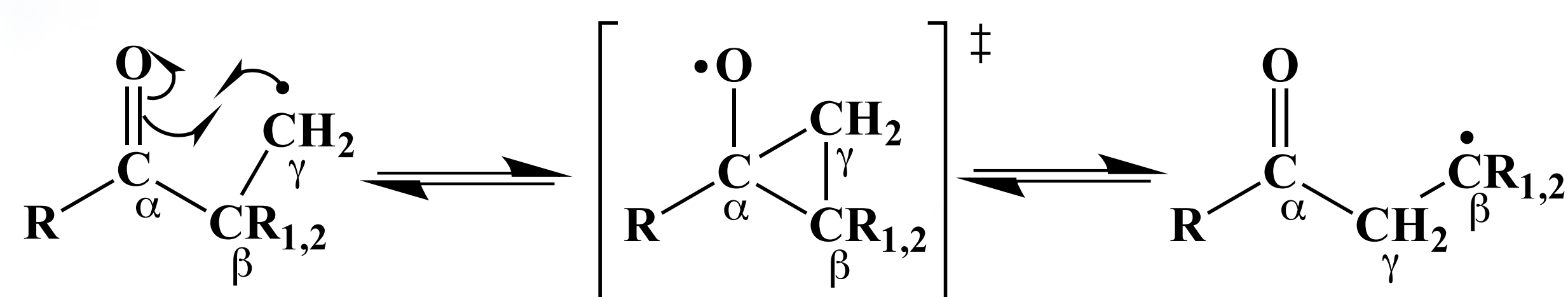


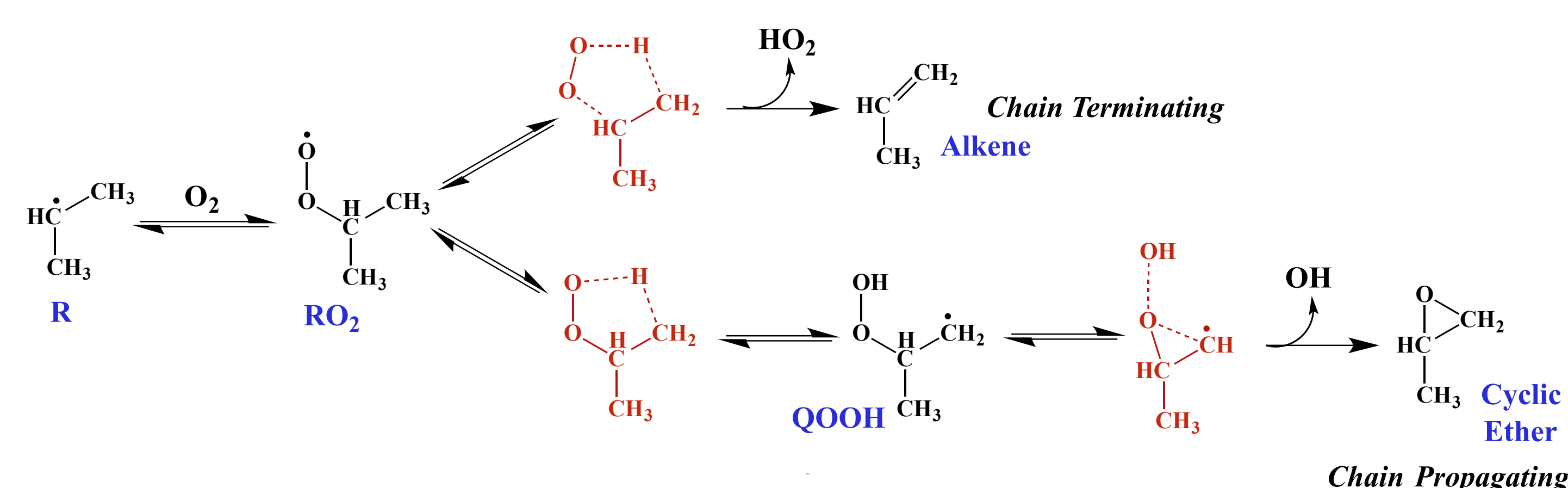
## Introduction and Experiment

**Rapid Radical Rearrangement** (1,2-acyl group migration<sup>1</sup>) is observed for  $\beta$  ketone radicals.



