



Fuel Cycle Research and Development

Volatile Gas Capture Development

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Developments in Nuclear Waste Forms Panel Discussion
Materials for Nuclear Waste Disposal and Environmental Cleanup
MS&T 2011
Columbus, OH October 19-20, 2011



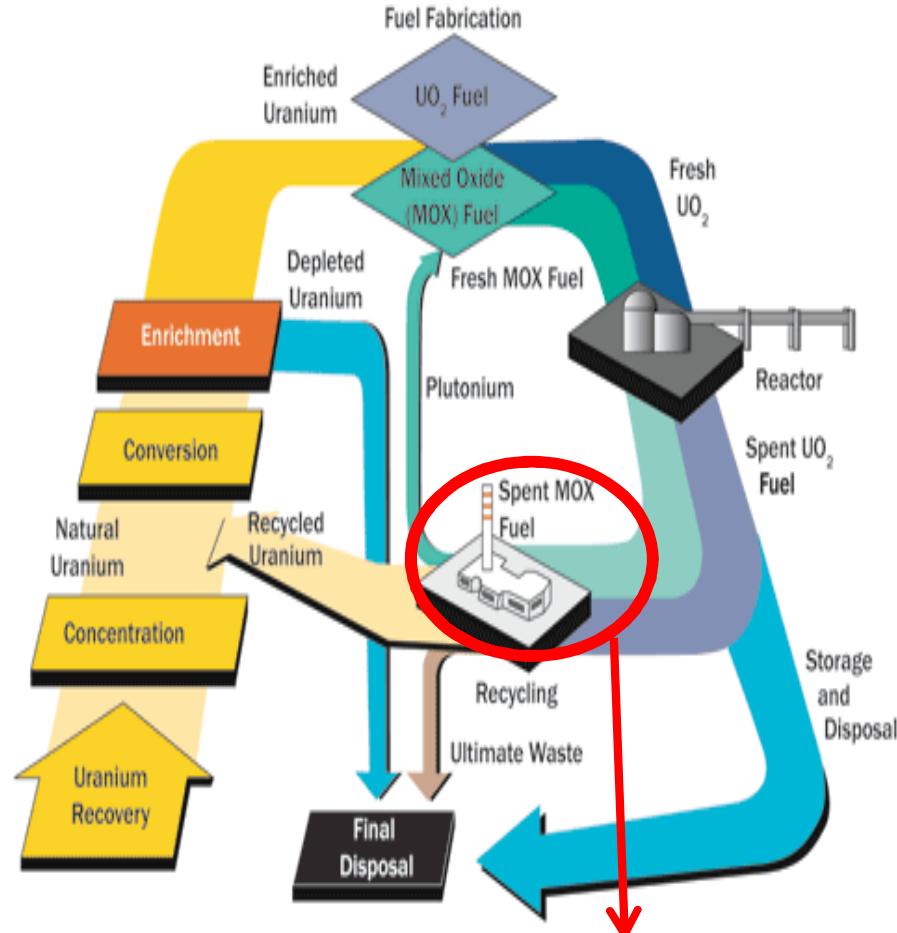
Overview – Separations & Storage

I. Nuclear Waste Legacy (EM) for USA and the world

*Fukushima Daiichi Nuclear Power Plant explosion March 11, 2011: I^{129} , I^{131} volatile gas released; Cs^{135} , Cs^{137} aqueous released
(www.IAEA.org)*



