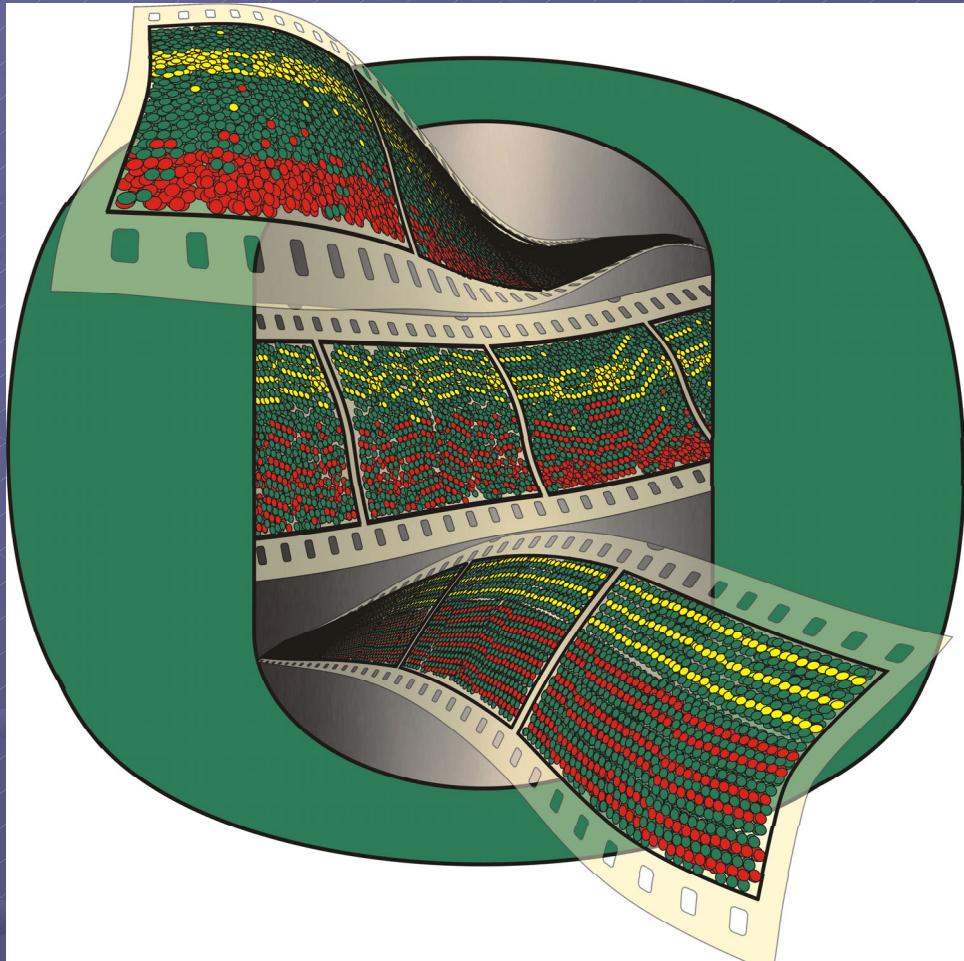


Crystal Engineering of Novel Thermoelectric Materials



David Johnson

Polly Berseth

Fred Harris

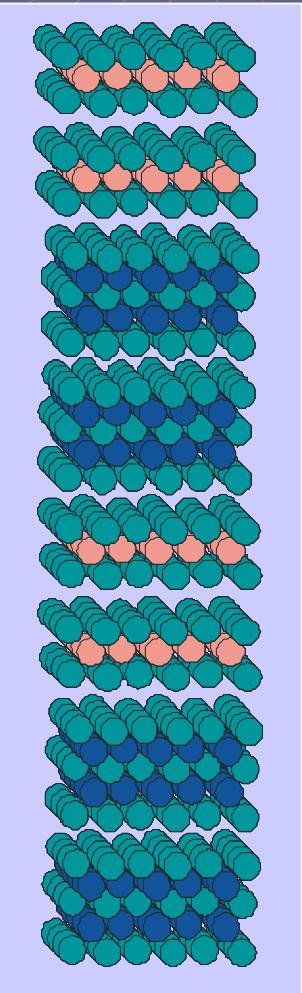
Ngoc Nguyen

University of Oregon

18 Feb 2004

What is a superlattice?

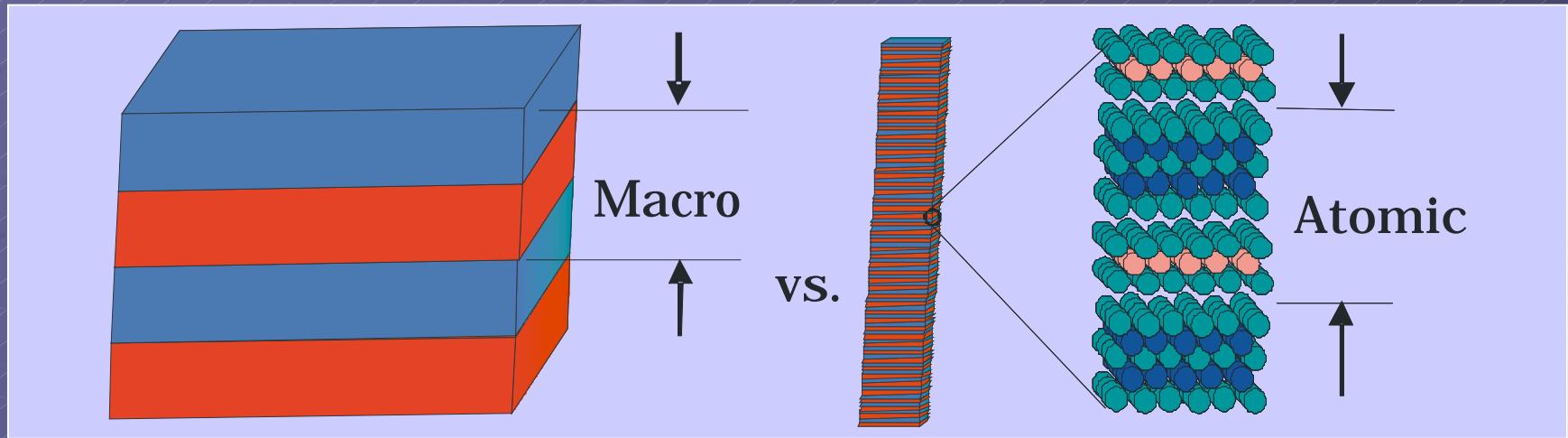
Quick Review



- Two or more materials are alternated
- The resulting structure is then repeated
- Layers within a superlattice are crystallographically aligned with one another

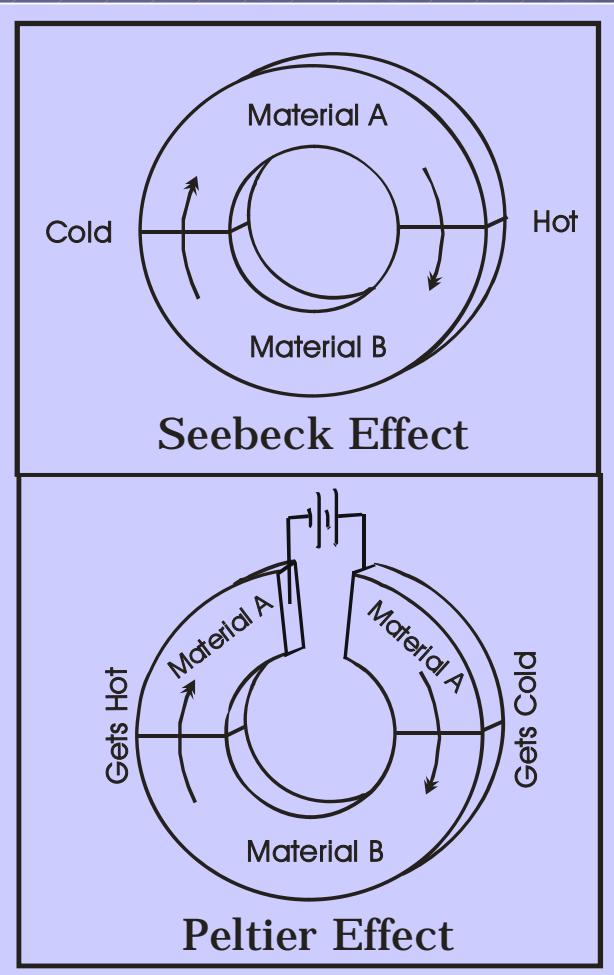
Why are Superlattices Interesting

Fundamental Understanding



- Composites are well understood on the macro scale
- What happens to the above alternating material as the layers get thinner?
- One unit cell of each material → new properties?

Thermoelectric Materials

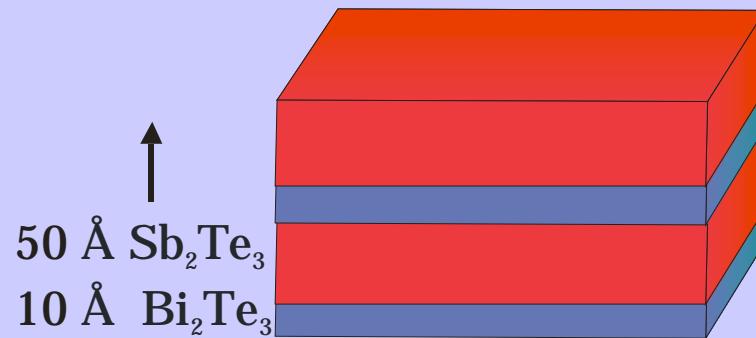


$$ZT = \left(\frac{\sigma S^2}{K_e + K_l} \right) T$$

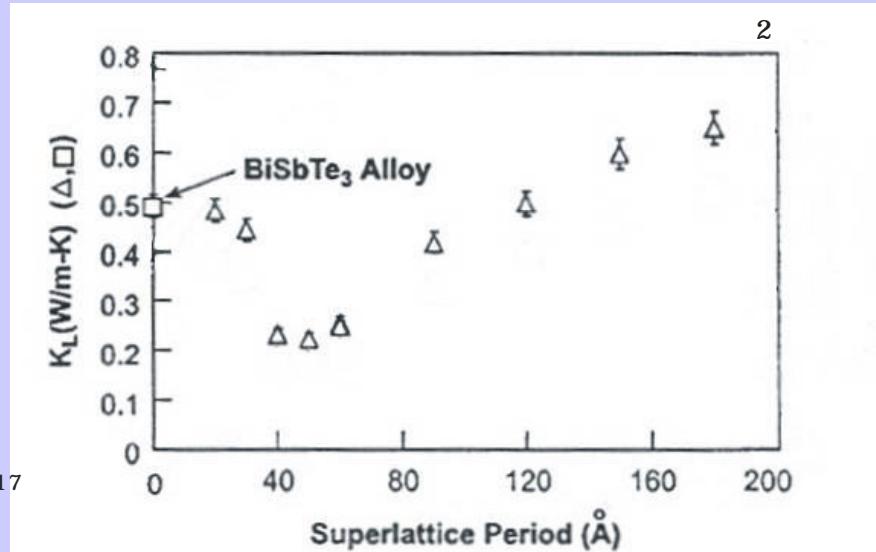
- ZT = Normalized Thermoelectric Eff.
- σ = electrical Conductivity
- S = Seebeck Coefficient
- K_e = Thermal Conductivity (Electronic)
- K_l = Thermal Conductivity (Lattice)
- T = Temperature

Superlattices and Thermoelectrics

Venkatasubramanian, et al. have shown evidence that superlattice structures lower thermal conductivities from the values of bulk components



$\text{ZT} \sim 2.4$ in p-type $\text{Bi}_2\text{Te}_3 / \text{Sb}_2\text{Te}_3$
 $\text{ZT} \sim 1.4$ in n-type $\text{Bi}_2\text{Te}_3 / \text{Bi}_{2.83}\text{Se}_{0.17}$

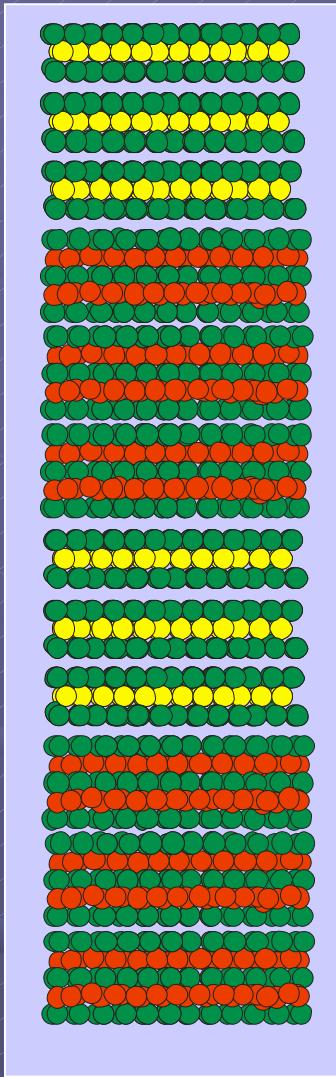


Highest ZT ever reported in a bulk alloy at 300K:
 $\text{ZT} \sim 1.14$ - p-type $(\text{Bi}_2\text{Te}_3)_{0.25}(\text{Sb}_2\text{Te}_3)_{0.72}(\text{Sb}_2\text{Se}_3)_{0.03}$

(1) Venkatasubramanian, R.; Slivola, E.; Colpitts, T.; O'Quinn, B. Nature 2001, 413, 597

(2) Venkatasubramanian, R. Phys. Rev. B: Condens. Matter Mater. Phys. 2000, 61, 3091.

