

TECHNICAL AREA STATUS REPORT
FOR
WASTE DESTRUCTION AND STABILIZATION

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ACRONYMS AND ABBREVIATIONS

| | |
|------------------|---|
| ALARA | as low as reasonably achievable |
| Ag | Silver |
| As | Arsenic |
| ATI | Alternative to Incineration |
| Ba | barium |
| BDAT | Best Demonstrated Available Technology |
| Br | bromine |
| C | carbon |
| °C | degrees Celsius |
| Cd | cadmium |
| CEP | Catalytic Extract Process |
| CEPOD | Catalytic Electrochemical Production Devices |
| Cl | chlorine |
| CO | carbon monoxide |
| CO ₂ | carbon dioxide |
| Cr | chromium |
| D&D | decontamination and decommissioning |
| DOE | Department of Energy |
| DRE | Destruction and Removal Efficiency |
| EM | Office of Environmental Restoration and Waste Management |
| EM-10 | Office of Planning and Resource Management |
| EM-20 | Office of Environmental Quality Assurance & Resource Management |
| EM-30 | Office of Waste Operations |
| EM-40 | Office of Environmental Restoration |
| EM-50 | Office of Technology and Development |
| EPA | Environmental Protection Agency |
| ER | Environmental Restoration |
| ETCAP | Environmental Technology Cost Savings Analysis Project |
| F | fluorine |
| °F | degrees Fahrenheit |
| FFCA | Federal Facility Compliance Agreement |
| F&ORs | Functional and Operational Requirements |
| ft | feet |
| FY | fiscal year |
| HAZWRAP | Hazardous Waste Remedial Actions Program |
| H ₂ | hydrogen |
| H ₂ O | water |
| HCl | hydrogen chloride |
| HEPA | high efficiency particulate air |
| Hg | mercury |
| HOC | Halogenated Organic Compounds |
| INEL | Idaho National Engineering Laboratory |

| | |
|---------------------------------|--|
| LANL | Los Alamos National Laboratory |
| LDR | Land Disposal Restrictions |
| MEO | Mediated Electrochemical Oxidation |
| MIT | Massachusetts Institute of Technology |
| MLLW | mixed low-level waste |
| mm | millimeter |
| MMT | Molten Metal Technology |
| MS | molten salt |
| MSW | municipal solid waste |
| MWIP | Mixed Waste Integrated Program |
| MWTP | Mixed Waste Treatment Project |
| Na ₂ CO ₃ | sodium carbonate |
| NEPA | National Environmental Policy Act |
| NESHAP | National Emissions Standards for Hazardous Air Pollutants |
| NO _x | nitrate compounds |
| NRC | Nuclear Regulatory Commission |
| ORNL | Oak Ridge National Laboratory |
| OTD | Office of Technology Development |
| Pb | lead |
| PCB | polychlorinated bi-phenols |
| pH | measure of acidity |
| PI | principal investigator |
| PIC | Products of Incomplete Combustion |
| PNL | Pacific Northwest Laboratory |
| POHC | Principal Organic Hazardous Constituents |
| ppm | parts per million |
| PVC | polyvinyl chloride |
| QA | Quality Assurance |
| QA/QC | Quality Assurance/Quality Control |
| QAMS | Quality Assurance Management System |
| RCRA | Resource Conservation and Recovery Act |
| R&D | Research and Development |
| RDDT&E | Research, Development, Demonstration, Testing and Evaluation |
| RFP | Rocky Flats Plant |
| Se | selenium |
| SO _x | sulfate compounds |
| TASR | Technical Area Status Report |
| TCLP | Toxicity Characteristic Leaching Procedure |
| TRU | transuranic |
| TSCA | Toxic Substance Control Act |
| TSG | Technical Support Group |
| TTP | Technical Task Plan |
| TTTDP | Thermal Treatment Technology Development Program |
| TTWG | Thermal Treatment Working Group |

| | | |
|--------|-------|-------------------------------------|
| U.S.S. | | U.S. Steel |
| UV | | ultraviolet |
| UVP | | Ultraviolet Photo Oxidation |
| WDS | | Waste Destruction and Stabilization |
| WMIS | | Waste Management Information System |

TECHNICAL AREA STATUS REPORT FOR WASTE DESTRUCTION AND STABILIZATION

1. INTRODUCTION

The Office of Environmental Restoration and Waste Management (EM) was established by the Department of Energy (DOE) to direct and coordinate waste management and site remediation programs/activities throughout the DOE complex. In order to successfully achieve the goal of properly managing waste and the cleanup of the DOE sites, the EM was divided into five organizations: the Office of Planning and Resource Management (EM-10); the Office of Environmental Quality Assurance and Resource Management (EM-20); the Office of Waste Operations (EM-30); the Office of Environmental Restoration (EM-40); and the Office of Technology and Development (EM-50).

The mission of the Office of Technology Development (OTD) is to develop treatment technologies for DOE's operational and environmental restoration wastes where current treatment technologies are inadequate or not available. The Mixed Waste Integrated Program (MWIP) was created by OTD to assist in the development of treatment technologies for the DOE mixed low-level wastes (MLLW). Throughout the DOE complex, mixed waste is a problem because definitive treatment standards and capacity have not been established and few disposal facilities are available. Currently, DOE sites are storing mixed waste for future disposal, despite the fact that regulations governing the hazardous constituents of the mixed waste requires treatment by specific deadlines.

Statutory requirements which provide the driving force for the development of mixed waste treatment technologies are the Land Disposal Restrictions (LDR) within the Resource Conservation and Recovery Act (RCRA) and the new Federal Facility Compliance Act of 1992 (FFCA). The LDR regulations have mandated schedules for treatment regardless of whether treatment standards and applicable treatment technologies exist. The FFCA requires that plans for technologies and treatment be agreed to by the states. There is a clear need for technologies designed to meet the unique requirements for mixed waste processing and for a system-wide integrated strategy to develop treatment technology and deploy the capability to treat mixed waste (Reference 1).

The mission of the MWIP is to plan and manage the national Research, Development, Demonstration, Testing and Evaluation (RDDT&E) program qualifying emerging and existing technologies on a systems basis for waste treatment and disposal of MLLW, in coordination with the Office of Waste Management's needs and schedules. The MWIP is developing a unified approach for the treatment of all DOE MLLW presently in inventory and that which is being generated. Only a small percentage of DOE's MLLW can be treated by available DOE treatment facilities. The development of integrated process systems capable of effectively treating all of the various types of DOE wastes is the ultimate goal of the MWIP. The treatment facility must include capabilities for front end waste handling, physical and chemical pre-

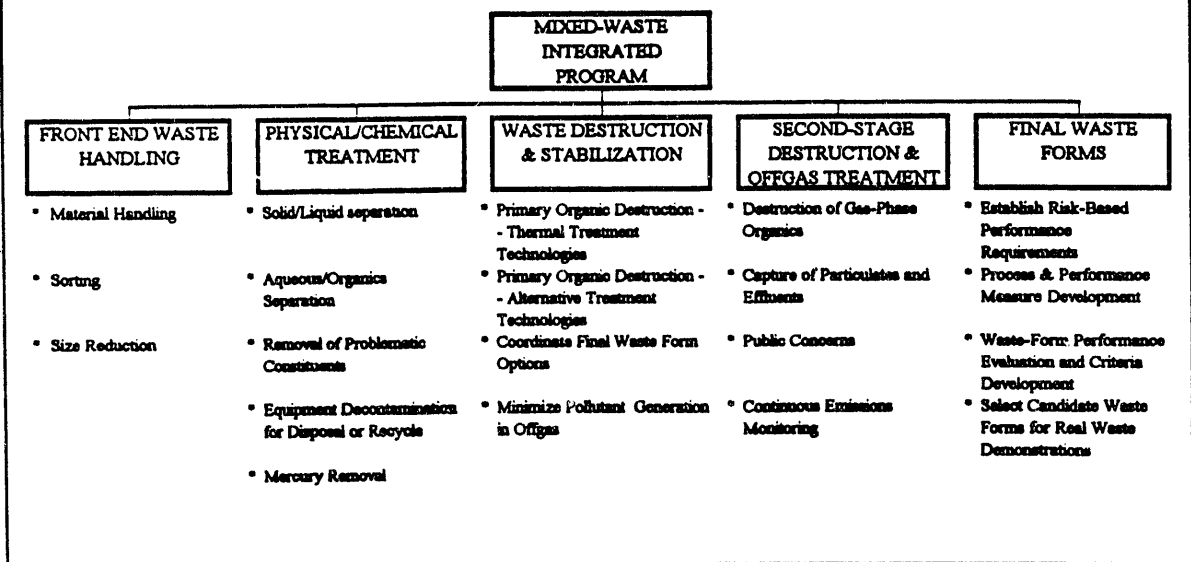
treatment as required, offgas treatment, and the generation of stable final waste forms suitable for disposal.

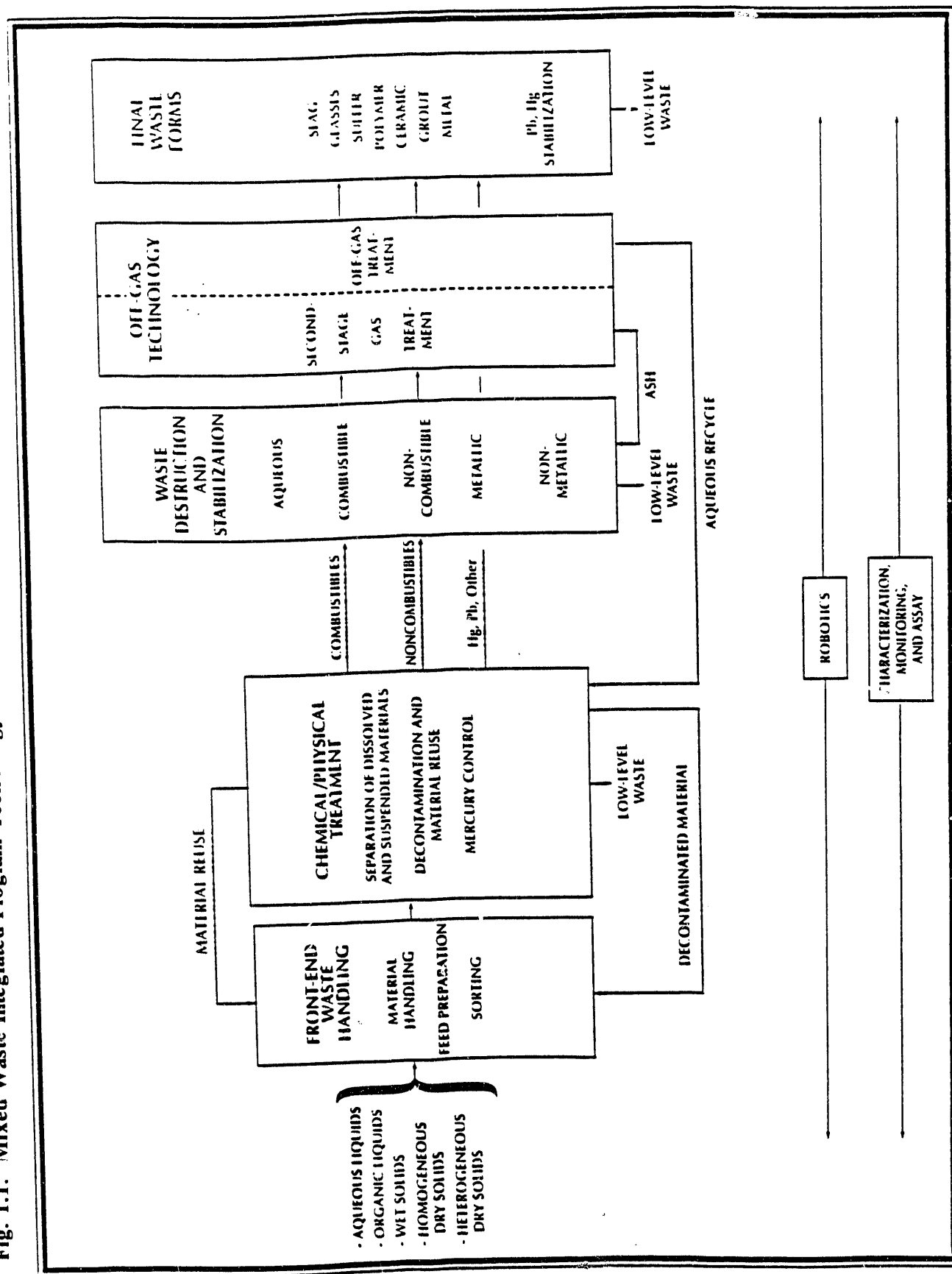
The DOE has established general criteria that successful treatment technologies must achieve. The treatment technologies must be socially and politically acceptable. They must be cost effective, and air emissions, any liquid effluents, and final waste forms must meet or exceed all applicable DOE, EPA, and state regulatory requirements. In addition, the treatment technologies developed must allow the DOE to meet the requirements in the Federal Facility Compliance Agreements (FFCAs). Finally, these process systems must be developed and demonstrated within a time frame that allows the DOE to meet their goal of remediation of all DOE sites within 30 years.

1.1 TECHNICAL SUPPORT GROUP RESPONSIBILITIES

The MWIP has established five Technical Support Groups (TSGs) whose purpose is to identify, evaluate, and develop treatment technologies within five general technical areas representing waste treatment functions from initial waste handling through generation of final waste forms (Figure 1.1). These TSGs are: (1) Front-End Waste Handling, (2) Physical/Chemical Treatment, (3) Waste Destruction and Stabilization, (4) Second-Stage Destruction and Offgas Treatment, and (5) Final Waste Forms. Each TSG consists of a team of experts in the particular technical area covered by the TSG (Table 1.1). These experts come from the DOE sites, contractors,

Table 1.1. Mixed Waste Integrated Program Work Breakdown Structure





EPA, and the NRC. Each TSG must communicate and work closely with the other TSGs to insure that development efforts result in technically successful integrated process systems for the DOE's waste streams.

The MWIP is responsible for issuing solicitations for Technical Task Plans (TTPs). The TTP process is the mechanism that allows researchers to submit proposals and receive funding for the development and/or improvement of treatment technologies as determined to be needed by the Office of Waste Management (EM-30). The function of the TSGs is to identify technology development needs for these solicitations, review the TTPs submitted as a result of the solicitation, prioritize them, and make recommendations for funding. The intent of this process is to identify the range of options proposed to address the MLLW treatment needs and to judge the relative ability of these options to meet the needs of the DOE, including schedule constraints. If any needs are not addressed by submitted TTPs, the TSGs are responsible for follow-up solicitation of proposals, within the DOE or from the commercial sector, to fill the unaddressed needs.

Once selected TTPs are funded, the TSGs are responsible to evaluate development progress and ensure that quality data is generated. With quality data, MWIP can then select the technologies with the greatest potential for successfully treating DOE MLLW. Pilot-scale demonstrations of these technologies will determine necessary design and operational parameters. Finally, MWIP will assist Waste Operations (EM-30) and Environmental Restoration (EM-40) in applying these technologies at full-scale, providing technical support throughout the design, construction, start-up, and operation of the treatment systems. The end result of this work will be to develop and demonstrate technologies that are safe, cost effective, and can adequately treat the DOE MLLW to meet all appropriate requirements.

1.2 WASTE DESTRUCTION AND STABILIZATION TECHNICAL SUPPORT GROUP

The Waste Destruction and Stabilization (WDS) TSG is responsible for MWIP technology development for mixed waste destruction, reduction, and stabilization as identified in Table 1.1.

1.2.1 Responsibilities and Strategy for the WDS TSG

The responsibilities and strategy of the WDS TSG consist of the following key elements. These elements are further discussed in Sections 1.2.4, 1.2.5, and 1.2.6.

- Formulate specific technology requirements for mixed waste destruction and stabilization in support of EM-30 treatment needs as identified by such organizations as the Mixed Waste Treatment Project (MWTP) or by individual sites.
- Identify gaps or deficiencies in existing potentially applicable technologies.

- Evaluate emerging innovative technologies for application to DOE mixed waste treatment.
- Develop a set of commonly accepted standards and criteria by which all technologies can be evaluated.
- Identify similar work in progress and establish a means of sharing information with other similar projects to avoid duplication of effort and avoid funding of specific dedicated development projects.
- Identify at an early stage those technologies that offer the greatest potential for rapid deployment (2 year time frame) versus promising emerging technologies that offer potential improvements over existing technologies for the longer term by using the systems analysis task as a tool.
- Prepare and issue requests for technical proposals that include statements of work for specific technology development areas that focus on identified EM-30 needs.
- Evaluate all proposals and recommend for funding those projects whose proposals are most responsive to the statements of work and which the WDS TSG judges to offer the highest potential for success in the particular application evaluation criteria.
- Review/approve test plans for individual TTPs. Authorize work to be initiated based on test acceptability.
- Assure that all development projects generate quality data.
- Monitor funded development projects and periodically, based on a consensus judgment of the TSG, determine if funding of the project should continue based on the probability of the technology's success within the MWTP flowsheet. The potential for project success will be determined by evaluation of performance, cost, and risk by the MWIP systems analysis.
- Communicate closely with other TSGs to ensure that the candidate waste destruction and stabilization technologies interface acceptably with front end waste handling, chemical/physical treatment, generation of final waste forms, and second stage destruction/offgas treatment to form an integrated process system.
- Develop specifications for deployment of the chosen technologies in the field or pilot plant applications..

Projects that are proposed as cooperative efforts with the private sector are encouraged by the WDS TSG. Such cooperative efforts generally lower costs both through cost sharing and by leveraging the technical expertise found outside the DOE laboratories. Close ties with the private sector reduce the likelihood that an individual project is reinventing an already developed process. Finally, technology developers may have a limited perspective of issues like manufacturability, cost containment, operability, materials of construction, and ultimate market for the technology of which commercial entities must be highly conscious.

Engineering modeling of proposed WDS technologies should be an integral part of a selected technology development program from project inception. A clear methodology for chemical and process engineering evaluation, cost/benefit analysis, and risk analysis is being conducted by MWIP. These analysis are conducted by the System Analysis Group of MWIP. Such efforts will allow MWIP to make definitive comparisons of alternative technologies to the baseline technologies.

1.2.2 Surrogate Waste Streams for Testing

The ability to evaluate candidate treatment technologies on a common basis is of utmost importance to the success of the MWIP. The key to achieving a common test basis is establishment of a standard set of waste streams. A list of candidate waste feed streams of concern to the WDS task area will be produced in collaboration with the Final Waste Forms TSG waste stream effort.

From this list, an effort is underway within the MWIP to define a set of surrogate waste compositions in the form of "recipes" to be used in all future treatment technology development activities. The surrogate waste compositions for each major waste category address the quantities of bulk matrix components, hazardous organic constituents, metals, and surrogate radionuclides. When finalized, these surrogate waste recipes will be included in the requests for proposals as the waste streams on which the development work must focus. Efforts related to DOE waste characterization and the establishment of surrogate waste streams are described in Section 2.0.

1.2.3 Definitions of Current and Emerging Treatment Technologies

With the assistance of TSG members, consultants, and other experts in the field, the WDS TSG will compile and maintain a list of current and emerging technologies for addressing the applicable waste streams. Current technologies include those that are currently in use in the commercial sector and within the DOE. Emerging technologies are those that are under development and have shown promise for implementation by DOE.

1.2.4 Determination of Treatment Technology Requirements

WDS technologies cannot be fully evaluated except as part of a complete

treatment system that considers waste input and output requirements. For purposes of developing a straw man process flowsheet and providing a basis for technology comparisons, the Functional and Operational Requirements (F&ORs) document developed by EM-30's MWTP has made an initial selection of existing waste destruction and stabilization technologies for specific waste streams (Figure 1.2). This selection currently serves as a technology baseline for the MWIP. The current technology baseline will continue to be updated and expanded upon via sources within the DOE, EPA, commercial sector, and from international experience on treatment of similar waste streams.

Technology development requirements will be evaluated based on deficiencies in current options due to baseline technology flowsheet incompatibility or concerns related to safety, performance, permissibility, or cost. The treatment technology evaluation and selection process is described in the next two sections.

1.2.5 Technology Evaluation Criteria

The technology evaluation criteria will address two key activities. The first activity is evaluating proposed technologies for their ability to conform to minimum performance requirements and to improve on baseline technologies for determination of initial funding. The second activity is establishing a basis for correlating progress on funded technology projects with recommendations for continued funding.

The degree to which the criteria are applied will clearly depend upon where the proposed project stands on a maturity scale ranging from proof-of-principal validation to prototype demonstration. The application of these criteria is intended to allow timely decisions on which projects to fund and then to push funded projects as rapidly as possible toward demonstration and subsequent down-selection.

Those R&D projects proposed for DOE funding that fall within the mission of the MWIP and are recommended for funding by the Waste Destruction and Stabilization TSG should incorporate the criteria listed below as a major part of their work scope. The deliverables for each project will specify the degree to which each criterion is satisfied.

- 1) **Waste Feed Acceptance Criteria:** What are the feed compositions and forms that can be accepted by this process and what fraction of the total mixed waste stream does that represent? A detailed inventory of the applicable DOE waste streams collected into waste forms by physical form and matrix composition will be supplied by the MWIP. Standardized surrogate matrix and hazardous compositions for each waste will also be supplied by the MWIP to the WDS. The Principal Investigator (PI) shall incorporate into his scope of work the full range of waste forms which his process will potentially address, and the test plan shall use the surrogate waste description as the test base for each waste form addressed. Feasibility tests must be conducted early in the project to determine the validity of the postulated waste feed capability. Difficulties anticipated or encountered in handling the surrogate waste in a test

This flowchart illustrates the various thermal treatment processes for hazardous waste, categorized by waste type and treatment stage. The processes are as follows:

- Heavy Organics Thermal Treatment:** Heavy organics enter a Car-Bottom Furnace, which produces off-gas and feeds into a Rotary Kiln Incinerator. The Rotary Kiln Incinerator produces off-gas and feeds into a Low-Residue Thermal Treatment unit.
- Heavy Combustibles Thermal Treatment:** Heavy combustibles enter a Rotary Kiln Incinerator, which produces off-gas and feeds into a Low-Residue Thermal Treatment unit.
- Light Combustibles Thermal Treatment:** Light combustibles enter a Refractory-Lined Controlled Air Incinerator, which produces off-gas and feeds into a Low-Residue Thermal Treatment unit.
- Mercury Wastes Thermal Treatment:** Mercury wastes enter a Roaster Bakeout unit, which produces heterogeneous dry solids and feeds into a Car-Bottom Furnace. The Car-Bottom Furnace produces off-gas and feeds into a Mercury Removal unit, which then feeds into a Thermal Treatment Unit for Nonmetals.
- Nonmetals, Ferrous Metals, and Nonferrous Metals Thermal Treatment:** These materials enter a Roaster Bakeout unit, which produces heterogeneous dry solids and feeds into a Tunnel Dryer. The Tunnel Dryer produces off-gas and feeds into a Thermal Treatment Unit for Nonmetals. The Thermal Treatment Unit for Nonmetals produces off-gas and feeds into a Joule Melter, which then feeds into a Ferrous Metals Melt/Slag unit. The Ferrous Metals Melt/Slag unit produces off-gas and feeds into a Nonferrous Metals Melt/Slag unit. The Nonferrous Metals Melt/Slag unit produces off-gas and feeds into a final unit labeled 'TO BE DETERMINED'.
- Final Products:** The final products of the thermal treatment processes are Ash and Slag, Off-Gas Treatment, Glass/Ceramic Final Form, Ferrous Metals Final Form, and Nonferrous Metals Final Form.

will be discussed with the WDS TSG and modifications to the test base will be considered. It must be recognized that deletion of any components from the surrogate matrix or hazardous components list will imply a narrower scope for the application of the technology. The TSG will ascertain whether the narrower waste scope adequately addresses a DOE waste stream and whether the volume and uniqueness of that stream represents a sufficient problem that the cost versus benefit derived justifies the investment in the technology development.

2) **Process Effectiveness:** The key questions are how well does the technology perform and what are the anticipated advantages over existing alternatives. Feasibility studies with key waste constituents should be conducted early in the project to define expected process effectiveness. For the primary treatment unit under development, characterization of the treatment environment as to the range of pressures, temperatures, temperature profiles, and flow rates is required. The DREs for the POHCs in the surrogate will be measured at the exit to the last destruction stage of the process, as will the identity and concentration of all PICs and the concentration, composition, and size distribution of particles in the offgases. For some technologies, the TSG may deem it important to measure some or all of the above parameters at the exit to the primary destruction chamber as well. Mass and energy balances will be developed for all constituents in the waste feed. In particular, non-organic hazardous components in the secondary waste will be determined. Although any appropriate methods for characterization and analysis for the above determinations may be employed, the standard EPA approved methods will be the baseline and will be used to verify the accuracy and reliability of any nonstandard method. The MWIP will provide direction to the PIs regarding the characterization protocol to be used.

3) **Secondary Waste:** At the earliest possible project stage, the nature and quantity of any secondary wastes generated should be determined on the basis of the experimental test results. This will serve as an important point of comparison with alternate technologies and will be used as a justification for further funding of the development project. Strategies for controlling the composition of secondary waste streams will be an integral part of each project.

4) **Offgas Composition:** The nature and quantity of offgas produced by the primary technology and any secondary units is of paramount importance because this is often the primary issue for public and environmental impact. Toxic and radioactive constituents should be determined at an early stage to assure that an assessment of cleanup needs is made and that regulatory/public concerns are addressed as early as possible.

5) **ALARA Concerns:** Any potential concerns associated with maintaining

operator exposure to radioactivity and toxic substances as low as reasonably achievable (ALARA) should be addressed at an early stage. Such factors translate to the reliability and maintainability of the technology while treating potentially highly radioactive and toxic waste streams. The technology should a) minimize buildup of such materials in the process, b) require minimum personnel access into the process internals for any purpose, and c) be potentially adaptable to remote operation in the case of penetrating radiation and to secondary containment in the case of treatment of alpha-contaminated wastes.

6) **Permittability:** One of the major issues for implementing technology is the extensive and often exhaustive permitting process. The permitting requirements for all technologies being developed must consider, but not be limited to, requirements under RCRA, TSCA, NEPA, NESHAPs, DOE Orders, as well as state and local requirements. This is a time-consuming and costly process and, since the outcome or time frame cannot be readily predicted, pursuit of an alternate technology that may offer regulatory and public acceptance advantages in addition to technical advantages is justified. Many proposals submitted for consideration tout regulatory or public acceptance advantages without substantiating documentation. The WDS TSG intends to have an EPA representative with extensive experience in permitting as a member of its advisory team. Other advisory team members have extensive experience in permitting as well as technology development. A review of the claims by the PI on these issues will be accomplished as early in the development process as is practical to attempt to substantiate any claims to easier permitting. Additional requirements for deliverables may become necessary to accomplish this review and these will be determined as experience in the review process is gained.

7) **Risk Assessment:** A risk assessment protocol will be developed based on input from the MWIP Risk Assessment team. This analysis will determine the risks associated with implementing and operating facilities incorporating a given technology. Early risk and/or hazards assessments will focus on basic operating parameters and obvious concerns regarding those parameters. Information on how a technology would integrate into a treatment plant will be requested as the technology matures. Flow sheets and the specification of full scale equipment designs will be accomplished as early in the development process as practical to allow timely risk assessments and to identify potential show-stoppers early in the development process. Input information required to perform risk assessments for a technology will be specified by the Risk Assessment team.

8) **Cost/Benefit Analysis:** A major concern is how a technology compares to other options in cost of implementation and operation, including treatment

costs, costs associated with the management of generated secondary waste, and the cost of meeting regulatory and permitting requirements. A life cycle cost breakdown will be performed, the depth of the analysis dependent on the maturity of the technology development and the level of uncertainty in key elements of the analysis. The analysis will be performed by the MWIP Cost/Benefit team with input from the TSG. Specific information requirements for accomplishing these analyses, and how uncertainties in the values of parameters which are dependent on technology maturity will be treated, will be provided by the Cost/Benefit team.

9) **Compatibility in an Integrated Process System:** During the development process, a waste destruction or stabilization technology should focus on the ultimate requirement to interface acceptably with front end waste handling, generation of final waste forms, and second stage destruction and offgas treatment to form an integrated process system. This interface will be analyzed by the MWIP performance assessment team. Specific requirements for data necessary to complete the performance assessment will be provided by the performance assessment team.

1.2.6 Selection of Development Projects to be Funded

The most important single mission of the WDS TSG is to match specific technology requirements identified by the TSG with proper selection of development projects to be funded. The criteria by which treatment technologies will be evaluated were presented in the previous section. For each particular application within the technical area, a screening of existing applicable technologies versus promising emerging technologies will be made. The technology selection process should follow a two-tiered approach. At an early stage, those technologies that offer the highest potential for rapid deployment in the field will receive highest initial emphasis. Concurrently, those emerging technologies judged by the TSG to offer clear potential for improvement over existing technologies in the longer term will be funded for further development. To avoid duplication of effort and unnecessary funding of work, any similar work in progress both within and outside the DOE will be identified and arrangements made to share the information generated from that project with the WDS TSG.

For technology areas that are identified by the TSG to require development, requests for technical proposals will be prepared and issued by the TSG that include a very specific statement of work. The request for proposal package will include a questionnaire called the WDS Technology Proposal Fact Sheet, to assist in providing a common basis for proposal evaluation. The fact sheet will address the issues upon which the evaluation criteria are based (Section 1.2.5). Proposals received by the TSG will be evaluated and projects will be recommended for funding on the following basis: 1) responsiveness to the proposal's statement of work, and 2) highest potential for success as judged by the WDS TSG using the evaluation criteria outlined in

Section 1.2.5.

1.2.7 Experimental Plan

The PI for a project funded in the MWIP WDS technical area shall develop an experimental plan for the project. The experimental plan is a document which outlines planned tests, the objectives of the tests, a schedule for testing, the technical approach and major parameters for each test, and the criteria to be used to evaluate the outcome of a test in terms of demonstrating success. The intent of the experimental plan is to ensure that the PI's planned experiments support the specific goals of the project and provide a logical means of evaluating the results of tests relative to established WDS criteria. Administrative controls will be developed to ensure that the test plan is approved by the TSG before experimental/development work is conducted. The experimental plan should include the following sections:

- Objectives
- Background
- Schedule
- Description of:
 - Technical Approach/Problems
 - Major Parameters
 - Surrogate(s) to be Tested
- Success Criteria
- References

1.2.8 Quality Assurance/Quality Control

The QA/QC program for any funded TTP shall describe the provisions that will be implemented to provide confidence that a development project activity will be performed in an acceptable and consistent manner. The experimental plan submitted with any funded TTP must satisfy all applicable and relevant QA requirements established by codes, standards, and regulations governing waste treatment development efforts imposed by federal, state and local authorities. The plan must follow DOE Order 5700.6C for all experimental operations. It is recommended that the EPA document QAMS 004, "Guidelines and Specifications for Preparing QA Program Plans" be followed for WDS projects. A plan for verifying and documenting appropriate operation and calibration of all measuring instrumentation will be completed by the PI and submitted to the MWIP. Adequate recording and documentation of the performance of each experiment shall be provided and shall follow general QA procedures. Proposed data analysis methods shall describe, as a minimum, the following requirements:

- Acceptance and implementation of general QA/QC principals by the personnel and facility performing the work.

- The PIs and their management accepts responsibility for the implementation, assessment, and improvement of a Quality Assurance Program. They also accept responsibility for the accomplishment of the experimental plan.
- Performance readiness evaluation should be performed prior to major scheduled or planned work to verify at least the following characteristics sufficient for the type and level of development:
 - Work prerequisites have been satisfied;
 - Technical and QA procedures have been reviewed for adequacy and completeness;
 - Personnel have been suitably trained and qualified;
 - The proper equipment, materials, and resources are available;
 - A system for test control, inspection, and equipment calibration exists;
 - A QA record system exists.

Technical expert(s) from applicable Technical Support Groups (TSGs) may be selected to participate in the Performance Readiness Evaluation Process (Reference 2).

1.2.9 Project Deliverables

The following are examples of deliverables to which each funded project may be required to commit:

- A list that defines DOE waste streams applicable to the technology;
- Descriptions of surrogate wastes to be used by the project for evaluating the technology;
- Success criteria for evaluating performance of proposed demonstrations;
- A comprehensive list of competing technologies;
- Test reports of completed demonstrations;
- Detailed project schedules;
- Periodic status reports; and
- Technology specifications for the deployment of the technology in a waste treatment system.

1.3 Scope of Technical Area Status Report

Technical Area Status Reports (TASR) will be prepared and updated annually by each MWIP TSG. The purpose of this Technical Area Status Report is to detail the responsibilities and to document the efforts of the Waste Destruction and Stabilization TSG. Because the cornerstone for this program is the accurate identification and description of DOE mixed waste streams, the current status of waste characterization efforts will be reviewed. This review will identify EM-30 and EM-40 waste treatment needs. The current state-of-the-art in treatment technologies for waste destruction and

stabilization will be defined, existing industrial treatment technologies that can be applied to DOF waste streams will be identified and described, and the advantages and disadvantages of these technologies will be reviewed. New or emerging technologies will be discussed and the current stage of development of these technologies will be examined. Previous endeavors in identifying and evaluating thermal treatment technologies will also be reviewed to ensure that OTD-sponsored RDDT&E is not duplicating technologies developed elsewhere and that on-going RDDT&E builds upon the work of others. A comparison of waste treatment needs with treatment technologies will identify technology development needs. Based upon the resulting conclusions, recommendations for future actions will be made.

2. DOE WASTE STREAM IDENTIFICATION

The basic precondition for the efficient and economical processing and disposal of the DOE MLLW is a thorough knowledge of the quantities and compositions of the waste streams. The MWTP recognized the importance of accurate waste characterization information for technology evaluations and therefore funded the MWTP data-gathering activities. The objective of this group was to expand upon the waste volume reports and data bases that had been previously prepared (References 3-6). More detailed waste characterization information was obtained by visiting all the major DOE sites. The following information was desired for each waste stream:

- Waste location;
- Description of how the waste was generated;
- RCRA information (LDR category and EPA waste codes);
- Hazardous constituents;
- Chemical matrix and concentration of major components;
- Radionuclide types, activities, and handling category;
- Physical form and size, container sizes, special packaging information;
- Waste inventory volumes, locations, and expected generation rates.

All new information obtained from these visits were used to update the DOE Waste Management Information System (WMIS) data base. However, much of the desired information was not available for many waste streams.

The waste descriptions are divided into nine classes as can be noted in the Mixed Waste Categories Figure 2.1. The major categories are aqueous liquid wastes, organic liquids, solid process residues, soils, debris materials, special wastes, inherently hazardous wastes, unknown wastes and treated materials. Most liquid wastes (1000's and 2000's) and solid process residues (3000's) are generated in a routine manner from process operations and are subject to full RCRA regulations. They are generally homogeneous within a specific waste stream; however, it has been regularly reported that other foreign materials such as a failed pump may have been included within the process waste drums. Soils (4000's) are to be the subject of a future regulation and will likely be further subdivided at that time.

Debris materials (5000's), as classified by the EPA, do not need characterization before processing because of their heterogeneity and the difficulty in characterization. The debris materials only require treatment by an appropriate technology with the process residues being included in the appropriate process residue categories. The regulations state that debris must be solid materials with an average particle size larger than 60 mm. Debris are materials that:

- 1) have been originally manufactured or processed (specific examples include tanks, pipes, valves, appliances, scrap metal, paper, plastic, rubber, glass, concrete, brick and crushed drums);
- 2) are plant or animal matter; or

- 3) are natural geologic materials.

Debris can be an inseparable mixture (by simple mechanical removal processes) of such materials with soil, liquid, sludge, or other solid waste materials, but the debris must be more than 50% of the total material. Debris cannot be process residuals (1000's and 2000's as described above).

Special wastes (6000's) are those that need specific treatments or are not currently considered in the Mixed Waste Treatment Project because of safety or regulatory concerns. Inherently hazardous wastes (7000's) are those waste streams that cannot be treated to remove the hazardous contamination. The first objective is for the recycle of these materials within the DOE or general commerce if very low level radioactive materials become releasable for general use. Unknown wastes (8000's) are those wastes that cannot reasonably be included into a specific waste category because of a lack of information on that stream. The treated wastes (9000's) will result from treatment processes and should be suitable for disposal. A more detailed description of the subcategories making up a waste category and the constituents of those waste streams is included in Appendix A.

In addition to the matrix category defined in this document, the wastes need to be defined as to radioactive content as either contact handled, alpha wastes, or remote handled. Contaminants will also be defined for each of the streams and will be hazardous organic or toxic metals with appropriate subdivisions. EPA codes provide one method for identifying the hazardous contents.

The last category for each of the groups is an uncategorized group. These groups represent the presence of either multiple waste types or the lack of sufficient information to adequately classify the wastes into a more specific group. It is anticipated that, as this categorization matrix is utilized by the sites, additional waste types will be identified. Therefore open categories remain within the matrix and will be defined as needed.

Waste data gathering activities are continuing and updates to this waste stream information will be published in FY93. The quality of waste data is expected to improve as the sites complete characterization, classification, and treatment activities. However, improvements in data quality and consistency are needed. The waste data would be improved with uniformity in definitions of waste streams and classification of specific streams for each site. Currently, there are differences with respect to classification of high-level waste, transuranic waste, low-level waste, and PCB waste streams. Resolution of these differences would further improve the waste information.

Although the DOE mixed waste information is not sufficient for design purposes, it can be used on a general basis to determine what type of treatment processes are needed and which waste streams have a high priority. Once this determination is made, the TSGs will assess the performance of currently developed technologies to

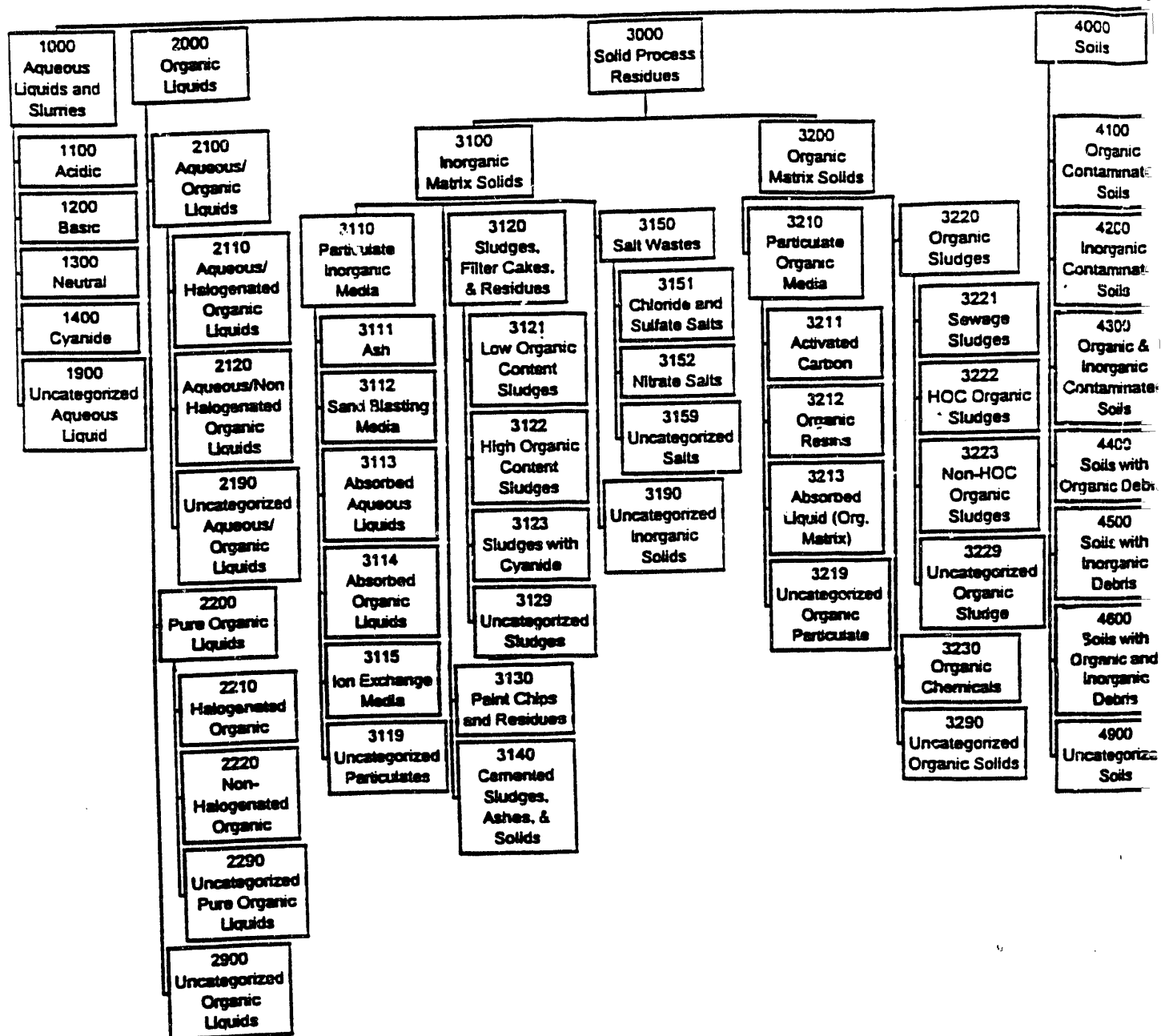
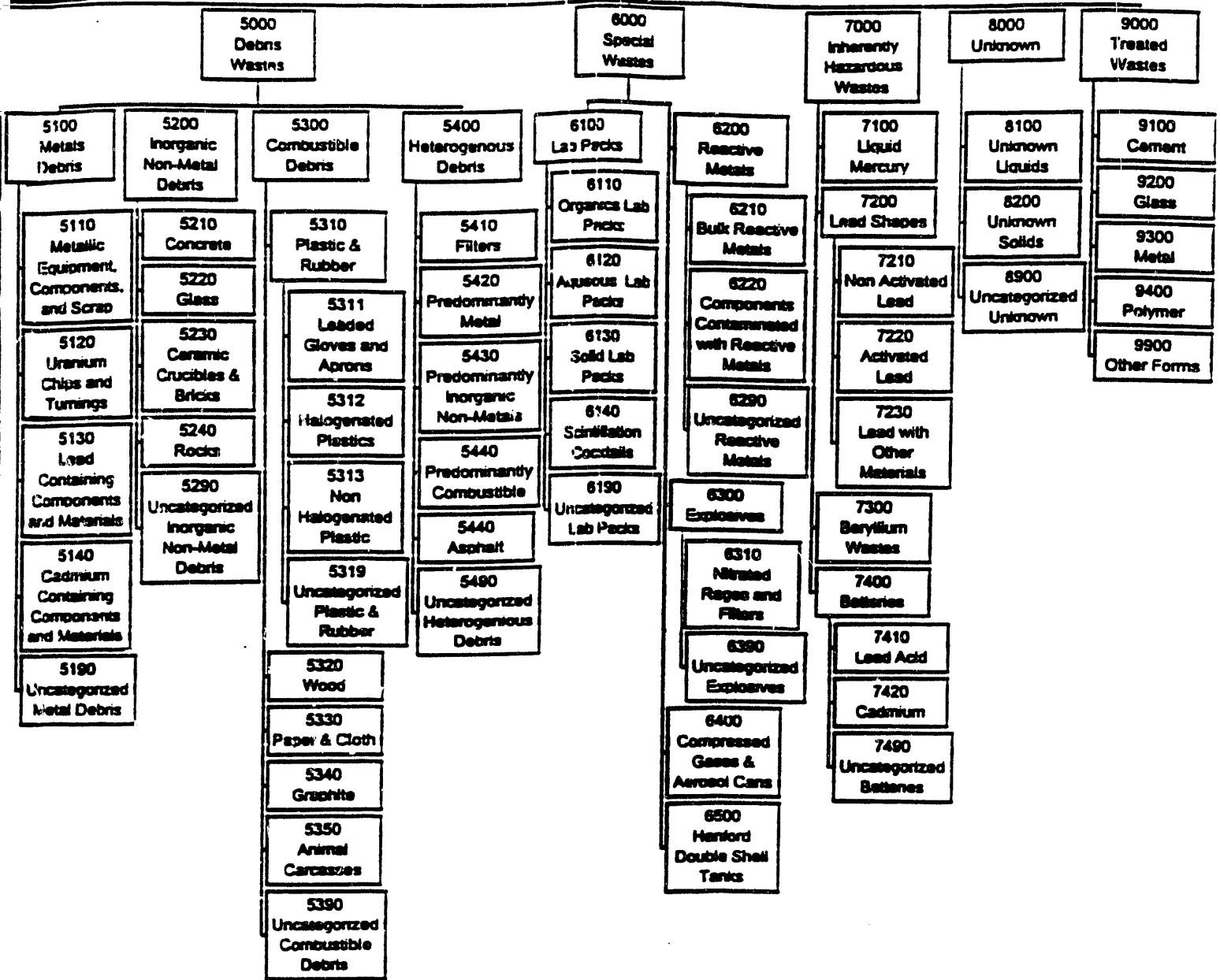


Fig. 2.1. Mixed Waste Categories Figure

Mixed
Wastes



identify which waste streams can adequately be treated with currently available treatment technologies. By matching up the treatment needs with currently available technologies, the TSGs are able to identify those waste streams where treatment technologies have not yet been developed or where existing technologies are inadequate so that improvements or alternatives are warranted.

