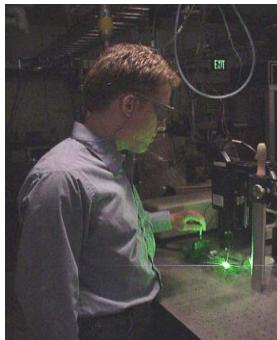


Exceptional service in the national interest



Detection of Trace Explosives by Ion Mobility-Mass Spectrometry

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Explosive Technologies Group

Unclassified-Unlimited Release



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Explosives Detection



Ion Mobility Spectrometry

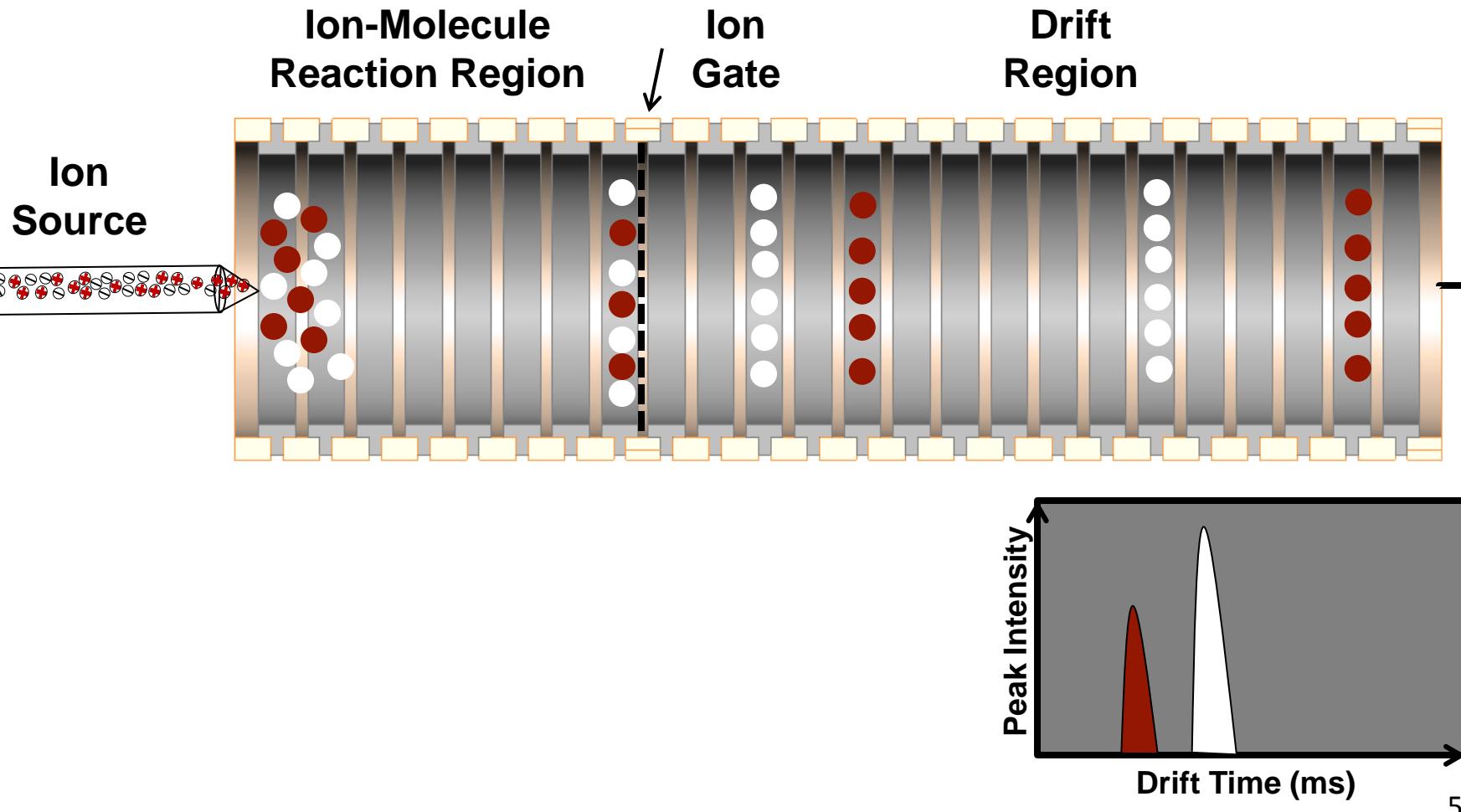
Commercial ^{63}Ni -IMS System



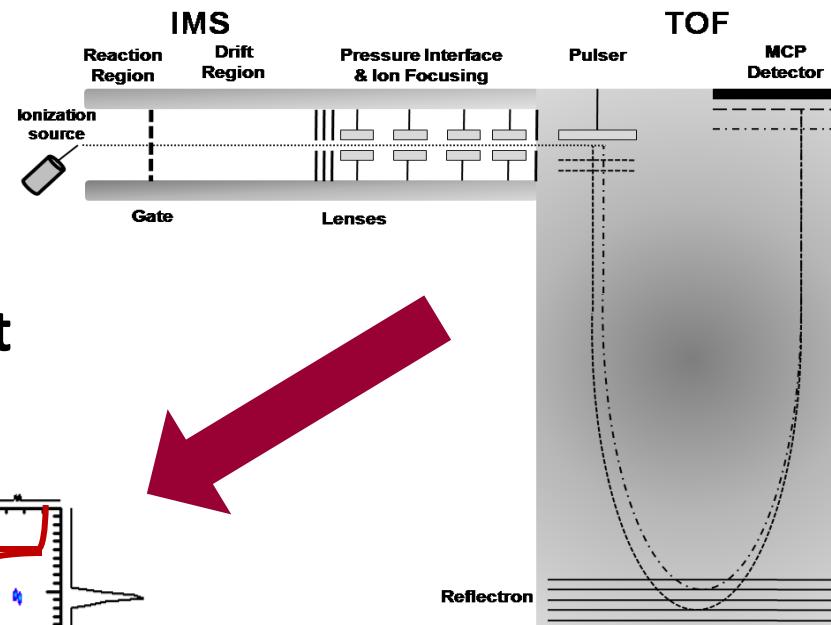
Advantages of IMS

- **Gas phase ion separation by size/charge**
- **Quick analysis cycle (under 6 seconds)**
- **Commercial systems easy to operate; currently deployed worldwide**
- **Air is carrier gas**
- **Good qualitative instrument**
- **Sub ppb detection limit for many explosives**