II. Nuclear Fuel Reprocessing (NE)



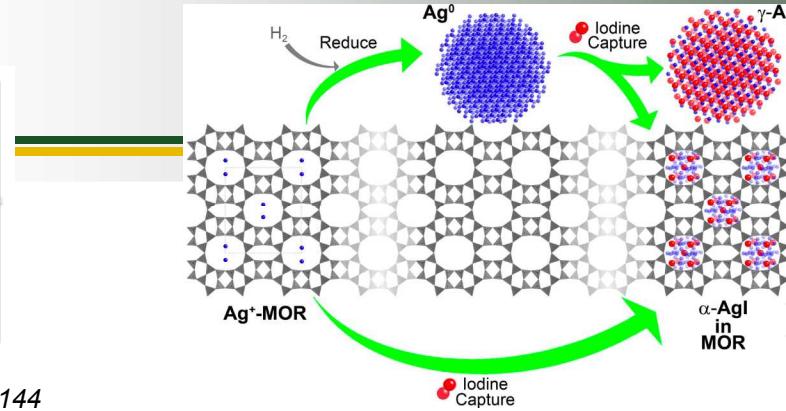
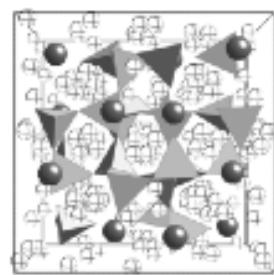
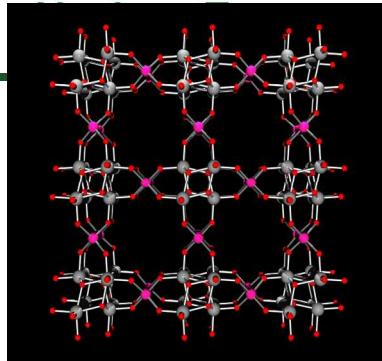
Source: U.S. Nuclear Regulatory Co

MS&T 2011 Panel Discussion

Separations of non-burnable volatile fission products and lesser actinides



SNL Overview – Capture/Storage

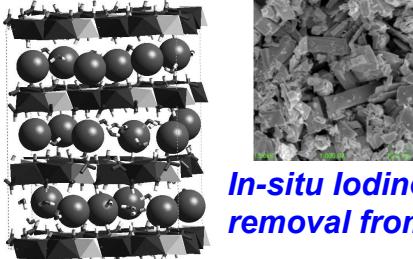


Ag-MOR
 $I_2(g)$ capture & mechanisms

JACS, 2010, 132(26), 8897
J Phys Chem Letters, 2011, 2, 2742

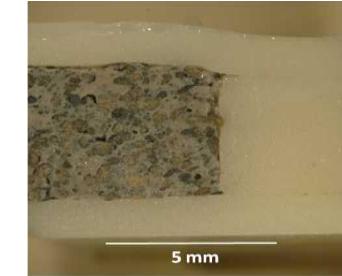
CST, Cs^+ removal from water to Pollucite Waste Form

Applied Geochem, 2011, 26, 57



In-situ Iodine removal from water

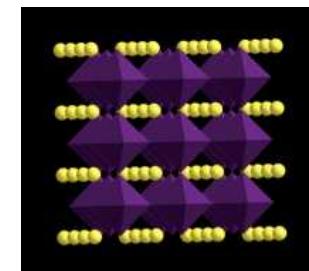
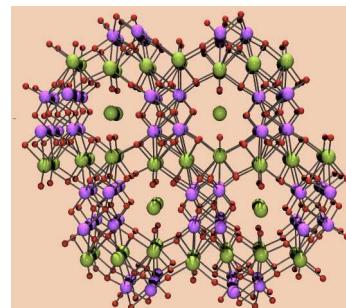
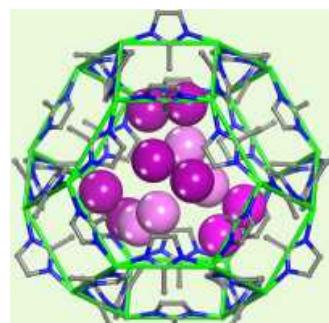
Fundamental Research to Applied to Commercial Products
Design the Separation Material To Develop the Waste Form