**Chain-Propagating Oxidation Channels** via resonance-stabilized QOOH are highly favored in open-chain ketones.

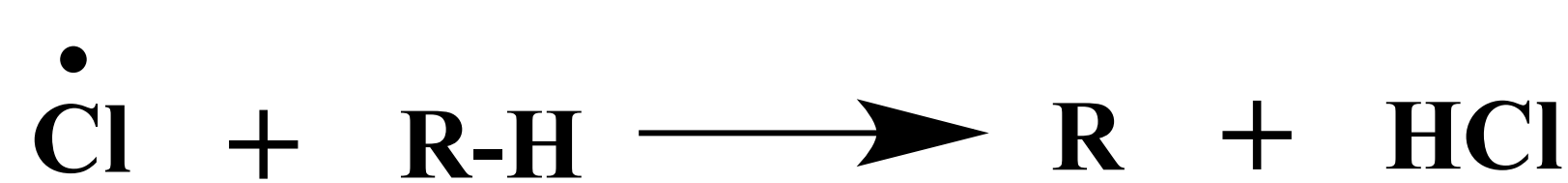
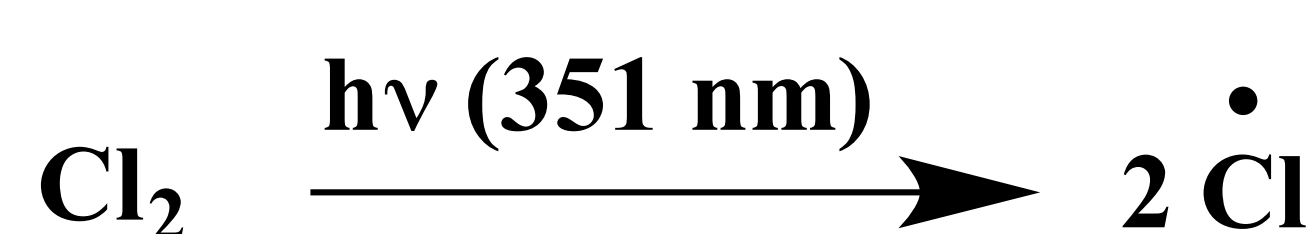