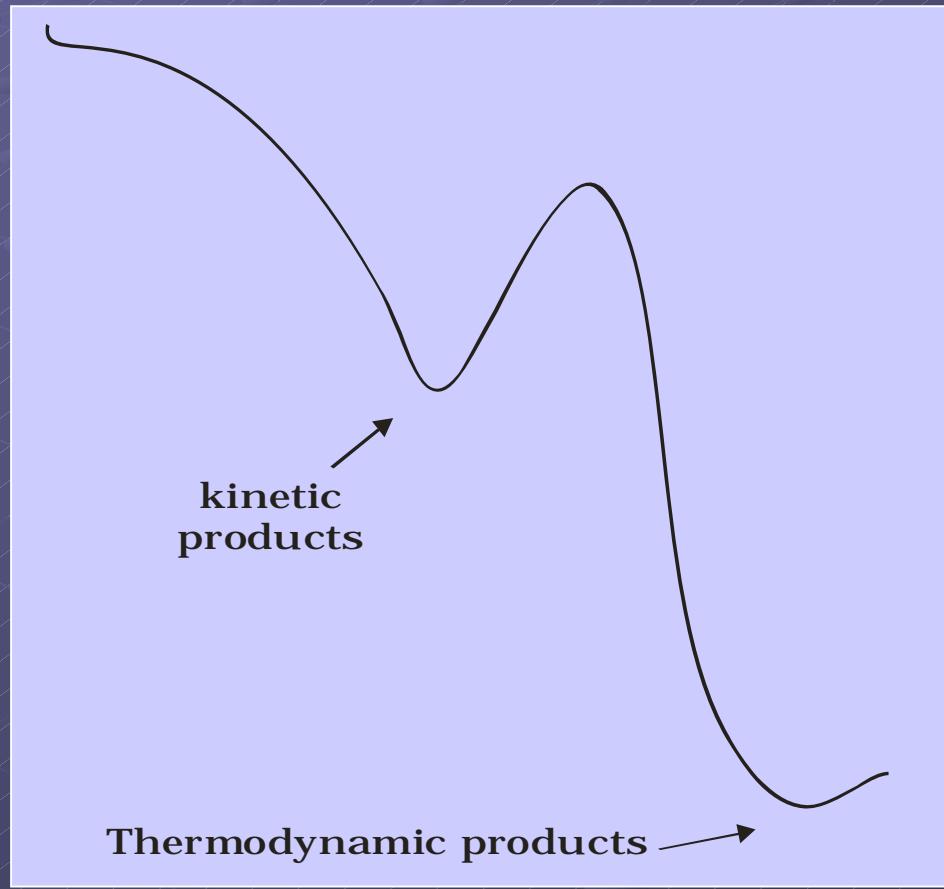
$[(\text{Bi}_2\text{Te}_3)_x(\text{TiTe}_2)_y]$ Superlattices



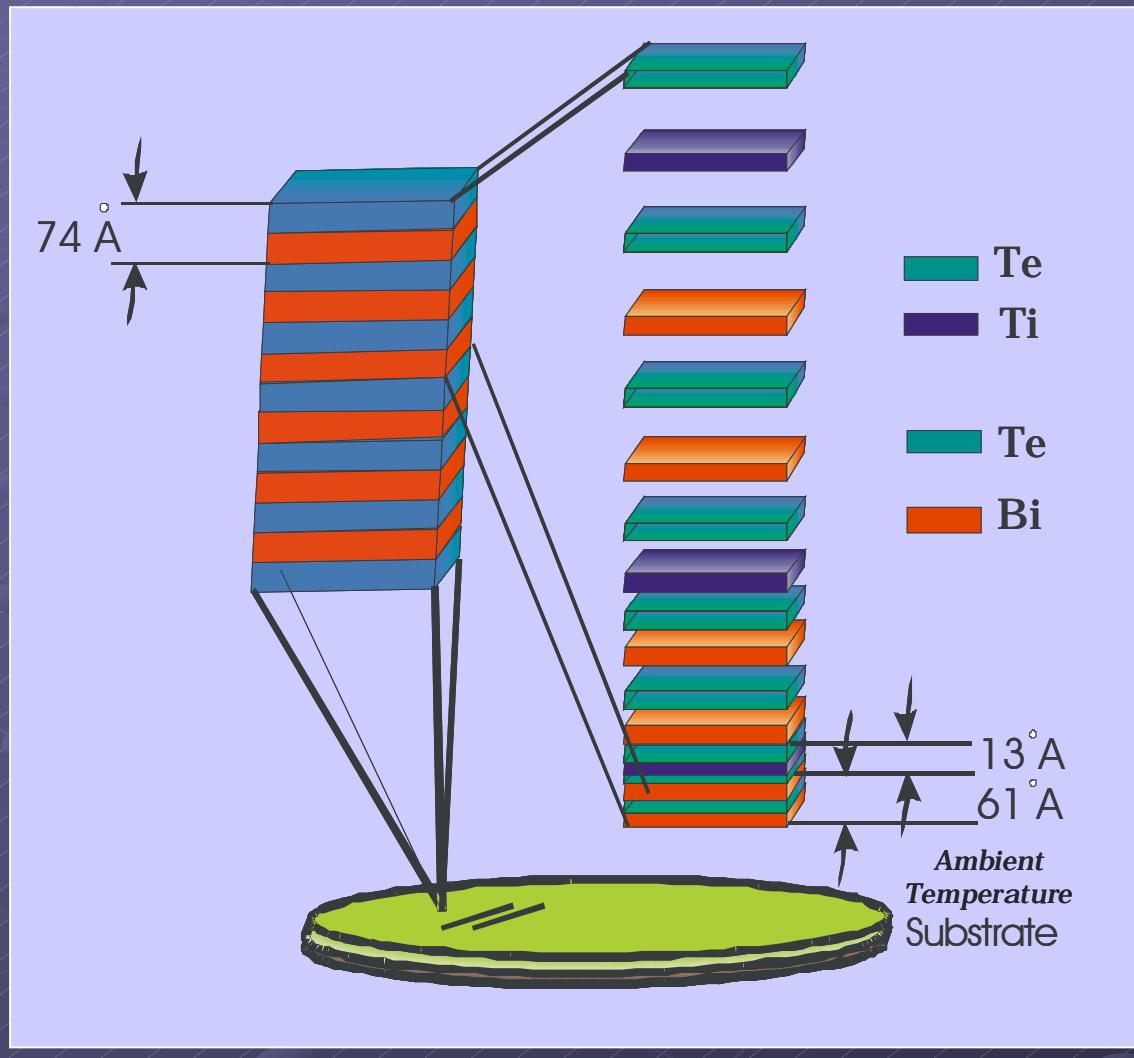
- Thermal conductivity reduction should hold true for other superlattice structures
- Van der Waals gaps should make clean interfaces between layers
- Interesting Properties?
 - Bi_2Te_3 – Semiconductor
 - TiTe_2 – Semi-Metal

A Low Temperature Technique is Required

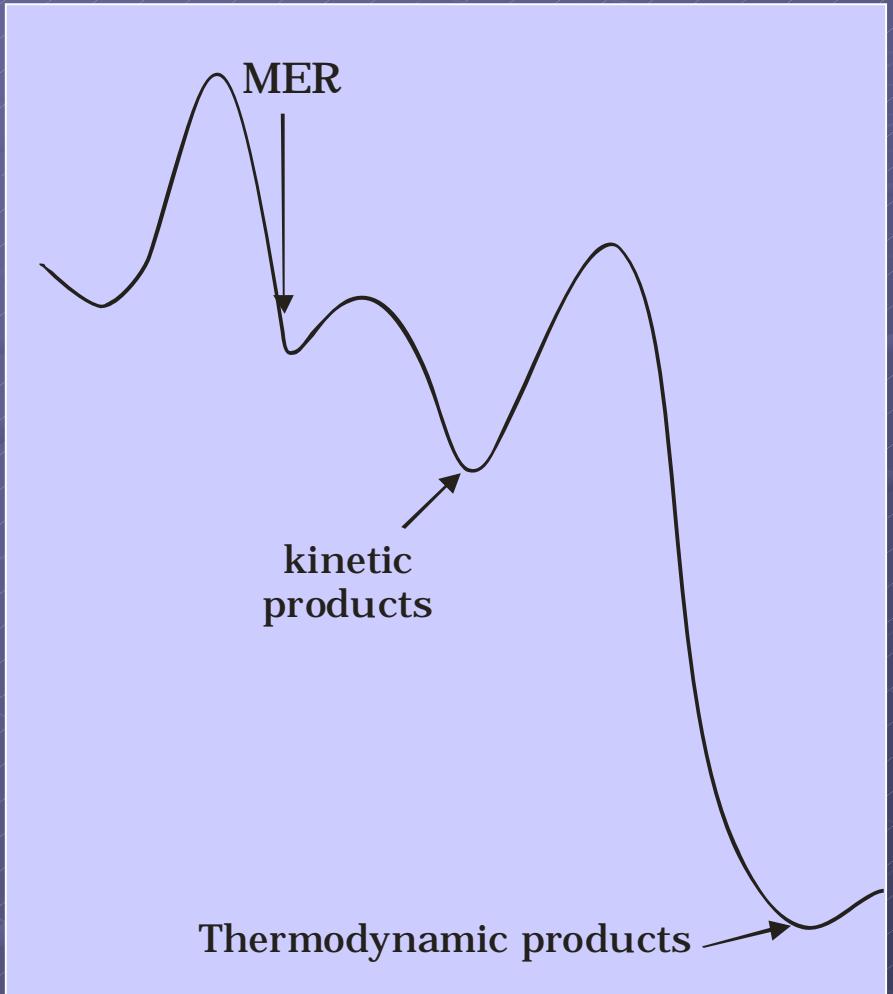
One of the main problems for trapping kinetic products in the solid state is overcoming diffusion



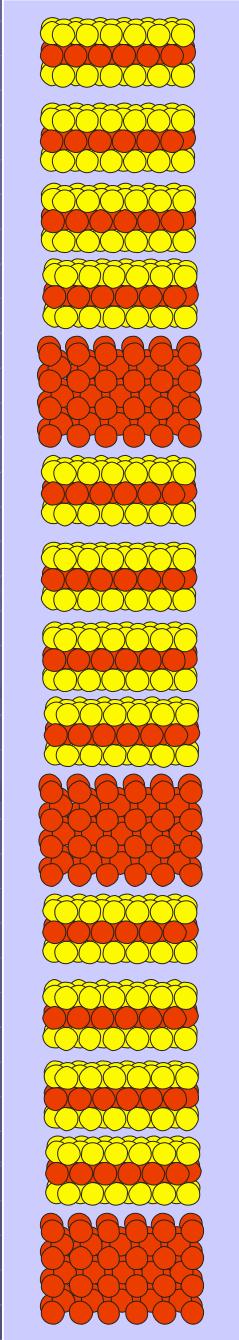
Modulated Elemental Reactants (MER)



