

# Characterization of electrodeposited nano-crystalline $\text{Sb}_2\text{Te}_3$ films

**J.L. Lensch-Falk<sup>1</sup>, D. Banga<sup>1</sup>, V. Stavila<sup>1</sup>, P. Sharma<sup>2</sup>, D.L. Medlin<sup>1</sup>**

<sup>1</sup>Sandia National Laboratories, Livermore, CA

<sup>2</sup>Sandia National Laboratories, Albuquerque, NM

# Electrodeposition of $\text{Sb}_2\text{Te}_3$

- Electrodeposition is promising for synthesis of thermoelectric materials such as  $\text{Sb}_2\text{Te}_3$

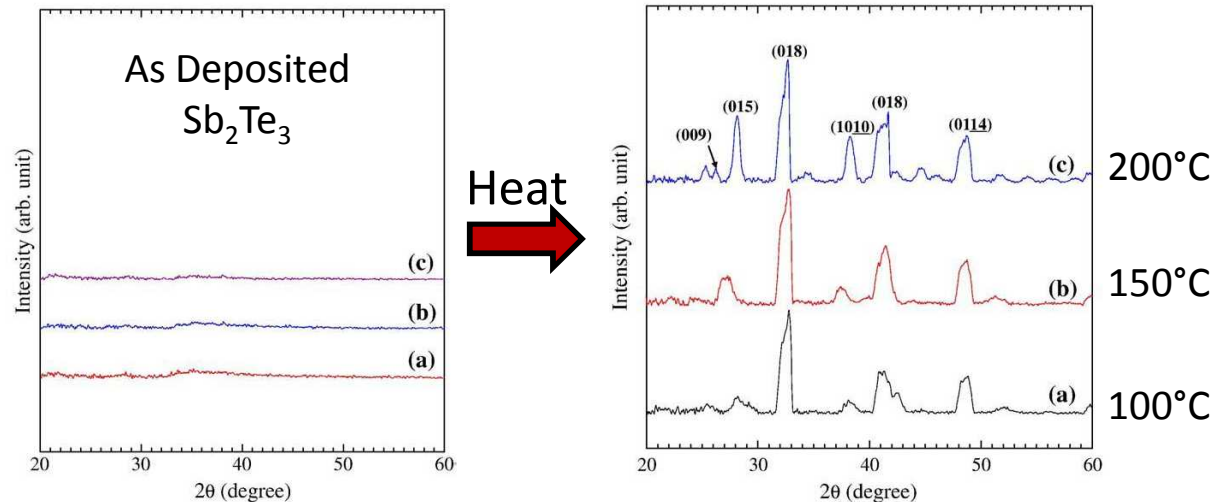
**Low cost**

**Room temperature**

**Scalable**

**Rapid**

*Post-synthesis processing has generally been required to achieve high quality, crystalline deposits*

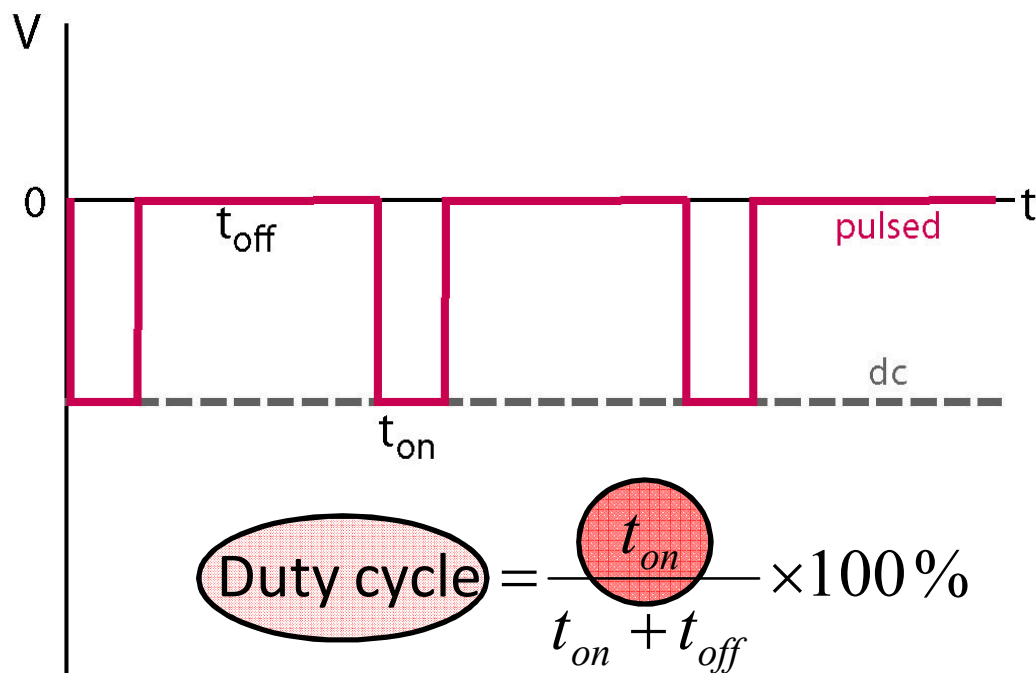


M-Y. Kim & T-S. Oh, *Thin Solid Films*, **518**, 6550-6553, 2010.

- Pulsed plating can improve **morphology**, **density**, and **crystallinity**

*How do the pulsed electrodeposition parameters affect the crystallinity, microstructure, and texture in  $\text{Sb}_2\text{Te}_3$  thin films?*

