

Biofuel alternatives to ethanol: pumping the microbial well*

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Engineered microorganisms are currently used for the production of food products, pharmaceuticals, ethanol fuel and more. Even so, the enormous potential of this technology has yet to be fully exploited. The need for sustainable sources of transportation fuels has generated a tremendous interest in technologies that enable biofuel production. Decades of work have produced a considerable knowledge-base for the physiology and pathway engineering of microbes, making microbial engineering an ideal strategy for producing biofuel. Although ethanol currently dominates the biofuel market, some of its inherent physical properties make it a less than ideal product. To highlight additional options, we review advances in microbial engineering for the production of other potential fuel molecules, using a variety of biosynthetic pathways.

Introduction

Conversion of biomass to biofuels has been the subject of intense research efforts since the 1970s. This work has recently gained significant political and scientific momentum owing to concerns about climate change, global energy security and petroleum supply. Global energy usage is projected to nearly double in the next two decades [1–3], and biological fuel production might serve as a sustainable, carbon-neutral energy source compatible with current engine technology. In an effort to offset increases in consumption and to limit the fossil fuel-related negative impacts on the environment, the US Department of Energy has established the goal of supplanting 30% of gasoline consumption with cellulosic ethanol by 2030 [2]. Similarly, a European Union Directive (2003) aims to replace 5.75% of all gasoline and diesel transport fuels with biofuels by 2010 [4].

Owing to physical and political limitations on arable land, it is believed that future biofuels will, by necessity, originate from abundantly available lignocellulosic bio-mass [5]. The liberation of monomeric sugars from this biomass is the subject of intensive research efforts that might be the key to the sustainability of any biofuels process. This extremely important area of research has been reviewed elsewhere [6,7]. Here, we concentrate on another important factor for the cost-effective production of biofuels from lignocellulosic feedstocks – namely the conversion of biomass hydrolysates (monosaccharides) to target molecules. We present some recent developments in

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microbial pathway modifications as a means of producing both currently used and prospective biofuels from agriculturally prevalent sugars.

The use of microbial systems for the production of industrially relevant compounds has seen substantial gains in the past decade as a direct result of the genomics revolution. Further advances in gene regulation, protein engineering, pathway portability, synthetic biology and metabolic engineering will propel the development of cost-efficient systems for biofuel production. The current range of biofuels consists primarily of microbially derived ethanol and plant-based biodiesel ([Box 1](#)). Although bio-diesel is favored in several European countries, ethanol dominates the majority of the world biofuel market, including that of the United States. The pros and cons associated with any potential biofuel present challenges and opportunities that are best addressed on a case-by-case basis, and these issues are therefore beyond the scope of this review. Instead, we cover advances in metabolic engineering to produce biomolecules that are currently being explored for the development as transportation fuels ([Figure 1](#)).

Pathway Engineering

Sugar catabolism and the fermentation pathway

Agriculturally derived lignocellulosic hydrolysates contain predominantly glucose and xylose and small amounts of arabinose, galacturonic acid and rhamnose, in addition to a large percentage of lignin residues (not addressed here). For the economical bioconversion of these feedstocks to commodity chemicals, efficient use of a high percentage of available monosaccharides is crucial. Although glucose metabolism is a nearly universal trait of living organisms, the ability to metabolize other sugars varies widely within microbes. Therefore, catabolism of monosaccharides has been an important target in the development of an optimized microbial platform. Pathways for xylose and arabinose use have recently been engineered into *Saccharomyces cerevisiae* [8–10] and *Zymomonas mobile* [11]. Although developments in sugar catabolism were made in the context of homo-ethanol production, the advances are applicable to the production of a variety of fuel molecules through multiple biosynthetic pathways (discussed below).

Microbial production of ethanol has been reviewed extensively [12–14]. *S. cerevisiae*, the most commonly used organism for ethanol production, and *Z. mobilis* are known to be homo-ethanol fermenters, and wild-type *Escherichia coli* ferments sugars to a mixture of ethanol and organic acids ([Figure 1](#)). A vast amount of work has been done to engineer *E. coli* strains capable of homo-ethanol production [14]. By chromosomal integration of the production-of-ethanol (PET) operon from *Z. mobilis*, Ingram and coworkers [15] were able to engineer *E. coli* strain KO11 for homo-ethanol production [12]. An evolved version of this strain is currently being tested for commercial applicability [16]. More recently, Kim et al. [17] reported the construction of an *E. coli* strain SE2378 that relies only on native enzymes to produce ethanol from glucose and xylose.

