

High Temperature Solar Selective Coatings for Solar Power Central Receivers

Andrea Ambrosini

Timothy N. Lambert, Aaron Hall, Clifford Ho,
Cheryl Ghanbari

Sandia National Laboratories

ASME 2013 7th International Conference on Energy Sustainability
Session1-14-6: Receiver Modeling and Thermal Storage Analysis

Project Description

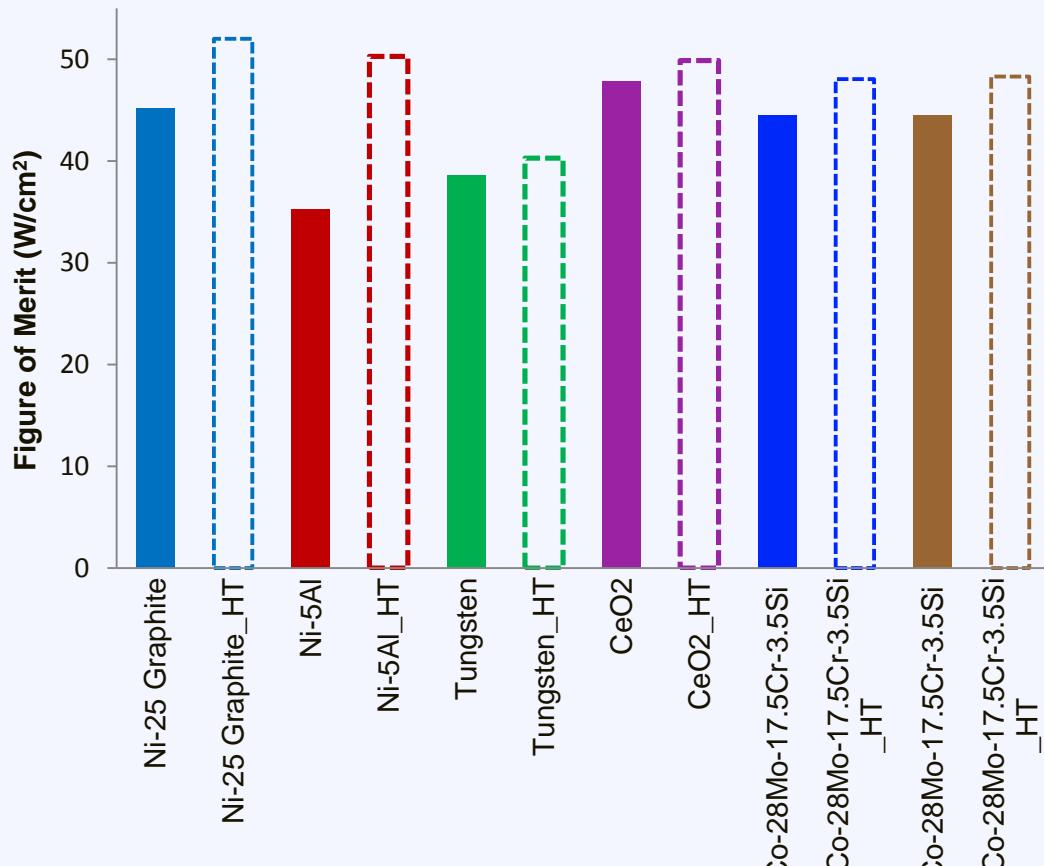
- Next-generation power tower temperatures will likely operate at temperatures ≥ 650 °C
- The efficiency of a power tower plant can be increased if the energy absorbed by the receiver is maximized while the heat loss from the receiver to the environment is minimized
- Pyromark® has a high solar absorptance ($\alpha > 0.95$), but also high emittance ($\epsilon \sim 0.87$) at the temperatures of interest
- Cermet coatings currently used in troughs have excellent optical properties, but are not well-suited for power tower applications: they are sensitive to oxidation and suffer performance degradation at temperatures > 500 °C



Improved selective absorber coatings for receivers must maintain high absorptance in the solar spectrum but lower emittance in the infrared spectrum. It must also be stable in air, easily applied at large scales, cost effective, and survive thousands of heating and cooling cycles

- At 650 °C, a reduction in ϵ from 0.88 to 0.4 will increase the thermal efficiency by 4%
- At 800 °C, the same reduction increases the thermal efficiency by 7%
- Levelized cost of energy (LCOE) estimated to be reduced at least 0.25¢/kWh

