

Characterization of electrodeposited nano-crystalline Sb_2Te_3 films

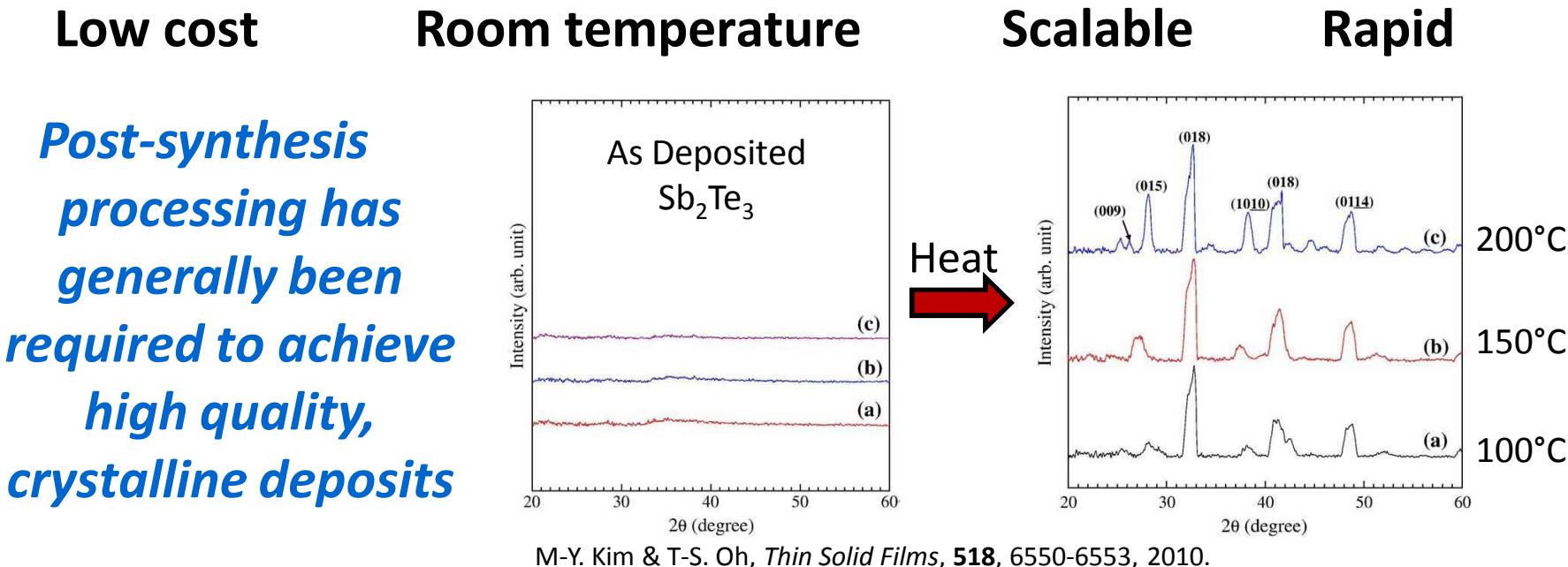
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Electrodeposition of Sb_2Te_3

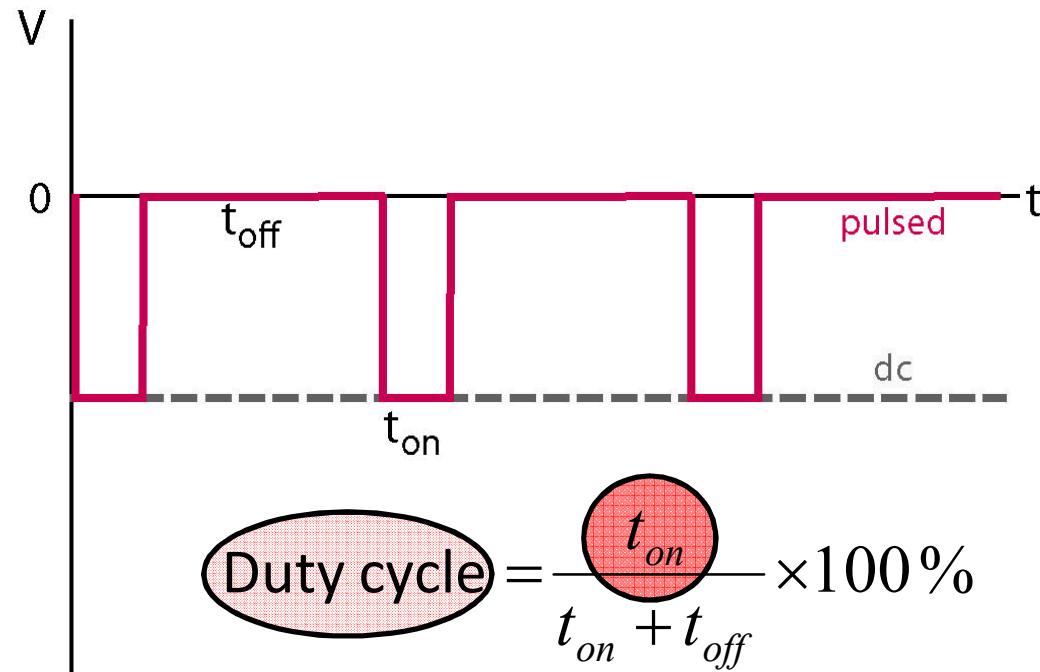
- Electrodeposition is promising for synthesis of thermoelectric materials such as Sb_2Te_3



- Pulsed plating can improve **morphology, density, and crystallinity**

How do the pulsed electrodeposition parameters affect the crystallinity, microstructure, and texture in Sb_2Te_3 thin films?