Several of the DOE waste categories are identified as problematic. This means that treatment technologies have not been identified for these waste streams due to constituents contained within the waste stream or because of their physical waste form. Treatment of these problematic waste streams to render them compliant with land disposal restrictions is the responsibility of the WDS. Detailed waste stream descriptions, hazardous waste components, and quantity information for these problematic waste streams listed below can be found in Appendix B. These problematic waste streams include:

- Cemented sludges;
- Beryllium wastes;
- Reactive metals;
- Tritium contaminated wastes;
- Compressed gases; and
- Chlorine containing wastes.

The DOE has large quantities of mixed cemented wastes. Since these mixed cemented waste streams are listed for both the radioactive and hazardous components of the waste, it is assumed that the cementation process did not adequately stabilize the hazardous component. Therefore, these waste streams will require treatment since they currently do not pass TCLP tests. Since the cement mixed with these waste streams will not burn, very high temperatures will need to be obtained if a thermal treatment process is selected for reprocessing.

Beryllium fines were also identified as a problematic waste stream. Since beryllium is a listed carcinogen, the Clean Air Act has set very restrictive feed and emission limits on any thermal treatment of beryllium containing wastes. In fact, these limits are so low that the thermal treatment of high concentrations of beryllium waste is prohibited. The Waste Destruction and Stabilization TSG will need to identify possible treatment technologies for this waste stream.

Additional problematic waste streams identified are those which contain reactive metals. The use of a thermal treatment process for treating this waste stream creates some problems, since reactive metals can cause extensive damage to the refractories. These waste streams not only create problems in the treatment process, but also pose a waste handling problem due to the ignitability of the reactive metals.

The addition of tritium contaminated wastes to the problematic waste list is due to the fact that this waste stream creates an offgas problem (creates tritiated water which will pass through pollution control systems to the environment). Compressed gases were added because currently the MWTP has not identified a treatment system that

can handle this waste stream. Chlorine-containing wastes are included because these materials create problems in offgas treatment and waste form stability.

Environmental restoration wastes, particularly those resulting from the decontamination and decommissioning of many DOE facilities, are anticipated to be of a large volume, although very little data are available. Specific information about many of these wastes is needed so that more comprehensive evaluations and plans can be completed.

Conclusions drawn from the MWTP Data-Gathering Activities indicates several important factors:

- Most wastes are stored/generated at only a few of the 30 DOE sites. Some of the major DOE facilities are listed below:
 - Fernald Environmental Management Project
 - Hanford
 - Idaho National Engineering Laboratory
 - K-25 Site
 - Los Alamos National Laboratory
 - Oak Ridge National Laboratory
 - Paducah Gaseous Diffusion Plant
 - Pantex Plant
 - Portsmouth Gaseous Diffusion Plant
 - Rock Flats Plant
 - Sandia National Laboratory Albuquerque
 - Savannah River Site
 - Y-12 Plant;
- The most significant waste volumes for treatment considerations are the inorganic solids and the aqueous waste streams.
- The types of waste being stored and being generated at each site are different. Flexible processes are needed to minimize the number of treatment processes required by EM-30 and EM-40.

The WDS TSG and the MWIP staff are establishing waste feed streams that represent a significant portion of the mixed waste in the DOE Complex and will provide a first cut determination of the ability of new technologies to meet destruction and stabilization guidelines. Two surrogate sludges and one bulk combustible surrogate waste form have been specified to present a first cut at a standard waste description for a significant portion of the DOE Complex's waste streams. The description of these waste forms follows:

- 1) Inorganic Sludge (MWTP 3100 Series): Matrix is 20% vermiculite, 20% diatomaceous earth, 20% ion exchange resin, 25% ferrous hydroxide sludge, 15-20% dry portland cement, and up to 5% water. Sludge will contain 5% to 10%

sulfate and chloride salts and will be contaminated with up to 1% hazardous organics, including halogenated hydrocarbons and toxic metals. EPA waste codes that should be exhibited or contained are: D002, D006, D007, D008, D009, D019, D022, D040, F001, F003, F004.

- 2) Organic Sludge (MWTP 3200 Series): Matrix is 25% activated carbon from filters, 25% organic resins, 25% absorbed organics on cellulosic materials (i.e. pulverized corn cobs) and 15-20% dry portland cement. Sludge will contain 5% to 10% sulfate and chloride salts and will be contaminated with up to 1% hazardous organics, including halogenated hydrocarbons, and toxic metals. EPA waste codes that should be exhibited or contained are: D001, D002, D006, D007, D008, D009, D019, D022, D040, F001, F003, F004.
- 3) Bulk Combustible Waste (MWTP 5440 Series -- Heterogeneous debris): Matrix is 30% low density cellulosic materials and wood, 20% polyethylene plastic and sheeting, benelex, and polyvinyl chloride plastic, 10% rubber, and 20-50% tramp metal and glass. Waste will be contaminated with up to 1% hazardous organics, including halogenated hydrocarbons and toxic metals. EPA waste codes that should be exhibited or contained are: D001, D002, D006, D007, D008, D009, D019, D022, D040, F001, F003, F004.

Percentages of each waste constituent are approximate and should be adjusted as constituents such as portland cement, inorganic salts, water, and tramp metal and glass are varied. All waste streams will contain surrogate materials to represent the expected behavior of plutonium, uranium, and/or fission products in concentrations expected for low level waste streams. The concentrations of chlorides that may appear as hydrogen chloride in the emissions, and the hazardous organic and toxic metals concentrations will be adjusted to meet the current regulatory constraints on the system's emissions.

3. SUMMARY OF PREVIOUS EFFORTS

There have been several previous studies to identify, review, and assess thermal treatment technologies for MLLW. Of these studies, two in particular have provided significant assistance in the various tasks assigned to the Waste Destruction and Stabilization TSG. Both of these studies involve treatment technologies that can be used to thermally treat DOE MLLW. In addition to these reports, there have been numerous research and development projects involving waste treatment technologies. The results of these study projects can have a major impact on the decisions and recommendations of the Waste Destruction and Stabilization TSG.

3.1 HAZARDOUS WASTE REMEDIAL ACTIONS PROGRAM REPORT

Reference 7, the Hazardous Waste Remedial Actions Program (HAZWRAP) documents a study conducted for the OTD addressing the thermal treatment of DOE waste streams. The report is organized into two volumes. Volume 1 identifies several important issues related to the thermal treatment of waste. A summary of the DOE MLLW quantities and characteristics was assembled. Six generic waste categories were defined and utilized to assess the processing capabilities of the technologies identified. A discussion on the appropriate regulations concerning the thermal treatment of MLLW was also included. The report identifies 35 thermal treatment technologies and gives a brief description of each. All technologies listed in this report were evaluated and ranked according to LDR compliance, operability, applicability, radioactive contamination control, and cost.

Volume 2 contains much more detailed information concerning each treatment technology (including descriptions, the theory of operation, advantages and deficiencies, process support requirements, cost data, developmental status, and reference information). The thermal treatment technologies reviewed encompass operational conventional and unique incinerators, industrial process co-firing units, municipal waste treatment technologies, and emerging treatment technologies.

There are four significant conclusions that are reached in this report and restated here:

- None of the evaluated technologies ranked high in treating all of the general waste categories. This fact illustrates one reason why additional research and development is needed.
- The technologies with the highest ratings were fluidized bed incinerators, rotary kiln incinerators, plasma arc furnaces, and glass furnaces (this is due to the versatility of these treatment technologies). It is not surprising that these are the technologies that are currently used by the DOE, are under construction or are planned for construction in the near future, and/or have significant demonstrations in progress.

- The third important conclusion discussed in the HAZWRAP report concerns an overall approach to treating the waste. Instead of trying to process all of the various combustible waste streams in one thermal treatment unit, a more effective facility would result if at least two thermal processes are utilized. For example, a facility employing a controlled air incinerator and a plasma arc furnace would be highly effective at treating all of the general waste categories. This facility would likely have a lower life-cycle cost, would be more effective at destroying all of the waste types, and would produce very stable final waste form(s).
- The report points out that the evaluations of the waste technologies are highly dependent on the actual circumstances involved. Care must be taken not to overextend the results from a general analysis to a more specific situation or from one specific situation to another.

3.2 Thermal Treatment Working Group Report

The second report (Reference 8) is a product of the MWIP's Thermal Treatment Working Group (TTWG). The TTWG was comprised of a group of thermal treatment experts from throughout the DOE system. The TTWG was tasked with assisting the MWIP in the evaluation and prioritization of thermal treatment options. In addition, this group assisted in developing strategies to expedite the development of technologies that were recommended for further investigation.

The purpose of the TTWG report was to establish DOE mixed waste treatment needs, describe the approach of the TTWG to meet these needs, prioritize DOE mixed waste streams, identify candidate thermal treatment technologies and their associated requirements for near term application. This report covers several important concepts that can aid the Waste Destruction and Stabilization TSG in their efforts to develop waste treatment technologies. Several important issues related to determining a strategy for developing thermal technologies were discussed.

In content, this report is very similar to the HAZWRAP Report (Reference 7). Regulatory issues and public acceptance issues were discussed. Several sources of information concerning DOE mixed waste quantities and characteristic were reviewed. Based upon this information, the waste was grouped into general categories according to similar characteristics that affect how the waste is to be treated. A methodology to prioritize the waste streams was developed and 21 waste streams were ranked accordingly. Thermal treatment technologies were evaluated against applicable DOE needs. The technologies were then ranked according to their ability to treat the general waste categories. The report finishes with information concerning the developmental needs of the treatment technologies and provides several recommendations and conclusions. Ranking of near term technologies in this study based on the criteria of maintainability, safety risk, operability, flexibility, effluent, and maturity, differed from those developed in the HAZWRAP report. Such discrepancies demonstrate that, until operating and performance data are obtained for technologies

on a common basis including treatment of standardized waste streams, clear choices for down-selection cannot be made with sufficient reliability.

It should be noted that although both of these OTD reports identify waste streams that were grouped into general categories, there are still significant efforts needed in characterizing the waste streams throughout the DOE system. More detailed chemical and physical characterization of waste feed streams will be required before ultimate selection of treatment technologies for the MWTP flowsheet can be made. For example, transport properties, such as density and viscosity are needed for liquid wastes. Likewise, to process solid waste, the size of the solids and how they are packaged must be determined. Properties that pertain to waste destruction and stabilization, such as heat of combustion (or heat of reaction) and elemental and chemical analyses must be determined. However, the information that is presented in these reports is an excellent starting point in the efforts to develop and improve treatment technologies.

3.3 RESEARCH AND DEVELOPMENT PROJECTS

For several years, many sites throughout the DOE complex have been conducting research and development of treatment technologies to provide solutions to waste management problems. Demonstrations of many of these technologies have led to successful applications of waste destruction facilities at various DOE sites. These facilities can effectively treat several different types of wastes, thereby reducing the waste volume. Unfortunately, many waste streams are still unable to be treated. Consequently, research and development efforts for these waste streams is continuing at both the individual sites and the DOE complex wide MLLW. The status and results of each of these efforts are too voluminous to be discussed in this report; however, the status of many of the technologies currently being considered are discussed in Section 4.

4. SUMMARY OF WDS TREATMENT TECHNOLOGIES

This section contains descriptions of both innovative and demonstrated commercial treatment technologies that may be applicable to DOE MLLW. The technologies presented in this section are not meant to be inclusive of all waste destruction and stabilization technologies, but rather only to highlight some of the more promising technologies for application to DOE waste streams. Appendices C-E contains a more exhaustive, but still not complete, listing of waste destruction technologies. These technologies include those that are currently being researched or considered for research as TTPs, as well as those that are now commercially available. In many cases, the available technologies will adequately treat a large percentage of DOE MLLW; however, it may still be desirable to expend research funds on these technologies to improve upon them, especially in terms of applying them to radioactive service.

The technologies presented here are divided into two groups, innovative and conventional. These groups are then further divided by similar principles of operation. On first appearance, it would seem that by separating the technologies on the basis of principal of operation that the technologies are also separated on the basis of the types of waste that the technologies can effectively treat. However, this is not the case. In all cases, there are multiple principles of operation that can be used to treat one particular waste stream. For example, organic liquids can be destroyed by using thermal, biological, chemical, and electrochemical treatment processes.

It should be mentioned here that thermal treatment technologies whose primary application is production of a stable vitreous waste form (such as joule melters and microwave melters) rather than waste destruction come under the purview of the Final Waste Forms TSG. These technologies are also being pursued by the MWIP, and overlapping of functions is addressed by close coordination between the two TSGs.

A sub-Technical Support Group has been formed as part of the WDS TSG to focus on technology alternatives to incineration (ATI). This entity is called the ATI sub-TSG. Section 4.3 lists the currently identified alternatives to incineration, some of which are described in Sections 4.1 and 4.2, as well as a discussion of the issues associated with arriving at a clear definition of "alternatives to incineration."

A few research projects were started prior to the organization of the Waste Destruction and Stabilization TSG. These have been considered by the WDS TSG and been determined as technology development that should continue but, in a few cases, with a refocus of their mission. An example is the development of aqueous based organic destruction technology. This effort should primarily focus on aqueous waste clean-up and only secondarily on bulk organic waste destruction. These projects include the following topics:

- Testing of a centrifugal plasma furnace;
- Plasma arc furnace treatment of compacted wastes;
- Catalytic destruction of organics;
- Catalyzed Electrochemical Production Devices (CEPOD);

- Supercritical water oxidation;
- High temperature packed-bed reactor;
- Silent discharge plasma;
- Studies on the volatility of actinides species.

Descriptions, current status, and other information concerning these technologies, as well as other treatment technologies, are included in this section or can be found in References 7 and 8.

4.1 INNOVATIVE WASTE DESTRUCTION AND STABILIZATION TECHNOLOGIES

Innovative technologies are loosely defined as those technologies that are not currently being used on a large scale to treat wastes. These technologies are generally in the development stage. In some cases, there are technologies that have been utilized on a large scale for many years for purposes other than waste treatment. These technologies are still classified as innovative as they have not been effectively demonstrated for waste destruction and stabilization and/or there are several issues still remaining to be worked out before they can be used in full-scale applications.

4.1.1 Thermal Waste Destruction and Stabilization Technologies

Rather than reiterating all of the information contained in References 7 and 8, only a brief discussion of thermal processing is presented here. These references include extensive listing of thermal treatment technologies, detailed information concerning these technologies, additional references where more information can be obtained, and a listing of commercial companies and DOE facilities investigating each thermal treatment technology. These companies and laboratories can be excellent sources of information. Appendices C-E provide summary information on several thermal technologies.

Current WDS TTP activities involving thermal treatment processes include: metal melting technologies and plasma arc incineration. Joule melter technologies and other thermal technologies whose primary function is generation of a final waste form are being addressed by the Final Waste Forms TSG. The metal melting technologies are basically adapted from the metals industry (e.g., induction furnaces and plasma arc melters) and the glass industry (e.g., fuel-fired and joule heated melters). There are also a few new melting processes that are being researched as a waste management tool (e.g., the microwave melter). Although most metal melting technologies are not considered new, these technologies are discussed in the innovative technology section because they have only limited operational experience in the radioactive waste management area.

Melter technologies are currently receiving attention because they offer the potential for requiring no additional treatment of the generated inorganic residues. The waste forms resulting from metal melting of the input waste streams tested so far

appear to meet the identified final waste form standards. Because of the high temperature of the melting operations, melters can be used to destroy organics, although for highly contaminated waste streams an incineration system will likely be more efficient. The use of melters for the destruction of organics are likely to require significant process modifications.

Melters are ideally suited for inorganic waste streams such as inorganic oxides and elemental metals. Furthermore, the chemistry in the melt can be reducing or oxidizing depending on the type of waste form desired. When processing oxides, the final waste form will be a glass or a ceramic, depending on the rate of cooling as well as other parameters. When processing metals, the melt will form a top layer of slag and a bottom layer of molten metal. The slag can then be separated from the molten metal, allowing for recycle of the molten metal. Depending on how the melter is operated, it is possible to oxidize the majority of the radionuclides in the waste so that they will become part of the slag, thereby decontaminating the molten metal. The Waste Destruction and Stabilization TSG will evaluate molten metal technologies to determine their applicability to DOE MLLW.

Many thermal treatment devices share similar research needs. The common needs for some of the more promising thermal devices are shown in Table 4.1. This table identifies both the advantages and disadvantages for each technology. Many of the disadvantages can be eliminated with additional research. In terms of melters, there are several research requirements that need to be resolved and disadvantages that need to be overcome to ensure safe and proper operation in a radioactive environment. In general, these research needs can be classified as requiring more operational experience, better materials of construction, improved materials handling techniques, less waste pretreatment, control of chemistry in the process, and detailed analysis of the resulting residue and off-gas to determine the constituents that are in these effluents.

Innovative thermal treatment technologies are primarily directed towards the destruction of organics and producing a stable final waste form. In addition to these waste streams, thermal technologies can be used to destroy other compounds such as nitrates and cyanides. At the current time, the majority of funded development in innovative thermal treatment is associated with plasma systems. Based on the conclusions of the HAZWRAP report (Reference 7), this technology was selected as a high priority for research, development, demonstration, testing and evaluation for four main reasons. First, previous efforts to identify and evaluate thermal treatment technologies rated plasma treatment as a strong candidate for its application to many of the DOE waste streams. Second, a plasma system has demonstrated the ability to treat drums of some types of waste without first removing the waste. This capability can be beneficial to the DOE because there is currently a large portion of DOE wastes that are stored in drums, and it is undesirable to remove this waste from the drums and sort through the waste because of the associated hazards and costs. Another important benefit of the plasma technology may be its ability to simultaneously treat organics, metals, and inorganic oxides which also precludes the need to remove waste from drum for sorting. Another element that led to the decision to concentrate at this

time on plasma technology was the fact that there are already ongoing efforts in other DOE organizations to investigate other melters such as joule heated melters and arc melters.

There are several variations in plasma systems that can be utilized for waste treatment. One of the significant variations is in how the plasma is generated. There are different methods that can be used to create a plasma, and therefore there are various proposed TTPs to investigate these different types of plasmas. There has also been significant work in the type of chamber used for the plasma system. At this point, one of the important TTPs that has been approved for funding is to build a fixed hearth plasma system using a transferred arc plasma. This system will be used to test various subsystems and to verify the treatment of various types of waste.

4.1.2 Chemical Waste Destruction and Stabilization Technologies

There are no studies identified by the WDS TSG that investigate the applicabilities of innovative chemical destruction technologies to DOE MLLW.

4.1.3 Electrochemical Waste Destruction and Stabilization Technologies

Two TTPs were recommended by the TSG for FY93 funding on a lower priority. These TTPs investigate the feasibility of treating DOE MLLW streams with electrochemical destruction technologies. The Mediated Electrochemical Oxidation (MEO) process is designed to convert organic materials to CO₂, H₂O, and inorganic nontoxic ions. The CEPOD system uses a powerful regenerated oxidative catalyst to dissolve, destroy, or decontaminate organic and inorganic materials. Due to funding constraints, these two electrochemical treatment technologies were not supported.

4.1.4 Radiolytic Waste Destruction and Stabilization Technologies

There are no studies identified by the WDS TSG that investigate the applicabilities of innovative radiolytic waste destruction technologies to DOE MLLW.

4.2 CURRENTLY AVAILABLE WASTE DESTRUCTION AND STABILIZATION TECHNOLOGIES

These technologies are loosely defined as those technologies with significant waste treatment experience. As a result of this experience, these processes are well understood, the types of waste that can be treated with these process is known, and the limitations are clearly established. Consequently, once a waste stream is characterized, it can be determined if these processes are appropriate and can adequately treat the waste streams without a significant amount of research and development. In addition, these processes can normally be designed and operated without a great deal of testing. However, this definition does not imply that improvements to these technologies are not needed for successful MLLW service.

Table 4.1. Thermal Treatment Technologies Advantages and Disadvantages

| TYPE OF DEVICE | ADVANTAGES | DISADVANTAGES |
|----------------------------|--|---|
| All Thermal Units | <p>High organic destruction efficiency.</p> <p>Can operate primary chamber in substoichiometric or excess air mode.</p> <p>High volume reduction factor.</p> | <p>Feed addition subsystems are prone to have problems and are difficult to seal.</p> <p>Ash removal subsystems are prone to have problems and are difficult to seal.</p> <p>Requires large energy input.</p> <p>Fuel-fired units have a large quantity of offgas.</p> <p>Continuous emissions monitoring techniques are limited.</p> |
| All Incineration Units | <p>Proven technologies that are well understood. DOE already has various types of incinerators installed and capable of processing a variety of wastes.</p> <p>Mandated destruction technology by EPA for many types of wastes.</p> <p>Can have large waste feed changes.</p> <p>Can have variations in waste composition.</p> | <p>Difficult and costly to permit.</p> <p>Lack of public knowledge has made general public opposed to incineration.</p> <p>Ash residue may contain leachable heavy metals.</p> |
| Controlled-Air Incinerator | <p>Calm primary chamber reduces particulate carry over.</p> <p>Proven in radioactive service.</p> | <p>High metals content can be troublesome.</p> <p>Only moderate throughput capabilities.</p> |
| Rotary Kiln Incinerator | <p>High throughput capacity.</p> <p>Capable of handling trash noncombustible materials.</p> | <p>Poor seals at both ends of kiln.</p> <p>Tumbling action results in high particulate loading in offgas.</p> |
| Fluidized Bed Incinerator | <p>Excellent heat and mass transfer.</p> <p>Very effective for liquids and sludges.</p> | <p>High particulate loading in offgas.</p> <p>Requires extensive size reduction for solids.</p> |
| All Melter Units | <p>Can add a flux to lower the melting point.</p> <p>For waste streams that are corrosive or require excessive temperatures for destruction and melting, a freeze wall (skull) can be used to protect the chamber refractory.</p> <p>Produces a stable waste form that may not require additional treatment.</p> <p>Can control the redox state in the melt and in the gas space above the melt.</p> <p>Can use additives and melt cooling rate to result in a more stable final waste form.</p> | <p>Poor characterization of process offgas.</p> <p>Materials of construction for electrodes, refractory, etc. are subject to a harsh chemical environment and high temperatures which result in rapid degradation.</p> <p>Volatility of radionuclides when using melters is unknown.</p> <p>Slower organics throughput compared to incineration.</p> |
| Joule-Heated Melters | <p>Heavy metals and nongaseous radionuclides are dissolved or encapsulated by the melt limiting the emissions of these waste constituents.</p> <p>Containment of heavy metals or non-gaseous radionuclides in the melt can be enhanced by oxidation of these constituents or by using a cold cap over the melt.</p> | <p>Metallics are undesirable and may need to be sorted out of the waste.</p> <p>Requires a long residence time for the melted material.</p> <p>Waste sizing will be required.</p> <p>The chemistry of the melt and ultimately the stability of the glass will be affected by variations in feed composition.</p> <p>Chemistry must be determined for various types of waste feeds.</p> <p>Salts in waste or as a result of the destruction of waste will cause problems with glass stability.</p> <p>Concern about high nitrogen oxides generation.</p> <p>Application to waste processing is very limited and requires testing on more types of waste.</p> <p>Need to identify optimum process configuration including air pollution control equipment and melt tapping system.</p> <p>Need to determine if additives are needed.</p> <p>Metal volatility unknown.</p> |
| Plasma Melter Unit | <p>Will handle bulk metals and other inorganics.</p> <p>Can be operated at very high temperatures to effectively destroy organics.</p> <p>Plasma produces free radicals that promote destruction of organics.</p> <p>Plasma system can feed large containers such as drums.</p> <p>Electrodes are not in contact with the melt.</p> <p>Can handle variations in waste composition so that potentially little or no waste sorting is necessary.</p> | |

The EPA has been developing and tracking technologies for hazardous waste treatment. The EPA also encourages development in the private sector. Many of these activities are directed toward Environmental Restoration (ER) problems at Superfund sites. Some of these technologies can be used to clean up DOE ER problems but, since their primary function is the destruction of hazardous organic constituents, they may also have applications in MLLW treatment.

4.2.1 Thermal Waste Destruction and Stabilization Technologies

As was previously stated, the conventional thermal treatment technologies are incinerators or processes that are very similar to incineration. Waste incineration is well established and has been in use for over 100 years. With the recent concern about the environment and waste management, many improvements have been made in this technology. Currently, incineration is one of the best waste management tools available. Destruction and removal efficiencies of hazardous constituents in excess of 99.9999% can be obtained. High quality turnkey systems are available that can be tailored to a specific situation. Radioactive waste incineration has been successfully utilized in the U.S. and in many other countries.

The large incineration experience base allows for the successful use of this technology on a wide variety of waste streams. A variety of incinerator types and configurations have been developed to accept specific combustible waste types. Three of the most commonly used incinerator types for waste treatment applications have been controlled air, rotary kiln, and fluidized bed incinerators. Controlled air incinerators are typically used for low density packaged waste streams but have been adapted to other applications such as medical waste treatment, cremation, and liquid waste combustion. Rotary kiln incinerators are considered the workhorse of the hazardous waste incineration industry due to their ability to accept a wide variety of waste physical forms and sizes. Fluidized bed incinerators have been adapted to accept liquid and slurry waste, sludges, and solids that have been reduced in size.

In addition to the above mentioned incinerator types, other incinerator configurations are available to treat more specific waste streams. Multiple hearth incinerators are primarily used for treatment of sludges containing organics. Liquid injection incinerators are generally designed to accept liquids and slurries, and with some modification gaseous wastes also. Large (> 1000 Ton/day) incinerators are used for municipal solid waste combustion. Industrial combustion processes originally developed for other applications are now also being used for waste treatment. Such systems include waste heat boilers, furnaces, and kilns and each takes advantage of the heat content of organic waste for their primary process application. References 7 and 8 provide more detailed lists and descriptions of these incineration technologies.

The disadvantages for waste incineration that could be addressed by research are tabulated in Table 4.1. Specific research needs for the individual types of incinerators are detailed in Reference 8. In addition to these research needs, there are a few other issues to consider.

For some waste streams, more than one treatment technology may be required in series to effectively destroy the waste's organic fraction, stabilize the inorganic fraction, and minimize overall offgas emissions. For example, it may be more cost effective and result in a better system if waste with a high organic content is first processed in an incinerator and the resulting ash is then processed in a melter. Incinerators are more efficient at processing wastes with a high organic content and will likely result in a higher waste processing rate. The small volume of resulting ash can then easily be handled in a small melter to produce a stable final waste form. In addition, the chemistry in the melt and in the offgas is much easier to control in this scenario and may, therefore, result in a better final waste form and fewer emissions in the offgas.

Likewise, an aqueous waste that has a high nitrate and dissolved metals content could be first treated in a fluid-bed calciner. The calciner converts the nitrates to nitrogen oxides, which leave as a gas that can be treated in the pollution control system. Simultaneous to destroying the nitrates, the calciner converts the dissolved metals to metal oxides which can then be vitrified in a melter. This two step process is more efficient than using the melter alone as calciners can oxidize the metals much more efficiently than other technologies and can also evaporate off the moisture at a rapid rate and in a controlled manner. There may be other benefits as well, such as easier temporary storage because of the resulting volume reduction and safer form of the waste. There are limitations to calcination such as the presence of certain species, e.g. sodium nitrates, makes the process unacceptable because the calcined waste forms a viscous material.

Thermal treatment technologies have been instrumental in DOE's current waste management practices for two main reasons. First, thermal treatment is a very effective method for waste destruction and stabilization. In fact, for many types of wastes, the EPA has determined that incineration is the Best Demonstrated Available Technology (BDAT), and therefore federal regulations require that these wastes be treated by incineration. Secondly, the DOE presently has incinerators (many more are under construction or are planned for construction, and/or in the permitting stage). These future facilities will have a major impact on DOE waste management strategy and practices but there will continue to be many required technology improvements and other issues to be resolved.

It has been suggested, particularly by some developers of alternative technologies, that incinerators are not permissible and that, only by substituting such alternative technologies, will treatment of MLLW be allowed. However, the commercial sector has demonstrated the permissibility of incinerators. A key to implementation of incineration and alternative technologies for treatment of MLLW will be the involvement of the public and regulatory agencies at an early stage in the process.

4.2.2 Chemical Waste Destruction and Stabilization Technologies

Chemical treatment destruction technologies are processes in which hazardous wastes are altered by chemical reactions. These chemical reactions can destroy the

hazardous components. In other cases, the resulting product can still be hazardous, although in a more convenient form for further processing or disposal. Possible chemical treatment processes includes oxidation, reduction, ozonation, and electrolysis. Limitations in the use of chemical treatment processes are the low solubilities of some metals, impurities in the waste that can inhibit reactions, and the potential for generating equally hazardous byproducts.

The Waste Destruction and Stabilization TSG will evaluate the applicability of these waste treatment technologies to DOE MLLW streams.

A determination needs to be made between the Waste Destruction and Stabilization and the Chemical/Physical Treatment TSGs as to which technical area these technologies should reside in, on a case-by-case basis.

4.2.3 Biological Organic Destruction and Stabilization Technologies

A discussion on organic waste destruction technologies would not be complete without including biological treatment processes. Biological processes have been successfully used by industrial wastewater treatment facilities for years. Oil, chlorinated hydrocarbons, and heavy metals can be reduced within a biological treatment facility. Limitations of this technology include sensitivity to changes in waste stream concentration, metal salts concentration, and pH changes. The available information from development efforts to date must be reviewed by the WDS TSG to evaluate the applicability of biological treatment technologies to the DOE's waste streams.

A determination needs to be made between the Waste Destruction and Stabilization and the Chemical/Physical Treatment TSGs as to which technical area these technologies should reside in, on a case-by-case basis.

4.3 ALTERNATIVES TO INCINERATION

As previously mentioned, the WDS TSG has created a sub-TSG to focus on technology alternatives to incineration in support of the DOE's waste management efforts. The ATI sub-TSG has identified alternatives to incineration and broadly characterized these technologies as either thermal or non-thermal in nature. Following is a current list of technology alternatives to incineration assembled by the ATI sub-TSG:

Thermal Treatments:

- Wet Air Oxidation - Catalyzed
- Wet Air Oxidation - Non-catalyzed
- Calcination
- Supercritical Water Oxidation
- Steam Reforming
- Microwave Processing

Metal Melting and Recycle

Non-Thermal Treatments¹:

Biotreatments

Electrochemical Oxidation

Electron Beam and Other Radiolytic methods

Silent Discharge Plasma or other ozonation

Corona Discharge

UV Mediated Oxidation

Sonochemical Destruction

Supercritical CO₂

Extraction

The general types of wastes streams for which the above listed technologies will focus are as follows. It is not assumed that all the listed technologies will be applicable to all of these waste types.

- Nitrates
- Chlorides
- Radioactively contaminated
- Tritium contaminated mixed wastes
- PCB contaminated materials
- Ion Exchange resins
- Plastics and other room trash (primarily cellulose)

A significant challenge to the ATI sub-TSG is to arrive at a consensus concerning what technologies should be identified as an "alternative to incineration" and the justification for their inclusion as such. For example, is a technology an alternative because it will address a waste stream that is not amenable to incineration or is it an alternative to the process of incineration? An example of the latter is the treatment of vermiculite which is non-combustible and therefore does not burn at typical incinerator conditions.

To be considered an alternative to incineration a prospective technology could show an advantage in one or more of the following areas and be equal in those areas not exceeded:

- handle wastes not handled by incineration, or handle a wide range of waste equal to those handled by incineration;

¹ A determination needs to be made between the Waste Destruction and Stabilization and the Chemical/Physical Treatment TSGs as to which technical area these technologies should reside in, on a case-by-case basis.

- applicable to small problematic waste streams;
- technically surpass incineration capabilities;
- produce more benign wastes, smaller volumes, or no secondary waste;
- reduce or eliminate fugitive emissions;
- operate at lower costs; and
- meet or exceed all regulatory requirements.

5. CURRENTLY FUNDED WDS ACTIVITIES

The current efforts in the Waste Destruction and Stabilization TSG are focused on the TTPs funded in FY93. These activities are summarized in Table 5.1. Two thermal treatment technology categories have been funded in FY-93, the plasma hearth furnace and molten metal/slagging technologies.

Table 5.1. Summary of FY93 Funded TTP Activities

| TITLE | RESPONSIBLE ORGANIZATION | TTP NO. |
|---|---------------------------------|----------------|
| The Plasma Hearth Process | MSE/SAIC | PE021202 |
| The Plasma Hearth Process | ORNL/SAIC | OR132020 |
| Liquid Metal Recycle and Waste Treatment | LANL | AL132001 |
| Magnetic and Nonmagnetic Melt/Slag Treatment of Mixed Waste | PNL | RL332014 |
| Waste Stream Diagnostics and Control for Treatment | ORNL | OR-NEW |
| Control of Metal Emissions from Mixed Waste Incinerators | ORNL | OR132006 |

The plasma hearth development program has been funded by various EM-50 organizations since FY-91. The plasma hearth process is perceived as a highly versatile WDS technology that could treat a wide variety of DOE's mixed wastes. Demonstrations have been performed on simulated compacted mixed waste and on buried mixed waste. This project will demonstrate processing of as many of the surrogate waste streams as possible, using the existing plasma hearth unit, and will also upgrade the system to more closely represent a full-scale production unit.

Molten metal/slagging technologies are being investigated for their potential to treat high metal- content mixed waste streams without requiring significant feed segregation. Metals processed could be recycled for limited uses within the DOE, and the non-metallics will be destroyed or bound up in the slag for subsequent disposal. FY-93 efforts will investigate the technology options, select a preferred molten metal process, and perform limited demonstrations on selected surrogate waste streams. The effort will include significant involvement by a commercial partner.

Two TTPs have been funded to investigate the characteristics of thermal treatment device effluents to aid in proper sampling and analysis of effluents. An effort to characterize the spectrum of effluents from thermal treatment devices, beginning with the plasma arc furnace, is being managed by ORNL. This task will define the surrogate formulations for MWIP-identified waste streams, sampling and analysis

requirements, identify standardized sampling and analysis protocols, and established QA/QC requirements for measurements to be taken for all thermal treatment demonstrations. A second effluent activity will investigate the control of metal emissions from mixed waste thermal treatment units.

6. RECOMMENDATIONS FOR FUTURE ACTIVITIES

6.1 Improved Waste Data

The WDS TSG will continue to direct and develop waste treatment technologies in support of DOE EM-30 needs. There are numerous efforts currently in progress and planned for the near future that are directed at solving the DOE waste management problems. One area that is currently being worked is the identification of waste streams; however, significant efforts in obtaining more detailed chemical and physical characteristics is necessary for the detailed design of full-scale treatment systems. With the waste information that is now being gathered, it is possible to determine which existing technologies can be used, where improvements to existing technologies are needed, and where altogether new technologies must be developed.

It is important to continue working on identifying waste streams where existing treatments are not sufficient to meet DOE needs. This information will serve as the basis for the developments and improvements that are required. It is also important to identify where DOE facilities may already exist. Efforts to identify currently available technologies that can be used to treat DOE waste should continue. Likewise, it would be beneficial to identify the commercial organizations that are processing wastes similar to DOE wastes. These commercial organizations may be able to provide valuable information about various technologies and possibly even process DOE waste at an accelerated schedule.

6.2 Communications in Technology Development

One of the critical elements that must be incorporated in this process is communications. In order to eliminate redundant efforts, ensure effective use of resources, and to make sure that there are no technology gaps, each of the TSGs must communicate with the each other. Coordination as to which TSG should deal with a particular technology is not always straightforward as there is some overlap of the functions for the TSGs. For example, metal melting technologies can be used to destroy organic wastes, to produce an enhanced final waste form, or to recycle metals by producing shielding block, etc. Consequently, a decision must be made as to which technologies and tasks will be covered by the Destruction and Stabilization TSG, the Final Waste Form TSG, and the Physical/Chemical Treatment TSG. Likewise, vitrification technologies can be used to destroy organics and to produce an enhanced final waste form. Therefore, it is important to maintain communications to prevent the duplication of efforts.

It is equally important to make sure that no development needs are overlooked. Without active communications between all TSGs, there is a concern that some areas that require research and development will not be addressed. For example, for certain waste streams the Waste Destruction and Stabilization TSG must specify to the

Physical/Chemical Treatment TSG what pretreatment is required for the destruction and stabilization processes that are being considered. Likewise, the Second-Stage Destruction and Off-Gas Treatment TSG and the Final Waste Form TSG are affected by the processes selected by the Waste Destruction and Stabilization TSG. Therefore, different air pollution control capabilities may be required, and different performance standards may need to be developed for the final waste form. This interaction between the TSGs is particularly important in developing a systems approach to waste management rather than concentrating only on the individual technologies. In addition to the communications between the TSGs, there are other Integrated Programs and Integrated Demonstrations in the DOE that are related to or can use the information gained from the MWIP. The MWIP should also maintain communications with these organizations.

6.3 Future Development Activities

The MWIP's FY-94 Call for Proposals was issued to the field in February 1993 (Reference 9). The specific needs identified by the WDS TSG are contained in this call, portions of which are reproduced as Appendix F.

7. REFERENCES

1. *Mixed Waste Integrated Program Technology Needs Statement and Call for Proposals*, Mixed Waste Integrated Program, FY-94.
2. *Guidance for Conducting a DT&E Demonstration Performance Readiness Evaluation*, EM-55, January 1993.
3. *Land Disposal Restrictions Case-by-Case Extension Application for Radioactive Mixed Wastes*, U.S. Department of Energy, Washington D.C., 1991.
4. *Integrated Data Base for 1991: U.S. Spent Fuel and Radioactive Waste Inventories, Projections, and Characteristics*, DOE/RW-0006, Rev. 7, U.S. Department of Energy, Washington D.C., 1991.
5. *Waste Management Information System Data Base (WMIS)*, maintained by HAZWRAP, Oak Ridge, TN.
6. W.A. Ross, C.L. Warner, et al, *Locations, Volumes, and Characteristics of DOE's Mixed Low-Level Wastes*, Waste Management '92, March 1-5, 1992, Tucson, Arizona.
7. *An Assessment of Thermal Destruction Technologies for Application to Department of Energy Mixed Wastes Volume 1 & 2*, DOE/HWP-106, Hazardous Waste Remedial Actions Program, Oak Ridge, TN., August 1991.
8. *Mixed Waste Integrated Program Interim Evaluation Report on Thermal Treatment Technologies*, DOE/MWIP-2, U.S. Department of Energy, September 1992.
9. G. Coyle, EM-542, to TPOs and TPMs, Subject: *Requests for FY-94 Short Form Technical Task Plan Proposals for the Department of Energy Mixed Waste Integrated Program*, February 2, 1993.

APPENDIX A

Mixed Waste Characterization Code Descriptions¹

¹ Information contained in Appendix A was obtained from the following document: Ross, W., Warner, C., et al, *Locations, Volumes, and Characteristics of DOE's Mixed Low-Level Wastes*, Waste Management '92, March 1-5, 1992, Tucson, Arizona.

