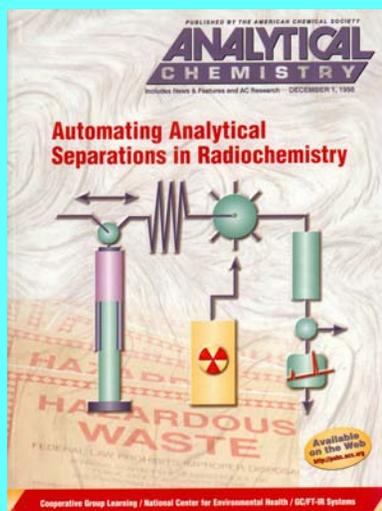




# Project #42370. Radioanalytical Chemistry for Automated Nuclear Waste Process Monitoring

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# Radioanalytical Chemistry for Automated Nuclear Waste Process Monitoring

- n Research program directed toward rapid, sensitive and selective determination of  $\beta$ - and  $\alpha$ - emitters in LAW processing streams:
  - specific targets  $^{99}\text{Tc}$  (previous focus),  $^{90}\text{Sr}$ ,  $^{137}\text{Cs}$  and TRU
- n Automated radiochemical measurements via integration of A) sample preparation/treatment, B) rapid selective separation, and C) on-line radiometric detection steps within a single functional analytical instrument
- n Significant knowledge gaps exist relative to designing chemistries for such instruments so that analytes can be quantitatively and rapidly separated and analyzed in challenging matrixes
- n Specific focus for the renewal period:  $^{90}\text{Sr}$ , TRU-gross alpha,  $^{137}\text{Cs}$



# Project Objectives

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- n Introduce automated radiochemistry to the field of process monitoring of LAW matrices
- n Develop fundamental understanding of the chemical processes that must occur within the analyzer instrument (i.e., matrix modification/speciation control and selective separation chemistries)
- n Investigate matrix modification chemistries for chemically complex, variable matrices
- n Investigate separation chemistries suitable for rapid selective separation of target analytes
- n Investigate a new flow-through radiometric detector concept based on solid state diode detectors for quantification of TRU in process solutions
- n Investigate new data acquisition and signal processing techniques to enhance performance of scintillation and diode detectors for use in process monitoring
- n Characterize candidate chemistries and instrument configurations with regard to detection limits, sensitivity, selectivity, analyses time
- n Define the limitations and the potential of the proposed monitoring approach. New knowledge and expertise will provide information for designing effective instruments
- n Educate students in modern radiochemistry and DOE cleanup challenges

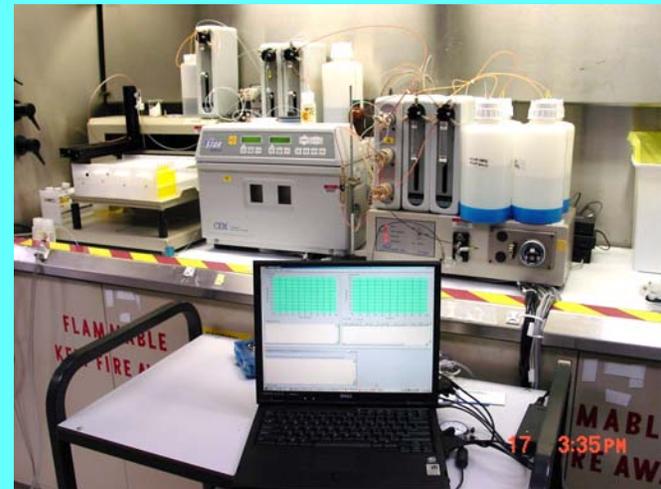
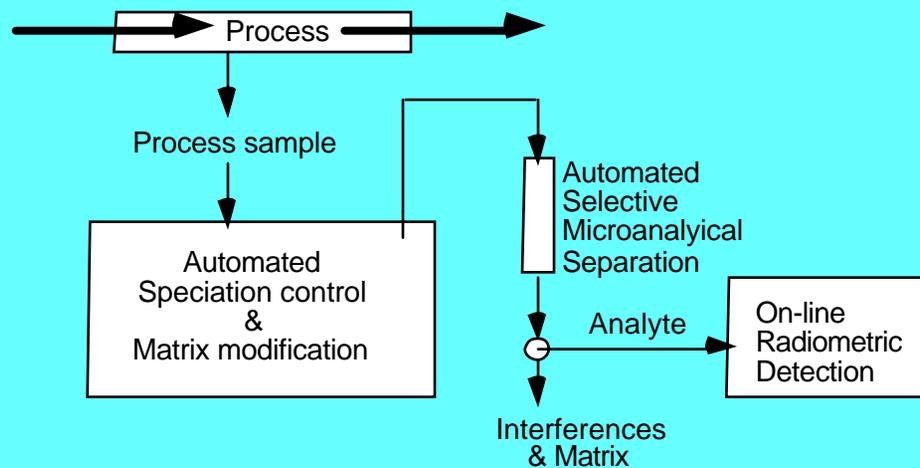


# Process Monitoring Challenges and Reality

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- n Generic Challenges:
  - Complex, high-ionic strength, caustic brine sample matrixes
  - Undefined or varying chemical speciation
  - Low mass concentrations, need for isotopic information
  - Radioactive and stable matrix interferences
  - Detection limits, robustness, sample turnaround time
- n Challenges/requirements limit the choice of analytical approaches
- n Reality: "Radiochemistry has always been and still is a crucial tool in the field of radionuclide determination, ...particularly in the case of alpha and beta emitters".
- n Current baseline: costly and lengthy manual laboratory measurements

# Automated Radiochemistry: Approach



- n Automated radiochemical measurements via integration of the following steps within a single functional instrument:
- sample preparation/treatment,
  - rapid selective separation,
  - on-line radiometric detection



