

## **X-ray nanodiffraction studies of ionically-controlled nanoscale phase separation in cobaltites**

Geoffery Rippy<sup>1</sup>, Lacey Trinh<sup>1</sup>, Alexander M. Kane<sup>1</sup>, Aleksey L. Ionin<sup>1</sup>, Michael S. Lee<sup>1</sup>, Rajesh V. Chopdekar<sup>1</sup>, Joyce M. Christiansen-Salameh<sup>1</sup>, Dustin A. Gilbert<sup>2</sup>, Alexander J. Grutter<sup>2</sup>, Peyton D. Murray<sup>3</sup>, Martin [V. Holt](#)<sup>4,5</sup>, Zhonghou Cai<sup>4,5</sup>, Kai Liu<sup>3</sup>, Yayoi Takamura<sup>1</sup>, Roopali Kukreja<sup>1</sup>

1 – Department of Materials Science and Engineering, University of California Davis, Davis, CA 95616

2 – NIST Center for Neutron Research, National Institute of Standards and Technology, Gaithersburg, MD 20899

3 – Physics Department, University of California Davis, Davis, CA 95616

4 – Center for Nanoscale Materials, Argonne National Laboratory, Lemont, IL 60439

5 – Advanced Photon Source, Argonne National Laboratory, Lemont, IL 60439

### **Abstract**

*Complex oxide heterostructures provide access to emergent functional and structural phases which are not present in the bulk constituent materials. In this letter, we focus on Gd/La<sub>0.67</sub>Sr<sub>0.33</sub>CoO<sub>3</sub> (LSCO) heterostructures due to the high oxygen ion conductivity, as well as the coupled magnetic and electronic properties of LSCO, which are strongly dependent on the oxygen stoichiometry. This combination of properties enable the ionic control of the functional properties of LSCO thin films through the presence of oxygen getter layers such as Gd. We utilize x-ray nanodiffraction to directly image the nanoscale morphology of LSCO thin films as they are progressively transformed from the equilibrium perovskite phase to the metastable brownmillerite (BM) phase with increasing Gd thickness. Our studies show the coexistence of perovskite and BM phases with a critical oxygen vacancy concentration threshold which leads to the formation of extended BM filaments. These filaments were oriented at either ~10° or ~50-60° angles from the [100] direction. In addition to lateral phase separation, we observed phase separation within the film thickness possibly due to pinning of the perovskite or BM phase by the substrate/LSCO or Gd/LSCO interface, respectively. Furthermore, strain maps reveal that the perovskite phase changes from compressive to tensile strain on opposite sides of a BM filament.*

Perovskite oxides ( $ABO_3$ ) exhibit a wide array of tunable functional properties which can be manipulated by tailoring ionic distribution and stoichiometry. Numerous studies have focused on understanding and controlling functionalities via changing  $A/B$  cation stoichiometry in the perovskite structure [1-6]. Recently, there has been a growing interest in utilizing anion stoichiometry [7-13] or substitution i.e. formation of oxyfluorides [14, 15], to significantly alter and tune the electronic, magnetic, and structural properties. In addition, the presence of the closely related, oxygen-deficient brownmillerite (BM) phase ( $ABO_{2.5}$ ) provides a rich phase diagram with a wide range of functional properties, which are highly sensitive to the oxygen stoichiometry and thus can be tailored via ionic control. Specifically, structural parameters such as  $BO_6$  octahedral bond distances and angles [8, 16], as well as electronic parameters such as cation valence states [17, 18] are sensitive to the local anion environment and can be modified using epitaxial strain [19, 20], external fields [17, 21] or ionic migration [22]. Large oxygen non-stoichiometry has been demonstrated in  $SrCoO_{3-\delta}$  (BM phase), and methods such as oxygen/vacuum anneals, ionic liquids or electrochemical differences in an electrochemical cell geometry have been utilized to control the oxygen stoichiometry and manipulate functional properties [21, 23, 24]. Their highly tunable ionic and electronic transport properties also makes them potential candidates for application in energy devices such as solid oxide fuel cells, information storage, and computing technologies including magneto-ionic devices and memristors [25, 26].

The deposition of a strong oxygen getter on top of an oxide thin film has recently emerged as a novel way to tailor oxygen stoichiometry and nanoscale functional properties [7, 27, 28]. For example, the presence of a strong oxygen getter (e.g. Gd) led to the suppression of the magnetization of  $La_{0.67}Sr_{0.33}CoO_3$  (LSCO)/Gd heterostructures throughout the full film thickness (36 nm) [27]. In contrast, applying a similar approach to material systems such as (Ni,Co)O/GdFe heterostructures produced a thin (3 nm) layer of ferromagnetic NiCo at the interface [29]. The resulting antiferromagnetic-to-ferromagnetic conversion allowed control of exchange bias at low magnetic fields. In both cases, this phenomenon was associated with the oxidation of the getter layer during which oxygen ions were extracted from the oxide underlayer. The vertical extent of the reaction was postulated to depend on the relative stabilities of the oxide and metal layers, as well as the oxygen ion conductivity and the work function of the getter layer. However, the role played by nanoscale morphology, i.e. phase separation and defects in these oxide thin films and heterostructures remains largely unknown. (Scanning) transmission electron microscopy

(STEM/TEM) provides a direct method to access atomic lengthscales, however it is restricted to extremely thin samples ( $< 10$  nm) and the sample preparation, as well as the imaging process itself, can modify the nanoscale morphology and interfacial properties.[30-32] Thus, while the observation of ordered oxygen vacancies in  $\text{La}_{1-x}\text{Sr}_x\text{CoO}_3$  thin films in STEM images can be explained as a means alleviate epitaxial strain, non-destructive x-ray diffraction-based characterization of long-range oxygen vacancy ordering is still lacking.[33-36]