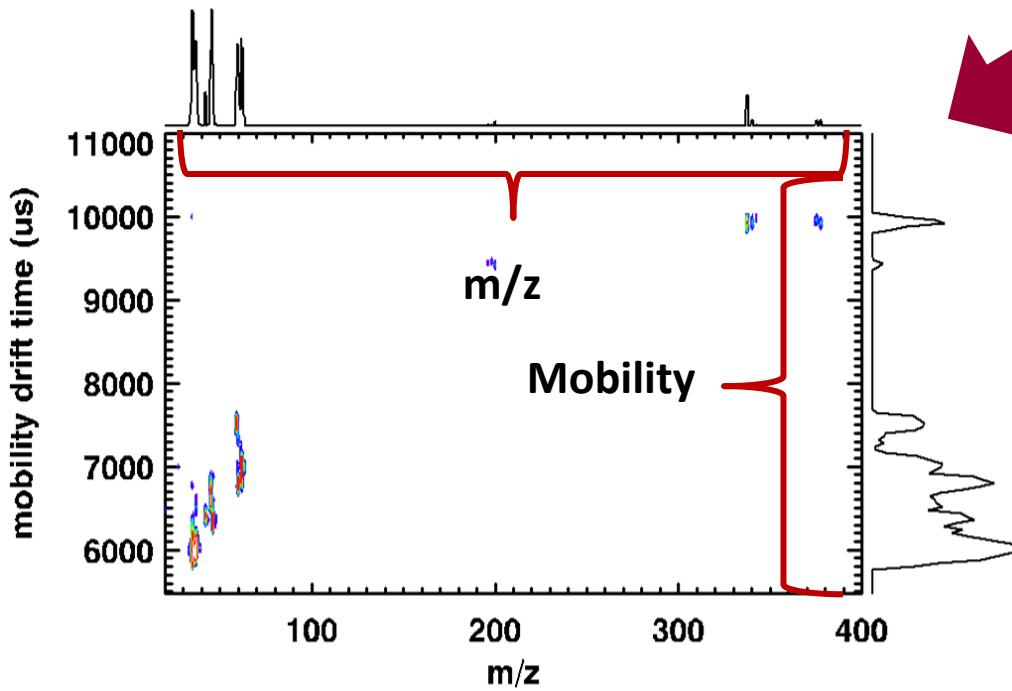
Ion Mobility Spectrometry



Simultaneous Mobility-Mass ID

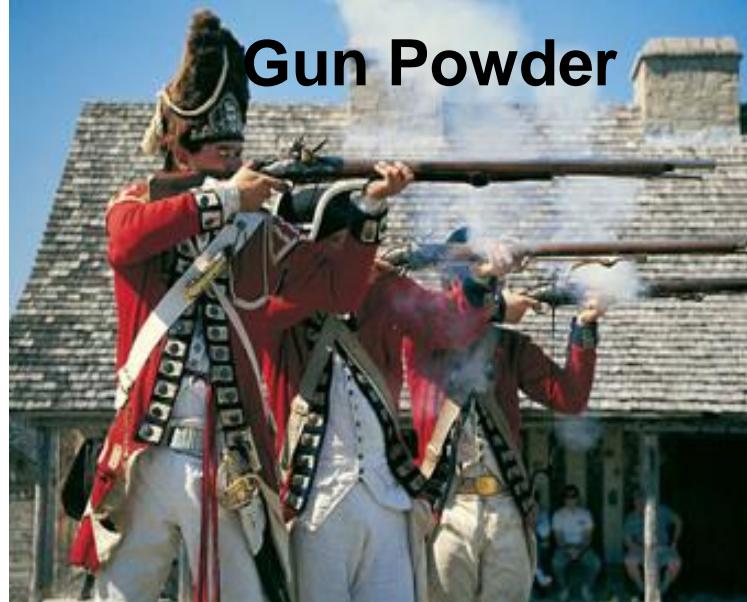


2D Mobility-Mass Plot



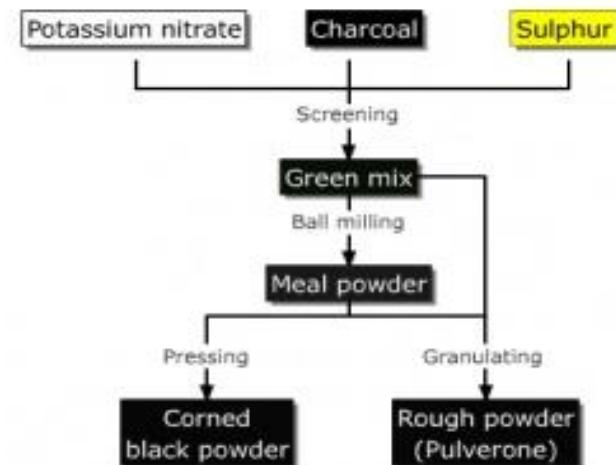
*Reduced Mobility values
identify analytes in IMS*

$$K_0 = \left(\frac{L^2}{Vt_d} \right) \left(\frac{P}{P_0} \right) \left(\frac{T_0}{T} \right)$$



Black Powder by IMMS

- Obtain a K_0 value for black powder
- Mass ID the ion species that make up the mobility peak
- Determine differences in K_0 values for black powders from differing origins
- Published in *Analytical Chemistry*



Previous Black Powder Detection Methods

Direct Chemical Methods

- Chemical spot testing: using a reagent that confirms presence of potassium nitrate or sulfur

Spectroscopic Methods

- Refractive Index measurements, x-ray powder diffraction, emission spectrography, polarizing microscopy