# Automated Radiochemistry: Research Focus

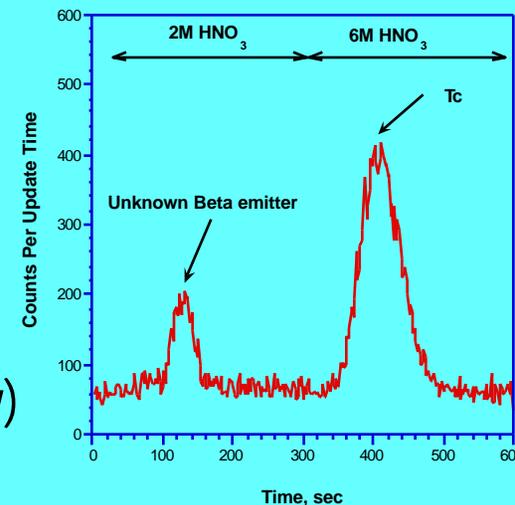
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- n Sample modification/ speciation control chemistries for  $^{90}\text{Sr}$ , TRU,  $^{137}\text{Cs}$  (acidification, digestion, oxidation, speciation control)
- n Separation chemistries for  $^{99}\text{Tc}$  and  $^{90}\text{Sr}$ ,  $^{137}\text{Cs}$  and TRU as necessary
  - sorbent material selection and characterization (uptake selectivity, kinetics, elution chemistries, stability etc.
  - development of rapid, automated separation schemes
- n On-line radiometric detection
  - solid state diode detection for in-situ alpha detection/spectroscopy
  - thin film scintillation detection for in-situ alpha detection in corrosive matrices
  - light detection, radiation detector design and signal processing for beta scintillation detection (Clemson)
- n Analytical and radiochemistry of the LAW waste matrices

# Automated Radiochemistry Challenges

Example: Tank AN-102 -Radiochemical interference issue in the automated Tc analysis

Inventory in	uCi/mL
Gross alpha:	0.18
Gross beta:	590
Tc-99	8.23(ppm)
(at least 60% as non-pertechnetate)	
Cs-137	335
Cs-135	3.02
Sr-90	84
Co-60	0.12
Pu-239/240	0.01
Am-241	0.15
Unknown Activity	158



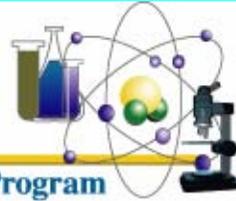
$^{90}\text{Sr}$ -high abundance of  $^{137}\text{Cs}/^{137\text{m}}\text{Ba}$  (separation chemistry)  
TRU- corrosive matrix, high  $\beta/\gamma$  backgrounds

