

$$\frac{\mu_n}{\mu_e} = 1.04066884 (26) \times 10^{-3} \quad (0.25 \text{ ppm}) \quad (4)$$

$$\frac{\mu_n}{\mu_p} = -0.68497935 (17) \quad (0.25 \text{ ppm})$$

Our result can also be expressed in terms of the nuclear magneton  $\mu_N$ . In this case, however, the accuracy is slightly degraded by the uncertainty in the electron to proton mass ratio. Using the value of  $m/M$  obtained by combining results from Philips et al [17] and Cohen and Taylor [18] we have

$$\frac{\mu_n}{\mu_N} = -1.91304184 (88) \quad (0.45 \text{ ppm}) \quad (5)$$

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