Fermentation is a source of numerous other products. The diversity of the end-products from fermentation depends on the reactions by which pyruvate, a primary intermediate of this pathway, is metabolized and how NAD is regenerated from NADH.

Two products of bacterial fermentation have gained prominence as potential biofuels in recent years – hydrogen and butanol.

In a recent study, Kawaguchi and coworkers [18] reported the construction of a hydrogen-producing strain of *E. coli* that lacks the lactate- and succinate-producing pathways found in the wild-type strain and that directs glucose metabolism toward hydrogen production by over-expression of the formate hydrogen lyase (FHL) complex (Figure 1). This strain is capable of producing hydrogen at a yield of 1.82 mol/mol glucose, significantly above that of the wild-type strain (1.08 mol/mol glucose) [18]. Efforts in the microbial photobiological production of hydrogen from water have been reviewed extensively elsewhere [19].

Butanol has been proposed as a viable alternative to ethanol owing to the fact that it has a higher energy content and lower solubility in water. It can be transported through existing pipelines and can be used to supplement both gasoline and diesel fuels. Historically, butanol has been produced biologically using *Clostridium acetobutylicum*. *Clostridium beijerinckii* BA101 has also been studied for its acetone–butanol–ethanol (ABE) fermentation capability [20]. In contrast to much of what is covered in this review, many of the butanol production capabilities of these solventogenic Clostridia have been achieved through conventional mutagenesis techniques [21,22]. Total ABE production by strains of *C. beijerinckii* BA101, *C. acetobutylicum* PJC4BK and *C. acetobutylicum* P260 is in the range of 25–33 g/L [20], with specific yields of butanol from *C. beijerinckii* BA101 being reported at 11.9–14.3 g/L [23]. Improvements in yields and a better understanding of butanol toxicity are among the potential benefits of engineering the butanol production pathway into a well known host, such as *E. coli* or *S. cerevisiae*.

The butanol example illustrates the pros and cons of any fuel choice. Because of its higher energy content and lower hygroscopicity and corrosivity, butanol appears to be superior to ethanol. However, as mentioned earlier, it is produced at a lower titer and is much more toxic than ethanol. Also, because of its high boiling-point, butanol would require more energy than ethanol for distillation-based purification from a fermentation broth. Various biological and physical engineering approaches reviewed elsewhere [20] are helping to address and overcome many of these challenges.

The fatty acid pathway

Currently, biodiesel production uses plant oils for the vast majority of its starting material [24]. Transesterified fatty acids such as fatty acid methyl ester (FAME) and ethyl ester (FAEE) provide carbon chain lengths compatible with compression ignition engines and have consequently been developed for use as biodiesel. However, if biodiesel is to replace a significant portion of current petro-diesel needs, microbes might serve to provide a more consistent and scalable source for this commodity.

The variety of fatty acids available from microbial sources can potentially provide the mixture of chain lengths and branching required for an ideal fuel-blend (Box 1). Microbial fatty acid biosynthesis and the diversity of naturally occurring fatty acids are well understood [25]. One high-yield source is the oleaginous algae *Botryococcus braunii* [26,27]. In this organism, up to 40% of dry cell weight consists of fatty acids.

Different microalgal species have been shown to accumulate different amounts of lipid, ranging from 40–70% of their biomass. Manipulating growth conditions such as CO₂ supplementation, and nitrogen and light limitation can increase the lipid content of these algae [28]. A more direct replacement of conventional diesel fuel might come from n-alkane production in bacteria. This unusual trait has been reported to occur in *Vibrio furnissii* [29]. Available studies of alkane biosynthesis in this bacterium indicate that the enzymatic mechanism is a novel sequential reduction of n-hexadeca-noic acid to n-hexadecane catalyzed by the enzyme FahR [29]. This work reported that the yield of the C₁₆ alkane from the C₁₆ fatty acid is greater than 70%. However, a recent report disputes these findings [30]. Alkane production can also be facilitated by the decarbonylation of fatty aldehydes (Figure 1) [31,32]. However, the rates of conversion of decarbonylases described to date are too slow for commercial application.