In this letter, we present x-ray nanodiffraction studies to investigate the nanoscale morphology of cobaltite thin films with Gd getter layers. Our nanodiffraction measurements provide an alternative structural characterization method with nanometer spatial resolution ( $\sim 25$  nm) in a non-destructive manner to directly image the nanoscale structure [37]. X-ray measurements were performed by tuning to the Bragg reflections which are sensitive to each of the structural rearrangements induced by variations in oxygen stoichiometry of the LSCO film as the Gd layer thickness increases. We observed phase separation of the perovskite and oxygen-deficient BM phases in all samples including the as-grown LSCO reference sample, although, the amount of BM phase varied dramatically. For the as-grown sample, we observed few  $0.5$ - $3$   $\mu\text{m}$  clusters of the BM phase, while for the LSCO/Gd ( $0.5$  nm) heterostructure, BM clusters grew to lengthscales of  $4$   $\mu\text{m}$ . Surprisingly, for the LSCO/Gd ( $3$  nm) heterostructure, we observed long filaments in the range of  $10$ - $75$   $\mu\text{m}$  oriented at either  $\sim 10^\circ$  or  $\sim 50$ - $60^\circ$  angles from the  $[100]$  direction. This result indicates that a critical oxygen vacancy may be required to form the BM filaments. Furthermore, in addition to the lateral phase separation, phase separation was observed vertically along the thickness of the film possibly due to pinning of perovskite or BM phase by substrate/LSCO or LSCO/Gd interface, respectively. These results not only advance our understanding of nanoscale morphology in oxides, but also demonstrate powerful approach to investigate structural and electronic properties at relevant lengthscales.

Epitaxial  $\text{La}_{0.67}\text{Sr}_{0.33}\text{CoO}_3$  ( $36$  nm) thin films were grown by pulsed laser deposition on (001)-oriented  $(\text{La}_{0.18}\text{Sr}_{0.82})(\text{Al}_{0.59}\text{Ta}_{0.41})\text{O}_3$  (LSAT) substrates. The laser energy density utilized for the deposition was  $0.9$   $\text{J}/\text{cm}^2$  at a  $1$  Hz repetition rate. The samples were deposited at  $700$   $^\circ\text{C}$  with an oxygen pressure of  $300$  mTorr and then were slowly cooled to room temperature in  $300$  Torr oxygen pressure. Thin films of elemental Gd ( $0.5$  and  $3$  nm)/Au ( $3$  nm) were then deposited by sputtering at room temperature in an Ar environment ( $5$  mTorr). Due to the large negative heat of

formation ( $\Delta H_f$ ) for  $\text{GdO}_x$ , the Gd layer spontaneously oxidized upon deposition [38], with the LSCO film acting as the source of the oxygen ions. In order to control the oxygen ion distribution in the LSCO films, two different Gd getter layer thickness were deposited (0.5 and 3 nm). The Au layer was deposited to prevent oxygen exchange with the ambient environment.

X-ray nanodiffraction experiments were conducted at Hard X-ray Nanoprobe (26-ID-C) beamline of the Advanced Photon Source and Center for Nanoscale Materials at Argonne National Laboratory. Figure 1(a) shows a schematic of the x-ray nanodiffraction experimental geometry. The monochromatic (10.4 keV) incident x-ray beam was focused by a Fresnel zone plate down to 25 nm spot size on the sample. The focusing led to an overall effective beam divergence of  $0.24^\circ$ . Zone plate center stop and order sorting aperture were used to remove all but the first order diffracted x-ray beam. As the zone plate center stop blocks the central part of the x-ray beam, the resulting first order diffracted beam after the zone plate was hollow in the middle. This first order diffracted x-ray beam was utilized to perform nanodiffraction measurements in reflection geometry. The diffracted beam from the sample was acquired using a Pixel Area Detector (PAD) detector located 70 cm away from the sample with a pixel size of 55  $\mu\text{m}$ . The sample theta was tuned to access the  $(002)_p$  Bragg reflection for the perovskite phase and the  $(008)_{bm}$  Bragg reflection for the BM phase simultaneously, thus allowing clear identification of both phases as shown in Figure 1(b). Here the subscripts  $p$  and  $bm$  refer to the Miller index notation for the perovskite and BM structures respectively, and will be used throughout the article. Note that the BM phase is composed of alternating layers of Co ions in octahedral and tetrahedral coordination, resulting in a quadrupling of the unit cell in the  $c$ -direction in comparison to the parent perovskite phase [2, 39]. Therefore, the BM phase can be characterized in an x-ray diffraction (XRD) experiment by a shift of the main film Bragg reflection to lower  $2\theta$  values as well as the appearance of half order reflections. Figure 1(b) shows the XRD patterns collected near the  $(002)_p$  Bragg reflection at 10.4 keV. Bragg reflections from both the BM and perovskite phases were observed for the LSCO/Gd heterostructures, while only the perovskite phase was observed for the as-grown LSCO thin film without any metal capping. The lattice parameters obtained for the perovskite phase were  $3.803 \pm 0.002 \text{ \AA}$  for the as-grown LSCO sample, and  $3.805 \pm 0.002 \text{ \AA}$  and  $3.819 \pm 0.002 \text{ \AA}$  for the LSCO/Gd(0.5 nm) and Gd(3 nm) heterostructures, in agreement with literature values. Detailed characterization information for these films can be found in Gilbert [27].

