



U.S. DEPARTMENT OF
ENERGY

Nuclear Energy

SAND2011-7848C

Fuel Cycle Research and Development

Volatile Gas Capture Development

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Developments in Nuclear Waste Forms Panel Discussion
Materials for Nuclear Waste Disposal and Environmental Cleanup
MS&T 2011
Columbus, OH October 19-20, 2011



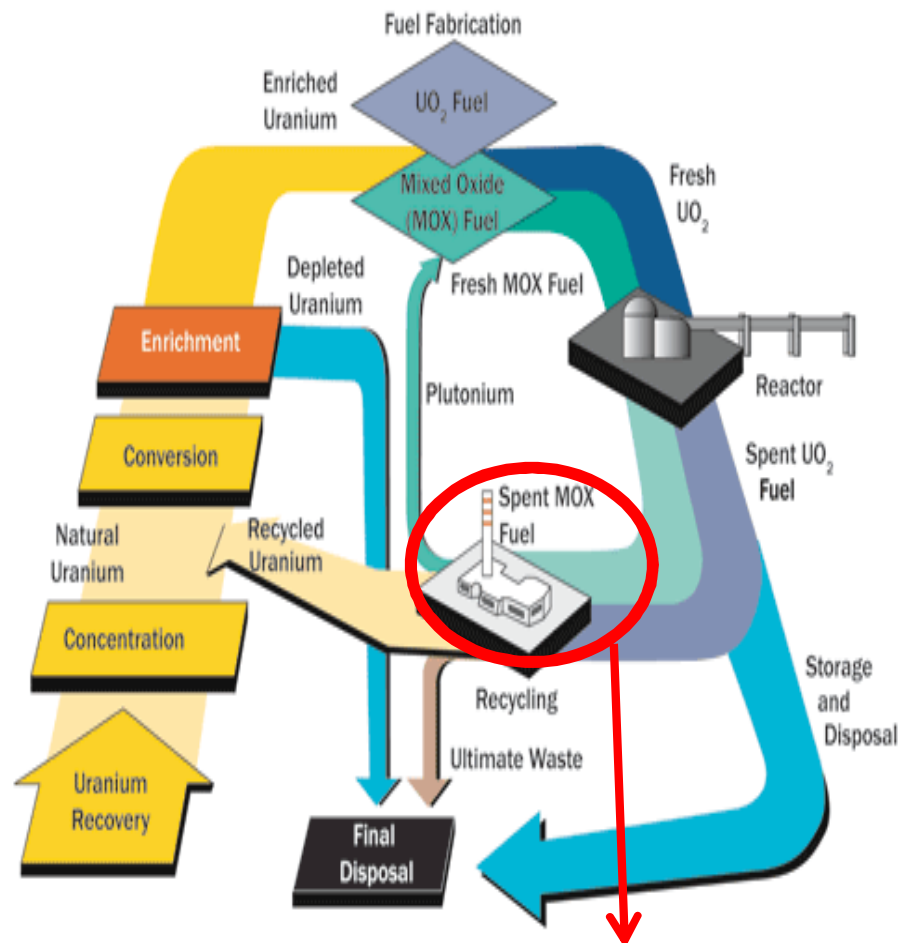
Overview – Separations & Storage

I. Nuclear Waste Legacy (EM) for USA and the world

*Fukushima Daiichi
Nuclear Power
Plant explosion
March 11, 2011:
 I^{129} , I^{131} volatile
gas released;
 Cs^{135} , Cs^{137}
aqueous released
(www.IAEA.org)*



II. Nuclear Fuel Reprocessing (NE)

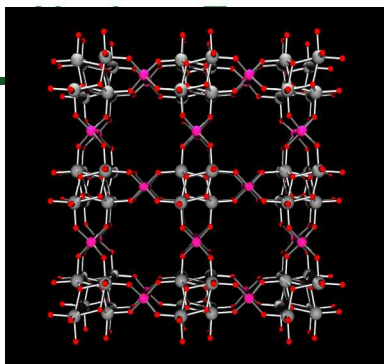


Source: U.S. Nuclear Regulatory Commission

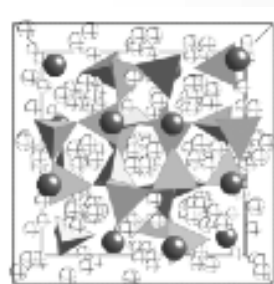
Separations of non-burnable volatile fission products and lesser actinides