Universal Core-Shell Glass Waste Form Iodine & Getter
JACerS, 2011, 94(8), 2412

I_2 /MOF, Isolation to Waste Form

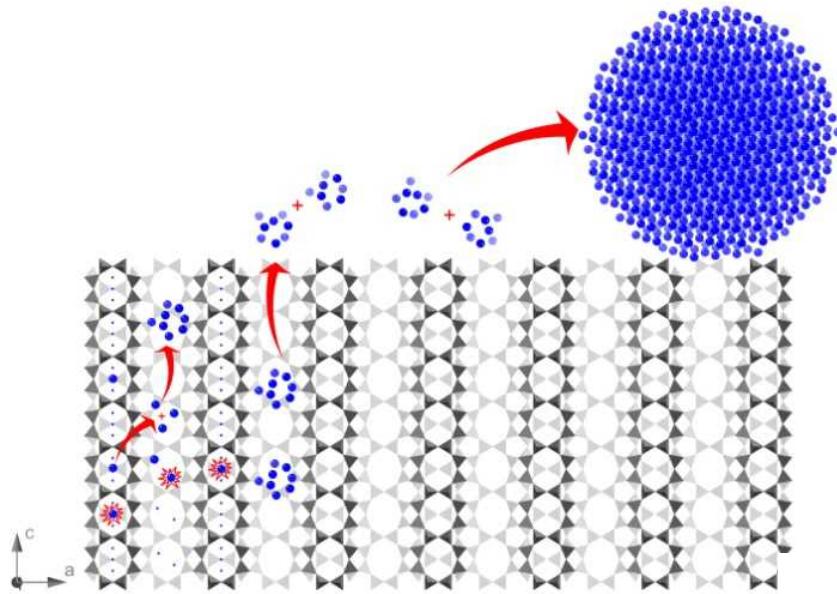
JACS, 2011, 133(32), 12398
Ind. Eng. Chem. Res, 2011, ASAP
US Patent Application, 2011



Sr getter, 1-step to Perovskite waste form
JACS, 2002, 124(3), 1704



Ag-MOR zeolite for I_2 Capture and Temporary Storage

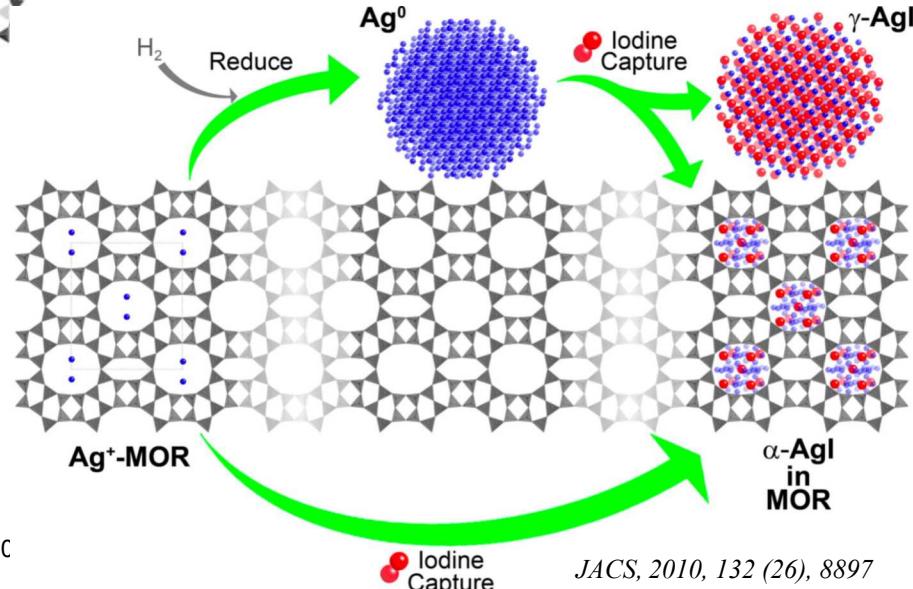


Ag^+ -MOR ion reduction to Ag^0 clusters and migration out of pores upon nanoparticle formation

J Phys Chem Letters, 2011, 2, 2742

Ag^0 -MOR + I_2 yields a mixture of γ -AgI bulk surface nanoparticles and sub-nanometer α -AgI.