In order to generate high-resolution nanodiffraction maps, the sample was raster scanned using hybrid opto-mechanical nano-positioning stages [37], recording the diffraction image at each step. The intensity of each pixel of the map represents the integrated intensity on the PAD detector over the given Bragg reflection for a particular sample position. Figure 2(a) shows a nanodiffraction map for the LSCO/Gd (3 nm) heterostructure tuned to the  $(002)_p$  Bragg condition. Perovskite-rich regions (yellow/white) are evident by the measured diffraction pattern with higher intensity for the  $(002)_p$  reflection as shown in Figure 2(b). Figure 2(c) shows the same nanodiffraction map but with integrated intensity for the  $(008)_{bm}$  Bragg reflection. The reversal of contrast compared to Fig. 2(a) indicates that perovskite-deficient regions indeed consist of BM-rich regions. This fact is also evident by the diffraction pattern shown in Figure 2(d), where an increase in intensity for the  $(008)_{bm}$  Bragg reflection is observed. A zoomed-in scan of the BM filament showing the complex interface between the perovskite and BM phases and the contrast reversal for the integrated intensity in the same region are presented in the insets of Figure 2(a) and 2(c). These images as well as multiple other nanodiffraction maps (not shown) measured on this sample are consistent with sample-averaged XRD measurements, where Bragg reflections from both perovskite and BM phase are observed indicating that the film is not fully transformed to BM phase. The measured nanodiffraction maps also reveal that even in perovskite-rich regions further away from BM-rich areas, the  $(008)_{bm}$  reflection, albeit weak is still present (Fig. 2(b)). Similarly, in BM-rich areas, the  $(002)_p$  reflection is always observed, revealing that the phase separation exists not only in the lateral direction but also along the film thickness. This fact indicates that the perovskite phase never completely transforms into the BM phase at any given location, and that some BM phase is formed everywhere on the sample surface.

Figure 3 compares the nanodiffraction maps for the LSCO/Gd heterostructures as well as the as-grown LSCO reference sample. The LSCO/Gd (0.5 nm) heterostructure shows the presence of BM regions with size ranging from 2-4  $\mu\text{m}$ , while for the LSCO/Gd (3 nm) heterostructures, long BM filaments in the range of 10-75  $\mu\text{m}$  were observed. For the LSCO/Gd(0.5 nm) heterostructure, only one area showed filament formation and the rest showed similar distribution as Fig. 3(b). In contrast to sample-averaged XRD measurements, even the as-grown LSCO sample showed the presence of BM phase, although the BM regions were fairly scattered across the sample with sizes ranging from 0.5-3  $\mu\text{m}$ . These measurements show that the LSCO/Gd (3 nm) heterostructure progressed the furthest in the phase transformation with multiple regions of BM-rich filaments

compared to the other samples.

Figure 3(d) presents  $\theta/2\theta$  scan for the *local* perovskite- and BM-rich areas for all three samples, measured at the highlighted region in Figure 3(a), (b), and (c). The BM phase in the LSCO/Gd (3 nm) heterostructure was highly mosaic and the lattice parameter varied from region to region as confirmed by the shift of the  $(008)_{\text{bm}}$  Bragg reflection between two different areas (solid blue diamonds in Figure 3(c)). The variation in the BM lattice parameter ranged from  $15.36 \pm 0.01 \text{ \AA}$  to  $15.32 \pm 0.01 \text{ \AA}$ , a difference of  $0.04 \text{ \AA}$  across the sample (in comparison to difference of  $0.01 \text{ \AA}$  for perovskite phase within a given sample). Similarly, for the LSCO/Gd(0.5 nm) heterostructure and as-grown LSCO sample, lattice parameters of  $15.36 \pm 0.01 \text{ \AA}$  and  $15.40 \pm 0.01 \text{ \AA}$  were obtained for the BM phase resulting in a sample to sample variation of  $0.04 \text{ \AA}$ . In contrast, the sample to sample variation obtained for the perovskite phase is  $0.03 \text{ \AA}$ . As shown in Fig. 3(d), with increasing Gd thickness, the perovskite lattice parameter increases which is consistent with the incorporation of an increasing concentration of randomly distributed oxygen vacancies and agrees with the trend observed in parallel-beam  $\theta/2\theta$  measurements (Fig. 1(b)). Here we note that while the BM phase and LSAT substrate reflections are close to each other in a sample-averaged XRD pattern, by locally measuring  $\theta/2\theta$  scans over the BM phase, we were able to isolate the  $(008)_{\text{bm}}$  reflection from the substrate as shown in Fig. 3(d), which allowed precise calculation of the BM lattice parameters discussed above.

In order to depict the local variation of the lattice parameter of the perovskite-rich phase, we calculated the strain maps for the LSCO/Gd (3 nm) heterostructure as shown in Figure 4(a) and (b). The strain is defined as  $(a_{x,y} - a_{\text{ref}})/a_{\text{ref}}$ , where  $a_{x,y}$  is local lattice parameter obtained from nanodiffraction maps, and  $a_{\text{ref}}$  is the sample-averaged lattice parameter for the perovskite phase calculated by averaging the lattice parameter measured in all nanodiffraction scans for the sample. The BM phase is also highlighted in the figure and the strain for the residual perovskite phase in BM-rich region is plotted as well. Both the strain maps clearly reveal that the perovskite phase exists in two different strain states on opposite sides of BM filament. Figure 4(c) and 4(d) plots line scans adjacent to BM phase showing that the compressive strain in perovskite phase changes to tensile strain across the BM filament. This sign reversal of the strain was present consistently in all regions with BM filaments measured for the LSCO/Gd (3 nm) heterostructure

The fact that these extended filaments were prominently observed for the LSCO/Gd(3 nm) heterostructure, and were rarely observed in the LSCO/Gd(0.5 nm) heterostructure, as well as their complete absence in the as-grown LSCO sample suggests the presence of a critical oxygen vacancy concentration in order to achieve growth of long BM filaments. Once the oxygen vacancy concentration exceeds this threshold, small BM clusters grow to form long filaments as observed in the LSCO/Gd(3 nm) heterostructure. The filament orientation is typically  $50^\circ$ - $60^\circ$  or near  $10^\circ$  from the [100] in-plane direction. In the BM structure, the oxygen vacancies are typically arranged within the BO layer resulting in chains of vacancies parallel to the [110] direction of the perovskite lattice. This rearrangement results in chains of apex-linked  $\text{BO}_4$  tetrahedra which also are along the [110] direction of the perovskite lattice. Distortions to these tetrahedral chains have been observed in  $\text{La}_{(1-x)}\text{A}_x\text{MnO}_{2.5}$  as a function of composition and size of cation with deviations in the range of  $60^\circ$ - $70^\circ$  [39]. These tetrahedral chain distortions as well as the fact that we are imaging these vacancy channels in diffraction geometry, i.e. at an angle of about  $\sim 20^\circ$  from the sample, can explain the observed orientation of  $50^\circ$ - $60^\circ$  for the BM filaments in the nanodiffraction maps. The presence of BM filaments with orientation of  $\sim 10^\circ$ , however is still puzzling and could potentially imply much higher distortions of the emerging BM phase.

