

LA-UR-12-22286

Approved for public release; distribution is unlimited.

Title: Biogeochemistry in Sea Ice: CICE model developments

Author(s): Jeffery, Nicole
Hunke, Elizabeth
Elliott, Scott
Turner, Adrian

Intended for: 17th Annual CESM Workshop, 2012-06-18/2012-06-21 (Breckenridge, Colorado, United States)



Disclaimer:

Los Alamos National Laboratory, an affirmative action/equal opportunity employer, is operated by the Los Alamos National Security, LLC for the National Nuclear Security Administration of the U.S. Department of Energy under contract DE-AC52-06NA25396. By approving this article, the publisher recognizes that the U.S. Government retains nonexclusive, royalty-free license to publish or reproduce the published form of this contribution, or to allow others to do so, for U.S. Government purposes. Los Alamos National Laboratory requests that the publisher identify this article as work performed under the auspices of the U.S. Department of Energy. Los Alamos National Laboratory strongly supports academic freedom and a researcher's right to publish; as an institution, however, the Laboratory does not endorse the viewpoint of a publication or guarantee its technical correctness.

Biogeochemistry in Sea Ice: CICE model developments

N. Jeffery¹, E. Hunke, S. Elliott¹, and A. Turner¹

¹*Los Alamos National Laboratory, Los Alamos, USA*

Polar primary production unfolds in a dynamic sea ice environment, and the interactions of sea ice with ocean support and mediate this production. In spring, for example, fresh melt water contributes to the shoaling of the mixed layer enhancing ice edge blooms. In contrast, sea ice formation in the fall reduces light penetration to the upper ocean slowing primary production in marine waters. Polar biogeochemical modeling studies typically consider these types of ice-ocean interactions. However, sea ice itself is a biogeochemically active medium, contributing a significant and, possibly, essential source of primary production to polar regions in early spring and fall. Here we present numerical simulations using the Los Alamos Sea Ice Model (CICE) with prognostic salinity and sea ice biogeochemistry. This study investigates the relationship between sea ice multiphase physics and sea ice productivity. Of particular emphasis are the processes of gravity drainage, melt water flushing, and snow loading. During sea ice formation, desalination by gravity drainage facilitates nutrient exchange between ocean and ice maintaining ice algal blooms in early spring. Melt water flushing releases ice algae and nutrients to underlying waters limiting ice production. Finally, snow loading, particularly in the Southern Ocean, forces sea ice below the ocean surface driving an upward flow of nutrient rich water into the ice to the benefit of interior and freeboard communities. Incorporating ice microphysics in CICE has given us an important tool for assessing the importance of these processes for polar algal production at global scales.



U.S. DEPARTMENT OF
ENERGY

Office of
Science



Biogeochemistry in Sea Ice: CICE model developments

Nicole Jeffery, Elizabeth Hunke, Scott Elliott, Adrian Turner

This Talk

- ▶ Motivation: Sea Ice, Polar biogeochemistry, ocean, atm, climate ...
- ▶ Constituents of sea ice
- ▶ Salt

Impacts at the micro scale

Impacts at climate scale: Arctic vs Antarctic

Nitrogen tracers in a simple algal model

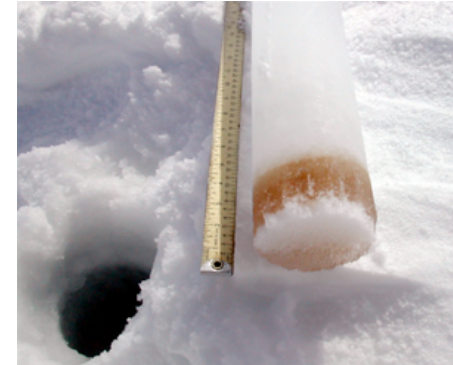
- 1) Arctic vs. Antarctic
 - 2) Nitrification on/off
- ▶ Conclusions

Sea Ice Algae

(a)



(c)



(b)



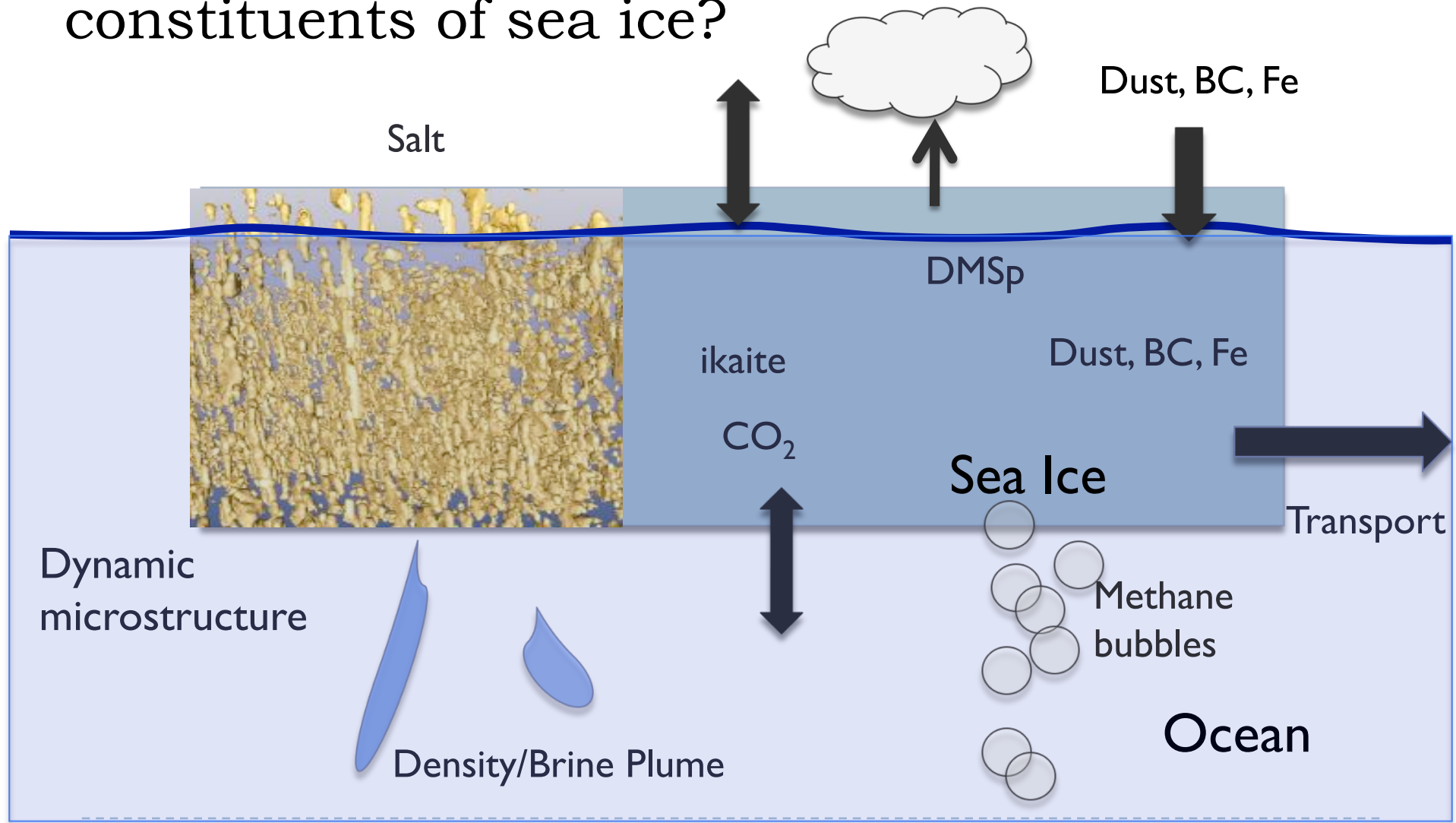
(d)



