

April 5, 2013

Electrochemical Solution Growth of Gallium Nitride

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Michael Russel, Viswanath
Krishnamoorthy, Ana-marie Mollo,
Linda E. Johnson



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Project Objective



You are here

To develop a novel, scalable, cost-effective growth technique for producing high quality, low dislocation density bulk gallium nitride for substrates for GaN-based LEDs, lasers, power electronics, and other interesting devices.

Motivation for Bulk Growth

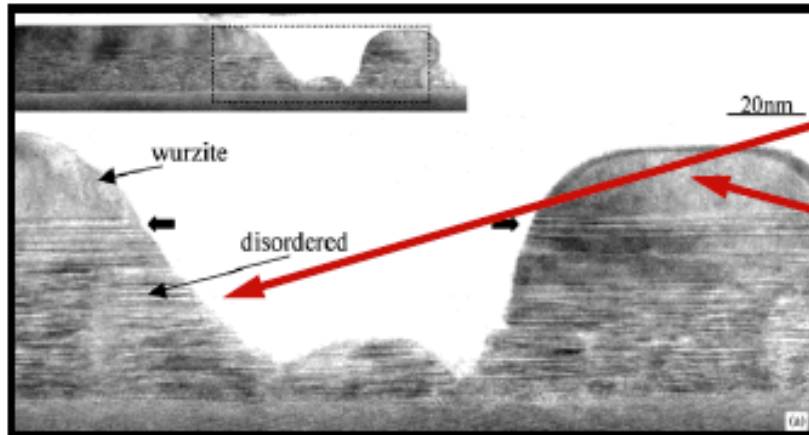
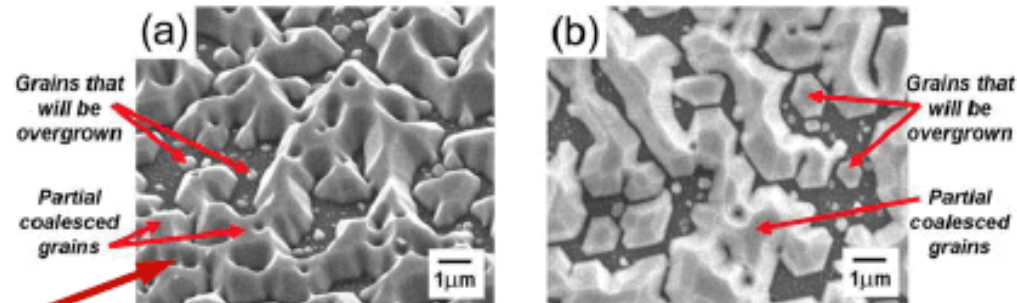


Figure from Lada et al., J. Crystal Growth 258, 89 (2003).

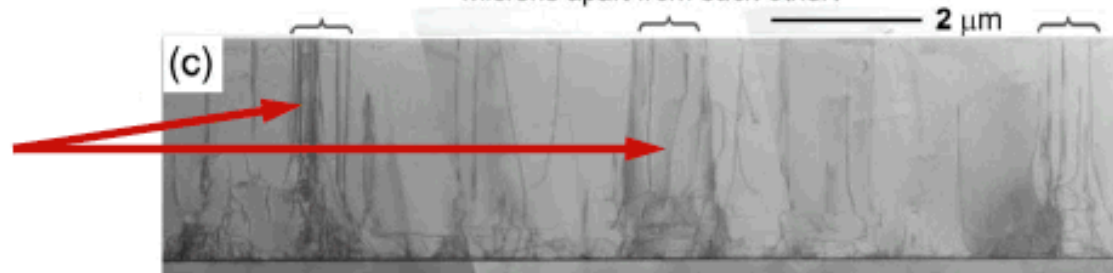
- High temperature growth on the GaN nuclei produces GaN grains.
- Growth conditions can be varied to enhance the pyramidal growth mode or lateral coalescence. Dislocations are bent laterally on pyramidal facets.
- Dislocations are concentrated in bunches located microns apart.

- As grown GaN nucleation layers contain disordered GaN with many stacking faults.
- Once annealed, wurtzite GaN forms on top of disordered GaN NL, forming nano-sized GaN nuclei from which further high temperature GaN growth occurs.

SEM Images of 3D GaN grain growth



The threading dislocation appear in bunches which are located a few microns apart from each other.

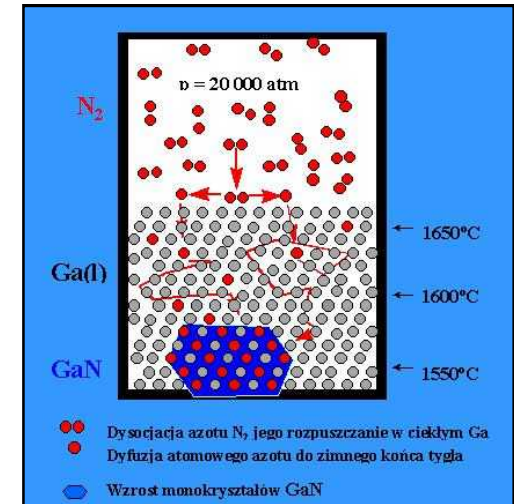


TEM cross section

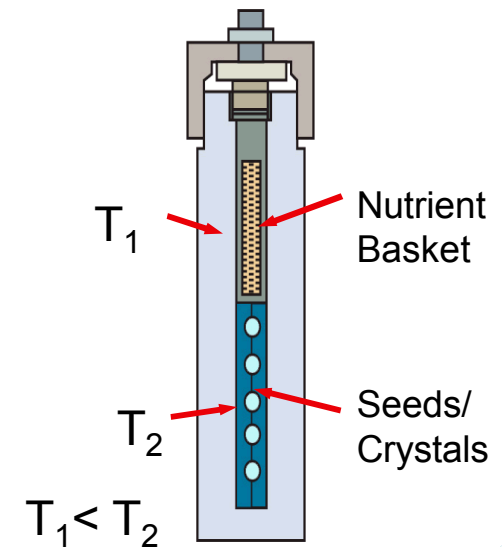
TEM courtesy of D. M. Folstaedt

State of the Art in Bulk GaN Growth

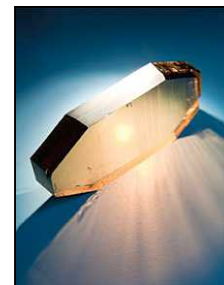
True bulk GaN not yet readily available



High Nitrogen Pressure
Solution Growth



Ammonothermal



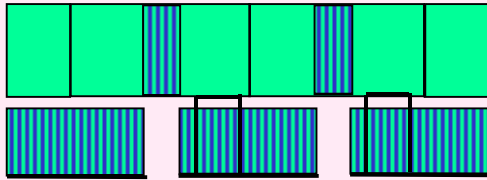
Ammono
IEEE Spectrum

Scalability Limited, Cost-Prohibitive

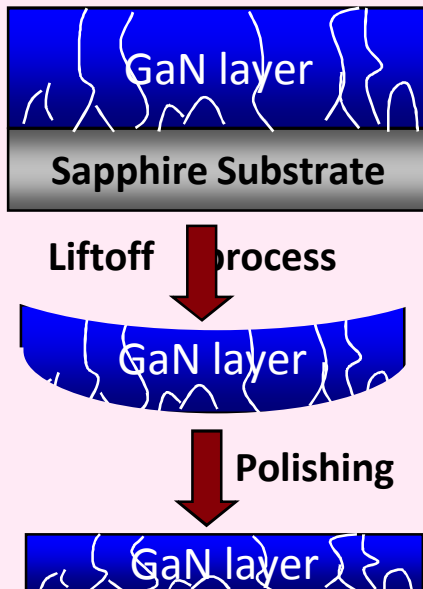
Hydride Vapor Phase Epitaxy (HVPE)

Dislocation Filtering Techniques

Lateral Overgrowth



HVPE



- **Gas phase approaches require high quality substrate**
 - *Sapphire or SiC, or bulk seeds*
 - *Quality not high, limited in size*
 - *Very expensive*
- **Difficulty with impurity incorporation on different crystal facets**

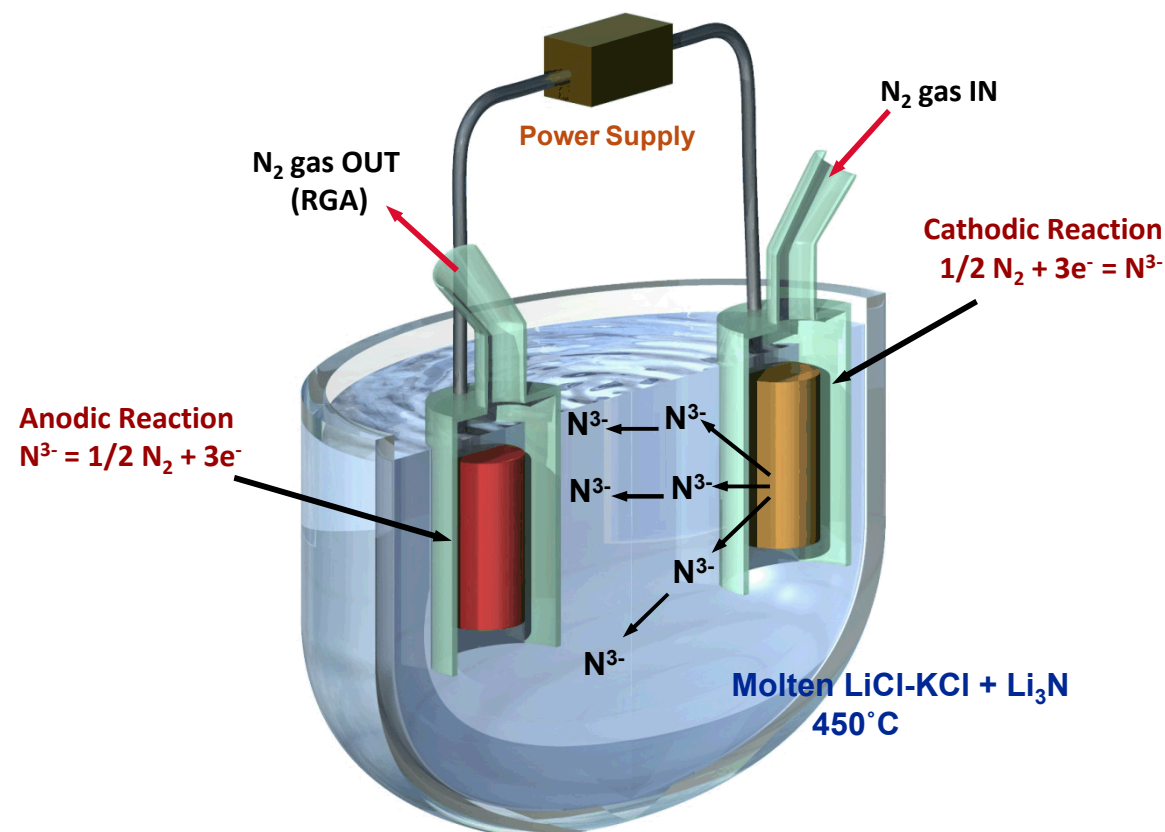
Desires/Requirements for a Bulk GaN Growth Technique

- Good crystalline quality ($\rho \leq 1 \times 10^5 \text{ cm}^{-2}$)
- High growth rate ($\sim \text{mm/hr}$): high throughput, high volume production
- Low impurity content
- Low background carrier concentration ($\sim 1 \times 10^{16} \text{ cm}^{-3}$)
- Scalable
- Controllable
- Manufacturable
- Reasonably inexpensive
- Applicable to InN, GaN, AlN, and III-N alloys

$1/2\text{N}_2 + 3\text{e}^- \rightarrow \text{N}^{3-}$: The Reactive Intermediate

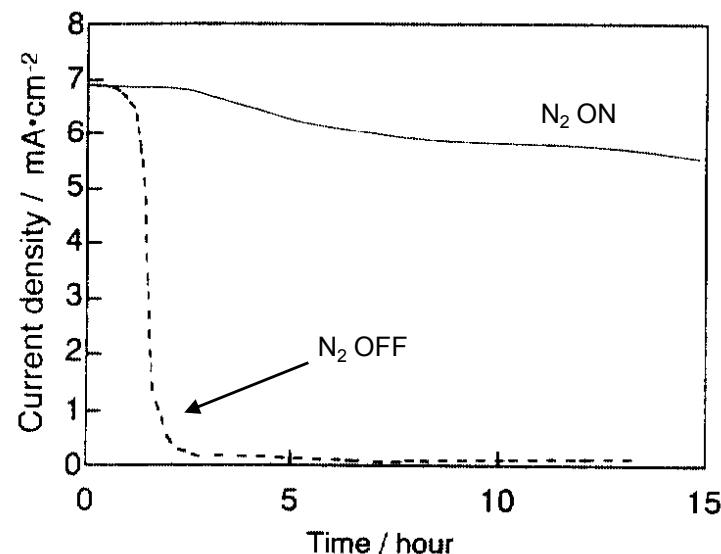
T. Goto and Y. Ito, "Electrochemical reduction of nitrogen in a molten chloride salt"
Electrochimica Acta, Vol. 43, Nos 21-22, pp 3379-3384 (1998).

