

1 **Analysis of permeability change in dissolving rough fractures**
2 **using depth-averaged flow and reactive transport models**

3 Lichun Wang^{1, 2, 3*}, M. Bayani Cardenas¹

4 ¹ Department of Geological Sciences, University of Texas at Austin, Austin, Texas, USA

5 ² now at Institute of Surface-Earth System Science, Tianjin University, Tianjin, 300072, China

6 ³ Tianjin Key Laboratory of Earth Critical Zone Science and Sustainable Development in Bohai

7 Rim

8 * corresponding author: wanglichun@tju.edu.cn

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10 **Highlights:**

11 (1) A 2D depth-averaged flow and reactive transport model is developed for dissolving
12 rough fractures.

13 (2) The model showed that the linear time-dependence of permeability due to self-
14 reinforcing dissolution is ubiquitous.

15 (3) The time rate of change of permeability is constant which is consistent with theory and
16 experimental observations.

17
18 **Key words:** Permeability, Fracture, Time, Heterogeneity, Acidic fluid, CO₂

19

20 **Abstract:**

21 Many geophysical problems and engineering applications involve fluid flow through
22 evolving faults and fractures. These situations are typical in CO₂ injection scenarios where
23 supercritical CO₂ or CO₂-saturated fluids are able to dissolve the solid matrix. Dissolution or
24 precipitation within a fracture could lead to self-reinforcing or self-limiting behavior due to
25 feedbacks from changes in permeability (k). There is limited predictive understanding of this
26 phenomenon. Here, we investigated the case of fracture dissolution using a two-dimensional
27 depth-averaged reactive transport model. The model simulated an acidic fluid dissolving a three-
28 dimensional rough fracture. We observed different dissolution patterns across a wide range of
29 geometric properties and different Peclet and Damkohler numbers (960 cases total). Despite this
30 variation, a persistent linear increase in k with time (t) roughly manifested once the dissolution
31 front advanced through the fracture. The time rate of change of k (k/t) is roughly constant for all
32 simulations. The modeled rate is consistent with a theoretical value based on the parallel plates
33 model and with experimentally derived values. The results confirm the common presence of
34 linear time-dependence of fracture k due to self-reinforcement.

35 **1. Introduction**

36 Fractures are ubiquitous across different scales in subsurface geologic materials (Candela
37 et al., 2012). Mass and energy transport through fractures play a critical role in many geophysical
38 problems, including geothermal exploitation, oil and gas recovery, hydraulic fracturing, and
39 subsurface CO₂ sequestration (Berkowitz, 2002). Transport through fractures is controlled by its
40 intrinsic permeability (k) that is inherently determined by the aperture field (Nicholl et al., 1999;
41 Witherspoon et al., 1980). Fracture aperture fields however are dynamic over broad time scales.
42 Fracture geometry can be altered by mechanical deformation (Kang et al., 2016; Pyrak-Nolte and

43 Nolte, 2016; Wang and Cardenas, 2016) and mineral dissolution or precipitation (Huerta et al.,
44 2012; Szymczak and Ladd, 2004; Szymczak and Ladd, 2012). Thus, fractures exhibit a broad
45 range of permeability partly due to their evolution.

46 Here we focused on the problem where an acidic fluid dissolves a rough fracture and
47 leads to k and flux increases. This problem is relevant, for example, when vast amounts of CO₂
48 are injected into a brine-filled reservoir and where potential CO₂ pathways are limited by
49 caprocks (Kampman et al., 2016). However, if fractured caprocks are potentially abundant with
50 calcite or other soluble carbonates, fractures may expand due to dissolution of its walls by the
51 acidic CO₂-saturated brine and reactivity of calcite (Deng et al., 2015). The acidity of brine can
52 be persistently high if CO₂ acts as a separate phase flowing through rough fractures (Fitts and
53 Peters, 2013); this might promote calcitic fracture dissolution, and consequently compromise
54 permanent CO₂ storage (Altman et al., 2014).

55 Understanding the behavior of calcite fractures exposed to CO₂ and the resultant fracture
56 geometry evolution at the field scale remains challenging; this is due to the scarcity of
57 monitoring data (Fitts and Peters, 2013). Thus, numerous studies turned to physical and
58 numerical experiments to interrogate this problem. Previous experimental studies demonstrated
59 that fracture k remains fairly constant unless the reaction front breaks through to the end of
60 fracture (Deng et al., 2015). Regardless of varying CO₂ pressures, dissolution experiments all
61 suggested that fracture k increased linearly with time after the reaction front reached the outlet of
62 three-dimensional (3D) fractures. Note that this linear time-dependence of k was directly shown
63 in our recent computational study (Wang and Cardenas, 2017) rather than the original
64 experimental study, which only showed that k increased with expanding fracture volume instead
65 of directly with time (Deng et al., 2015). These experimental observations motivated our recently

66 proposed theory showing that a linear time-dependence of k manifests after the dissolution front
67 reaches the outlet of idealized two-dimensional (2D) fractures (Wang and Cardenas, 2017).
68 However, whether, when, and at what rate k increases with time via a linear relationship for
69 natural, 3D, rough, and spatially-correlated fractures remain unaddressed. Moreover, various
70 reactive transport regimes encapsulated by Peclet and Damkohler numbers might affect the
71 dissolution process and thereby alter the response of k to fracture dissolution over time. In these
72 situations, the characteristics of k evolution are largely unknown. This study addresses these
73 broad unsolved problems related to the fracture k evolution.

74 **2. Methods**

75 The approach taken here is to numerically simulate the coupled flow through and
76 dissolution of 3D fractures that consist of pure calcite via a 2D depth-averaged (in-plane)
77 reactive transport model representing Ca^{2+} concentration. The simulations kept track of k while
78 the fracture evolves. Synthetic fractures that covered a broad range of geometry (240 cases) with
79 varying roughness and aperture correlation length were analyzed. In total, we implemented 960
80 reactive transport simulations since we used increasing pressure gradients to achieve different
81 Peclet and Damkohler numbers for each fracture. Note that the Peclet and Damkohler numbers
82 evolve with the fracture aperture field. In this study, the Peclet and Damkohler numbers referred
83 to are the initial values prior to when dissolution begins. For simplicity, a 2D depth-averaged
84 reactive model rather than a 3D model was used here. This is justified by two reasons. First,
85 although a 2D model is unable to generate out-of-plane dissolution, the 2D model can reasonably
86 represent and replicate the reactive transport process within a 3D domain (Detwiler and Rajaram,
87 2007) (3D computationally-intensive simulations might produce more accurate results
88 (Starchenko et al., 2016)). Therefore, the simulation results of and insights arising from this

89 study can still largely hold. Second, solving a 3D problem involves prohibitive computational
90 expense and instability issues while the 2D model is free of these issues. More detailed
91 explanations follow in section 2.2.

92 **2.1. Generation of synthetic fractures with lognormally distributed aperture**

93 Natural fractures typically have complex spatially-correlated aperture fields following
94 normal and lognormal distributions (Brush and Thomson, 2003; Renshaw, 1995). Here, the
95 spatially-correlated aperture (b) fields with varying dimensionless correlation length ($0.01 \leq \lambda L$
96 ≤ 1 with increment ≈ 0.1 , where λ is the correlation length and L ($L = 0.1\text{m}$) is total fracture
97 length) were generated through the geostatistical simulator SGeMS via the Sequential Gaussian
98 Model (Remy et al., 2009). The aperture fields originally followed a normal distribution but were
99 converted into a lognormal distribution according to the method proposed by Brush and
100 Thomson (2003), because the lognormal distribution represents a broader range of roughness
101 ($0.1 \leq \sigma_b / \langle b \rangle \leq 1.6$ with increment ≈ 0.2) than the normal distribution could have, where σ_b and
102 $\langle b \rangle$ are the standard deviation and the arithmetic mean of the b field (Figure 1). The b was set to
103 be 1×10^{-12} m wherever contact points were identified. The synthetic fractures were used for
104 assessing the effect of heterogeneity on acidic fluid-induced dissolution patterns via the reactive
105 transport model.