Chromatographic Methods

- TLC, ion chromatography/conductivity detection

IMS

- Barringer IONSCAN Model 200

MS

- TOF-SIMS & LDI-TOF-MS

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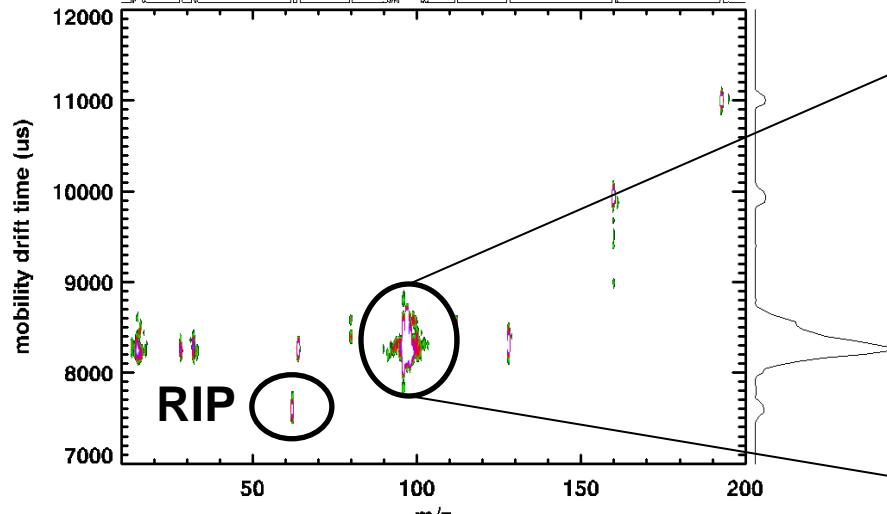
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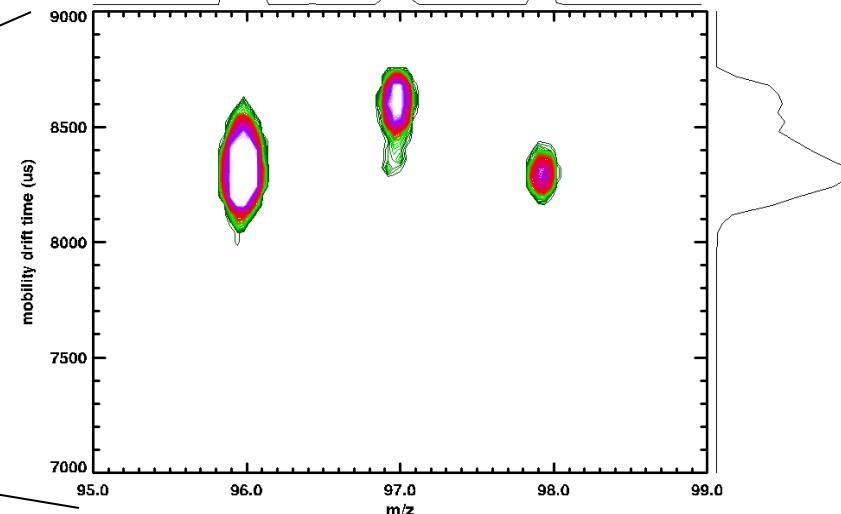
Results & Conclusions

- Black powder successfully detected using ^{63}Ni -IM-TOFMS
- *First reported* K_0 value for the primary black powder mobility peak at $2.28 \pm 0.02 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$
 - S_n^- clusters where $n=1-4$
- Regardless of origin, each BP sample produced the same product ions
- ^{63}Ni -IM-TOFMS should be used in future black powder analyses