Found that nitrogen was **continuously** and **nearly quantitatively** reduced to nitride ions



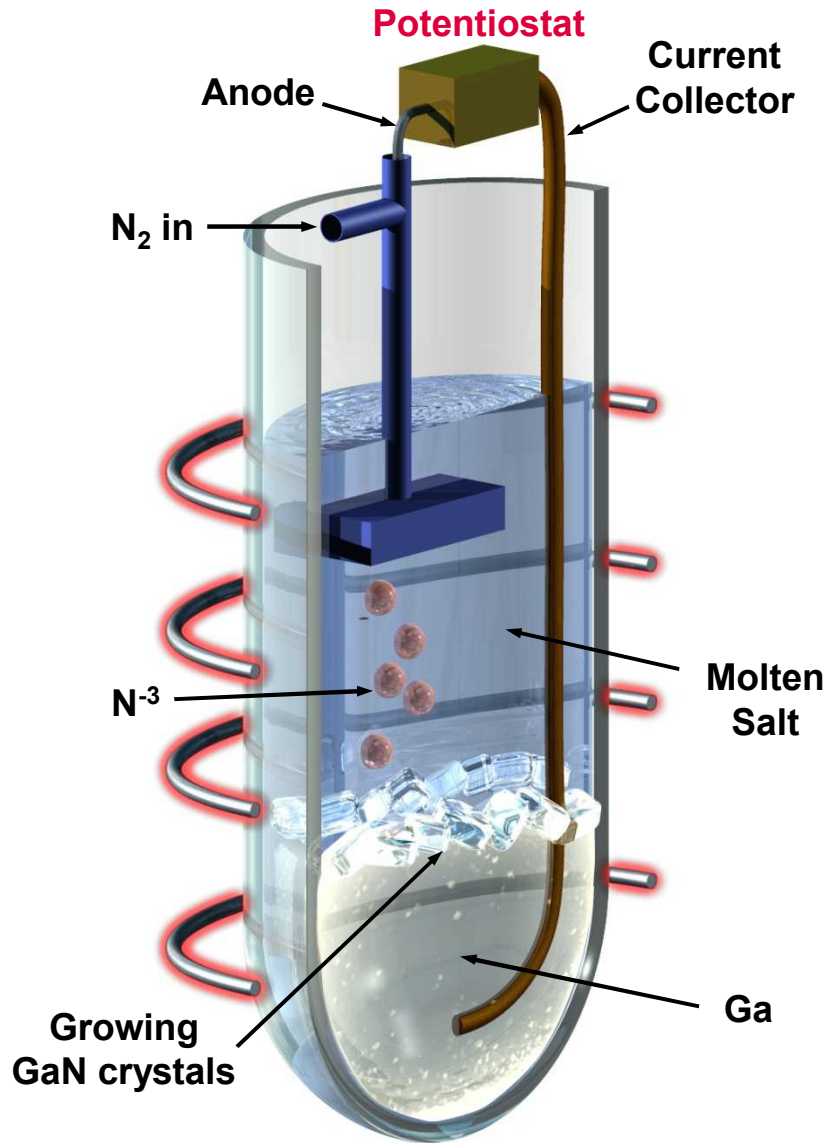
Advantages of using N_2 gas:

- Clean
- Inexpensive
- Control over precursor conc.
- Continuous, controlled supply



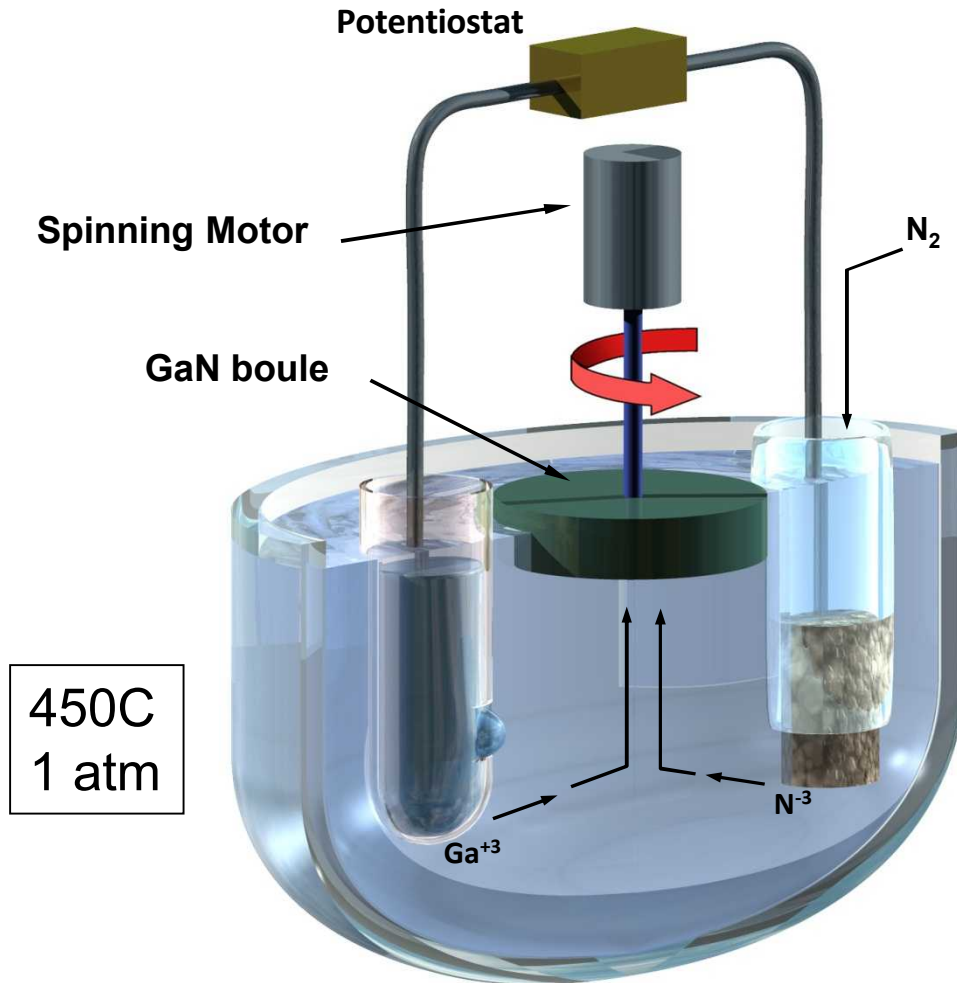
Report of nitride concentration in LiCl
in literature: 12 mole %

Electrodeposition is Not a Viable Technique for Thick, Insulating, High Quality Crystals

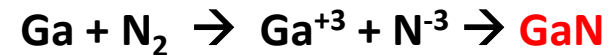


- Tends to produce rough deposits for low-conductivity materials
- Gallium source on one side of crystal, nitride source on the other
 - Diffusion of either species through crystal very, very low
 - How to get Ga and N ions to same surface??
- However, it was a quick and inexpensive method for demonstrating viable chemistry and favorable kinetics
- Nitride ion available in large quantities... use this capability to our advantage

Sandia's Patented New Growth Technique: Electrochemical Solution Growth (ESG)



Create Ionic Precursors
Electrochemically:



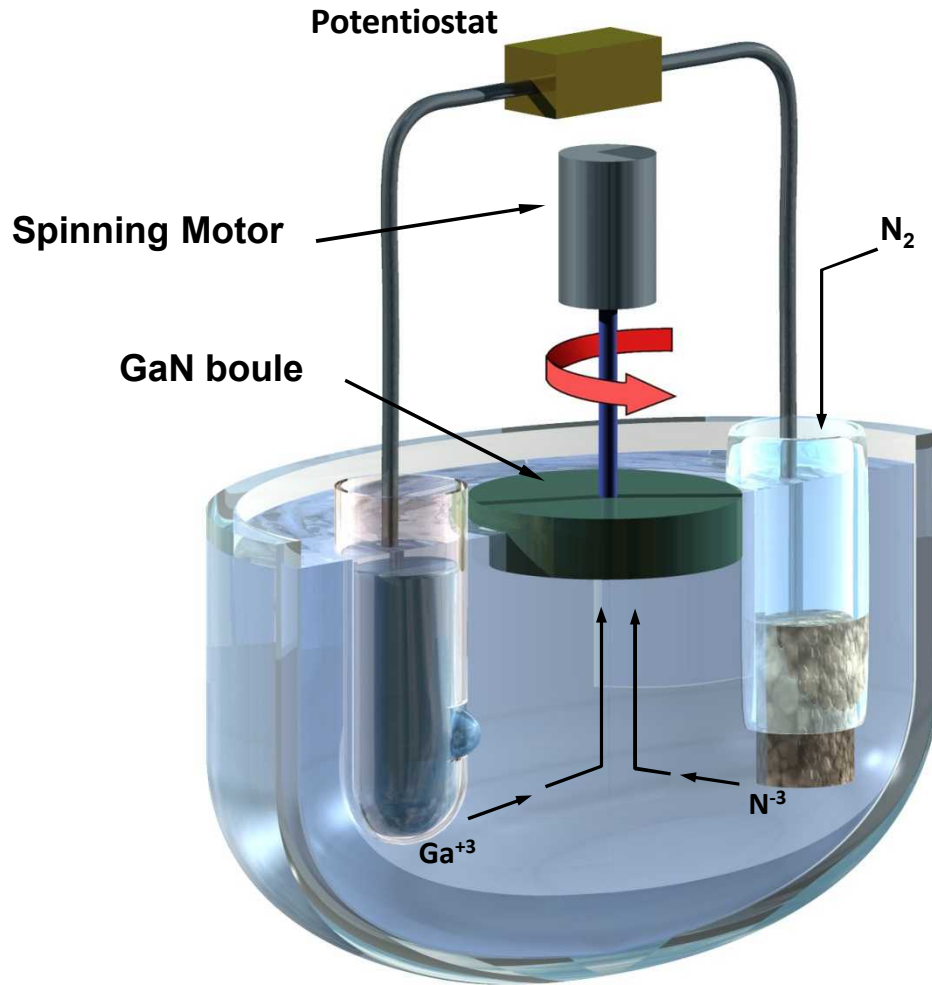
Use salt flow to deliver precursors to
seed crystal surface

Increase growth rate through flux of
reactants (increase spin rate)

Precursors can be replenished as they are consumed
Advantage: Continuous, isothermal or steady-state growth

U.S. Patent Issued October 2008

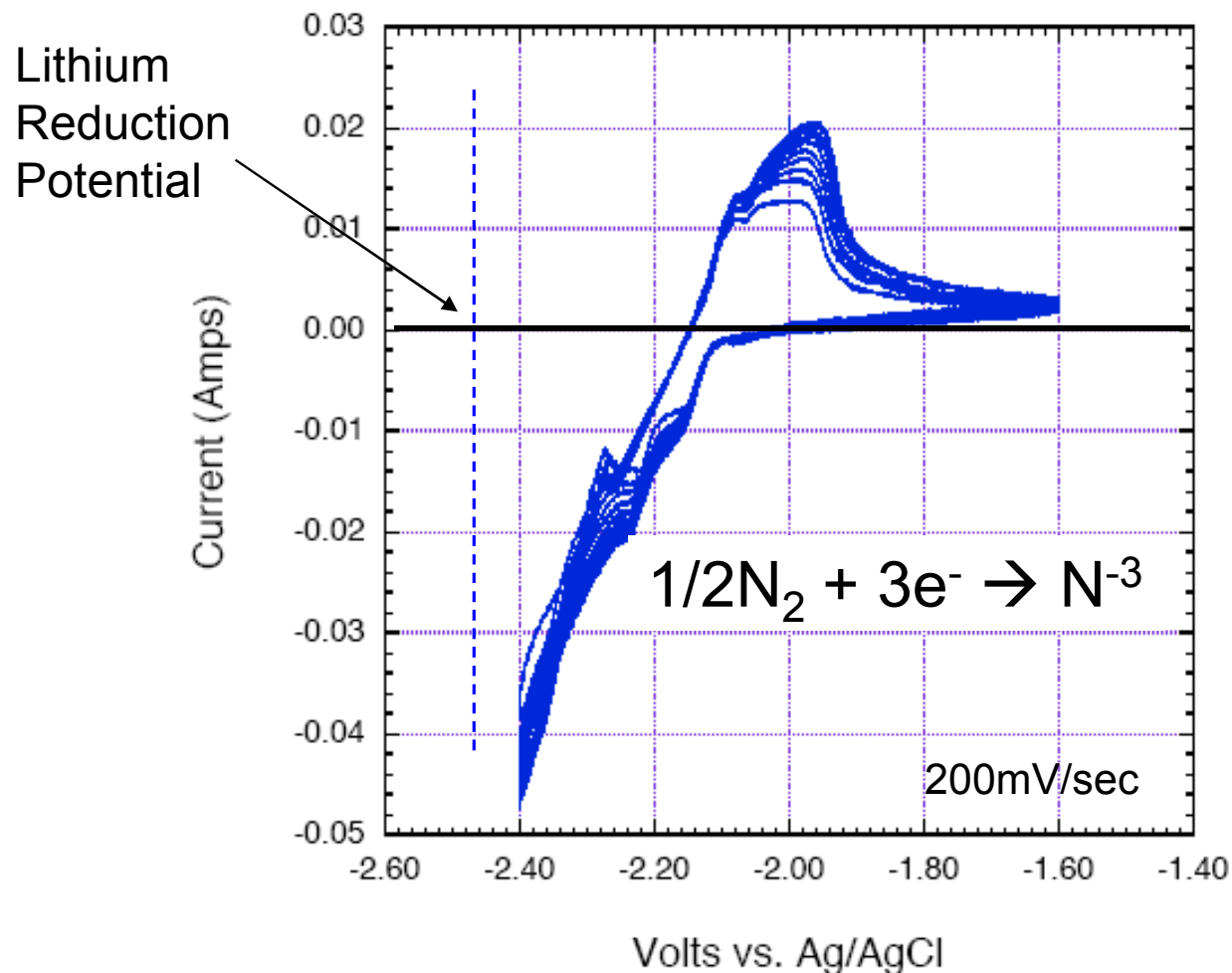
Electrochemical Solution Growth (ESG): Current Status