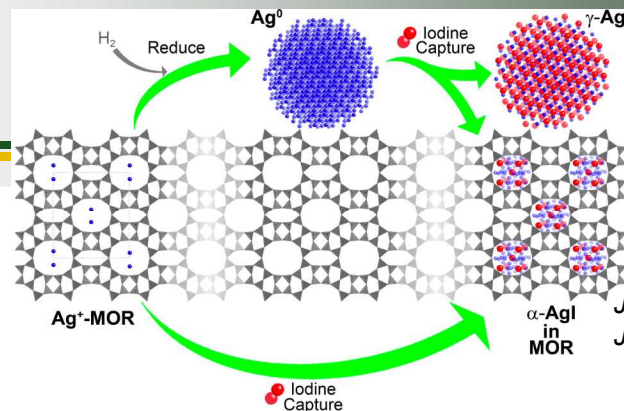
SNL Overview – Capture/Storage



CST, Cs⁺ removal from water to Pollucite Waste Form



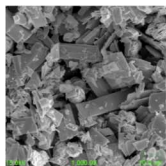
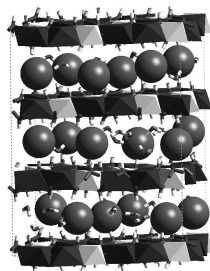
R&D100 1996
JACerS, **2009**, 92(9), 2144
JACerS, **2011**, 94(9), 3053
Solvent Extr. & Ion Exch, **2011**, submitted



**Ag-MOR
I₂(g) capture & mechanisms**

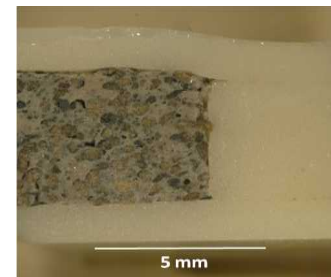
JACS, **2010**, 132(26), 8897
J Phys Chem Letters, **2011**, 2, 2742

Applied Geochem, **2011**, 26, 57



In-situ Iodine removal from water

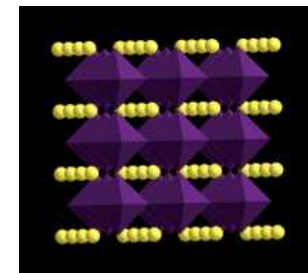
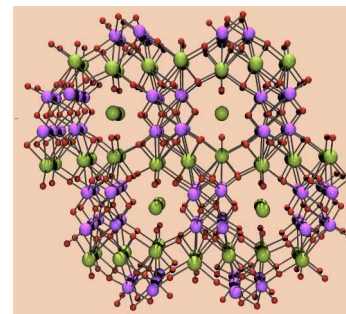
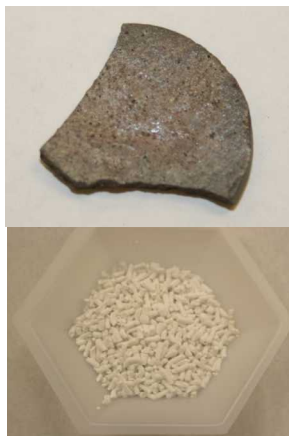
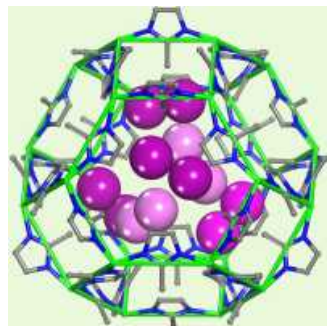
**Fundamental Research to
Applied to Commercial Products
Design the Separation Material
To Develop the Waste Form**



