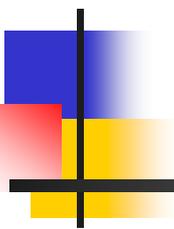


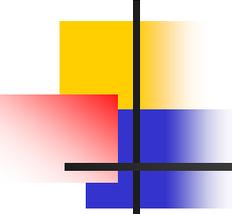
# *Research on Fuels & Lubricants*



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Joseph M. Perez,  
Tribology Group, Chemical Engineering Dept.,  
Penn State University, University Park, PA 16802

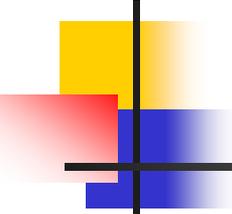
Diesel Engine Emissions Reduction Conference  
Newport, RI  
August 24-28, 2003



## Penn State's Slippery Bunch:

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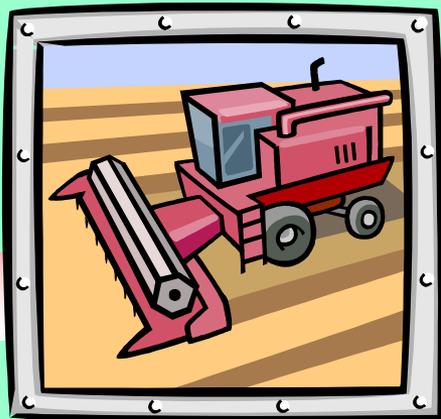
- 1950's Dew Line Lubricants, New Base Oil & Additive Technology
- 1960's SR 71 Blackbird Hydraulic Fluids, Super Refined Lubricants (Type II)
- 1970's Oxidation, Greases, Metals
- 1980's VPO, Adiabatic Engine, MeOH Oils
- 1990's Environmentally Friendly Fluids  
Extended Drain Oils



## *Current Projects*

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- Fuel Studies
  - DME
  - Biodiesel
  - ULSF
- Vegetable Oils
- High Temperature Liquid Lubricants
- Coatings & Lubricants
- Role of Chemical Structure



## Penn State "Green" Project

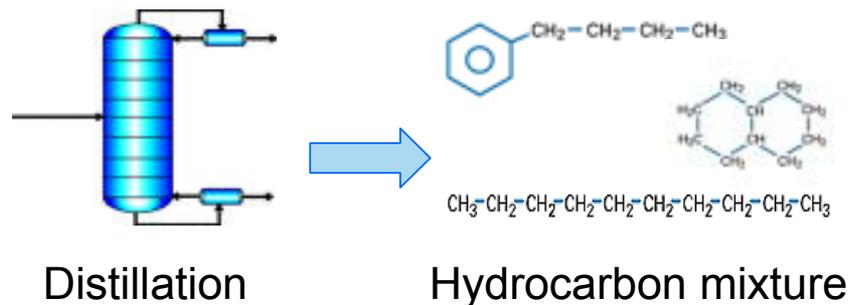
---

1. Over 200 pieces of farm & construction equipment on campus.
2. Conversion to Environmentally Friendly Lubricants initiated.
3. Use of Biodiesel in farm equipment.
4. Conversion of waste oils to Biodiesel – Undergraduate Engineering Project

# FUELS

## Diesel Fuels

Petroleum cut boiling ~ 282-338°C, #2, LSDF and ULSD  
 300 ppm S  
 32 ppm S  
 ULSD (< 15ppm S)



Soybeans

+ ROH  $\xrightarrow{\text{catalyst}}$



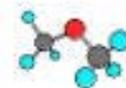
## Biodiesel Fuels

Blends of methyl esters made from vegetable oils

## Dimethyl Ether

Converted from Syngas

Hydrocarbon  $\longrightarrow$  Syngas  $\longrightarrow$  DME



# DME Research

## DME is environmentally benign

Decomposes rapidly

Doesn't harm ozone layer



DME Methane + H<sub>2</sub> + CO

## Reduces diesel engine emissions

Addition of oxygen into combustion zone

## Engine and Vehicle Tests

Problems include low viscosity (wear), high vapor pressure, and material compatibility

➔ Laboratory Tests

Viscosity Studies

Injector Studies

O-Ring Studies

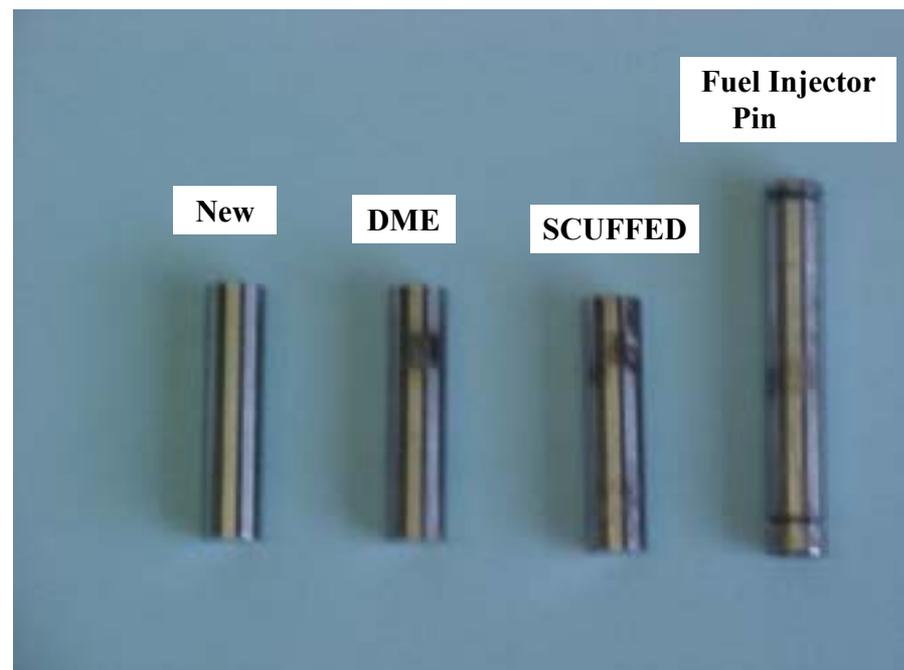


# Fuel Injector Studies



Modified Cameron –  
Plint Machine

## Test Pins



# Biodiesel Fuel Studies

Previous work involved study of VPO of diesel and biodiesel fuels in pilot plant <sup>(10)</sup>

Demonstrated in laboratory tests that addition of oxygen to biodiesel resulted in improvement in friction