MIXED WASTE CHARACTERIZATION CODE DESCRIPTIONS

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|-------------------------------|---|---------------------------------|
| 1000 | Aqueous Liquids and Slurries | Aqueous solutions or slurries that have less than 1% organic content. Solids must be pumpable, but can be up to about 35-40% of the mass. | |
| 1100 | Acidic | These solutions have a pH of less than 2 and a RCRA code of D002. They are typically general waste waters, plating line solutions, and electropolishing activities. The acid may be of any type. Most common acid types are nitric, sulfuric, and hydrochloric. | The acid nature of the wastes. |
| 1200 | Basic | These are solutions with a pH of greater than 12.5 and a RCRA code of D002. They are generated from a variety of activities. They can result from neutralization of acidic streams. | The basic nature of the wastes. |
| 1300 | Neutral | Solutions with a pH from 2 to 12.5. One source is condensate from evaporators. | |
| 1400 | Cyanide | Any stream that contains cyanide as a significant component. Solutions will generally be basic. | Cyanide gas generation. |
| 1900 | Uncategorized Aqueous Liquids | Aqueous liquid streams for which insufficient information is available to further characterize or streams that contain mixtures of waste categories. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--|---|------------------------------|
| 2000 | Organic Liquids | Liquid streams that are either essentially a pure organic stream or those streams that contain both aqueous and organic materials. Solids must be pumpable but can be up to 35-40% of the mass. | |
| 2100 | Aqueous/Organic Liquids | Liquid streams that contain mixtures of aqueous and organic liquids with contents of 1% or more of organic liquids but less than about 99% organic. | |
| 2110 | Aqueous/ Halogenated Organic Liquids | Liquid streams that contain mixtures of aqueous and halogenated organic liquids with contents of 1% or more of organic liquids but less than about 99% organic. | |
| 2120 | Aqueous/Non- Halogenated Organic Liquids | Liquid streams that contain mixtures of aqueous and non-halogenated organic liquids with contents of 1% or more of organic liquids but less than about 99% organic. | |
| 2190 | Uncategorized Aqueous/Organic Liquids | Liquid streams that contain mixtures of aqueous and uncategorized organic liquids. | |
| 2200 | Pure Organic Liquids | Liquids with only organic materials. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|------------------------------------|--|---|
| 2210 | Halogenated Organic | Nearly pure organic liquids containing more than trace (~1000 ppm) levels of halogens (e.g., F, Cl, Br, etc.). Contaminated freon is one specific stream. High level PCB wastes are also a potential stream. | Treatment must provide for PCB destruction. |
| 2220 | Non-Halogenated Organic | Nearly pure organic liquids free of more than trace (~1000 ppm) levels of halogens (e.g., F, Cl, Br, etc.). Oils, hexon, and methanol are typical streams. | |
| 2290 | Uncategorized Pure Organic Liquids | Organic liquids that can not be categorized as halogenated or non-halogenated. | |
| 2900 | Uncategorized Organic Liquids | Organic liquids for which insufficient information is available to determine the aqueous content and whether the stream is more than trace halogenated. | |
| 3000 | Solid Process Resides | These materials are typically residues from process operations such as waste water clean up or process operations, or are listed wastes. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|-----------------------------|---|--|
| 3100 | Inorganic Matrix Solids | These are materials that have an inorganic matrix or content such that they would have a high residue from incineration. They may contain both hazardous organics and metals. They may also contain either aqueous or organic interstitial liquids. | Destruction or removal of organics in solids and immobilization of toxic metals. |
| 3110 | Particulate Inorganic Media | Fine particulate wastes. Typical sources are ash from incinerators, dusts, sand blasting residue, vermiculite, and ion exchange media. | Dispersibility of the wastes. |
| 3111 | Ash | Materials generated from the incineration of radioactive wastes. It includes both bottom ash, fly ash, and other solid residues. | May contain residual carbon, heavy metals in a very dispersible form. |
| 3112 | Sand Blasting Media | Particulate material (generally course sand or glass) used to decontaminate or clean radioactively contaminated materials. | |
| 3113 | Absorbed Aqueous Liquids | Inorganic materials such as clay, vermiculite, or diatomaceous earth added to absorb aqueous liquids or placed in drums to absorb liquids if internal containers leak. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|------------------------------------|---|---|
| 3114 | Absorbed Organic Liquids | Inorganic materials such as clay, vermiculite, or diatomaceous earth used to absorb organic liquids or placed in drum to absorb liquids if internal containers leak. | |
| 3115 | Ion Exchange Media | Inorganic materials that have been used to remove ions from liquid streams. | |
| 3119 | Uncategorized Particulates | Particulate materials that can not be assigned to any of the above categories or is a mixture of such materials including absorbed liquids or materials that have some trap materials. | |
| 3120 | Sludges, Filter Cakes and Residues | These materials are generally from waste water cleanup or from settling ponds. They may contain organic materials in limited quantities from laundry or other sources. Heavy metals are present in some sludges. Equipment, filters, and other materials have occasionally been included with some drums. | Identification of foreign items in drums. |
| 3121 | Low Organic Content Sludges | Sludges with less than 1% of hazardous organic materials. | |
| 3122 | High Organic Content Sludges | Sludges with greater than 1% hazardous organic material. The organics can be either halogenated or non-halogenated materials. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|-------------------------------------|---|--|
| 3123 | Sludges with Cyanide | Sludges such as those in 211 that contain cyanide as more than a trace concentration. | Destruction of cyanide. |
| 3129 | Uncategorized Sludges | Sludges with unknown levels of organic content or sludges with other components such as paper filter media. | |
| 3130 | Paint Chips & Residues | New or removed paint. The paint may have some liquids content either as original paint or as a paint stripper; it may also be paint chips. Painting equipment would be a debris waste. | It will likely have a high residue after incineration. |
| 3140 | Cemented Sludges, Ashes, and Solids | Sludges or solids that contain cement or other solidifying agents either as a water absorber or that are mixed with cement or solidifying agents to produce a homogenous solid waste, but do not yet meet disposal requirements. Final treated cemented materials are a 9100 waste. | Residual or free water in waste. |
| 3150 | Salt Wastes | Salt Wastes | |
| 3151 | Chloride and Sulfate Salts | Evaporated or process salt that may contain more than trace concentrations of sulfate and chlorides or other halogens. | The corrosion potential for process equipment and the limited solubility in glass waste forms. |
| 3152 | Nitrate Salts | Evaporated or process salts that are predominantly nitrate salts. | Destruction of NOx. |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|-----------------------------------|---|--|
| 3159 | Uncategorized Salt Wastes | Salt wastes with unknown salts, a combination of 3151 and 312 salts, or are non chloride, sulfate, or nitrate. | |
| 3190 | Uncategorized Inorganic Solids | Inorganic solid waste streams that can not be further characterized or is a mixture of 211 to 217 materials. | Insufficient characterization information. |
| 3200 | Organic Matrix Solids | These are materials that have an organic matrix or base structure. They may have some liquid present, but will not leave a large residue when incinerated. | Destruction of the organic content. |
| 3210 | Particulate Organic Media | Particulate organic media including spent organic resins, spent carbon filters used in waste water cleanup, or particulate organic material used to absorb organic aqueous liquids. | Destruction of the organic content. Some resins are not compatible with cement or grout systems. |
| 3211 | Activated Carbon | Particulate activated carbon that has often been used for removal of organic materials from off-gas streams or liquid streams. | |
| 3212 | Organic Resins | Organic based resins that have been used in waste water treatment or other applications. | |
| 3213 | Absorbed Liquids (Organic Matrix) | Liquids absorbed on a particulate organic matrix such as cellulose or pulverized corn cobs. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--------------------------------------|--|---------------------------------------|
| 3219 | Uncategorized Organic Particulate | Particulate materials or media that can not be categorized above or that is a mixture of the above materials. | |
| 3220 | Organic Sludges | Organic based sludges of various types. | |
| 3221 | Sewage Sludges | Sludges generated in treating waste water from animals or people. | Streams may contain PCB's |
| 3222 | HOC Organic Sludges | Halogenated organic containing materials that can not be poured from a drum at room temperature for treatment as an organic liquid. | Transfer of contents from containers. |
| 3229 | Uncategorized Organic Sludges | Organic sludges which could not be categorized as a 3221, 322, or 3223 because of it being a combination of materials or because of its unknown chemistry. | |
| 3230 | Organic Chemicals | Drums of unused organic chemicals. | |
| 3290 | Uncategorized Organic Solids | Materials that can not be included into any of the above organic solid categories or is a combination of categories. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--|---|---|
| 4000 | Soils | Soils contaminated with hazardous materials and radioactivity. Codes 4100 to 4300 may contain up to 5% debris materials. Codes 4400 to 4600 may have 5 to 50% debris. Streams with more than 50% debris would be a debris waste (5000). This group is subdivided as noted below however it may be modified when specific EPA regulations are promulgated. | |
| 4100 | Organic Contaminated Soils | Soils contaminated with hazardous organics from activities such as spills, drains, and waste water treatment. | Removal or destruction of the hazardous organic materials. |
| 4200 | Inorganic Contaminated Soils | Soils contaminated with hazardous inorganics from activities such as spills, drains, and waste water treatment. | Removal or immobilization of the hazardous inorganic materials. |
| 4300 | Organic and Inorganic Contaminated Soils | Soils contaminated with both hazardous organic and inorganic materials from activities such as spills, drains, and waste water treatment. | |
| 4400 | Soils with Organic Debris | Soils contaminated with more than 5% hazardous organics debris from activities such as spills, drains, D&D, and previous waste disposal actions. These streams may also contain organic or inorganic hazardous materials. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--|---|------------------------------|
| 4500 | Soils with Inorganic Debris | Soils contaminated with more than 5% hazardous inorganics debris from activities such as spills, drains, D&D, and previous waste disposal actions. These streams may also contain organic or inorganic hazardous materials. | |
| 4600 | Soils with Both Inorganic and Organic Debris | Soils contaminated with more than 5% of either or both hazardous organic and inorganic debris from activities such as spills, drains, D&D, and previous waste disposal actions. These streams may also contain organic or inorganic hazardous materials. | |
| 4900 | Uncategorized Soils | Soils contaminated with unknown materials or with a combination of 4000 series waste categories. | |
| 5000 | Debris Wastes | Wastes that meet the EPA criteria for Debris. Debris materials are divided into four groups as either metal, inorganic non-metal, combustible, or mixtures of materials (heterogeneous). If the wastes is dominate in one type of material it should be classed as that material otherwise it is heterogeneous. | |
| 5100 | Metal Debris | Streams that generally have a metal content greater than about 95%. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|---|--|--------------------------|
| 5110 | Metallic Equipment, Components, and Scrap. | General metallic items that have been used in process operations or maintenance. Typical items include piping, pumps, metal filters, traps, wire, and fixtures. It is anticipated that the metal content is greater than about 95%. | |
| 5120 | Uranium Chips and Turnings | Uranium metal components or particulate materials with other types of materials to reduce uranium reactivity. | |
| 5130 | Lead Containing Components and Materials | Metallic lead containing materials, including gloveboxes, lead wool, lead base solder materials, or lead components used in radioactive processes. It also includes lead shapes mixed with other materials or lead encapsulated in other metals. This stream does not include lead acid batteries which are a 7410 waste or pure lead shapes used as shielding (7200). | |
| 5140 | Cadmium Containing Components and Materials | Components formed from cadmium or contain a significant fraction of cadmium. This stream does not include cadmium batteries which are a 7420 waste. | |
| 5190 | Uncategorized Metal Debris | Metallic components that can not be classed as a specific 5100 waste stream because of lack of knowledge or because it contains multiple categories. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|--|---|---|
| 5200 | Inorganic Non-Metal Debris | Debris streams composed of about 95% inorganic nonmetal materials. | |
| 5210 | Concrete | Concrete materials removed from buildings or roadways. | |
| 5220 | Glass | Items composed primarily of glass. It may be process equipment, laboratory equipment, window materials, vessels, bottles, or light bulbs if metallic components are removed. This stream may include leaded glass. The glass may contain small amounts of organic, metal, or other inorganic materials. | Glass from florescent bulbs may contain some Hg or PCB contamination. |
| 5230 | Ceramic Crucibles and Bricks | Crystalline or glass materials used as crucibles or refractories. | |
| 5240 | Rocks | Rocks and gravel that have a particle size greater than 60 mm. | |
| 5290 | Uncategorized Inorganic Non-Metal Debris | Non-Metallic items that can not be classed as a specific 5200 waste stream because of lack of knowledge or because it contents multiple categories. | |
| 5300 | Combustible Debris | Specific waste streams that contain more than 95% combustible materials. | |
| 5310 | Plastic and Rubber | Plastic and rubber such as sheeting, containers, gloves, gaskets, and components of benelex or plexiglass. | The halogen content during combustion. |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|--|--|---|
| 5311 | Leaded Gloves and Aprons | Rubber materials that contain a high fraction of lead and lead compounds. | |
| 5312 | Halogenated Plastics | Plastics, such as PVC, that contain halogens as part of their chemical structure. | The halogen content during combustion. |
| 5313 | Non-Halogenated Plastics | Plastics, such as polyethylene, that are free of halogenated materials. | |
| 5319 | Uncategorized Plastic and Rubber Materials | Materials that can not be separated into any of the above categories or that contain a mixture of such materials types. | |
| 5320 | Wood | Wood items such as structural timbers, boxes, or pallets. | Residence time required for complete combustion. |
| 5330 | Paper and Cloth | Paper and cloth items such as protective clothing, and items used to wipe up contamination or absorb liquids. Wipes may contain some absorbed organic and aqueous liquids. | Nitrate contain rags and rapid combustion rates. |
| 5340 | Graphite | Crucibles or components of graphite or carbon. | Sufficient residence time to totally oxidized. |
| 5350 | Animal Carcasses | Dead animals or parts of animals. Most animals will have been used in testing and may contain chemical agents such as lime or formaldehyde to stabilize them. | Biological hazards may be present. May have a high residue from stabilizing agents. |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|-----------------------------------|--|--|
| 5390 | Uncategorized Combustible Debris | Uncharacterized combustible or mixtures of combustible materials in the above categories. | |
| 5400 | Heterogeneous Debris | Mixtures of metals, non-metals, combustibles, soils, and process residues that can be classified as debris. | Sorting of these materials may be necessary for treatment. |
| 5410 | Filters | HEPA filters, and other process filters. Filters are contaminated with fine particulate. HEPA filters may be either wood or metal frame. Particulate filter media would be process residues. | Mixtures of materials that can not be easily separated. |
| 5420 | Predominantly Metal | Debris materials that contain more than about 50% but less than 95% metals with other debris. | |
| 5430 | Predominantly Inorganic Non-Metal | Debris materials that contain more than about 50% but less than about 95% inorganic non-metals with other debris. | |
| 5440 | Predominantly Combustible Debris | Debris materials that contain more than about 50% but less than 95% combustible materials with other non-combustible debris. | |
| 5450 | Asphalt | Asphalt materials from roadways or other sources and contain both tar and gravel. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--|--|---|
| 5490 | Uncategorized Heterogeneous Debris | Heterogeneous debris that can not be further characterized. or does not contain a dominance of metals, non-metals, or combustible debris. | |
| 6000 | Special Wastes | These waste streams will require specific treatment methods that are not expected to be common with other waste types. It also includes wastes streams for which treatment capacity may not be initially established as part of the Mixed Waste Treatment Project. | |
| 6100 | Lab Packs | This category includes more than just the conventionally identified lab packs. It includes all wastes that contain liquids in container with surrounding packing materials such as vermiculite. Lab packs contain mixtures of chemicals in drums. Chemicals are mostly solid but can contain liquids in bottles. | Sorting of materials. Wide variety of chemicals. Ability to characterize. |
| 6110 | Organic Lab Packs | Lab packed materials that contain organic liquids. | |
| 6120 | Aqueous Lab Packs | Lab packed materials that contain aqueous liquids. | |
| 6130 | Solid Lab Packs | Lab packed materials that do not contain liquid materials. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|--|--|--|
| 6140 | Scintillation Cocktails | Solutions used for scintillation counting. Solutions are most often in the original glass or plastic analysis bottles. | Removal of the solutions from the bottles and containers and subsequent treatment of the containers. |
| 6190 | Uncategorized Lab Packs | Lab packed materials that could be classified as more than one waste code from 6100 series or that have insufficient information available to further classify. | |
| 6200 | Reactive Metals | Reactive metals are typically sodium metal or alkali metal alloys, but can also be particulate fines of aluminum, uranium, zirconium, or other pyrophoric materials and may be mixed with stabilizing materials. | Reactive nature of materials and potential for hydrogen gas generation. |
| 6210 | Bulk Reactive Metals | Nearly pure reactive metals in containers. They may have various types of impurities, but the bulk of materials is reactive metal and can be treated in a bulk processing system. | Same as above. |
| 6220 | Components Contaminated with Reactive Metals | Piping, pumps, and other materials that have reactive metal contamination. The bulk of the material is not reactive metals, but the reactive metals require treatment before disposal. | Some of the reactive materials may not be readily accessible to chemical reaction. |
| 6290 | Uncategorized Reactive Metals | Non alkali metal reactives or mixtures of reactives. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|-----------------------------------|---|---|
| 6300 | Explosive | Any material that may explode during normal or extreme handling. | Serious handling problem and worker safety problem. |
| 6310 | Nitrated Rags & Filters | Rags that have absorbed nitric acid and then been left in storage. | During storage nitrocellulose may form which is explosive. |
| 6390 | Uncategorized Explosives | Materials that may be explosive that are not nitrated rags or filters. | |
| 6400 | Compressed Gases and Aerosol Cans | Aerosol cans and gas cylinders with gases of any composition. It is expected that the containers will be pressurized. Non-pressurized containers would be a debris waste. | Rapid gas release from container failure during processing. |
| 6500 | Hanford Double Shell Tanks | Wastes in Hanford Double Shell Tanks. | |
| 7000 | Inherently Hazardous Materials | Materials that are composed of inherently hazardous materials. | |
| 7100 | Liquid Mercury | Liquid mercury pourable from containers. Will include drums of waste that contain elemental mercury within a container. | Separation of Hg from other waste materials. |
| 7200 | Lead Shapes | Bulk Lead materials that are separable from other wastes in the form of bricks, sheets, or unique components. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|--------------------|---------------------------|--|---|
| 7210 | Non-Activated Lead | Lead bricks, shipping casks, or shielding materials. The lead should only be surface contaminated. | Recycle of the materials. |
| 7220 | Activated Lead | Lead activated from it's use in radiations fields such as in reactors or accelerators where it can be activated. | Radiation levels are generally sufficient to require shielding and remote handling. |
| 7230 | Lead with Other Materials | Lead bricks or shapes in drums of waste with other inorganic or organic type materials. The lead should be a shape that can be decontaminated and can be easily separated from the other wastes. | |
| 7300 | Beryllium Wastes | Beryllium metal chips or materials contaminated with more than trace levels of beryllium. It should be a P014 RCRA waste. | BDAT specifies recycle. |
| 7400 | Batteries | These are generally lead, and cadmium-type batteries, but will include other types as well. | The BDAT specifies recycle of the metals. |
| 7410 | Lead Acid | Lead Acid batteries. | |
| 7420 | Cadmium | Cadmium type batteries. | |
| 7490 | Uncategorized Batteries | Other types of batteries or unclassifiable batteries. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--------------------------|--|---|
| 8000 | Unknown | Waste materials that can not reasonable be classified into one of the other categories based on available information or streams for which the available information is conflicting. | Inadequate information to allow classification. |
| 8100 | Unknown Liquids | Liquid wastes that can not be categorized as aqueous or organic liquids. | |
| 8200 | Unknown Solids | Solid Materials that can be categorized. | |
| 8900 | Uncategorized Unknown | Materials that can not be categorized. | |
| 9000 | Treated Wastes | Wastes that have been treated to meet the Land Disposal Restriction. | |
| 9100 | Cement | Cement type waste forms including grouts and cements of various types. | |
| 9200 | Glass | Wastes that have been converted to a vitreous type waste form. | |
| 9300 | Metal | Metal waste forms that have been consolidated or decontaminated and are ready for disposal or recycle. | |
| 9400 | Polymer | Organic type waste forms including polyethylene, bitumen, resins, and other organic binders. | |

| <u>Code Number</u> | <u>Title</u> | <u>Description</u> | <u>Treatment Concern</u> |
|------------------------|--------------|---|------------------------------|
| 9900 | Other Forms | Final waste forms that are not classed as either a cement, glass, metal, or polymer. It would include amalgamated mercury and microencapsulated lead. | |

APPENDIX B

Department of Energy's Low-Level Mixed Waste Inventory¹

1 Information contained in Appendix B was obtained from the following document: Ross, W., Warner, C., et al, *Locations, Volumes, and Characteristics of DOE's Mixed Low-Level Wastes*, Waste Management '92, March 1-5, 1992, Tucson, Arizona.

Appendix B. Department of Energy's MLLW

| WASTE FORM | EPA WASTE CODES | DESCRIPTIONS | QUANTITIES STORED | QUANTITIES GENERATED YEARLY | SUSPECTED CONTAMINANTS | BASELINE TREATMENT OPTIONS (MWTF) |
|------------------------------|--|--|--------------------|-----------------------------|---|--|
| Aqueous Liquids | D001, D002, D003, D004, D006, D007, D008, D009, D011, D018, F001, F002, F003, F005, F006, F009 | Aqueous solutions with less than 1% organic content. Solids must be pumpable (30-40% of the mass). Waste stream may contain multiple characteristics. Stream may contain some of the following: As, Cd, Cr, Pb, Hg, Ag, & organics | 1,975 cubic meters | 249 cubic meters | ignitable characteristic; reactive characteristic; toxic characteristic; As, Cd, Cr, Pb, Hg, Ag, benzene; degreasing solvents; solvents; electroplating sludges; cyanides | Treatment option will be determined after better waste characterization of this waste stream |
| Corrosive -- Acids | D002, D004, D006, D007, D008, D009, D011, F005, U044, U122, U123, U134, U151, U196 | Principle sources of these wastes are general waste water cleanup, plating line solutions, and electroplating activities. Streams may contain some of the following: Cr, Pb, Hg, Ag, & organics | 102 cubic meters | 3,726 cubic meters | corrosive characteristic; As, Cd, Cr, Pb, Hg, Ag, sodium azide; chloroform, formaldehyde, formic acid, hydrofluoric acid, pyridine | To Be Determined by the Physical/Chemical Treatment TSG |
| Corrosive -- Bases | D002, D004, D006, D007, D008 | These are solutions with a high pH generated from a variety of activities. Streams may contain some of the following: As, Cd, Cr, & Pb. | 50 cubic meters | 22 cubic meters | corrosive characteristic; As, Cd, Cr, Pb | To Be Determined by the Physical/Chemical Treatment TSG |
| Reactive -- Cyanides | D002, D004, D006, F007, F009 | Any waste stream that contains cyanide as a significant component. Solutions will generally be basic. Streams may contain some of the following: As, Cd, & Cr | 2 cubic meters | 0.3 cubic meters | corrosive characteristic; As, Cd, electroplating solutions; cyanides | To Be Determined by the Physical/Chemical Treatment TSG |
| Toxic Metals | D002, D006, D007, D008, D009, D010, F001, F003, F005, F006 | Streams containing toxic metals. Streams contain some of the following: cadmium, chromium, lead, mercury, selenium. Contains trace hazardous organics. | 3,991 cubic meters | 197 cubic meters | corrosive characteristic; Cd, Cr, Pb, Hg, Se; degreasing solvent; solvents; electroplating sludges | To Be Determined by the Physical/Chemical Treatment TSG |
| Trace Organics | D001, D008, D039, F001, F002, F003, F005 | Streams with organics at levels of less than 1%. Some streams also contain some lead | 64 cubic meters | 69 cubic meters | ignitable characteristic; Pb; tetrachloroethylene; degreasing solvents; solvents | Wet Air Oxidation Process -- Aqueous/Organic Treatment Unit |
| Mercury Containing Solutions | D008, D009 | Any aqueous liquids that are contaminated with mercury | 102 cubic meters | 0.02 cubic meters | Pb, Hg | To Be Determined by the Physical/Chemical Treatment TSG |
| Non-Halogenated Organics | D001, D002, D006, D007, D008, D009, D010, D018, D019, D028, D029, D039, D040, F001, F002, F003, F005, U019, U154, U120, U139, U161 | Solvents free of F, Cl, Br, etc., e.g. oils, hexon, methanol. Streams may contain Cd, Cr, Pb, Hg, & Se as trace contamination | 239 cubic meters | 226 cubic meters | ignitable characteristic; corrosive characteristic; Cd, Cr, Pb, Hg, Se, benzene, carbon tetrachloride, 1,2-dichloroethane, 1,1-dichloroethylene, tetrachloroethylene, trichloroethylene, degreasing solvents; solvents; methanol; toluene, xylene, Benzene, etc | Car Bottom Furnace -- Heavy Organics Thermal Treatment |

Appendix B. Department of Energy's MLLW

| WASTE FORM | EPA WASTE CODES | DESCRIPTIONS | QUANTITIES STORED | QUANTITIES GENERATED YEARLY | SUSPECTED CONTAMINANTS | BASILINE TREATMENT OPTIONS (MWTE) |
|----------------------------------|--|---|---------------------|-----------------------------|---|--|
| Halogenated Organics | D001, D002, D003, D006, D008, D009, D011, D018, D019, D021, D022, D028, D029, D039, D040, F001, F002, F003, F004, F005, U044, U226, U228, U080 | Solvents containing F, Cl, Br, etc. It includes all PCB contaminated organic streams. Streams may contain Cd, Cr, Pb, Hg, & Ag as trace contamination. | 795 cubic meters | 112 cubic meters | ignitable characteristic; corrosive characteristic; reactive characteristic; Cd; Cr; Pb; Hg; Ag; benzene; carbon tetrachloride; chlorobenzene; chloroform; 1,1-dichloroethane; 1,1-dichloroethylene; tetrachloroethylene; trichloroethylene; solvents, etc. | Car Bottom Furnace – Heavy Organics Thermal Treatment |
| Scintillation Cocktails | D001, D002, D006, D007, D008, D022, F001, F002, F003, F004, F005 | Solvents used for scintillation counting. Solutions are most often in the original glass or plastic analysis bottles. Streams may contain Cd, Cr, & Pb as trace contamination. | 146 cubic meters | 27 cubic meters | ignitable characteristic; corrosive characteristic; Cd; Cr; Pb; chloroform; degreasing solvents, solvents. | Car Bottom Furnace – Heavy Organics Thermal Treatment |
| Mercury Containing Organics | D001, D009, D010, F002 | Organic liquids that are contaminated with mercury. | 4 cubic meters | 1 cubic meters | ignitable characteristic; Hg; Se; solvents. | * PROBLEMATIC WASTE STREAM |
| Sludges, Filter Cakes & Residues | D001, D002, D003, D006, D007, D008, D009, D011, D019, D029, D039, D040, F001, F002, F003, F004, F005, F006, F007, F008, F009, F029, F030, F098, P106, P120, U123 | These materials are generally from waste water cleanup or from settling ponds. They may contain organics in limited quantities from laundry or other sources. Heavy metals are present in some sludges. Cemented sludges are not included in this category. | 11,812 cubic meters | 683 cubic meters | ignitable, corrosive, & reactive characteristics; Cd; Cr; Pb; Hg; Ag; carbon tetrachloride; 1,1-dichloroethylene; tetrachloroethylene; trichloroethylene; degreasing solvents; solvents; electroplating sludges; cyanides; vanadium oxide; formic acid | Rotary Kiln Incinerator – High Residue Thermal Treatment |
| Absorbed Liquids | D001, D002, D003 | Aqueous or organic liquids absorbed onto a solid such as vermiculite or clay. | 321 cubic meters | 0.06 cubic meters | ignitable, corrosive, & reactive characteristics | To Be Separated by Front End Waste Handling TSG. Organics will be treated in the High Residue Rotary Kiln Incinerator. |
| Salt Cake | D003 | Evaporated salt solutions either nitrate and chloride that may contain a high residual water content. Streams may contain Ba as trace contamination. | 128 cubic meters | 0 cubic meters | inertium | To Be Determined by the Physical/Chemical Treatment TSG |
| Inorganic Dry Solids | D001, D002, D004, D005, D008, D010, F001, F002, F003, F004 | Inorganic solids without notable free liquids. Most wastes will contain some chemically combined water or absorbed water. Liquid fractions will generally be less than 5% of the mass of the material. | 1,408 cubic meters | 0.07 cubic meters | ignitable & corrosive characteristic; As; Ba; Pb; Se; degreasing solvents, solvent contamination. | Treatment option will be determined after better waste characterization of this waste stream |
| Processing Salts | D001, D004, D005, D006, D007, D008, D010, D016, D028, D035, F001, F003 | Salts that have been used in processing. They are mostly F, Cl, NO3 based salts. Streams may contain As, Ba, Cd, Cr, Pb, Se, & organics as trace contamination. | 133 cubic meters | 4 cubic meters | ignitable characteristic; As; Ba, Cd; Cr; Pb; Se; 1,2-dichloroethane; methyl ethyl ketone; degreasing solvents; solvent contamination; electroplating sludge | To Be Determined by the Physical/Chemical Treatment TSG |

Appendix B. Department of Energy's MLLW

| WASTE FORM | EPA WASTE CODES | DESCRIPTIONS | QUANTITIES STORED | QUANTITIES GENERATED YEARLY | SUSPECTED CONTAMINANTS | BASELINE TREATMENT OPTIONS (MAYTP) |
|-----------------------------------|--|---|--------------------|-----------------------------|--|---|
| Cemented Sludges | D006, D007, D008, F001, F002, F003, F006 | Sludges that contain cement either as a water absorber or that are mixed with cement to produce a homogeneous solid waste. Streams may contain Cd, Cr, Pb, & organics as trace contamination. | 9,236 cubic meters | 230 cubic meters | Cd, Cr, Pb, degreasing solvents, solvent contamination, electroplating sludges | PROBLEMATIC WASTE STREAM |
| Ash, Dust & Particulate | D004, D006, D007, D008, D010, D011, F001, F002, F003, F005, F006 | Fine particulate wastes. Typical sources are ash from incinerators, dusts, and paint chips. Streams may contain As, Cd, Cr, Pb, Ag, Se, & organics as trace contamination. | 1,017 cubic meters | 36 cubic meters | As, Cd, Cr, Pb, Se, Ag, degreasing solvents, solvent contamination, electroplating sludge | Joule Melter - Thermal Unit for Non-Metals |
| Soils | D001 thru D0043, F001, F002, F003, F005 | Contaminated soils from spills, leaks, cleanups, and waste burial. Streams may contain As, Ba, Cd, Cr, Pb, Hg, Ag, Se, & organics as trace contamination. | 364 cubic meters | 38 cubic meters | ignitable, corrosive, reactive & toxicity characteristics; degreasing solvents, solvent contamination | Rotary Kiln Incinerator - High Residue Thermal Treatment |
| Glass | D001, D002, D005, D006, D008, D009, F001, F002, U056 | Items composed primarily of glass. It may be process equipment, laboratory equipment, window materials, vessels, bottles, light bulbs, or glass boards or forms used within process equipment or for abrasion of surfaces. | 577 cubic meters | 8 cubic meters | ignitable & corrosive characteristic; Ba, Cd, Pb, Hg, degreasing solvents, solvent contamination, benzene | Joule Melter - Non-Metallic Thermal Treatment |
| Ceramic Crucibles, Bricks & Media | D004, D007, D008, D009, D011, F001, F002, F005 | Oxide materials generally used as crucibles or refractories. They may also be beads or shapes used for catalysts, reactor beds, or for milling or grinding. Streams may contain As, Cr, Pb, Hg, Ag, & organics as trace contamination. | 873 cubic meters | 1 cubic meters | As, Cr, Pb, Hg, Ag, degreasing solvents, solvent contamination | Joule Melter - Non-Metallic Thermal Treatment |
| Metal Wastes | D002, D006, D008, D009, F001, F002 | Inorganic solid materials generated in routine process operations and classed as metals. This category will include wastes that are mixed ferrous and nonferrous metals or defined metals. Streams may contain Cd, Cr, Pb, Hg, & organic trace contamination. | 9,976 cubic meters | 36 cubic meters | corrosive characteristic; Cd, Pb, Hg, degreasing solvent, solvent contamination | This waste stream will need to be sorted by the Front End Waste Handling TSG. Metals will be treated in the Metal Melt/Slag treatment process. BASELINE TECHNOLOGY TO BE DETERMINED |
| Non-Ferrous Metals | D006, D008 | Non-ferrous metals are the principle component such as aluminum, copper, and cadmium. May have trace Pb contamination. | 5 cubic meters | 1 cubic meters | Cd, Pb | Non-Ferrous Metals Melt/Slag - BASELINE TECHNOLOGY TO BE DETERMINED |
| Mercury | D009, U151 | Liquid mercury pourable from containers. | 3 cubic meters | 0.7 cubic meters | Hg | Car Bottom Furnace - Roaster Bakeout |
| Lead Shapes | D002, D003, D007, D008 | Lead bricks, shipping casks, or shielding materials. The lead should only be surface contaminated. May have some Cr or organic contamination. | 755 cubic meters | 108 cubic meters | corrosive & reactive characteristic; Cr, Pb, non-halogenated solvents | Decontamination Technologies To Be Determined by the Physical/Chemical Treatment TSG. |
| Activated Lead | D008 | Lead activated from its use in radiation fields such as in reactors or accelerators. | 1 cubic meter | 0.1 cubic meter | Pb | To Be Determined by the Final Waste Form TSG. |
| Organic Wet Solids | D001, D002, D003, D004, D005, D006, D007, D008, D010, D011, F001, F002, F003, F005, U002, U080, U161 | These are organic solids that have significant content of liquids generally combined with the waste matrix. Streams may contain As, Ba, Cd, Cr, Pb, Ag, & Se as trace contamination. | 33 cubic meters | 10 cubic meters | ignitable, corrosive & reactive characteristics; As, Ba, Cd, Cr, Pb, Se, Ag, degreasing solvent, solvent contamination | Rotary Kiln Incinerator - High Residue Thermal Treatment |

Appendix B. Department of Energy's MLLW

| WASTE FORM | EPA WASTE CODES | DESCRIPTIONS | QUANTITIES STORED | QUANTITIES GENERATED YEARLY | SUSPECTED CONTAMINANTS | BASILINE TREATMENT OPTIONS (MWTE) |
|--------------------------|---|---|--------------------|-----------------------------|--|---|
| Absorbed Combustibles | D001, D002, D003, D006, D007, D008, D010, D012, F001, F002, F003, F005 | Rags or paper used to wipe up spills. Organic liquid spill cleanup materials. Streams may contain Ba, Cr, Pb, Se, & organics as trace contaminants. | 113 cubic meters | 3; cubic meters | ignitable & corrosive characteristics, Ba, Cd, Cr, Pb, Se, Ag, degreasing solvents, solvent contamination | Refractory Lined Controlled Air Incinerator -- Low Residue and/or Rotary Kiln Incinerator -- High Residue |
| Paint & Residues | X001, X003, D003, X007, D008, D009, F002, F003, F005 | New or removed paint. The paint may be liquid or have a liquid content either as original paint or a paint stripper. It may also only be paint chips. | 31 cubic meters | 10 cubic meters | ignitable & reactive characteristics, Ba, Cr, Pb, Hg, halogenated solvents, non-halogenated solvents. | Car Bottom Furnace -- Heavy Organics Thermal Treatment |
| Resins | X001, F001, F002 | Spent organic resins including spent carbon filters used in waste water cleanup. | 10 cubic meters | 0 cubic meters | ignitable characteristics, halogenated solvents | Joule Melter -- Non-Metallic Thermal Treatment |
| Animal Carcasses | D004, D005, D009, D010, D011 | Dead animals or parts of animals. Most animals will have been used in testing and may contain chemical agents such as lime to stabilize them. | 0 cubic meters | 0.02 cubic meters | As, Ba, Hg, Se, Ag | Rotary Kiln Incinerator -- High Residue Thermal Treatment |
| Organic Dry Solids | D001, D002, D003, D006, D008, D009, D011, D018, D029, D039, D040, F001, F002, F003, F005, F007 | These are organic solids that do not have significant content of liquids. Streams that may contain a variety of solid organic wastes such as mixtures of wood, plastic, and paper. Waste stream may contain Ba, Cd, Pb, Ag, & organics as trace contaminants. | 7,731 cubic meters | 115 cubic meters | ignitable & corrosive characteristics, Ba, Cd, Pb, Hg, Ag, benzene, degreasing solvents, halogenated solvents, non-halogenated solvents, electropolishing solutions | Refractory Lined Controlled Air Incinerator -- Low Residue |
| Plastic & Rubber | D001, D002, D008, F001, F002 | Plastic sheeting or components such as benzolex or plexiglass. It also includes non-leaded glovebox gloves. Waste stream may contain traces of Pb and organics. | 75 cubic meters | 14 cubic meters | ignitable & corrosive characteristics, degreasing solvents, halogenated solvents | Refractory Lined Controlled Air Incinerator -- Low Residue |
| Paper, Cloth & Rags | D001 thru D009, D011-D043, F001, F002, F003, F004, F005, F012, P015, P048, P113, P120, U002, U032, U052, U080, U133, U134, U144, U151, U154, U161, U209, U211, U220, U226, U239 | Materials used in spill and process cleanups. They may contain some absorbed organic and aqueous liquids. | 3,648 cubic meters | 3 cubic meters | ignitable, corrosive, reactive & toxic characteristics, degreasing solvents, halogenated solvents, non-halogenated solvents, beryllium, thallium oxide, P & U series contaminants. | Refractory Lined Controlled Air Incinerator -- Low Residue |
| Heavy Sludges & Asphalts | D001, D007, D008, D009, D010, D035, D039, D040, F001, F002, F003, F005 | Heavy sludges are organic materials that can not be poured from a drum at room temperature. Also contains asphalt roadways or walkways that have become contaminated with radioactivity. Asphalt contains a high fraction of rock & organic binders. | 143 cubic meters | 1 cubic meter | ignitable characteristic, Cr, Pb, Hg, Se, degreasing solvents, halogenated solvents, non-halogenated solvents | Car Bottom Furnace -- Heavy Organics Thermal Treatment |
| Graphite and Carbon | F006 | Crucibles or components of graphite or carbon. | 0 cubic meters | 217 cubic meters | electroplating sludges | Refractory Lined Controlled Air Incinerator -- Low Residue |
| Heterogeneous Wastes | D001, D007, D008, D011, F001, F002, F003 | Solid materials that may contain a mixture (either as manufactured or in packaging) of organic, inorganic solid, and/or metallic materials. Streams may contain Cd, Cr, Hg, Se, Ag, and organics as trace contaminants. | 27 cubic meters | 6 cubic meters | ignitable characteristic, Cr, Pb, Ag, degreasing solvents, halogenated solvents, non-halogenated solvents | Rotary Kiln Incinerator -- High Residue Thermal Treatment |

Appendix B. Department of Energy's MLLW

| WASTE FORM | EPA WASTE CODES | DESCRIPTIONS | QUANTITIES STORED | QUANTITIES GENERATED YEARLY | SUSPECTED CONTAMINANTS | BASILINE TREATMENT OPTIONS (HWTTP) |
|---|--|---|--------------------|-----------------------------|---|--|
| Construction, Cleanup & Process Debris | D001, D002, D003, D004, D005, D006, D007, D008, D009, D010, D011, D018, D029, D039, D040, D043, F001, F002, F003, F005, F006, F009, U117, P015 | Construction debris is generated from the remodeling of radioactive facilities and could include piping, wiring, wall materials, and flooring. This waste stream should be considered a mixture of combustible organic, metal, and ceramic materials. | 7,324 cubic meters | 86 cubic meters | Ignitable, corrosive, & reactive characteristics; As, Ba, Cd, Cr, Pb, Hg, Se, Ag, degreasing solvents, halogenated solvents, non-halogenated solvents; electroplating sludge, cyanides. | This waste stream will require presorting activities – Front End Waste Handling TSG |
| Lab Pucks | D001, D002, D003, D004, D005, D006, D007, D008, D009, D010, D016, F001, F002, F003, F004, F005, F029, U117, U154, U211, U228, U094, U188 | Mixture of chemicals in drums. Chemicals are mostly solid but can contain liquids. Packing materials such as vermiculite are commonly included. Streams may contain As, Ba, Cd, Cr, Pb, Hg, Se, and organics as trace contamination. | 84 cubic meters | 26 cubic meters | Ignitable, corrosive, & reactive characteristics; degreasing solvents; halogenated solvents, non-halogenated solvents, P & U listed wastes. | Rotary Kiln Incinerator – High Residue Thermal Treatment |
| Equipment and Gloveboxes | D007, D008, F001, F002, F005 | Process equipment generally metallic, but that may contain oil, greases, or other process materials. The equipment may include some associated electrical wiring and piping. Packaged gloveboxes may contain windowns and gloves. | 3 cubic meters | 3 cubic meters | Cr, Pb, degreasing solvents; halogenated solvents; non-halogenated solvents | This waste stream will require presorting activities – Front End Waste Handling TSG Metals will be treated by the Metal Melt/Slag Technology/ Glass will be treated by the Joule Melter. |
| Filters | D001, D002, D004, D006, D007, D008, D009, D010, D011, D018, F001, F002, F003, F004, F005, F006 | HEPA filters and other process filters that may be metallic, organic or ceramic are included. Filters are contaminated with fine particulate. HEPA filters may be either wood or metal frame. Streams may contain As, Ba, Cd, Cr, Pb, Hg, Se, Ag, & organics. | 2,059 cubic meters | 61 cubic meters | Ignitable, corrosive, & reactive characteristics; As, Cd, Cr, Pb, Hg, Se, Ag, benzene; degreasing solvents; halogenated solvents; non-halogenated solvents; electroplating sludges. | This waste stream will require presorting activities – Front End Waste Handling TSG Metals to be treated by Metal Melt/Slag, Glasses in the Rotary Kiln Incinerator, Ceramics in the Joule Melter, & Wood in the Controlled Air Incinerator. |
| Reactor Equipment, Experimental Hardware & Fuel | D001, D003, D004, D006, D008, D009, F001, F002, F003 | Reactor experiments may contain listed chemicals and toxic metals. They can become highly radioactive during neutron irradiation in reaction. These materials may need special care since they contain activated metals and can not be cleaned by decons. | 0.2 cubic meters | 141 cubic meters | Ignitable & reactive characteristics; As, Cd, Pb, Hg, degreasing solvents; halogenated solvents, non-halogenated solvents. | This waste stream will require presorting activities – Front End Waste Handling TSG Metals will be treated by Metal Melt/Slag |
| Other Toxic Metal Containing Materials | D006, D008, D011, F001 | As, Ba, Cd, Cr, Se, & Ag contaminated materials. Lead may be present as a trace contamination. Contains trace hazardous organics. | 97 cubic meters | 13 cubic meters | Cd, Pb, Ag, degreasing solvents. | Rotary Kiln Incinerator – High Residue Thermal Treatment |
| Lead Containing Components & Materials | D008 | Glovebox gloves or shielding aprons containing lead or lead oxide in a rubber or plastic binder, which includes lead acid batteries. May also be lead wood, or lead base solder materials. | 7 cubic meters | 1 cubic meters | Lead | Rotary Kiln Incinerator – High Residue Thermal Treatment |
| Mercury Contaminated Materials | D001, D002, D003, D005, D006, D007, D008, D009, D010, D011, D018, U151 | Materials contaminated with Hg. Materials may be of any general type. Streams may contain Ba, Cd, Cr, Pb, Se, & Ag as trace contamination. | 109 cubic meters | 33 cubic meters | Ignitable, corrosive, & reactive characteristics; Ba, Cd, Cr, Pb, Hg, Se, Ag, benzene. | Car Bottom Furnace – Residue Baked out |

Appendix B. Department of Energy's MLLW

| WASTE FORM | EPA WASTE CODES | DESCRIPTIONS | QUANTITIES STORED | QUANTITIES GENERATED YEARLY | SUSPECTED CONTAMINANTS | BASELINE TREATMENT OPTIONS (MWTP) |
|-----------------------------|--|--|--------------------|-----------------------------|--|---|
| Tritium Waste | D001, D002, D003, D006, D007, D008, D009, D018, F001, F003 | Waste streams contaminated with tritium. They may be liquid or solids. Streams may contain Cr & Hg as trace contamination. Contains trace hazardous organics. | 37 cubic meters | 974 cubic meters | Ignitable, corrosive, & reactive characteristics; Cd, Cr, Pb, Hg, benzene, halogenated degrading solvents; non-halogenated solvents. | * PROBLEMATIC WASTE STREAM |
| Reactive Metals | D001, D003 | They are typically sodium metal or sodium metal alloys, but can also be particulate fines of aluminum, uranium, beryllium, zirconium, or other pyrophoric materials and may be mixed with reaction preventing materials. | 6 cubic meters | 0.5 cubic meters | Ignitable & reactive characteristics | * PROBLEMATIC WASTE STREAM |
| Nitrated Resins and Filters | D001 | Resins that have been used to absorb nitric acid and then been left in storage. | 12 cubic meters | 0 cubic meters | Ignitable characteristic. | Refractory Liquid Controlled Air Incinerator -- Low Residue |
| Compressed Gases | N/A | Aerosol cans and gas cylinders of any type of composition. | 8 cubic meters | 8 cubic meters | N/A | * PROBLEMATIC WASTE STREAM |
| Beryllium Waste | D001, D003, D004, D006, D007, D008, D009, D011, D018, D019, D022, D023, D024, D025, D026, P012, P015, P048, P113, P120, U002, U032, U032, U080, U133, U134, U144, U151, U154, U161, U209, U211, U220, U226, U239 | Beryllium metal chips, dusts, or materials contaminated with beryllium. Contains trace hazardous organics. | 1 cubic meters | 1 cubic meters | Ignitable & reactive characteristics; As, Cd, Cr, Pb, Hg, Ag, benzene, P & U listed wastes. | * PROBLEMATIC WASTE STREAM |
| PCB Contaminated | D001, D007, D008, D040, F001, F002, F003 | Solids and liquids contaminated with PCB. A wide range of material types. Streams may contain Cr & Pb as trace contamination. | 4,084 cubic meters | 149 cubic meters | Ignitable characteristics; Cr, Pb, degrading solvents; halogenated solvents; non-halogenated solvents. | Car Bottom Furnace -- Heavy Organics Thermal Treatment |

* MWIP has not assigned baseline technologies to these waste streams.

APPENDIX C

Thermal Treatment Technologies¹

1 Information contained in Appendix C was obtained from the *Mixed Waste Integrated Program Interim Evaluation Report on Thermal Treatment Technologies*, DOE/MWIP-2, U.S. Department of Energy, September 1992. Additional information was supplied by the Waste Destruction and Stabilization Technical Support Group.

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PREFACE

Appendix C presents additional information on each technology discussed in the text of this report. This appendix is divided into three parts: C1 - Incinerator Technologies; C2 - Miscellaneous Technologies; and C3 - Melter Technologies. Descriptions, waste applicability, advantages, disadvantages, and development needs are given for each technology. In addition, DOE laboratories involved in the technologies are listed, as well as commercial vendors. References used to assemble information for each technology are also listed.