Heat Treated Coatings



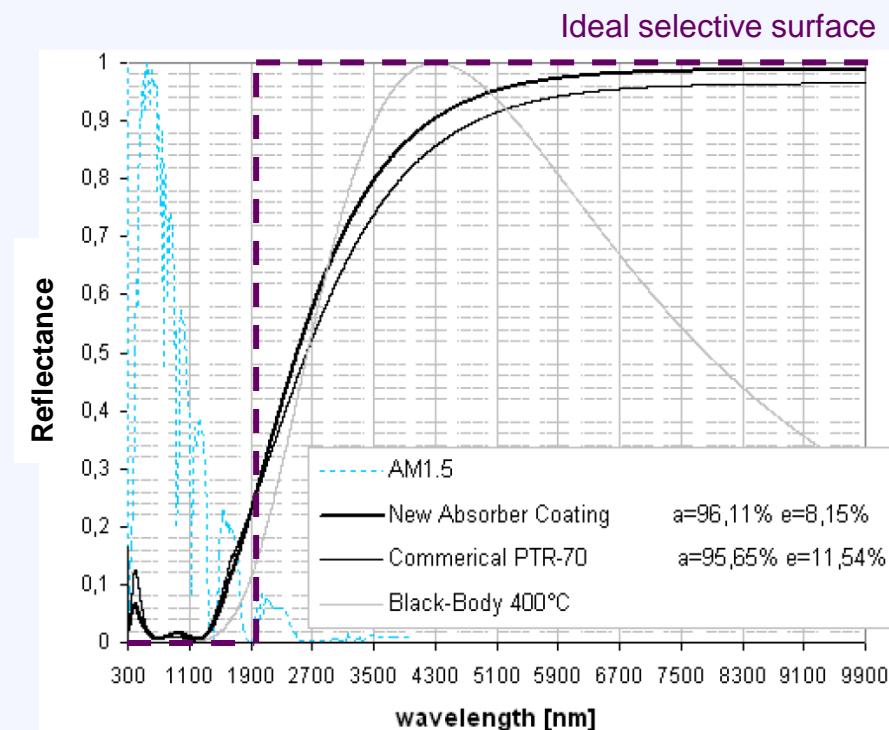
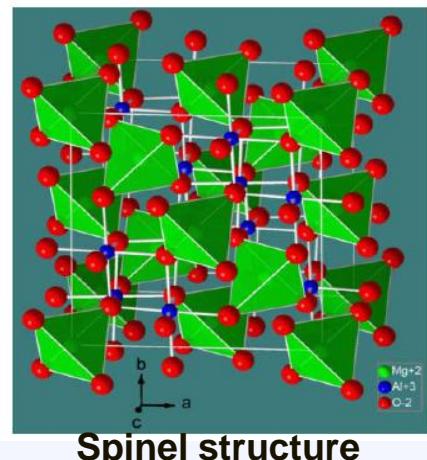
FOMs for as-deposited (filled bars) coating test coupons and coating test coupons heat treated for 6 hours at 600° C (open bars).



- Heating generally increased both α and ϵ , resulting in an overall increase in FOM
- Likely surface oxidation
 - Sometimes results in violent oxidation (tungsten) or spalling (tungsten carbide)
 - Effects thermal stability of films
- Surface modification (smoothing) may also occur over time; under investigation

Materials

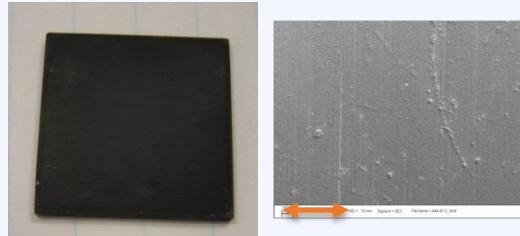
- Novel materials that are intrinsically solar selective: high α , low ϵ and stable in air and high temperatures for power towers
- Metal spinel oxides (AB_2O_4):
 - Inherently stable at high temperature and in air
 - Amenable to doping and substitution (e.g. Ni^{2+} , Mn^{2+} and Cu^{2+}), to chemically tailor their properties



N. Benz, et al, Advances In Receiver Technology For Parabolic Troughs, SolarPACES 2008.

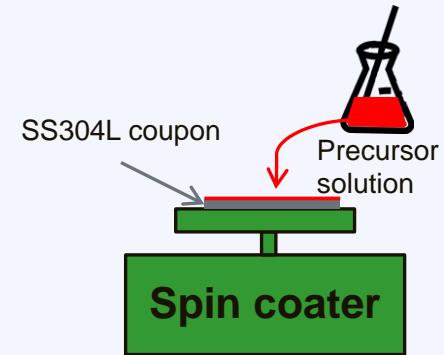
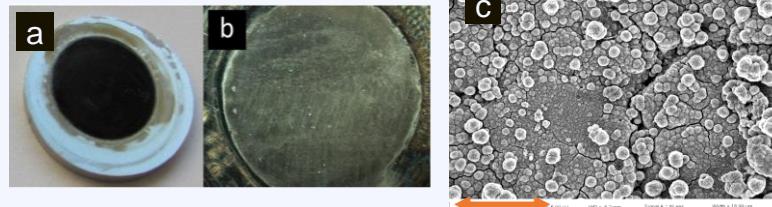
Deposition Methods

- *Spin coating*
 - Facile synthesis of coatings with varying formulations and dopant concentrations
 - Allows for rapid deposition and optical screening of a composition space
- *Electrodeposition*
 - Novel approach to screening solar selective materials
 - Can result in novel surface morphologies



FeCo_2O_4 coating (left) and SEM of FeCo_2O_4 surface (right). Scale bar = 500 μm .