106 **2.2. Two-dimensional (2D) depth-averaged reactive transport model**

107 Previous studies of fracture dissolution due to an acidic fluid have shown that the velocity
108 and concentration fields can reach steady-state much sooner than any appreciable change of
109 aperture field occurs (Detwiler and Rajaram, 2007; Elkhoury et al., 2013). In light of this, we
110 solved the steady-state flow and transport equations for velocity and solute concentration,
111 respectively, and used the resultant concentration for determining the dissolution rate and

112 updating the aperture field. The simulations were implemented sequentially at each time step.
 113 That is, we first solved the flow field, followed by the concentration field, and lastly using the
 114 Ca^{2+} concentration to update the aperture field. This 2D depth-averaged reactive transport model
 115 has been tested to be valid for capturing the dissolution pattern for real 3D fractures (Detwiler
 116 and Rajaram, 2007; Elkhoury et al., 2013). The detailed simulation framework is discussed
 117 below.

118 **2.2.1. 2D depth-averaged modified Local Cubic Law for fluid flow**

119 Single-phase fluid flow through a fracture is governed by the Navier-Stokes and
 120 continuity equations (NSE). Nonetheless, directly solving the NSE for a 3D rough and tortuous
 121 fracture flow field is computationally costly and sometimes lead to situations where numerical
 122 solvers do not converge, especially for non-Darcian flow situations (Zimmerman et al., 2004;
 123 Zimmerman et al., 1991). Vertically-integrating the NSE circumvents having to directly solve the
 124 NSE. This approach is valid under certain circumstances such as low pressure gradient cases
 125 with small velocity magnitude. In this case when the inertial force in the NSE is negligible, the
 126 NSE simplifies into the well-known Local Cubic Law (LCL) upon vertically integrating the local
 127 parabolic velocity profile (Nicholl et al., 1999). Thus the LCL is a so-called depth-averaged
 128 model (DAM) for fluid flow.

129 Fracture roughness and tortuosity were separately considered (Brush and Thomson, 2003;
 130 Ge, 1997) in the development of previous LCL models. Our recent study improved the prediction
 131 of local flow fields through the modified LCL (MLCL) that collectively considered roughness,
 132 tortuosity, and weak inertial force (Wang et al., 2015), and thereby the MLCL improves the
 133 solution compared to that using other LCL models. The MLCL is:

$$134 \quad \nabla \cdot \left[\frac{T_x}{M} \frac{\partial p}{\partial x} \cos(\phi_x) \vec{i} + \frac{T_y}{M} \frac{\partial p}{\partial y} \cos(\phi_y) \vec{j} \right] = 0 \quad (1)$$

135 where p [Pa] is total pressure, the transmissivity vector \mathbf{T} (T_x, T_y) [m^3] between cells can be
 136 estimated using a harmonic mean of adjacent apertures, ϕ (ϕ_x, ϕ_y) [-] is the flow orientation
 137 angle accounting for local tortuosity, M [-] is the correction factor considering roughness and
 138 weak inertial force, \vec{i} and \vec{j} represent the unit vectors in the x and y directions, respectively. T and
 139 ϕ are estimated based on the aperture field, M can be found at Wang et al. (2015).

140 In this study, we used the MLCL to solve steady-state fluid flow. We imposed a no-flux
 141 boundary for lateral boundaries, and applied a pressure gradient driving fluid flow in the x
 142 direction (Figure 2). The resultant p were used for estimating the depth-averaged flux \mathbf{q} [q_x, q_y]
 143 [m/s]:

$$144 \quad \mathbf{q} = -\frac{b^2}{12\mu} \nabla p \quad (2)$$

145 where b [m] is the local aperture, μ [Pa·s] is fluid dynamic viscosity. \mathbf{q} was further used for the
 146 transport problem which calculated advective and dispersive fluxes. Note that we implemented
 147 the simulations by incrementally increasing the pressure gradient (1 Pa/m \sim 1×10^3 Pa/m) to
 148 achieve progressively larger flow rates and thus different Peclet and Damkohler numbers for
 149 each initial fracture geometry.

150 **2.2.2. 2D depth-averaged advection-dispersion equation for transport**

151 Assuming that the Taylor-Aris' dispersion theory is locally valid (Aris, 1956; Taylor,
 152 1953), the solute transport in 3D rough fractures at steady state can be treated by the 2D depth-
 153 averaged model:

$$154 \quad \nabla \cdot (\mathbf{q}C) = \nabla \cdot (b\mathbf{D}\nabla C) + R \quad (3)$$

155 where b is local aperture field, C [kg/m^3] represents local depth-averaged concentration, i.e.,
 156 [Ca^{2+}], \mathbf{D} [m^2/s] is the dispersion tensor dependent on \mathbf{q} according to the Taylor dispersion theory
 157 (Wang et al., 2012):

158
$$D = D_m + \frac{q^2}{210D_m} \quad (4)$$

159 The depth-averaged dissolution flux R [kg/(m²s)] is described as (Detwiler and Rajaram, 2007;
160 Szymczak and Ladd, 2012):

161
$$R = \frac{2\kappa}{1 + \frac{4\kappa b}{Sh \cdot D_m}} (C_{eq} - C) \quad (5)$$

162 where κ is the local chemical reaction rate constant, and C_{eq} [kg/m³] is the saturated
163 concentration for [Ca²⁺] when CO₂-saturated brine equilibrates with calcite, $Sh \sim 7.6$ [-] is
164 Sherwood number defined as the ratio of dissolution rate to rate of diffusion (Sh is constant for a
165 specific geometry (Kays et al., 2012) given that the fracture length satisfies its threshold
166 requirement), and $D_m = 2.03 \times 10^{-9}$ m²/s is the molecular diffusion coefficient. In the case of CO₂-
167 saturated brine dissolving calcite in a relatively low pH environment, κ is on the order of 1×10^{-3}
168 m/s (Elkhoury et al., 2013; Pokrovsky et al., 2009). Given that the mean b is on the order of
169 1×10^{-3} m, the value of $4\kappa b / (Sh \cdot D_m) \gg 1$, which signifies that equation (5) can be reduced to
170 equation (6) by eliminating κ , i.e., the effect of κ on altering R diminishes (Detwiler and
171 Rajaram, 2007; Wang and Cardenas, 2017).

172
$$R = \frac{Sh \cdot D_m}{2b} (C_{eq} - C) \quad (6)$$

173 Equation (6) suggests that the entire problem becomes a diffusion-limited system, where R only
174 depends on Ca²⁺ concentration.

175 The C field was solved through equation (3) for a constant inlet concentration $C_{inlet} =$
176 1.22 kg/m³. The C_{inlet} was smaller than C_{eq} , where C_{inlet} was set to be equilibrated with the
177 invading acidic fluid with a high CO₂ concentration and thus low pH at the inlet, and $C_{eq} = 2.10$
178 kg/m³ following Elkhoury et al. (2013) where pH was higher than that at the inlet. This ensured

179 that dissolution occurs along the invading flow path. Moreover, an open boundary (i.e., no
 180 dispersive flux) was specified at the outlet (right) (Figure 2), and the initial C was set to be C_{eq} .
 181 The above-mentioned 2D reactive transport scheme with a few assumptions (e.g., constant inlet
 182 concentration, the applicability of the Taylor dispersion theory to estimate the local dispersion
 183 coefficient) has been verified to be accurate for capturing dissolution patterns within 3D fractures
 184 (Detwiler and Rajaram, 2007; Elkhoury et al., 2013).

