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Electromagnetic Effects Due to Spin-Orbit Coupling

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The existence of strong spin-orbit coupling in the single-particle model of the nucleus implies the existence of a term

$$-f(r) (\vec{\sigma} \cdot \vec{\mathcal{L}}) = -f(r) (\vec{\sigma} \cdot [\vec{r} \times \vec{p}]) \quad (1)$$

in the single-particle Hamiltonian. This gives rise to an interaction of charged nucleons with external electromagnetic fields with vector potentials  $A$ . Replacing the momentum in the Hamiltonian by  $\vec{p} - \frac{e}{c} \vec{A}$ , we obtain from the spin-orbit coupling term the interaction energy:

$$\frac{e}{c} f(r) \cdot (\vec{\sigma} \cdot [\vec{r} \times \vec{A}]) \quad (2)$$

One consequence of this effect is that the magnetic moments of odd-proton nuclei should deviate from the Schmidt lines.<sup>1</sup> In a constant magnetic field with  $\vec{A} = \frac{1}{2} [\vec{H} \times \vec{r}]$  the interaction term (2) becomes:

$$\frac{e}{2c} f(r) (\vec{\sigma} \cdot [\vec{r} \times [\vec{H} \times \vec{r}]]) = \frac{e}{2c} f(r) \left\{ r^2 (\vec{\sigma} \cdot \vec{H}) - (\vec{r} \cdot \vec{\sigma})(\vec{r} \cdot \vec{H}) \right\} \quad (3)$$

For a state with orbital angular momentum  $\rho$ , total angular momentum  $j = \rho \pm \frac{1}{2}$ , and magnetic quantum number  $m_j$  the expectation value of the interaction energy is:

$$\pm \frac{e}{2c} \cdot \overline{r^2 f(r)} \cdot \frac{2j+1}{2j+2} \frac{m_j}{j} \cdot H \quad (4)$$

with the positive sign for  $j = \rho + \frac{1}{2}$ , the negative sign for  $j = \rho - \frac{1}{2}$ .

The factor in front of  $H$ , for  $m_j = j$ , is the negative additional magnetic moment. The effect consequently places the magnetic moments inbetween the Schmidt lines, as is experimentally observed. Taking the spin-orbit splitting energy,  $\overline{f(r)}(2\rho + 1) \cdot \hbar$ , to be about 2 Mev at  $\rho = 4$ ,  $A = 100$ ,

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1. The fact that velocity dependent two-body forces give rise to additional moments has been pointed out by Blanchard, Avery, and Sachs, Phys. Rev. 78, 292 (1950).

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