Run #1

Temp- 325°C

Feed Rate- 1000 g/hr

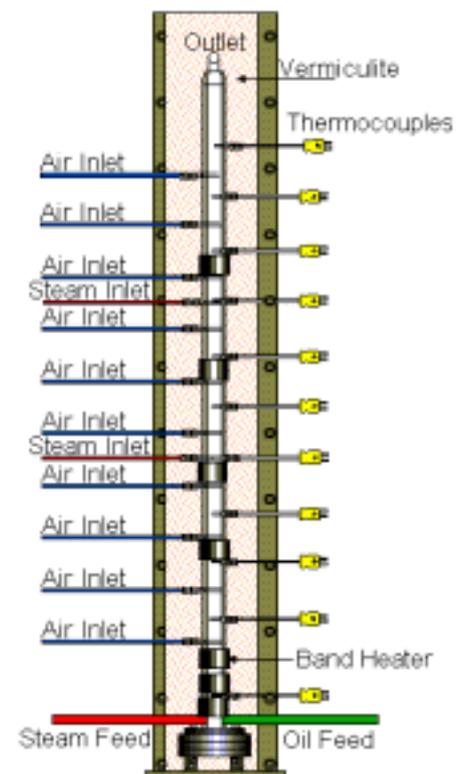
O<sub>2</sub>/Feed Mole Ratio- 1.0

Run #2

Temp- 375°C

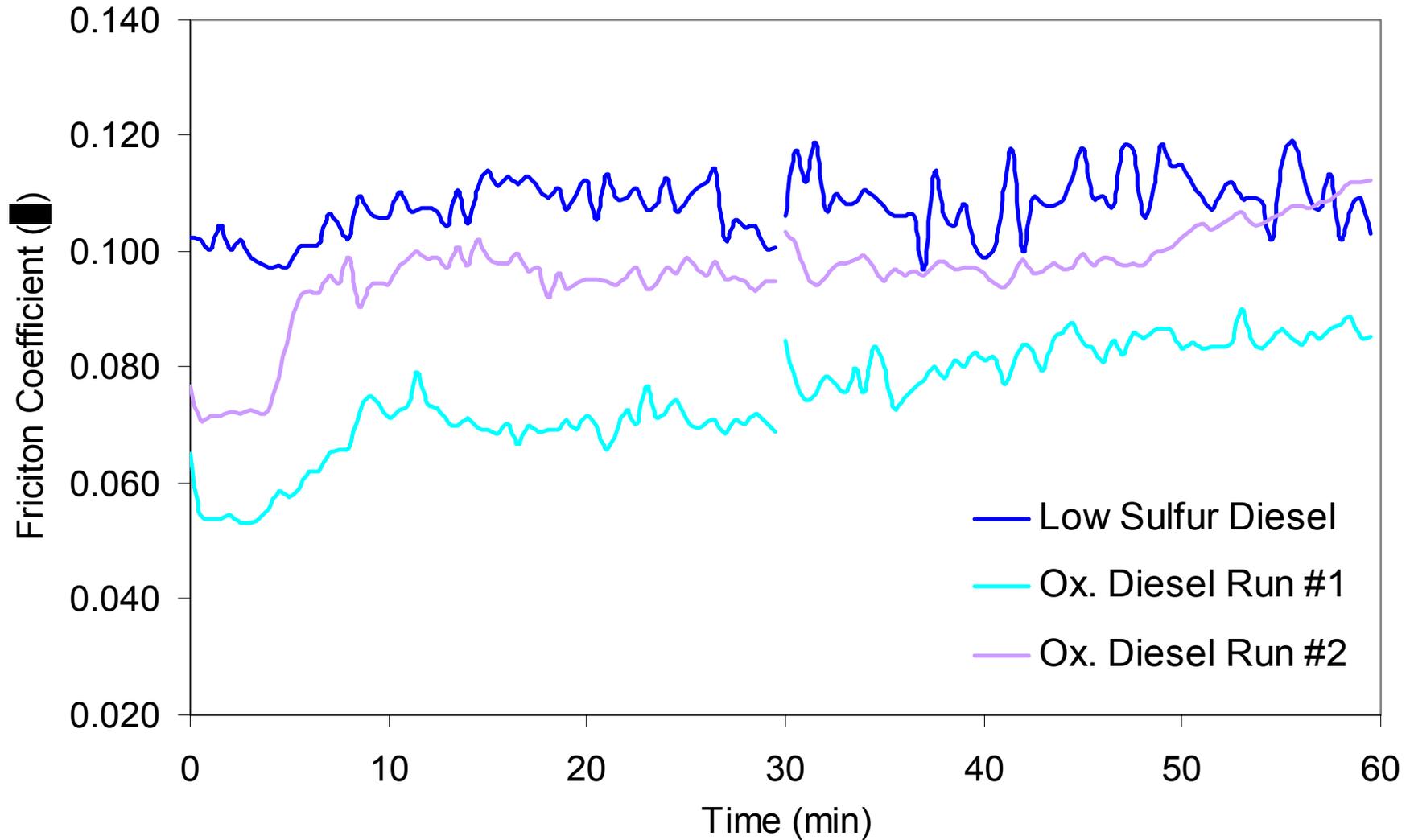
Feed Rate- 1000 g/hr

O<sub>2</sub>/Feed Mole Ratio- 1.0

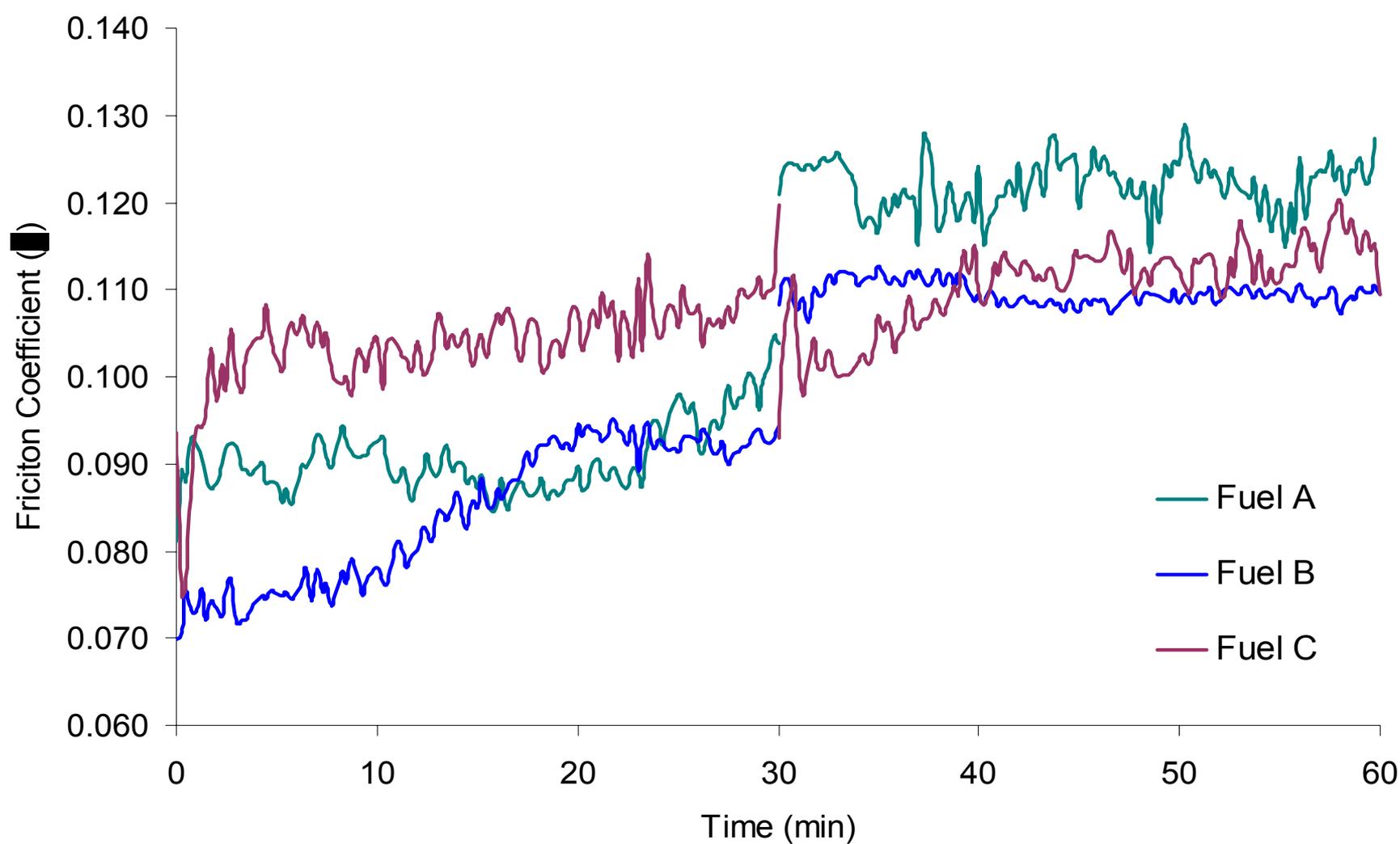


(10) Wain, K. Perez, J. "Oxidation of Biodiesel Fuels for Improved Fuel Lubricity" Proceedings of the Internal Combustion Engine Division, Lubrication and Friction Committee ASME Rockford, IL #2002-ICE-447 (2002)

# Low Sulfur & Oxidized Diesel Fuels



# Friction Traces for ULSDFs



# Fuel Deposit Tests

JMP13@PSU.EDU



## Micro-oxidation test

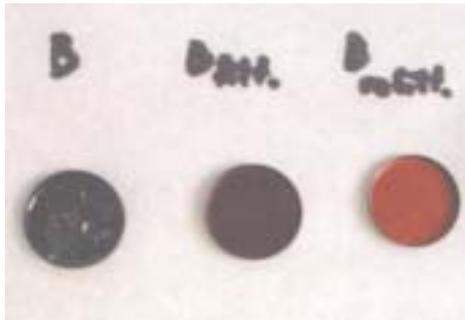
10ml of test fuel into glass test tube  
One stainless steel pan  
Heat to 150°C for 7 days  
Weigh and characterize deposits on pan

Fuel also filtered through Al  
column to remove additives and  
analyzed

## Test Fuels

A,B,C	Ultra low sulfur fuels, different manufacturers
D	Low sulfur diesel
E	Kerosene
G	#2 diesel

# Fuel Deposits



Progressively less deposits as B is filtered



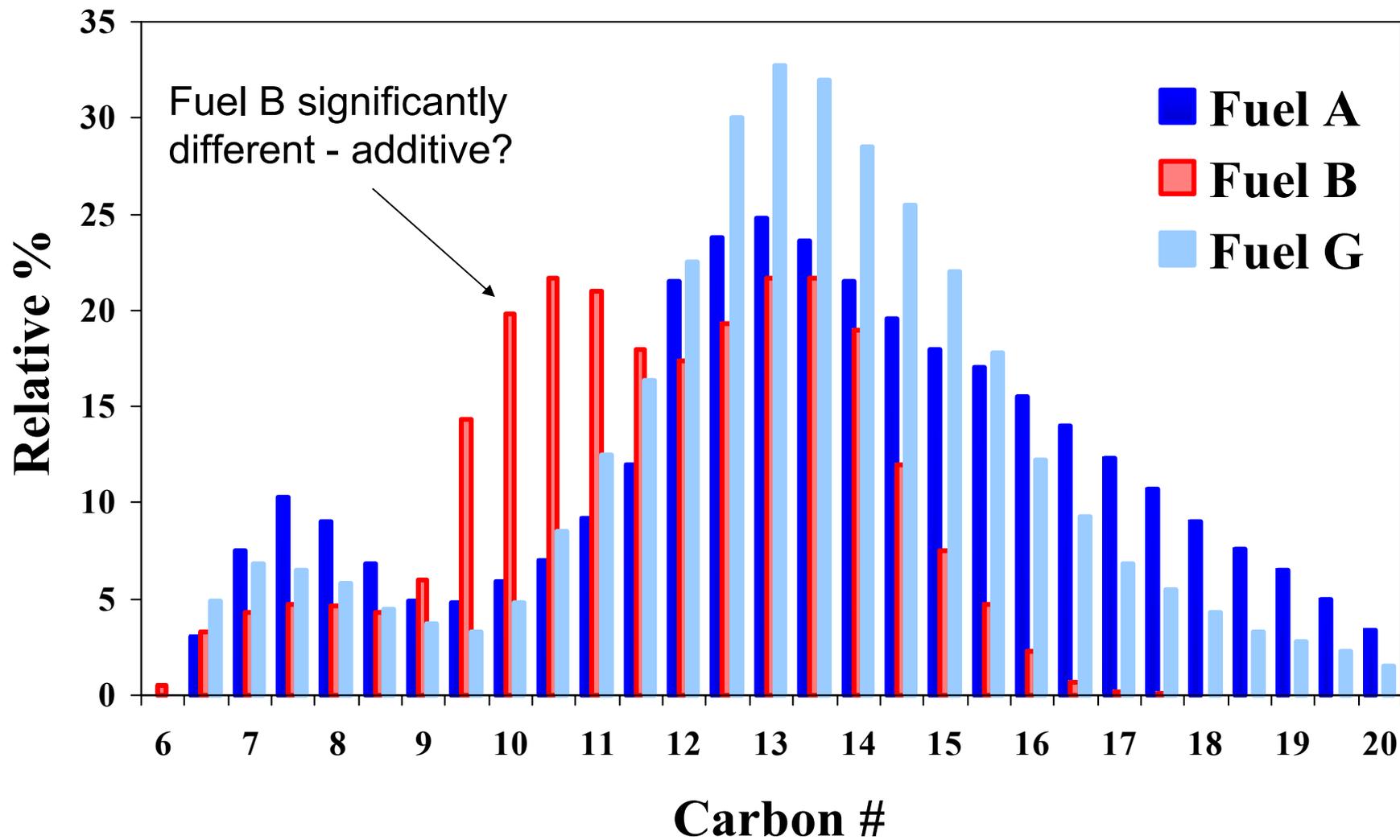
Order of deposit thickness, most to least:  
**B >> A > D > C > G > E**



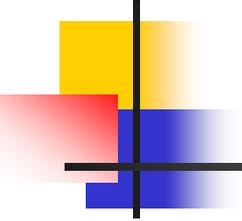
Progressively less deposits as G is filtered (not as dramatic as B)

\*Filtered fuel shows little or no deposits on walls of glass micro-oxidation tubes as well as on coupons

# GC Analyses - Fuels



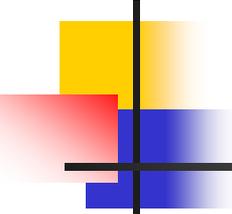
Fuel C, D similar to A



# LUBRICANT RESEARCH

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Does the Chemical Structure of the Base Fluid affect its effectiveness in protecting the surface against wear?