A functional process diagram is given for each technology, which gives a general illustration of the waste sort/processing/treatment train for each technology. Additional clarifications to be considered when reviewing the functional process diagrams are discussed below. The portion on each functional process diagram listing nontreatable waste categories should not be taken as absolute information; rather, it should be reviewed along with the attached waste applicability section, as well as Tables 4.2 and 7.1. Waste streams with low applicability to a given technology were listed in the nontreatable category since they are, in general, ineffectively treated by that technology. In most cases, there are exceptions within each waste stream which are applicable to a given technology.

C1 - Incinerator Technologies

In general, incinerator technologies are not applicable to aqueous liquids, noncombustibles, or solids which are large in size. However, some incinerators are amenable to limited amounts of these waste types.

Most incinerators would be equipped with a shredder for large dry waste types. Incinerators, in general, produce an ash or ash-like residue which would require further treatment if produced from mixed waste processing. Therefore, an ash treatment step is included. Many of the incinerator processes will also produce an offgas residue resulting from offgas scrubbing, which would also require treatment.

C2 - Miscellaneous Technologies

Most of the technologies in this category are highly applicable to one or two waste types, rather than a variety of waste types. For example, five of the technologies are applicable only to aqueous liquids, two are highly applicable to dry homogeneous solids, and one is highly applicable to combustible liquids or solids only.

The high temperature reactor/furnace technologies produce ash residues which would require a treatment step to stabilize the leachable species. The three technologies which serve to oxidize organics from aqueous solutions produce not only an ash-like precipitate, but also the treated water stream which would require conversion to a solid form, in most cases, before disposal.

C3 - Melter Technologies

These technologies are applicable to a wide variety of waste streams. However, most of them would require a shredder for some of the larger solids, since they do not have the capability of accepting, for a variety of reasons, large waste constituents. Generally, melters are not amenable to large heterogeneous solids due to the metal content, with the exception of those melters that are operated at very high temperatures, such as the electric arc furnace, plasma arc furnace, and slagging kiln technologies. Since the melter technologies produce a molten slag which then solidifies to a glassy matrix, it will not be necessary, in most cases, to include an additional treatment step. However, if the glass matrix does not pass leach tests, the matrix will require further treatment, most likely including remelting.

APPENDIX C1

INCINERATOR TECHNOLOGIES

Technology Name: Rotary Kiln Incinerator

Maturity: Operational - Conventional

Description:

The rotary kiln is a cylindrical refractory-lined shell mounted on a slight incline. Rotation of the kiln provides for movement of waste through the kiln as well as for enhancement of waste mixing. Rotary kilns normally require a secondary combustion chamber to assure complete destruction of hazardous constituents. The primary chamber functions to pyrolyze or combust solid waste to gases. The gas-phase combustion reaction is completed in the secondary. Both primary and secondary chambers are generally supplied with auxiliary fuel systems. An extensive offgas system is generally required to control the high volume of emissions.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to both sludge and pumpable organic liquid waste streams. |
| Wet Solids: | High applicability to all subcategories of wet solids. |
| Dry Homogeneous Solids: | High applicability to dry solids, especially soils. |
| Dry Heterogeneous Solids (Small): | High applicability to both the combustible and non-combustible fractions of heterogeneous solid wastes. |
| Dry Heterogeneous (Large): | High applicability to large wood items and low applicability to non-combustibles such as equipment and gloveboxes. |

Advantages:

The rotary kiln incinerator is the most versatile type of conventional incinerator. It can handle a wide variety of solid and liquid waste types, and is capable of a wide range of physical waste feed configurations. Ash is removed continuously and does not interfere with waste oxidation. A rotary kiln incinerator can be operated at very high temperatures to handle difficult to destroy constituents, and has a good turndown ratio.

Disadvantages:

Rotary kiln incinerators generally have high capital costs for installation. High particulate loadings are often experienced. Drying of some aqueous sludge waste or melting of some solid wastes can result in clinker or ring formation on refractory walls. Spherical or cylindrical objects may roll through the kiln before complete combustion. Rotary kiln incinerators are not very thermally efficient, and cannot be thermally cycled often (shutdown/startup cycle). The large volumes of air required for combustion give rise to large costly offgas treatment systems.

Development Needs:

Better kiln seal design, advanced offgas systems, better stack monitoring and other real-time performance assurance capabilities; control of heavy metal emissions; combustion by-product formation: sub-micron particulate emissions.

Vendor List:

| | |
|----------------------------|--------------------------------------|
| ABB Raymon | Harper Electric Furnace |
| ABB Environmental Services | International Waste Energy Systems |
| AMETEK Process Systems | International Energy System |
| Allis Chalmers | Joy Energy Systems |
| Anderson 2000 Inc. | Kennedy Van Saun Corp. |
| Aqua-Guard Technologies | Lurgi Corp. |
| Bigelow-Liptak | M&S Engineering & Manufacturing Co. |
| Brule CE&E Inc. | McGill Pollution Control Systems |
| Cleansoils | Soil Purification Inc. |
| Cleeve Brooks Div. | Surface Combustion |
| College Research Corp. | Texcel Environmental Systems |
| Combustion Engineering | Thermal Inc. |
| Combustion Technologies | Thermal Process Construction |
| Conservtherm Systems | Trofe Inc. |
| DRE Technologies | Von Roll |
| Ford-Bacon-Davis | Vulcan Waste Systems |
| Fuller Power | Westinghouse Resource Energy Systems |
| | Williams Environmental Services |
| | John Zinc |

DOE Laboratories Involved in Technology:

Rocky Flats Plant
Oak Ridge K-25 Plant
Savannah River Site
Idaho National Engineering Laboratory

References:

C.C. Lee, G.L. Huffman, D.A. Oberacker, "An Overview of Hazardous/Toxic Waste Incineration", Hazardous Waste Management, Vol. 36, No. 8, August 1986, pp.922-931.

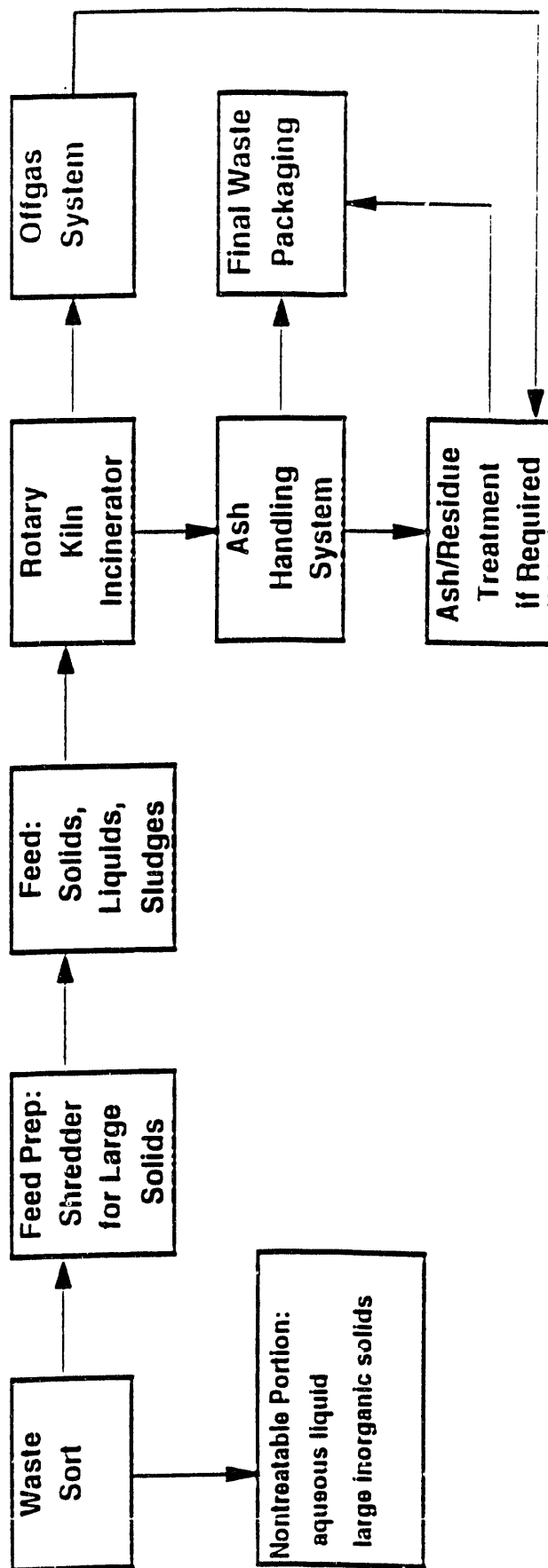
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G. Rich and K. Cherry, Hazardous Waste Treatment Technologies, Pudvan Publishing Co., 1987.

R.J. McCormick et al., Costs for Hazardous Waste Incineration, Noyes Publications, Park Ridge, NJ, 1985.

J. Frankel, N. Sanders, G. Vogel, "Profile of the Hazardous Waste Incinerator Industry", MITRE Corp. report, 1982.

ROTARY KILN INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Fluidized Bed Incinerator

Maturity: Operational-Conventional

Description:

A vertical refractory lined vessel containing a bed of an inert granular material. The bed is "fluidized" by passing air, which serves as combustion air, through a perforated plate at the bottom of the vessel. Waste is fed to the hot bed for combustion where the high thermal mass and turbulent mixing action of the bed material rapidly transfers the heat to the waste. Auxiliary fuel is often used to maintain bed temperature. A secondary chamber may be required to ensure complete combustion for hazardous wastes. Limestone is usually added to the bed to provide capability for in-bed acid gas scrubbing capability (no scrubber required). Offgas particulate removal is required. A variation of fluidized bed technology is a circulating bed system where higher air velocities cause high carryover rates. The carryover material is recovered and returned to the system.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | Medium applicability to organic liquid sludges and high applicability to pumpable organic liquids. |
| Wet Solids: | High applicability to resins with only moderate to low applicability to sludges, absorbed liquids, and cemented sludges respectively. |
| Dry Homogeneous Solids: | Medium applicability to homogeneous dry solids. |
| Dry Heterogeneous Solids (Small): | High applicability to combustible wastes within this category and low applicability to non-combustible wastes. |
| Dry Heterogeneous Solids (Large): | Medium applicability to wood waste (with size reduction). |

Advantages:

The fluidized bed incinerator is relatively simple in design, as it has few moving parts. Its capital and maintenance costs are relatively low, and the incinerator is long-lived. A fluidized bed incinerator is simple to operate, and has ease of process control and high thermal efficiency. Lower operating temperatures lead to lower NO_x formation and metal emission rates, and the capability for in-bed scrubbing eliminates the need for

an offgas scrubber system. Fluidized bed incinerators are versatile in that they can accept solids, liquids, sludges, and gases.

Disadvantages:

Fluidized bed incinerators have a relatively low throughput capacity, and it is difficult to remove residuals from the bed. Operating costs of fluidized bed incinerators are relatively high. Solid wastes will likely have to be pre-treated (shredded or sized) prior to introduction. Residence times are non-uniform, and particulate entrainment rates are high. The vessel and related components are subject to erosion. Low melting point materials in the bed may cause the bed material to fuse.

Development Needs:

Advanced offgas systems capable of removing higher percentages of the radioactive constituents; better stack monitoring and other real-time performance assurance capabilities; control of heavy metal emissions; combustion by-product formation; sub-micron particulate emissions.

Vendor List:

| | |
|---------------------------|-----------------------------------|
| ARI Technologies | GA Technologies |
| AWT Systems | Hankin Environmental Systems Inc. |
| Aerojet Energy Conversion | Keeler/Dorr Oliver |
| Anderson 2000 Inc. | Lurgi Corp. |
| Combustion Power company | Niro Atomizer Inc. |
| Conversion Technologies | Process Combustion Corp. |
| Copeland Associates | Texcel Environmental Systems Co. |
| Fuller Company | Waste Tech Services, Inc. |
| | Zimpro/Passavant Inc. |

DOE Laboratories Involved in Technology:

Rocky Flats Plant

References:

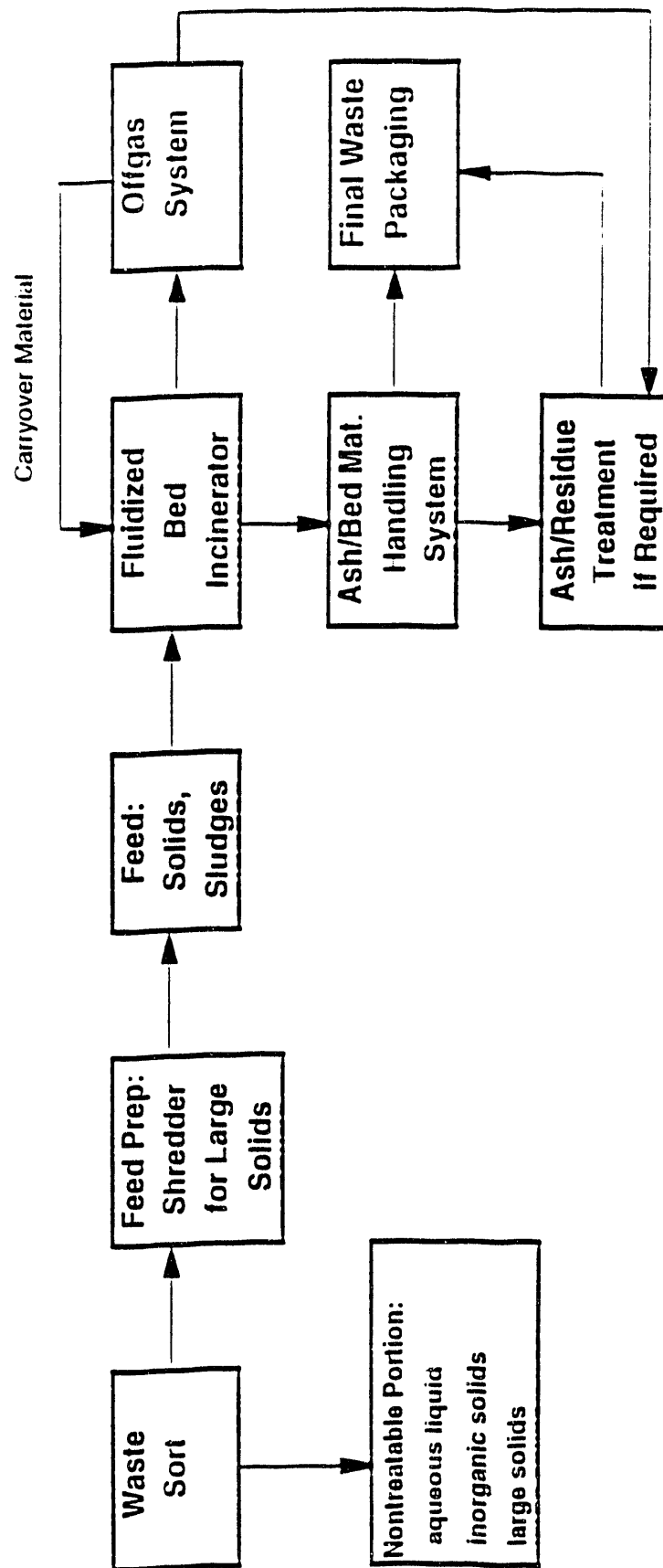
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Gerald Rich and Kenneth Cherry, Hazardous Waste Treatment Technologies, Pudvan Publishing Co., 1987.

R.A. Koenig, J. McFee, J.S. Vavruska, "Incineration Systems", Incineration Conference 1990, San Diego, CA, May 1990.

FLUIDIZED BED INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Agitated Hearth Incinerator

Maturity: Operational-Unique

Description:

The agitated hearth incinerator is a vertical, cylindrical chamber with a rabble arm that is rotated around the incinerator hearth to slowly agitate the waste pile, spreading out the waste material and exposing unburned material to the combustion air supply. The shaft for the rabble arm penetrates up through the bottom of the incinerator in the center of the hearth. Solid waste is fed in from the side through a ram feeder. Multiple burners are located at approximately mid-height in the cylindrical walls for start-up and auxiliary heat input. These burners could also be used for liquid waste disposal. Combustion gases pass from the primary chamber to the secondary chamber for extended residence time. In operation, waste is slowly fed into the chamber over a period of time. During this period, the rabble arm continually mixes the burning waste. Eventually, the waste pile builds up and waste feeding is stopped. The incinerator goes through a burn-out cycle where the rabble arm continues to mix the waste and stir the ashes until all combustible material is consumed. As the heat input from the burning waste begins to fall, the burners are ignited to maintain the temperature in the primary chamber at approximately 800°C. When the waste is completely consumed, an ash discharge door in the floor of the hearth is opened and the ash is raked out of the chamber by the rabble arm. When the ashes have been removed, the ash discharge door is closed, the feeding cycle begins, and the process is repeated.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability. |
| Organic Liquids: | High applicability. |
| Wet Solids: | High applicability. |
| Dry Homogeneous Solids: | Medium applicability. |
| Dry Heterogeneous Solids (Small): | High applicability for combustible waste components. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

This is a simple system that can achieve an excellent burn-out of combustible matter because of the long solids retention time and the mixing of the waste. The process is easy to monitor and control, and can incinerate a wide range of combustible waste types including solids, liquids, and sludges.

Disadvantages:

Proper operation depends on movement of the rabble arm, which is subject to mechanical, chemical, and thermal stresses. Operation is also limited by the size of waste constituents that are too heavy to be moved by the rabble arm or which could jam the rabble arm.

Development Needs:

Conversion and long-term operation tests on radioactive waste. Characterization of offgas, especially in terms of toxic metals and fine particulate. Better stack monitoring and other real-time performance assurance capabilities.

Vendor List:

Environmental Tech (it is unknown if this company still exists)

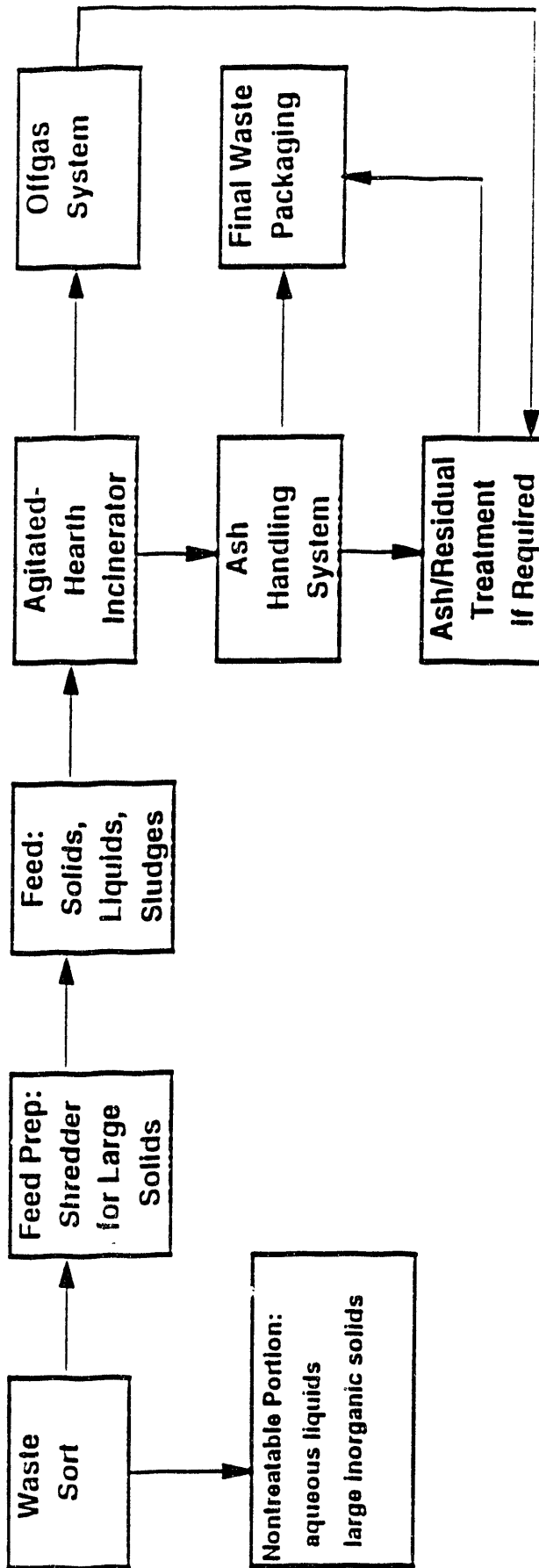
DOE Laboratories Involved in Technology:

Rocky Flats Plant

References:

D. L. Ziegler, "Incineration Process Fire and Explosion Protection", presented at the 13th AEC Air Cleaning Conference, San Francisco, California, August 1974.

AGITATED-HEARTH INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Multiple-Hearth Incinerator

Maturity: Operational-Conventional

Description:

A multiple-hearth incinerator consists of a refractory-lined steel shell with a series of circular hearths arranged in a vertical design. A series of rotating, air-cooled rabble arms conveys the solid waste from upper to lower hearths. As the waste is conveyed down through the incinerator the successive hearths are used for drying, heating, combustion, burnout, and cooling of the waste. Fuel burners are mounted on the side of the vessel in the hearths where combustion and burnout occur. These burners can be used for high heat value hazardous liquids if desired. A secondary chamber may be required for complete destruction of hazardous wastes. Some form of air pollution control equipment will be required, and will vary with the waste being processed. This type of incinerator has been used principally for sludges, tars, or other low-heat value solids requiring long solids retention times, and has been commonly used for disposal of de-watered activated waste-water treatment sludges. Use of this type of incinerator has been largely abandoned.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | Medium applicability to organic liquid sludges and high applicability to pumpable organic liquids. |
| Wet Solids: | High applicability to all subcategories of wet solids waste with the exception of medium applicability to cemented sludges. |
| Dry Homogeneous Solids: | Low to medium applicability to homogeneous dry solids and soils respectively. |
| Dry Heterogeneous Solids (Small): | Medium applicability to combustible wastes within this category and low applicability to non-combustible wastes. |
| Dry Heterogeneous Solids (Large): | Low applicability to wood waste only, not applicable to remainder of category. |

Advantages:

The long solids retention times achieved in multiple hearth incinerators increase the complete destruction of waste materials. Multiple hearth incinerators can handle a wide

range of wastes, including solids, sludges, liquids, and gases, and are capable of evaporating large amounts of water. A wide range of fuels may be utilized to operate multiple hearth incinerators.

Disadvantages:

Multiple hearth incinerators cannot handle wastes that fuse into large chunks during incineration, and are not good for wastes requiring high destruction temperatures. The incinerators are susceptible to thermal shock. The large volumes of air required for combustion give rise to large, costly, and difficult-to-operate offgas treatment systems. Solid wastes may have to be pre-treated (shredded) before processing.

Development Needs:

Advanced offgas systems adapted to remove high percentages of radioactive constituents; better stack monitoring and other real-time performance assurance capabilities; control of heavy metal emissions; combustion by-product formation; sub-micron particulate emissions.

Vendor List:

| | |
|-----------------------------------|----------------------------------|
| Bethlehem Corp. | Texcel Environmental Systems Co. |
| BSP Thermal Systems | Thermal Process Construction |
| Hankin Environmental Systems Inc. | Zimpro/Passavant Inc. |
| Kennedy Van Saun Corp. | |

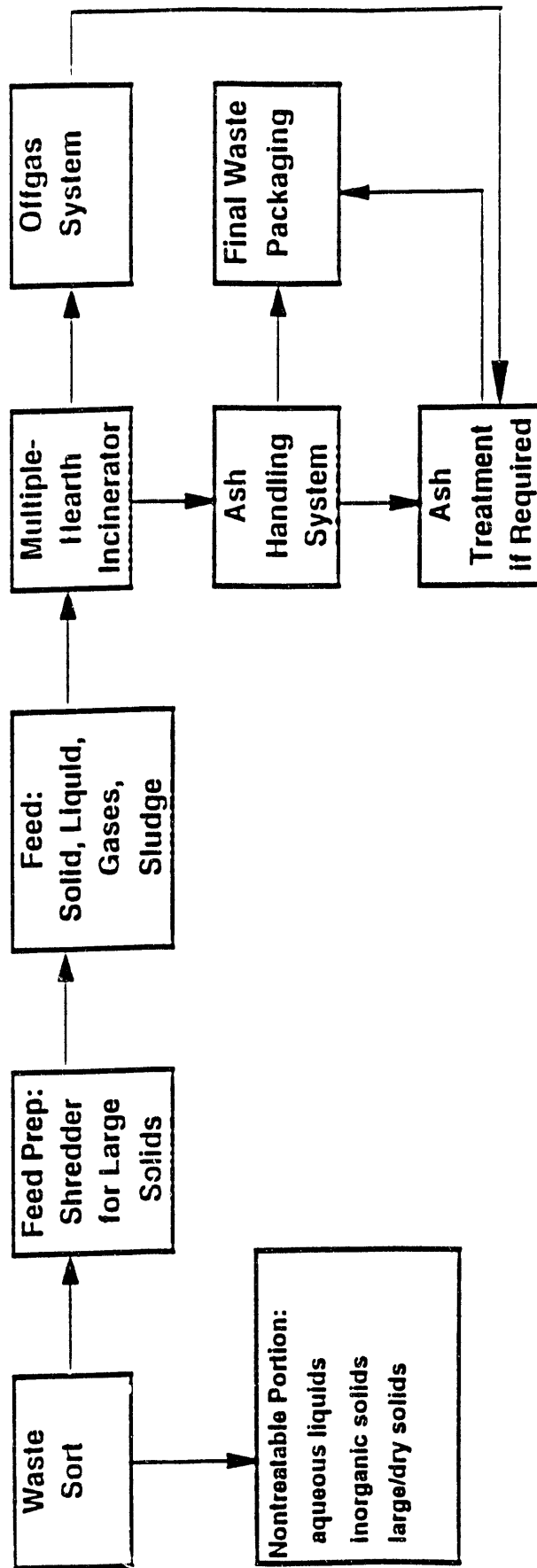
DOE Laboratories Involved in Technology:

None

References:

- C.W. Modres, "Waste-to-Energy '87: Exploring the Total Market", Badger Engineers, Inc. report, 1987.
- W.E. Sweet, R.D. Ross, G.V. Velde, "Hazardous Waste Incineration: A Progress Report", Journal of the Air Pollution Control Association, Vol. 35, No. 2, February 1985.
- G. Rich and K. Cherry, Hazardous Waste Treatment Technologies, Pudvan Publishing Co., Northbrook, IL, 1987.
- J. Cudahy, T. Eicher, "Hazardous Waste Incineration Course", prepared by IT Corp., August, 1988.

MULTIPLE-HEARTH INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: KFK Excess Air Incinerator

Maturity: Operational-Conventional

Description:

The KFK Incinerator is a vertical shaft excess air incinerator. The incinerator consists of a cylindrical shaft furnace with a refractory lining. The bottom section of the furnace is constructed in a cone shape. An afterburning chamber, two hot gas filters and a two-stage flue gas scrubbing system are located downstream of the furnace. The scrubbing system consists of a jet scrubber, a Venturi scrubber, a HEPA filter, and an exhaust fan. Charging of the waste takes place via a feeding system which is accommodated in glove boxes. The furnace is charged automatically, depending on the O₂ content as well as on the furnace temperature. A double closure serves to ensure the contamination-free supply of waste from the drums. The cylindrical shaft furnace is operated at a temperature of at least 850°C. The minimum temperature is attained by means of a propane burner. For incineration, air is supplied in a controlled manner via several inlets oriented tangentially to the furnace walls. The temperature of the ash bed in the cone-shaped bottom part of the furnace is maintained at ≤800°C by means of steam addition. As a result, heat is removed from the ash bed and the slag is prevented from adhering to the furnace. Ash discharge also takes place within a glovebox system which is equipped with a double closure to avoid contamination.

Waste Applicability:

Aqueous Liquids: Low applicability.

Organic Liquids: High applicability.

Wet Solids: High applicability, except possibly for resin wastes.

Dry Homogeneous Solids: Low to medium applicability to homogeneous dry solids and soils respectively.

Dry Heterogeneous Solids (Small): High applicability to combustible wastes in this category, low applicability to non-combustible wastes.

Dry Heterogeneous Solids (Large): Low applicability to wood waste only.

Advantages:

This technology has over 20 years operating experience in Germany and Japan, incinerating both beta and alpha contaminated wastes (liquid wastes since 1988).

Disadvantages:

Large volumes of combustion air result in complex, costly offgas treatment systems. High particulate carryover will entrain radioactive components into downstream components.

Development Needs:

Advanced offgas systems, combustion by-product formation, and determination of optimal secondary chamber operating points.

Vendor List:

Fahrholf (Denmark)
NGC (Japan)
NUKEM (Germany)

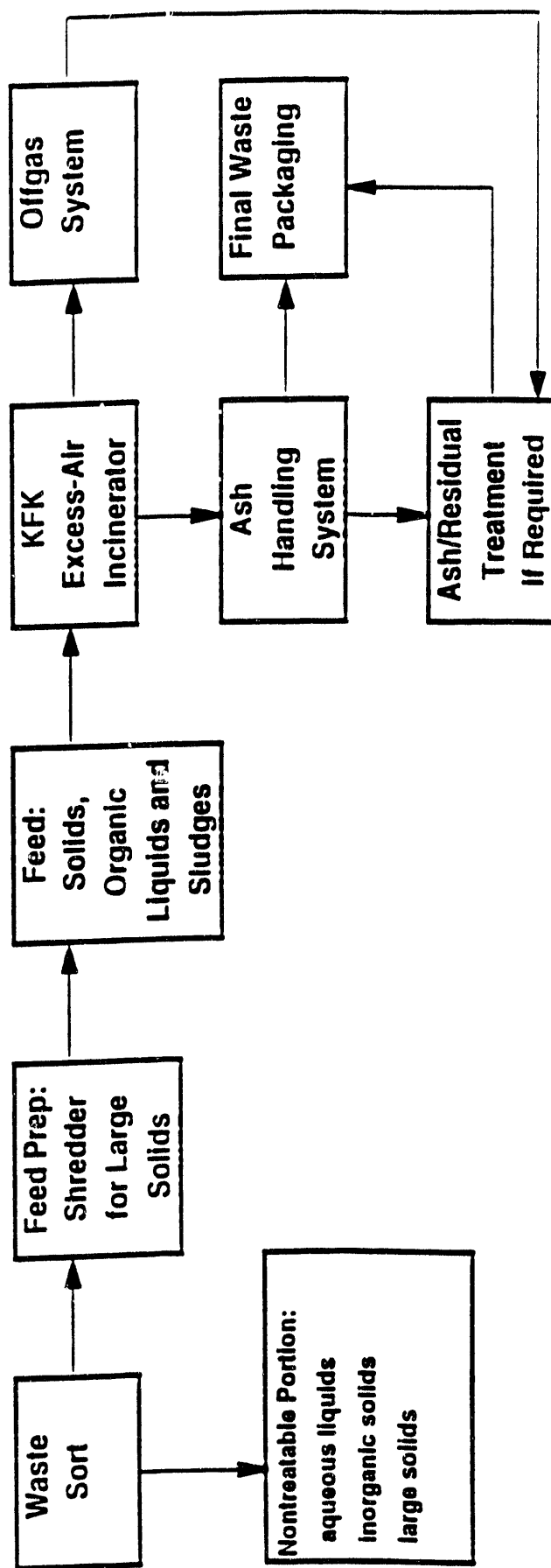
DOE Laboratories Involved in Technology:

None identified

References:

F. Dirkz, W. Hempelmann, W. Pfeifer, G. Steinhaus, "Incineration of Radioactive Residues Further Development of KFK Incineration Plants, Plant Performance and Test Results", presented at 1991 Incineration Conference, May 1991.

KFK EXCESS-AIR INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Liquid Injection Incinerator

Maturity: Operational-Conventional

Description:

A type of incinerator designed to process liquid wastes only. They are usually simple, refractory-lined cylinders equipped with one or more waste burners. Only one combustion chamber is generally used (secondary chamber not necessary for proper destruction). An offgas treatment system may be required depending on the application, and will vary in design based on the types of waste being processed. Most commercial type liquid injection incinerators do not use offgas equipment; however, a liquid injection incinerator for radioactive waste will likely require some type of offgas particulate filter. Sometimes used as a secondary chamber for other incinerator types.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids that are pumpable only. |
| Wet Solids: | Not applicable to any wet solid subcategories with the exception of low applicability to resin wastes. |
| Dry Homogeneous Solids: | Not applicable. |
| Dry Heterogeneous Solids (Small): | Not applicable. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

No secondary combustion chamber is needed if the primary combustor has enough residence time. Liquid injection incinerators can incinerate a wide range of liquid hazardous waste. No continuous ash removal system is required other than for downstream air pollution control systems. Their simple design is thermally efficient, entails virtually no moving parts, and enables fairly high turndown ratios. Maintenance costs for liquid injection incinerators are low.

Disadvantages:

Wastes which can be accepted in a liquid injection incinerator are restricted to only those that can be atomized through a burner nozzle (liquids with low or no solids content, and no sludges or solids). The incinerator system is sensitive to waste composition changes. The incinerator burners may be susceptible to plugging. The offgas systems necessary for a liquid injection incinerator generate secondary byproduct wastes that are often difficult to handle.

Development Needs:

Advanced offgas systems capable of removing a higher percentage of radioactive constituents; better stack monitoring and other real-time performance assurance capabilities; control of heavy metal emissions; combustion by-product formation; sub-micron particulate emissions.

Vendor List:

| | |
|-----------------------------------|---------------------------------|
| Anderson 2000 Inc. | Lurgi Corp. |
| Bayco Industries | McGill Pollution Control Sys. |
| Bedford Industries Inc. | Met-Pro |
| Bigelow-Liptak | NOA Inc. |
| Brule CE&E | North American Manufact. Co. |
| Burn-Zol | Peabody |
| B&W | Preco |
| Combustion Technologies | Process Combustion |
| Copen | Product Recovery and Energy Co. |
| Durr Engineering & Management | Pyro Industries Inc. |
| Entech | Texcel Env. Systems Inc. |
| Energy Development Assoc. | T-Thermal Inc. |
| Epscon Industrial Systems Inc. | Trecan Combustion Inc. |
| Fuel & Combustion Technology Inc. | Selas Fluid Processing Corp. |
| Hirt Combustion Engineers | Smith Eng. & Environmental |
| John Zink | Sure-Life |
| Kelly | Surface Combustion Inc. |
| Liquid Injection | United |
| Lotepro Corp. | UPO Solid Waste Systems |

DOE Laboratories Involved in Technology:

Savannah River Site

References:

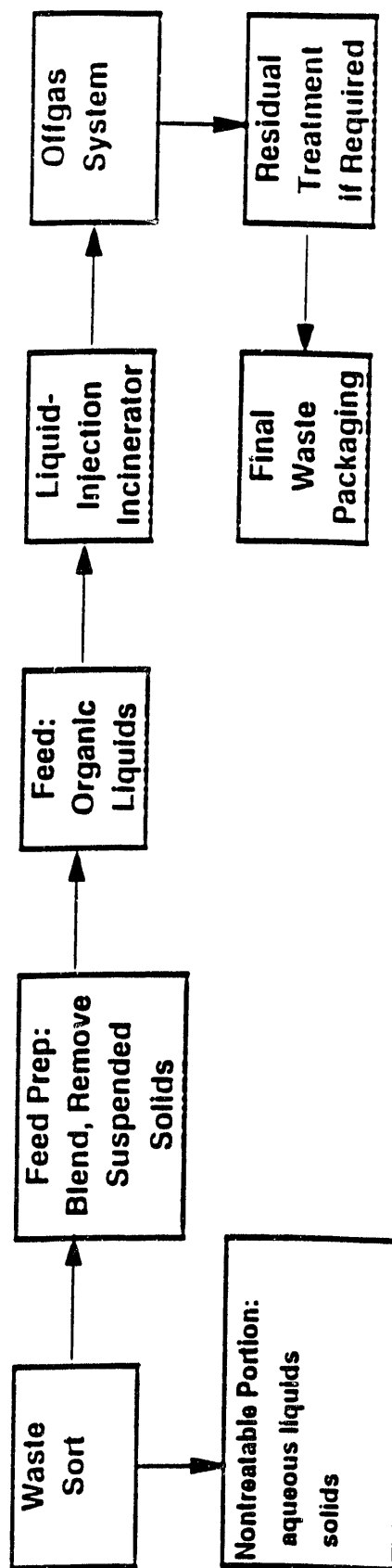
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G. Rich and K. Cherry, Hazardous Waste Treatment Technologies, Pudvan Publishing Co., 1987.

R.J. McCormick et al., Costs for Hazardous Waste Incineration, Noyes Publications, Park Ridge, NJ, 1985.

I. Frankel, et al, "Profile of the Hazardous Waste Incinerator Manufacturing Industry", MITRE Corp. report, 1982.

LIQUID-INJECTION INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Controlled Air Incinerator

Maturity: Operational - Conventional

Description:

A controlled air incinerator is the name often used for the stationary hearth class of incinerator. This type of incinerator is usually designed as a two-stage combustion process with some systems using three chambers. Solid waste is fed into the primary chamber and burned at roughly 50 to 80% of the stoichiometric air requirement (starved air condition). This pyrolyzes the waste, thus emitting a volatile fraction with the required heat supplied by partial combustion and oxidation of the fixed carbon. The resultant smoke and pyrolytic products, consisting primarily of volatile hydrocarbons and carbon monoxide along with some combustion products, pass to the secondary chamber. Excess air is provided in the secondary chamber to assure complete combustion. Liquid waste can be incinerated in either the primary or secondary chambers. An offgas treatment system is required to provide emission control, dependent on the application and waste type.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids that are pumpable and medium applicability to sludges. |
| Wet Solids: | Low applicability to sludges and medium applicability to the remaining wet solids subcategories. |
| Dry Homogeneous Solids: | Low applicability. |
| Dry Heterogeneous Solids (Small): | High applicability to combustible heterogeneous solids and low applicability to noncombustible solids. |
| Dry Heterogeneous Solids (Large): | Medium applicability to wood waste and not applicable to noncombustible equipment and metal type wastes. |

Advantages:

The starved air condition in the primary chamber leads to a lower air velocity, thus minimizing particulate entrainment and carryover. Controlled air incinerators can be used to process a wide variety of wastes including solids, liquids, and sludges, and can handle

wastes with high water content. These incinerators have a low cost modular design, can utilize a wide range of supplementary fuels, and are easy to control.

Disadvantages:

Solid wastes generally have to be pre-treated or packaged in some fashion before they can be fed to the incinerator. Controlled air incinerators are not well suited for wastes containing fusible ash, large bulky solid wastes, or large quantities of essentially non-combustible materials (i.e. metal and glass). Batch feeding of waste can lead to pressure spikes in the primary chamber. The large volumes of air required for secondary combustion give rise to large, costly, and difficult to operate offgas treatment systems. Offgas systems generate secondary byproduct wastes that are often difficult to handle.

Development Needs:

Advanced offgas systems capable of retaining a higher percentage of radioactive constituents, better stack monitoring and other real-time performance assurance capabilities; control of heavy metal emissions; combustion by-product formation; sub-micron particulate emissions.

Vendor List:

AER

Aerojet Energy Conversion
American Energy Waste System
Anderson 2000 Inc.
Basic Environmental Engineering
Besser-Wasteco Corp.
Burney The Burner
Cil Incineration Systems
Cleever-Brooks Div.
Consumat Systems
Econo-Therm Energy Systems Corp.
International Waste Energy Systems
Fuller Company

Joy Energy Systems Inc.
Kennedy Van Saun Corp.
Koch Process Systems Inc.
Simonds Manufacturing Corp.
Stock Equipment Co.
Thermall Inc.
Thermal Process Constr. Co.
Trecan Combustion Ltd.
Vent-O-Matic Incineration Corp.
Vulcan Waste Systems Inc.
John Zink Co.

DOE Laboratories Involved in Technology:

| | |
|--|---------------------|
| Brookhaven National Laboratory | PANTEX |
| Idaho National Engineering Laboratory | Rocky Flats Plant |
| Lawrence Livermore National Laboratory | Savannah River Site |
| Los Alamos National Laboratory | |
| Oak Ridge Y-12 Plant | |
| Oak Ridge K-25 Site | |

References:

C.C. Lee, G.L. Huffman, D.A. Oberacker, "An Overview of Hazardous/Toxic Waste Incineration", Hazardous Waste Management, Vol. 36, No. 8, August 1986, pp. 922-931.

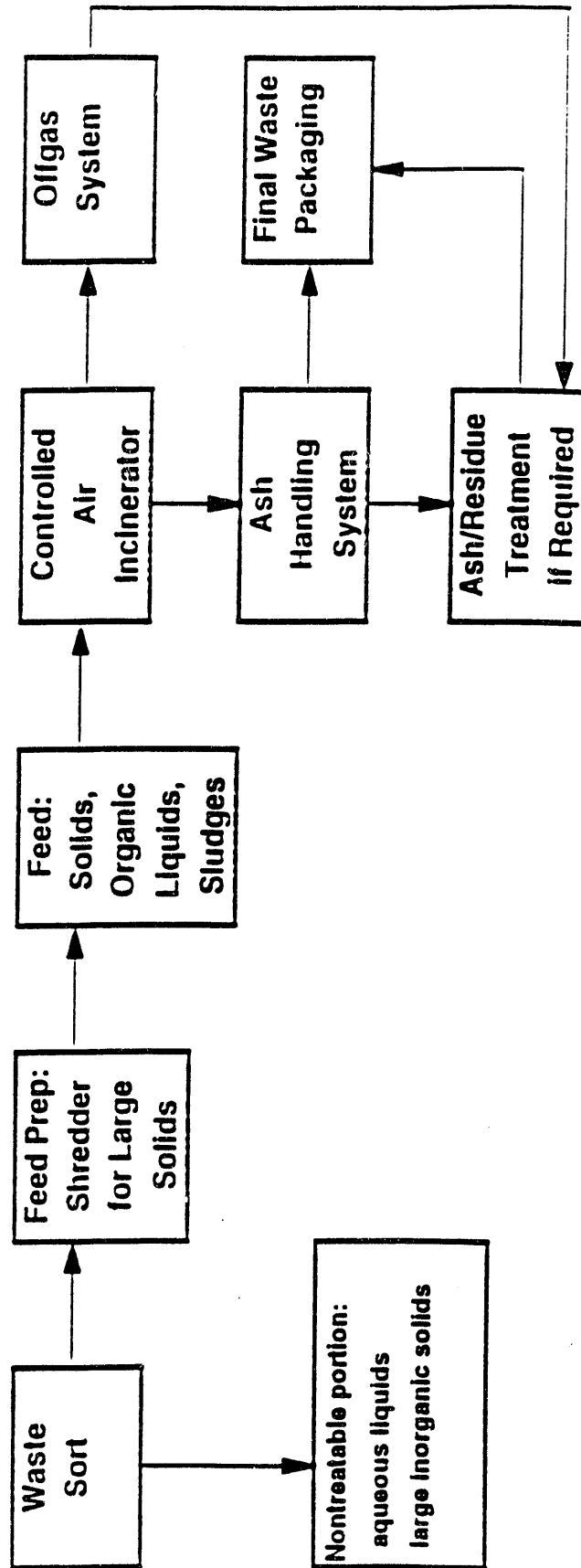
R.J. McCormick et al., Costs for Hazardous Waste Incineration, Noyes Publications, Park Ridge, NJ, 1985.

R. A. Koenig, L. C. Borduin, D. A. Hutchins, J. S. Vavruska, C. L. Warner, "The Los Alamos Controlled Air Incinerator for Radioactive Waste", Vols. I-III, LA-9427, 1982-1987.