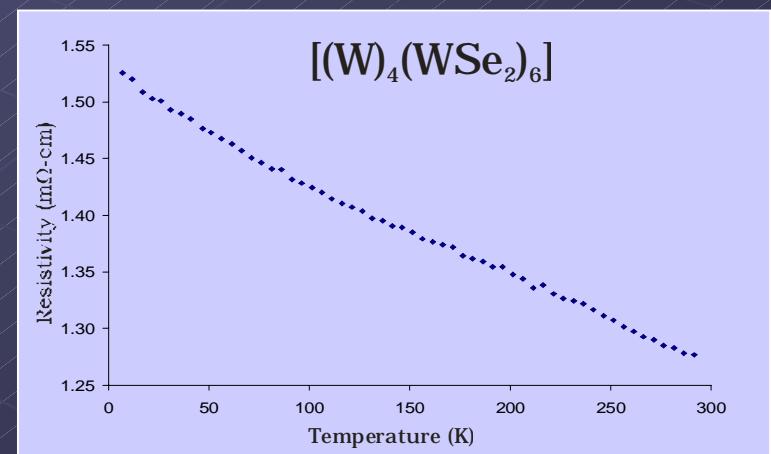
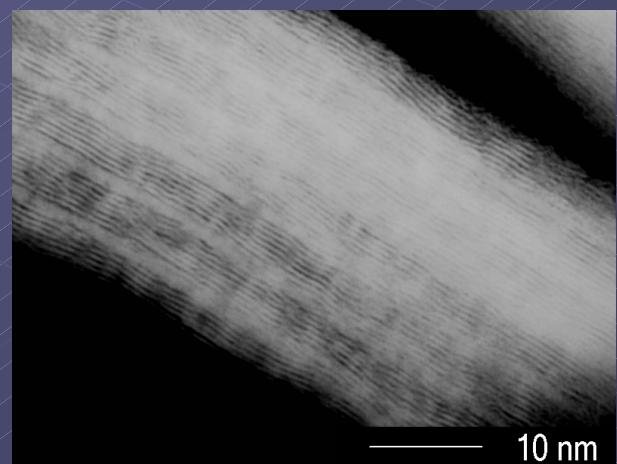
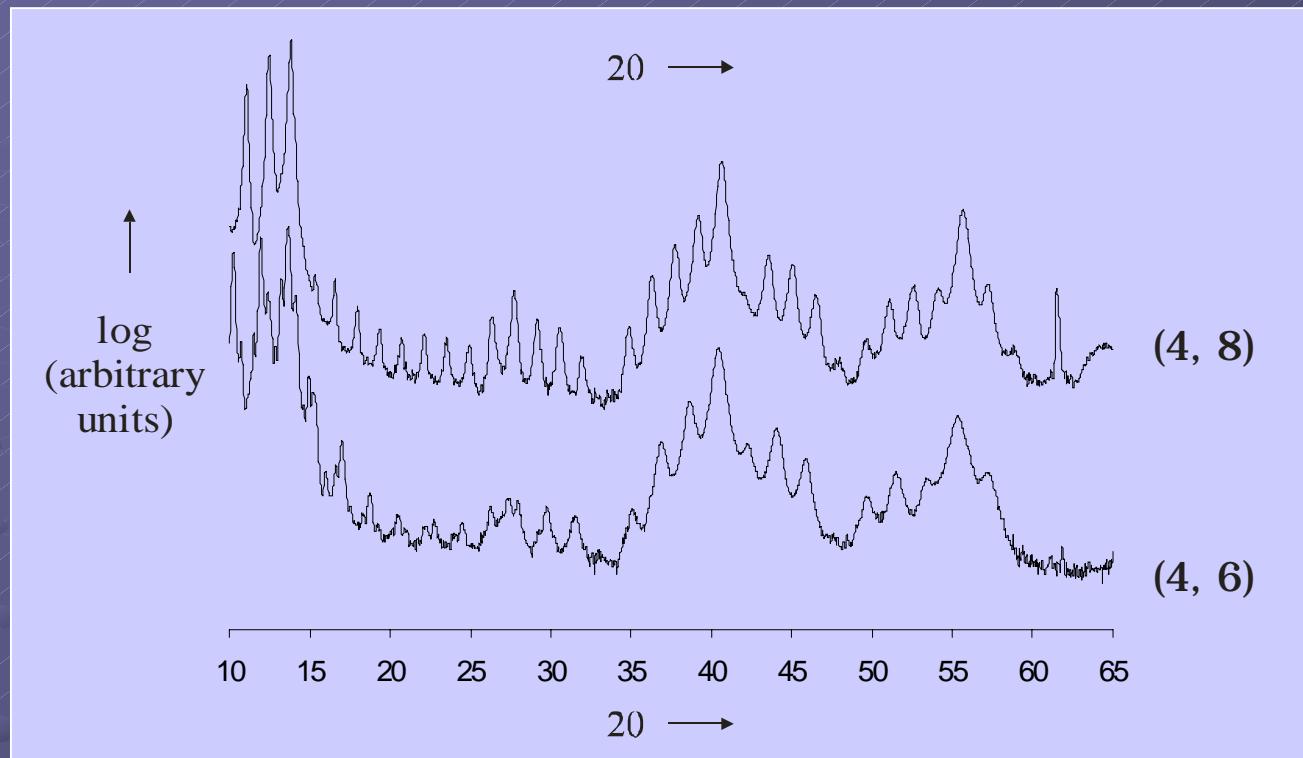
Modulated Elemental Reactants (MER)



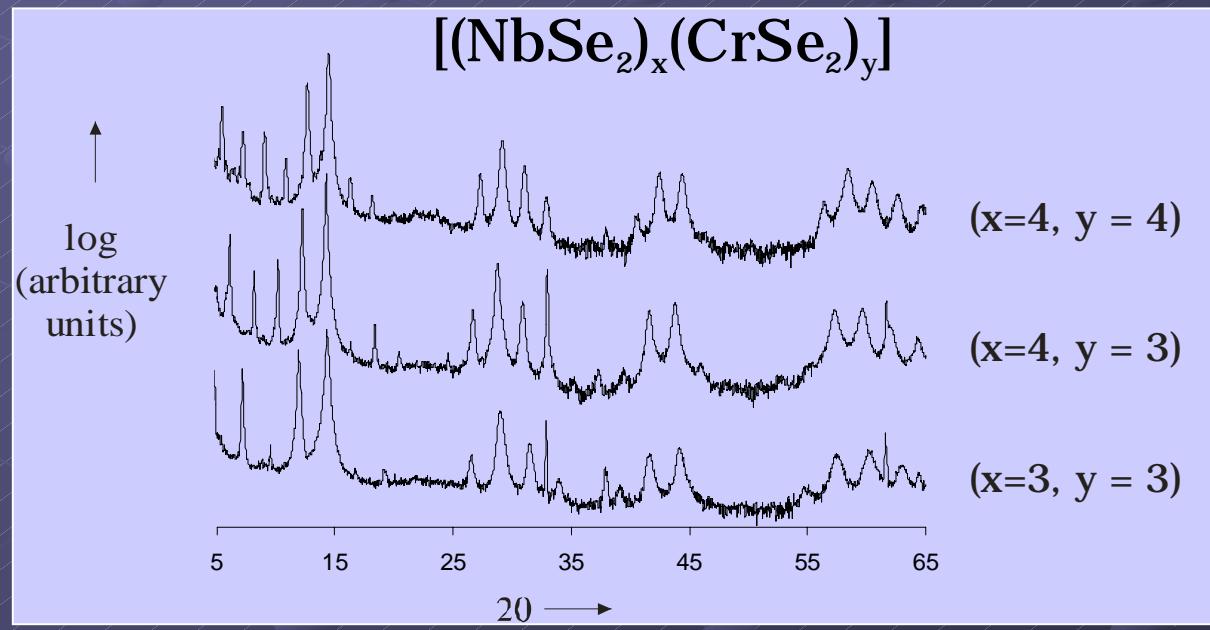
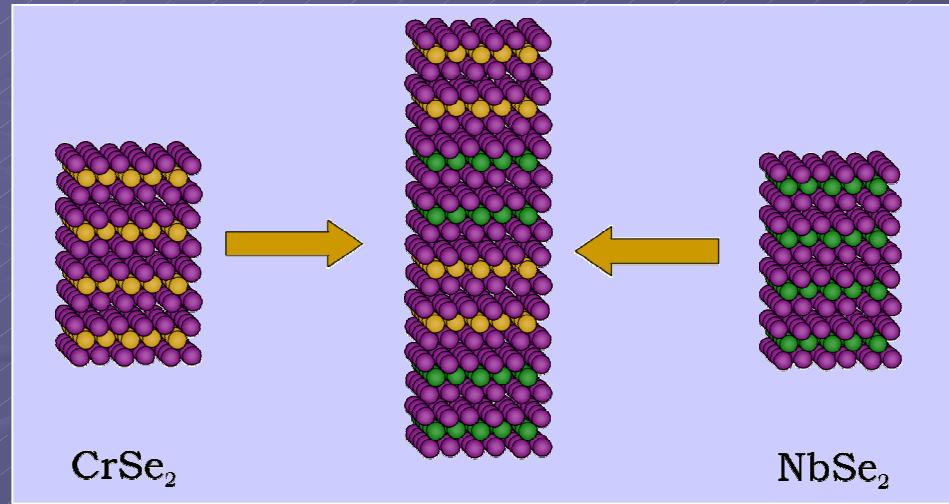
- Gentle annealing of the reactant stack ($<500^{\circ}\text{C}$) causes the kinetic product to form