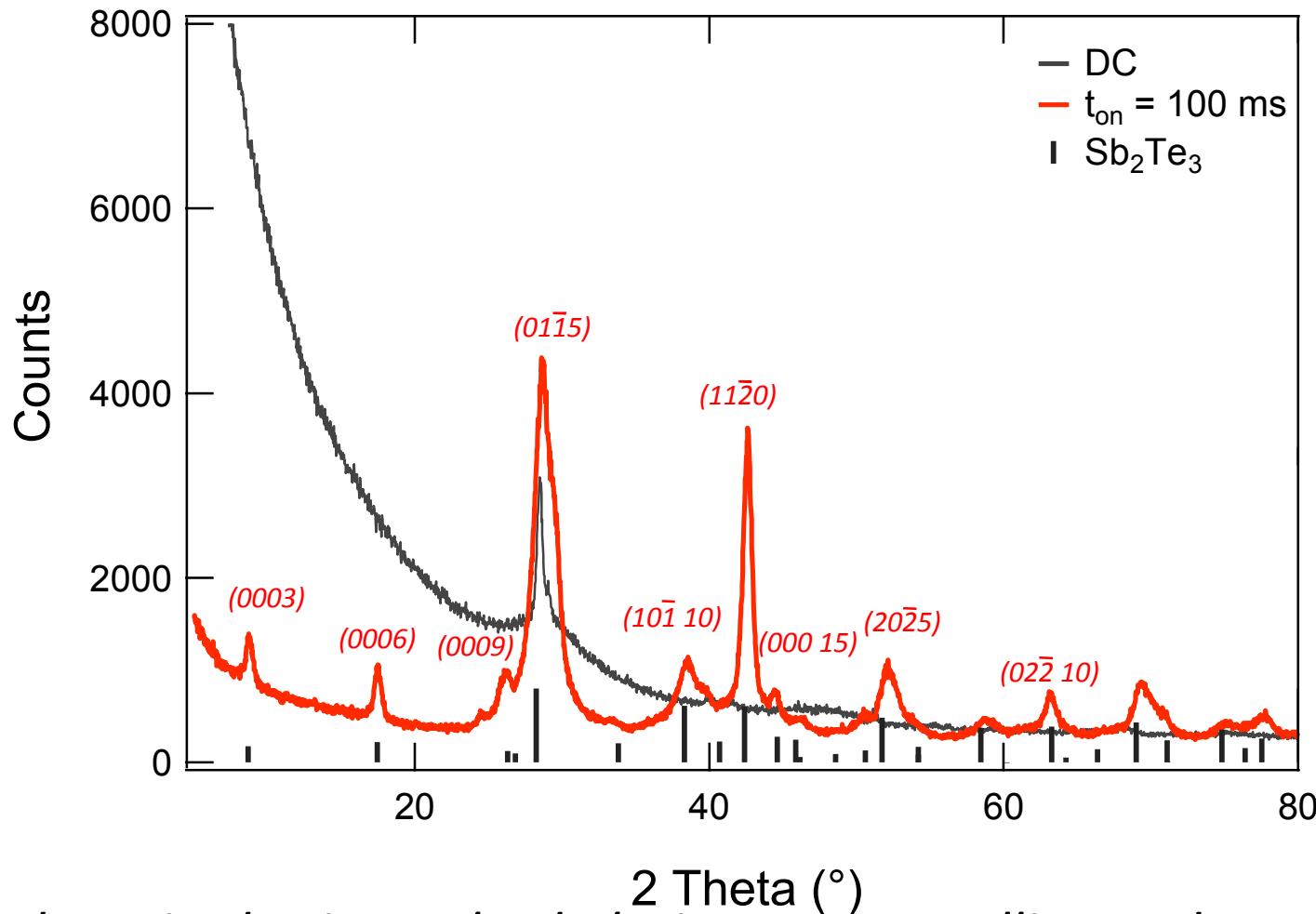
Electrodeposition Conditions



Electroplating Conditions	
Bath	$\text{HNO}_3 + \text{C}_4\text{H}_6\text{O}_6 + \text{SbO}^+ + \text{HTeO}_2^+$
t_{on} for pulsed plating	25 ms – 100 ms
Duty cycle	2 % -100 % (dc plating)
Temperature	$\approx 20^\circ\text{C}$
Deposition Potential	-0.3 V

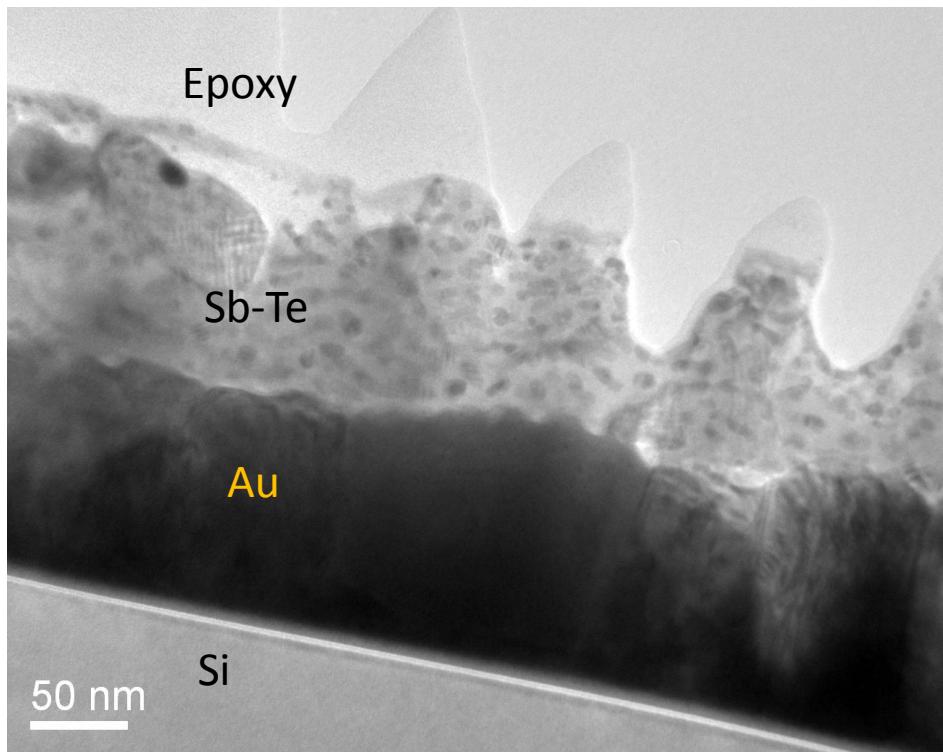
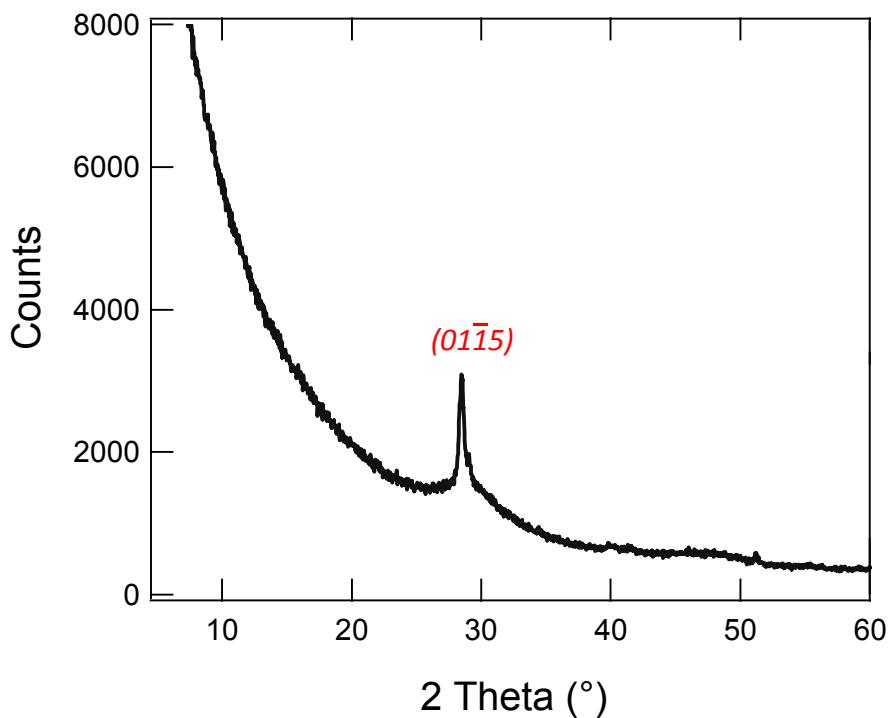
Working electrode (substrate): Au/Ti/Si; Reference electrode: Ag/AgCl; Counter electrode: Pt