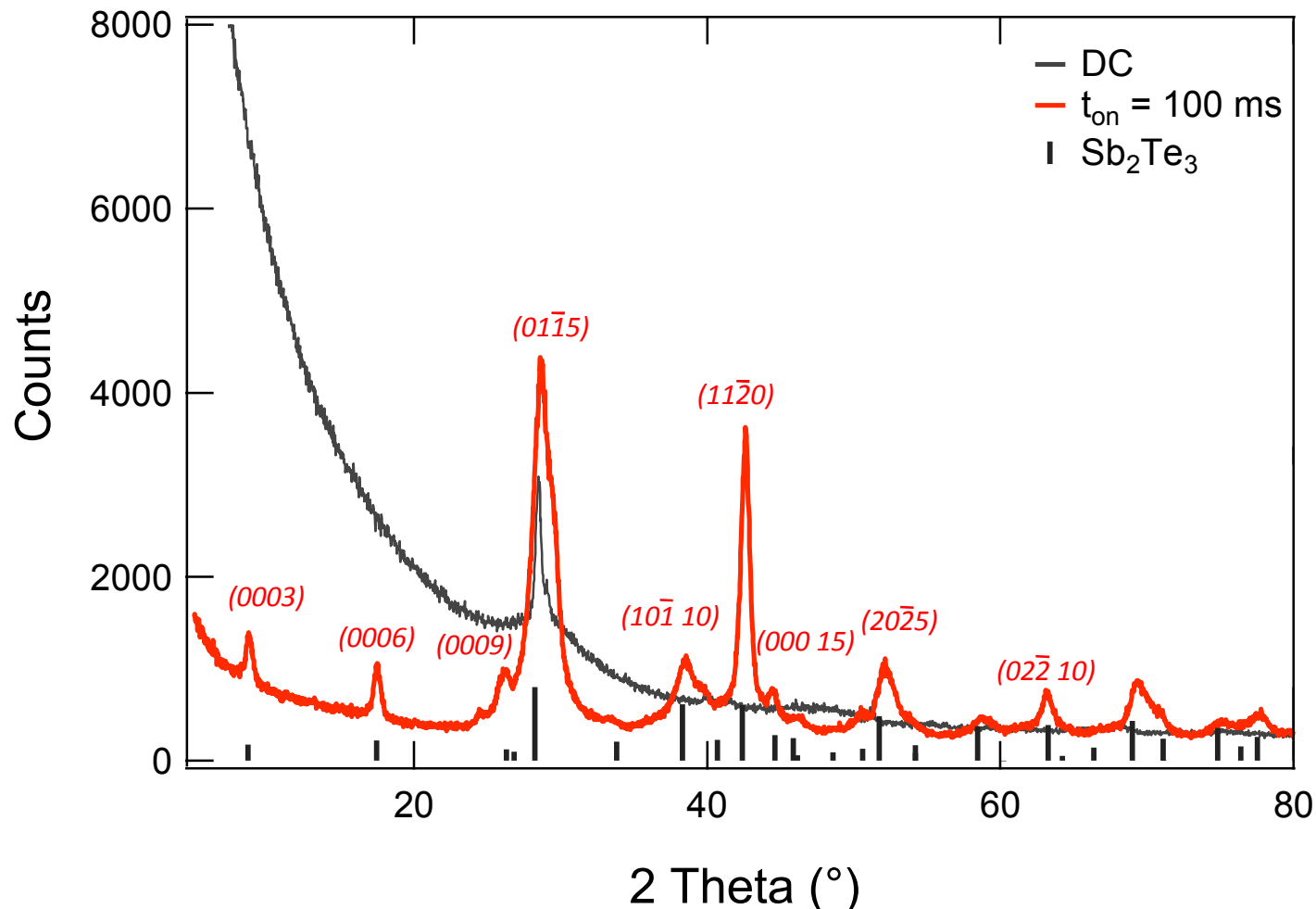
# Electrodeposition Conditions



Electroplating Conditions	
Bath	$\text{HNO}_3 + \text{C}_4\text{H}_6\text{O}_6 + \text{SbO}^+ + \text{HTeO}_2^+$
$t_{on}$ for pulsed plating	25 ms – 100 ms
Duty cycle	2 % -100 % (dc plating)
Temperature	$\approx 20^\circ\text{C}$
Deposition Potential	-0.3 V

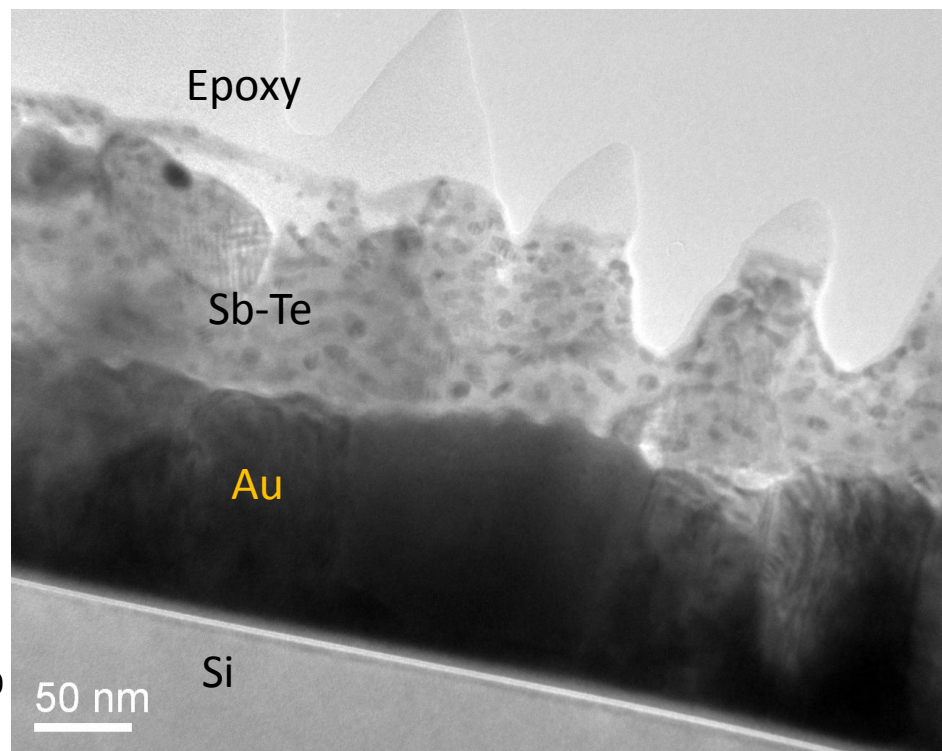
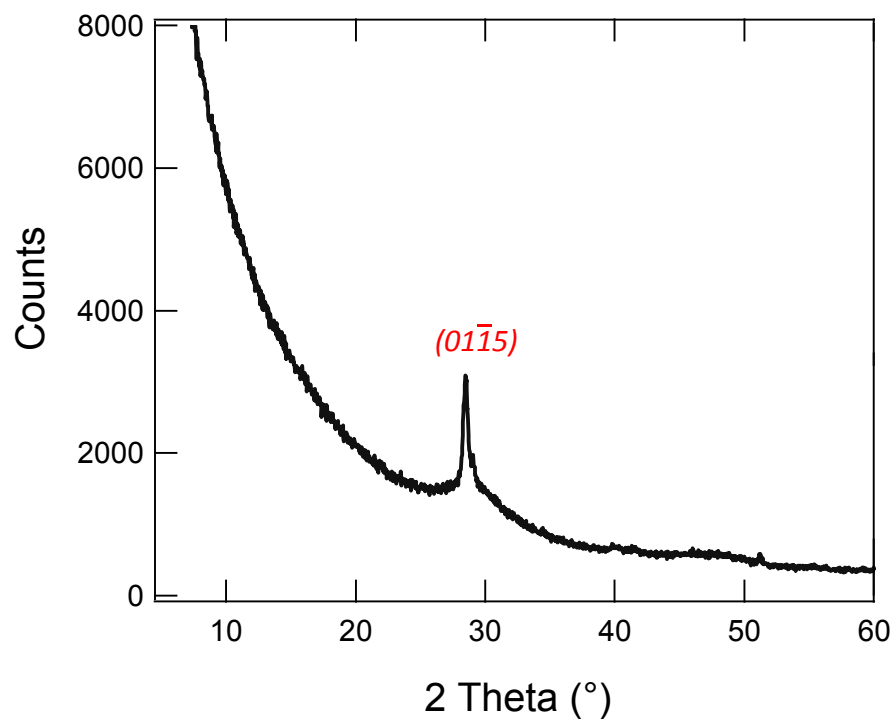
Working electrode (substrate): Au/Ti/Si; Reference electrode: Ag/AgCl; Counter electrode: Pt