Universal Core-Shell Glass Waste Form Iodine & Getter
JACerS, **2011**, 94(8), 2412

I₂/MOF, Isolation to Waste Form

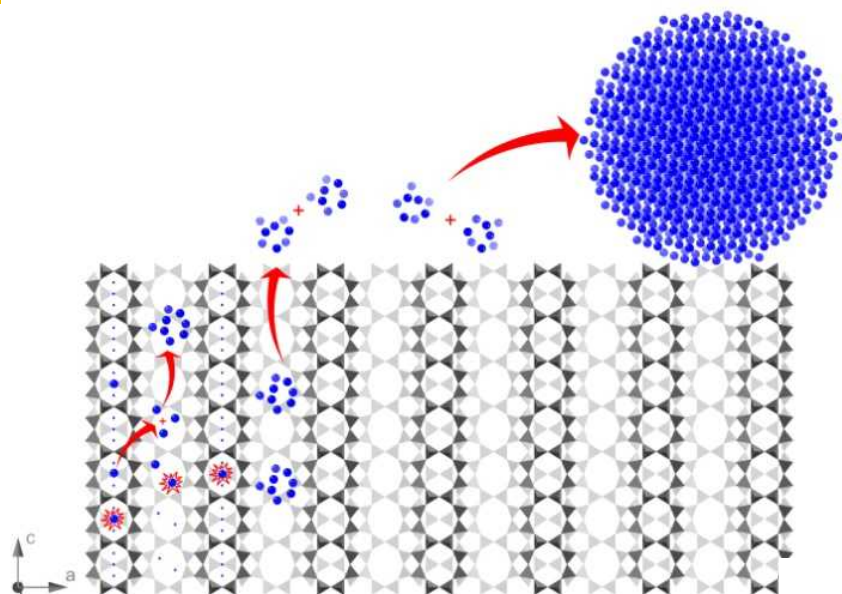
JACS, **2011**, 133(32), 12398
Ind. Eng. Chem. Res., **2011**, ASAP
US Patent Application, 2011



Sr getter, 1-step to Perovskite waste form
JACS, **2002**, 124(3), 1704



Ag-MOR zeolite for I_2 Capture and Temporary Storage

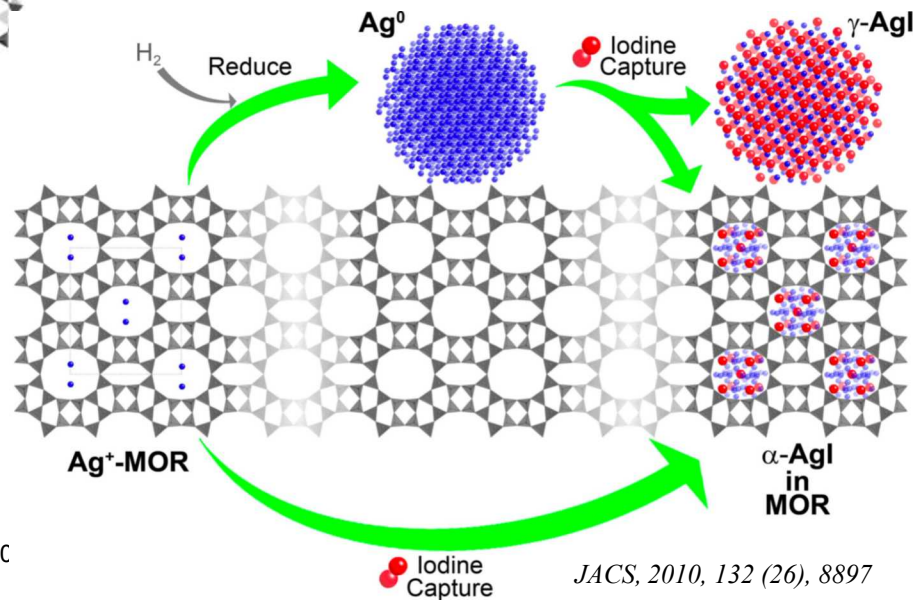


Ag^+ -MOR ion reduction to Ag^0 clusters and migration out of pores upon nanoparticle formation

J Phys Chem Letters, 2011, 2, 2742

Ag^0 -MOR + I_2 yields a mixture of γ -AgI bulk surface nanoparticles and sub-nanometer α -AgI.