Fatty acid biosynthesis draws from the pool of acetyl-CoA produced by several central metabolic pathways. To improve the free fatty acid levels in microbes, the main approaches adopted to date have focused either on diverting the pool of acetyl-CoA towards fatty acid biosynthesis or on decreasing the cellular consumption of fatty acids. An example of the former method includes an engineered *E. coli* co-expressing several enzymes: an acetyl-CoA carboxylase, the first enzyme in the fatty acid pathway; a truncated thioesterase; and the fatty acid synthase. This leads to an increased production of free fatty acids [33]. Alternatively, to minimize fatty acid catabolism, genes such as *yfcYX-ydiD* in *E. coli* [34], and *faa1*, *faa2* and *faa3* in *S. cerevisiae* [35] have been targeted for deletion or inactivation. Also, deletion of the four genes associated with neutral lipid synthesis in *S. cerevisiae* (i.e. *are1*, *are2*, *dga1* and *lro1*) results in a 2.5-fold increase in the production of neutral fatty acids during late exponential and stationary growth phase [36].

Although fatty acids are the direct products of the fatty acid biosynthesis pathway, reduced and modified fatty compounds have also been made in bacteria. An example of microbial production of fatty alcohols and esters is the production of Jojoba oil-like wax esters and butyl esters in an engineered *E. coli* [37]. A recently isolated enzyme has also been shown to esterify fatty acids to isoprenyl alcohols, further expanding the range of possible bio-esters [38]. These fatty alcohols and esters can be produced with the appropriate chain length required to serve as biodiesel candidates (Figure 1).

Biodiesel from microalgae-derived lipids has been explored extensively [26]. However, these studies still rely on a chemical transesterification to produce the fuel molecules. Recently, Kalscheuer et al. [39] reported an *E. coli*-based process for in vivo transesterification. This FAEE biosynthesis was accomplished in recombinant *E. coli* by co-expression of the ethanol production genes from *Z. mobilis* in combination with the acyltransferase WS/ DGAT (wax ester synthase/acyl-coenzyme A:diacylglycerol acyltransferase) gene from *Acinetobacter baylyi* ADP1 [39]. Although this is an exciting proof of concept, the yields from this system need to be significantly improved for viable industrial application.

The isoprenoid pathway

The isoprenoids are a family of natural products that are synthesized with the use of the activated hydrocarbon monomers isoprenyl pyrophosphate (IPP) and its isomer dimethylallyl pyrophosphate (DMAPP) [40]. It is possible to produce several branched-chain alcohols, alkanes, alkenes and cyclic hydrocarbons through the isoprenoid biosynthetic pathway. For decades, this family of compounds has been posited as a potential source of biofuels [28,41]. In addition to branched-chain hydrocarbons, this pathway can be used to produce isopentanol (isoamyl alcohol) and its acetate ester, compounds proposed as additives for spark ignition fuels [42]. Sources of isoprenoids previously suggested for fuel production have included oil-producing algae such as *B. braunii*. These organisms produce large amounts of isoprenoids as well as fatty acids [27]. Owing to the complex mixture of hydrocarbons, the extract is more akin to a biocrude and therefore would require hydrogenation and cracking, in much the same manner as fossil-derived counterparts [43]. Advances in the understanding and engineering of isoprenoid biosynthesis pathways might facilitate the production of either a similar biocrude or perhaps engine-compatible molecules from microorganisms.

Isoprenoid monomers are produced either from acetyl-CoA through the mevalonate pathway or from pyruvate and glyceraldehyde-3-phosphate (G3P) through the deoxyxylulose phosphate (DXP) (also known as the methylerythritol phosphate [MEP]) pathway [40,44,45]. These five-carbon monomers (i.e. IPP, DMAPP) can be polymerized to chains ranging in length from one to more than a dozen units [44,45]. Viable production of fuel molecules relies on both increasing precursor biosynthesis as well as optimizing several downstream steps.

The majority of the research in this area has come from studies of pharmaceutically important compounds such as lycopene, carotenoids, sterols, taxol and artemisinin [44,46–48]. High-level production of IPP has been achieved in both *E. coli* and *S. cerevisiae*, and many different isoprenoids have been produced with the use of these engineered hosts [49–53]. In the engineering of *E. coli* hosts, the existence of two separate biosynthetic routes has been exploited by introducing the non-native mevalonate pathway. In yeast, the native mevalonate pathway has been improved by several approaches [50,54–56].