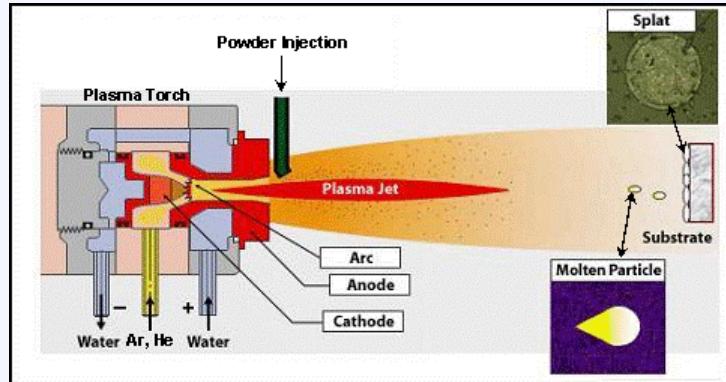
(a) Co_3O_4 /SS304L prepared high-T ED method (middle).
(b) Image of Co_3O_4 film (deposited for 4h) on SS304L.
(c) SEM image is shown at far right (scale bar = 5 μm)



Spin coating (top) and electrodeposition (bottom)

Deposition Methods

- *Thermal Spray*
 - High-surface area coating technique
 - Ability to coat in the field
 - Development of thermal spray techniques to apply pore formers to modify surface morphology in an efficient and cost-effective manner

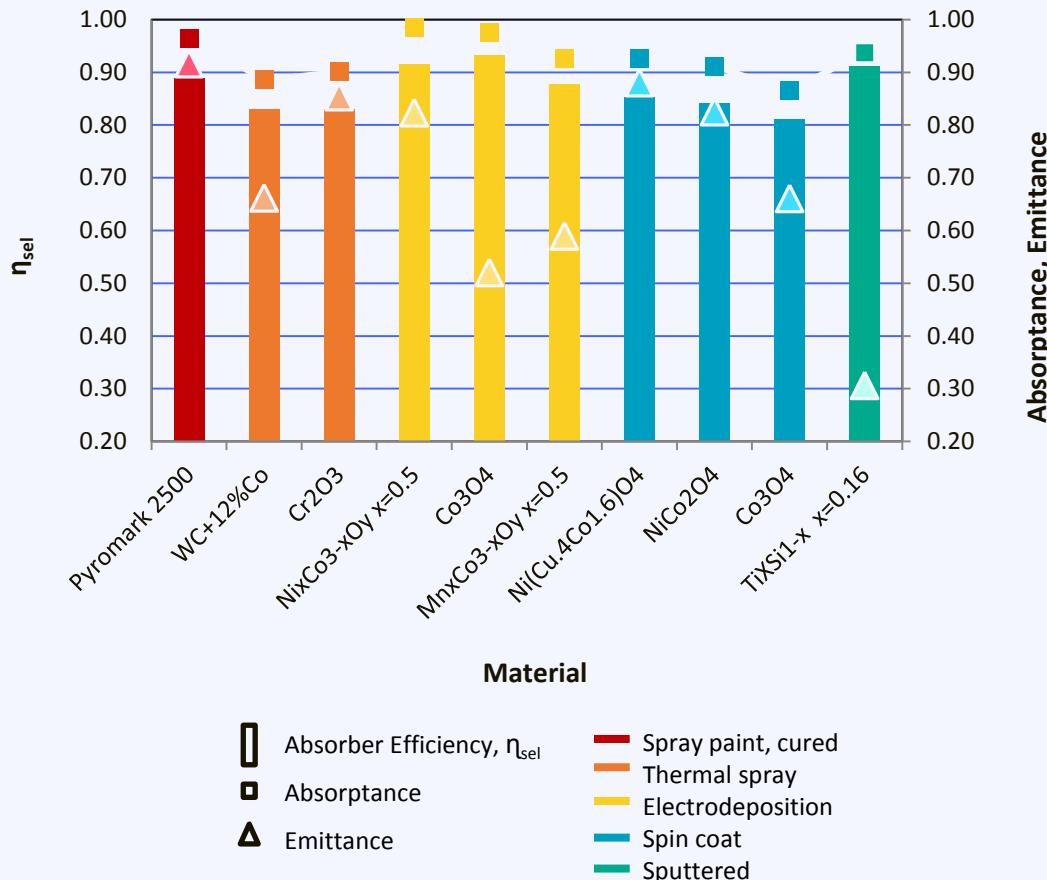


Air-plasma thermal spraying process for absorber coatings

Figure of Merit, n_{sel}

$$\eta_{sel} = \frac{\alpha_s Q - \varepsilon \sigma T^4}{Q}$$

- α_s = solar absorptance
- Q = irradiance on the receiver (W/m^2)
- ε = thermal emittance
- σ = Stefan-Boltzmann constant ($5.67 \times 10^{-8} \text{ W/m}^2/\text{K}^4$)
- T = surface temperature (K)



Absorber efficiency, absorptance, and emittance of coatings developed at Sandia and NREL via various deposition techniques. (AOP FY12)

Levelized Cost of Coating (LCOC)

- Similar to the leveled cost of electricity (LCOE)
- Defined as the ratio of the total annualized coating costs (\$) to the annual thermal energy absorbed (kWh_{th}):