Ag^+ -MOR + I_2 produces exclusively sub-nanometer α -AgI ("perfect fit", confined in pores)

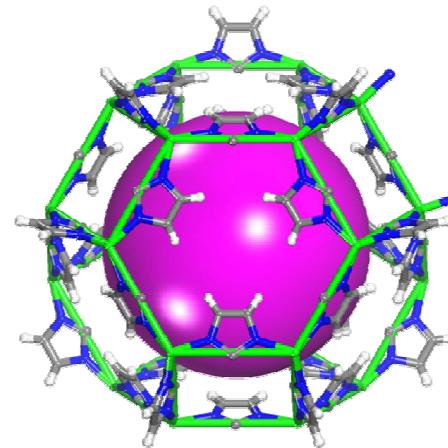




Separations Materials Studies - MOFs

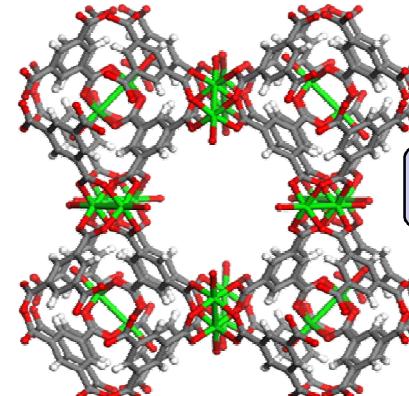
Understand the Scientific Basis for the Capture and
Immobilization Technologies

Basolite Z1200, ZIF-8
Constricted Pore Opening ($\approx 3.4\text{\AA}$)
 $1100 - 1600 \text{ m}^2/\text{g}$
Pore Volume = 0.636 cc/g
stable in Air & H_2O



$\text{I}_2@\text{ZIF-8} \sim 125 \text{ wt.\% I}_2$

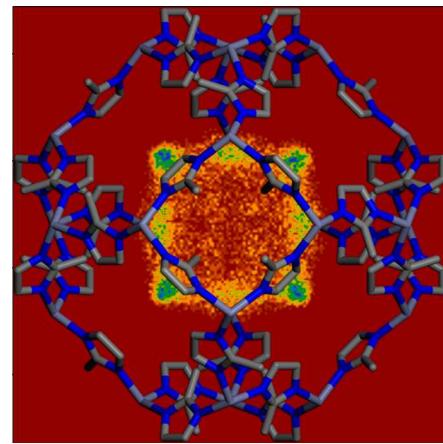
Basolite C300, HKUST-1
Open Channels, $\approx 1\text{nm}$ in 3D
 $1500-2100 \text{ m}^2/\text{g}$
Exposed Metal Sites of Framework



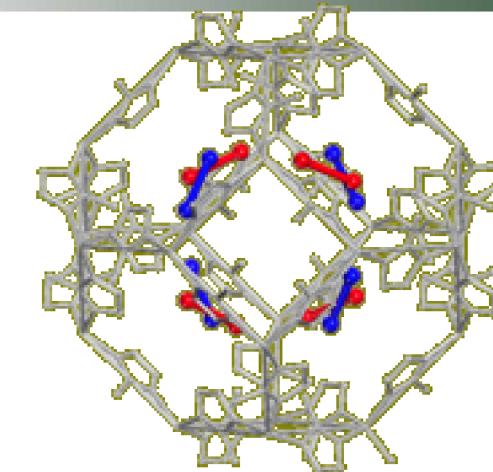
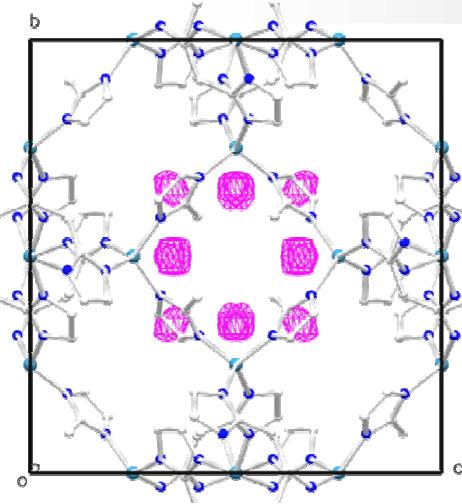
$\text{I}_2@\text{HKUST-1} \sim 150 \text{ wt.\% I}_2$