## Background: Isopropyl Radical Oxidation<sup>2</sup>



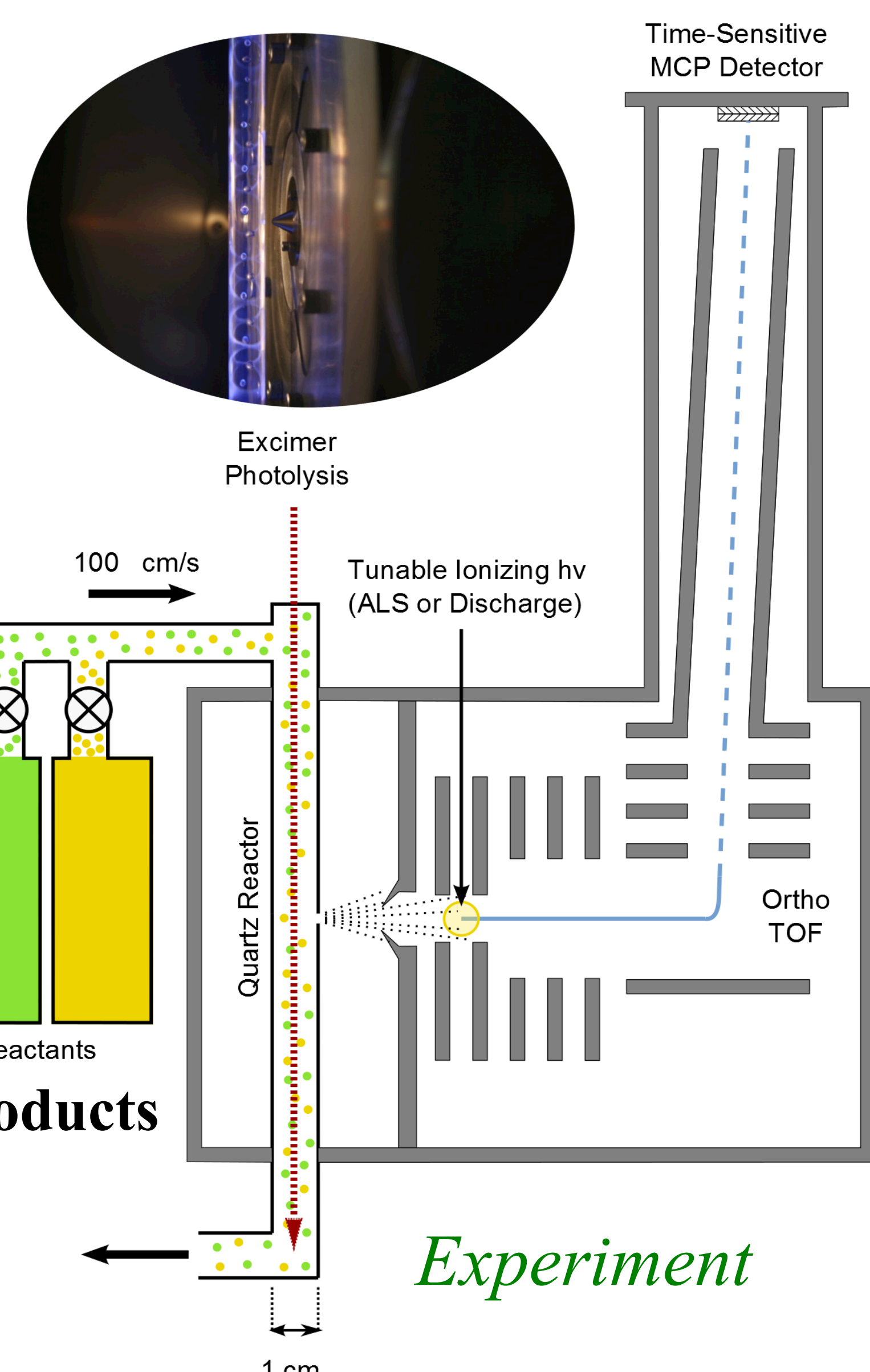
• **Multiplexed Photoionization TOF Mass Spectrometry** with tunable synchrotron light generates three-dimensional data in the mass, energy and time domains.