These advances in monomer production can be used to produce isoprenoid biofuels in an industrially tractable organism. Isoprenoids (e.g. phytoene [C40]) serve as intermediates in the production of large isoprenoid compounds, and they might be suitable as biocrude. Both *E. coli* and *S. cerevisiae* have been successfully engineered for the production of the phytoene-based carotenoids [51,57–59]; unfortunately, the current yields of these compounds fall far short of what is required if they are to produce viable fuel alternatives [47,50,60]. However, if sufficient titers of phytoene are reached, it could be refined into motor and jet fuel through the infrastructure currently used to process petroleum.

In addition to biocrude production, recent work suggests the isoprenoid pathway can be used to produce the proposed gasoline additives isopentanol and isoamylacetate [42]. For example, researchers have demonstrated that a pyrophosphatase isolated from *Bacillus subtilis* can dephosphorylate IPP to form isopentenol [61]. The acetylation of isopentenol by an engineered *E. coli* has also been demonstrated [62,63]. Additionally, saturated, mono- or diunsaturated monoterpenes and sesquiterpenes might be useful as

diesel and jet fuels. The ability to modify terpene cyclases for the production of novel terpenes will greatly expand the number of potential fuels molecules that can be synthesized microbially [64–66].

Major challenges and future directions

Although known metabolic pathways offer several possible avenues for the biosynthesis of fuel molecules, several other factors need to be addressed before they can be applied in an industrial setting. First and foremost are the physical properties of the potential biofuel molecule. These properties have an impact on everything, from its suitability as a fuel to the purification processes and the mode of its transport to consumers. Although discussion of the properties for the biofuels mentioned herein is far beyond the scope of this review, this idea is nicely exemplified by the brief comparison of ethanol and butanol presented above.

The primary challenge in biofuel production is achieving yields that make these fuels cost-competitive with petroleum-based products. For any of the engineering approaches outlined in this review to be successful, the development of efficient lignocellulosic breakdown to monosaccharides is crucial. The development of this linchpin technology will enable the production of microbial-produced biofuels from crops that have higher growth-rates, better yields, lower soil-impacts, and lower water, fertilizer and pesticide requirements than currently used crops such as oil palms, corn or soybeans [67]. These characteristics should serve to make biofuels more competitive with petroleum, while making them less competitive with food production for arable land.

To date, pathway manipulations have largely been limited to the use of existing genes and conventional gene expression techniques. Although these approaches have yielded increases in target molecule production, a cell-wide approach to metabolic engineering will be required to maximize the rate of biosynthesis in addition to the yield of the desired compound. To this end, there now exists a large body of work on the alteration of transcriptional and translational rates in addition to the development of evolved enzymes to maximize throughput. This knowledge is poised for use in the systematic design of highly efficient biofuel production pathways. This global approach is nicely illustrated by recent publications showing the effect of various transcription factors on ethanol tolerance and production in *E. coli* and *S. cerevisiae* [68,69].

Another important challenge in metabolic engineering is satisfying cellular energetic concerns (i.e. thermodynamic constraints). For the production of bulk chemicals and biofuels such as ethanol, fermentation is ideal. However, commercial production of 1,3-propanediol has been accomplished aerobically using engineered *E. coli* optimized for the redox demands of the corresponding pathway [70]. An important parameter in strain optimization is the balance of energy and cofactor requirements for the metabolic pathways used to generate the target molecule. One example of this is from the production of polyhydroxybutyrate (PHB) in *E. coli* under fermentative conditions. In this study [71], PHB yield was improved by the introduction of UdhA, an NAD⁺–NADP⁺ transhydrogenase, to balance NADH and NADPH levels [71]. This work also illustrates the potential of engineered microbes in producing replacements for petroleum-based consumer products other than fuel (see [Box 2](#) for more examples). Supplanting high-

value petroleum products, such as plastics, with renewable products is an important first step towards creating sustainable sources for the petroleum products upon which we have come to rely in our daily life. With continued improvements in metabolic engineering, we expect microbial production of bulk commodities to become economically competitive with petrochemicals.