185 **2.2.3. Updating the dynamic aperture field during dissolution**

186 Based on mass conservation, the expansion rate of the local aperture due to dissolution is:

$$187 \quad \frac{\partial b}{\partial t} = \frac{R}{F\rho_s} \quad (7)$$

188 where t is time, F ($F = 0.4$ [-]) is the stoichiometric coefficient for converting a solid phase
 189 mass, i.e. calcite, to aqueous phase, i.e. $[Ca^{2+}]$, and ρ_s [kg/m^3] is solid density. The time step for
 190 updating b was determined to ensure that the velocity and concentration fields reach steady-state
 191 prior to any appreciable aperture change. That is, at each computational time step, we first solved
 192 the steady-state flow field first, and then following this solved the transport equation, and then
 193 afterwards used the resultant R for updating b according to equation (7). The simulation kept
 194 running until a linear relationship between k and t was observed. This linear relationship is
 195 expected for 2D idealized fractures (Wang and Cardenas, 2017). It should follow:

$$196 \quad k/t = \frac{Sh \cdot D_m}{12F\rho_s} (C_{eq} - C) \quad (8)$$

197 The simulations were implemented in MATLAB via the finite difference method.
 198 Specifically, we adopted a central difference approximation for handling the second derivative to
 199 achieve second-order accuracy, and applied the upwind scheme for handling the advection term
 200 that considers the direction of propagation with first-order accuracy. The simulations were run in

201 a memory-share workstation; each run took minutes to hours depending on the pressure gradients
202 and heterogeneity of aperture fields. The goal of this study is to assess the linear time-
203 dependence of k [m^2] suggested above for 3D rough fractures and to determine whether it
204 represents a common phenomenon at various reactive transport regimes demonstrated by Peclet
205 and Damkohler numbers. This was achieved by solving the 2D reactive DAM for 3D fractures.

206 **3. Results and discussions**

207 **3.1. Impact of fracture heterogeneity on dissolution patterns**

208 The reactivity of calcite induced by acidic brine leads to the fracture wall's dissolution;
209 this generally leads to an expanding aperture field with channelized/preferential dissolution paths
210 demonstrated by the dissolution fingers (Figure 3) and enlarged flux (Figure 4). These
211 preferential dissolution flow paths in turn enhance the concentration gradient acting as a
212 geochemical driving force that further dissolves fractures; this is the driver for self-reinforcing
213 fracture dissolution (Li and Einstein, 2017). The variety of patterns resulting from the potentially
214 self-reinforced dissolution of 3D rough-walled fractures is due to variations in the local
215 Damkohler number (Da) defined as:

$$216 \quad Da = \frac{RL}{(C_{eq} - C)\mathbf{q}} \quad (9)$$

217 where L [m] is fracture length (the grid cell size was not used here for calculating Da since we
218 are more concerned about the area-averaged $\langle Da \rangle$ [-] as discussed below). The Da inherently
219 hinges on the b field and pressure gradient that dictates the magnitude of R , \mathbf{q} , and C . Here, we
220 only show the dissolution patterns where dissolution fingers are prominent (Figure 3), many
221 other patterns with and without apparent dissolution fingers are not shown here.

222 For short-range, spatially-correlated fractures, the dissolution fingers gradually became
223 more apparent with increasing $\langle Da \rangle$ (the area-averaged Da) (Figures 3a-3f); this observation is

224 congruent with previous studies (Iyer and Elkhoury, 2015; Szymczak and Ladd, 2004). This is
225 because at low pressure gradient with large $\langle Da \rangle$ the localized dissolution fingers keep
226 advancing forward with a strong positive feedback between flow and reaction; this self-
227 reinforcement results in an almost constant finger width (Figures 3c and 3f) - a similar
228 phenomenon was found in partially molten porous media with developed melting channels for
229 magma flow (Jordan and Hesse, 2015). However, for high pressure gradient cases with small
230 $\langle Da \rangle$, the localized fingers disappear due to the fairly uniform reactivity that widens the aperture
231 field at more or less the same dissolution rate, where the original aperture structure was
232 maintained (Figures 3a and 3d).

233 A similar trend or pattern, i.e., the preferential dissolution fingers gradually establish,
234 with increasing $\langle Da \rangle$ was also found for fractures with relatively long λ (Figure 3g-3l). Unlike
235 fractures with short λ where dissolution fingers with a relatively uniform spacing developed,
236 dissolution fingers were mostly concentrated in the preferential flow paths determined by
237 connected large aperture zones within the originally unaltered spatially-correlated aperture field;
238 this is again the self-reinforcing mechanism as demonstrated by the focused flux (Figure 4). The
239 number of dissolution fingers was much less and the width was wider in the fractures with long λ
240 than the dissolution fingers in the fractures with short λ (Figure 3). These dissolution patterns
241 agree with previous results (Iyer and Elkhoury, 2015). While $\sigma_b/\langle b \rangle$ affects the self-reinforcing
242 dissolution by altering q and the resultant C , the role of $\sigma_b/\langle b \rangle$ in determining the overall
243 dissolution patterns is trivial compared to the effects caused by λ as discussed before; this was
244 also suggested in Iyer and Elkhoury (2015).

245 **3.2. Linear time-dependence of fracture permeability**

246 In general and aligned with experimental results (Deng et al., 2015; Noiriél et al., 2013),

247 the dissolution of rough fractures in an acidic environment led to a linear time-dependence of k
248 (Figures 5-6) across the analyzed Peclet (Pe [-]) and $\langle Da \rangle$ number regimes considering a
249 varying degree of reactive transport regimes; the Pe and $\langle Da \rangle$ considered were within the
250 validity thresholds theoretically proposed in Wang and Cardenas (2017) regarding the fracture
251 aspect ratio. This linear k increase with time persists across different dissolution patterns
252 illustrating a common relationship.

253 Despite different dissolution patterns, the onset of the linear relationship roughly
254 occurred once the dissolution front reaches the outlet of 3D rough fractures (Figure 5); this was
255 also demonstrated in previous idealized numerical model experiments and laboratory
256 observations (Deng et al., 2015; Fitts and Peters, 2013). The onset of the linear relationship is
257 attributed to the following reasons: (a) the dissolution front migrates and leaves behind an
258 expanding and usually large b , (b) the unaltered b ahead of the dissolution front is often smaller
259 than the continuously expanding b , (c) the narrowest b (i.e., the unaltered b usually located at the
260 outlet) limits the magnitude of k according to the Cubic Law (Witherspoon et al., 1980;
261 Zimmerman et al., 1991). All these factors imply that the bulk k remains fairly constant until the
262 unaltered and narrowest b located at the outlet expands due to dissolution; this eventually
263 initiates the linear k increase with t .