Chukchi Sea, Alaska, June 2009

More generally...

What are the climatically important
constituents of sea ice?



A physics based classification of sea ice constituents:

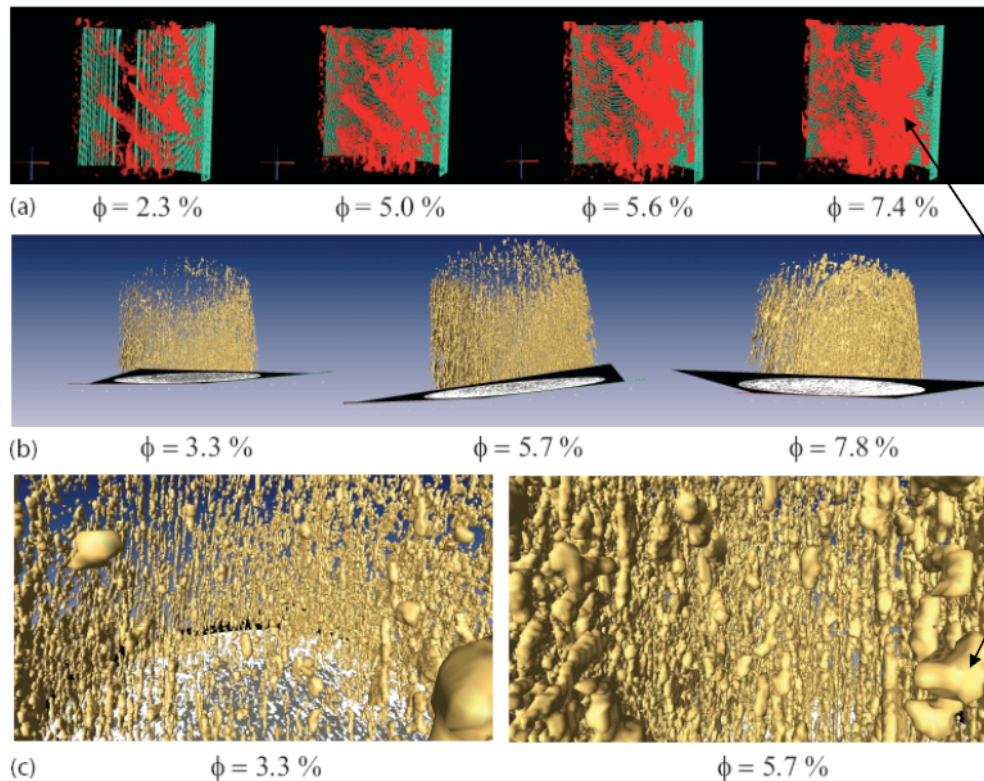
- 1) Salt – modifies T_{melt} , conductivity, heat capacity, ice density, ice porosity, and vertical transport physics
- 2) Tracers that move like salt – nitrate, silicate, ammonium, phosphate
- 3) Tracers that cling to the ice crystals – algae, DMSp
- 4) Tracers that precipitate – DOC (Ikaite)

* Dust (fe), Black Carbon, Methane bubbles

Atmospheric interactions through DOE Polar Project -- Steve Ghan (PNNL), Phil Rasch (PNNL), Hailong Wang (PNNL), Natalie Mahowald (Cornell), Lynn Russel (Scripps)

Impact of Salt (micro scale)

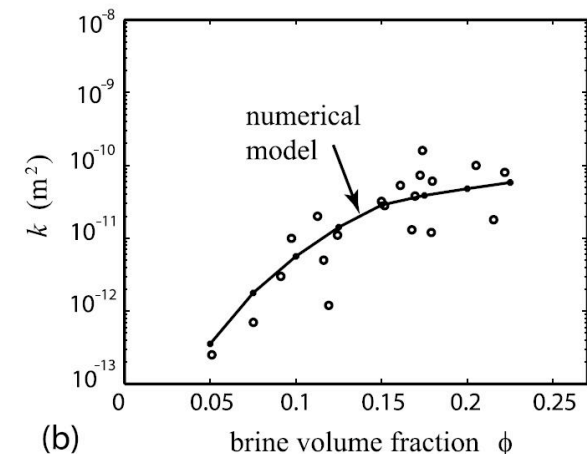
Imaged Sea Ice Structures



Lab-grown sea ice: reconstructions of X-ray CT of 1 cm cores
Heaton, Miner, Eicken, Zhu, Golden, *in prep* (2006)