# Effect of Structure

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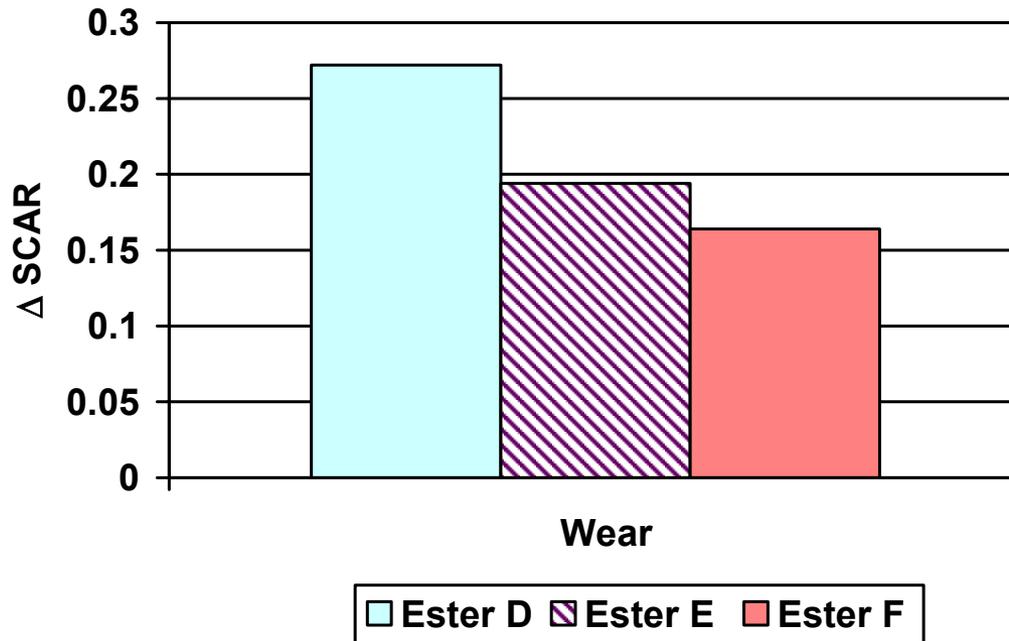


**To evaluate structure effect use same acid (2-ethylhexanoic) and different alcohols**

- **Neopentyl Glycol**  $(\text{CH}_3)_2\text{C}(\text{CH}_2\text{OH})_2$
- **Trimethylol propane**  $\text{CH}_3\text{CH}_2\text{C}(\text{CH}_2\text{OH})_3$
- **Pentaerythritol**  $\text{C}(\text{CH}_2\text{OH})_4$

# Effect of Acid Chain Length on Wear

■ Trimethylol propane = alcohol



**Acids:**

**D = nC5**

**E = nC7**

**F = mixture of  
nC8 & C10**

# Wear Index =

$$\frac{(\text{Total Carbons})(\text{Effective Chain Length})}{(\text{Polar Value} + \text{Branching Value})}$$

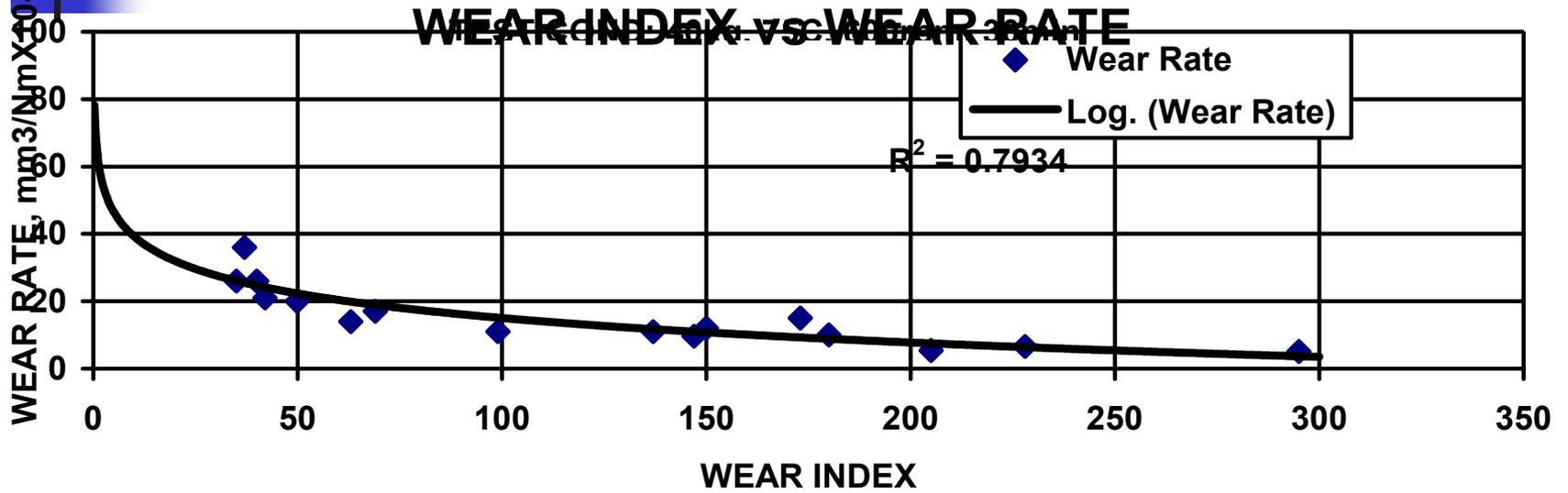
**where:**

**Total Carbons = total carbons in the molecule**

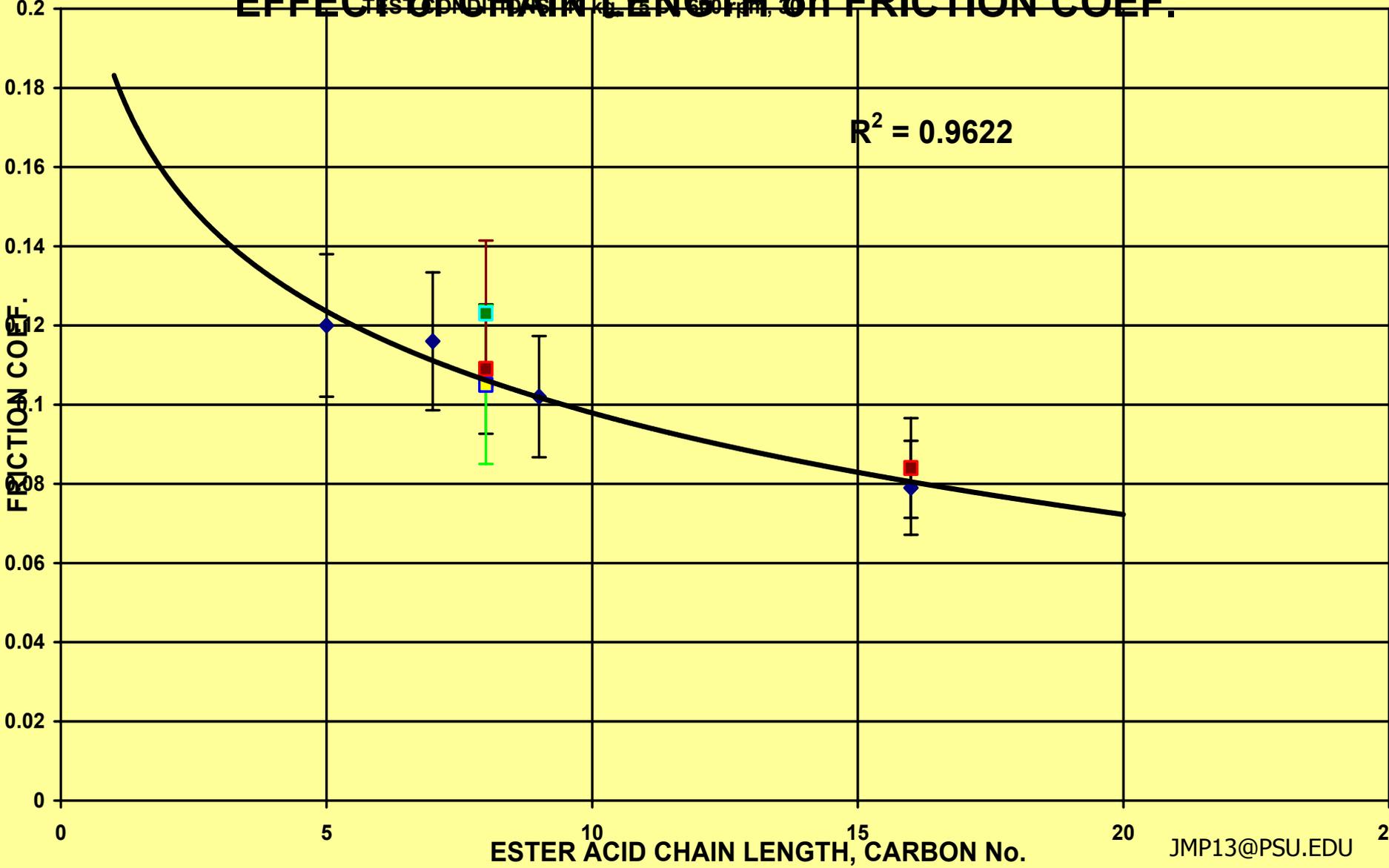
**Effective Length = longest free chain of carbons available to form a film.**