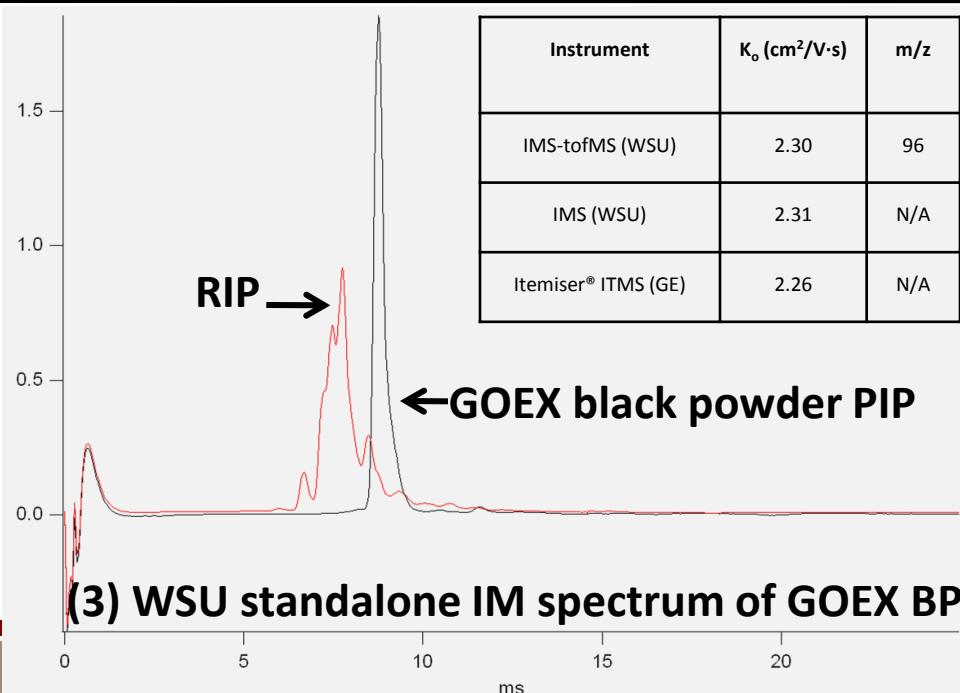
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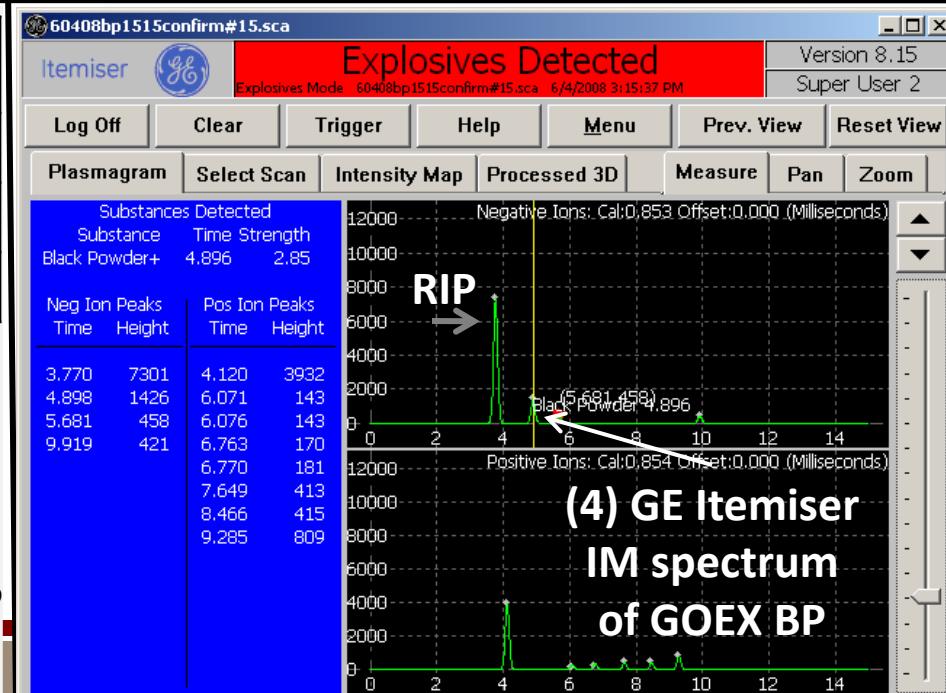
(1) 2D Mobility-Mass plot of GOEX Black Powder



(2) Inset showing most intense peak from plot (1)



(3) WSU standalone IM spectrum of GOEX BP



Reproducibility of BP Measured K_0 Values

- GOEX black powder on three IMS systems: IM-TOFMS, standalone IMS and GE Security's ITEMISER IMS
- Intense broad mobility peak sample with a K_0 of $2.29 \pm 0.03 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$