- Electrochemistry studies
- Preliminary fluid dynamics
 - Growth rates $\sim \text{mm/hr}$
- Chemistry studies
- ESG autonucleation:
 - mm-sized crystals in 2 hrs
 - Bandedge photoluminescence
- ESG boule growth:
 - Deposition of GaN at seed surface

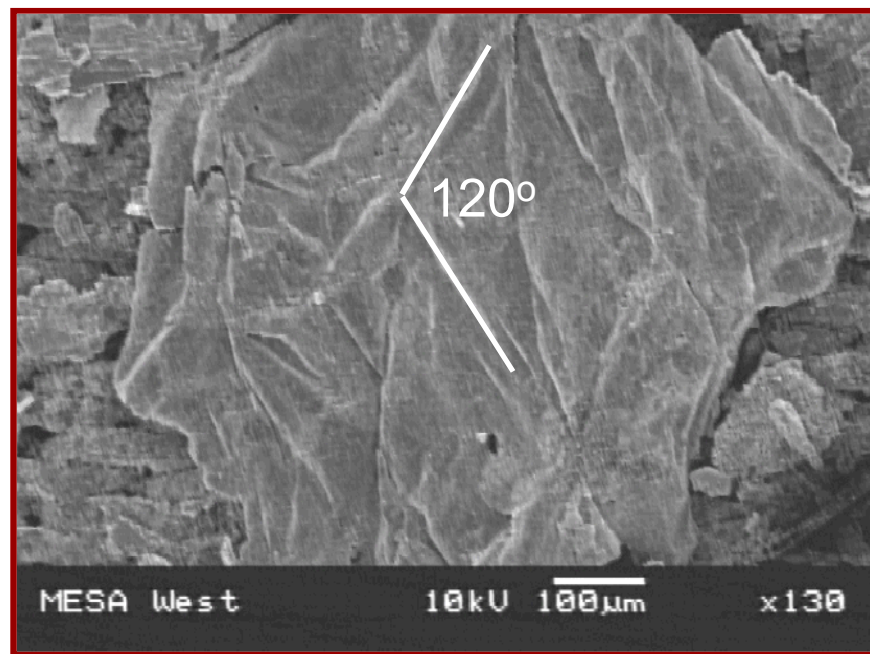
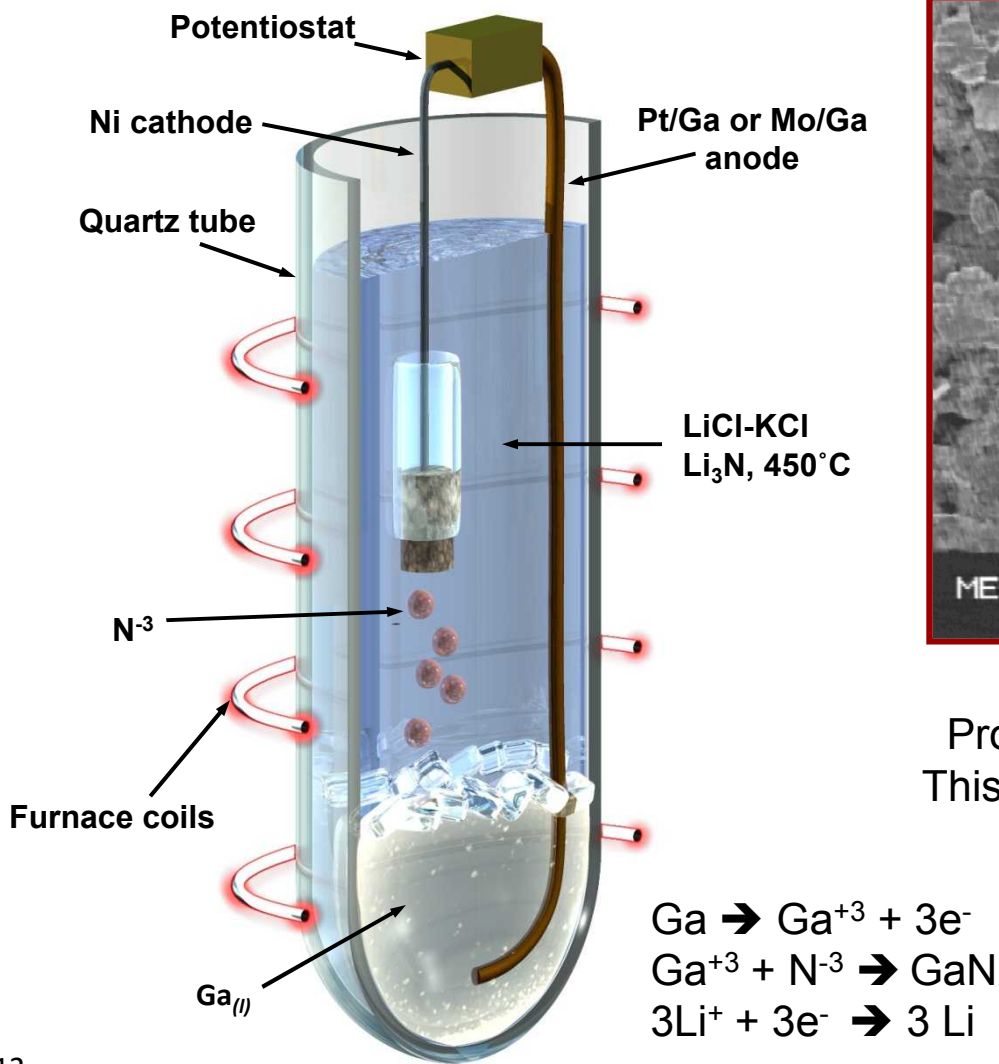
Next step: Developing quality of crystals

Example of Nitrogen Gas Reduction Cyclic Voltammograms



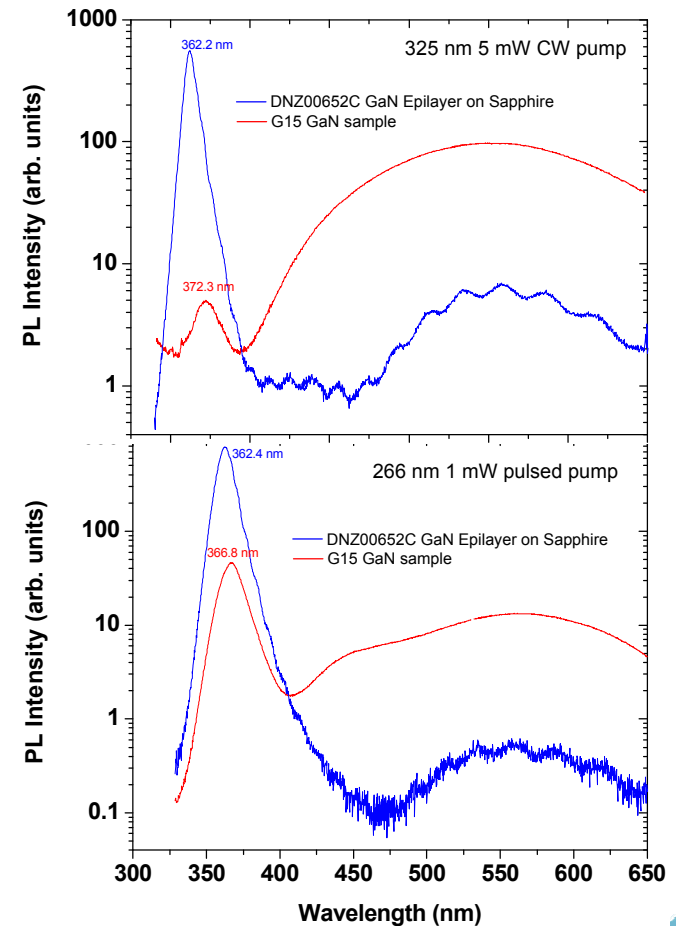
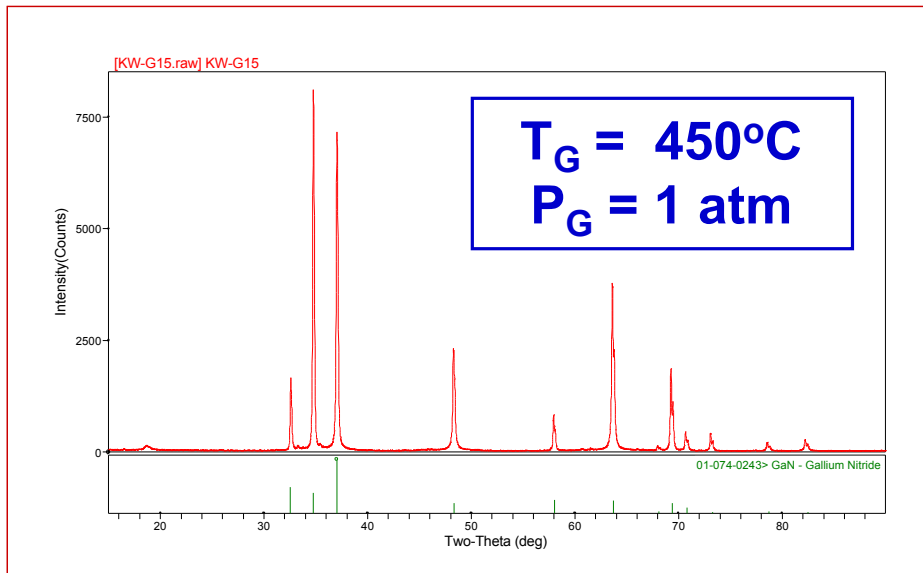
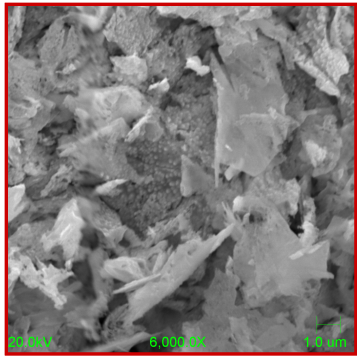
Autonucleation Experiments of GaN Produce mm-sized crystals in <2hrs.

$\text{Li}_3\text{N} + \text{Ga}$, 450°C



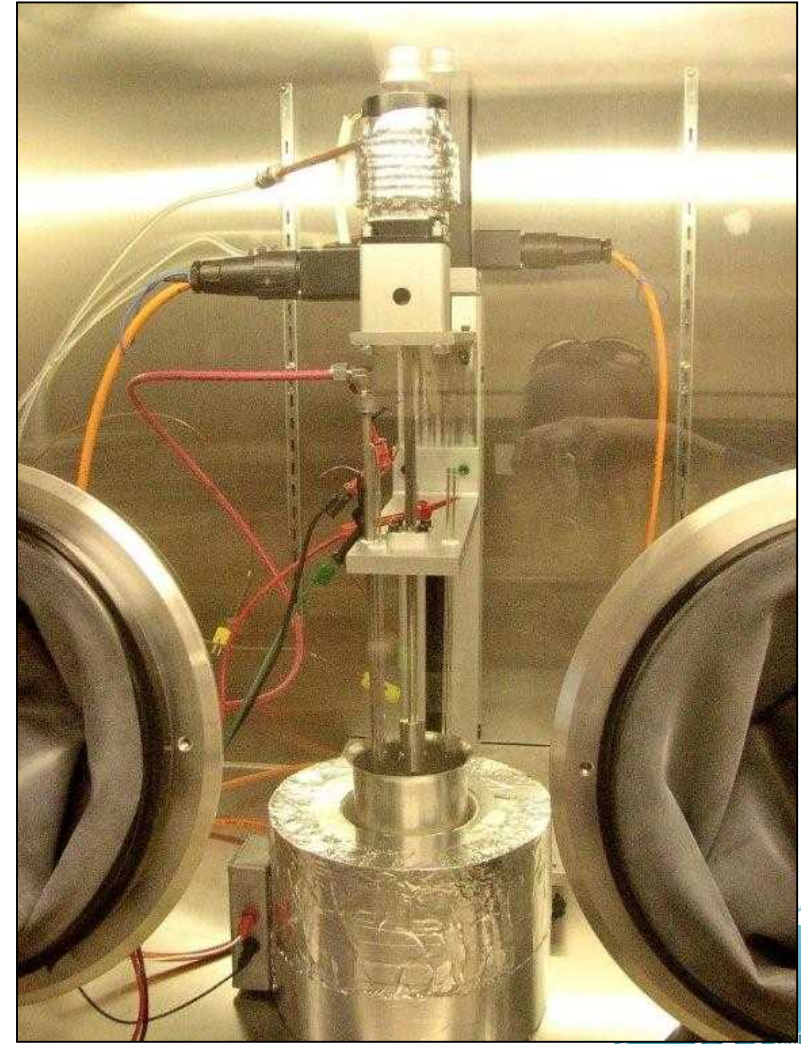
Produced numerous wurtzite GaN crystals;
This crystal was $\sim 1.25\text{mm}$ long x 0.8mm wide