To date, the optimization of bioethanol production has involved considerable metabolic engineering, including the use of efficient alternate routes and enzymes from multiple and often non-model organisms. One such widely used strategy is the high-efficiency PET cassette derived from *Z. mobilis* [14]. For success in the production of alternative biofuels, similar metabolic engineering strategies will be necessary; for example, the development of targeted and efficient transport systems, the improvement of the resistance of biofuels producers to the toxic effects of accumulating biomolecules, the optimization of carbon flux to the desired products, and the construction of strains that are robust under industrial process conditions [14,72–74]. Some advances that might have enormous potential in the field of microbial biofuel production include the engineering of a reduced-genome *E. coli* strain that can be used for the systematic design of desired phenotypes [75], and the inter-microbial genome transplantation demonstrated in *Mycoplasma caprilorum* [76]. These examples represent the first steps towards engineering an entire biological system from the ground up. Massive sequencing efforts exploring the biodiversity of culturable organisms as well as metagenomes, such as the Sargasso Sea project [77], have generated a wealth of genetic information. These endeavors are providing novel biosynthetic pathway information that can be used to design optimized systems for every fuel category. The rapid advances seen in the development of these technologies will almost undoubtedly facilitate the efficient and reliable production of systems for novel biofuels. Ultimately, political forces, process economics, engine technology, and supply infrastructure will dictate the widespread acceptability and use of these alternate fuel sources.

Box 1. Current motor fuels and their biological counterparts

Gasoline is a complex mixture of hydrocarbons composed primarily of branched-chain alkanes and aromatics ranging from 4 to 12 carbons in length [78]. Similarly, petroleum diesel (No. 2) is a mixture of hydrocarbons ranging from 9 to 23 carbons in length, with an average length of 16 carbons. The types of hydrocarbons in gasoline and diesel have a strong impact on the properties of the fuel. For example, branching and unsaturation leads to greater octane numbers in gasoline [79] and lower cetane numbers in diesel [80]. Conversely, n-alkanes have higher cetane numbers and lower octane values. These differences highlight the need for multiple alternative fuels to supplant petroleum-based counterparts.

To date, research into gasoline substitutes has focused largely on ethanol. However, biosynthesis pathways in microbial systems, both natural and engineered, yield molecules that are similar or identical to those currently found in gasoline. These

include straight- and branched-chain alkanes and alkenes from fatty acid and isoprenoid pathways, in addition to several alcohols and esters ([Figure 1](#)). Establishing high titer-production of these molecules through the use of microbial cultures could serve to replace some or all of the stock currently coming from petroleum.

Currently, plant oils constitute the major biological source for diesel fuel alternatives (biodiesel). However, even the most efficient plant-based production is insufficient to meet current diesel usage levels without a dramatic increase in cultivation [\[81–83\]](#). Research into the production of compounds from fatty acid pathways suggests that straight-chain alkanes, identical to those found in petroleum diesel, and fatty alcohols can also be derived from microbial sources. Although currently far from industrial application, the production of biodiesel by an engineered strain of *E. coli* [\[39\]](#) exemplifies an alternative means of producing fatty esters that has been enabled by microbial engineering.

In addition to the hydrocarbons that constitute the bulk of transportation fuels, additives are used to improve the performance and stability of both gasoline and diesel fuel. Gasoline additives include anti-knock agents, anti-oxidants, lead scavengers, lubricants and detergents. Diesel fuel additives are generally used to eliminate the gelling of the fuel in cold weather. Two classes of commonly used additives or oxygenates are alcohols and ethers, both of which reduce emissions and replace the previously used toxic octane enhancers, such as benzene, and anti-knock agents such as lead.

In addition to ethanol, other biologically derived alcohols, such as butanol and isopentanol, can increase the octane rating of gasoline without significantly decreasing the energy density [\[42\]](#). In the past, butanol was commercially produced through the use of microbial cultures, and isopentanol is currently accessible at the laboratory scale. Advances in bioengineering technologies might eventually provide an economically competitive means of producing these fuel additives.