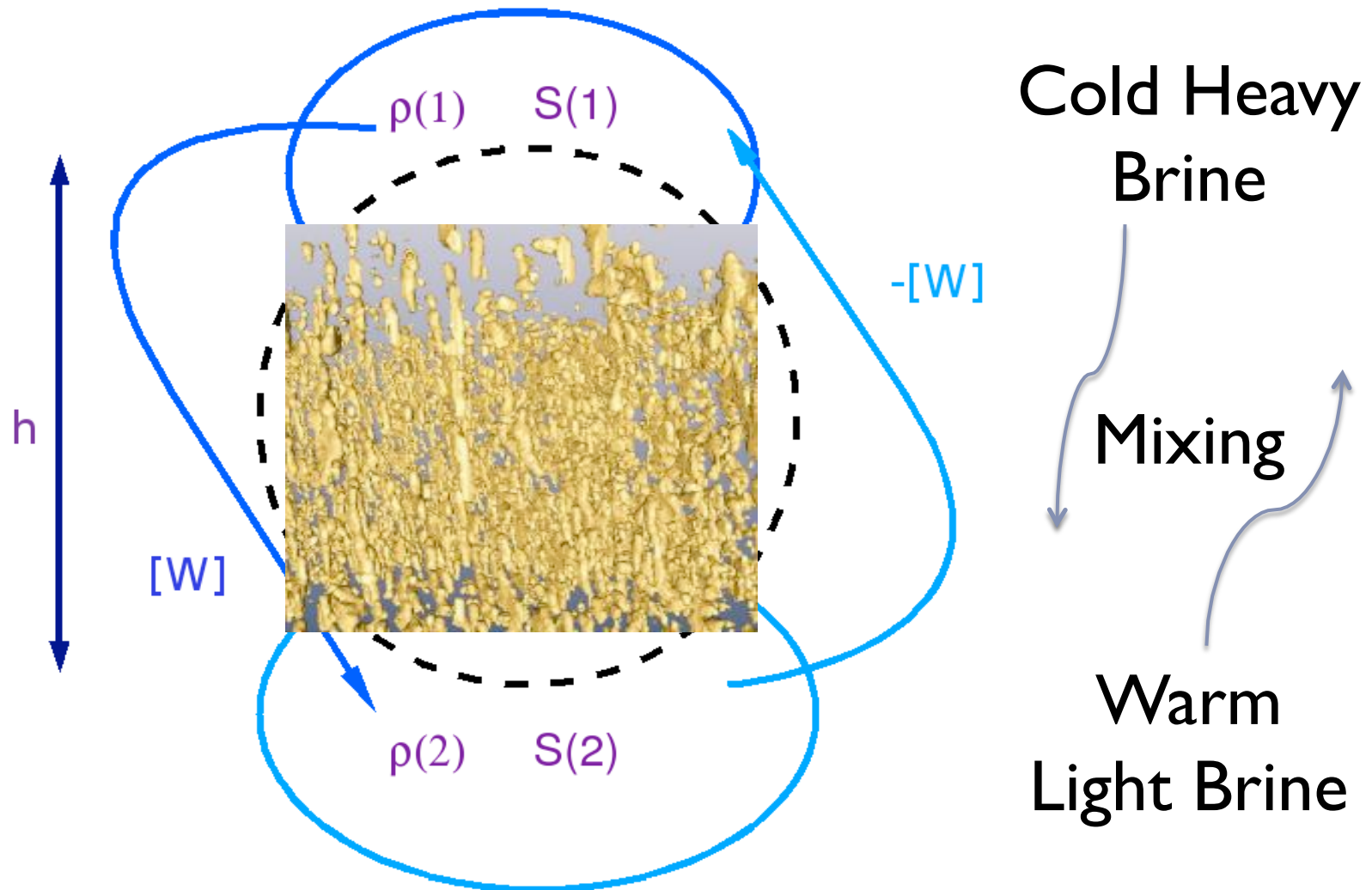
Brine
inclusions

Salinity and
Temperature determine
the porosity,
permeability and the
desalination rate.