- Knowledge gaps exist and are being filled as an outcome of this research

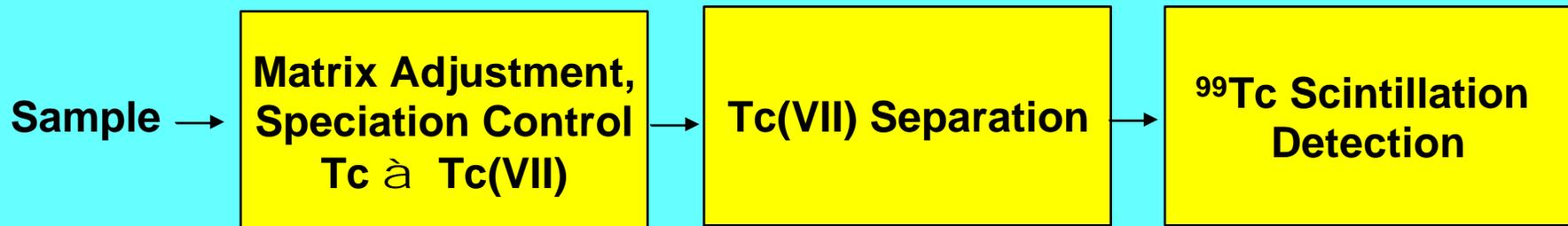


# EMSP

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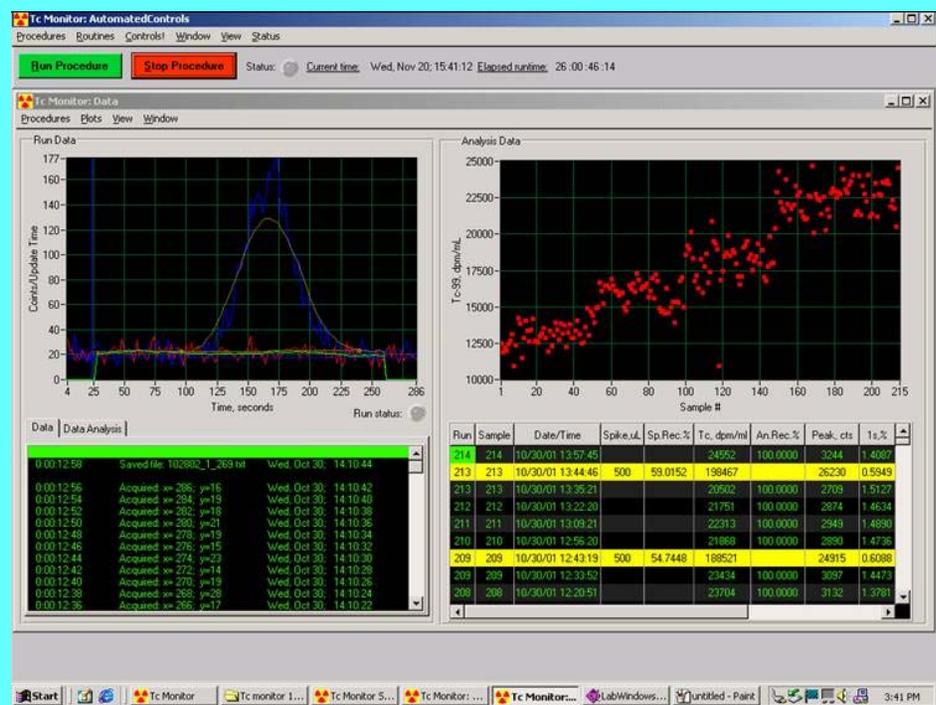
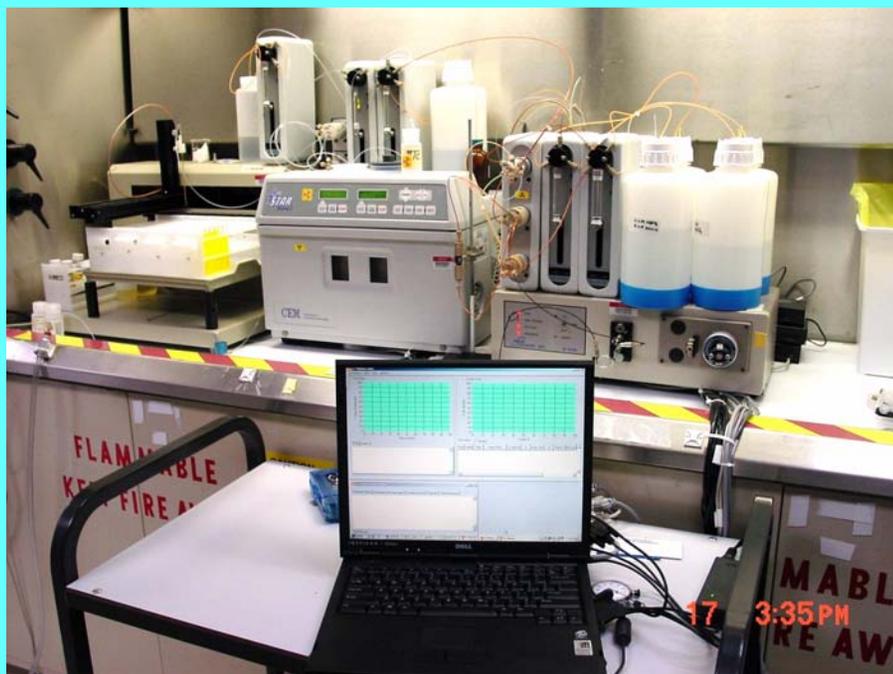
## Automated Radiochemistry Example: Tc Process Monitoring



- **Flow-through scintillation detection of the beta-emitting <sup>99</sup>Tc**  
(radiometric detection is sensitive to meet DL with short counting times)
  - Detection using glass scintillator flow cell
- **Automated, rapid radiochemical separation of the <sup>99</sup>Tc(VII) from sample matrix and radioactive interferences (e.g. <sup>90</sup>Sr/<sup>90</sup>Y, <sup>137</sup>Cs, etc.)**  
(separation is required to measure <sup>99</sup>Tc in the presence of other radionuclides)
  - Small column separation using anion exchange sorbent
- **Automated, rapid sample treatment/oxidation to control Tc speciation**  
(oxidation is required to convert all <sup>99</sup>Tc to <sup>99</sup>Tc(VII) for total <sup>99</sup>Tc analysis)
  - Sample acidification and microwave assisted oxidation using persulfate
- **Automated, digital fluid handling for sample/solution delivery and system integration**
  - Computer-controlled valves and syringe pumps



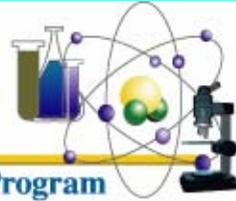
# Prototype Instrument Built on EMSP Science: <sup>99</sup>Tc Analyzer for WTP