# Continuous vs. pulsed electrodeposition of $\text{Sb}_2\text{Te}_3$

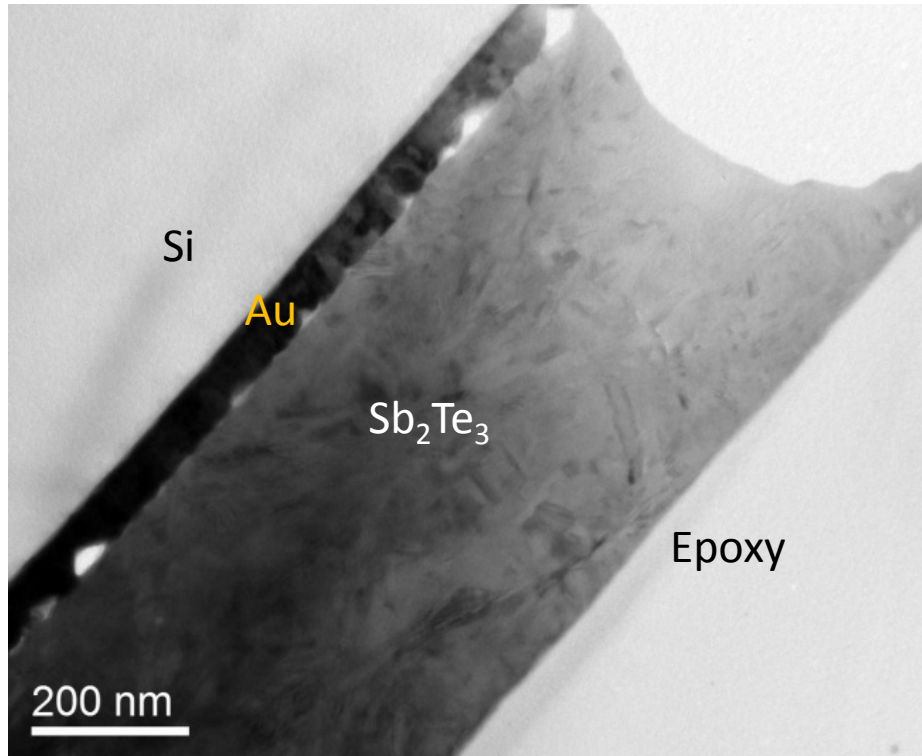
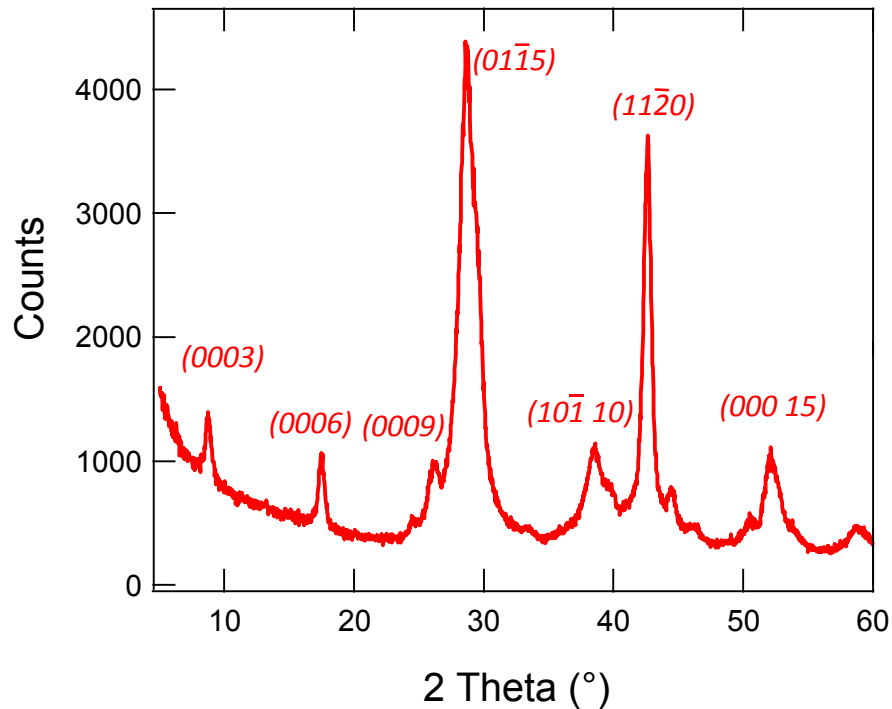


*Films deposited using pulsed plating are crystalline and consistent with the  $\text{Sb}_2\text{Te}_3$  tetradymite crystal structure*

# Continuous (dc) electrodeposition of $\text{Sb}_2\text{Te}_3$



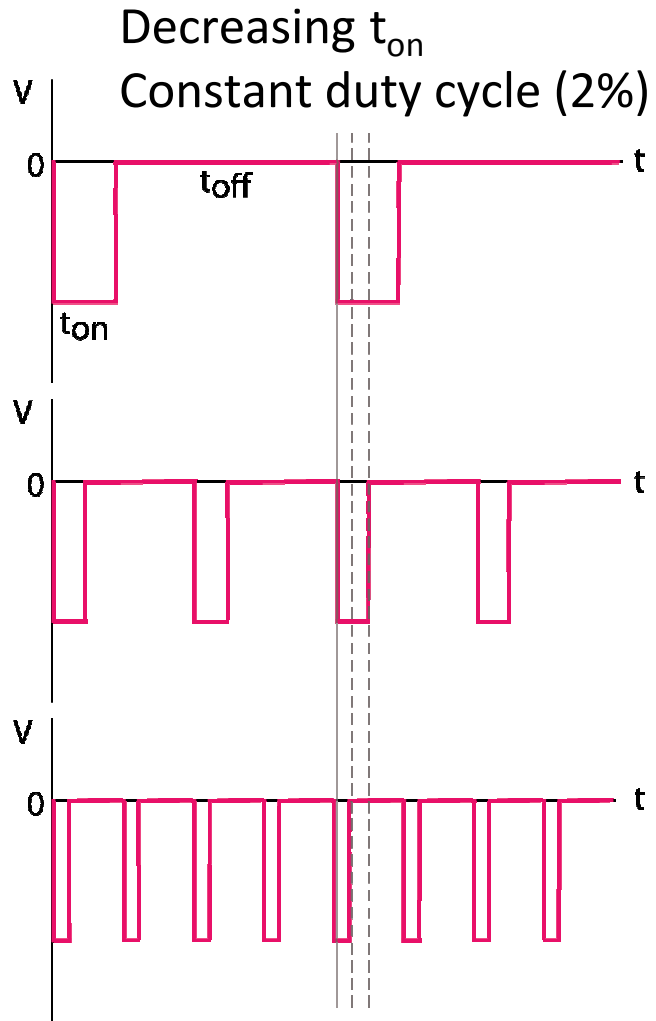
# Pulsed electrodeposition of $\text{Sb}_2\text{Te}_3$