Ag^+ -MOR + I_2 produces exclusively sub-nanometer α -AgI (“**perfect fit**”, confined in pores)

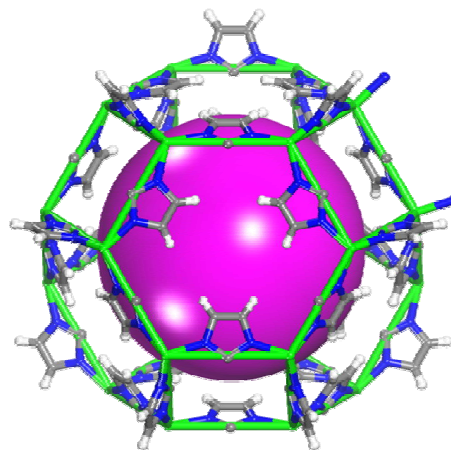




Separations Materials Studies - MOFs

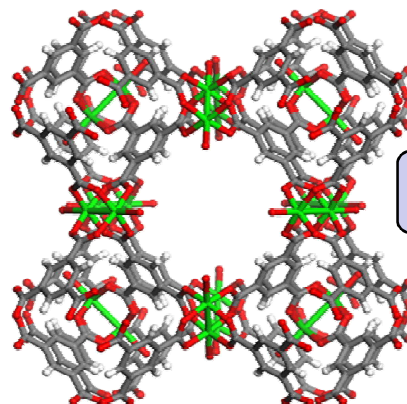
Understand the Scientific Basis for the Capture and Immobilization Technologies

Basolite Z1200, ZIF-8
Constricted Pore Opening ($\approx 3.4\text{\AA}$)
1100 – 1600 m^2/g
Pore Volume = 0.636 cc/g
stable in Air & H_2O



$\text{I}_2@\text{ZIF-8} \sim 125 \text{ wt.\% I}_2$

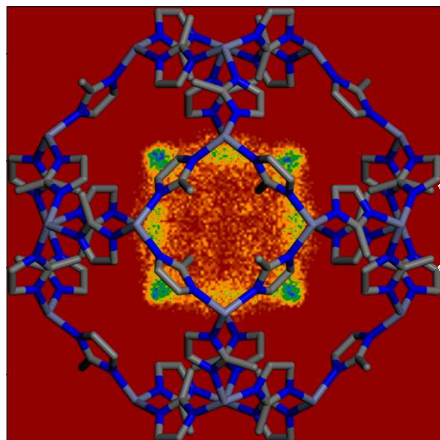
Basolite C300, HKUST-1
Open Channels, $\approx 1\text{nm}$ in 3D
1500-2100 m^2/g
Exposed Metal Sites of Framework



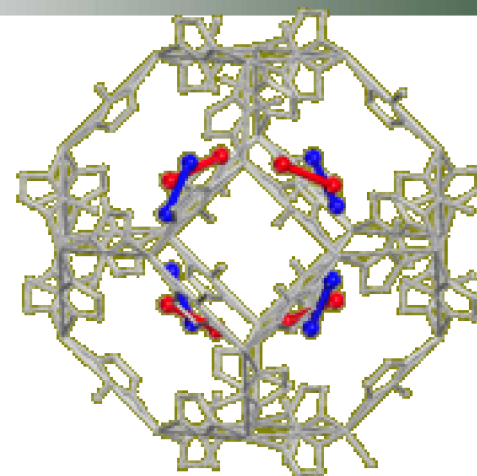
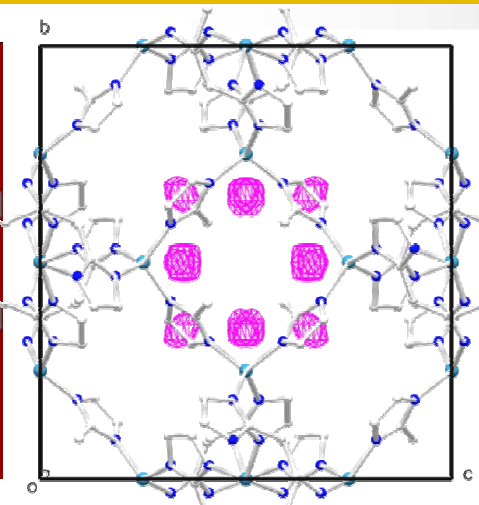
$\text{I}_2@\text{HKUST-1} \sim 150 \text{ wt.\% I}_2$