It is also noticeable that the BM phase grows to alleviate the strain in perovskite phase due to consistent observation of change in sign of strain across the BM filament. Additionally, as discussed above, the perovskite phase never completely transforms into BM phase at any given location and vice-versa. Thus, even in the BM filaments, a small amount of perovskite is present potentially due to pinning of the perovskite by the LSAT substrate. Similarly, for the perovskite phase, a small amount of BM phase is always observed which potentially implies a complete depletion of the oxygen in the top LSCO layer, resulting in the BM transformation at the Gd/LSCO interface throughout the LSCO film. These measurements thus allow us to not only discern and image the phase separation in Gd/LSCO heterostructures laterally, but also provide indication of phase separation across the thin film depth.

In summary, our x-ray nanodiffraction measurements shed light on the structural transformation in LSCO heterostructures as a function of Gd getter layer thickness in a nondestructive manner. This transformation proceeds by phase separation, resulting in an evolution of lateral distribution of perovskite and BM phase with Gd thickness. Phase separation across the depth is also observed

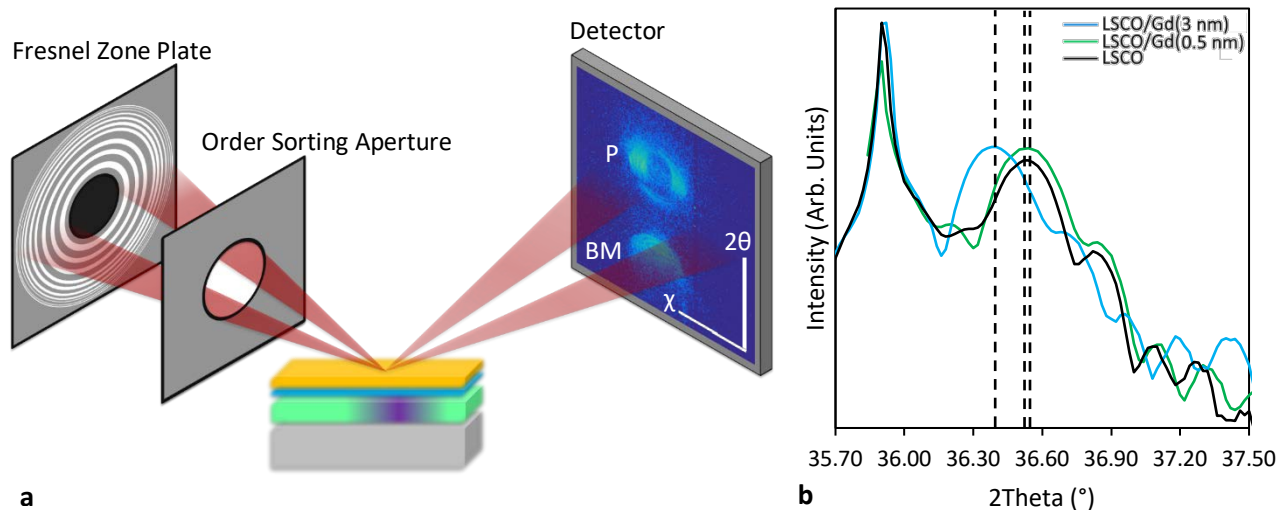
due to pinning of perovskite and BM phase at the LSCO/substrate and LSCO/Gd interface, respectively. Above a critical thickness of Gd layer (3 nm), long BM filaments are observed due to ordering of oxygen vacancies at 50°-60° or 10° angles from the [100] direction. The emerging BM phase is highly distorted as observed by the deviations of tetrahedral chains from the [110] directions, high mosaicity, and variations in lattice parameter of 0.04 Å. The perovskite phase near the BM regions shows a change in strain state from compressive to tensile on opposite sides of BM filaments. These measurements highlight local structural effects in Gd/LSCO heterostructures and provide insights regarding nanoscale strain behavior near phase boundaries. Manipulation of these nanoscale heterogeneities i.e. BM filaments by varying getter layer or electric field can provide a unique method to tailor the electronic and ionic conductivity in an ionically-gated device. These nanoscale filaments can provide channels across which oxygen ion conduction can occur without the need of bulk structural transformation and thus will be more energetically favorable. Reversible control of these filaments will open new avenues of controlling anion conduction in applications such as solid oxide fuel cells and magneto-ionic devices.