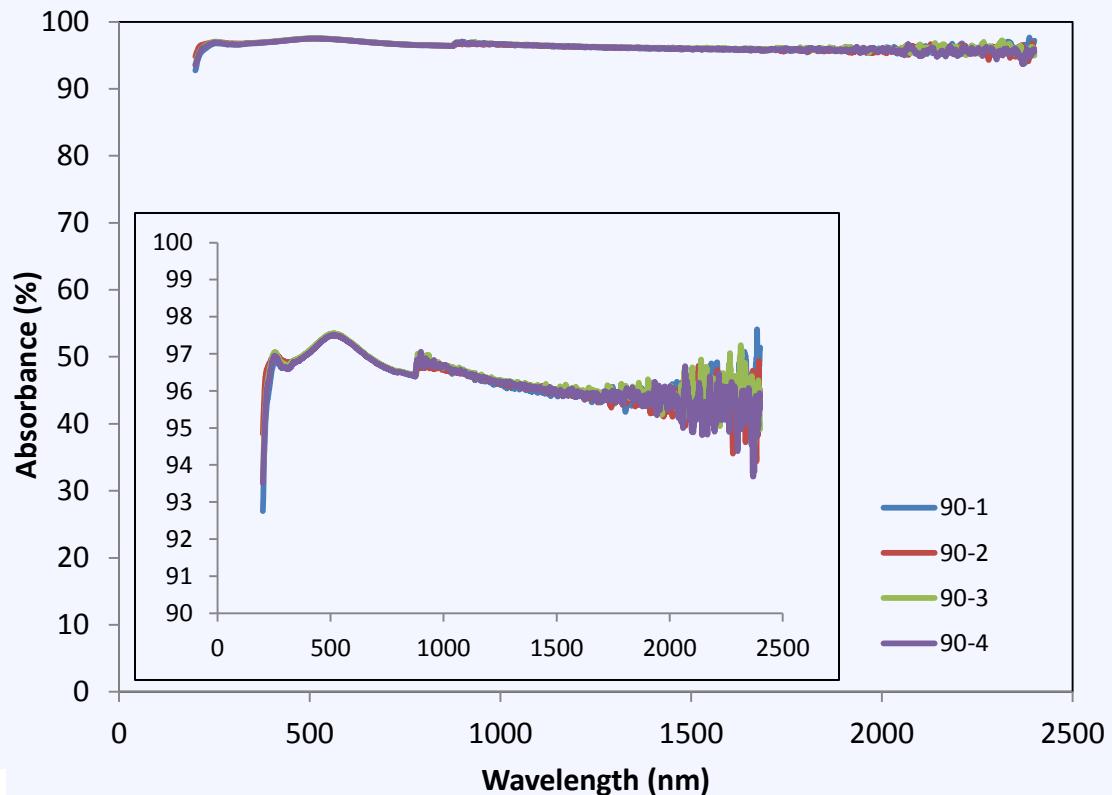
$$LCOC = C_{annual}/E_{thermal}$$

- C_{annual} = Initial coating cost/life of plant + recoating costs/recoating interval + lost revenue due to down time/recoating interval + annualized lost revenue due to degradation
- $E_{thermal}$ = Annual thermal energy absorbed (new) – Lost energy absorbed due to degradation – Lost energy absorbed due to recoating down time (annualized)

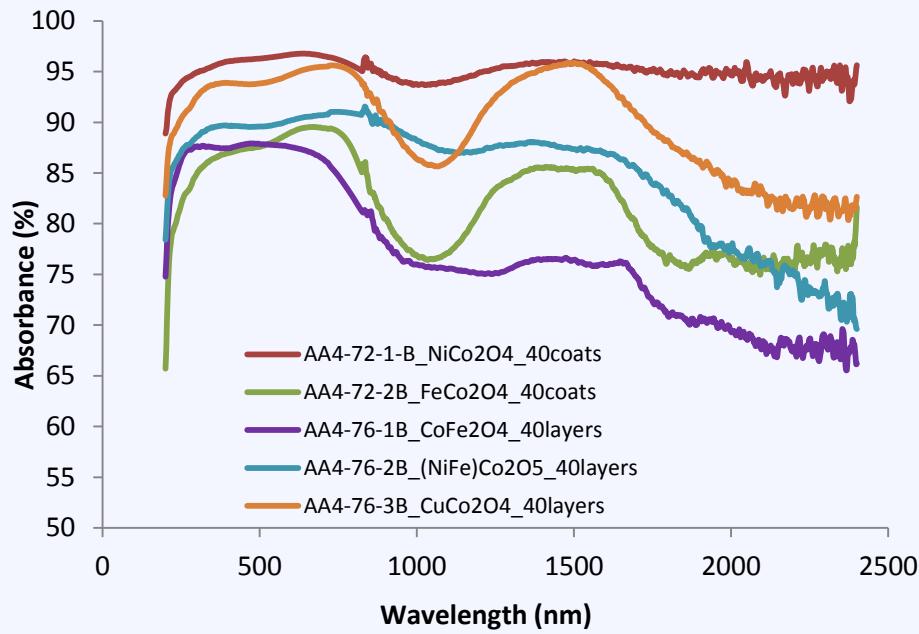
These parameters depend not only on the selective absorber efficiency, η_{sel} , which impacts the thermal energy absorbed and revenue costs, but also on degradation rate, material costs, and reapplication costs

