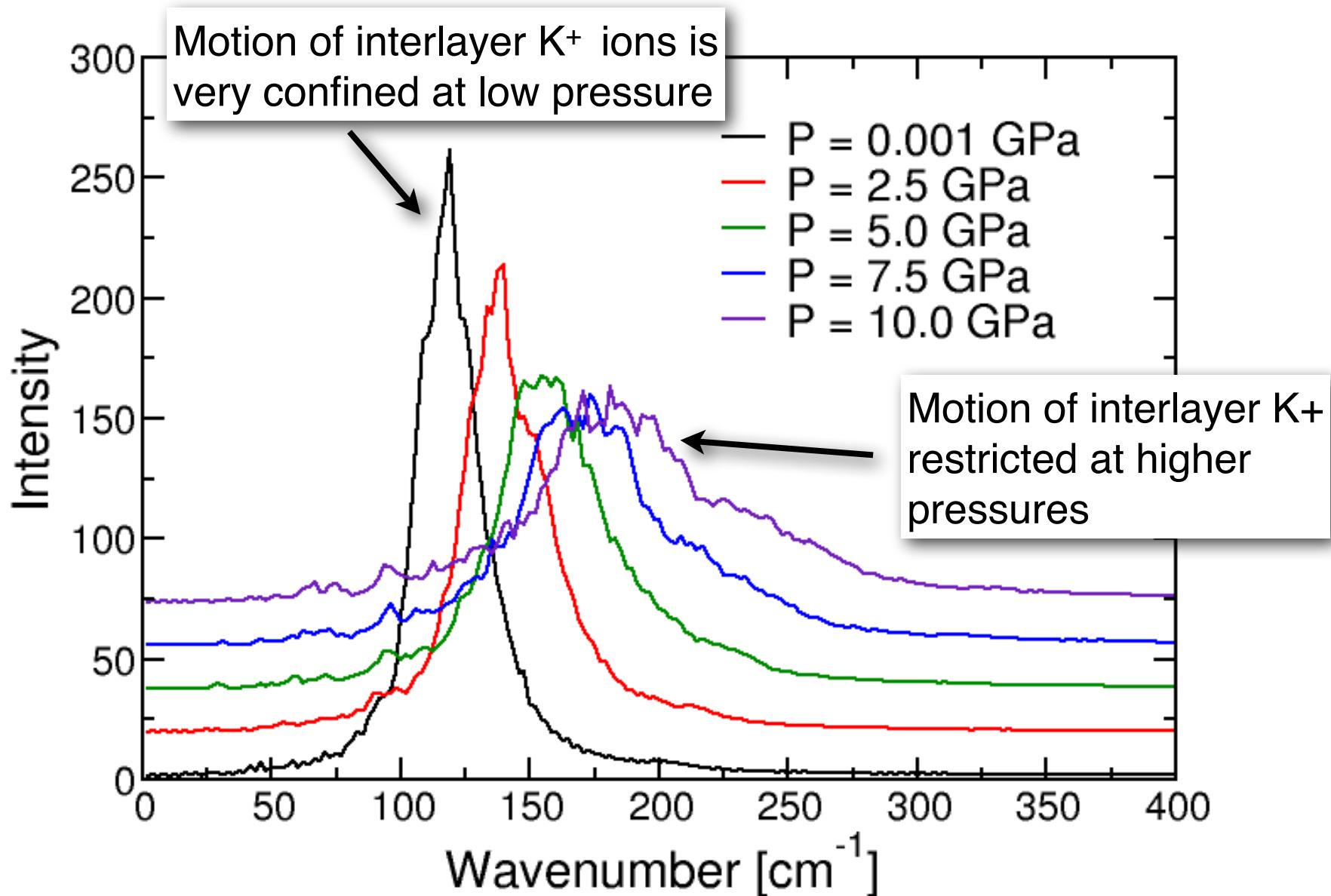
K-O radial distribution function
$$g(r) = \int 4\pi r^2 \rho dr$$



At high pressure see change in first neighbor peak

This change is reduced at higher temperatures

Power spectra - interlayer K⁺



Conclusions

- Our simulation results matches well with experiment
- The clayff force field does a good job at capturing the mechanical properties of muscovite in the elastic regime
- Temperature does not greatly impact the mechanical properties of muscovite in the elastic regime
- Pressure has a greater effect on the mechanical properties of muscovite than does temperature