Fundamental Studies to Understand Iodine Separations and Capture

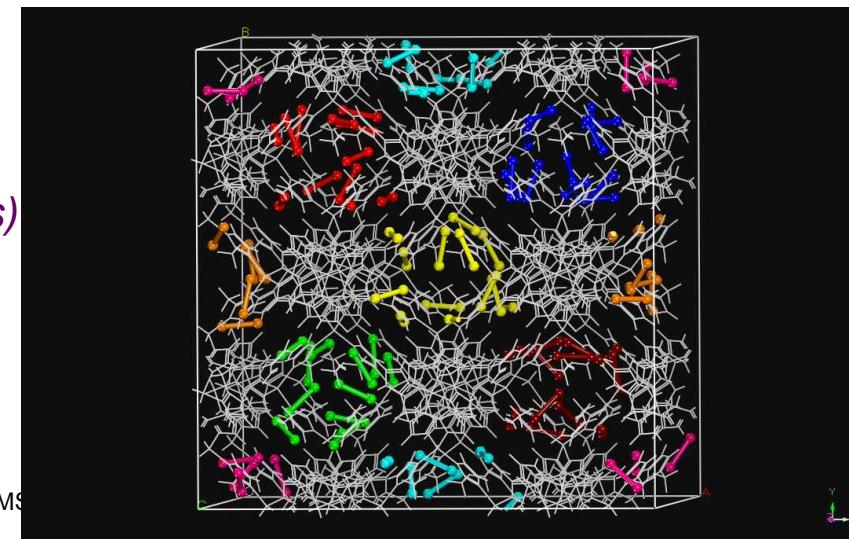


Modeling + crystallography



ZIF-8/I₂ Structure Determination

*NEAMS Modeling (MD simulations)
Indicate cage trapment of I₂*





Waste Forms – Low Temperature Sintering Glass for GCM Waste Form

JACerS, 2011, 94(8), 2412

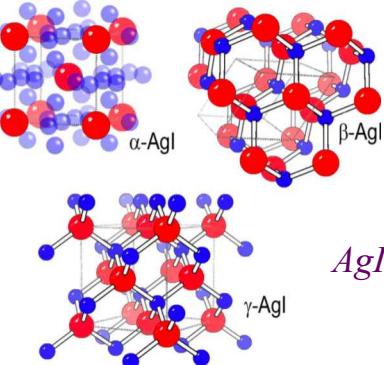
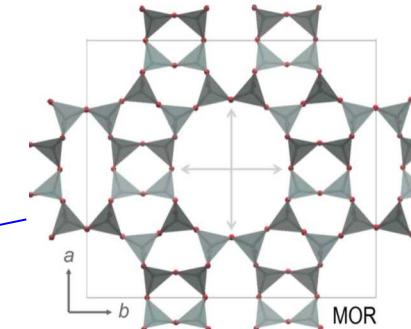
Homogenous Glass GCM: for
AgI or AgI-MOR off-gas capture and storage