### **Acknowledgements**

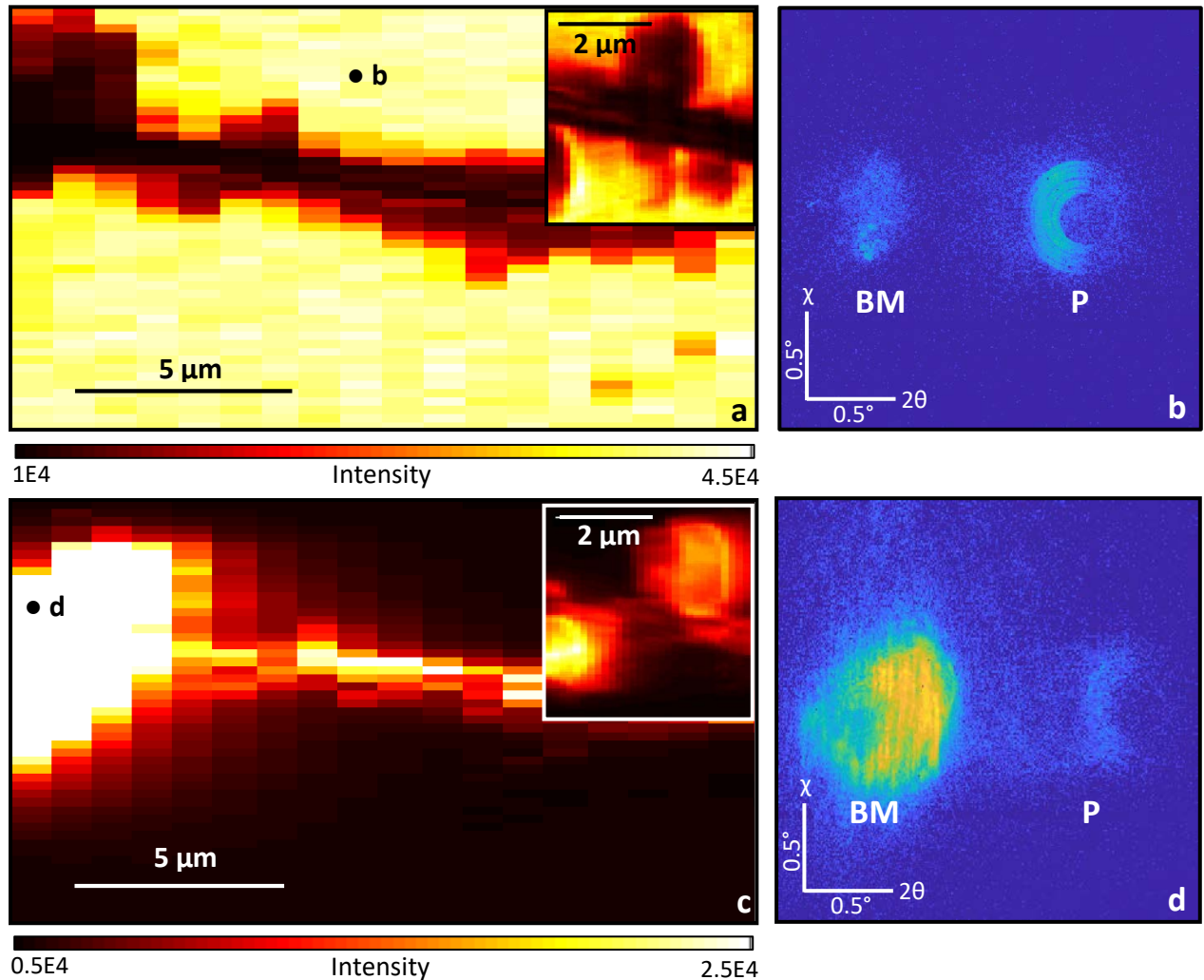
G. R., L.T., J.M.C.-S., R.K. acknowledges the support by University of California Davis for this research while A.M.K., A.L.I., M.S.L., R.V.C., and Y.T acknowledge the University of California Multicampus Research Programs and Initiatives (Grant No. MR-15-328528). This research used resources of the Advanced Photon Source and the Center for Nanoscale Materials, both U.S. Department of Energy (DOE) Office of Science User Facilities operated for the DOE Office of Science by Argonne National Laboratory under Contract No. DE-AC02-06CH11357.