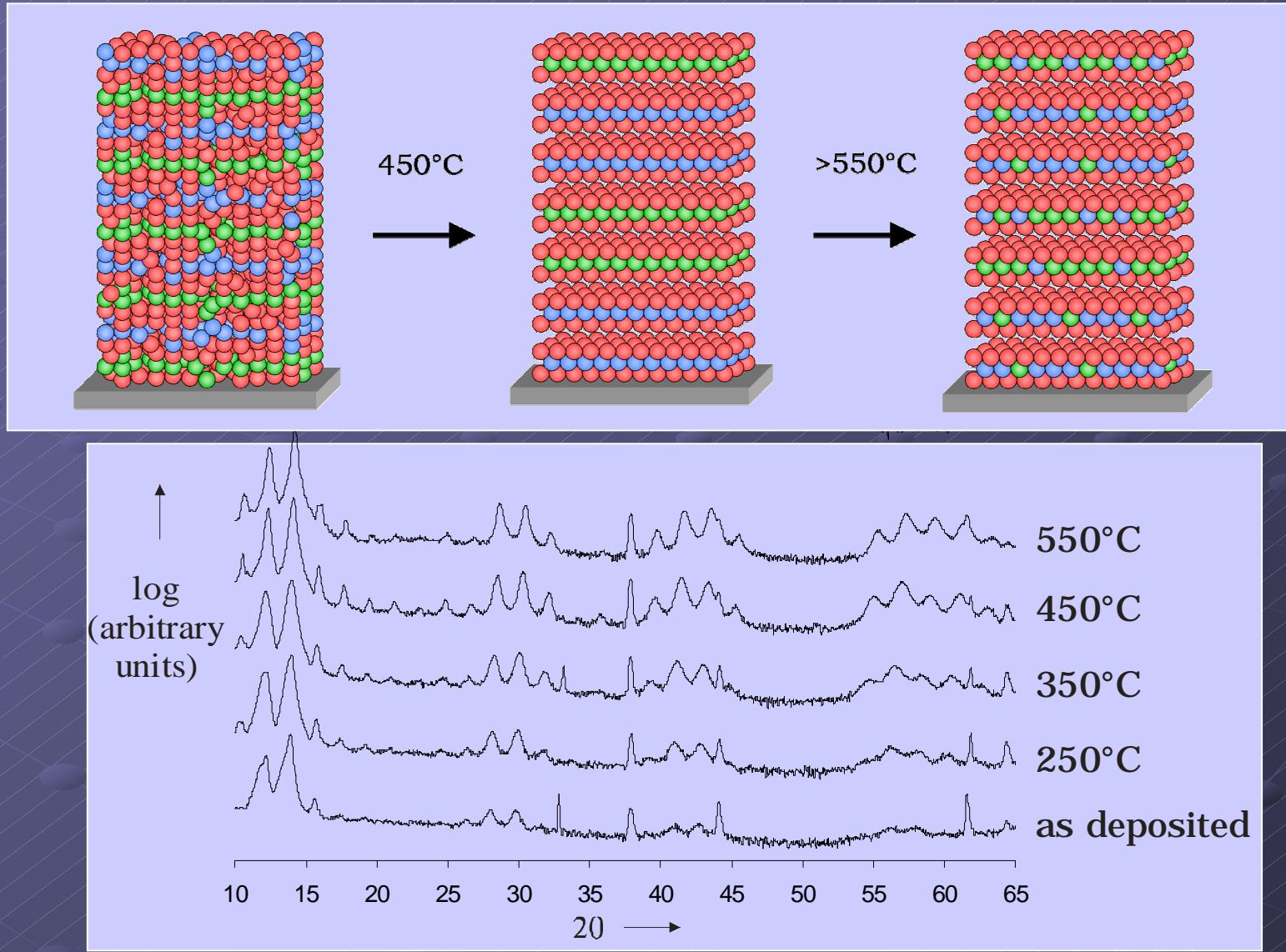
$[(W)_x(WSe_2)_y]$ Superlattices



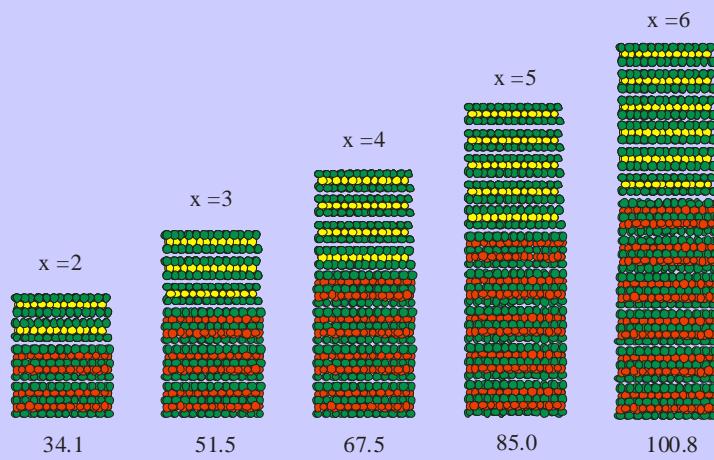
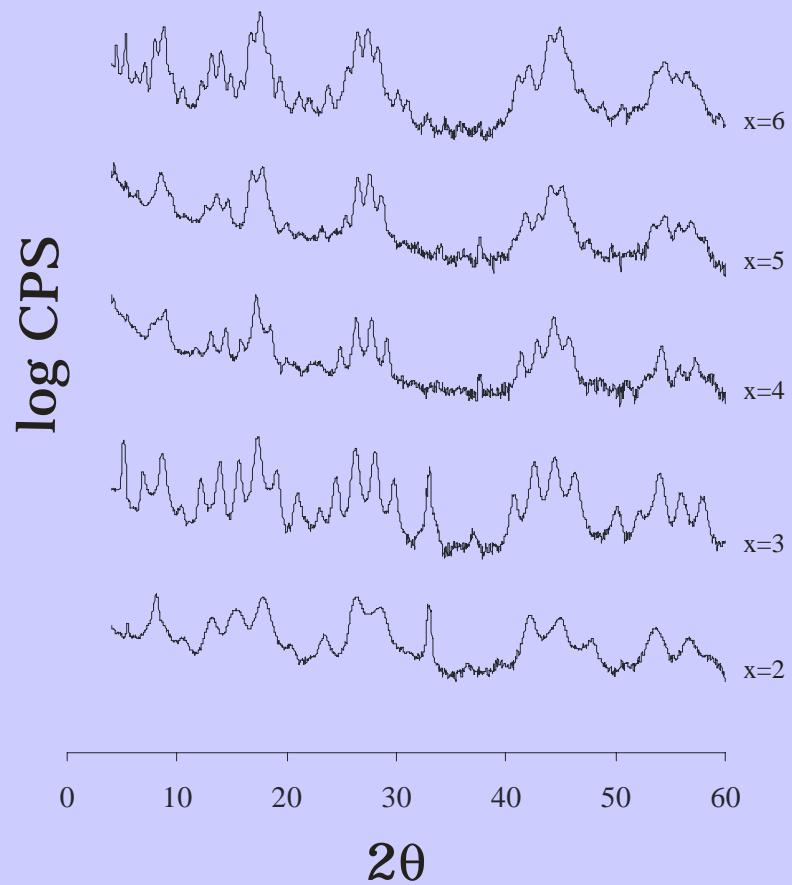
$[(\text{NbSe}_2)_x(\text{CrSe}_2)_y]$