Instrument	System Type	Drift Time	K_0 ($\text{cm}^2 \text{V}^{-1} \text{s}^{-1}$)	Nominal m/z value
IM-TOFMS (WSU)	Hybrid	8.331 ± 0.025	2.30	128
IMS (WSU)	Standalone	8.750 ± 0.000	2.31	Not applicable
Itemiser® ITMS (GE)	Standalone	4.176 ± 0.022	2.26	Not applicable

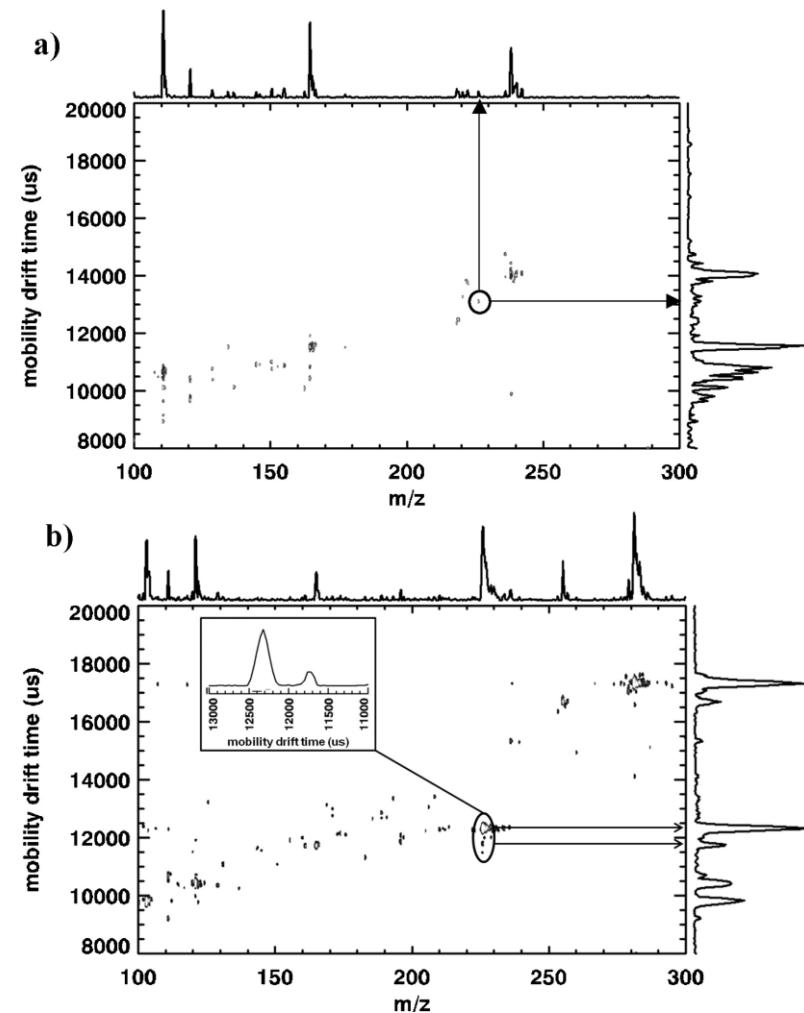
Evaluation of Interferences with Explosives by IMMS

- Evaluate ingredients in common household products for mass and mobility interferences with explosives
- Demonstrate a need for simultaneous mobility-mass measurements for accurate threat detection in complex mixture analysis
- Published in *Analytica Chimica Acta*



Results & Conclusions

- Two common household ingredients influenced IMMS response for TNT and RDX
 - Enhanced and suppressed IMMS response
- Ions from the ingredients shared a similar m/z and drift time range for the common explosives
- The mobility separation often produced several K_0 values for each m/z identified ion species
- The samples were rich in isomeric species
 - The mobility separation found several more examples of isobaric species for the identified ions
- Reported new K_0 values for PCP ingredients



Summary & Overall Conclusions

- Ion Mobility Spectrometry (IMS) is a proven trace explosives detection technology
- Mass spectrometry coupled to IMS enhances both the IMS and MS data output
- The future of IMS and IMMS
 - Complex mixture characterization
 - Detection of trace, non-explosive threat compounds
 - Further hybridization with separation techniques (GC,LC, CE, etc.) and mass spectrometers (TOFs, etc.)

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