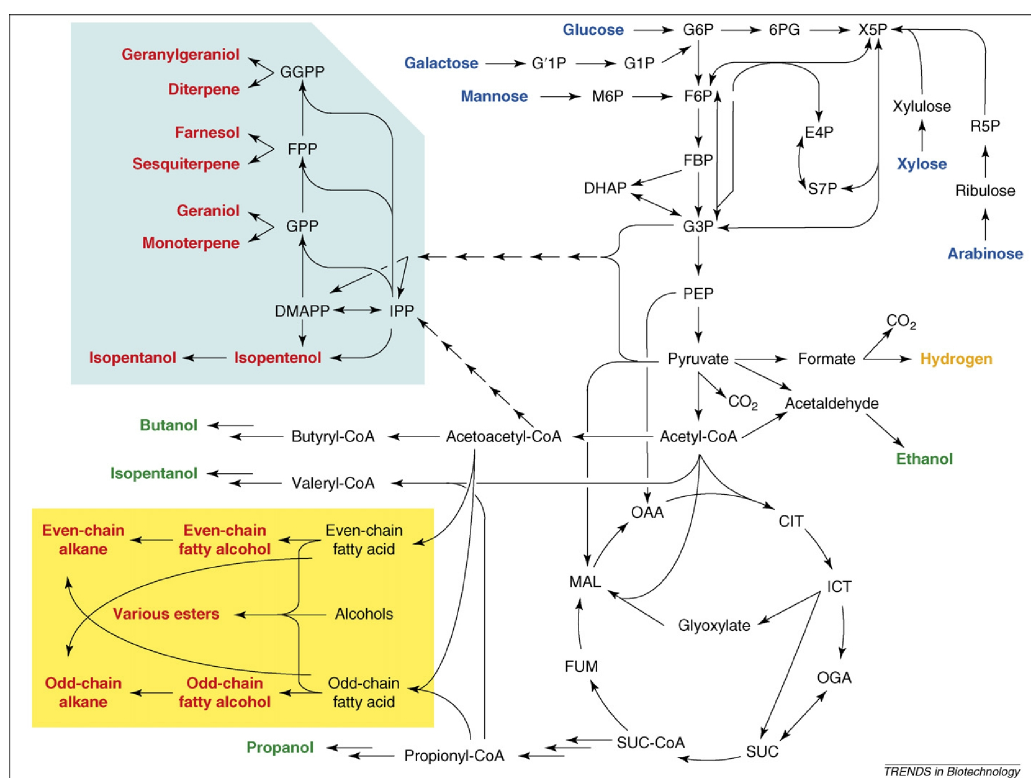


Figure 1. Central metabolic pathways and the potential fuel molecules derived from them. The blue box indicates isoprenoid pathways and isoprenoid-derived molecules. The yellow box indicates fatty acid pathways and fatty acid-derived molecules. Short-chain alcohols are shown in green text. Biomass-derived sugars are indicated in blue text. Abbreviations: 6P,G, 6-phosphogluconate; CIT, citrate; DHAP, dihydroxyacetone phosphate; DMAPP, dimethylallyl pyrophosphate; E4P, erythrose-4-phosphate; F6P, fructose-6-phosphate; FBP, fructose-1,6-bisphosphate; FPP, farnesyl pyrophosphate; FUM, fumarate; G⁰ 1P, galactose-1-phosphate; G1P, glucose-1-phosphate; G3P, glyceraldehyde-3-phosphate; G6P, glucose-6-phosphate; GGPP, geranylgeranyl pyrophosphate; GPP, geranyl pyrophosphate; ICT, isocitrate; IPP, isopentenyl pyrophosphate; M6P, mannose-6-phosphate; MAL, malate; OAA, oxaloacetate; OGA, 2-oxoglutarate; PEP, phosphoenolpyruvate; R5P, ribose-5-phosphate; S7P, sedoheptulose-7-phosphate; SUC, succinate; SUC-CoA, succinyl coenzyme A; X5P, xylulose-5-phosphate.

Box 2. Microbial sources of other petrochemical commodities

The petrochemical industry is a source of a variety of other commodities in addition to fuel, Industrial production using engineered microbes also has the potential to replace other crude-oil-derived products [14,84]. Historically, *Clostridium* strains have been used for the production of a variety of biosolvents, such as acetone, butanol and ethanol [21]. More recently, *E. coli* strains have been engineered by Dupont/Genencor for the industrial production of 1,3-propanediol from glycerol through an aerobic process [70]. Microbial production of both biodegradable plastics (e.g. PHA (polyhydroxyalkanoate), polylactic acid [85,86]) and non-biodegradable plastics (e.g. polythioesters [87]) provides a range of biobased polymers and represents an emerging field. For example, Metabolix Inc (Cambridge, MA) has commercialized a cost-effective method to produce a PHA-based plastic (Mirel™) through the fermentation of corn sugar [88]. Similarly, adipic acid, which serves as a precursor in nylon production, has been produced with the use of an engineered strain of *E. coli* [89].