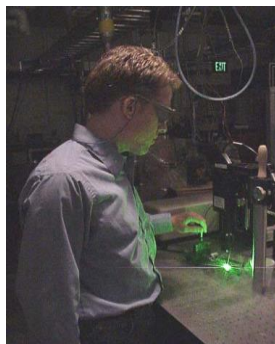


Exceptional service in the national interest



Detection of Trace Explosives by Ion Mobility-Mass Spectrometry

C.L. Beppler, Ph.D.
Explosive Technologies Group

Unclassified-Unlimited Release



Explosives Detection



Ion Mobility Spectrometry

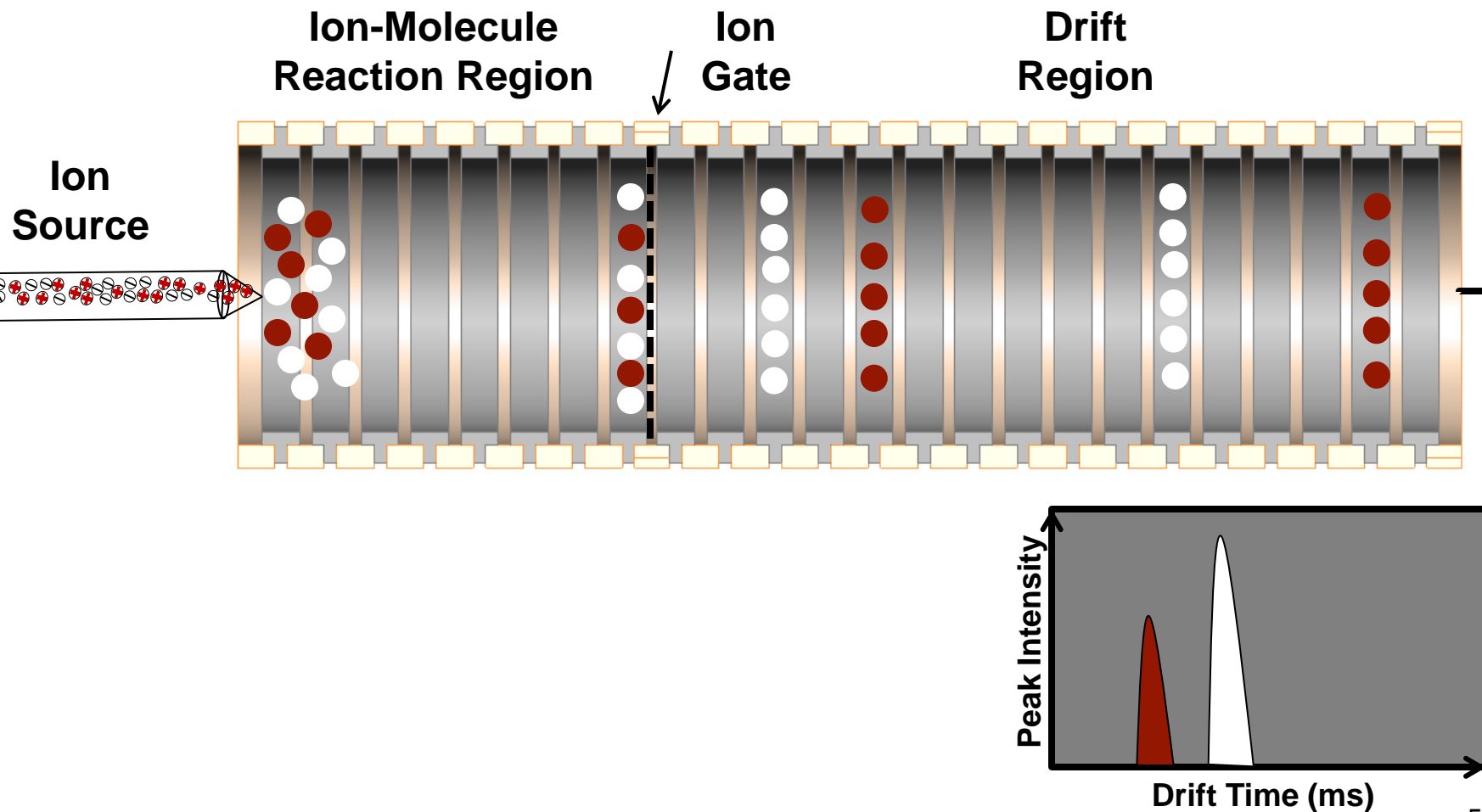