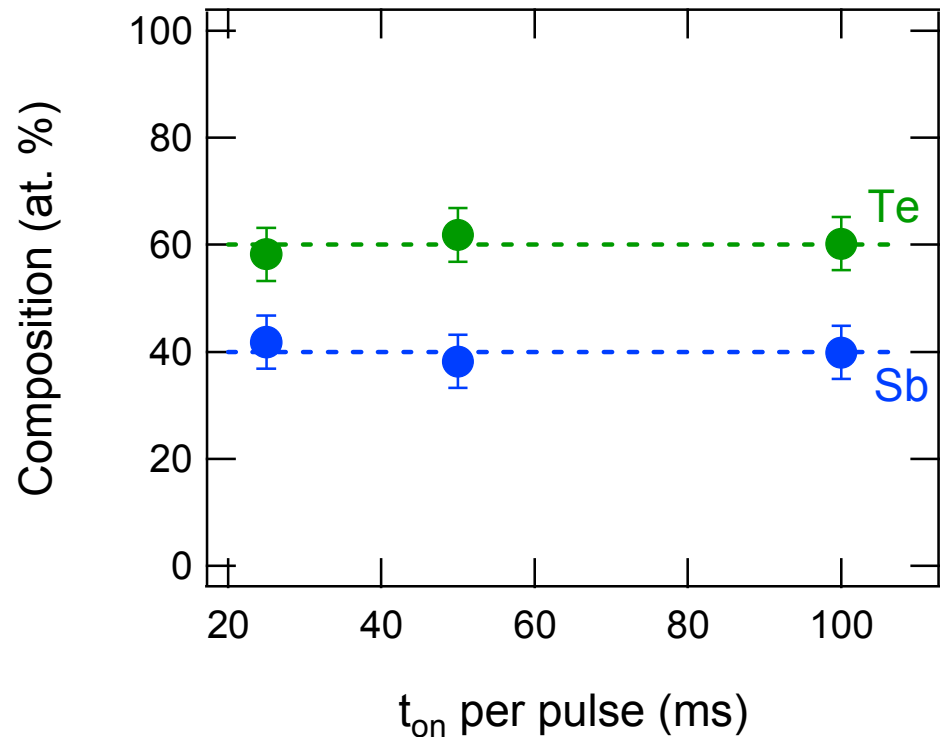
Duty cycle = 2 %,  $t_{\text{on}} = 100$  ms

- Polycrystalline
- Stoichiometric  $\text{Sb}_2\text{Te}_3$
- Uniform composition throughout thickness

# How does the composition vary with $t_{on}$ ?



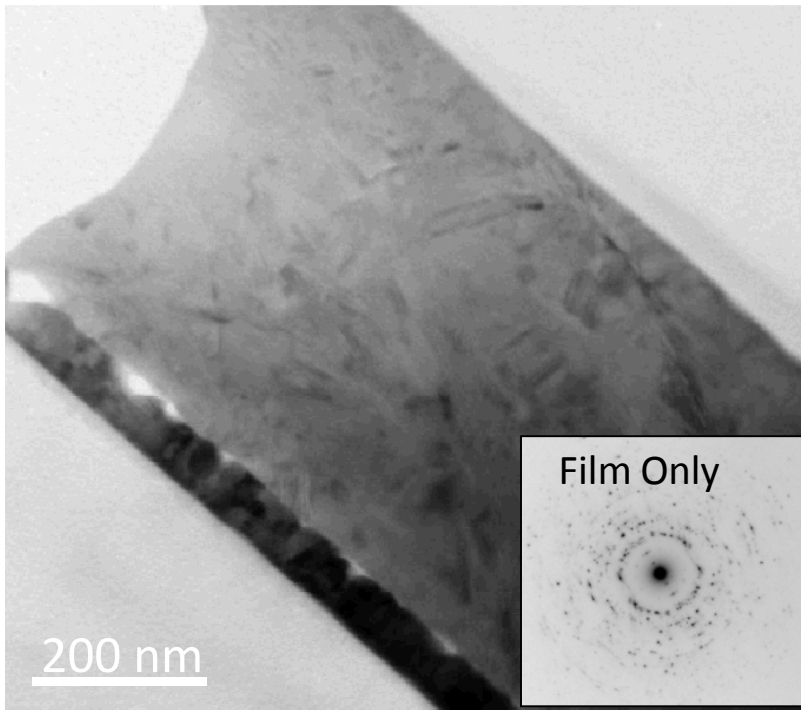
$$\text{Duty cycle} = \frac{t_{on}}{t_{on} + t_{off}} \times 100\%$$



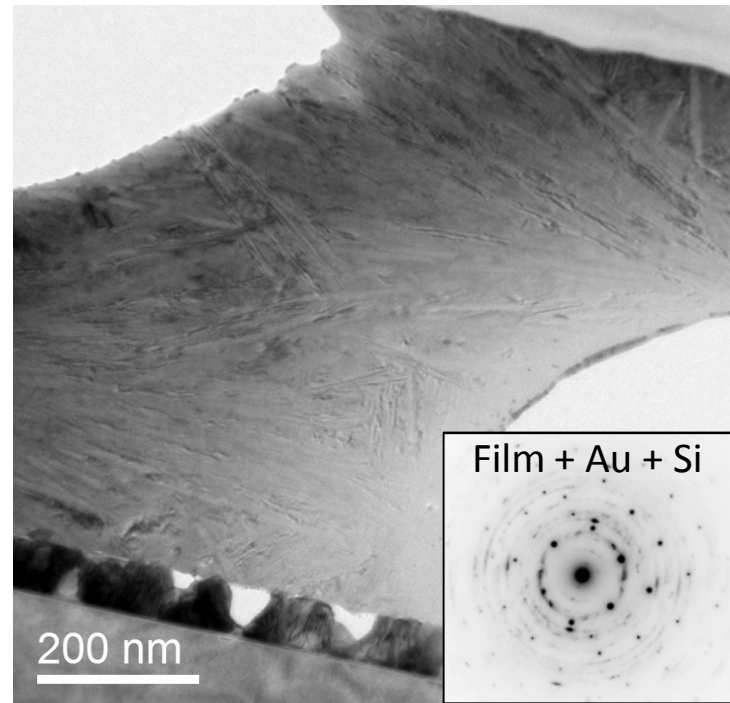
*Composition is uniform and stoichiometric for pulse on times of 25-100 ms*

