

# GENERALIZED MODULAR REPRESENTATION FRAMEWORK FOR THE SYNTHESIS OF EXTRACTIVE SEPARATION SYSTEMS

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## *Abstract*

In this work, a systematic process synthesis and intensification method is presented for extractive separation based on the Generalized Modular Representation Framework (GMF). GMF employs a mass/heat exchange module-based superstructure representation, incorporating detailed thermodynamic model, to investigate conventional or intensified process options for nonideal azeotropic separation without a pre-postulation of plausible unit/flowsheet configurations. Orthogonal Collocation is also applied to enhance GMF representation to obtain intra-module operation information and module dimensionality estimation while maintaining model size compactness. Thus, entrainer selection, process synthesis, design, and intensification are examined within a single mixed-integer nonlinear optimization (MINLP) problem. A case study on the separation of ethanol-water mixture is presented to highlight the potential of the proposed approach in deriving optimal and verifiable extractive separation systems.

## *Keywords*

Process intensification, Process synthesis, Extractive separation, Distillation, Modular framework

## **Introduction**

1 The separation of azeotropic mixtures is a frequent but  
2 challenging task in chemical processes (Mahdi et al., 2015)  
3 Conventional distillation, as the most widely utilized liquid  
4 separation technology, has limitations to achieve a higher  
5 purity beyond the azeotrope point in these systems.  
6 Extractive separation technologies based on process  
7 intensification principles (e.g., extractive distillation,  
8 membrane-assisted extraction) offer the solution to  
9 overcome the physical equilibrium by introducing an extra  
10 mass separating agent and by exploiting the synergy  
11 between multiple phenomena (Tian et al., 2018).  
12 Extensive efforts have been made to investigate the  
13 synthesis and design of extractive separation systems, also  
14 with consideration of entrainer selection (Kossack et al.,  
15 2008). Recent advances in process synthesis for process  
16 intensification have been leveraging phenomenological  
17 representation methods to generate process options without  
18 the pre-postulation of plausible unit-operation-based  
19 flowsheets which may hinder the discovery of novel

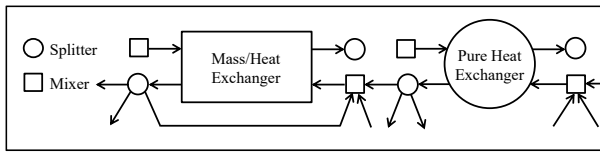
solutions (Tula et al., 2017; da Cruz and Manousiouthakis,  
2017; Demirel et al., 2017). This provides the opportunity  
to synthesize chemical process from a lower aggregated  
level and to investigate the effect of functional materials  
from process fundamental perspective. However, key open  
questions remain for these phenomena-based synthesis  
approaches on how to systematically derive intensified  
processes and how to explore the combinatorial design  
space in a computationally efficient way.

As an attempt to address these challenges, in this work,  
we present a systematic framework for the synthesis of  
process intensification systems based on the Generalized  
Modular Representation Framework (GMF), with specific  
focus on extractive separation processes. The rest of the  
paper is organized as follows: In Section 2, the main  
principles and model formulation of the GMF, coupled with  
Orthogonal Collocation (OC) to enhance representation  
accuracy without sacrificing model size compactness, are  
introduced in detail. Section 3 showcases the proposed

39 GMF and GMF/OC approaches on an ethanol-water  
40 separation example.

## 41 Synthesis Framework

42 Originally proposed by Papalexandri and Pistikopoulos  
43 (1996), the Generalized Modular Representation  
44 Framework (GMF) is built on the idea that process  
45 operations can be characterized as a set of mass- and heat-  
46 transfer phenomena, such as the mass transfer of one  
47 component from one phase to another (e.g., distillation) or  
48 from one substance to another (e.g., reaction) due to the  
49 difference in their chemical potential. Thus, two types of  
50 phenomenological building blocks are used in GMF to  
51 represent chemical processes from a lower aggregated level  
52 than unit operations, namely a multifunctional mass/heat  
53 exchange module and a pure heat exchange module (Figure  
54 1).