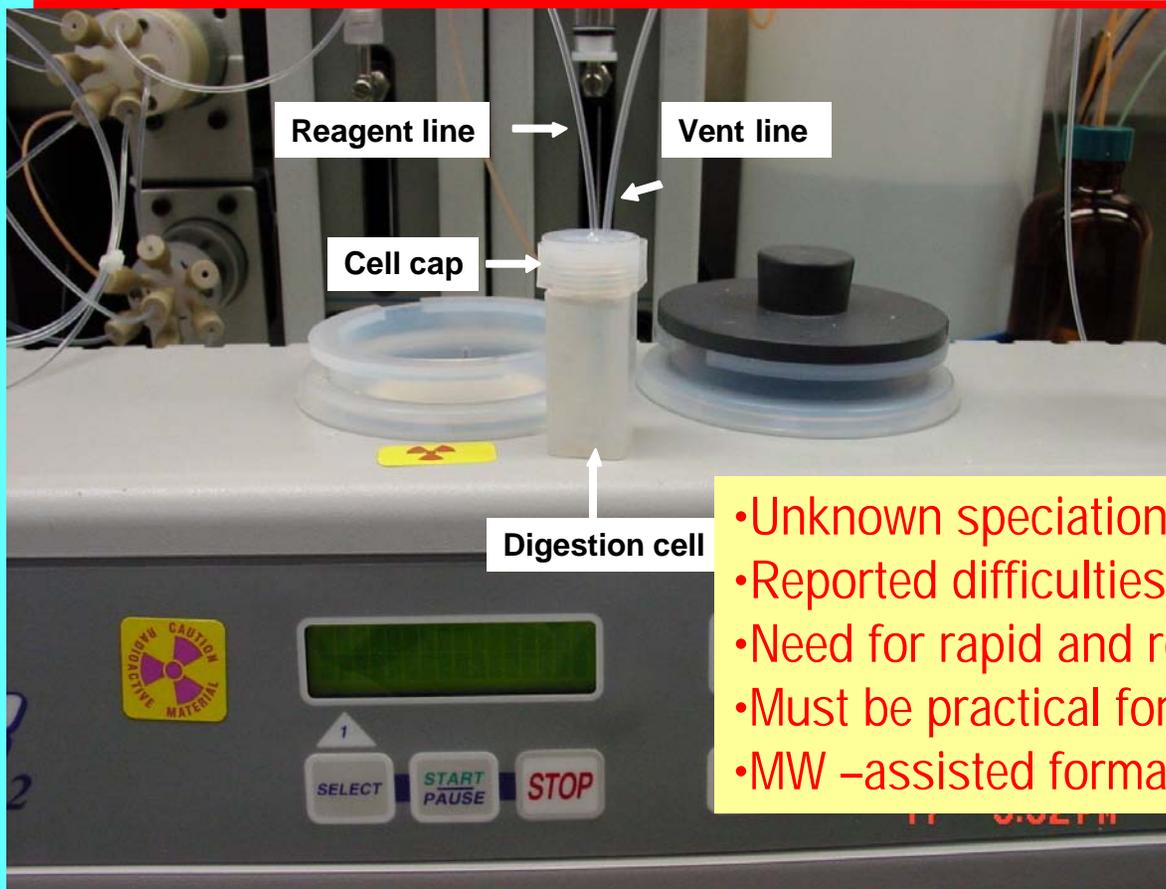


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## Microwave Assisted Sample Oxidation/Digestion for Automated Measurements

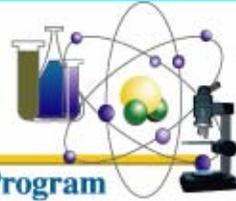


- Unknown speciation
- Reported difficulties for converting Tc to Tc(VII)
- Need for rapid and reliable methodology
- Must be practical for use in plant settings
- MW –assisted format using acidification and oxidation



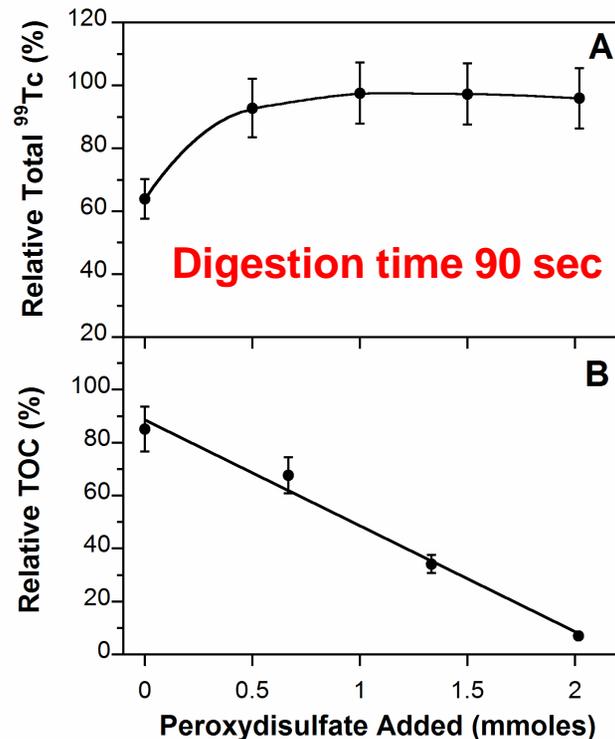
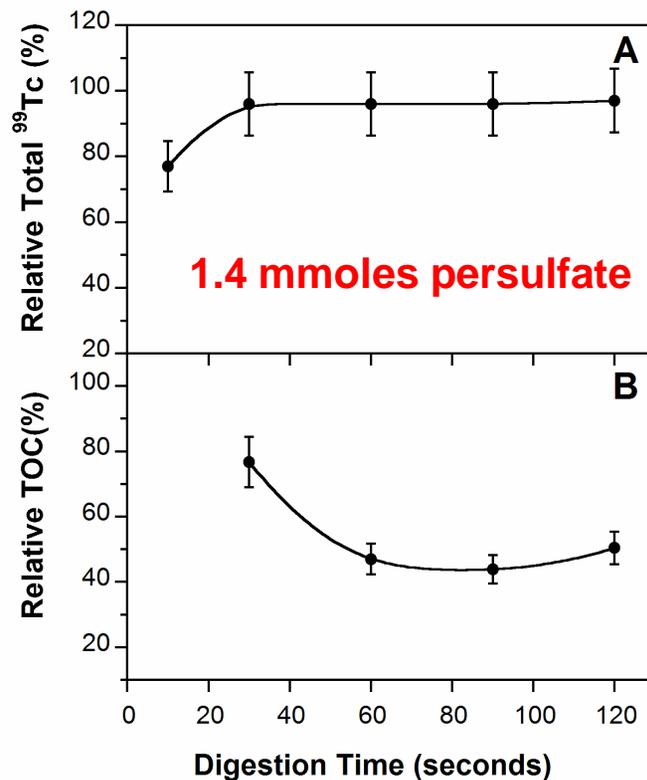
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## Microwave-Assisted Oxidation of Non-Per technetate

Envelope C AN-107 waste: ~ 60% of non-per technetate  
Chemical speciation is unknown

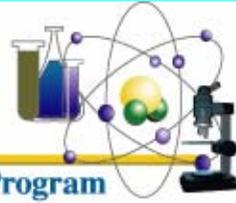


• **Rapid, reliable Tc oxidation is possible using MW-assisted  $\text{Na}_2\text{S}_2\text{O}_8$  treatment**