GaN ESG Produces Photoluminescent GaN Crystallites

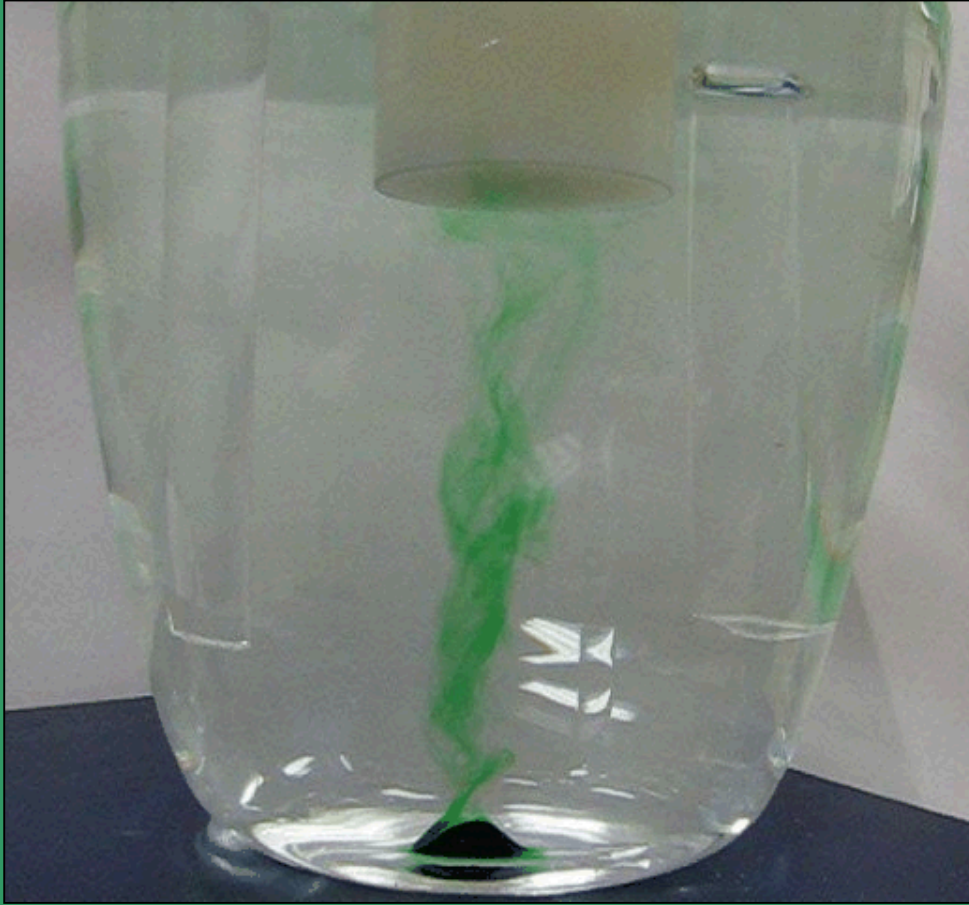


Mary Crawford, SNL

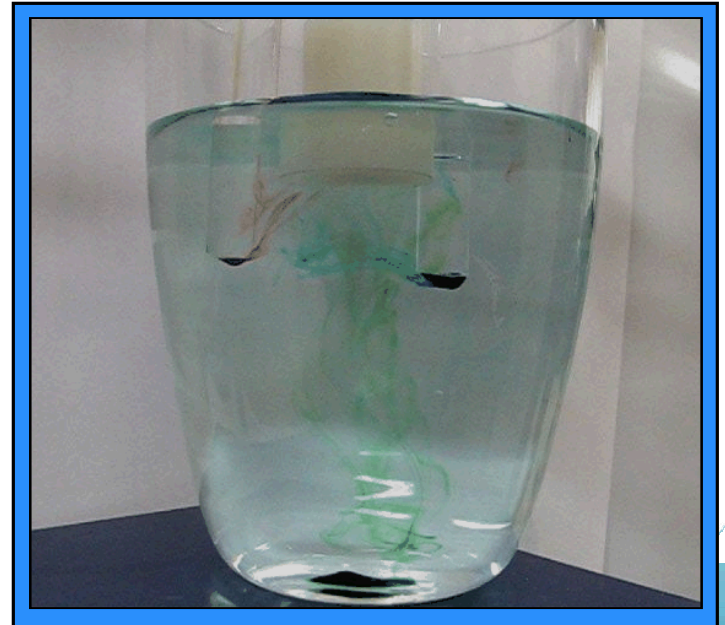
Gen-2 Boule Growth Reactor Up to 1.5" diameter



Room Temperature Fluid Dynamics Experiments



Under laminar flow conditions, the spinning seed will draw nutrient-containing fluid to the surface— if the electrodes are properly shaped and located.



Innovative Manufacturing Initiative Award from the Department of Energy's Advanced Manufacturing Office (AMO)

- Sun Edison (Michael Seacrist)/Sandia (Karen Waldrip)/GTRI (Prof. Russ Dupuis)
- 3 year, \$4.6M award with cost share, began 9/1/2012
- Major Year 1 goals:
 - Develop reproducible ESG process on 1 cm² MOCVD-grown GaN seed crystal, evaluate growth rate
 - Develop and validate ANSYS growth models including electrochemistry
- Major Year 2 goals:
 - Scale up to 50 mm diameter seed crystal
 - Sustain growth to > 1mm
 - Grow and characterize epi layers on seed surfaces
- Major Year 3 goals:
 - Scale to 100 mm diameter
 - Demonstrate functional optical and electrical devices on ESG-grown material
 - Design, build and install 150 mm reactor

Conclusions

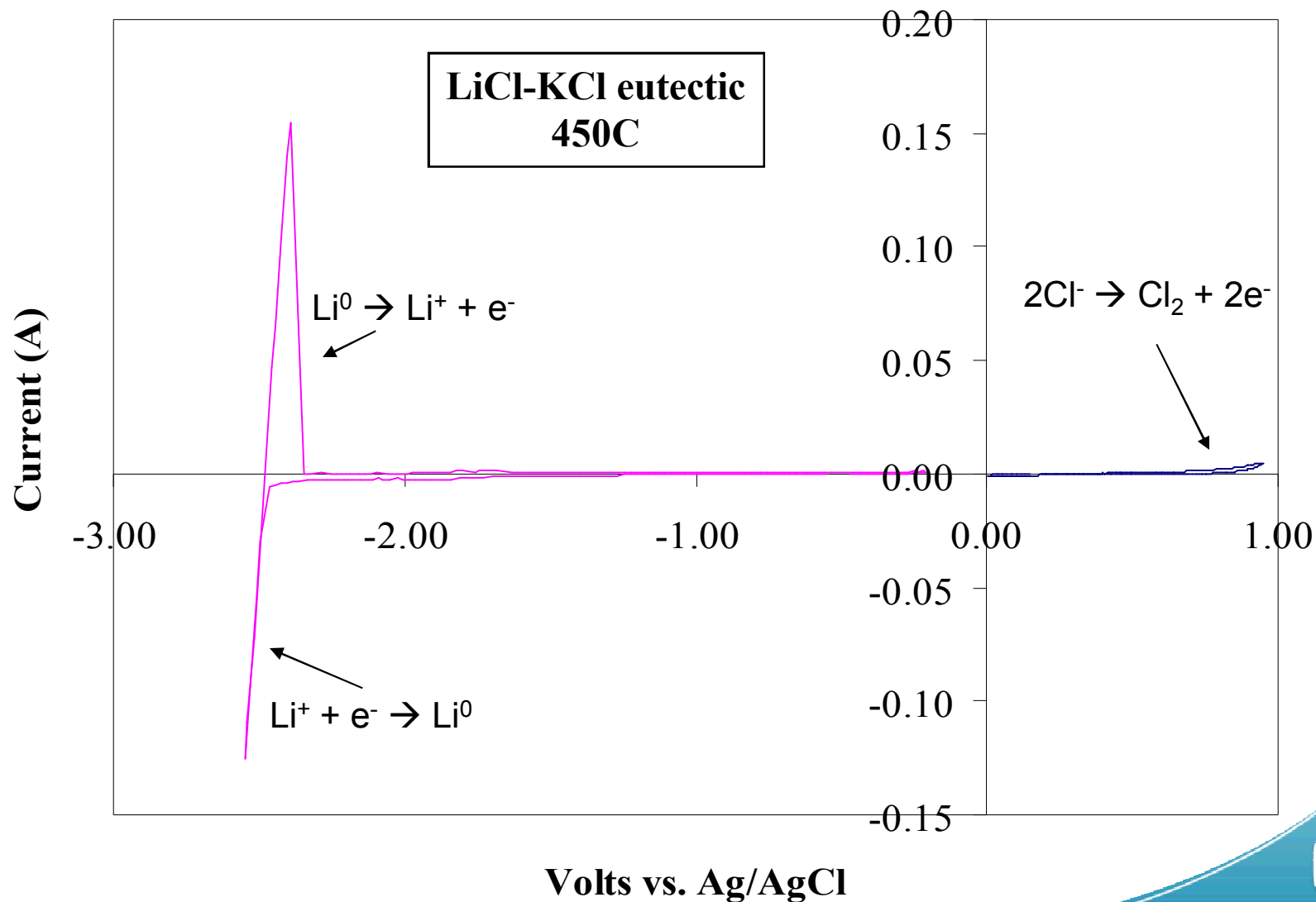
- **Electrochemical reduction of N_2 and autonucleation of GaN in eutectic LiCl/KCl salt demonstrated**
- **Modeling and experiments show growth of GaN on seed substrate possible**
- **Scaling up process to grow GaN on 1 cm² (and greater) seed substrates**

Acknowledgements:

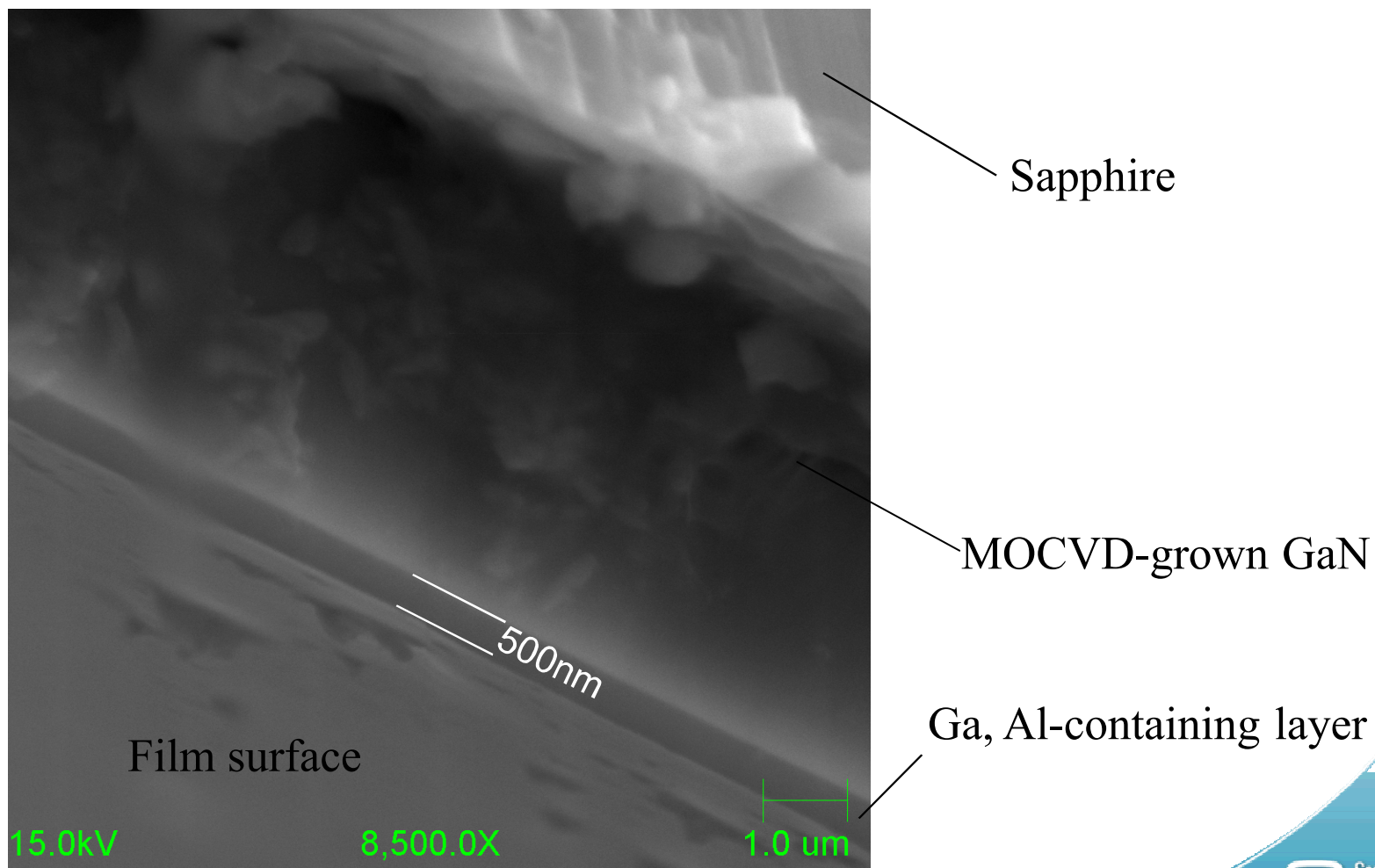
- **DOE Office of Electricity Delivery and Energy Reliability**
- **DOE Advanced Manufacturing Office (AMO)**