**Polar Value = No. of carboxyl groups + No. of hydroxyl groups**

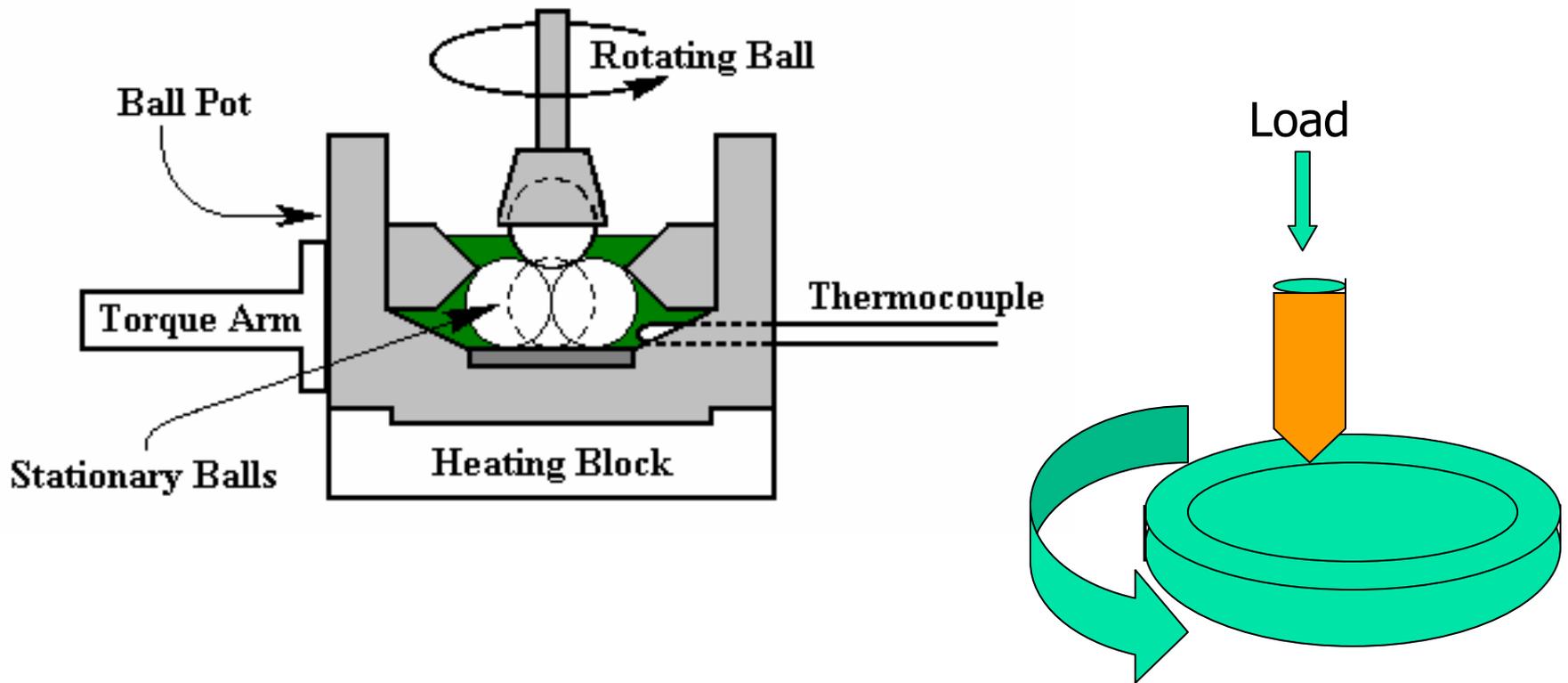
**Branching Value = ( 0.5 x No. of branches) + No. of double bonds.**

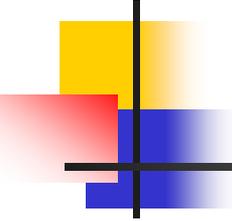


# EFFECT OF CHAIN LENGTH on FRICTION COEF.



# Test Methods





# Test Conditions

## Four Ball Wear Tester:

**ANSI 52100 stainless steel balls**

**Test Time:**

**30 min Run-in**

**30 min Steady State**

**30 min Surface Eval'n**

**Test temp. = RT, 60°C, 75°C**

**Speed = 600, 1200 RPM**

**Loads = 1,10, 40 Kg**

## Pin – on – Disc:

Variable Speed

Variable Load

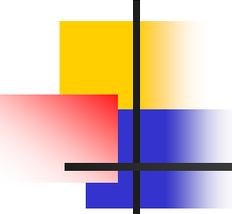
This study:

10 RPM

20 N

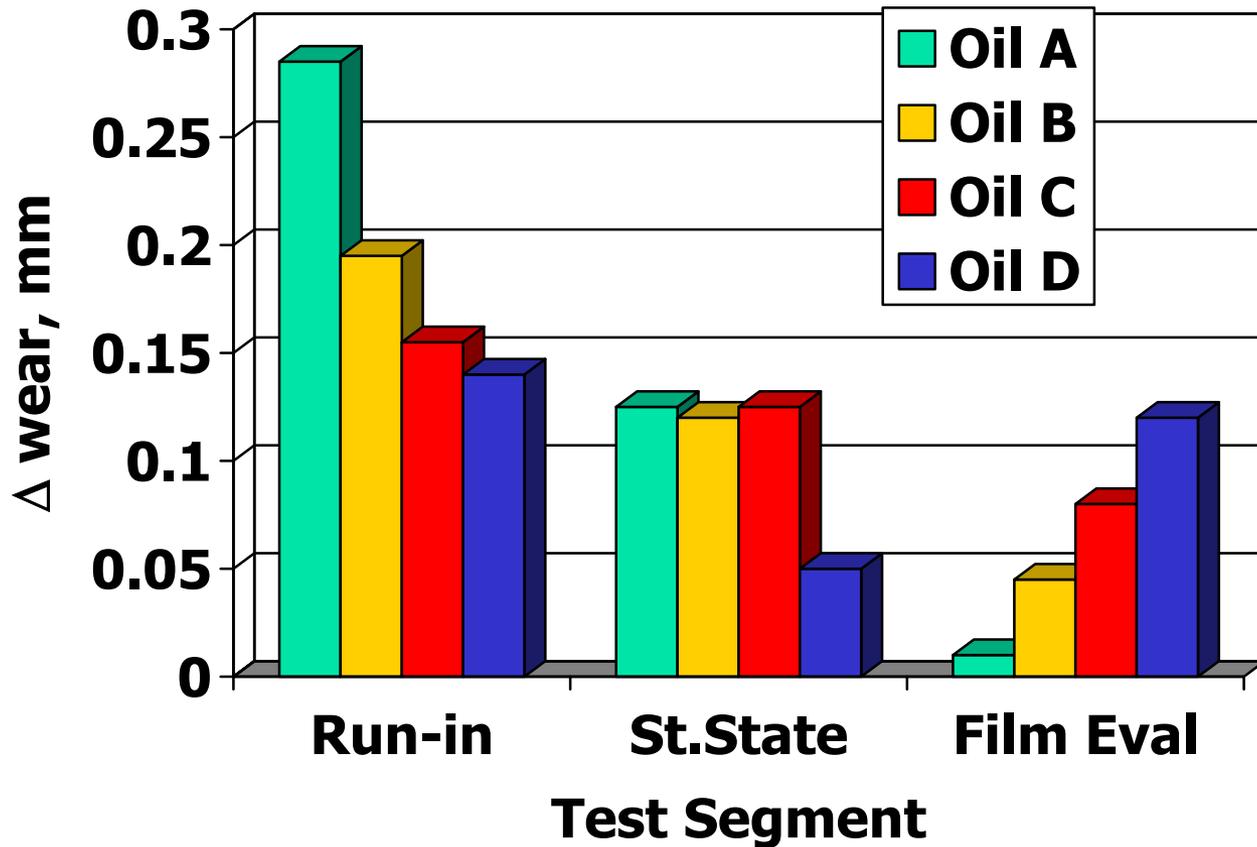
Room Temp.

## Properties of Test Oils

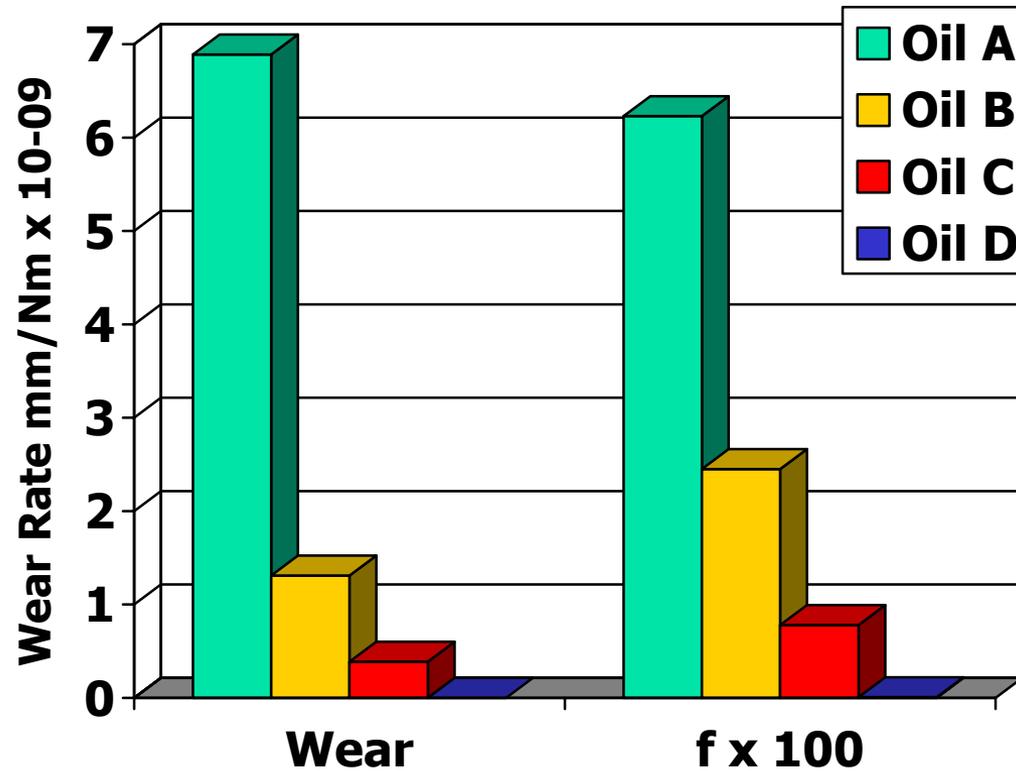


Oil Properties	Oil A	Oil B	Oil C	OIL D
cSt Visc @ 100 °C (ASTM D 445)	3.9	4.99	8.29	24.4
cSt Visc @ 40 °C	16.9	28.8	66.8	215
Viscosity Index (ASTM D2270)	123	97	91	120
Flash Point, °C (ASTM D 92)	219	226	254	>>200