Pyromark 2500

Sample	α (solar)	ϵ (80C rel)	ϵ (2400nm)	FOM
Pyromark 2500-1	0.965	0.861	0.972	0.889
Pyromark 2500-2	0.966	0.874	0.950	0.890
Pyromark 2500-3	0.965	0.865	0.950	0.889
Pyromark 2500-4	0.965	0.841	0.956	0.890



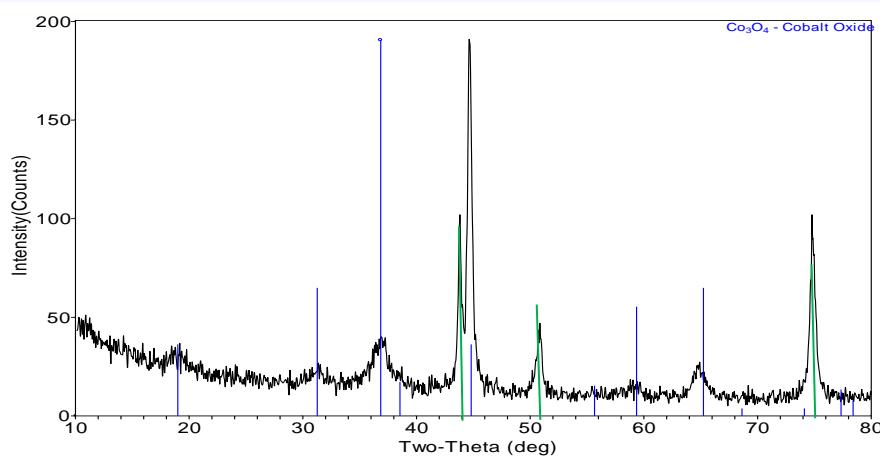
Results-Spin Coated Films



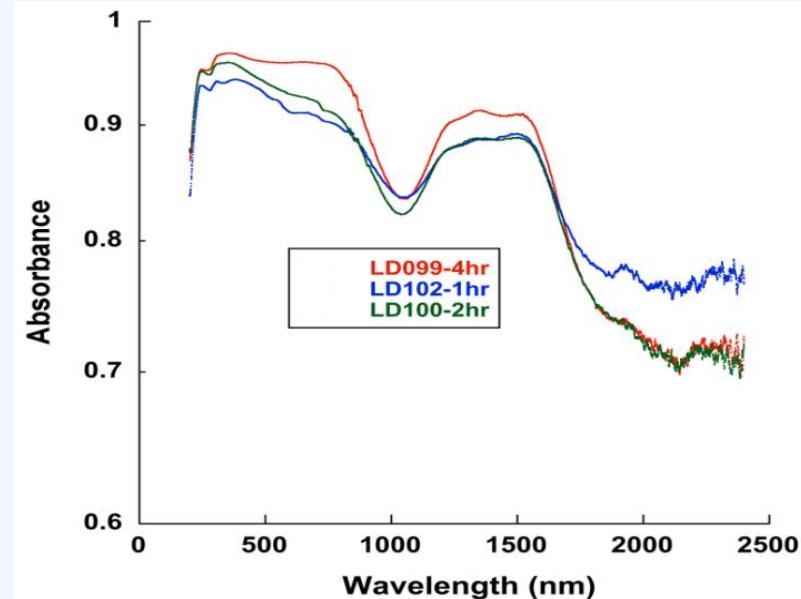
- NiCo_2O_4 shows high n_{sel} , due high α
 - However ϵ remains high
- Diffuse reflectance of some cobaltites show an undesirable absorptance “dip”, possibly due to a band gap transition
- CoFe_2O_4 does not have this dip and exhibits lower values of ϵ in the near-IR range
- Attempted to combine the high α of the cobaltite and the lower ϵ of the ferrites, several solid solutions were attempted
 - Some success in lowering ϵ versus NiCo_2O_4 , but α was also lowered

Material	α	ϵ_{80}	ϵ_{2400}	FOM (W/cm ²)
NiCo_2O_4	0.91	0.30	0.95	0.858
FeCo_2O_4	0.80	0.17	0.81	0.759
CoFe_2O_4	0.82	0.20	0.66	0.784
CuCo_2O_4	0.89	0.22	0.82	0.847
$(\text{NiFe})\text{Co}_2\text{O}_5$	0.88	0.34	0.70	0.837
SS304L coupon (no heat)	0.46	0.24	0.58	0.426
SS304L coupon	0.62	0.13	0.60	0.590

Results-Electrodeposition



PXRD on thin films $\text{Co}_3\text{O}_4/\text{SS304L}$ indicating the formation of the spinel (blue) phase directly. Strong peaks from the SS304L substrate are also observed (green).



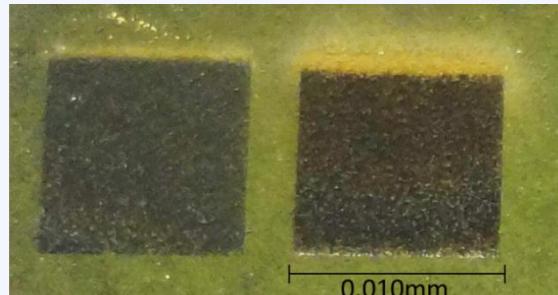
Diffuse reflectance UV-Vis spectra of a) three $\text{Co}_3\text{O}_4/\text{MS-SS304L-P}$ formed directly from high temperature electro-deposition.