Continuous vs. pulsed electrodeposition of Sb_2Te_3

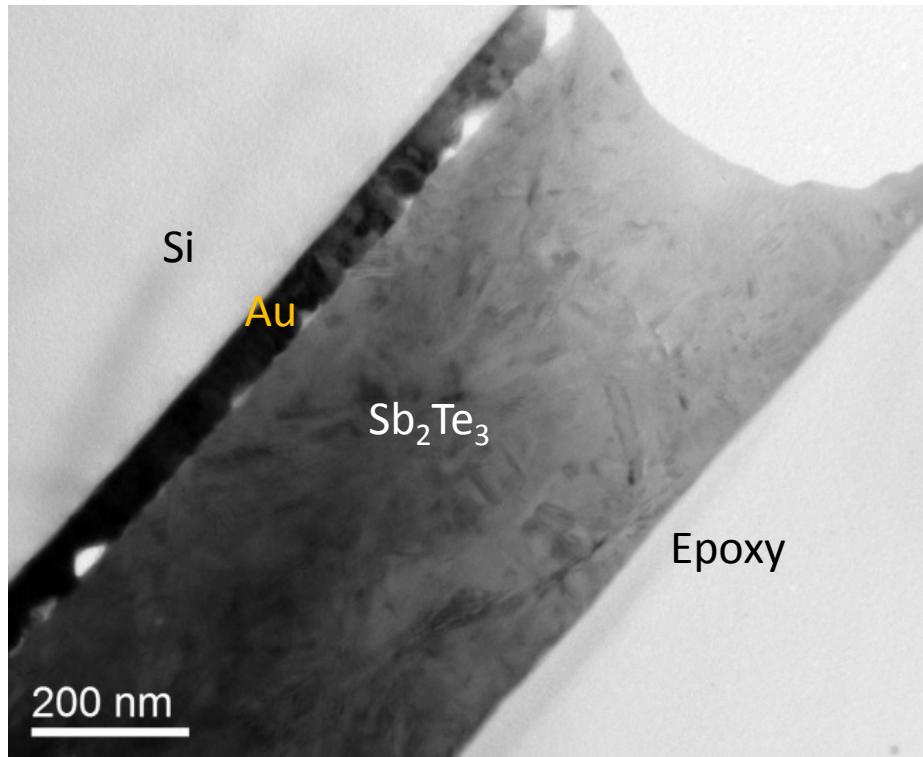
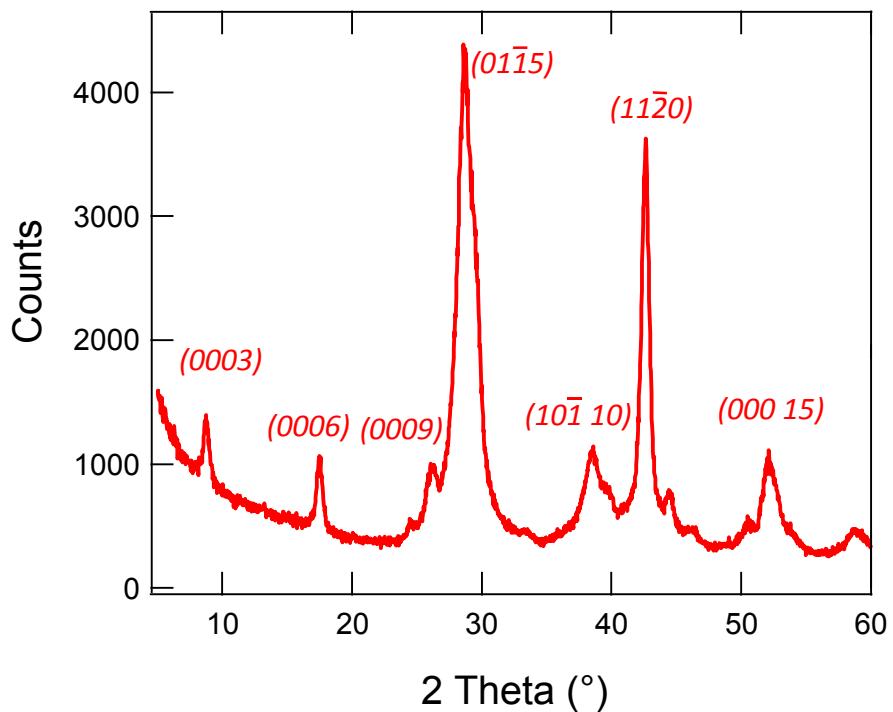


Films deposited using pulsed plating are crystalline and consistent with the Sb_2Te_3 tetradymite crystal structure