56 Figure 1. GMF modular building blocks

57 To ensure mass transfer feasibility within a mass/heat  
58 exchange module, “driving force constraints” are developed  
59 based on the change of total Gibbs free energy to exploit the  
60 “ultimate” thermodynamic space (see Ismail et al. (2001)  
61 for detailed derivation). Moreover, with a unified  
62 expression, the “driving force constraints” can  
63 systematically identify necessary reaction and/or separation  
64 phenomena taking place in each module without pre-  
65 postulation of plausible separation tasks, reaction tasks, or  
66 reactive separation tasks. However, since nonideal  
67 separation systems are of focus in the current work, readers  
68 are referred to Ismail et al. (2001) and Tian et al. (2019) for  
69 GMF application in reactive separation systems.

70 A superstructure network is constructed to allow for all  
71 possible interconnections between each GMF module via  
72 the use of auxiliary mixers and splitters. In this context, the  
73 GMF model consists of a physical model to represent  
74 underlying phenomena taking place in each module as well  
75 as a structural model to generate structural alternatives  
76 (Ismail, 1999). To further extract intra-module operation  
77 information and module design details while maintaining  
78 combinatorial compactness, the Orthogonal Collocation  
79 (OC) method (Seferlis and Hrymak, 1994) has been coupled  
80 with GMF physical model and has been demonstrated for  
81 the synthesis of advanced separation systems (Proios and  
82 Pistikopoulos, 2006; Aligusane et al., 2006). In what  
83 follows, we present in detail the mathematical formulation  
84 for each sub-model.

## GMF Physical Model

The physical model is applied to evaluate the physical  
feasibility in each GMF module and the overall flowsheet  
optimality with respect to design targets. Due to the space  
limitation, we present in the following section the major  
constraints and equations for the modeling of a GMF  
mass/heat exchange module, while the comprehensive  
mathematical formulation involving physical models for  
other GMF structural components (e.g., mixer, splitter, etc.)  
can be found in Ismail et al. (1999, 2001).

– Mass balances for each component around each  
mass/heat exchange module

$$f_e^{LI} x_{ei}^{LI} + f_e^{VI} x_{ei}^{VI} - f_e^{LO} x_{ei}^{LO} - f_e^{VO} x_{ei}^{VO} = 0 \quad (1)$$

– Energy balances around each mass/heat exchange  
module

$$f_e^{LI} h_e^{LI} + f_e^{VI} h_e^{VI} - f_e^{LO} h_e^{LO} - f_e^{VO} h_e^{VO} = 0 \quad (2)$$

– Energy balances around utility exchangers of module

$$Q h_e - f_e^H (h_e^{HO} - h_e^{HI}) = 0 \quad (3)$$

$$Q c_e - f_e^C (h_e^{CI} - h_e^{CO}) = 0 \quad (4)$$

– Summation of molar fractions

For streams  $s = LI, LO, VI, VO$

$$\sum_i x_{ei}^s - 1 = 0 \quad (5)$$

– Phase defining constraints

For liquid streams

$$\sum_i (\gamma_{ei}^s P_{ei}^{sat,s} x_{ei}^s) / (\phi_{ei}^s P_{tot}) \leq 1 \quad (6)$$

For vapor streams

$$\sum_i (x_{ei}^s \phi_{ei}^s P_{tot}) / (\gamma_{ei}^s P_{ei}^{sat,s}) \leq 1 \quad (7)$$

– Mass transfer driving force constraints for each  
component

$$G1_{ei} \times G2_{ei} \geq 0 \quad (8)$$

$$G1_{ei} = f_e^{LI} x_{ei}^{LI} - f_e^{LO} x_{ei}^{LO} \quad (9)$$

$$G2_{ei} = \ln \left[ \frac{\gamma_{ei}^{LO} x_{ei}^{LO} P_{ei}^{sat,LO}}{\phi_{ei}^{VI} x_{ei}^{VI} P_{tot}} \right] \quad (10)$$

– Thermodynamic property calculation

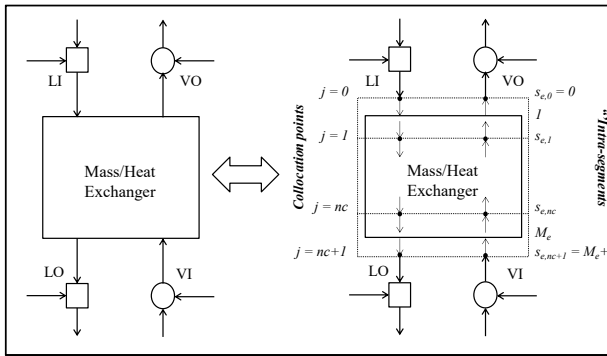
Thermodynamic properties (e.g., saturated vapor  
pressure ( $P^{sat}$ ), enthalpy ( $h$ ), activity coefficient ( $\gamma$ ),  
fugacity coefficient ( $\phi$ )) are calculated via detailed  
thermodynamic models (e.g., NRTL, UNIQUAC) for  
accurate capture of nonideal mixture properties.