# How does the microstructure vary with $t_{on}$ ?

$t_{on} = 100$  ms



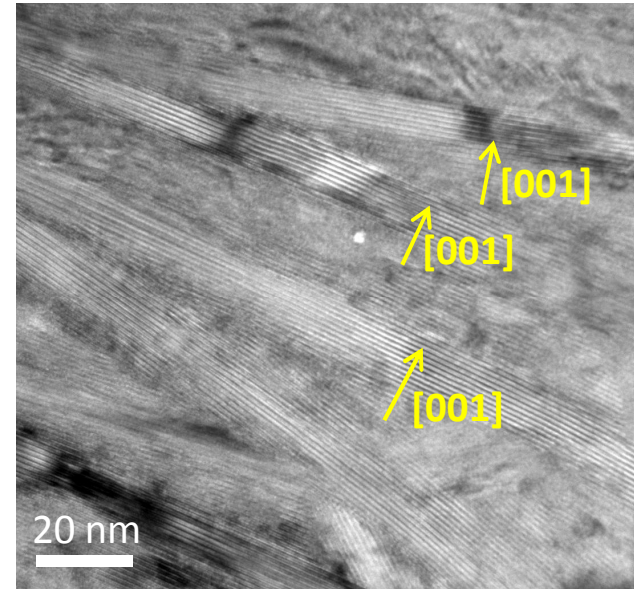
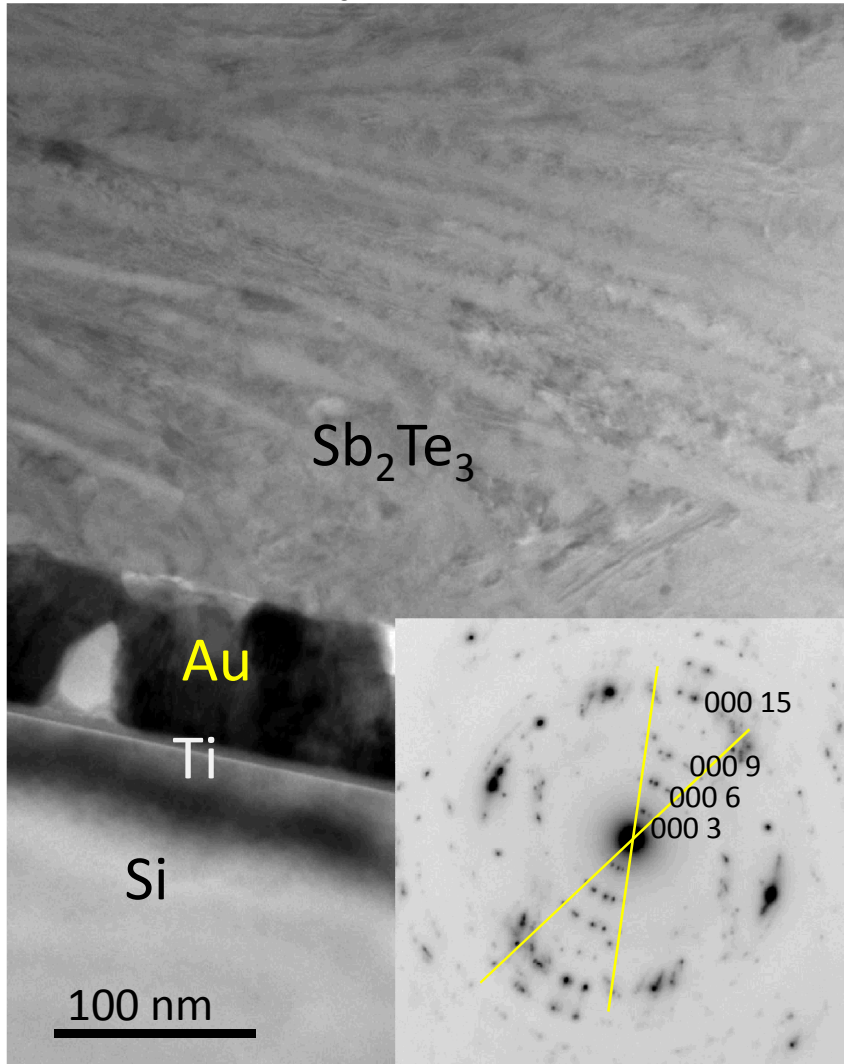
$t_{on} = 50$  ms





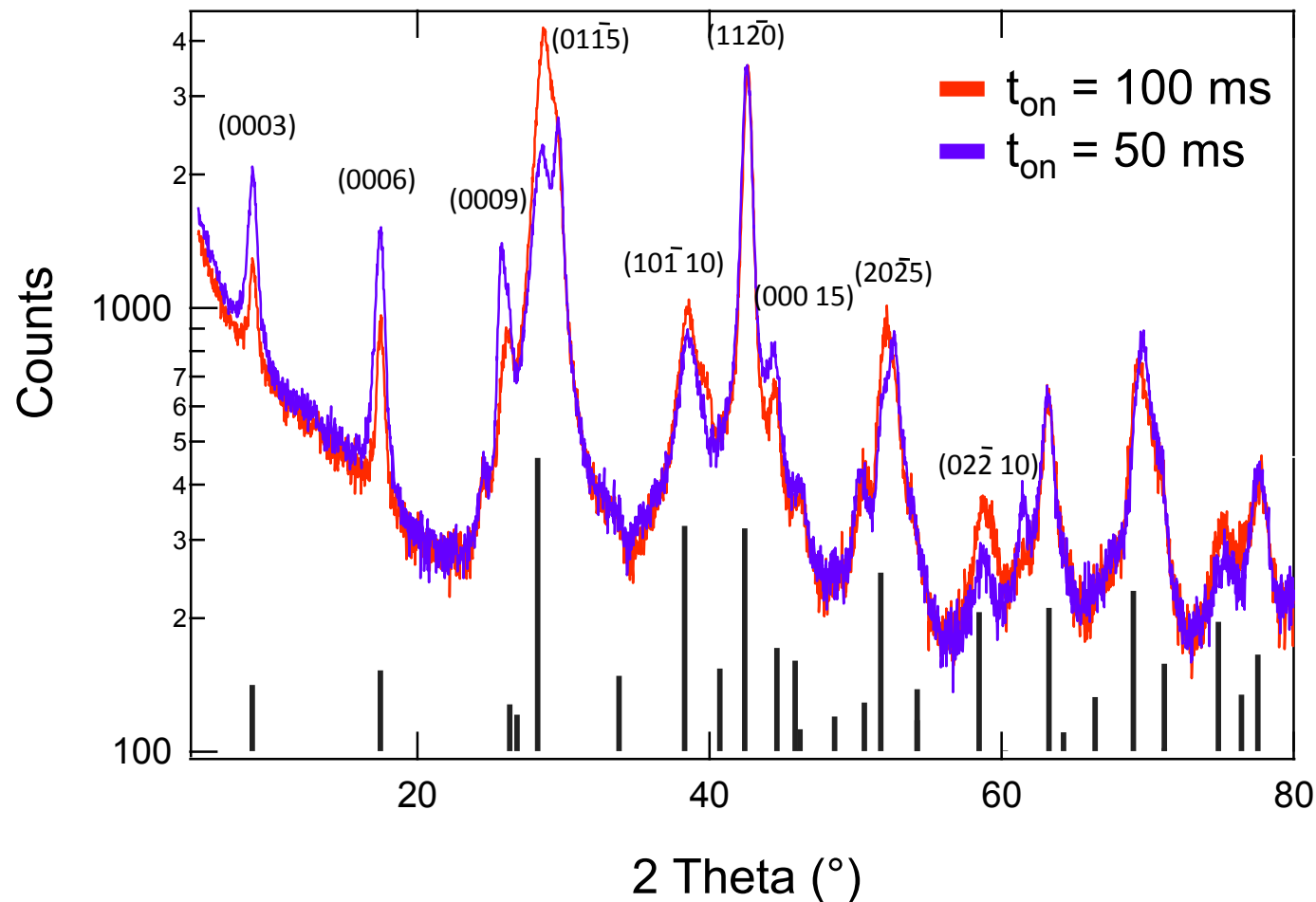
# Lamellar microstructure observed at short pulse times