Golden et al., 2007

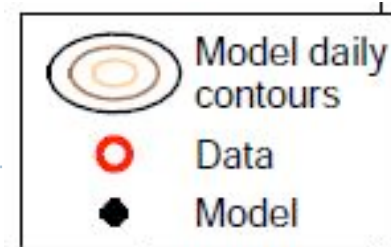
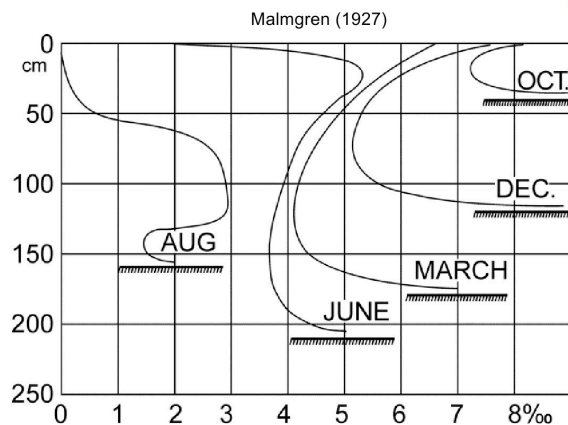
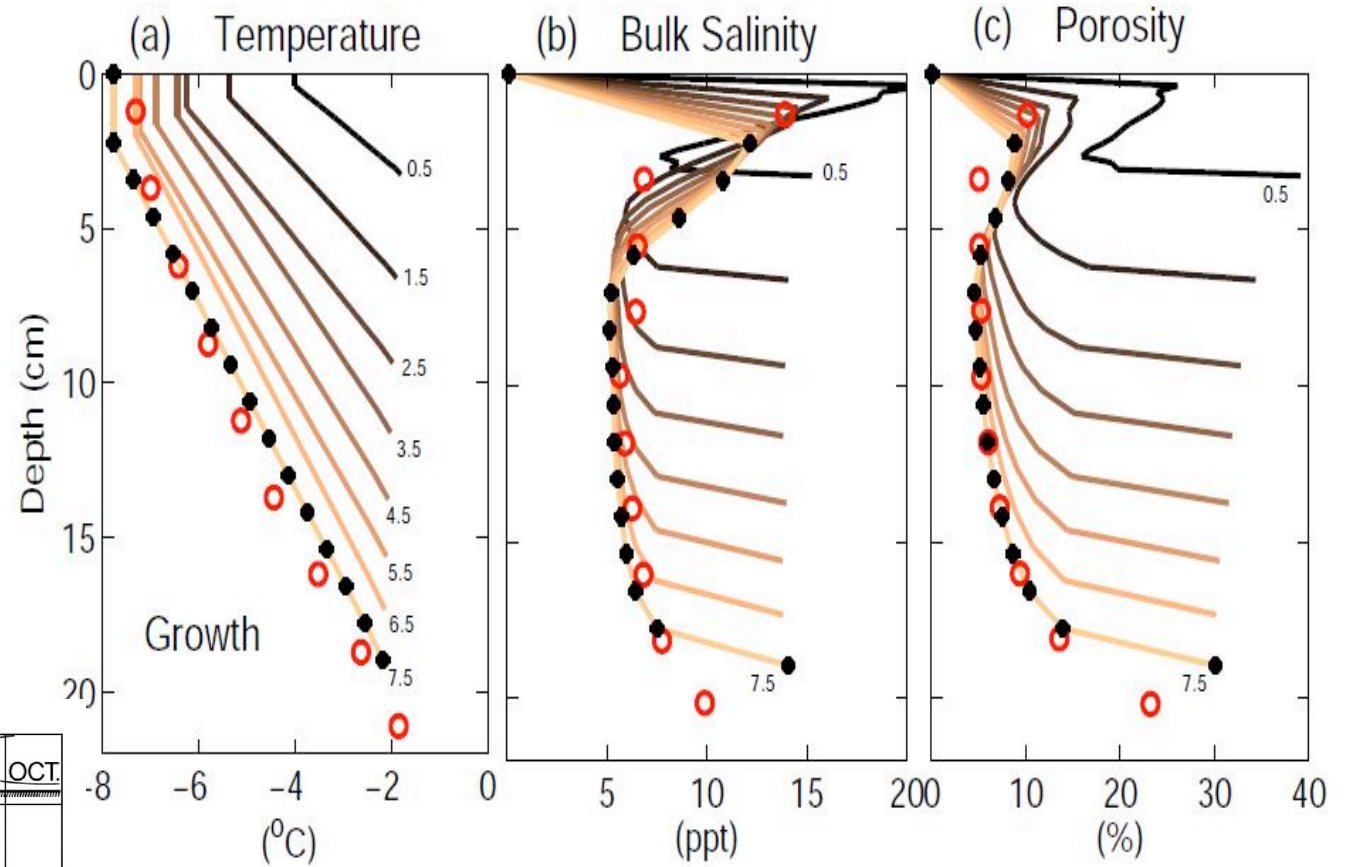
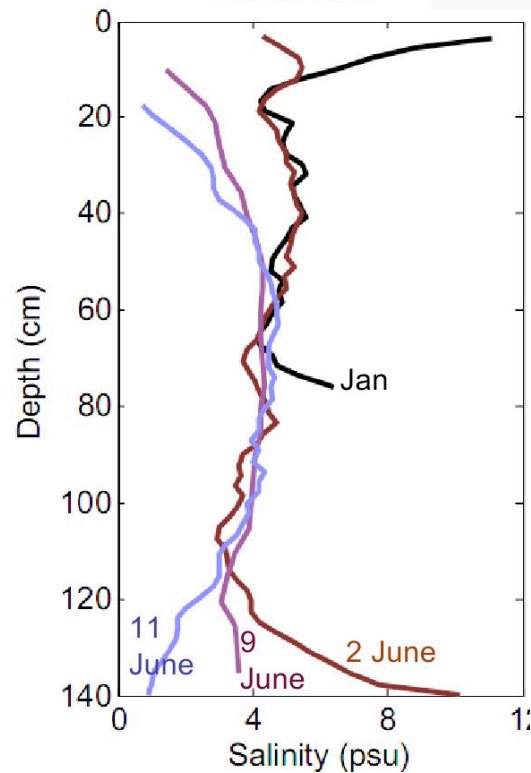
Brine Motion during Ice Growth: Gravity Drainage



As the ice desalinates, more sea ice solidifies, which decreases the permeability and slows gravity drainage.