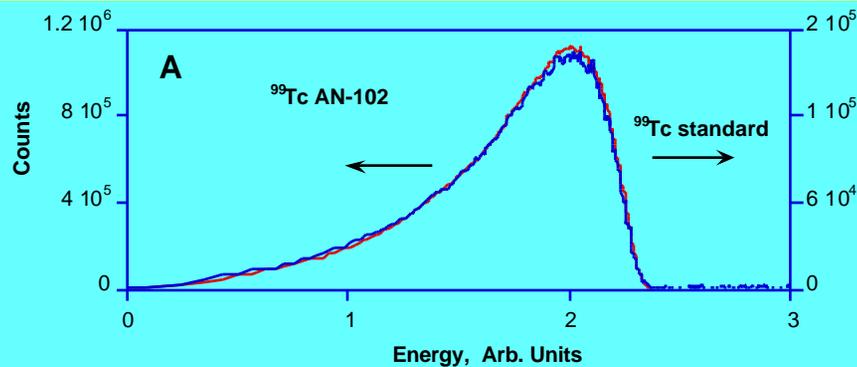
# EMSP

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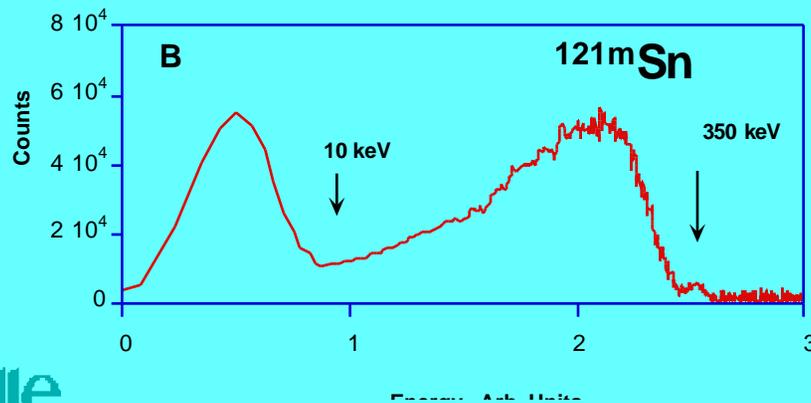
## EMSP Program Enabled Identification of Interferent in the Tc Measurement

- liquid scintillation spectrometry of the separated Tc fractions indicated good radiochemical purity for the analysis of radiochemically complex samples



**Additional verification:**

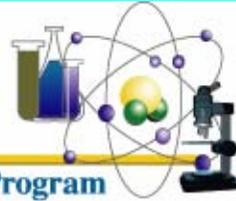
LEPS  
Gamma  
ICP MS



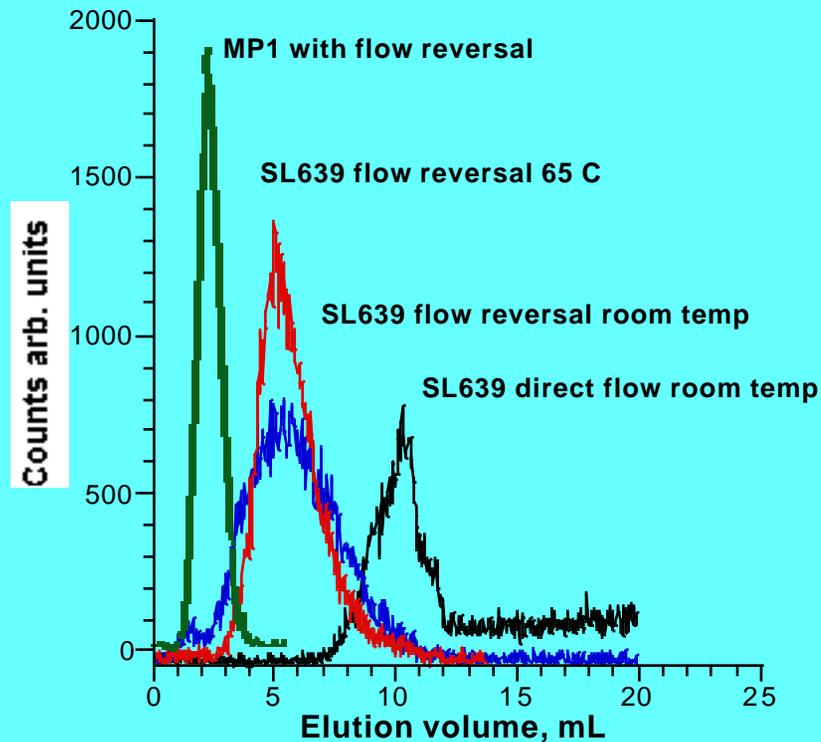
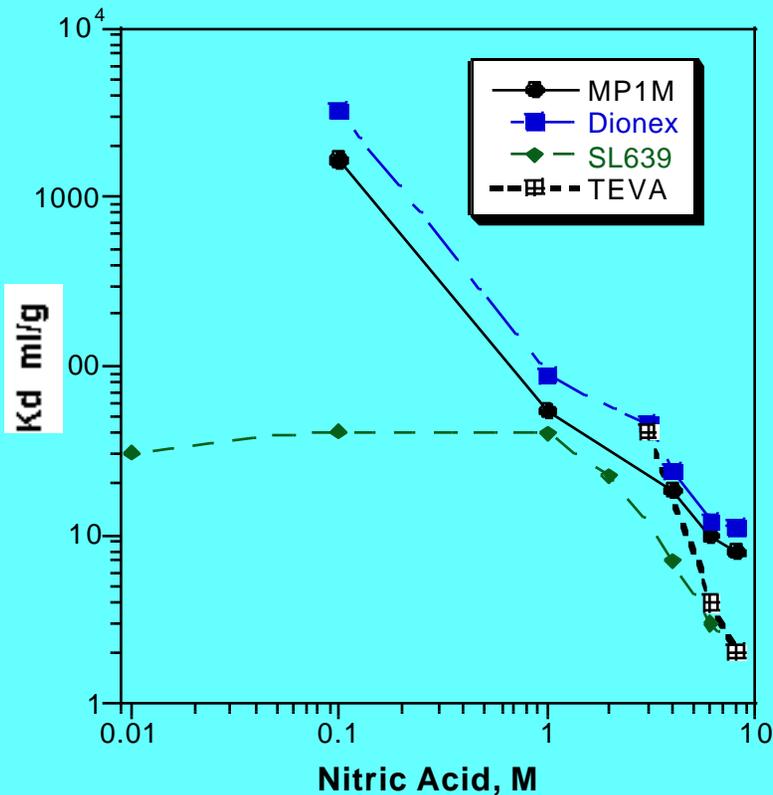