Commercial ^{63}Ni -IMS System



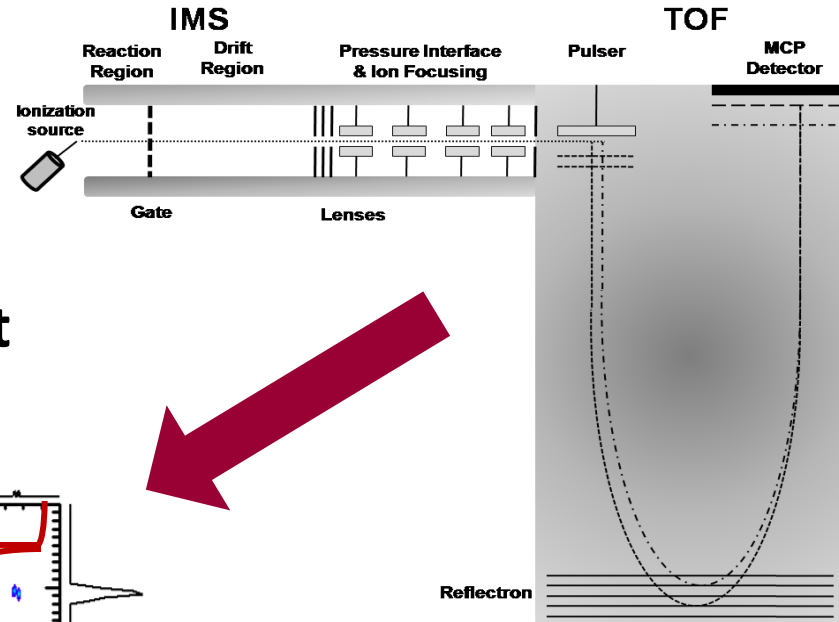
Advantages of IMS

- Gas phase ion separation by size/charge
- Quick analysis cycle (under 6 seconds)
- Commercial systems easy to operate; currently deployed worldwide
- Air is carrier gas
- Good qualitative instrument
- Sub ppb detection limit for many explosives