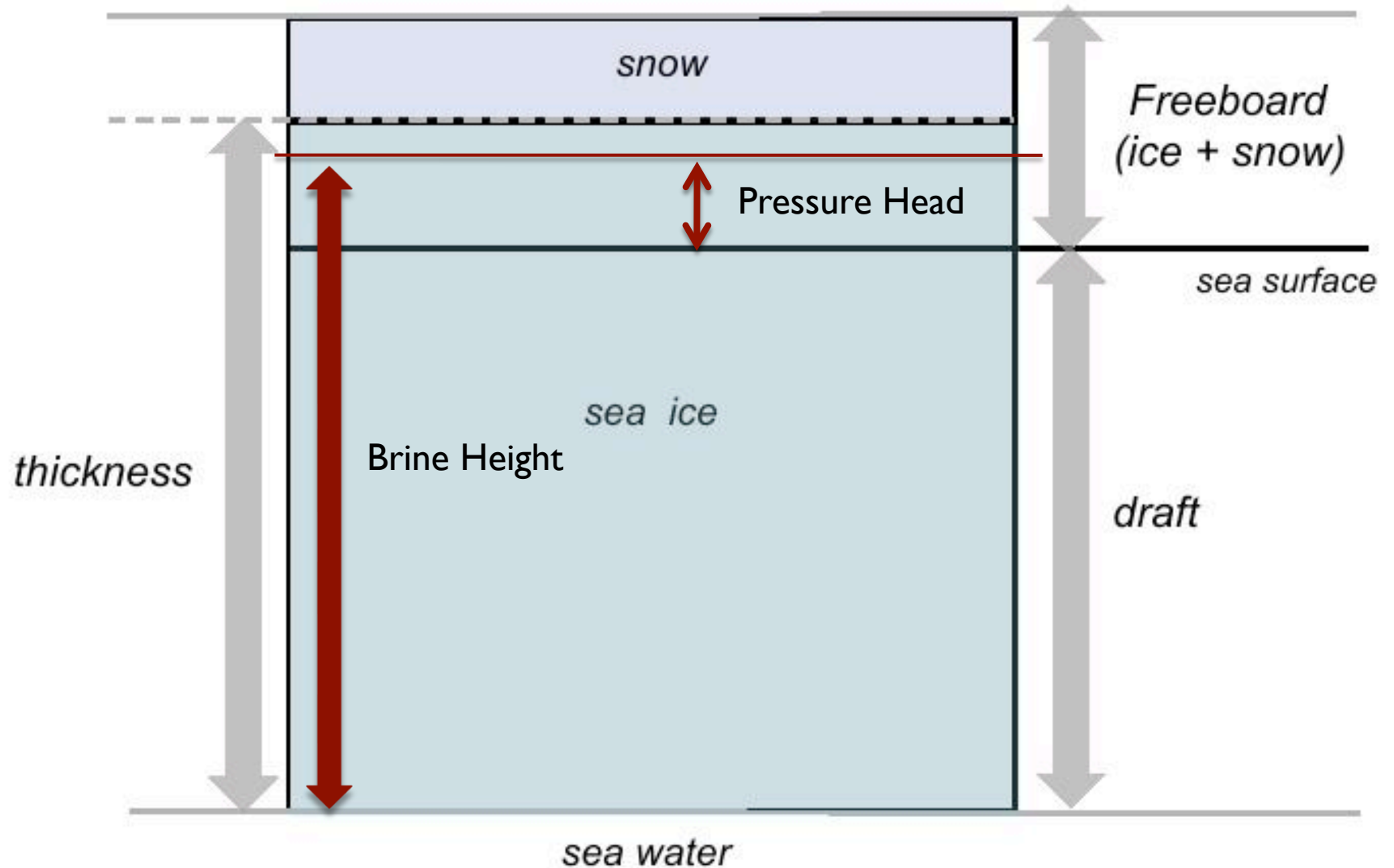
Barrow 2007 C. Petrich (UAF)