Sb-Te thin film:  $t_{\text{on}} = 50$  ms, duty cycle = 2 %

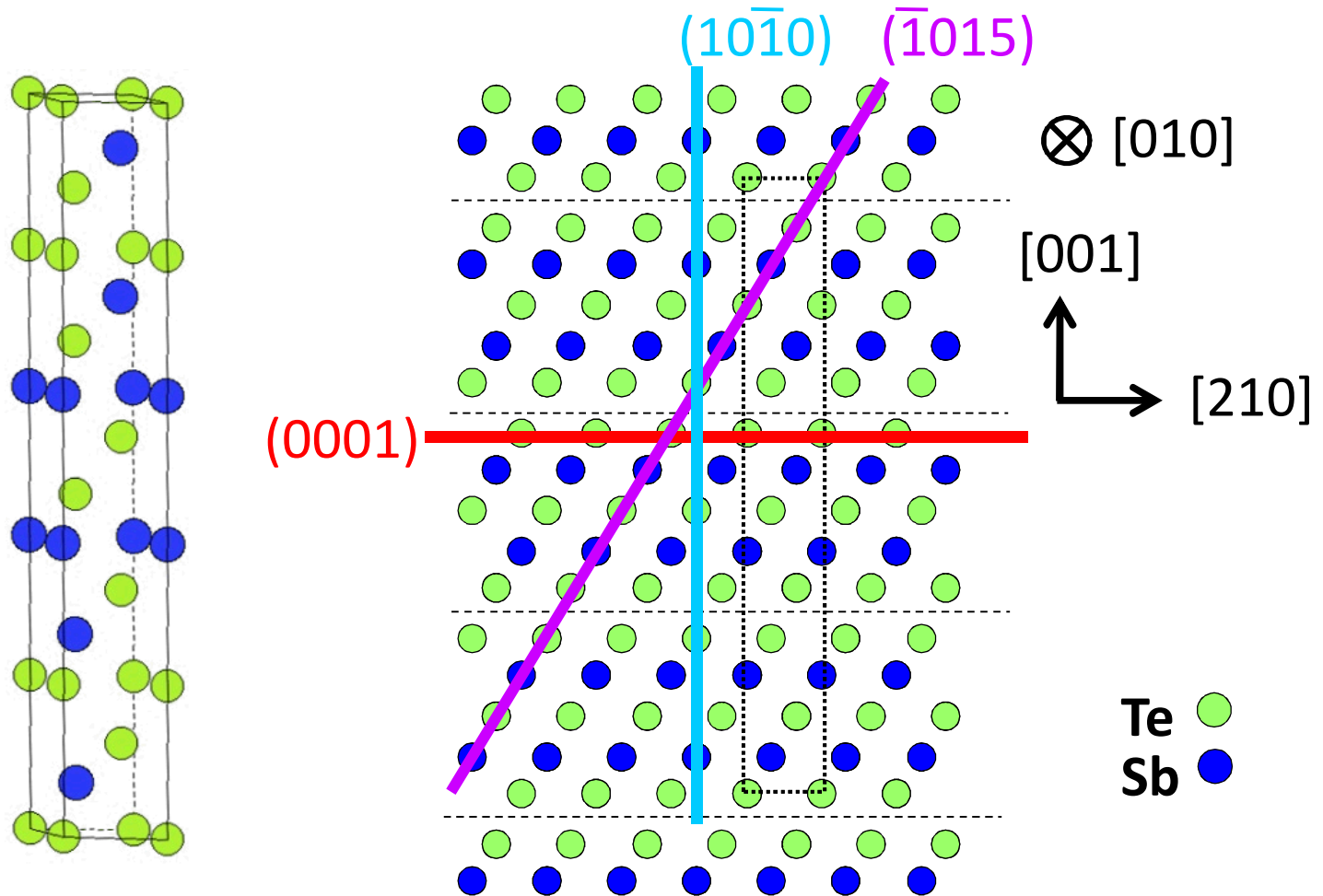


- For  $t_{\text{on}} \leq 50$  ms, lamellar features that are nearly parallel (within  $20^\circ$ ) to the substrate are commonly observed
  - Grains are up to 750 nm x 40 nm thick
  - Many pulses (10 – 40) are necessary to deposit the material in one lamellae
- These grains are oriented with the  $\text{Sb}_2\text{Te}_3$  (000/) planes parallel to the long axis (the [001] direction parallel to the short axis)