## GMF/OC Physical Model

Each GMF mass/heat exchange module is highly  
compact in nature, featuring a certain mass/heat transfer  
pattern (e.g., component A transfer from liquid to vapor  
while component B and C vice versa). However, in practice,  
an aggregation of “intra-segments” inside these modules is

129 often necessitated to overtake the transfer duty to achieve  
 130 such “intensified” performances, in analogy to the  
 131 distillation trays in a column section. To enhance intra-  
 132 module representation while maintaining model size  
 133 compactness, Orthogonal Collocation method is integrated  
 134 with GMF physical model which obtains operation and  
 135 design details at a minimum essential set of “intra-  
 136 segments” defined by the “collocation points”. It is worth  
 137 noting that, from computational perspective, the  
 138 incorporation of OC approximation does not increase the  
 139 combinatorial size of resulting synthesis problem, as  
 140 continuous variables, instead of binary variables, are  
 141 introduced for the description of “intra-segments”.

142 Based on OC principles, each mass/heat exchange  
 143 module is discretized into  $nc$  interior and 2 exterior  
 144 collocation points, as can be seen in Figure 2. The location  
 145  $s_{e,j}$ , of each interior collocation point corresponds to a  
 146 number of “intra-segments” and is determined by the roots  
 147 of Hahn polynomials obeying orthogonality conditions as  
 148 an explicit function of the total number of “intra-segments”,  
 149  $M_e$ , in the corresponding module (Stewart et al., 1985).  
 150 After determining the discretization scheme (i.e., the  
 151 number of  $nc$ ), liquid and vapor component flowrate and  
 152 enthalpy variables are approximated at these collocation  
 153 points using Lagrange polynomials to further constitute  
 154 intra-module physical constraints. The GMF/OC physical  
 155 model thus comprises: (i) collocation point locations and  
 156 Lagrange polynomial weighting functions, (ii) physical  
 157 constraints enforced at interior collocation points (e.g.  
 158 mass/energy balances, driving force constraints, phase  
 159 defining constraints), and (iii) physical constraints to link  
 160 discretized collocation points with the overall module  
 161 inlet/outlet streams (Proios and Pistikopoulos, 2006). An  
 162 indicative list of modeling constraints is as follows.



164 *Figure 2. Mass/heat exchange module in*  
 165 *GMF and GMF/OC representation*  
 166 *(adapted from Proios and Pistikopoulos, 2006)*

167 – Collocation point locations

$$168 \quad s_{e,0} - 1 = 0$$

$$169 \quad s_{e,j} - s_j(M_e) = 0$$

$$170 \quad s_{e,nc+1} - (M_e + 1) = 0$$

171 – Lagrange polynomial weighting functions

$$172 \quad W_{j'}^V(s) - \prod_{z=1, z \neq j'}^{nc_e+1} \frac{s-s_{e,z}}{s_{e,j'}-s_{e,z}} = 0 \quad (18)$$

$$173 \quad W_{j'}^L(s) - \prod_{z=0, z \neq j'}^{nc_e} \frac{s-s_{e,z}}{s_{e,j'}-s_{e,z}} = 0 \quad (19)$$

– Component mass and energy balances

$$174 \quad \sum_{j'=1}^{nc_e+1} W_{j'}^V(s_{e,j} + 1) f c_c^V(s_{e,j'}) - f c_c^V(s_{e,j})$$

$$175 \quad + \sum_{j'=0}^{nc_e} W_{j'}^L(s_{e,j} - 1) f c_c^L(s_{e,j'}) - f c_c^L(s_{e,j}) = 0 \quad (20)$$

$$176 \quad \sum_{j'=1}^{nc_e+1} W_{j'}^V(s_{e,j} + 1) f^V(s_{e,j'}) h^V(s_{e,j'})$$

$$177 \quad + \sum_{j'=0}^{nc_e} W_{j'}^L(s_{e,j} - 1) f^L(s_{e,j'}) h^L(s_{e,j'})$$

$$178 \quad - f^V(s_{e,j}) h^V(s_{e,j}) - f^L(s_{e,j}) h^L(s_{e,j}) = 0 \quad (21)$$