Characteristic 'C' Profile

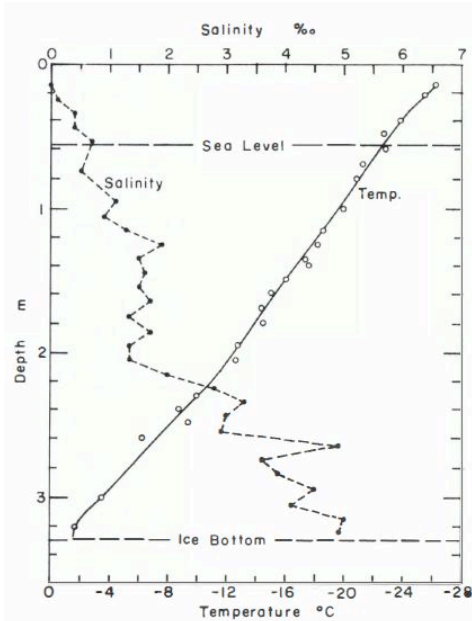


Brine Motion during Melt Flushing

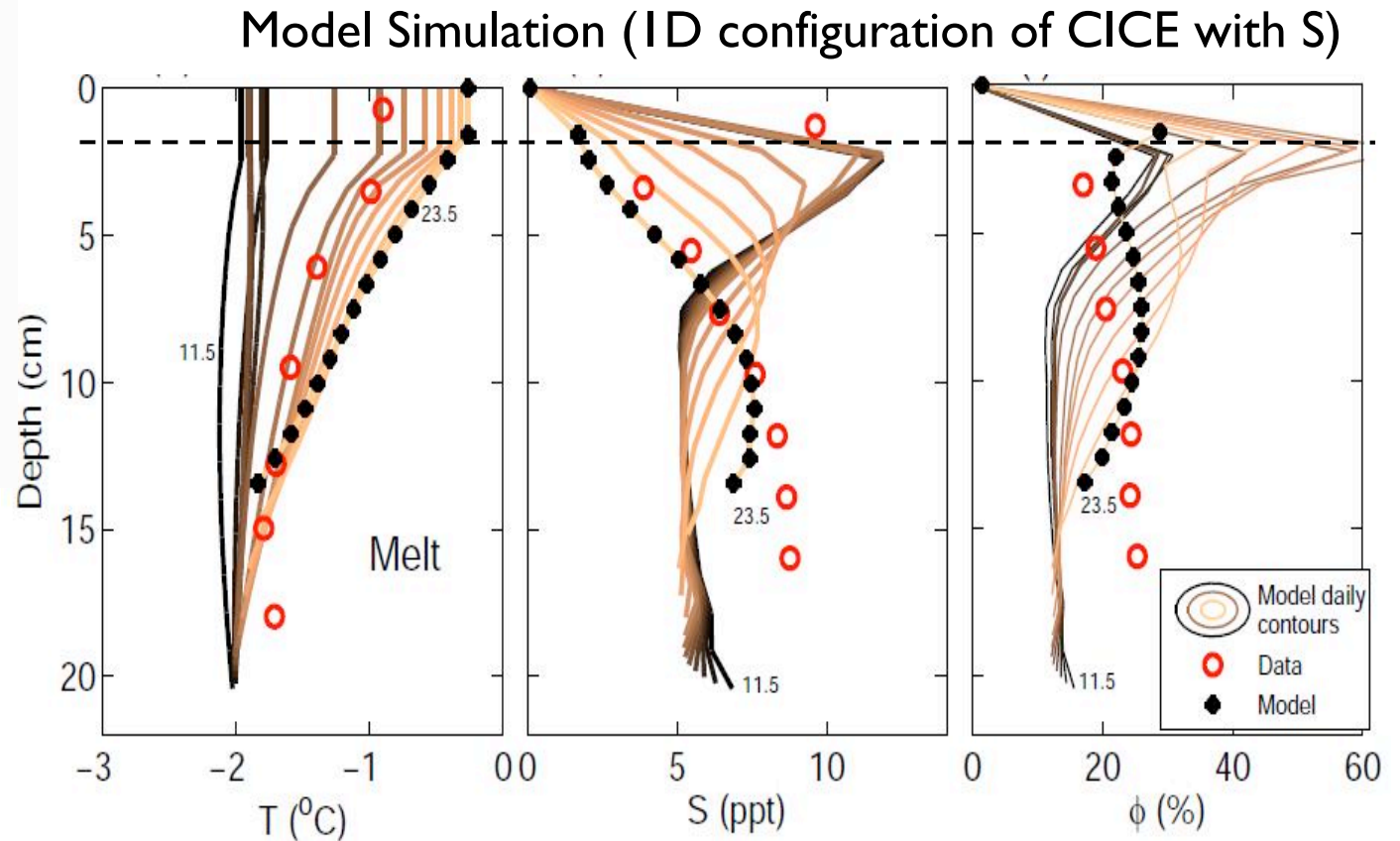
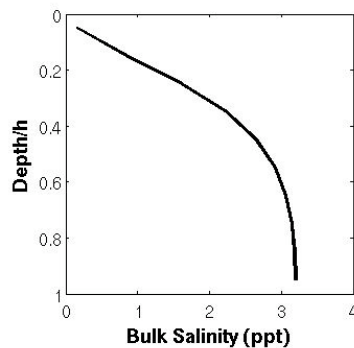
- 1) **Flushing — downward flow:** low salinity meltwater desalinates ice
- 2) **Snow accumulation — upward flow:** ocean water replenishes depleted nutrients



Melting and Multi-year Ice profile



Cox & Weeks (1974)

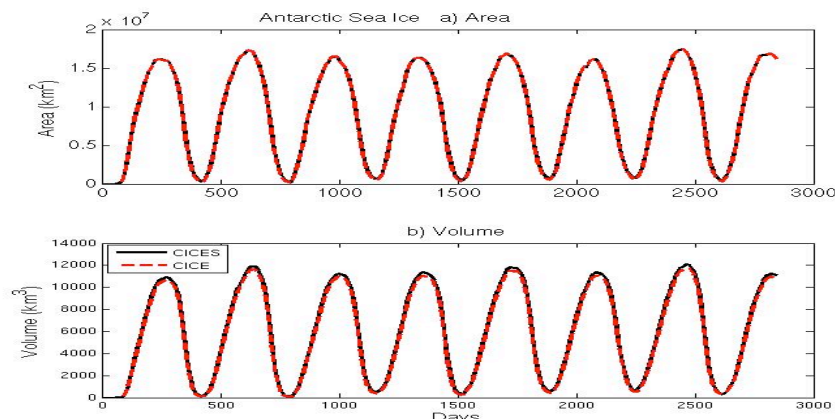


Standard CICE currently
assumes a fixed S profile

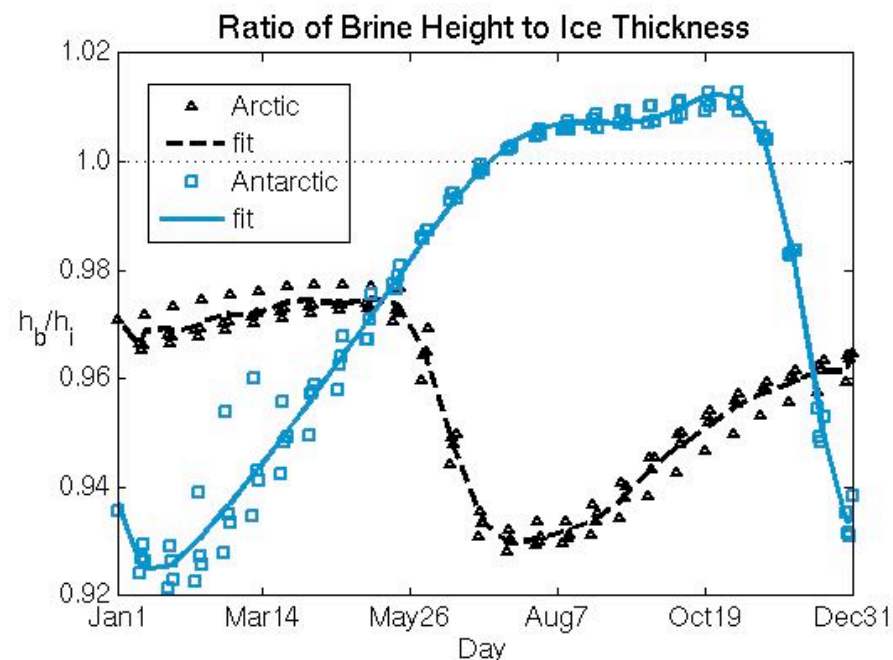
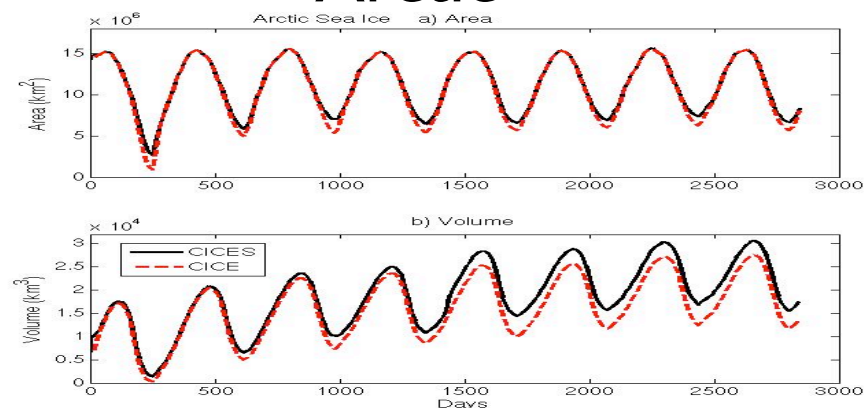
Impact of Salinity (climate scale)