264 After the initiation of the linear relationship between k and t , we observed a persistent
265 linear time-dependence of k (Figures 5 and 6a) for 3D rough fractures with varying $\sigma_b/\langle b \rangle$ and λ
266 regarding varying degrees of dissolution patterns in response to different reactive transport
267 regimes. This is however beyond our expectation because the theory of the linear time-
268 dependence was developed based on 2D fractures (i.e., parallel plates) with a single flow channel
269 which is essentially a straight a line when viewed orthogonally from the fracture plane, so

270 everything is forced to flow along this line only (Wang and Cardenas, 2017). Unlike 2D
271 fractures, the flow field in 3D rough fractures is far more complex and is potentially
272 characterized by multiple preferential flow curves from the top orthogonal view rather than a
273 single line (Elkhoury et al., 2013); this may lead to violation of the linear relationship theory.
274 However, our numerical experiments within 3D fractures demonstrated that the collective flow
275 behavior with possible multiple flow channels (Figure 5) is identical to that for a single flow
276 channel, i.e., k increases linearly with time due to self-reinforcement. This was also identified in
277 physical experiments for 3D rough fractures (Deng et al., 2015). This similarity in k evolution
278 holds because the main conduits in 3D fractures exhibiting positive feedback between flow and
279 transport dominate over smaller channels in terms of flux (Figure 4). That is, the dominant
280 conduits in 3D fractures behave in a similar way as the single channel in 2D fractures. The
281 theory of the linear relationship based on a 2D idealized model is therefore applicable for
282 describing the k evolution of 3D rough fractures with more complex flow patterns and
283 phenomena such as channelization.

284 To further interrogate the common occurrence of linear time-dependence of k , we
285 quantified the time rate of change of permeability (k/t) based on the slope of $k(t)$ curves (Figure
286 6a). The slope was computed and chosen once the linear relationship manifested over 200
287 simulation time steps. To exclude the effect of $\langle b \rangle$, we created and conducted simulations
288 through two sets of fractures with different $\langle b \rangle$; each set of given $\langle b \rangle$ had the same range of
289 $\sigma_b/\langle b \rangle$ and λ/L . Not surprisingly and regardless of Pe and $\langle Da \rangle$ (i.e., different reactive transport
290 regimes), all k/t are on the order of magnitude of 10^{-12} m²/s with a median value ($\sim 1.3 \times 10^{-12}$
291 m²/s) that is slightly greater than the theoretical value of 1.07×10^{-12} m²/s according to equation
292 (8) (Figure 6b). This also holds true for the two sets of $\langle b \rangle$ (results are indistinguishable in

293 Figure 6b). The narrow variations in k/t might be attributed to the Sh because it is dependent on
294 fracture geometry (Kays et al., 2012). The aperture fields and main flow conduits were
295 expanding over the course of the dissolution process (Figures 3 and 5) and thus Sh was slightly
296 varying.

297 Aside from slight variations in k/t caused by variation in Sh , the slope of k/t also depends
298 on the concentration difference according to equation (8). This difference is related to the acidity
299 of brine (e.g., pH values under different supercritical CO₂ pressures). Our simulations employed
300 a constant concentration ($[Ca^{2+}]$) that was based on previous experiments (Elkhoury et al., 2013)
301 and thus excluded the concentration effects. Deng et al. (2015)'s experiments explored the calcite
302 fracture's dissolution patterns at high and low supercritical CO₂ pressures. They found that the
303 saturated $[Ca^{2+}]$ varied in response to the CO₂ solubility induced by different CO₂ pressures. We
304 calculated k/t by estimating the right hand term in equation (8); these calculations considered the
305 effects of CO₂ pressures on C_{eq} that was derived from effluent concentration. Although C_{eq} was
306 variable, Deng et al.'s two experiments all confirmed a linear time-dependence of k and the
307 experimentally-derived k/t were equally close to 1.8×10^{-12} m²/s, whose value is slightly higher
308 than the median value ($\sim 1.3 \times 10^{-12}$ m²/s) based on our numerical experiments (Figure 6b). The k/t
309 derived by different methods, including theory, physical experiments, and numerical simulations,
310 are on the same order of magnitude albeit with up to 30% difference between them. This fact not
311 only confirms the common occurrence of linear time-dependence of k for 3D rough fractures at
312 varying reactive transport regimes, but also demonstrates the robustness of the DAM in capturing
313 k evolution in a self-reinforcing reactive environment.

314 **4. Implications**

315 The subsurface injection of CO₂ promotes both the dissolution and precipitation of

316 minerals within and around the CO₂ plume (Altman et al., 2014; Steefel et al., 2013). In the case
317 of fractured calcite or acid-soluble caprock, the migration of CO₂-saturated brine could lead to
318 either self-sealing or self-reinforcing phenomena depending on the supply of acidic fluid. In the
319 worst case with a persistent source of acidic fluid, e.g., CO₂ and brine two-phase flow (Fitts and
320 Peters, 2013), our study implies that the acidic fluid migration can promote the fractured calcite
321 dissolution with increasing k and flux over time; this might compromise the subsurface CO₂
322 storage in a relatively shorter time than expected. Note that our simulations only considered a
323 pure calcite matrix. For fractures containing relatively inert materials (quartz, clay, etc.), the
324 extent of permeability increase is limited by the residual materials in the fractures. It may even
325 be the case that these residual materials are mobilized during calcite dissolution and migrate
326 downstream resulting in clogging and a decrease in fracture permeability. These scenarios are
327 beyond the scope of this study.

328 Although our study focused on the fracture dissolution at the single fracture scale, the
329 evolving k path observed in our simulations was proposed to manifest at the field scale. An
330 appropriate example is karst conduit development (Andre and Rajaram, 2005) wherein a slow
331 continuous k increase is followed by an rapid k increase due to the positive feedback between
332 flow and reactive transport processes in a H₂O-CO₂-CaCO₃ system. The k evolution could also
333 be vital for mass transport in many other self-reinforcing systems, including partially-molten
334 magma flow in porous media (Hewitt, 2010; Jordan and Hesse, 2015) and warm water flowing
335 through fractured ice (Das et al., 2008).

336 **5. Conclusion**

337 We simulated flow and reactive transport processes for dissolving three-dimensional
338 rough fractures using a two-dimensional depth-averaged model. In this self-reinforcing

339 dissolution system, the fracture expands followed by mass flux increase. This leads to an
340 increasing geochemical driving force that further dissolves the fracture. Based on numerical
341 simulations through rough-walled fractures with different roughness and aperture field
342 correlation length, we observed a persistent linear time-dependence of fracture permeability (k)
343 after the dissolution front reaches the fracture outlet. This linear relationship between k and time
344 (t) holds across a range of Peclet and Damkohler numbers which correspond to different
345 dissolution patterns from distributed, evenly-spaced fingers to very focused flows with a
346 dominant channel. The linear time-dependence of k is consistent with experimental observations.
347 The estimated time rate of change of k (i.e., k/t) further illustrates the linear relationship between
348 k and t . k/t is on the same order of magnitude for all the studied fractures despite their varying
349 degrees of dissolution patterns. Our numerical model-derived k/t values for rough fractures at
350 varying reactive transport regimes are consistent with values based on carefully-designed
351 experiments and on theory for idealized two-dimensional geometry. All these collectively
352 suggest the common presence of linear fracture permeability increase with time due to self-
353 reinforcing dissolution.

354

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361

362 **References**

363 Altman, S.J., 2014 Aminzadeh, B., Balhoff, M.T., Bennett, P.C., Bryant, S.L., Cardenas, M.B.,
364 Chaudhary, K., Cygan, R.T., Deng, W., Dewers, T., DiCarlo, D.A., Eichhubl, P., Hesse,
365 M.A., Huh, C., Matteo, E.N., Mehmani, Y., Tenney, C.M., Yoon, H.