• **Photoionization Spectra (PI)** help determine products and branching.

• **Photolysis** generates  $\text{Cl}\cdot$ , which abstract H, leaving a radical (R) that can add  $\text{O}_2$ .

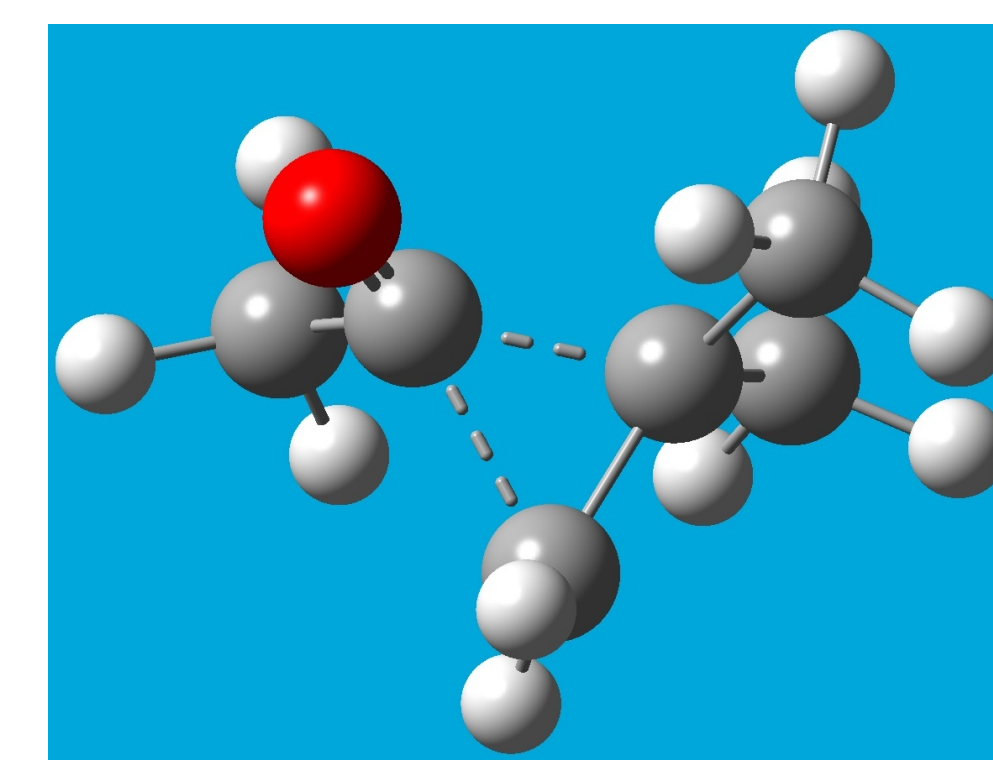
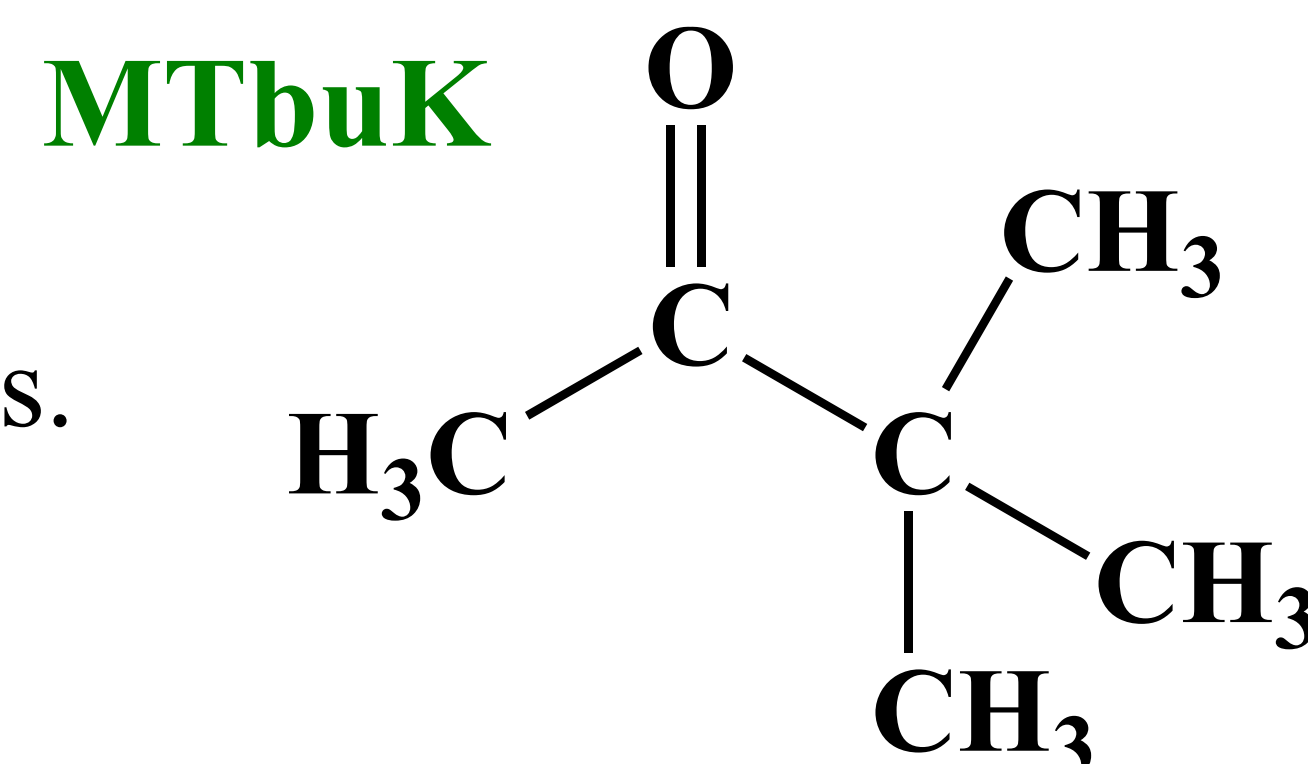