# Effect of decreasing pulse on time: structure & texture



# Crystallography of $\text{Sb}_2\text{Te}_3$

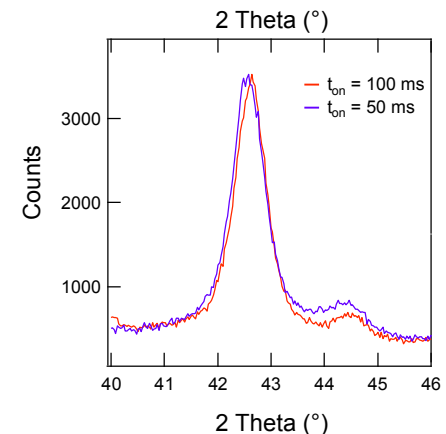
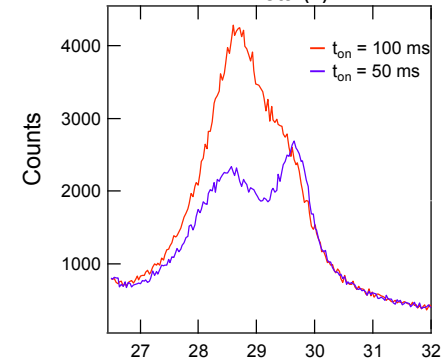
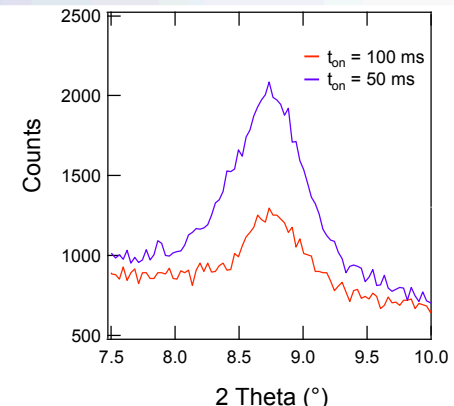
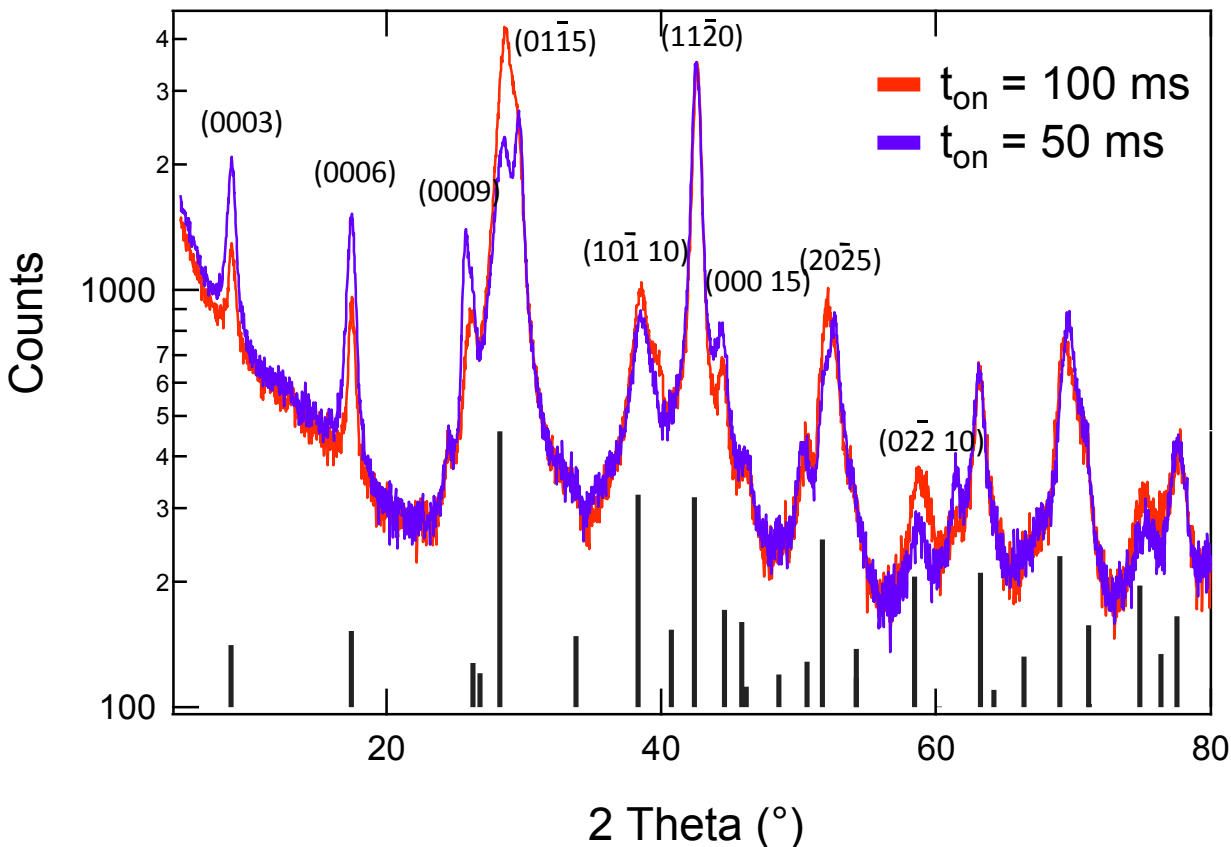


## Tetradymite

Hexagonal unit cell:

$$a = b = 4.264 \text{ \AA}, c = 30.458 \text{ \AA}$$

# Changes in preferred orientation observed with decreasing $t_{on}$



- Deposition on (000/) planes is favored over (011̄5) planes at shorter pulse times

# Summary

- $\text{Sb}_2\text{Te}_3$  thin films were grown by a pulsed potentiostatic electrodeposition process
  - For short pulse times, electron diffraction and XRD indicates that the films are somewhat textured with the (000/) planes nearly parallel to the substrate when deposited at short pulse times
  - A unique microstructure with lamellar grains that measure 10-40 nm thick by up to 750 nm long present in films deposited at short pulse times (< 50 ms) was observed. Within these grains, the  $\text{Sb}_2\text{Te}_3$  (000/) planes are parallel to the long axis giving the overall film a (000/) texture which is not observed at longer pulse times
    - Short pulses followed by off time may allow for more atoms to be in more stable configurations
    - (000/) are low index planes with reasonable high atomic density and are likely stable
- *The ability to directly grow crystalline  $\text{Sb}_2\text{Te}_3$  thin films by electrodeposition at room temperature without the need for high temperature anneals may lead to new opportunities for nanostructure fabrication where diffusion and coarsening must be suppressed*