366 **Chemical and Hydrodynamic Mechanisms for Long-Term Geological Carbon**
367 **Storage.**

368 J. Phys. Chem. C, 118(2014), pp. 15103-15113, 10.1021/jp5006764.

369 Andre, B.J., 2005 Rajaram, H.

370 **Dissolution of limestone fractures by cooling waters: Early development of**
371 **hypogene karst systems.**

372 Water Resour. Res., 41(2005), 10.1029/2004WR003331.

373 Aris, R., 1956.

374 **On the Dispersion of a Solute in a Fluid Flowing through a Tube.**

375 P. Roy. Soc. Lond. A. Mat., 235(1956), pp. 67-77, 10.1098/rspa.1956.0065.

376 Berkowitz, B., 2002.

377 **Characterizing flow and transport in fractured geological media: A review.**

378 Adv. Water Resour., 25(2002), pp. 861-884, 10.1016/s0309-1708(02)00042-8.

379 Brush, D.J., 2003 Thomson, N.R.

380 **Fluid flow in synthetic rough-walled fractures: Navier-Stokes, Stokes, and local**
381 **cubic law simulations.**

382 Water Resour. Res., 39(2003), pp. 1085, 10.1029/2002wr001346.

383 Candela, T., 2012 Renard, F., Klinger, Y., Mair, K., Schmittbuhl, J., Brodsky, E.E.

384 **Roughness of fault surfaces over nine decades of length scales.**

385 J. Geophys. Res.-Sol. Ea., 117(2012), B08409, B0840910.1029/2011jb009041.

386 Das, S.B., 2008 Joughin, I., Behn, M.D., Howat, I.M., King, M.A., Lizarralde, D., Bhatia, M.P.

387 **Fracture Propagation to the Base of the Greenland Ice Sheet During Supraglacial**

388 **Lake Drainage.**

389 Science, 320(2008), pp. 778-781, 10.1126/science.1153360.

390 Deng, H., 2015 Fitts, J.P., Crandall, D., McIntyre, D., Peters, C.A.

391 **Alterations of Fractures in Carbonate Rocks by CO₂-Acidified Brines.**

392 Environ. Sci. Technol., 49(2015), pp. 10226-10234, 10.1021/acs.est.5b01980.

393 Detwiler, R.L., 2007 Rajaram, H.

394 **Predicting dissolution patterns in variable aperture fractures: Evaluation of an**

395 **enhanced depth-averaged computational model.**

396 Water Resour. Res., 43(2007), W0440310.1029/2006wr005147.

397 Elkhoury, J.E., 2013 Ameli, P., Detwiler, R.L.

398 **Dissolution and deformation in fractured carbonates caused by flow of CO₂-rich**

399 **brine under reservoir conditions.**

400 Int. J. Greenh. Gas Con., 16(2013), pp. S203-S215, 10.1016/j.ijggc.2013.02.023.

401 Fitts, J.P., 2013 Peters, C.A.

402 **Caprock fracture dissolution and CO₂ leakage.**

403 Rev. Mineral. Geochem., 77(2013), pp. 459-479, 10.2138/rmg.2013.77.13.

404 Ge, S., 1997.

405 **A governing equation for fluid flow in rough fractures.**

406 Water Resour. Res., 33(1997), pp. 53-61, 10.1029/96wr02588.

407 Hewitt, I.J., 2010.

408 **Modelling melting rates in upwelling mantle.**
409 Earth Planet. Sci. Lett., 300(2010), pp. 264-274, 10.1016/j.epsl.2010.10.010.
410 Huerta, N.J., 2012 Hesse, M.A., Bryant, S.L., Strazisar, B.R., Lopano, C.L.
411 **Experimental evidence for self-limiting reactive flow through a fractured cement**
412 **core: Implications for time-dependent wellbore leakage.**
413 Environ. Sci. Technol., 47(2012), pp. 269-275, 10.1021/es3013003.
414 Iyer, J., 2015 Elkhoury, J.E.
415 **Effect of Fracture Heterogeneities on Reactive Flow.**
416 49th US Rock Mechanics/Geomechanics Symposium. American Rock Mechanics
417 Association.
418 Jordan, J.S., 2015 Hesse, M.A.
419 **Reactive transport in a partially molten system with binary solid solution.**
420 Geochem. Geophys. Geosy., 16(2015), pp. 4153-4177, 10.1002/2015GC005956.
421 Kampman, N., 2016 Busch, A., Bertier, P., Snippe, J., Hangx, S., Pipich, V., Di, Z., Rother, G.,
422 Harrington, J.F., Evans, J.P., Maskell, A., Chapman, H.J., Bickle, M.J.
423 **Observational evidence confirms modelling of the long-term integrity of CO2-**
424 **reservoir caprocks.**
425 Nat. Commun., 7(2016), 12268, 10.1038/ncomms12268.
426 Kang, P.K., 2016 Brown, S., Juanes, R.
427 **Emergence of anomalous transport in stressed rough fractures.**
428 Earth Planet. Sci. Lett. 454(2016), pp. 46-54. 10.1016/j.epsl.2016.08.033.
429 Kays, W.M., 2012 Crawford, M.E., Weigand, B.
430 **Convective heat and mass transfer.**

431 Tata McGraw-Hill Education.

432 Li, W., 2017 Einstein, H. H.

433 **Theoretical and Numerical Investigation of the Cavity Evolution in Gypsum Rock.**

434 Water Resour. Res., 53(2017), pp. 9988-10001. 10.1002/2017WR021776

435 Nicholl, M.J., 1999 Rajaram, H., Glass, R.J., Detwiler, R.

436 **Saturated flow in a single fracture: evaluation of the Reynolds Equation in**

437 **measured aperture fields.**

438 Water Resour. Res., 35(1999), pp.3361-3373, 10.1029/1999wr900241.

439 Noiriél, C., 2013 Gouze, P., Madé, B..

440 **3D analysis of geometry and flow changes in a limestone fracture during dissolution.**

441 J. Hydrol. 486 (2013), pp. 211-223, 10.1016/j.jhydrol.2013.01.035.

442 Pokrovsky, O.S., 2009 Golubev, S.V., Schott, J., Castillo, A.

443 **Calcite, dolomite and magnesite dissolution kinetics in aqueous solutions at acid to**

444 **circumneutral pH, 25 to 150 C and 1 to 55 atm pCO₂: New constraints on CO₂**

445 **sequestration in sedimentary basins.**

446 Chem. Geol., 265(2009), pp. 20-32,10.1016/j.chemgeo.2009.01.013.

447 Pyrak-Nolte, L.J., 2016 Nolte, D.D.

448 **Approaching a universal scaling relationship between fracture stiffness and fluid**

449 **flow.**

450 Nat. Commun., 7(2016), pp. 10663, 10.1038/ncomms10663.

451 Remy, N., 2009 Boucher, A., Wu, J.

452 **Applied geostatistics with SGeMS: A user's guide.**

453 Cambridge University Press.

454 Renshaw, C.E., 1995.
455 **On the relationship between mechanical and hydraulic apertures in rough-walled**
456 **fractures.**
457 J. Geophys. Res.: Sol. Ea., 100(1995), pp. 24629-24636, 10.1029/95jb021596.

458 Starchenko, V., 2016 Marra, C.J., Ladd, A.J.C.
459 **Three-dimensional simulations of fracture dissolution.**
460 J. Geophys. Res.-Sol. Ea., 121(2016), pp. 6421-6444, 10.1002/2016JB013321.

461 Steefel, C.I., 2013 Molins, S., Trebotich, D.
462 **Pore scale processes associated with subsurface CO2 injection and sequestration.**
463 Rev. Mineral. Geochem., 77(2013), pp. 259-303, 10.2138/rmg.2013.77.8.