Continuous (dc) electrodeposition of Sb_2Te_3



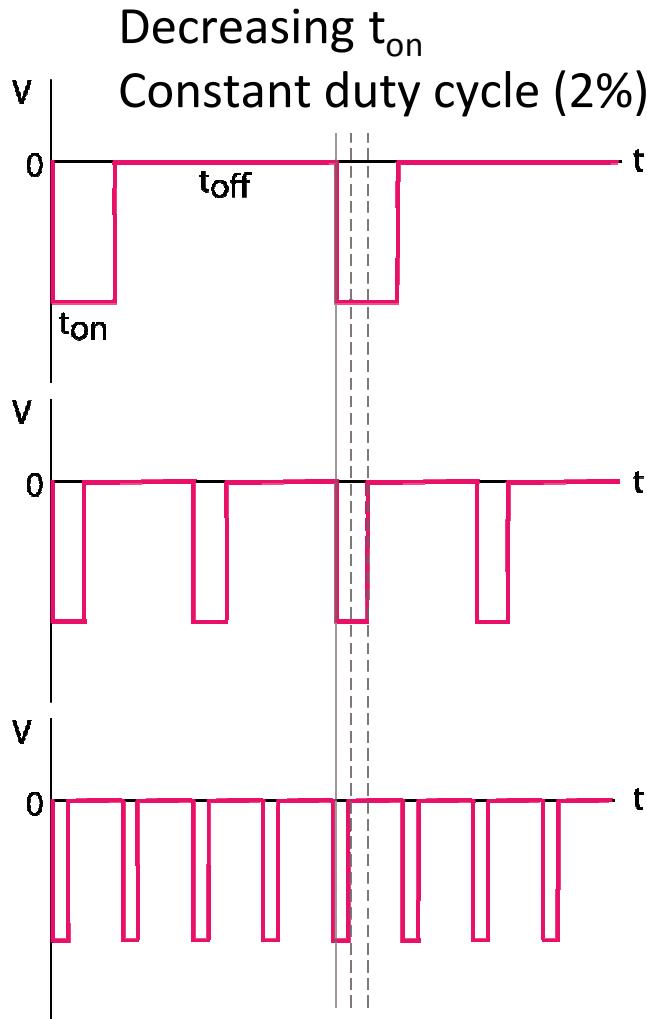
Pulsed electrodeposition of Sb_2Te_3



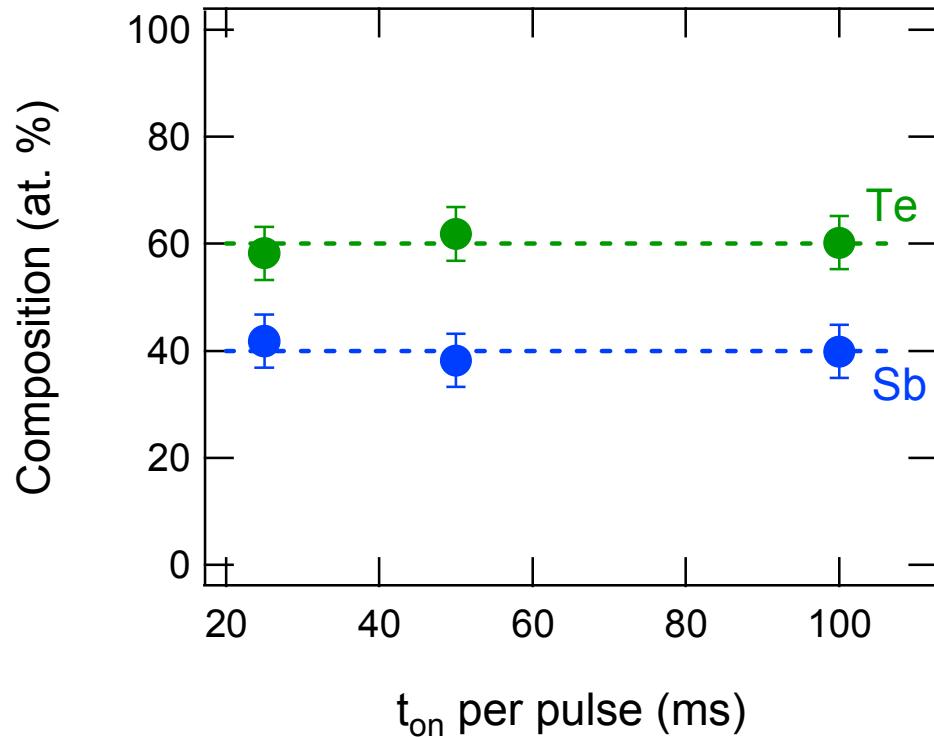
Duty cycle = 2 %, $t_{\text{on}} = 100$ ms

- Polycrystalline
- Stoichiometric Sb_2Te_3
- Uniform composition throughout thickness

How does the composition vary with t_{on} ?



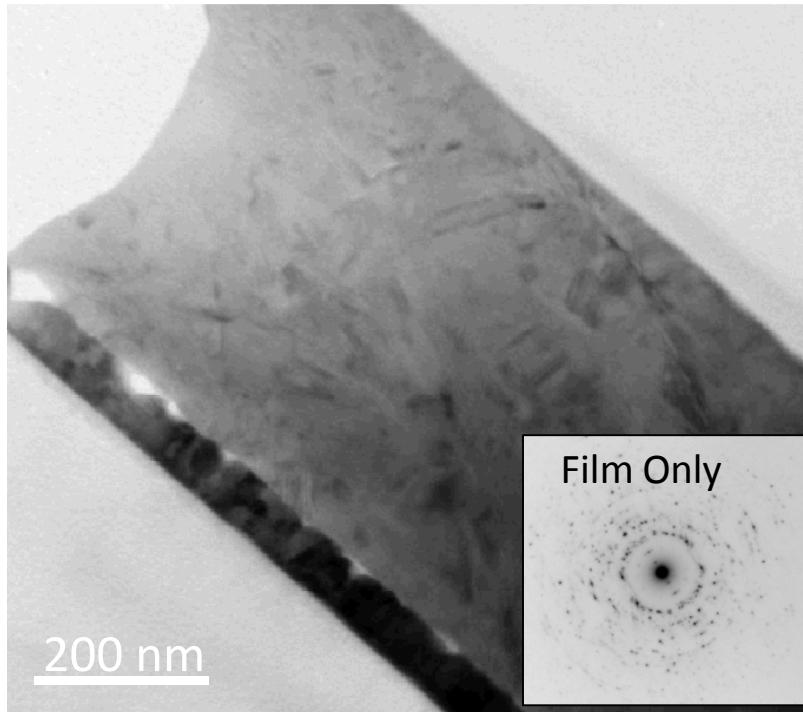
$$\text{Duty cycle} = \frac{t_{on}}{t_{on} + t_{off}} \times 100\%$$



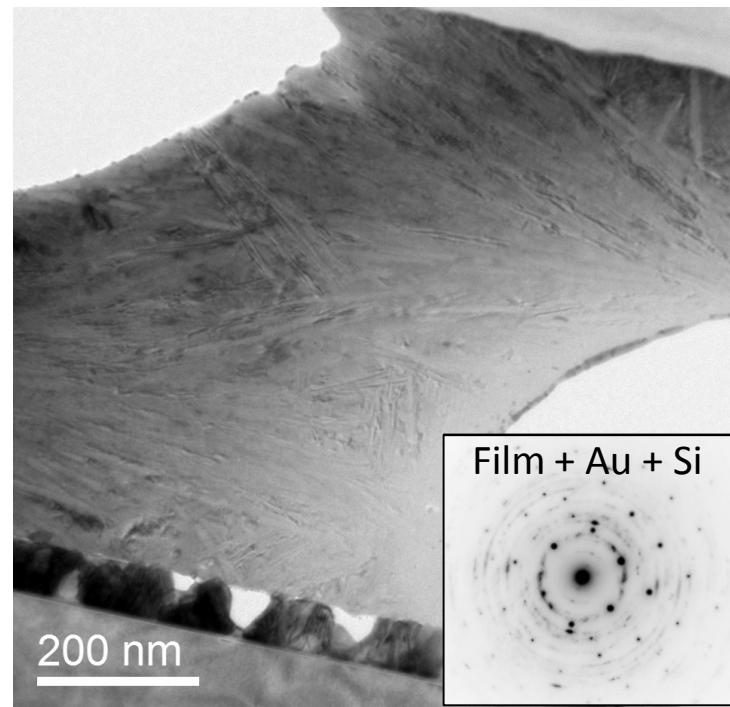
Composition is uniform and stoichiometric for pulse on times of 25-100 ms

How does the microstructure vary with t_{on} ?