50 wt% AgI/50 wt% Glass
500°C for 3 hr



50 wt% AgI/50 wt% Glass,
500°C for 3 hr



AgI mp 556°C

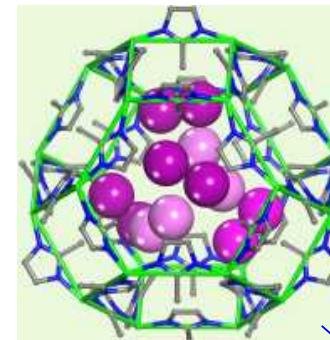
Core-Shell GCM Glass Waste Forms



Glass shell, AgI/glass core,
75/25



Glass shell,
AgI-MOR/Ag/Glass core 80/20/5



I_2 /MOF, Isolation
to Waste Form

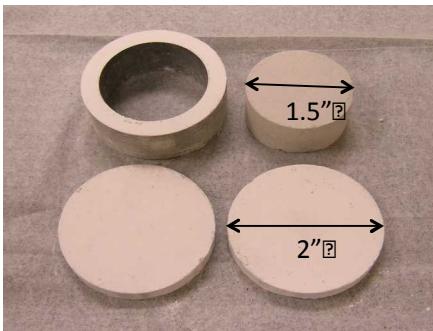
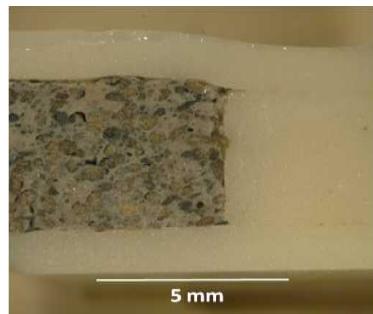
JACS, 2011, 133(32), 12398
Ind. Eng. Chem. Res.,
2011, ASAP
US Patent Application, 2011





Waste Forms – Low Temperature Sintering Glass Core-Shell Scalable Composition

*Universal Core-Shell Glass Waste
Form Iodine & Getter*



Scale-up

Proof of Concept Completed

Industrialization?: Fabrication and stockpile of core/shell components for as-needed use

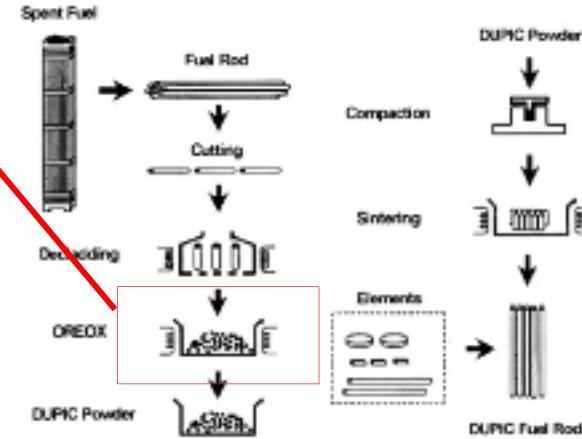
Better understanding of performance in real-world conditions needed



Related MOC Project: Ceramic Nanofiber Membranes

Removal of Fission Products and Poisons for Fuel Recycle:

- crushing the fuel
- performing a simplified heating process and/or hot gas extraction process ($\approx 650-1500^{\circ}\text{C}$)
- separate out the volatile fission products (eg., Cs, Sr, Rb) and poisons (eg., Xe)
- recycle of the fuel through mixing with fresh fuel and re-sintering

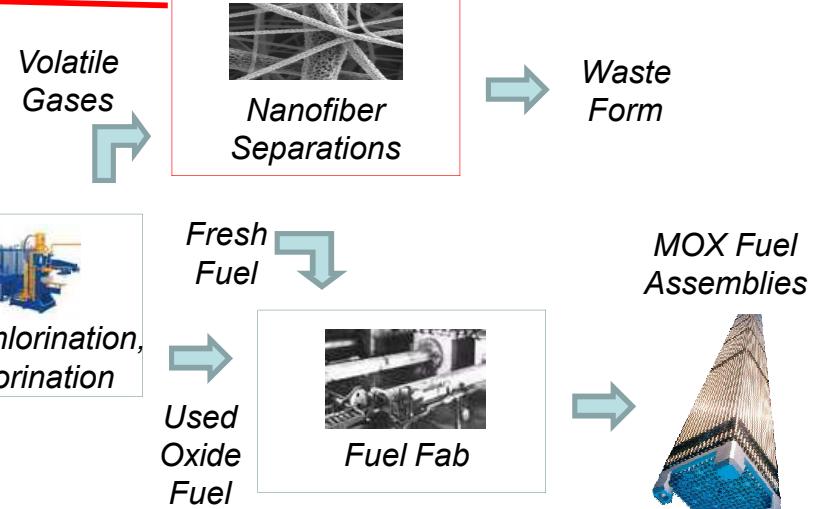


Kang, et.al., *Metal.Mater.* 2000, 6(6), 583

DUPIC-like Fuel Fabrication Process:

Model process system for insertion of nanofiber membrane & capture products during heating