7 Year Control Run: **CICE** vs CICES

Antarctic



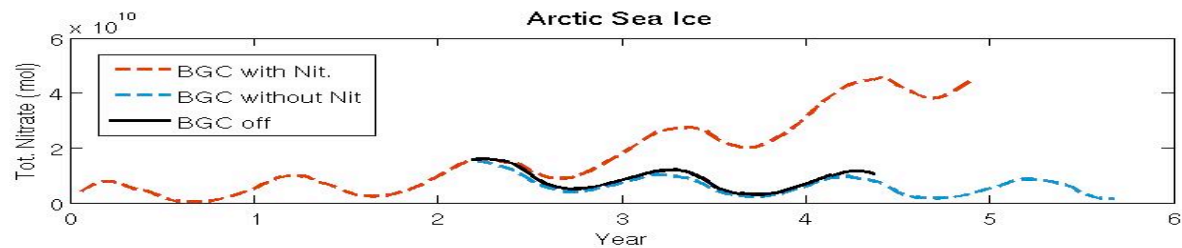
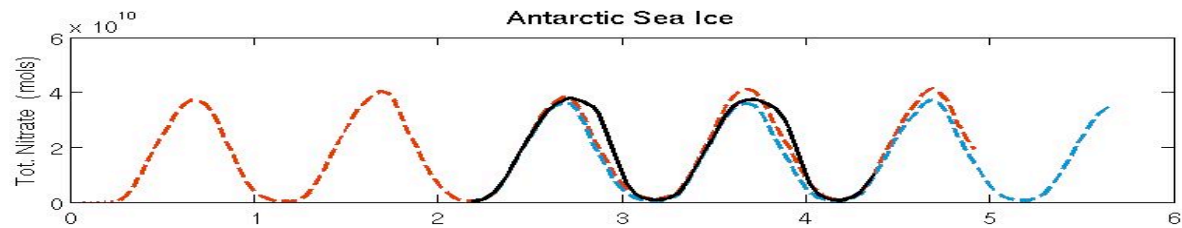
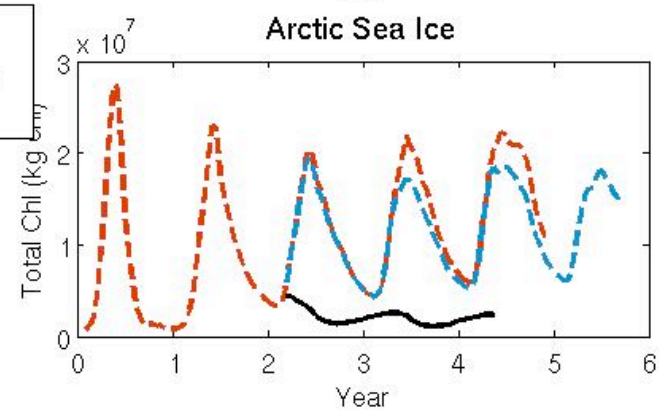
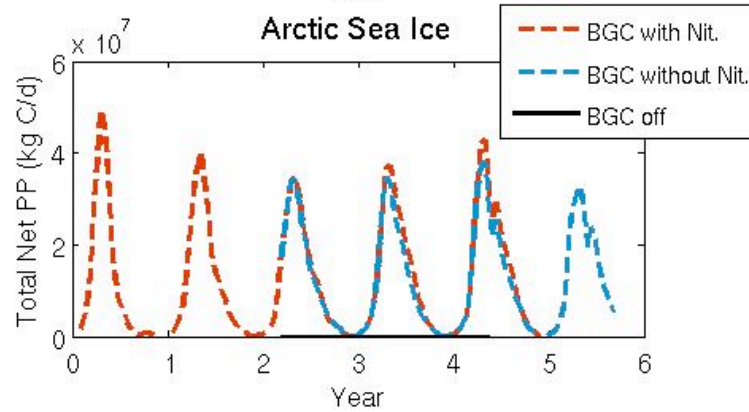
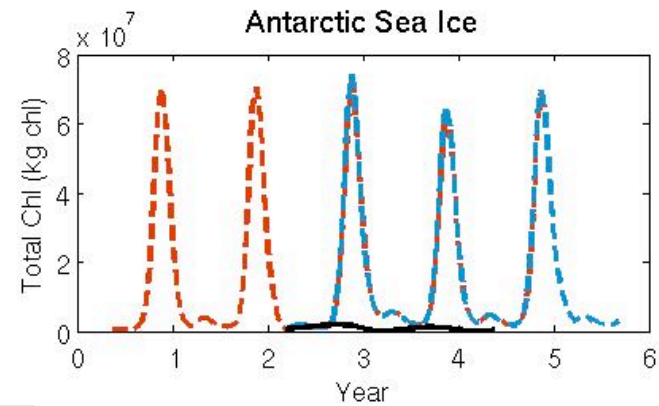
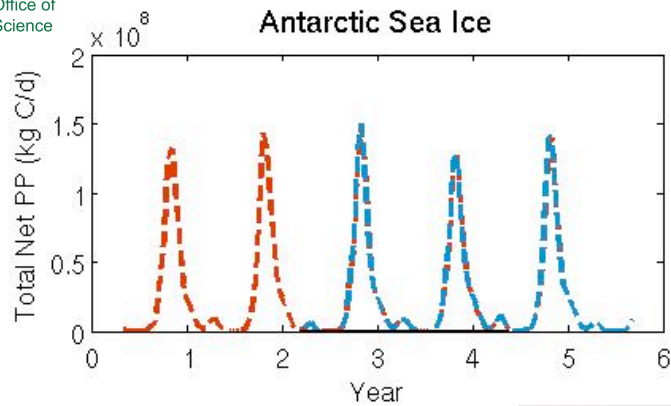
Arctic