R. E. McRee, "Operation of Controlled Air Incinerators and Design Considerations for Controlled Air Incinerators Treating Radioactive Wastes", presented to the Conference on Incineration of LLRW, Tucson, AZ, March 1985.

R. L. Gillins, H. A. Bohrer, "Progress Report on Contaminated Solid Waste Incineration at the Waste Experimental Reduction Facility", prepared for the U.S. DOE, EGG-WM-7162, February 1986.

CONTROLLED AIR INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Cyclone Incinerator

Maturity: Operational-Unique

Description:

The cyclone incinerator is a single hearth, vertical cylindrical vessel in which cyclonic flow is induced through the tangential introduction of fuel and air. The high-shear cyclonic flow provides intense mixing and complete combustion. Cyclone incinerators are primarily used for solid fines and dried sludges, but special furnaces have also been designed for gases or liquids. Typically, the hearth rotates with stationary rabble teeth for moving ash to a center discharge. Horizontal cyclone furnaces without hearths are also employed. These units carry the ash away with the offgas for downstream collection.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids that are pumpable and medium applicability to sludges. |
| Wet Solids: | Medium applicability to sludges, absorbed liquids, and resins and low applicability to cemented sludges. |
| Dry Homogeneous Solids: | Low applicability to soils and not applicable to other dry solids waste such as concrete, bricks, and salts. |
| Dry Heterogeneous Solids (Small): | Medium applicability to combustible heterogeneous solids and low applicability to noncombustible solids. |
| Dry Heterogeneous Solids (Large): | Medium applicability to wood waste (size reduced) and not applicable to noncombustible equipment and metal type wastes. |

Advantages:

Cyclone incinerators are inexpensive and mechanically simple. The low temperature requirements allow for fast startup and cool down. The combustion in cyclone incinerators is stable and efficient, and the combustion volume is small. The refractory is long-lived. The offgas and particulate loading are separated centrifugally. The high energy density of the process results in high destruction efficiencies at moderate temperatures.

Disadvantages:

Cyclone incinerators are limited to processing gaseous, liquid, and sludge wastes.

Development Needs:

Destruction and removal efficiency determination.

Vendor List:

Babcock & Wilcox
International Gas Technology
York-Shipley Inc.

DOE Laboratories Involved in Technology:

Mound

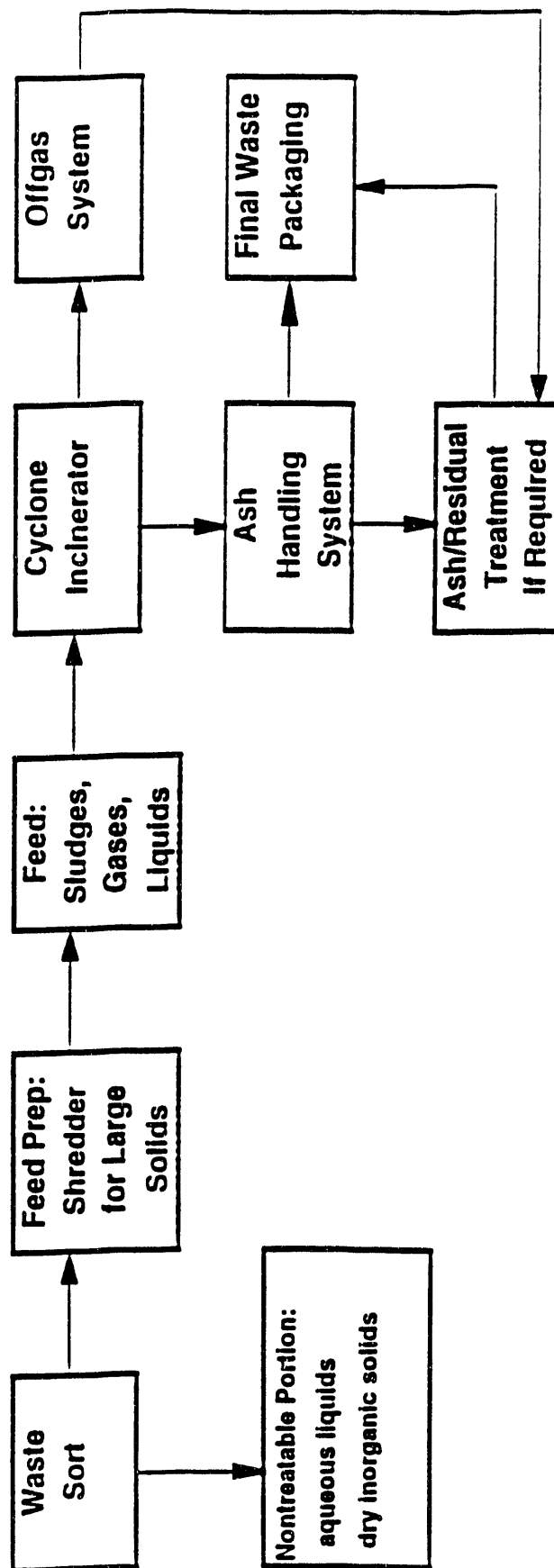
References:

H.M. Freeman, "Innovative Thermal Hazardous Waste Treatment Processes", PB85-192847, April 1985.

C.R. Brunner, "Incineration Systems", Incinerator Consultants Inc., Reston, VA, 1988.

Contact: Institute of Gas Technology (Headquarters), 3424 South State Street, Chicago, IL, 60616, (312) 567-3650.

CYCLONE INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Indirect Fired Pyrolysis Incinerator

Maturity: Operational-Unique

Description:

A thermal treatment process consisting of a low temperature, indirect-fired furnace for pyrolyzing waste followed by a rich fume reactor to complete combustion and destruction. The pyrolysis process achieves chemical decomposition of waste materials by applying heat in the absence of oxygen resulting in high DREs and low NO_x levels and particulate carryover. The process is available in continuous feed for granular or liquid materials, or batch feed for liquids, solids, or sludges in open containers. Wastes are pyrolyzed at relatively low temperatures (1000°-1600°F) for 15-30 minutes for the continuous system and 4-6 hours for the batch system. The resulting fumes are then completely combusted in a rich-fume reactor chamber at 1800°-2200°F for 1-2 seconds. Heating in the pyrolyzing chamber is provided by natural gas or fuel oil. A widely used commercial application is the destruction of organic contamination on metals and equipment. One application would collect the pyrolysis fumes for utilization as a fuel gas.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquid sludges and low applicability to pumpable liquids. |
| Wet Solids: | Medium applicability to sludges and absorbed liquids, high applicability to resins, and low applicability to cemented sludges. |
| Dry Homogeneous Solids: | Low applicability to soils and other dry solids waste such as concrete, bricks, and salts. |
| Dry Heterogeneous Solids (Small): | Medium applicability to combustible heterogeneous solids and high applicability to noncombustible solids (removal of organic contamination). |
| Dry Heterogeneous Solids (Large): | High applicability to noncombustible equipment and metal type wastes (removal of organic contamination). |

Advantages:

Inert materials are not melted or vaporized. The indirect heating and pyrolyzing mode of indirect fired pyrolysis incinerators minimizes particulate carryover. These incinerators produce low volumes of offgas with low NO_x concentrations. Excellent control of thermal rates can be achieved.

Disadvantages:

Indirect fired pyrolysis incinerators are inefficient for processing high Btu liquid wastes, and the process is not applicable for inert solids, except to remove organic contamination. Batch system process rates are low. Removal of waste containers from batch system presents high contamination risk when processing radioactive wastes.

Development Needs:

Adaptation to radioactive service.

Vendor List:

Bryant Incinerator
Midland-Ross Corporation

DOE Laboratories Involved in Technology:

None

References:

Marc Breton et al., "Technical Resource Document: Treatment Technologies for Solvent Containing Wastes", U.S. Environmental Protection Agency, EPA/600/2-86/095, October 1986.

Harry Freeman, Innovative Thermal Hazardous Organic Waste Treatment, Noyes Publications, Park Ridge, NJ, 1985.

T.J. Schultz et al., "Pyrolytic Incineration of Hazardous and Toxic Wastes", American Institute of Chemical Engineers, Houston, TX, March 1989.

"Treatment Technology Briefs: Alternatives to Hazardous Waste Landfills", U.S. Environmental Protection Agency, EPA/600/8-86/017, July 1986.

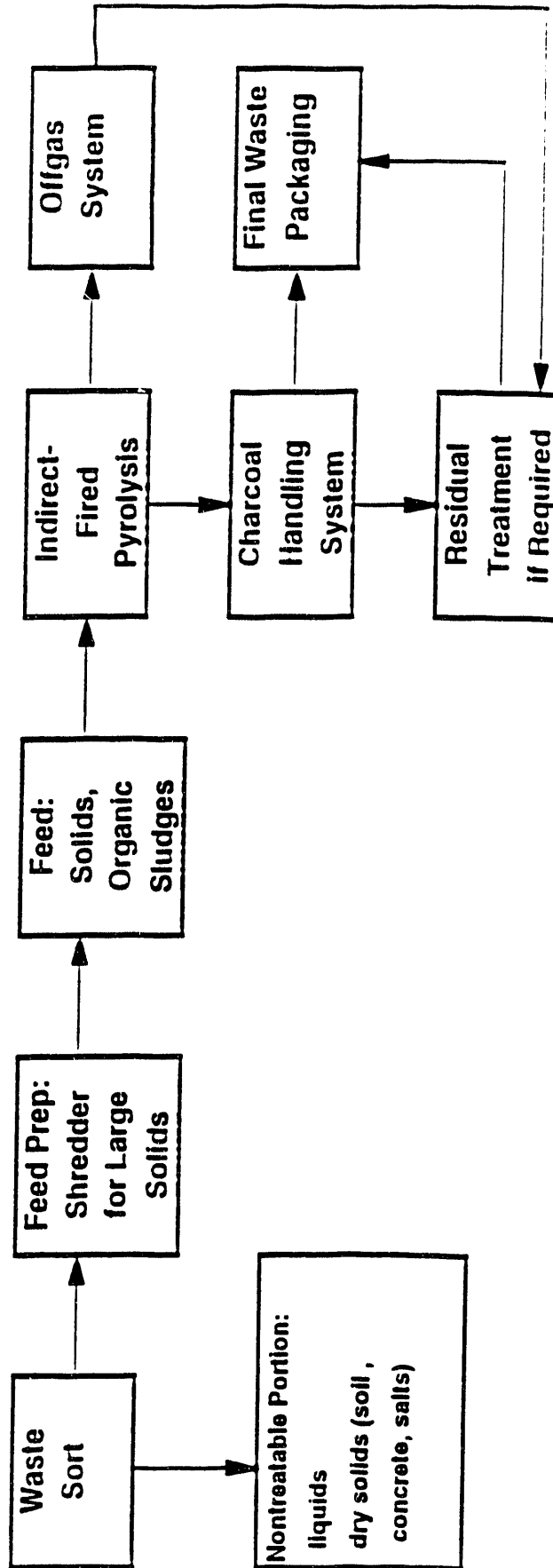
J.C. Shah et al., "Thermal Treatment for Disposal of Containerized Hazardous Wastes", American Institute of Chemical Engineers 1988 Spring National Meeting, New Orleans,

LA, March 1988.

"A Background Paper on Pyrolytic Incineration, Surface Combustion". Inc., Maumee, OH, August 1988.

Contact: Tom Schultz, Surface Combustion, Inc., 1700 Indian Wood Circle. Maumee, Ohio, 43537-0428, (419) 891-7150.

INDIRECT FIRED PYROLYSIS INCINERATOR FUNCTIONAL PROCESS DIAGRAM



APPENDIX C2

MISCELLANEOUS TECHNOLOGIES

Technology Name: Infrared Furnace

Maturity: Operational-Unique

Description:

A thermal waste processing unit employing direct radiant (infrared) heat in a primary chamber to desorb organics from soils followed by fossil fuel fired secondary combustion chamber. The primary chamber has several heating zones with increasing temperatures to initially dry and finally combust the waste passing through. The primary chamber can be operated in a pyrolysis or combustion mode. Waste is transported through the primary chamber on a mesh metal alloy conveyor belt. Variable residence time is provided by adjusting the belt speed. The process is designed to treat organically contaminated soils and sludges. Most solid wastes require size reduction to insure maximum exposure to the radiant energy. Sludges require pretreatment drying before feeding to the incinerator. The waste is stirred by rotary rakes to ensure adequate exposure. The ash is quenched by water sprays. Available in stationary and mobile applications.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Not applicable to aqueous liquids. |
| Organic Liquids: | Not applicable to organic liquids. |
| Wet Solids: | Low applicability to wet sludges, absorbed liquids, resins, and cemented sludges. |
| Dry Homogeneous Solids: | High applicability to soils and other fine dry solid waste. |
| Dry Heterogeneous Solids (Small): | Not applicable to combustible solids or noncombustible wastes. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

Infrared furnaces have high throughput capacities. They operate with non-flame combustion, and therefore have low NO_x and PIC generation rates. Offgas requirements for such furnaces are minimal. Infrared furnaces are easy to operate, and can be purchased as mobile units (6 trailers).

Disadvantages:

Wastes which can be processed with an infrared furnace are limited to solid fines and sludges, and drying and other pretreatment is usually required for sludges. Feed handling equipment is prone to clogging. Infrared furnaces are expensive. Sticky ash clinging to the conveyor belt can become a problem when the furnace is operated at high combustion temperatures.

Development Needs:

Optimization, effect of treatment on the leachability of metals in matrix; improved transport method through furnace (belt); and improved mixing mechanism.

Vendor List:

ECOVA
Harper Electric Furnace
National Applied Science Systems Inc.
OHM
Shirco Infrared Systems, Inc.
Westinghouse HazTech

DOE Laboratories Involved in Technology:

Savannah River Site

References:

Howard O. Wall et al., "The SITE Demonstration of the Shirco Electric Infrared Incinerator", Journal of the Air Pollution Control Association, Vol. 39, No. 6, June 1989.

"Technology Evaluation Report, SITE Program Demonstration Test", Shirco Infrared Incineration System, Peak Oil, Brandon, Florida, EPA/540/5-88/002a, September 1989.

"Technology Demonstration Summary SITE Program Demonstration Test", Shirco Infrared Incineration System at the Peak Oil Superfund Site, EPA/540/55-88/002, January 1989.

David Charlesworth and Mike Hill, "Electrically Fired Incineration of Combustible Radioactive Waste", 1985 National Conference on Environmental Engineering, Boston, July 1985.

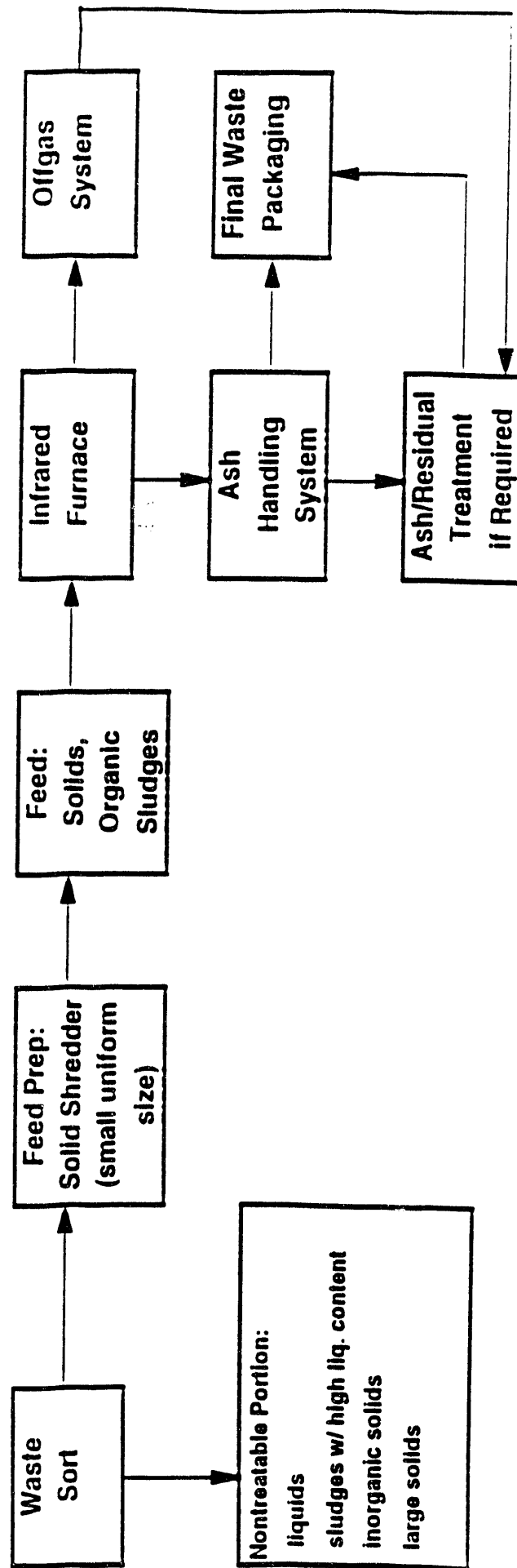
Final Report, On Site Incineration Testing of Shirco Infrared Systems Portable Pilot Test Unit, Times Beach Dioxin Research Facility, Times Beach, MO, Shirco Report No. 815-85-2, November 1985.

A. Judson Hill, "Hazardous Waste Treatment Capabilities of the Shirco Infrared Demonstration and Full Scale Mobile Waste Processing Systems". 2nd National Symposium on the Leading Edge of Incineration, Washington, DC. October 1987.

Shirco Infrared Incineration System, Applications Analysis report, EPA/540/A5-89/010, June 1989.

Contact: Shirco Infrared Systems, Inc., 1195 Empire Central, Dallas, TX 75247, (214) 630-7511.

INFRARED FURNACE FUNCTIONAL PROCESS DIAGRAM



Technology Name: Wet Air Oxidation

Maturity: Operational-Unique

Description:

The aqueous phase oxidation of dissolved or suspended organic substances at elevated temperatures and pressures. Oxygen (air) and a dilute organic/water mixture are introduced into a reactor vessel at subcritical conditions (350°-650°F and 20-200 atm.) where oxidation of the organics occurs. The process, once started, is thermally self-sustaining and is maintained above the vapor pressure of water to minimize evaporation. The process reduces the organics to H₂O, CO₂, and various biodegradable acids. Reaction times of 60 minutes are typical.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | High applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids with < 10% organics. |
| Wet Solids: | Not applicable to solids. |
| Dry Homogeneous Solids: | Not applicable. |
| Dry Heterogeneous Solids (Small): | Not applicable. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

The wet air oxidation process is thermally self-sustaining. It is suited for non-incinerable dilute wastes, and requires small equipment volumes. Its low off-gas volumes are free of NO_x, SO_x, PICs, and particulate.

Disadvantages:

Since wet air oxidation does not generally meet EPA treatment standards, the process is predominantly used for pretreatment. The wet oxidation process is not highly predictable. Existing full-scale units are largely tailored to bench-scale results on specific compounds. Wastes which can be processed using wet air oxidation are limited to weak aqueous organic solutions. High-pressure system hardware is required. Offgas scrubbing

is required. The wet air oxidation process is not effective on halogenated species.

Development Needs:

Improve systems corrosion and corrosion monitoring; evaluate oxyhydroxide formation with actinides; evaluate ash content limits; and adaptation to radioactive applications

Vendor List:

Zimpro/Passavant, Inc.
Oxidyne
Ver Tech

DOE Laboratories Involved in Technology:

None

References:

Marc Breton et al., "Technical Resource Document: Treatment Technologies for Solvent Containing Wastes", U.S. Environmental Protection Agency, EPA/600/2-86/095, October 1986.

Harry Freeman, Innovative Thermal Hazardous Organic Waste Treatment, Noyes Publications, Park Ridge, NJ, 1985.

H.M. Freeman et al., "Thermal Destruction of Hazardous Waste -A State-of-the-Art Review, Journal of Hazardous Materials, Vol. 14, 1987, pp. 103-117.

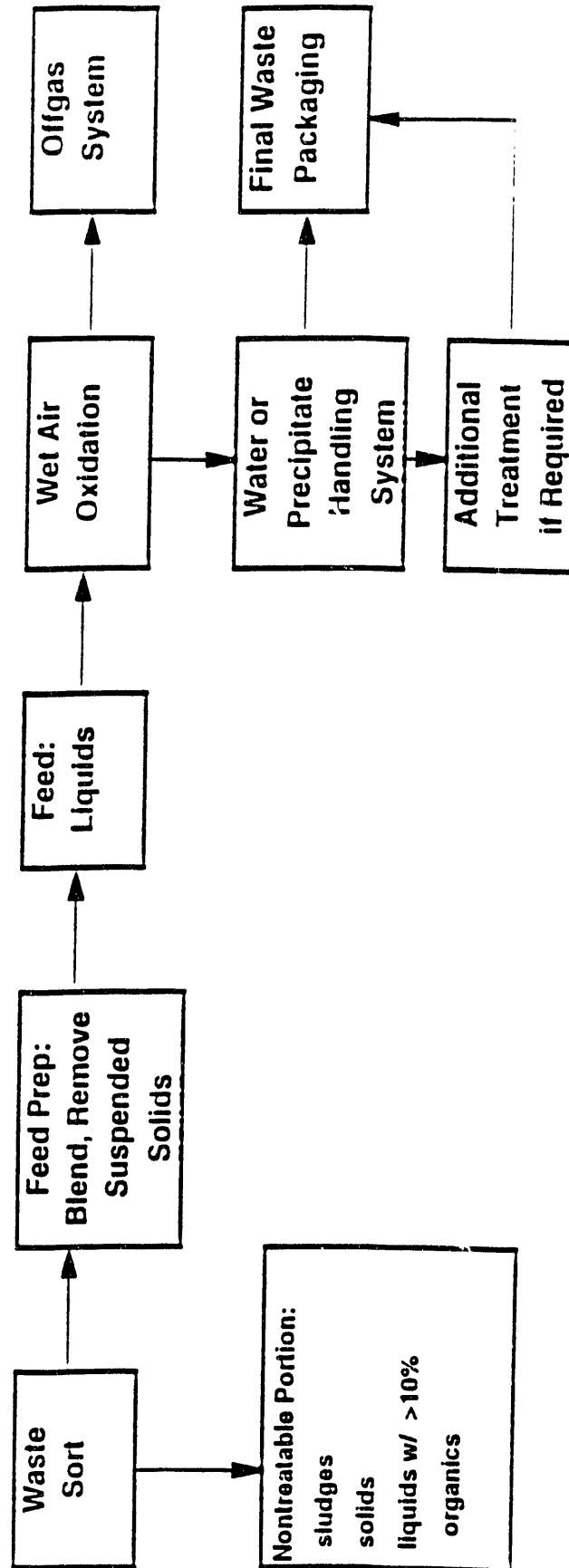
Gerald Rich and Kenneth Cherry, Hazardous Waste Treatment Technologies, Pudvan Publishing, Northbrook, IL, 1987.

J.P. Wilks et al., "Wet Oxidation of Mixed Organic and Inorganic Radioactive Sludge Wastes from Water Reactor", The 1989 Incineration Conference, Knoxville, TN, May 1989.

Contact: William Copa, ZIMPRO, Inc., Military Road, Rothschild, Wisconsin, 54474, (715) 359-7211.

Contact: Gerald C. Rappe, Vertech Treatment Systems, 12000 Pecos-Third Floor, Denver, CO 80234, (303) 452-8800.

WET AIR OXIDATION FUNCTIONAL PROCESS DIAGRAM



Technology Name: Steam Gasification Detoxifier

Maturity: Operational-Unique

Description:

A two-stage thermal process in which hydrocarbons are vaporized at 700°-1100°F in an autoclave, and then injected into a reaction chamber (detoxifier) with superheated steam where the organics are decomposed via steam hydrocarbon reforming chemistry. Typical detoxifier operating conditions are 2100°-3000°F at a slightly negative pressure. Organics can be vaporized in-drum, minimizing waste handling requirements, or by pumping from large tanks. Non-volatiles remain behind in the drum for subsequent disposal. The system consists of two boxes, evaporator and gasifier, which are small enough (4 ft x 6 ft x 7 ft) to be located inside many existing building spaces. All process monitors and controls are located inside these boxes, and the system is designed for automatic, hands-off operation. The offgas is processed through halogen absorbers, carbon absorbers, and catalytic carbon monoxide converters to remove metals, methane, carbon monoxide, hydrogen, and HCl, which are normal exhaust gas constituents from this process. The offgas from this process has potential value as a fuel gas. Process rates are 1-5 drums per 24-hour day. A number of these units were manufactured and sold.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Medium applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids. |
| Wet Solids: | Low applicability to sludges. |
| Dry Homogeneous Solids: | Not applicable. |
| Dry Heterogeneous Solids (Small): | Medium applicability to combustible solids and not applicable to noncombustibles. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

Steam gasification detoxifiers can achieve a high DRE. Their low offgas volumes are free of NO_x, SO_x, PICs, and particulate. extremely small process equipment; remote and automatic operations; low waste handling requirements.

Disadvantages:

Steam reforming detoxifiers are best for organic liquid wastes, but application for organically contaminated solids has also been demonstrated. The batch processing rate is 1-3 drums/day.

Development Needs:

Demonstration of continuous feed; offgas-catalyst improvement; oxyhydroxide formation with actinides; Hydrogen gas buildup; radioactive contamination/exposure provisions.

Vendor List:

Synthetica Technologies, Inc.

DOE Laboratories Involved in Technology:

Hanford
Savannah River Site
Sandia National Laboratory

References:

Terry R. Galloway, "Destroying Hazardous Waste On Site Avoiding Incineration", Environmental Progress, Vol. 8, No. 3, August 1989.

Charles A. Wentz and Terry R. Galloway, "Public Impact on Technical Research: The Dissimilar Fates of Two Waste Gasification Projects", Environmental Progress, Vol. 8, No. 3, August 1989.

T. R. Galloway, "Renew Carbon On-site by Steam Reforming", Chem Eng, December 1991, p. 11.

T. R. Galloway and F. Sidney Howard, "On-site Reactivation of Granular Carbon with the Synthetica Detoxifier", Annual AIChE Meeting, Los Angeles, November 17-22, 1991.

T. R. Galloway and Jerry L. Sprung, "Waste Destruction by Very High Temperature Steam Reforming", National Academy of Sciences/National Research Council, Committee on Potential Applications of Concentrated Solar Photons, Solar Energy Research Institute, November 7-8, 1990, Golden, Colorado.

T. R. Galloway, "Synthetica Detoxifier", The Hazardous Waste Consultant. McCoy & Associates, Colorado, November/December 1990.

T. R. Galloway, "The Need for New Technical Approaches to Environmental Control & Management", Annual Chemical Marketing Research Association Meeting, San Francisco. conference vol. pp. 13-16, February 5-7, 1990.

T. R. Galloway, "Destroying Hazardous Waste On-site -- Avoiding Incineration", Environmental Progress, 8, 176-185 (1989).

C. A. Wenz and T. R. Galloway, "Public Impact on Technical Research: The Dissimilar Fates of Two Waste Gasification Projects", 8, 186-189 (1989).

T. R. Galloway, "The Role of Steam in Lowering PICs in a Thermal Detoxifier", Annual AIChE Meeting, San Francisco, November 5-10, 1989.

T. R. Galloway, "The Destruction of Infectious Waste in the Thermolytica Detoxifier", Proceedings of the HazMat West 89 Conference and Exhibition, Long Beach. November 7-9, 1989.

T. R. Galloway, "Thermal Treatment with the Thermolytica Detoxifier", Chapter 8 in Book entitled: "Thermal Processes, Volume 1: Innovative Thermal Processes for Treating Hazardous Waste", pp. 77-93, Technomic Publishing Co., Lancaster, Pennsylvania, 1990.

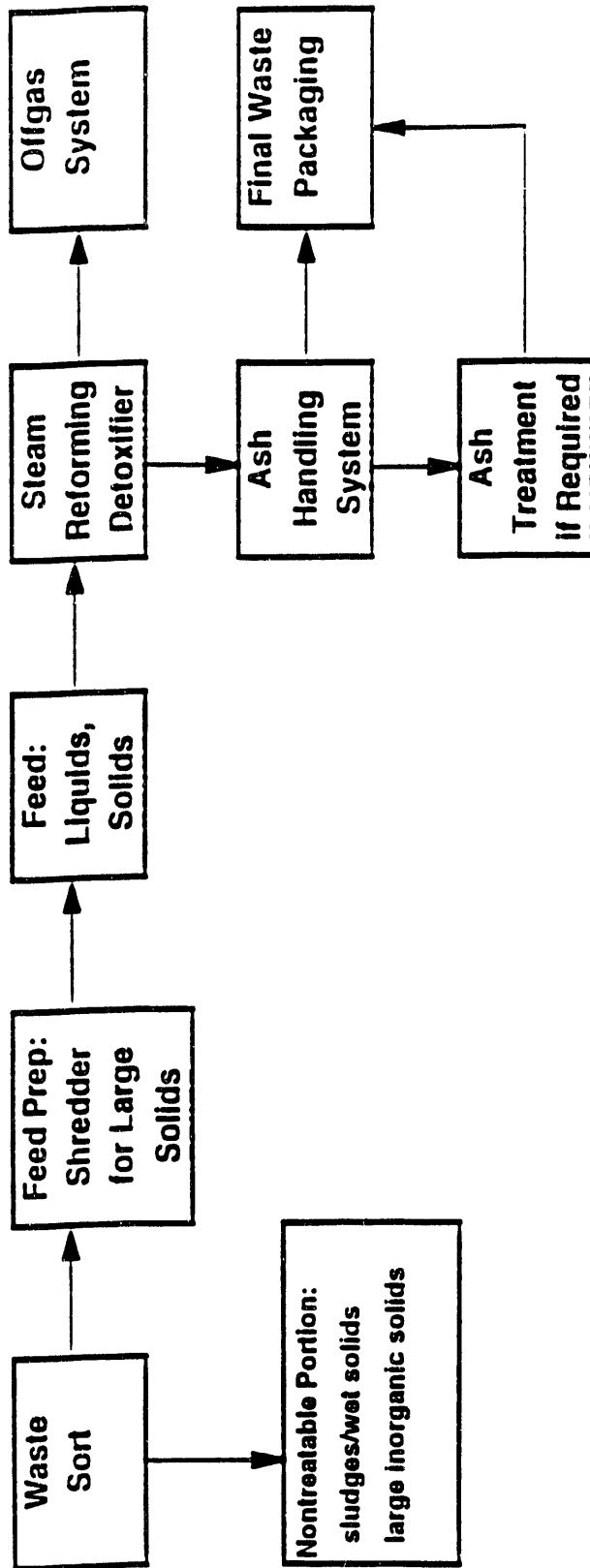
T. R. Galloway, "Destroying Hazardous Waste On-Site", AIChE Annual Meeting, Pub. Symposium Vol., Washington, D.C., November 27-December 2, 1988.

T. R. Galloway, "Thermolytica Detoxifier", The Hazardous Waste Consultant, McCoy & Associates, Colorado, May/June 1988.

T. R. Galloway, "Achieving Reduced Risk -- The Thermolytica Detoxifier Destroying Hazardous Waste On-Site", International Conference on Incineration of Hazardous/Radioactive Wastes, San Francisco, CA, May 3-6, 1988.

T. R. Galloway, "Economical On-Site Waste Detoxification: An Exercise in Heat Recovery", American Institute of Chemical Engineers Symposium Series No. 257, Vol. 83, pp. 418-424 (1987), Presented at 1987 National Heat Transfer Conference, Pittsburgh, PA, August 9-12, 1987.

STEAM GASIFICATION DETOXIFIER FUNCTIONAL PROCESS DIAGRAM



Technology Name: Supercritical Water Oxidation

Maturity: Emerging-Demonstration

Description:

The aqueous phase oxidation of dissolved or suspended organic contaminants at temperature and pressure conditions that are supercritical for water (above 705°F and 218 atm). Oxygen (air) and a dilute organic/water mixture are introduced into a reactor vessel where oxidation of the organics occurs. In supercritical water, oxygen and organics are totally miscible and oxidation proceeds rapidly and completely. Inorganic compounds are nearly insoluble and precipitate out. The process reduces the organics to H₂O, CO₂, and various biodegradable acids. Reaction times of less than one minute are required. The process, once started, is thermally self-sustaining, as well as providing a source of high-temperature process heat. One application employs a deep well and static head to generate supercritical pressures.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | High applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids with < 10% organics. |
| Wet Solids: | Not applicable to solids. |
| Dry Homogeneous Solids: | Not applicable to soils or other solids. |
| Dry Heterogeneous Solids (Small): | Not applicable. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

The super-critical water oxidation process is thermally self-sustaining. It is suited for processing non-incinerable dilute wastes. Its low off-gas volumes are free of NO_x, SO_x, PICs, and particulate. The super-critical water oxidation process can achieve complete oxidation of organics, and has a high DRE. Short (one minute) residence times allow a smaller reactor; hence, small equipment volumes are required. The process provides efficient precipitation of inorganics. Offgas scrubbing is not required. Process provides a source of high temperature process heat.

Disadvantages:

This technology has high cost and potential equipment limitations due to stringent temperature and pressure requirements. The technology is limited to weak aqueous organic solutions. There may be equipment fouling problems, especially with pumps fouling from particulate matter. Precipitated salts are difficult to remove. The supercritical water oxidation technology has not been demonstrated for solid content wastes.

Development Needs:

Materials of construction for high temperature/pressure conditions and abrasion problems; high pressure pumps which are not susceptible to fouling; corrosion control and monitoring; scale-up; solid effluent handling; investigate phase behavior-precipitation.

Vendor List:

ABB Lummus Crest
A. H. Halff Associates
Ecowaste

Genesyst Inc.
Modar Inc.
Modell Development Corp.

DOE Laboratories Involved in Technology:

Los Alamos National Laboratory
Sandia National Laboratory
NIST
Idaho National Engineering Laboratory
Rocky Flats Plant
Westinghouse Hanford Company

References:

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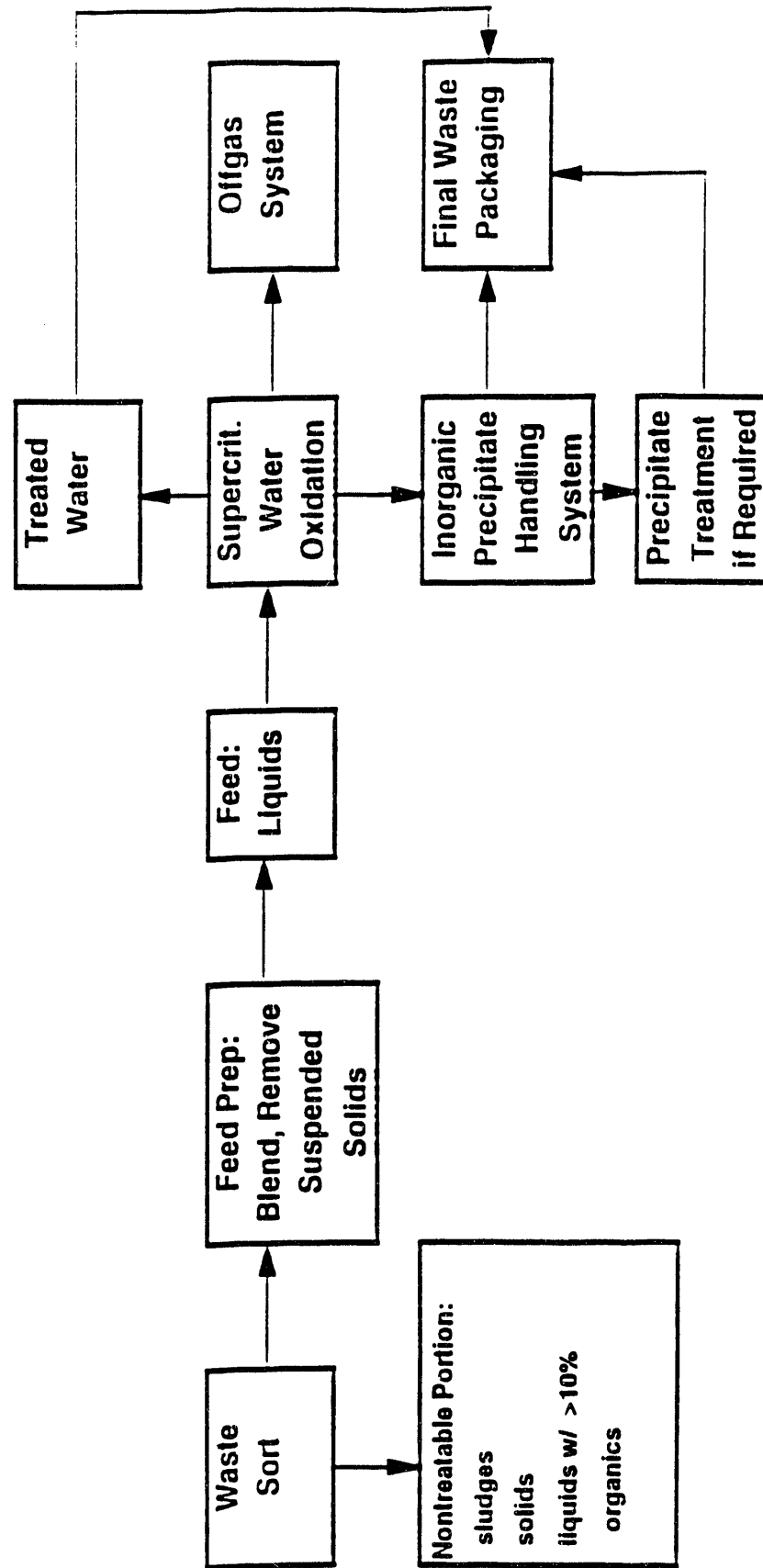
"Phase II Final Report: Oxidation of Hydrocarbons and Oxygenates in Supercritical

Water", HAZWRAP report No. DOE/HWP-90, September 1989.

Contact: Prof. Jeff Tester, Energy Laboratory, MIT, 77 Massachusetts Ave., Room E40-45, Cambridge, MA 02139-4307, (617) 253-3401.

Contact: Dr. Michael Modell, MODEC, 39 Loring Drive, Framingham, MA 01701, (508)820-09213.

SUPERCritical WATER OXIDATION FUNCTIONAL PROCESS DIAGRAM



Technology Name: Ultraviolet Photo-Oxidation

Maturity: Operational - Unique

Description:

Ultraviolet photo-oxidation (UVP) is a process that destroys or detoxifies hazardous chemicals in aqueous solutions utilizing UV radiation from various sources. UV radiation, ozone, and hydrogen peroxide combine to oxidize organic compounds including chlorinated hydrocarbons and aromatic compounds.

The UVP unit consists of a reactor module, air compressor/ozone generator module, and a hydrogen peroxide feed system. Offgas from the reactor passes through an ozone destruction (Decompozon) unit. The Decompozon unit destroys all gaseous volatile organic compounds stripped off in the reactor. UVP operation is based on the theory that adsorption of energy in the UV spectrum results in a molecule's elevation to a higher energy state, thus increasing the ease of bond cleavage and subsequent oxidation of the molecule.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | High applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids with <10% organics. |
| Wet Solids: | Not applicable to solids. |
| Dry Homogeneous Solids: | Not applicable to solids. |
| Dry Heterogeneous Solids (Small): | Not applicable to solids. |
| Dry Heterogeneous Solids (Large): | Not applicable to solids. |

Advantages:

UVP is skid-mounted, portable, and permits on-site treatment of a wide variety of liquid wastes. UVP is not a thermal technology, therefore it does not pose the risks or perception problems normally associated with thermal treatment. While UVP is effective at all concentrations, it does so without giving off any air emissions. The unit can be used as a stand alone, or combined with other treatment units in a system.

Disadvantages:

UVP is not a very versatile technology. The inability of UV light to penetrate and destroy pollutants in soil or in turbid or opaque solutions is a limitation to this approach. UVP is only capable of treating clear liquid wastes, and the reaction rate is dependent upon the pH of the input solution. During the process, the catalyst is susceptible to degradation and some harmless organics can produce competing reactions. Maintenance of UVP units is required on a routine basis.

Development Needs:

UVP is a fully developed technology and is widely available in the commercial market. There has been some history of failure in the heater element of the Decompozon unit, however this was a minor problem. Other areas in need of development include improved efficiency of the light source with respect to bandwidth; decreased catalyst degradation; decreased competing organic reaction; less dependency on pH with respect to reaction rate; and a more in-depth look at large-scale operations/economics.

Vendor List:

Artech Incorporated
DeGussa
ECOVA
Kerr McGee
Peroxidation Systems
Syntex Chemicals
Ultrox International

DOE Laboratories Involved in Technology:

Sandia National Laboratory
Lawrence Livermore National Laboratory
DOE Kansas City Plant

References:

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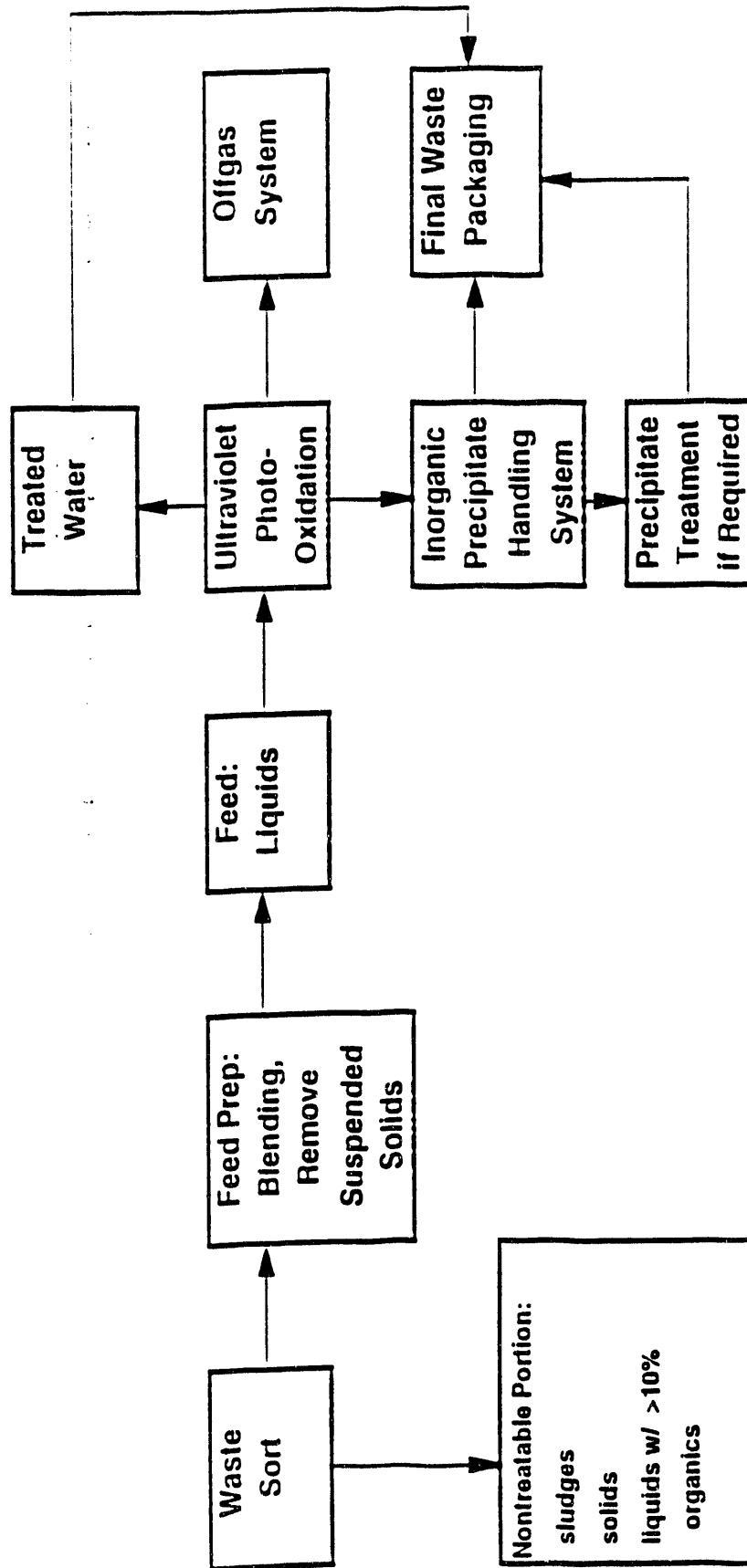
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EPA, EPA/625/8-87/014, September 1987.