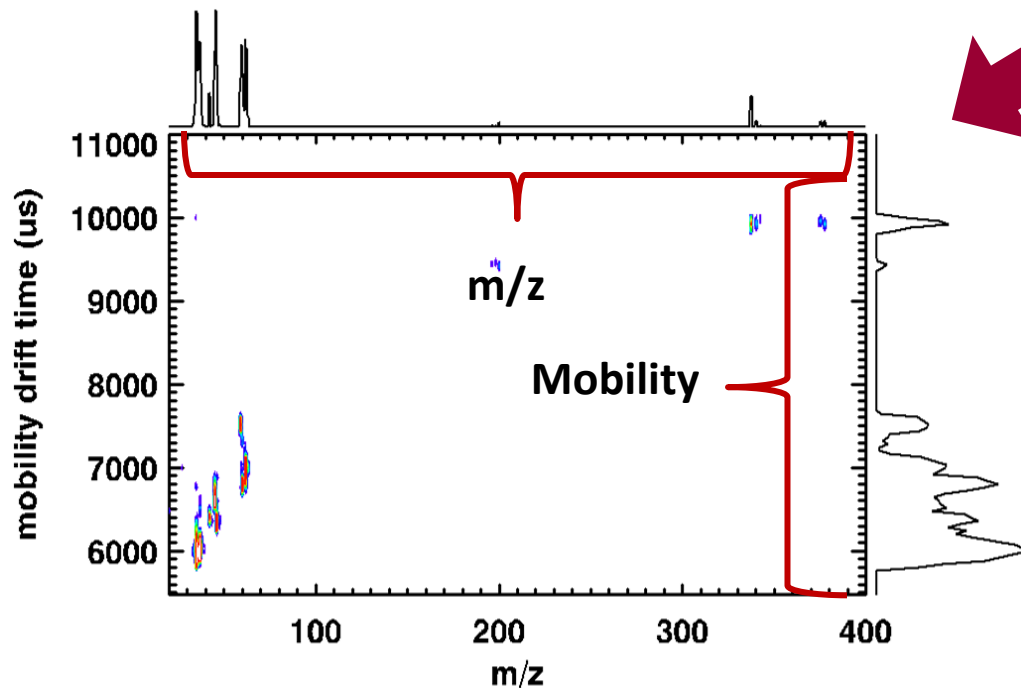
Ion Mobility Spectrometry



Simultaneous Mobility-Mass ID

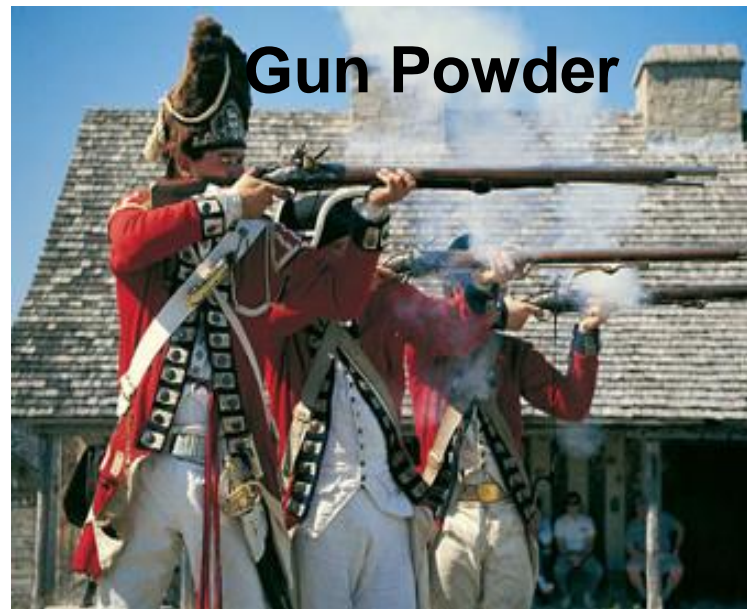


2D Mobility-Mass Plot



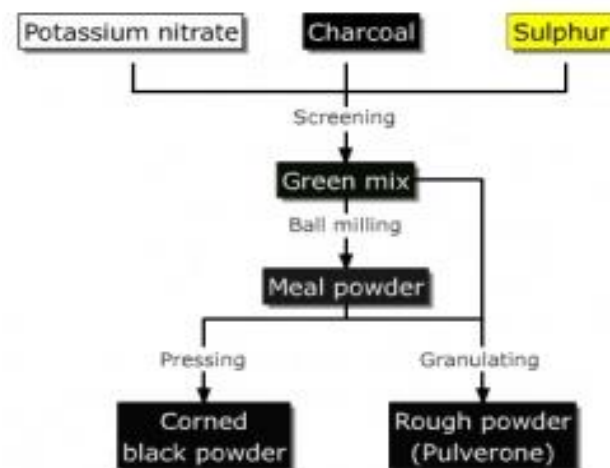
*Reduced Mobility values
identify analytes in IMS*

$$K_0 = \left(\frac{L^2}{V t_d} \right) \left(\frac{P}{P_0} \right) \left(\frac{T_0}{T} \right)$$



Black Powder by IMMS

- Obtain a K_0 value for black powder
- Mass ID the ion species that make up the mobility peak
- Determine differences in K_0 values for black powders from differing origins
- Published in *Analytical Chemistry*



Previous Black Powder Detection Methods

Direct Chemical Methods

- Chemical spot testing: using a reagent that confirms presence of potassium nitrate or sulfur

Spectroscopic Methods

- Refractive Index measurements, x-ray powder diffraction, emission spectrography, polarizing microscopy

Chromatographic Methods

- TLC, ion chromatography/conductivity detection

IMS

- Barringer IONSCAN Model 200

MS

- TOF-SIMS & LDI-TOF-MS

(1) Hoffman, C. M., M.S. Byall, E.B. Byall *Journal of Forensic Science* 1974, 19, 54-63.