Fundamental Studies to Understand Iodine Separations and Capture

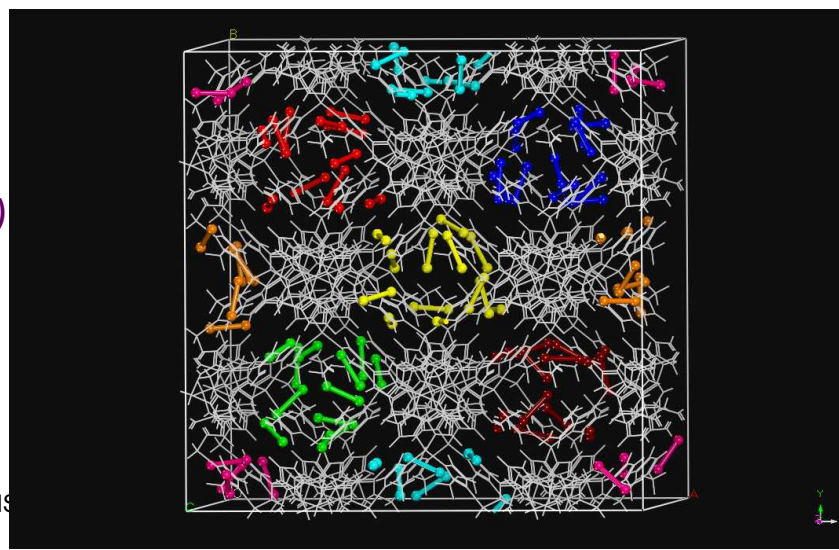


Modeling + crystallography



ZIF-8/I₂ Structure Determination

*NEAMS Modeling (MD simulations)
Indicate cage trapment of I₂*





Waste Forms – Low Temperature Sintering Glass for GCM Waste Form

JACerS, 2011, 94(8), 2412

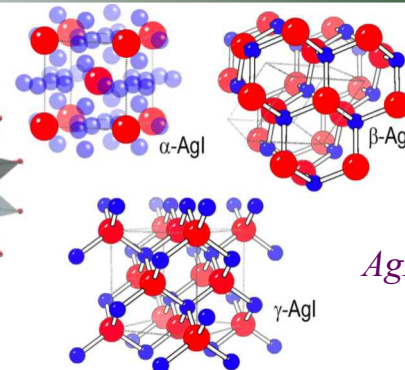
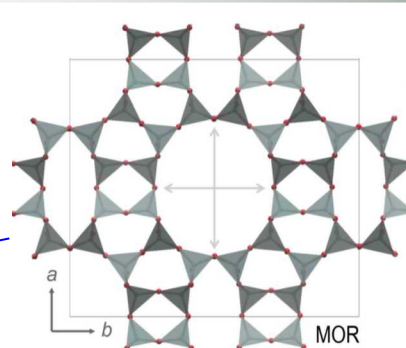
Homogenous Glass GCM: for
Agl or AgI-MOR off-gas capture and storage