"Ultrox International Ultraviolet Radiation/Oxidation Technology", US EPA, EPA/540/A5-89/012, September 1990.

ULTRAVIOLET PHOTO-OXIDATION FUNCTIONAL PROCESS DIAGRAM



Technology Name: Plasma Pyrolysis Reactor

Maturity: Emerging-Demonstration

Description:

A horizontal reactor chamber in which liquid waste molecules are pyrolyzed by passing through a thermal plasma plume. The plume is generated by passing an electric charge through an atmospheric airstream which ionizes the gas molecules and generates temperatures up to 18,000°F. The collinear electrodes of the plasma device act as a plug-flow atomization zone for the liquid waste feed, and the pyrolysis chamber serves as a mixing zone where the atoms recombine to form H₂, CO, HCl, and particulate C. Residence times in the residence zone and recombination zone are 500 microseconds and one second, respectively. Temperature in the recombination zone is maintained at 1200°-2400°C. After offgas scrubbing, the residual gases are electrically ignited in a flare stack.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids. |
| Wet Solids: | Not applicable to absorbed liquids or sludges with organics. |
| Dry Homogeneous Solids: | Not applicable to solids. |
| Dry Heterogeneous Solids (Small): | Not applicable. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

The small equipment size required for this technology allows for portability; minimal setup is required after delivery to new site. The plasma pyrolysis reactor technology has a high throughput and can process highly toxic and refractory compounds, as well as wastes with low heating values. rapid on/off cycle times; high DREs; high destruction temperatures; produces a fuel gas for energy recovery.

Disadvantages:

The technology can treat only liquids with light particulate loading. Plasma pyrolysis reactors are energy intensive to operate.

Development Needs:

Limited long term operational data; electrode life uncertainties, significantly affects peak electrical use (peak charge may increase); heating value limits of waste streams; power needs v.s. feed properties.

Vendor List:

Pyrolysis Systems Inc.
Westinghouse Research & Development Center

DOE Laboratories Involved in Technology:

None

References:

Marc Breton et al., "Technical Resource Document: Treatment Technologies for Solvent Containing Wastes", U.S. Environmental Protection Agency, EPA/600/2-86/90, October 1986.

Harry Freeman, "Innovative Thermal Hazardous Organic Waste Treatment", Noyes Publications, Park Ridge, NJ, 1985.

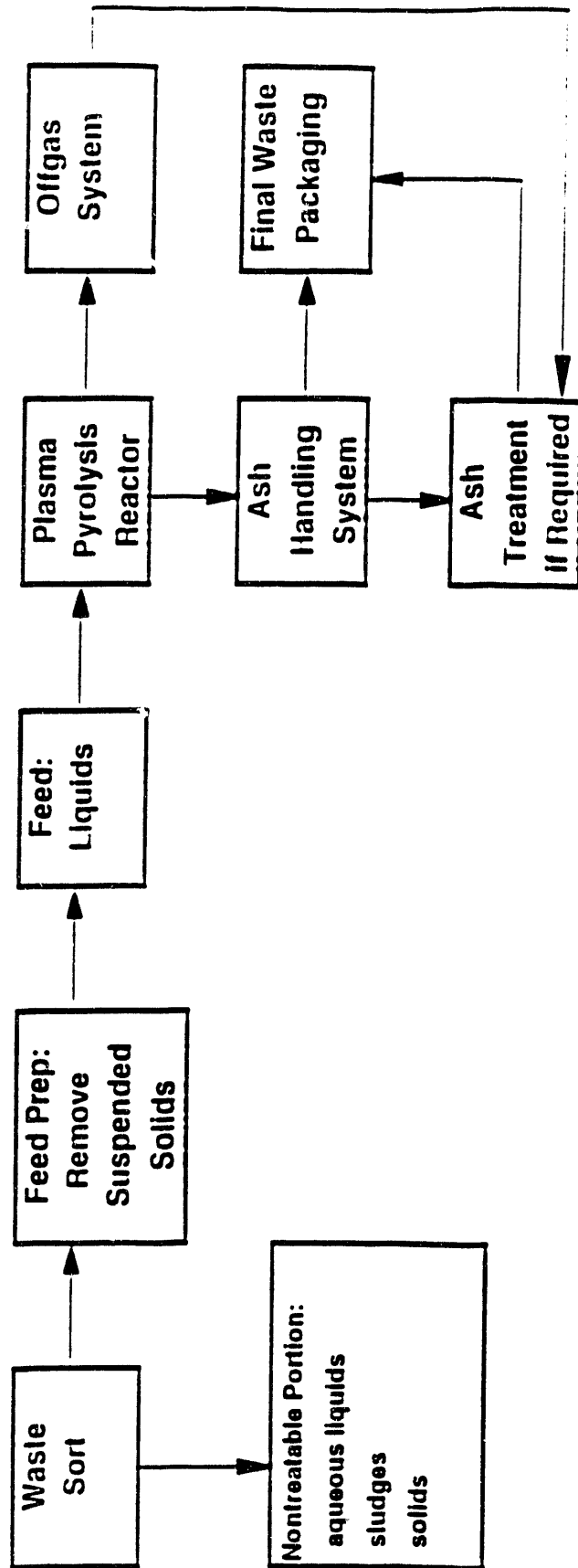
Michael F. Joseph, Thomas G. Barton, "Waste Destruction by Plasma Arc Pyrolysis", Pyrolysis Systems Inc., Kingston, Ontario, Canada.

The Hazardous Waste Consultant, McCoy & Associates, Vol. 4, Issue 3, May/June 1986.

Contact: E.S. Fox, Jr., Pyrolysis Systems, Inc., 61 Thorold Road, Welland, Ontario, L3B 5P1, Canada, (416) 735-2401.

Contact: Westinghouse Plasma Systems, P.O. Box 350, Madison, PA 15663, (412) 722-5275.

PLASMA PYROLYSIS REACTOR FUNCTIONAL PROCESS DIAGRAM



APPENDIX C3

MELTER TECHNOLOGIES

Technology Name: Molten Salt Furnace

Maturity: Emerging-Pilot/Emerging-Demonstration

Description:

In the molten salt (MS) process, waste and air are continuously introduced beneath the surface of a sodium carbonate (Na_2CO_3) melt at a temperature of 750° to 1000°C . Supplemental fuel may be required if the waste is not sufficiently combustible. Rapid destruction of the waste results from the catalytic effect of the salt, and from the intimate contact of the waste with air and the hot molten salt, which provides rapid transfer of heat to the waste. The molten salt forms chemical complexes with toxic metals and radionuclides which reduces their thermodynamic activity and thus retains them in the salt. Sodium carbonate is used because it prevents emission of acidic gasses, such as HCl (ordinarily produced from organic chloride compounds) and SO_2 (from organic sulfur compounds). Also, it is stable, nonvolatile, inexpensive, and nontoxic. The carbon and hydrogen of the waste are converted to CO_2 and steam; halogens form their corresponding sodium halide salts; phosphorus, sulfur, arsenic, and silicon (from glass or ash in waste) form oxygenated salts; and the iron from metal containers forms iron oxide. The ash is trapped in the melt. The melt is removed periodically or batch-wise to prevent excessive build-up of halide salts or ash. The ash can be separated from the salt in an aqueous separations process with the sulfates and chlorides scrubbed out and the carbonates recycled to the melt. The CO_2 and water can be captured and stored in liquid form to be analyzed prior to release.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Low applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids. |
| Wet Solids: | Not applicable to solids. |
| Dry Homogeneous Solids: | Not applicable. |
| Dry Heterogeneous Solids (Small): | High applicability to combustible dry solids only. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

High waste destruction efficiency, high heat transfer rates. Liquid waste effluent is not produced. The molten salt combustor is versatile, handling a wide variety of wastes. Excellent temperature control may be maintained due to the thermal inertia of melt bed. Acid gases are not produced, nor are they emitted. The radioactive elements of heavy metals are retained in the salt. Potentially good public acceptance.

Disadvantages:

High ash waste requires greater salt make-up than liquid wastes (e.g. solvents), and salt/ash separation is difficult. Feedstock must be size-reduced, as large forms, e.g. 55 gal-drum, cannot be accepted. The molten salt is corrosive to most metals. The system complexity is high because of salt recycling needs to make the process cost-effective.

Development Needs:

Performance of materials of construction over range of salt chemical compositions and temperatures. Develop process for treatment of spent melt, e.g. process to separate ash from salt; develop process to recover radioactive elements or heavy metals from salt.

Vendor List:

Rockwell International

DOE Laboratories Involved in Technology:

Lawrence Livermore National Laboratory
Los Alamos National Laboratory
Hanford (using zinc chloride salts)

References:

Marc Breton et al, "Technical Resource Document: Treatment Technologies for Solvent Containing Wastes", U.S. Environmental Protection Agency, EPA/600/2-86/095, October 1986.

H.M. Freeman et al., "Thermal Destruction of Hazardous Waste A State-of-the-Art Review", Journal of Hazardous Materials, Vol. 14, 1987, pp. 103-117.

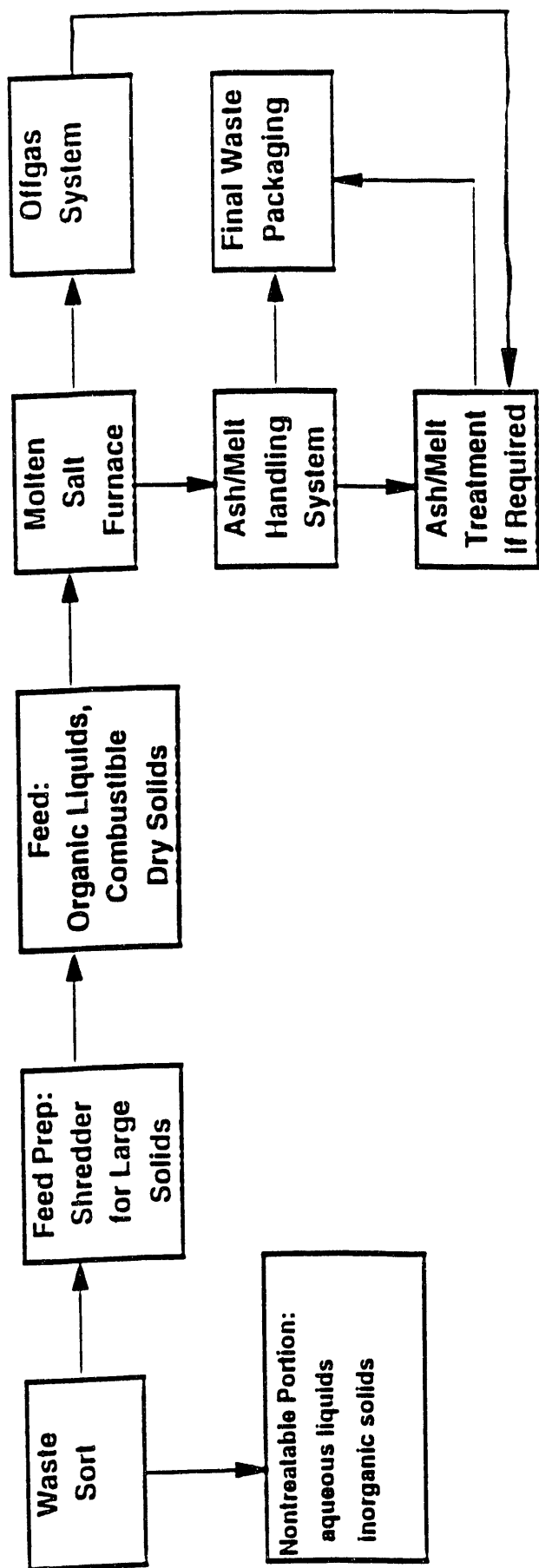
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R.L. Gay et al., "Destruction of Toxic Wastes Using Molten Salts". Technical Meeting of the American Institute of Chemical Engineers. Anaheim, CA, April 1981.

J.G. Johanson et al., "Destruction of Hazardous Wastes by the Molten Salt Destruction Process", Seminar of the American Society of Testing Materials Committee D-27, Nashville, TN, March 1982.

Contact: Richard L. Gay, Rocketdyne Division, Rockwell International Corp., 6633 Canoga Ave., Canoga Park, CA 91303, (818) 700-3505.

MOLTEN SALT FURNACE FUNCTIONAL PROCESS DIAGRAM



Technology Name: Joule Heated Melter

Maturity: Operational on Mixed High-Level Radioactive Waste

Description:

A refractory lined reactor in which a pool of glass is initially melted by auxiliary heating, then maintained in a molten state by joule heating (alternating electric current passing through the glass between submerged electrodes dissipates energy due to bulk glass resistivity). The technology described here is distinguished from the High-Temperature Joule Melter described later by its nominal operating temperature of 1200°C or less. This class of process equipment includes a broad range of designs. It is the base technology for vitrifying high-level radioactive waste at Savannah River's Defense Waste Processing Facility, the West Valley Vitrification Facility and the Hanford Waste Vitrification Plant. This general technology has been deployed internationally and operated under remote radioactive conditions for over six years (1985 to 1991) in the PAMELA plant at Mol, Belgium. High level mixed wastes are typically fed in a slurry form to facilitate transfer of waste to the process. Glass formers or premelted glass is mixed in with the waste to provide the silica and fluxes needed to melt at the operating temperature limit of 1200°C.

For non-slurried waste applications, waste is introduced into the furnace above the molten glass pool along with the combustion air. Combustion is achieved by exposure to the radiant heat above the pool or by contact with the molten glass. Exhaust gases flow out the opposite end of the furnace. Solid products of combustion and noncombustible materials are encapsulated in the glass, which can be continuously removed or batch discharged to solidify into a nonleachable matrix. A feeding variation by one developer introduces the waste and air under the surface of the molten glass via a drop tube to confine most of the combustion below the surface of the pool, enhancing intermixing of the waste and combustion gases with the glass and attaining higher particulate retention. Typical mean glass residence times range from 24 to 48 hours. This assures homogeneity of the glass material being discharged even with variations in the waste stream.

Waste Applicability:

- | | |
|------------------|---|
| Aqueous liquids: | Medium applicability to aqueous liquids. The technology can and has been fed dilute liquid waste streams where evaporation and vitrification of the residue occurs. High level waste may have at least 40 wt% solids with the balance being liquid. |
| Organic liquids: | High applicability to organic liquids. Organics present at up to 100 grams/liter in high level waste have been destroyed with high destruction efficiencies (>99.99%). For |

strictly organic hazardous wastes, destruction efficiencies in excess 99.999% have been demonstrated at Mound and Pacific Northwest Laboratory.

- Wet Solids:** High applicability to wet solids. This report's definition of wet solids is consistent with the primary application of this technology, the processing of waste slurries and sludges.
- Dry Homogeneous Solids:** High applicability to dry homogeneous solids. Some size reduction may be required to facilitate the feeding of the unit but processing of dry solids is an adaptation of the conventional glass industry use of this technology.
- Dry Heterogeneous (Small and Large):** This technology is judged not to be applicable to Solids heterogenous solids because of the presumed metal content. Metals will precipitate to the floor of the melter, not be dissolved, and ultimately lead to electrical shorting between the power electrodes.

Advantages:

This adaptation of the glass industry technology has been thoroughly tested for slurries and sludges typical of high-level, mixed wastes. The operating conditions for successfully producing a chemically durable product is well documented. The ability to destroy organics has been demonstrated in melters that employ plenum heaters. At the prescribed operating temperatures ($<1200^{\circ}\text{C}$), a broader spectrum of electrodes and glass contact refractory can be used. A long reliable operating life, in excess of two to five years, should be expected without failure. This technology has been designed for totally remote operation. For mixed wastes that pose a significant chemical or radioactive hazard during operations, these designs can be confidently employed. Use of joule heating minimizes possible safety issues regarding combustible gases for radioactive application.

Disadvantages:

The relatively low operating temperature of 1200°C limits the waste loading in the product glass. For high-level waste a loading of 25 to 35 wt% of wastes is typical. For contaminated soils or similar compositions, 60 to 80 wt% waste loading may be more typical. However, the relatively high density of the glass (2.5 to 2.8 gm/cc) may result in high volume reduction. The operating temperature essentially precludes the opportunity to process high metal containing waste streams. Here, the metals can settle to the floor, collect and cause an electrical short between the power electrodes. The capital cost for these high-level waste melters is relatively high and operating life may be limited to several years. Process control necessitates feed characterization and periodic sampling and analysis of the glass stream. In general, the joule heated melter is more suitable to

feed streams where there is a substantial amount of relatively homogeneous feed. The joule melter can be idled at a lower temperature, but must be drained of glass if a complete shutdown is required.

Development Needs:

The specifics of the waste stream need to be defined and an acceptable glass needs to be tailored for its processing. After laboratory development of these waste glasses, demonstration of the technology with the specific waste stream(s) is needed to quantify the specific throughput and to identify any unforeseen issues. Joule heated melters are available within the DOE Complex and industry, so that demonstrations can be performed without large capital investments. However, testing may require modification of melter feed systems.

Vendor List:

American Environmental Management Corp.
Frazier-Simplex Inc.
Penberthy Electromelt International, Inc.
Recomp, Inc.
Sorg Engineering
Toledo Engineering Co., Inc

DOE Laboratories Involved in Technology:

Pacific Northwest Laboratory
Savannah River Laboratory
Mound Laboratory

References:

Janke, 1990, D.S. Janke and C. C. Chapman, Characteristics of Fernalds K-65 Residue Before, During and After Vitrification, FMPC/SUB-035, Pacific Northwest Laboratory, Richland, WA.

Brouns, 1988, R. Brouns, A. A. Balasco, et. al., Bench Scale Glassification Test on Rocky Mountain Arsenal Basin F Material, AMXTH-TE-CR-88015, Pacific Northwest Laboratory, Richland, WA.

Klinger, 1989, Joule Heated Glass Furnace Processing of a Highly Aqueous Hazardous Waste Stream, L. M. Klinger, P. L. Abellera, March 17, 1989, MLM-3577, Mound Applied Technologies, Miamisburg, OH.

Larson, 1983, Assessment of power reactor waste immobilization by vitrification, D. E.

Larson, et. al., August, 1983, EPRI-NP-3225, Pacific Northwest Laboratories, Richland, WA.

Klinger, 1985, Glass Furnace Project Final Report: An Evaluation of Operating Experience for Low-Level Nuclear Waste Processing, L. M. Klinger, K. M. Armstrong, February, 28, 1985, MLM-3229, Mound Laboratories, Miamisburg, OH.

Armstrong, 1985, Nitrate Waste Processing by means of a Joule-Heated Glass Furnace, K. M. Armstrong, L. M. Klinger, October 18, 1985, MLM-3304, Mound Laboratories, Miamisburg, OH.

Klinger, 1988, Glass Furnace Processing of Rocky Flats Plant Wastes - an Evaluation, L.M. Klinger, P.L. Abellera, April 29, 1988, MLM-3493, Mound Laboratories, Miamisburg, OH.

Wiese, 1988, Industrial Vitrification of High Level Liquid Waste with the PAMELA Plant in Belgium, Wiese, H. and E. Ewest, Proceedings of the International Topical Meeting on Nuclear and Hazardous Waste Management, Spectrum '88, September, 1988, pp 75-77.

Barnes S.M., J.M. Pope and C. C. Chapman, Three Year's Progress of the West Valley Demonstration Project Vitrification System, March 1988, Waste Management '88, Tucson, Arizona, Waste Management '88 Proceedings

Brouns, R. A. and M. S. Hanson, 1984, "The Nuclear Waste Glass Melter--An Update of Technical Progress", Fuel Processing and Waste Management, Vol. 1, p. 101. American Nuclear Society, Inc., La Grange Park, Illinois.

Buelt, J. L. 1985, A Mobile Encapsulation and Volume Reduction System for West Low-Level Wastes. PNL-5533, Pacific Northwest Laboratory, Richland, Washington.

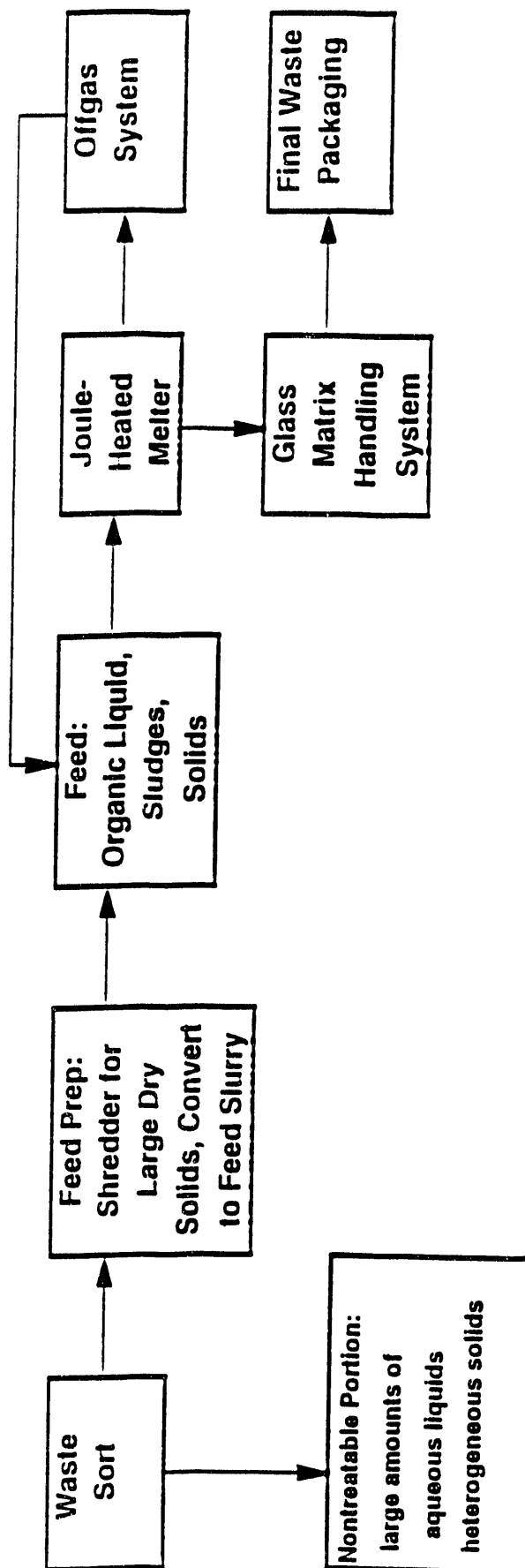
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Holton, L. K., Jr., D. N. Berger, W. J. Bjorklund, and R. D. Dierks, 1984, "Design Features of a Radioactive Liquid-Fed Ceramic Melter System", Transactions of the American Nuclear Society, Vol. 46, p. 785. American Nuclear Society, Inc., La Grange Park, Illinois.

McElroy, J. L., W. J. Bjorklund, and W. F. Bonner, 1982, "Waste Vitrification: A Historical Perspective", The Treatment and Handling of Radioactive Wastes, A. G. Blasewitz, J. M. Davis, and M. R. Smith, Editors, p. 171. Battelle Press, Richland, Washington.

JOULE-HEATED MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: Plasma - Arc Furnace

Maturity: Emerging-Pilot

Description:

A plasma arc furnace uses the energy from a thermal plasma arc, generated by joule heating of a gaseous electrical conductor between two high voltage electrodes, to combust organics and melt inert waste components. The plasma arc is generated within the furnace primary chamber by a removable plasma torch. Two types of plasma torches are available; transferred and non-transferred arc. The transferred arc uses a conductive hearth for maintaining the voltage differential for arc generation. In the non-transferred plasma torch, the arc is generated and maintained within the torch body. Waste is introduced into the furnace into a molten bath of material, which could be inert waste or other material. The high temperature plasma zone and the molten bath (in excess of 3000°F) combust (or pyrolyze) the organics and melt all other inert materials into the bath. Volatile organics are further treated in a secondary combustion chamber. Very small gas volumes are required for the plasma arc, resulting in low offgas volumes. Molten solid material can be removed continuously by overflow or poured by batch and forms a leach-resistant, vitrified (glassy) waste form. Furnace operation is similar to a dual chamber controlled air incinerator with the substitution of a plasma arc torch for a burner in the primary chamber. The plasma arc furnace can reprocess all of its byproducts such as flyash, filters and scrubber residues.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Medium applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids. |
| Wet Solids: | High applicability to absorbed liquids and sludges with organics. |
| Dry Homogeneous Solids: | High applicability. |
| Dry Heterogeneous Solids (Small): | High applicability. |
| Dry Heterogeneous Solids (Large): | High applicability. |

Advantages:

Solid byproduct is a vitrified "glassy" slag that is excellent for stabilization of toxic metals and radionuclides. Quiescent combustion in primary chamber results in reduced particulate emissions. Plasma energy assists carbon burnout. Reduced offgas volume decreases air pollution control equipment costs. Byproducts such as flyash, filters, and scrubber residue can be reprocessed through the furnace. The process requires minimal waste characterization and pretreatment and has been demonstrated on an industrial scale for metal production. The furnace can be shut down and restarted with little difficulty. It can operate with a skull to reduce reaction of the melt with the metal or refractory hearth. Use of an electrical plasma avoids possible safety issues regarding combustible gases for radioactive application. This is an emerging technology for a variety of waste types.

Disadvantages:

The non-transferred arc allows operation over a wide range of power settings and has relatively simple controls, however, operation, start up, and control of a transferred arc plasma furnace is more complex than conventional incineration. If high temperatures are maintained, this may lead to high NO_x levels and increased volatilization of heavy metals.

Development Needs:

Optimization of slag chemistry for metals stabilization; evaluation of variation in slag chemistry resulting from variations in the input stream; re-introduction of condensed volatile metals into slag phase; electrode life studies; DRE of hazardous organics; safety assessments for heterogeneous waste processing; determination of radionuclide partitioning in slag/metal phases.

Vendor List:

ABB
Plasma Energy Corp.
Retech Inc.
Westinghouse

DOE Laboratories Involved in Technology:

Idaho National Engineering Laboratory
Pacific Northwest Laboratories
Westinghouse Hanford Company

References:

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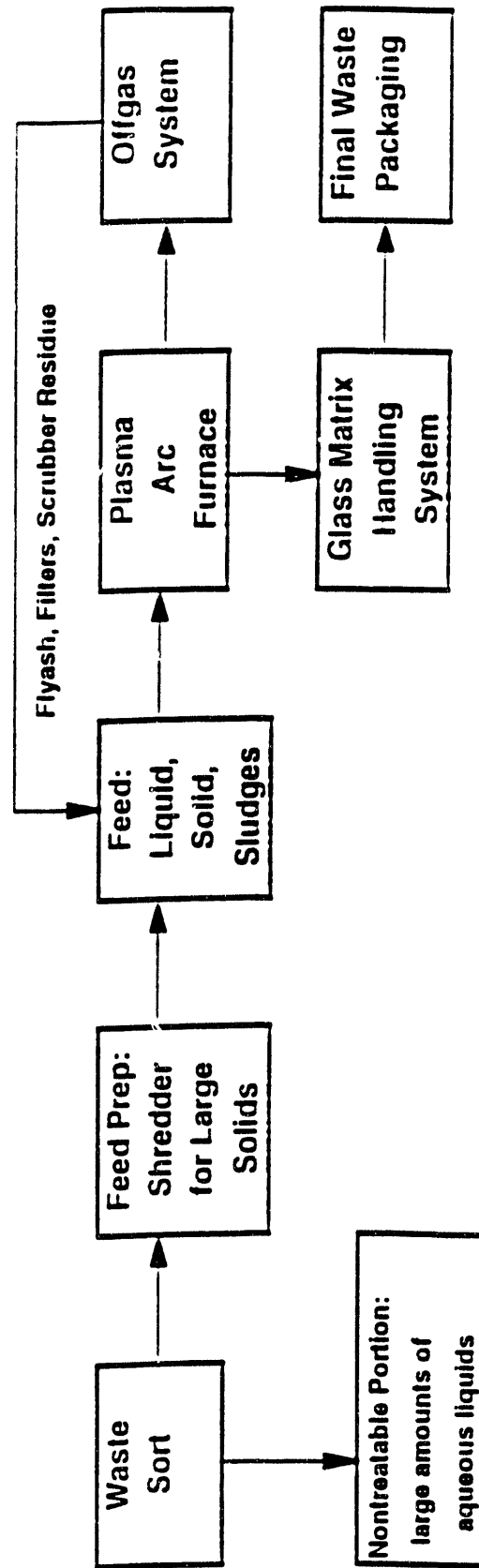
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C.R. Brunner, "Incineration Systems-Selection and Design", Incinerator Consultants Incorporated, Reston, VA 1988.

J.A. Batdorf, private communication, September 1990, Haz Answers, Inc., Idaho Falls, ID.

R. D. Peters and W. A. Ross, "Plasma Melting of Non-Irradiated Fuel Assembly Hardware: Initial Testing and Evaluation", Radioactive Waste Management and the Nuclear Fuel Cycle, 1989, Vol. II(4), pp. 333-345.

PLASMA ARC FURNACE FUNCTIONAL PROCESS DIAGRAM



Technology Name: Microwave Melter

Maturity: Demonstration

Description:

This process utilizes microwave energy for in-container solidification/stabilization of radioactively contaminated non-organic wastes such as incinerator ash, sludges, or soils. Waste moisture is removed in a belt-driven microwave dryer prior to treatment. The dry waste materials are vitrified inside a metal disposal container in either a batch or continuous feed mode. Melt temperatures range from 1800°-2600°F and the resulting product is a glassy monolith that meets radioactive disposal criteria for liquid and particulate content, and RCRA LDR requirements for leaching of toxic hazardous constituents. The process results in volume reductions on the order of 80% with waste loadings on the order of 60%.

Waste Applicability:

| | |
|-----------------------------------|--|
| Aqueous Liquids: | Medium applicability to aqueous liquids. |
| Organic Liquids: | Medium applicability to organic liquids with a secondary combustion system added (not currently part of system). |
| Wet Solids: | High applicability to absorbed liquids and sludges with organics. |
| Dry Homogeneous Solids: | High applicability. |
| Dry Heterogeneous Solids (Small): | Not applicable to heterogeneous wastes. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

Direct application of energy to the wastes - surrounding equipment remains relatively cool; process occurs inside disposal container, minimizing waste handling.

The waste form will meet applicable waste acceptance criteria for the disposal facilities; equipment is inexpensive and easy to maintain; process requires short heating time to achieve operational temperature (on the order of 30 minutes); heating can be instantaneously interrupted; heating is uniform in the waste material; energy can be selectively directed to the waste and not the equipment, preventing thermal cycling of the

equipment; the waste form is processed "in drum," reducing the material handling and generation of additional waste: "in-drum" processing eliminates the requirement of producing a pourable, low viscosity melt; waste volumes are reduced up to 80% compared to current cementation processes.

Disadvantages:

Applicable to dry or near-dry non-organic wastes only; waste must be relatively homogeneous fines; low throughput results in high unit operating costs; uneven melting of the wastes, especially near the bottom and sides; some oxidation of the metal waste container; melt-through of the container at hot spots.

Development Needs:

Evaluation of the process for other applications, i.e. destruction of hazardous wastes; leachability of the vitrified waste; develop dielectric property models; study container corrosion problems; control heat profiles in small heterogeneous wastes; perform volatility studies on liquid organics.

Vendor List:

Japanese manufacturers

DOE Laboratories Involved in Technology:

Rocky Flats Plant
Oak Ridge National Laboratory
Oak Ridge K-25 Plant
Los Alamos National Laboratory

References:

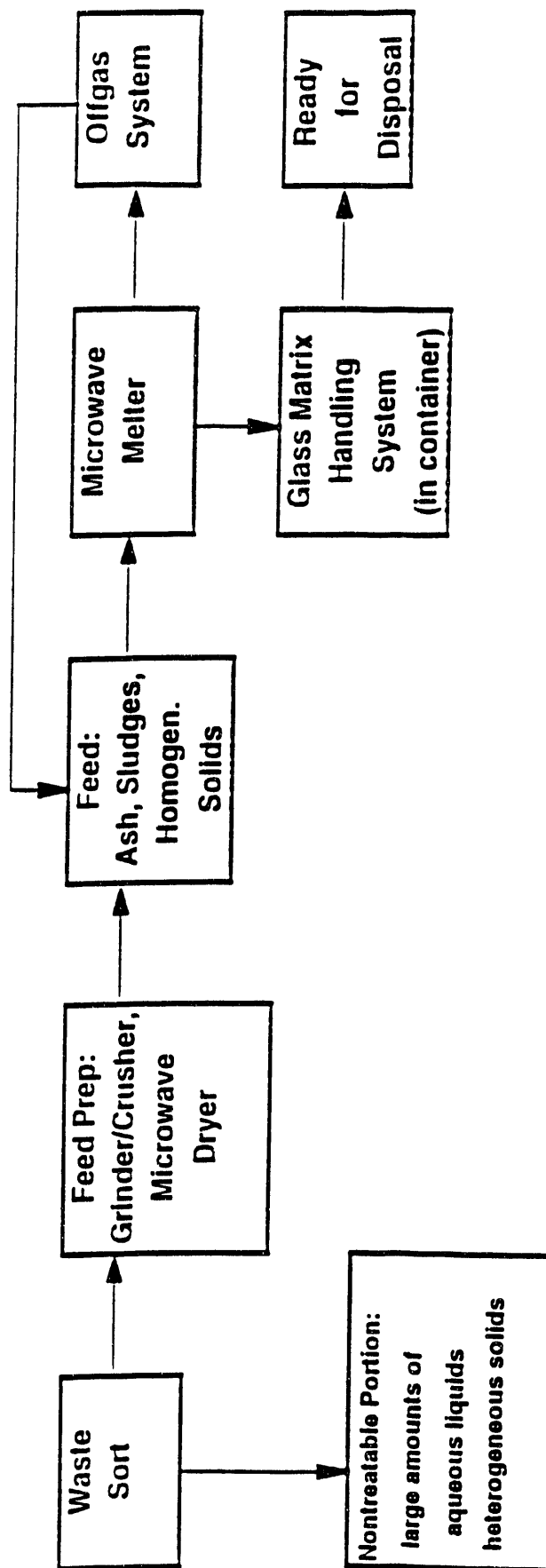
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- F. Komatsu et al., "Application of Microwave Treatment Technology to Radioactive Waste", Proceedings of the 1989 Incineration Conference, Knoxville, TN.
- R.D. Petersen, "Microwave Vitrification of Rocky Flats TRU Sludge", American Nuclear Society 1989 Winter Meeting, San Francisco, CA, November 1989.

C-80

Contact: Robert D. Peterson, EG&G Rocky Flats, Rocky Flats Plant, Denver, CO (303) 966-4051.

Contact: Greg Sprenger, EG&G Rocky Flats, Rocky Flats Plant, P.O. Box 464, Golden, CO 80402-464, (303) 955-3159

MICROWAVE MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: Slagging Kiln

Maturity: Operational - Unique

Description:

A slagging kiln is an incinerator designed to operate at sufficiently high temperatures so that the entire charge of waste material essentially melts into a "slag". Almost all slagging kilns are based on an improved rotary kiln incinerator design (see "Rotary Kiln" entry for details), requiring more attention to the refractory lining and the slag handling equipment. Other designs are possible, however, as evidenced in one particular application of a German-designed multi-chamber slagging kiln. Slagging kilns normally require a secondary combustion chamber to assure complete destruction of hazardous constituents. The primary chamber functions to combust solid waste to gases at temperatures of 2000° to 2200° F, thus leaving a melted slag residue of the noncombustible components (i.e. alumina and silica compounds, metal, glass). The slag melt progresses through the kiln into a water quench, where it solidifies and fractures into small pieces, and is then drawn from the process. Both primary and secondary chambers are generally supplied with auxiliary fuel systems which can be used for liquid waste incineration. An extensive offgas system is generally required to control the high volume of emissions. Slagging kilns are generally used in applications involving high calorific value type wastes.

Waste Applicability:

| | |
|--------------------------------------|---|
| Aqueous Liquids | Medium applicability to aqueous liquids. |
| Organic Liquids: | High applicability to organic liquids. |
| Wet Solids: organics. | High applicability to absorbed liquids and sludges with organics. |
| Dry Homogeneous Solids: | High applicability. |
| Dry Heterogeneous Solids (Small): | High applicability to heterogeneous wastes. |
| Dry Heterogeneous Solids (Large): | High applicability. |

Advantages:

Can handle a wide variety of solid, liquid and sludge waste types; can accept whole

metal drums of waste without breaching or shredding; slag is removed continuously and does not interfere with waste oxidation; when operated at very high temperatures leads to more complete burning and better destruction of difficult to destroy compounds; reduced offgas particulate loading due to adsorption into the slag; lower excess air requirements.

Disadvantages:

High capital cost for installation; spherical or cylindrical objects may roll through the kiln before complete combustion; need to replace the refractory lining more often; higher temperatures increase probability of volatilizing heavy metals; not efficient for low calorific wastes; cannot be thermally cycled often (shutdown/startup cycle); feed composition must be tightly controlled; maintaining seals difficult; large volumes of air required for combustion give rise to large, costly, and difficult to operate offgas treatment systems.

Development Needs:

Better kiln seal design, slag chemistry, advanced offgas systems, stack monitoring and other real-time performance assurance capabilities; control of heavy metal emissions; combustion by-product formation; sub-micron particulate emissions.

Vendor List:

Allis Chalmers
Combustion Engineering Co.
Ford, Bacon, and Davis
Rollins Environmental Services
Von Roll, Ltd (Switzerland)
John Zink Co.

DOE Laboratories Involved in Technology:**References:**

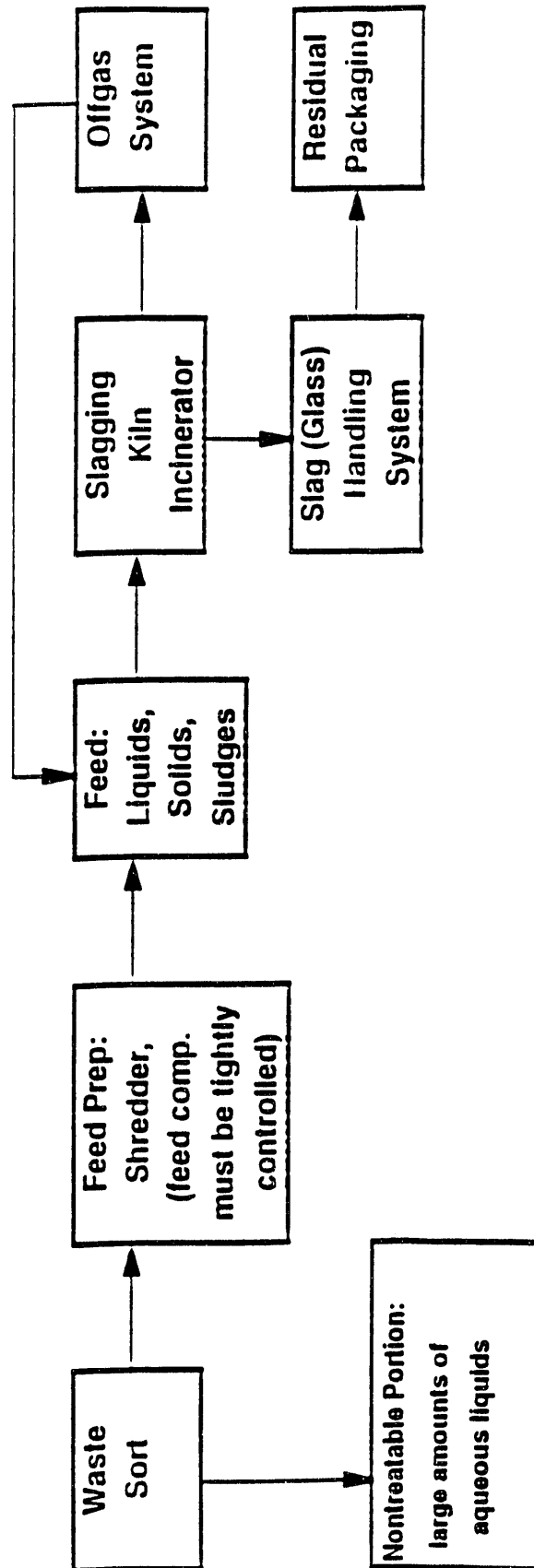
D.A. Tillman et al., "Rotary Incineration Systems for Solid Hazardous Wastes". Chemical Engineering Progress, July 1990.

P.W. Falcone and R.J. Buchanan, "Hazardous Waste Incineration by Slagging - Mode Rotary Kiln", 20th Annual Mid-Atlantic Industrial Waste Conference, Washington, D.C., June 1988.

N. Van de Voarde et al., "High Temperature Incineration of Radioactive Waste", Nuclear Science and Technology, Commission of the European Communities, 1986.

N. Van de Voarde et al., "High Temperature Slagging Incineration -Recent Operating Experience", Spectrum '86. American Nuclear Society International Topic Meeting, Niagara Falls, NY, 1986.

SLAGGING KILN INCINERATOR FUNCTIONAL PROCESS DIAGRAM



Technology Name: Electric Furnace Melter

Maturity: Operational - Conventional (has not been demonstrated on waste processing)

Description:

Electric furnaces have been used as smelters in the steel industry for several years. The electric furnace melter uses graphite electrodes to melt inorganic waste components into a glassy slag and pyrolyze or combust organic waste components. The electrodes may be submerged in the molten bath where the resistance to the electrical current passing between the electrodes creates the temperatures necessary to melt the material, or the electrodes may remain above the surface of the bath, creating an arc plasma zone of high temperatures. Temperatures of 1650°C are routinely maintained within the furnace chamber and higher temperatures are achievable. Waste can be fed to the furnace through chutes, hollow electrodes, or a series of doors which form an airlock. Depending on the type of feeding system used and the size of the waste, some size reduction may be necessary. It may also be beneficial to pretreat the waste with a fluxing agent, such as lime.

Waste Applicability:

Aqueous Liquids: Medium applicability, although there is a concern of a steam explosion if liquids get below the melt surface.

Organic Liquids: High applicability. The chamber temperatures are typically 550°C higher than conventional incinerators.

Wet Solids: High applicability.

Dry Homogeneous Solids: High applicability.

Dry Heterogeneous Solids (Small): High applicability.

Dry Heterogeneous Solids (Large): High applicability.

Advantages:

This technology can handle a wide variety of waste streams, such as organics, inorganics, and bulk metals. As with all melter technologies, a leach-resistant final waste

form is generated. In addition, the high temperatures should provide excellent destruction of organics. As with all electrically-heated systems, the offgas volume is reduced, as are the associated pollution control equipment sizes.

Disadvantages:

The high temperatures result in a high volatilization of toxic, heavy metals present in the waste stream, especially in a reducing environment. There is a heavy consumption of electrodes, especially in an oxidizing environment. If melting bulk metals, there is a possibility of steam explosions if liquids get below the surface.

Development Needs:

Testing with various types of waste feed is needed to gain experience, verify applicability, and identify potential problems. Operational and physical parameters must be optimized and methods utilized to keep the heavy metals from volatilizing from the melt.