$[(\text{TaSe}_2)_x(\text{VSe}_2)_y]$



Unit Cell Control

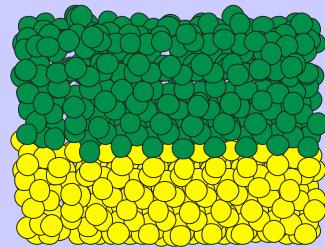


F. R. Harris, S. Standridge, C. Feik, D. C. Johnson,
Angew. Chem. Int. Engl. **2003**, *42*, 5296.

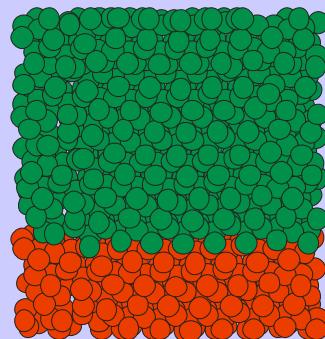
Two Main Calibrations

Stoichiometric

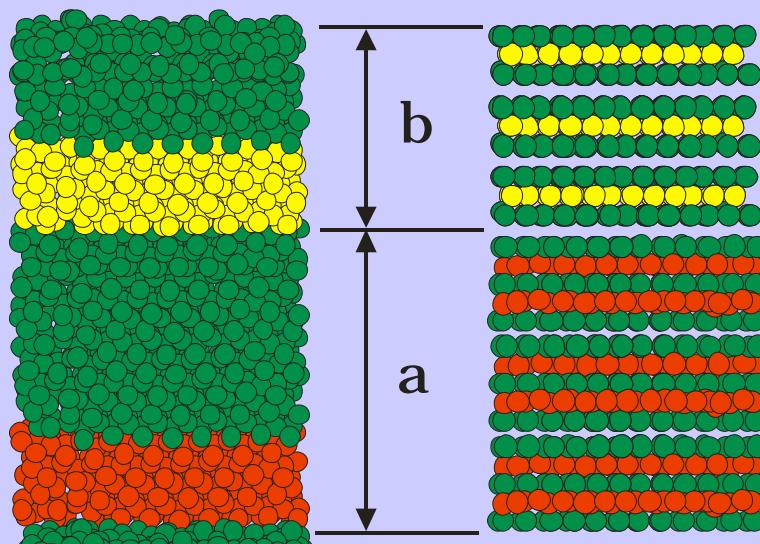
1:2 Ti : Te

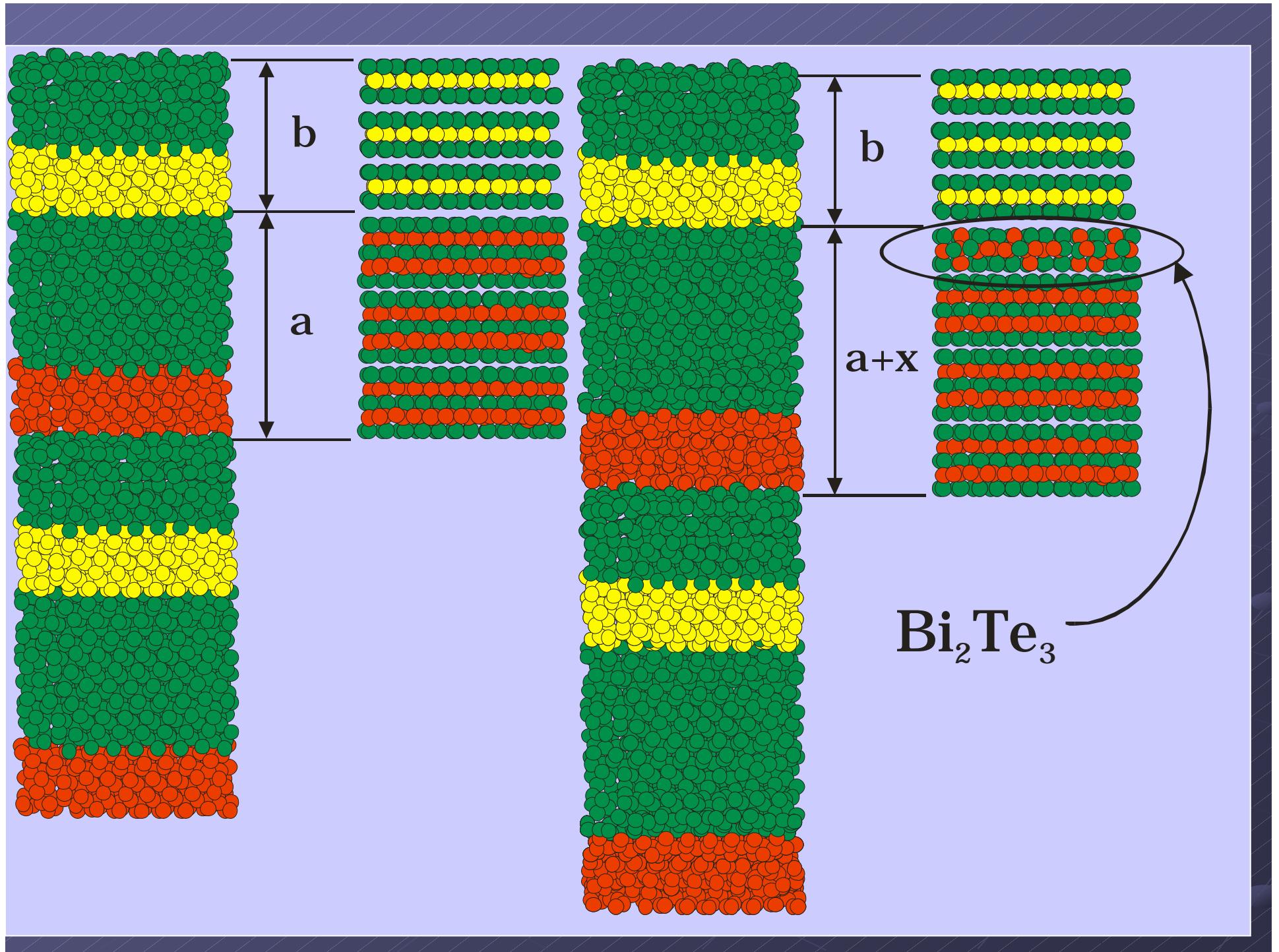


2:3 Bi : Te



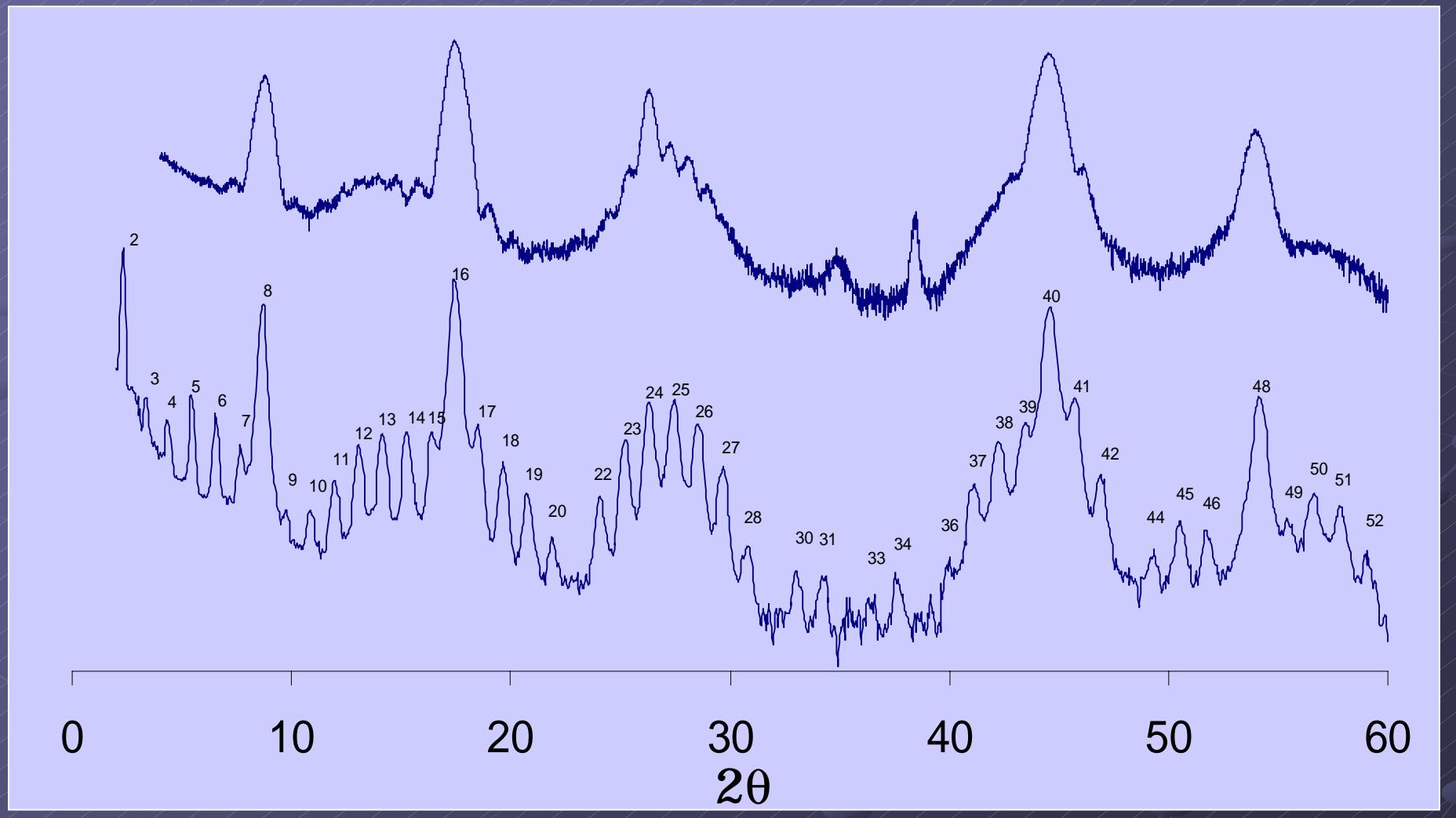
Absolute Amount



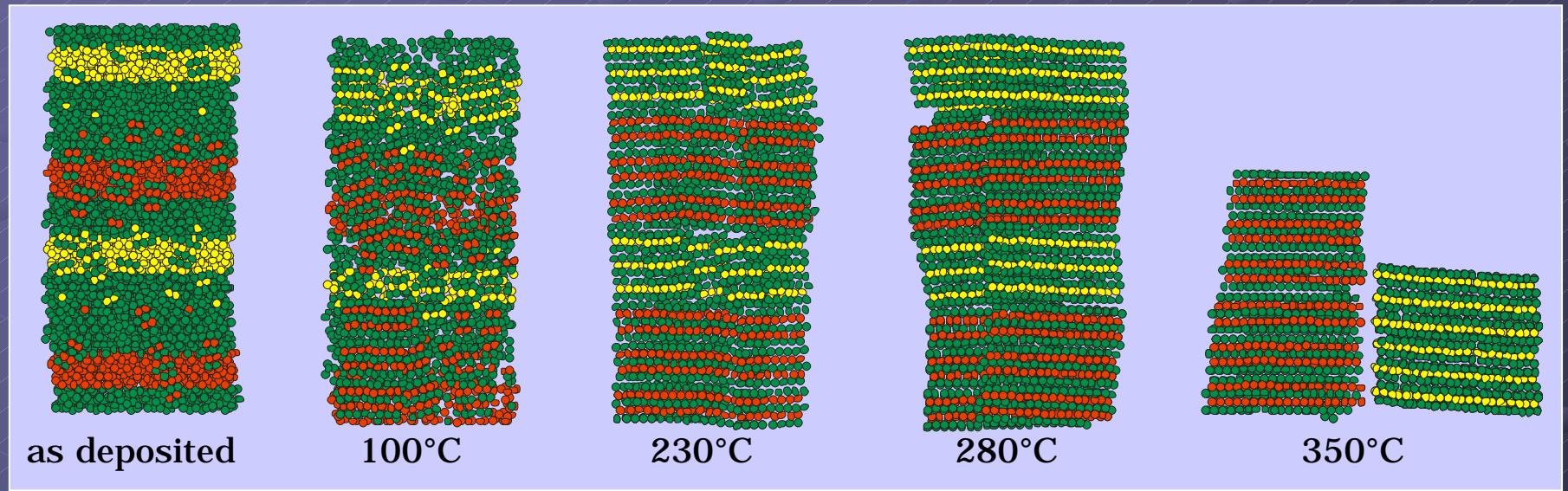


Calibration of Component Layers

Absolute Thickness is Critical

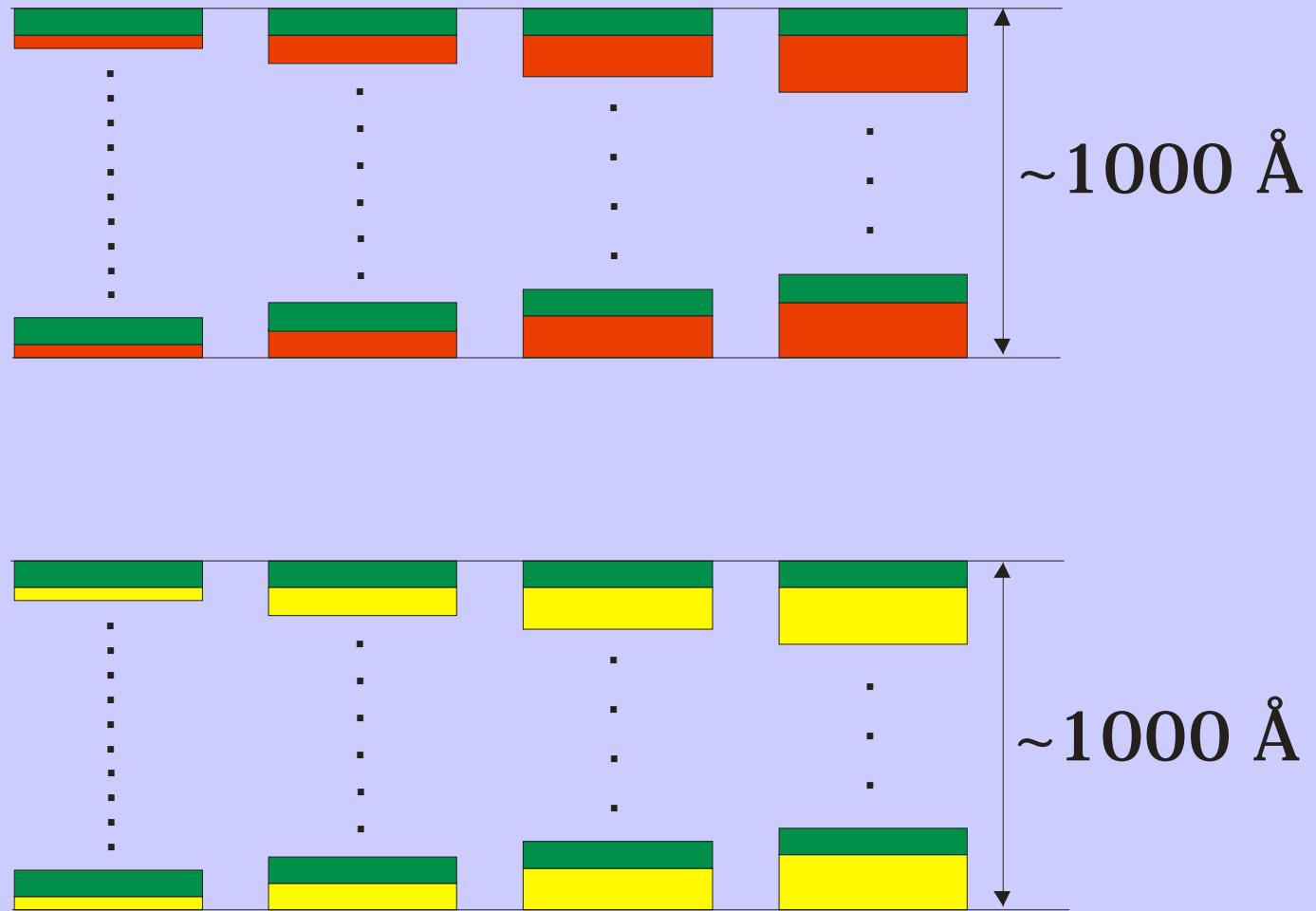


Proposed Mechanism

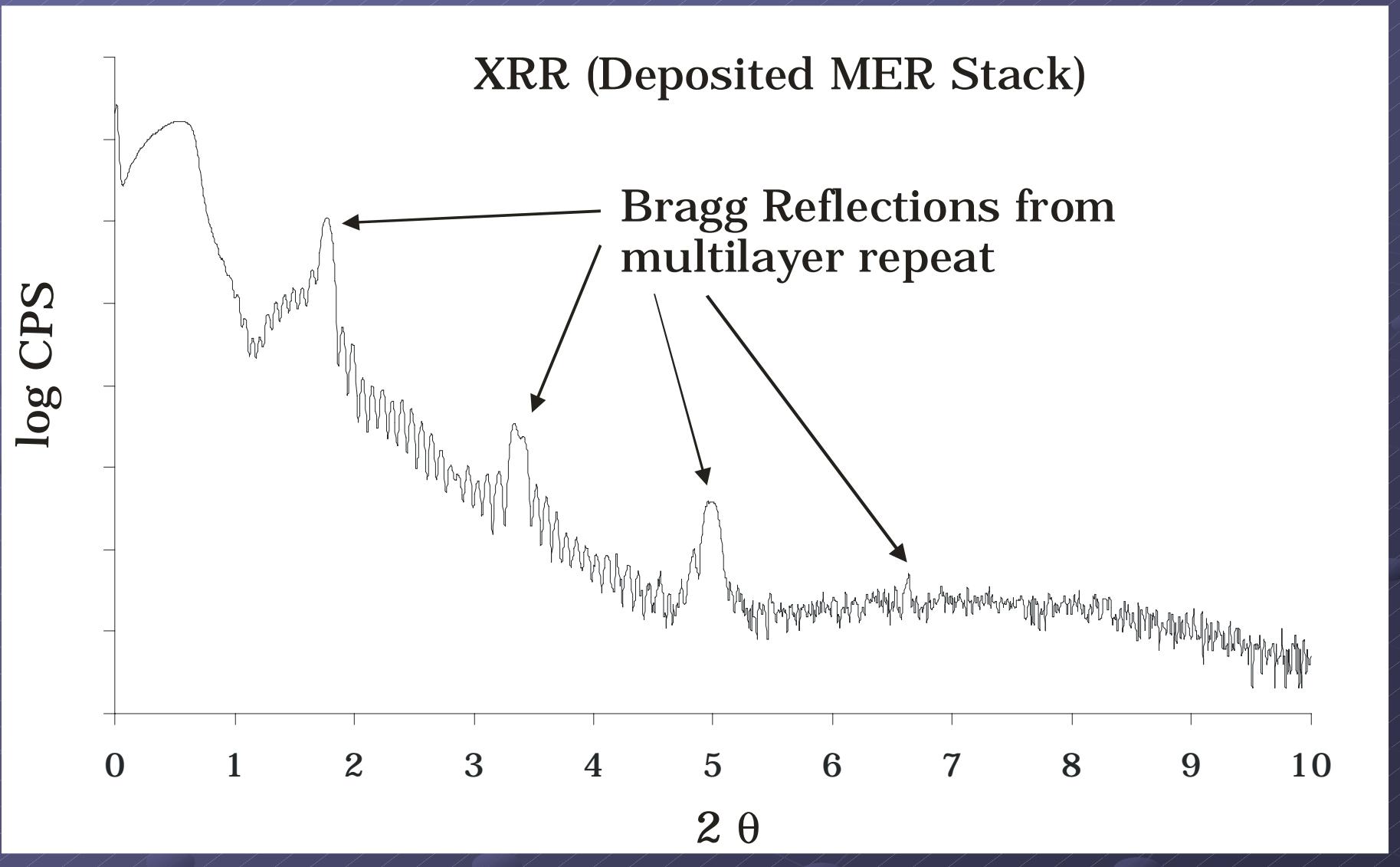


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Angew. Chem. Int. Engl. **2003**, *42*, 5296.

■ Bismuth ■ Titanium ■ Tellurium



Deposited Reactant Stacks



Calibrating Absolute Thickness



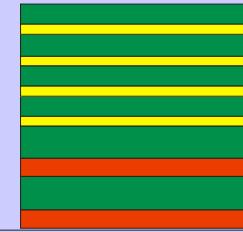
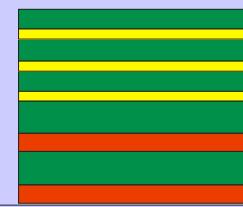
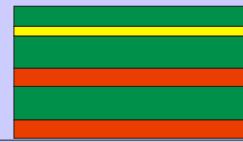
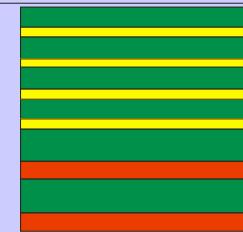
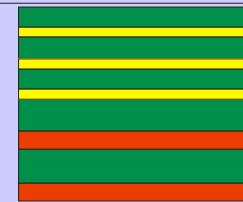
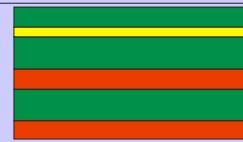
Bismuth