50 wt% AgI/50 wt% Glass
500°C for 3 hr



50 wt% AgI/50 wt% Glass,
500°C for 3 hr



AgI mp 556°C

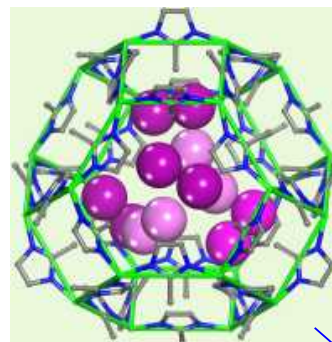
Core-Shell GCM Glass Waste Forms



Glass shell, AgI/glass core,
75/25



Glass shell,
AgI-MOR/Ag/Glass core 80/20/5



***I₂/MOF, Isolation
to Waste Form***

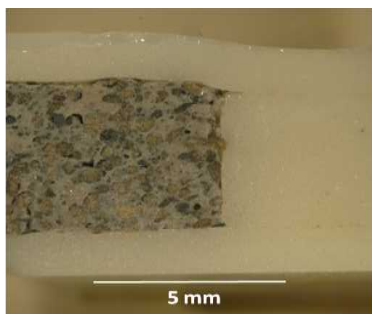
*JACS, 2011, 133(32), 12398
Ind. Eng. Chem. Res,
2011, ASAP
US Patent Application, 2011*





Waste Forms – Low Temperature Sintering Glass Core-Shell Scalable Composition

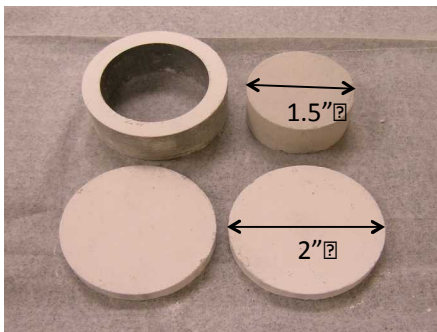
*Universal Core-Shell Glass Waste
Form Iodine & Getter*



Proof of Concept Completed

Industrialization?: Fabrication and
stockpile of core/shell components for
as-needed use

Better understanding of performance in
real-world conditions needed



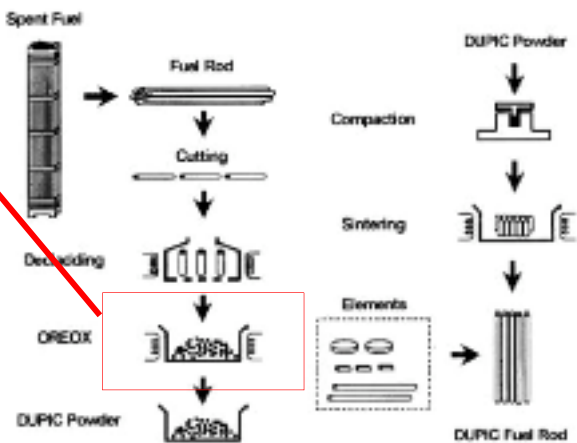
Scale-up



Related MOC Project: Ceramic Nanofiber Membranes

Removal of Fission Products and Poisons for Fuel Recycle:

- crushing the fuel
- performing a simplified heating process and/or hot gas extraction process ($\approx 650-1500^{\circ}\text{C}$)
- separate out the volatile fission products (eg., Cs, Sr, Rb) and poisons (eg., Xe)
- recycle of the fuel through mixing with fresh fuel and re-sintering



Kang, et al., Metal Mater. 2000, 6(6), 583

DUPIC-like Fuel Fabrication Process:

Model process system for insertion of nanofiber membrane & capture products during heating

Del Cul, et al., Advanced Head-End Processing of Spent Fuel, 2005; ANS Annual Meeting; DOE/NE-SWG meeting, Albuquerque, NM Dec. 2010

Volatile Gases

Nanofiber Separations

Waste Form

Dry Chlorination



Fuel Assemblies



Hardware Removal & Decladding

Used Bare Fuel



Volox, Chlorination, or Fluorination

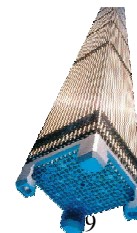
Fresh Fuel

Used Oxide Fuel



Fuel Fab

MOX Fuel Assemblies





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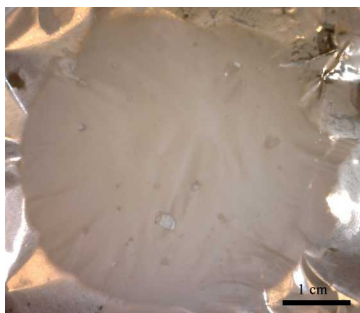
Off Gas Separations Membranes for MOC Fuel Cycle Applications