– Conditions linking collocation points to module inlet/outlet streams

$$179 \quad f_e^{VO} - \sum_c (\sum_{j=1}^{nc_e+1} W_j^V(1) f c_c^V(s_{e,j})) = 0 \quad (22)$$

$$180 \quad f_e^{VO} h_e^{VO} - \sum_{j=1}^{nc_e+1} W_j^V(1) f^V(s_{e,j}) h^V(s_{e,j}) = 0 \quad (23)$$

### 181 GMF Structural Model

The structural model generates structural alternatives via the use of logical constraints to denote the existence of GMF modules, process streams, and their interconnection schemes. Some indicative constraints are presented below.

– To define the existence of each module

$$182 \quad y h_e - y_e \leq 0 \quad (24)$$

– To define the existence of module inlet/outlet streams (i.e., no streams exist if the module does not exist)

$$183 \quad (f_e^{LI} + f_e^{LO} + f_e^{VI} + f_e^{VO}) - y_e F^{max} \leq 0 \quad (25)$$

– To define the existence of interconnecting streams between modules

$$184 \quad f_{e'e}^s - y_{e'e}^s F^{max} \leq 0 \quad (26)$$

– To define that if a module exists, there will be an inlet/outlet flow

$$185 \quad y_e - (\sum_{n^{liq}} y l l_{n_e} + \sum_{e'} y l l_{e'e} + \sum_{e'} y c l_{e'e}) \leq 0 \quad (27)$$

$$186 \quad y_e - (\sum_{e'} y l l_{e'e} + \sum_{e'} y l h_{e'e} + \sum_{p^{liq}} y l p_{ep}) \leq 0 \quad (28)$$

### 187 Objective Function

In this work, the objective function is formulated to optimize cost performances. Utility cost can be explicitly considered based on the utility consumption given by Eq. (3) and Eq. (4). However, capital costing correlations cannot be directly applied as no equipment is pre-postulated within the GMF and GMF/OC approaches. Thus a pseudo-capital module cost function is adapted based on the design information provided by OC approximations, as developed in Proios and Pistikopoulos (2006). The overall objective function is formulated as:

$$188 \quad Obj = \sum_e C_{cw} Q c_e + \sum_e C_{st} Q h_e + \sum_e Cost_{mod,e} y_e \quad (29)$$

214 In this work, the resulting mixed-integer nonlinear  
 215 programming (MINLP) problem is solved using the  
 216 Generalized Benders Decomposition (GBD) method  
 217 (Geoffrion, 1972) implemented in the General Algebraic  
 218 Modeling System (GAMS). The primal NLP subproblem is  
 219 solved with solver CONOPT and the master MIP with  
 220 CPLEX. Nonetheless, due to the highly nonlinear and  
 221 nonconvex nature of the model, the problem is not solved to  
 222 global optimality. However, arguably even “intermediate”  
 223 solutions can provide useful information on the process.  
 224 Integer cuts can also be introduced to generate alternative  
 225 solution structures.

## 226 Case Study: Separation of Ethanol and Water

227 In this section, the ethanol-water separation problem in  
 228 Ismail et al. (1999) is revisited with GMF and GMF/OC  
 229 approaches. The resulting optimal GMF configurations are  
 230 also translated to equipment-based flowsheet alternatives  
 231 and validated with steady-state simulation, thus closing the  
 232 loop for the synthesis of intensified and verifiable process  
 233 systems.

### 234 Problem Statement

235 Ethanol (EtOH) and water (H<sub>2</sub>O) can form a minimum  
 236 boiling azeotrope of 89.43 mol% ethanol at a temperature  
 237 of 351.15 K and a pressure of 1 atm. Based on initial  
 238 analysis of residue curve maps, ethylene glycol (EG), a  
 239 heavy entrainer, and methanol (MeOH), a light entrainer,  
 240 are selected as candidate mass separating agents, both of  
 241 which can facilitate the separation by forming a  
 242 homogenous mixture with the feed components.