Titanium

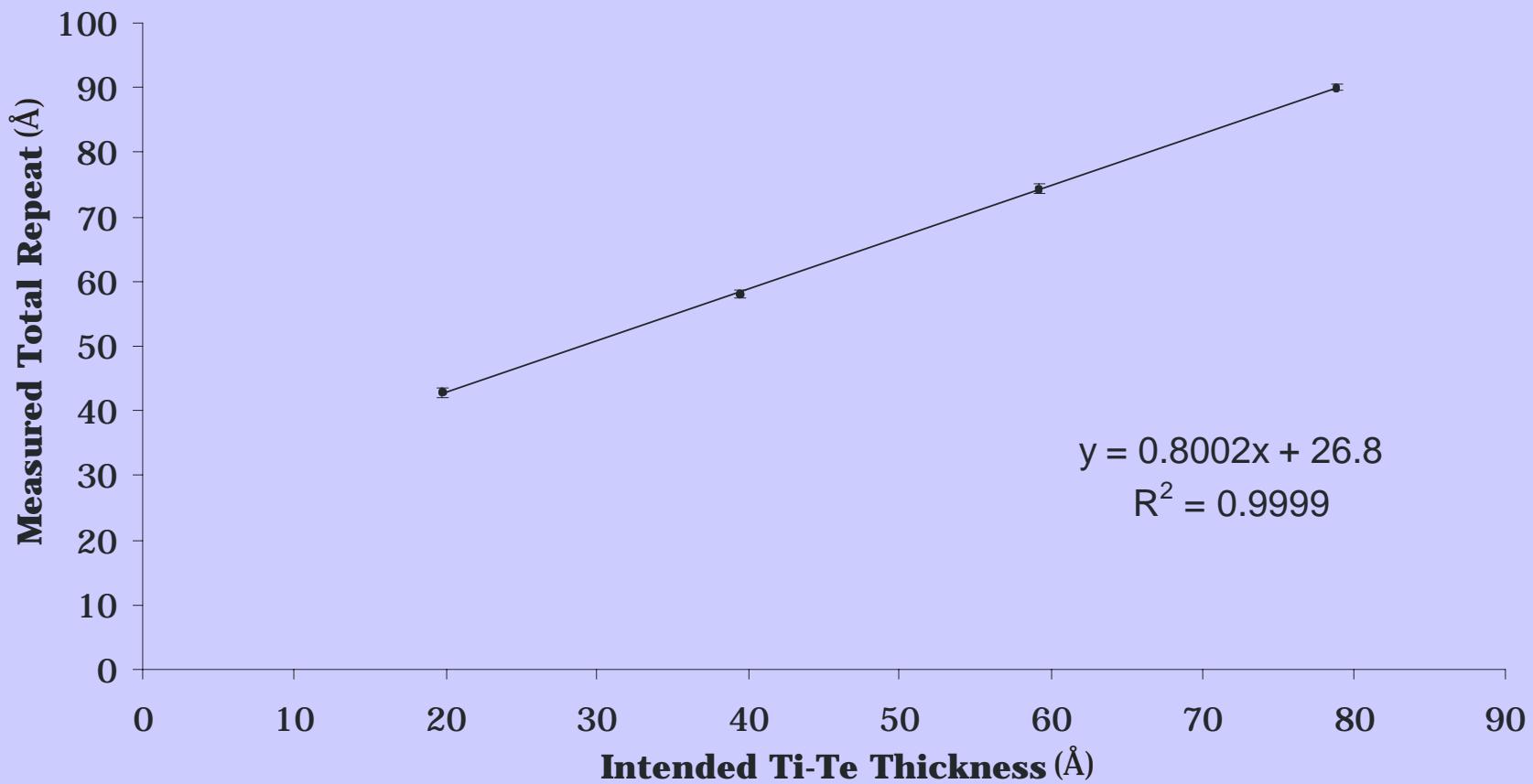


Tellurium



~1000 Å

Calibrating Absolute Thickness

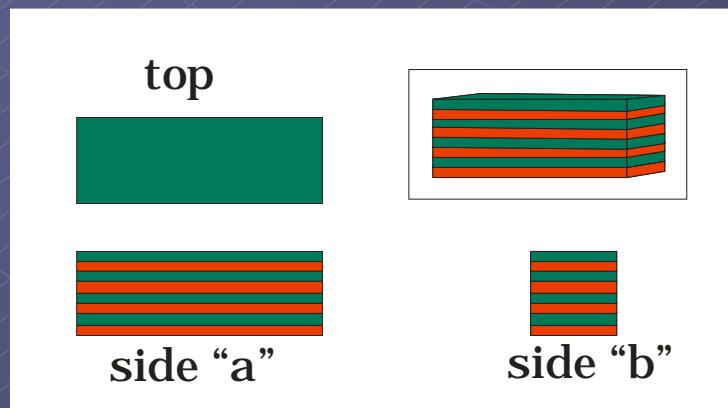
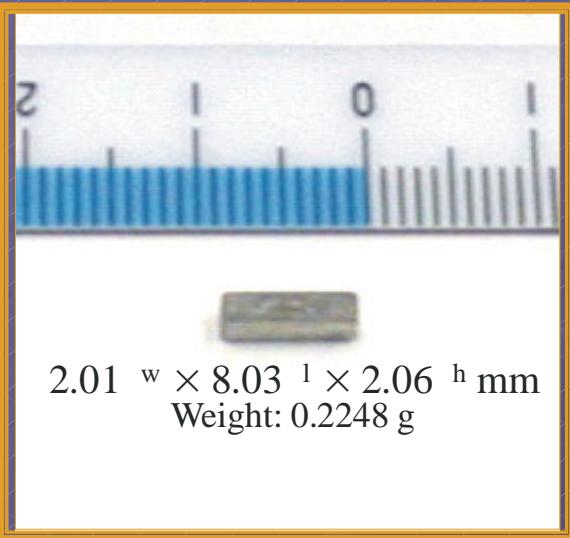


$[(\text{Bi}_2\text{Te}_3)_6(\text{TiTe}_2)_3]$

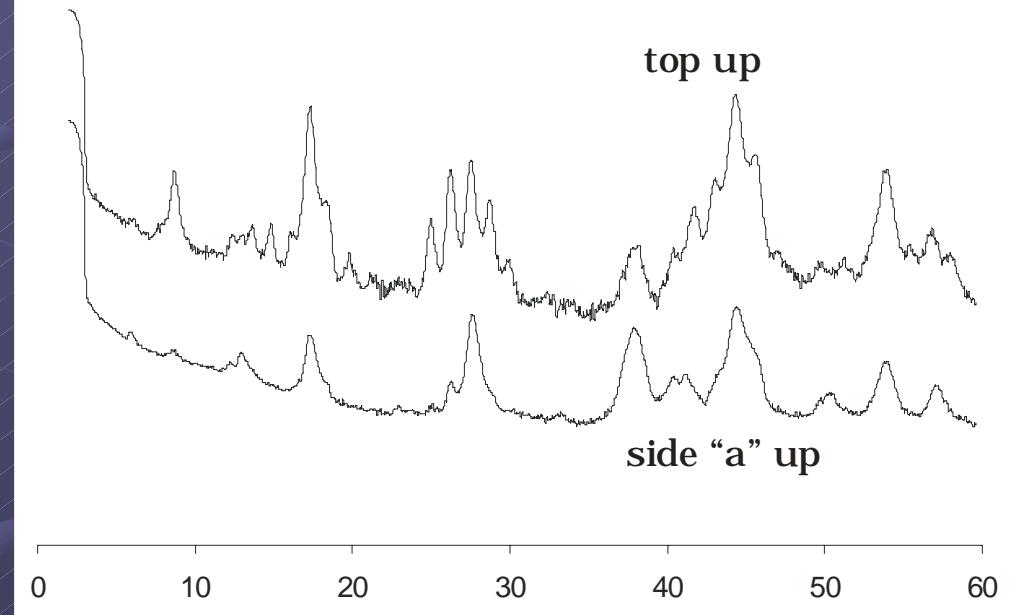
TiTe_2 →
 Bi_2Te_3 →

10 nm
—

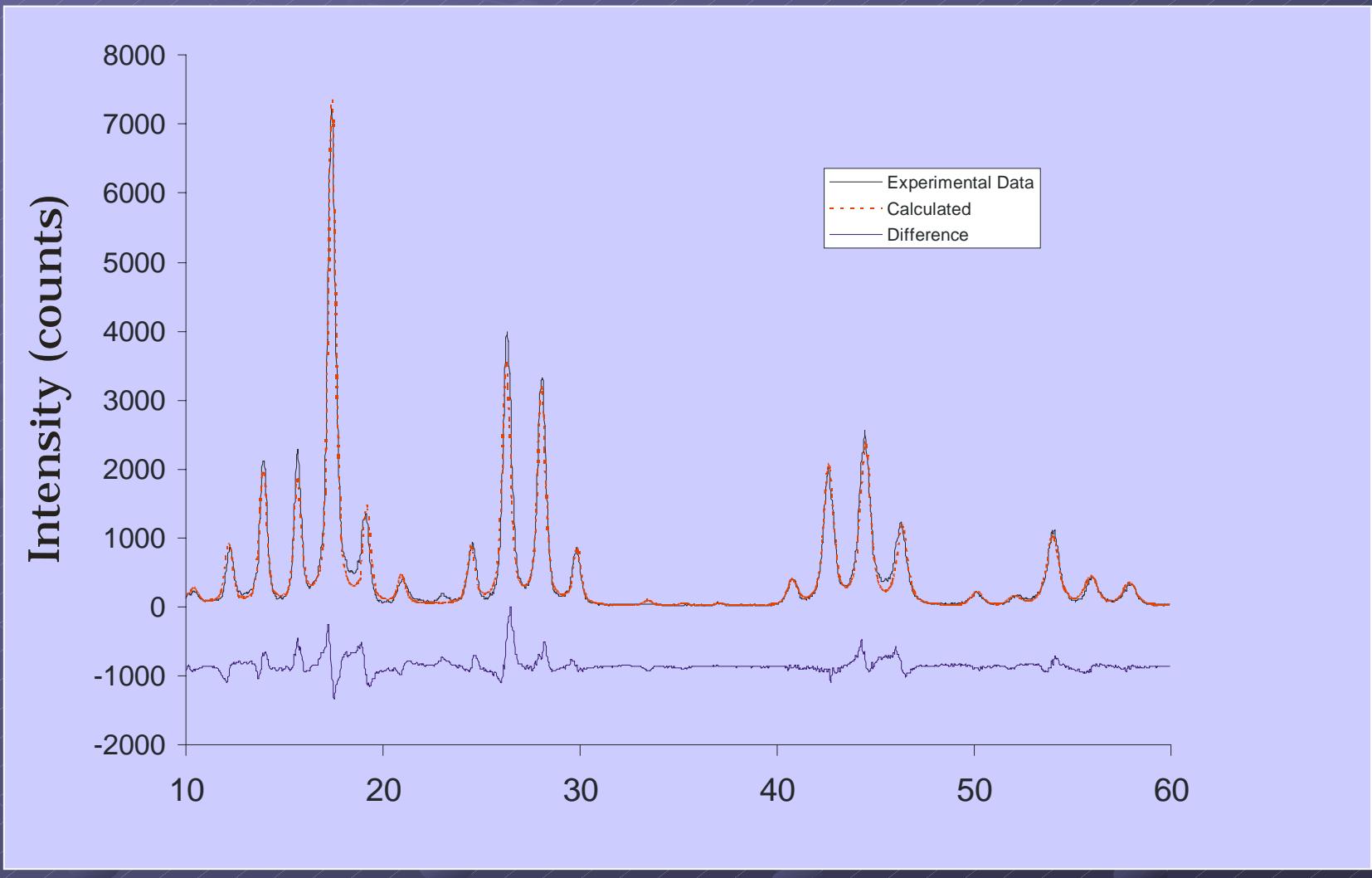
Bulk Samples



MER can make bulk amounts



Structure Analysis via Rietveld Refinement



Summary of Rietveld Refinement Results

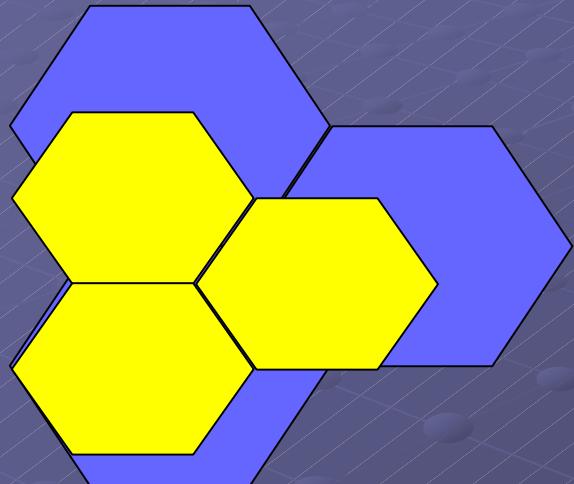
- Bi_2Te_3 layers spacings similar to the bulk
- TiTe_2 layer spacings similar to the bulk
- Van der Waals gap between Bi_2Te_3 and TiTe_2 resembles more closely the van der Waals gap in TiTe_2
- The relative density of the layers are close to bulk values ($\text{TiTe}_2 : \text{Bi}_2\text{Te}_3 = 1.36 : 1.00$)

An Interesting Question...

There is a wealth of data that suggests each component layer looks like its respective bulk material...

What is the crystallographic relationship between layers in the *a-b* direction?