464 Szymczak, P., 2004 Ladd, A.J.C.
465 **Microscopic simulations of fracture dissolution.**
466 Geophys. Res. Lett., 31(2004), pp. L23606, 10.1029/2004gl021297.

467 Szymczak, P., 2012 Ladd, A.J.C.
468 **Reactive-infiltration instabilities in rocks. Fracture dissolution.**
469 J. Fluid Mech., 702(2012), pp. 239-264, 10.1017/jfm.2012.174.

470 Taylor, G., 1953.
471 **Dispersion of Soluble Matter in Solvent Flowing Slowly through a Tube.**
472 P. Roy. Soc. Lond. A. Mat., 219(1953), 186-203, doi: 10.1098/rspa.1953.0139.

473 Wang, L., 2016 Cardenas, M.B.
474 **Development of an empirical model relating permeability and specific stiffness for**
475 **rough fractures from numerical deformation experiments.**
476 J. Geophys. Res.-Sol. Ea., 121(2016), pp. 4977-4989, 10.1002/2016JB013004.

477 Wang, L., 2017 Cardenas, M.B.
478 **Linear permeability evolution of expanding conduits due to feedback between flow**
479 **and fast phase change.**
480 Geophys. Res. Lett., 44(2017), pp. 4116-4123, 10.1002/2017gl073161.

481 Wang, L., 2012 Cardenas, M.B., Deng, W., Bennett, P.C.
482 **Theory for dynamic longitudinal dispersion in fractures and rivers with Poiseuille**
483 **flow.**
484 Geophys. Res. Lett., 39(2012), pp. L05401, 10.1029/2011gl050831.

485 Wang, L., 2015 Cardenas, M.B., Slotke, D.T., Ketcham, R.A., Sharp, J.M.
486 **Modification of the Local Cubic Law of fracture flow for weak inertia, tortuosity,**
487 **and roughness.**
488 Water Resour. Res., 51(2015), pp. 2064-2080, 10.1002/2014WR015815.

489 Witherspoon, P.A., 1980 Wang, J.S.Y., Iwai, K., Gale, J.E.
490 **Validity of Cubic Law for fluid flow in a deformable rock fracture.**
491 Water Resour. Res., 16(1980), pp. 1016-1024, 10.1029/WR016i006p01016.

492 Zimmerman, R.W., 2004 Al-Yaarubi, A., Pain, C.C., Grattoni, C.A.
493 **Non-linear regimes of fluid flow in rock fractures.**
494 Int. J. Rock Mech. Min., 28(2014), pp. 325-331, 10.1016/0148-9062(91)90597-f.

495 Zimmerman, R.W., 1991 Kumar, S., Bodvarsson, G.S.
496 **Lubrication theory analysis of the permeability of rough-walled fractures.**
497 Int. J. Rock Mech. Min., 41(1991), Supplement 1(0), pp. 163-169,
498 10.1016/j.ijrmms.2004.03.036.
499

500 **Figure Captions**

501 **Figure 1.** Examples of log-normally distributed aperture fields across a broad range of roughness
502 ($0.1 \leq \sigma_b / \langle b \rangle \leq 1.6$) and dimensionless correlation length ($0.01 \leq \lambda / L \leq 1$), where b is the
503 aperture field with standard deviation σ_b and arithmetic mean $\langle b \rangle$, λ is the correlation length of
504 the b field, and L is total fracture length and width. The right column shows the relative
505 frequency distribution of the aperture field corresponding to similarly colored outlines for the
506 fields. Note that the right column plots are for the log-normally distributed aperture fields, where
507 the integrated area under each of the curve increases with roughness.

508

509 **Figure 2.** Conceptual diagram of two-dimensional, depth-averaged model for fluid flow and
510 reactive transport in a rough-walled fracture. The colors represent the aperture magnitude, fluid
511 flow boundary conditions are in red, transport boundary conditions are in blue, and white arrow
512 indicates the mean flow direction driven by the pressure gradient ∇p .

513

514 **Figure 3.** Aperture dissolution patterns for four fractures with increasing dimensionless
515 correlation lengths (λ / L) from top to bottom at time = 1000 initial pore volume; each column
516 represents different area-averaged Damkohler numbers ($\langle Da \rangle$) with the increasing pressure
517 gradient ∇p from left to right. λ is the aperture correlation length, L is total fracture length, the
518 initial pore volume = volume / flux at time = 0, and Da is defined in equation (9).

519

520 **Figure 4.** Self-reinforcing dissolution patterns with channelized streamlines (red) over time for
521 two fractures with increasing dimensionless correlation length (λ / L) from top to bottom. λ is the
522 aperture correlation length, L is total fracture length, and the initial pore volume $PV_0 = \text{volume} /$

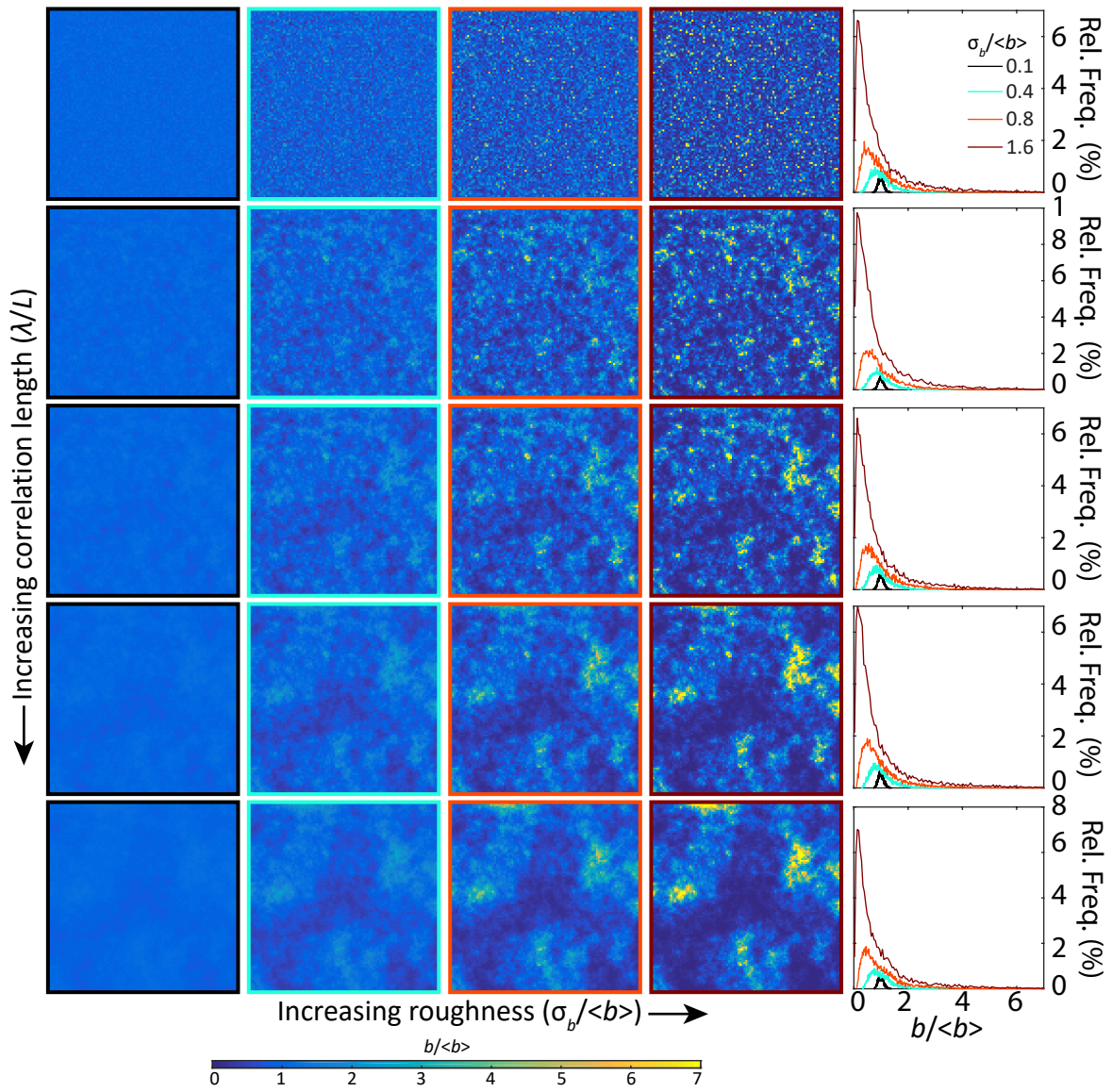
523 flux at time = 0. The streamlines connect the fluid path from the inlet to outlet, where flux is the
524 same in the region between streamlines. The tortuous flow path with a zigzag shape (especially
525 in the first column) is due to the fluid flowing in the undissolved fracture zone with randomly-
526 distributed apertures. The number of streamlines depends on the total flux which increases over
527 time.