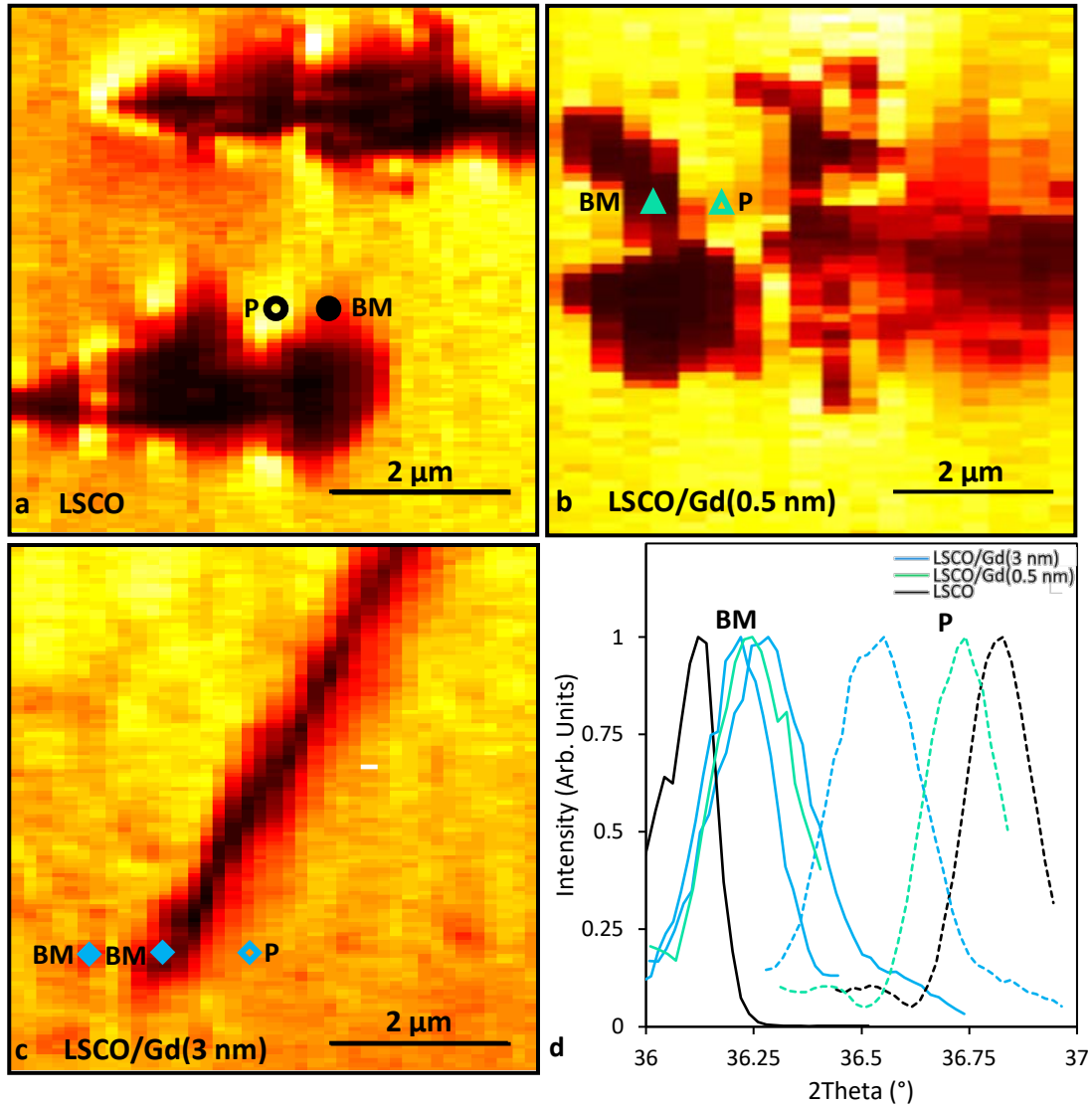
## Figures



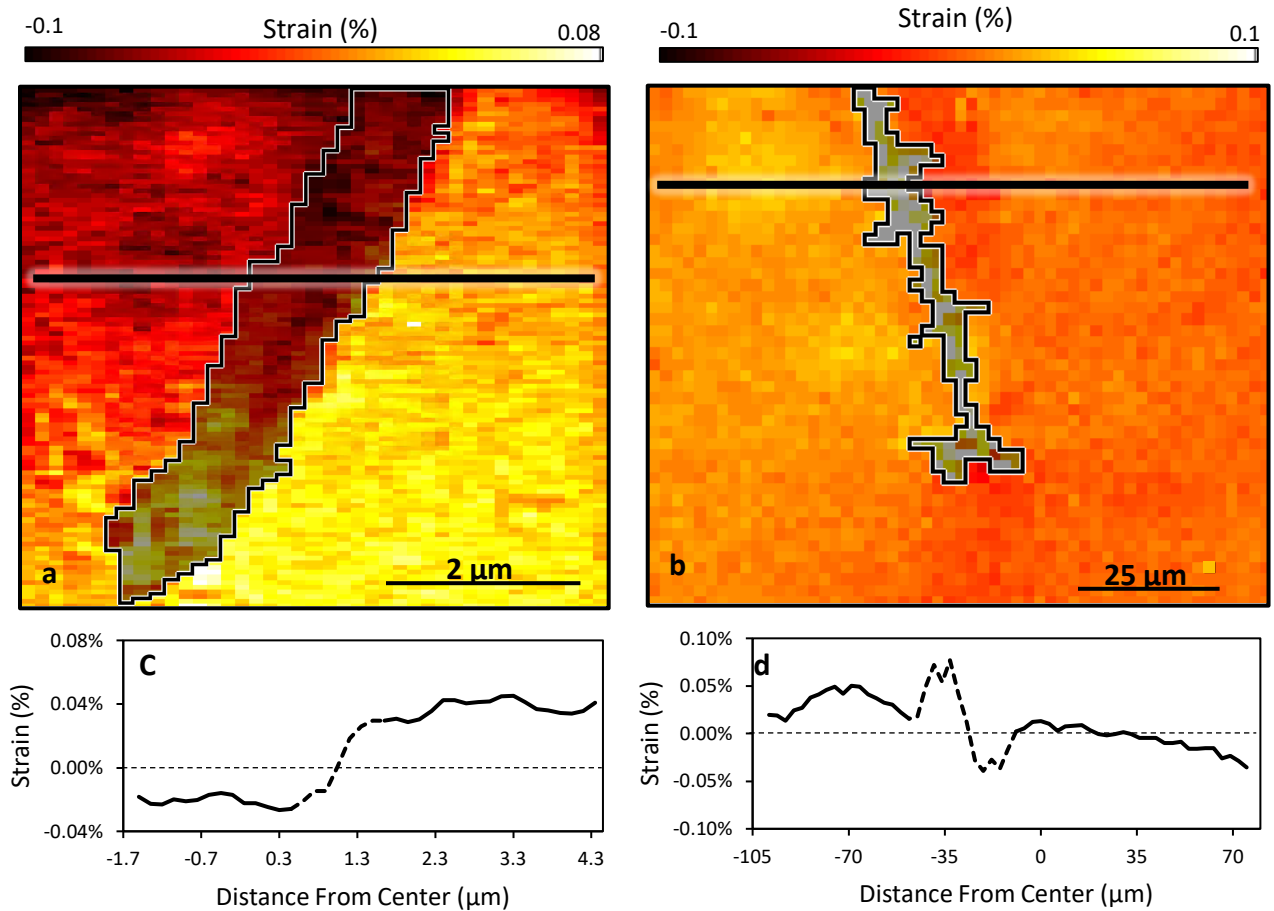
**Figure 1.** (a) Schematic of x-ray nanodiffraction experimental setup: Fresnel zone plate focuses x-rays down to a 25 nm spot size and the diffracted x-rays are collected by a two-dimensional PAD detector. An order sorting aperture and zone plate central stop are used to block the zero-order x-ray beam. Nanodiffraction images are recorded by scanning the sample. The diffraction pattern captured by the CCD detector shows both  $(002)_p$  and  $(008)_{bm}$  Bragg reflections for all LSCO/Gd heterostructures. (b) XRD patterns near the  $(002)_p$  Bragg reflection for the LSCO/Gd heterostructures and as-grown reference LSCO sample collected using unfocused x-ray beam.