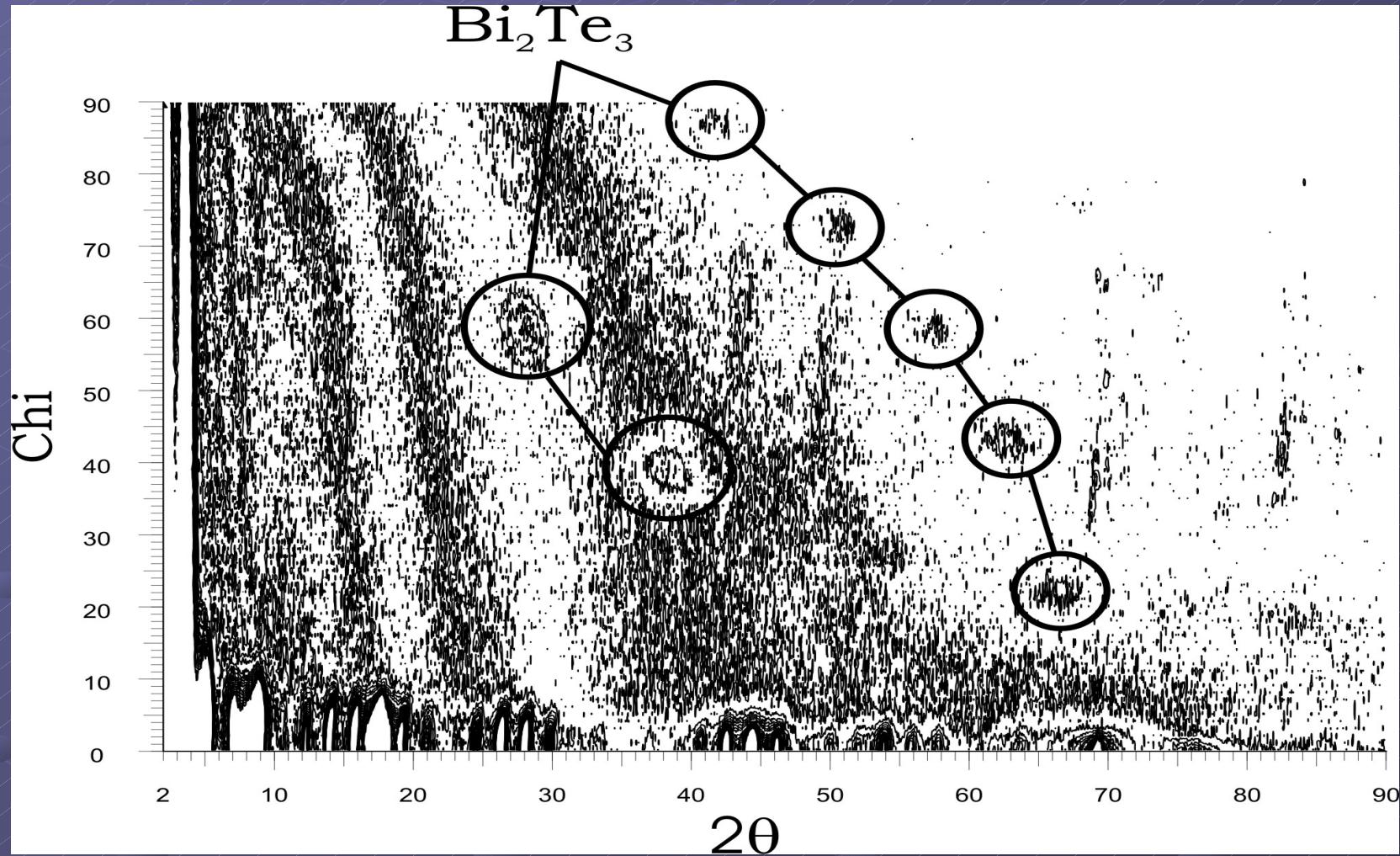
19% Mismatch in the *a* lattice parameter between the end of a TiTe_2 layer and the start of a Bi_2Te_3 layer



a lattice parameters

Bi_2Te_3	= 4.4 Å
TiTe_2	= 3.7 Å

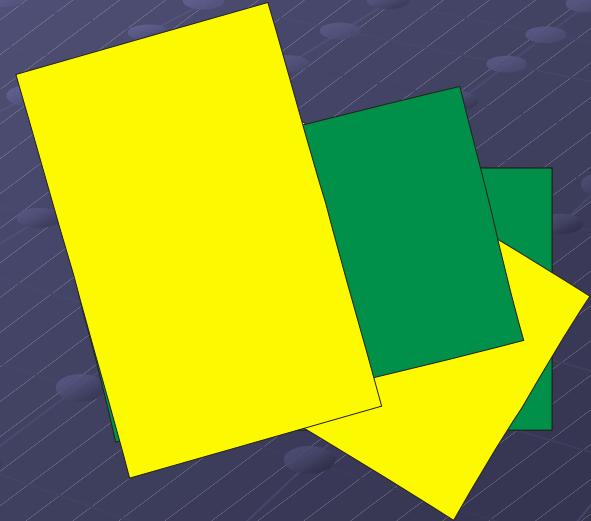
A K-space Map of [(Bi₂Te₃)₃(TiTe₂)₃]

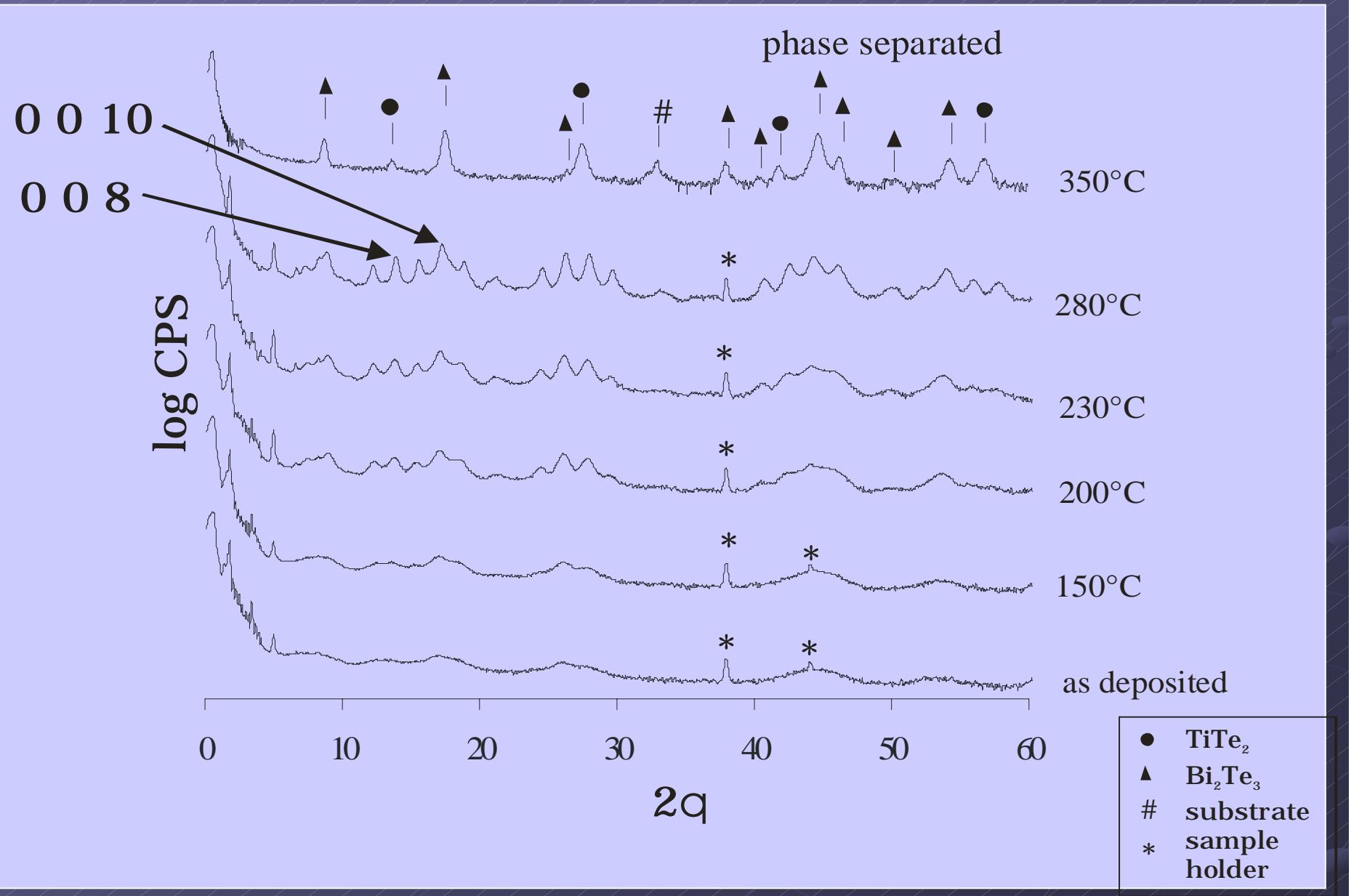


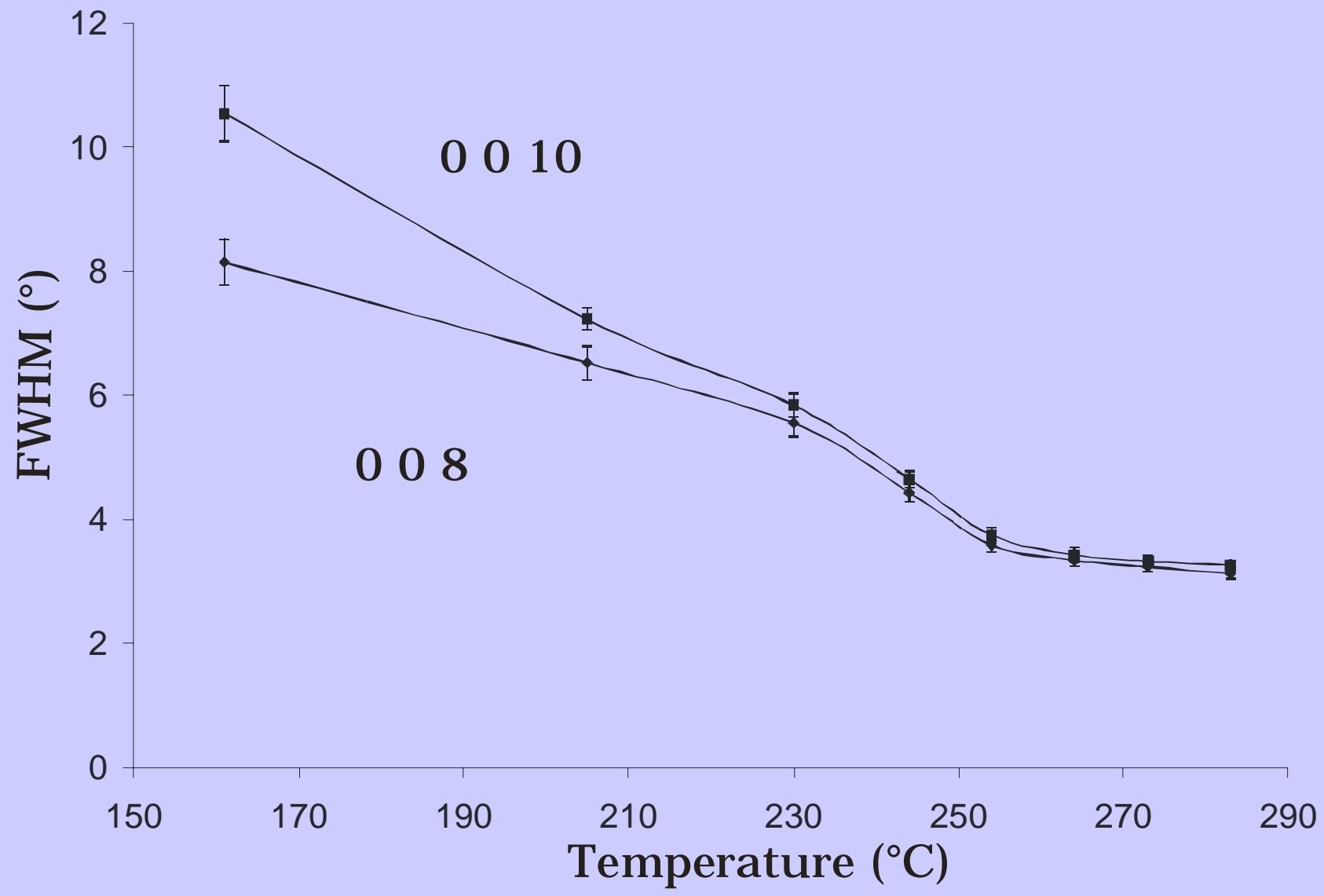
The Short Story

XRD data suggests that the layers are turbostratic

Much like a deck of cards carelessly thrown on a table, there is crystallographic alignment in the c direction, but random alignment between the layers in the a - b plane.



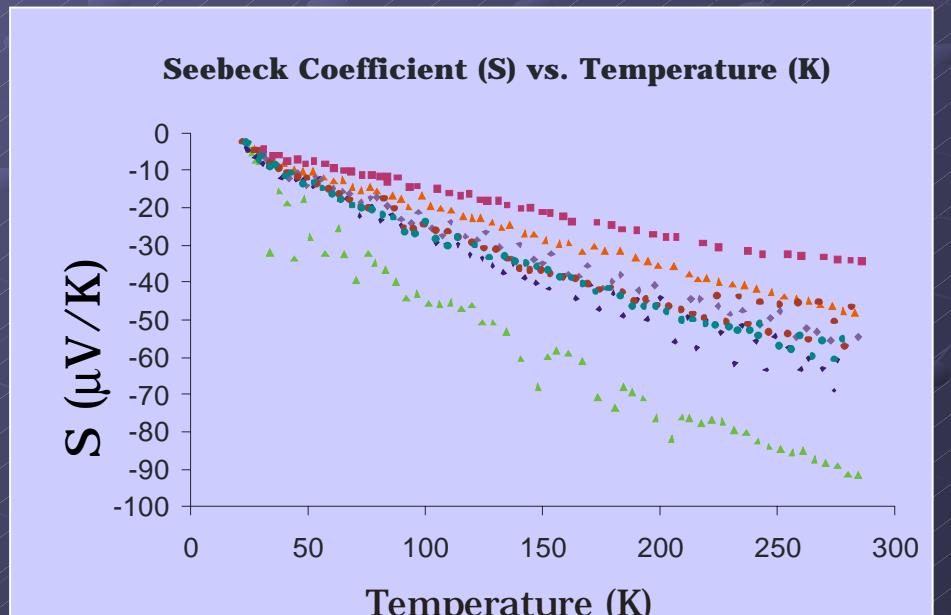
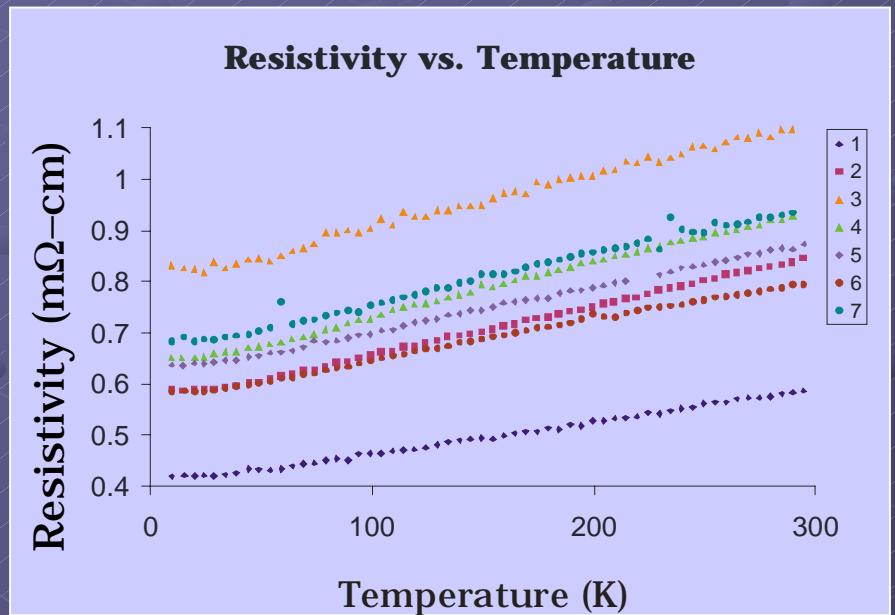