$t_{on} = 100$ ms

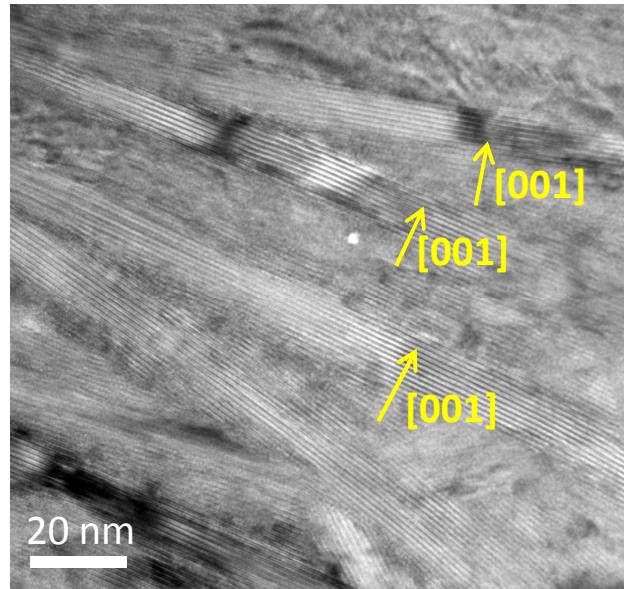
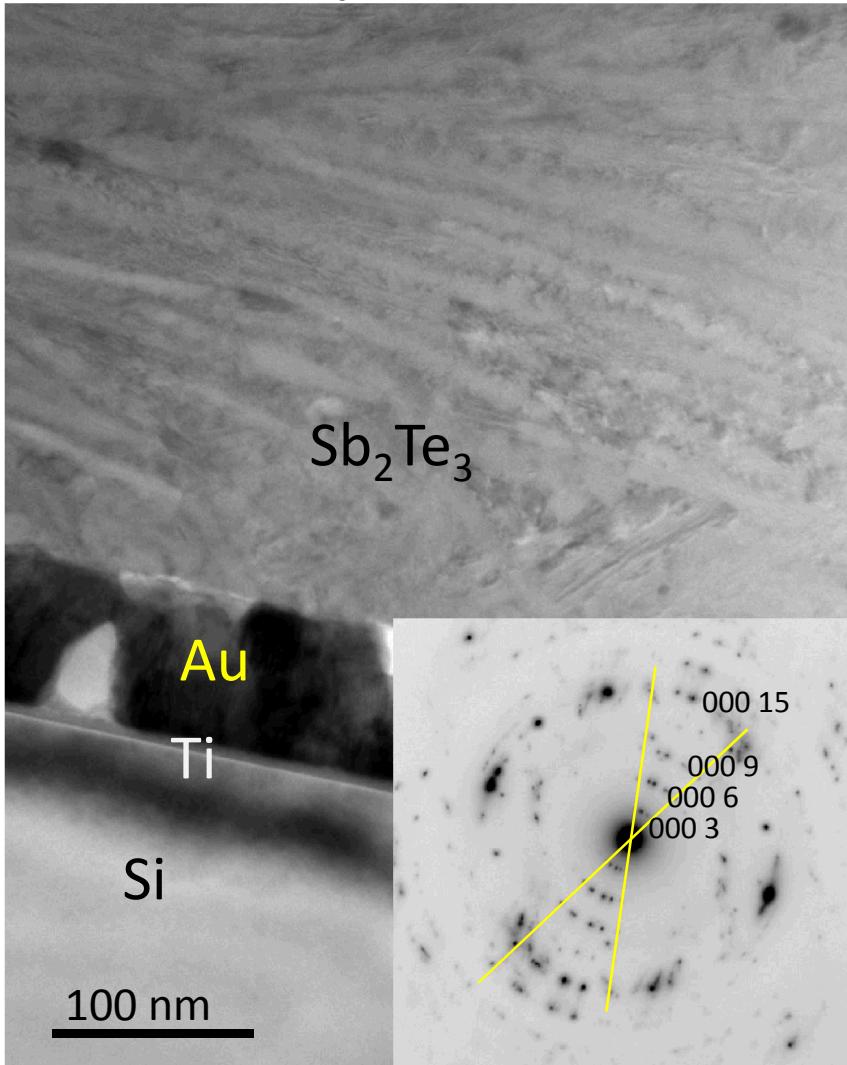


$t_{on} = 50$ ms



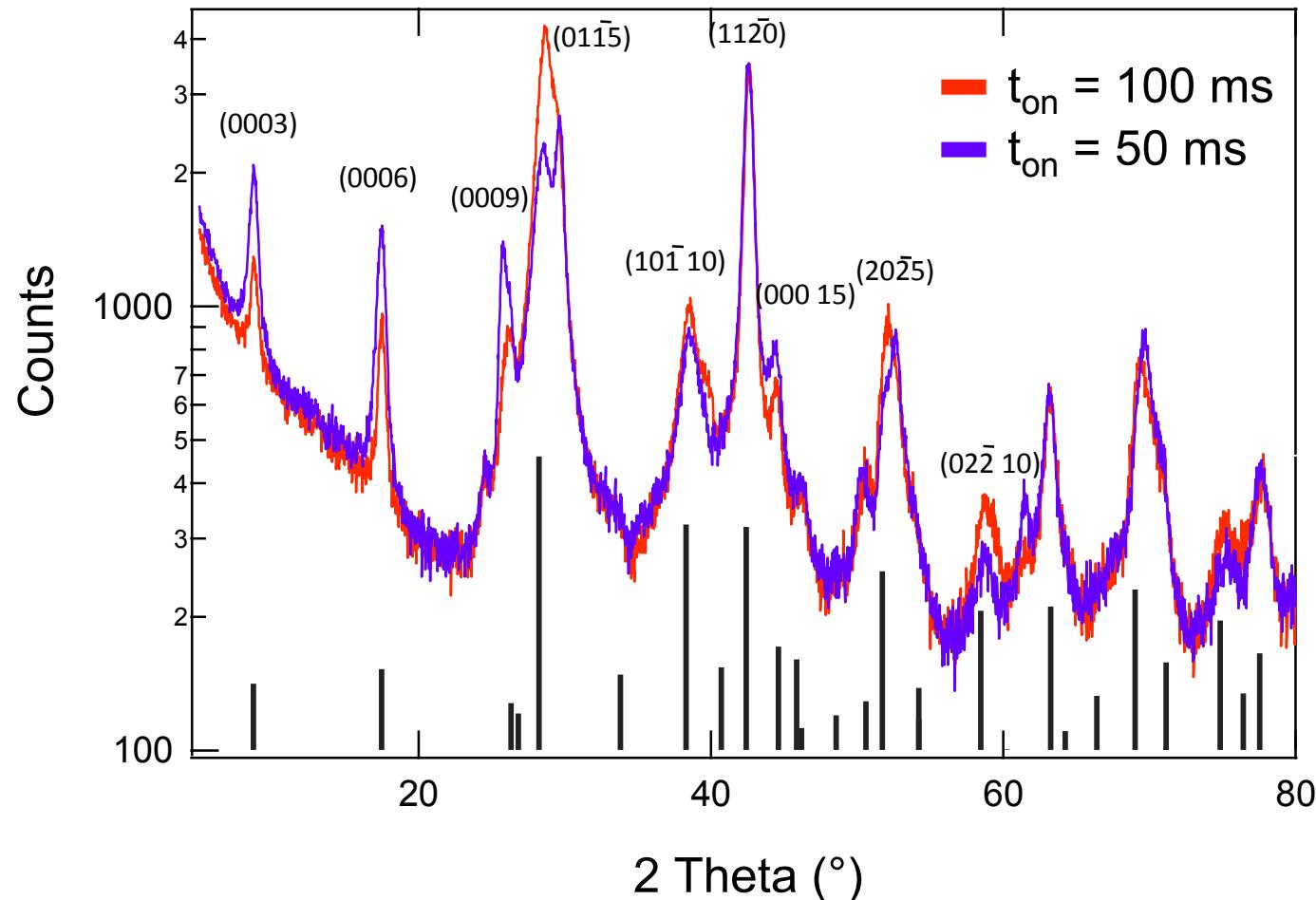
Lamellar microstructure observed at short pulse times