Extra Slides

Complete Electrochemical Background of Salt



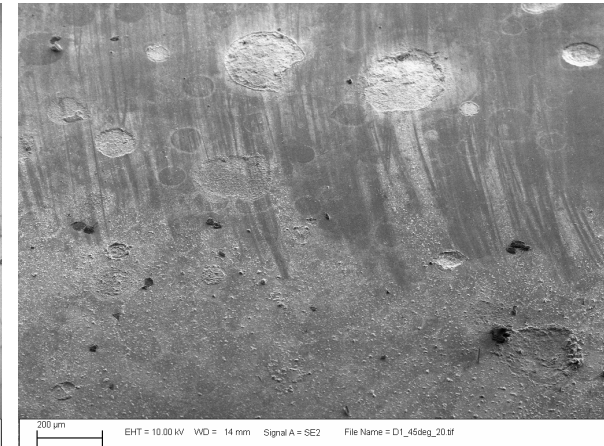
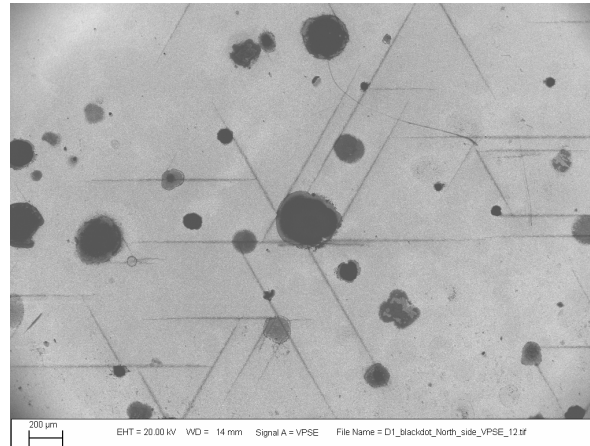
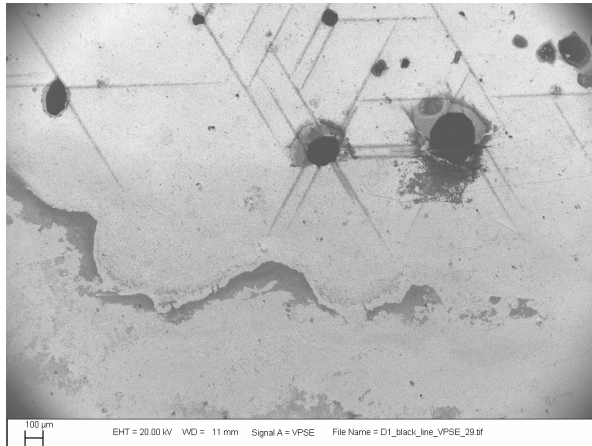
SEM of SNL RD-ESG Growth Run #1



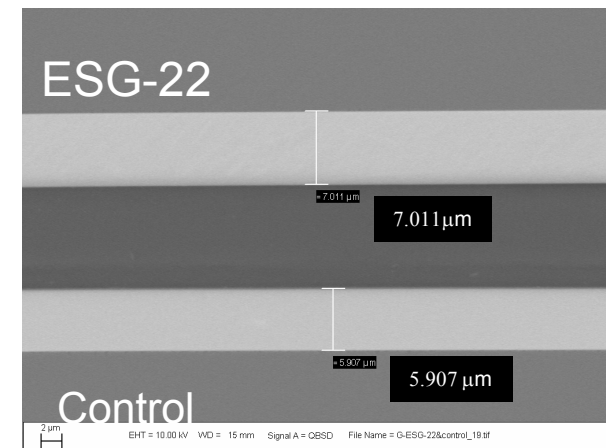
XPS and SIMS of SNL RD-ESG #1

- XPS showed Ga, N, C, O, Si, Al, etc.
- SIMS revealed the layer to be a graphitic carbon layer, with Ga, N, and GaN clusters inside
 - Ga,N content was about 10%
 - Profile was consistent with an increasing concentration
- Problem with salt purity from supplier
 - Developing in-house purification technique for reagent grade salt

First GNOEM RD-ESG Experiment

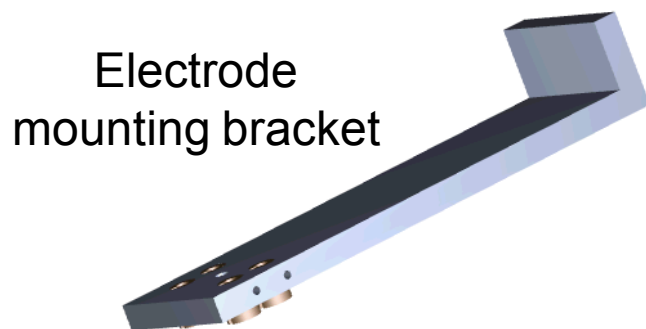


- Growth at GNOEM Systems
- Hardware failure—susceptor sheared, not sure when
- Black line on sample surface delineated a thicker, specular region and thinner, roughened area
- *Defect selective etching observed* (several microns/hr)
- Highly encouraging for crystal quality
- Polished cross sections of control and experiment sample consistently measure about $1\mu\text{m}$ thicker for experiment



Bonnie MacKenzie, SNL

Boule Growth Reactor Design



Previous Setup

Ring Stand & Clamps --
irreproducible and clumsy

Temperature Control through
variac power supply

Quartz crucibles and electrodes

Cement mounting of seed crystal

Reactor located in secure area



New Reactor

Positional and rotational control through
computer-controlled stepper motors and
machined mounting bracket

Temperature Control through reactor temperature
feedback to power supply

p-BN and/or stainless steel materials

Precision-machined SS mechanical susceptor

Accessible to uncleared personnel

Acknowledgements: M.J. Russell, P. Michel

Nature Deposits $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ Single Crystals from Solutions of Ionic Precursors



Giant crystal caves, Naica, Mexico
Photo from National Geographic

Growth Rate Calculations

From MBE:

$$\phi = \text{\#atoms}/(\text{surface unit/monolayer thickness}) * \text{GR}$$

$$\phi = (n/S/c/2) * \text{GR}$$

$$S = \frac{a^2 c (3)^{1/2}}{4n}$$

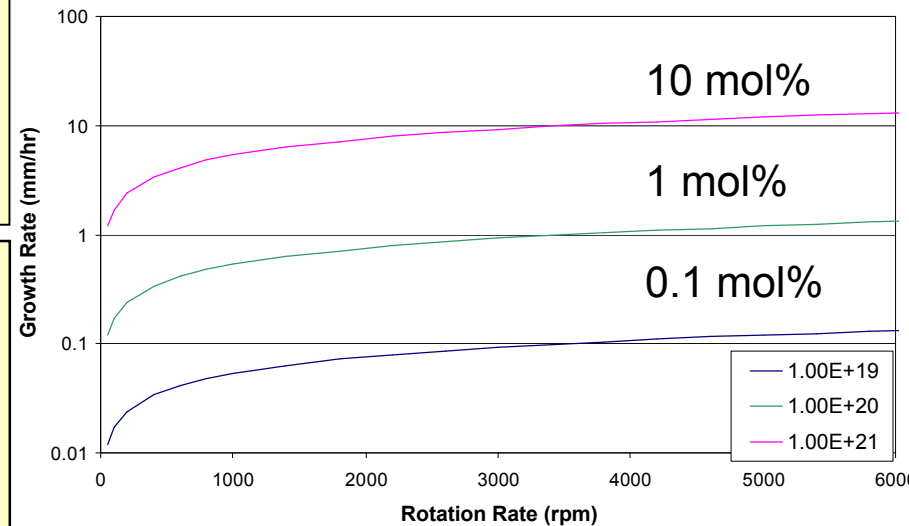
$$\text{GR} = \frac{a^2 c (3)^{1/2} \phi}{4}$$

Levitch Equation (from RDE):

$$i_{l,c} = 0.62nFAD_o^{2/3}\omega^{1/2}\nu^{-1/6}C_o^*$$

$$\frac{1}{A} \frac{dN}{dt} = \phi = 0.62D_o^{2/3}\omega^{1/2}\nu^{-1/6}C_o^*$$

Growth Rate vs. Rotation Speed and Concentration



$$\text{GR (mm/hr)} = \frac{a^2 c (3)^{1/2} * 0.62D_o^{2/3}\omega^{1/2}\nu^{-1/6}C_o^*}{4}$$

Growth Rate Calculations

- Growth Rate = GR
- a, c = lattice constants
- ω = rotation rate (rpm)
- C_o^* = bulk concentration
- D_o = diffusion constant
- y_h = hydrodynamic boundary layer thickness
- δ_o = diffusion layer thickness
- m_o = mass transfer coefficient
- ϕ = flux (at/cm²-sec)
- n (or N) = # atoms
- ν = kinematic viscosity

Growth Rate Equations

From MBE:

$$\phi = \text{\#atoms}/(\text{surface unit}/\text{monolayer thickness}) * \text{GR}$$

$$\phi = (n/S/c/2) * \text{GR}$$

$$S = \frac{a^2 c (3)^{1/2}}{4n}$$

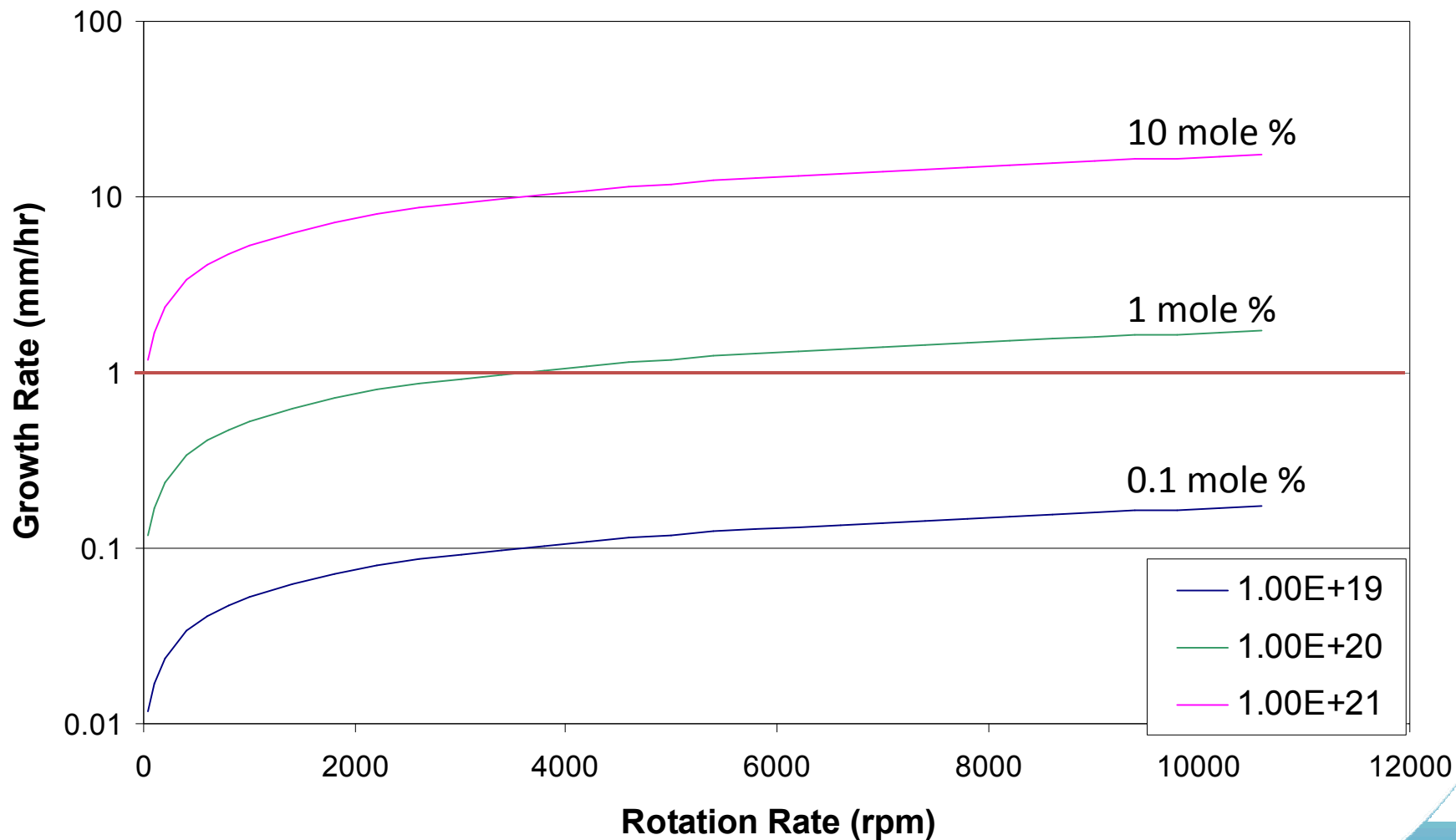
$$\text{GR} = \frac{a^2 c (3)^{1/2} \phi}{4}$$

Levitch Equation (from RDE): $i_{l,c} = 0.62nFAD_o^{2/3}\omega^{1/2}\nu^{-1/6}C_o^*$

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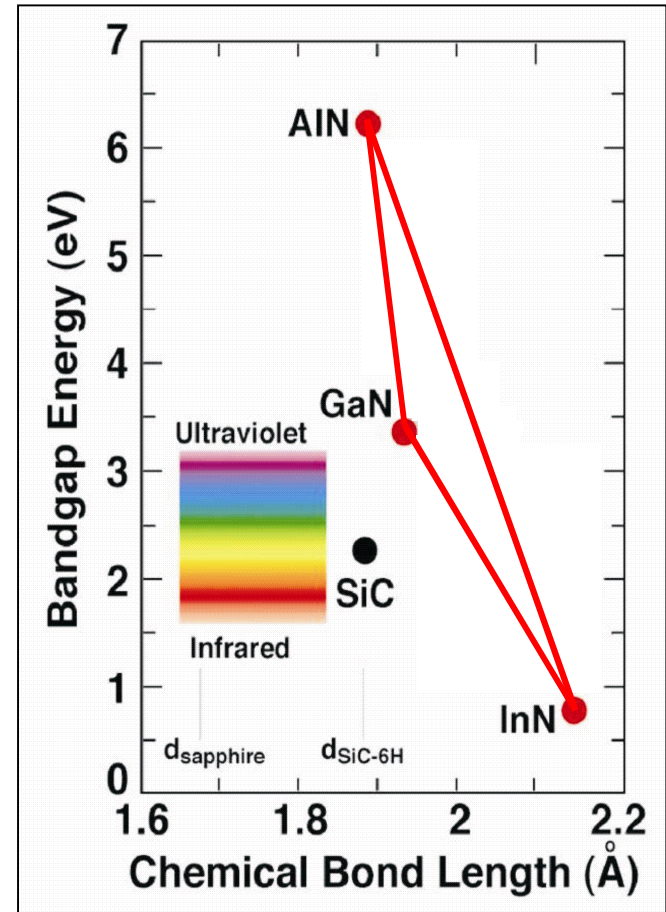
$$\text{GR (mm/hr)} = \frac{a^2 c (3)^{1/2} * 0.62D_o^{2/3}\omega^{1/2}\nu^{-1/6}C_o^*}{4} \times \frac{10\text{mm} * 3600\text{s}}{\text{cm} * \text{hr}}$$