*High Temperature
Volatile Gas Capture
Materials*

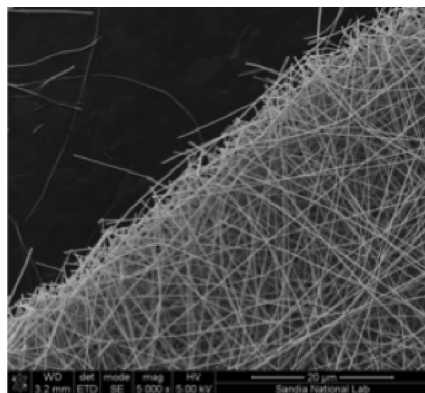
Nanofiber Membranes:

- **High flux** (accessibility to high open surface area)
- **Easy Insertion in off-gas streams** (membrane)
- **Compatible with innovative separation technologies**
- **Chemical, mechanical, radioactive durability**
- **High capacity** (high waste loading)
- **Selectivity** (catch trace amount of residue)
- **Functionalization:** Bulk Surface and Internal Pore System

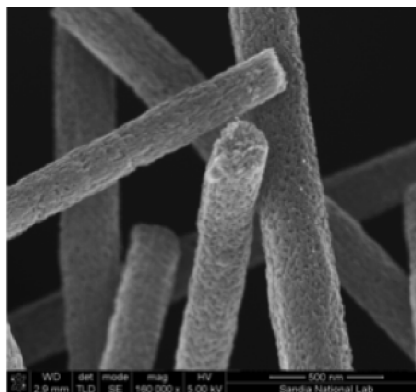
To Date:



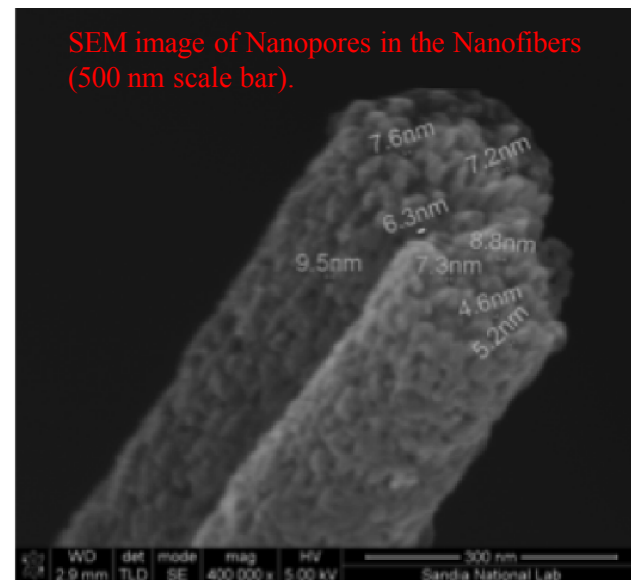
Optical image of
membrane



*SEM image of
Nanofiber Membrane
(20 micron scale bar)*



*SEM image of Nanofibers
(500 nm scale bar)*



*SEM image of Nanopores in the Nanofibers
(500 nm scale bar).*



- Understanding the fundamentals to better design the capture material and waste form
- The Capture Material and the Waste Form can be designed for each other to ensure materials compatibility and processing simplicity
- Understanding of Waste Form performance criteria will greatly aid in design and implementation

Acknowledgements

- Sandia National Laboratories is a multi-program lab managed and operated by Sandia Corp., a wholly owned subsidiary of Lockheed Martin Corporation, for the US DOE's NNSA under contract DE-AC04-94AL85000.
- Continued financial support from DOE/NE-FCRD-SWG