- New high temperature electrodeposition method results in direct deposition of Co_3O_4 w/o need for additional sintering step
- Initial η_{sw} (0.849-0.871) look promising compared to Pyromark® 2500 (0.892)
- Amorphous phase may be present
 - Annealing studies (followed by XRD and SEM) will be performed to compare with previous ED results and to detect any change in crystallinity
- Mechanical stability seems improved on as-deposited coatings vs. rt deposition

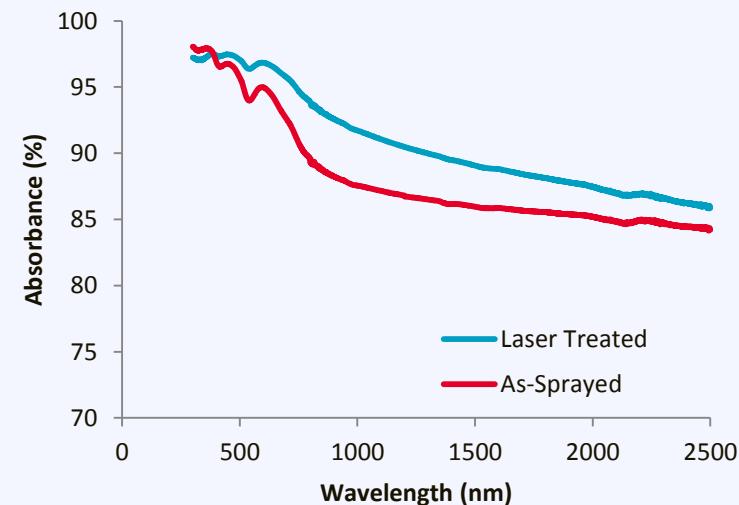
Results-Surface Modification of Thermal Sprayed Cr_2O_3

Cr_2O_3

- Melts at 2435 °C
- Extreme
 - thermal stability
 - chemical stability
 - hardness
 - wear resistance
- $\eta_{\text{sel}} = 0.83$
(as-deposited)



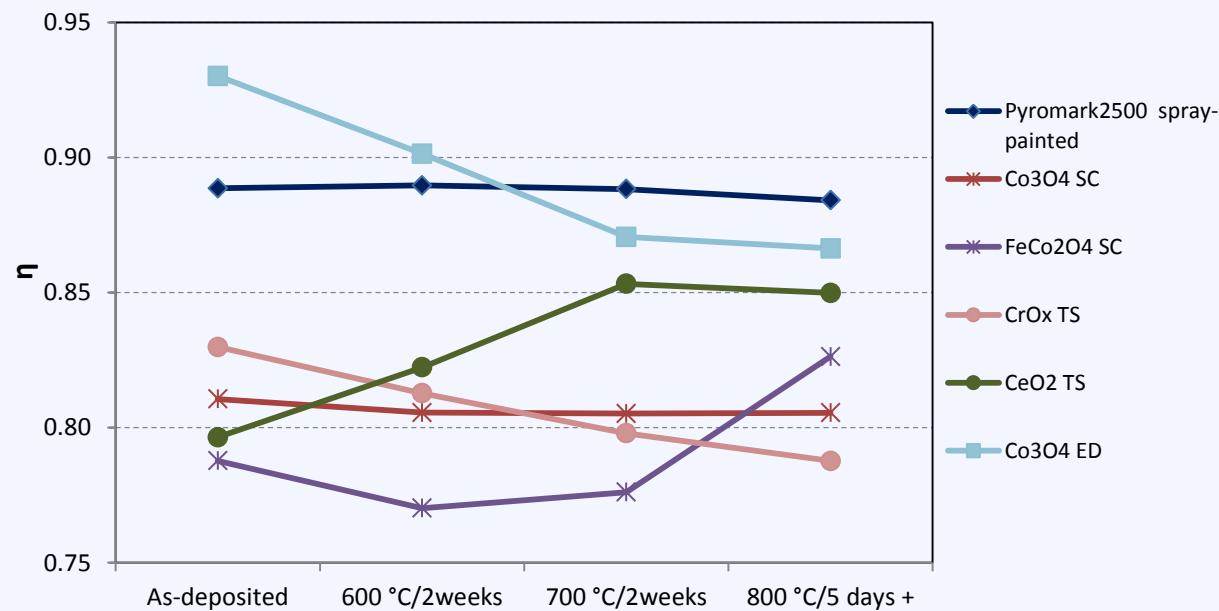
Microscopy image of Cr_2O_3 coating after laser surface treatment.