Growth Rate vs. Rotation Speed and Concentration



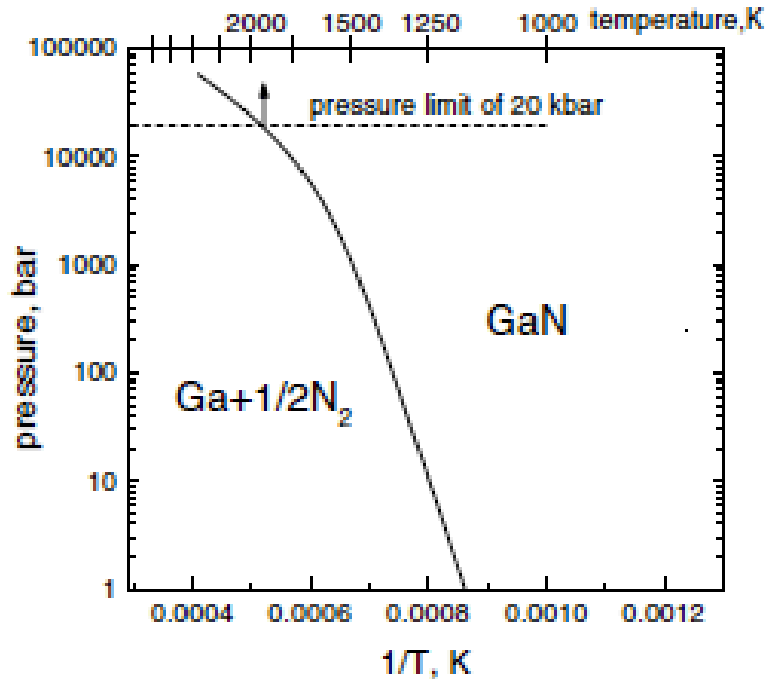
Motivation for GaN Power Devices

- **Footprint:** WBG power electronics offer advantages over silicon
 - No active cooling systems
- **Flexibility:** GaN offers additional device design options due to ability to alloy with AlN (higher standoff voltages) and InN (higher switching frequencies), new device architectures
- **Cost:** SiC expensive; GaN has market pull from solid-state lighting to reduce cost

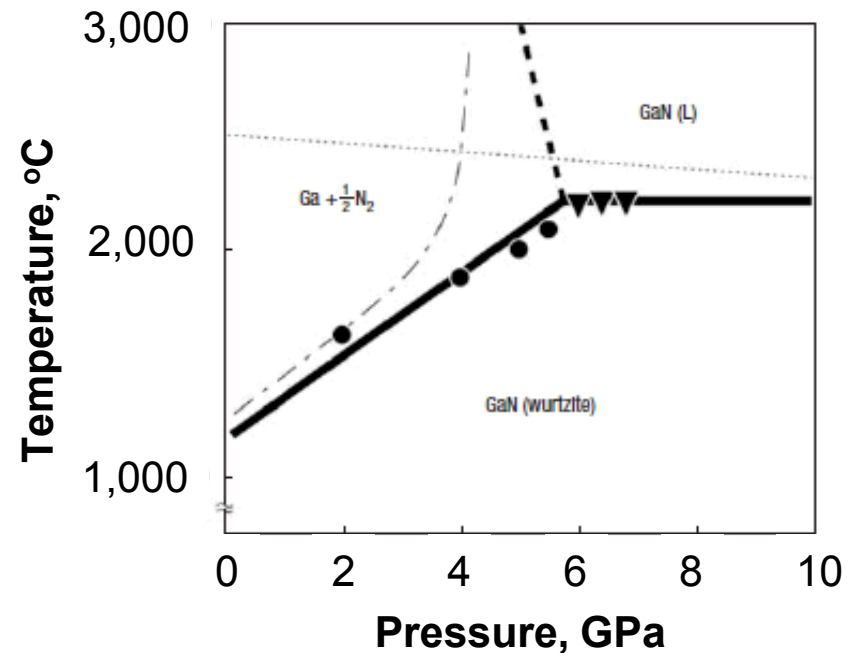


GaN Thermodynamics

Melting point: ~2,225°C at 6GPa



Grzegory et al, *J. Phys. Cond. Matt.* (14), 2002



Utsumi et al, *Nature Mat.* Nov 2003

- One report of congruent melting of GaN ca. 2003
- Many calculations of GaN melting conditions 2500C, 45,000 atm

Historical Evolution of Power Electronics

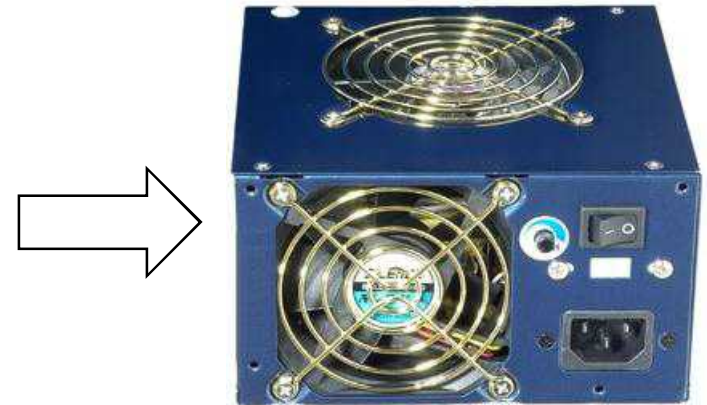


1942

Hg-arc rectifier

2012

Si thyristor



20??

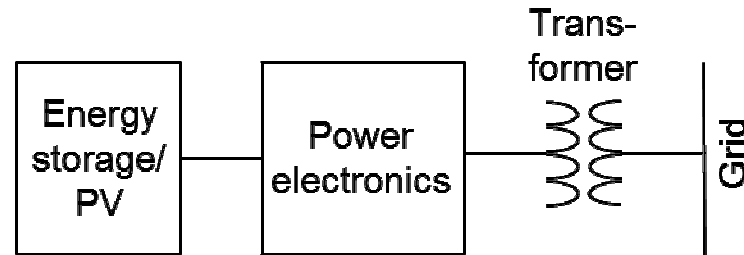
SiC/GaN version

More compact

More efficient

Less cooling

Power Conversion System



- Renewable energy sources and energy storage need to interface with the electric grid through a power conversion system
- Heart of the power conversion system consists of semiconductor switches that determine overall cost, reliability, and performance (efficiency, power quality)
- Most systems today utilize silicon-based semiconductor switches (IGBTs, thyristors) – mature technology, incremental improvements
- Recent advances in the wide-bandgap materials Silicon Carbide (SiC) and Gallium Nitride (GaN) offer potential for improved performance, reduced cooling requirements, smaller system size, lower cost

Application Classes of Power Devices

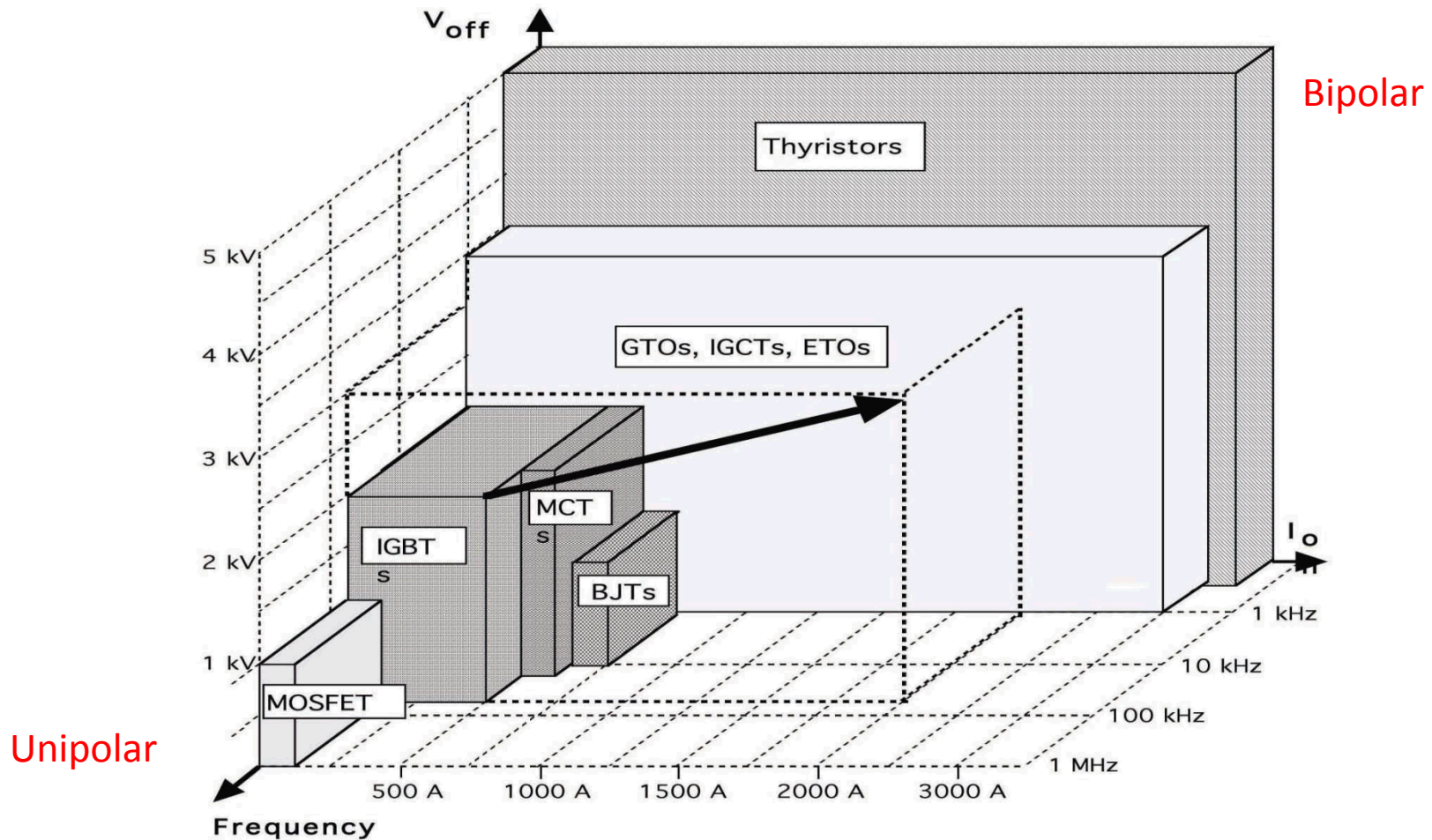
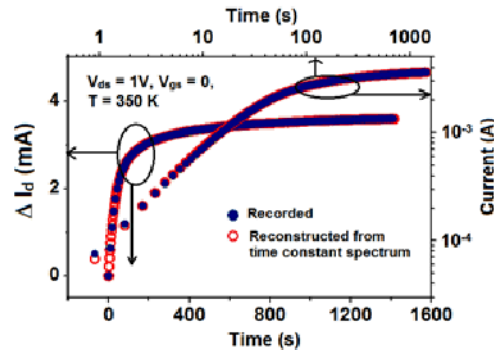
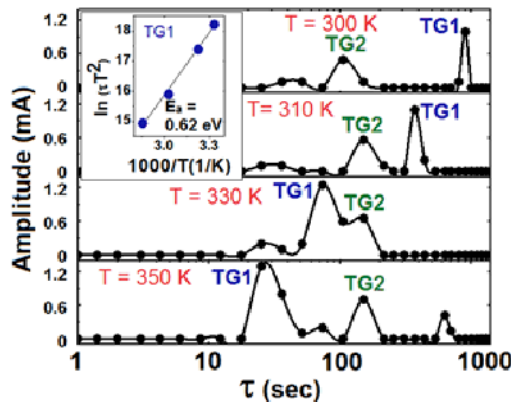


Figure from Mohan et al., "Power electronics: Converters, Applications, and Design" (Wiley, 2003).