$T = 300 - 700 \text{ K}; P = 2 - 10 \text{ Torr}$

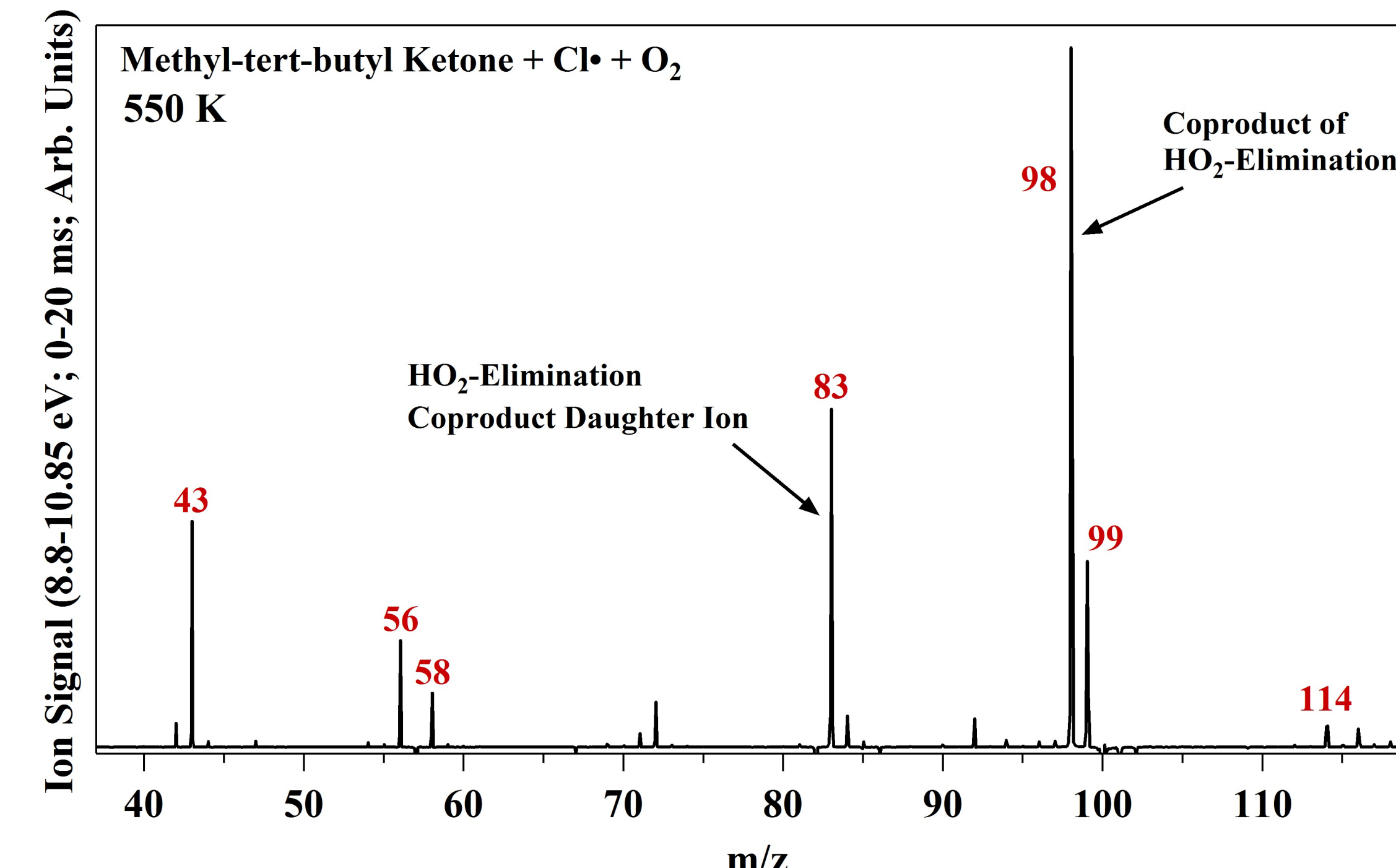


## Oxidation of Methyl-Tert-butyl Ketone (MTbuK)

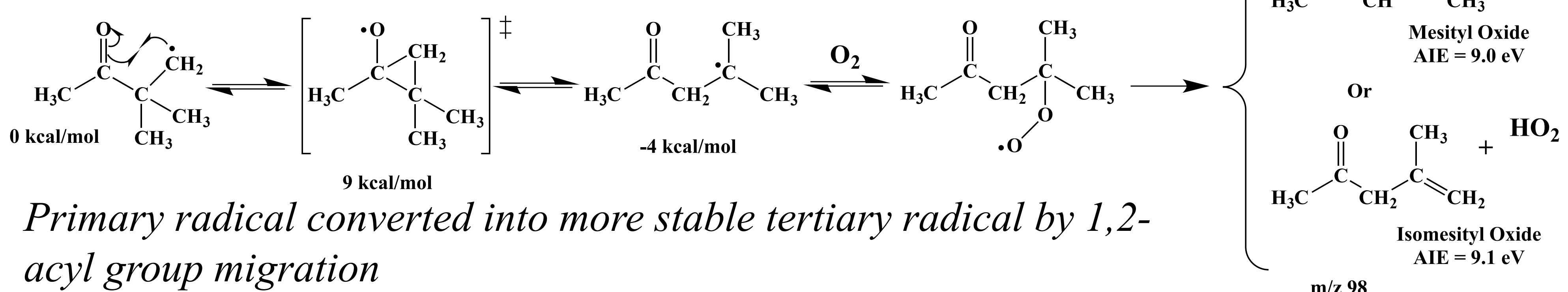
All MTbuK carbons are either quaternary or terminating methyl groups. Thus  $\text{HO}_2$  + alkene channels analogous to propane are not possible.