243 An atmospheric ethanol-water liquid feed (EtOH  
 244 85mol%, H<sub>2</sub>O 15 mol%) at a flow rate of 10 kmol/s and a  
 245 temperature of 351.3 K is considered. The separation task is  
 246 to obtain liquid ethanol at a rate of at least 8 kmol/s with a  
 247 purity higher than 99 mol%. The system pressure is fixed at  
 248 1 atm. The NRTL equation is utilized to capture the  
 249 nonideal liquid mixture properties using the thermodynamic  
 250 parameters given in Ismail et al (1999). The synthesis  
 251 objective is to identify a suitable entrainer and to determine  
 252 a cost-optimal design configuration to meet the afore-  
 253 mentioned production specifications.

### 254 Synthesis of EtOH-H<sub>2</sub>O Separation System with GMF

255 In this case study, a maximum of 8 GMF mass/heat  
 256 exchange modules and 16 associated pure heat exchange  
 257 modules are allowed for the representation/synthesis of this  
 258 process. The resulting synthesis model involves 437  
 259 constraints, 4864 continuous variables, and 478 binary  
 260 variables. Only operating cost, comprising hot and cold  
 261 utility cost, is considered in the objective function to  
 262 determine an optimal design configuration with respect to  
 263 energy consumption. Capital cost estimation will be  
 264 incorporated later in GMF/OC synthesis to capture the  
 265 interaction of operating and capital cost in these systems.

The optimal separation scheme is illustrated in Figure  
 3 with an operating cost of  $\$6.34 \times 10^7/\text{y}$ . The heavy  
 entrainer ethylene glycol is selected to facilitate the  
 separation of desired ethanol product from overhead. Three  
 mass/heat exchange modules are utilized, interconnected  
 with two pure heat exchange modules, suggesting a  
 configuration of extractive distillation (to be verified later  
 with equipment-based simulation). This design is consistent  
 with the optimal structure identified in Ismail et al. (1999).

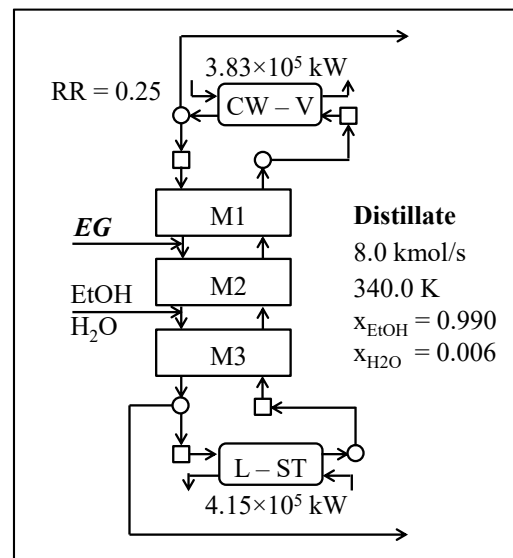


Figure 3. GMF configuration:  
 EtOH-H<sub>2</sub>O-EG extractive distillation

To illustrate the ability of GMF to identify versatile  
 process alternatives, two intermediate solutions with higher  
 operating costs are shown in Figure 4. Alternative 1 also  
 employs EG as entrainer. However, instead of being  
 intensified into a single “column” configuration as that in  
 Figure 3, Alternative 1 consists of a side-module coupled  
 with the main “column”. Comparing to the optimal solution,  
 larger reflux flowrates are necessitated in this system due to  
 the less integrated and thus less efficient separation scheme,  
 resulting in an operating cost of  $\$6.98 \times 10^7/\text{y}$ .

Alternative 2 in Figure 4(b) illustrates the process  
 structure with MeOH as entrainer. In the ternary mixture of  
 MeOH-EtOH-H<sub>2</sub>O, ethanol is the intermediate component  
 with respect to volatility. Therefore, two separation steps  
 are identified in this process solution to obtain pure EtOH  
 products. The first separation module removes water via the  
 bottom liquid flow, while the vapor flow is directed to the  
 other module for purification of ethanol from methanol. In  
 a sequential manner of operation, this flowsheet requires an  
 operating cost of  $\$11.4 \times 10^7/\text{y}$ .