Del Cul, et al., *Advanced Head-End Processing of Spent Fuel*, 2005; *ANS Annual Meeting*; *DOE/NE-SWG meeting*, Albuquerque, NM Dec. 2010





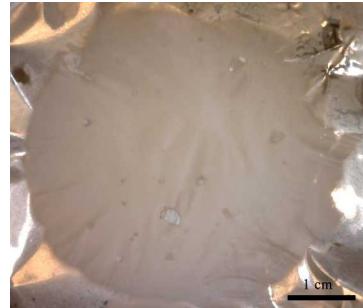
Off Gas Separations Membranes for MOC Fuel Cycle Applications

*High Temperature
Volatile Gas Capture
Materials*

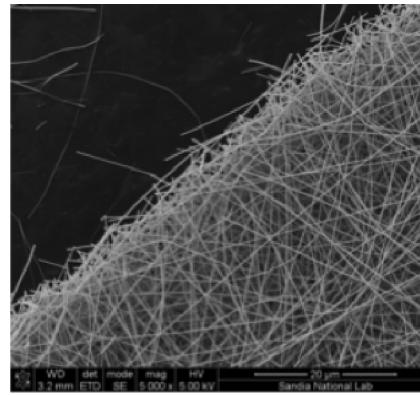
Nanofiber Membranes:

- **High flux** (accessibility to high open surface area)
- **Easy Insertion in off-gas streams** (membrane)
- **Compatible with innovative separation technologies**
- **Chemical, mechanical, radioactive durability**
- **High capacity** (high waste loading)
- **Selectivity** (catch trace amount of residue)
- **Functionalization:** Bulk Surface and Internal Pore System

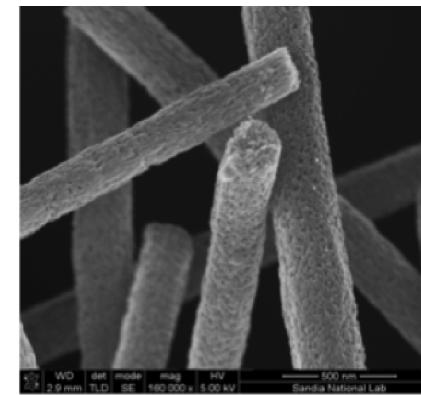
To Date:



Optical image of
membrane

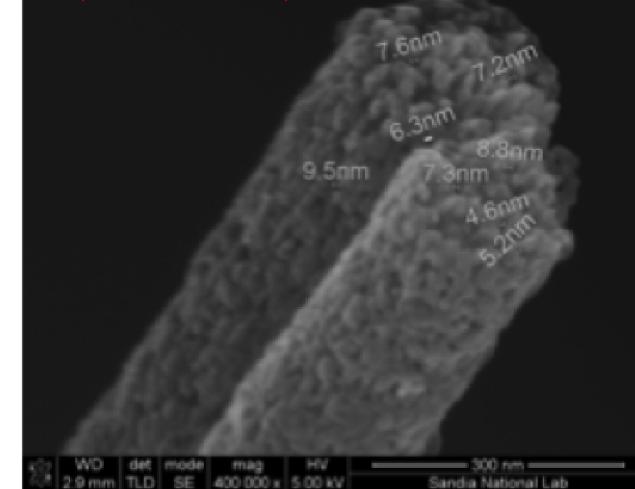


SEM image of
Nanofiber Membrane
(20 micron scale bar)



SEM image of Nanofibers
(500 nm scale bar)

SEM image of Nanopores in the Nanofibers
(500 nm scale bar).





Conclusions

- Understanding the fundamentals to better design the capture material and waste form
- The Capture Material and the Waste Form can be designed for each other to ensure materials compatibility and processing simplicity
- Understanding of Waste Form performance criteria will greatly aid in design and implementation

Acknowledgements

- Sandia National Laboratories is a multi-program lab managed and operated by Sandia Corp., a wholly owned subsidiary of Lockheed Martin Corporation, for the US DOE's NNSA under contract DE-AC04-94AL85000.
- Continued financial support from DOE/NE-FCRD-SWG