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## Tc(VII) Separation Chemistries



- **Strongly basic AnIX is a preferred material for Tc separation**



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## Separation of $^{99}\text{Tc}$ from Interferences in LAW

- Sequence of column scrub steps is required to remove interferences

Column: MP-1 resin 4.6 x 50 mm (0.83 mL bed volume)

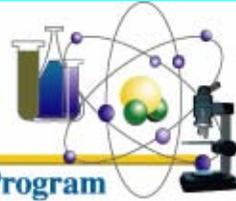
Step/Description	Reagent	Volume	Flow Rate
1. Column Conditioning	0.2M $\text{HNO}_3$	2 mL	7.5 mL/min
2. Sample load	Digested sample	1.6 mL	5 mL/min
3. Column scrub I	0.2 M $\text{HNO}_3$	5 mL	7.5 mL/min
4. Column scrub II	1 M $\text{NaOH}$	6 mL	7.5 mL/min
5. Column scrub III	0.2 M $\text{HNO}_3$ -0.5 M $\text{H}_2\text{C}_2\text{O}_4$	6 mL	7.5 mL/min
6. Column scrub IV	2 M $\text{HNO}_3$	5 mL	7.5 mL/min
7. Pertechnetate elution <sup>a</sup>	8 M $\text{HNO}_3$	5.6 mL	1.5 mL/min

a Flow direction through the column is reversed during elution step.

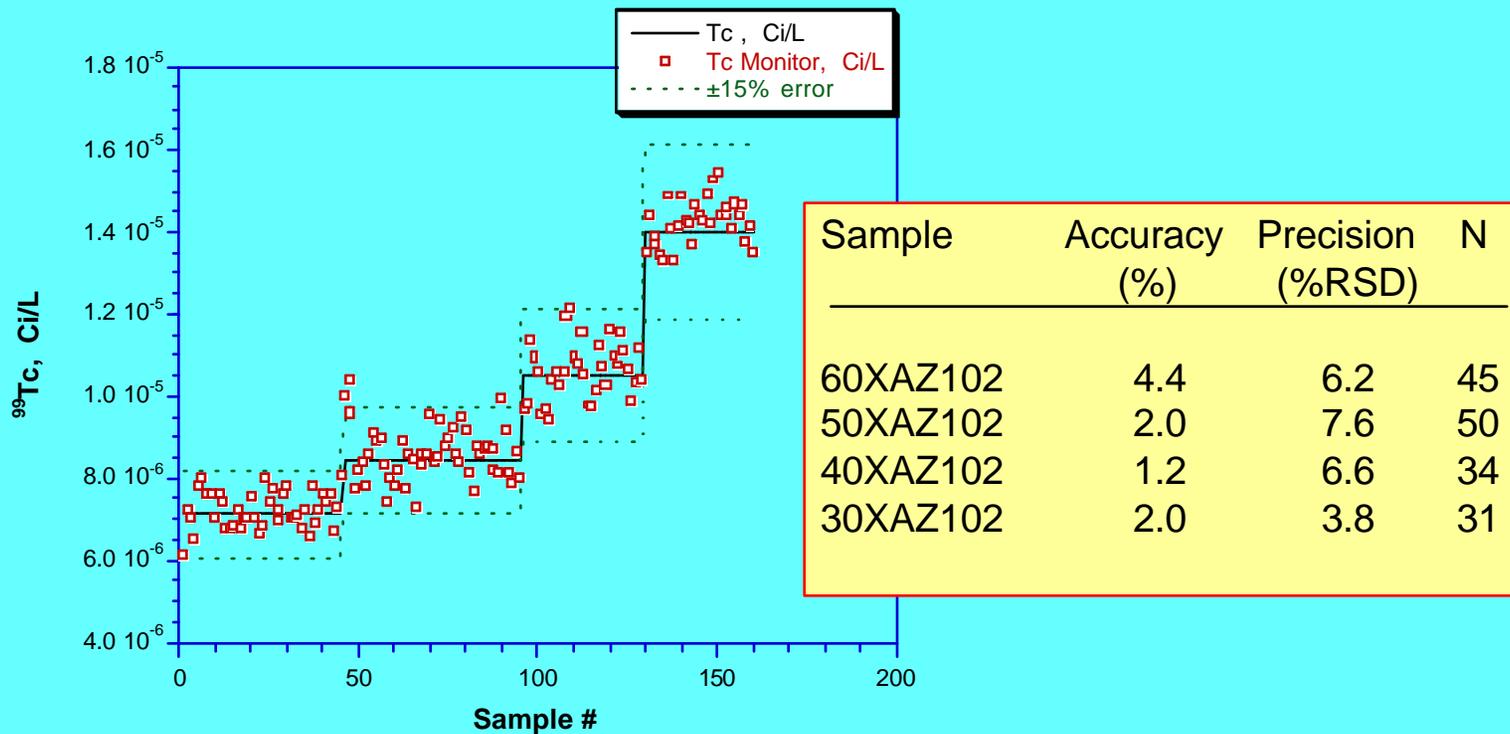


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## Tc Process Monitor Instrument Testing



**• Instrument selected for Hanford WTP in peer reviewed, competitive selection process!!!**

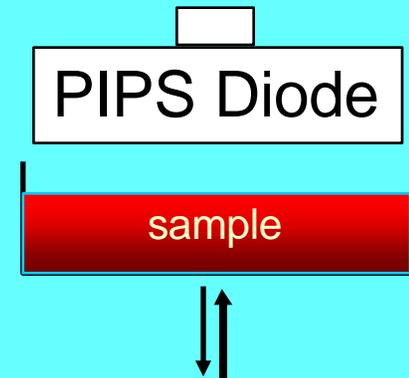


U.S. Department of Energy  
Pacific Northwest National Laboratory

# Direct Detection of Alpha Emitters Using PIPS Diodes

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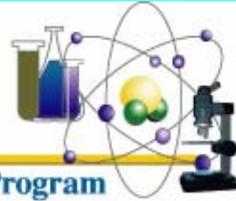
- n Low noise and good resolution
- n Operation in ambient light
- n Passivated surface w/o metal contacts
  - Rugged
  - Chemically resistant
  - Surface can be touched, cleaned--modified