**Figure 2. Nanodiffraction maps for the LSCO/Gd(3 nm) heterostructure:** a) Nanodiffraction map of the integrated intensity of the  $(002)_p$  Bragg reflection for the LSCO/Gd(3 nm) heterostructure showing phase separation into perovskite-rich regions (yellow/white) and BM-rich regions (red/brown). The inset image is a zoomed-in map of a section of the feature, showing the complex interface between the two regions. (b) Diffraction pattern at a perovskite-rich region which shows higher intensity of perovskite Bragg reflection compared to the BM Bragg reflection. (c) Nanodiffraction map of the same region, but with the integrated intensity of the  $(008)_{bm}$  Bragg reflection. The inset image is the zoomed-in map of the same region as that of inset in (a). (d) Diffraction pattern at a BM-rich region, showing an increase in intensity of the BM Bragg reflection.



**Figure 3. Comparison of LSCO/Gd heterostructures:** Nanodiffraction maps of the integrated intensity of the  $(002)_p$  Bragg reflection for LSCO/Gd heterostructures with Gd thicknesses of a) 0 nm, b) 0.5 nm, and c) 3 nm. The yellow regions represent the perovskite-rich phase and red/brown regions indicate BM-rich phase. (d) Local  $\theta$ - $2\theta$  scans from the regions indicated in panels (a), (b), and (c) which allow for the determination of the local lattice parameter for each phase as discussed in the text. The closed symbol represent measurement on the BM phase, while open symbol represents perovskite phase. A variation in *local* lattice parameter for the BM phase within a given sample is observed, as well as within the perovskite phase between samples due to the increase in oxygen vacancy concentration as the Gd thickness increases.



**Figure 4. Strain maps of LSCO/Gd(3 nm) heterostructure:** Strain maps around two BM filaments in the LSCO/Gd(3 nm) heterostructure. These strain maps show that on opposite sides of the BM filaments, the strain in the perovskite phase differs for (a) a smaller 6 x 6 μm region shown in Figure 3(c), and (b) a larger 80 x 80 μm region. The gray shading highlights the BM filaments. The strain of the residual perovskite phase present in the BM filaments is also plotted. (c) and (d) show line scans across the BM phase clearly showing the reversal from compressive to tensile strain in perovskite phase across the BM filament. The dashed lines refer the BM filaments while the solid lines refer to the perovskite phase.

## References

1. Wu, J. and C. Leighton, *Glassy ferromagnetism and magnetic phase separation in  $La_{1-x}Sr_xCoO_3$* . Physical Review B, 2003. **67**(17): p. 174408.
2. Dixon, E., J. Hadermann, and M.A. Hayward, *The synthesis and complex anion-vacancy ordered structure of  $La_{0.33}Sr_{0.67}MnO_{2.42}$* . Journal of Solid State Chemistry, 2011. **184**(7): p. 1791-1799.
3. Setter, N. and L.E. Cross, *The role of B-site cation disorder in diffuse phase transition behavior of perovskite ferroelectrics*. Journal of Applied Physics, 1980. **51**(8): p. 4356-4360.
4. Ogale, A.S., et al., *Octahedral cation site disorder effects on magnetization in double-perovskite  $Sr_2FeMoO_6$ : Monte Carlo simulation study*. Applied Physics Letters, 1999. **75**(4): p. 537-539.
5. Nagai, T., W. Ito, and T. Sakon, *Relationship between cation substitution and stability of perovskite structure in  $SrCoO_{3-\delta}$ -based mixed conductors*. Solid State Ionics, 2007. **177**(39): p. 3433-3444.
6. Caciuffo, R., et al., *Structural details and magnetic order of  $La_{1-x}Sr_xCoO_3$  ( $x \leq 0.3$ )*. Physical Review B, 1999. **59**(2): p. 1068-1078.
7. Ferguson, J.D., et al., *Epitaxial oxygen getter for a brownmillerite phase transformation in manganite films*. Adv Mater, 2011. **23**(10): p. 1226-30.
8. Jeen, H., et al., *Topotactic phase transformation of the brownmillerite  $SrCoO_{2.5}$  to the perovskite  $SrCoO_{3-\delta}$* . Adv Mater, 2013. **25**(27): p. 3651-6.
9. Krick, A.L. and S.J. May, *Evidence for oxygen vacancy manipulation in  $La_{1/3}Sr_{2/3}FeO_{3-\delta}$  thin films via voltage controlled solid-state ionic gating*. APL Materials, 2017. **5**(4).
10. Kalabukhov, A., et al., *Effect of oxygen vacancies in the  $SrTiO_3$  substrate on the electrical properties of the  $LaAlO_3/SrTiO_3$  interface*. Physical Review B, 2007. **75**(12).
11. Dass, R.I., J.Q. Yan, and J.B. Goodenough, *Oxygen stoichiometry, ferromagnetism, and transport properties of  $La_{2-x}NiMnO_{6+\delta}$* . Physical Review B, 2003. **68**(6).
12. Biegalski, M.D., et al., *In situ examination of oxygen non-stoichiometry in  $La_{0.80}Sr_{0.20}CoO_{3-\delta}$  thin films at intermediate and low temperatures by x-ray diffraction*. Applied Physics Letters, 2014. **104**(16): p. 161910.
13. Gazquez, J., et al., *Lattice mismatch accommodation via oxygen vacancy ordering in epitaxial  $La_{0.5}Sr_{0.5}CoO_{3-\delta}$  thin films*. APL Materials, 2013. **1**(1): p. 012105.
14. Moon, E.J., et al., *Fluorination of Epitaxial Oxides: Synthesis of Perovskite Oxyfluoride Thin Films*. Journal of the American Chemical Society, 2014. **136**(6): p. 2224-2227.
15. Moon, E.J., et al., *Comparison of topotactic fluorination methods for complex oxide films*. APL Materials, 2015. **3**(6): p. 062511.
16. Torrance, J., et al., *Systematic study of insulator-metal transitions in perovskites  $RNiO_3$  ( $R=Pr, Nd, Sm, Eu$ ) due to closing of charge-transfer gap*. Physical Review B, 1992. **45**(14): p. 8209-8212.
17. Lu, N., et al., *Electric-field control of tri-state phase transformation with a selective dual-ion switch*. Nature, 2017. **546**(7656): p. 124-128.
18. Moritomo, Y., et al., *Magnetic and electronic properties in hole-doped manganese oxides with layered structures:  $La_{1-x}Sr_xMnO_4$* . Physical Review B, 1995. **51**(5): p. 3297-3300.