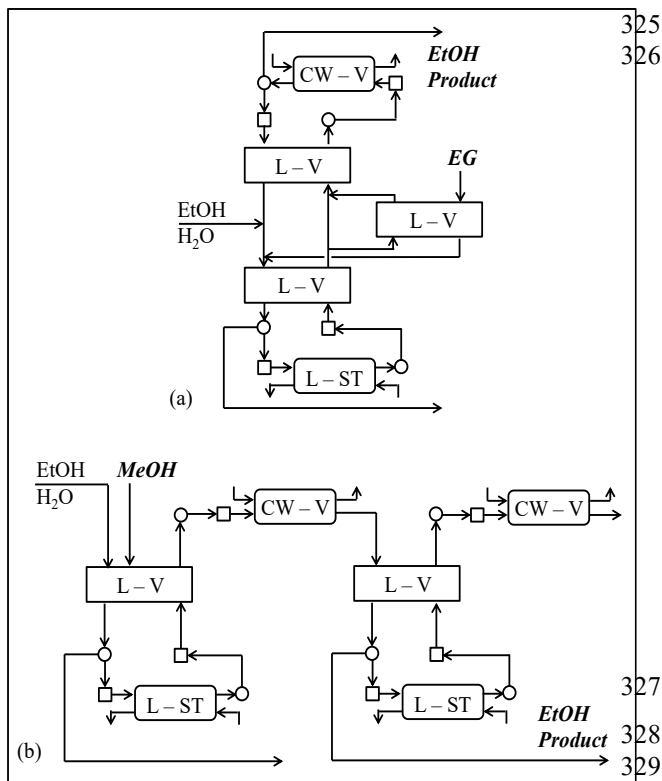


Figure 4. Intermediate GMF configurations:  
 EtOH-H<sub>2</sub>O extractive separation  
 (a) Alternative 1 (b) Alternative 2

### Synthesis of EtOH-H<sub>2</sub>O Separation System with GMF/OC

In this section, GMF/OC approach is applied to investigate the afore-introduced ethanol-water separation problem. However, the synthesis task is driven by the minimization of total annualized cost (TAC) accounting for utility cost and capital cost.

The pre-specified discretization scheme (i.e., number of collocation points) in each mass/heat exchange module plays a key role to enhance physical representation accuracy as well as to monitor model size compactness. For this reason, different numbers of interior collocation points have been used to discretize each GMF module. As presented in Table 1, discretization with 2 interior collocation points provides the best cost performance for the design of this extractive separation process. The optimal number of “intra-segments” in each module is also given in Table 1.

The resulting optimal GMF/OC solution features the same modular configuration as the optimal GMF solution (Figure 3). This is consistent with the observation in Proios and Pistikopoulos (2006) for a broader scope of distillation systems, as operating cost constitutes the major part of TAC in these systems and thus serves as the dominant driving force for cost optimization.

Table 1. GMF/OC synthesis for EtOH-H<sub>2</sub>O Separation with different discretization schemes

| Interior collocation points | 1                     | 2                    | 3                    |
|-----------------------------|-----------------------|----------------------|----------------------|
| “Intra-segments”            | M <sub>1</sub> = 10.0 | M <sub>1</sub> = 2.9 | M <sub>1</sub> = 3.3 |
|                             | M <sub>2</sub> = 2.0  | M <sub>2</sub> = 6.0 | M <sub>2</sub> = 3.0 |
|                             | M <sub>3</sub> = 10.0 | M <sub>3</sub> = 2.4 | M <sub>3</sub> = 3.0 |
|                             |                       |                      |                      |
| Reflux ratio                | 0.30                  | 0.25                 | 0.27                 |
| Q <sup>H</sup> (kW)         | 3.86×10 <sup>5</sup>  | 3.70×10 <sup>5</sup> | 3.87×10 <sup>5</sup> |
| Q <sup>C</sup> (kW)         | 4.12×10 <sup>5</sup>  | 4.03×10 <sup>5</sup> | 4.05×10 <sup>5</sup> |
| Operating Cost (\$/y)       | 6.37×10 <sup>7</sup>  | 6.13×10 <sup>7</sup> | 6.38×10 <sup>7</sup> |
| Capital Cost (\$/y)         | 0.73×10 <sup>7</sup>  | 0.63×10 <sup>7</sup> | 0.63×10 <sup>7</sup> |

### Steady-state Simulation and Validation

In this step, we validate the above-derived optimal GMF and GMF/OC configurations with equipment-based simulation using Aspen Plus® to verify if extractive distillation is suggested as the process option for this ethanol-water separation problem. As per the estimation of “intra-segments” by the GMF/OC synthesis with 2 interior collocation points, the modular configuration is translated to an 11-tray extractive distillation column. Accordingly, EtOH-H<sub>2</sub>O feed is introduced onto the 10<sup>th</sup> tray while EG entrainer enters the 4<sup>th</sup> tray.