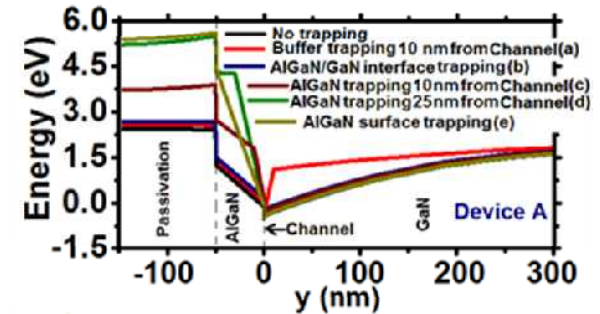
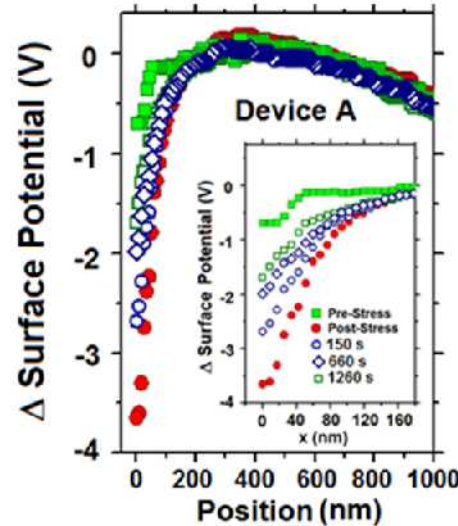
Analysis of Slow Transient Recovery of Drain Current – Deep Level Defects



$$\Delta I_d = \sum_i A_i \left[1 - \exp\left(-\frac{t}{\tau_i}\right) \right]$$



Passivated
Al_{0.15}Ga_{0.85}N/GaN HEMT

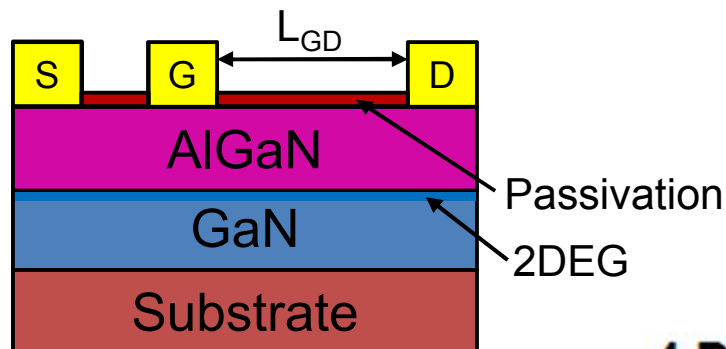


S. Dasgupta et al., submitted to IEDM

KFM surface potential
measurements and device
modeling to determine
spatial location of traps

More understanding needed of defects
responsible for current collapse and
the dependence on composition,
growth, and device structure;
better understanding of surface effects
and surface field engineering for
high breakdown voltage

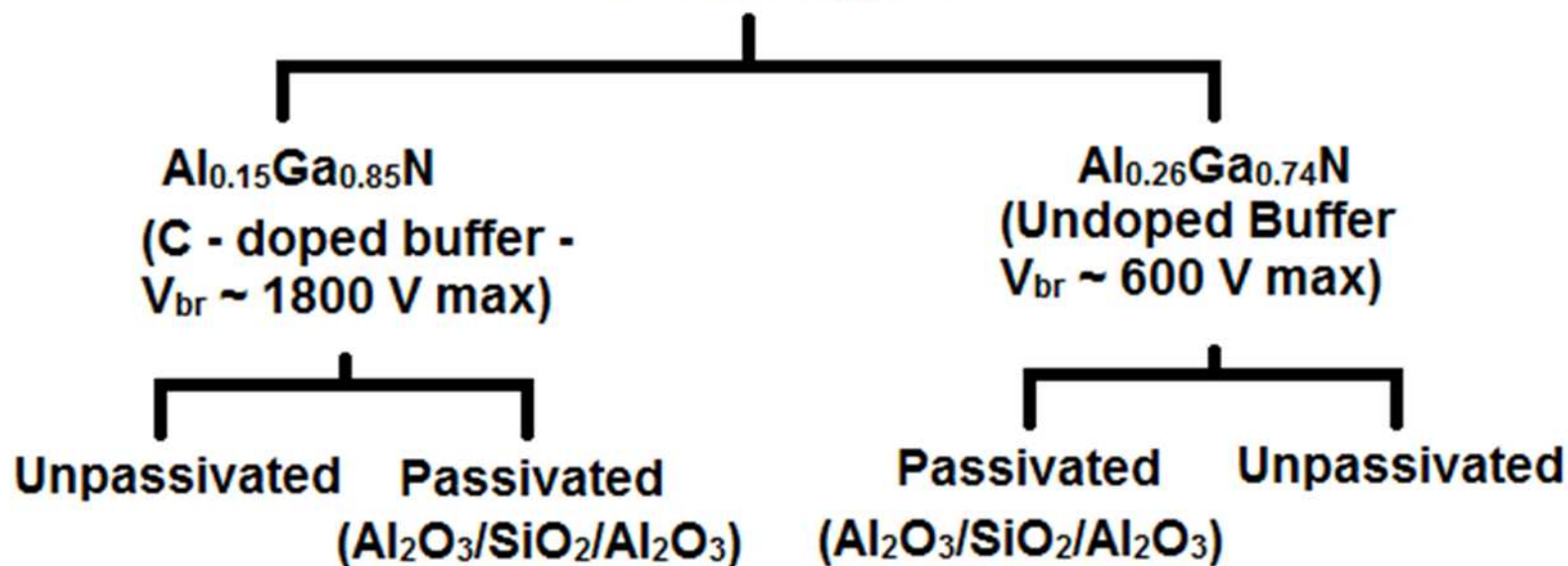
Lateral AlGaN/GaN Power HEMT



High Electron Mobility Transistor:

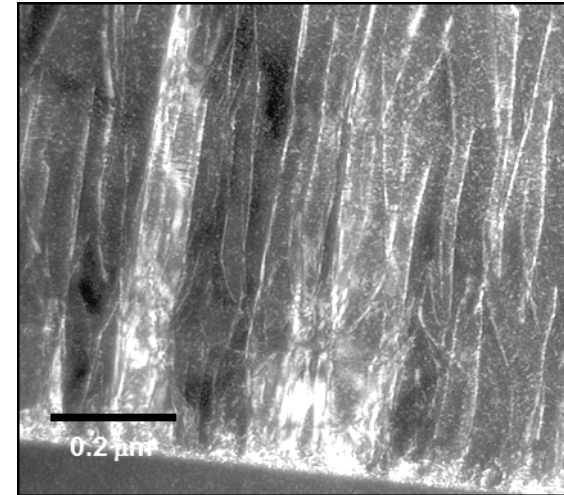
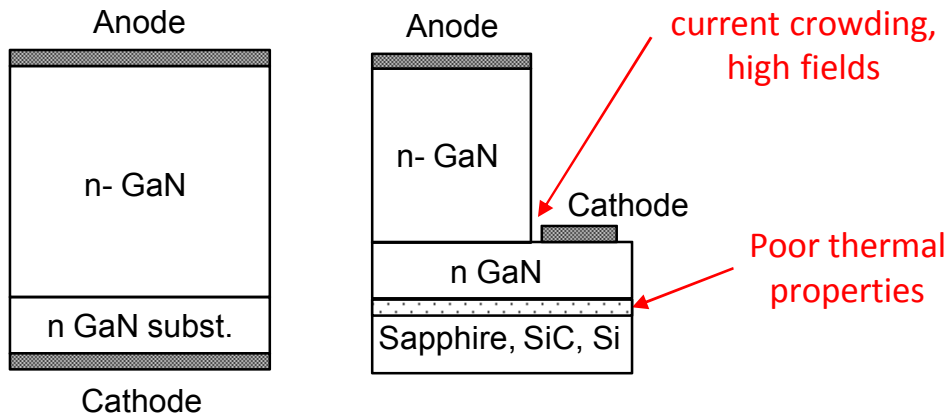
- Designed and fabricated at MIT
- Polarization induces high-m channel
- Normally-on device
- L_{GD} and Al% control V_{BD}

4 Device Types



GaN Vertical Device on GaN Substrate

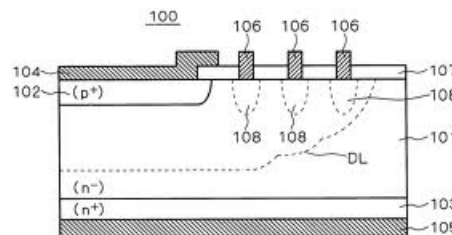
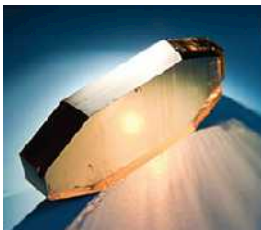
GaN substrate



The science-based materials physics foundation for vertical GaN power devices grown on native GaN substrates is needed (e.g. breakdown, carrier lifetime and transport)

High dislocation density for heteroepitaxial GaN (10^9 - 10^{10} cm^{-2})

(Ammono)
GaN



- Premature breakdown
- Enhanced impurity incorporation
- Point / extended defect interaction

GaN Vertical Device – Previous Attempts

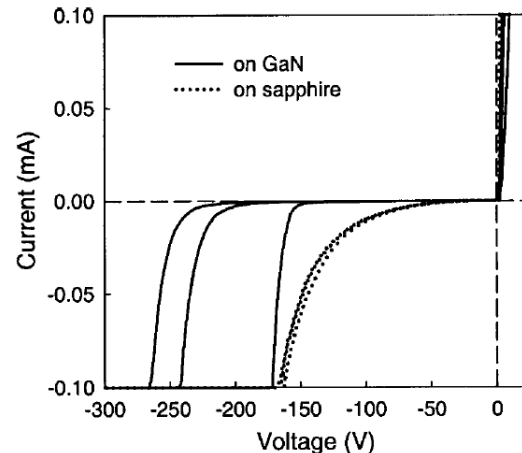
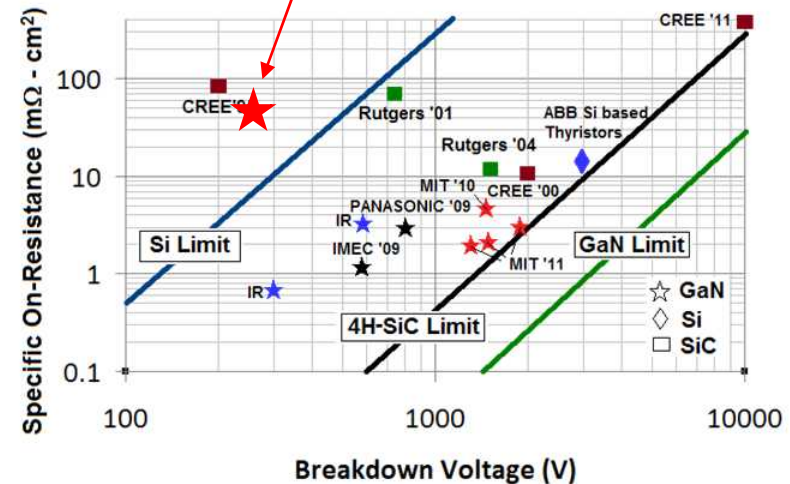
X. Cao et al., APL 87 053503 (2005)

- MOCVD on HVPE GaN substrate
- PiN diode
- $n \sim 1 \times 10^{17} \text{ cm}^{-3}$

In summary, homoepitaxy of GaN PiN rectifiers on a free-standing GaN yielded markedly improved material with a reduced dislocation density and impurity concentration, leading to better breakdown characteristics than similar devices grown on sapphire. However, the rectifiers still show relatively high reverse leakage, premature breakdown, and negative temperature dependence of the breakdown voltage. All of these findings reveal the presence of point and micro-defects in the voltage-blocking layer. The microdefects are believed to originate from the substrate surface, and thus may be eliminated by improving the bulk GaN quality and surface preparation prior to the growth. Point defects in GaN, including common impurities (C, H, and O), N and Ga vacancies and antisites may act as dopants or compensation centers and need to be further reduced. To achieve high reverse-blocking capability, it is critical to precisely control the doping concentration in the drift region at low levels. Further work should focus on lowering the unintentional doping and compensation in homoepitaxial GaN.

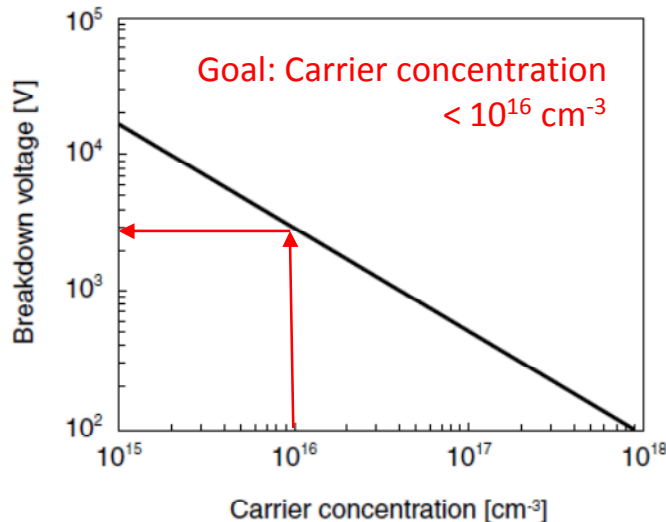
Device problems are due to fundamental materials issues

Device is far from theoretical limit



Significant variation between devices grown on GaN

Low and Controllable n-type Doping is Critical for Power Device



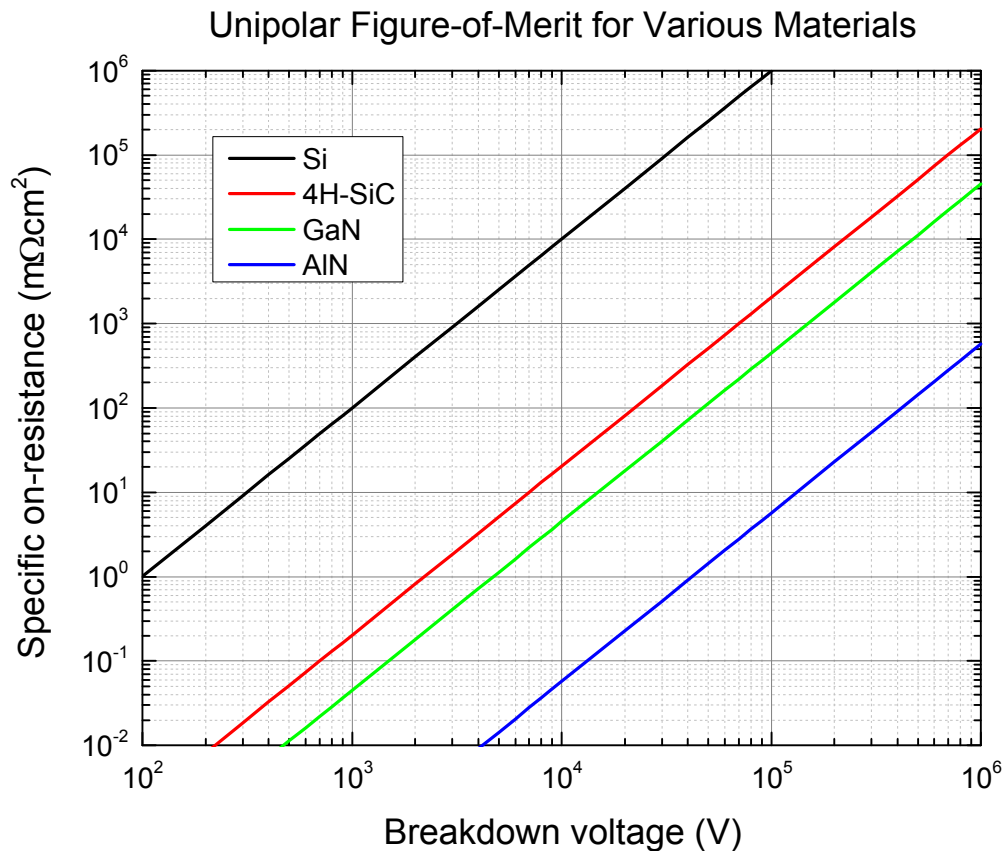
T. Tanabe et al., SEI Tech. Rev. **64**, 21 (2007).