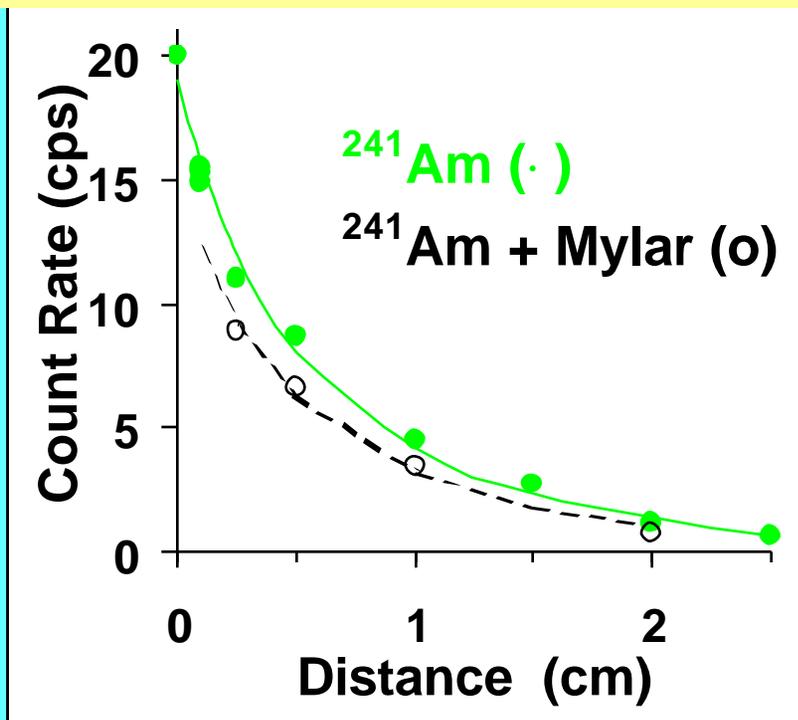
# EMSP

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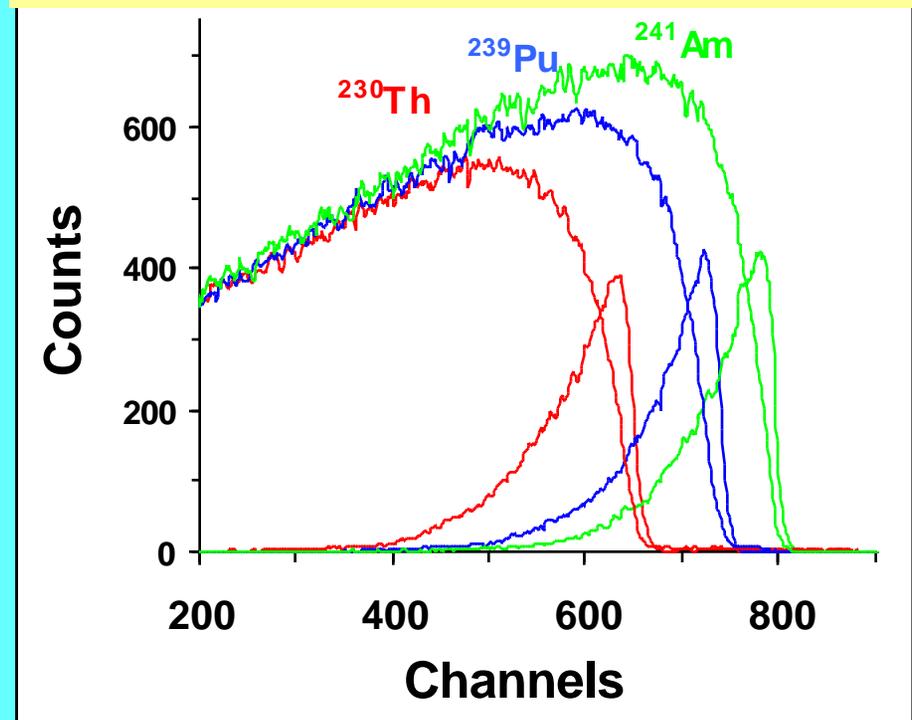
## Direct Analysis of Alpha Emitters In Solution

### Activity vs. Distance\*



\* Distance between solution surface and detector face. 125 nCi/ml  $^{241}\text{Am}$  activity.

### Alpha Energy "Spectra"†



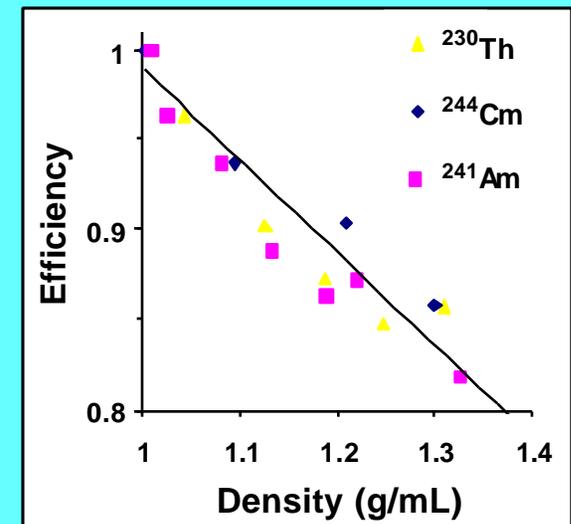
† 125 nCi/ml, 10 hr counts, 0.5 cm standoff from surface.