*Rearrangement TS*

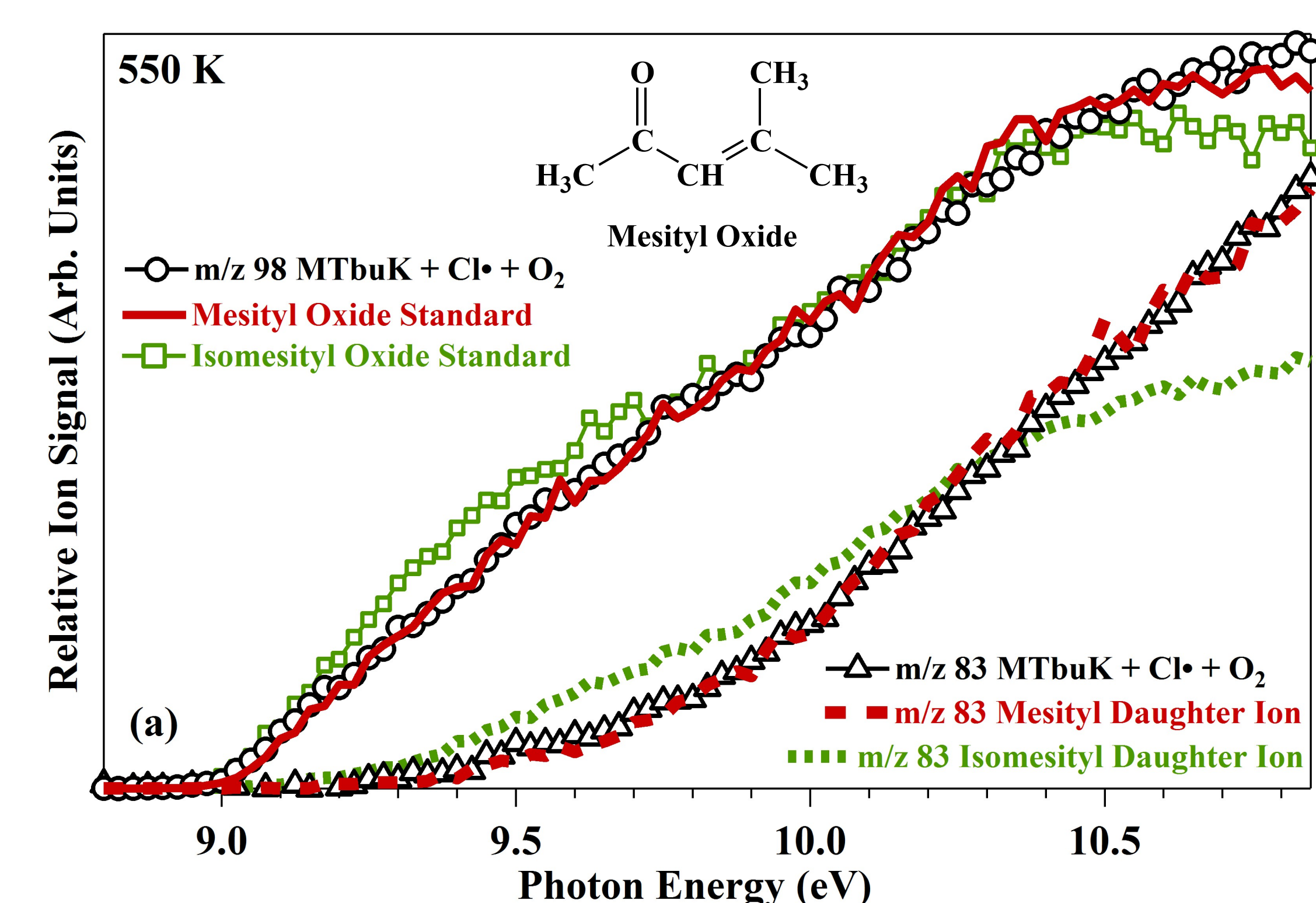


Yet the  $m/z$  98  $\text{HO}_2$ -elimination coproduct is **dominant** and results from radical rearrangement that enables formation of a double bond:



*Primary radical converted into more stable tertiary radical by 1,2-acyl group migration*