Vendors:

Electrolysis Inc.
Heat Engineering Corp.
Koch Process Systems
Lectromelt
Mannesmann Demag Corp.
Whiting Corporation

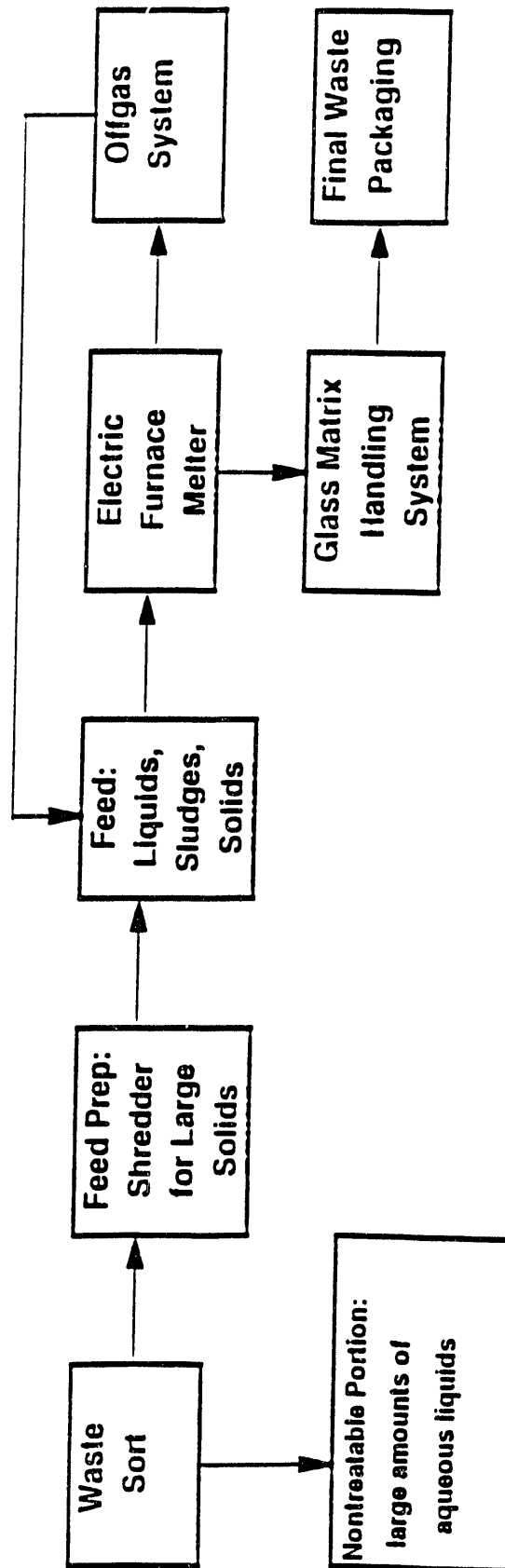
DOE Laboratories Involved in Technology:

Idaho National Engineering Laboratory

References:

"Metalcaster's Reference and Guide", First Edition, E. L. Kotzin, Editor, American Foundrymen's Society, Des Plaines, IL, August 1972.

ELECTRIC FURNACE MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: Fuel-Fired Melter

Maturity: Operational - Conventional (not demonstrated for waste processing)

Description:

The fossil fuel hearth melting technology is a thermal smelting technology consisting of a molten slag bath into which metal ore, blast furnace slag, and other waste materials are introduced. The specific process variation addressed here is an adaptation of a proprietary commercial metal smelting technology, known as "Sirosmelt." The sirosmelt process utilizes a lance through which air and fuel can be injected under the surface of the slag bath. This injection of air fuel mixture creates high turbulence within the bath, providing good mixing and combustion of the waste. The system is flexible in producing an oxidizing or reducing environment, depending on the waste being processed. Operational temperatures of the molten bath of as much as 1600°C destroy the organics and melt the inert fractions into a vitrified slag product. Fluxing agents can be introduced into the bath through the lance while larger particle waste forms are fed through an auxiliary feed port. The resulting slag of melted inert material is removed and cast into 1 to 2 ton blocks.

Waste Applicability:

Aqueous Liquids: Medium applicability.

Organic Liquids: High applicability.

Wet Solids: High applicability.

Dry Solids: High applicability.

Dry Heterogenous Solids (Small and Large): Low applicability to heterogeneous solids due to poor mixing with these feed materials. Lance may enhance mixing and increase applicability.

Advantages:

Lance injection of air creates excellent waste/bath mixing for maximum combustion, relatively simple operating concept with few moving parts, high temperatures resulting in high waste destruction efficiencies.

Disadvantages:

High temperatures within the system will volatilize metals and will generate NO_x, the

high turbulence will cause high particulate carryover, and contamination control problems in a non-sealed furnace configuration.

Development Needs:

The primary need is to develop and demonstrate the operability of the technology as a waste treatment process. The Sirosmelt process has had very limited testing. As with other melting technologies, effects of heterogeneous waste streams on slag chemistry and process operations, as well as the fate of heavy metals and radionuclides, need further investigation.

Vendor List:

Ausmelt Pty, Ltd. (Australia)
General Glass Equipment Co.
Surface Combustion, Inc.
Toledo Engineering Co.

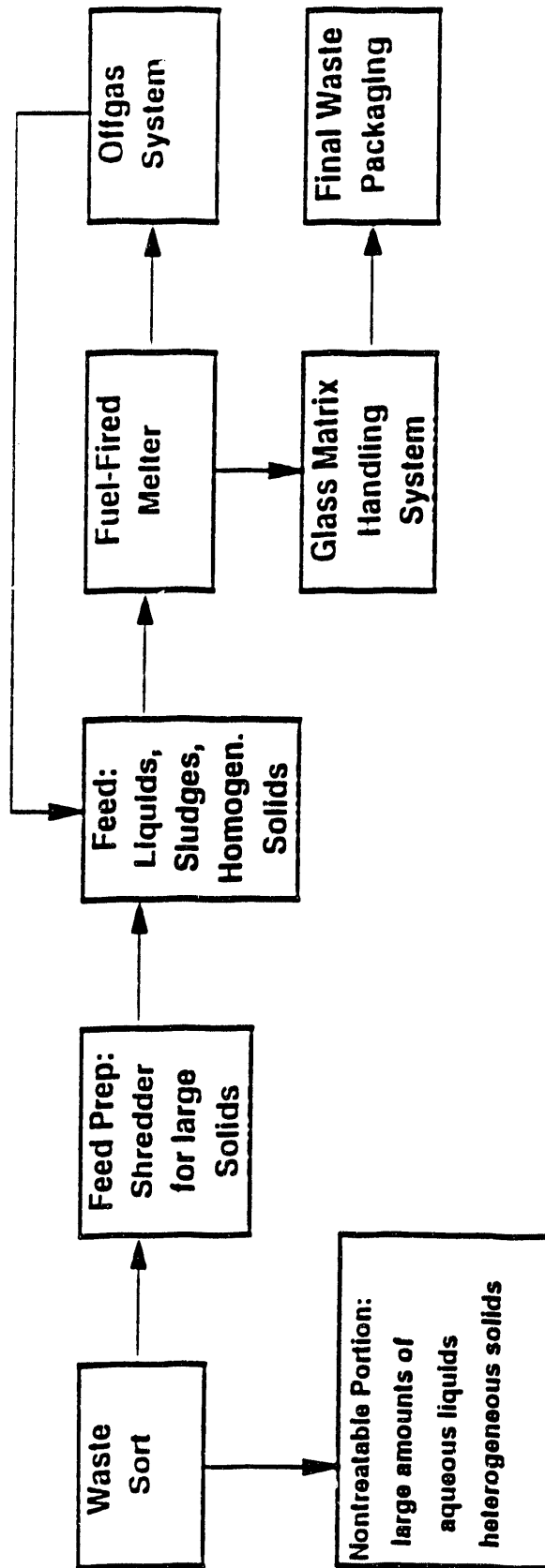
DOE Laboratories Involved in Technology:

None

References:

SAIC, "An Assessment of Incineration and Melting Treatment Technologies for Application to the RWMC Buried Waste", 1991.

FUEL FIRED MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: High Temperature Joule Melter

Maturity: Emerging - Pilot Scale

Description:

The high temperature joule heated melter can take many different forms. A specific design is directed toward the overall operational production objectives. This generalized technology is the foundation for nearly all high quality glass produced in the glass industry. The unit has a processing chamber which contains the molten glass and is lined by refractory. This versatile device can process a broad spectrum of wastes. Organic liquids, wet solids, dry solids, and heterogeneous solids can all be fed to this generalized process if the appropriate off gas treatment system is connected. The material is fed through a central location. If the waste contains combustible solids or organics, oxidation air is directed into the pile. After the material heats, combusts and oxidizes, it settles to the molten glass surface, where it melts and is homogenized with the balance of the material in the molten pool. In this arrangement, top entering electrodes are immersed in the molten pool and provide the joule heating. This allows renewal of the consumable electrodes which are usually either graphite or molybdenum. Operating temperatures in excess of 3000°F can be sustained using conventional materials. The joule heating induces natural convection around the electrodes resulting in good mixing and nearly uniform temperatures within the majority of the bulk glass. The high temperature allows metals such as iron and stainless steel to be included in the waste. Metals settle into the pool, melt and collect at the bottom. These molten materials can then be oxidized and incorporated into the bulk glass before being discharged into the waste box or a post-treatment system. Separation of about two to three feet between the end of the power electrodes and the molten metal avoids significant electrical shorting.

Waste Applicability:

- | | |
|------------------|--|
| Aqueous Liquids: | Medium applicability to aqueous liquids. Direct aqueous liquid feeding onto the pool can consolidate unit operations and may be attractive for certain waste streams. |
| Organic Liquids: | High applicability to organic liquids. Demonstrated destruction efficiencies in excess of 99.999% have been demonstrated at Mound laboratory and Pacific Northwest Laboratory. |
| Wet Solids: | High applicability to wet solids. |
| Dry Solids: | High applicability to dry solids because it can oxidize and melt the feed material into a molten pool within the same device. |

| | |
|---|--|
| Dry Heterogenous Solids (Small and Large): | High applicability. This technology can accommodate metals contained in the waste. The collection of molten metals at the bottom of the molten pool may be oxidized in place or tapped off periodically. |
|---|--|

Advantages:

With the high temperature capability of this technology, metals that may be found in the waste feed can be melted, collected at the bottom, and oxidized. High waste loading can be realized at higher temperatures so 80 to 100 wt% of the waste may be incorporated into a chemically durable material before being discharged for disposal. The large inventory of molten material allows high variations in the instantaneous composition being fed. The large molten pool which may represent 4 to 5 days of feeding, can be used to average the waste composition over time and can allow large variations over significantly long periods of time without adversely impacting the quality of the discharged material. The ability to oxidize feed materials directly as indicated or process slurries, solutions, and sludges, without pretreatment allows a very broad range of material to be considered for processing. The configuration is readily adaptable to radioactive operation, because all key replaceable systems can be accessed and replaced from the top.

Disadvantages:

This device is best suited for long term, continuous operation. Therefore, rapid shut down and intermittent operation are not recommended. High temperatures within the system will volatilize heavy metals and generate high NO_x.

Development Needs:

The key development need is the demonstration of the different waste streams in a unit of this style. This will allow measurement of instantaneous and specific processing rates to be defined and the identification of phase separation, if any. Tailoring of acceptable glasses may also be required for acceptance. Effects of heterogeneous waste streams on slag chemistry and process operations, as well as the fate of heavy metals and radionuclides, and NO_x production need further study.

Vendor List:

American Environmental Management Corp.
Frazier-Simplex Inc.
Penberthy Electromelt International, Inc.
Recomp, Inc.
Sorg Engineering

Toledo Engineering Co., Inc

DOE Laboratories Involved in Technology:

Pacific Northwest Laboratory
Mound Laboratory

References:

Armstrong, 1985, Nitrate Waste Processing by means of a Joule-Heated Glass Furnace, K. M. Armstrong, L. M. Klingler, October 18, 1985, MLM-3304, Mound Laboratories, Miamisburg, OH.

Brouns, 1988, R. Brouns, A. A. Balasco, et. al., Bench Scale Glassification Test on Rocky Mountain Arsenal Basin F Material, AMXTH-TE-CR-88015, Pacific Northwest Laboratory, Richland, WA.

Chapman, 1991, C. C. Chapman, Evaluation of Vitrifying Municipal Incinerator Ash, Proceedings of the Fifth International Symposium on Ceramics in Nuclear and Hazardous Waste Management, American Ceramic Society, April 29 -- May 3, 1991.

Janke, 1990, D.S. Janke and C. C. Chapman, Characteristics of Fernalds K-65 Residue Before, During and After Vitrification, FMPC/SUB-035, Pacific Northwest Laboratory, Richland, WA.

Klinger, 1989, Joule Heated Glass Furnace Processing of a Highly Aqueous Hazardous Waste Stream, L. M. Klingler, P. L. Abellera, March 17, 1989, MLM-3577, Mound Applied Technologies, Miamisburg, OH.

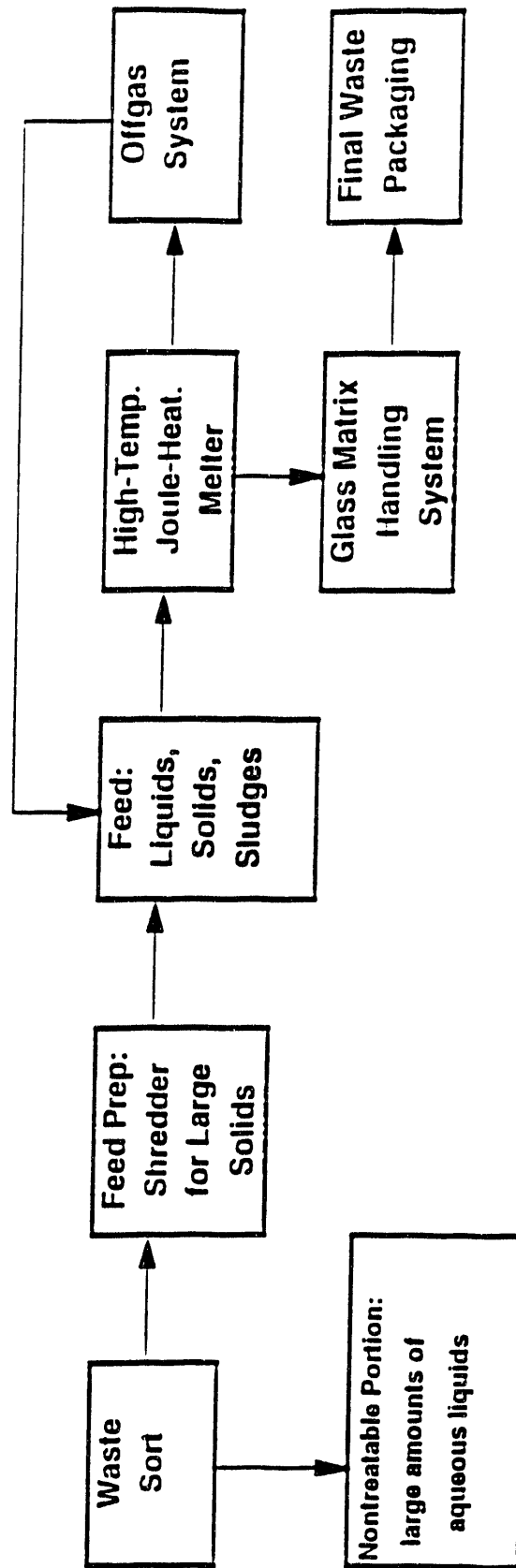
Klinger, 1988, Glass Furnace Processing of Rocky Flats Plant Wastes - an Evaluation, L.M. Klingler, P.L. Abellera, April 29, 1988, MLM-3493, Mound Laboratories, Miamisburg, OH.

Klinger, 1985, Glass Furnace Project Final Report: An Evaluation of Operating Experience for Low-Level Nuclear Waste Processing, L. M. Klingler, K. M. Armstrong, February, 28, 1985, MLM-3229, Mound Laboratories, Miamisburg, OH.

Larson, 1983, Assessment of power reactor waste immobilization by vitrification, D. E. Larson, et. al., August, 1983, EPRI-NP-3225, Pacific Northwest Laboratories, Richland, WA.

Moore, 1989, All Electric Furnace Works Well for Gallo Glass, R. D. Moore, R. E. Davis, Glass Industry, March, 1989, pp 10-15.

HIGH-TEMPERATURE JOULE HEATED MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: In-Can Resistance Melter

Maturity: Developmental

Description:

An alloy canister or can is used as both the melting crucible and the disposal container. The can is placed inside a resistance-heated furnace and heated up to 1050°-1070°C. Waste and glass frit are added simultaneously in the desired proportions by gravity feed through a drop tube. The tube can be submerged below the melt surface to increase the absorption of inorganic matter into the melt. If the waste is to be combusted as well as vitrified, oxygen is also added through the drop tube. As the waste and glass frit are added to the melter, the level in the can will rise. When the can is full, the waste and frit feed is diverted into a second in-can melter while the filled can in the first melter is cooled, removed from the furnace, and capped before transportation to a disposal facility. The critical process parameters are temperature, rate of waste/frit addition, the ratio of frit to waste, and, for waste combustion, the amount of oxygen in the system.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Low applicability because of heat input constraints, treatment via calcination or evaporation may be desirable for large quantities of aqueous liquid waste with the remaining residue treated by in-can melting. |
| Organic Liquids: | Low applicability. |
| Wet Solids: | Medium applicability. |
| Dry Homogeneous Solids: | Medium applicability. |
| Dry Heterogeneous Solids (Small): | Low applicability. Poor mixing of melt is a concern. |
| Dry Heterogeneous Solids (Large): | Not applicable. |

Advantages:

This process is fairly simple and does not require transfer of the molten material from vessel to vessel. With the exception of the volatile matter that becomes part of the offgas, all the waste material is fed to the final disposal container. Consequently, the melter is not degraded by the corrosiveness of the melt, and the furnace interior should not be

contaminated to the degree that other melters are. These characteristics enhance the remote operability of the melter.

Disadvantages:

The in-can melter has a slower processing rate than other melters. The maximum melting rate is dependent on the can diameter, which is determined by the heat load which the alloy canister can handle. In short, the processing temperature and time are limited by the durability of the canister alloy. At operating temperature, the alloy canister can be subjected to a severe environment. A corrosive molten glass and high temperature oxidation will degrade the canister unless an expensive alloy is used. In addition, because the alloy can has a higher thermal expansion than the glass melt, the can contraction from cooling would normally be greater than the glass. As a result, after the can and glass are cooled, the hardened glass will keep the can in an expanded condition with severe mechanical stresses. There can also be some control problems with the in-can melter. The rising molten glass level must be continuously monitored, which is difficult at operating temperatures. The glass frit-to-waste ratio can also be difficult to control if the waste is added directly from the discharge of another waste treatment unit such as a calciner. This method of feed addition can also result in poor blending of the waste and the glass frit.

Development Needs:

Improvements in heat and mass transfer are needed to reduce the melt time. Longer term testing is needed to identify and solve operational problems. Application to radioactive waste must be verified and the offgas characterized.

Vendor List:

Not commercially available.

DOE Laboratories Involved in Technology:

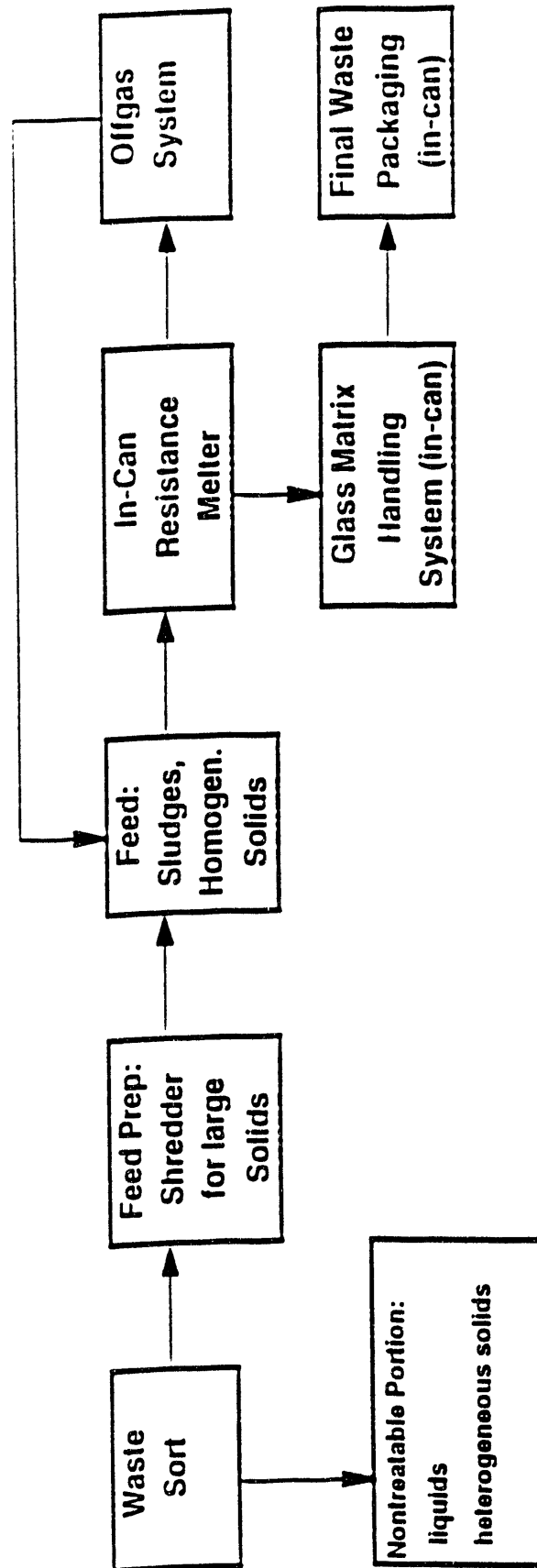
Pacific Northwest Laboratory

References:

H. T. Blair, "In-Can Melting Process and Equipment Development from 1974 to 1978", PNL-2925 UC-70, prepared by Pacific Northwest Laboratory for the U.S. DOE under contract EY-76-C-06-1830, August 1979.

J. L. Buel, "The Feasibility of Incinerating and Vitriifying Organic Resins in a Single Step", The Handling and Treatment of Radioactive Wastes, edited by A. G. Blasewitz, J. M. Davis, and M. R. Smith.

IN-CAN RESISTANCE MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: Stirred Joule Melter

Maturity: Emerging - Pilot

Description:

Stirred joule melters are joule-heated melters in which the molten material is agitated by a stirrer. Depending on the type of waste feed, different stirrers can be utilized to optimize the process. The waste can be fed in a dry form or in an aqueous slurry; however, a lower throughput results from an aqueous feed. A two-zone melter is used with the top zone highly mixed by the stirrer. The bottom zone is less turbulent so that gas bubbles can separate and rise out of the zone, resulting in a dense glass. Electric resistant heaters are used to pyrolyze organic materials and provide startup heat until electrically conductive temperatures are reached so that joule heating can be established.

Waste Applicability:

| | |
|----------------------|---|
| Aqueous Liquids: | Low applicability for aqueous liquids. |
| Organic Liquids: | High applicability. |
| Wet Solids: | High applicability. |
| Dry Solids: | High applicability. |
| Heterogenous Solids: | Not applicable because of potential damage to the stirrer by large solid objects. |

Advantages:

Because the stirrer increases efficiency in heat distribution, stir melters have a high throughput rate for their size. Throughput rates with the stirrer operating have been eight times greater than those without the stirrer operating. The greater efficiency in heat distribution also permits operation of the stir-melter at lower temperatures, thus allowing increased flexibility in selection of materials for melter components and increased contaminant incorporation into the waste glass. The smaller size and lower operating temperatures also reduce costs by reducing heat losses.

Disadvantages:

Because this technology is basically a variation of high temperature joule melters, there are the same types of disadvantages for the stirred joule melter as for the high temperature joule melter. There is concern about damaging the stirrer if large metallic objects are added to the melter. There is also a concern with heavy metal carryover from

volatilization because of the high temperature, long residence time, and potentially reducing environment. As with other vitrifiers, chloride and sulfate salts in the waste are not tolerated very well.

Development Needs:

More work is needed to demonstrate this type of unit on various types of waste streams. In conjunction with this work, different types of glasses can be tested and the chemistry verified. Characterization of the offgas is needed, and when appropriate, efforts to minimize reduction of metal oxides and thereby minimize volatilization of metals would be beneficial. If organics are to be processed in this type of melter, the unit must be mated to a secondary combustion chamber.

Vendor List:

Glasstech

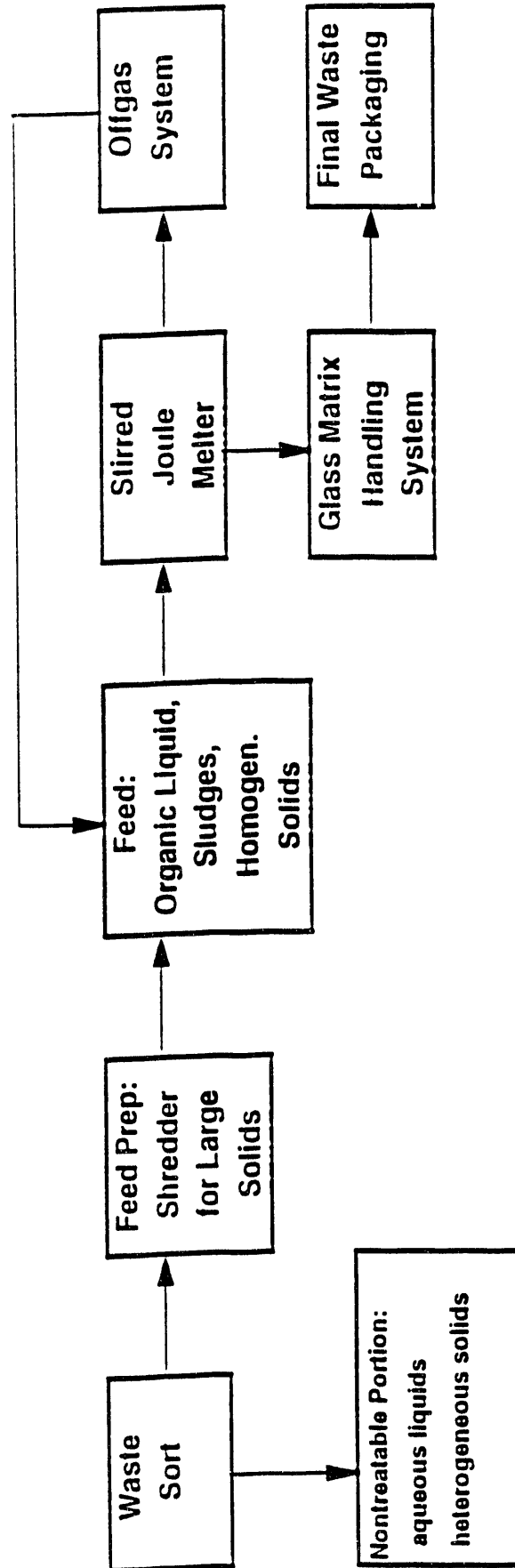
DOE Laboratories Involved in Technology:

Savannah River Site

References:

R. S. Richards and J. W. Lacksonen, "Stir-Melter Vitrification of Simulated Radioactive Waste, Fiberglass Scrap, and Municipal Waste Combustor Flyash", presented at the 93rd Annual Meeting of the American Chemical Society, Cincinnati, OH, April/May 1991.

STIRRED JOULE MELTER FUNCTIONAL PROCESS DIAGRAM



Technology Name: Induction Melter

Maturity: Operational-Conventional

Description:

An induction melter consists of a refractory-lined crucible with an electrical induction coil encircling the crucible for a heat source. A high frequency power supply provides the electrical input and a cooling water system is needed to cool the induction coil and the power supply. An induction melter can be used to melt metals or vitrify inorganic materials such as incinerator ash in order to volume reduce the waste and obtain a more stable final waste form. Waste material is placed in the crucible and the power supply is turned on. The material in the crucible begins to melt, forming a molten mass which flows down into the bottom of the crucible, filling the void spaces between the unmelted waste. Once the waste in the crucible is melted, additional waste material is slowly added to the crucible and allowed to melt before the next batch of material is added. When vitrifying inorganic material, an additive may be used to lower the melting point of the waste material. When melting metals, a slag coagulant is added to the top of the molten mass to aid in slag removal. Once the waste is completely melted and at the desired temperature, the melter is tilted so that the molten mass can be poured into a refractory-lined mold.

Waste Applicability:

| | |
|-----------------------------------|---|
| Aqueous Liquids: | Low applicability, with only small amounts of moisture present in the waste. If moisture is added after the melting begins, steam explosions can occur. Even trace quantities of moisture can cause splattering of the molten matter. |
| Organic Liquids: | Low applicability. In metal melting operations, it is undesirable to enhance contact between the waste and oxygen, or more slag will be formed. |
| Wet Solids: | Low applicability; can only be charged before the system is heated up. |
| Dry Homogeneous solids: | Low applicability. |
| Dry Heterogeneous Solids (Small): | Medium applicability (metals only). |
| Dry Heterogeneous Solids (Large): | Medium applicability (metals only). |

Advantages:

This technology is used commercially in the foundry industry and is well understood. It provides high density of final waste form, which results in a good volume reduction ratio. The final waste form is highly resistant to leaching. The only waste pretreatment necessary is size reduction of large components to fit in the melter.

Disadvantages:

A slag bridge can form when melting metals that can result in an insulating effect, which will lead to higher temperatures in the melt, which can damage the refractory lining. The slag bridge can also prevent the release of smoke and gases, resulting in a pressure buildup and a possible eruption of the molten material if the pressure breaks the bridge. Moisture can cause steam explosions. High frequency power supply can result in generation of a large amount of electrical "noise" throughout the electrical distribution system. The temperature of the molten mass must be carefully controlled to ensure proper transfer of material into the mold.

Development Needs:

Efforts to verify treatment of various types of wastes are needed. Incorporation of a nuclear-grade offgas treatment system and possibly a secondary combustion chamber may be needed as well. Improvements in monitoring the process, including the melt temperature, would be beneficial. Characterization of offgas is necessary, and improvements in offgas monitoring would be beneficial.

Vendors:

ABB Industrial Systems
Ajax Magnethermic Corp.
Inductotherm Corp
Industrial Furnace Systems
Leco Corporation
Omega
Pillar Industries
Radyne Corporation

DOE Laboratories Involved in Technology:

Hanford
Idaho National Engineering Laboratory

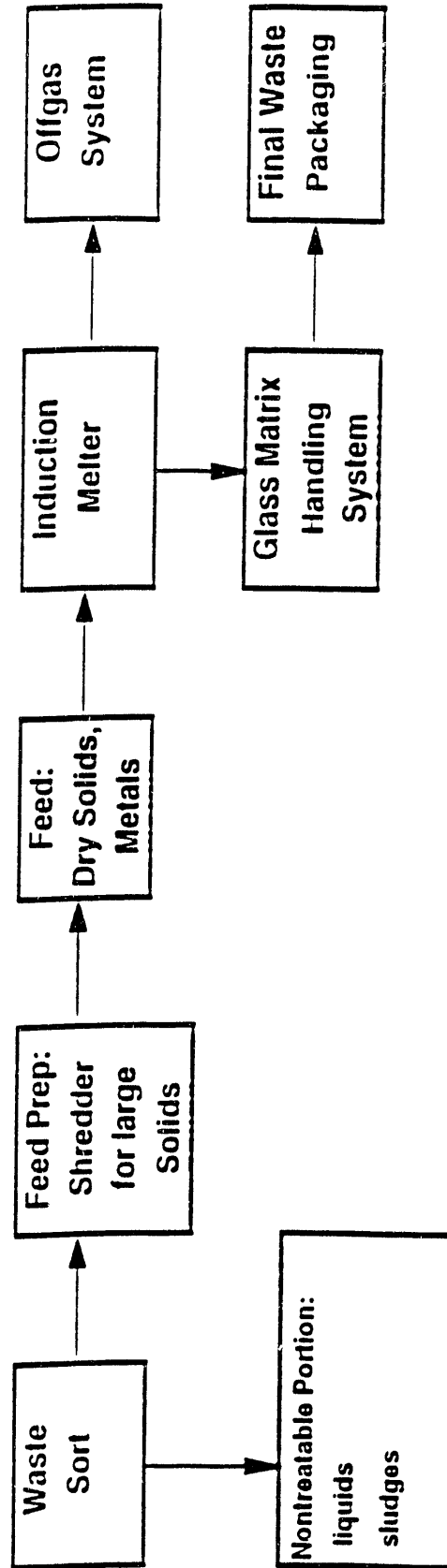
References:

K. Katoh, S. Fujisaki, and K. Hirayama, "Solidification Tests of Radioactive Incineration Ash with Induction Heat Melting Process", presented at the International Conference on Incineration of Hazardous, Radioactive, and Mixed Waste, San Francisco, California, May 1988.

M. M. Larsen and J. A. Logan, "Sizing and Melting Development Activities Using Noncontaminated Metal and the Waste Experimental Reduction Facility", prepared by EG&G Idaho for the U. S. DOE, EGG-2319, May 1984.

R. L. Gillins and R. Y. Maughan, "Progress Report on Metal Sizing and Melting Activities at the Waste Experimental Reduction Facility", prepared for the U.S. DOE, EGG-2434, November 1985.

INDUCTION MELTER FUNCTIONAL PROCESS DIAGRAM



APPENDIX D

Additional Thermal Treatment Technologies¹

¹ Information contained in Appendix D was obtained from the following document: Geimer, R., Hertzler, T., Gillins, R., Anderson, G., *Assessment of Incineration and Melting Treatment Technologies for RWMC Buried Waste*, EG&G-WDT-1035, EG&G Idaho, Inc., February 1992.

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| Fossil Fueled Shaft Furnace (Cupola) | D-9 |
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| Graphite Electrode Shaft Furnace | D-13 |

CYCLONE MELTER

Technology Description

A cyclone melting system utilizes cyclonic flow patterns to create high turbulence, which produces good mixing and heat transfer, enhancing combustion of organics and promoting melting, agglomeration, and separation of inorganics. Preheated combustion air (800-1100°F), fuel (natural gas, fuel oil, or powdered coal) and the waste is injected tangentially into a cyclone barrel. Temperatures within the combustion chamber can reach 3000°F. At this operating temperature the inorganic components of the feed material are melted and transported to the outer wall of the cylinder. The slag formed is tapped from the bottom of the chamber and quenched in a water bath. The resulting residue composition and quality is dependent on the waste feed composition and the addition of any glass making constituents. The organics are oxidized either within the primary chamber or in a secondary chamber with a separate burner.

Past/Current Applications

Original development centered around the combustion of crushed coal in electrical generating facilities. Past waste applications of the cyclone furnace/melting technology have been for the incineration of sewage sludge, liquid waste destruction, and the combustion of high inorganic (ash) coal. Current applications for incineration/vitrification of municipal solid waste (MSW) ash, incinerator ash, soils with particle size reduction, asbestos, and waste insulation fiberglass are being studied and are in various stages of development.

Advantages

The Cyclone melter is capable of achieving high destruction and removal efficiencies of organics mainly due to the high turbulence/mixing of the waste with the fuel/air mixture. This turbulence and mixing also promotes rapid heat transfer, allowing for the melting of inerts in a short residence time. The resulting slag residue is a glass-like material that should prove leach resistant.

Disadvantages

Chemistry of the residual solid product is difficult to control due to up-front estimation of waste feed characteristics, prefeeding estimated quantities of chemistry control additives, and short residence time effecting waste mixing and homogeneity. Other disadvantages include: restricted waste feed particle size, little control of fate of bulk metals, high temperatures and turbulence in the system generates thermal NO_x, and

high temperatures also volatilizes some heavy metals (e.g., Cd, Pb, Cr) which are transported into the off-gas system.

Research Needs

Research is needed to better define the slag characteristics resulting from the processing of various waste streams. Physical characteristics and chemical leaching characteristics are dependent on the types of waste processed and additives that may be added. Advanced testing needs to be performed on various anticipated waste compositions.

Process Data

Status: Emerging-pilot

RCRA Handling Code: T18

Thermal Capacity Range: Low-- 0.70 MMBtu/hr
High-- 425 MMBtu/hr

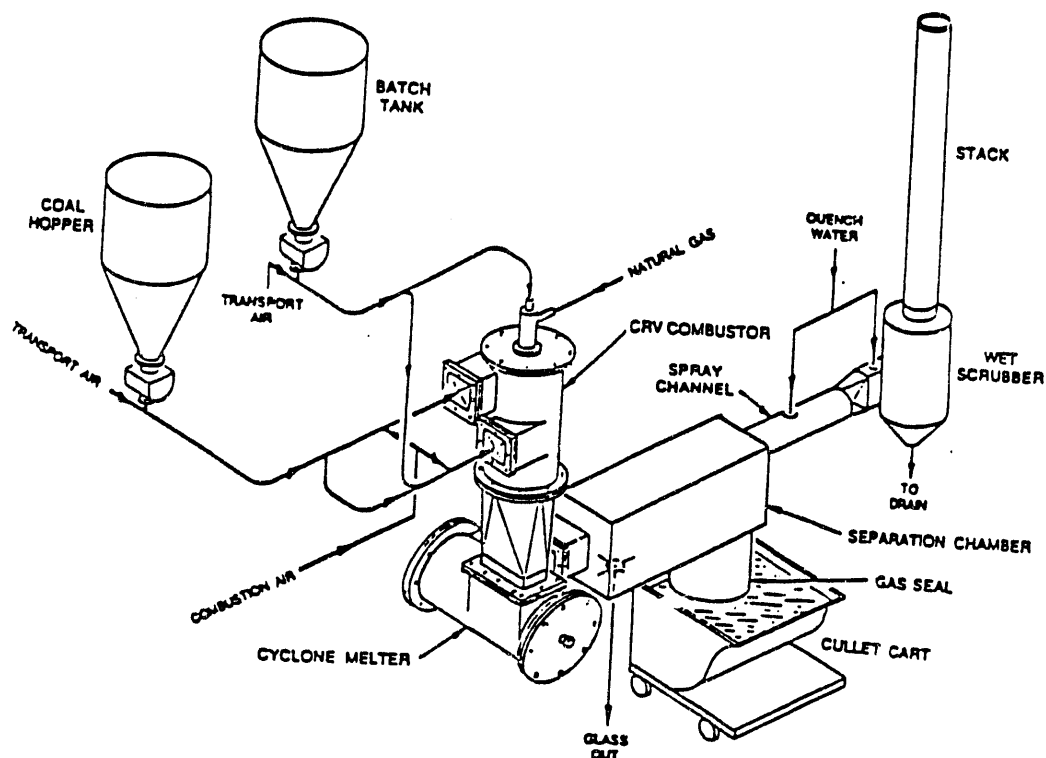


Fig. D.1. Cyclone Melter.

GRAPHITE ELECTRODE HEARTH MELTER^c

Technology Description

The graphite electrode hearth melter technology encompasses a variety of electrically powered furnaces/melters. These furnaces use electrical energy transferred through graphite electrodes to melt raw materials (ore, slag waste, scrap, etc.) charged into a molten bath. The electrodes may be submerged in the molten bath where the resistance to the electrical current passing between the electrodes creates the temperatures necessary to melt the material, or they may remain above the surface of the bath creating an arc plasma zone of high temperature. Temperatures up to 3000°F are routinely maintained within the furnace chamber with even higher temperatures achievable. The high temperatures of the molten bath quickly destroy the organic components while melting the inert material into a glassy slag. A closed-furnace (sealed) system design would provide better offgas contamination control when dealing with radioactive and hazardous waste. Material feed for the closed furnace arrangement can be provided through feed handling chutes, hollow electrodes, or a series of air-lock doors. Material with grain sizes from 1/4" to 2" diameters can be fed through the chutes while fines (< 1/4") can be fed through the electrodes. Larger containers and objects can be fed through the air-lock door system.

Past/Current Applications

The electric-arc melting technology has been used in the steel industry for many years. Applications have focused on secondary smelting and metals recovery from steel making dusts and scrap iron. Current steel applications for sealed systems are for pig-iron smelting in a reducing atmosphere. New applications in the waste processing area are being demonstrated. An Arc Pyrolysis unit using a direct current electric arc furnace has been designed specifically for the treatment of solid hazardous waste. Limited data on this process is available at this time.

Advantages

Electric-arc furnaces can handle a wide variety of waste forms and sizes, depending on whether an open or closed system is used. The use of electrical energy rather than a fossil fuel for melting and combustion of the waste material significantly reduces the volumes of offgas and associated pollution control equipment requirements. Additionally the slag residue produced is a vitrified material that is likely to be a very stable waste form.

Disadvantages

The high operating temperatures of the system volatilize heavy metals such as cadmium, lead, and chromium. Electrode consumption is high, particularly in an oxidizing environment. The potential for accidental feed of water below the molten surface of the melt could cause a steam explosion. Additionally high turbulence causing offgas control difficulties may result from the introduction of high combustible waste streams.

Research Needs

Optimization of slag chemistry for heavy metals stabilization; evaluation of variations in slag chemistry resulting from variations in the input waste; reintroduction of condensed volatile metals into the slag phase. Additional testing and evaluation of technology performance on anticipated waste composition must be performed.

Process Data

Status: Emerging-pilot scale

RCRA Handling Code: T18

Thermal Capacity Range: Low Range-- 0.2 MMBtu/hr

High Range-- 164 MMBtu/hr

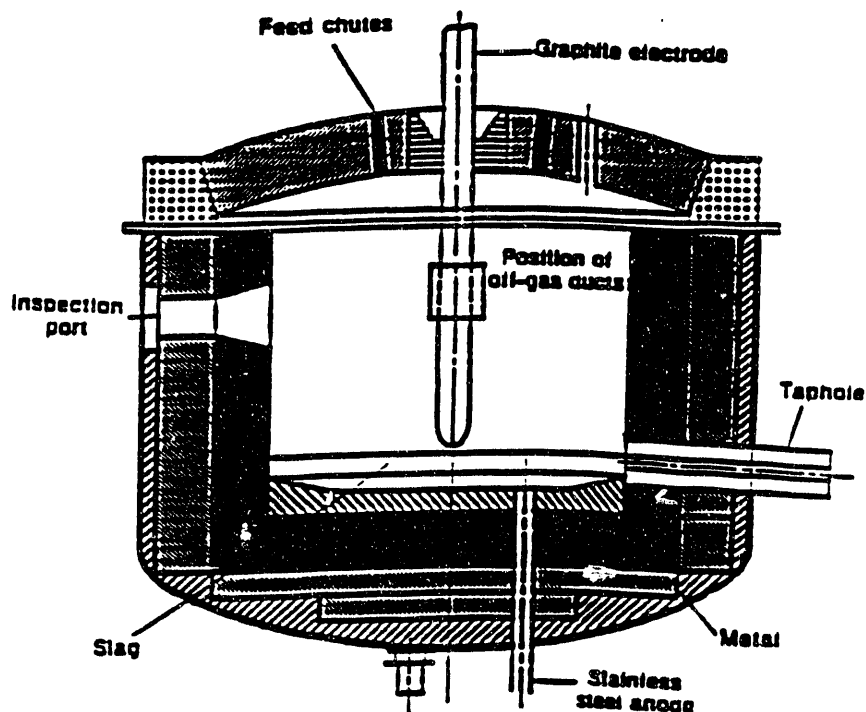


Fig. D.2. Graphite Electrode Hearth Melter

Fossil Fueled Shaft Furnace (Cupola)

Technology Description

The traditional cupola is a vertical cylindrical water-cooled shaft furnace commercially used in the foundry industry. The process feed material, typically consisting of scrap iron and steel along with coke and limestone, is fed into the furnace from the top, and the combustion (blast) air is introduced through tuyeres at the base of the vessel. The combustion of the coke and the counter current flow of the product gases through the charge material supply the heat necessary to melt the iron. The cupola furnace is operated in a reducing atmosphere to promote the reduction of metallic oxides associated with the iron scrap charge. The primary product gases are carbon monoxide (CO) and hydrogen (H₂). Limestone is the primary fluxing agent added to slag the coke ash and other non-combustibles. Metal and slag are tapped from the bottom of the furnace and the CO and H₂ are fed into an air enriched afterburner to complete combustion of the CO, H₂, and any volatilize. Conventional cupola furnaces operate with a metal to coke weight ratio of approximately 10:1 with 90% of the coke is consumed in heating and 10% added as carbon to the metal.