(2) Meng, H., B. Caddy *Journal of Forensic Science* 1997, 42, 553-570.

(3) Wolten, G. M., R. S. Nesbitt, A. R. Galloway, G. L. Loper *Journal of Forensic Science* 1979, 24, 423-430.

(4) Mahoney, C. M. G. G., Albert J. Fahey *Forensic Science International* 2005, 158, 39-51.

(5) Hestley, A. K., B. P. G. Johnson, J. S. Moridue, D. G. Tuck *Organica Chimica Acta* 2002, 324, 105-112.

(6) Parker, R. G., M. O. Stephenson, J. M. McOwen, J. A. Cherolis *Journal of Forensic Science* 1975, 20, 133-140.

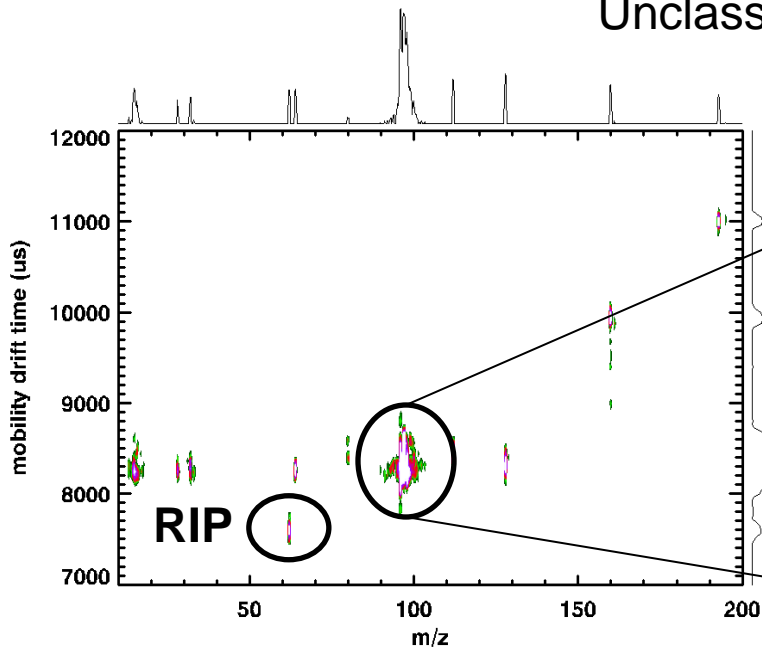
(7) Beveridge, A. D., S. F. Payton R. J. Audette A. J. Lambertus R. C. Shaddick *Journal of Forensic Science* 1975, 20, 431-454.

(8) Hall, K. E., B.R. McCord *Journal of Forensic Science* 1993, 38, 928-934.

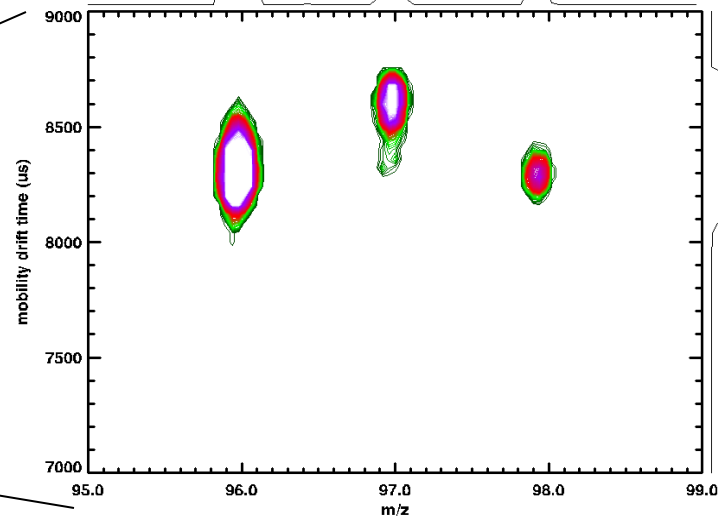
(9) Fetterolf, D. D., T.D. Clark *Journal of Forensic Science* 1993, 38, 28-39.