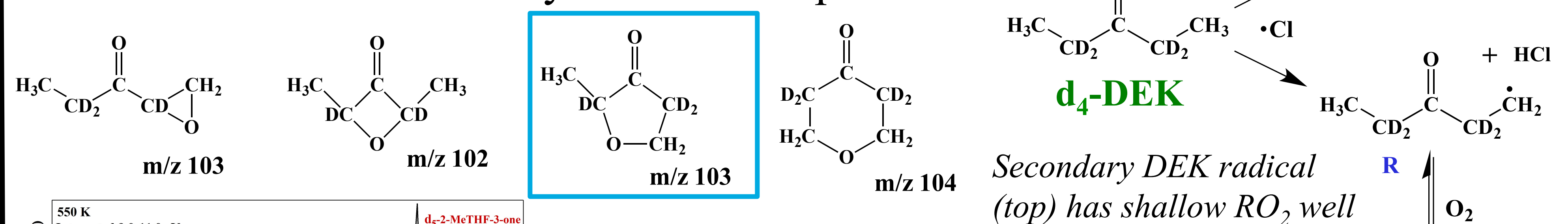
PI spectra show mesityl oxide is  $m/z$  98 product.



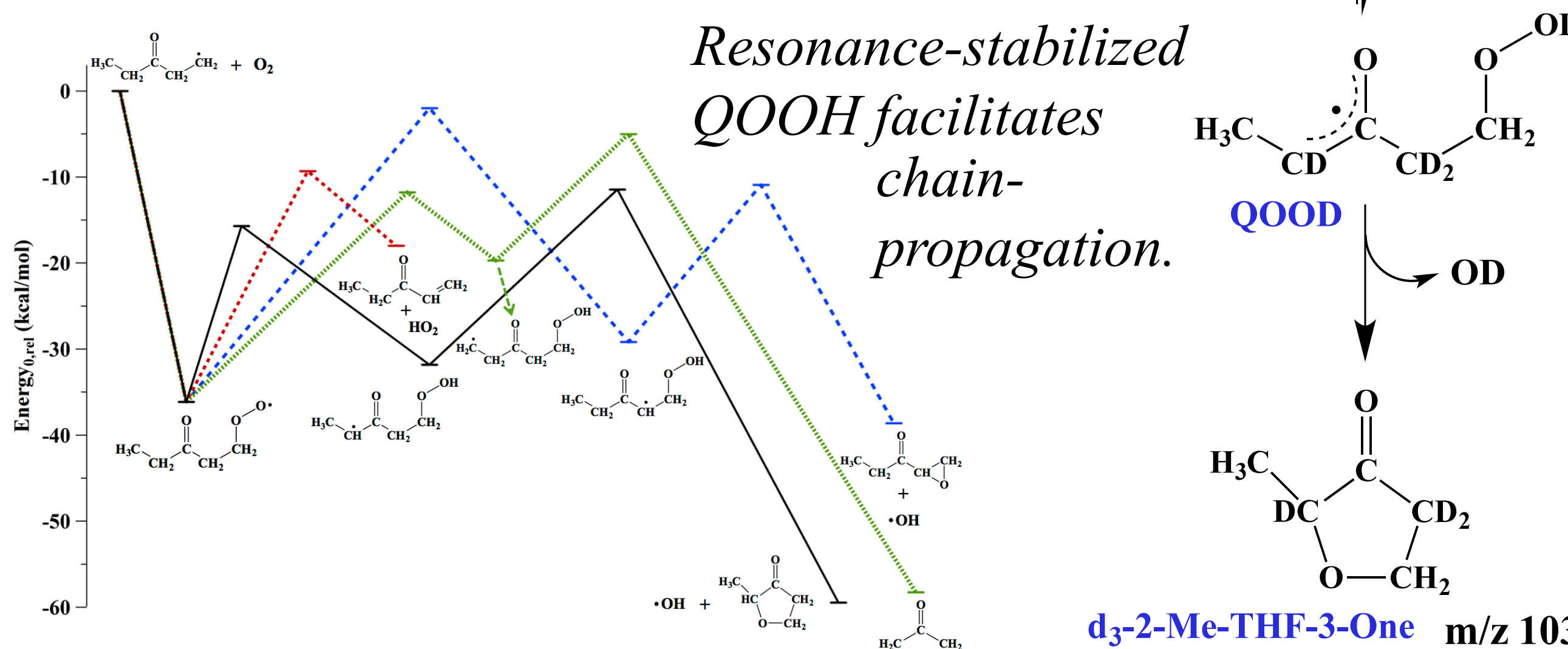
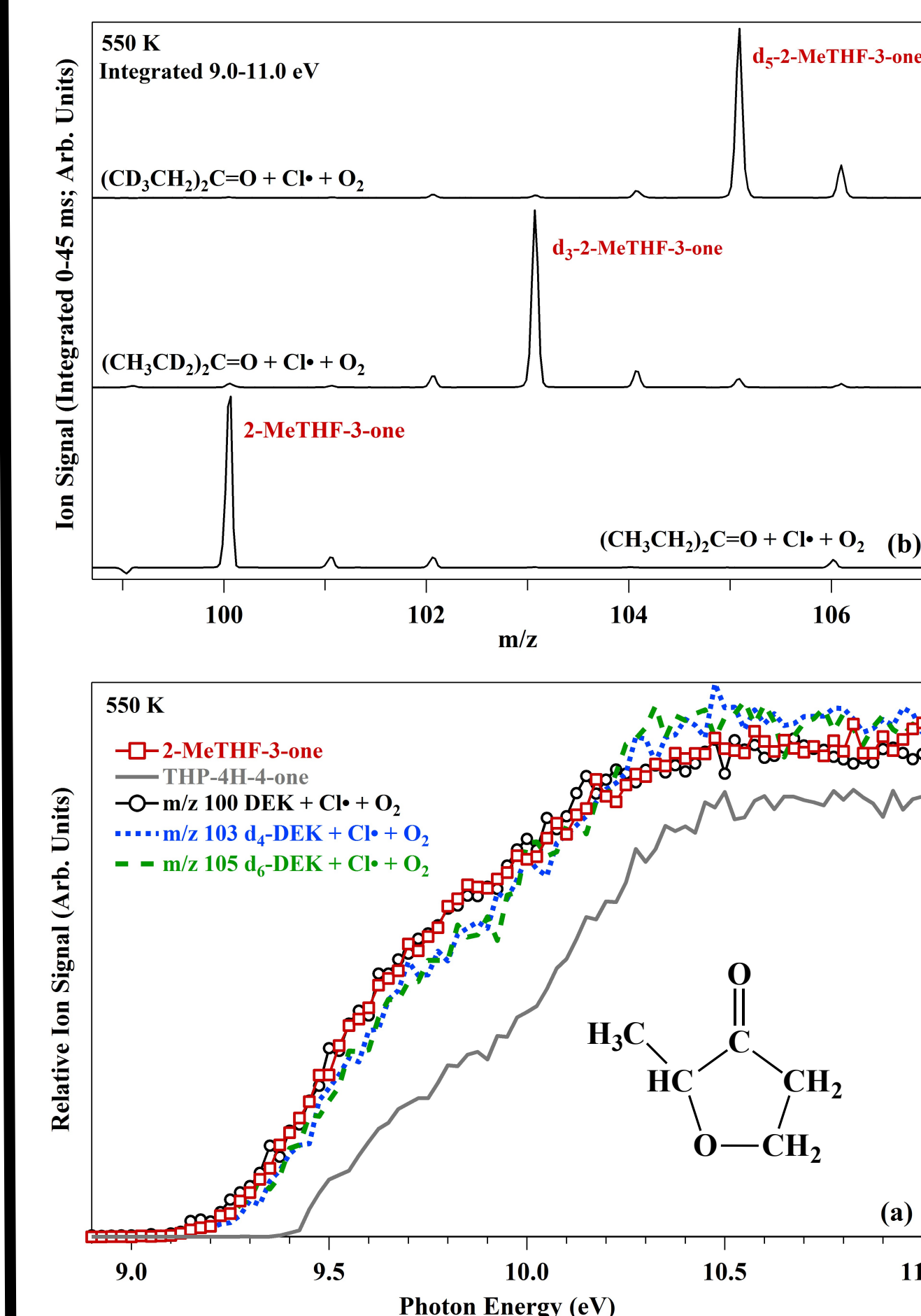
Rearrangement is fast relative to  $\text{O}_2$ -addition under our conditions (550 K, 8.0 Torr;  $[\text{O}_2] = 3 \times 10^{16} \text{ cm}^{-3}$ )

## Oxidation of Diethyl Ketone (DEK; $\text{d}_4$ -DEK and $\text{d}_6$ -DEK)

Formation of four distinct cyclic ethers is possible.



$\text{d}_4$ - and  $\text{d}_6$ -DEK data show the 5-membered ring, 2-Me-THF-3-one is highly favored. The product PI spectra are nearly identical with the standard.



[1]. e.g. Karl et al. *JOC*, **37**, 2834 (1972).  
[2]. DeSain et al. *JPCA*, **107**, 4415 (2003).