$$V_B = \frac{\epsilon E_C^2}{2qN_D}$$

High breakdown voltage requires low carrier concentration

- *Controllable* incorporation of Si donors in GaN
- Minimization of donor impurities / defects (O , V_N)
- Minimization of compensating acceptors (C , V_{Ga} – places a lower limit on effective carrier concentration under high electric fields)
- Requires scientific understanding of growth on GaN substrates and the impact of growth conditions on native point defects
- *Most knowledge of these issues to date is for heteroepitaxy on SiC, sapphire, or Si*

Theoretical Performance of Ultra-Wide-Bandgap AlN Unipolar Device



Outstanding performance *in theory* – but numerous material physics problems must be solved, similar to GaN:

- Growth on AlN substrates for vertical devices
- Low, non-compensated n-type doping
- Minority-carrier transport
- Gate oxide

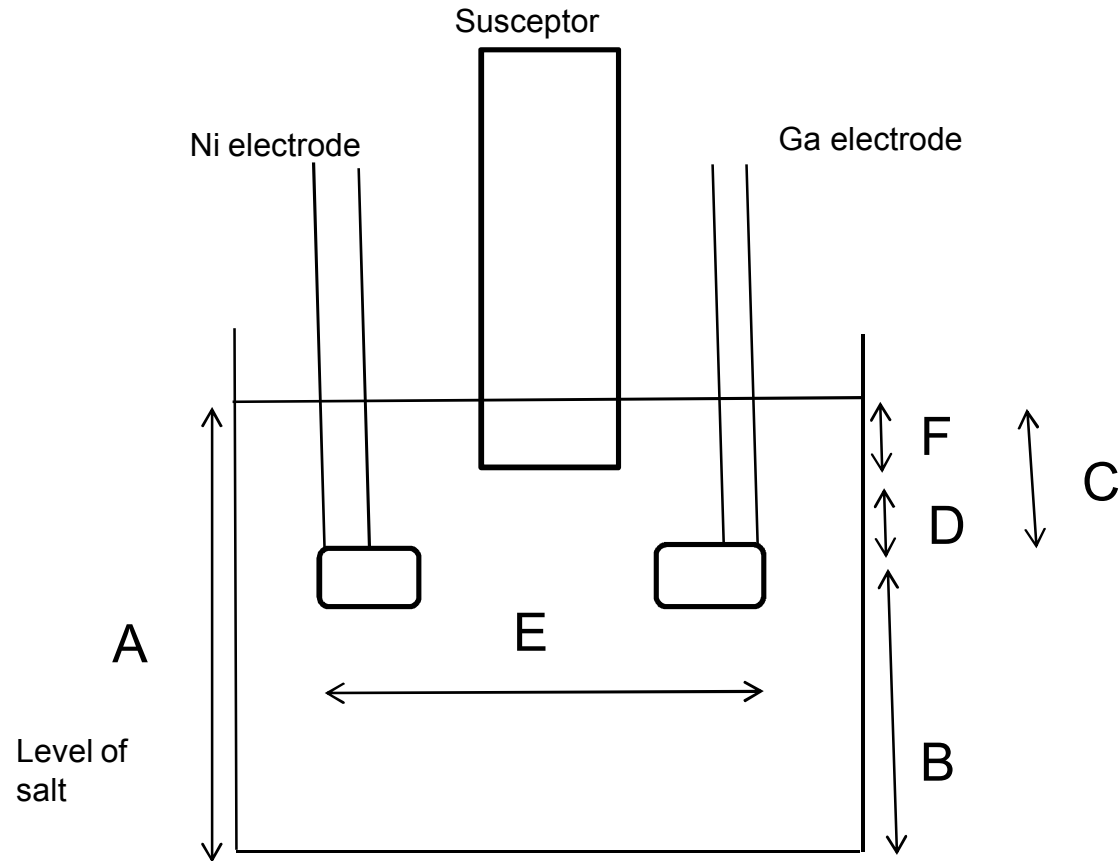
Conclusion: Key Materials Physics Issues for WBG Power Electronics

- Native substrates: Need large-area, defect-free wafers to achieve yield and reliability approaching that of Si devices
 - Enable high voltage and current for grid-scale devices
 - GaN and AlN substrates for vertical III-N devices
- Fundamental III-N material properties
 - Controllable low-doped n-type drift layers
 - Ultra-wide bandgap and high critical field
 - Bipolar devices: Carrier lifetime, mobility
 - Charge trapping, deep level defects, and extended defects
 - Dependence of switching speed on materials properties
 - Breakdown mechanisms (avalanche vs. others, e.g. surface)
 - Novel heterojunction engineering (e.g. graded composition)
- Gate oxides for MOSFETs and IGBTs
 - SiC/SiO₂ low channel mobility, bias-temperature instability
 - Gate oxides on III-N materials

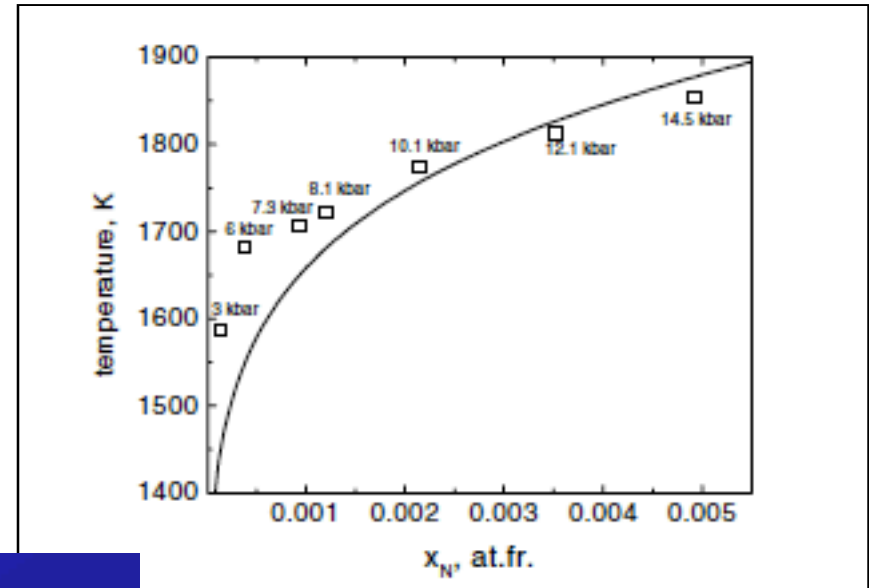
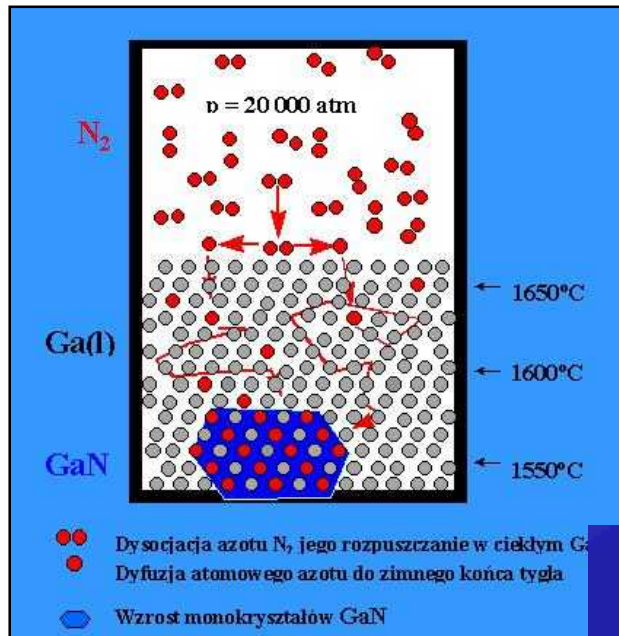
Thermal Battery Salt (no filtration)



Experimental Set Up

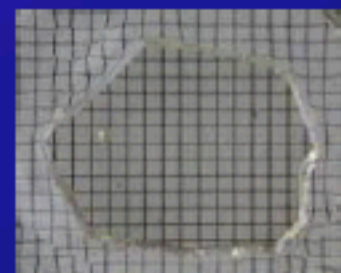
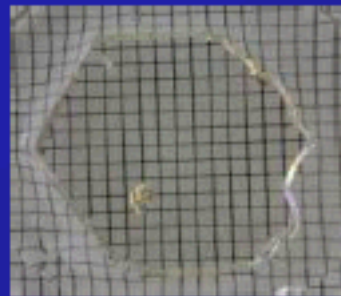


High Nitrogen Pressure Solution Growth



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Physics
Warsaw, Poland**



**Dissolve atomic N in molten Ga
(1500C, 20,000 atm)**

2" diameter is physical limit

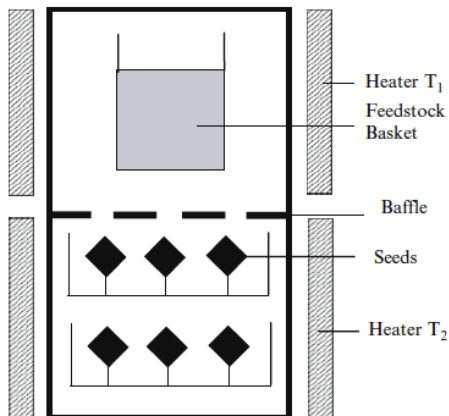
First blue LD grown (Nakamura)

Dislocations $< 100\text{ cm}^{-2}$

Ammonothermal Growth



- **GaN is extremely difficult to dissolve**
 - Supercritical ammonia (500-800C, 4,000-5,000 atm)
 - Requires additional mineralizers
 - Acidic or basic approaches
- **Major players:**
 - Dwilidinski & Doradzinski: Ammono
 - Nakamura: Sora
 - Haskell & Fini: Inlustra
- Up to 60 $\mu\text{m}/\text{day}$ growth rates



$$T_2 > T_1$$

- Another solvent: Molten LiF (850C, 240atm)
- B. Fiegelson, NRL

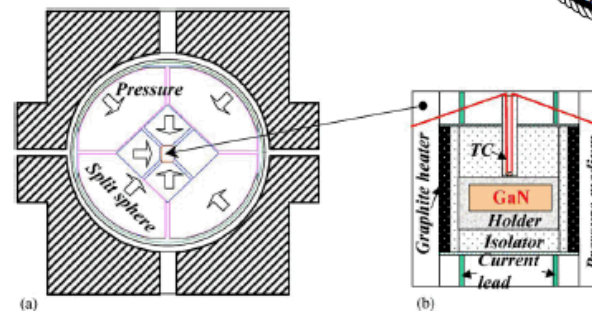
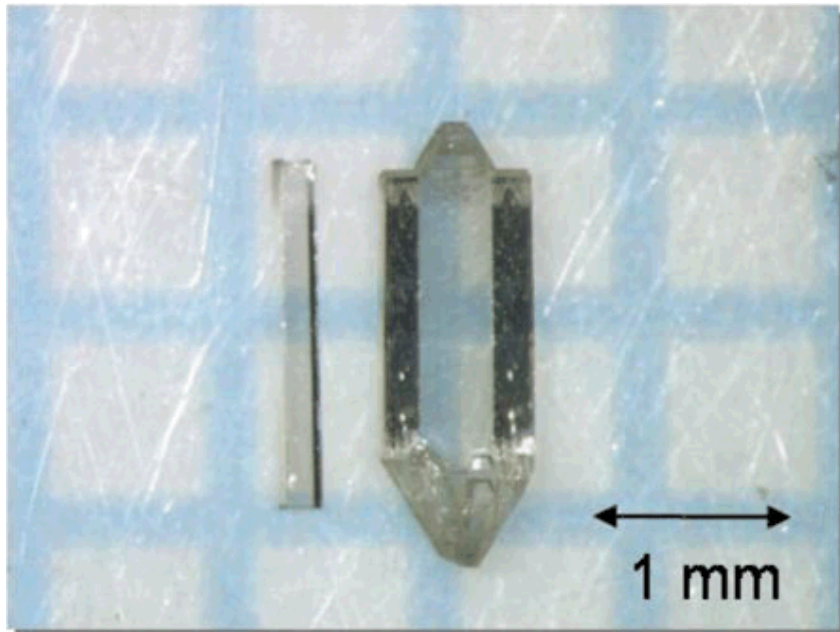


Fig. 1. Schemes of the (a) multi-anvil split sphere high-pressure apparatus, and (b) high-pressure cell.

Alkali Metal Flux/LPE

- Sodium Flux/LPE
- Na dissociates N_2
- 750C, 50 atm
- Growth rates $\sim 2\mu\text{m/hr}$
- Additives



Yamane et al., 2006