Results & Conclusions

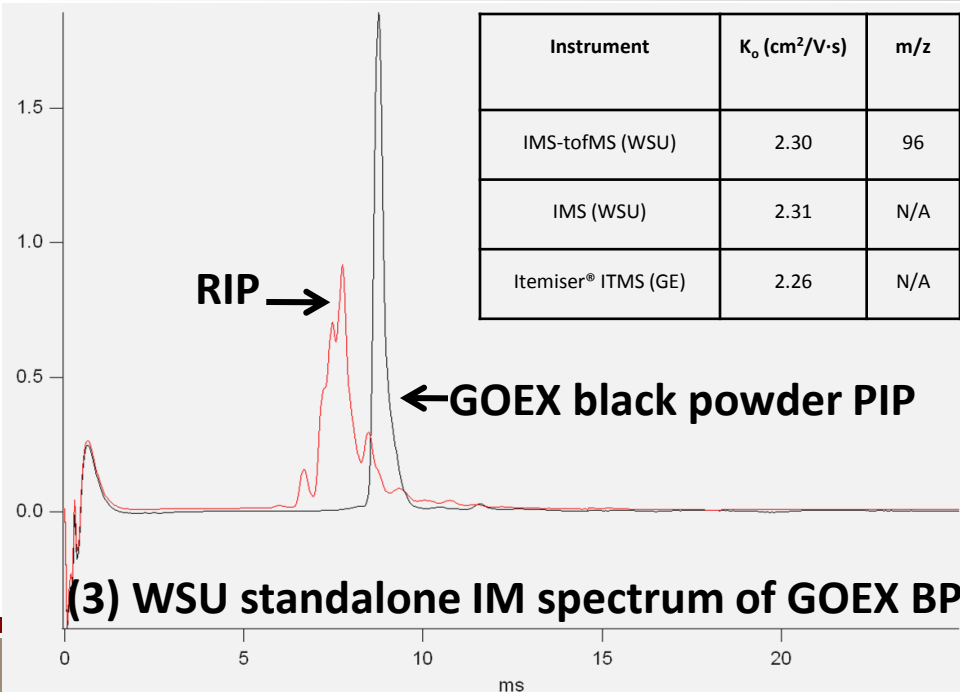
- Black powder successfully detected using ^{63}Ni -IM-TOFMS
- *First reported* K_0 value for the primary black powder mobility peak at $2.28 \pm 0.02 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$
 - S_n^- clusters where $n=1-4$
- Regardless of origin, each BP sample produced the same product ions
- ^{63}Ni -IM-TOFMS should be used in future black powder analyses



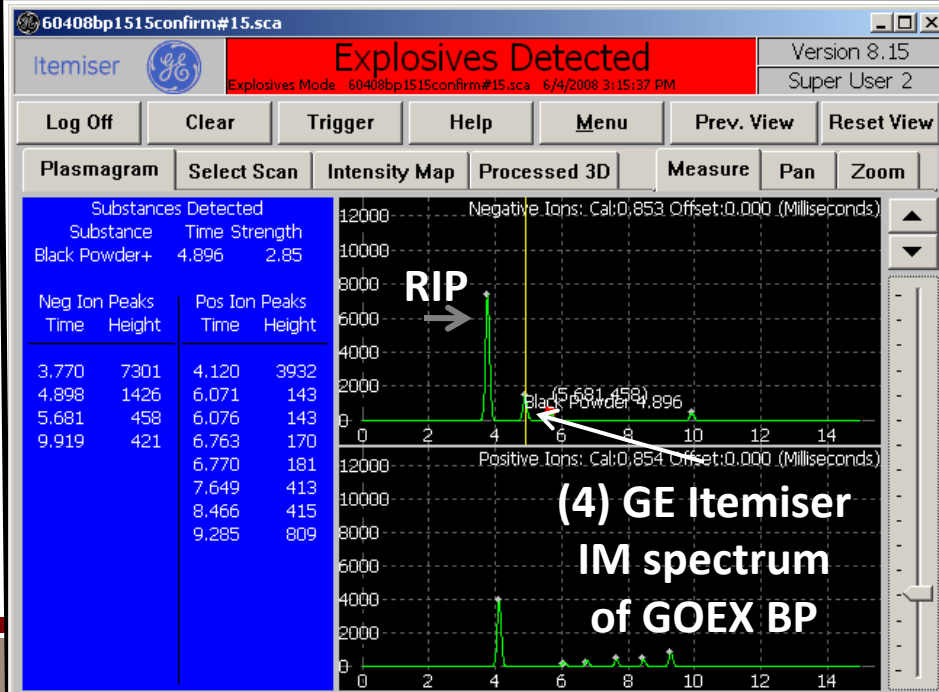
(1) 2D Mobility-Mass plot of GOEX Black Powder