Sb-Te thin film: $t_{\text{on}} = 50 \text{ ms}$, duty cycle = 2 %

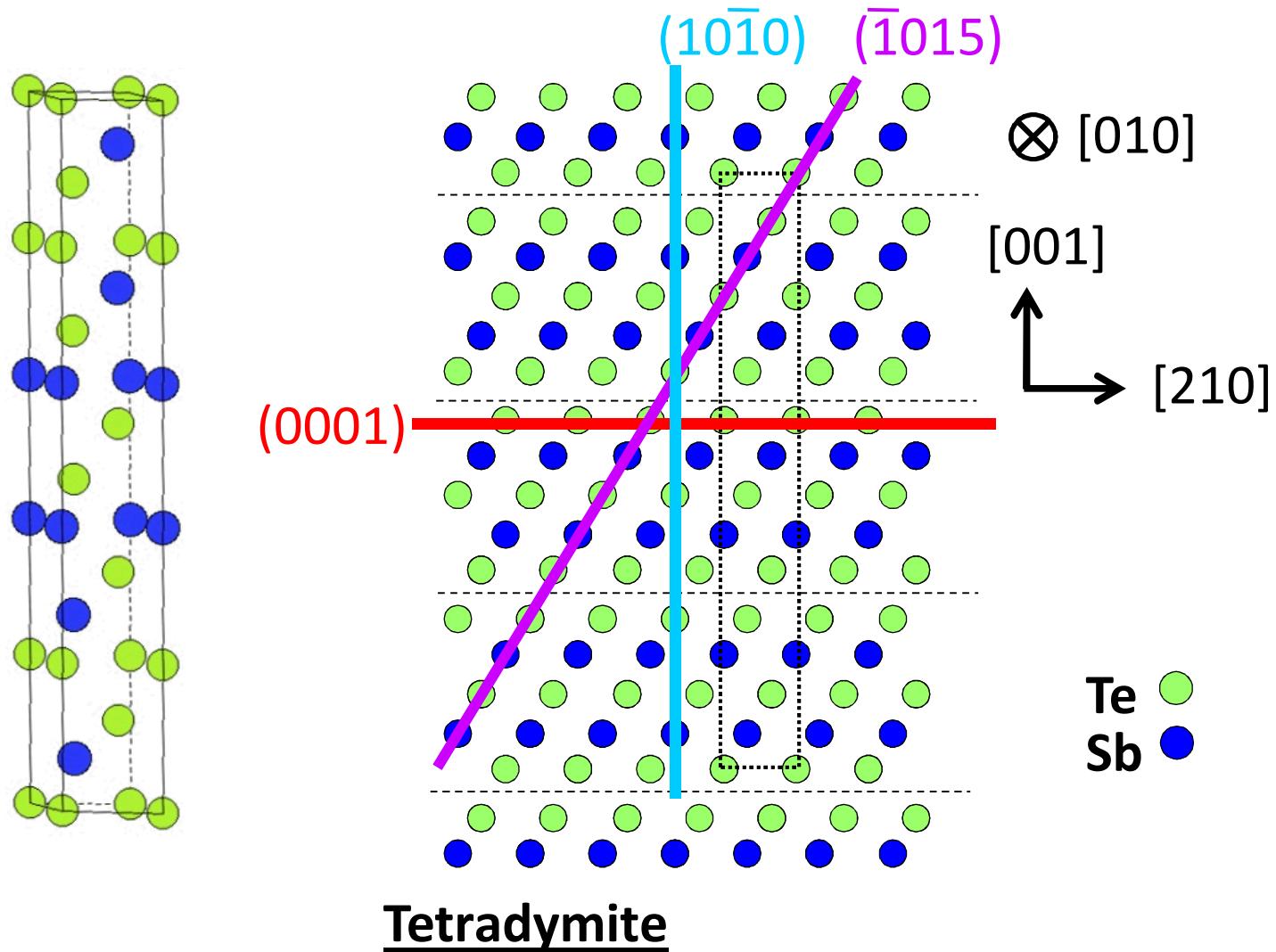


- For $t_{\text{on}} \leq 50 \text{ ms}$, lamellar features that are nearly parallel (within 20°) to the substrate are commonly observed
 - Grains are up to 750 nm x 40 nm thick
 - Many pulses (10 – 40) are necessary to deposit the material in one lamellae
- These grains are oriented with the Sb_2Te_3 (000 \bar{l}) planes parallel to the long axis (the [001] direction parallel to the short axis)