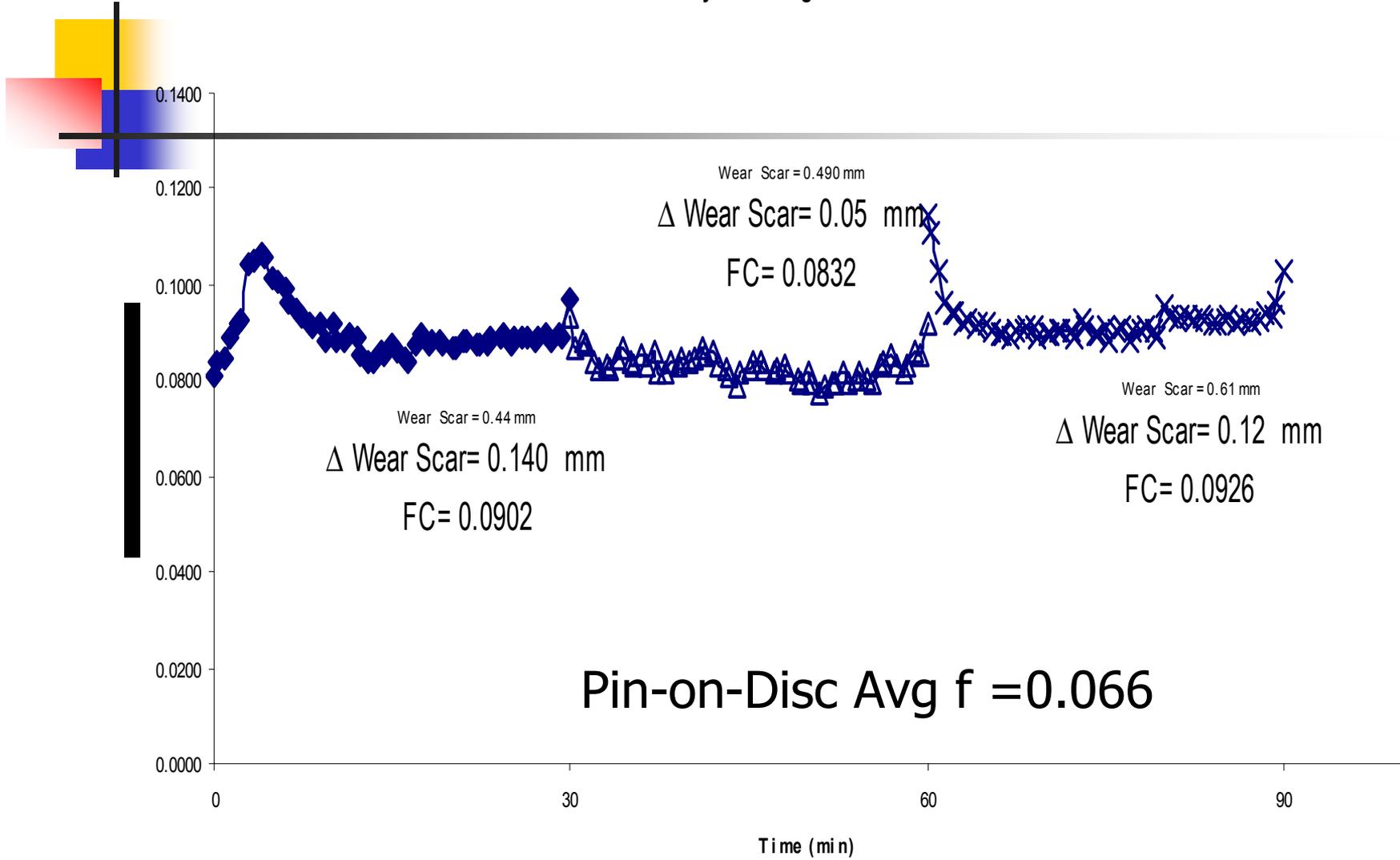
# Effect of "Chain Length" of Hydrocarbon Oils on Wear – 4Ball Test



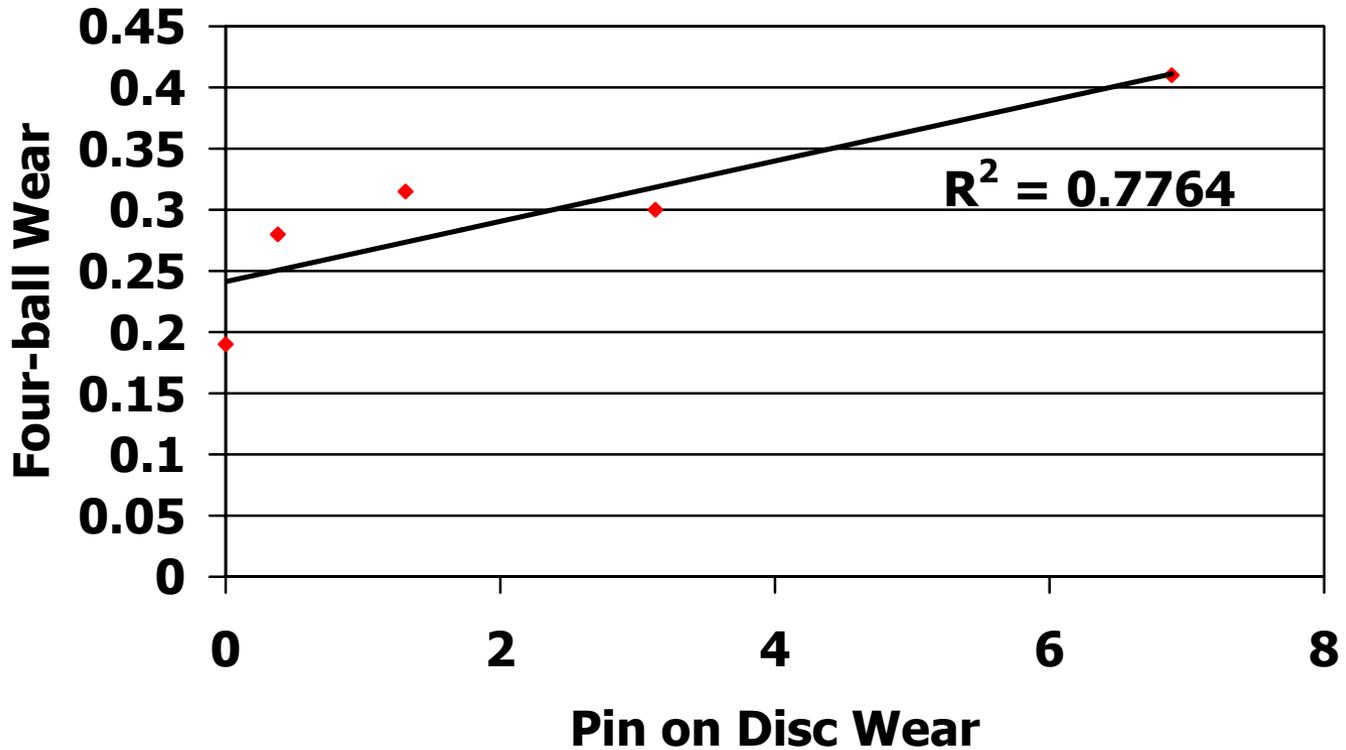
# Effect of Oil "Chain Length" – Tribometer (CSEM)



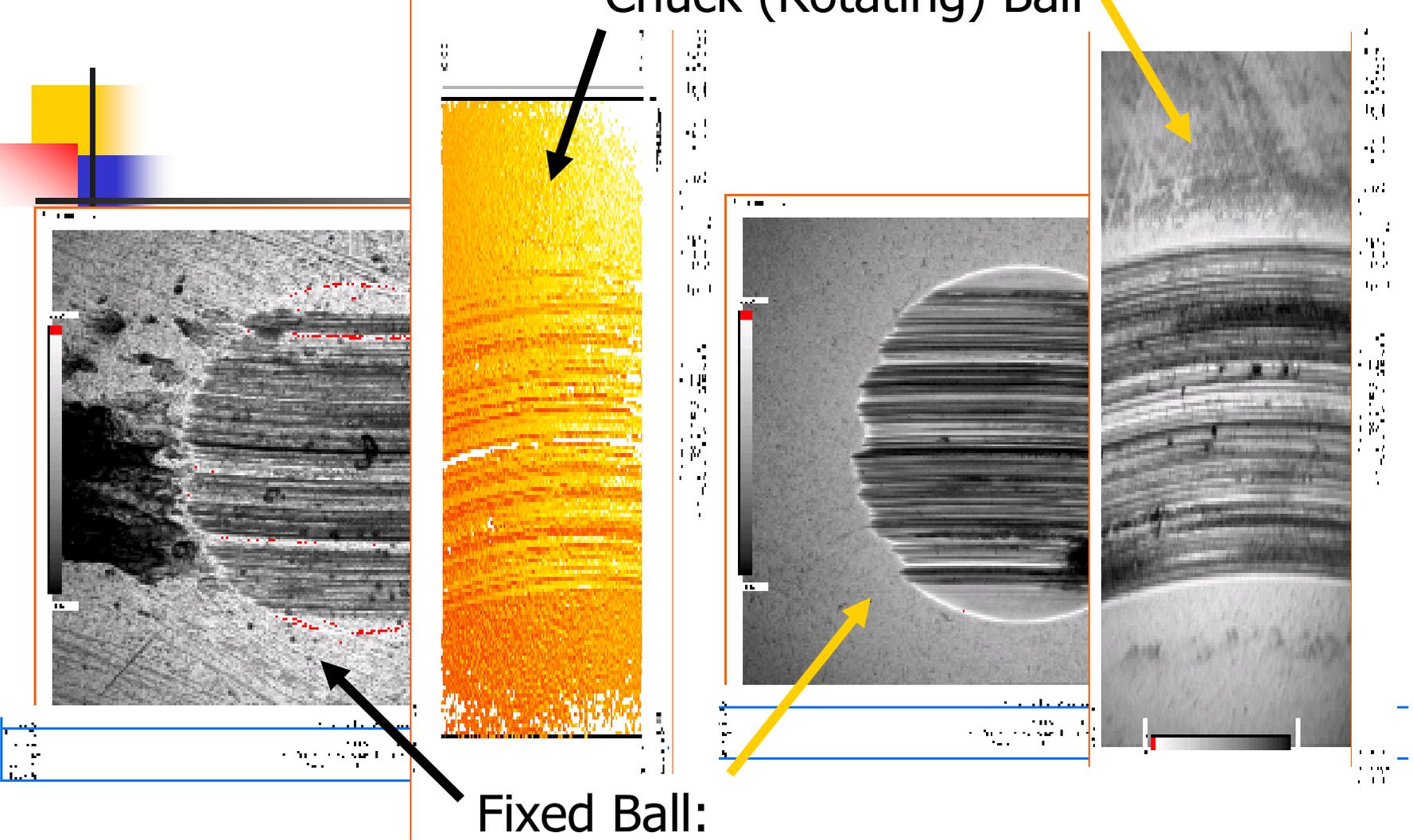
# Friction coefficient of HMW Synth - Veg Oil without antiwear additive