Nitrogen-Based Ice Algal Biogeochemistry (Tracers that move like S, Tracers that cling)

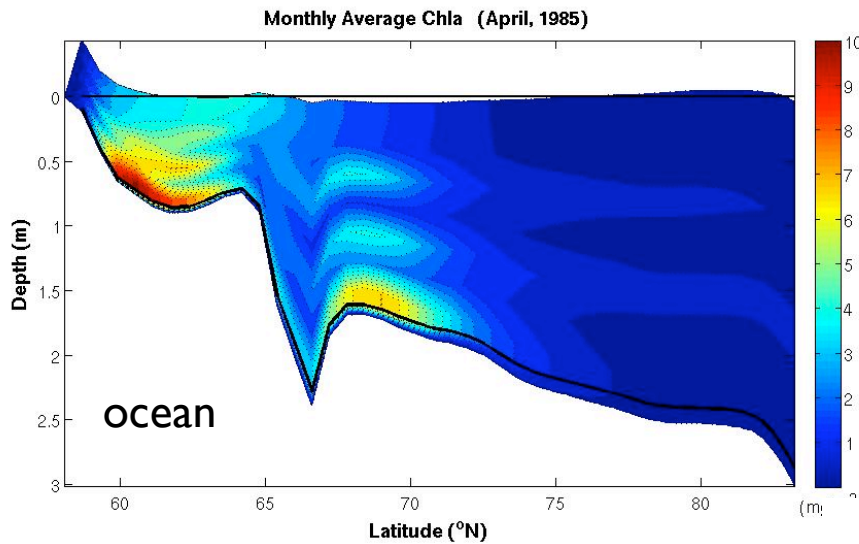
Tracers : Algal nitrogen, Nitrate, Ammonium and Silicate

- ▶ Nitrate/Silicate Arctic ocean climatology
- ▶ Maximum growth-rate is 0.5 doublings/day
- ▶ Light, Silicate, and Nitrate/Ammonium can limit growth
- ▶ Nitrification timescale 67 days
- ▶ No additional salinity, light, or temperature Inhibition.

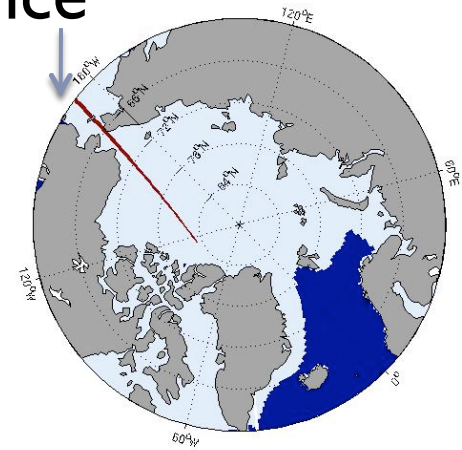


Year 4; Monthly Average

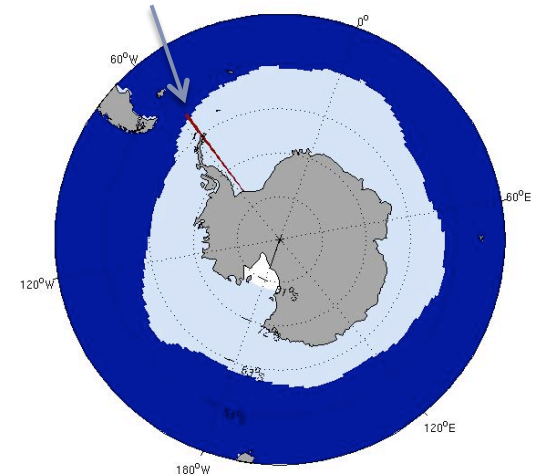
Arctic Chla
April



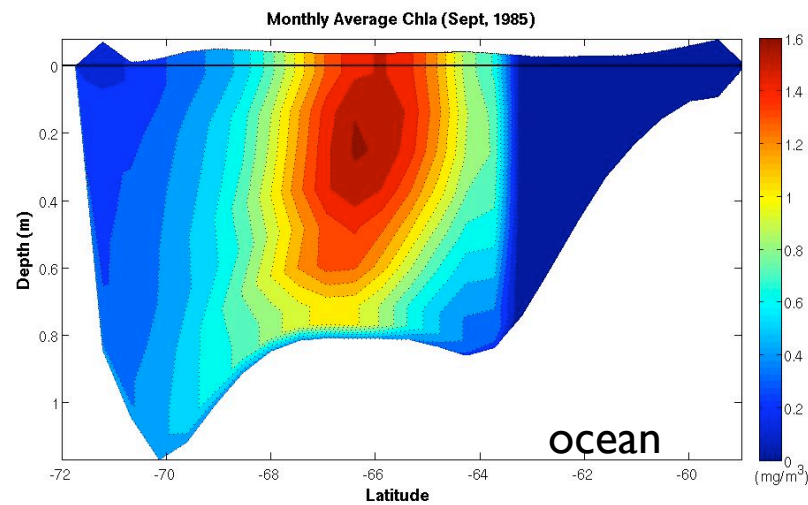
Slice



Slice



Antarctic Chla
September



Future work

- ▶ Coupling with POP bgc/CESM
- ▶ Feedbacks between ice constituents/bgc and ice radiative transfer
- ▶ Dust (fe), black carbon