# Analysis of Actual Samples

Sample Matrix	Density (g/mL)	PIPS Assay <sup>a</sup> (Bq/ml)	Lab Assay <sup>b</sup> (Bq/ml)	Difference (%)
Water <sup>c</sup>	1	4.6	4.0	-13.9
Water <sup>c</sup>	1	392.7	403.0	2.6
Hanford Tank AZ-10 <sup>f,d</sup>	1.22	17.1	15.7	-9.2
Hanford Tank AN-10 <sup>f</sup>	1.249	70.4	82.1	14.2
Hanford Tank AZ-10 <sup>f</sup>	1.260	120.0	154.9	22.5
Savanna River Tank 2 <sup>3c,d</sup>	1.031	817.6	745.0	-9.7
Savanna River Tank 3 <sup>d</sup>	1.171	6.6	8.6	22.7
Savanna River Tank 4 <sup>d</sup>	1.204	224.8	205.5	-9.4

- a) Based upon <sup>241</sup>Am, 10 min count, efficiency corrected
- b) Precipitation plating and AES
- c) Spiked with <sup>241</sup>Am
- d) supernate

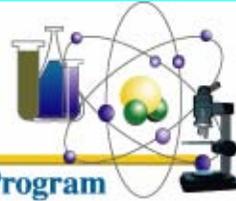
- n Rapid total alpha assay
- n Limited sensitivity ~5 Bq/mL
- n Spectral information....



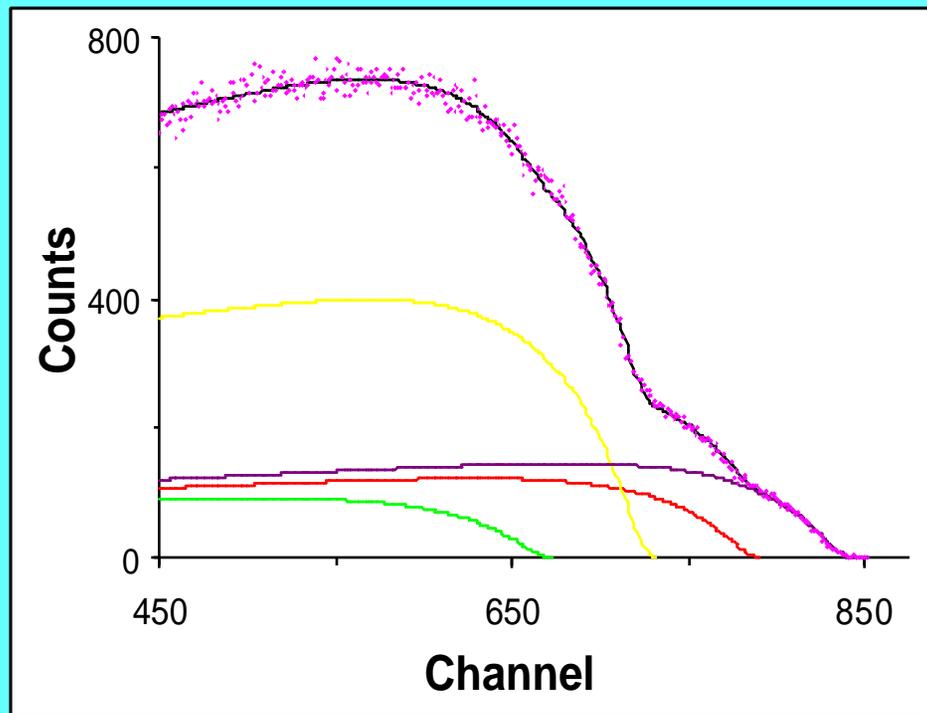


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## Direct AEA in Liquids Using Diodes: Spectral Deconvolution



20%  $^{244}\text{Cm}$ , 15%  $^{241}\text{Am}$ , 50%  $^{239}\text{Pu}$ , 15%  $^{233}\text{U}$   
Total solution activity 125 nCi/mL

Isotope	Solution Activity <sup>a</sup>	Calculated Activity <sup>b</sup>
$^{244}\text{Cm}$	<b>25.0</b>	<b>24.8</b>
$^{241}\text{Am}$	<b>18.8</b>	<b>20.2</b>
$^{239}\text{Pu}$	<b>62.5</b>	<b>60.9</b>
$^{233}\text{U}$	<b>18.8</b>	<b>19.3</b>

a) nCi/mL, mixture of standards

b) nCi/mL, spectral deconvolution,  
least squares optimization



# Relevance

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- n  $^{90}\text{Sr}$ ,  $^{99}\text{Tc}$  and TRU process measurements will be required at WTP at Hanford and Savannah River sites
  - Better process control, plant throughput and regulatory compliance
- n Lack of analytical methodologies for these radionuclides
- n Baseline is not significantly different from PUREX plant days
- n Plants are being constructed: this research is timely
- n This analytical methodology has dual use:
  - on-line/at-line measurements
  - automated tool for the analytical laboratory
- n Potential for significant cost savings over operational cycle of the waste treatment plants



# Implementation/Relevance

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## n Hanford WTP:

- On-line Tc monitoring is required for the operation of the Tc removal columns (if process is implemented)
- Our technology was selected for the Hanford WTP
- Sr and TRU monitoring: automated laboratory measurements on grab samples Salt processing facility at SRS
- Need for rapid Sr and TRU analysis in the lab to support process control
- Indication that such technology may be a preferred method for this need
- Goal is to pursue prototype instrumentation development with SRS using new scientific knowledge developed under this EMSP program



## Conclusions

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- n EMSP support enabled development of the scientific knowledge lacking for the development of process analyzers based on automated radiochemistry
- n Tc monitor instrumentation/ approach developed and characterized in detail and was selected for use at the Hanford WTP
- n Work with WTP would have not been possible without the science and knowledge base developed under EMSP support
- n The ongoing research is directed at  $^{90}\text{Sr}$ , TRU and  $^{137}\text{Cs}$  monitoring
- n Invaluable opportunity for young investigators to develop scientific career while working on relevant problems