Electrical Properties of [(Bi₂Te₃)_x(TiTe₂)_y] Superlattices

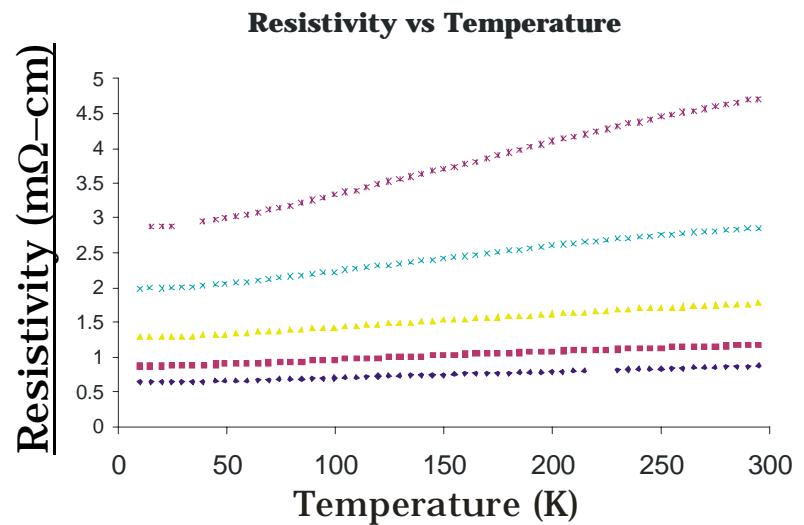
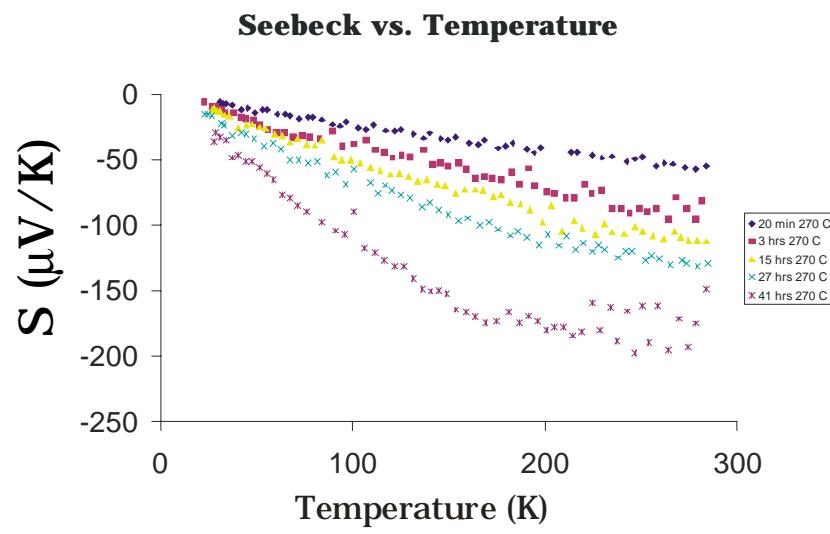
Precision Experiment

7 different samples of [(Bi₂Te₃)₅(TiTe₂)₄] were made

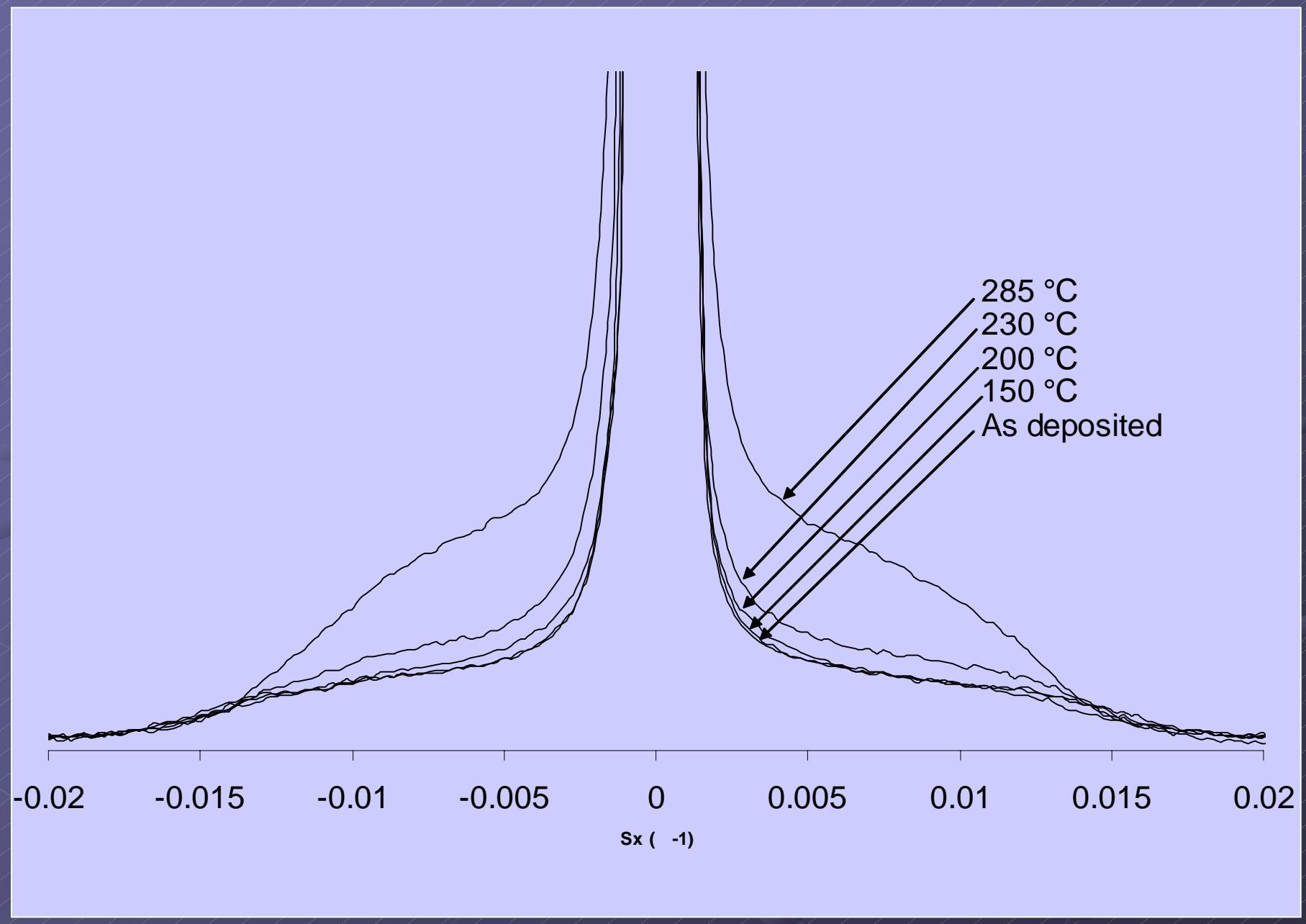


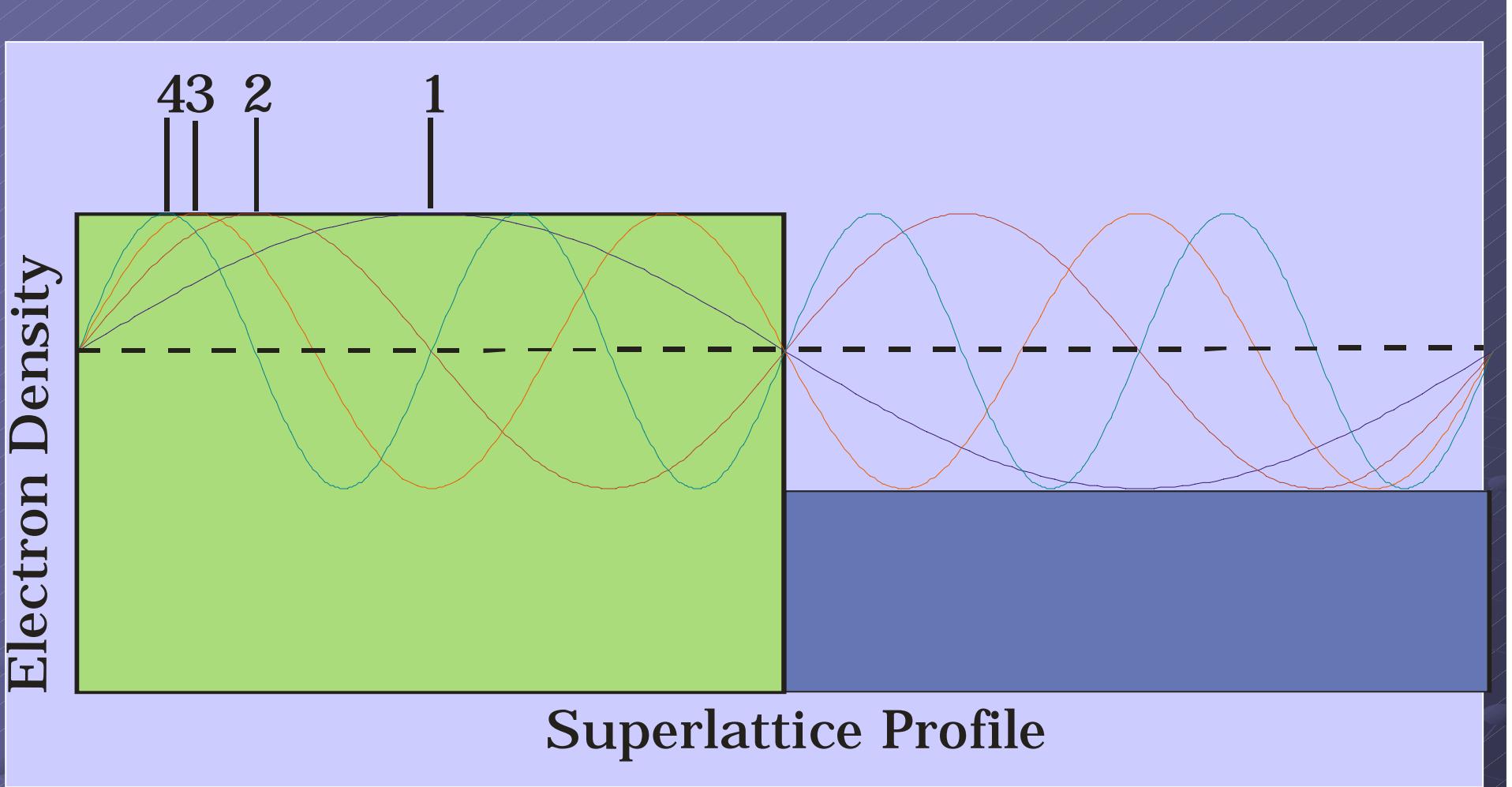
Annealing Study

Annealing makes electrical properties more desirable



- 20 Minutes 270
- 3 Hours 270
- 15 Hours 270
- 27 Hours 270
- 41 Hours 270





X-ray diffraction is nothing more than a Fourier transform of the electron density normal to the scanning axis

Future Work

- Analyze Electrical properties of $[(\text{Bi}_2\text{Te}_3)_x(\text{TiTe}_2)_y]$ as a function of x and y
- Analyze order between a-b planes as a function of x and y
- Analyze thermal conductivity as a function of x and y

Acknowledgements

**Polly Berseth
Evan Foster
Stone Kim
Heike McNeil
Ngoc Nguyen
Arwyn Smalley
John Thompson
Josh Williams**



National Science Foundation

- DMR 9813726
- DMR 0103409
- DGE 0114419



Prof. Terry M. Tritt