- Laser surface treatment has significantly darkened the coating
- Because of the small size of these treated areas it was not possible to acquire ϵ and α to determine η_{sel}
- Diffuse reflectance shows a measureable increase in absorbance post-laser treatment
- The mechanism for this change in reflectance is under investigation

Results-Durability

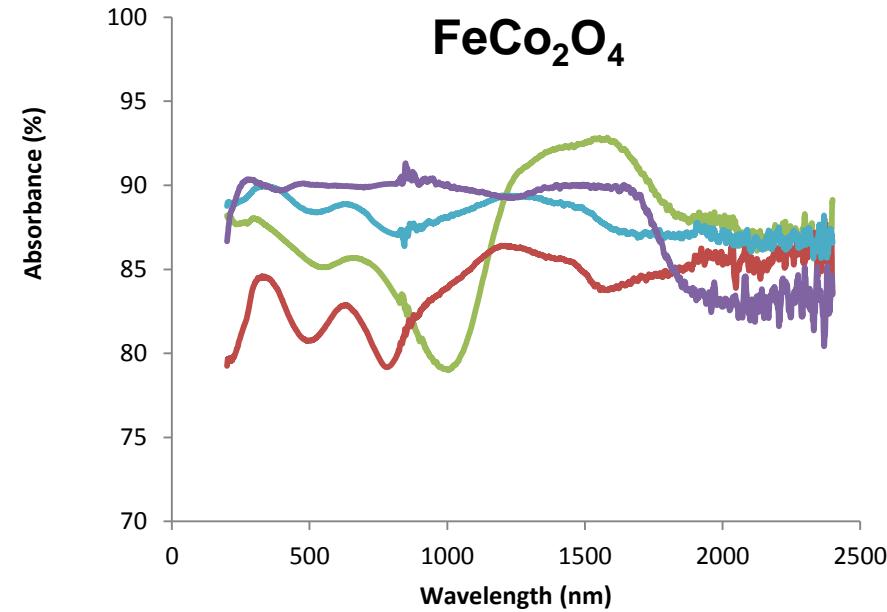
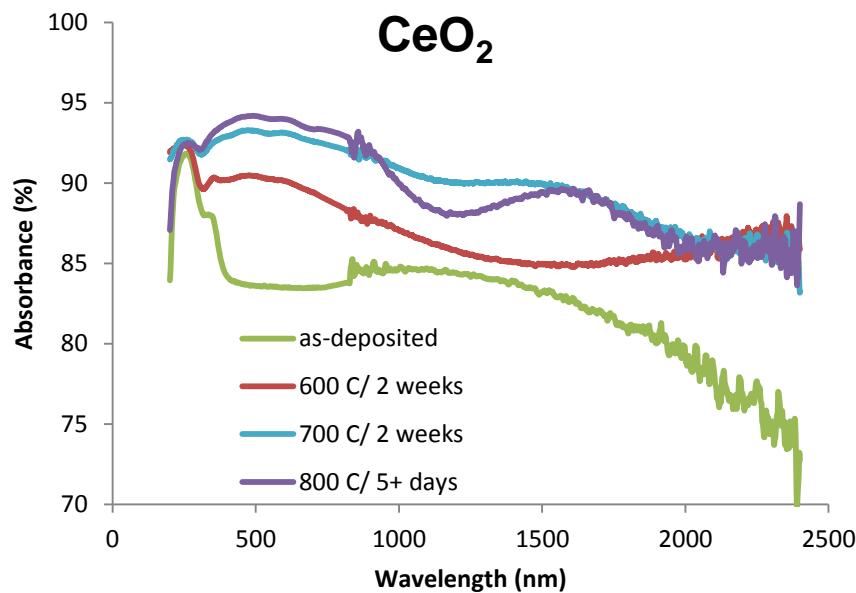
- Several coatings with high η were aged at various temperatures to investigate durability: 600 °C for 2 weeks, 700 °C for 2 weeks, and 800 °C for 5+ days
 - Heating time for 800 °C differs due to a furnace failure
- Pyromark® 2500 remains stable during aging, though α begins to decline after heating at 800 °C heating
- Electrodeposited (rt) Co_3O_4 samples decline in performance, but remain competitive with Pyromark®
 - However these films are not mechanically robust



FOM, η , for various coatings as a function of aging.