(2) Inset showing most intense peak from plot (1)



(3) WSU standalone IM spectrum of GOEX BP



Reproducibility of BP Measured K_0 Values

- GOEX black powder on three IMS systems: IM-TOFMS, standalone IMS and GE Security's ITEMISER IMS
- Intense broad mobility peak sample with a K_0 of $2.29 \pm 0.03 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$

Instrument	System Type	Drift Time	K_0 ($\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$)	Nominal m/z value
IM-TOFMS (WSU)	Hybrid	8.331 ± 0.025	2.30	128
IMS (WSU)	Standalone	8.750 ± 0.000	2.31	Not applicable
Itemiser [®] ITMS (GE)	Standalone	4.176 ± 0.022	2.26	Not applicable

Evaluation of Interferences with Explosives

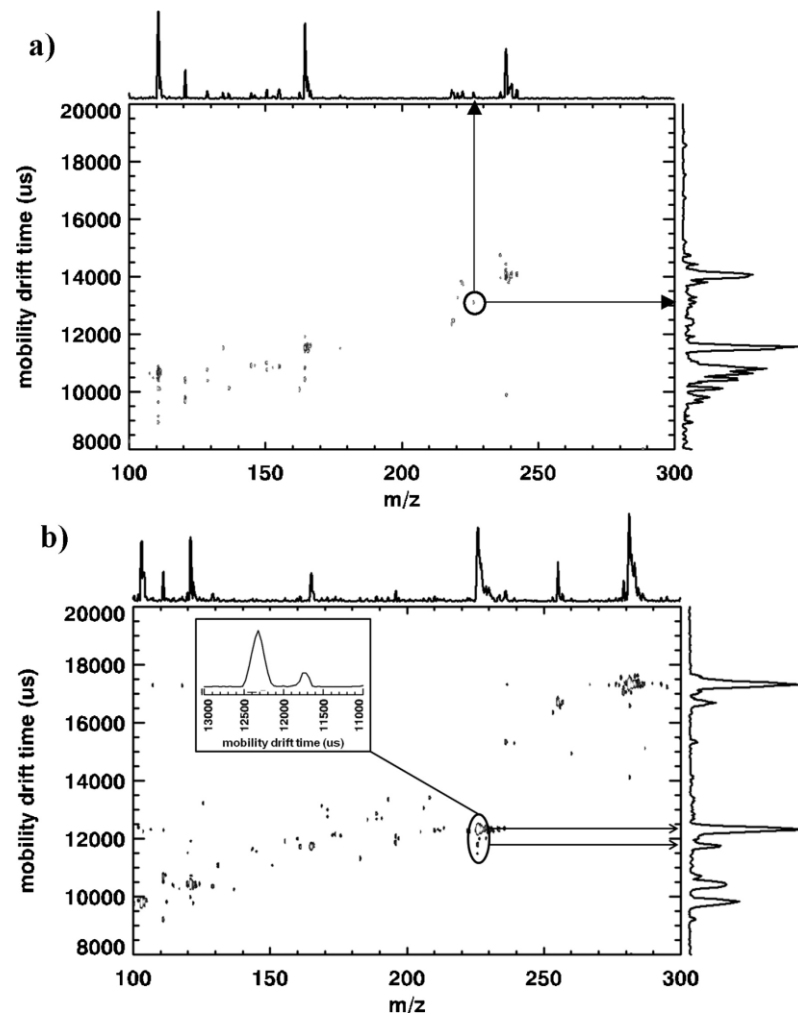
by IMMS

- Evaluate ingredients in common household products for mass and mobility interferences with explosives
- Demonstrate a need for simultaneous mobility-mass measurements for accurate threat detection in complex mixture analysis
- Published in *Analytica Chimica Acta*