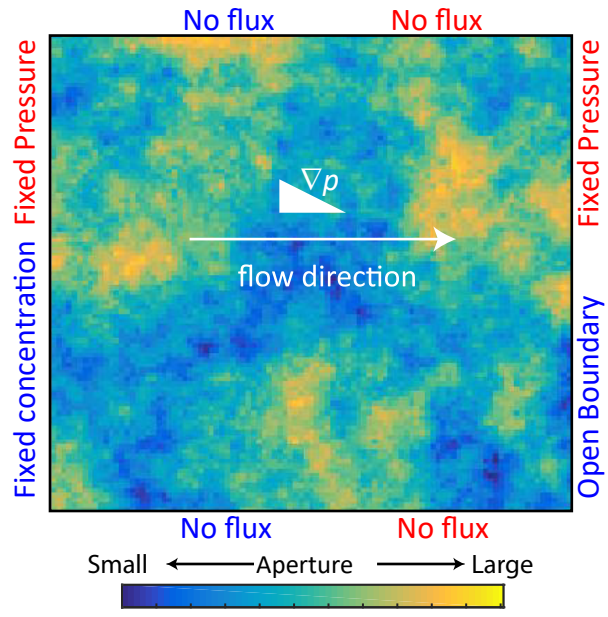
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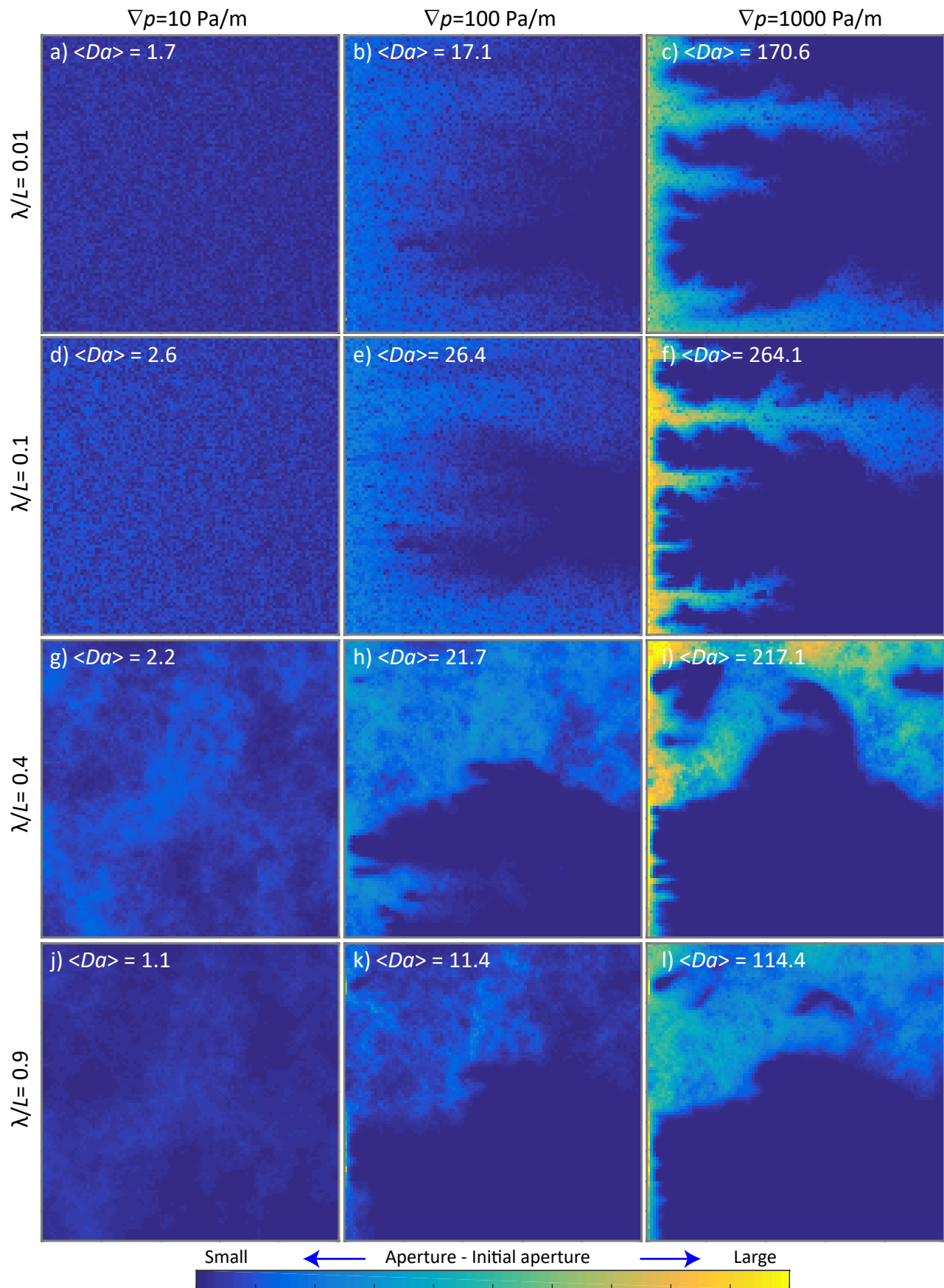
529 **Figure 5.** Linear permeability increase (blue line) with time through 3D rough fractures as
530 exemplified by a) $\lambda/L=0.01$, $\sigma_b/\langle b \rangle=0.8$ and b) $\lambda/L=0.6$, $\sigma_b/\langle b \rangle=0.4$, where b is the aperture field
531 with standard deviation σ_b and arithmetic mean $\langle b \rangle$, λ is the aperture correlation length, and L is
532 total fracture length. Insets show the dissolution patterns = expanding aperture field – the initial
533 aperture field.