Effect of decreasing pulse on time: structure & texture



Crystallography of Sb_2Te_3

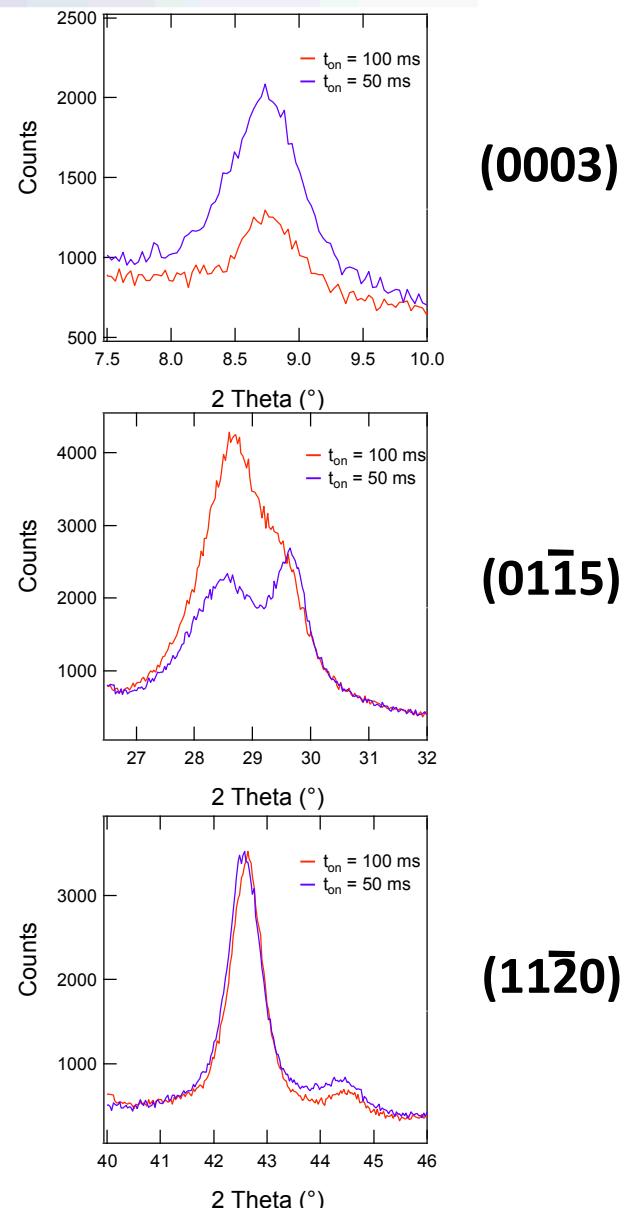
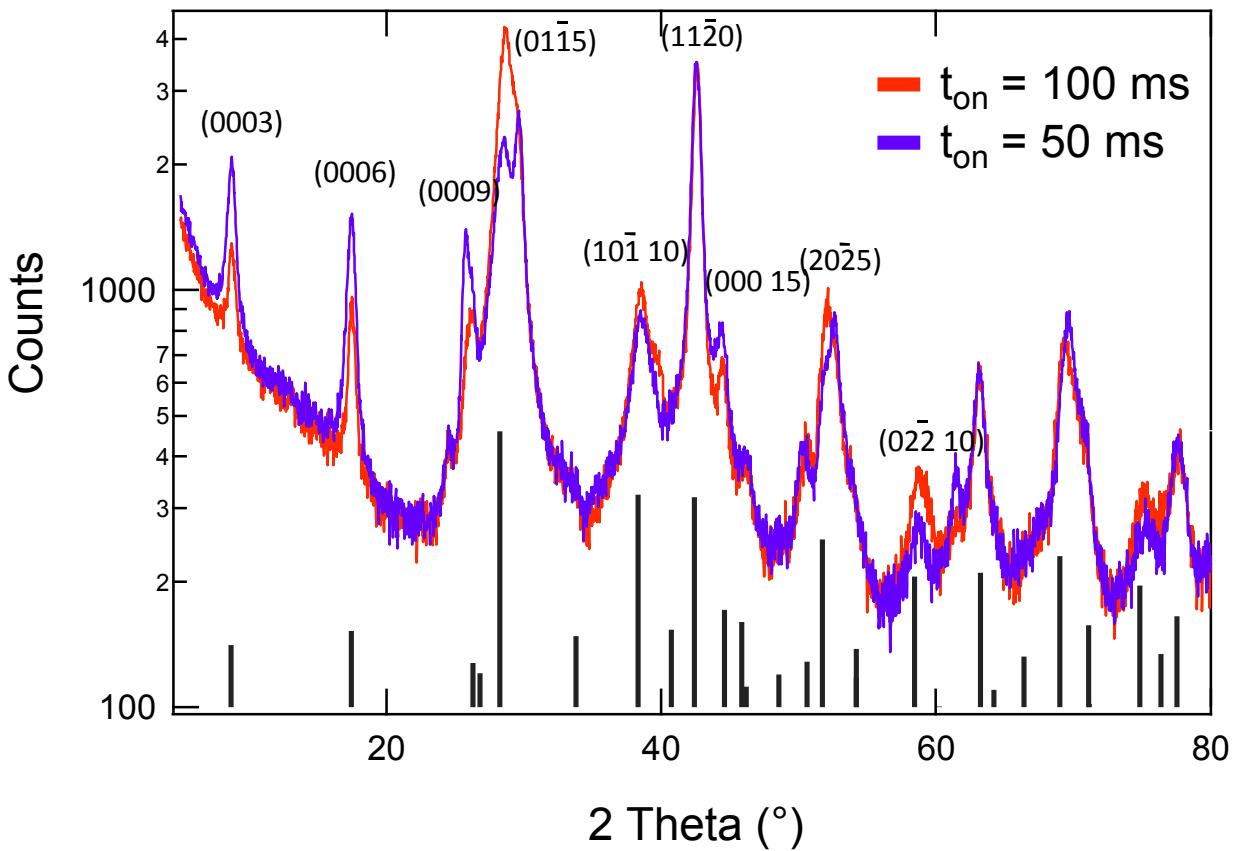


Tetradymite

Hexagonal unit cell:

$$a = b = 4.264 \text{ \AA}, c = 30.458 \text{ \AA}$$

Changes in preferred orientation observed with decreasing t_{on}



- Deposition on (000 \bar{l}) planes is favored over (011 $\bar{5}$) planes at shorter pulse times

Summary

- Sb_2Te_3 thin films were grown by a pulsed potentiostatic electrodeposition process
 - For short pulse times, electron diffraction and XRD indicates that the films are somewhat textured with the (000 \bar{l}) planes nearly parallel to the substrate when deposited at short pulse times
 - A unique microstructure with lamellar grains that measure 10-40 nm thick by up to 750 nm long present in films deposited at short pulse times (< 50 ms) was observed. Within these grains, the Sb_2Te_3 (000 \bar{l}) planes are parallel to the long axis giving the overall film a (000 \bar{l}) texture which is not observed at longer pulse times
 - Short pulses followed by off time may allow for more atoms to be in more stable configurations
 - (000 \bar{l}) are low index planes with reasonable high atomic density and are likely stable
- *The ability to directly grow crystalline Sb_2Te_3 thin films by electrodeposition at room temperature without the need for high temperature anneals may lead to new opportunities for nanostructure fabrication where diffusion and coarsening must be suppressed*