Results & Conclusions

- Two common household ingredients influenced IMMS response for TNT and RDX
 - Enhanced and suppressed IMMS response
- Ions from the ingredients shared a similar m/z and drift time range for the common explosives
- The mobility separation often produced several K_0 values for each m/z identified ion species
- The samples were rich in isomeric species
 - The mobility separation found several more examples of isobaric species for the identified ions
- Reported new K_0 values for PCP ingredients



Summary & Overall Conclusions

- Ion Mobility Spectrometry (IMS) is a proven trace explosives detection technology
- Mass spectrometry coupled to IMS enhances both the IMS and MS data output
- The future of IMS and IMMS
 - Complex mixture characterization
 - Detection of trace, non-explosive threat compounds
 - Further hybridization with separation techniques (GC, LC, CE, etc.) and mass spectrometers (TOFs, etc.)

Acknowledgements



Institutional Support

- Washington State University (Pullman, WA)
- Dr. Herbert H. Hill, Jr. Research Group (Chemistry Dept.)



(Formerly GE Security)

Funding Support

- National Science Foundation
- SAIC
- Savannah River National Laboratory
- Safran Morpho (formerly GE Security)

Relevant References

1. C.L. Crawford, H. Boudries, R.J. Reda, K.M. Roscioli, K.A. Kaplan, W.F. Siems, and J. H.H. Hill, *Analysis of Black Powder by Ion Mobility-Time of flight-Mass Spectrometry*. Analytical Chemistry, 2010. 82: p. 387-393.
2. C.L. Crawford, S. Graf, M. Gonin, K. Fuhrer, X. Zhang, and J. H.H. Hill, *The Novel Use of Gas Chromatography-ion mobility, time of flight mass spectrometry with secondary electrospray ionization for complex mixture analysis*. International Journal of Ion Mobility Spectrometry, 2011. 14: p. 23-30.
3. C.L. Crawford, *Improving Ion Mobility Mass Spectrometry (IMMS) for National Security Threat Detection*, Washington State University: UMI/ProQuest.
4. C.L. Crawford, B.C. Hauck, J.A. Tufariello, C.S. Harden, V. McHugh, W.F. Siems, and H.H. Hill Jr, *Accurate and reproducible ion mobility measurements for chemical standard evaluation*. Talanta, 2012. 101(0): p. 161-170.
5. C.L. Crawford and J. Herbert H. Hill, *Mass Spectrometry in Homeland Security*, in *Mass Spectrometry Handbook*, M.S. Lee, Editor. 2012, John Wiley & Sons: New Jersey.
6. C.L. Crawford, G.A. Fugate, P.R. Cable-Dunlap, N.A. Wall, W.F. Siems, and H.H. Hill Jr, *The novel analysis of uranyl compounds by electrospray-ion mobility-mass spectrometry*. International Journal of Mass Spectrometry, 2013(0).
7. C.L. Crawford and J. H.H. Hill, *Evaluation of false positive responses by mass spectrometry and ion mobility spectrometry for the detection of trace explosives in complex samples*. Analytica Chimica Acta, 2013. 795: p. 36-43.
8. C.L. Crawford and H.H. Hill, *Comparison of reactant and analyte ions for ⁶³Nickel, corona discharge, and secondary electrospray ionization sources with ion mobility-mass spectrometry*. Talanta, 2013. 107(0): p. 225-232.

Contact Information

Christina L. Beppler (Crawford), Ph.D.
Org. 02555-MS1455, Sandia National Labs
(505) 284-9966, clcrawf@sandia.gov