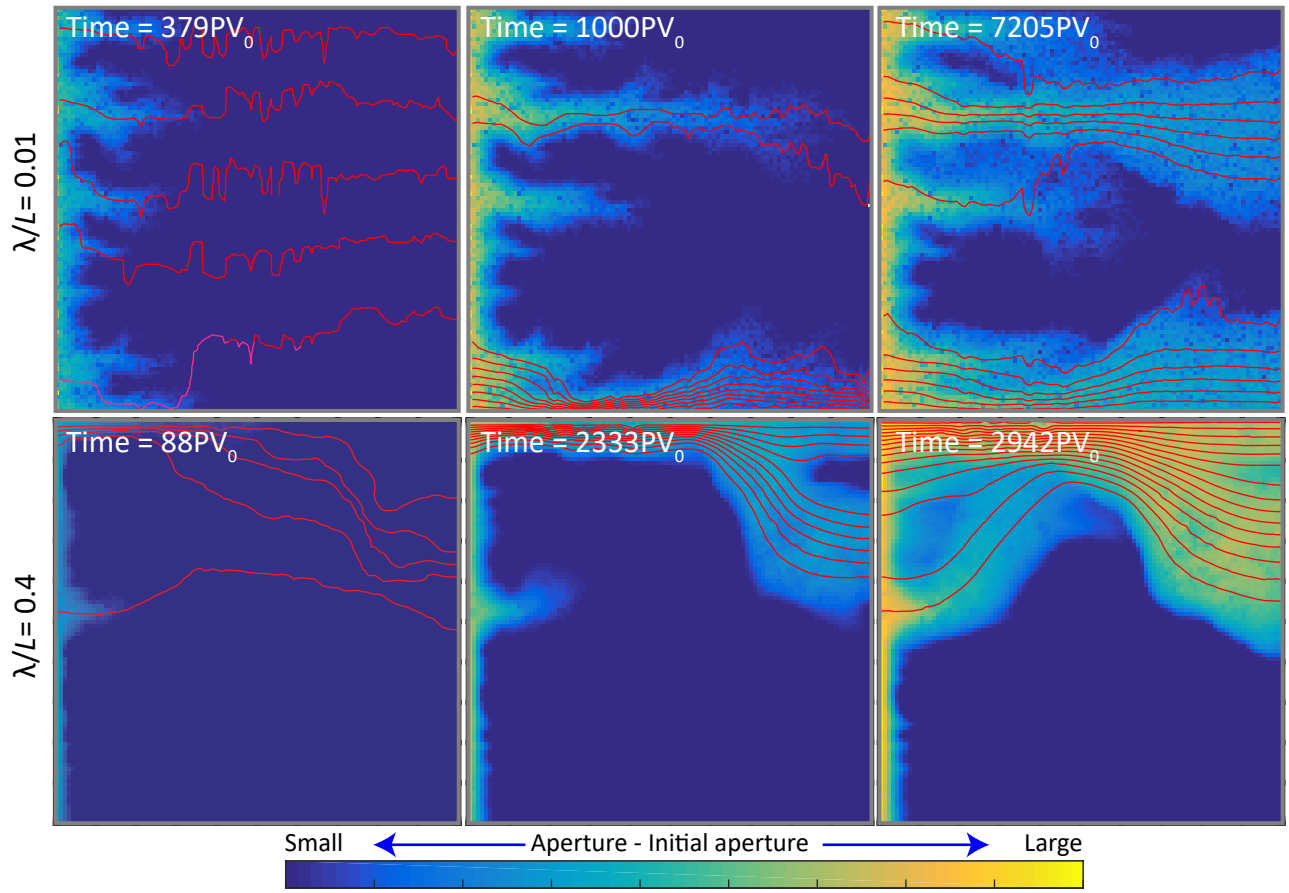
534

535 **Figure 6.** The linear relationship between permeability (k) and time (t) across a range of Peclet
536 number ($Pe=10^2\sim 10^5$), area-averaged Damkohler number ($\langle Da \rangle=10^{-1}\sim 10^3$), fracture roughness
537 ($\sigma_b/\langle b \rangle=0.1\sim 1.6$), and aperture field correlation length ($\lambda=0.1L\sim 1L$). This linear relationship is
538 demonstrated by: a) linear dimensionless permeability (k/k_0) increase with dimensionless time
539 for all studied fractures, and b) small range in the time rate of change of k (k/t) as demonstrated
540 by a histogram of k/t . k_0 is permeability (k) at $t=0$, initial pore volume = volume/flux at $t=0$,
541 $Pe=q/D_m$, q is fluid flux, D_m is molecular diffusion coefficient, Da is defined in equation (9), Sh
542 is Sherwood Number that is constant for given geometry, b is the aperture field with standard
543 deviation σ_b and arithmetic mean $\langle b \rangle$, and L is total fracture length. The dashed line is the
544 theoretical value ($k/t = 1.07\times 10^{-12} \text{ m}^2/\text{s}$) according to equation (8).

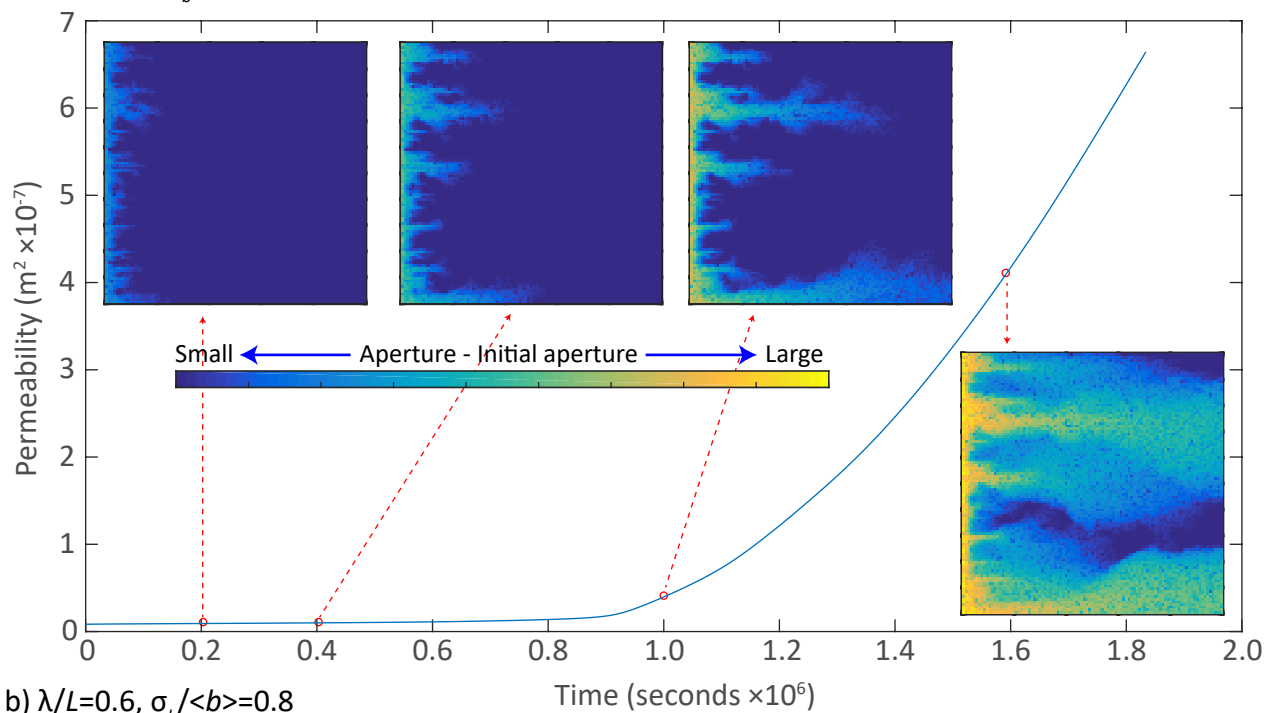








a) $\lambda/L=0.01, \sigma_y/\langle b \rangle=0.8$



b) $\lambda/L=0.6, \sigma_y/\langle b \rangle=0.8$