The simulation results for the equipment-based flowsheet alternative are presented in Figure 5. Figure 6 shows the comparison profiles of column temperature and liquid composition between GMF, GMF/OC, and Aspen simulation. As shown, both GMF and GMF/OC match well with the major temperature and composition trends inside the extractive column, while GMF/OC provide further design and operation details within each column section. Up to this stage, the extractive distillation configuration shown in Figure 5 is identified as the optimal process solution in this case study.

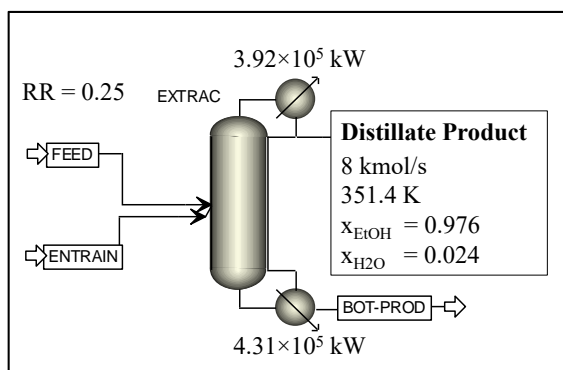
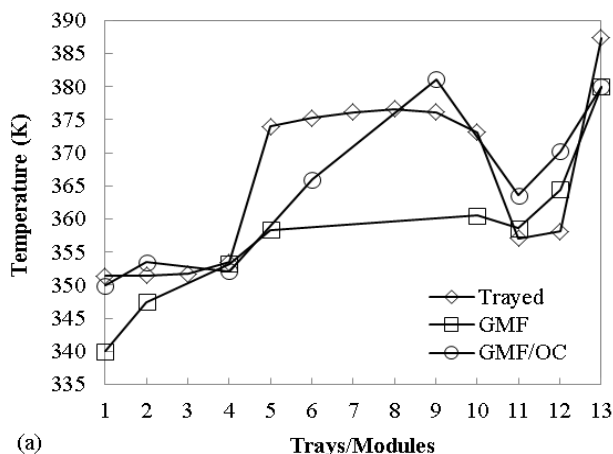
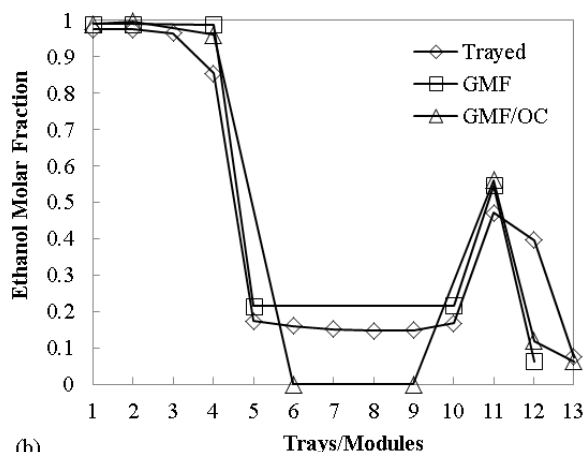


Figure 5. Aspen Plus simulation & validation of EtOH-H<sub>2</sub>O-EG extractive distillation



(a)



(b)

Figure 6. Steady-state validation: GMF & GMF/OC configurations (a) Temperature profile (b) EtOH composition

### 358 Conclusion

359 In this paper, we have presented a systematic framework, i.e. the Generalized Modular Representation

361 Framework, for the synthesis of process intensification  
 362 systems, with specific focus on extractive separation  
 363 process. It is demonstrated, through the synthesis and design  
 364 of an ethanol-water separation system, that GMF is able to  
 365 simultaneously address entrainer selection, unit/flowsheet  
 366 synthesis, design, and intensification in a single problem  
 367 formulation. Moreover, GMF/OC allows for detailed intra-  
 368 module physical representation and provides module  
 369 dimensionality information, while maintaining the GMF's  
 370 main advantage of combinatorial compactness. Ongoing  
 371 work focuses on applying the GMF and GMF/OC  
 372 approaches to membrane-assisted systems and cyclic  
 373 operation systems. Extensions of the synthesis framework  
 374 to include operability, safety, and dynamic controllability  
 375 assessment are also under current investigation (Tian,  
 376 2019).