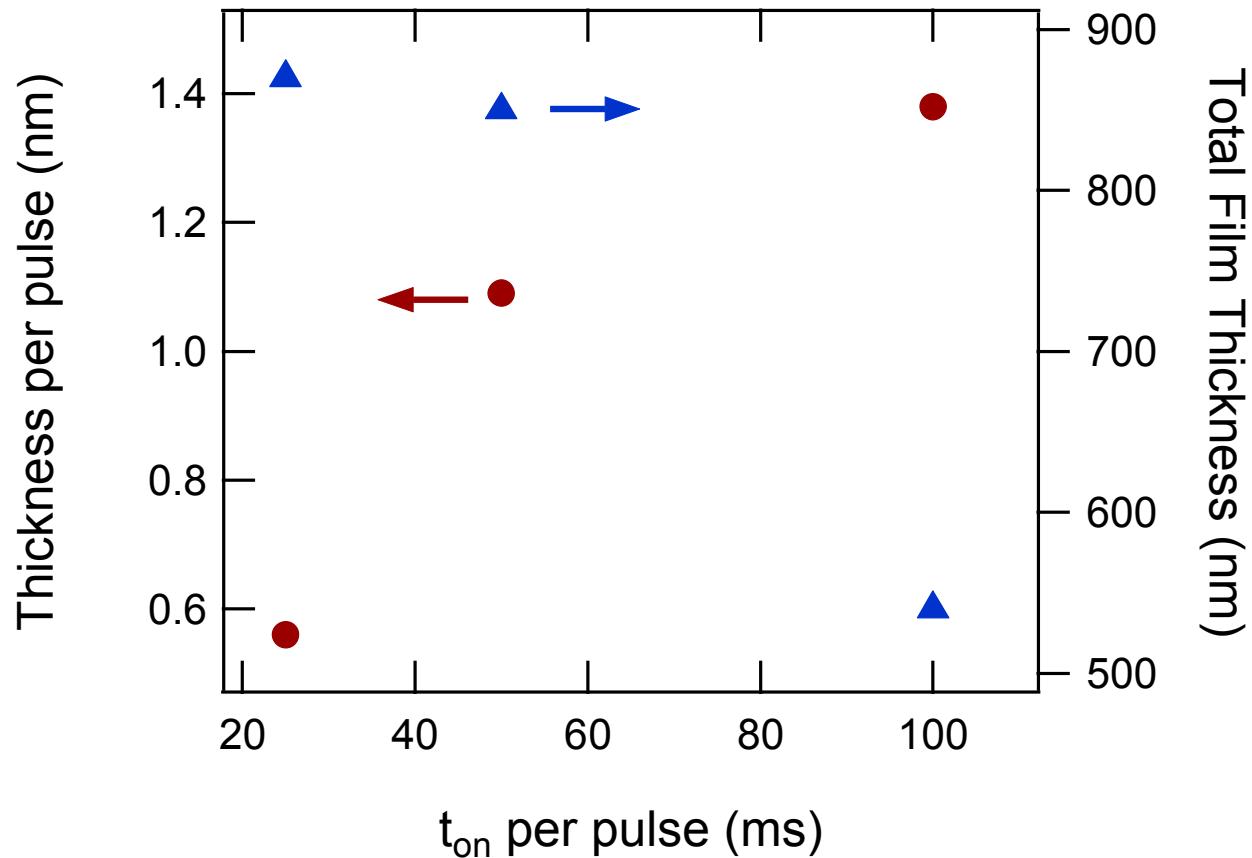
Acknowledgements

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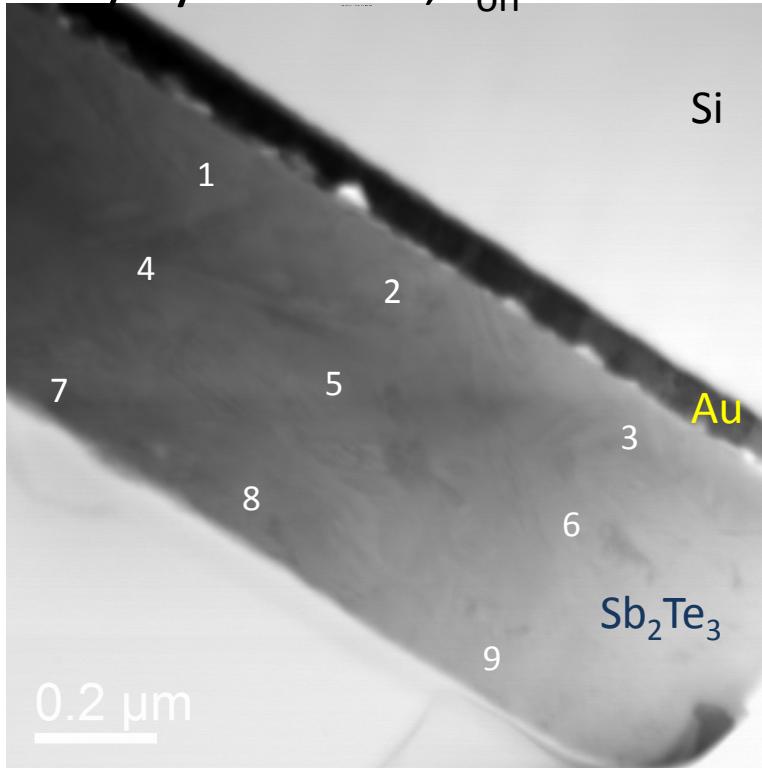


How does the film thickness vary with t_{on} ?



Compositional uniformity in pulse plated Sb_2Te_3

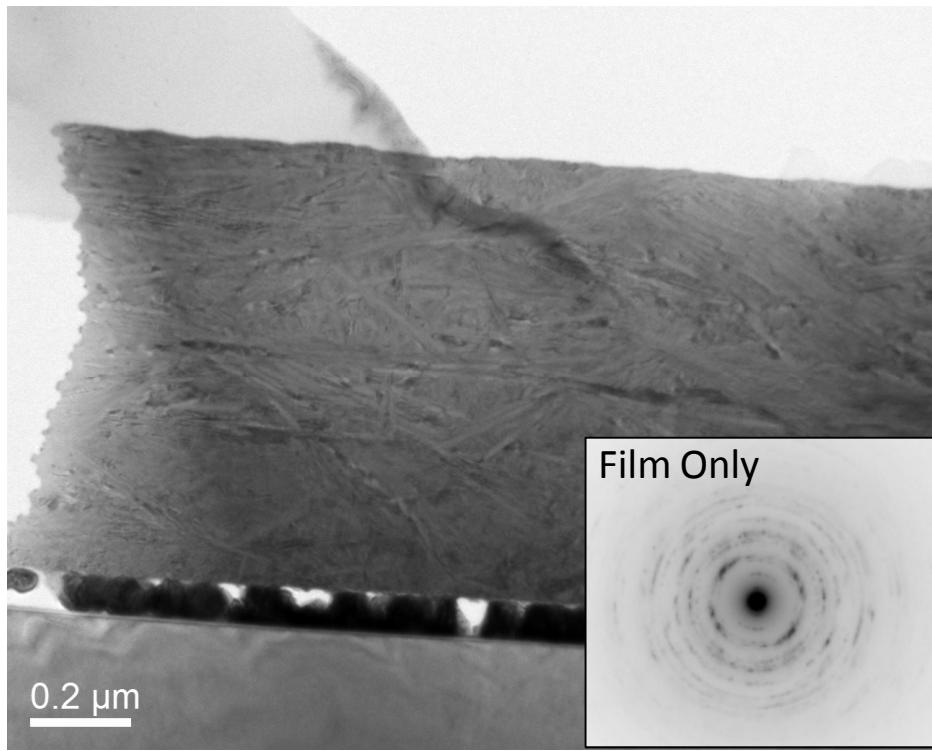
Duty cycle = 2 %, $t_{\text{on}} = 100$ ms



Semi-Quantitative STEM/EDS Analysis

	at. % Sb	at. % Te
1	40.0	60.0
2	38.1	61.9
3	38.3	61.7
4	39.6	60.4
5	39.5	60.5
6	37.8	62.2
7	35.8	64.2
8	39.7	60.4
9	41.7	58.3

The composition is uniform through the thickness of the pulse electrodeposited film at 2% duty cycle



$$t_{\text{on}} = 25 \text{ ms}$$

Hypotheses for microstructural change

Shorter pulse times lead to fewer atoms deposited per pulse

Off times allow atoms in less stable positions to desorb into solution

(001) Lamellae may be a stable configuration

- Van der waals bonding between 5-layer packets
- Low index, high density planes generally stable