19. Fuchs, D., et al., *Tuning the magnetic properties of LaCoO<sub>3</sub> thin films by epitaxial strain*. Physical Review B, 2008. **77**(1): p. 014434.
20. Choi, W.S., et al., *Strain-Induced Spin States in Atomically Ordered Cobaltites*. Nano Letters, 2012. **12**(9): p. 4966-4970.
21. Lu, Q. and B. Yildiz, *Voltage-Controlled Topotactic Phase Transition in Thin-Film SrCoO<sub>x</sub> Monitored by In Situ X-ray Diffraction*. Nano Lett, 2016. **16**(2): p. 1186-93.
22. Cui, B., et al., *Direct imaging of structural changes induced by ionic liquid gating leading to engineered three-dimensional meso-structures*. Nat Commun, 2018. **9**(1): p. 3055.
23. Jeon, H., et al., *Reversible redox reactions in an epitaxially stabilized SrCoO<sub>x</sub> oxygen sponge*. Nature Materials, 2013. **12**: p. 1057-1063.
24. Lu, N., et al., *Electric-field control of tri-state phase transformation with a selective dual-ion switch*. Nature, 2017. **546**: p. 124-128.
25. Kim, Y.M., et al., *Probing oxygen vacancy concentration and homogeneity in solid-oxide fuel-cell cathode materials on the subunit-cell level*. Nat Mater, 2012. **11**(10): p. 888-94.
26. Yang, J.J., et al., *Memristive switching mechanism for metal/oxide/metal nanodevices*. Nat Nanotechnol, 2008. **3**(7): p. 429-33.
27. Gilbert, D.A., et al., *Ionic tuning of cobaltites at the nanoscale*. Physical Review Materials, 2018. **2**(10).
28. Grutter, A.J., et al., *Reversible control of magnetism in La<sub>0.67</sub>Sr<sub>0.33</sub>MnO<sub>3</sub> through chemically-induced oxygen migration*. Appl. Phys. Lett., 2016. **108**: p. 082405.
29. Gilbert, D.A., et al., *Controllable positive exchange bias via redox-driven oxygen migration*. Nat Commun, 2016. **7**: p. 11050.
30. Yao, L., et al., *Electron-Beam-Induced Perovskite-Brownmillerite-Perovskite Structural Phase Transitions in Epitaxial La<sub>2/3</sub>Sr<sub>1/3</sub>MnO<sub>3</sub> Films*. Advanced Materials, 2014. **26**: p. 2789-2793.
31. Srot, V., et al., *Influence of TEM specimen preparation on chemical composition of Pb(Mg<sub>1/3</sub>Nb<sub>2/3</sub>)O<sub>3</sub>-PbTiO<sub>3</sub> single crystals*. Micron, 2014. **62**: p. 37-42.
32. Eberg, E., et al., *Comparison of TEM specimen preparation of perovskite thin films by tripod polishing and conventional ion milling*. J Electron Microsc (Tokyo), 2008. **57**(6): p. 175-9.
33. Choi, W.S., et al., *Strain-Induced Spin States in Atomically Ordered Cobaltites*. Nano Letters, 2012. **12**: p. 4966-4970.
34. Sharma, M., et al., *Coercivity enhancement driven by interfacial magnetic phase separation in SrTiO<sub>3</sub>(001)/Nd<sub>0.5</sub>Sr<sub>0.5</sub>CoO<sub>3</sub>*. Phys. Rev. B, 2011. **84**: p. 024417.
35. Gazquez, J., et al., *Lattice mismatch accommodation via oxygen vacancy ordering in epitaxial La<sub>0.5</sub>Sr<sub>0.5</sub>CoO<sub>3-δ</sub> thin films*. APL Materials, 2013. **1**: p. 012105.
36. Walter, J., et al., *Perpendicular magnetic anisotropy via strain-engineered oxygen vacancy ordering in epitaxial La<sub>1-x</sub>Sr<sub>x</sub>CoO<sub>3-δ</sub>*. Physical Review Materials, 2018. **2**(11): p. 111404.
37. Hruszkewycz, S.O., et al., *Coherent Bragg nanodiffraction at the hard X-ray Nanoprobe beamline*. Philos Trans A Math Phys Eng Sci, 2014. **372**(2010): p. 20130118.
38. Posadas, A.B., et al., *Scavenging of oxygen from SrTiO<sub>3</sub> during oxide thin film deposition and the formation of interfacial 2DEGs*. Journal of Applied Physics, 2017. **121**(10).

39. Parsons, T.G., et al., *Synthesis and Structural Characterization of  $La_{1-x}A_xMnO_{2.5}$  ( $A = Ba, Sr, Ca$ ) Phases: Mapping the Variants of the Brownmillerite Structure*. *Chemistry of Materials*, 2009. **21**(22): p. 5527-5538.