# Acknowledgements

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Michelle Hekmaty

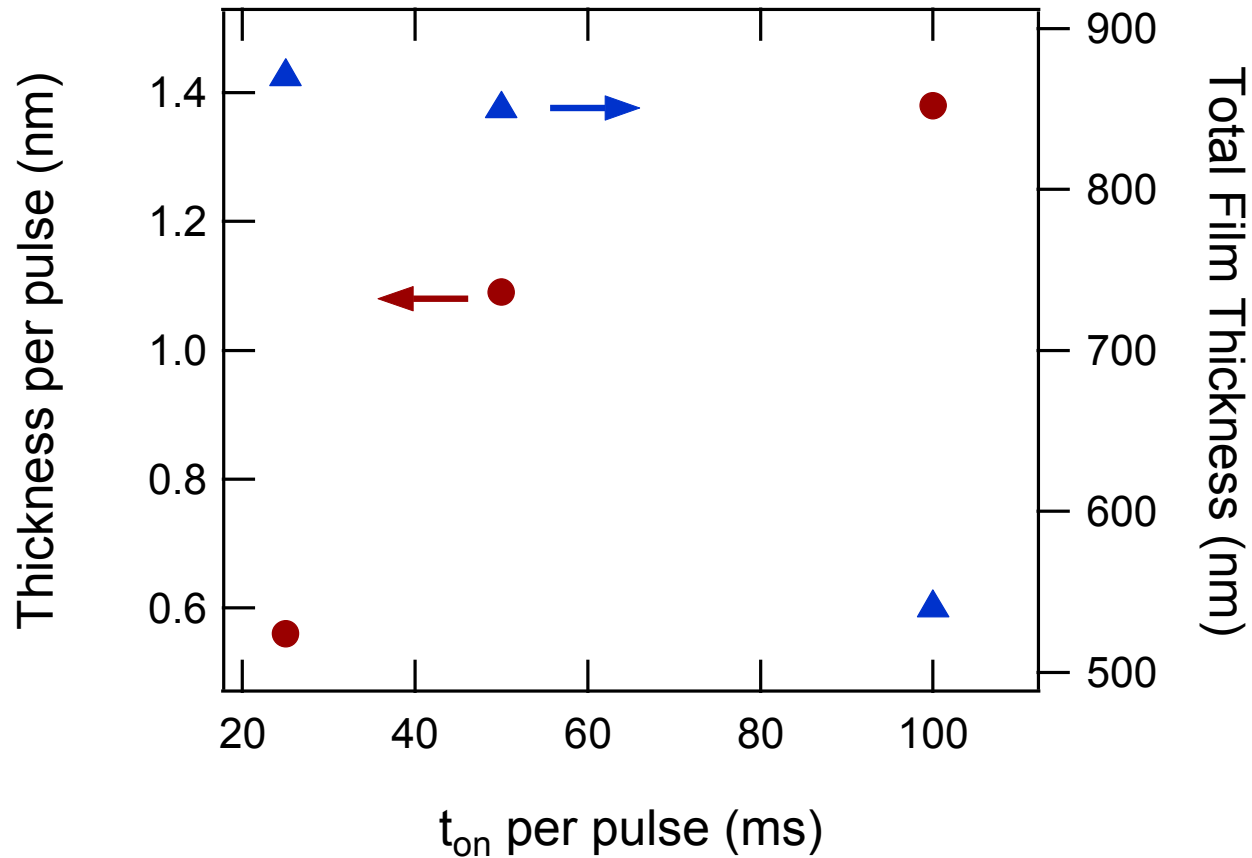
Mark Homer

David Robinson

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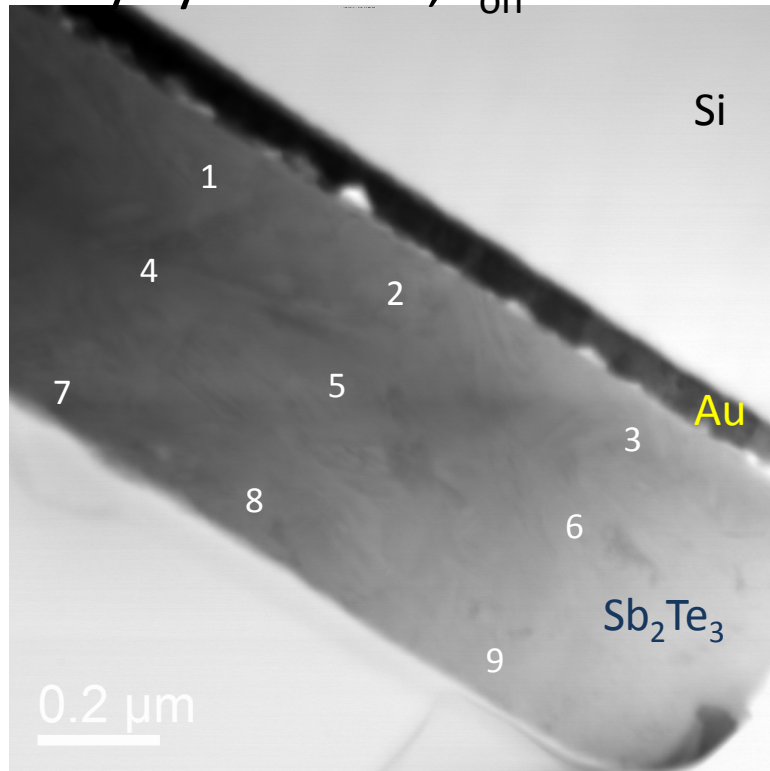


# How does the film thickness vary with $t_{on}$ ?



# Compositional uniformity in pulse plated $\text{Sb}_2\text{Te}_3$

Duty cycle = 2 %,  $t_{\text{on}} = 100$  ms

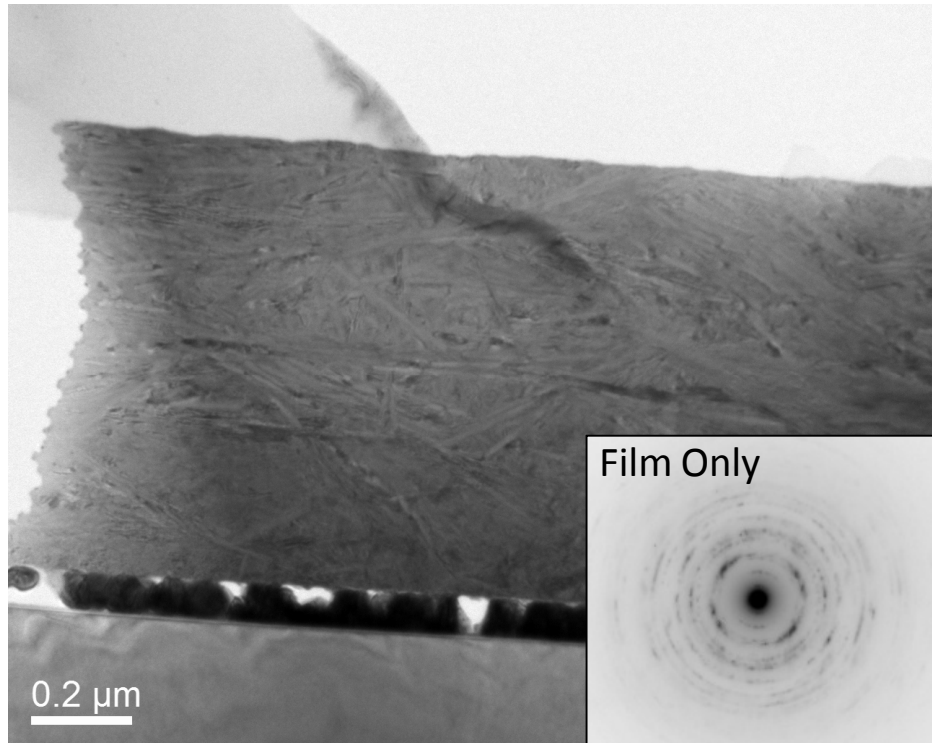


## Semi-Quantitative STEM/EDS Analysis

	at. % Sb	at. % Te
1	40.0	60.0
2	38.1	61.9
3	38.3	61.7
4	39.6	60.4
5	39.5	60.5
6	37.8	62.2
7	35.8	64.2
8	39.7	60.4
9	41.7	58.3

*The composition is uniform through the thickness of the pulse electrodeposited film  
at 2% duty cycle*





$$t_{\text{on}} = 25\ \text{ms}$$

# Hypotheses for microstructural change

Shorter pulse times lead to fewer atoms deposited per pulse

Off times allow atoms in less stable positions to desorb into solution

(001) Lamellae may be a stable configuration

- Van der Waals bonding between 5-layer packets
- Low index, high density planes generally stable