# Effect of "Chain Length" of Hydrocarbon Oils on Wear – 4Ball Test vs Tribometer



Chuck (Rotating) Ball

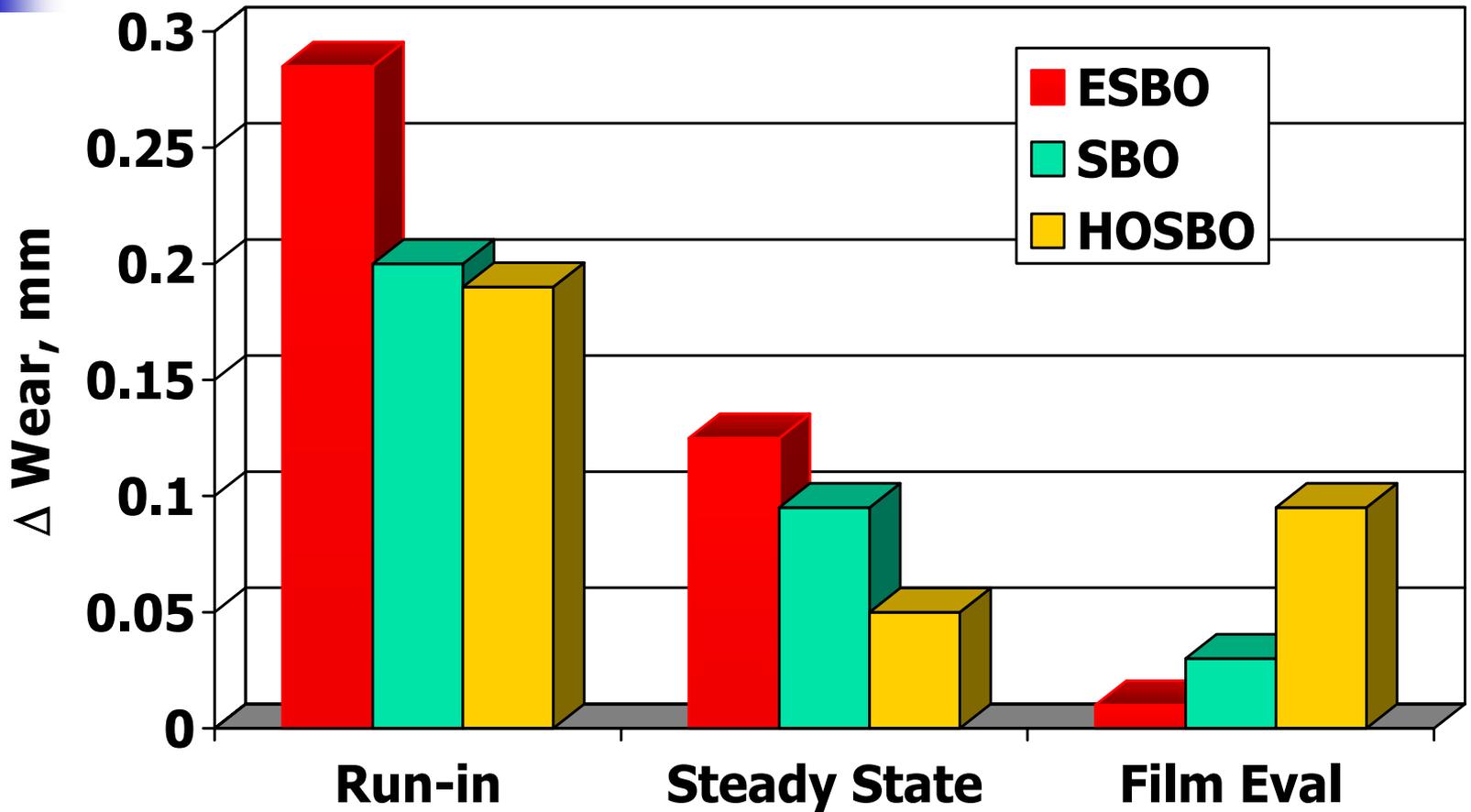


Fixed Ball:

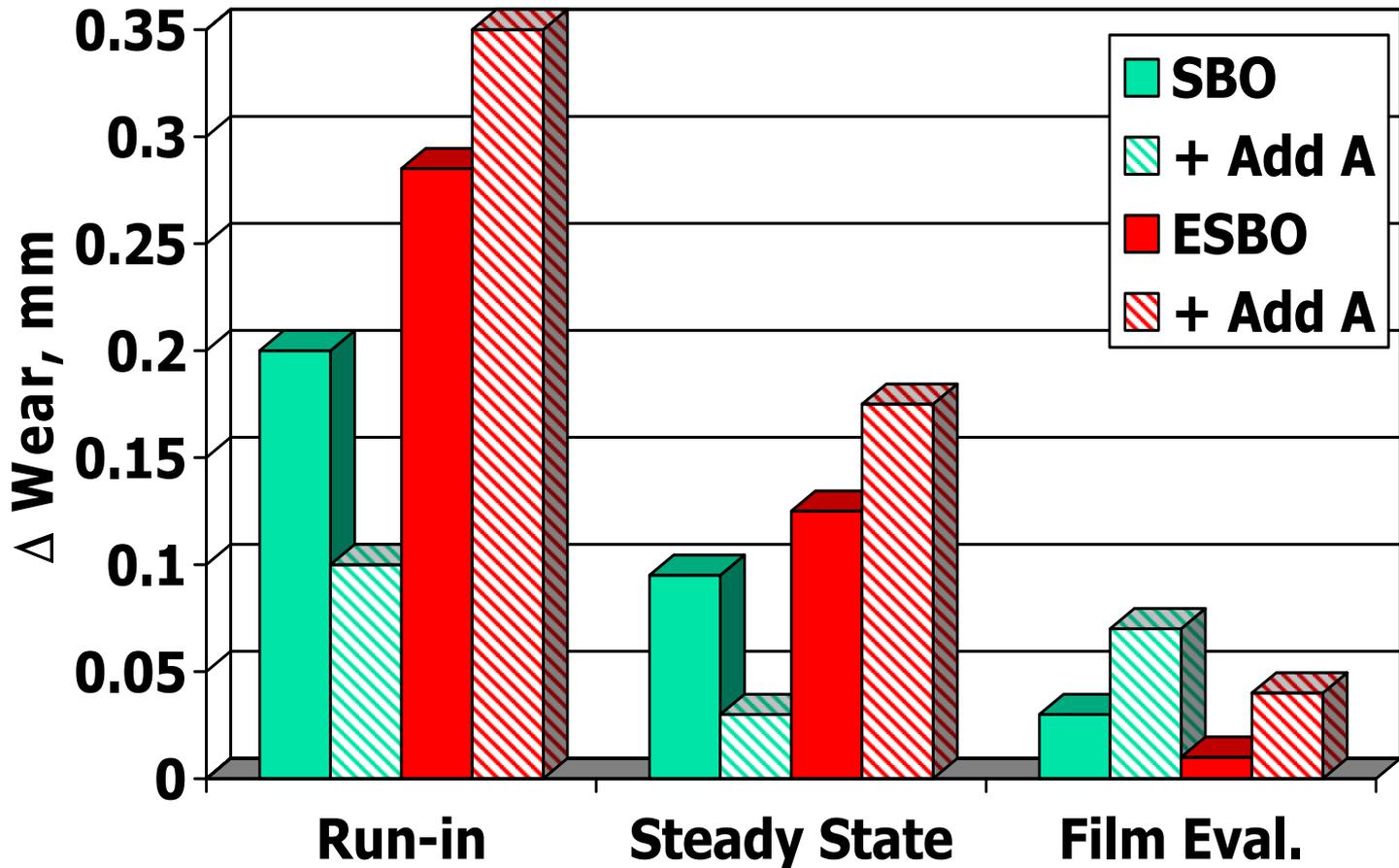
Before Cleaning

After Cleaning

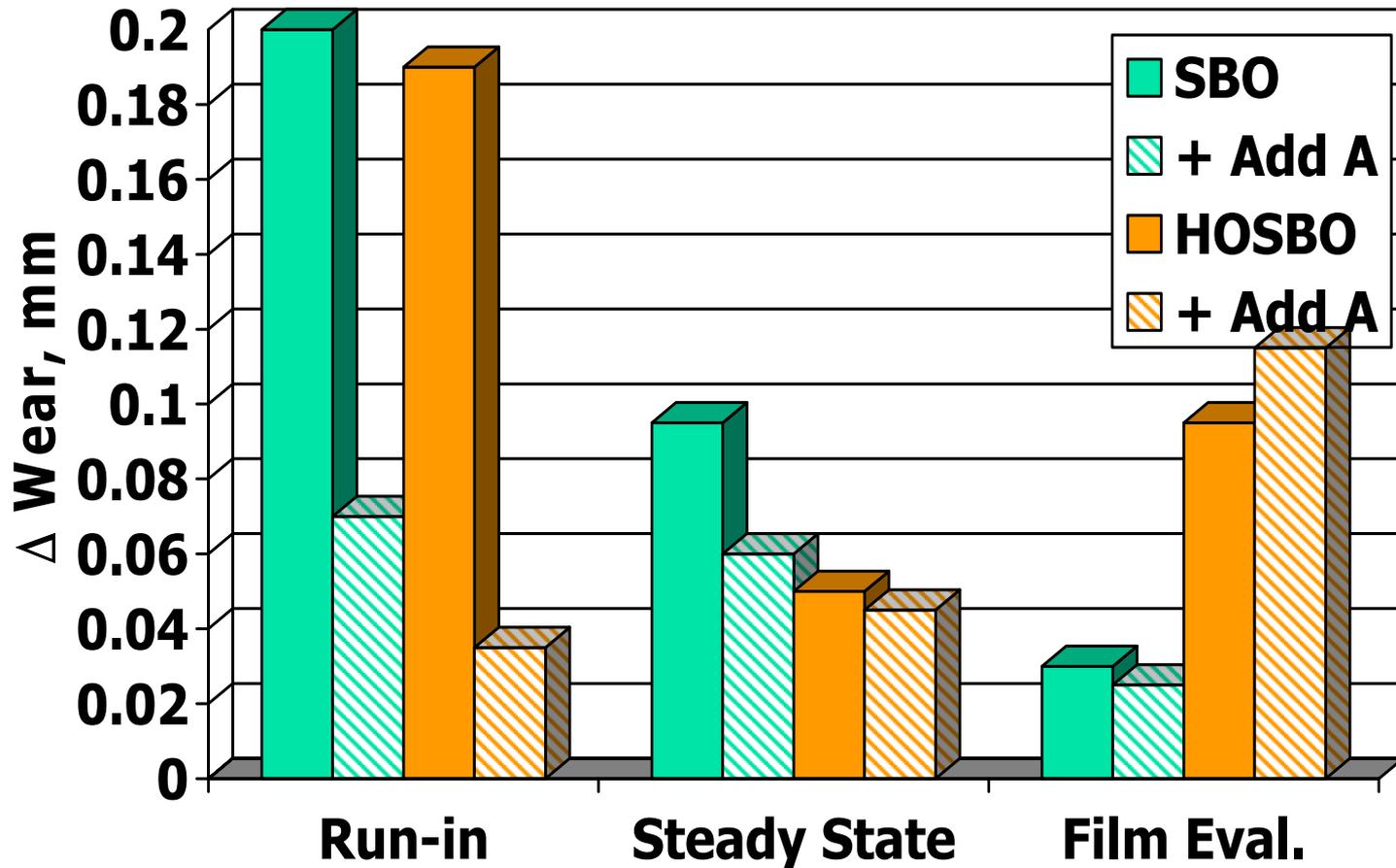
# Effect of Double Bonds – Veg Oils



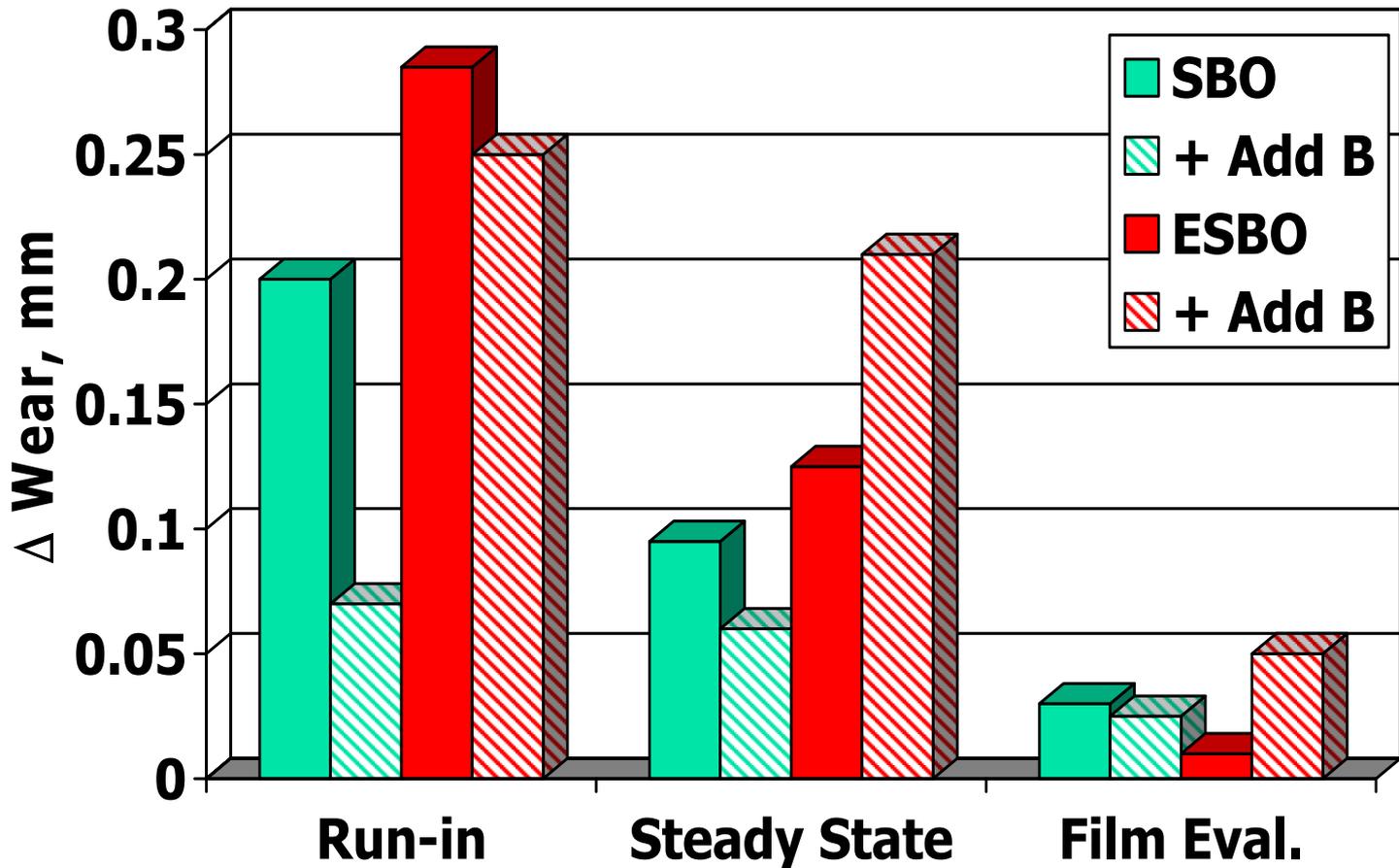
# Additive Effectiveness - Additive A



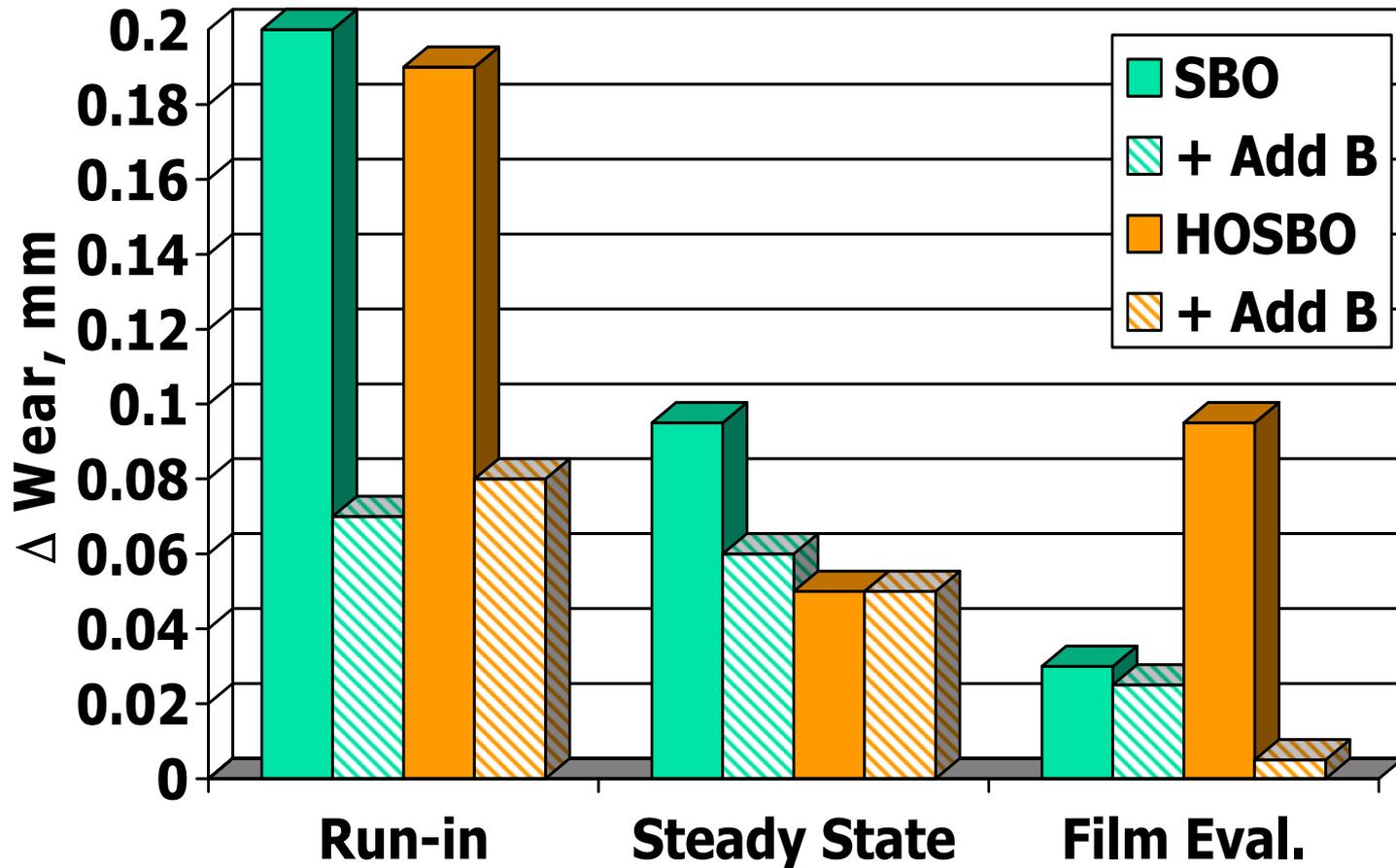
# Effect of Unsaturation - Additive A



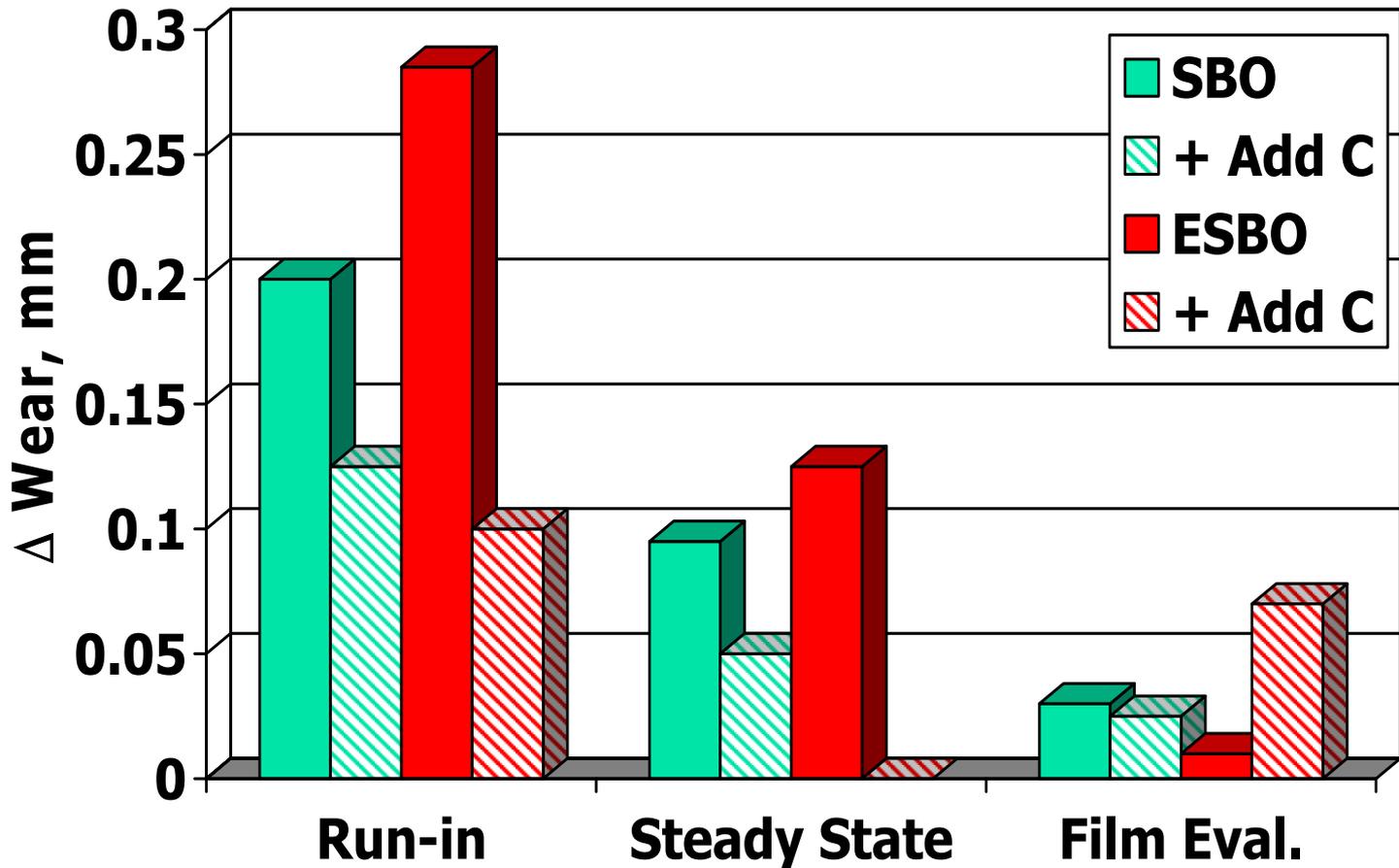
# Additive Effectiveness - Additive B



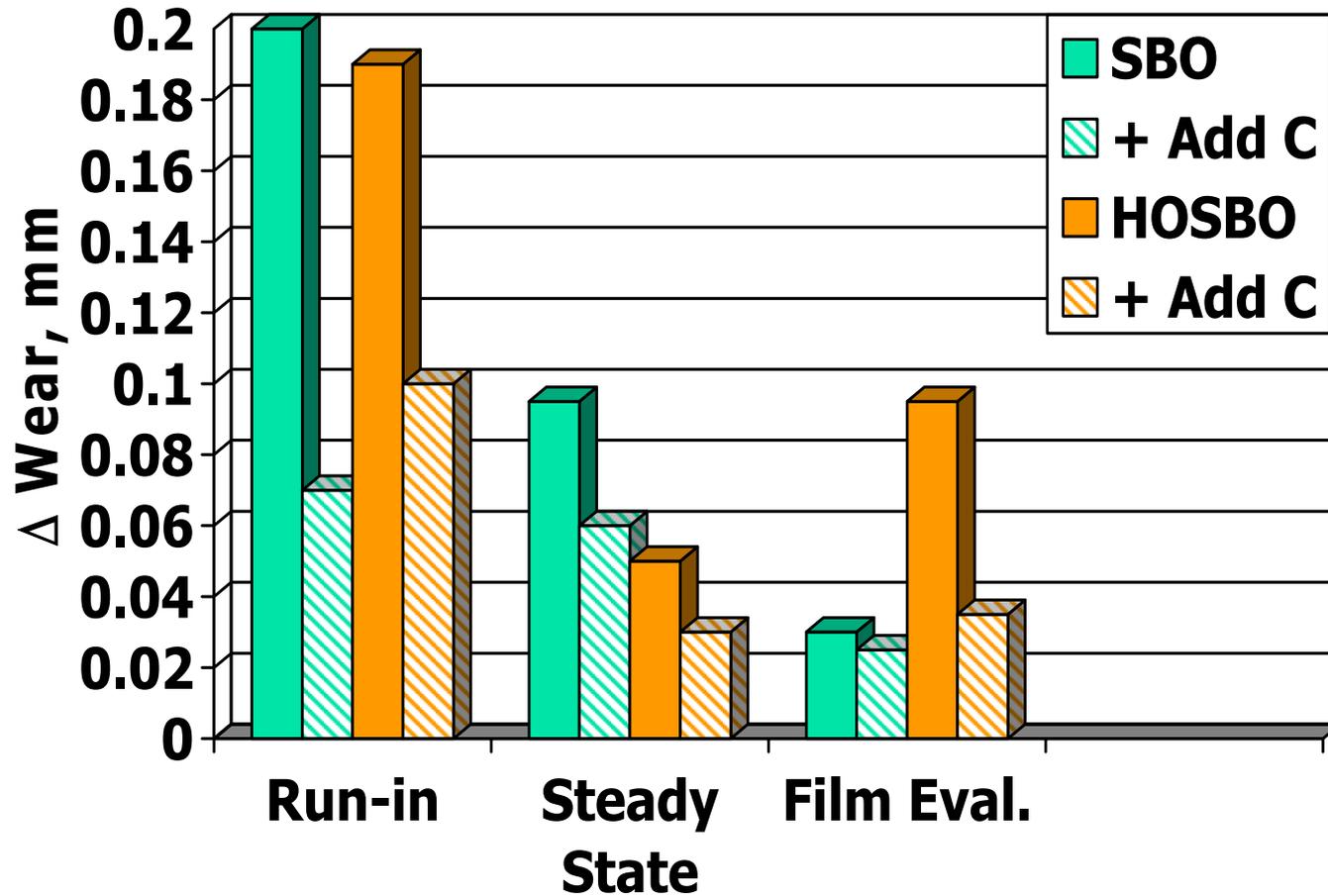
# Effect of Unsaturation - Additive B

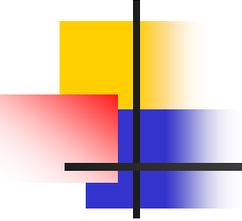


# Additive Effectiveness - Additive C



# Effect of Unsaturation - Additive C

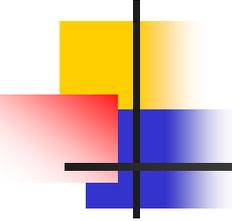




# Summary

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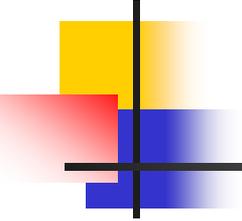
- EFF & L - research studies & demonstration projects.
- Oxygenated Alternative Fuels - reduce particulates.
- DME - potential wear problems.
- VPO Biodiesel - effective f & wear additive.
- ULSF's – wear, deposits, filter plugging.
- Chemical structure of base fluids and additives - significant factor in future lubricant formulation.
- New test methods - key to understanding surface interactions. (Optical, Advanced Photon Source, etc.)
- Surface engineering—materials, coatings & lubricants.