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### 81 References

- 82 Alguasane, T. Y., Proios, P., Georgiadis, M. C., & Pistikopoulos, E.  
 83 N. (2006). A framework for the synthesis of reactive  
 84 absorption columns. *Chemical Engineering and  
 85 Processing: Process Intensification*, 45(4), 276-290.  
 86 da Cruz, F. E., & Manousiouthakis, V. I. (2017). Process  
 87 intensification of reactive separator networks through  
 88 the IDEAS conceptual framework. *Computers &  
 89 Chemical Engineering*, 105, 39-55.  
 90 Demirel, S. E., Li, J., & Hasan, M. F. (2017). Systematic process  
 91 intensification using building blocks. *Computers &  
 92 Chemical Engineering*, 105, 2-38.  
 93 Geoffrion, A. M. (1972). Generalized benders  
 94 decomposition. *Journal of optimization theory and  
 95 applications*, 10(4), 237-260.  
 96 Ismail, S. R., Pistikopoulos, E. N., & Papalexandri, K. P. (1999).  
 97 Modular representation synthesis framework for  
 98 homogeneous azeotropic separation. *AIChE  
 99 journal*, 45(8), 1701-1720.  
 400 Ismail, S. R., Proios, P., & Pistikopoulos, E. N. (2001). Modular  
 401 synthesis framework for combined separation/reaction  
 402 systems. *AIChE Journal*, 47(3), 629-649.  
 403 Kossack, S., Kraemer, K., Gani, R., & Marquardt, W. (2008). A  
 404 systematic synthesis framework for extractive  
 405 distillation processes. *Chemical Engineering Research  
 406 and Design*, 86(7), 781-792.  
 407 Mahdi, T., Ahmad, A., Nasef, M. M., & Ripin, A. (2015). State-  
 408 of-the-art technologies for separation of azeotropic  
 409 mixtures. *Separation & Purification Reviews*, 44(4),  
 410 308-330.  
 411 Papalexandri, K. P., & Pistikopoulos, E. N. (1996). Generalized  
 412 modular representation framework for process  
 413 synthesis. *AIChE Journal*, 42(4), 1010-1032.  
 414 Proios, P., & Pistikopoulos, E. N. (2006). Hybrid generalized  
 415 modular/collocation framework for distillation column  
 416 synthesis. *AIChE journal*, 52(3), 1038-1056.  
 417 Seferlis, P., & Hrymak, A. N. (1994). Optimization of distillation  
 units using collocation models. *AIChE journal*, 40(5),  
 813-825.

- 420 Stewart, W. E., Levien, K. L., & Morari, M. (1985). Simulation of  
421 fractionation by orthogonal collocation. *Chemical*  
422 *Engineering Science*, 40(3), 409-421.
- 423 Tian, Y., Demirel, S. E., Hasan, M. F., & Pistikopoulos, E. N.  
424 (2018). An overview of process systems engineering  
425 approaches for process intensification: State of the  
426 art. *Chemical Engineering and Processing-Process*  
427 *Intensification*, 133, 160-210.
- 428 Tian, Y., Mannan, M. S., & Pistikopoulos, E. N. (2018). Towards  
429 a systematic framework for the synthesis of operable  
430 process intensification systems. In *Computer Aided*  
431 *Chemical Engineering* (Vol. 44, pp. 2383-2388).  
432 Elsevier.
- 433 Tian, Y.; Pistikopoulos, E. N. Synthesis of Operable Process  
434 Intensification Systems - Steady-state Design with  
435 Safety and Operability Considerations. Submitted.
- 436 Tula, A. K., Babi, D. K., Bottlaender, J., Eden, M. R., & Gani, R.  
437 (2017). A computer-aided software-tool for sustainable  
438 process synthesis-intensification. *Computers &*  
439 *Chemical Engineering*, 105, 74-95.