Variations of the basic cupola "cold-blast" (no waste heat recovery) furnace have been developed over the years. A hot-blast cupola furnace was developed to recover the sensible heat in the exhaust gases and increase the blast temperatures to 400°C. Additional developments have centered around the use of auxiliary fuels such as natural gas, oil, anthracite coal and plasma arc (electrical) to minimize the quantity of coke needed as well as increase the productivity while decreasing the environmental impact of off gas products.

Past/Current Applications

The cupola (shaft) furnace is a very mature secondary smelting foundry technology. New areas of application are focusing on the waste treatment potential of these furnaces. With the addition of plasma torches and other auxiliary sources of heat energy, the cupola furnace may have potential waste processing applications.

Advantages

Shaft furnace smelting is a mature technology and considerable experience in adapting auxiliary heat sources (e.g., plasma arc) to enhance system flexibility have been tested and verified in the foundry industry. This experience will influence future development, however the technology is limited by the lack of development for waste treatment.

Disadvantages

Conventional fossil fuel-fired cupola furnaces have significant gas velocities which prevent the charging of fine material. Fines and small particles are easily entrained and carried into the pollution control equipment. Bridging and freezing of the charge material can occur. Non-homogeneity of feed material can cause slag chemistry control problems and temperature excursions within the furnace.

Process Data

Status: Emerging technology

RCRA Handling Code: T18

Thermal Capacity Range: Low-- 4 MMBtu/hr
High-- 250 MMBtu/hr

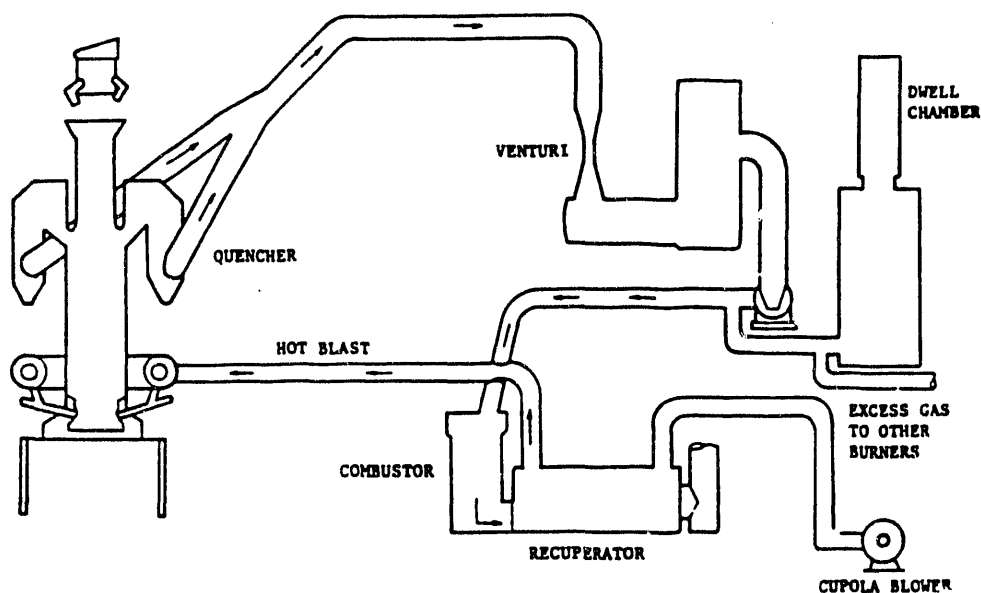


Fig. D.3. Fossil Fuel Shaft Furnace.

Plasma Torch Fired Shaft Furnace^d

Technology Description

The plasma-fired cupola operates in much the same way as the conventional fossil fuel-fired cupola except it utilizes plasma torches as an auxiliary source of thermal energy. The plasma torches are symmetrically located at the base of the vessel in the areas of the tuyeres. The plasma torches superheat the blast air up to 1000°C. This heat source decreases the amount of coke necessary in the feed, which greatly reduces the volume of offgas generation. The decrease in offgas generation reduces the gas velocities within the furnace allowing finer material to be charged and processed. Process gases (e.g., CO, H₂) from the top of the furnace are recycled through the plasma torches additionally reducing offgas volumes. The product gases not recycled are fully oxidized in an air enriched afterburner as in the conventional cupola.

Past/Current Applications

Plasma-fired cupola furnaces have been commercially applied in the foundry industry. Plants applying this technology both in the U.S. (General Motors) and abroad (Peugeot) have verified the benefits of using plasma arc fired cupolas in foundry applications. Westinghouse Electric is presently involved in extending the application of this technology to hazardous waste treatment. Limited information is currently available on the testing of this pilot scale system.

Advantages

The plasma-fired cupola has a number of advantages over the conventional coke fired cupola. The reduction of coke use reduces cost, decrease offgas generation and particulate emission by reducing superficial velocities within the furnace. The auxiliary heat source provided by the plasma allows flexibility in waste input characteristics and melt temperatures can be controlled rapidly reducing bridging problems.

Disadvantages

As with conventional cupola furnaces the "bridging" of the charge material can occur. Residual solid product chemical and physical characteristics is very difficult. Slag chemistry is only controllable based on initial feed characteristics, waste density and waste mixtures. Estimation of feed parameters, fluxing additives, and non-homogeneity of waste will cause final waste form variations. Additionally channeling of offgas through the charge material causes variation in bed temperatures and other operational parameters.

Research Needs

Continued testing and development for waste treatment applications is necessary. Characteristics and mixes of charge material variations and resulting slag chemistry needs further study.

Process Data

Status: Emerging-bench scale

RCRA Handling Code: T18

Thermal Capacity Range: Low-- 4 MMBtu/hr
High-- 41 MMBtu/hr

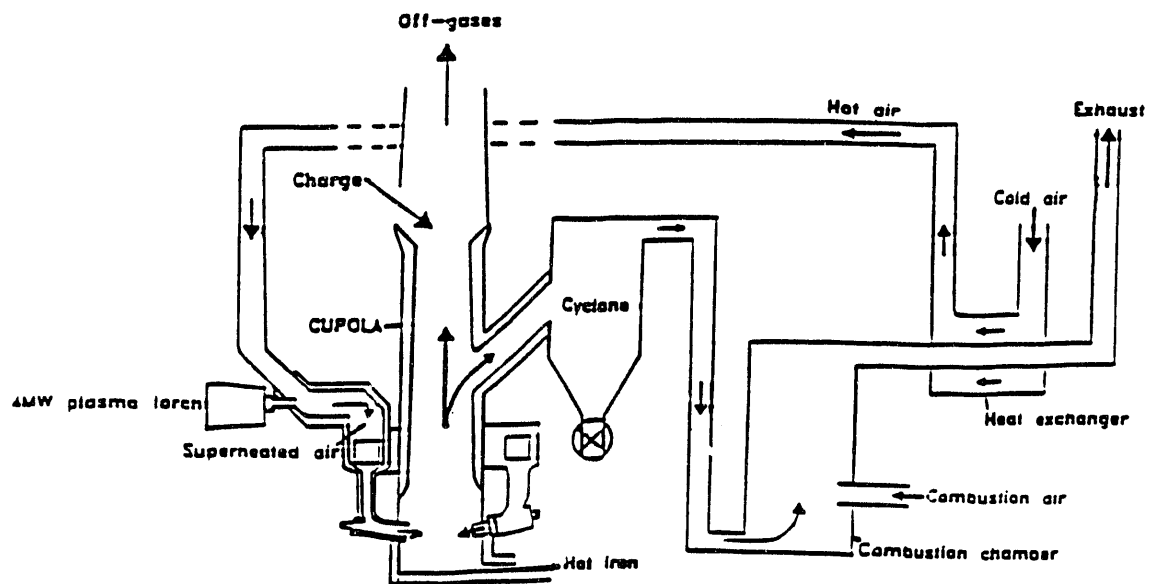


Fig. D.4. Plasma Torch-Fired Shaft Furnace.

Graphite Electrode Shaft Furnace

Technology Description

The graphite electrode shaft furnace technology focuses around the use of electrical energy rather than fossil fuels, transmitted through an electric arc to pyrolyze hydrocarbons into partially oxidized gaseous components (CO , H_2 , tars, gums, etc) which are subsequently fully oxidized in an air rich afterburner. The pyrolyzing concept of incineration utilizes a starved air environment in the primary chamber to obtain the partially oxidized effluent.

A variation to the typical pyrolysis is known as the Skygas process. The skygas process depends on an electric arc initiating the dissociation and activation of water within the reactor and the feed to create gaseous radicals of H_2 and OH . The gaseous radicals pass up through the waste initiating a chain reaction and propagation of additional free radicals and non-radicals. The large hydrocarbon molecules are thus broken down into partially oxidized gaseous components. Predominate gas products from the skygas process are the same as the basic process and consist primarily of CO , and H_2 .

The gases exiting the primary chamber are again subjected to an electric arc and passed through a coke bed for additional refining and breakdown of tars and gums.

Past/Current Applications

Controlled air pyrolysis using fossil fuels as the heat source have been used as a waste treatment process for many years. The use of a controlled air electric arc processing a waste treatment application is new. Limited data is currently available on the application of this technology to waste treatment.

Advantages

The use of an electric arc rather than fossil fuels to pyrolyze and/or initiate radical chain reaction chemistry will reduce the offgas volume generation which will reduce pollution control handling requirements. Multiple sets of reactor units in the skygas process would further reduce carbon dioxide and tars/gums resulting in a higher quality of gas product for reuse.

Disadvantages

As with other shaft furnace technologies bridging of the charged material within the shaft is a problem, especially if high percentages of low melting materials (e.g., glass) exist within the feed material. The resulting residue from the primary reactor of the Skygas process is a non-vitrified ash. Limited feed handling capabilities were noted in testing of chopped tire feed and high moisture content sludge.

Research Needs

Limited data exists on the application of this technology on waste streams of mixed composition. Additional testing and documentation of process data, operational parameters and waste acceptability needs to be performed. Final waste form (ash) stabilization not well addressed in the Skygas process.

Process Data

Status: Emerging technology

RCRA Handling Code: T18

Thermal Capacity Range: Low-- 4 MMBtu/hr
High-- 41 MMBtu/hr

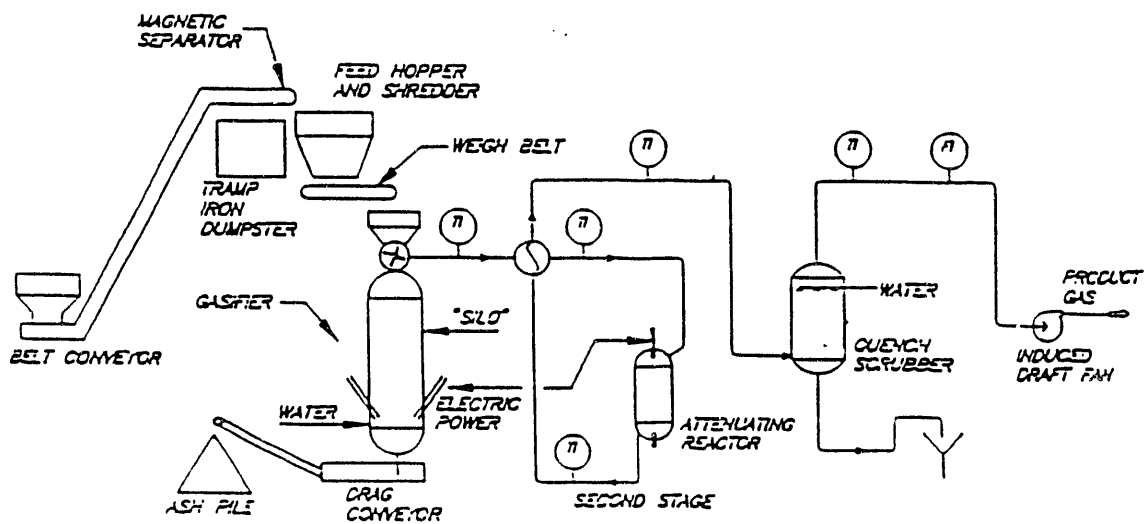


Fig. D.5. Graphite Electrode Shaft Furnace.

APPENDIX E

Additional Furnace Technologies¹

¹ Information contained in Appendix E was obtained from the following document:
Batdorf, J., Gillins, R., Anderson, G., *Assessment of Selected Furnace Technologies for
RWMC Waste*, EGG-WTD-10036, EG&G Idaho, Inc., March 1992.

5.0 MOLTEN METAL BATH TECHNOLOGY

The term molten metal bath refers to a treatment approach that utilizes a pool of molten metal as the medium to transfer heat to waste materials to break down the organics and vitrify the inorganics into a slag material. The slag is drawn off the top of the melt and the bath is maintained at a relatively constant level by tapping off or adding metal, as required. Steel is typically used as the bath material, but any of a variety of metals could be used to tailor the process to meet specific treatment requirements.

5.1 DEVELOPMENT HISTORY

Development of the technology is relatively recent. U.S. Steel (U.S.S.) developed a molten metal bath technology between 1982 and 1986 termed the catalytic extraction process (CEP)(Reference 1). A technology innovation group at U.S.S. developed the process looking for ways to use the stored energy in the molten steel and slag in the steel plant's processing units. The group concluded that they could use the molten metal as a solvent and run high temperature chemical reactions on waste materials and byproducts. U.S.S. received broad patent coverage for the technique. However, U.S.S. opted not to pursue it, and transferred the rights in 1989 to a new development company, Molten Metal Technology (MMT) of Cambridge, Massachusetts, via the Massachusetts Institute of Technology (MIT) (Reference 1).

MMT subsequently formed alliances with a variety of commercial companies to help market the technology. The technical entities that have agreements in place with MMT include L'Air Liquide of France, DuPont, and Rollins Environmental Services of the United States. The CEP is illustrated in Figure E.1.

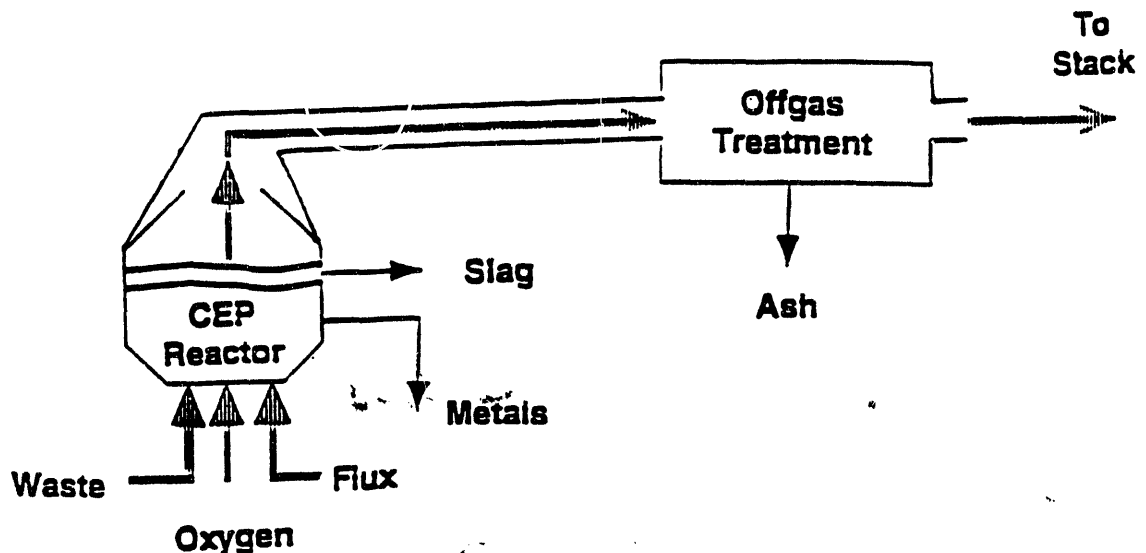


Fig. E.1. Simplified Flow Diagram for Catalytic Extraction Process.

A separate molten metal bath project was initiated in 1986 as a joint German/Swedish (KFA-KTH) investigation (Reference 2 and 3). The project examined various metal industry processes for use in the destruction of hazardous wastes. This investigation leaned toward adapting the iron bath coal gasification process, rather than the scrap metal conversion process that MMT is pursuing, though the processes are very similar. This project was reported as a paper study and is currently in the pilot demonstration stage.

Applicable waste treatment data on this process are extremely limited. MMT is treating the details of its process as proprietary, making assessment of the technology highly speculative. However, some general conclusions can be drawn.

5.2 PROCESS DESCRIPTION

5.2.1 Current Waste Applications

The application for which the technology was originally intended was introduction of chemical energy into a scrap steel converter to increase the capacity for scrap steel processing. This was accomplished by adding carbon from various waste forms, such as waste oil and tires, to reduce the iron oxide and produce iron and carbon monoxide. MMT has subsequently applied the process to a wide variety of chemical byproducts and industrial waste materials. The technology primarily aimed at the resource recovery market, where valuable materials can be recovered from a waste or byproduct stream within an existing commercial process, thereby avoiding RCRA regulations (Reference 4). Examples include the recovery of valuable metals such as nickel from a nickel contaminated organic stream, recovery of cobalt from spent catalysts, or generation of synthesis gas from hydrocarbon bearing wastes. Some work has also been done on waste processing to demonstrate destruction of organics such as PCBs.

The preferred means of feeding is by injecting liquid, gas, or fluidized solid waste into the bottom of the metal bath along with reagent gases and fluxes. However, the vendor reports that large solids such as whole PCB transformers were fed over the bath from a conveyer system.

5.2.2 Theory of Operation

Catalytic extraction processing utilizes standard "off the shelf" equipment from the steel processing industry. The process vessel is a steel converter, which is an enclosed, airtight cylinder with dimensions that can vary widely to yield the combination of surface area and retention time required for specific waste characteristics.

To initiate the process, the metal catalyst, which is typically steel, is melted in the reactor by one of three standard methods; induction heating, electric arc, or plasma arc. Of the three methods, plasma arc is least used. Temperatures in the 2500-3500°F range are used, with 3000°F most typical. A reducing atmosphere is maintained in the reactor.

Waste material in the form of gas, liquid, sludge, or solid is introduced into the molten metal bath beneath the surface, generally utilizing a pneumatic transport

approach for the solids. Because of rapid heat transfer from the molten steel to the waste, the waste material dissociates into its elemental constituents. The dissociated atoms become evenly distributed throughout the bath by the natural convection currents present. Oxygen is injected into the bath to strip the carbon absorbed by the steel. This reaction forms carbon monoxide.

The molten inorganic material (slag) introduced by the waste rises to the top of the bath and is eventually skimmed or tapped off. The slag acts as an insulator to increase the bath's thermal efficiency and as a primary scrubber to remove such materials as volatile heavy metals, phosphates, and sulphur. Other reactants may be added to the bath to optimize slag chemistry. Typically, the slag is cast into shapes for ease of handling, though it can be quenched and collected in a water bath where the thermal shock creates a relatively small aggregate waste form. The metal components are heavier than the slag, collect in the metal bath and are tapped separately from the slag. If the metals are mixed, subsequent processing would be necessary for separation and recovery of the metals of interest (Reference 1).

The CEP design does not provide for secondary combustion of organics because MMT believes complete organic destruction will occur in the bath and in the high temperature reactor space above it. The KFA-KTH pilot plant will have either oxygen injection in the reactor above the bath or a separate secondary combustion chamber (Reference 2). The CEP unit could be provided with a secondary chamber as well.

5.2.3 Material and Energy Balances

Figure E.2. is a graphical representation of the mass flow for a molten bath furnace treating 1000 lb/hr of an inert waste. In this case, the waste contains no combustible material, however, some offgas is generated from the vaporization of moisture in the feed. The only gases entering the furnace are from air leaks and purges on the feed system. Electrical energy is supplied through submerged electrodes or induction heating. This process results in a minimum offgas volume. During processing of an actual waste, the offgas flow is likely to be higher because of small amounts of organic materials in the feed. Also, decomposition of inorganic materials such as nitrates and carbonates will release gases.

Figure E.3. is a graphical representation of the energy flow for the molten metal bath process shown in Figure E.2. Approximately 77% of the electrical energy input is used to melt the solid inert material. The remaining energy is lost as heat through the furnace walls and as hot offgas.

Treatment of a waste with even a small organic content will dramatically alter the material and energy balances. Waste with only 8% carbon, resulting in a heating value of 1120 Btu/lb, would release energy equivalent to the electrical input shown in Figure E.3. Complete combustion of this waste would result in 10 times the offgas volume. Thus, although this is a low volume offgas process, the offgas flow rate can easily be dominated by a small amount of combustible material in the feed.

5.3 TECHNOLOGY ANALYSIS

5.3.1 Advantages

The CEP operates at high temperatures and with sufficient residence time within the bath to achieve highly efficient destruction of organics. The high energy density of the molten metal allows shorter residence times for complete destruction. The high temperature operation results in vitrification of inorganic solids and produces a slag that is likely to be a highly stable and leach resistant waste form. Operation of the technology in a reducing mode minimizes the high NO_x that would otherwise be generated by the high temperatures. The technology has a long history of the building and operation of large scrap steel converts. The use of oxygen injection instead of air results in low volumes of offgas emissions and the introduction of waste below the surface of the bath results in high particulate retention and thus low offgas carryover. The large thermal mass of the metal bath results in a more stable and uniform temperature environment than open flames and a greater likelihood of complete organic destruction. Nonflame combustion ensures that the offgas is free of contamination from products of incomplete fuel combustion.

CEP reactor vessels are simple and compact. As in most vitrification devices, the fly ash may be returned to the reactor for reprocessing. Radiologically contaminated metal could be used for the bath, providing a method of decontamination, since radionuclides tend to migrate to the slag and volume reduction by eliminating void spaces.

5.3.2 Disadvantages

The high operating temperatures of the CEP will result in relatively high volatilization of heavy metals and radionuclides into the offgas stream, requiring special attention for the removal of these constituents in the offgas treatment system.

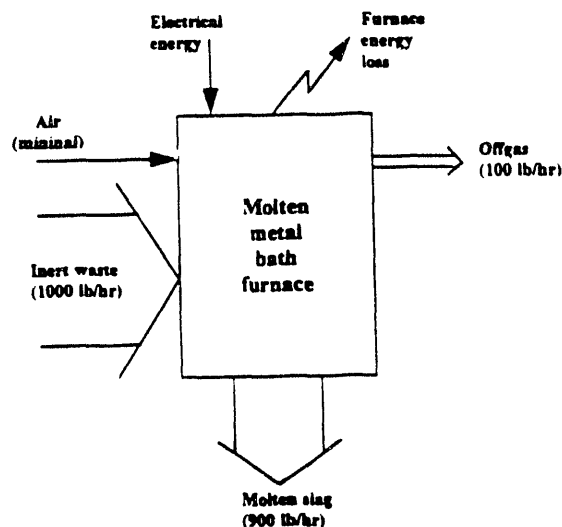


FIG. E.2. Graphical Representation of Mass Flows for a Simplified Molten Metal Bath Process with Inert Feed (HHV = 0.0 Btu/lb).

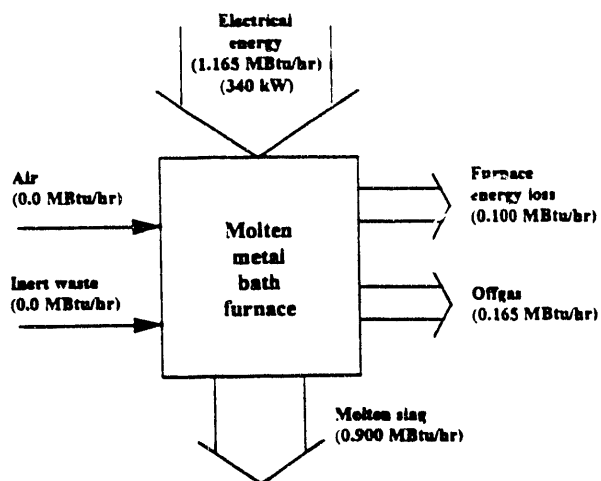


Fig. E.3. Graphical Representation of Energy Flows for a Simplified Molten Metal Bath Process with Inert Feed (HHV = 0.0 Btu/lb).

Solid waste must be size reduced extensively for subsurface injection, or else dropped into the bath from above, which may reduce the effectiveness of the process to completely treat the waste (i.e. the waste may volatilize and be carried directly into the offgas without complete destruction). Introduction of significant quantities of noncombustible liquids or wet solids directly into the metal bath has the potential of causing a steam explosion. Slag chemistry is dependent on waste characteristics and may be difficult to optimize with a heterogeneous waste feed. If the metal in the bath is not radioactively contaminated to start with, it will be considered contaminated after exposure to mixed waste.

5.3.3 State of Development

The development work performed at U.S.S. between 1982 and 1986 verified the principles of operation and could be considered bench scale development, though feed rates of up to 22,000 tons/year were demonstrated on a variety of carbon-containing streams. The German/Swedish effort is in the pilot demonstration stage, but no data are available on demonstration results.

MMT claims their technology is commercially available and has developed designs for facilities ranging in capacity from 5,000 to 100,000 tons/year. However, no commercial units have been constructed to date. Realistically, the technology will require actual waste processing at pilot scale before it can be considered commercially available.

5.3.4 Research, Development, or Demonstration Needs

Primary research needs to demonstrate the technology's effectiveness in hazardous waste destruction. Other needs include the following: investigation of partitioning of radionuclides and heavy metals in the metal bath and slag; determination of slag chemistry sensitivity to variations in waste content; and, leachability characteristics of the slag. The technology has not been demonstrated for soil applications.

5.4 POTENTIAL FOR INEL APPLICATIONS

The potential application of this technology to the INEL buried wastes is difficult to assess given the state of development. If the technology proves to be effective at processing large solid wastes without extensive pretreatment by introduction into the metal bath from above the surface, then it is likely that most buried wastes could be processed. However, the technology has been demonstrated to be most effective when the waste is gaseous, liquid, or fluidized solid fines and is injected under the bath. Extensive pretreatment of most buried wastes would be required to yield solid fines capable of fluidization in order to pneumatically transport the solids for introduction into the molten metal bath beneath the surfaces.

5.5 REFERENCES

1. Smith, Jeffrey D., "Molten Metal Technology", EI Digest, pp. 8-13, July 1991.
2. Axelsson, Carl-Lennart, "KFA-KTH Joint Investigation on a Process for Hazardous Waste Destruction in an Iron Melt", Stockholm, Sweden, May 1988.
3. Zimmer, Erich, "Treatment of Hazardous Wastes in an Iron Melt", KFA Julich GmbH, Julich, Germany, September 1988.
4. Yates, Ian C. and Johnston, James E., "Resource Recovery With Catalytic Extraction Processing", Environmental Waste Management, pp. 30-31, May 1991.

APPENDIX F

Mixed Waste Integrated Program Technology Needs Statement and Call for Proposals FY94¹

¹ Information contained in Appendix F was obtained from the following document: *Mixed Waste Integrated Program Technology Needs Statement and Call for Proposals*, Mixed Waste Integrated Program, FY-94.

WASTE DESTRUCTION/STABILIZATION

BACKGROUND INFORMATION

The Waste Destruction/Stabilization TSG is identifying and evaluating innovative technologies that fill key waste treatment needs or offer significant improvements over baseline technologies for waste destruction, recovery, or stabilization. The TSG provides guidance to the MWIP and industry so that technology development can be effectively focused on DOE identified needs. Specifications and criteria used to identify technology needs and evaluate proposed technologies for meeting those needs are listed in the last section of this report. New and/or developing technologies will be compared with existing technology by using accepted standards.

With the assistance of members, consultants, and experts in the field, the TSG is compiling a list of current and emerging technologies for addressing applicable waste streams. Current technologies will comprise those that are presently industrial use and will serve as baseline methods. Emerging technologies will comprise those that are still under development, but have shown promise for addressing needs and concerns.

A baseline technology inventory for mixed waste destruction, reduction, or stabilization will be generated for input from the MWTP waste treatment plant flow sheets. Additional baseline technologies extracted from DOE, EPA, commercial, and foreign experience on similar waste will be evaluated to ascertain the current best state-of-the-art technology. Technology needs are then evaluated based on either a technology hole in the flow sheets or a baseline technology for which significant concerns about performance, economics, or permitting are a major concern. Technology development activities that may pertain to mixed waste from all available sources will be evaluated and those that meet the needs criteria will be recommended for support. Proposals for additional new and innovative technologies that may offer significant improvements over the baseline will be solicited.

The use of thermal treatment technologies for waste destruction and stabilization is being investigated. Innovated technologies, loosely defined as those technologies that are not currently being used on a large scale to treat wastes and/or significant extensions of current technology beyond its demonstrated capabilities must be developed, and issues regarding full-scale operation must be resolved. Development of the following processes is currently in progress: metal-melting technologies, and plasma-arc incineration. Metal-melting technologies are basically adapted from the metals industry (e.g., induction furnaces and plasma-arc melters). Although the operating principles for these processes are not new, there is only limited operational experience in the waste management area. In addition, new concepts in melting processes should be researched as waste management tools (e.g., the microwave melter).

Melter technologies hold the promise of being highly effective for waste treatment because the residue may require little or no additional treatment prior to disposal. The treated waste form may be physically and chemically stable enough to pass regulatory

standards. Because of the high temperature of melting operations, melters can be used to remove or destroy regulated organics. Process modifications will be required to ensure the destruction of organic material during the metal-melting process. Consistency and reproducibility of treated waste treatment processes is required for regulatory approval.

Melters are ideally suited for inorganic waste streams such as inorganic oxides and elemental metals. Furthermore, the chemistry in the melt can be reducing or oxidizing depending on the type of waste form desired. When processing oxides, the final waste form will be a glass or a ceramic, primarily depending on the rate of cooling. When processing metals, the melt will form a top layer of slag and a bottom layer of molten metal. The slag can then be separated from the molten metal, allowing for the recycle of the molten metal. Depending on operating parameters, it is possible to oxidize the majority of the radionuclides in the waste so the radionuclides become part of the slag, resulting in a decontaminated molten metal.

The baseline technologies for organic destruction and volume reduction for all the waste streams except aqueous waste and metals are various forms of incineration. Alternative processes need to be evaluated and developed as a potentially more effective method of treatment, a more economical method, particularly for small volume waste streams, and, because of the permitting concerns for incineration, as a potential fall-back process. Currently the state of the art for alternatives to incineration is being evaluated to expand DOE options on waste treatment. A preliminary list of technology development options is included in section entitled "Alternatives to Incineration."

Research and development are needed in the waste destruction area. A primary need in the area of waste destruction and stabilization is to develop thermal treatment technologies that are capable of treating a wide variety of DOE mixed wastes. Ideally, these technologies will minimize the need for waste characterization, handling, sorting, and pretreatment. These technologies will destroy or stabilize the hazardous constituents so that all applicable DOE orders and regulatory requirements are met. Generally, these needs can be classified as requiring more operational experience, better materials of construction, improved materials handling techniques, less waste pretreatment, control of the chemistry in the process, and detailed analysis of the resulting residue and off-gas to determine the constituents that are in the process effluent streams. A secondary consideration is to produce an enhanced final waste form that will pass EPA leach tests and meet the Land Disposal Restrictions. The technologies must be acceptable to the general public and must be permitable by appropriate regulatory agencies. Additionally, it is desirable to minimize the generation of secondary wastes and to minimize pollutant releases to the offgas.

Ultimately, technologies that meet these criteria and that are selected for funding will be demonstrated from their current level of development through pilot-scale tests. When the technology has been successfully demonstrated as a single unit, the process will be incorporated into a pilot-plant system to simulate performance within the context of the prototypical treatment scheme. This phase of the work will be conducted in conjunction with Office of Waste Management staff. Initial testing

should be conducted on a wide variety of waste types to demonstrate the versatility of the technology. Further testing will be used to identify and solve problems associated with the technology and to characterize secondary wastes and effluents. Testing in out years will determine important parameters such as equipment configuration, capital and operating costs, and operational variables.

The following needs are listed in priority order. Principal investigators should refer to the evaluation criteria in the last section of this document for guidance on information that is of interest to the MWIP. Each TTP directed to development and demonstration of technology hardware or processes must also include a simple process flow diagram illustrating how the technology fits into the baseline (Mixed Waste Treatment Project) flow diagram. This simplified flow diagram should clearly identify:

- technology location in the prototype process flow
- generic waste streams that are highly applicable for the technology and required pretreatment
- input waste pretreatment requirements
- input/output waste streams
 - solid wastes and byproducts
 - liquid wastes and byproducts
 - consumable materials (e.g., catalysts, filters, electrodes)
- process volume ratio (effluent volume divided by influent volume)
- byproduct/effluent treatment requirements
- decontamination/decommissioning and disposal requirements
- potential for process automation

SPECIFIC NEEDS

1. Metals processing

Metal recovery and recycle for use within the DOE complex is a treatment objective for mixed waste processing. A determination of the ability to substantially partition the radionuclides and possible the heavy metals into the slag phase needs to be accomplished. A technology status review for this process is currently in progress. A technology development program to measure actual partition coefficients and methods for enhancing those coefficients is needed but standards for acceptable levels of contaminants for recycled metals will be need as guidance from DOE.

RDDT&E on melter technologies (e.g., induction) that can be used to treat inorganic mixed wastes is needed. These melters will be used to decontaminate, volume reduce, and/or stabilize ferrous and nonferrous metals, oxides, noncombustible solids, sludge and potentially small amounts of organic. This RRDT&E work will involve engineering studies and technology demonstrations to bring the technology demonstrations to implementation on a full-scale basis. This work may be conducted in any or all of three phases, depending on the development status of the technology, and will consist of all or some of the following tasks:

Phase 1 -- Demonstrate the technology's viability and determine the technology's capabilities

- Perform proof of principle tests including preliminary mass balances
- Demonstrate the technology on a wide variety of inert wastes to determine the ability to destroy or stabilize hazardous constituents and organics (if present)
- Determine waste loading that can be used
- Determine characteristics of residues and additional treatment requirements of residue
- Investigate the possibility of decontaminating metals for recycling
- Determine characteristics of offgas and the types of pollutants in offgas to be treated
- Determine partitioning of radionuclides and toxic metals in the metal, slag, and offgas.
- Identify problems with technology and investigate potential solutions to problems

Phase 2 -- Establish Operational Parameters

- Evaluate system thermodynamics
- Determine melt chemistry and the need to use additives
- Through long term testing, optimize operational parameters (e.g., temperature, redox state)
- Determine throughput capabilities
- Determine quantity of secondary wastes produced
- Demonstrate a consistent product over a long term operations
- Identify limitations to technology
- Test and verify solutions to operational problems previously identified
- Establish oxidation conditions to completely convert metals to oxides

Phase 3 -- Determine Conceptual Design Requirements for use in Pilot Plant Tests

- Identify and demonstrate materials of construction to maximize component life
- Develop pilot-scale configuration of system including subsystems and components such as waste feeding, residuals removal and handling, and pretreatment requirements
- Estimate life-cycle costs

- Deliverables:
- a. Test plan and technology development approach including preliminary data on comparison of proposed process with available alternatives.
 - b. Results of laboratory-scale test including radionuclide data for selected waste stream(s) as appropriate.

- c. Plans for pilot-scale testing including assessment of the potential use of Pacific Northwest National Laboratory pilot-plant.
- d. Results of pilot-scale tests.

EM-30's earliest schedule for development of prototypical treatment and Federal Facility Compliance Agreement-based regulatory drivers require that deliverable "a" be drafted by early FY94. Deliverable "b" should be nearing completion by the end of FY94, and "c" must be in progress early in FY95 and continued as appropriate. Preliminary results for deliverable "d" must be available by the end of FY95 and will continue as necessary.

2. Flexible thermal treatment technologies

Research, development, demonstration, testing, and evaluation (RDDT&E) of emerging versatile thermal technologies that can treat a wide variety of DOE waste streams is needed to simplify waste processing, simplify permitting requirements, and lower life cycle costs. Processes that are capable of handling heterogeneous raw waste such as mixes of metals, inerts, and organics are of interest. This work will involve engineering studies and technology demonstrations to bring the technology to implementation on a full-scale basis. Development of a plasma-based process is currently in progress; other robust thermal treatment technologies should be considered for inclusion in this development effort. Experience using flexible thermal treatment technologies should be compared and "lessons learned" from waste processing should be applied to mixed waste treatment. One alternative process is the glass melter; this technology is being developed under the Final Waste Forms technical area. The two processes will be compared as to their effectiveness and the advantages and disadvantages of each.

Phase 1 --Demonstrate the technology's viability and determine the technology's capabilities

- Perform proof of principle test including preliminary mass balances
- Demonstrate the technology on a wide variety of waste types to determine the range of wastes that can be treated
- Determine ability to destroy and/or stabilize hazardous constituents
- Determine characteristics of residues and additional treatment requirements of residues (an enhanced final residue which does not require additional treatment is most desirable)
- Investigate feasibility of recycling residues through process if additional treatment is required
- Determine characteristics of offgas and the types of pollutants in offgas to be treated
- Determine partitioning of radionuclides and toxic metals

- Identify problems with technology and investigate potential solutions to problems
- Demonstrate ability to destroy large quantities of combustibles

Phase 2 -- Establish Operational Parameters

- Evaluate systems thermodynamics
- Investigate process chemistry and the need for additives
- Through long-term testing, optimize operational parameters (e.g., temperature, pressures)
- Determine throughput capabilities
- Determine quantity of secondary wastes produced
- Demonstrate a consistent product over long term operations
- Identify limitations to technology
- Test and verify solutions to operational problems previously identified

Phase 3 -- Determine Conceptual Design Requirements for use in Pilot Plant Tests

- Identify and demonstrate materials of construction to maximize component life
- Develop pilot-scale configuration of system including subsystems and components such as waste feeding, residuals removal and handling, and pretreatment requirements
- Estimate life-cycle costs

- Deliverables:
- a. Test plan and technology development approach including preliminary data on comparison of proposed process to baseline.
 - b. Results of laboratory-scale tests including radionuclide data for selected waste stream(s) as appropriate.
 - c. Plans for pilot-scale testing including assessment of the potential use of Office of Waste Management pilot-plant.
 - d. Results of pilot-scale tests.

EM-30's earliest schedule for development of prototypical treatment requires that deliverable "a" be drafted by early FY94. Deliverable "b" should be nearing completion by the end of FY94, and "c" must be in progress early in FY95 and continued as appropriate. Preliminary results for deliverable "d" must be available by the end of FY95 and will continue as necessary.

3. Sampling and characterization of thermal treatment emissions

A consistent methodology for monitoring combustion processes should be developed and implemented for all experiments conducted using thermal treatment under the MWIP. This methodology will ensure that data are comparable and that alternative processes can be directly compared as to performance, risk and life-cycle

cost. Commercial capability in this regard should be surveyed and utilized as appropriate.

Emissions of metals and organics (i.e., products of incomplete combustion) from incinerators are of public concern and can prevent this technology from being permitted. An aggressive program to understand the mechanisms involved in the metal emission process and in devising methods to reduce those emission to regulatory and publicly acceptable levels is needed. An initial program to model the emission process and to verify that model should be a primary objective for FY94.

4. Alternatives to Incineration

The MWIP is interested in proposals on alternatives to incineration such as wet air oxidation (catalyzed and non-catalyzed), calcination, supercritical water oxidation, steam reforming, metal and glass melting, and microwave processing. Non-thermal treatment technologies of interest include: biotreatment, electrochemical oxidation, electron beam and other radiolytic methods, silent discharge plasma and other ozonation methods, corona discharge, UV mediated oxidation, sonochemical destruction, and supercritical CO₂ extraction. Waste streams that are considered candidates for these technologies are: nitrates, chlorides, radioactively contaminated metals, tritium contaminated mixed wastes, PCB contaminated materials, ion exchange resins, fluid organics, aqueous waste contaminated with organics, and plastics and other room trash (including cellulose). This is a broad category of need that is not as strongly linked to the implementation of the MWTP flowsheet as other areas. Principal investigators ideas are of interest to the MWIP.

An example of the need for alternatives to incineration is the need to consider technologies to replace wet air oxidation. The aqueous waste treatment streams baseline treatment technology for MWTP is a conventional wet air oxidation process. This process suffers the disadvantages of high pressure, low organic feed concentration capabilities, limited waste stream acceptability and lack of predictability, and typically not meeting EPA treatment standards. More robust and effective technologies need to be explored to replace a baseline technology for which major concerns as to performance and permitting are issues. Significant effort to find mature technology improvements or replacements for the wet air oxidation process are needed. Catalyzed wet air oxidation is a mature technology that may better meet the MWTP needs. Commercial availability should be pursued and the state-of-the-art assessed to determine if additional development is required.

The proposed technology should be clearly identified as to why it is an alternative to incineration. That is, is it an alternative because it will address a waste stream that is not amenable to treatment by incineration or is it an alternative to the process of incineration. An example of the waste-stream specific alternative is the treatment of vermiculite which does not burn. TTPs should discuss the improvement of proposed technology over baseline technology (i.e., incineration and wet air oxidation). An alternative process may be more applicable to small problem waste streams; the process need not handle a wide range of waste streams to be of interest to the MWIP.

Desirable advantages for prospective alternatives to incineration are: handle wastes not handled by incineration or handle a wide range of wastes equal to those handled in incineration; technically surpass incineration capabilities; produce more benign wastes, smaller volumes, or no secondary waste; reduce or eliminate fugitive emissions; and operate at lower risk.

This RDDT&E work will involve engineering studies and technology demonstrations to bring the technology to implementation on a full-scale basis. This work may be conducted in any or all of three phases, depending on the development status of the technology, and will consist of all or some of the following tasks:

Phase 1 -- Demonstrate the technology's viability and determine the technology's capabilities

- Perform proof of principle tests including preliminary mass balances
- Demonstrate the technology on a wide variety of waste types to determine the range of wastes that can be treated
- Determine ability to destroy and/or stabilize hazardous constituents
- Determine characteristics of residues and additional treatment requirements of residues
- Investigate potential to purify water so it can be recycled or meet regulatory requirement for release to the environment
- Determine characteristics of offgas and the types of pollutants in offgas to be treated
- Identify problems with technology and investigate potential solutions to problems

Phase 2 -- Establish Operational Parameters

- Investigate process chemistry and the need for additives
- Through long-term testing, optimize operational parameters (e.g., temperature, pressures)
- Determine throughput capabilities
- Determine quantity of secondary-wastes produced
- Demonstrate a consistent product over long term operations
- Identify limitations to technology
- Test and verify solutions to operational problems previously identified

Phase 3 -- Determine Conceptual Design Requirements for use in Pilot Plant Tests

- Identify and demonstrate materials of construction to maximize component life
- Develop pilot-scale configuration of system including subsystems and components such as waste feeding, residuals removal and handling, and pretreatment requirements
- Estimate life-cycle costs

- Deliverables:
- a. Test plan and technology development approach including preliminary data on comparison of proposed alternative process with available technology.
 - b. Results of laboratory-scale tests including radionuclide data for selected waste stream(s) as appropriate.
 - c. Plans for pilot-scale testing including assessment of the potential use of Lawrence Livermore National Laboratory pilot-plant.
 - d. Results of pilot-scale tests.

EM-30's earliest schedule for development of prototypical treatment requires that deliverable "a" drafted by the end of FY94. Deliverable "b" should be nearing completion by the end of FY95, and "c" must be in progress early in FY95 and continued as appropriate. Preliminary results for deliverable "d" must be availability by the end of FY96 and will continue as necessary.

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