# Acknowledgement

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- Appreciation is given for partial funding of these projects by Air Products; Cargill Corp; Caterpillar, Inc.; Cummins Engine Co.; USDA Laboratory (Peoria, Il) and Valvoline, Inc. Their financial support is appreciated.
- A special thanks to Dr. George Fenske and the Tribology Group at Argonne National Laboratory for their continued interest and support of this research.



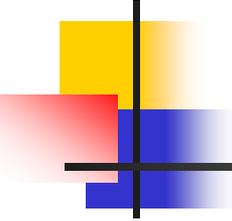
# The research contributions of the following Graduate Students is acknowledged:

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- Penn State University:
  - Kimberly Wain – Biodiesel Fuels, DME
  - Elana Chapman–DME, Oxygenated Fuels
  - Waleska Castro – Veg. Oils, f & wear tests
  - Kraipat Cheenkachorn – Vegetable Oils
  - David Weller – Chemical Characterization
- Northwestern University:
  - Ashlie Martini - Pin-on-disc studies,
  - Mark Sturino - Pin-on-disc studies, Optical Microscopy

No Not  
JOEPA





# Is Tribology Important?

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- Lack of Tribological Solutions results in Big Business:
  - 1980 Survey - Over 20 Billion lost due to friction and wear annually
  - 1995 - Over 1.5% of the gross national product is lost due to friction and wear

ASME Research Committee, circa 1980

Amato, Ivan, “ Better ways to Grease Industry’s Wheels” Fortune, Sept 1995; 256 [B]-256[K]