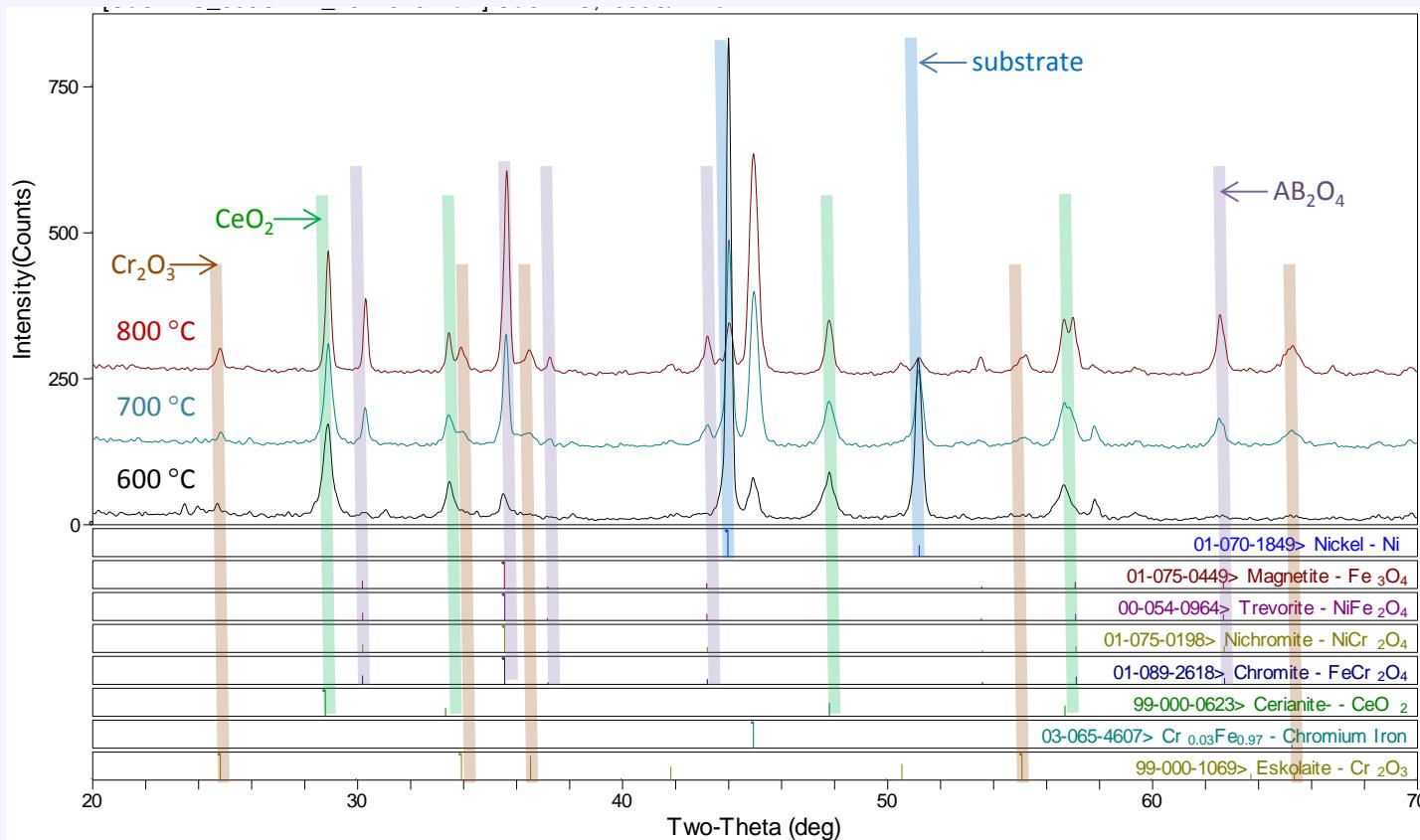
Ambrosini, ASME ES2013

Results-Durability



- Thermal sprayed CeO₂ coating actually increases in η_{sel} when aged
 - Visible darkening of coating, increase in α
 - Possibly reduction of the CeO₂ to CeO_{2- δ}
 - Appearance of dip near 1200 nm after 800 °C may imply formation of a band gap
 - Inadvertent doping via cation migration from the stainless steel substrate may also influence the coating properties of CeO₂
- Conversely, the dip present in as-deposited FeCo₂O₄ disappears upon heating
 - Increase in α , decrease in the near-IR range

Durability of CeO_2

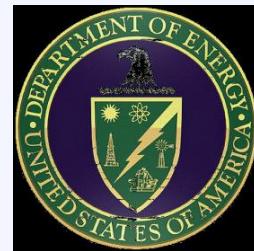


- Peaks corresponding to CeO_2 change little with heating, but
- Peaks corresponding to the substrate (“Ni”) decrease, while those corresponding to “ AB_2O_4 ” and oxidized Cr increase with temperature
 - Phase likely forms upon the oxidation of the stainless steel substrate
 - Either of these phases, which are more absorptive than ceria, may account for the increase in absorptivity.

Summary

- High-temperature electrodeposition used to deposit Co_3O_4 coatings directly onto stainless steel coupons
 - Coatings show a figure of merit competitive with Pyromark[©]
- Thermal durability examination (600-800 °C) of coatings underway
 - Spin-coated and thermal-sprayed coatings remain robust
 - Most materials show a decline in optical properties, except for CeO_2 and FeCo_2O_4
 - Reaction with substrate at higher temperatures remains a concern
- Thermal-sprayed Cr_2O_3 coatings were laser-treated to change surface morphology
 - Initial results show an increase in absorptance after treatment
- Levelized cost of coating (LCOC) (a LCOE-like metric) defined as the ratio of the total annualized coating costs (\$) to the annual thermal energy absorbed (kWh_{th})

Acknowledgements



- Cheryl Kennedy (NREL)
- Matthew Gray (NREL)
- Emma Johnson, Malynda Hungate (Spin coating)
- Landon Davis, Marlene Bencomo (Electrodeposition)
- Marlene Knight (Thermal Spray)
- Bonnie McKenzie (SEM)
- National Solar Thermal Test Facility at Sandia National Labs

Sandia National Laboratories is a multi-program laboratory managed and operated by Sandia Corporation, a wholly owned subsidiary of Lockheed Martin Corporation, for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-AC04-94AL85000.

Thank you for your attention. Questions?

